

Supporting Information

3D-Printed Hierarchical Porous Frameworks for Sodium Storage

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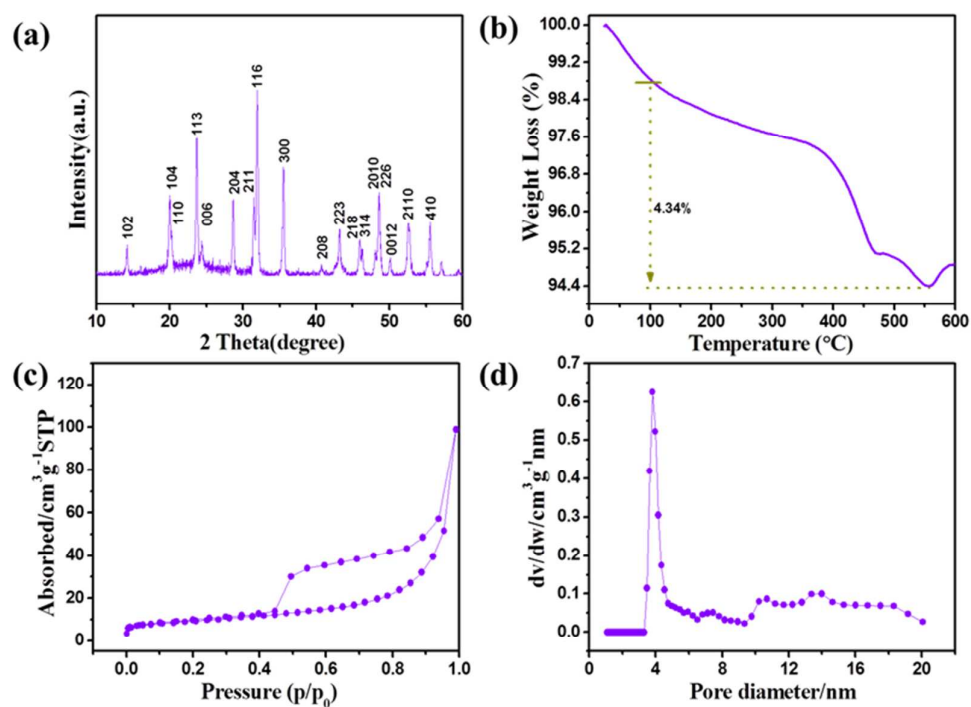


Figure S1. (a) XRD pattern, (b) TGA curve, (c) Nitrogen adsorption-desorption isotherm, and (d) pore size distribution curve of NVP.

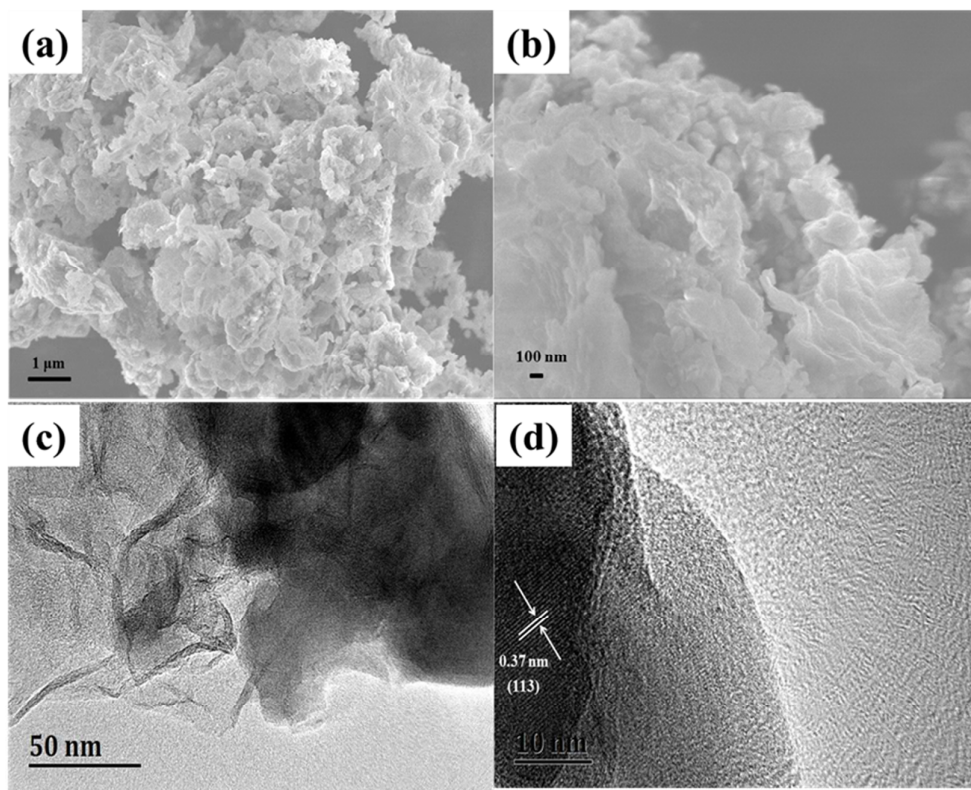


Figure S2. (a, b) SEM images with different magnifications, (c) TEM image, and (d) HRTEM image of NVP, showing a significant portion of NVP are encapsulated within rGO nanosheets.

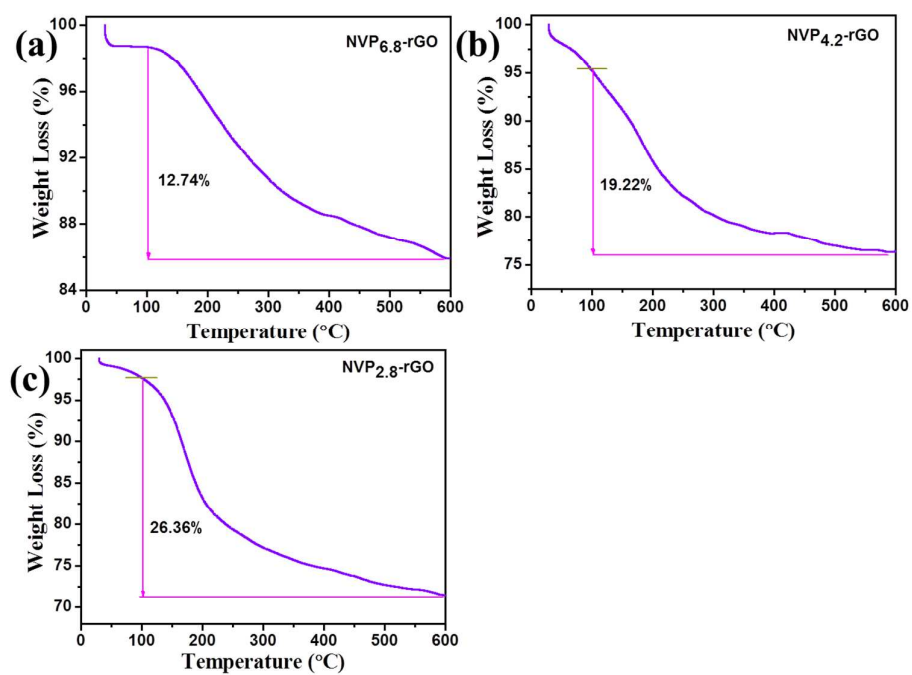


Figure S3. (a-c) TGA curves of different hierarchical porous frameworks, showing different rGO contents.

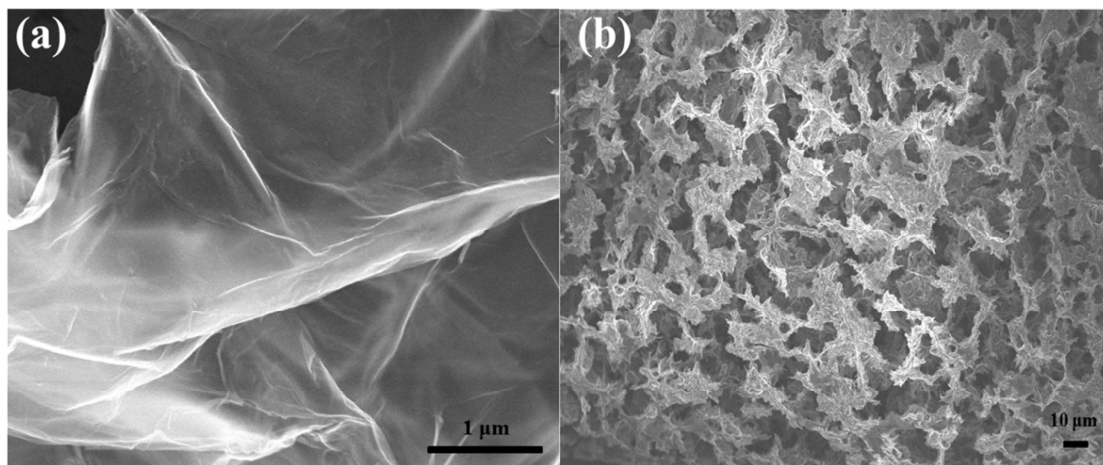


Figure S4. SEM images of GO sheet (a) and GO-NVP sheets of 3D-printed hierarchical porous frameworks without thermal annealing (b). Showing the adopted GO sheets with the average size of $\sim 20 \mu\text{m}$.

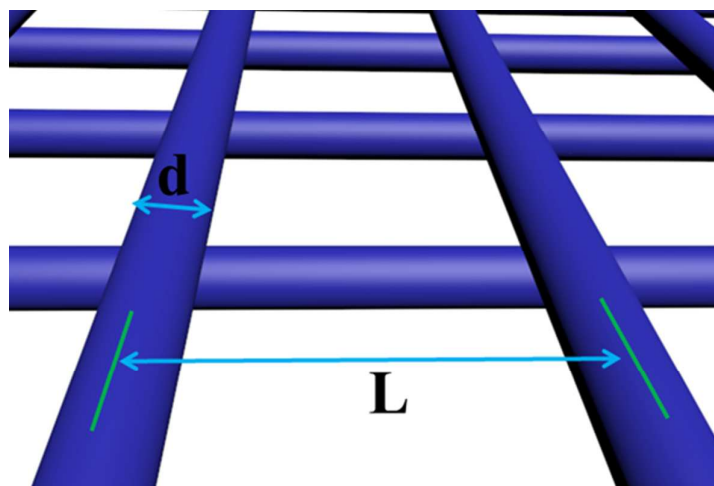


Figure S5. Schematic illustration of the 3D-printed frameworks with a spacing-to-diameter ratio of ~ 4 . The in-plane center-to-center rod spacing is defined as L , and the filament diameter is defined as d .

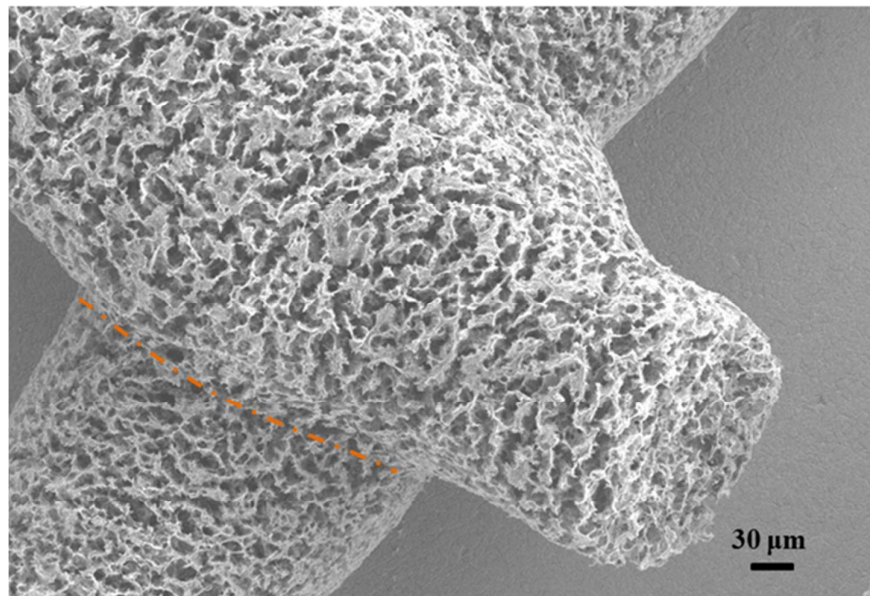


Figure S6. Cross-sectional SEM image of 3D printed frameworks, showing continuous filaments and hierarchical porous structure.

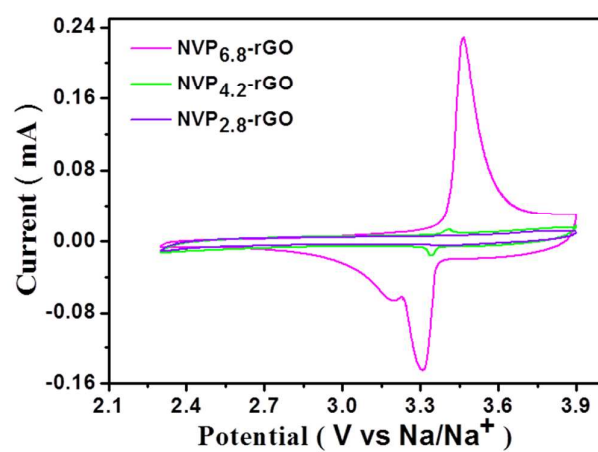


Figure S7. The CV profiles of 3D printed NVP-rGO frameworks.

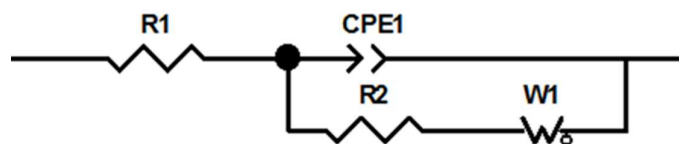


Figure S8. Randles equivalent circuit for 3D printed NVP-rGO frameworks electrode/electrolyte interface. $R1$ is the bulk resistance, $R2$ is the charge transfer resistance, $CPE1$ is the constant phase element, and $W1$ is the Warburg impedance related to the diffusion of sodium ions into the bulk electrodes.

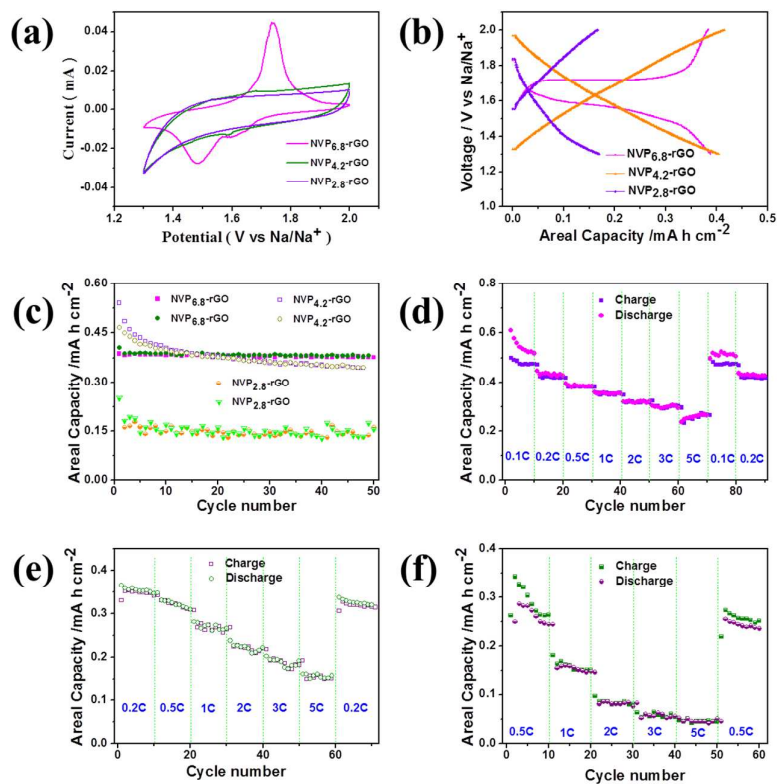


Figure S9. The electrochemical performances of 3D printed NVP-rGO frameworks.

(a) CV profiles. (b) Selected 10th charge and discharge profiles at 1C. (c) Cycle performance at 1C. Cycling and discharge rate performance of (d) NVP_{6.8}-rGO, (e) NVP_{4.2}-rGO, and (f) NVP_{2.8}-rGO.

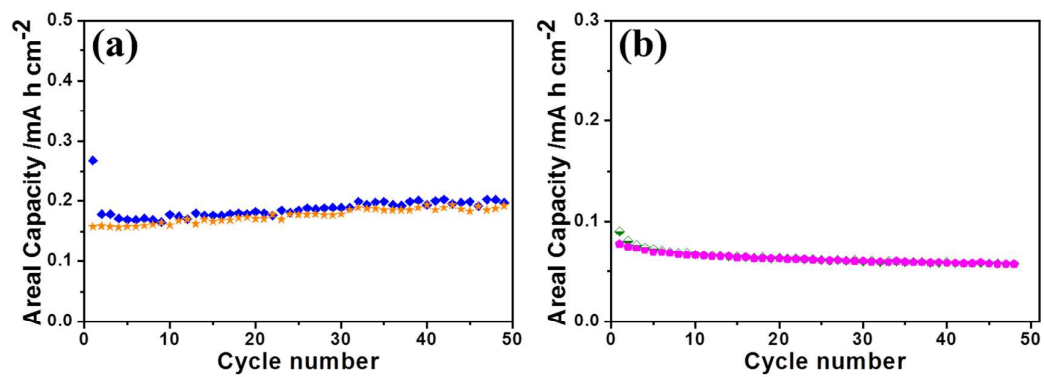


Figure S10. The electrochemical performances of 3D printed NVP_{6.8}-rGO frameworks without freeze-drying process. (a) Cycle performance at 2C as the cathode material. (b) Cycle performance at 1C as the anode material.

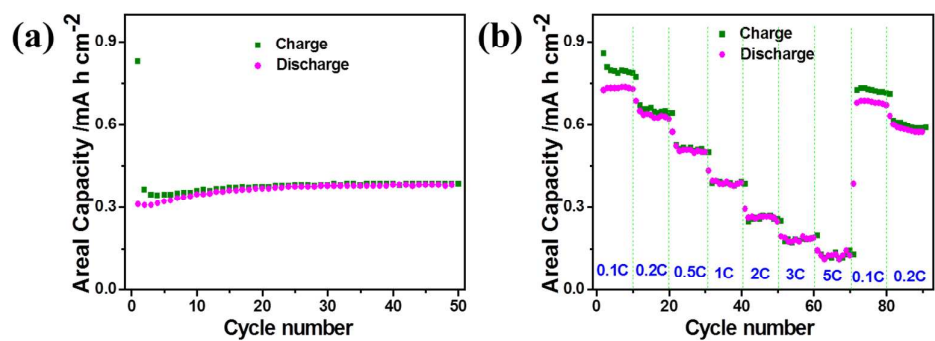


Figure S11. The electrochemical performances of 3D printed NVP_{6.8}-rGO frameworks symmetric full-cell. (a) Cycle performance at 1C. (b) Rate profiles.

Table S1. Comparison of the electrochemical performances of 3D printed electrodes with reported sodium ion batteries.

Electrode materials	Specific capacity (mAh/cm ² @mA/cm ²)	Rate performance (mAh/cm ² @mA/cm ²)	References
SnS ₂ @rGO	1.49@0.23	0.78@30	1
SnS-G	1.04@0.03	0.31@7.3	2
Sb@TiO _{2-x} Nanotubes	1@0.2	0.73@20	3
Ge nanowire	0.05@0.02	0.01@0.44	4
TiO ₂ @G nanocomposites	0.66@0.13	0.23@30	5
Expanded graphite	0.14@0.01	0.09@0.05	6
NVP _{6.8} -rGO frameworks	1.26@0.42 cathode 0.43@0.42 anode	0.65@42 0.32@4.2	This work

Table S2. Kinetic parameters of 3D printed NVP-rGO frameworks.

Samples	R_1 (Ω)	R_2 (Ω)	σ_w (S ⁻¹ s ^{-1/2})	D_{Na^+} (cm ² s ⁻¹)
NVP _{6.8} -rGO	5.8	157	13.59	6.12×10^{-13}
NVP _{4.2} -rGO	18.3	417	69.84	2.32×10^{-14}
NVP _{2.8} -rGO	12.3	802	98.27	1.17×10^{-14}

References

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- (2) Zhou, T.; Pang, W. K.; Zhang, C.; Yang, J.; Chen, Z.; Liu, H. K.; Guo, Z. Enhanced Sodium-Ion Battery Performance by Structural Phase Transition from Two-Dimensional Hexagonal-SnS₂ to Orthorhombic-SnS. *ACS Nano* **2014**, *8*, 8323-8333.
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