SUPPORTING INFORMATION

Toward A Soluble Model System for the Amyloid State

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β-Arch Consensus Sequences

Kajava and coworkers classified β -arches by the backbone dihedral angles of their loop regions (' β -arcs'), and identified several classes that produce 180° turns: three-residue (*ppl*, *xbl*), four-residue (*bepl*, *bebl*, *gbpl*, *bgpp*), five-residue (*blbbl*), and six-residue (*baepep*, *bllpbl*) β -arcs, where a = (right-handed) α -helix, $b = \beta$ -strand, d = bridge (δ), e = extended chain (ϵ), $g = \gamma$ ', l =left-handed α -helix, p = polyproline-II, and x = "around $\Phi = -80^{\circ}$ and $\Psi = -140^{\circ}$ " regions of the Ramachandran plot.¹ We mined the RCSB Protein Data Bank (PDB) for structures containing these backbone dihedral angles, using DSSP to assign dihedral angles.²⁻⁴ The PDB was culled by sequence identity (30%), resolution (2.0 Å), and R-factor (0.25), on December 30, 2012.⁵ After visual inspection of the culled set, we obtained the following sets of β -arches: 26 *ppl*, 0 *xbl*, 0 *bepl*, 8 *bebl*, 5 *gbpl*, 2 *bgpp*, 12 *blbbl*, 7 *bllpbl*, and 1 *baepep* β -arches. Of these, 3 *ppl*, 3 *bebl*, 5 *gbpl*, 1 *bgpp*, 12 *blbbl*, and 7 *bllpbl* β -arches produced 180° turns (**Figure S1**). Sequence logos in **Figure S1** were constructed with WebLogo.⁶



Figure S1. Sequence logos (residue frequency vs. position) for *ppl*, *gbpl*, *bebl*, *bgpp*, *blbbl*, and *bllpbl* β -arcs that produce 180° turns.

Using a set of 16 *blbbl* β -arcs – the 12 *blbbl* β -arcs in our culled dataset, plus 4 *blbbl* β -arcs that were retrieved from our culled PDB upon slight broadening of our backbone angle constraints for the β -strands that flank *blbbl* (i.e., slight relaxation of the definition of *b*, for positions "1B1"-"1B3" and "2B1"-"2B3") – a new sequence logo was generated (**Table S1 and Figure S2**). These data were normalized to account for the natural abundance of amino acids.⁷

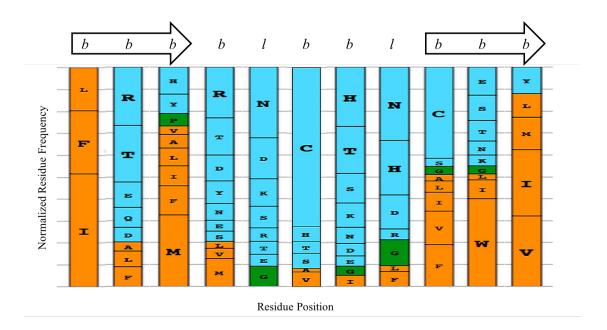


Figure S2. Sequence logo for *blbbl* β -arcs, where $b = \beta$ -strand, and l = left-handed α -helix backbone dihedral angles. Residues of the β -arc are typically hydrophilic (blue), while residues of flanking β -strands are typically hydrophobic (orange).

Table S1. The redundant set (culled set in red font) of *blbbl*-containing sequences in December 30, 2012 PDB. The sequences that had a *blbbl* backbone conformation, but that were not 180° β -arcs on visual inspection, are in gray boxes.

PDB code		blbbl		PDB code		bbb		bbb	PDB code			blbbl		PDB code	A1 res#	blbbl	
1a9x	4171			1qq1			RECIG		2qy1			KNKSD		3ir5		WGEKG	
1ag8	453 TVW			1 1			RGCHF		2-1-1			ENIKN		3ir6		WGEKG	
1bhe "	106 ITA 157 IQI			1qrb "			RECIG RGCHF		2rhp 2uvf			FSQEM LENHN		3ir7 3it5		WGEKG YHMDQ	
	207			1qrc			RECIG		2011			GNSON		3iuj		KDVKF	
1clw	265 LEI						RGCHF		"			SDLRD		3iur		KDVKF	
	285 FLF			1qxo			SGVRH		"	594	WHF	SEVKN	VKV	3iwk	447	 INCAQ	
1cs0	171			1r53			SGTEF		2v5i			RSASR		3jqy		VNSKE	
1czf	185 FDV			1rmg			TDTTD					RSCHH		3jur		LDCEN	
1edq 1ehn	452 452						TDVTH DTCSD		2vbk 2vfm			LRSGG RECIG				YRCRN VLSEN	
1eib	452			1tyv			RECIG		201111			RGCHF		"		ESCKY	
lezg		TNSOH		"			RGCHF		2vfn			RECIG		3k5i		TMAYD	
		TGSTD		1u79			AGCKG		"			RGCHF		3km5		YNCSD	
		TNSKD		1um0			SGVFE		2vfo			RECIG		31j5		LSQYT	
		TDSTN		1uun			TGAAG					RGCHF		31jq		ACMEH	
1ffr	452	TNSSG		1vbl "			KNVDN EGSSH		2vfp			RECIG RGCHF		3m1h		ACMEH FGCTD	
lgo7		TNVFG		lvz2			ERVKF		2vfq			RECIG		3n80		VNCYD	
1go8		TNVFG		1vz3			ERVKF		"			RGCHF		3n81		VNCYD	
1gv4	578			1 whu			EYKRC		2vle			VNCYD		3ogl		DKCSG	
1h54	490			1x6l			NGYQN		2wk2			NGYQN		3qcw		ENKAG	
1hg8	157			1y4z			WGEKG		2wme			INTWG		3rhj		INTYN	
1ia5	196 FDI 110			1y5i			WGEKG		2wox			INTWG VDSDR		3rhl 3rho		INTYN INTYN	
11a5	133 FSV			1yo8 1zum			FSQEX VNCYD		2x65			VDSDR		3rhq		INTYN	
	164 FDI			2a8j			GALSG		2xc1			RECIG		3riq		NSANN	
1 idk	159 ITL			2a81			GALSG		"			RGCHF				TLCNH	
1 iru	156			2a8m			GALSG		2yuh	14		SNLES		3s2k		GHLDY	
1ja9	170			2b0r			ENHVN					TELSN		3s94		GGLED	
1k4z	1404,2404 IFI						ENCVG					TKAHS		2		EDIRH	
	1423,2423 ISL 1442,2442 MDV						VNCDD DKSNK					EDCSD HSTKD		3suc		RACHN HHSQY	
	1482,2482 IYT						SKSSE					EDCSG		"		MGSTS	
1k7g		TNVFG					RCSQN		2zah			GTVVA		"		LTSSS	
1k7i		TNVFG		2b39			VSCKK		3bh6			SGLKD		"		LGSLG	
1k7q		TNVFG		2bx6			QDCEN					QDCEN		"		ISASQ	
1k8f	354						DDCTN					DDCTN		2450		LSSYG	
	373 ITV 392 VEI						RNCRD RDCRK					RNCRD RDCRK		3th0		RECIG RGCHF	
	432 IVS						ESSSN					ESSSN		3ven		RDFHD	
1k9t	452			2c1d			DGFVR		3bh7			QDCEN		3ver		RDFHD	
1kee	171			2cu2			LDTFG		"			DDCTN		3vex		RDFND	
1kq5	1404,2404			"			LGVSG					RNCRD		3vez		RDFHD	
	1423,2423 ISL			2ewe 2f8o			EKAIN FHASD					RDCRK		3vf2 3vf4		RDFHD RDFHD	
	1442,2442 MDV 1482,2482 IYT			2180 2g85			SGIRH		3ddu			ESSSN ERVKF		3vn4 3vmv		SNAHN	
1m6i	579			2gaw			ACMEH		3egw			WGEKG		3zqa		INTWG	
1mhs	818			2g19			ACMEH		3eqn			KNAAN		3zsc		KDAQN	
1nh6	452			2gp1			SNIMW		"			QNSHG		"		ENSHH	
1nhc	138			2inu			ENHEG		3fby			FSQEN		4a0p		KDLDS	
1004	191 FDI 453 TVW			2:-7			TGCNR HSLKS		3gd3			WNVFN		4a2a 4a2b		RGLDE RGLDE	
1004	258			2iq7 "			NSATT		3gd4 3gq8			WNVFN RACHN		4a20 4ap6		IHQVR	
1000 109j	453						GSSTG		5540			HHSQY		4dg6		QELDQ	
1p4k	80,380			202p	860		INTYN		"			LTSSS		"		SEIYW	
1pxz	111 LFM	RKVSH	VIL	2onp	453	$\mathbf{T}\mathbf{V}\mathbf{W}$	VNCYD	VFG	"	602	SLI	AGSST	SEA	4ecd	61	 TGVRH	
"	156 ITM			2pyg			DYLVD					LGSLG		4fdc		WNIFN	
1q11		SGVRF					KMTSD					ISASQ		4fr8 4gdy		VNCYD	
lqa1 "	265 LEI 285 FLF			2ahf			YGAQD SGIRH		3hb2			LSSYG TNVFG		4gdv 4h7n		SAVQC INDAA	
1qa2	265 LEI			2qx3			KNKSD		3hbu			TNVFG		9gaa		ACMEH	
1	285 FLF			"			ENIKN		3hbv			TNVFG		9gac		ACMEH	
1qa3	265 LEI			2qxz	92		KNKSD		3hwc			AVGTG		"	380	 ACMEH	
"	285 FLF	RGCHF	CKM		270		ENIKN		3inl	453		VNCYD		9gaf "		ACMEH	
																 ACMEH	

Of note in our *blbbl* β -arc consensus sequence, is the occurrence of L-asparagine (Asn, N), and less commonly L-aspartate (Asp, D), at the two positions that adopt backbone dihedral angles in the left-handed α -helix (*l*) region of the Ramachandran plot. This result is consistent with the observation that asparaginyl residues adopt conformations in partially-allowed regions of the Ramachandran plot more readily than any other non-glycyl amino acids.⁸ One theory is that the *l* conformation of an asparaginyl residue is stabilized by non-covalent attractive forces between the dipoles of the asparaginyl residue's side chain carbonyl, and the backbone carbonyl of either that residue or the previous residue.^{9,10}

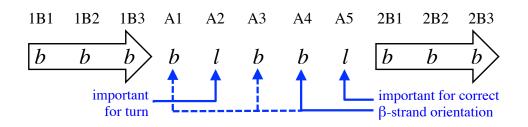


Figure S3. Contributions of backbone dihedral angles to *blbbl* β -arc.

To better characterize the influence of each element of the *blbbl* sequence on the β -arc's conformation, we mined the PDB (via DSSP) for sequences in which one element of *blbbl* was allowed to adopt any backbone conformation (n): nlbbl, bnbbl, blnbl, blbnl, and blbbn. In addition to the *blbbl* set, these searches returned culled sets of 1 *nlbbl*, 30 *bnbbl*, 2 *blnbl*, 3 *blbnl*, and 24 blbbn sequences. Of these, 1/1 nlbbl, 1/30 bnbbl, 2/2 blnbl, 1/3 blbnl, and 3/24 blbbn sequences were β -arcs (not necessarily 180°). For comparison, 12/13 of the *blbbl* sequences were 180° β-arcs. Examination of all these sequences revealed possible contributions of backbone angles to the β -arc (Figure S3). Many *bnbbl* sequences were parts of extended β -strands, which suggests that the *l* at position "A2" is important for creating a turn. Both *blbnl* sequences were β-turns, and many *blbbn* sequences created turns with flanking β-strands in orientations other than that of a β -arch; these results suggest that the *b* at position "A4" and the *l* at position "A5" are important for creating the distinctive β-strand orientation in a β-arch, which supports interactions between hydrophobic side chains on facing β -strands. With the importance of the *l* angles at positions "A2" and "A5" established, the culled PDB was searched for *nlnnl* sequences. Not counting *blbbl* sequences, the *nlnnl* search returned 20 sequences, 8/20 of which were 180° β -arcs. The *nlnnl* sequences that were not 180° β -arcs were often β -turns, or β -arcs with a < 180° change of chain direction; this suggested that the b angles at positions "A1," "A3," and "A4" contribute to the orientation of the β-strands in a 180° β-arch. However, the collective contribution of the *b* positions of *blbbl* to the orientation of the β -strands is probably not as important as the contribution of the "A5" l angle, given that 8/20 nlnnl sequences are still 180° β-arcs, while only 3/24 *blbbn* sequences are 180° β-arcs. Our set of 12 *blbbl* β-arches produced a consensus sequence with asparaginyl residues in both of the *l* positions. In contrast, in the consensus sequence for the 12 blbbl, 1 bnbbl, 3 blbbn, and 8 nlnnl (including 1 nlbbl, 1 blnbl, and 1 *blbnl*) β -arches, the most common amino acid at the *l* positions was glycine (Gly).

Kajava and coworkers identified single cases of non-*blbbl*, five-residue, 180° β -arcs: *blbbx*, *blbpl*, and *bldpl*.¹ The Ramachandran basins *p* and *d* neighbor the β -strand (*b*) basin. In this work we define a region x ($\Phi \in [30^\circ, 110^\circ]$, $\Psi \in [90^\circ, 120^\circ]$) that neighbors the left-handed α -helix (*l*) basin. Our searches identified no *blbbx* β -arches in our *blbbn* set, and one *blbpl* β -arch in our *blbnl* set. Mining the culled PDB for *blnnl* (including *bldpl*) sequences did not return any β -arches (besides the aforementioned two *blnbl* β -arches and one *blbnl* β -arch).

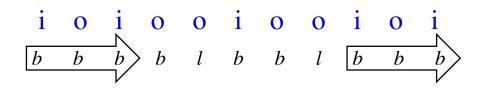


Figure S4. Solvent accessibility of residue side chains in the *blbbl* β -arc. Outside (o) residues are solvent-accessible; inside (i) residues are solvent-inaccessible.

For each β -arch in our culled dataset, we used established methods to determine the solvent-accessible surface area (ASA) of the residue side chains in the PDB structure;^{11,12} assigned "outside (o)" or "inside (i)" to side chains that were solvent-accessible or inaccessible, respectively; and generated the o/i consensus sequence in **Figure S4**. A side-chain was deemed to be solvent-accessible, if its relative accessibility was >5%, as calculated using NACCESS.¹²

Synthesis and Characterization of Peptides

The OAllyl-Gly-(1*R*,2*S*)-CHDA-Val-OH and Alloc-Glu-Val-^DPro-DADME-Fmoc linkers were synthesized via previously described routes.¹³ Peptide **1** was synthesized via the route depicted in **Figure S5**, and all other peptides were synthesized by analogous routes. Peptides were purified by reversed-phase high-performance liquid chromatography (HPLC), using a C18 preparatory column, and a mobile phase of CH₃CN:H₂O:CF₃COOH. Peptide identities were confirmed using matrix-assisted laser desorption/ionization time-of-flight mass spectrometry (MALDI-TOF MS). MALDI-TOF MS showed the following m/z values for the $[M+H^+]^+$ species: peptide **1** calculated = 3488.93, observed = 3488.0; peptide **2** calculated = 2722.50, observed = 2722.1; peptide **3** calculated = 3488.93, observed = 3489.5; peptide **4** calculated = 3574.94, observed = 3574.8; peptide **5** calculated = 3460.90, observed = 3460.6. Purity was determined by analytical HPLC or ultra performance liquid chromatography (UPLC), and was >95% for all purified peptides. Analytical HPLC and UPLC runs were performed using a C18 column on a 10-60% B solvent gradient, where solvent A is 99.9:0.1 H₂O:TFA and solvent B is 99.9:0.1 acetonitrile:TFA. All chromatograms recorded at 220 nm. (**Figures S6**).

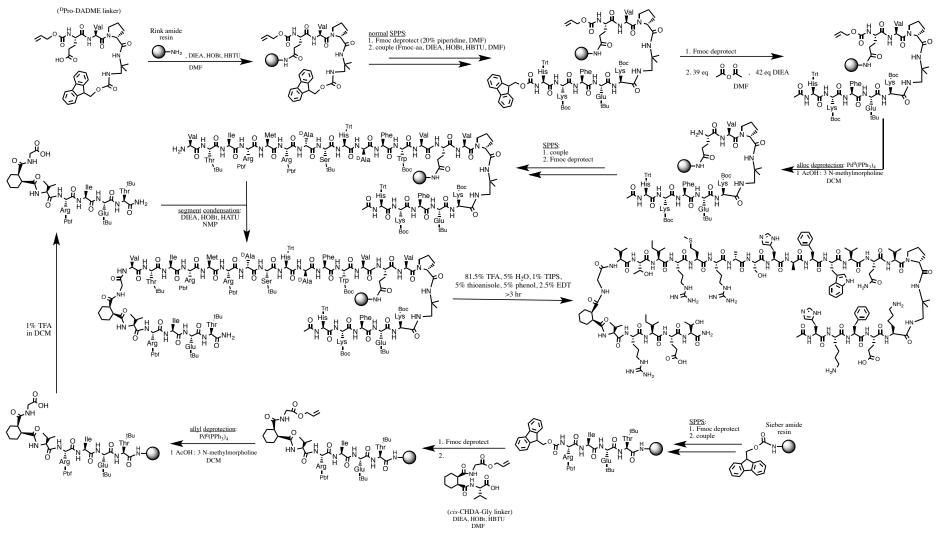


Figure S5. Synthetic route to peptide 1. Analogous routes were employed to synthesize other peptides.

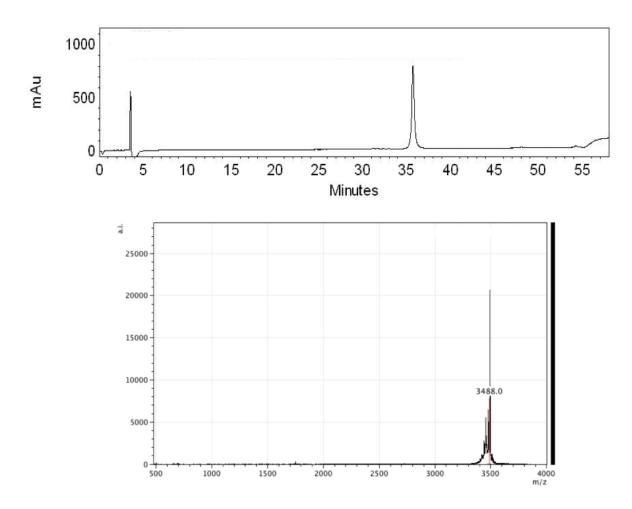
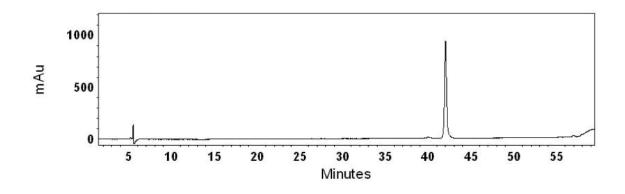


Figure S6A. Analytical HPLC trace and MALDI-TOF for peptide 1. [M+H]_{calc} = 3488.93



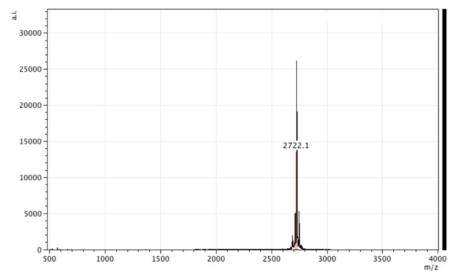


Figure S6B. Analytical HPLC trace and MALDI-TOF for peptide 2. [M+H]_{calc} = 2722.50

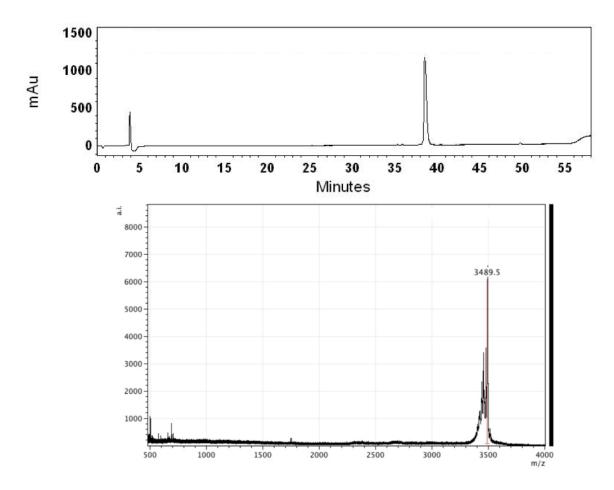


Figure S6C. Analytical HPLC trace and MALDI-TOF for peptide 3. [M+H]_{calc} = 3488.93

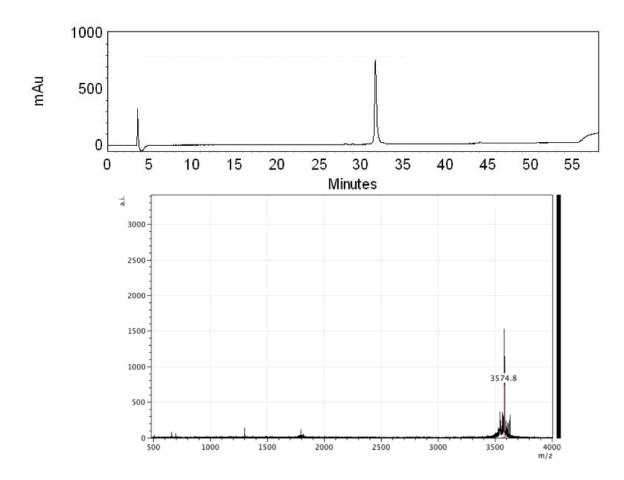


Figure S6D. Analytical HPLC trace and MALDI-TOF for peptide 4. $[M+H]_{calc} = 3574.94$

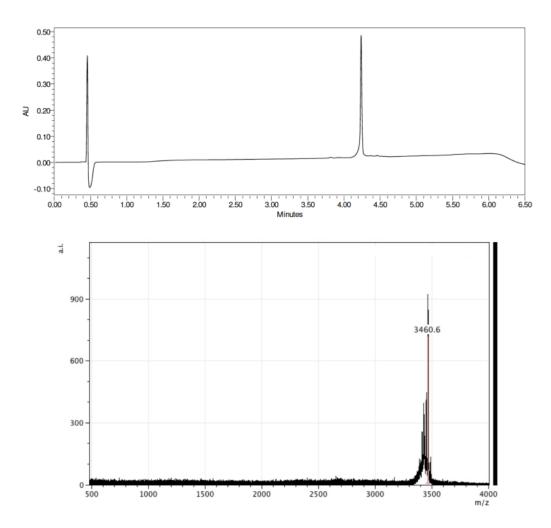


Figure S6E. UPLC trace and MALDI-TOF for peptide 5. $[M+H]_{calc} = 3460.90$

2D-NMR data acquisition and NMR structure determination

Each NMR sample was prepared by dissolving lyophilized peptide to 0.2 mM concentration in 9:1 H₂O:D₂O, pH 3.8, 2.5 mM NaOAc-d₃ buffer, with trace amounts of 2,2-dimethyl-2-silapentane-5-sulfonate (DSS) and/or dioxane as internal references. Peptide concentrations were determined by mass. Samples were prepared with total volumes of approximately 450 μ L in 5 mm NMR tubes. Peptide samples were stable in solution for at least 1 week at 4-5 °C, showing no visible precipitation of peptide, NMR line broadening, or decrease in NMR signal intensity over the entire period of study. Low salt conditions were employed to facilitate observation of weaker NMR signals. For each peptide, the following sets of 1D ¹H spectra were superimposable: [1] before and after collection of 2D NMR spectra (COSY, TOCSY, NOESY), and [2] 2 mM peptide in low salt buffer (9:1 H₂O:D₂O, 2.5 mM NaOAc-d₃, pH 3.8) and 2 mM peptide in high salt buffer (9:1 H₂O:D₂O, 100 mM NaOAc-d₃, pH 3.8).

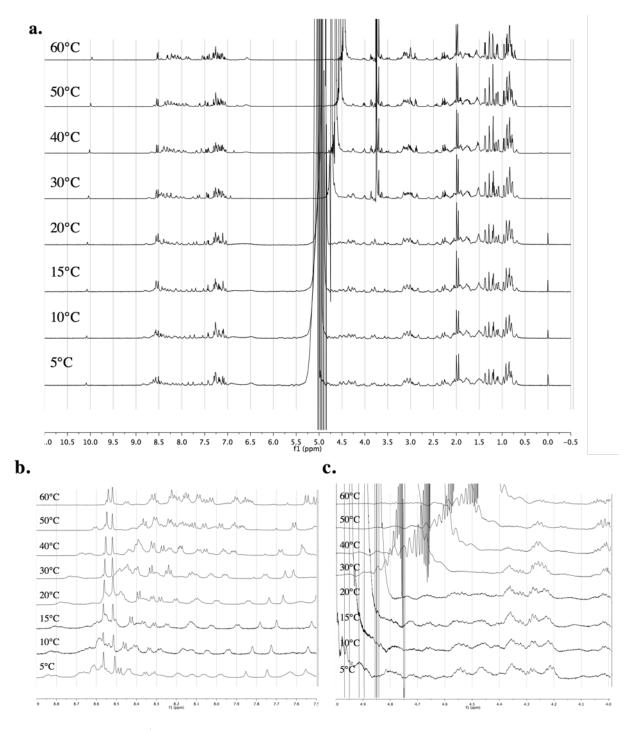


Figure S7. Increased ¹H NMR peak dispersion as temperature is decreased from 60 °C to 5 °C, for 0.2 mM peptide **1** in aqueous solution (2.5 mM NaOAc-d₃, pH 3.8, \pm DSS or dioxane). Panels (b) and (c) show the H α and HN regions, respectively.

Spectra were acquired at 5.0 °C on a Bruker Avance 600 MHz spectrometer, equipped with a 5 mm, z-axis gradient, triple resonance, cryogenic probe. The following standard Avance pulse programs were employed: 1D with water suppression using excitation sculpting (zgesgp),¹⁴ phase-sensitive 2D COSY-DQF with WATERGATE using 3-9-19 (cosydfgpph19),¹⁵⁻¹⁷ phase-

sensitive 2D TOCSY with WATERGATE 3-9-19 and mixing using DIPSI-2 (dipsi2gpph19),^{16,17} and phase-sensitive 2D NOESY with WATERGATE using water flip-back and 3-9-19 (noesyfpgpph19)¹⁶⁻¹⁸. Spectral widths of 8013-8418 Hz were used, with collection of 2048 points in the direct (f2) and 256 points in the indirect (f1) dimensions. TOCSY experiments used a mixing time of 100 ms. NOESY experiments used a mixing time of 250 ms. For peptide **1**, increasing the NOESY mixing time to 500 ms did not increase the number of NOEs observed. Note also that variable temperature studies suggested peptide unfolding at higher temperatures, with less peak dispersion (**Figure S7**), less downfield HN and H α chemical shifts, and fewer NOEs at higher temperatures. Data were processed using NMRPipe.¹⁹ Data were analyzed using Sparky,²⁰ with employment of sequential assignment procedures to assign chemical shifts of protons (**Tables S2**).²¹ Inter-residue NOEs are diagramed in **Figures S8**.

[NH	αH	β	н	γ	н	δ	н	ε	н		others	
1	Thr	8.41	4.35	3.	81	1.20								
2	Glu	8.51	4.34	1.65 1.37		2.82	2.82							
3	lle	8.62	4.53	2.03		1.28 1.09						0.89 (γ ² H)		
4	Arg	8.69	4.62	1.75	1.75	1.55	1.55	3.09	3.09			7.21 (εΗ)		
5	Val	8.38	4.37			0.87 0.87								
6					2.62 (0	xΗ), 2.	07 (βł	+), 1.9	3 (βH)	, 1.83	(βH),	1.80 (γH), 1.62 (γH), 1	39 (γH), 1.27 (γH)	
7			4.84,4.04											
8		8.12	4.22	2.3		0.96								
9	Thr	8.68	4.80	4.0	02	1.	09							
10		8.82	4.36	1.								0.81 (γ ² Η)		
11	-	8.55	4.42			1.54		3.13	3.13			7.25 (εΗ)	7.11 (εΗ)	
12		8.85	4.56			2.32								
13		8.36	4.20		1.17	1.45	1.45	2.84	2.84			7.54 (εΗ)		
14		8.45	4.38	1.										
15	Ser	8.41	4.32	3.79										
16		8.66	4.67	3.31				7.27	(δ²Η)			8.56 (ε¹H)		
17		8.19	4.23	1.										
18	Phe	8.32	4.78	3.09					29	7.	75	7.61 (zH)		
19	Trp	8.45	4.72	3.12				7.11	(δ¹Η)			7.56 (ε ³ Η) 7.17 (z ³ Η)	7.42 (h ² H) 7.05 (z ² H) 10.09 (ε ¹ H)	
20		8.24	4.10	1.9			0.87				_			
21		8.58	4.33			3.53				7.21	7.86			
22		7.98	4.41	1.9		0.84								
23	D-Pro		4.28			2.02			3.79					
24		<u> </u>	NH), 3.56											
25	,	8.62	4.49	2.42	2.02	1.53	1.53	1.78	1.78	3.02	3.02	7.63 (zH)		
26		8.56	4.51											
27		8.49	4.55	2.98					50		25	7.07 (zH)		
28	Lys	8.58	4.46		1.84	1.41	1.41			3.02	3.02	7.63 (zH)		
29	His	8.44	4.90	3.10	2.95			7.19	(δ²Η)			8.51 (ε¹Η)		
30	Ac		1.78											

Table S2A. Proton resonances (ppm) for peptide 1.

		NH	Η αΗ βΗ		γн		δ	н	εН		others	
7	Ac		1.52									
8	Val	8.58	4.51	2.	03	0.93	0.93					
9	Thr	8.33	4.24	3.	78	1.	16					
10	lle		4.14		82	1.17					0.84 (γ²H)	
11	Arg	8.33	4.34		1.18			2.83	2.83		7.54 (εH)	
12	Met		4.66		3.12							
13	Arg	8.55	4.56	1.85	1.85	1.61	1.61	2.93	2.93		7.58 (εH)	
14	D-Ala	8.37	4.40	1.	17							
15	Ser	8.27	4.35	3.83	3.83							
16	His	8.35	4.69	3.12	3.12			7.25	(δ²H)		8.55 (ε¹H)	
17	D-Ala	8.29	4.23	1.	12							
18	Phe	8.45	4.56	3.02	3.02			7.	11	7.25		
19	Trp	8.30	4.71	3.07	2.87			7.11	(δ¹H)			7.15 (h ² H) 7.43 (z ² H) 10.10 (ε ¹ H)
20	Val	8.18	4.08	1.	90	0.83	0.83					
21	Gln	8.52	4.26	2.31	2.31	1.30	1.19			7.84 7.84	1	
22	Val	8.32	4.13	2.	05	0.95	0.95					
23	D-Pro		4.37	2.24	2.24	2.05	1.93	3.78	3.78			
24	DADME	8.06 (NH), 3	3.52 (c	αH)							
25	Lys	8.56	4.47					1.80	1.73	3.01 3.0	L 7.64 (zH)	
26	Glu	8.53	4.45	1.50	1.50	1.80	1.71					
27	Phe	8.44	4.90	2.98	2.98			7.	04	7.41	7.54 (zH)	
28	Lys	8.47	4.25	2.44	2.44	1.76	1.76	1.55	1.55	3.10 3.1) 7.19 (zH)	
29	His	8.45	4.43	3.07	2.95			7.09	(δ²H)		8.50 (ε¹H)	
30	Ac		1.36									

 Table S2B. Proton resonances (ppm) for peptide 2.

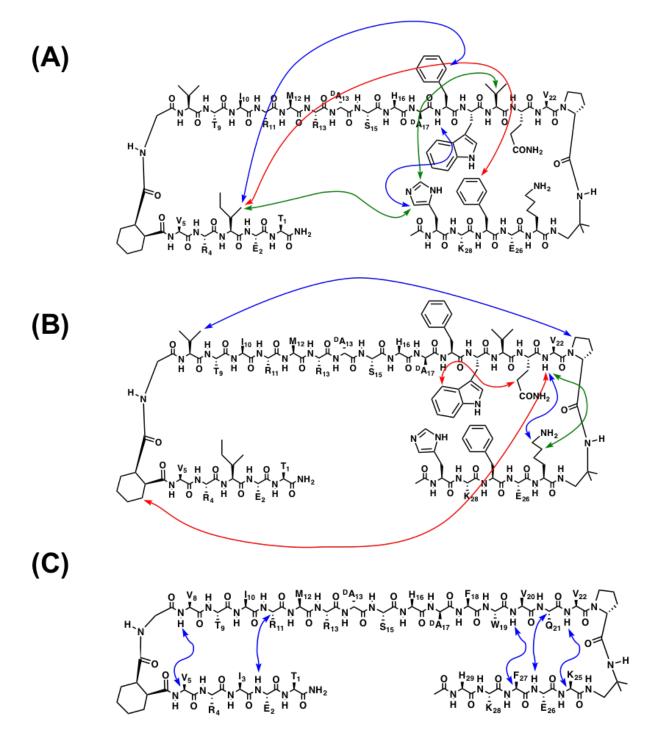


Figure S8. Graphical representation of NOEs observed for peptide **1** between protons on residues that are not adjacent in sequence. These medium- and long-range NOEs are highly diagnostic in terms of folding. (A) A subset of the NOEs observed for peptide **1** in which at least one proton is on a side chain. This subset features NOEs that include protons on the side chains of Ile-3, Phe-18, Val-20, Phe-27 and His-30. Different colors are used to help the reader follow individual arrows. (B) Another subset of the NOEs observed for peptide **1** in which at least one proton is on a side chain. This subset features NOEs that include protons on the side chains of the N-to-N linker (cyclohexyl ring), Val-8, Trp-19, Gln-21, Pro-23 and Lys-25. Different colors

are used to help the reader follow individual arrows. Collectively parts A and B of this figure show all NOEs between residues that are not adjacent in sequence and involving at least one side chain proton. (C) Inter-residue C α H - HN NOEs observed for peptide 1. Collectively, parts A-C show all NOEs that could be assigned for 1 involving protons on residues that are not adjacent in sequence. All of these NOEs are consistent with a single compact conformation (or family of related conformations) that corresponds to the mini-amyloid design goal (V. Kung, Ph.D. thesis, University of Wisconsin-Madison, 2016).

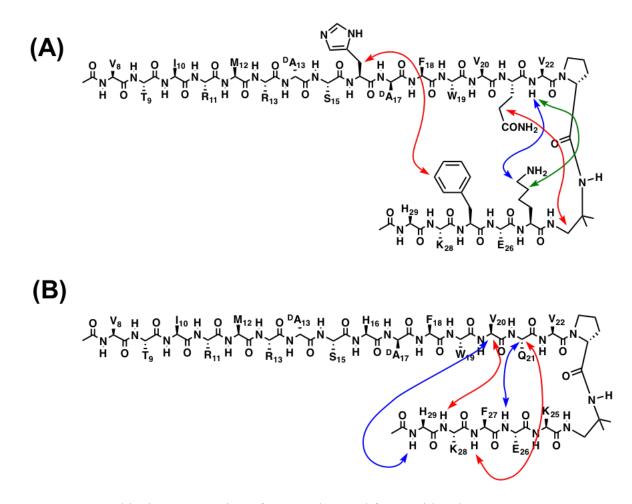


Figure S9. Graphical representation of NOEs observed for peptide **2** between protons on residues that are not adjacent in sequence. (A) NOEs observed for peptide **2** between protons from residues that are not adjacent in sequence, in which at least one proton is on a side chain. Different colors are used to help the reader follow individual arrows. (B) Inter-residue C α H - HN NOEs observed for peptide **2**. Different colors are used to help the reader follow individual arrows. It seems unlikely that all of these NOEs could arise from a single parallel β -sheet H-bonding pattern.

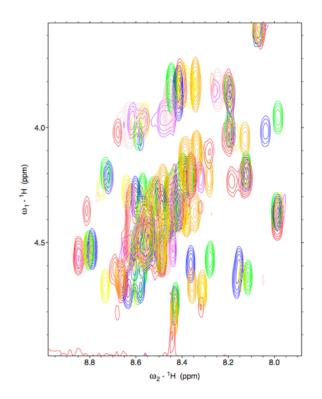


Figure S10. Overlay of H α -HN regions of 2-D TOCSY spectra for peptide 1 (red), 2 (orange), 3 (green), 4 (blue) and 5 (yellow). Also shown are data for two other variants of 1: replacement of β -arc residues ^DAla-14, His-16 and ^DAla-17 with Gly (pink), and replacement of all five β -arc residues, Arg-13 through ^DAla-17, with Gly (purple). More peak dispersion and more downfield chemical shifts are observed for H α and HN for peptide 1 relative to the other peptides. Spectra were taken at 5.0 °C, with 0.2 mM peptide, in 9:1 H₂O:D₂O, 2.5 mM NaOAc-d₃, pH 3.8.

Circular Dichroism (CD).

Circular dichroism can be used to assess protein folding in solution, with the association of α -helix (208 nm, 222 nm) and β -sheet (218 nm) secondary structures with characteristic minima in CD spectra.²¹ CD measurements were taken with an AVIV Model 420 Circular Dichroism Spectrometer, using 1.00 nm bandwidths and 10.000 sec averaging times, in 1.00 mm quartz cuvettes. For each sample measurement, the corresponding buffer blank measurement was subtracted, and the baseline CD signal at 320 nm was set to zero. Peptide solutions were prepared in aqueous solution with low (2.5 mM NaOAc, pH 3.8) or high (2.5 mM NaOAc, 100 mM NaCl, pH 3.8) concentrations of a charge-shielding salt. Peptide concentrations were determined by mass, and verified with UV absorption measurements. CD spectra were obtained at 20.0 °C (293.2 K).

Effects of added NaCl.

Introduction of 100 mM NaCl caused a profound change in the CD spectrum of peptide **1** at 20°C (Figure S11A). The residue-normalized CD signal became significantly more intense in the presence of added salt, and the position of the minimum shifted slightly to higher wavelength. Diffusion-ordered NMR spectroscopy (DOSY) and analytical ultracentrifugation

(AUC) measurements in the presence of 100 mM NaCl (2.5 mM NaOAc, pH 3.8) indicated that the added salt causes peptide 1 to self-associate. However, the AUC data could not be interpreted in terms of a single species and instead suggested a mixture of aggregated forms. These solutions showed no sign of precipitation over the course of several days.

Addition of NaCl to aqueous solution is known to screen Coulombic repulsions between solutes of like charge, such as 1 (an oligocation at pH 3.8). In addition, NaCl enhances the hydrophobic effect. Both of these effects could underlie the ability of added salt to induce self-association of 1 in aqueous solution. Because the CD minimum near 220 nm grows more intense in the presence of 100 mM NaCl (Figure S11A), we conclude that aggregation of 1 leads to enhanced formation of β -sheet secondary structure, and we speculate that there is enhanced population of an amyloid-like state.

Figure S11B shows that peptides 4 (two ^DAla \rightarrow Asn modifications in the β -arc) and 5 (two ^DAla \rightarrow Gly modifications in the β -arc) develop strong CD minima near 220 nm in the presence of 100 mM NaCl, even though neither of these peptides displays a distinct minimum in this region in aqueous buffer lacking salt (Fig. 4). We speculate that the enhanced population of β -sheet secondary structure indicated by CD for both 4 and 5 in the presence of NaCl arises because the salt has induced aggregation and formation of an amyloid like state, analogous to the behavior of 1.

Fig. S11C shows the effects of temperature on the per-residue CD signatures of peptides **1**, **4** and **5** in the presence of 100 mM NaCl in buffer was monitored at 220 nm. In each case, the CD intensity displays a cooperative transition over the observed temperature range (2° to 98°C); these transitions were fully reversible. The data for **4** and **5** display plateau regions at the lowest and highest temperatures, which suggest a fully folded/assembled state at the lowest temperatures and a fully unfolded/non-assembled state at the highest temperatures. The midpoints of these transitions ("melting temperatures") lie in the 45°-50°C range. The data for **1** show a low-temperature plateau similar to that seen for **4** and **5**; the data seem to approach a high-temperature plateau similar to that evident for **4** and **5**. Peptide **1** manifests a less cooperative transition (shallower slope at the midpoint) relative to **4** or **5**, but the data suggest that the assembly formed by **1** is more stable than those formed by **4** or **5** (higher mid-point temperature for **1**).

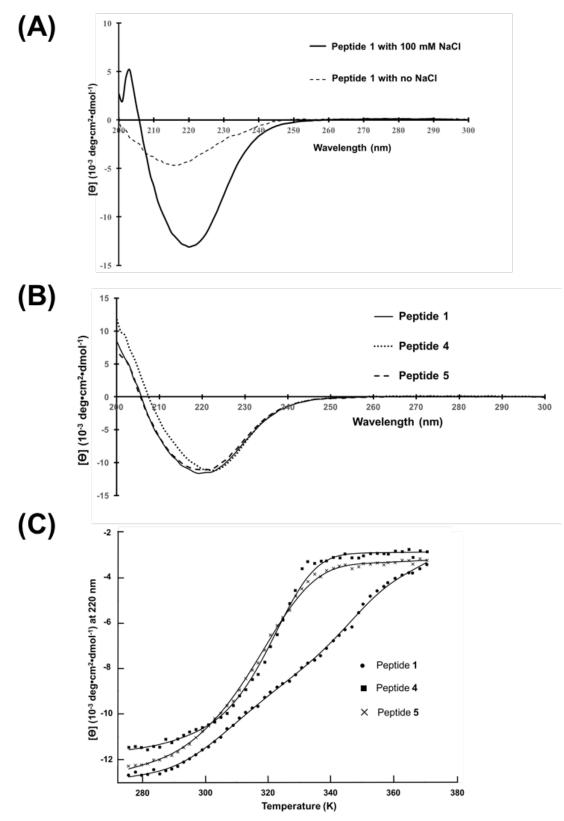


Figure S11. Effects of adding 100 mM NaCl on the per-residue circular dichroism of selected peptides. (A) Comparison of 0.2 mM 1 in aqueous 2.5 mM NaOAc, pH 3.8, 20° C (dashed line)

with the sample prepared under similar conditions but containing 100 mM NaCl (solid line). (B) Comparison of the per-residue circular dichroism of 0.1 mM peptide in aqueous 2.5 mM NaOAc, 100 mM NaCl, pH 3.8, 20° C: 1 (solid line), 4 (dotted line) and 5 (dashed line). (C) Mean residue ellipticity at 220 nm for 0.1 mM peptide in aqueous 2.5 mM NaOAc, 100 mM NaCl, pH 3.8, as a function of temperature: 1 (dots), 4 (squares) and 5 (crosses). Solid lines represent non-linear regression curve-fits for the thermal melt data.

Sedimentation Equilibrium Analytical Ultracentrifugation (AUC).

Our AUC experimental design and data analysis methods were reported in greater detail elsewhere.¹³ Sedimentation equilibrium data were collected using a Beckman Coulter Model XL-A Analytical Ultracentrifuge at 4 °C, for peptide **1**. Peptide solutions were prepared by simple dissolution in buffer (2.5 mM NaOAc, pH 3.8, with peptide concentrations of 0.09, 0.06, and 0.03 mM) and used without further manipulation. Initial peptide concentrations were determined from UV absorption spectra of samples diluted into 6 M GdmCl, using an extinction coefficient (ε) of 5690 M⁻¹ cm⁻¹.²² Gradients were monitored at the wavelength of maximal absorbance (276 nm for peptide **1**). Data were collected at rotor speeds of 10k, 15k, 23k, 32k, 42k. For each rotor speed, we assumed samples had reached equilibrium when gradients collected at least 3 hours apart were superimposable. After data collection at the highest rotor speed, gradients at one of the lower speeds were measured again and found to be superimposable with original measurements, indicating that there had been no irreversible loss of material during the course of the experiment.

The equilibrium data were analyzed using software written by Darrell R. McCaslin, in a manner analogous to previously-described analysis methods.²³ In the analysis, the solvent density (ρ) at 4 °C was computed using density increment functions to be 1.001 g/mL.²⁴ The partial specific volume ($\overline{\nu}$) of peptide **1** (0.736 cm³ g⁻¹) was calculated using consensus values for the natural amino acid residues, and corrections for acetyl, carboxamide, DADME, and CHDA groups.^{25,26} The molecular weight (M_S) of peptide **1** (3489.2 Da) was calculated based on the peptide sequence, including non-natural linkers and termini.

Plots of the logarithm of the measured absorbance as a function of the squared radial distance from the center of rotation were linear, at all speeds and for all loading concentrations (**Figure S12**). Linearity in such plots is indicative of the presence of a single molecular weight entity.

The final estimate of the weight-average molecular weight (M_P , or MW_{obs}), reported in **Table S3**, was obtained from global fits of the data at the varied loading concentrations and speeds to the distribution of a single macromolecular species. Inclusion of a term for a second macromolecular species worsened fits.

The AUC experiments were designed to examine the peptide association state at low-salt buffer conditions similar to those employed for NMR structural studies. For peptide 1, the weight-average molecular weight (MW_{obs}) was much smaller than the molecular weight based on the sequence (M_s). Furthermore, the apparent weight decreased with increasing concentration. These results suggest that at the low-salt conditions studied, the peptide is monomeric, and not forming dimers or larger molecular weight aggregates. We conclude that the observed $MW_{obs}/MW_{calc} \le 1$ indicates a monomeric state, with electrostatic nonideality accounting for $MW_{obs}/MW_{calc} << 1$.

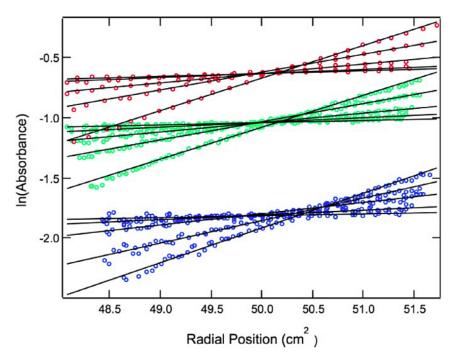


Figure S12. Equilibrium AUC data for peptide **1** in 2.5 mM NaOAc, pH 3.8. The data shown are for speeds of 10k, 15k, 23k, 32k, 42k, at initial loading concentrations of 0.03 mM (blue), 0.06 mM (green), and 0.09 mM (red).

Table S3. Sedimentation equilibrium results for peptide 1. Weight-average molecular weight (MW_{obs}) from fits to a single-species model, reported with least-squares fitting error in the fitted parameter.

buffer	${ m MW_{obs}}/{ m M_S}$	MW _{obs} (Da)		
2.5 mM NaOAc, pH 3.8	0.693 ± 0.004	2417		

NMR diffusion experiments

NMR experiments measuring peptide diffusion were performed. Samples were prepared by dissolving lyophilized peptide to ~2 mM concentration, with 0.6 mM dioxane, in aqueous buffer (9:1 H₂O:D₂O, 2.5 mM d₃-acetate, pH 3.8). Serial dilutions were made of these peptide solutions into buffer, to yield 0.2 mM and 0.02 mM peptide solutions. Spectra were acquired on a Bruker Avance 600 MHz spectrometer, equipped with a 5 mm, z-axis gradient, triple resonance, cryogenic probe. Self-diffusion rates were measured using a pulsed field gradient stimulated spin echo (PFGSTE) technique.²⁷ Spectra were acquired at 5.0 °C, with 75 pulsed field gradient (PFG) strengths in the 5-60 G/cm range, and acquisition of 16-240 transients per gradient experiment. Gradient pulses were applied for 5 ms, with a diffusion delay of 100 ms.

Data were analyzed using the variable gradient least square fitting routines in NMRPipe,¹⁹ and peak intensities were fit with a single exponential decay function to determine translational self-diffusion constants. For each condition studied, the self-diffusion constant of the peptide represented an average of values obtained from the intensities of at least 4 separate peaks, including peaks in the aromatic and aliphatic regions. The hydrodynamic radius (R_H) of

the peptide was extracted by comparison of peptide diffusion data with that of an internal standard, 1,4-dioxane ($R_H = 2.12$ Å).²⁸

The hydrodynamic radii derived from NMR diffusion experiments (obs R_H) are reported in **Table S4**. The predicted hydrodynamic radii of the monomeric peptides were calculated using the atomic-level (a = 2.84 Å) shell-model of HYDROPRO 10,²⁹ and are in good agreement with the hydrodynamic radii derived from NMR diffusion datas (**Table S4**). Furthermore, variable concentration NMR diffusion analysis of the peptides shows little to no increase in R_H values as peptide concentrations are increased from 0.02 mM to 0.2 mM. Also, no changes in chemical shifts, peak shapes, or splitting were observed on 1D ¹H NMR as peptide concentrations were increased from 0.02 mM to 0.2 mM. These results support the conclusion that peptides are monomeric (no peptide aggregation occurs) under the conditions used to obtain NMR structures of the peptides.

Table S4. Hydrodynamic radii of peptides, derived from NMR diffusion experiments (obs R_H) or HYDROPRO 10 calculations for monomeric peptides (calc R_H). All peptides are in 2.5 mM NaOAc, pH 3.8 buffer.

Peptide	Obs R_H (Å) At 0.2 mM peptide	Obs R_H (Å) At 0.02 mM peptide	Calc R _H (Å)
1	9.3 ± 0.6	8.3 ± 1.7	13.5
2	9.8 ± 0.4	9.5 ± 0.8	15.2
3	17.6 ± 0.4		
4	14.0 ± 1.1		
5	11.4 ± 0.4	8.4 ± 1.3	

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