

A Density Functional Theory Study on the Role of Polyacetylene as a Promoter in Selective Hydrogenation of Styrene on a Pd Catalyst

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Table S1. Bader charge analysis of the electronic rearrangement upon Pd adsorption

	type	X	Y	Z	CHARGE	MIN	ATOMIC
Pd25	Pd ₄ /PA	4.5968	8.5332	3.2393	9.891	1.0307	81.1384
Pd26	Pd ₄ /PA	4.9077	5.8752	3.3268	9.8984	1.0319	87.2486
Pd27	Pd ₄ /PA	3.4747	7.1618	5.2251	10.1023	1.1564	533.7344
Pd28	Pd ₄ /PA	5.9763	7.3728	5.2739	10.096	1.1635	546.0398
Pd25	Pd ₄ /PE	5.2188	9.3088	4.0953	10.0036	1.1238	88.0698
Pd26	Pd ₄ /PE	5.6721	6.7428	4.3966	10.0004	1.1068	120.908
Pd27	Pd ₄ /PE	4.2889	8.0633	6.2127	10.0142	1.149	451.9331
Pd28	Pd ₄ /PE	6.8138	8.4565	5.9898	10.0326	1.1295	419.8379

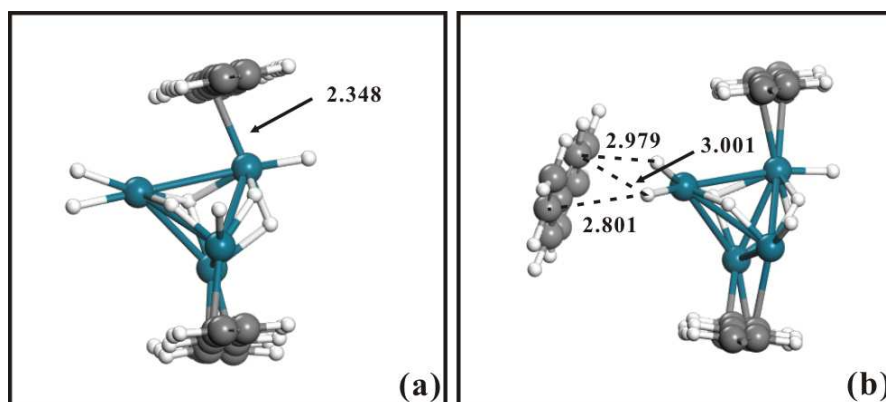


Figure S1. The optimized structures of (a) Pd₄H₈/2PA, and (b) styrene adsorbed Pd₄H₈/2PA, where the calculated styrene adsorption energy is -0.094 eV, significantly lower than the value on Pd₄H₈/PA (1.57 eV)

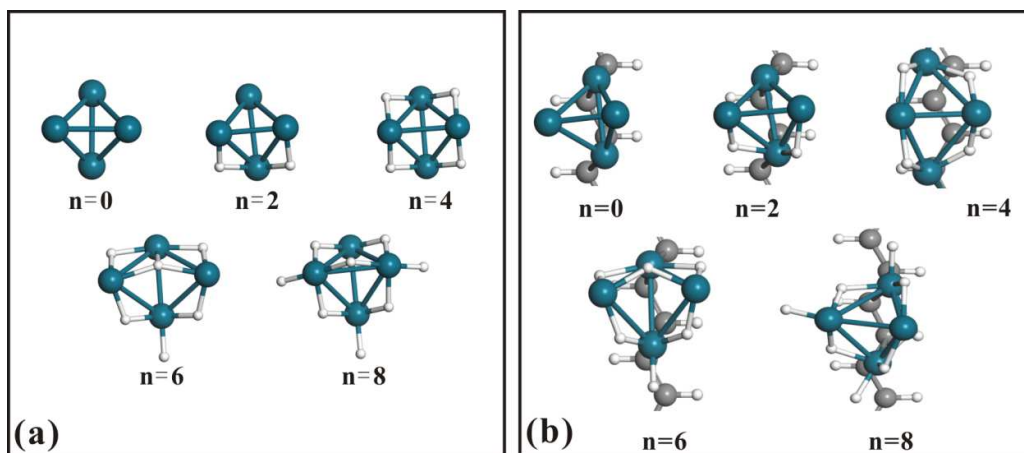


Figure S2. Optimized structure of H atoms adsorb on (a) Pd₄ cluster, and (b) Pd₄/PA cluster.

Table S2. Average bond length of Pd-Pd and Pd-H for the Pd₄ cluster and Pd₄/PA

Number of H atoms	Pd ₄		Pd ₄ /PA	
	Pd-Pd (Å)	Pd-H (Å)	Pd-Pd (Å)	Pd-H (Å)
0	2.608	-	2.657	-
2	2.741	1.785	2.756	1.686
4	2.743	1.714	2.996	1.717
6	2.995	1.778	2.910	1.734
8	2.889	1.717	2.860	1.728

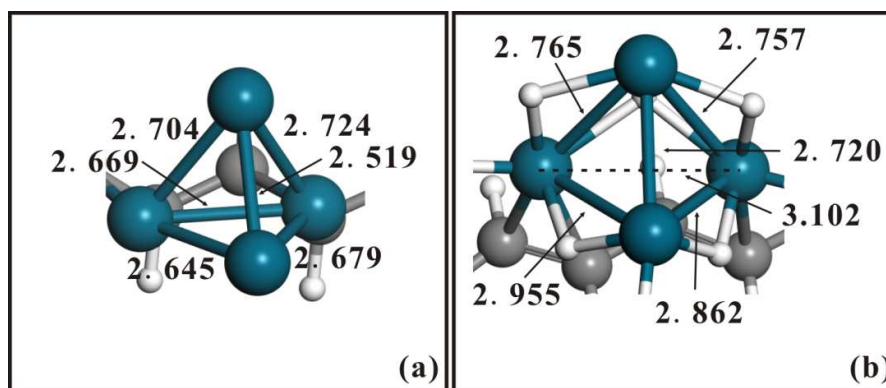


Figure S3. Optimized structure of (a) pure Pd₄ cluster on polyacetylene, (b) Pd₄H₈ supported on polyacetylene.

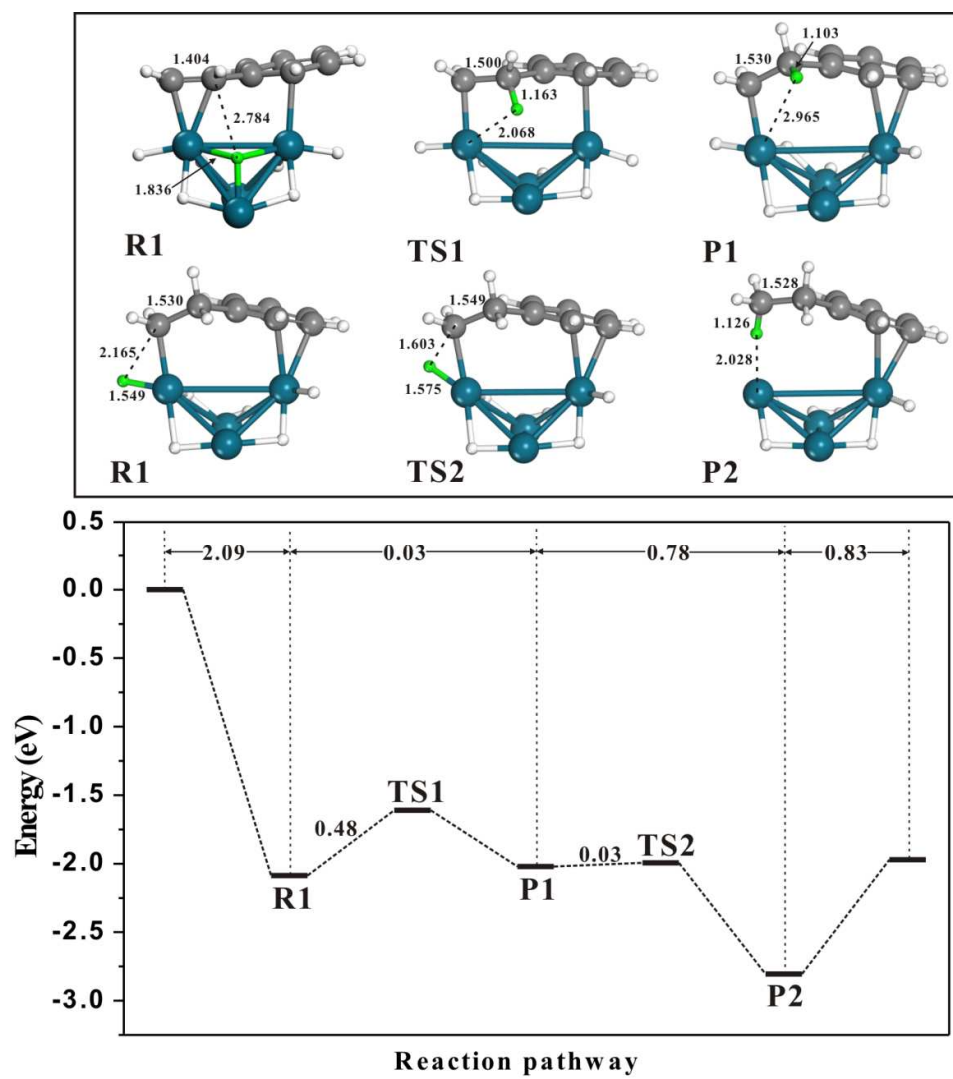


Figure S4. Calculated energy diagram of the hydrogenation process of the second route on the Pd_4H_8 cluster.

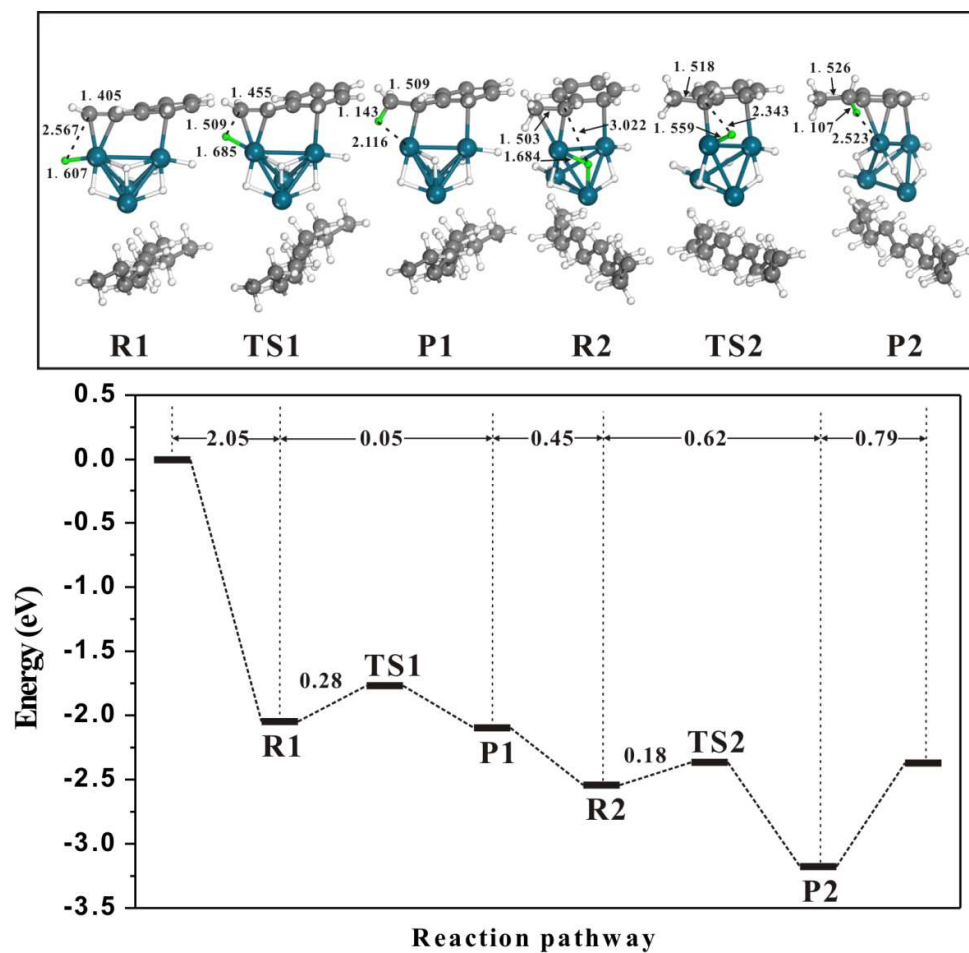


Figure S5. Calculated energy diagram of the hydrogenation process of the styrene on the Pd_4H_8 /polyethylene.