
Supporting Information: Proton Conductivity in Hydrogen Phosphate/Sulfates from a Coupled Molecular Dynamics/Lattice Monte Carlo (cMD/LMC) Approach

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Supporting Information

Convergence Behavior

The full numerical values of the diffusion coefficient with respect to the simulated AIMD length are shown in table 1. The diffusion coefficients obtained from the cMD/LMC approach only differ by a factor of two for variations of the underlying AIMD trajectory length. An excellent agreement up to a factor of two between diffusion coefficients obtained from AIMD and cMD/LMC approach can be reported for trajectory lengths larger than 30 ps. As expected, for simulated times in the range of 30 ps the AIMD diffusion coefficients strongly deviate from their corresponding values for longer simulations.

So we can state that the cMD/LMC approach enables us to estimate the diffusion coefficient from a very short AIMD simulation. A 30 ps AIMD trajectory is sufficient to decide if a compound shows a high proton conductivity or not.

Table 1: Diffusion Coefficients [$\text{\AA}^2/\text{ps}$] from different compounds with respect to different AIMD trajectory lengths.

	CsH ₂ PO ₄ cubic (HTP) 510K	CsH ₂ PO ₄ monoclinic (LTP) 490K	CsH ₂ PO ₄ cubic (HTP) 610K	CsHSO ₄ tetragonal (HTP) 420 K
30 ps underlying AIMD trajectory				
D _k cMD/LMC	$(4.9 \pm 1.0) \cdot 10^{-3}$	$\sim 0.0^a$	$(1.1 \pm 0.2) \cdot 10^{-2}$	$(2.0 \pm 0.5) \cdot 10^{-3}$
D _k AIMD	$(2.0 \pm 0.6) \cdot 10^{-2}$	$\sim 0.0^a$	$(1.0 \pm 0.2) \cdot 10^{-2}$	$(5.7 \pm 4.1) \cdot 10^{-3}$
60 ps underlying AIMD trajectory				
D _k cMD/LMC	$(5.9 \pm 1.0) \cdot 10^{-3}$	$\sim 0.0^a$	$(2.1 \pm 0.2) \cdot 10^{-2}$	$(2.0 \pm 0.5) \cdot 10^{-3}$
D _k AIMD	$(8.3 \pm 1.5) \cdot 10^{-3}$	$\sim 0.0^a$	$(2.0 \pm 0.3) \cdot 10^{-2}$	$(1.8 \pm 1.5) \cdot 10^{-3}$
120 ps underlying AIMD trajectory				
D _k cMD/LMC	$(5.4 \pm 1.0) \cdot 10^{-3}$	$\sim 0.0^a$	$(2.6 \pm 0.2) \cdot 10^{-2}$	$(2.6 \pm 0.5) \cdot 10^{-3}$
D _k AIMD	$(2.7 \pm 0.4) \cdot 10^{-3}$	$\sim 0.0^a$	$(1.8 \pm 0.1) \cdot 10^{-2}$	$(0.7 \pm 1.0) \cdot 10^{-3}$
experimental	$0.5 - 6.5 \cdot 10^{-3}^b$	$\sim 0.0^b$	$\sim 5.5 \cdot 10^{-2}^c$	$\sim 1 \cdot 10^{-3}^c$
D _k	$2.9 - 25 \cdot 10^{-3}^d$		(560 K)	

^aD_k very small ($\leq 3 \cdot 10^{-5}$) and error larger than the value itself

^bBaranov, Khiznichenko, Shuvalov, *Ferroelectrics*, 100, **1989**,135-141.

^cBelushkin, Carlile, Shuvalov, *J. Phys.: Condens. Matter*, 4, 2, **1992**, 389-398.

^dIshikawa, Maekawa, Yamamura, Kawakita, Shibata,Kawai, *Solid State Ionics*, **2008**, 179, 2345-2349.

Jump Rate of CsH₂PO₄ and CsHSO₄

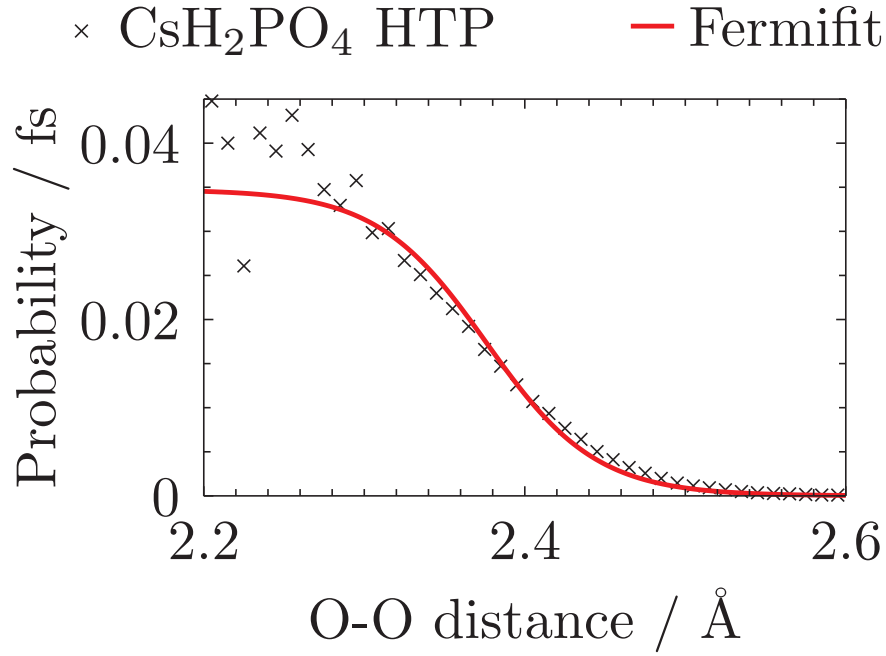
The distance dependent jump rates obtained from the AIMD and the resulting Fermi fit function are depicted in figure 1.

From this figure it becomes clear that the magnitude of the jump rate of CsHSO_4 is two orders decreased compared to CsH_2PO_4 . The Fermi fit was done according to:

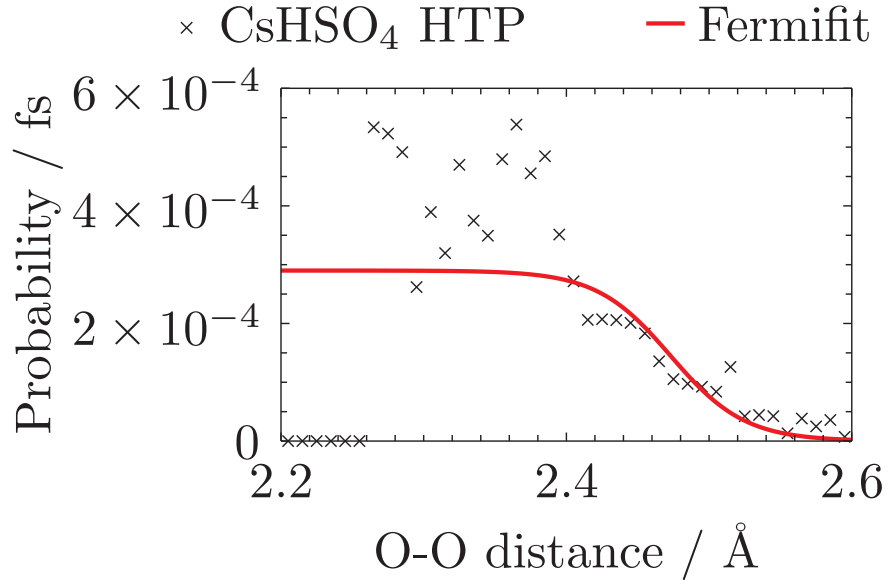
$$\omega(d) = \frac{a}{1 + \exp(\frac{d-b}{c})} \quad (1)$$

Thermostat Coupling and Jump Rate

The jump rates of CsH_2PO_4 (HTP) obtained from AIMDs with two different time constants of the thermostat (100 fs and 1000 fs) are depicted in figure 2. A ten fold increased time constants of the thermostat, corresponding to a drastically reduced coupling of the thermostat, does not significantly alter the jump rate. So we conclude a minor influence of the coupling strength of the thermostat on the proton transfer rates.



(a) Distance dependent jump rate of CsHSO₄ (HTP).



(b) Distance dependent jump rate of CsH₂PO₄ (HTP).

Figure 1: Comparison of distance dependent jump rates. The jump rates differs by a factor of 100.

× 510 K 1000 coupl. □ 510 K 100 coupl.

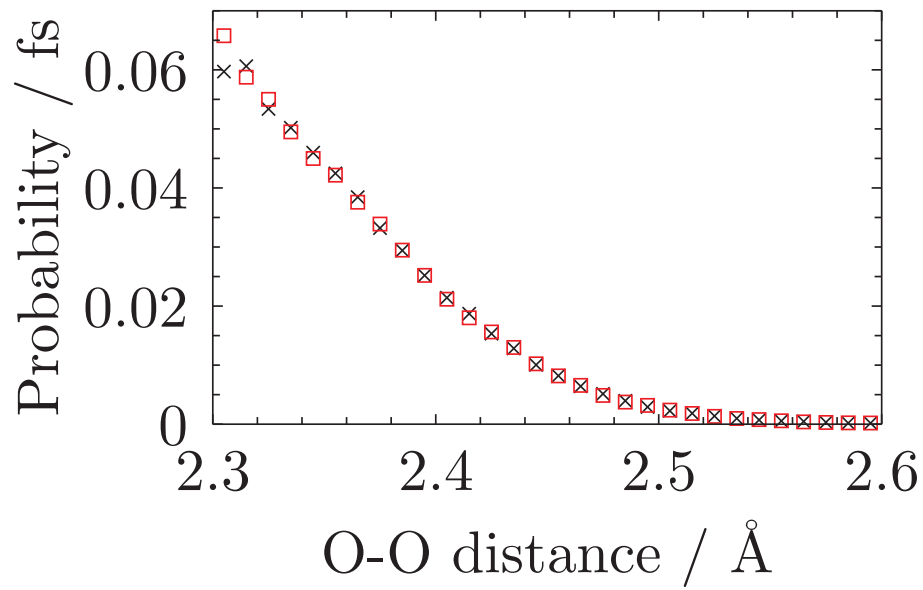


Figure 2: Comparison of the jump rates of CsH₂PO₄ (HTP) obtained from AIMDs with two different time constants of the thermostat (100 fs and 1000 fs).