

Anal. Chem., 1996, 68(18), 3194-3198, DOI:10.1021/ac960315q

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Figure S1 depicts the intensity (F_o/F) and lifetime (τ_o/τ) Stern-Volmer plots for BSA-Ac (top panel) and HSA-Ac (bottom panel). The fact that (F_o/F) and (τ_o/τ) are superimposable and that the relationships are linear demonstrate that dynamic (collisional) quenching alone governs both systems. The slopes ($K_{SV} = 7.1 \pm 0.3$ and 8.2 ± 0.2 M⁻¹ for BSA- and HSA-Ac, respectively) are presented in Table 1 of the paper.

Figure S2 depicts the Stern-Volmer plot for BSA-Ac covalently attached to APTES/glutaraldehyde-derivatized silica. The slope of this line is $9.0 \pm 0.7 \, \text{M}^{-1}$ and is also presented in Table 1. The error bars on the data points represent the standard deviation of at least three independent measurements. The reported uncertainty in the K_{SV} values are determined from the standard deviation of the slope.

Figure S3 represents the emission center of gravity^{S1,S2} as a function of O_2 concentration for native BSA- and HSA-Ac. It is clear that the emission spectrum of BSA-Ac (panel A) shifts to higher energies when quenched with O_2 while the HSA-Ac spectrum (panel B) remains essentially stationary. This is consistent with the K_{SV} vs emission wavelength plots that are presented in the paper (Figure 2).

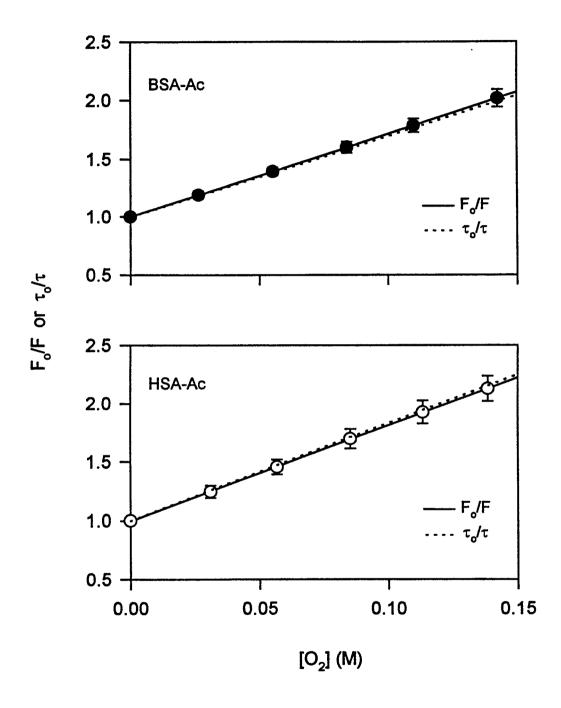
References

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- S2. Lakowicz, J.R.; Hogen, D. Biochemistry, 1981, 20, 1366-73.

Figure Captions

- Figure S1. Fluorescence Intensity (—, F_o/F) and lifetime (- - -, τ_o/τ) Stern-Volmer plots for BSA- and HSA-Ac quenched with O_2 .
- Figure S2. Stern-Volmer plot for BSA-Ac which is chemically bonded to APTES/glumodified silica quenched with O₂.
- Figure S3. Emission center of gravity as a function of O₂ concentration for BSA-Ac (panel A) and HSA-Ac (panel B) in buffer solution.



Ingersoll et al., Fig. S1

