Total Synthesis of (-)-Hennoxazole

Thomas E. Smith,* Wen-Hsin Kuo, Emily P. Balskus, Victoria D. Bock, Jennifer L. Roizen, Ashleigh B. Theberge, Kathleen A. Carroll, Tomoki Kurihara, and Jeffrey D. Wessler

Department of Chemistry, Williams College, Williamstown, Massachusetts 01267

Supporting Information

Table of Contents

General Information	S2
Experimental details and characterization data for 21	S3
Experimental details and characterization data for 22	\$3
Experimental details and characterization data for 23	S4
Experimental details and characterization data for 24b	S4
Experimental details and characterization data for 40	S5
Experimental details and characterization data for 5	S6
Experimental details and characterization data for 51	S7
Experimental details and characterization data for 54	S8
Experimental details and characterization data for 4a	S9
Experimental details and characterization data for 24a	S9
Experimental details and characterization data for 27	S10
Experimental details and characterization data for 28	S10
Experimental details and characterization data for 29	S11
Experimental details and characterization data for 30	S11
Experimental details and characterization data for 41	S12
Experimental details and characterization data for 42	S12
General procedure for TiCl ₄ /(<i>i</i> -Pr) ₂ NEt aldol-type reactions with acetals	S 13
General procedure for PhBCl ₂ /sparteine aldol-type reactions with acetals	S13
Characterization data for 45a	S14
Characterization data for 46a	S14
Characterization data for 45b	S14
Characterization data for 46b	S15
Characterization data for 47a	S15
Characterization data for 48a	S15
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 21	S17-18
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 22	S19-20
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 23	S21-22
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 24b	\$23-24
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 7	\$25-26
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 37	S27-28
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 40	S29-30
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 5	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for $47b$	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 51	\$35-36
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 53	\$37-38

¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 54	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 4a	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 55	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 1a	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (acetone- <i>d</i> ₆) for 1a	S47-48
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 24a	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 27	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 28	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 29	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 30	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 41	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 42	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 45a	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 46a	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 45b	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 46b	S69-70
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 47a	
¹ HNMR (500 MHz) and ¹³ CNMR (125 MHz) spectra (CDCl ₃) for 48a	

General Information. Oxygen- or moisture-sensitive reactions were carried out in flame-dried or ovendried glassware sealed with rubber septa under a positive pressure of dry nitrogen. Similarly sensitive liquids and solutions were transferred by gas-tight syringe or cannula. Unless indicated otherwise, reagents and solvents were purchased and used without purification. Ether, THF, and CH₂Cl₂, were purified by passage through a bed of activated alumina.¹ Analytical TLC was performed with 0.25 mm silica gel 60 plates with 254 nm fluorescent indicator from SiliCycle. Plates were visualized under UV light and treatment with either acidic *p*-anisaldehyde stain or aqueous ceric ammonium molybdate (CAM) solution followed by gentle heating. The term flash chromatography refers to preparative silica gel column chromatography as described by Still and co-workers.² Silica gel 60, 230-240 mesh, was purchased from SiliCycle (R10030B). ¹H NMR spectra are reported in ppm using tetramethylsilane (0.00 ppm) or solvent (CDCl₃: 7.24 ppm; acetone-d₆: 2.04 ppm) as an internal standard. Data are reported as (ap = apparent, s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, b = broad; coupling constant(s) in Hz; integration. Proton-decoupled ¹³C NMR spectra were recorded at 125 MHz and are reported in ppm using solvent as an internal standard (CDCl₃: 77.00 ppm, acetone-d₆: 206.00 ppm). Unless noted otherwise on the spectra, NMR spectra are recorded in CDCl₃. Infrared spectra were recorded as thin films on NaCl plates on a Fourier transform spectrometer (FTIR). Melting points are uncorrected. Optical rotations were measured using a sodium (589, D line) lamp and are reported as follows: $[\alpha]_{\lambda} T^{\circ C}$ (c = g/100 mL, solvent).

⁽¹⁾ Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. Organometallics 1996, 1518.

⁽²⁾ Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem 1978, 43, 2923.

Bisoxazole Methyl Ester 21.³



Following Wipf & Williams' method,⁴ a solution of serine amide **20** (1.000 g, 4.38 mmol, 1 equiv) in CH₂Cl₂ (40 mL) was cooled to -20 °C and Deoxo-Fluor (0.89 mL, 4.82 mmol, 1.1 equiv) was added dropwise. After 15 min, TLC showed complete consumption of starting material. Subsequent addition of BrCCl₃ (1.56 mL, 15.8 mmol, 3.6 equiv) followed by DBU (2.35 mL, 15.8 mmol, 3.6 equiv) led to an immediate darkening of color. The reaction mixture was allowed to warm to 0 °C and was stirred at that temp for 3 h. The reaction was quenched by the addition of sat NaHCO₃ (40 mL) at 0 °C, and the resulting suspension was extracted with EtOAc (3 × 50 mL). The combined organics were dried over MgSO₄, filtered, and the solvent removed *in vacuo* to provide the crude product as a light brown solid. Silica gel chromatography (60% EtOAc/hexanes; TLC: $R_f = 0.57$ in 100% EtOAc, CAM stain) gave bisoxazole methyl ester **21** (752 mg, 82%) as a white solid. ¹H NMR (500 MHz, CDCl₃) δ 8.30 (s, 1H), 8.28 (s, 1H), 3.95 (s, 3H), 2.56 (s, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 162.8, 161.3, 155.8, 143.5, 139.2, 134.2, 129.6, 52.2, 13.7 ppm.

Preparation of Bisoxazole Aldehyde 22.5



To a solution of bisoxazole ester **21** (328 mg, 1.58 mmol, 1 equiv) in CH₂Cl₂ (50 mL) at -78 °C was added DIBAL-H (3.94 mL of a 1.0 M solution in CH₂Cl₂, 3.94 mmol, 2.5 mmol) dropwise. The reaction mixture was stirred at -78 °C for 20 min and was then quenched by the addition of methanol (10 mL) and half-saturated aq Rochelle's salt. The resulting suspension was allowed to warm to rt and was stirred for 5 h until two distinct layers formed. The layers were separated and the aqueous layer was extracted with EtOAc (2 × 40 mL). The combined organics were dried over MgSO₄, filtered, and the solvent removed *in vacuo* to provide bisoxazole aldehyde **22** (287 mg, 100%) as a yellow solid, which was used without further purification (TLC: $R_f = 0.42$ in 40% Et₂O/CH₂Cl₂, UV, CAM stain). Mp 175.5–175.8 °C; ¹H NMR (500 MHz, CDCl₃) δ 10.00 (s, 1H), 8.32 (s, 1H), 8.24 (s, 1H), 2.57 (s, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 184.1, 163.1, 156.3, 143.5, 141.5, 139.3, 129.5, 13.7 ppm; IR (film) 3124, 1686, 1295, 1206 cm⁻¹; HRMS (EI): Exact mass calcd for C₈H₆N₂O₃ [M]+: 178.0378; Found: 178.0387. Elemental analysis calcd for C₈H₆N₂O₃: C, 53.94; H, 3.39; N, 15.73; Found: C, 54.08; H, 3.45; N, 15.70.

⁽³⁾ Chattopadhyay, S. K.; Kempson, J.; McNeil, A.; Pattenden, G.; Reader, M.; Rippon, D. E.; Waite, D. J. Chem. Soc., Perkin Trans. 1, 2000, 2415–2428.

⁽⁴⁾ Phillips, A. J.; Uto, Y.; Wipf, P.; Reno, M. J.; Williams, D. R. Org. Lett. 2000, 2, 1165–1168.

⁽⁵⁾ Williams, D. R.; Brooks, D. A.; Moore, J. L.; A. Stewart, A. O. Tetrahedron Lett. 1996, 37, 983–986.

Bisoxazole Dimethylacetal 23.



To a solution of bisoxazole aldehyde **22** (502 mg, 2.82 mmol, 1 equiv) in CHCl₃ (20 mL, it is essential to select a grade that is *not* stabilized with ethanol) at -30 °C was added TMSOMe (1.16 mL, 8.45 mmol, 3.0 equiv) followed by TMSOTf (153 µL, 0.845 mmol, 0.3 equiv) and the mixture turned slightly cloudy and took on a deeper orange color. After 14 h at -30 °C, an additional 1.0 equiv of TMSOMe and 0.1 equiv TMSOTf were added and the reaction was stirred for another 24 h before quenching with pyridine (1 mL). The mixture was warmed to rt and was poured into sat aq NaHCO₃ (40 mL) and was extracted with CH₂Cl₂ (2 × 40 mL). The combined organic layers were dried over a 1:1 mixture of K₂CO₃ and Na₂SO₄, filtered, and concentrated *in vacuo* to yield a white solid. Purification by silica gel chromatography (Et₃N-deactivated silica gel, 1:1:1 Et₂O/CH₂Cl₂/hexanes; TLC: R_r = 0.71, 100% Et₂O, UV) provided bisoxazole dimethylacetal **23** (609 mg, 96%) as a white solid. Mp 63.5–64.0 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.17 (s, 1H), 7.72 (s, 1H), 5.48 (s, 1H), 3.39 (s, 6H), 2.54 (s, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 162.6, 155.5, 139.5, 138.2, 136.8, 130.3, 98.3, 52.8, 13.7 ppm; IR (film) 3119, 1636, 1530, 1305, 1106, 1058, 984 cm⁻¹; HRMS (EI): Exact mass calcd for C₁₀H₁₂N₂O₄ [M]+: 224.0797; Found: 224.0794. Elemental analysis calcd for C₁₀H₁₂N₂O₄: C, 53.57; H, 5.39; N, 12.49; Found: C, 53.68; H, 5.41; N, 12.41.

TBS-Protected Bisoxazole Dimethylacetal 24b.



To bisoxazole dimethylacetal 23 (649 mg, 2.89 mmol, 1 equiv) in dry THF (29 mL) at -78 °C was added n-BuLi (2.5 M in hexanes, 1.27 mL, 3.18 mmol, 1.1 equiv). The reaction mixture gradually took on a bright yellow color. After stirring at -78 °C for 30 min, TBSOTf (0.696 mL, 3.03 mmol, 1.05 equiv) was added dropwise. The reaction mixture was stirred at -78 °C and the color slowly faded from yellow to almost colorless. After 40 min, TLC indicated some unreacted SM. An additional amount of TBSOTf (0.100 mL, 0.435 mmol, 0.15 equiv) was added. After 20 minutes, TLC showed complete consumption of SM. The reaction was quenched by the addition of sat aq NaHCO₃ (30 mL) and the resulting suspension was warmed to rt and extracted with CH_2Cl_2 (3 × 30 mL). The combined organic layers were dried over a 1:1 mixture of K₂CO₃ and Na₂SO₄, filtered, and concentrated *in vacuo* to yield a colorless solid. Purification by silica gel chromatography ($15 \rightarrow 55\%$ Et₂O/hexanes; TLC: R_f= 0.74, 80% ether/ hexanes, UV) provided TBS-protected bisoxazole dimethylacetal 24b (860 mg, 88%) as a colorless, crystalline solid. Mp 89.2–90.8 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.70 (s, 1H), 5.47 (s, 1H), 3.38 (s, 6H), 2.54 (s, 3H), 0.95 (s, 9H), 0.38 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 165.1, 156.6, 154.8, 139.6, 136.5, 98.6, 52.7, 26.4, 17.6, 13.7, -5.9 ppm; IR (film) 2929, 1611, 1114, 1101, 1061 cm⁻¹; HRMS (EI): Exact mass calcd for C₁₅H₂₃N₂O₃Si [M – OCH₃]⁺: 307.1478; Found: 307.1492. Elemental analysis calcd for C₁₆H₂₆N₂O₄Si: C, 56.78; H, 7.74; N, 8.28; Found: C, 56.98; H, 7.67; N, 8.29.

Trienyl Alcohol 40.



(a) Oxidation. A 25 mL concentration flask containing alcohol 37 (92.9 mg, 0.602 mmol, 1 equiv) in dry CH₂Cl₂ (6 mL) was cooled to 0 °C. Dess-Martin periodinane (DMP) (0.332 g, 0.783 mmol, 1.3 equiv) was added, and the solution became a cloudy white. After 5 min, the solution was warmed to rt. After 1 h, a 1:1 mixture of sat aq Na₂S₂O₄ and sat aq Na_HCO₃ (12 mL) was added to the crude mixture which was then stirred vigorously for 15 min. The layers were separated, and the aqueous layer was extracted with CH₂Cl₂ (2 x 12 mL). The combined organics were washed with sat aq NaHCO₃ (5 mL), dried over Na₂SO₄, filtered, and concentrated *in vacuo* to give a light yellow oil with a white precipitate. The residue was suspended in 20% Et₂O/pentane (5 mL) and filtered through a small plug of Celite (additional 5 mL rinse). The solvent was removed in vacuo to aldehyde 38 (116 mg, with DMP contaminants) as a light yellow oil (TLC: $R_f = 0.82$, 15% EtOAc/hexanes, anisaldehyde stain) which was carried on without further purification. Due to the volatility of aldehyde 38, it should not be exposed to pressures lower than 5 mm Hg: ¹H NMR (500 MHz, CDCl₃) δ 9.57 (t, J = 2.4 Hz, 1H), 5.42-5.34 (m, 2H), 5.32 (d, J = 10.2 Hz, 1H), 3.12 (d, J = 2.4 Hz, 2H), 2.95 (ddq, J = 10.2, 9.1, 6.8 Hz, 1H), 1.76 (d, J = 1.3 Hz, 3H), 1.64 (d, J = 5.6 Hz, 3H), 1.03 (d, J = 6.8 Hz, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 199.5, 135.3, 134.8, 124.7, 123.3, 47.5, 35.8, 24.4, 21.4, 17.8 ppm; IR (film) 3022, 2966, 2927, 2870, 2717, 1726, 1450, 1379, 1238, 1170, 1120, 1048, 1017, 970, 857 cm⁻¹.

(b) Horner–Wadsworth-Emmons Olefination. To a 10 mL concentration flask was added NaH (48 mg of a 60% dispersion in oil, 1.2 mmol, 2.0 equiv). The white solid was washed in dry pentanes $(3 \times 1 \text{ mL})$ and suspended in dry THF (0.4 mL). To this suspension was added triethylphosphonoacetate (0.251 mL, 1.26 mmol, 2.1 equiv) dropwise *via* syringe, and the solid dissolved to give a light yellow solution. After 1 h, the solution was cooled to 0 °C, and unpurified aldehyde 38 (theoretical from above, 0.602 mmol, 1 equiv) was added via cannula in dry THF (0.5 mL), producing a bright yellow/orange solution. After 30 min, the solution was warmed to rt and after an additional 30 min, sat aq NH₄Cl (20 mL) was added along with Et₂O (20 mL), and the layers were separated. The aqueous layer was further extracted with Et₂O (2 \times 10 mL), and the combined organic layers were dried over Na₂SO₄, filtered, and concentrated *in vacuo* to yield ethyl ester **39** (270 mg, with phosphate contaminants) as a clear, yellow oil (TLC: $R_f =$ 0.80 in 20% EtOAc/hexanes, anisaldehyde stain) which was carried on without further purification. Due to the volatility of ester 39, it should not be exposed to pressures lower than 5 mm Hg: ¹H NMR (500 MHz, CDCl₃) δ 6.90 (dt, J = 15.6, 6.7 Hz, 1H), 5.82 (dt, J = 15.6, 1.4 Hz, 1H), 5.41-5.31 (m, 2H), 5.12 1.0 Hz, 3H), 1.63 (d, J = 5.0 Hz, 3H), 1.28 (t, J = 7.1 Hz, 3H), 1.01 (d, J = 6.8 Hz, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) & 166.6, 146.4, 135.7, 132.3, 129.7, 123.0, 122.0, 60.2, 35.6, 35.0, 23.5, 21.4, 17.9, 14.3 ppm; IR (film) 3022, 2966, 2927, 2870, 2717, 1725, 1450, 1379, 1048, 970, 857 cm⁻¹.

(c) Reduction. To a 50 mL concentration flask containing ethyl ester 39 as an unpurified mixture (theoretical from above, 0.602 mmol, 1 equiv) in dry CH₂Cl₂ (8.6 mL) at 0 °C was added DIBAL-H (1.99 mL of a 1.0 M solution in CH₂Cl₂, 1.99 mmol, 3.3 equiv) dropwise via syringe. Some gas evolution was evident. After 1 h, half-saturated Rochelle's salt (20 mL) and Et₂O (20 mL) was added and the mixture stirred vigorously overnight and allowed to warm to rt. The layers were separated and the aqueous layer was further extracted with Et_2O (3 × 25 mL). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated *in vacuo* to yield a clear, yellow oil. The product was purified *via* automated silica column chromatography ($0 \rightarrow 15\%$ EtOAc/hexanes, 10 g column; TLC: R_f = 0.28 in 20% EtOAc/hexanes, CAM stain) to provide alcohol 40 (82.7 g, 76% yield for three steps) as a clear, colorless oil. Due to the volatility of alcohol 40, it should not be exposed to pressures lower than 5 mm Hg for more than several hours. When left overnight at 0.5 mm Hg for characterization purposes, partial product loss was observed: ¹H NMR (500 MHz, CDCl₃) δ 5.70-5.60 (m, 2H), 5.42-5.33 (m, 2H), 5.04 (d, J = 9.4 Hz, 1H), 4.13-4.07 (m, 2H), 3.05-2.97 (m, 1H), 2.82-2.69 (m, 2H), 1.67 (d, J = 1.4 Hz, 3H),1.64 (d, J = 3.9 Hz, 3H), 1.33 (br s, 1H), 1.01 (d, J = 6.7 Hz, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 136.1, 131.6, 130.9, 130.5, 129.9, 122.7, 63.7, 35.4, 35.0, 23.4, 21.5, 17.9 ppm; IR (film) 3325, 2963, 2870, 1449, 1377, 1042, 968, 868 cm⁻¹; $[\alpha]_D^{24} = -80.6^\circ$ (c = 1.00, CHCl₃); HRMS (EI): Exact mass calcd for C₁₂H₂₀O [M]+: 180.1514; Found: 180.1514; Anal calcd for C₁₂H₂₀O: C, 79.94%, H, 11.18%; Found: C, 79.72%, H, 11.35%.

Side Chain Allylic Bromide Coupling Fragment 5.



To a 25 mL concentration flask containing alcohol 40 (81.8 mg, 0.454 mmol, 1 equiv) in dry THF (2 mL) was added Et₃N (0.253 mL, 1.82 mmol, 4 equiv). The solution was cooled to -40 °C and MsCl (0.105 mL, 1.36 mmol, 3 equiv) was added dropwise. A white precipitate formed immediately. After 50 min at 40 °C, the suspension was warmed to 0 °C. After 30 min, a solution of LiBr (0.394 g, 4.54 mmol, 10 equiv) in THF (4 mL) was added dropwise via cannula. After 30 min, the flask was warmed to rt and stirred for an additional 30 min. The reaction was quenched with ice-cold sat aq NaHCO₃ (15 mL) and diluted with pentane (5 mL). The layers were separated and the aqueous layer was further extracted with pentane (3×10 mL). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated *in vacuo* to yield a clear, yellow oil. The product was purified *via* automated silica column chromatography (pentane, 4 g column; TLC: $R_f = 0.89$ in 20% EtOAc/hexanes, anisaldehyde stain) to provide bromide 5 (95.8 g, 87% yield) as a clear, colorless oil. Due to the volatility of bromide 5, it should not be exposed to pressures lower than 5 mm Hg for extended periods: ¹H NMR (500 MHz, CDCl₃) δ 5.75-5.67 (m, 2H), 5.42-5.32 (m, 2H), 5.06 (d, J = 8.9 Hz, 1H), 4.00-3.92 (m, 2H), 3.03-2.95 (m, 1H), 2.82-2.74 (m, 1H), 2.82-2.74 (m, 2H), 5.06 (m, 2H), 2H), 1.66 (d, J = 1.3 Hz, 3H), 1.64 (d, J = 4.5 Hz, 3H), 1.01 (d, J = 6.8 Hz, 3H) ppm; ¹³C NMR (125) MHz, CDCl₃) δ 136.0, 133.8, 131.4, 131.0, 127.2, 122.8, 35.4, 34.9, 33.3, 23.5, 21.5, 18.0 ppm; IR (film) 2964, 2927, 2868, 156, 1438, 1377, 1203, 965, 854, 580 cm⁻¹; $[\alpha]_D^{24} = -69.3^{\circ}$ (*c* = 1.00, CHCl₃); HRMS (EI): Exact mass calcd for C₁₂H₁₉Br [M]+: 242.0670; Found: 242.0663.

Thiazolidinethione Alcohol 51.



(a) **Reduction.** To a 10 mL flask containing bisoxazole methyl ether **47b** (248 mg, 0.388 mmol, 1 equiv) in CH₂Cl₂ (1.5 mL) under N₂ at –78 °C was added DIBAL-H (0.581 mL of a 1.0 M solution in CH₂Cl₂, 1.5 equiv) dropwise *via* syringe. After stirring for 1 h, the reaction was quenched by the addition of MeOH (7.5 mL). Rochelle's salt (10 mL of a 7:5 mixture of saturated solution and water) was added and the biphasic mixture was warmed to rt and stirred rapidly for 45 min. The layers were separated and the aqueous phase was extracted with CH₂Cl₂ (3 × 25 mL). The combined organic layers were dried over anhydrous Na₂SO₄, filtered, and concentrated *in vacuo* to give a colorless oil. This material was quickly subjected to flash chromatography (CH₂Cl₂ until the auxiliary was eluted then 30% Et₂O/CH₂Cl₂; TLC: R_r = 0.09 in 70% Et₂O/hexanes) to provide (*R*)-3-(2-(5-(*tert*-butyldimethylsilyl)-2-methyloxazol-4-yl) oxazol-4-yl)-3-methoxypropanal (**51a**) as a relatively unsable clear oil that was used immediately in the next step: ¹H NMR (500 MHz, CDCl₃) δ 9.83 (s, 1H), 7.67 (s, 1H), 4.81 (d, *J* = 7.6, 4.7 Hz, 1H), 3.38 (s, 3H), 3.03 (dd, *J* = 16.9, 7.7 Hz, 1H), 2.94 (dd, *J* = 16.9, 4.6 Hz, 1H), 2.55 (s, 3H), 0.96 (s, 9H), 0.34 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 200.0, 165.1, 156.8, 154.8, 140.8, 138.7, 135.2, 71.3, 56.9, 48.0, 26.3, 17.4, 13.6, -6.0 ppm.

(b) Aldol Reaction. To a 10 mL flask was added Sn(OTf)₂ (0.353 g, 0.847 mmol, 2.3 equiv) in a glove box. This tin compound was dissolved in CH₂Cl₂ (3.5 mL), the flask was cooled to -50 °C, N-ethyl piperidine (0.121 mL, 0.884 mmol, 2.4 equiv) was added dropwise via syringe, followed by (S)-Nacetyl-4-iPr-thiazolidinethione (ent-44a, 0.122 g, 0.552 mmol, 1.5 equiv). The solution was stirred at -50 °C for 4.5 h to form the tin enolate and then was cooled to -110 °C in a liquid nitrogen/hexanes slurry before aldehyde 51a (theoretical from above, 0.388 mmol, 1 equiv) was added via cannula in CH₂Cl₂ (1.5 mL). The solution was kept between -90 and -100 °C for 2 h and was then warmed to -78 ° C over 30 min and allowed to stir for an additional 30 min at -78 °C. The reaction was quenched with pH 7 buffer and filtered with CH₂Cl₂ (25 mL) through a plug of Celite into a separatory funnel containing sat aq NaHCO₃ (15 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 × 30 mL). The combined organic phases were dried over anhydrous Na₂SO₄, filtered, and concentrated in vacuo to give a clear yellow oil. ¹H NMR integration of the unpurified product indicated a 3:1 diastereomer ratio. This material was purified via automated silica column chromatography $(40 \rightarrow 70\% \text{ EtOAc/hexanes}, 110 \text{ g column}; \text{ TLC: } R_f = 0.28 \text{ in } 70\% \text{ EtOAc/hexanes})$ to provide diastereomerically pure alcohol 51 (214.2 mg, 46% yield over two steps) as a clear, yellow oil: ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3) \delta$ 7.65 (s, 1H), 5.15 (dd, J = 7.0, 6.8 Hz, 1H), 4.53 (dd, J = 7.2, 6.3 Hz, 1H), 4.33-4.26 (m, 1H), 3.57-3.47 (m, 3H), 3.39-3.30 (m, 1H), 3.34 (s, 3H), 3.03 (d, J = 11.1 Hz, 1H), 2.54(s, 3H), 2.42-2.31 (m, 1H), 2.20-2.12 (m, 1H), 2.11-2.03 (m, 1H). 1.06 (d, J = 6.7 Hz, 3H), 0.97 (d, J = 7.3 Hz, 3H), 0.95 (s, 9H), 0.38 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) & 202.8, 172.2, 165.0, 156.7, 154.7, 141.3, 138.9, 135.3, 75.1, 71.3, 66.4, 56.6, 45.4, 40.9, 30.7, 30.5, 26.5, 19.0, 17.7, 17.5, 13.7, -5.9 ppm; IR (film) 3420, 2958, 2930, 2855, 1696, 1616, 1583, 1469, 1363, 1314, 1157, 1094, 1036, 930, 844, 781, 730, 668 cm⁻¹; $[\alpha]_D^{25} = +212.1^\circ$ (c = 1.03, CHCl₃); HRMS (CI): Exact mass calcd for C₂₅H₄₀N₃O₅S₂Si [M + H]⁺: 554.2179; Found: 554.2181.

Glycal 54.



(a) Luche Reduction. To a 10 mL flask containing dihydropyranone **53** (218.7 mg, 0.5056 mmol, 1 equiv) under an Ar atmosphere was added a solution of CeCl₃•7H₂O (207 mg, 0.556 mmol, 1.1 equiv) in a 1:1 mixture of THF/MeOH (7.2 mL). The suspension was cooled to -78 °C and NaBH₄ (38.3 mg, 1.01 mmol, 2 equiv) was added as a solid in one portion. After 15 min, the flask was allowed to warm to -40 °C over 1 h.After an additional 45 min at -40 °C, the flask was recooled to -78 °C and the reaction mixture was poured into pH 7 buffer (150 mL) and was diluted with Et₂O (50 mL). The layers were separated and the aqueous phase was extracted with EtOAc (4 x 50 mL) The combined organic layers were filtered through a plug of silica gel and concentrated *in vacuo* to give the corresponding allylic alcohol (**53a**) which was carried on immediately without any further purification: TLC: $R_f = 0.36$ in 80% EtOAc/hexanes.

(b) TBS Protection. To a 5 mL flask containing allylic alcohol 53a (theoretical from above, 0.5056 mmol, 1 equiv) under an Ar atmosphere was added imidazole (103 mg, 1.52 mmol, 3 equiv) a catalytic amount of DMAP, and anhydrous DMF (2.0 mL). TBS-Cl (167.6 mg, 1.11 mmol, 2.2 equiv) was added and the reaction was stirred for 4 h. The solution was diluted with brine (50 mL) and EtOAc (50 mL), the layers were separated, and the aqueous layer was extracted with EtOAc (3×50 mL). The combined organics were washed with water $(2 \times 25 \text{ mL})$, dried over anhydrous Na₂SO₄, filtered, and concentrated *in vacuo*. This material was purified *via* flash chromatography $(5 \rightarrow 9\% \text{ Et}_2\text{O}/\text{CH}_2\text{Cl}_2, \text{TLC}: \text{R}_f = 0.31$ in 10% Et₂O/CH₂Cl₂) to provide allylic TBS ether 54 (177.0 mg, 61% yield over two steps) as a clear colorless oil: ¹H NMR (500 MHz, CDCl₃) δ 7.63 (s, 1H), 4.46 (dd, J = 8.0, 6.0 Hz, 1H), 4.42 (s, 1H), 4.35 (dd, J = 8.3, 7.0 Hz, 1H), 3.86-3.78 (m, 1H), 3.29 (s, 3H), 2.54 (s, 3H), 2.26 (ddd, J = 14.3, 8.7, 5.9 Hz, 1H), 2.15 (ddd, J = 12.7, 8.2, 4.3 Hz, 1H), 1.95 (dd, J = 13.0, 6.6 Hz, 1H), 1.69 (s, 3H), 1.68-1.59 (m, 1H), 0.95 (s, 9H), 0.88 (s, 9H), 0.39 (s, 3H), 0.38 (s, 3H), 0.06 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) & 165.1, 156.6, 154.7, 152.0, 141.2, 139.1, 135.9, 101.3, 72.6, 71.5, 64.3, 56.3, 39.5, 37.6, 26.5, 25.9, 19.8, 18.2, 17.6, 13.8, -4.7, -5.9 ppm; IR (film) 2928, 2857, 1675, 1586, 1463, 1383, 1252, 1062, 837, 837, 780, 495 cm⁻¹; $[\alpha]_D^{25} = +18.5^\circ$ (c = 1.00, CHCl₃); HRMS (CI): Exact mass calcd for C₂₈H₄₈N₂O₅Si₂Li [M + Li]⁺: 555.3262; Found: 555.3267.

Preparation of Pyran/Bisoxazole Coupling Fragment 4a (R = TBS).



To a 5 mL flask containing TBS ether 54 (170.0 mg, 0.3097 mmol, 1 equiv) in benzene (10 mL) under an argon atmosphere was added MeOH (0.376 mL, 30 equiv) and trimethyl orthoformate (0.170 mL, 5 equiv). A spatula tip of PPTS was added in one portion and the reaction was stirred for 90 min. The reaction was guenched with sat aq NaHCO₃ (50 mL) and diluted with EtOAc (40 mL). The layers were separated and the aqueous phase was further extracted with EtOAc (3×20 mL). The combined organics were dried over anhydrous Na₂SO₄, filtered, and concentrated *in vacuo*. This material was purified *via* flash chromatography (0 \rightarrow 40% EtOAc/hexanes, TLC: $R_f = 0.57$ in 40% EtOAc/hexanes) to provide mixed methyl acetal 4a (R = TBS, 144.4 mg, 80% yield) as a clear colorless oil: ¹H NMR (500 MHz, CDCl₃) δ 7.62 (s, 1H), 4.45 (dd, J = 7.7, 6.4 Hz, 1H), 4.02-3.93 (m, 1H), 3.51-3.42 (m, 1H), 3.29 (s, 3H), 3.00 (s, 3H), 2.54 (s, 3H), 2.18-2.06 (m, 2H), 1.95 (dd, J = 12.8, 4.5 Hz, 1H), 1.79 (d, J = 10.3 Hz, 1H), 1.31 (dd, J = 23.7, 11.6 Hz, 1H), 1.30 (s, 3H), 1.22 (dd, J = 23.6, 11.8 Hz, 1H), 0.95 (s, 9H), 0.87 (s, 9H), 0.39 (s, 3H), 0.38 (s, 3H), 0.04 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 165.1, 156.6, 154.8, 141.4, 139.1, 135.9, 99.6, 72.7, 65.8, 65.1, 56.3, 47.6, 45.3, 41.0, 40.3, 26.5, 25.8, 23.7, 18.0, 17.6, 13.8, -4.6, -5.9 ppm; IR (film) 2929, 2857, 1612, 1586, 1463, 1377, 1319, 1251, 1192, 1084, 1034, 930, 914, 870, 837, 779, 670, 580, 490 cm⁻¹; $[\alpha]_D^{25} = -20.1^\circ$ (c = 1.01, CHCl₃); HRMS (CI): Exact mass calcd for C₂₉H₅₂N₂O₆Si₂Li [M + Li]+: 587.3479; Found: 587.3523.

Methylated Bisoxazole Dimethylacetal 24a



To bisoxazole dimethylacetal **23** (78.8 mg, 0.351 mmol, 1 equiv) in dry THF (2.0 mL) at -78 °C was added *n*-BuLi (2.72 M in hexanes, 129 µL, 0.351 mmol, 1 equiv). The reaction mixture gradually took on a bright yellow color. After stirring at -78 °C for 30 min, MeI (24 µL, 0.386 mmol, 1.1 equiv) was added dropwise. The reaction mixture was stirred at -78 °C for 30 min and the color slowly faded from yellow to colorless. The reaction was quenched by the addition of sat. aq NaHCO₃ (2.5 mL) and the resulting suspension was warmed to rt and extracted with EtOAc. The combined organic layers were dried over a 1:1 mixture of K₂CO₃ and Na₂SO₄, filtered. and concentrated *in vacuo* to yield a colorless solid. Purification by silica gel chromatography (0 \rightarrow 40% EtOAc/hexanes; TLC: R_f= 0.60, 100% ether UV) provided methylated bisoxazole dimethylacetal **24a** (65.5 mg, 78%) as a colorless, crystalline solid. Mp 97.0–98.5 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.71 (s, 1H), 5.47 (s, 1H), 3.39 (s, 6H), 2.64 (s, 3H), 2.47 (s, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 160.1, 156.5, 149.6, 139.3, 136.2, 124.9, 98.5, 52.8, 13.6, 11.5 ppm; IR (film) 2937, 1593, 1197, 1097, 1055, 980 cm⁻¹; HRMS (EI): Exact mass calcd for C₁₁H₁₄N₂O₄ [M]⁺: 238.0954; Found: 238.0950; Anal calcd for C₁₁H₁₄N₂O₄: C, 55.46%, H, 5.92, N, 11.76%; Found: C, 55.57%, N, 11.66%.

Allylated Bisoxazole Dimethylacetal 27



To as solution of diethylamine (22.0 µL, 0.213 mmol, 1.5 equiv) in THF (1 mL) at -78 °C was added n-BuLi (133 µL of a 1.5 M solution in hexanes, 0.199 mmol, 1.4 equiv). After stirring for 5 min, warming to 0 °C for 10 min, and then recooling to -78 °C, this solution was added via cannula to protected bisoxazole dimethyl acetal 24b (48.0 mg, 0.142 mmol, 1 equiv) in dry THF (1 mL) at -78 °C. The reaction mixture immediately took on a bright red color and was stirred at -78 °C for 30 min. Allyl iodide (14.3 µL, 0.156 mmol, 1.1 equiv) was added dropwise, which caused a color change to light orange. After stirring at -78 °C for 15 min, sat aq NaHCO₃ (2 mL) was added and the reaction was warmed to rt. The resulting suspension was partitioned between CH₂Cl₂ (10 mL) and sat aq NaHCO₃ (10 ml) and The aqueous layer was extracted with CH_2Cl_2 (2 × 10 mL). The combined organic layers were dried over a 1:1 mixture of K_2CO_3 and Na_2SO_4 , filtered, and concentrated *in vacuo* to yield a colorless liquid. Purification by silica gel chromatography (20% Et₂O/hexanes; TLC: R_f= 0.77, 70% Et₂O/hexanes UV) provided allylated bisoxazole 27 (47.0 mg, 88%) as a clear colorless oil. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3) \delta 7.70 \text{ (s, 1H)}, 5.86 \text{ (ddt}, J = 17.1, 10.2, 6.5 \text{ Hz}, 1\text{H}), 5.47 \text{ (s, 1H)}, 5.08 \text{ (dd}, J = 17.1, 10.2$ 1.6 Hz, 1H), 5.01 (dd, J = 10.2, 1.5 Hz, 1H), 3.37 (s, 6H), 2.96 (t, J = 7.9 Hz, 2H), 2.57 (dt, J = 7.9, 6.7 Hz, 2H), 0.94 (s, 9H), 0.38 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 167.9, 156.6, 154.8, 139.6, 138.9, 136.5, 136.3, 115.9, 98.6, 52.8, 30.9, 27.5, 26.4, 17.6, -5.9 ppm; IR (film) 2931, 2858, 1612, 1580, 1470, 1251, 1391, 1193, 1103, 1062 cm⁻¹; HRMS (EI): Exact mass calcd for C₁₈H₂₇N₂O₃Si [M-OCH₃]⁺: 347.1791; Found: 347.1795; Anal calcd for C₁₉H₃₀N₂O₄Si: C, 60.29%, H, 7.99%, N, 7.40%; Found: C, 60.59%, H, 8.02%, N, 7.37%.

Deprotected Allylated Bisoxazole 28



To alylated TBS-protected bisoxazle **27** (16 mg, 0.042 mmol, 1 equiv) in dry THF (425 µL) at 0 °C was added tetrabutylammoium fluoride (TBAF) (51 µL of a 1.0 M solution in THF, 0.051 mmol, 1.2 equiv). After stirring at 0 °C for 10 min, the reaction was quenched by the addition of sat aq NaHCO₃ (5 mL) and CH₂Cl₂ (1 mL). The mixture was partitioned between CH₂Cl₂ (5 mL) and sat aq NaHCO₃ (5 mL) and the aqueous layer was extracted with CH₂Cl₂ (2 × 5 mL) The combined organic layers were dried over a 1:1 mixture of K₂CO₃ and Na₂SO₄, filtered, and concentrated *in vacuo* to yield a colorless liquid. Purification by silica gel chromatography (1:1:1 CH₂Cl₂/Et₂O/hexanes; TLC: R_f= 0.53, 70% Et₂O/hexanes, UV) provided deptotected allylated bisoxazole **28** (10.0 mg, 90%) as a clear colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 8.18 (s, 1H), 7.72 (s, 1H), 5.86 (ddt, *J* = 17.1, 10.2, 6.5 Hz, 1H), 5.48 (s, 1H), 5.09 (dd, *J* = 17.1, 1.6 Hz, 1H), 5.03 (dd, *J* = 10.2, 1.4 Hz, 1H), 3.39 (s, 6H), 2.95 (t, *J* = 7.4 Hz, 2H), 2.58 (dt, *J* = 7.4, 6.7 Hz, 2H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 165.4, 155.6, 139.5, 138.3, 136.8, 136.1, 130.3, 116.1, 98.3, 52.8, 30.7, 27.5 ppm; IR (film) 2939, 1103, 1059, 984, 916 cm⁻¹; HRMS (EI): Exact mass calcd for C₁₃H₁₆N₂O₄ [M]⁺: 264.1110; Found: 264.1114; Anal calcd for C₁₃H₁₆N₂O₄: C, 59.08%, H, 6.10%, N, 10.60%; Found: C, 58.94%, H, 6.16%, N, 10.46%.

Prenylated Bisoxazole 29



A 0.5 M solution of LiNEt₂ was prepared by addition of *n*-BuLi (1.00 mL of a 2.72 M solution in hexanes, 2.72 mmol) to diethylamine (0.310 mL, 3.00 mmol) in THF (4.13 mL) at -78 °C under an atmosphere of argon. After 5 min, the flask was warmed to 0 °C. In a separate flask, TBS-protected bisoxazole dimethylacetal 24b (100.8 mg, 0.298 mmol, 1 equiv) in THF (2.0 mL) was cooled to -78 °C under an atmosphere of argon and the LiNEt₂ solution prepared above was added dropwise until a yellow color persisted (to remove any adventitious acid source-about 4 drops). After this zero point, LiNEt₂ (0.893 mL of a 0.5 M solution in THF, 0.447 mmol, 1.5 equiv) was added dropwise via gastight syringe. The reaction took on an bright orange/red color. After stirring at -78 °C for 30 min, prenyl bromide (39 µL, 0.327 mmol, 1.1 equiv) was added dropwise causing the reaction to fade to light orange-yellow almost immediately. After 15 min, the reaction was quenched with sat aq NaHCO₃ (5 mL) and the color turned a very light yellow. The mixture was diluted with EtOAc (5 mL) and warmed to rt. The layers were separated and the aqueous phase was further extracted with EtOAc (3×5 mL). The combined organics were dried over a 1:1 mixture of anhydrous K₂CO₃ and Na₂SO₄, filtered, and concentrated *in vacuo*. This material was purified *via* automated silica column chromatography $(0 \rightarrow 15\%)$ EtOAc/hexanes, TLC: $R_f = 0.81$ in 70% Et₂O/hexanes) to provide prenylated bisoazole **29** (103.1 mg, 91% yield) as a clear colorless oil ¹H NMR (500 MHz, CDCl₃) δ 7.70 (d, J = 1.0 Hz, 1H) 5.47 (d, J = 0.8 Hz, 1H), 5.15 (tt, J = 7.2, 1.3 Hz, 1H), 3.37 (s, 6H), 2.87 (t, J = 7.4 Hz, 2H), 2.48 (dt, J = 7.6, 7.4 Hz, 2H), 1.67 (s, 3H), 1.60 (s, 3H), 0.95, (s, 9H), 0.38 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 168.3, 156.7, 154.6, 139.6, 138.9, 136.5, 133.4, 122.2, 98.6, 52.8, 28.3, 26.4, 25.7, 25.6, 17.62, 17.58, -5.9 ppm; IR (film) 2954, 2930, 1469, 1251, 1104, 1062, 843, 124 cm⁻¹; HRMS (EI): Exact mass calcd for C₂₁H₃₄N₂O₄Si [M]⁺: 406.2288; Found: 406.2273; Anal calcd for C₂₁H₃₄N₂O₄Si: C, 62.03%, H, 8.43%, N, 6.89%; Found: C, 62.17%, H, 8.56%, N, 6.94%.

Deprotected Prenylated Bisoxazole 30



To prenylated TBS-protected bisoxazole **29** (44.9 mg, 0.110 mmol, 1 equiv) in dry THF (2.0 mL) at 0 °C was added tetrabutylammoium fluoride (TBAF) (132 μ L of a 1.0 M solution in THF, 0.133 mmol, 1.2 equiv). After stirring at 0 °C for 10 min, the reaction was quenched by the addition of sat aq NaHCO₃ (2 mL) and CH₂Cl₂ (2 mL). The mixture was partitioned between CH₂Cl₂ (10 mL) and sat aq NaHCO₃ (10 mL) and the aqueous layer was extracted with CH₂Cl₂ (2 × 10 mL) The combined organic layers were dried over a 1:1 mixture of K₂CO₃ and Na₂SO₄, filtered, and concentrated *in vacuo* to yield a colorless liquid. Purification by silica gel chromatography (1:1:2 CH₂Cl₂/Et₂O/hexanes; TLC: R_f= 0.66, 70% Et₂O/hexanes, UV) provided deprotected prenylated bisoxazole **30** (31.8 mg, 99%) as a clear colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 8.18 (s, 1H), 7.72 (d, *J* = 1.0 Hz, 1H), 5.48 (s, 1H), 5.14 (tt, *J* =7.2, 1.3 Hz, 1H), 3.39 (s, 6H), 2.86 (t, *J* = 7.4 Hz, 2H), 2.50 (dt, *J* = 7.7, 7.4 Hz, 2H), 1.68 (s, 3H), 1.60 (s, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 165.8, 155.6, 139.5, 138.2, 136.8, 133.7, 130.2, 121.8, 98.3,

52.8, 28.3, 25.6, 25.5, 17.6 ppm; IR (film) 2933, 1103, 1059, 984 cm⁻¹; HRMS (EI): Exact mass calcd for $C_{15}H_{20}N_2O_4$ [M]⁺: 292.1423; Found: 292.1392; Anal calcd for $C_{15}H_{20}N_2O_4$: C, 61.63%, H, 6.90%, N, 9.58%; Found: C, 61.38%, H, 7.06%, N, 9.36%.

Trienyl Bisozaxole 41



To as solution of diethylamine (59 µL, 0.568 mmol, 2.1 equiv) in THF (1.5 mL) at -78 °C was added n-BuLi (340 µL of a 1.59 M solution in hexanes, 0.541 mmol, 2.0 equiv). After stirring for 5 min, warming to 0 °C for 10 min, and then recooling to -78 °C, this solution was added via cannula to protected bisoxazole dimethyl acetal 24b (93.1 mg, 0.271 mmol, 1 equiv) in dry THF (1.5 mL) at -78 ° C. The reaction mixture immediately took on a bright red color and was stirred at -78 °C for 30 min. Side chain allylic bromide 7 (106 mg, 0.436 mmol, 1.5 equiv) in THF (1 mL) was added dropwise via cannula, which caused a color change to light orange. After stirring at -78 °C for 15 min, sat aq NaHCO₃ (4 mL) was added and the reaction was warmed to rt. The resulting suspension was partitioned between CH_2Cl_2 (15 mL) and sat aq NaHCO₃ (15 ml) and The aqueous layer was extracted with CH_2Cl_2 $(2 \times 15 \text{ mL})$. The combined organic layers were dried over Na₂SO₄, filtered, and concentrated *in vacuo* to yield a clear yellow liquid. Purification by silica gel chromatography (25% Et₂O/hexanes; TLC: $R_{f=}$ 0.81, 70% Et₂O/hexanes UV) provided trienyl bisoxazole **41** (106.7 mg, 79%) as a clear colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 7.70 (d, J = 0.8 Hz, 1H), 5.50-5.39 (m, 2H), 5.47 (s, 1H), 5.35-5.33 (m, 2H), 4.99 (d, J = 9.2 Hz, 1H), 3.37 (s, 6H), 3.03-2.95 (m, 1H), 2.91 (t, J = 7.4 Hz, 2H), 2.71 (dd, J = 14.4, 6.2 Hz, 1H), 2.65 (dd, J = 14.4, 6.2 Hz, 1H), 2.51 (dt, J = 7.4, 7.4 Hz, 2H), 1.63 (d, J = 4.7 Hz, 3H), 1.60 (d, J = 1.2 Hz, 3H), 0.99 (d, J = 6.7 Hz, 3H), 0.95 (s, 9H), 0.38 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 166.0, 156.6, 154.7, 139.6, 138.9, 136.5, 136.2, 132.1, 130.4, 129.5, 128.8, 122.5, 98.6, 52.8, 35.3, 35.2, 29.9, 28.2, 26.5, 23.3, 21.4, 17.9, 17.6, -5.9 ppm; IR (film) 2957, 2930, 2858, 1612, 1579, 1464, 1448, 1376, 1318, 1251, 1193, 1161, 1104, 1063, 1005, 970, 929, 910, 843, 824, 812, 782, 748, 688, 582, 456 cm^{-1} ; HRMS (EI): Exact mass calcd for $C_{28}H_{44}N_2O_4Si$ [M]⁺: 500.3070; Found: 500.3080; Anal calcd for C₂₈H₄₄N₂O₄Si: C, 67.16%, H, 8.86%, N, 5.59%; Found: C, 67.43%, H, 8.94%, N, 5.40%.

Deprotected Trienyl Bisoxazole 42



To TBS-protected trienyl bisoxazole **41** (54.4 mg, 0.109 mmol, 1 equiv) in dry THF (2.2 mL) at 0 °C was added tetrabutylammoium fluoride (TBAF) (130 μ L of a 1.0 M solution in THF, 0.130 mmol, 1.2 equiv). After stirring at 0 °C for 10 min, the reaction was quenched by the addition of sat aq NaHCO₃ (2 mL). The mixture was partitioned between CH₂Cl₂ (10 mL) and sat aq NaHCO₃ (10 mL) and the aqueous layer was extracted with CH₂Cl₂ (2 × 10 mL) The combined organic layers were dried over Na₂SO₄, filtered, and concentrated *in vacuo* to yield an orange liquid. Purification by silica gel

chromatography (1:1:2 CH₂Cl₂/Et₂O/hexanes; TLC: R_f = 0.66, 70% Et₂O/hexanes, UV) provided deprotected trienyl bisoxazole **42** (39.8 mg, 95%) as a clear colorless oil. ¹H NMR (500 MHz, CDCl₃) δ ppm; ¹³C NMR (125 MHz, CDCl₃) δ 8.16 (s, 1H), 7.71 (d, *J* = 1.0 Hz, 1H), 5.49-5.41 (m, 2H), 5.47 (s, 1H), 5.37-5.30 (m, 2H), 4.98 (d, *J* = 9.2 Hz, 1H), 3.40 (s, 3H), 3.02-2.93 (m, 1H), 2.89 (t, *J* = 7.4, Hz, 2H), 2.70 (dd, *J* = 14.6, 6.2 Hz, 1H), 2.65 (dd, *J* = 14.6, 6.2 Hz, 1H), 2.51 (dt, *J* = 7.4, 7.4 Hz, 2H), 1.62 (d, *J* = 4.9 Hz, 3H), 1.60 (d, *J* = 1.2 Hz, 3H), 0.97 (d, *J* = 6.9 Hz, 3H) ppm; IR (film) 3135, 2962, 2832, 2730, 1635, 1579, 1829, 1501, 1448, 1377, 1310, 1266, 1193, 1158, 1102, 1061, 969, 917, 854, 781, 734, 653 cm⁻¹; HRMS (EI): Exact mass calcd for C₂₂H₃₀N₂O₄ [M]+: 386.2206; Found: 386.2190; Anal calcd for C₂₂H₃₀N₂O₄: C, 68.37%, H, 7.82%, N, 7.25%; Found: C, 68.52%, H, 8.04%, N, 7.24%.

General procedure for TiCl/Hünig's base aldol-type reactions with acetals

To the *N*-acetyl-thiazolidinethione (1.3 equiv) in CH₂Cl₂ (0.2 M) under argon atmosphere at 0 °C was added TiCl₄ (1.3 equiv) dropwise *via* syringe. After 10 min, the orange solution was cooled to -78 °C and (*i*-Pr)₂NEt (1.3 equiv) was added *via* syringe and the solution turned a characteristic blood-red color. After stirring for 30 min, the solution was warmed to -50 °C and stirred for 2 h. The dimethyl acetal (1 equiv) was added dropwise *via* cannula in CH₂Cl₂ (approx 0.3 M) followed by the dropwise addition of BF₃•OEt₂ (1.3 equiv). The solution was stirred at -78 °C for 2 h. The reaction was quenched by pouring into a rapidly stirring 1:1 mixture of CH₂Cl₂ and half-sat NH₄Cl. The layers were separated and the aqueous layer was re-extracted with CH₂Cl₂ and the combined organic layers were dried over anhydrous Na₂SO₄, filtered, and concentrated *in vacuo*. ¹H NMR integration of the unpurified product was used to determine the diastereomer ratio. Silica column chromatography provided diastereomerically pure methyl ether.

General procedure for PhBCl₂/sparteine aldol-type reactions with acetals

To the *N*-acetyl-thiazolidinethione (1.3 equiv) in CH₂Cl₂ (0.2 M) under argon atmosphere at 0 °C was added PhBCl₂ (1.3 equiv) dropwise *via* syringe. After 10 min, sparteine (2.6 equiv) was added *via* syringe and the resulting solution was warmed to rt and stirred for 30 min. The solution was cooled to – 78 °C and the dimethyl acetal (1 equiv) was added dropwise *via* cannula in CH₂Cl₂ (approx 0.3 M) followed by the dropwise addition of BF₃•OEt₂ (1.1 equiv). The solution was stirred at –78 °C for 1.5 h and then was warmed to rt over 2 h. After stirring at rt an additional 1 h, the reaction was quenched with sat aq NH₄Cl. The contents of the flask were transferred to a separatory funnel containing a 4:1 mixture of hexanes:CH₂Cl₂. The layers were separated and the aqueous layer was re-extracted with CH₂Cl₂ and the combined organic layers were washed with brine, dried over anhydrous Na₂SO₄, filtered, and concentrated *in vacuo*. ¹H NMR integration of the unpurified product was used to determine the diastereomer ratio. Silica column chromatography provided diastereomerically pure methyl ether.



Yellow crystals. Mp 92.3–92.7 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.37-7.33 (m, 4H), 7.32-7.26 (m, 1H), 5.00 (ddd, *J* = 7.7, 6.0, 1.5 Hz, 1H), 4.71 (dd, *J* = 7.6, 5.7 Hz, 1H), 3.86 (dd, *J* = 16.6, 7.6 Hz, 1H), 3.69 (dd, *J* = 16.6, 5.7 Hz, 1H), 3.30 (dd, *J* = 11.4, 8.1 Hz, 1H), 3.21 (s, 3H), 2.96 (dd, *J* = 11.4, 1.5 Hz, 1H), 2.40-2.31 (m, 1H), 1.04 (d, *J* = 6.9 Hz, 3H), 0.97 (d, *J* = 7.1, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 202.7, 171.0, 140.4, 128.4, 127.9, 126.9, 79.9, 71.7, 56.5, 46.1, 30.5, 30.2, 19.0, 17.4 ppm; IR (film) 2964, 2933, 2891, 2821, 1698, 1467, 1364, 1306, 1255, 1207, 1156, 1096, 1038, 1006, 987, 895, 701 cm⁻¹; [α]_D²⁴ = -214.2° (*c* = 1.00, CHCl₃); HRMS (EI): Exact mass calcd for C₁₆H₂₁NO₂S₂ [M]+: 323.1014; Found: 323.0997.

Methyl Ether 46a



Clear yellow oil: ¹H NMR (500 MHz, CDCl₃) δ 7.39-7.33 (m, 4H), 7.33-7.28 (m, 1H), 5.16 (dd, *J* = 6.9, 6.9 Hz, 1H), 4.81 (dd, *J* = 9.7, 3.0 Hz, 1H), 3.76 (dd, *J* = 17.1, 9.7 Hz, 1H), 3.53 (dd, *J* = 11.4, 8.1 Hz, 1H), 3.46 (dd, *J* = 17.1, 3.1 Hz, 1H), 3.22 (s, 3H), 3.03 (d, *J* = 11.4, 1H), 2.41-2.25 (m, 1H), 1.04 (d, *J* = 6.7 Hz, 3H), 0.96 (d, *J* = 6.9, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 202.8, 171.0, 140.5, 128.5, 128.0, 126.7, 79.5, 71.7, 56.7, 46.6, 30.8, 30.7, 19.0, 17.7 ppm; IR (film) 3029, 2961, 2930, 2822, 1678, 1455, 1365, 1313, 1275, 1161, 1096, 1039, 762, 701 cm⁻¹; [α]_D²⁴ = -266.0° (*c* = 1.34, CHCl₃); HRMS (EI): Exact mass calcd for C₁₆H₂₁NO₂S₂ [M]⁺: 323.1014; Found: 323.1053.

Methyl Ether 45b



Clear yellow oil: ¹H NMR (500 MHz, CDCl₃) δ 7.37-7.33 (m, 4H), 7.32-7.27 (m, 1H), 5.19 (dd, J = 8.1, 1.0 Hz, 1H), 4.72 (dd, J = 7.8, 5.5 Hz, 1H), 3.90 (dd, J = 16.7, 7.8 Hz, 1H), 3.65 (dd, J = 16.7, 5.5 Hz, 1H), 3.35 (dd, J = 11.4, 1.1 Hz, 1H), 3.30 (dd, J = 11.4, 8.2 Hz, 1H), 3.21 (s, 3H), 1.30 (s, 3H), 1.29 (s, 3H), 0.95 (t, J = 7.9 Hz, 6H), 0.61 (q, J = 7.9 Hz, 4H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 205.3, 170.5, 140.6, 128.4, 127.9, 127.0, 79.9 76.7, 72.6, 56.6, 45.8, 30.1, 28.1, 26.1, 7.0, 6.5 ppm; IR (film) 2955, 2937, 2876, 2823, 1736, 1700, 1456, 1369, 1312, 1272, 1242, 1169, 1150, 1102, 1036, 745, 724,

701 cm⁻¹; $[\alpha]_D^{24} = -74.5^\circ$ (*c* = 1.00, CHCl₃); HRMS (EI): Exact mass calcd for C₂₂H₃₅NO₃S₂Si [M]+: 453.1828; Found: 453.1829.

Methyl Ether 46b



Clear yellow oil: ¹H NMR (500 MHz, CDCl₃) δ 7.38-7.33 (m, 4H), 7.33-7.28 (m, 1H), 5.30 (d, *J* = 8.1, 1H), 4.82 (dd, *J* = 9.5, 3.3 Hz, 1H), 3.75 (dd, *J* = 17.0, 9.6 Hz, 1H), 3.49 (dd, *J* = 16.8, 3.6 Hz, 1H), 3.48 (dd, *J* = 8.1, 3.3 Hz, 1H), 3.42 (d, *J* = 11.3 Hz, 1H), 3.21 (s, 3H), 1.28 (s, 3H), 1.24 (s, 3H), 0.94 (t, *J* = 7.9 Hz, 6H), 0.60 (q, *J* = 7.9 Hz, 4H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 204.5, 170.3, 140.5, 128.5, 128.0, 126.8, 79.6 76.7, 72.6, 56.6, 46.0, 30.3, 28.1, 26.2, 7.1, 6.6 ppm; IR (film) 2955, 2910, 2876, 2822, 1736, 1700, 1456, 1368, 1312, 1242, 1148, 1102, 1036, 971, 934, 745, 724, 701 cm⁻¹; [α]_D²⁴ = -68.2° (*c* = 0.84, CHCl₃); HRMS (EI): Exact mass calcd for C₂₂H₃₅NO₃S₂Si [M]⁺: 453.1828; Found: 453.1855.

Methyl Ether 47a



Clear yellow oil: ¹H NMR (500 MHz, CDCl₃) δ 7.67 (s, 1H), 5.11 (dd, *J* = 7.4, 6.6 Hz, 1H), 4.84 (dd, *J* = 6.6, 6.0 Hz, 1H), 4.02 (dd, *J* = 17.8, 7.1 Hz, 1H), 3.80 (dd, *J* = 17.8, 5.5 Hz, 1H), 3.49 (dd, *J* = 11.4, 8.1 Hz, 1H), 3.37 (s, 3H), 3.02 (d, *J* = 11.4 Hz, 1H), 2.54 (s, 3H), 2.42-2.34 (m, 1H), 1.06 (d, *J* = 6.7 Hz, 3H), 0.98 (d, *J* = 6.9 Hz, 3H), 0.95 (s, 9H), 0.39 (s, 3H), 0.38 (s, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 202.3, 170.9, 164.9, 156.5, 154.7, 140.8, 138.8, 135.4, 72.2, 71.4, 56.8, 43.5, 30.6, 30.1, 26.3, 18.9, 17.5, 17.4, 13.7, -5.99, -6.01 ppm; IR (film) 2957, 2930, 2895, 2857, 1698, 1613, 1585, 1469, 1364, 1314, 1252, 1163, 1103, 1039, 930, 842, 782 cm⁻¹; $[\alpha]_D^{24} = -238.7^{\circ}$ (*c* = 1.00, CHCl₃); HRMS (EI): Exact mass calcd for C_{23H35}N₃O₄S₂Si [M]⁺: 509.1838; Found: 509.1840.

Methyl Ether 48a



Clear yellow oil: ¹H NMR (500 MHz, CDCl₃) δ 7.65 (s, 1H), 5.17 (dd, *J* = 6.9, 6.9 Hz, 1H), 4.89 (dd, *J* = 9.4, 2.9 Hz, 1H), 3.87 (dd, *J* = 17.9, 9.5 Hz, 1H), 3.65 (dd, *J* = 17.9, 2.9 Hz, 1H), 3.56 (dd, *J* = 11.4, 7.9 Hz, 1H), 3.38 (s, 3H), 3.04 (d, *J* = 11.7 Hz, 1H), 2.54 (s, 3H), 2.42-2.35 (m, 1H), 1.07 (d, *J* = 6.9 Hz, 3H), 0.98 (d, *J* = 7.1 Hz, 3H), 0.96 (s, 9H), 0.40 (s, 3H), 0.39 (s, 3H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 202.5, 170.7, 165.0, 156.7, 154.9, 141.0, 138.9, 135.1, 72.2, 71.5, 57.3 43.9, 30.8, 30.7, 26.5, 19.0, 17.8, 17.6, 13.8, -5.8 ppm; IR (film) 2956, 2929, 2857, 1698, 1466, 1364, 1315, 1250, 1165, 1104, 1040, 930, 841, 823, 811, 781, 680 cm⁻¹; [α]_D²⁴ = -111.3° (*c* = 1.02, CHCl₃); HRMS (EI): Exact mass calcd for C₂₃H₃₅N₃O₄S₂Si [M]⁺: 509.1838; Found: 509.1847.

0

0=





Ò

0=









-0

 \sim

-m

4

ഹ

9–0

- ~

bpm 8

	-0
	10_1
	50_ 50_
station of the second	02
	-40
	20
And a second	_09
	- -02
	-08
	_06
	-100
	110
	120
	130
	-140
	150
	160
	-02
\$22	-





-0

-∼

-ო

4

–ഗ

-0

- 1

-00













S26

-0

-4

-09

-02

-06











-0

-0

20

30

40

20

-09

-02

-8

-06

100

110

120

130

140

150



















C

TBS

OMe

0

က=



TBS














C

Z

ò

Ne

TBS

OMe

OTBS







ŌTBS





















¹HNMR: 500 MHz in acetone-d₆



-0

-~

-ო

4

ഹ

-0

- ~

bpm 8







-0

 \sim

-ო

4

–ഗ

-0

- ~

-00

	-0
	10
	20
	30
	-4
	50
	-09
	_02
	80-
	-06
	100
	110
	120
A series of the	130
	140
	150
	160
	170





-∾

- ന

4

-ഗ

-0

- ~

-00





0

TBS

MeO OMe



0

TBS

MeO OMe







-0

-∼

- m

4

ഹ

9-0

- ~

bpm 8

 \equiv





0

-10

20

30-

40

50

-09

-02

-80

06

100

110

140 130 120

170 160 150



 \sim

-ო

4

-ഗ

<u>ں</u>

- ~

-00













-0











-0

-6

-20

-09

-06

Me

0

MeO_OMe



1

-0

 \sim

-ო

4

-ഗ

9–0

- ~

bpm 8













-0

-~

- ന

4

-ഗ

-0

- 1

-00

OMe

0

က=

n





























C

TBS

<u>O</u>Me

0

က=






4

