

# **Supporting Information**

## **Synthesis, Characterization and Spectroscopic Investigation of Benzoxazole Conjugated Schiff Bases**

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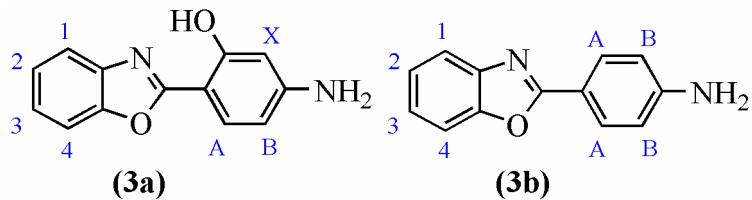
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## SI 1. Spectroscopic characterization

### SI 1.1. Precursors 3a-b



**Figure SI 1.** NMR attributions for the precursors **3a-b**.

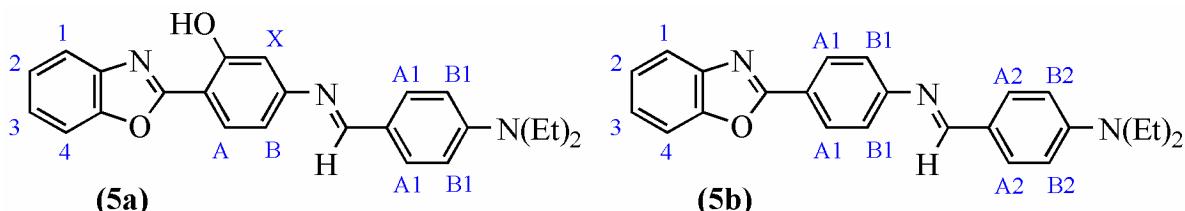
#### Precursor 3a

Yield: 70%. IR (ATR,  $\text{cm}^{-1}$ ): 3488 ( $\nu_{\text{asym}}$  NH<sub>2</sub>), 3382 ( $\nu_{\text{sym}}$  NH<sub>2</sub>), 3049 ( $\nu_{\text{arom}}$  =CH), 1578 e 1556 ( $\nu_{\text{arom}}$  C=C), 1620 (aromatic ring deformation). NMR-<sup>1</sup>H (300 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  (ppm) = 11.1 (s, 1H, OH), 7.70 (d,  $J_o$  = 8.6 Hz, 1H, H<sub>A</sub>), 7.60 (m, 2H, H<sub>1</sub> and H<sub>4</sub>), 7.30 (m, 2H, H<sub>2</sub> and H<sub>3</sub>), 6.25 (dd,  $J_o$  = 8.6 Hz,  $J_m$  = 1.3 Hz, 1H, H<sub>B</sub>), 6.00 (d,  $J_m$  = 1.3 Hz, 1H, H<sub>X</sub>), 6.10 (s, 2H, NH<sub>2</sub>).

#### Precursor 3b

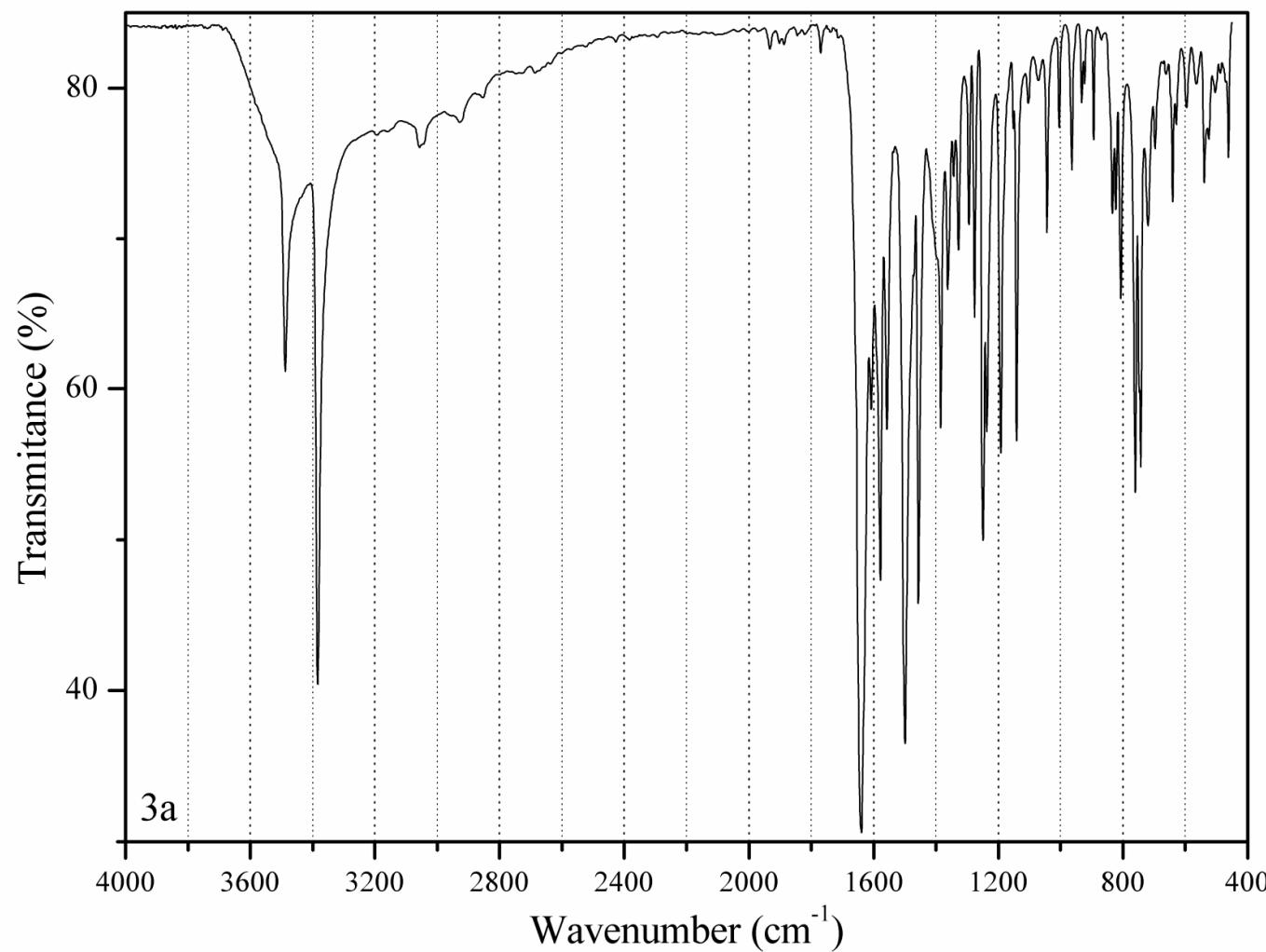
Yield: 72%. IR (ATR,  $\text{cm}^{-1}$ ): 3470 ( $\nu_{\text{asym}}$  NH<sub>2</sub>), 3294 ( $\nu_{\text{sym}}$  NH<sub>2</sub>), 3188 (Fermi's band), 3058 ( $\nu_{\text{arom}}$  =CH), 1577 and 1558 ( $\nu_{\text{arom}}$  C=C), 1602 (aromatic ring deformation). NMR-<sup>1</sup>H (300 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  (ppm) = 7.90 (d,  $J_o$  = 8.6 Hz, 2H, H<sub>A</sub>), 7.63-7.68 (m, 2H, H<sub>1</sub> and H<sub>4</sub>), 7.26-7.34 (m, 2H, H<sub>2</sub> and H<sub>3</sub>), 6.74 (d,  $J_o$  = 8.6 Hz, 2H, H<sub>B</sub>), 6.02 (s, 2H, NH<sub>2</sub>).

## SI 1.2. Attributions for NMR

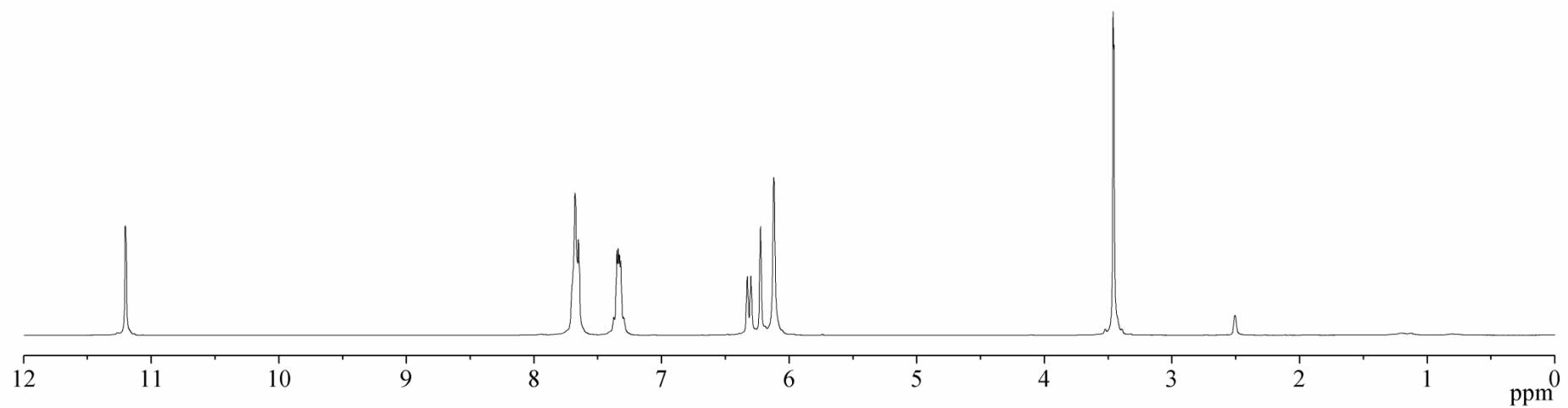


**Figure SI 2.** NMR attributions for the dyes **5a-b**.

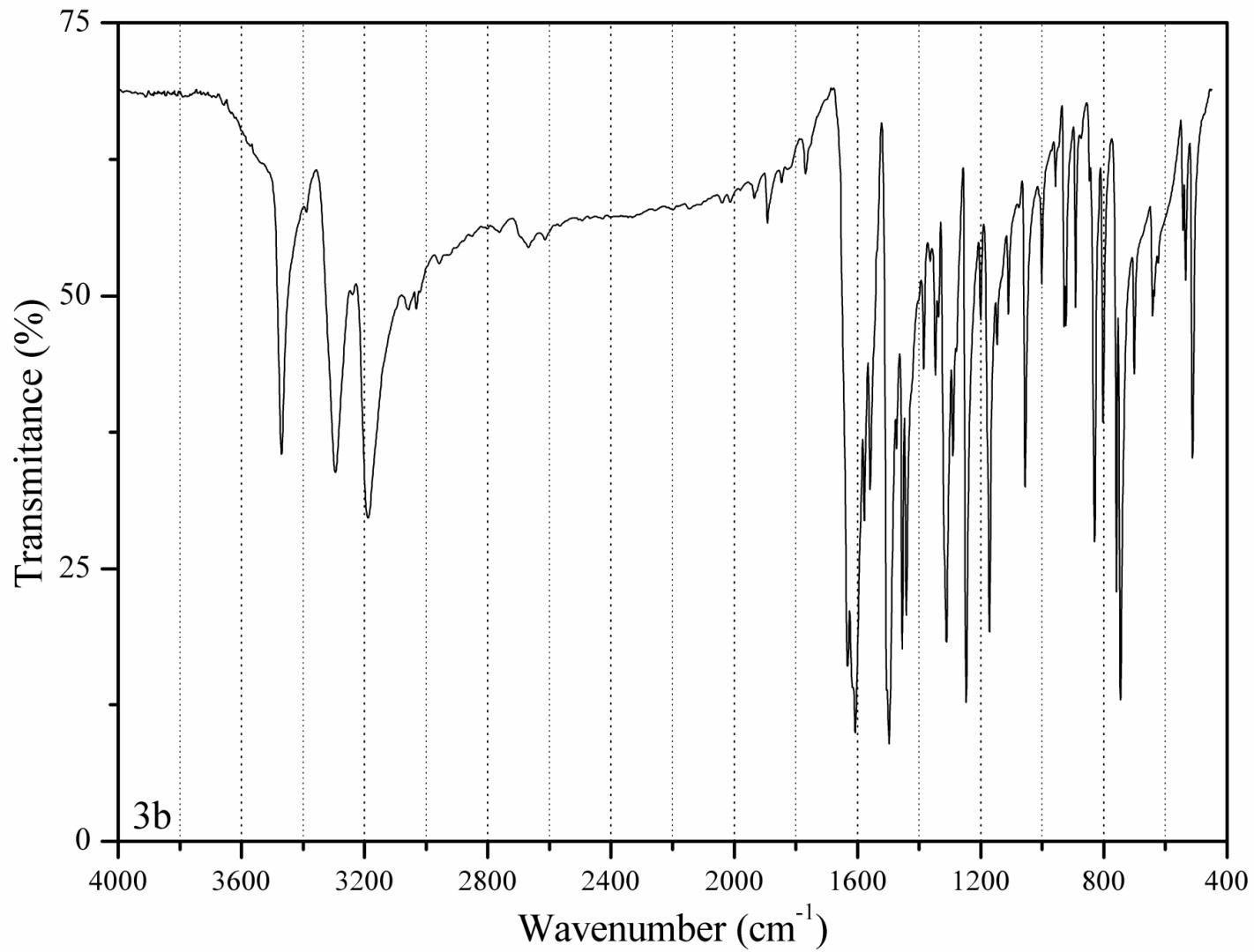
**SI 1.3. Original spectral data**



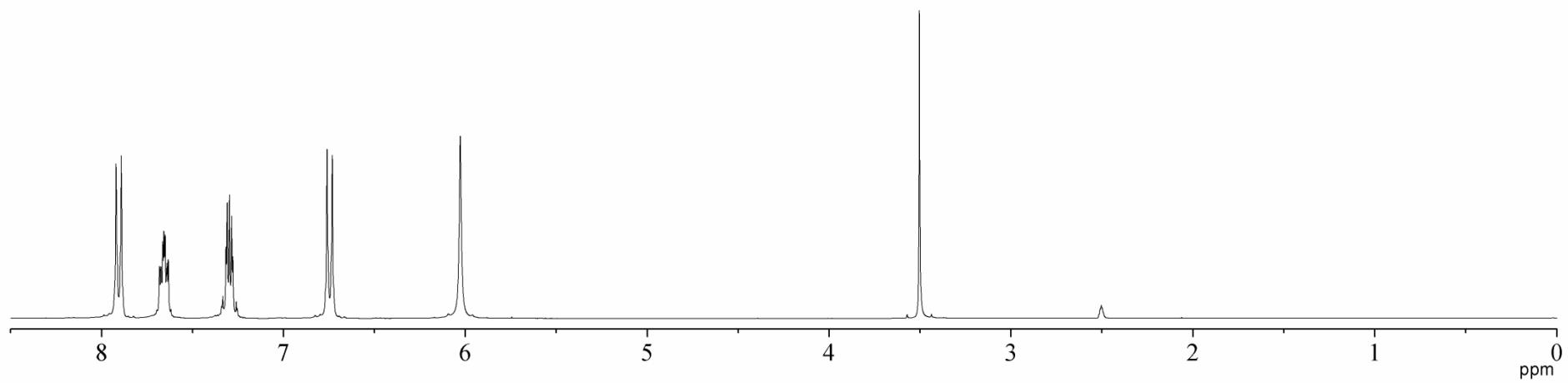
**Figure SI 3.** IR Absorption spectra (KBr) of the precursor **3a**.



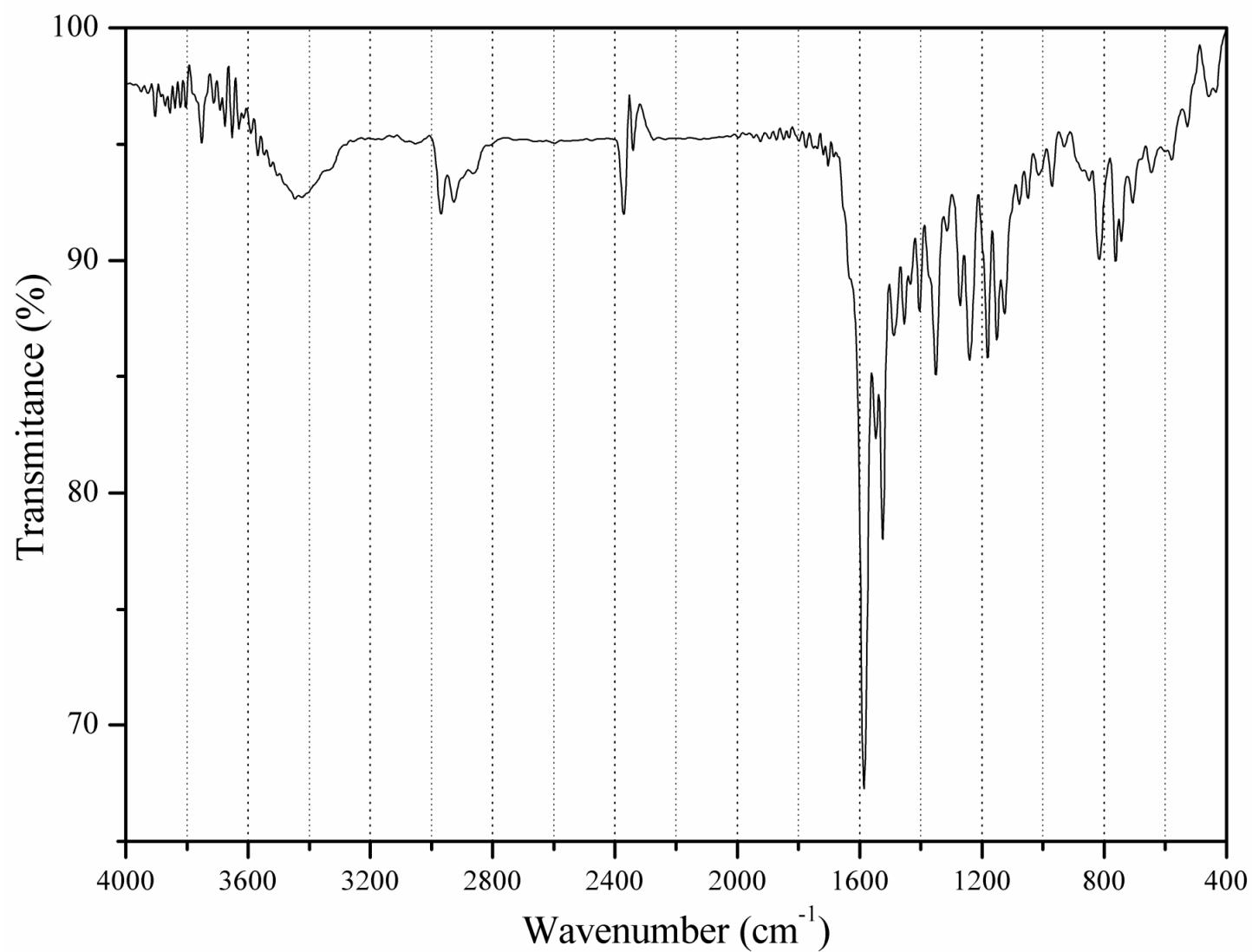
**Figure SI 4.**  $^1\text{H}$  NMR (300 MHz,  $\text{DMSO-d}_6$ ) spectra of the precursor **3a**.



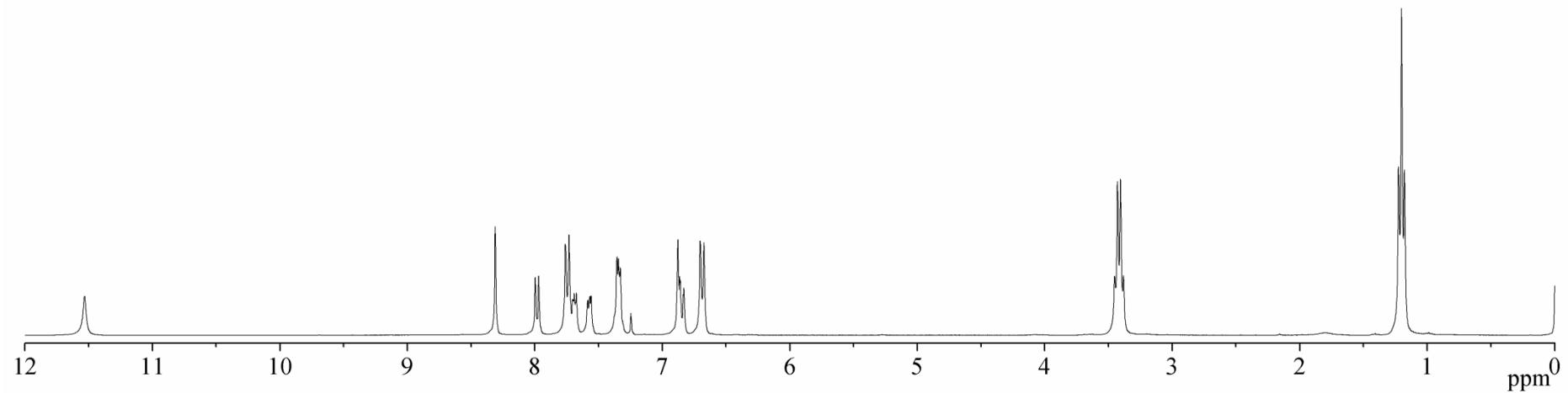
**Figure SI 5.** IR Absorption spectra (KBr) of the precursor **3b**.



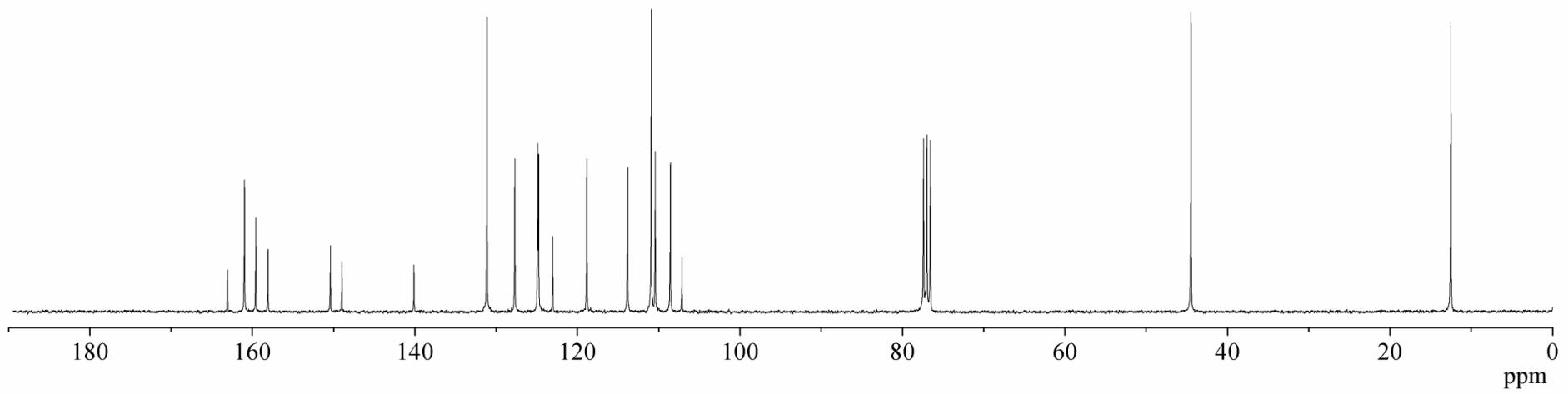
**Figure SI 6.**  $^1\text{H}$  NMR (300 MHz,  $\text{DMSO-d}_6$ ) spectra of the precursor **3b**.



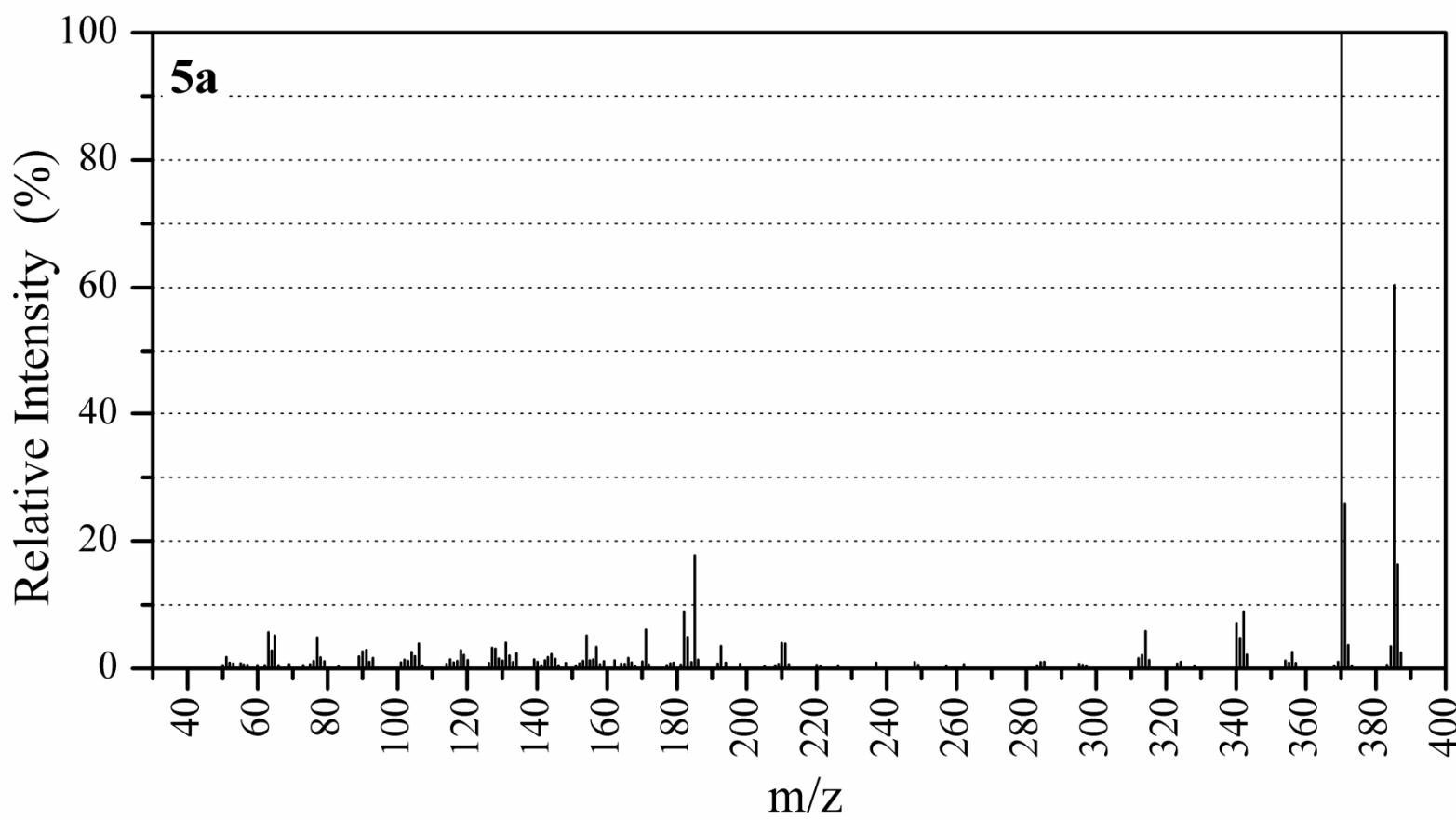
**Figure SI 7.** IR Absorption spectra (ATR) of the dye **5a**.



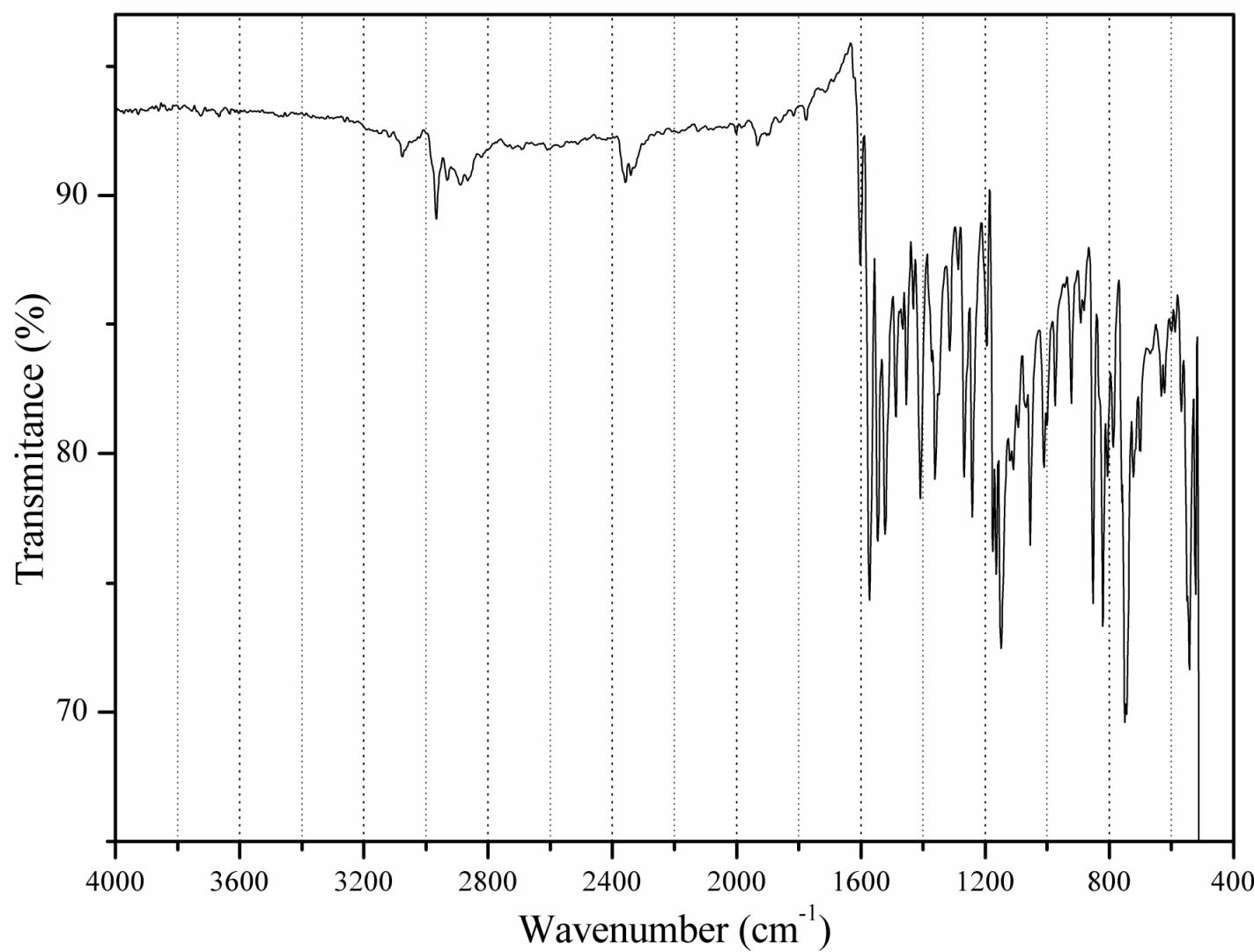
**Figure SI 8.**  ${}^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ) spectra of the dye **5a**.



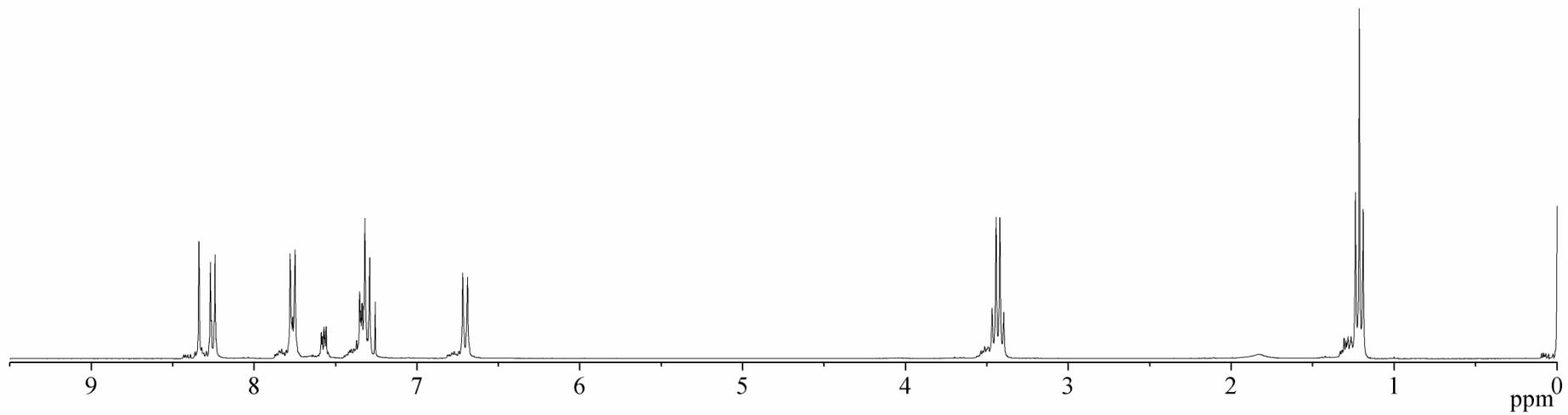
**Figure SI 9.**  $^{13}\text{C}$  NMR (75,4 MHz,  $\text{CDCl}_3$ ) spectra of the dye **5a**.



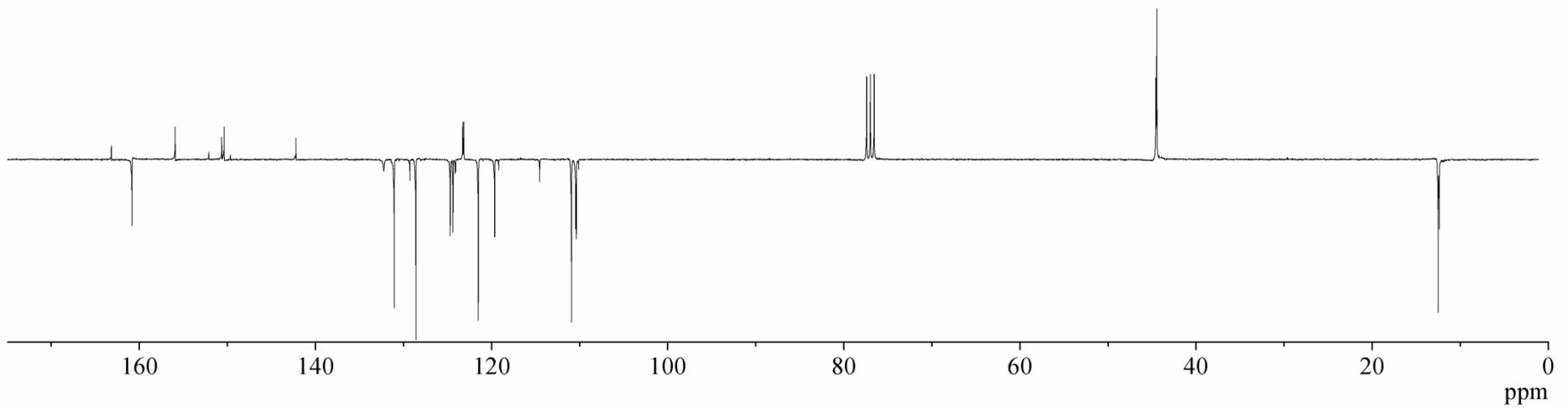
**Figure SI 10.** EIMS fragmentation pattern of the dye **5a**.



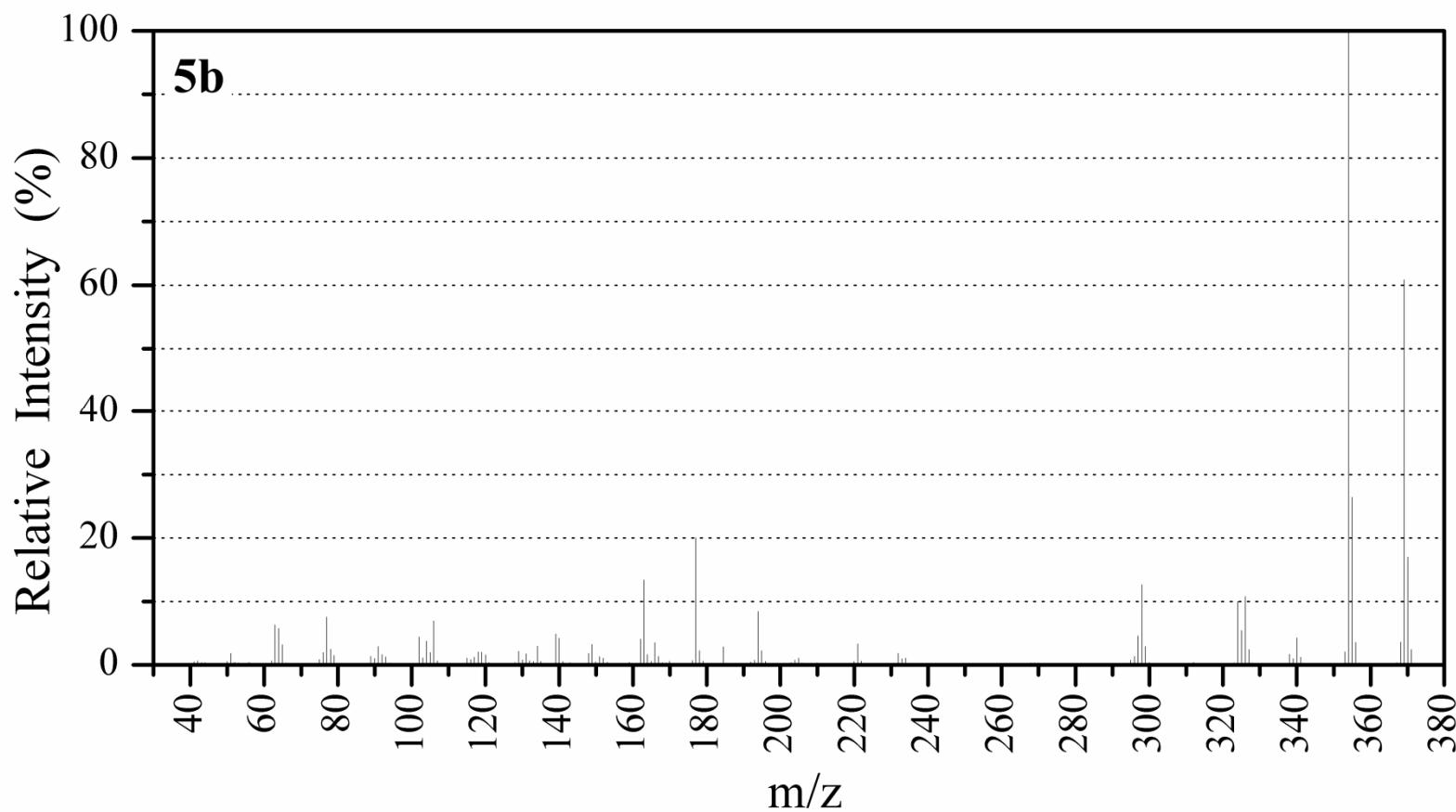
**Figure SI 11.** IR Absorption spectra (ATR) of the dye **5b**.



**Figure SI 12.**  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ) spectra of the dye **5b**.

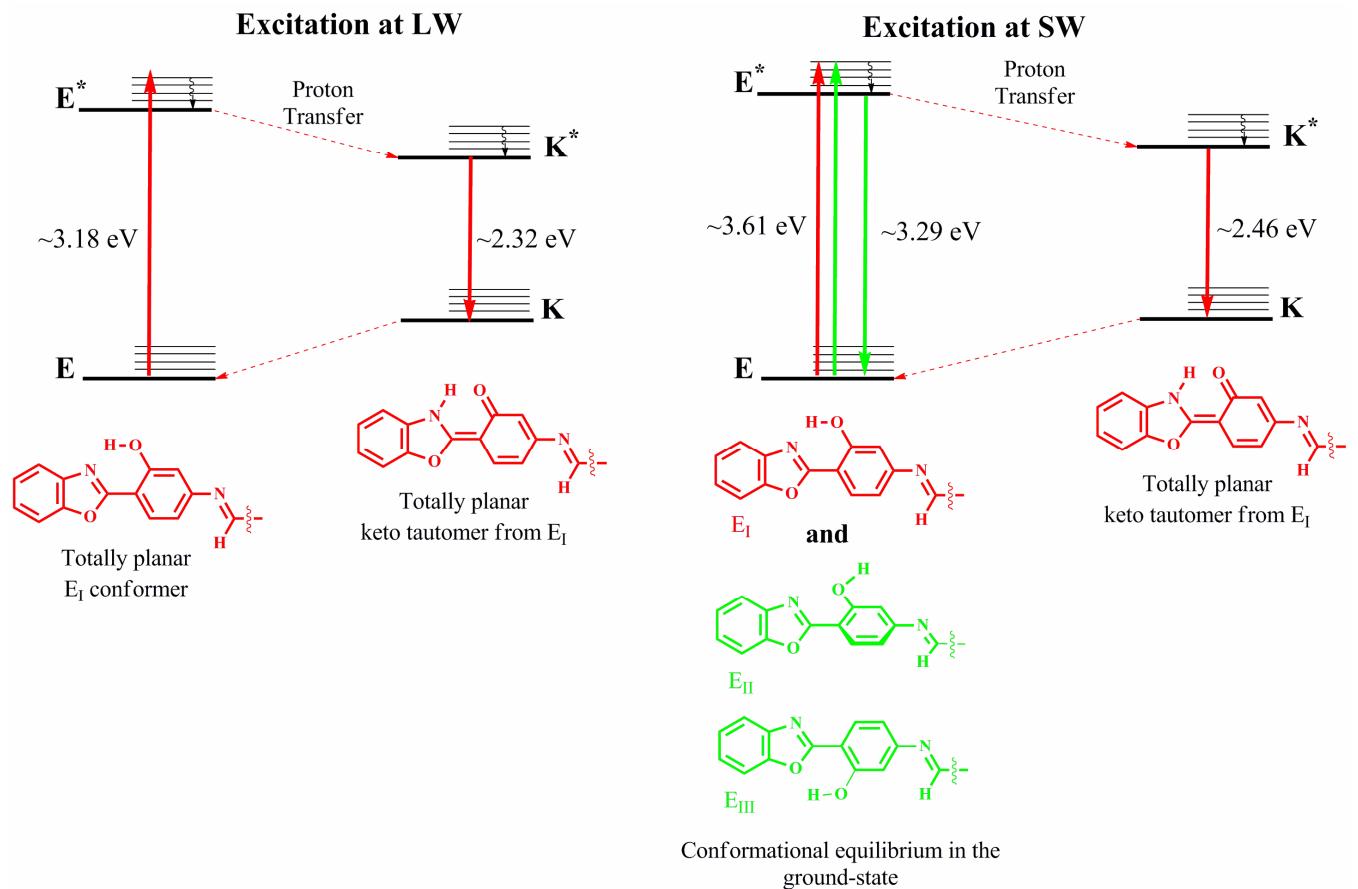


**Figure SI 13.** APT (75.4 MHz, CDCl<sub>3</sub>) spectra of the dye **5b**.

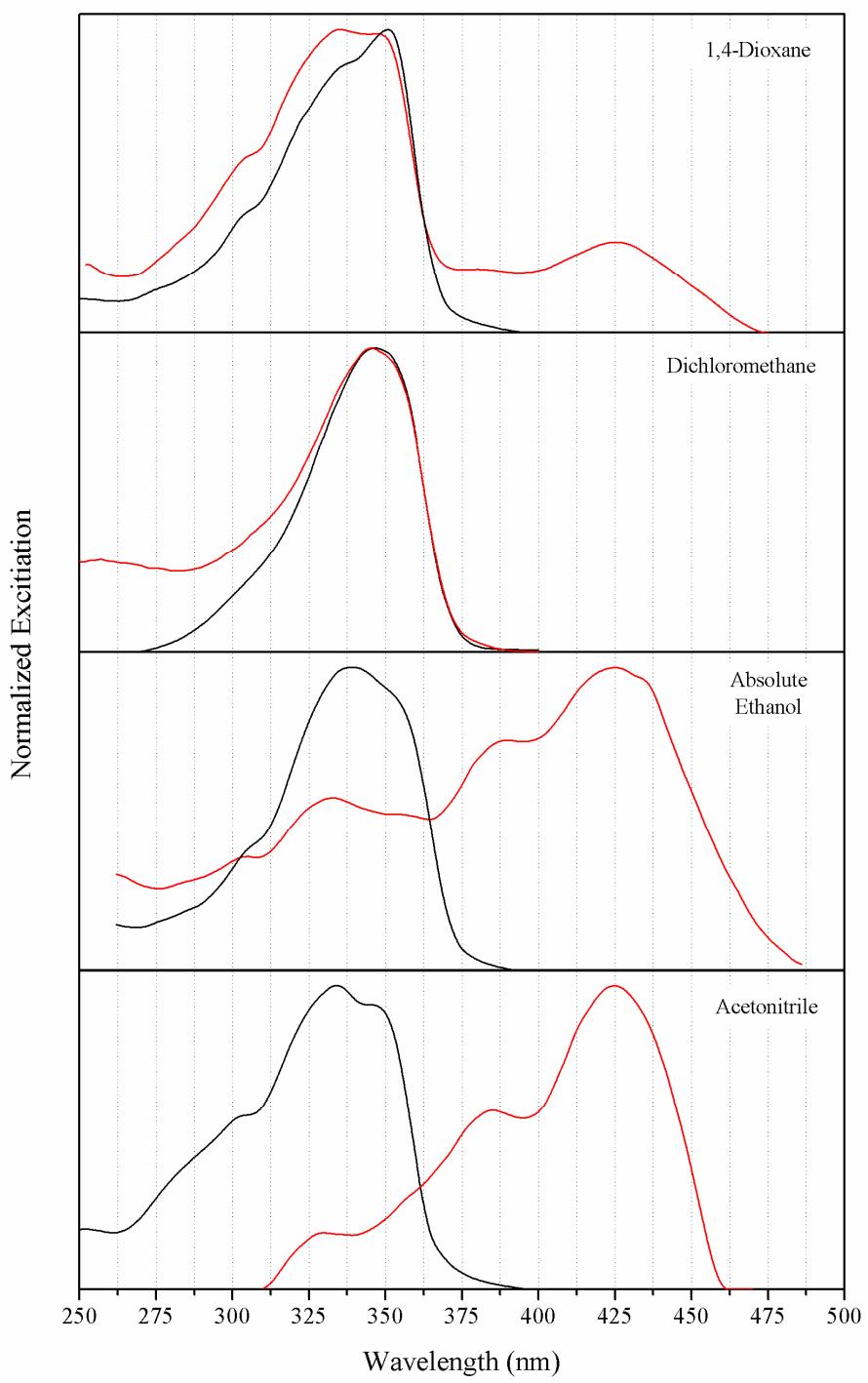


**Figure SI 14.** EIMS fragmentation pattern of the dye **5b**.

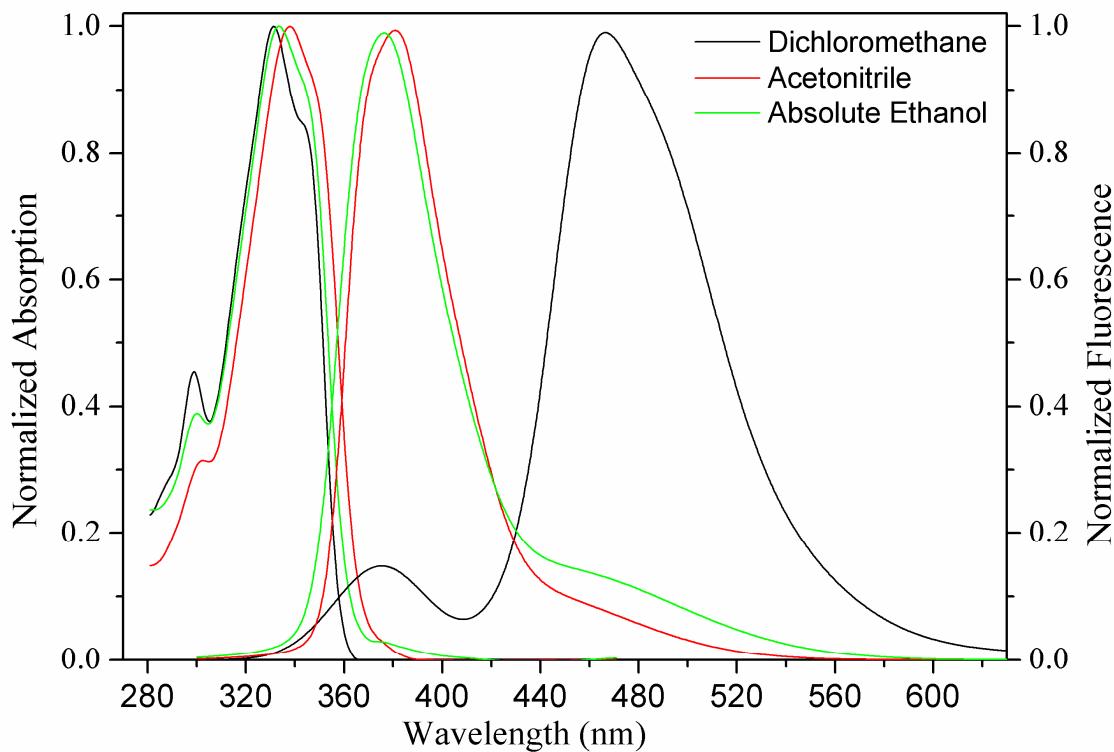
## SI 2. Photophysical Data



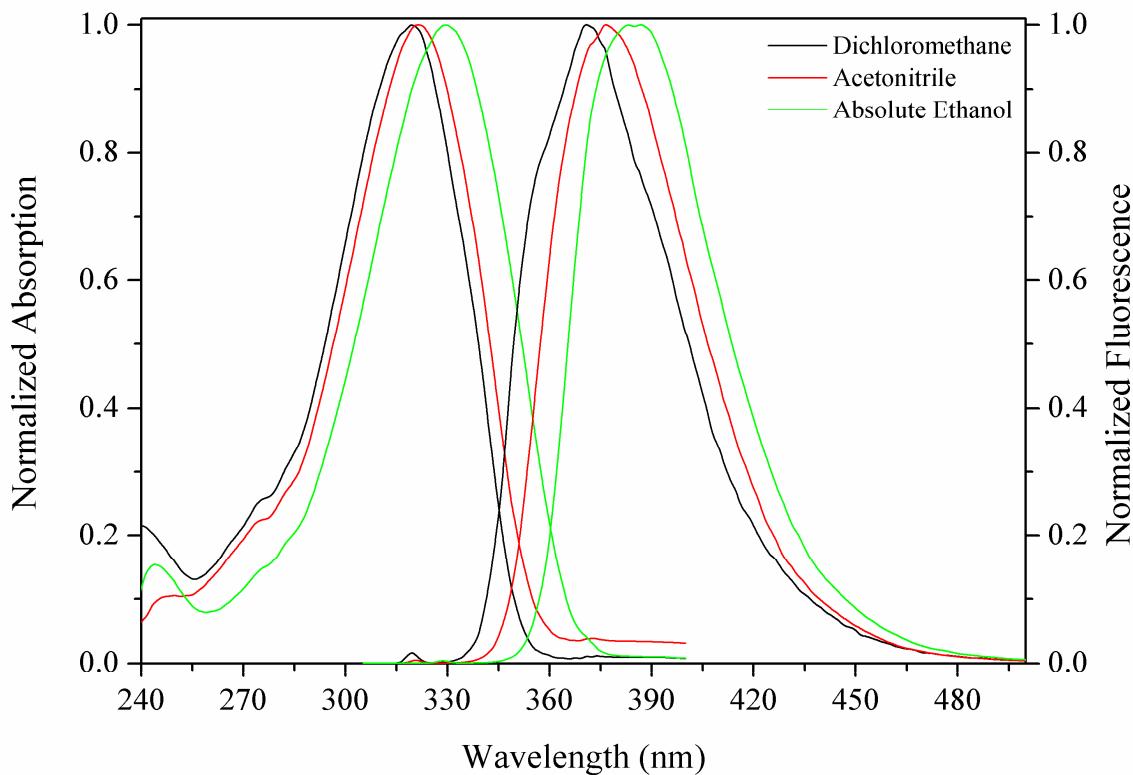
**Figure SI 15.** General Scheme of the ESIPT mechanism in the dye **5a**. The absorption and emission values are from the photophysical data (see Table 1 and 2 in the manuscript).



**Figure SI 16.** Normalized Excitation spectra of the dye **5a** using the normal emission (black solid line) and the keto emission (red solid line) as observation wavelengths.



**Figure SI 17.** Normalized UV-Vis absorption and fluorescence emission of **3a**.



**Figure SI 18.** Normalized UV-Vis absorption and fluorescence emission of **3b**.

**Table SI1.** Spectroscopic data from precursor **3a**.

Solvent	$\lambda_{\text{abs}}$ (nm)	$\epsilon \times 10^4$ ( $M^{-1} cm^{-1}$ )	$\lambda_{\text{em}}$ (nm)	$\Delta\lambda_{\text{ST}}$ (nm)	$\Phi_{\text{fl}}^*$	[ ] (mol/L)
CH <sub>2</sub> Cl <sub>2</sub>	332	2.8	467	135	0.012	1.6 x10 <sup>-5</sup>
CH <sub>3</sub> CN	334	1.5	378	42	0.023	3.2 x10 <sup>-5</sup>
EtOH	335	5.3	376	43	0.025	2.6 x10 <sup>-5</sup>

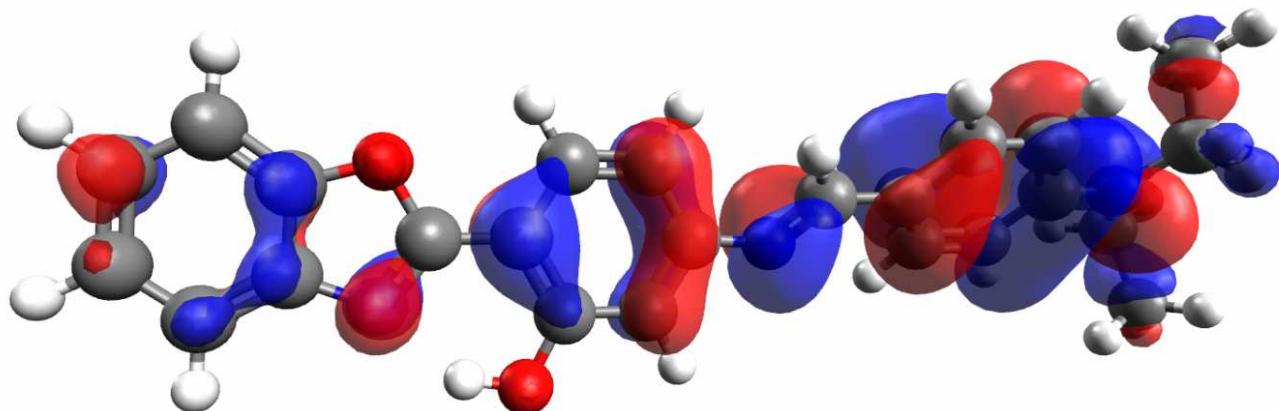
**Table SI2.** Spectroscopic data from precursor **3b**.

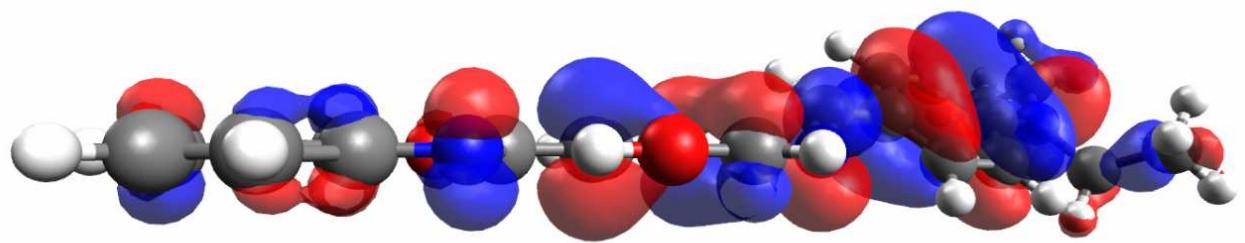
Solvent	$\lambda_{\text{abs}}$ (nm)	$\epsilon \times 10^4$ ( $M^{-1} cm^{-1}$ )	$\lambda_{\text{em}}$ (nm)	$\Delta\lambda_{\text{ST}}$ (nm)	$\Phi_{\text{fl}}^*$	[ ] (mol/L)
CH <sub>2</sub> Cl <sub>2</sub>	319	3.00	371	52	0.58	2.1x10 <sup>-5</sup>
CH <sub>3</sub> CN	321	3.13	376	55	0.51	2.3x10 <sup>-5</sup>
EtOH	330	0.35	385	55	0.48	1.22x10 <sup>-4</sup>

\* Using quinine sulphate as fluorescence quantum yield standard.

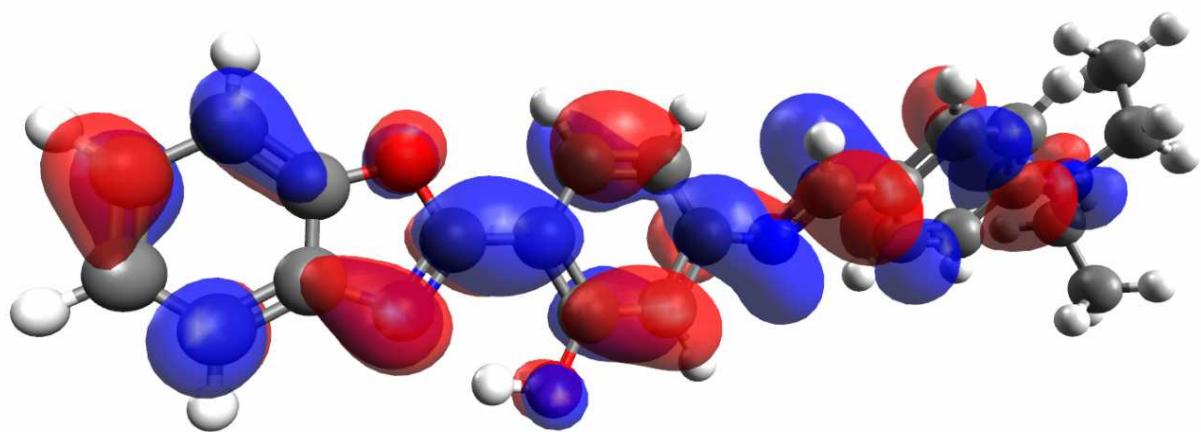
### SI 3. Theoretical information data

#### SI 3.1. Dye **5a**

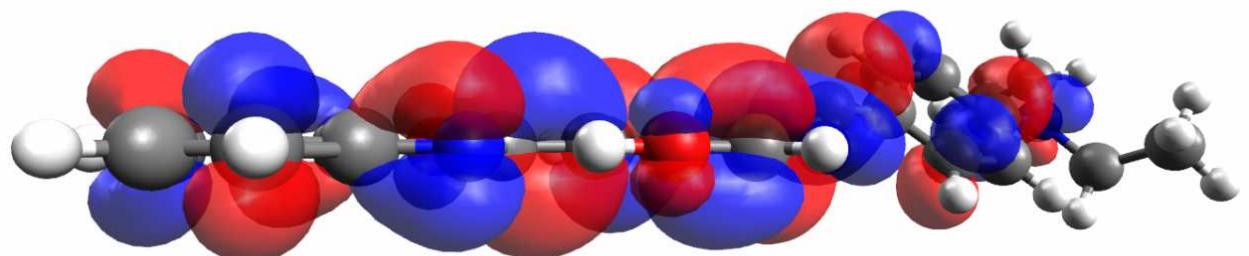
**Figure SI 19.** Superior view of the HOMO orbitals for **5a**.



**Figure SI 20.** Lateral view of the HOMO orbitals for **5a**.

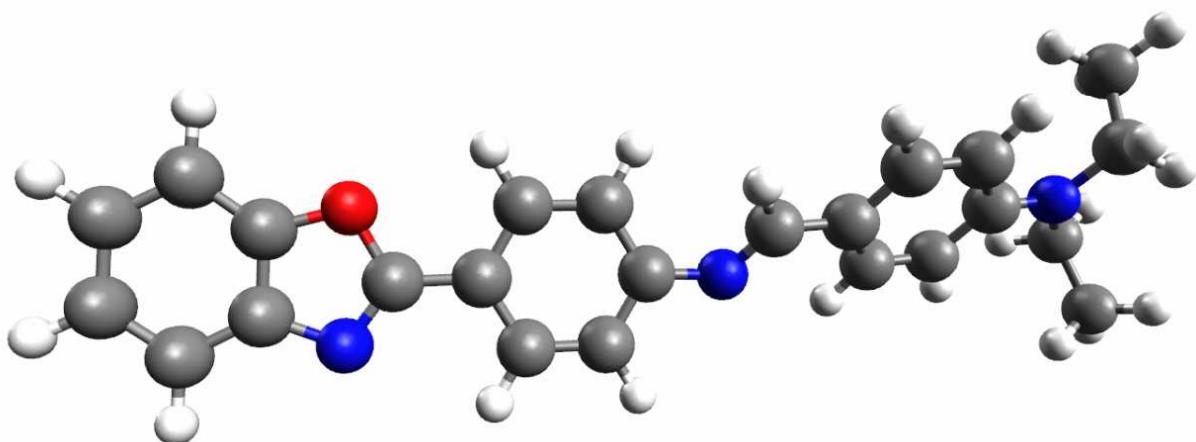


**Figure SI 21.** Superior view of the LUMO orbitals for **5a**.

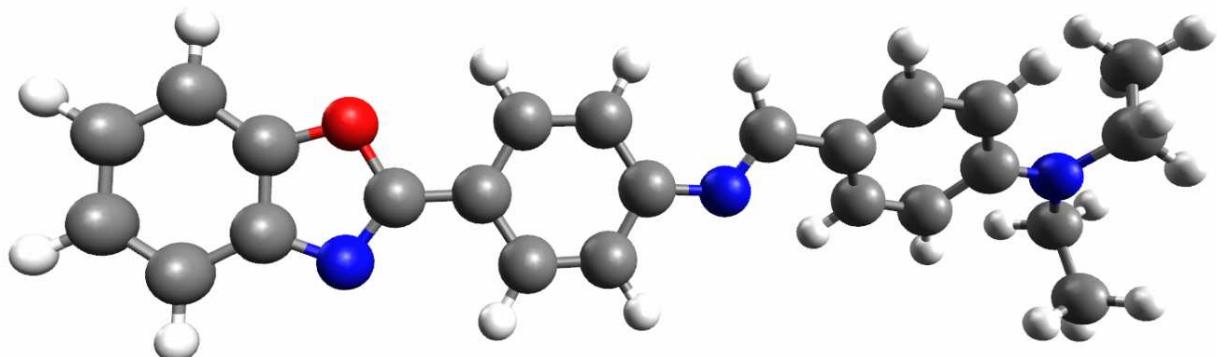


**Figure SI 22.** Lateral view of the LUMO orbitals for **5a**.

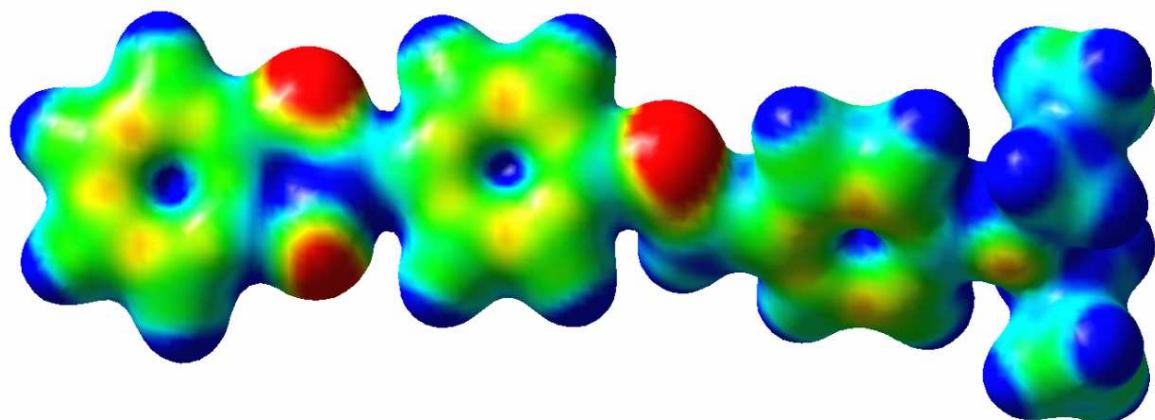
**SI 3.2. Dye **5b****



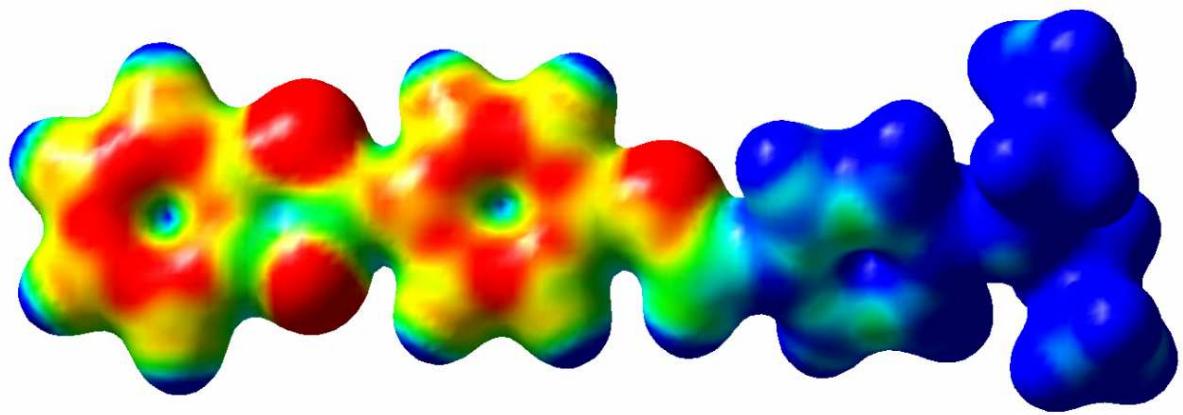
**Figure SI 23.** Structure of the Schiff base **5b** in the ground state.



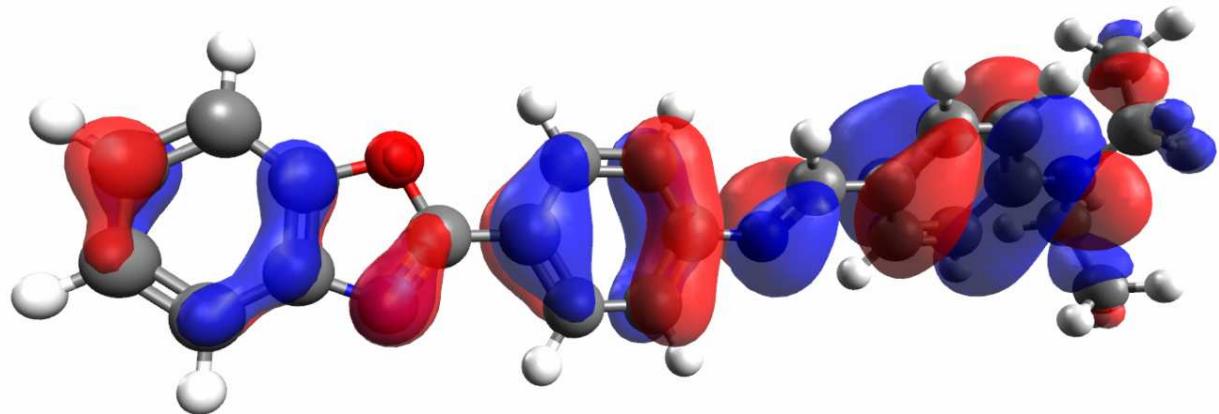
**Figure SI 24.** Structure of the Schiff base **5b** in the excited state.



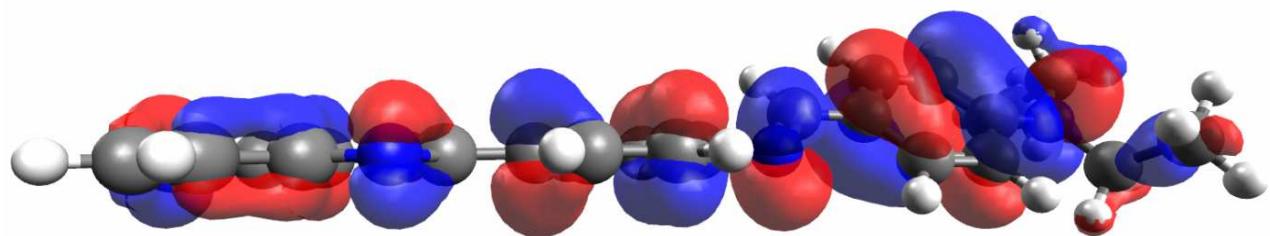
**Figure SI 25.** Electrostatic potential surfaces of the Schiff base **5b** in the ground state.



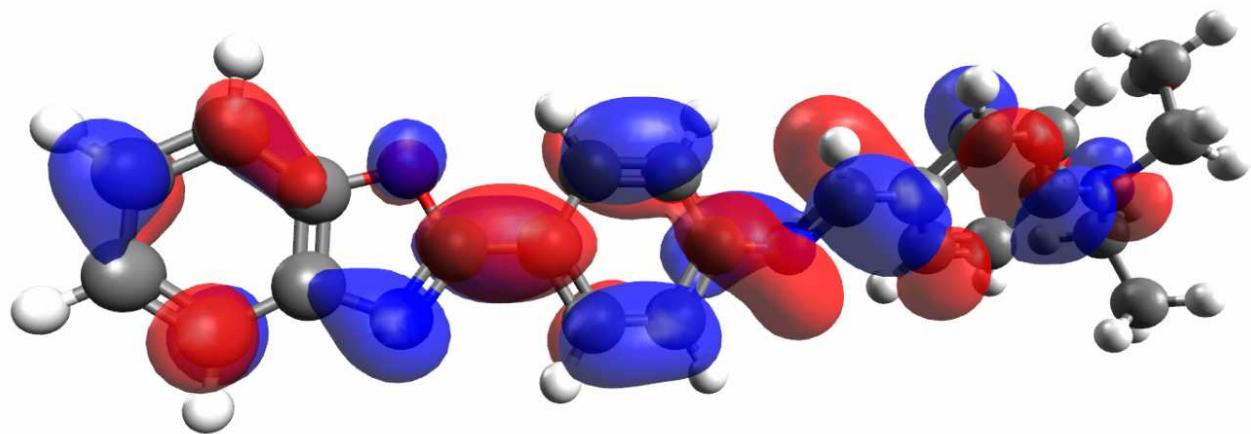
**Figure SI 26.** Electrostatic potential surfaces of the Schiff base **5b** in the excited state.



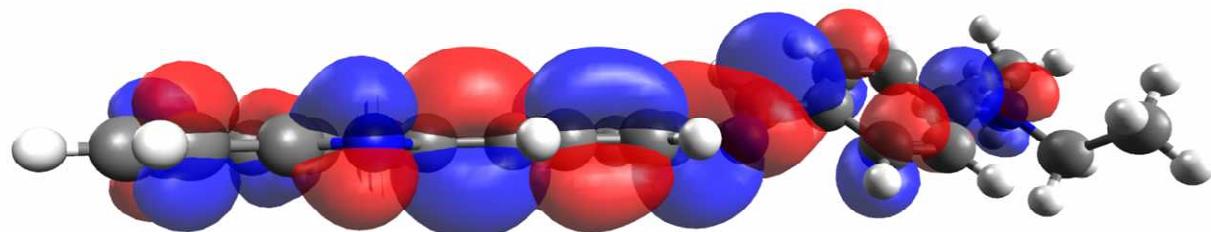
**Figure SI 27.** Superior view of the HOMO orbitals for **5b**.



**Figure SI 28.** Lateral view of the HOMO orbitals for **5b**.



**Figure SI 29.** Superior view of the LUMO orbitals for **5b**.



**Figure SI 30.** Lateral view of the LUMO orbitals for **5b**.