

Preparation and characterization of persistent maltose-conjugated triphenylmethyl radicals

Juan Antonio Mesa,[†] Amado Velázquez-Palenzuela,[‡] Enric Brillas,[‡] Josep Coll,[†] Josep Lluís Torres,[†] Luis Juliá^{†*}

[†]*Departament de Química Biològica i Modelització Molecular, Institut de Química Avançada de Catalunya (CSIC), Jordi Girona 18-26, 08034 Barcelona, Spain.*

[‡]*Departament de Química Física, Universitat de Barcelona, Martí i Franquès 1-11, 08028 Barcelona, Spain.*

SUPPLEMENTARY MATERIAL

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Bis(2,4,6-trichlorophenyl){2,6-dichloro-4-[2,3,4,6-tetra-O-acetyl- α -D-glucopyranosyl-(1 $\rightarrow$ 4)-2,3,6-tri-O-acetyl- β -D-glucopyranosyloxy]phenyl}methyl radical (1**)**

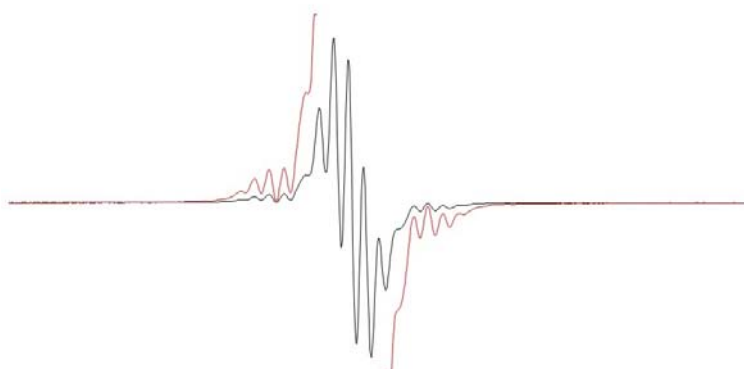


Figure S1. Experimental EPR of radical **1** in CH₂Cl₂ at 180 K. The red lines are y-axis expanded spectra.

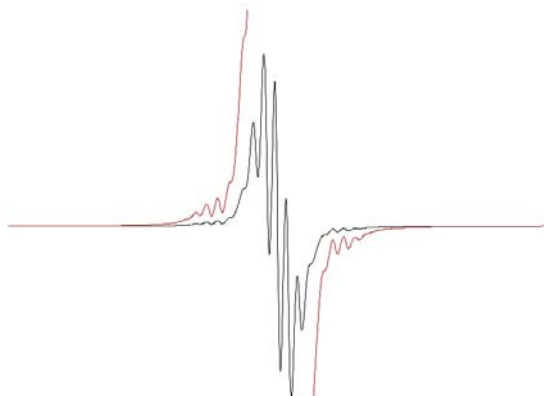


Figure S2. Simulated EPR of radical **1** in CH₂Cl₂ at 180 K. The red lines are y-axis expanded spectra.

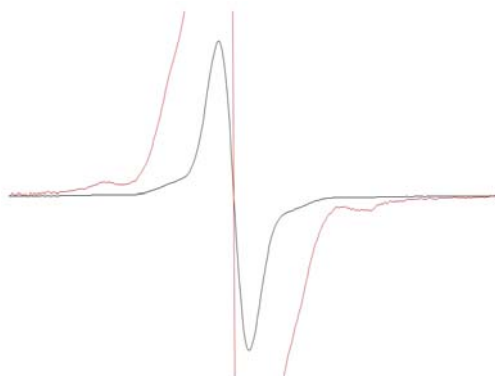


Figure S3. Experimental EPR of radical **1** in CH₂Cl₂ at room temperature. The red lines are y-axis expanded spectra.

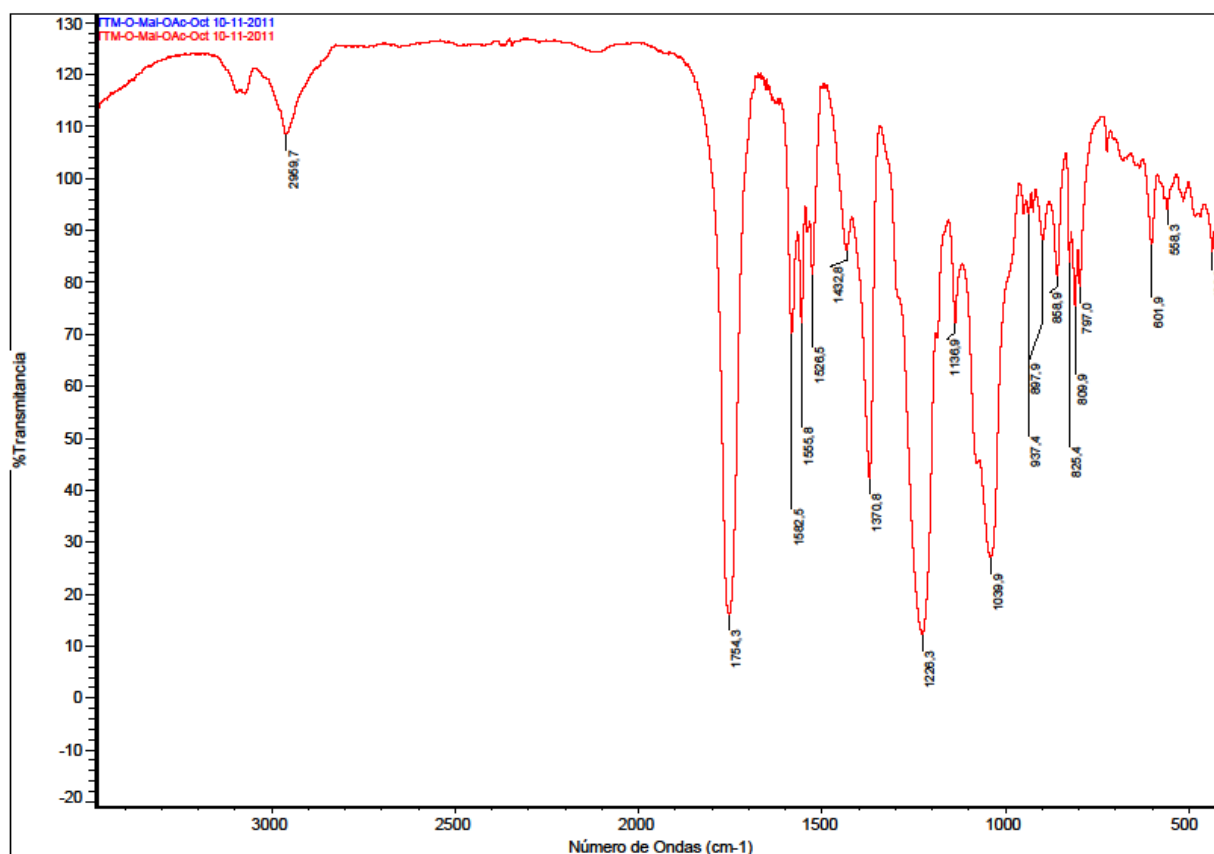


Figure S4. Infrared spectrum of radical **1**

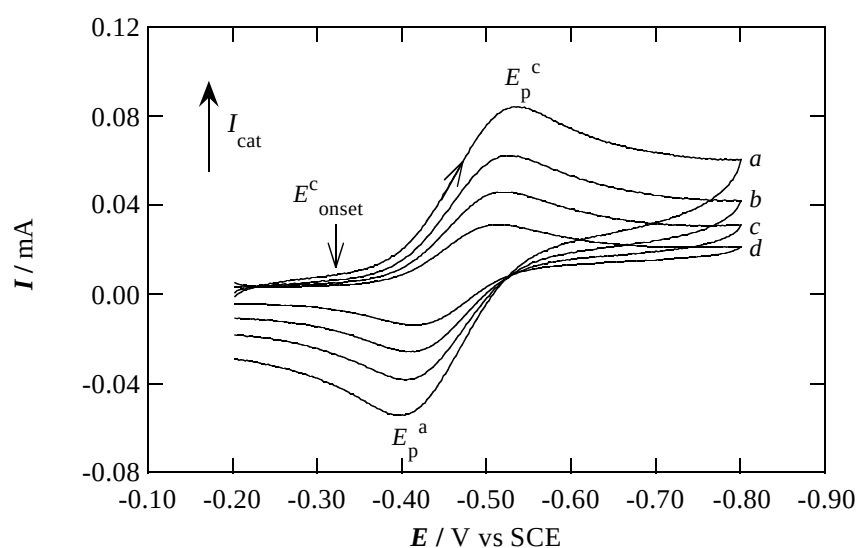


Figure S5. Cyclic voltammograms obtained for the reduction of radical **1** (10^{-3} M) in THF/water 10:1 (v/v) with TBPA (0.1 M) on a Pt electrode at 25 °C under N_2 atmosphere. Initial and final potential: -0.20 V, reversal potential: -0.80 V. Scan rate: (a) 200, (b) 100, (c) 50 y (d) 20 $mV s^{-1}$.

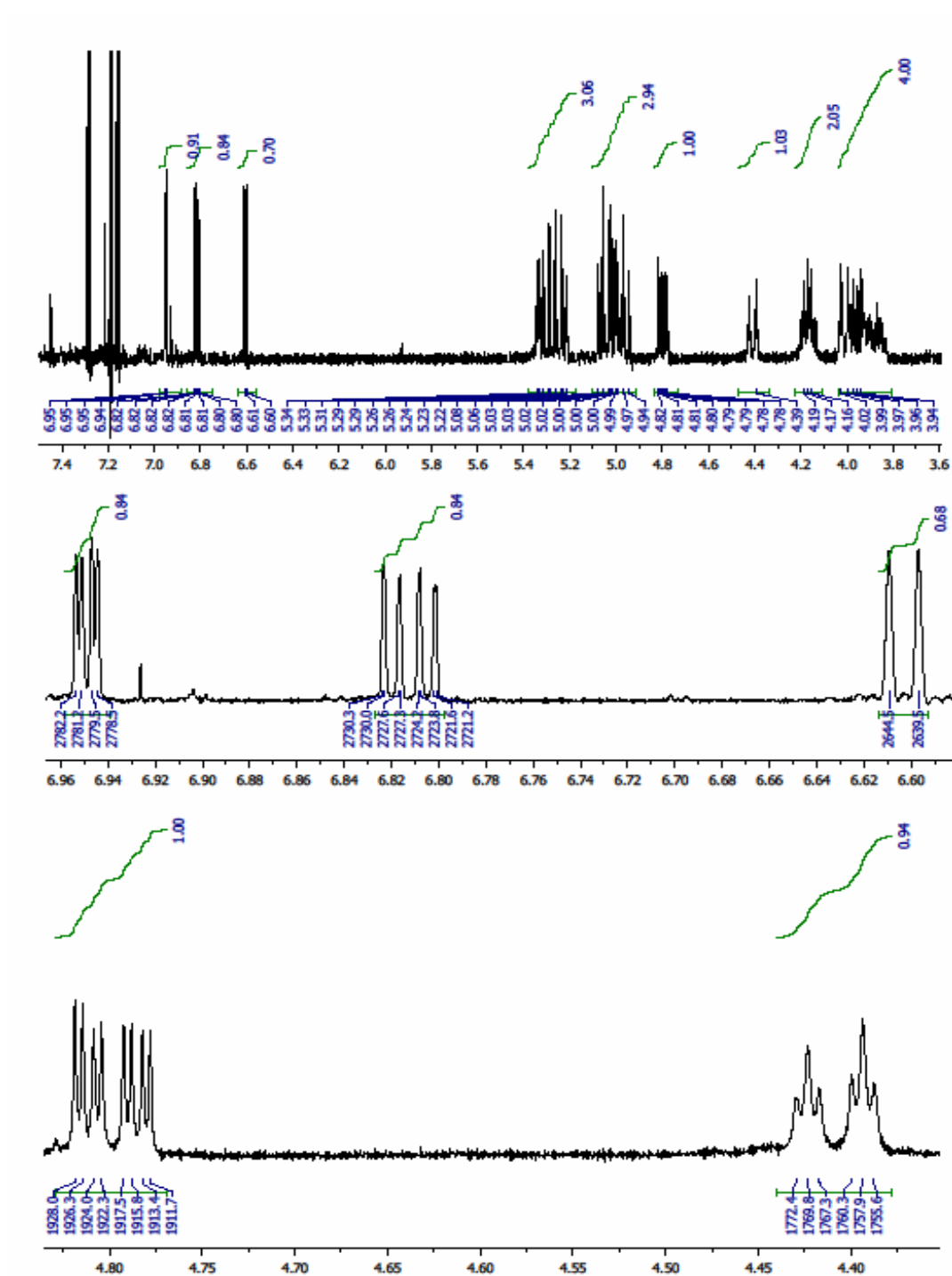


Figure S6. ^1H NMR (CDCl_3 , 400 MHz) spectrum and expansions of the same of compound **1H**

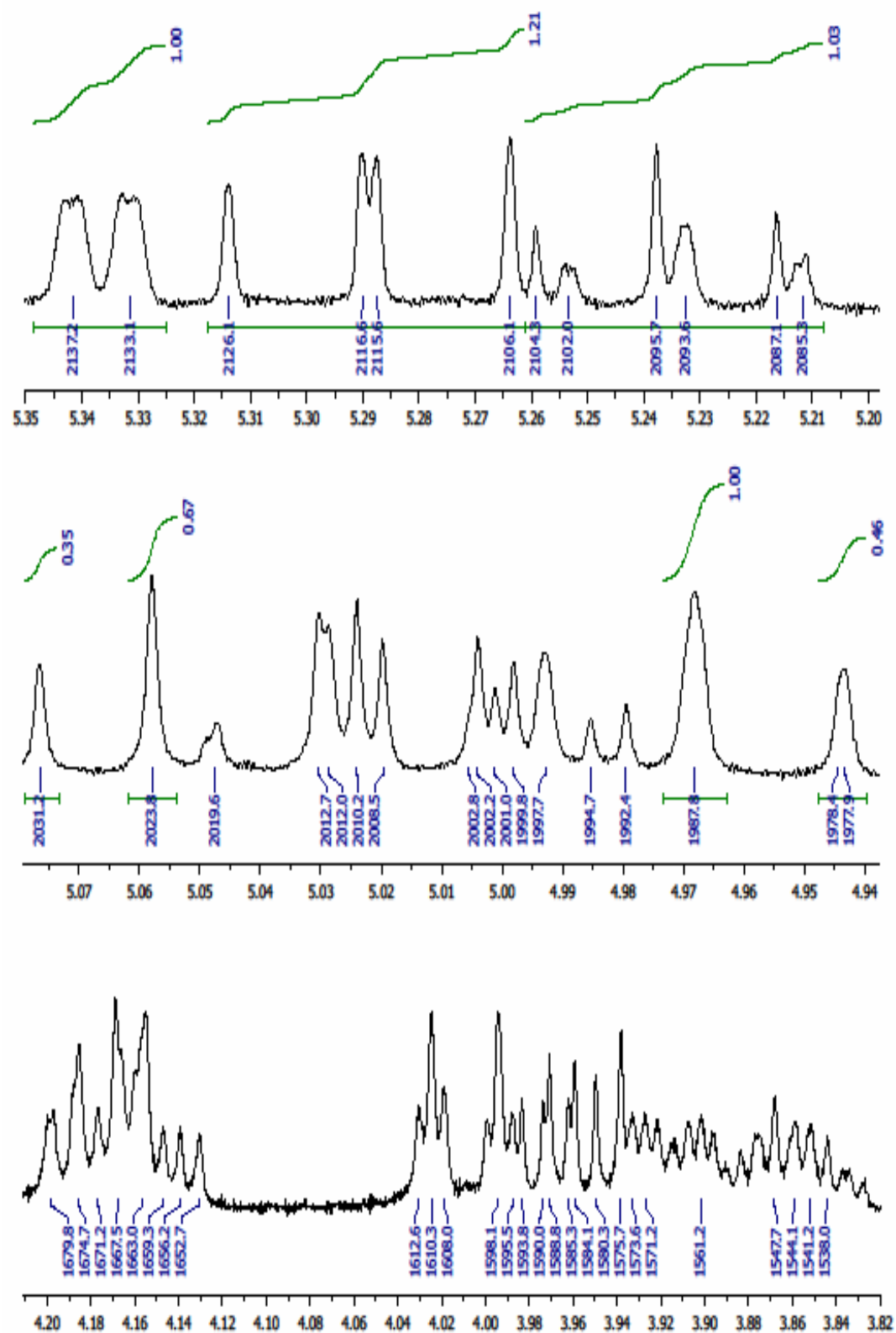


Figure S7. Expansions of the ^1H NMR (CDCl_3 , 400 MHz) of compound **1H**

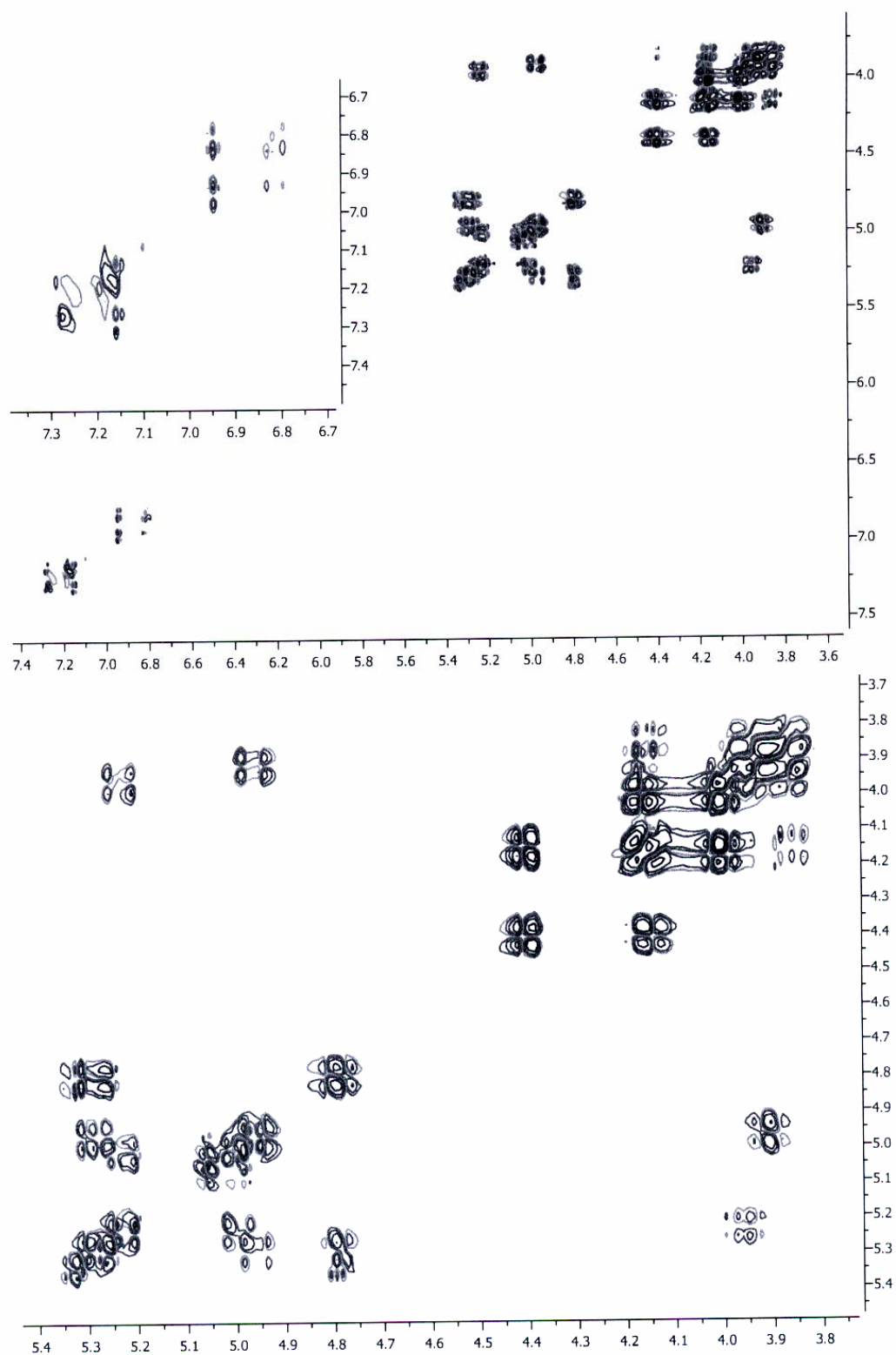


Figure S8. Details of the DQ-COSY NMR (CDCl₃, 400 MHz) spectrum of **1H**

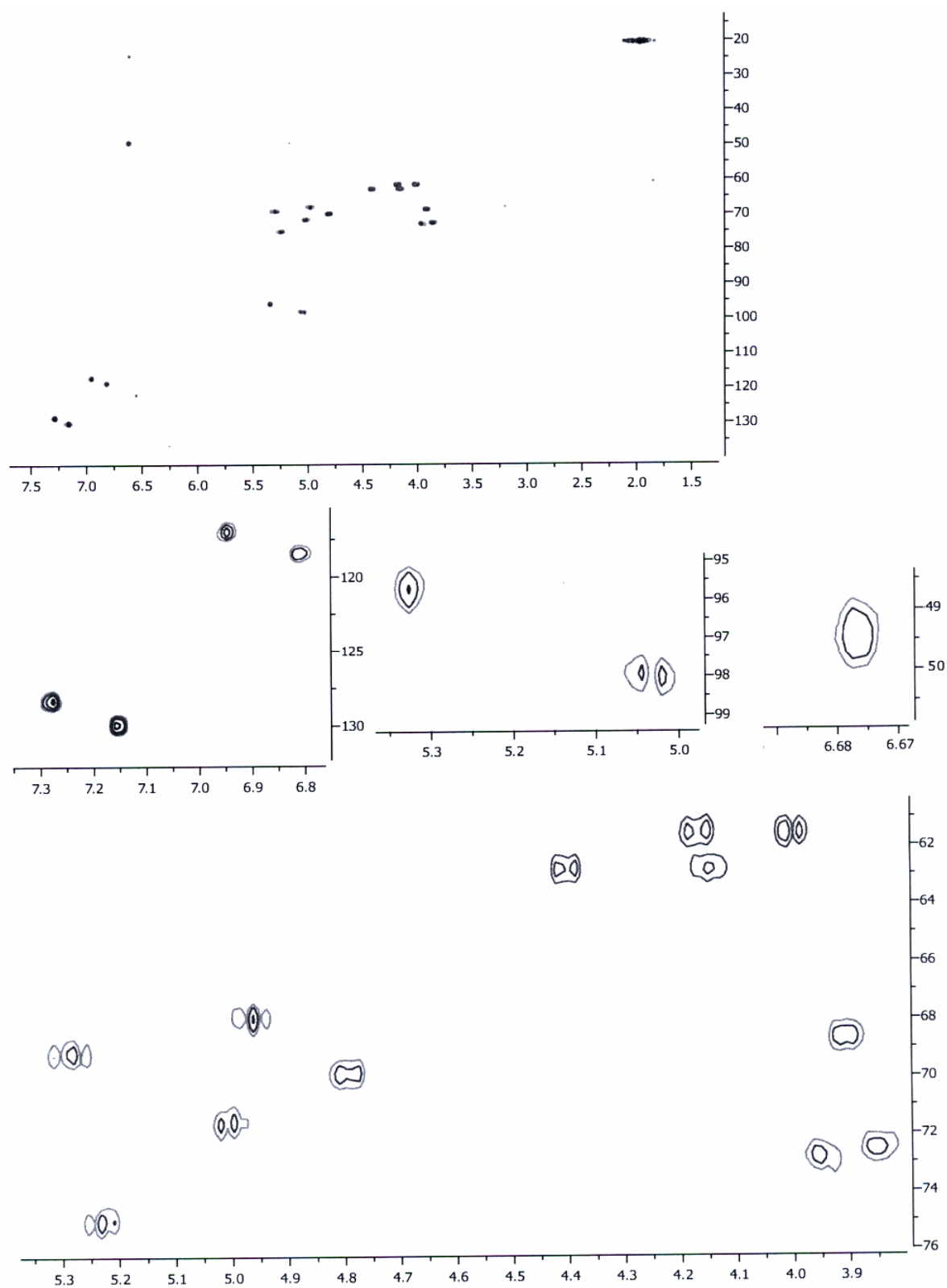


Figure S9. Details of the HSQC NMR (CDCl₃, 400 MHz) spectrum of **1H**

Bis(2,3,4,5,6-pentachlorophenyl){2,3,5,6-tetrachloro-4-[2,3,4,6-tetra-O-acetyl- α -D-glucopyranosyl-(1 $\rightarrow$ 4)-2,3,6-tri-O-acetyl- β -D-glucopyranosyloxy]phenyl}methyl radical (2)

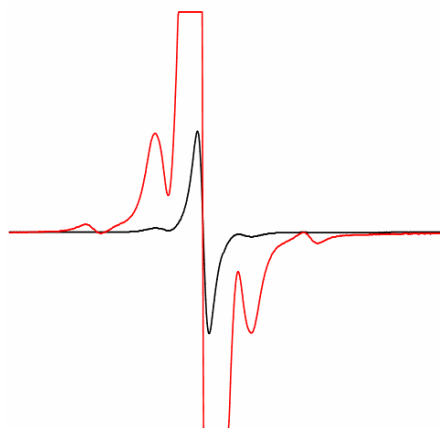


Figure S10. Experimental EPR of radical **2** in CH_2Cl_2 at room temperature. The red lines are y-axis expanded spectra.

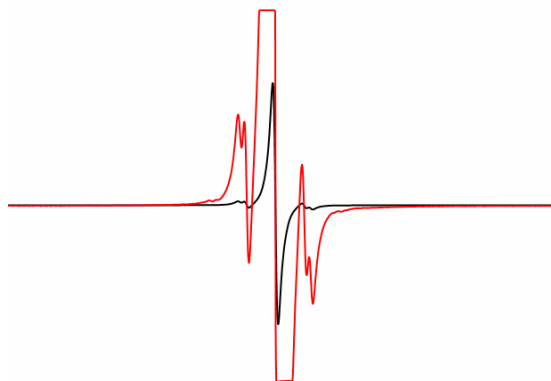


Figure S11. Experimental EPR of radical **2** in CH_2Cl_2 at 193 K. The red lines are y-axis expanded spectra.

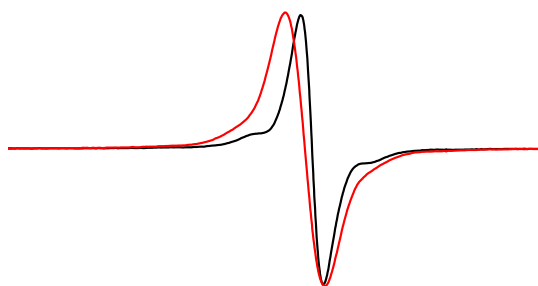


Figure S12. Experimental EPR of radical **1** (black) ($g = 2.0025 \pm 0.0001$) and **2** (red) ($g = 2.0030 \pm 0.0001$) in EtOH solution ($\sim 10^{-3}$ M) at room temperature

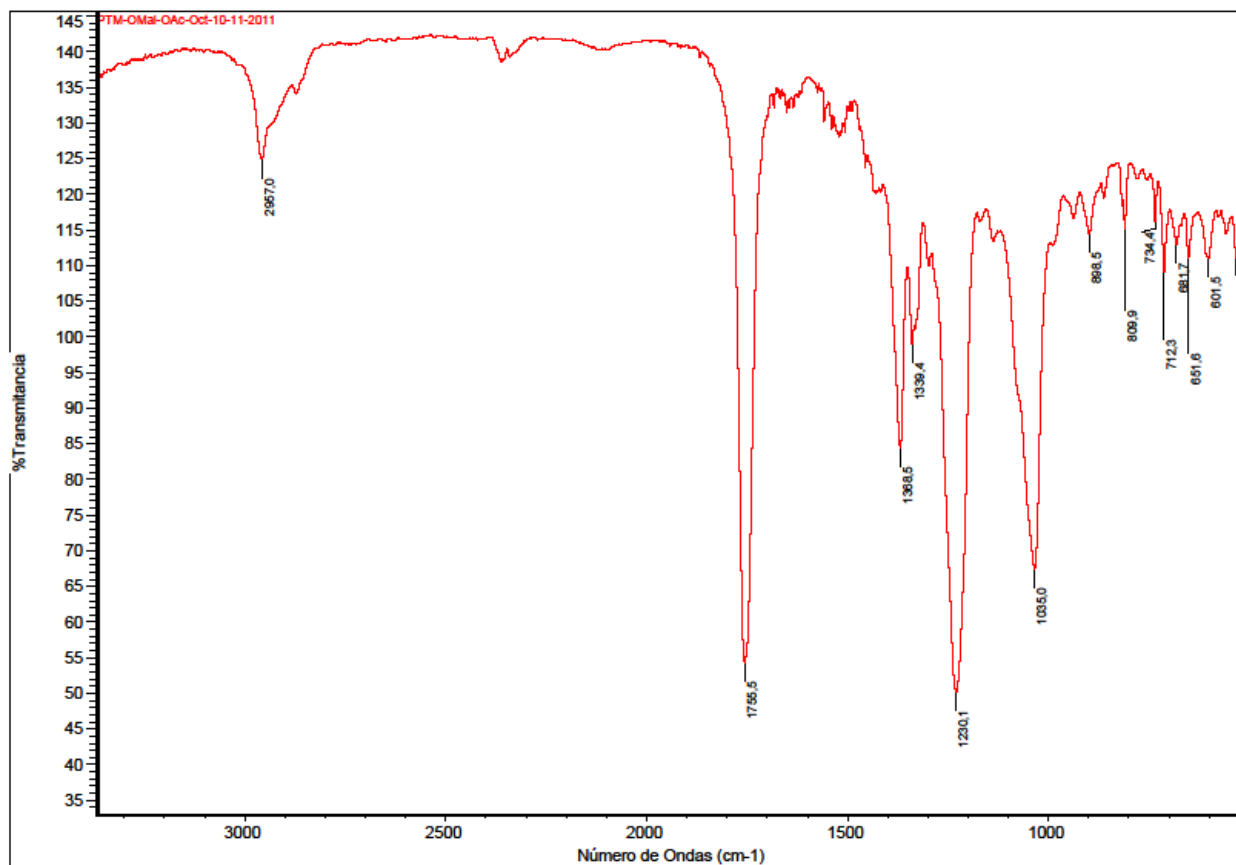


Figure S13. Infrared spectrum of radical **2**

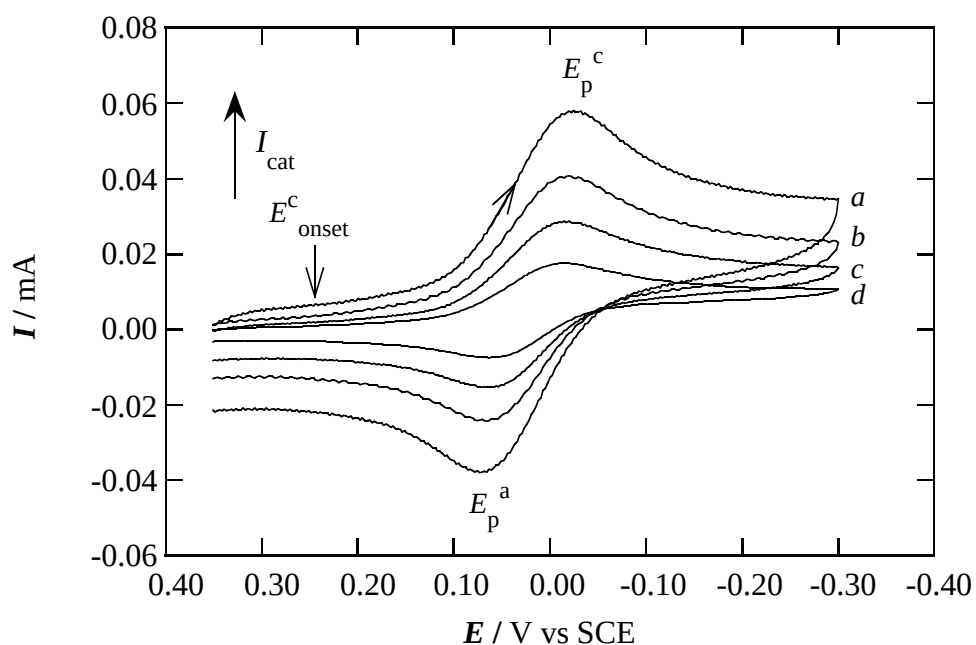


Figure S14. Cyclic voltammograms obtained for the reduction of radical **2** (10^{-3} M) in THF/water 10:1 (v/v) with TBPA (0.1 M) on a Pt electrode at 25 °C under N_2 atmosphere. Initial and final potential: 0.35 V, reversal potential: -0.30 V. Scan rate: (a) 200, (b) 100, (c) 50 y (d) 20 mV s^{-1} .

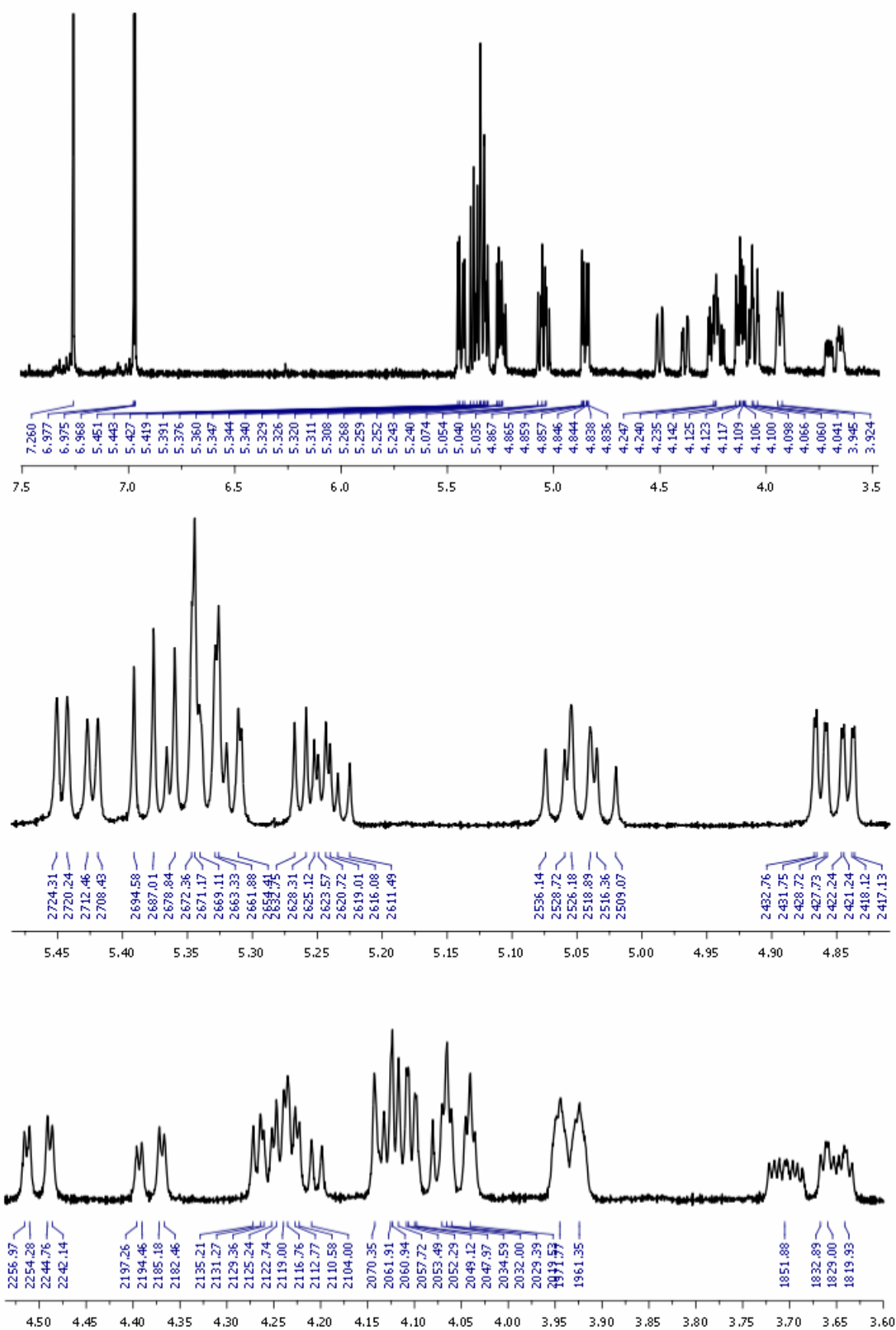


Figure S15. ^1H NMR spectrum and expansions (CDCl_3 , 400 MHz) of compound **2H**

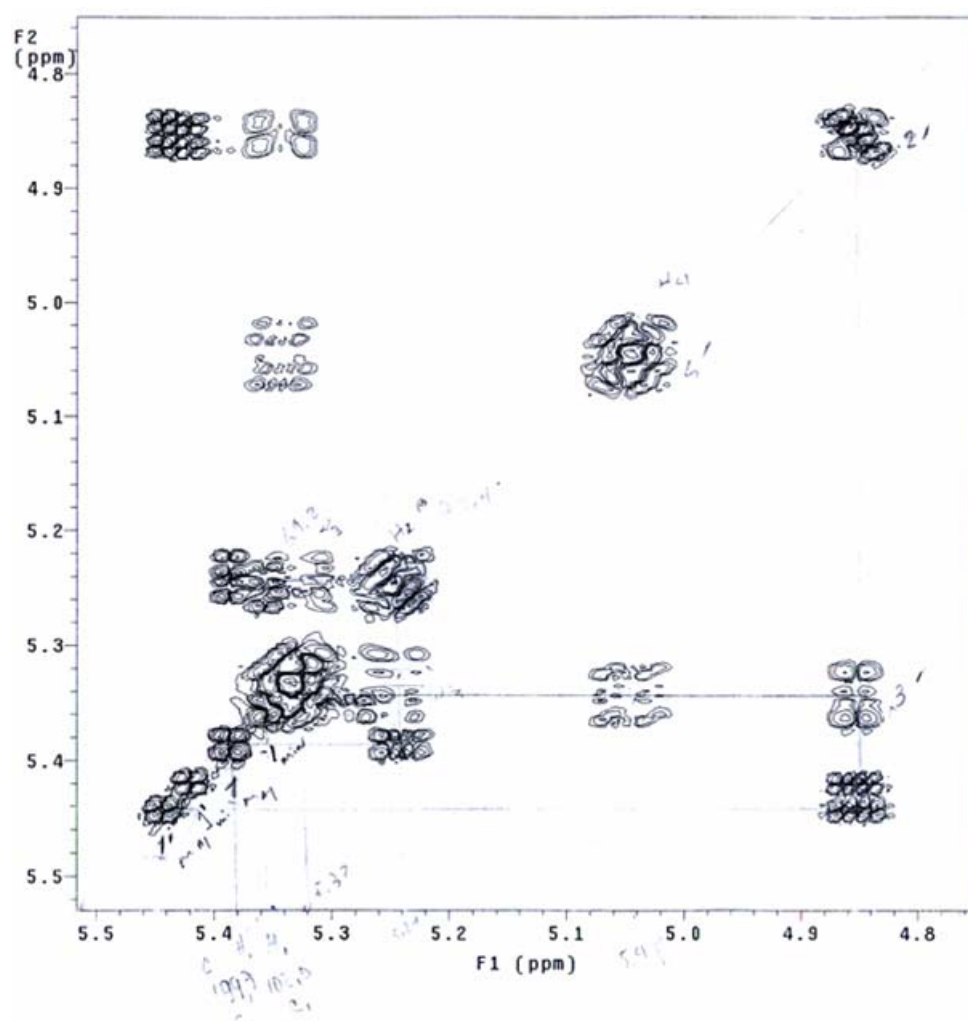


Figure S16. Details of the DQ-COSY NMR (CDCl₃, 400 MHz) spectrum of **2H**

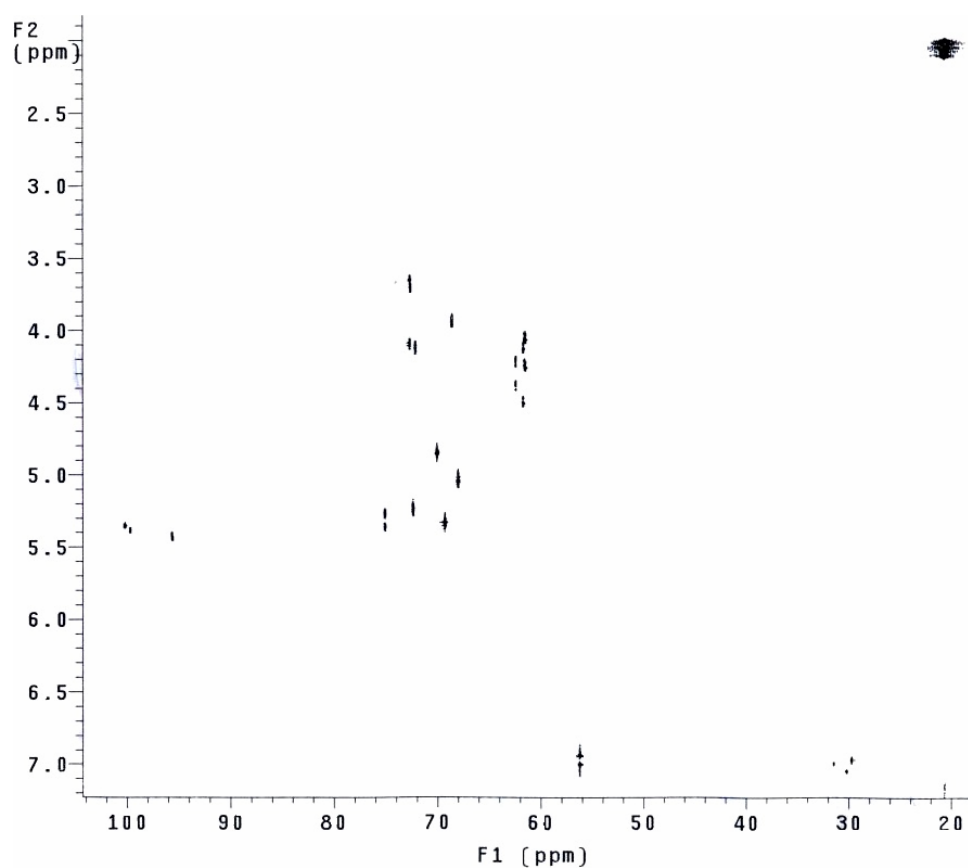


Figure S17. HSQC (CDCl_3 , 400 MHz) spectrum of **2H**

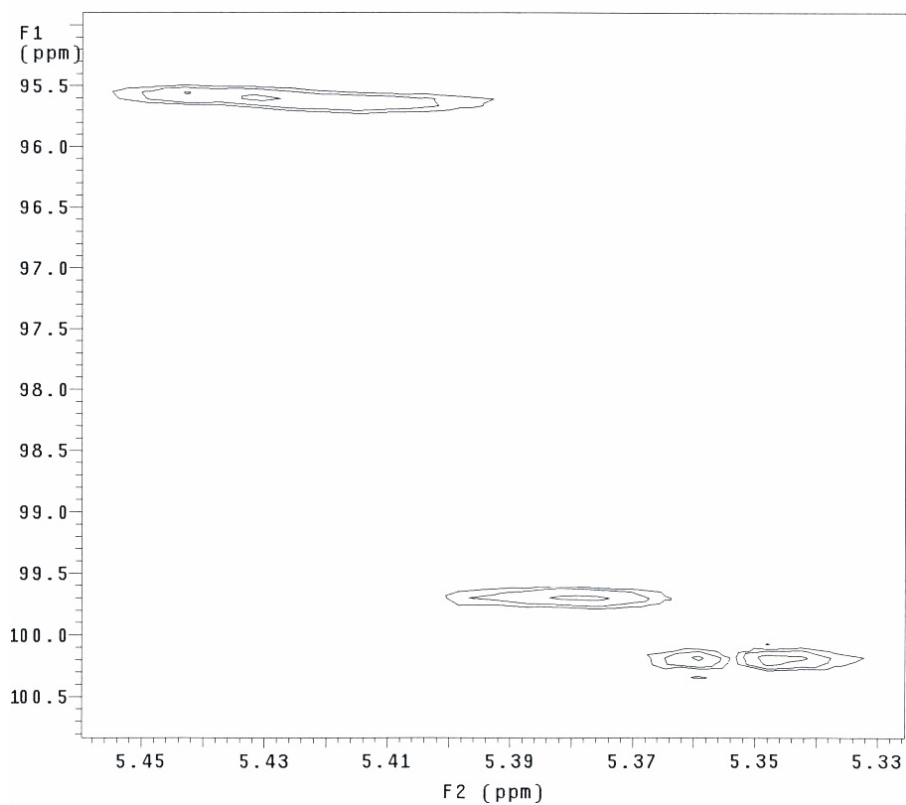


Figure S18. Details of the HSQC NMR (CDCl_3 , 400 MHz) spectrum of **2H**

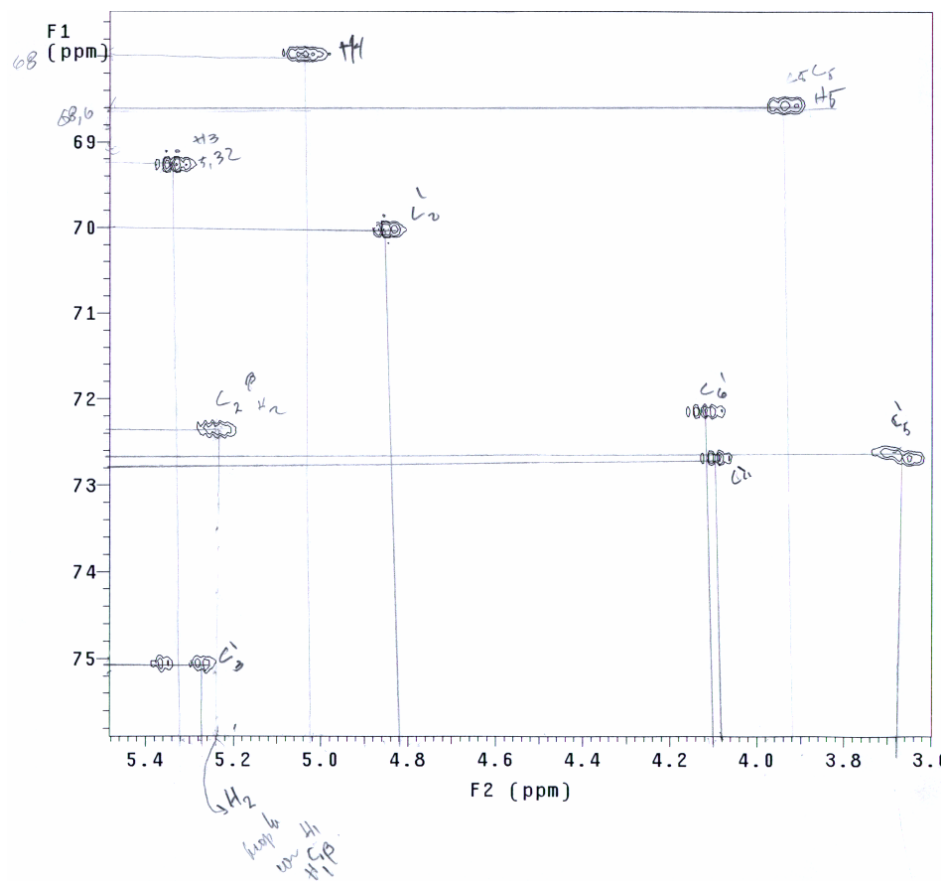


Figure S19. Details of the HSQC NMR (CDCl₃, 400 MHz) spectrum of **2H**

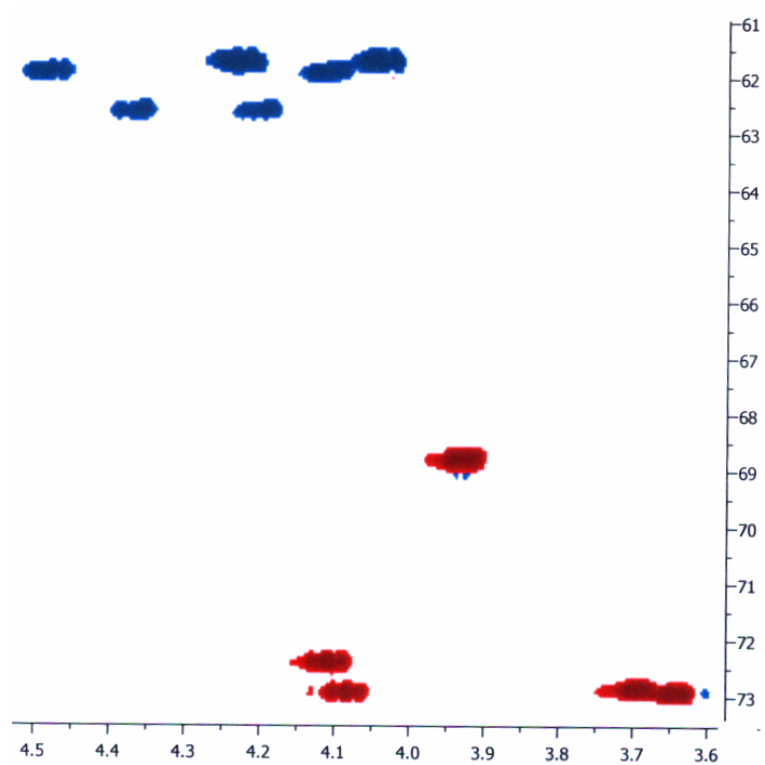


Figure S20. Details of the HSQC NMR (CDCl_3 , 400 MHz) spectrum of **2H**

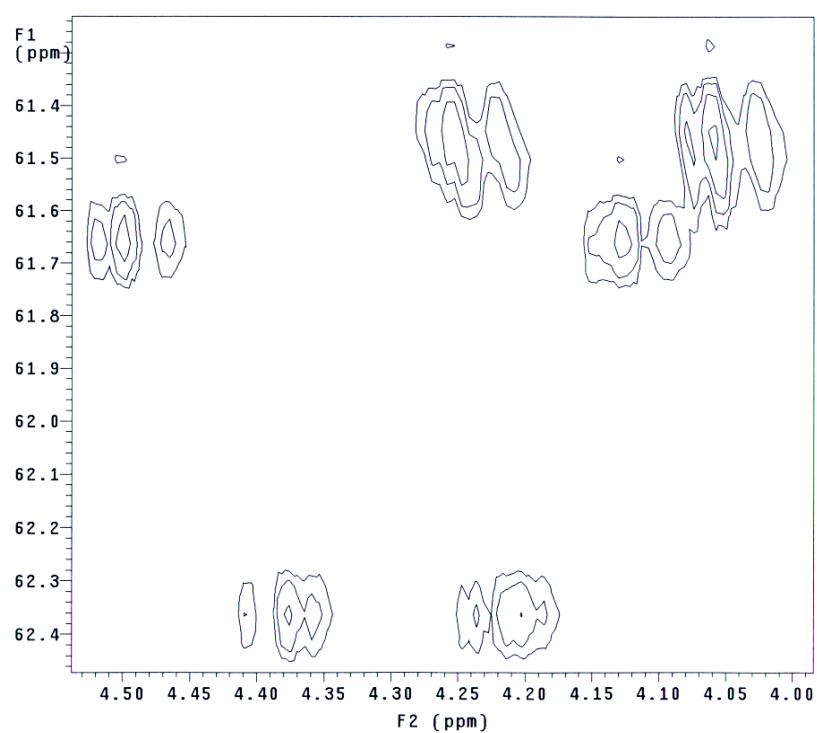


Figure S21. Details of the HSQC NMR (CDCl_3 , 400 MHz) spectrum of **2H**

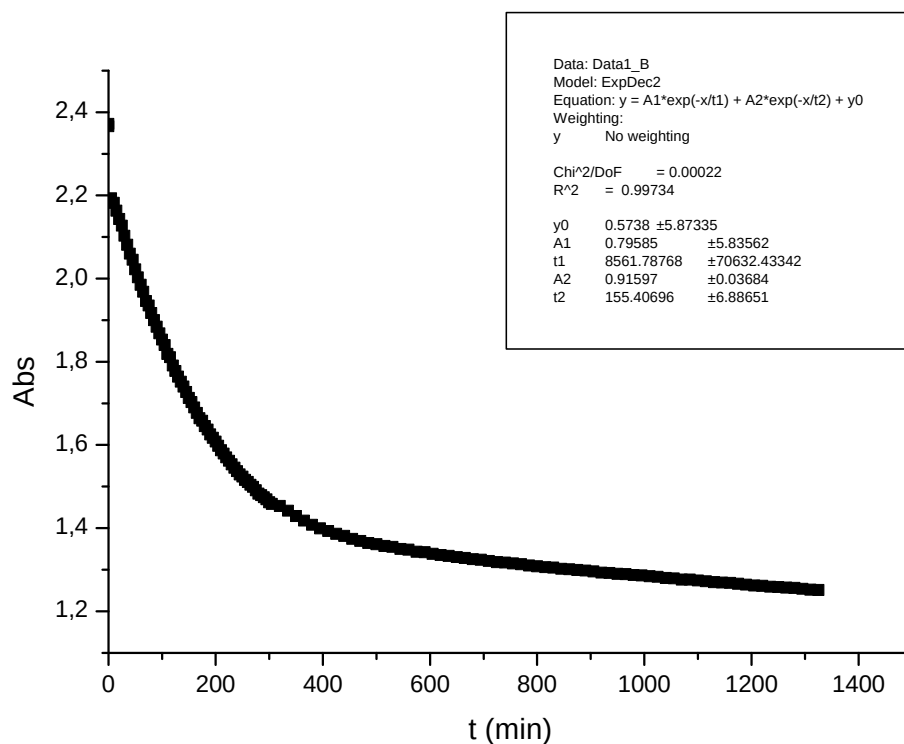


Figure S22. Decay of the UV absorbance (384 nm) of 2 (86 μ M) solution in THF/H₂O 10:1 v/v (rt) following addition of ascorbic acid (19 μ M)

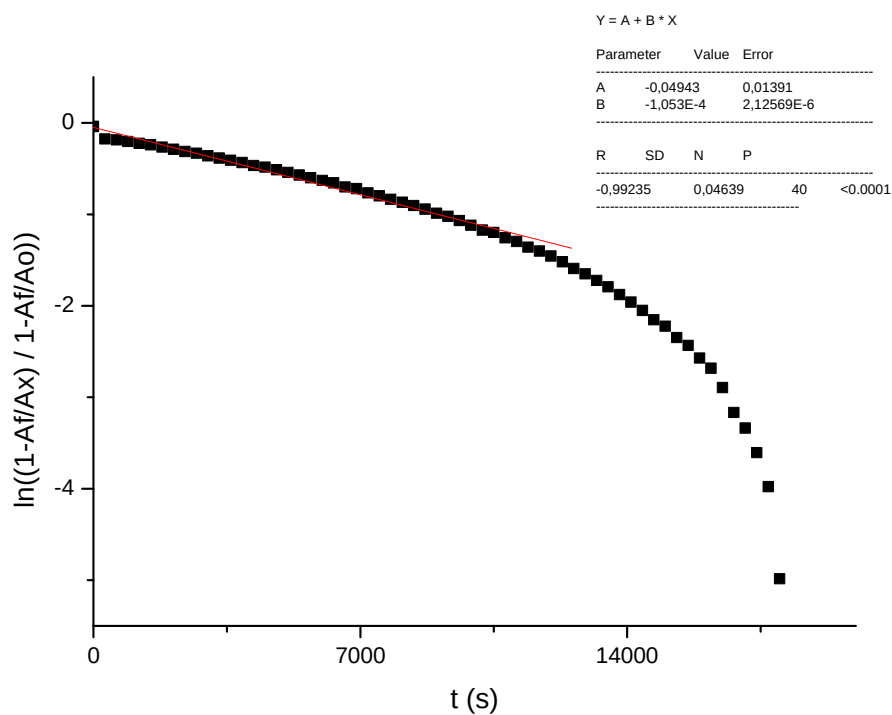


Figure S23. Estimation of the rate constant (k_1) using equation 2 in the manuscript.