

Coinage Metals-Catalyzed Cascade Reactions of Aryl Alkynylaziridines: Silver(I)-Single vs Gold(I)-Double Cyclizations

Nicolas Kern, Aurélien Blanc*, Solène Miaskiewicz, Michelle Robinette, Jean-Marc Weibel and Patrick Pale*

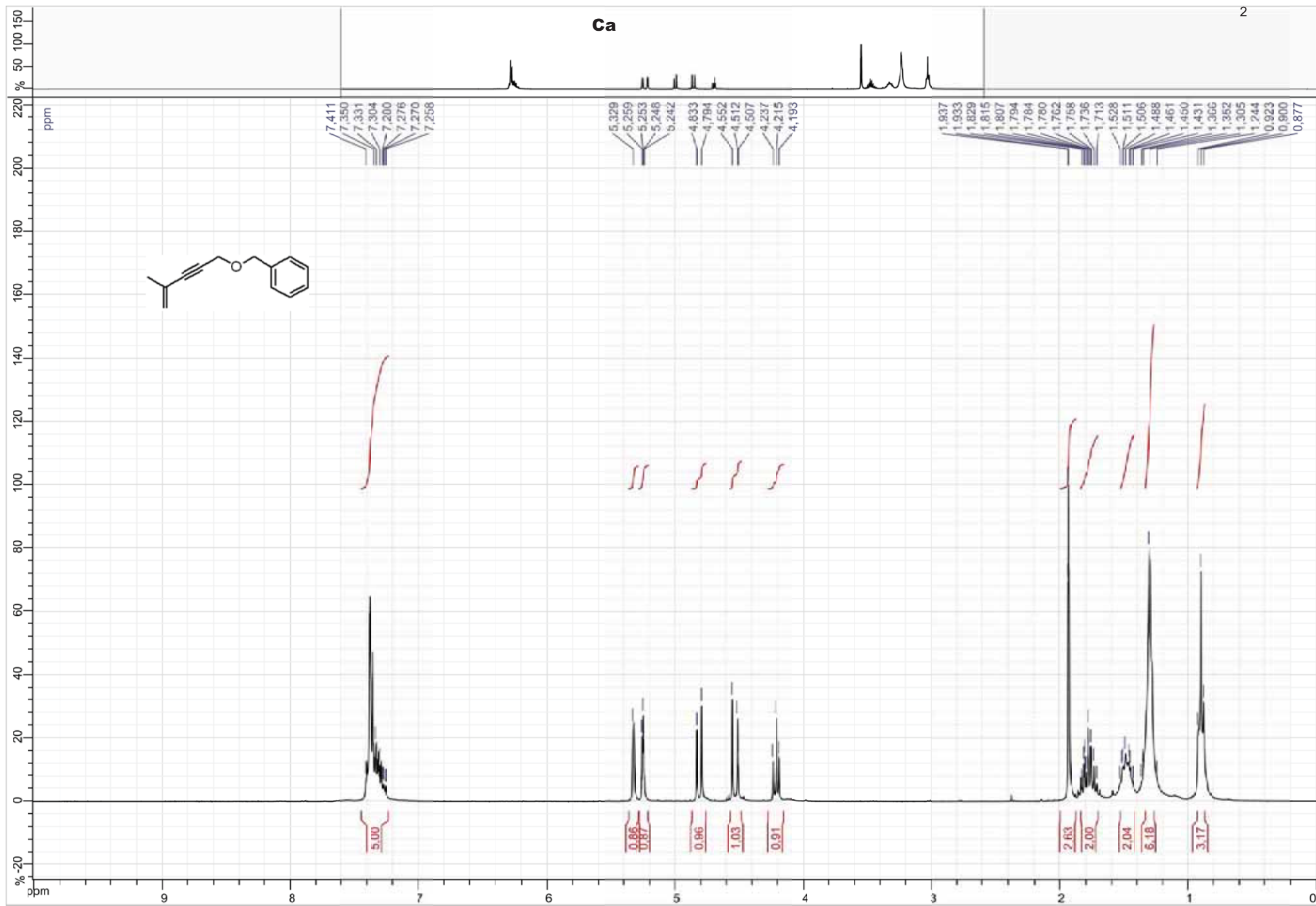
Laboratoire de Synthèse et Réactivité Organiques, Institut de Chimie, CNRS et Université de Strasbourg, 4 rue Blaise Pascal, 67070 Strasbourg France

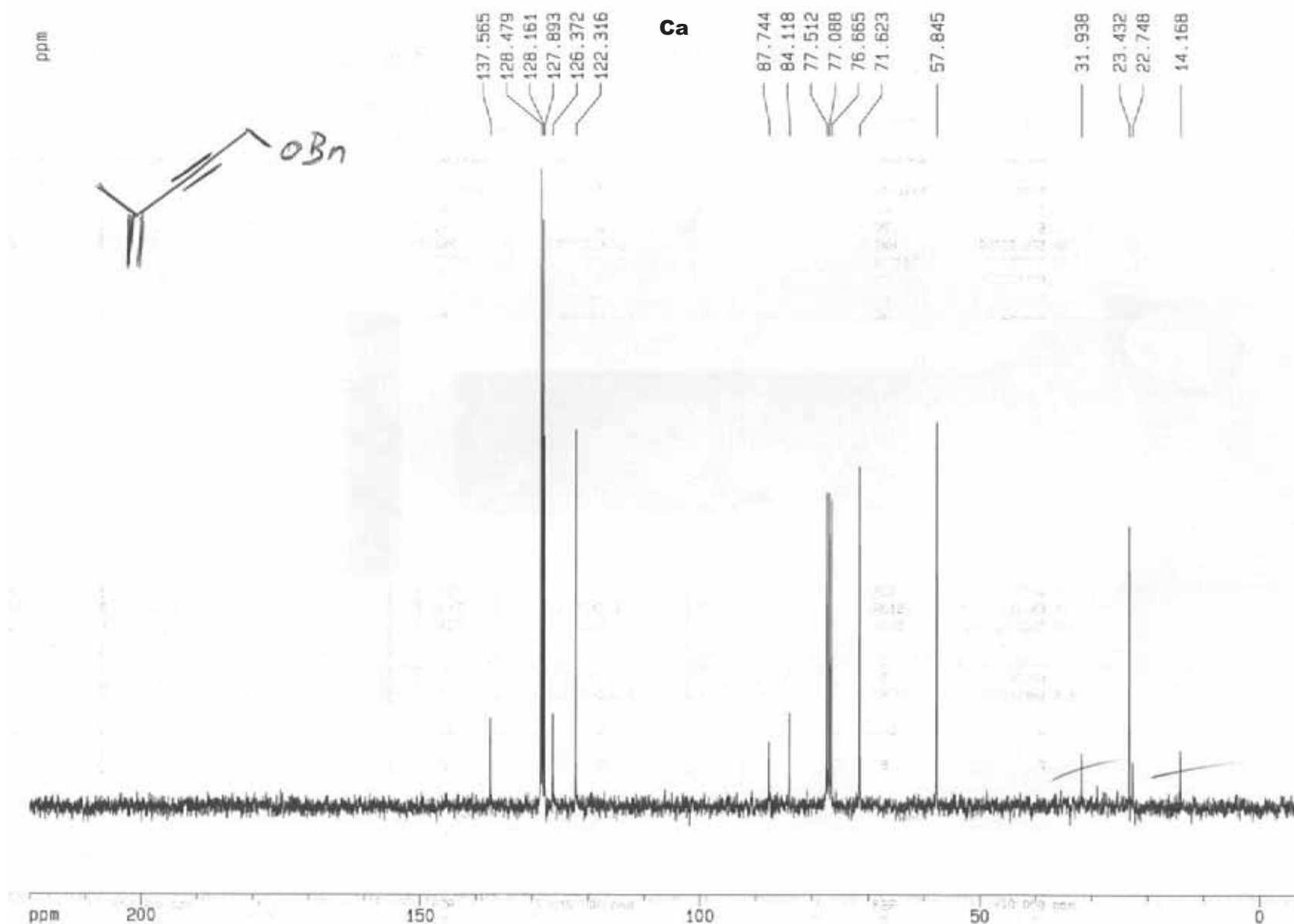
Supporting Information

Proton and Carbon NMR Spectra S2

Figure 2. ¹H NMR Spectra of the transient intermediates **3a** in gold(I)-catalyzed formation of spiro[isochroman-4,2-pyrroline] **2a** S157

X-ray structure and crystallographic data for **2h-maj** S158





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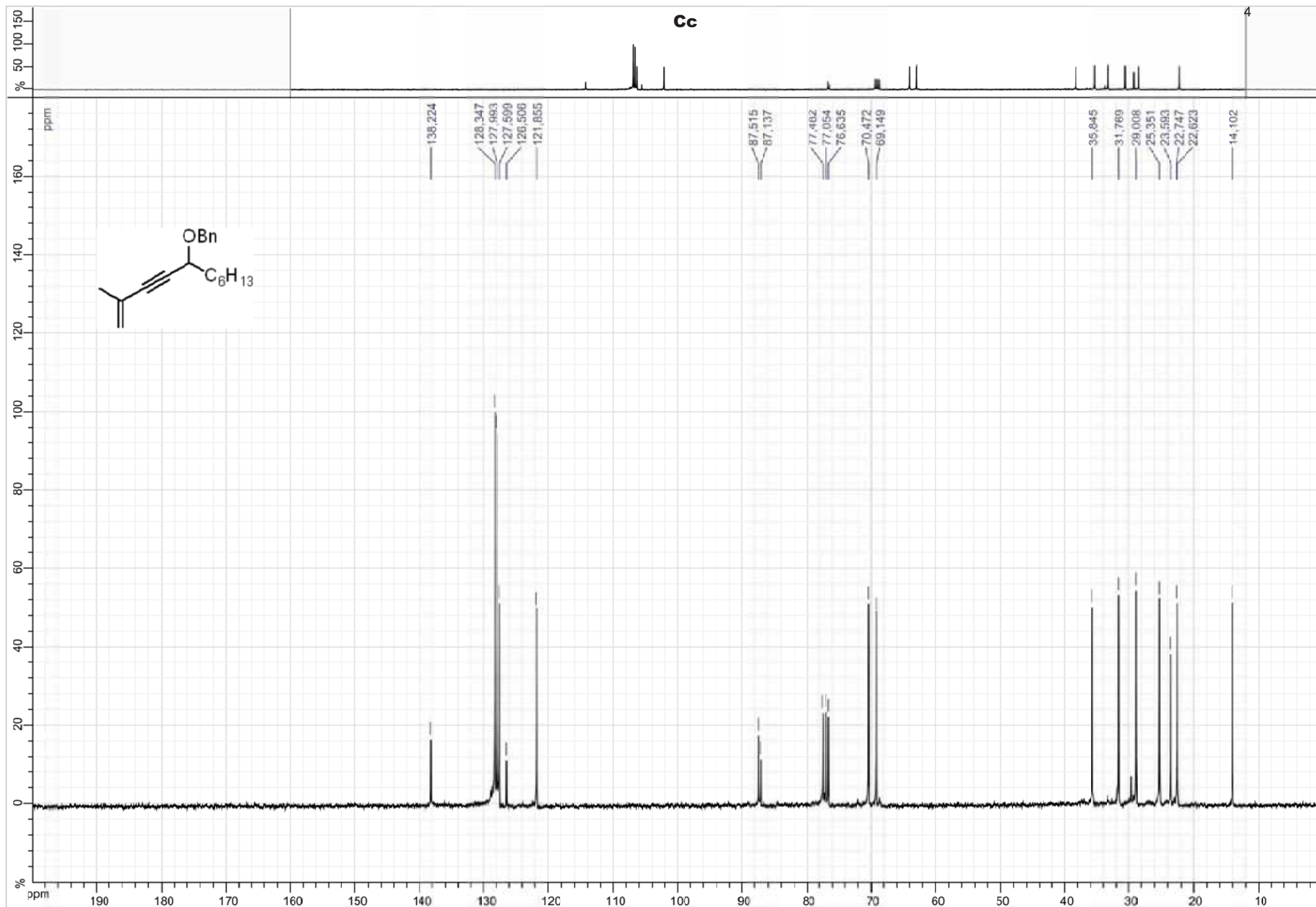
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FIDRES 0.620276 Hz
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RG 16384
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DE 10.00 usec
TE 300.0 K
D1 0.80000001 sec
d11 0.03000000 sec

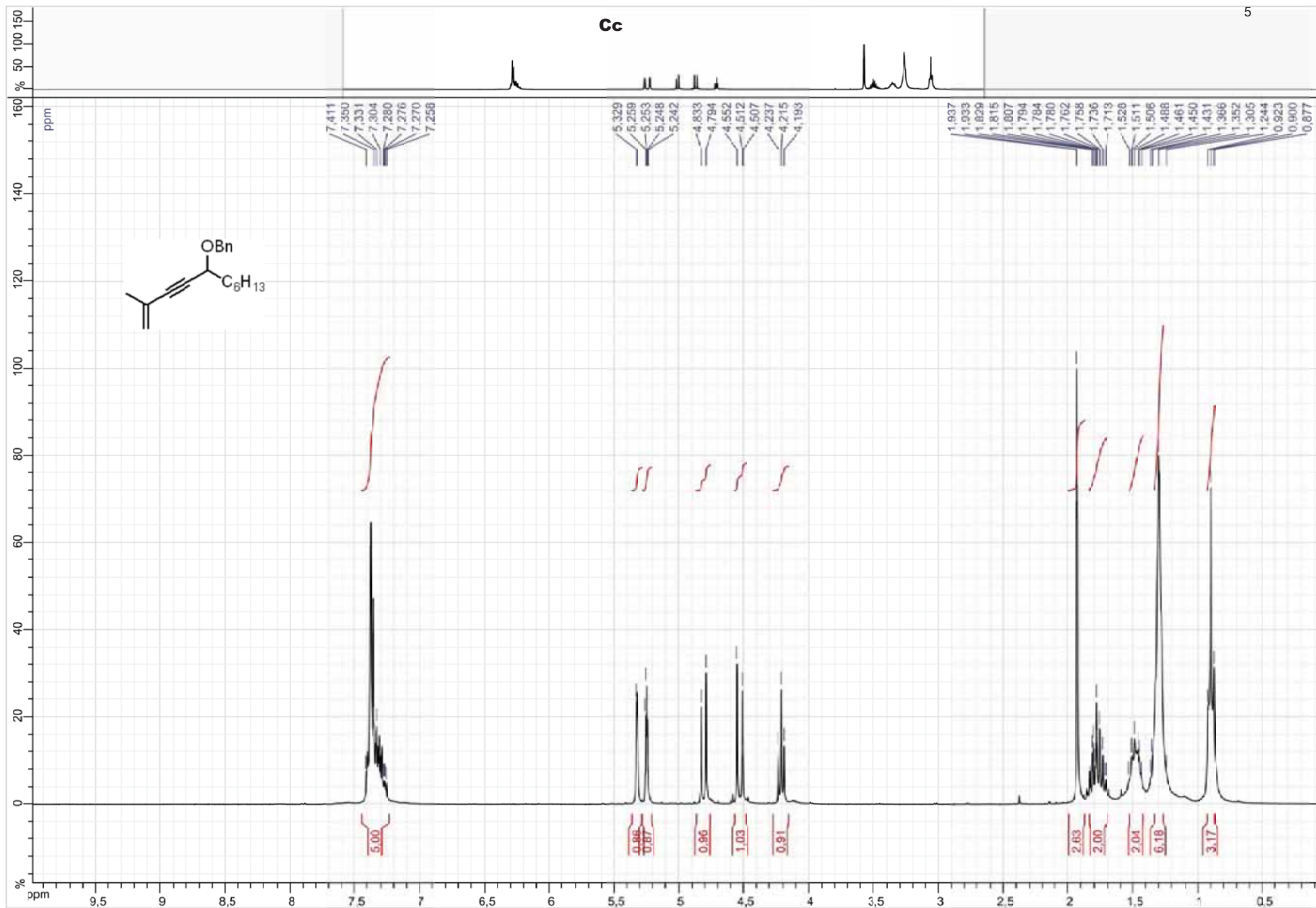
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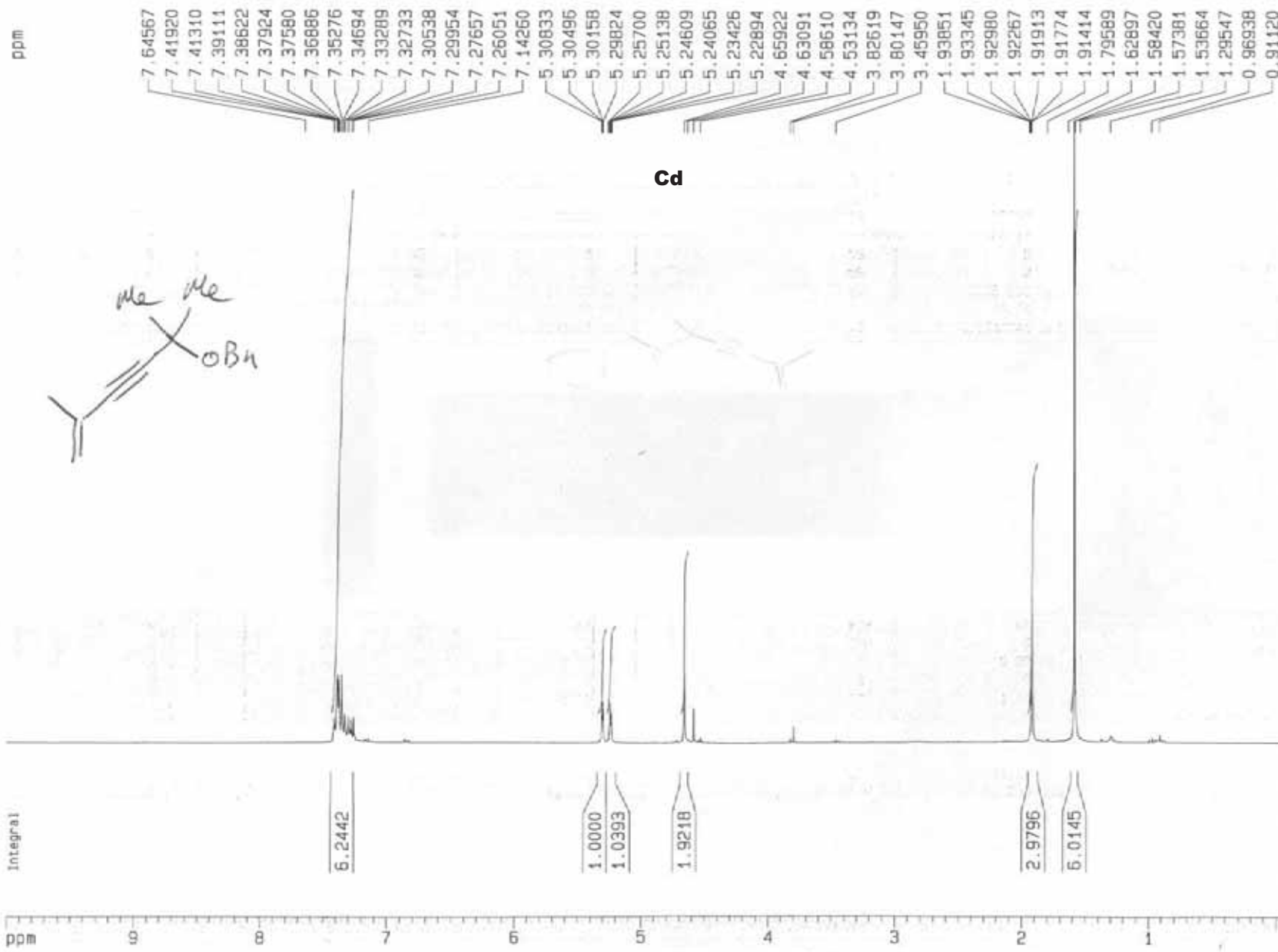
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F2 - Processing parameters
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WDW EM
SSB 0
LB 2.00 Hz
GB 0
PC 1.40

1D NMR plot parameters
CX 22.00 cm
CY 11.00 cm
F1P 220.000 ppm
F1 16602.90 Hz
F2P -10.000 ppm
F2 -754.68 Hz
PPNCH 10.45455 ppm/cm
HZCM 786.98106 Hz/cm







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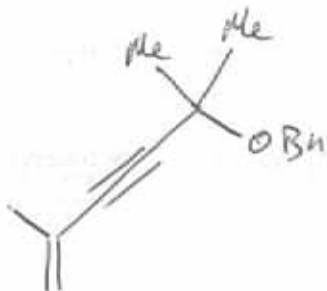
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FIDRES 0.119209 Hz
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DE 10.00 usec
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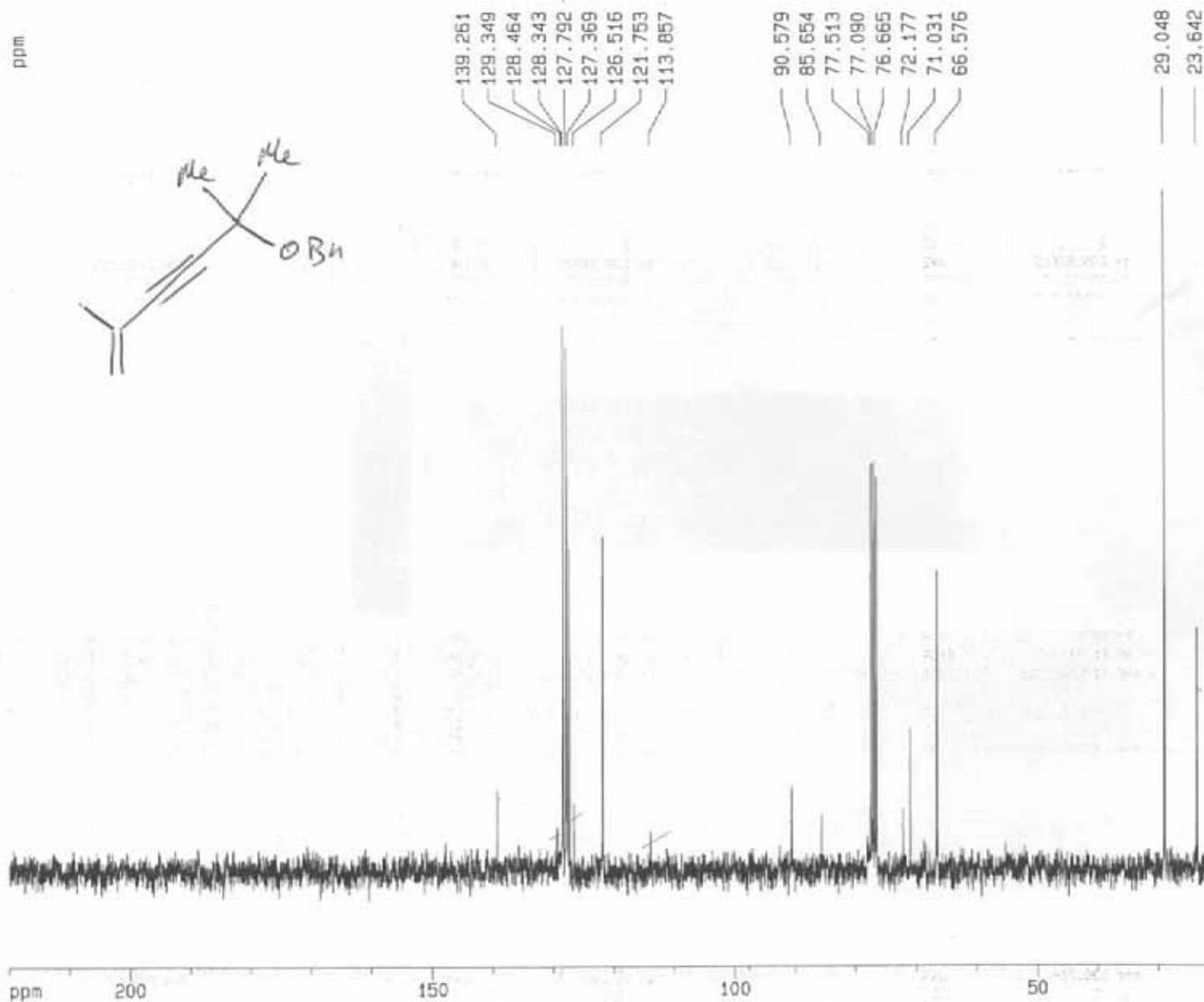
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1D NMR plot parameters
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F1 3001.30 Hz
F2P -0.100 ppm
F2 -30.01 Hz
PPMCM 0.45909 ppm/cm
HZCM 137.78696 Hz/cm

ppm



Cd



Current Data Parameters
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 EXPNO 11
 PROCNO 7 1

F2 - Acquisition Parameters

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 SWH 20325.203 Hz
 FIDRES 0.620275 Hz
 AQ 0.8061428 sec
 RG 16364
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 DE 10.00 usec
 TE 300.0 K
 D1 0.80000001 sec
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***** CHANNEL f1 *****

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***** CHANNEL f2 *****

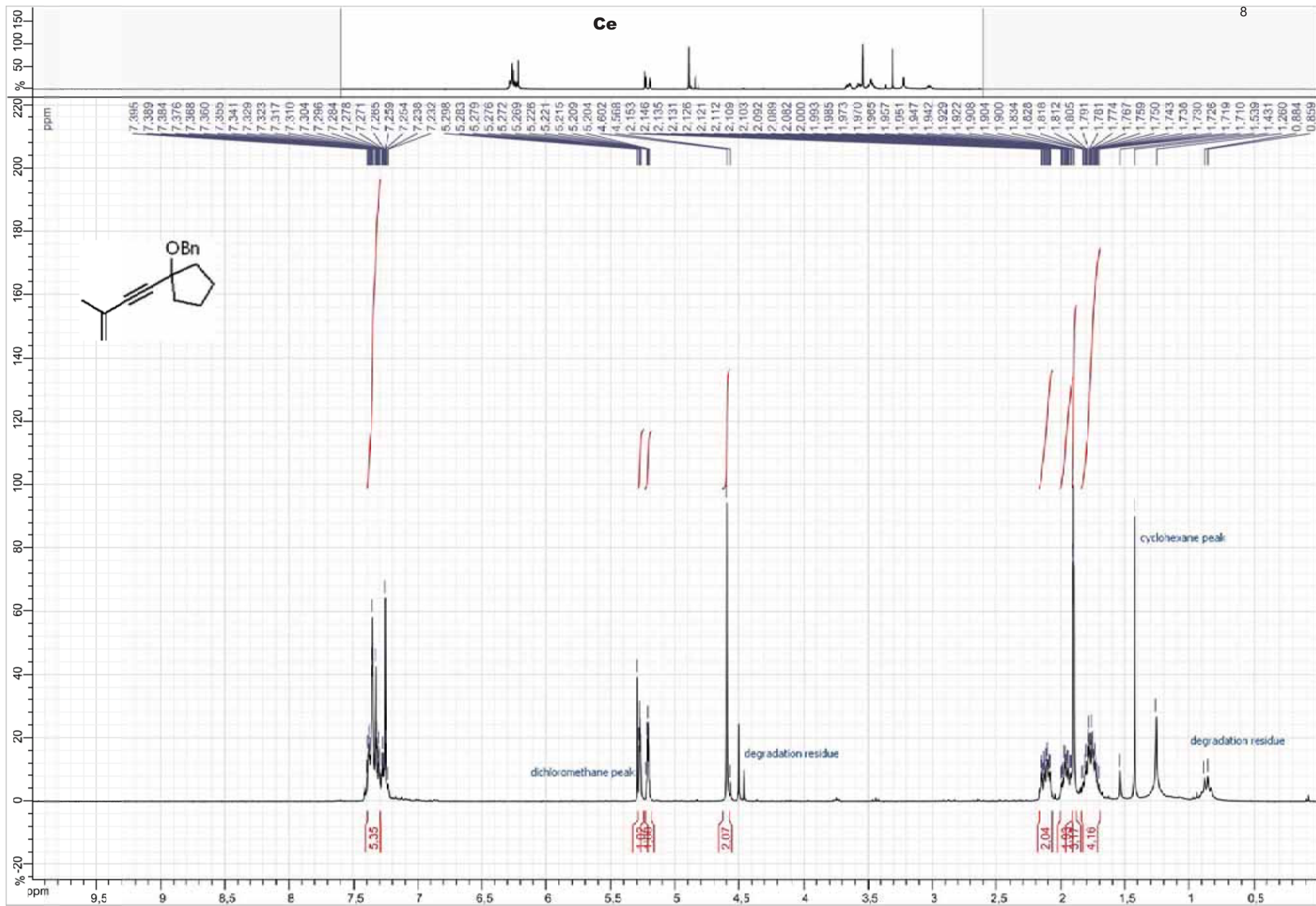
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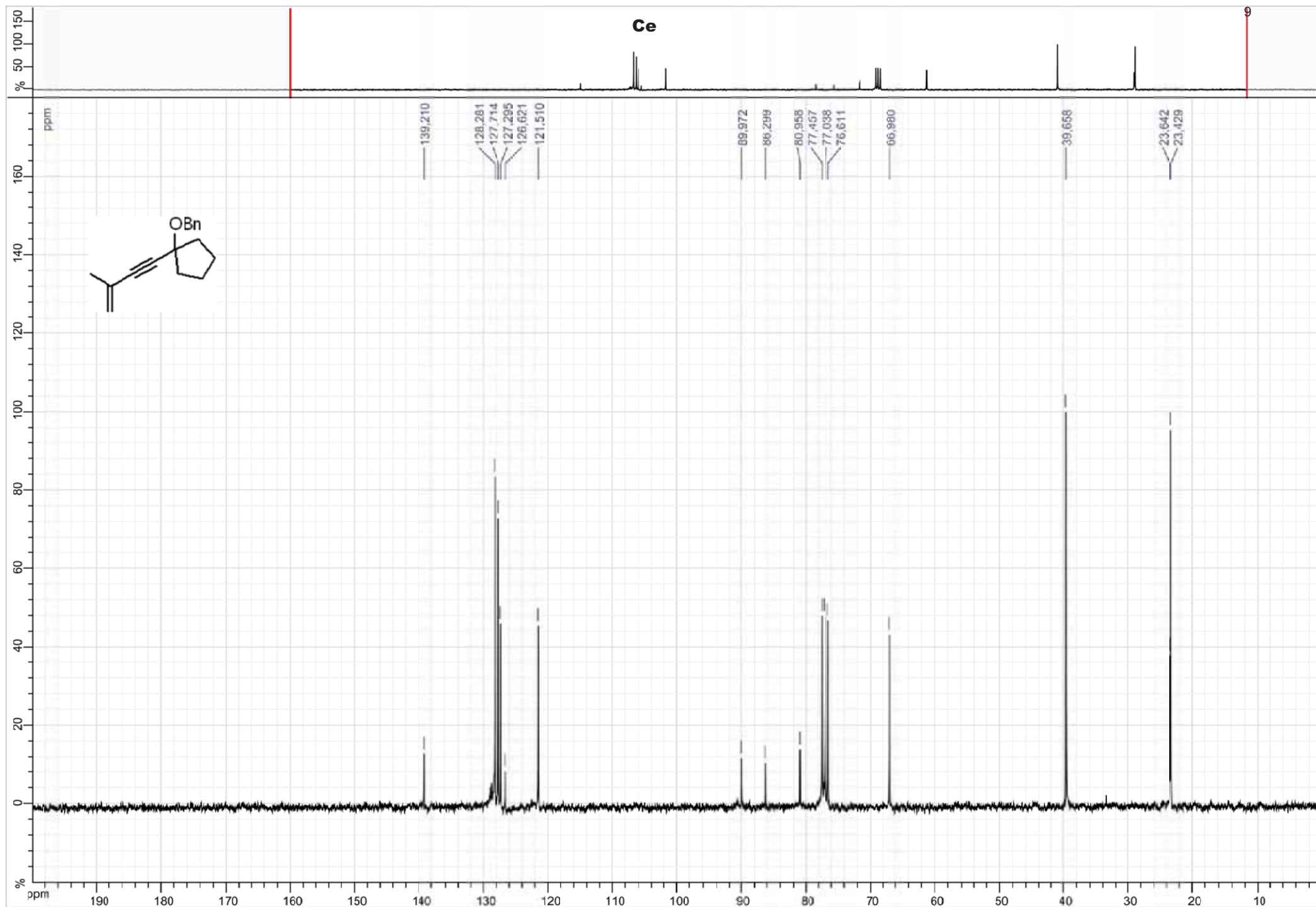
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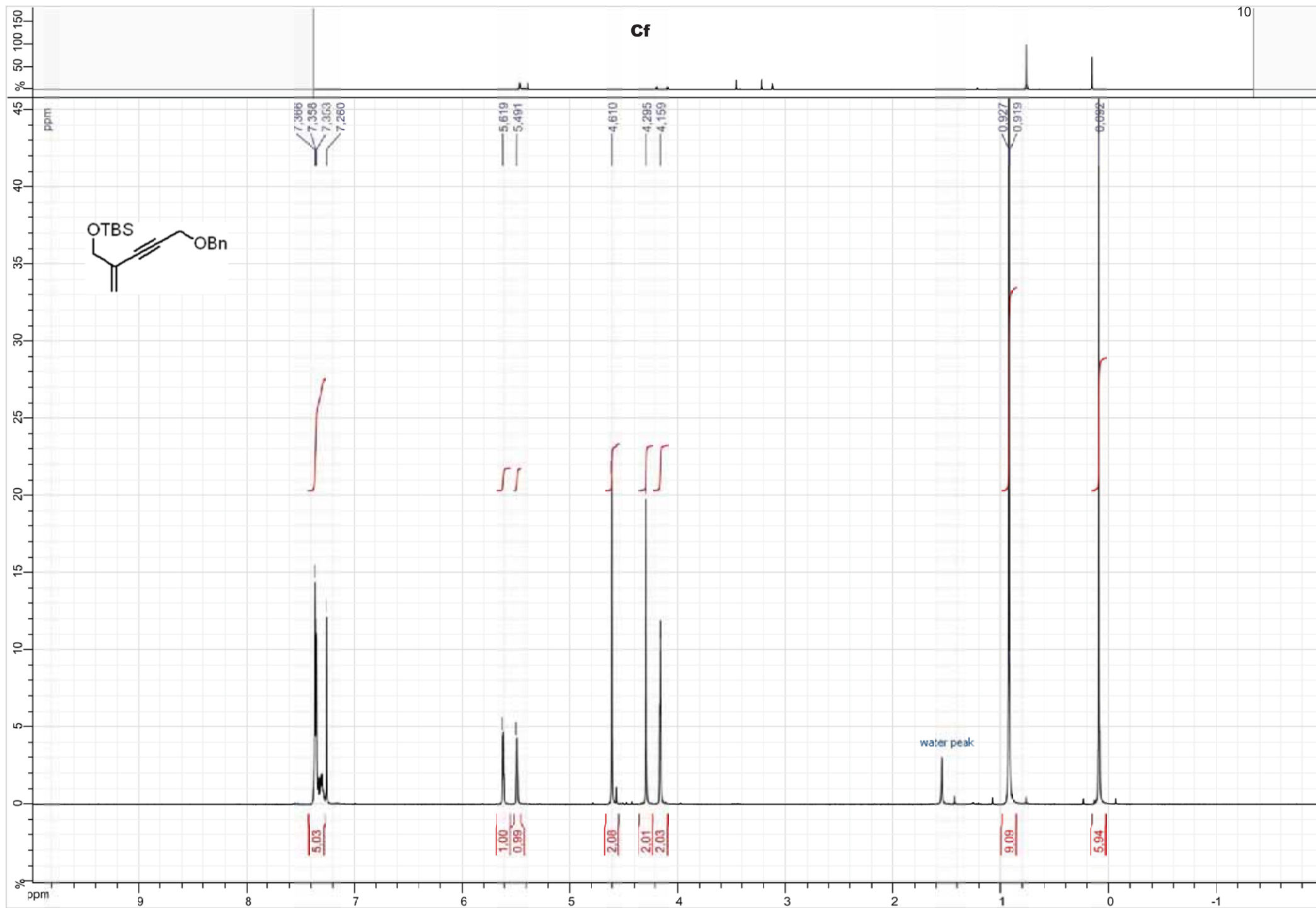
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 GB 0
 PC 1.40

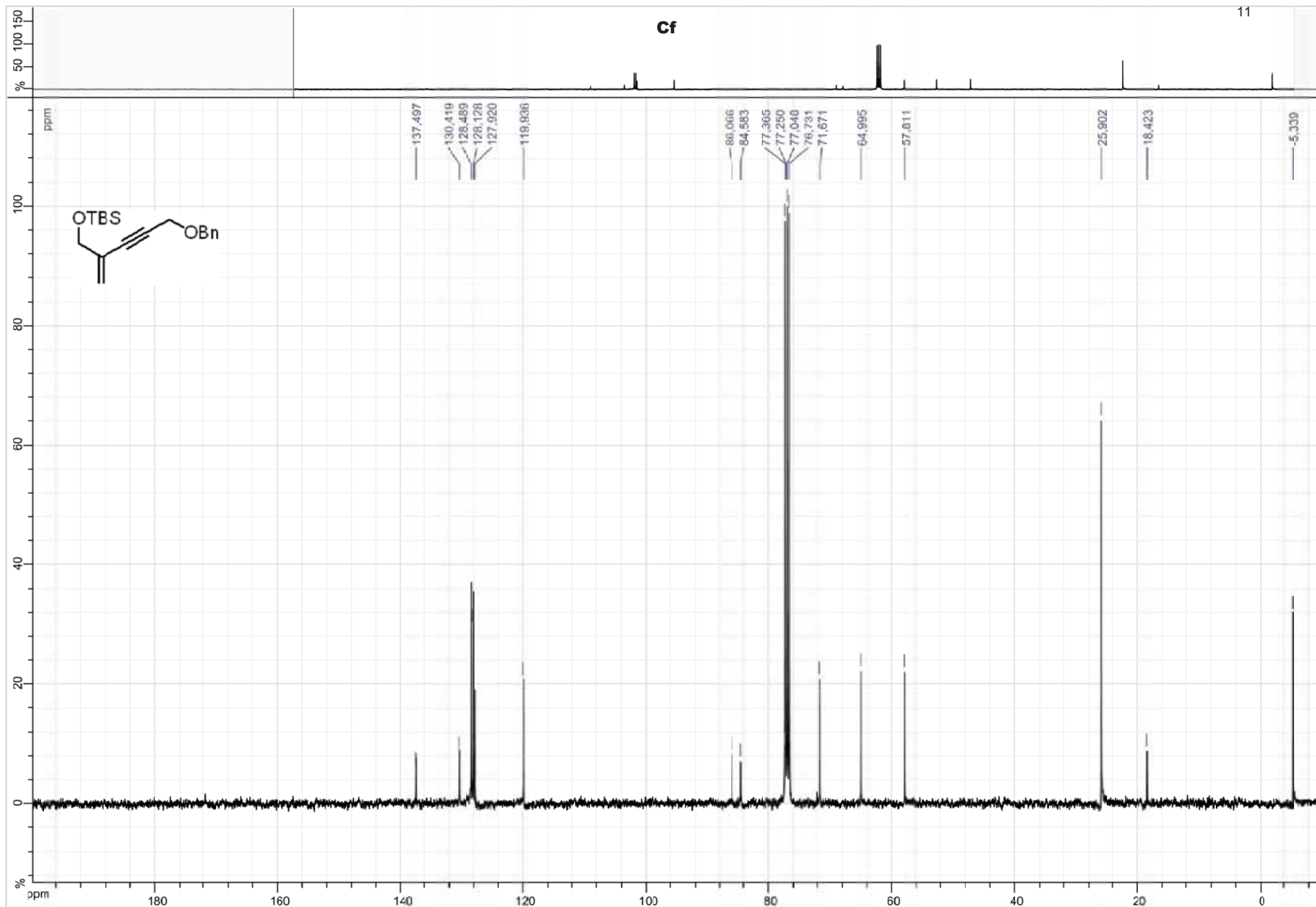
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CX 22.00 cm
 CY 11.00 cm
 F1P 220.000 ppm
 F1 16502.90 Hz
 F2P -10.000 ppm
 F2 -754.68 Hz
 PPMCM 10.45455 ppm/cm
 HZCM 786.98106 Hz/cm

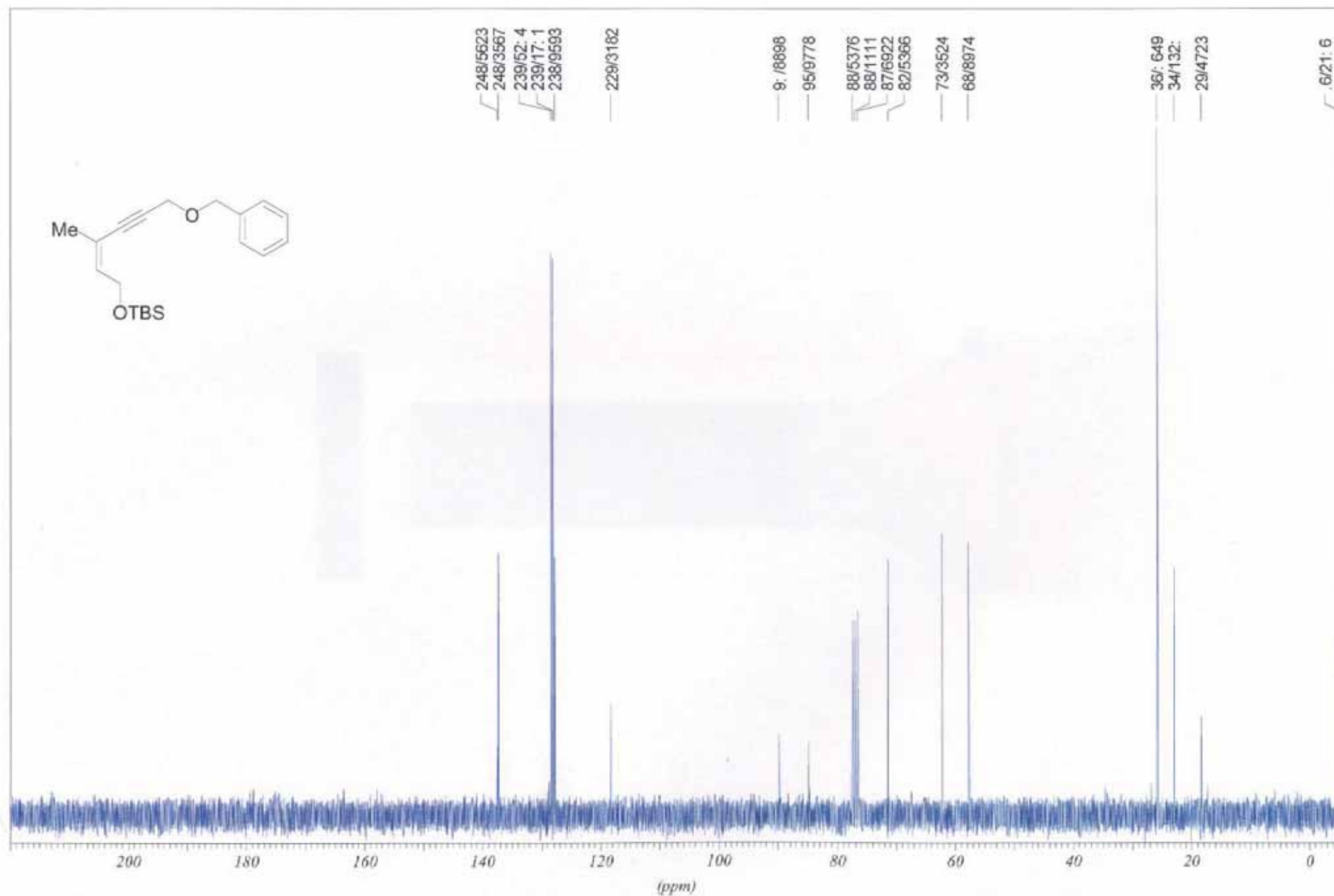


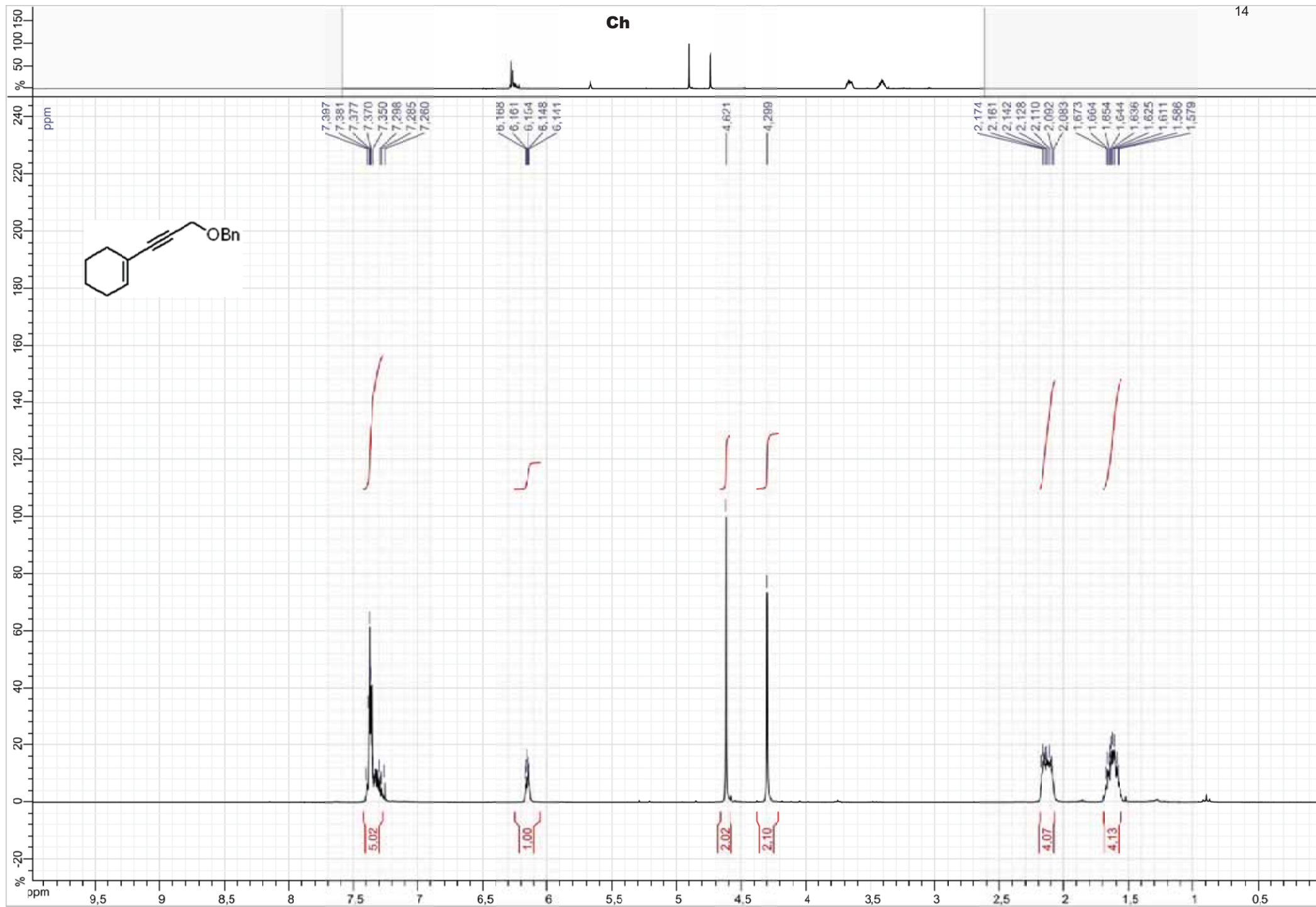


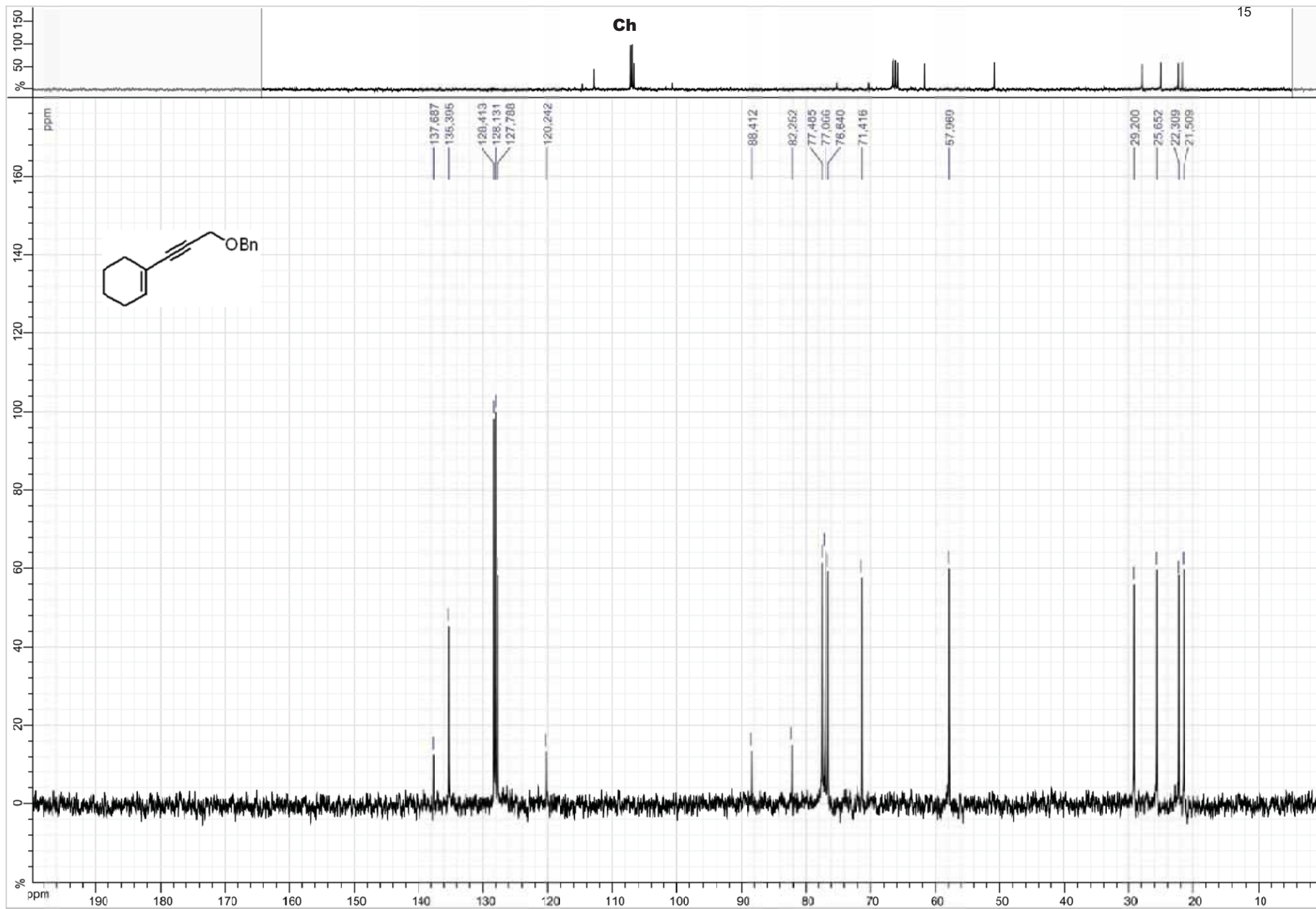


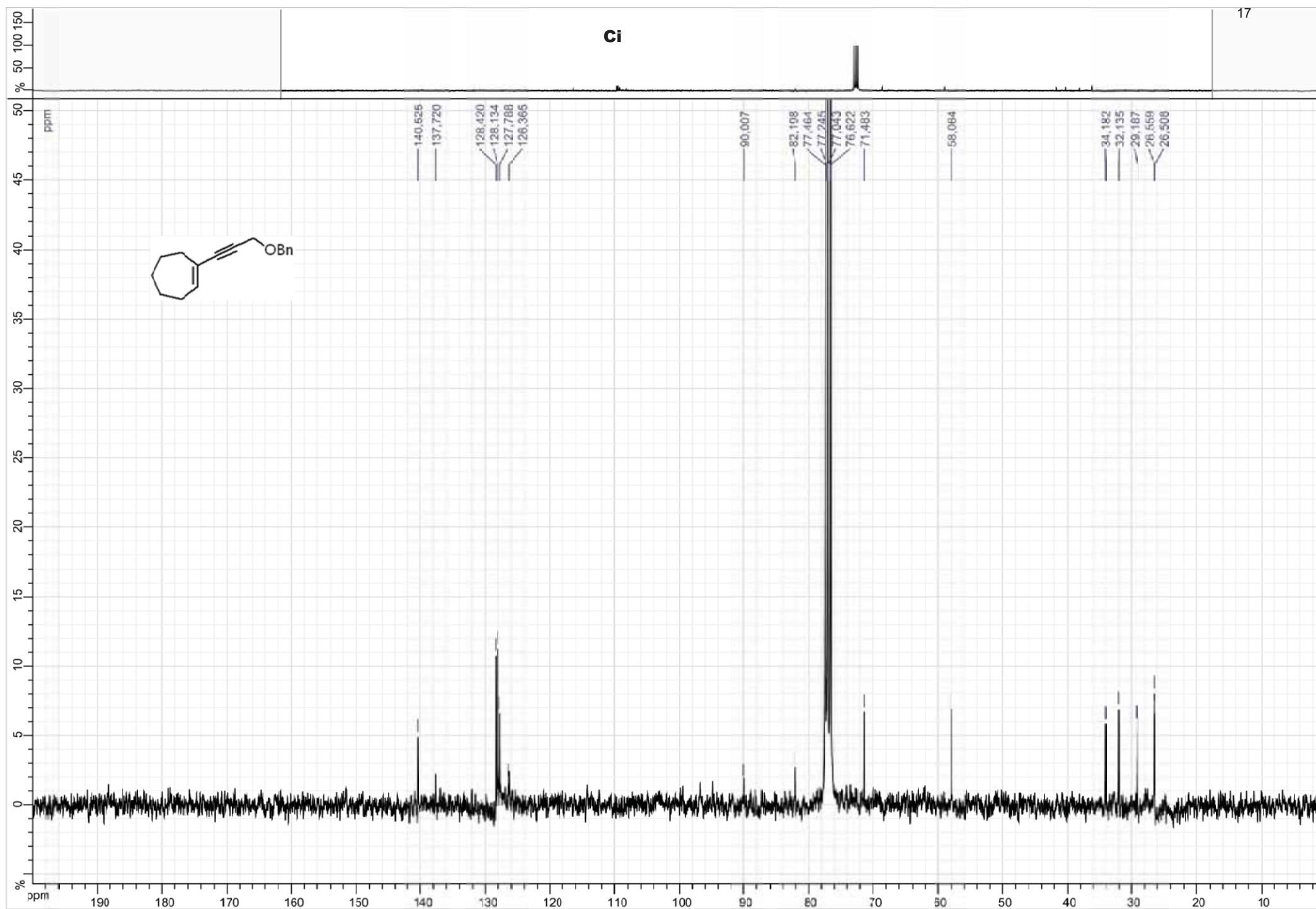






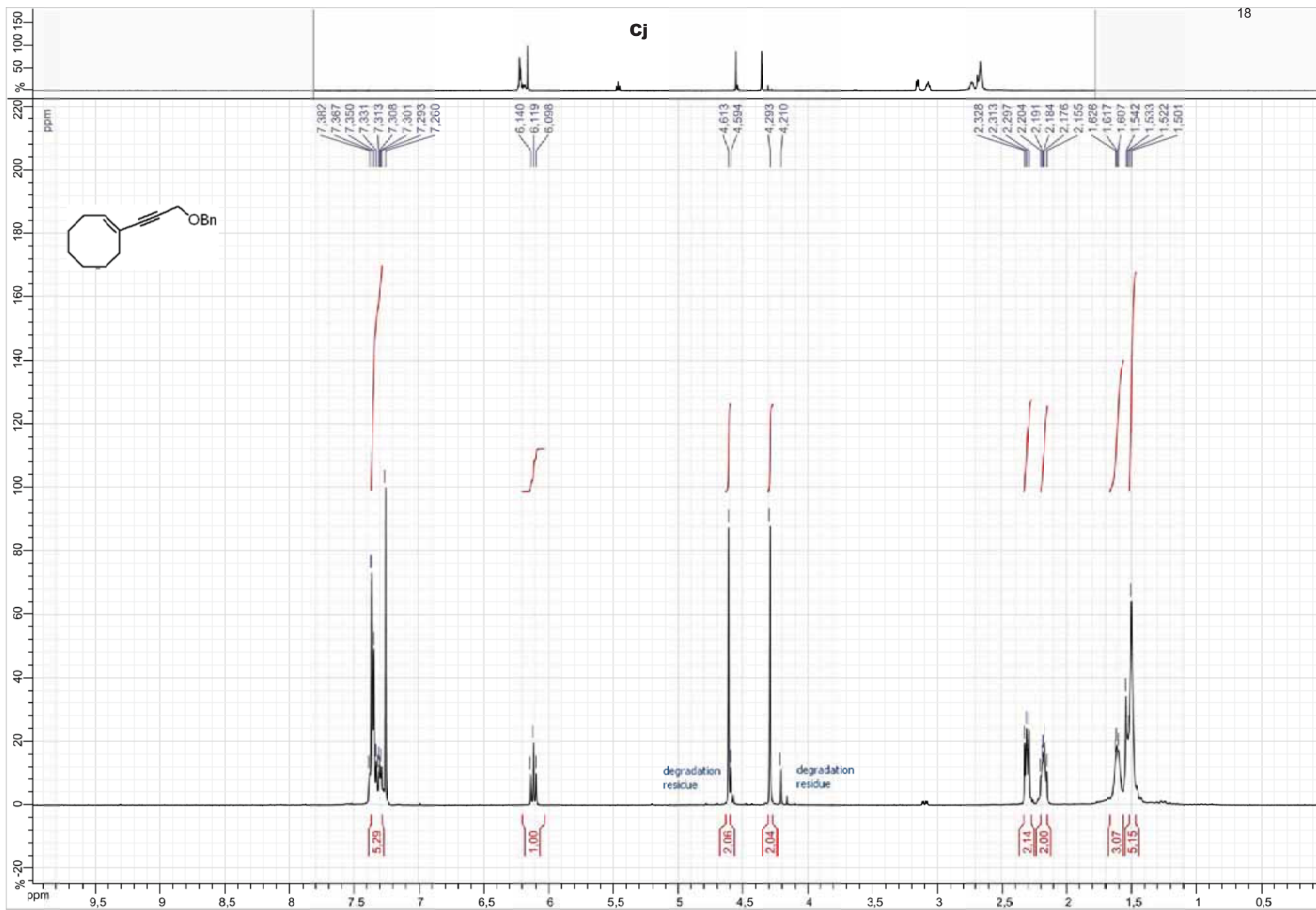


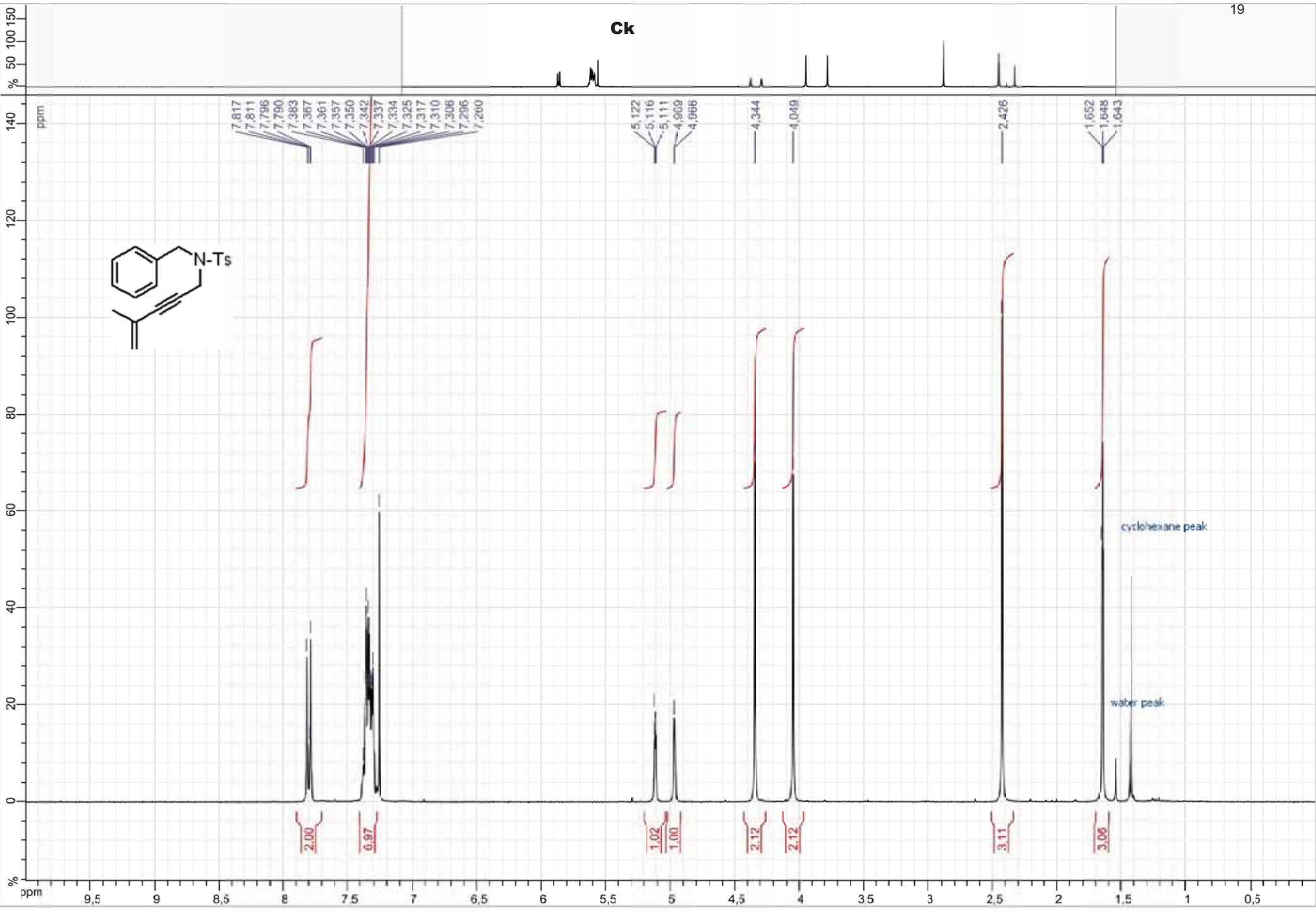


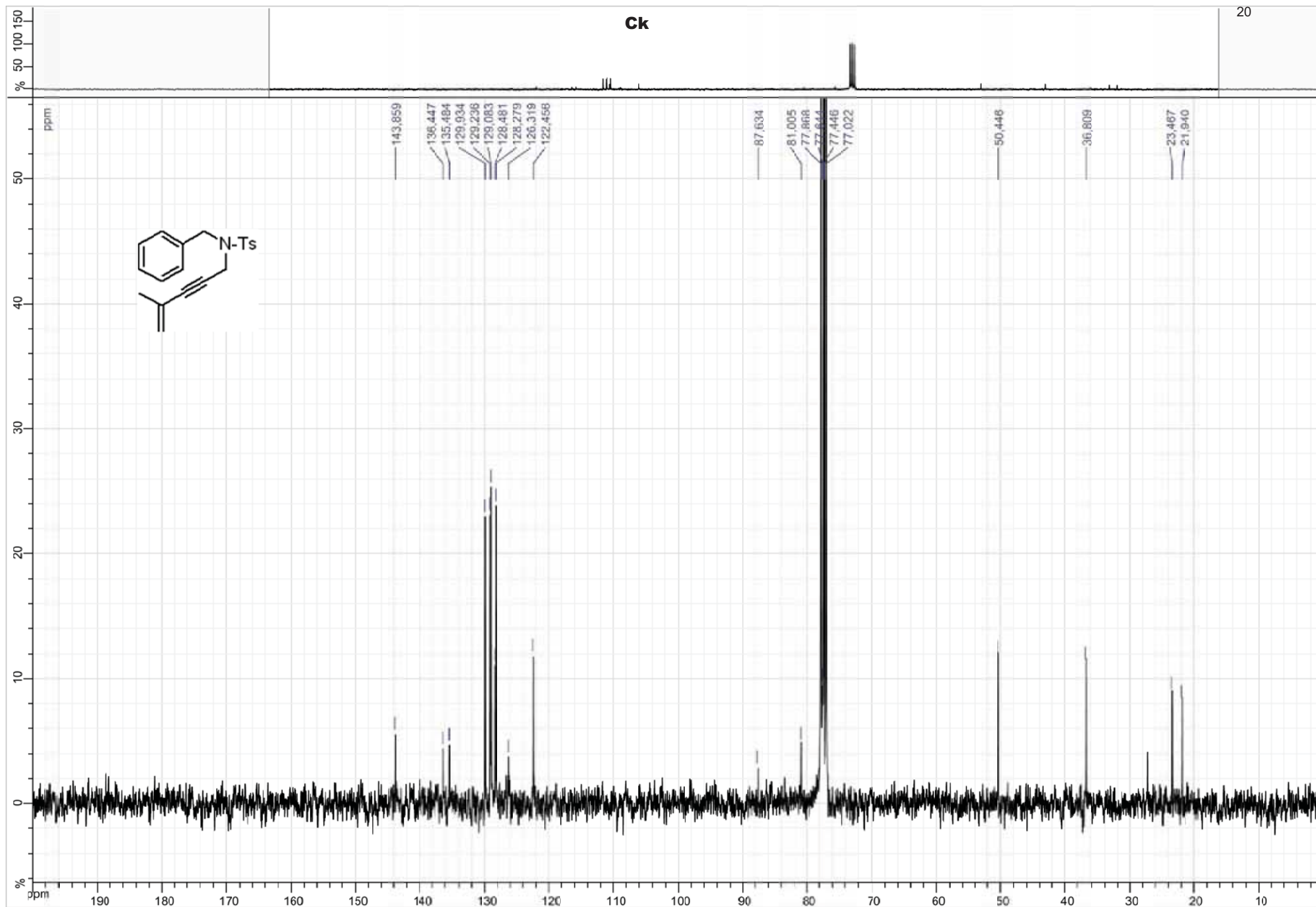


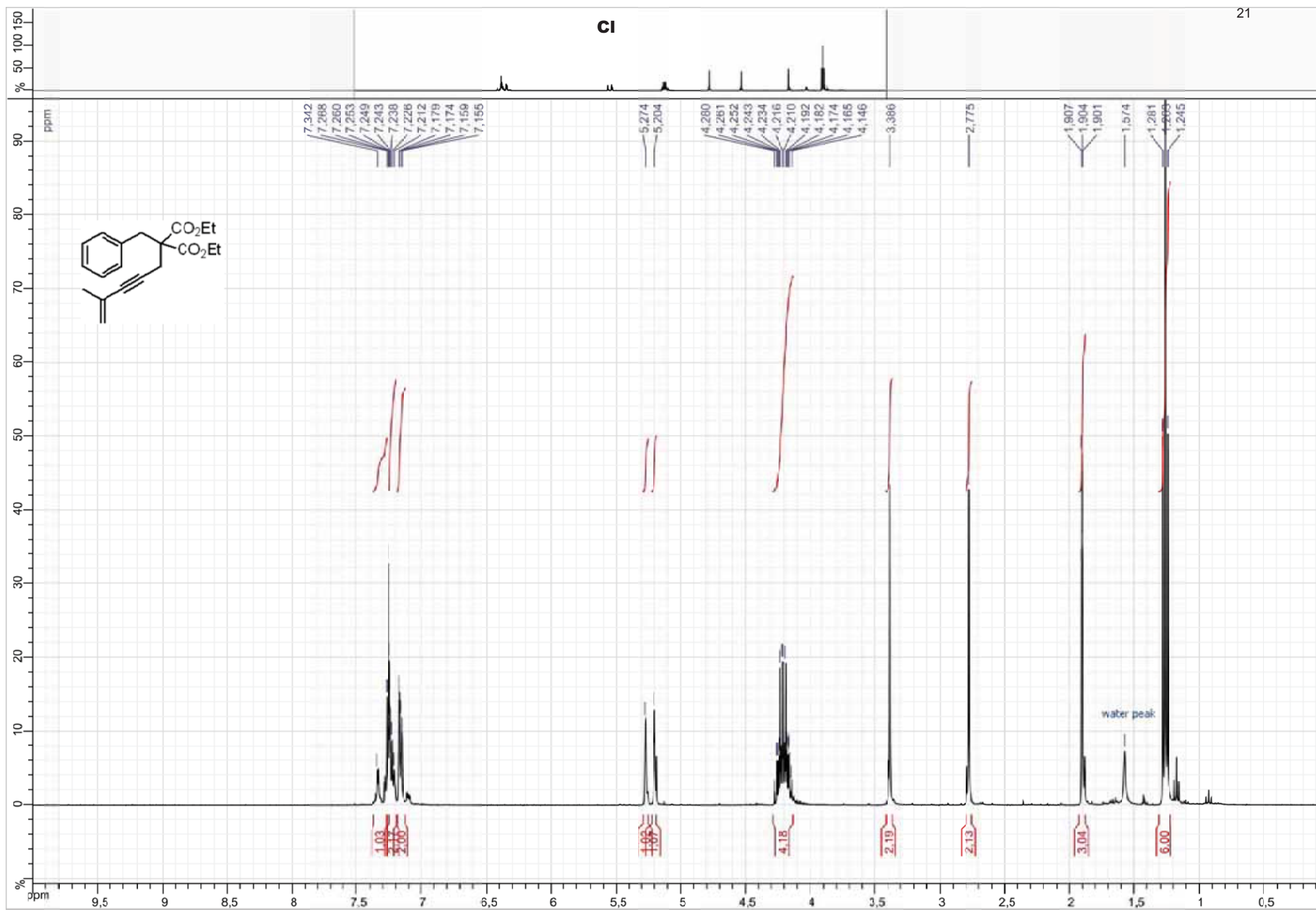
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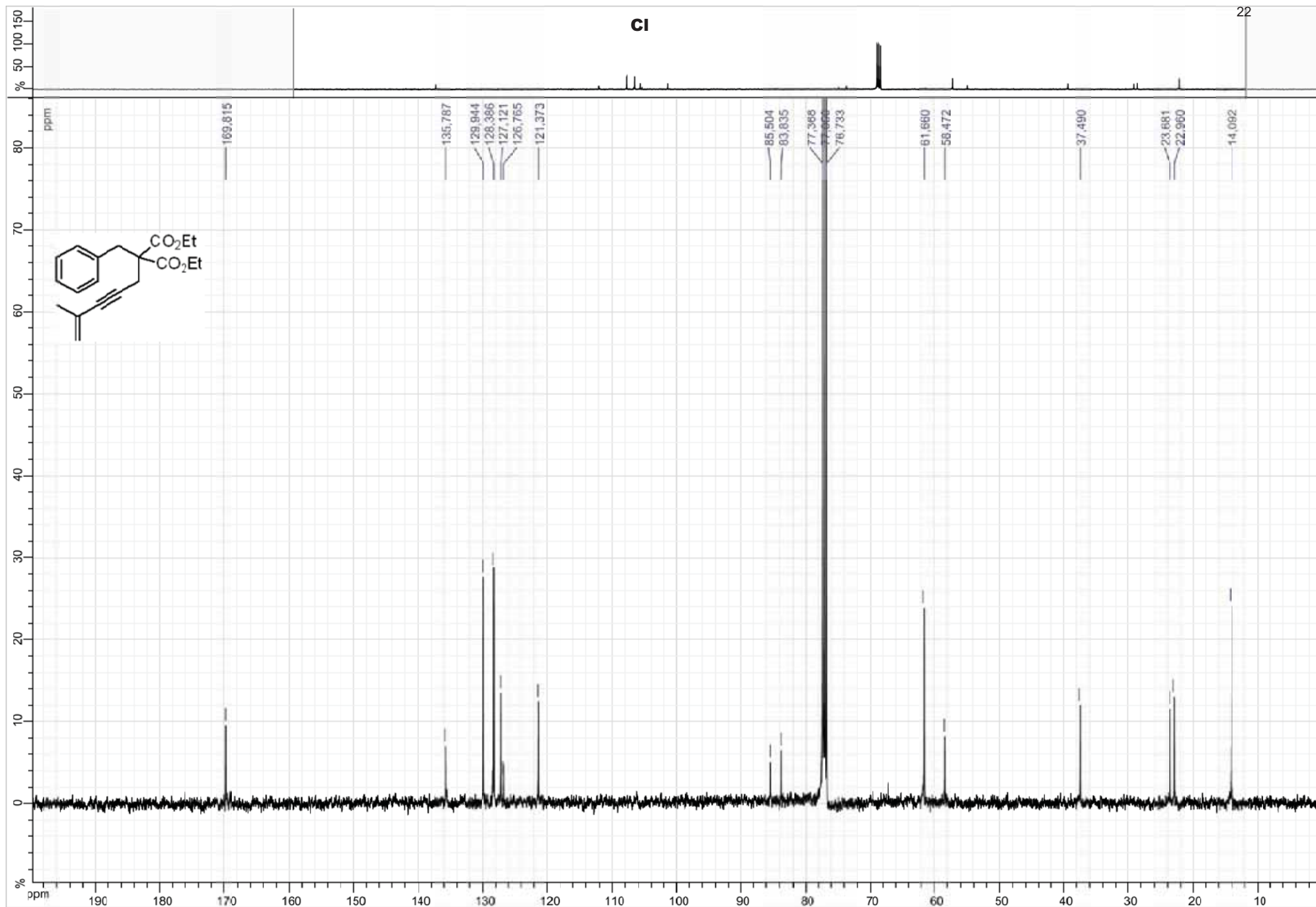
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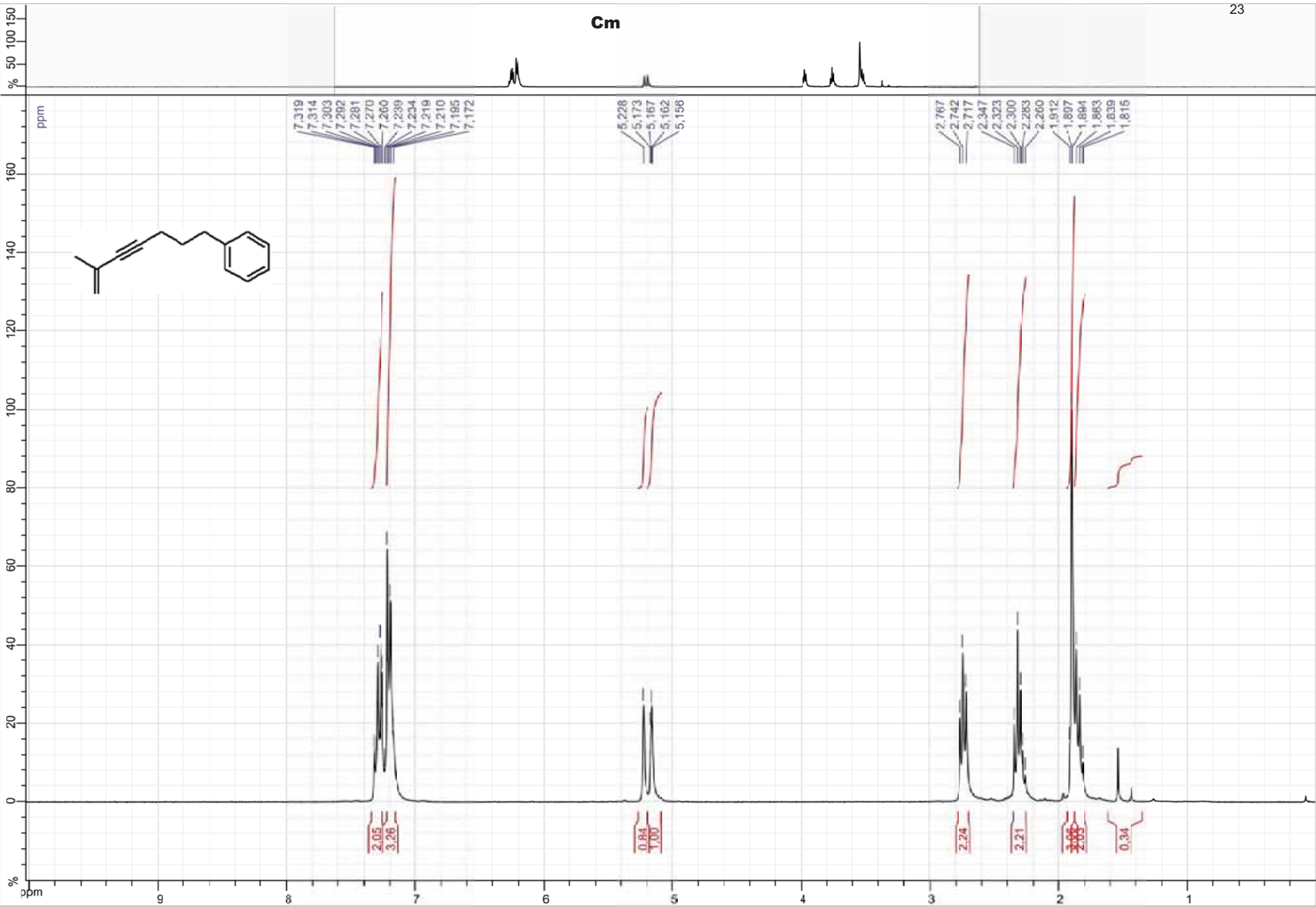


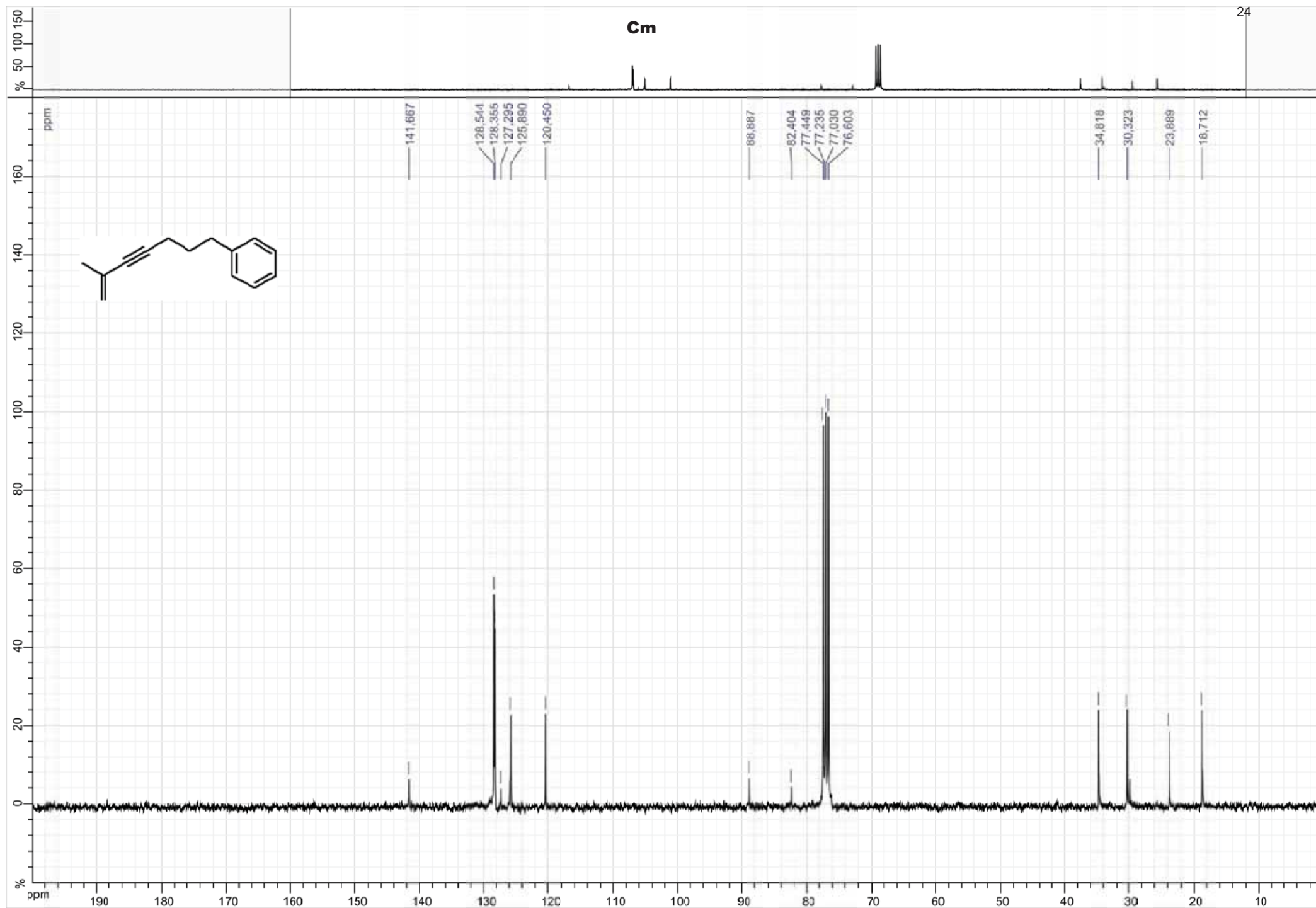


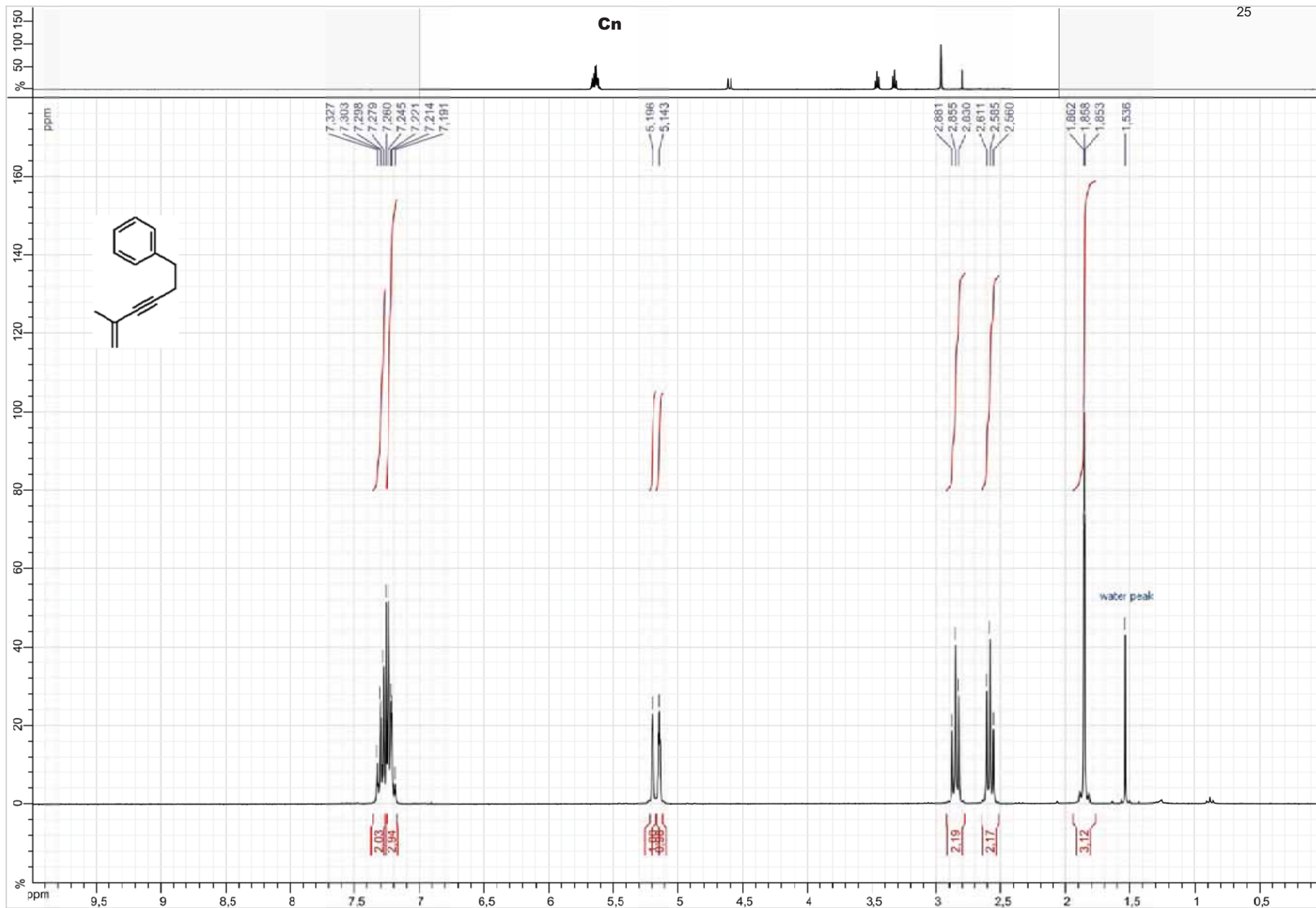


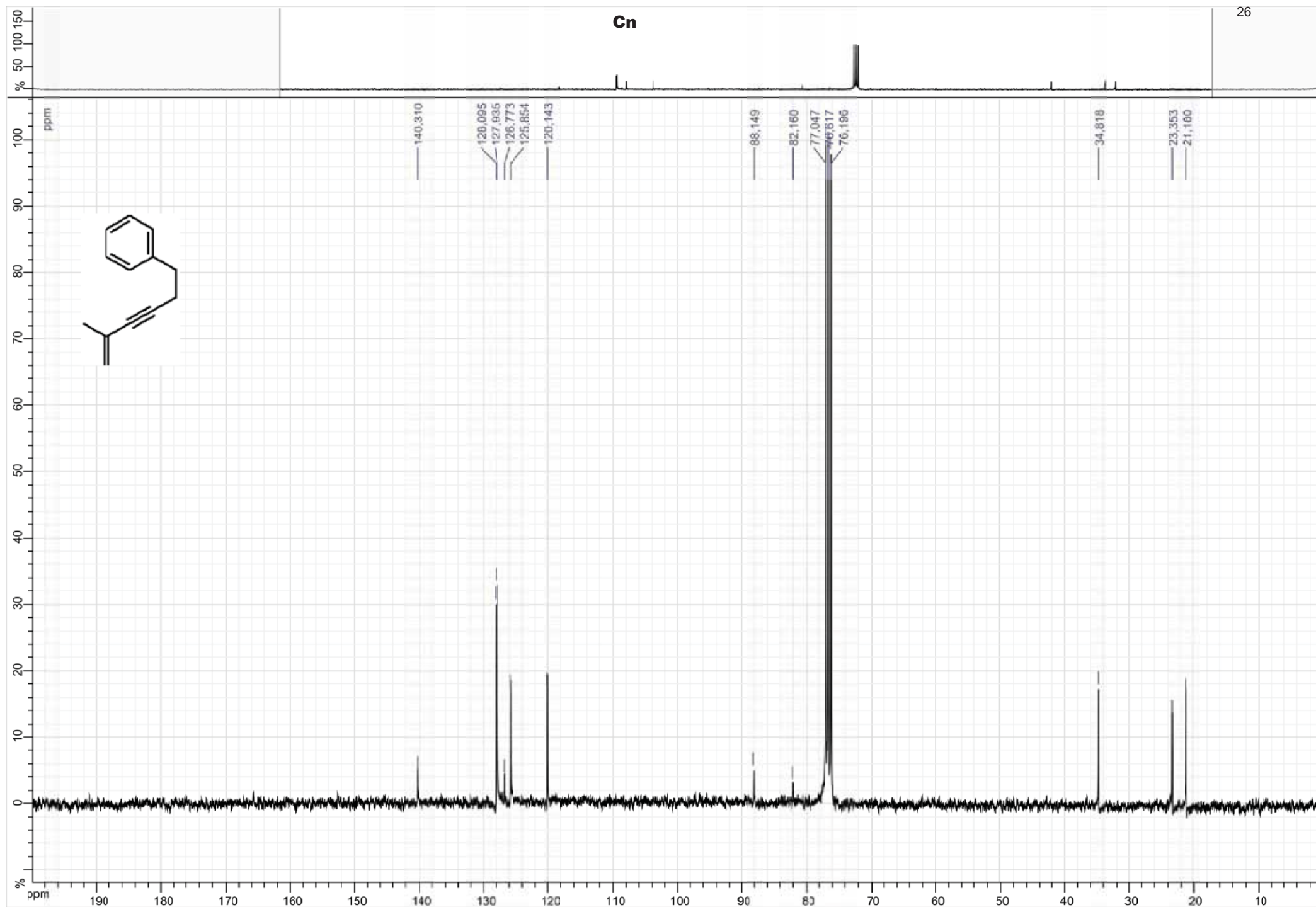


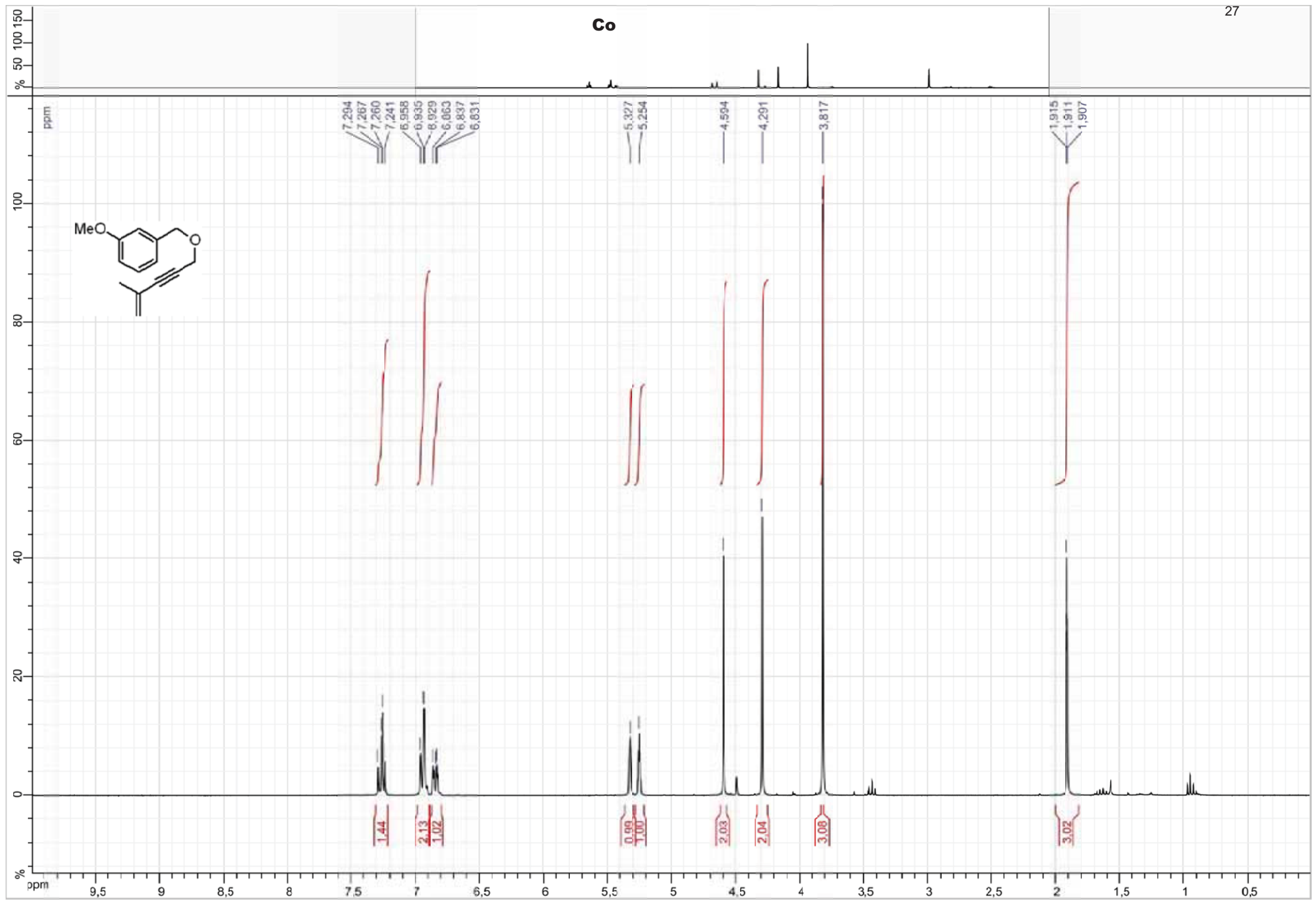


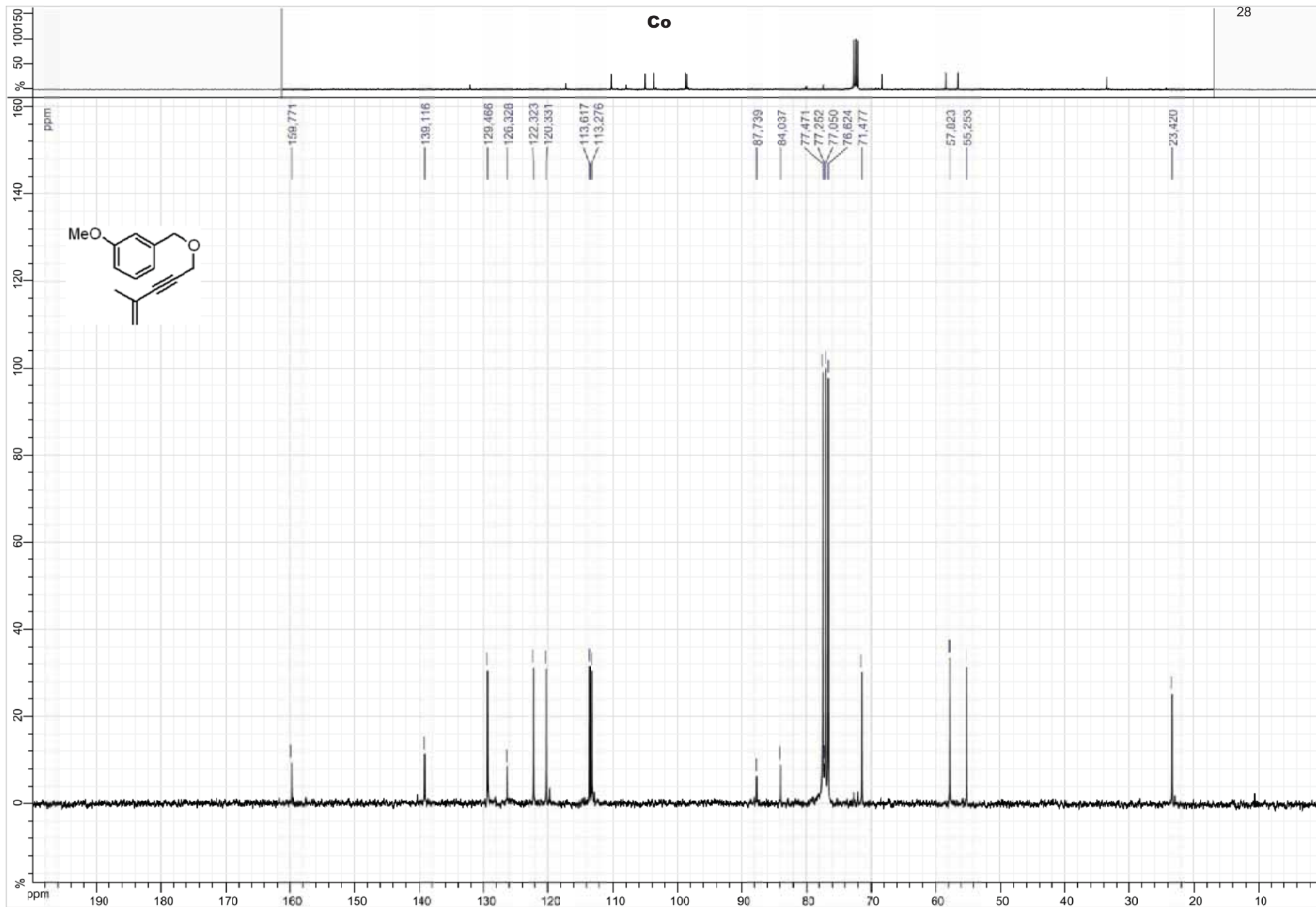


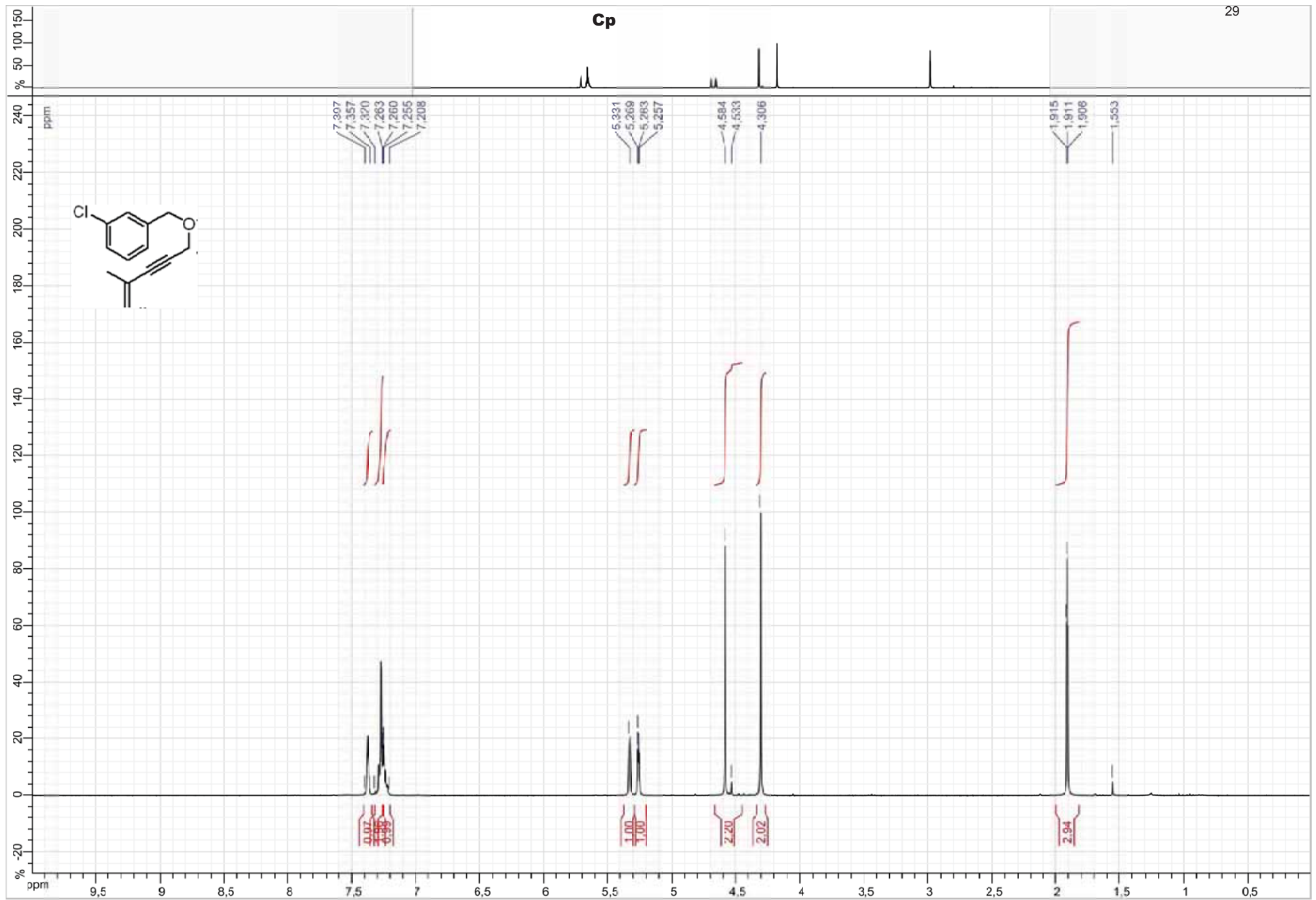


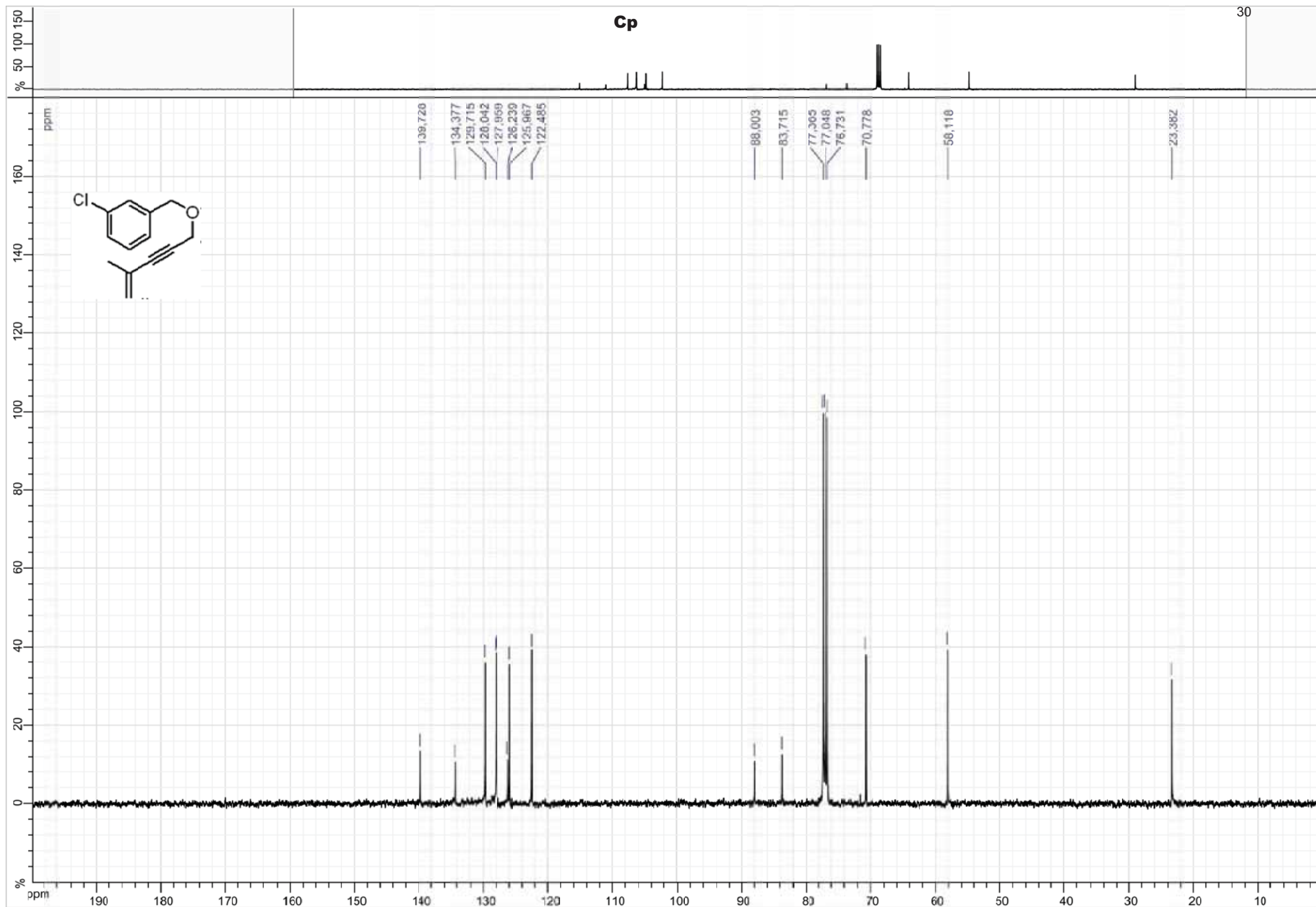


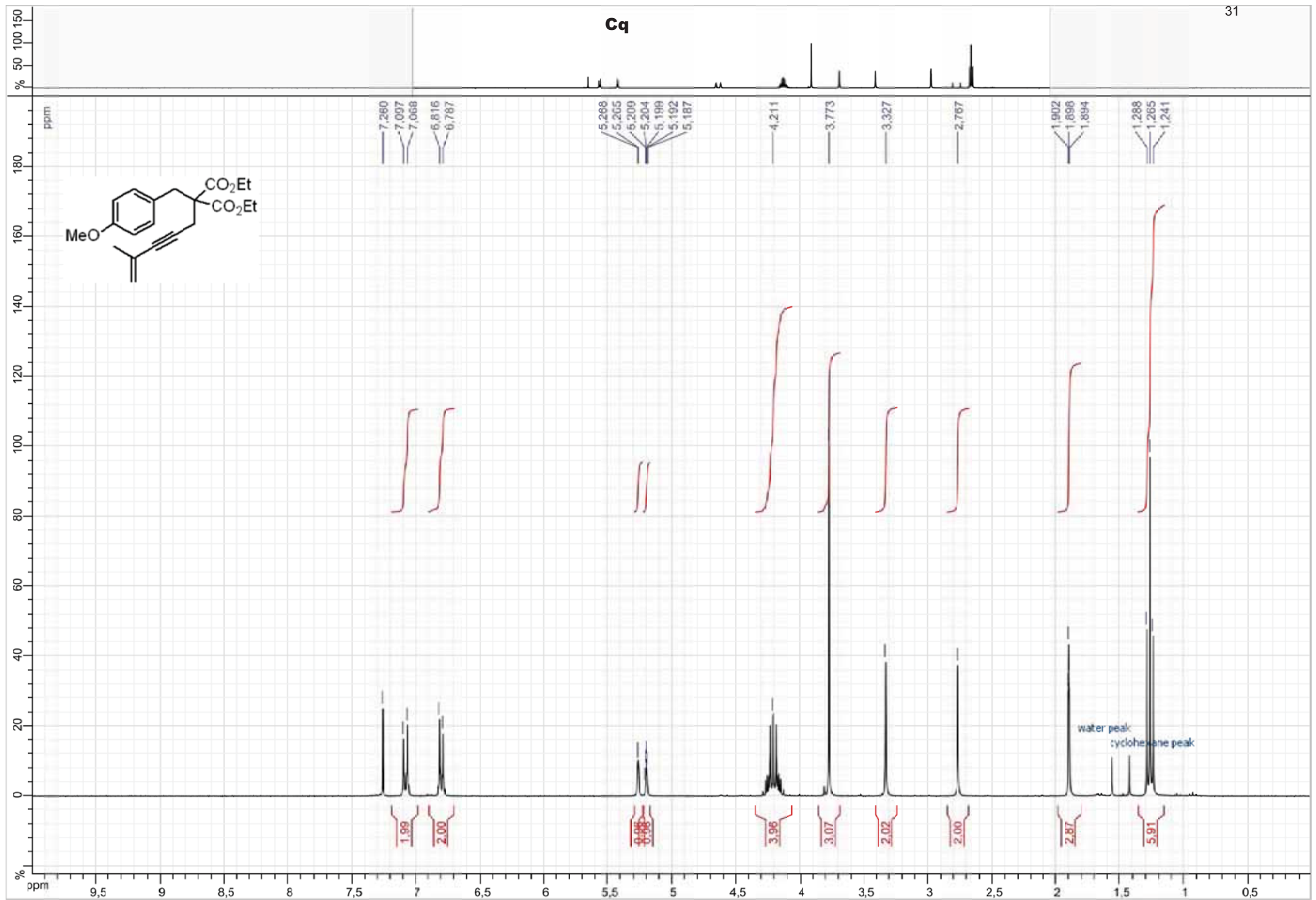


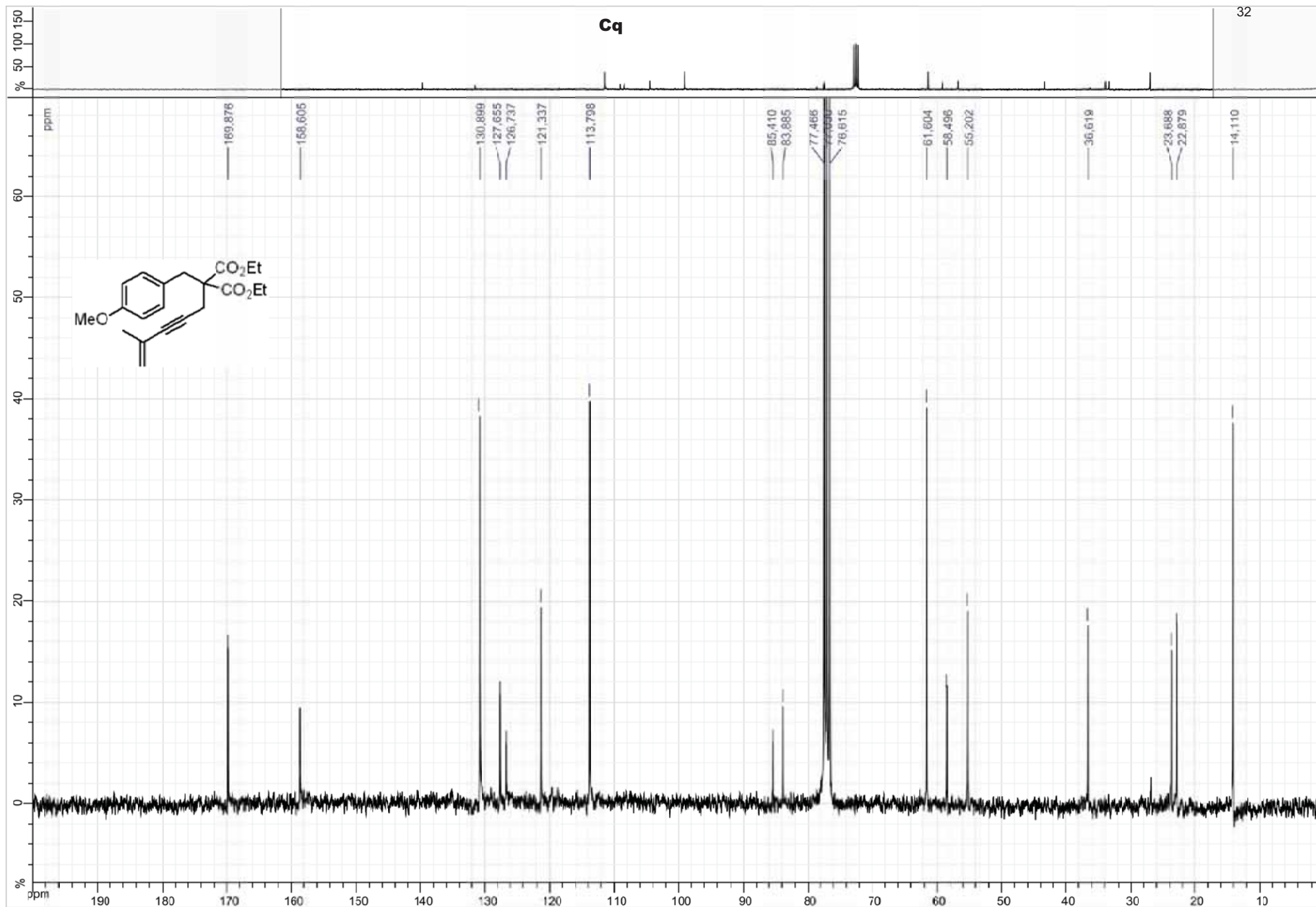


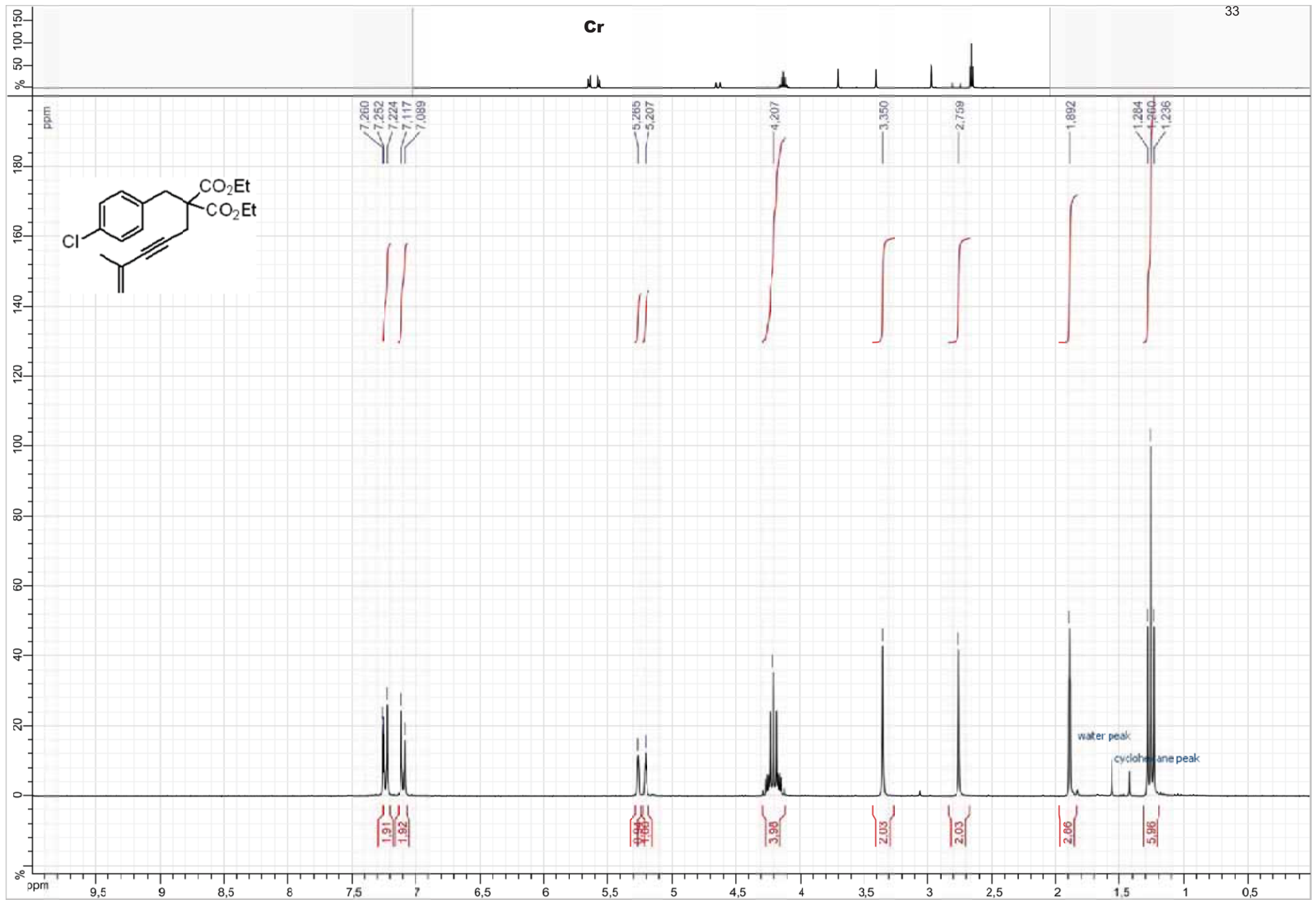


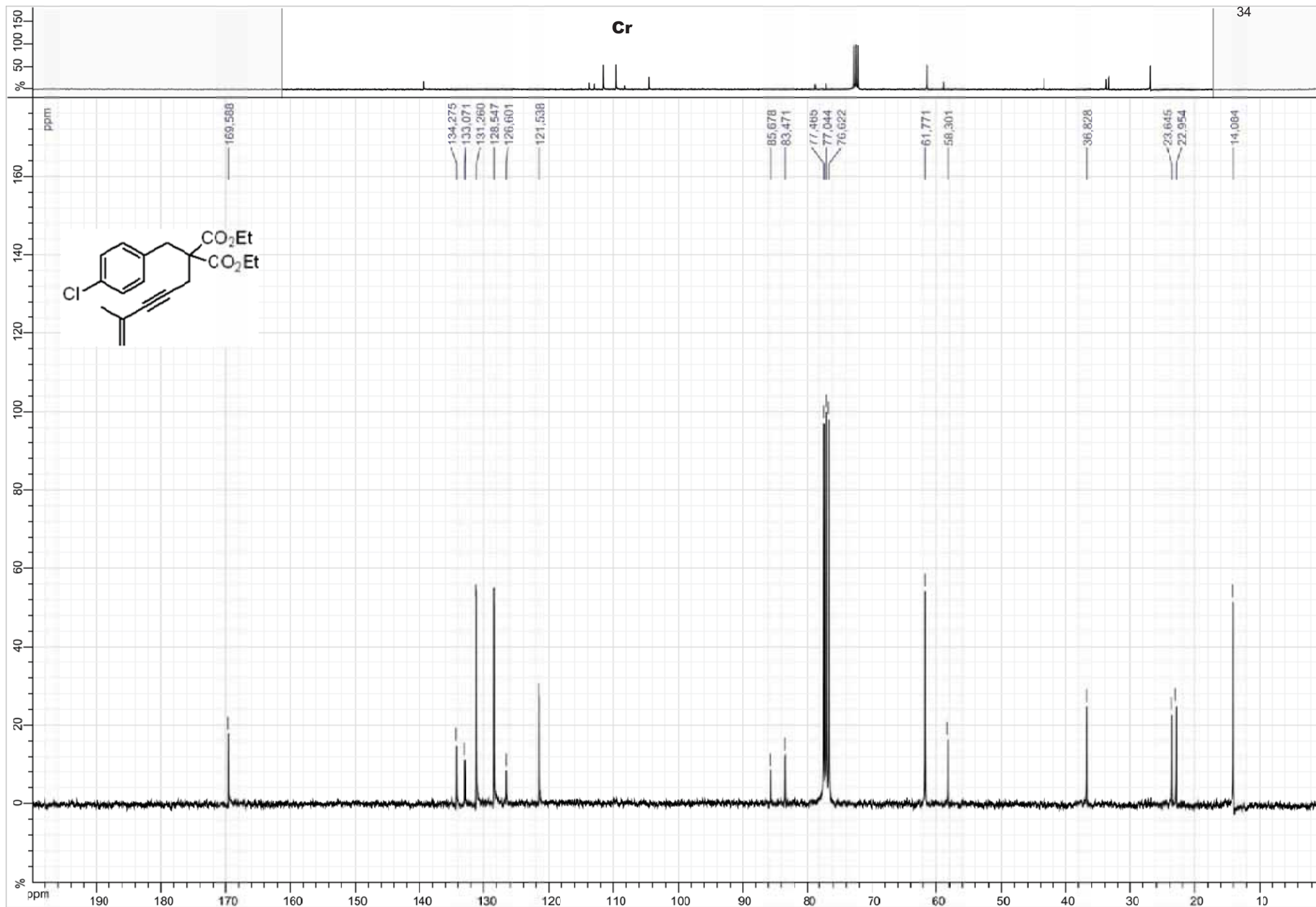


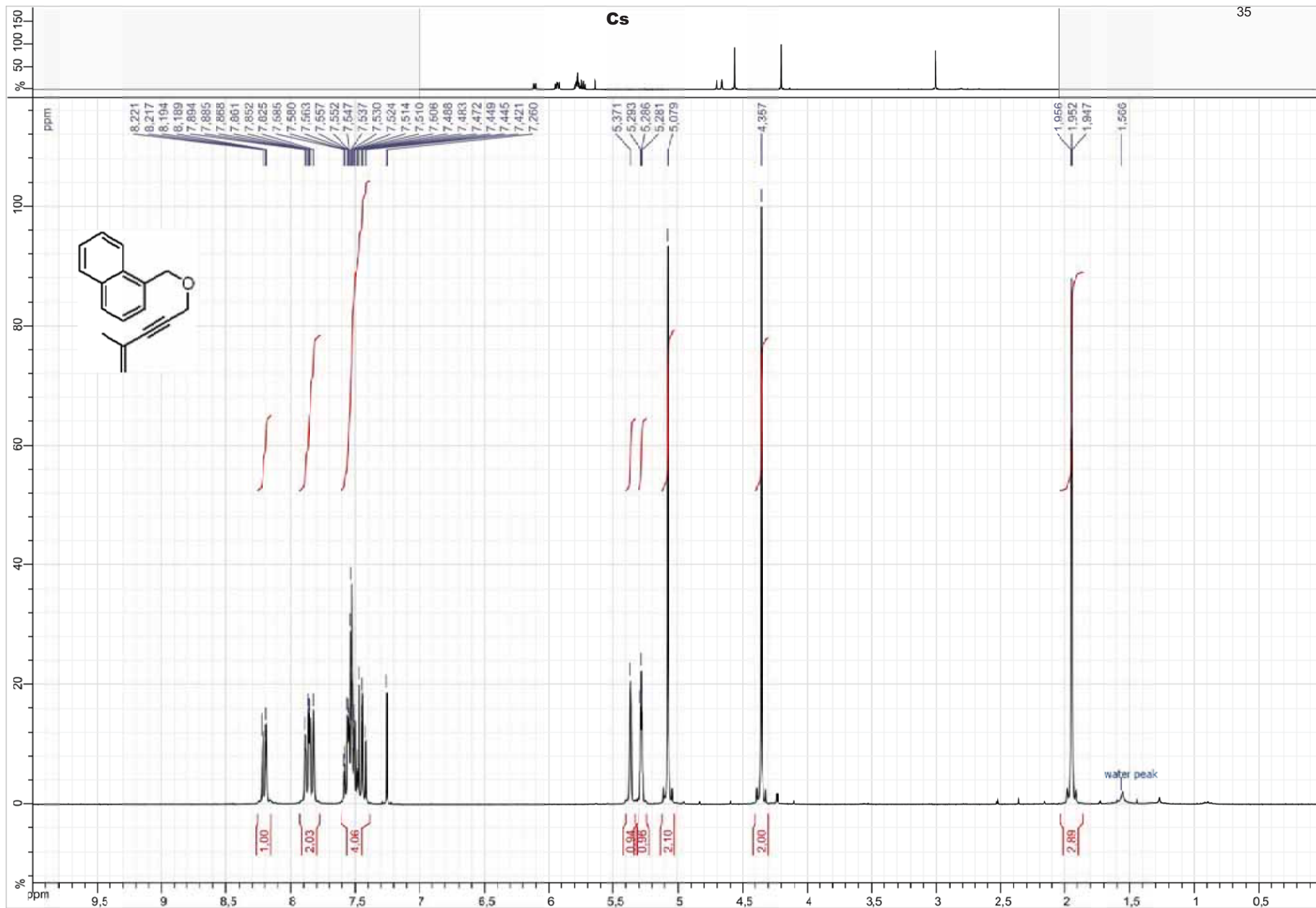


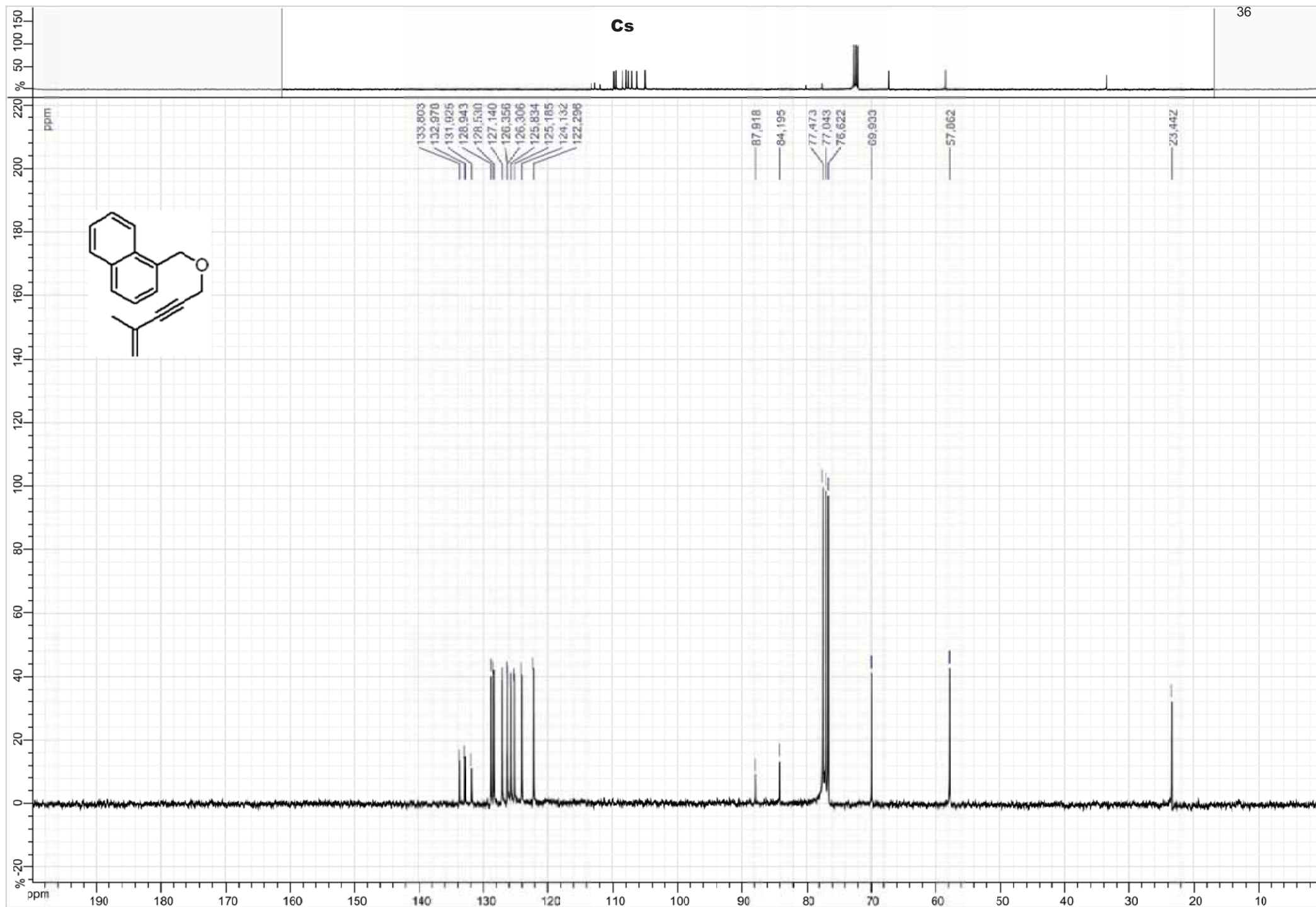


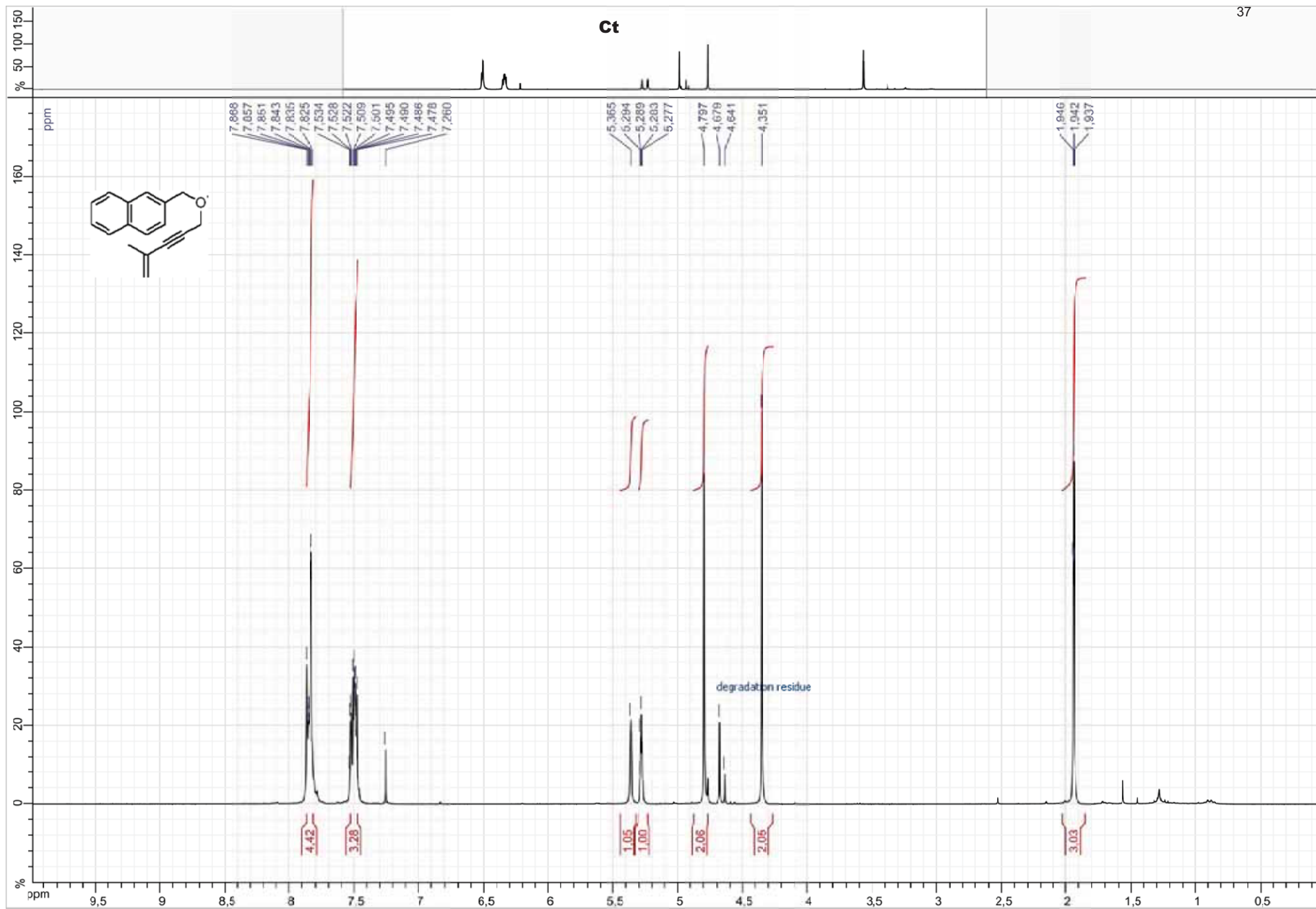


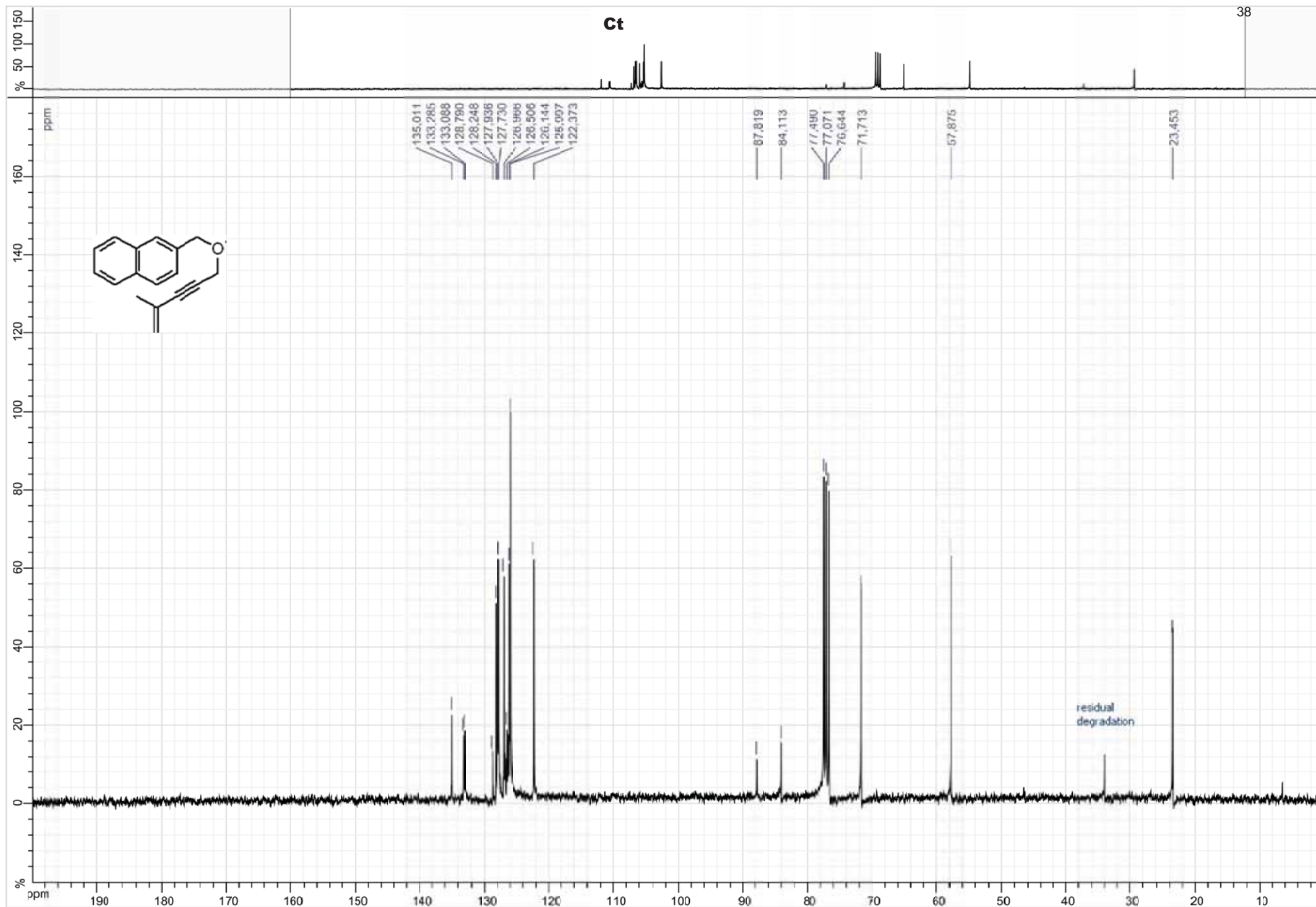


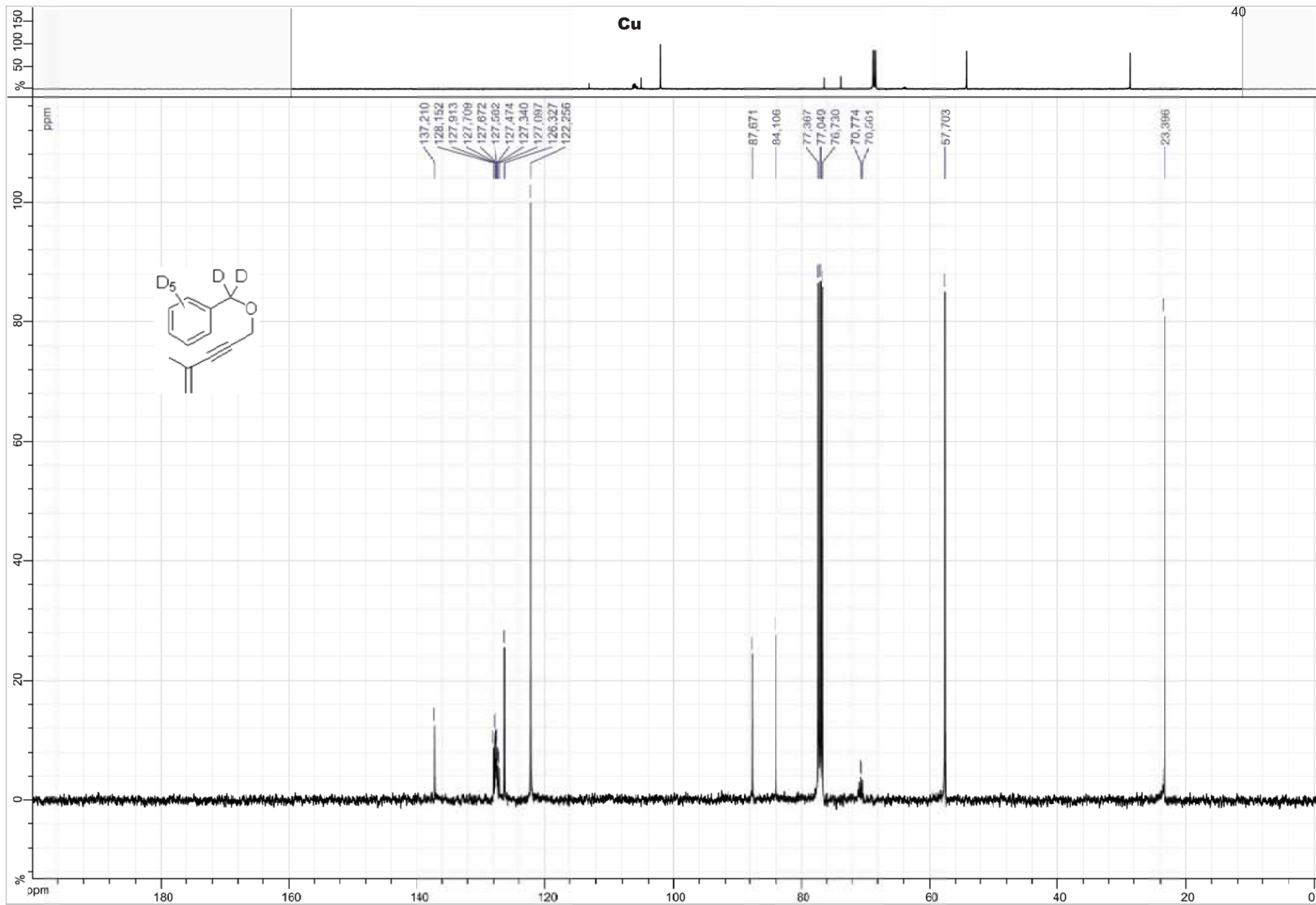


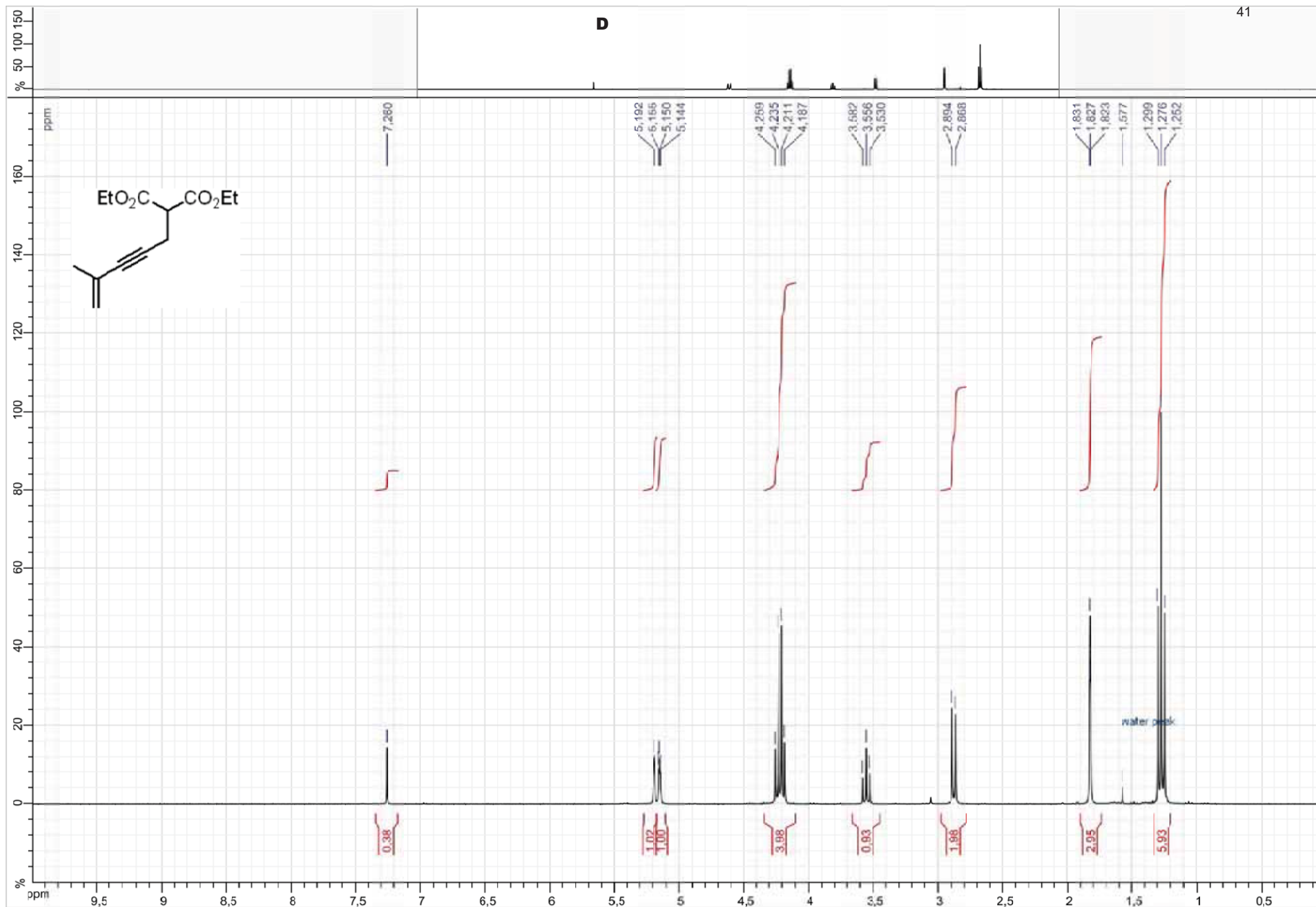


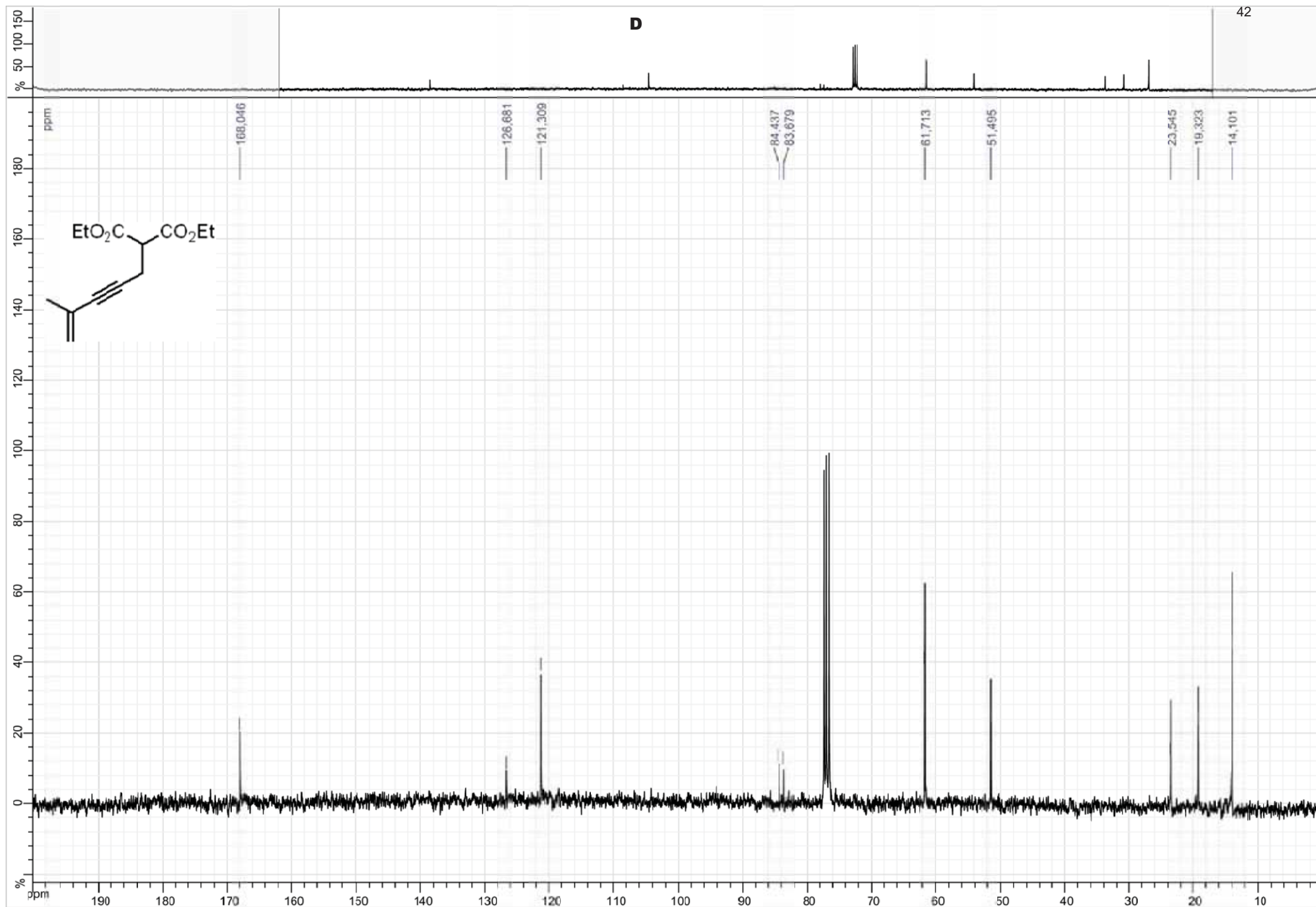


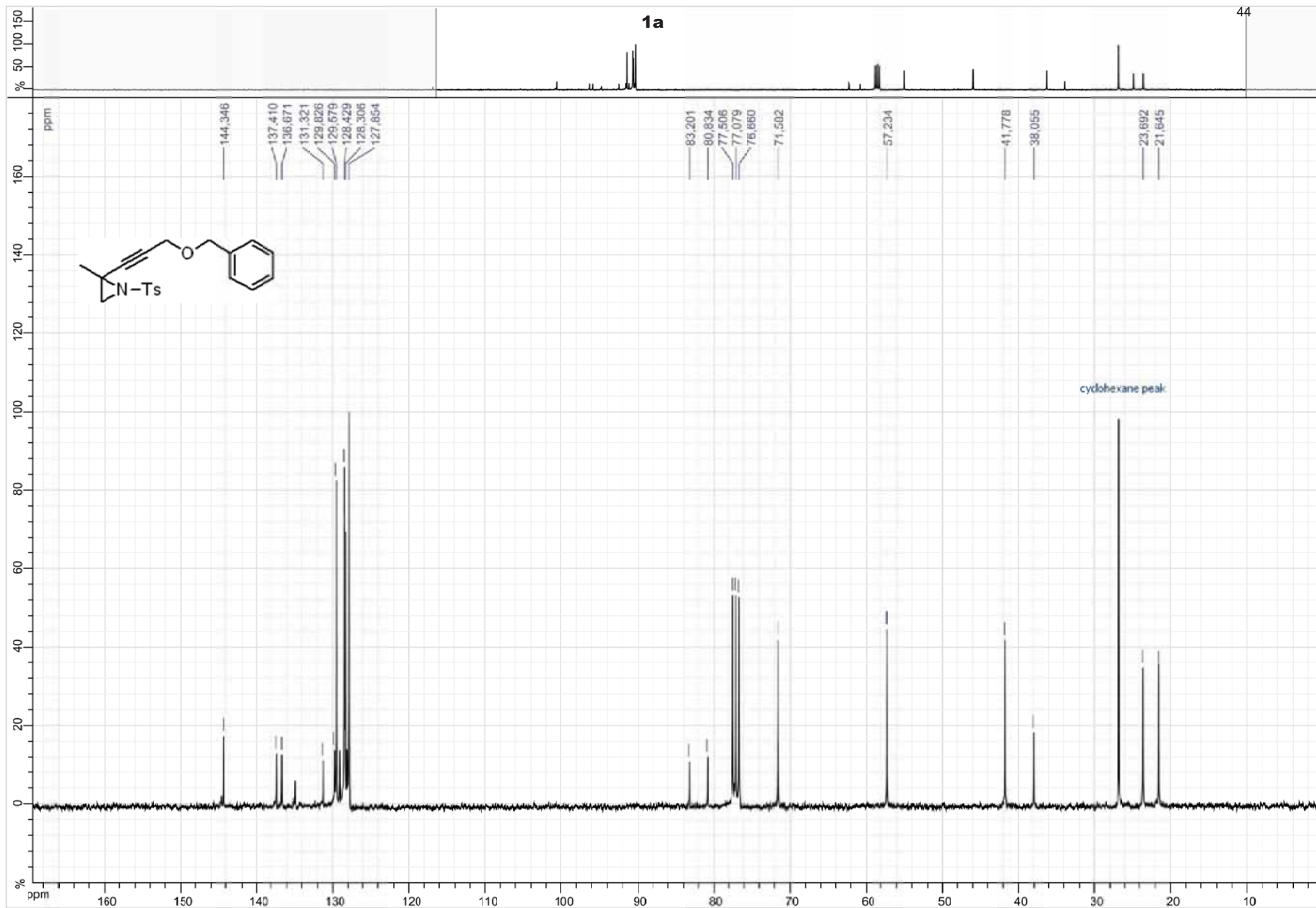


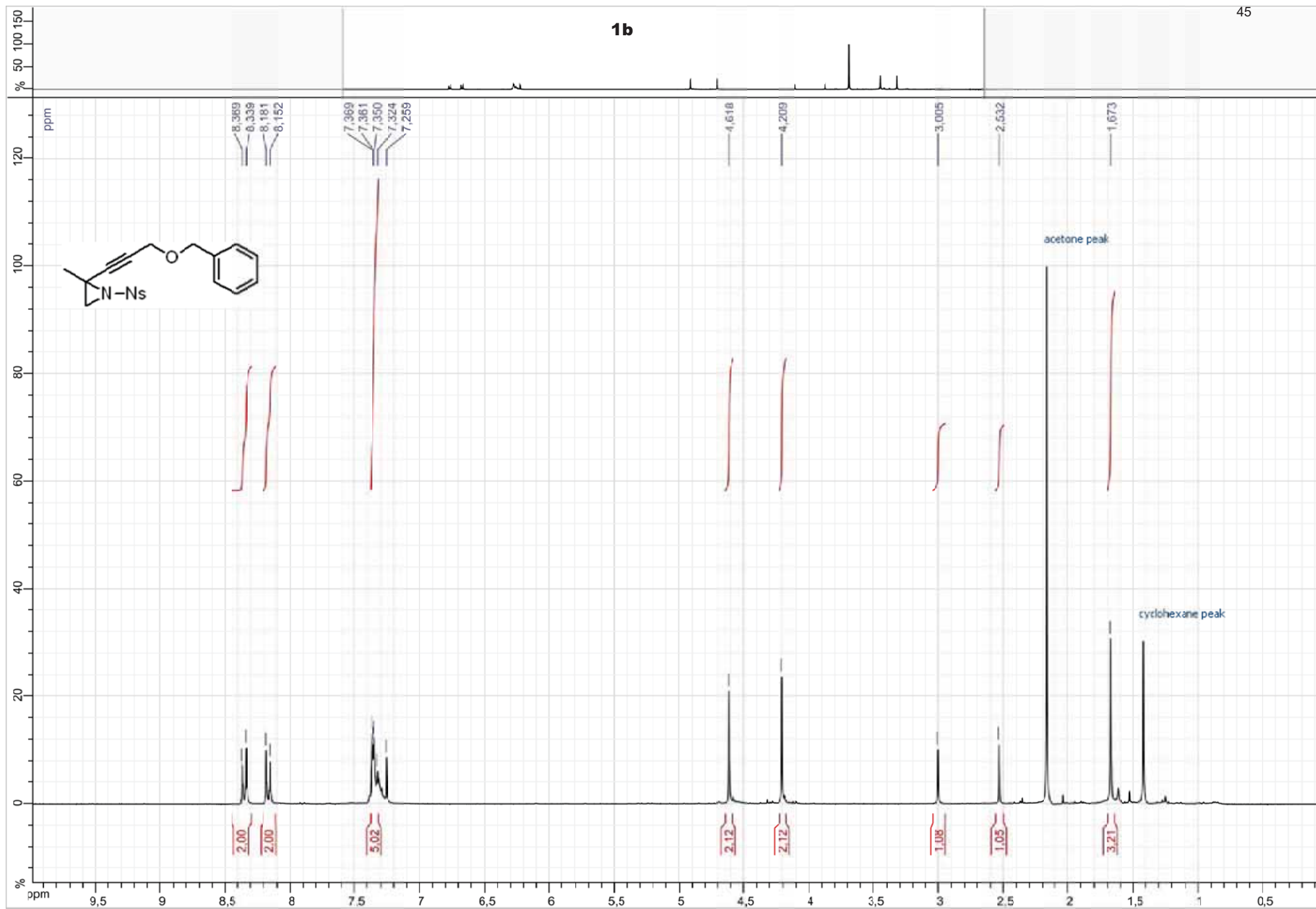


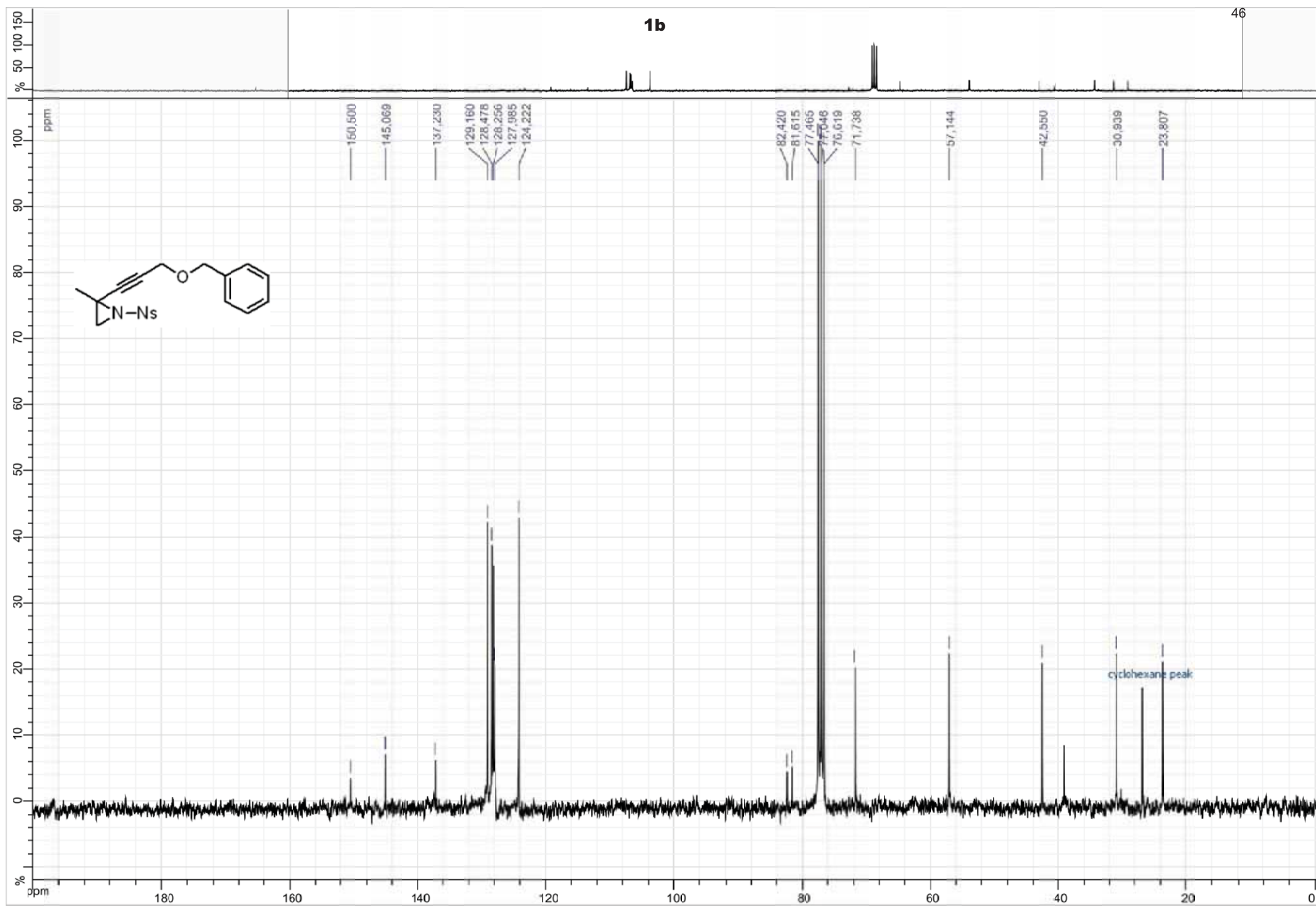






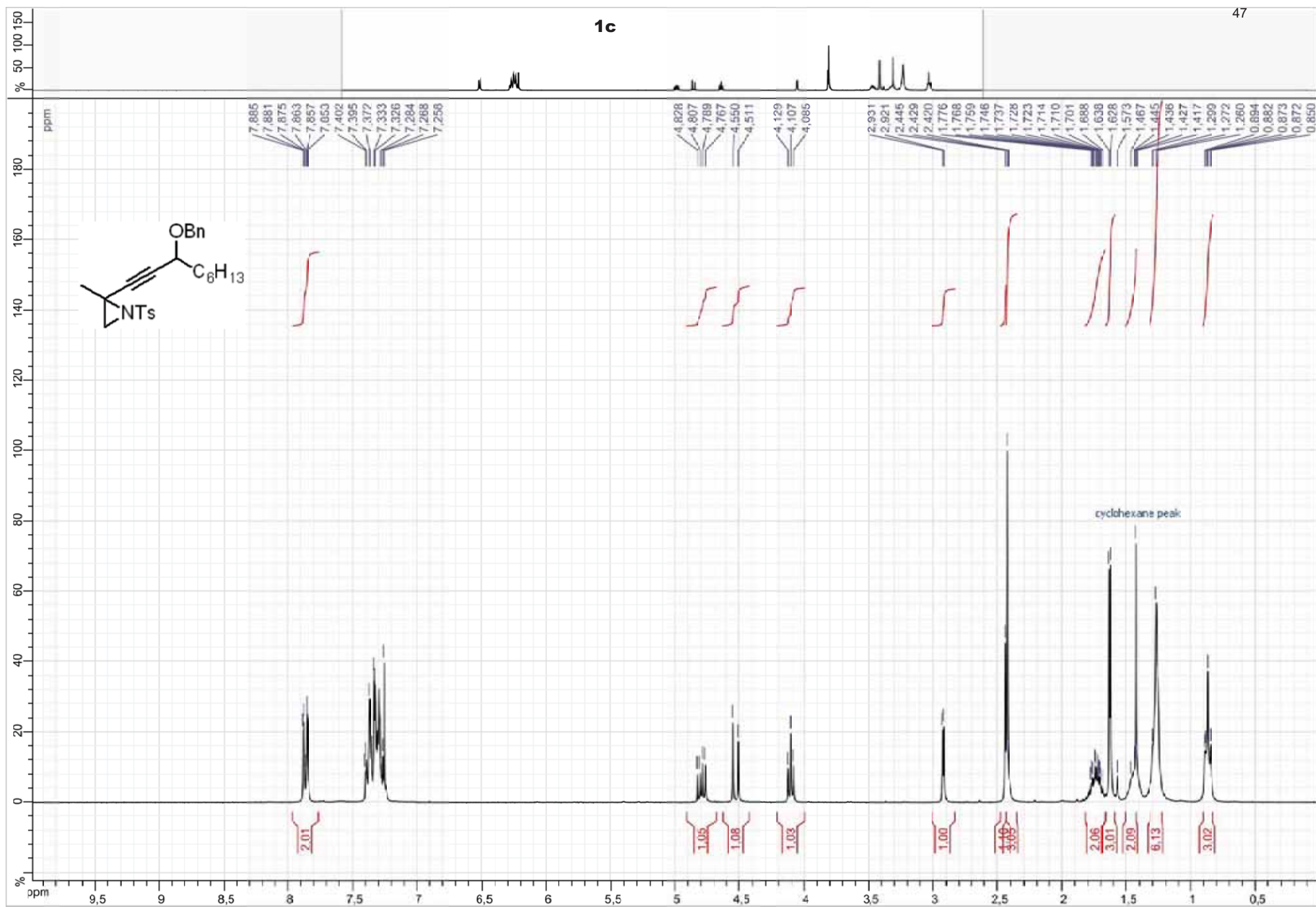


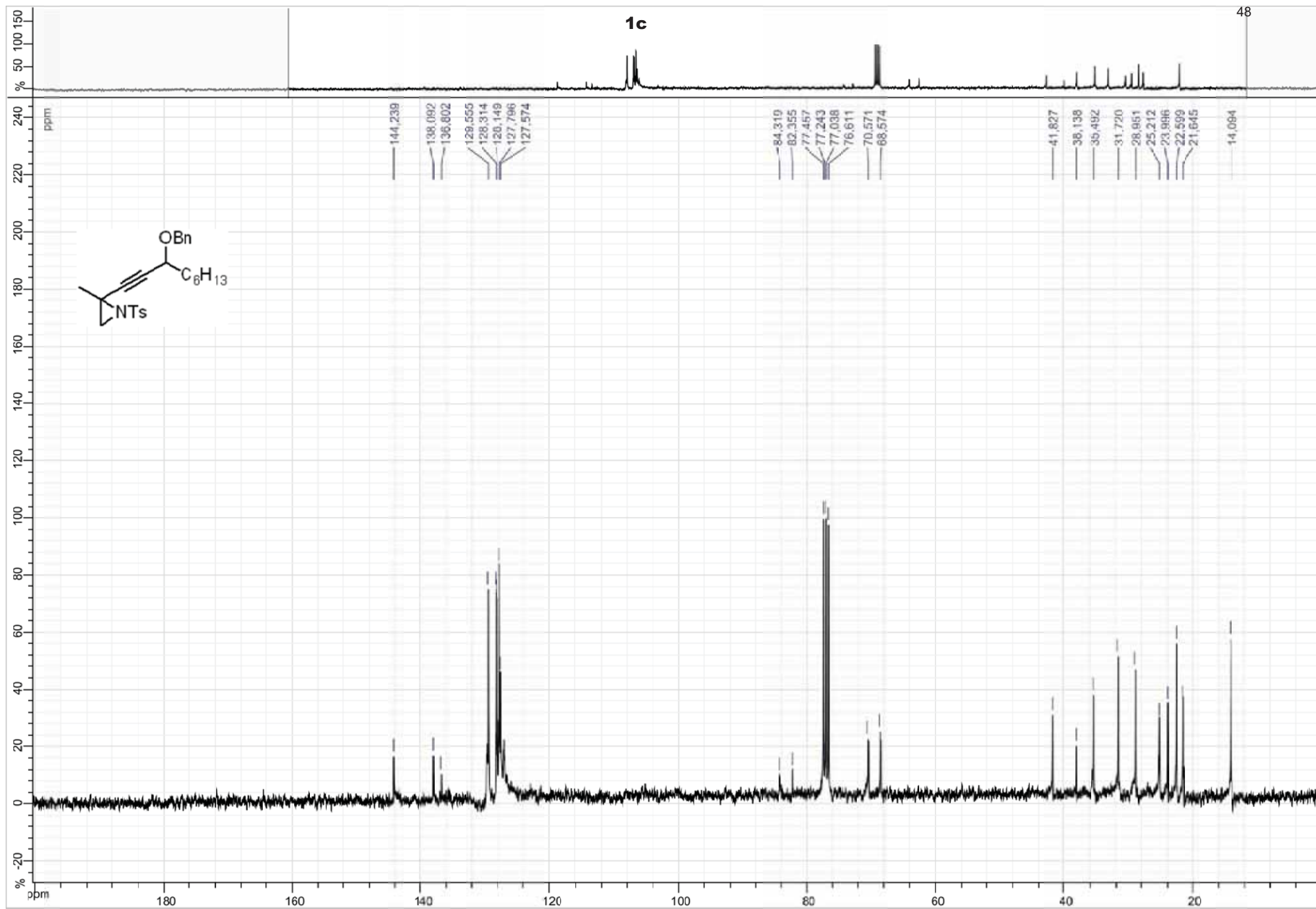




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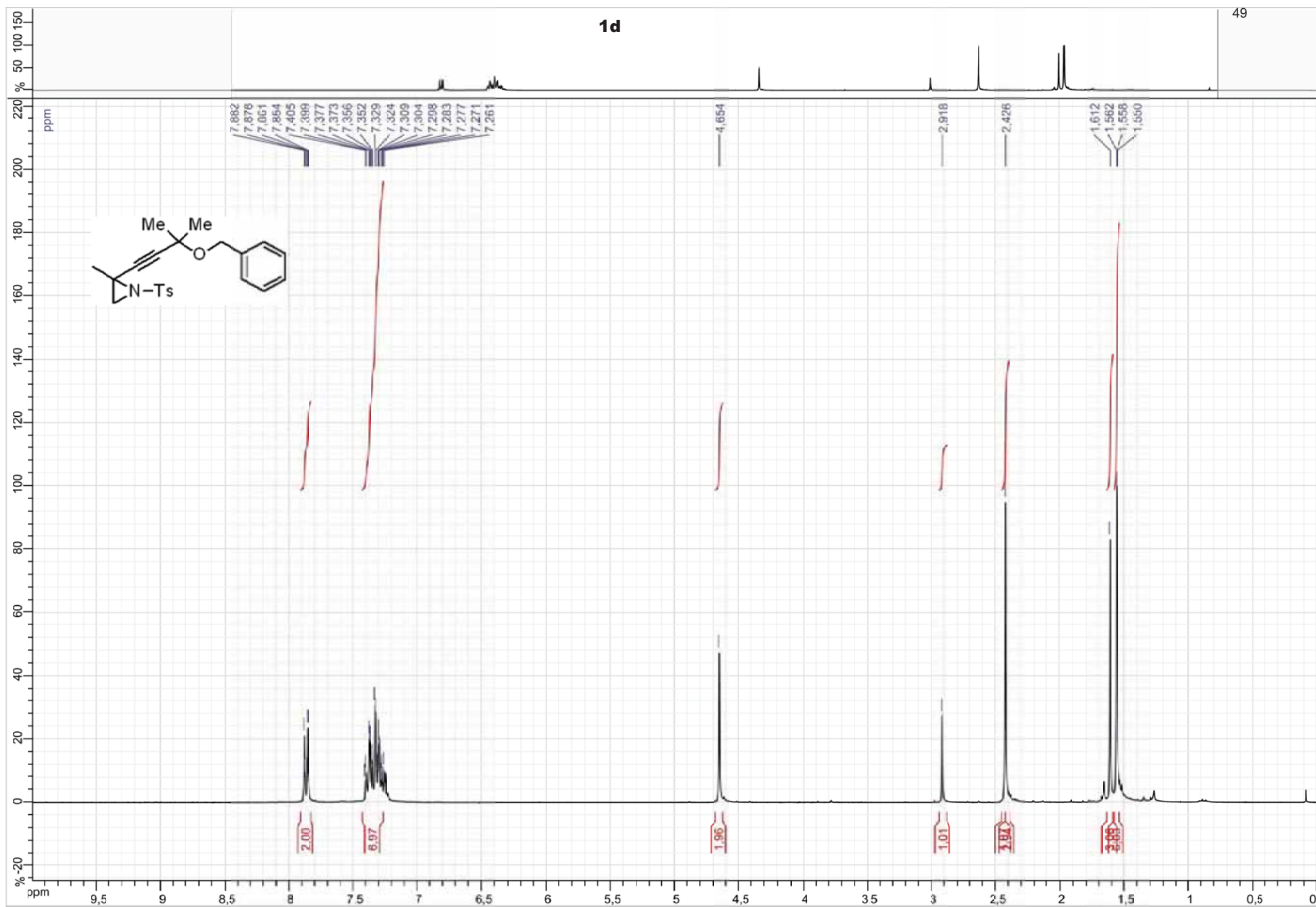
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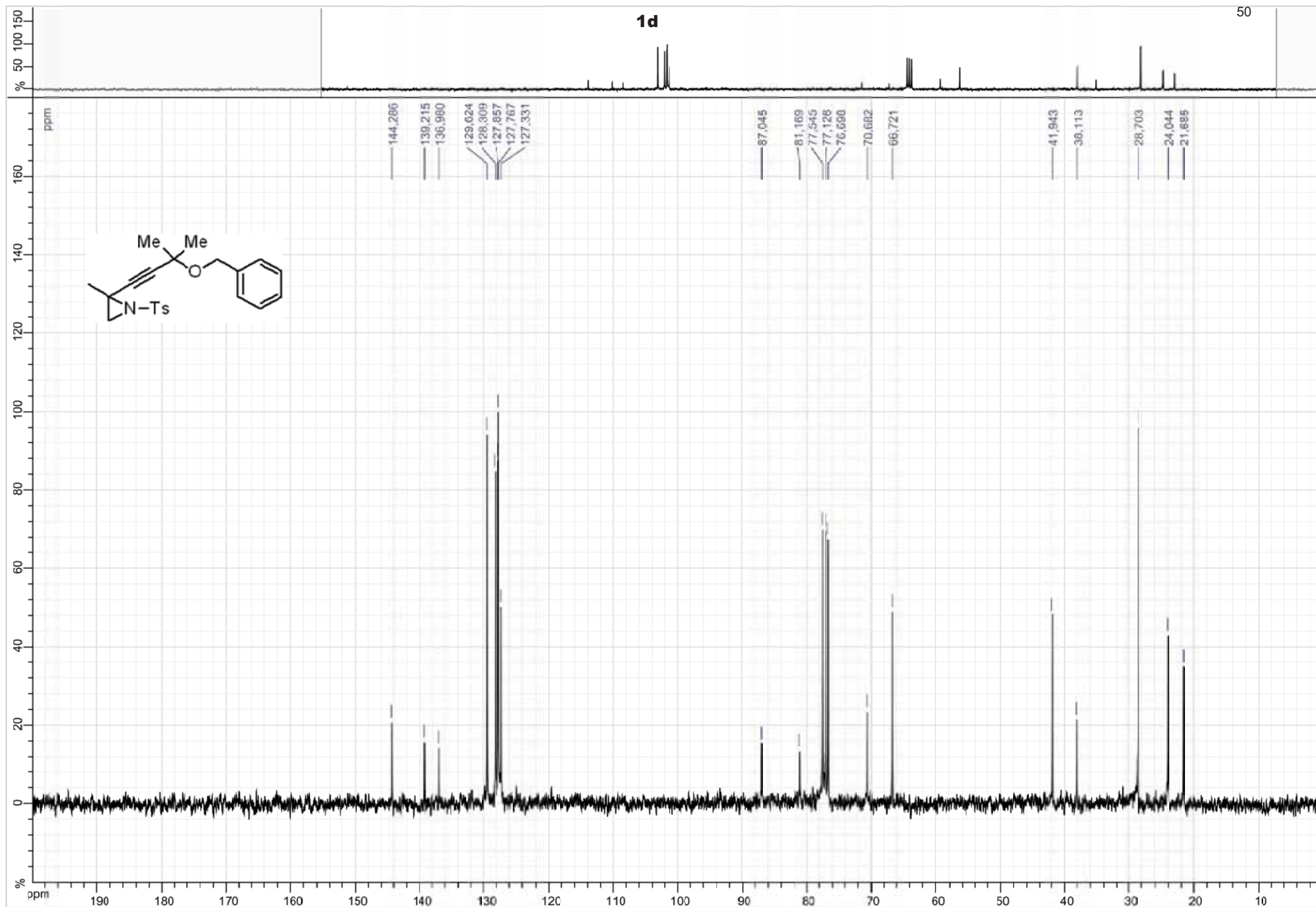


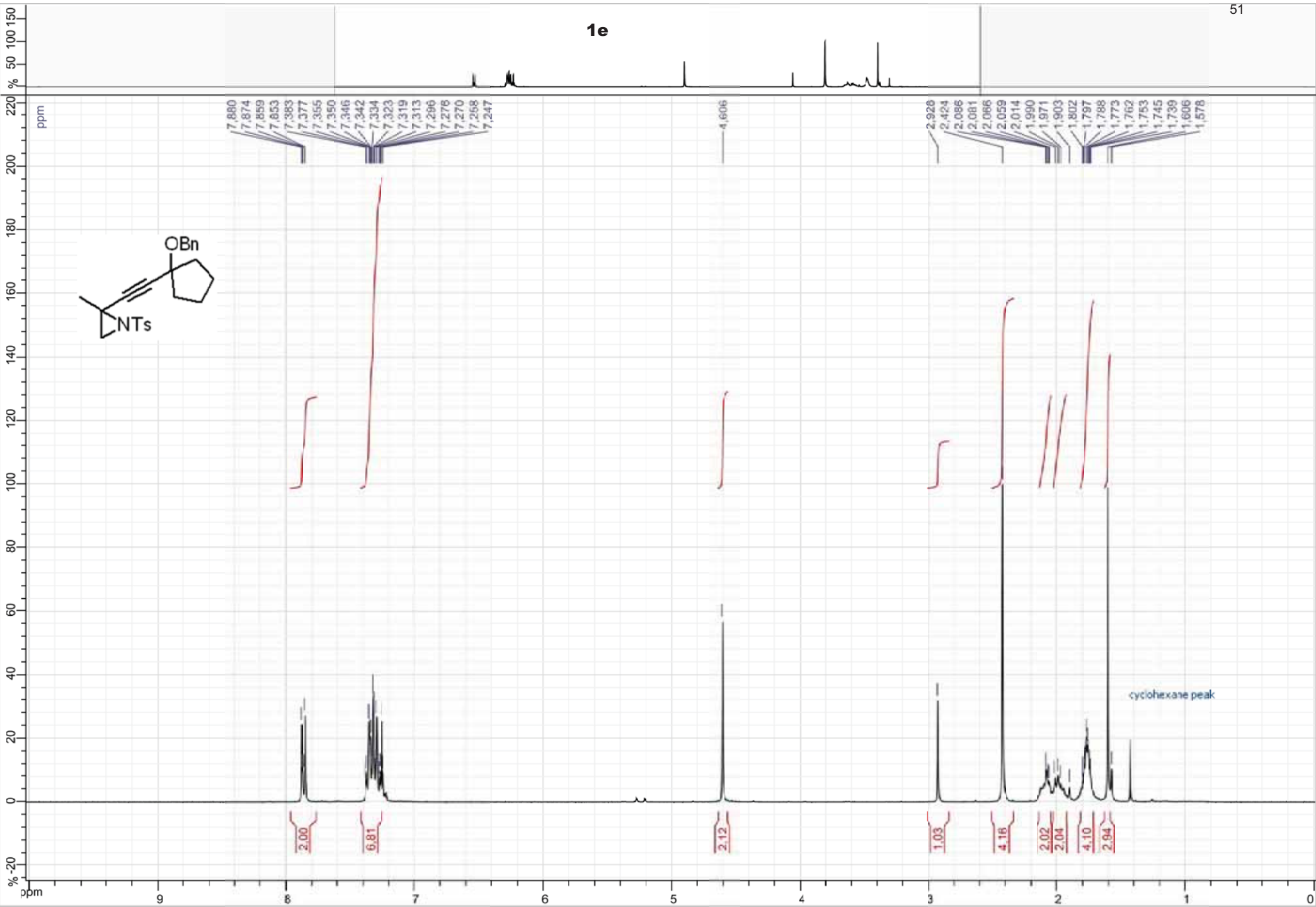


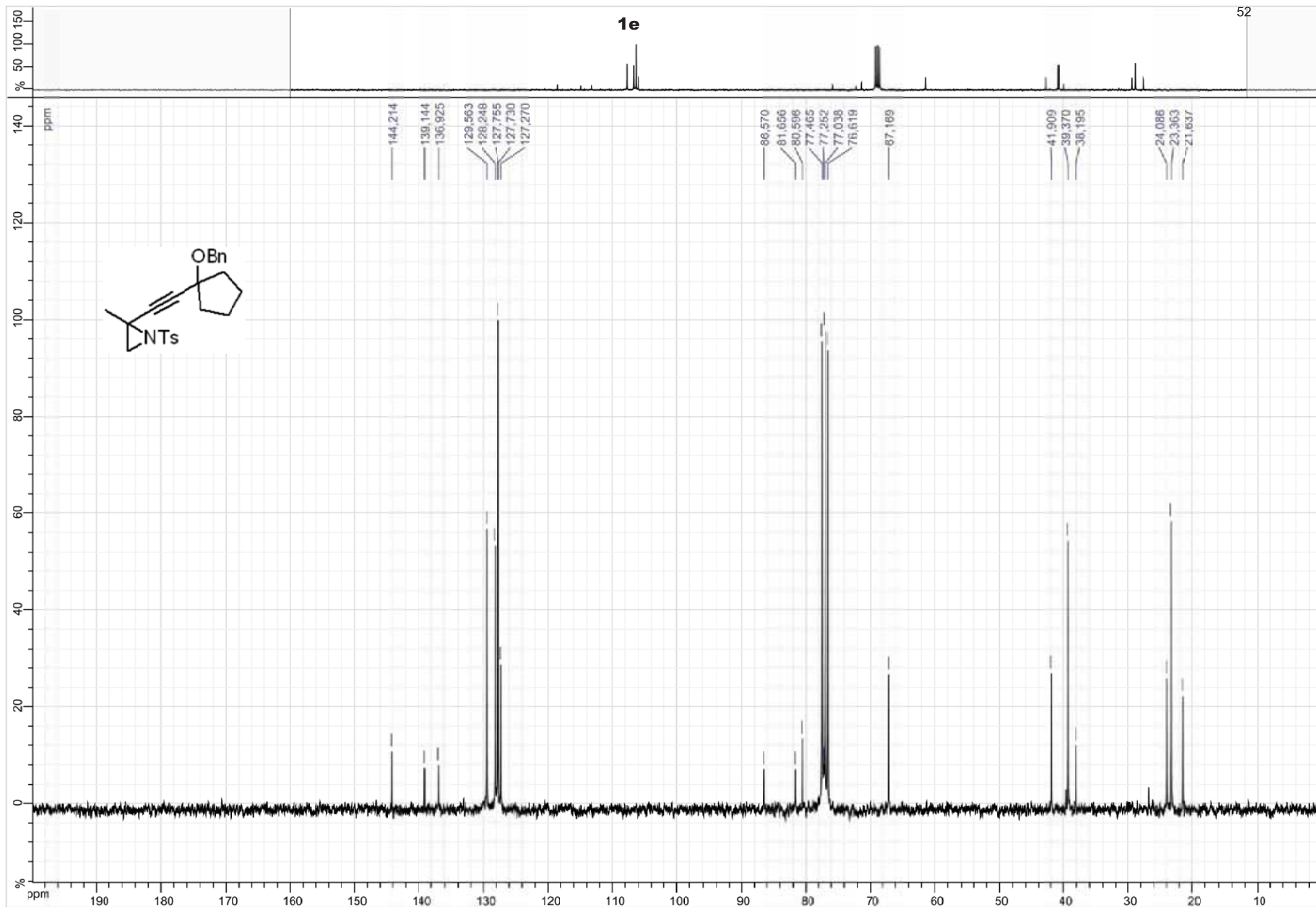
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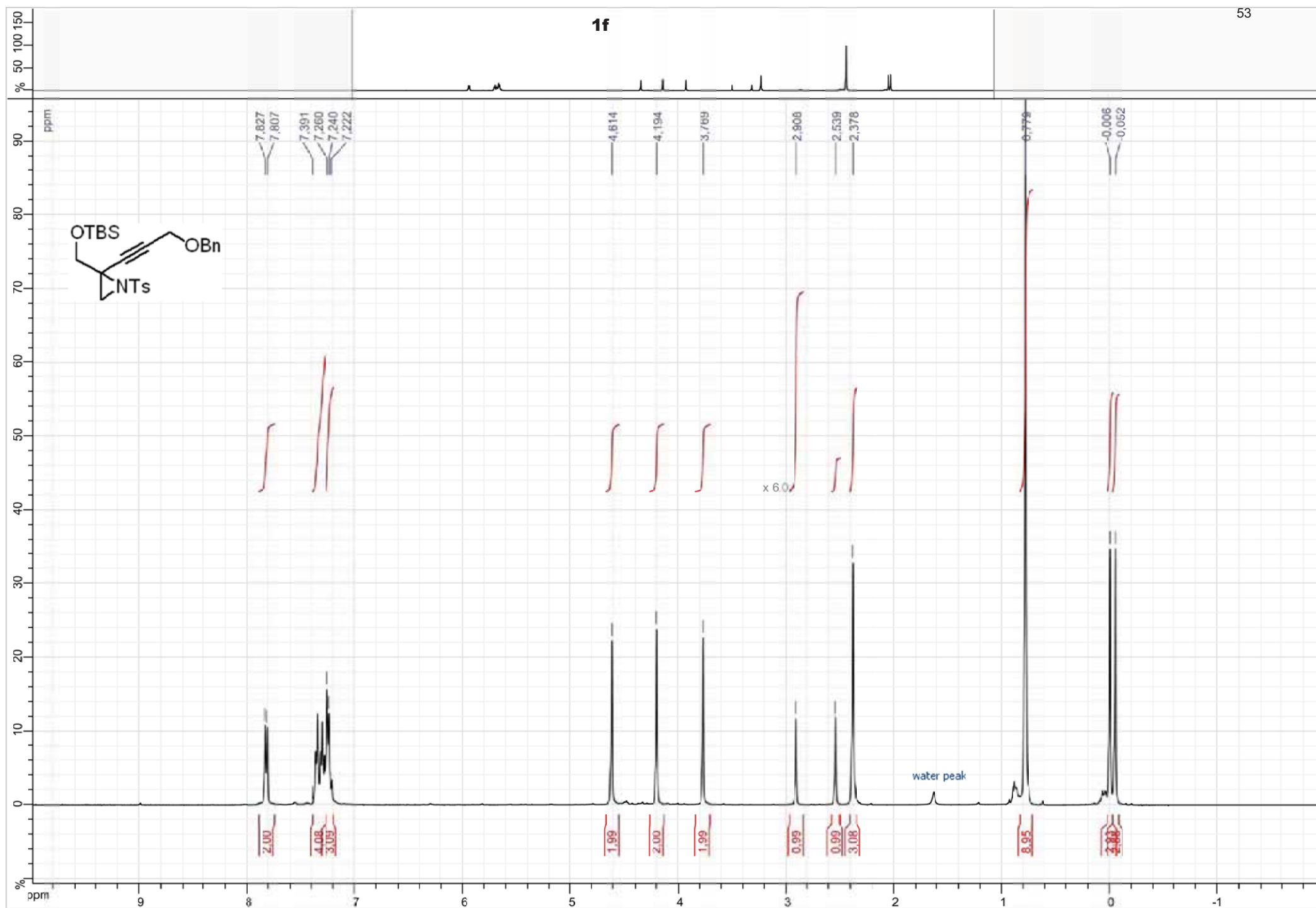
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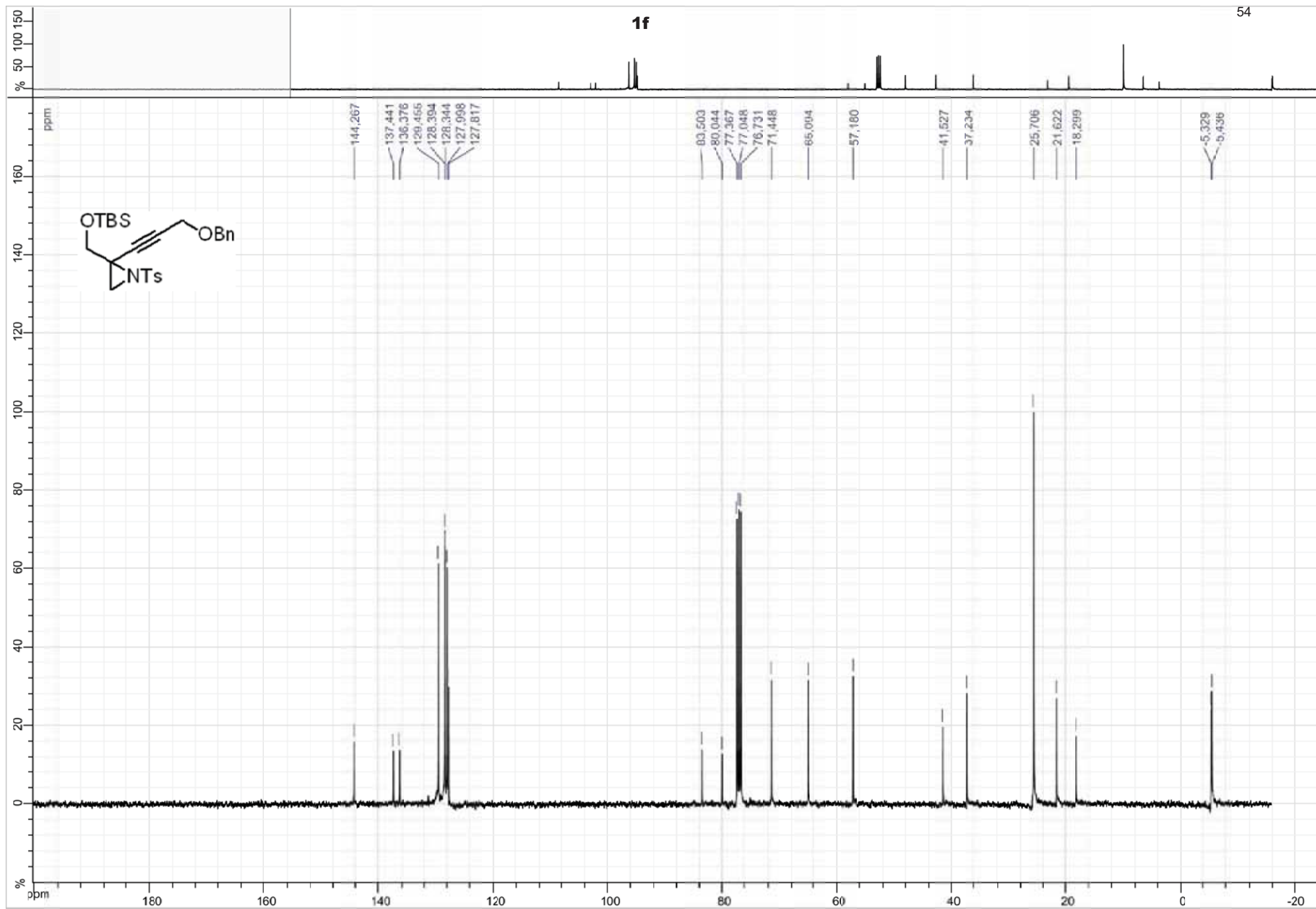


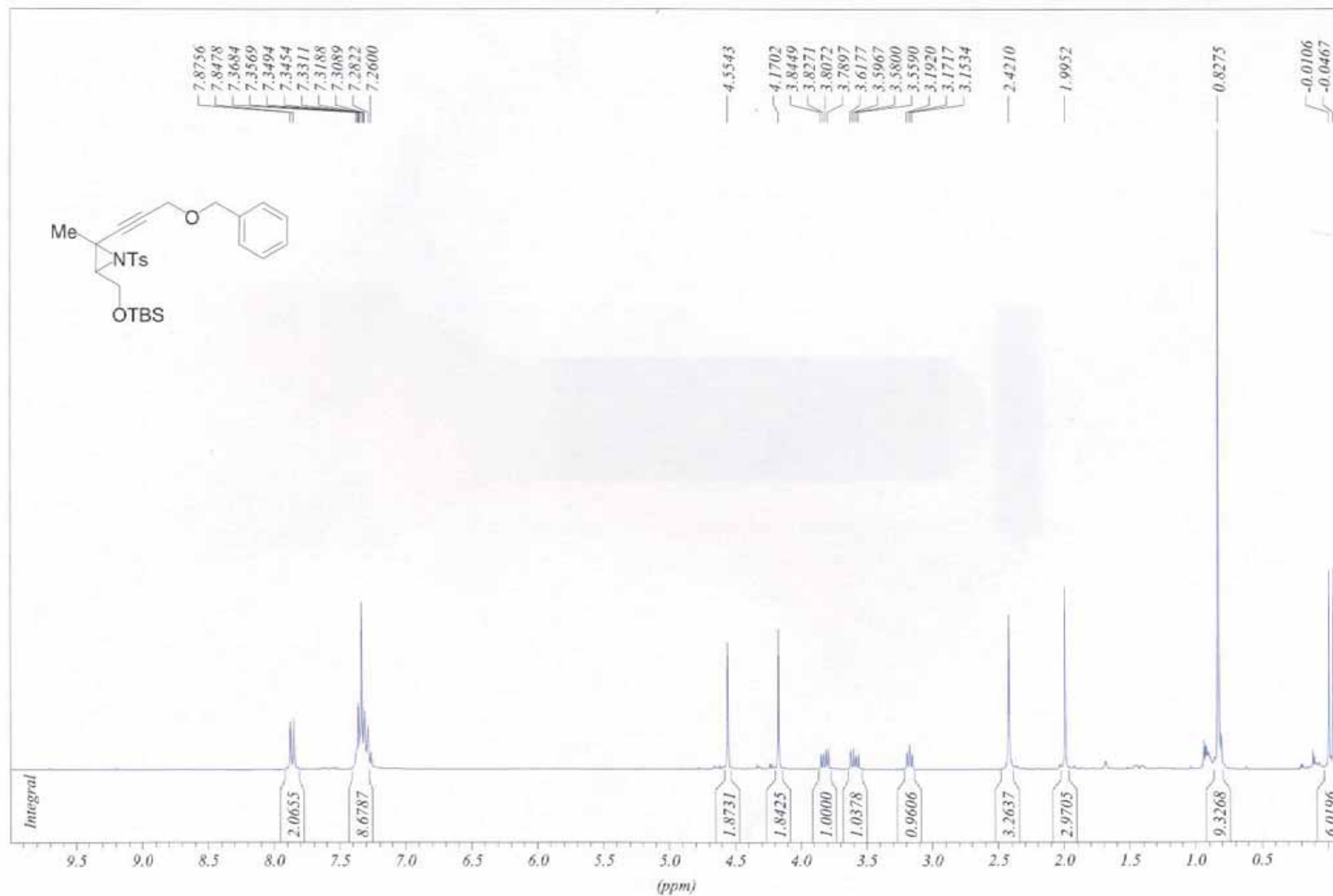


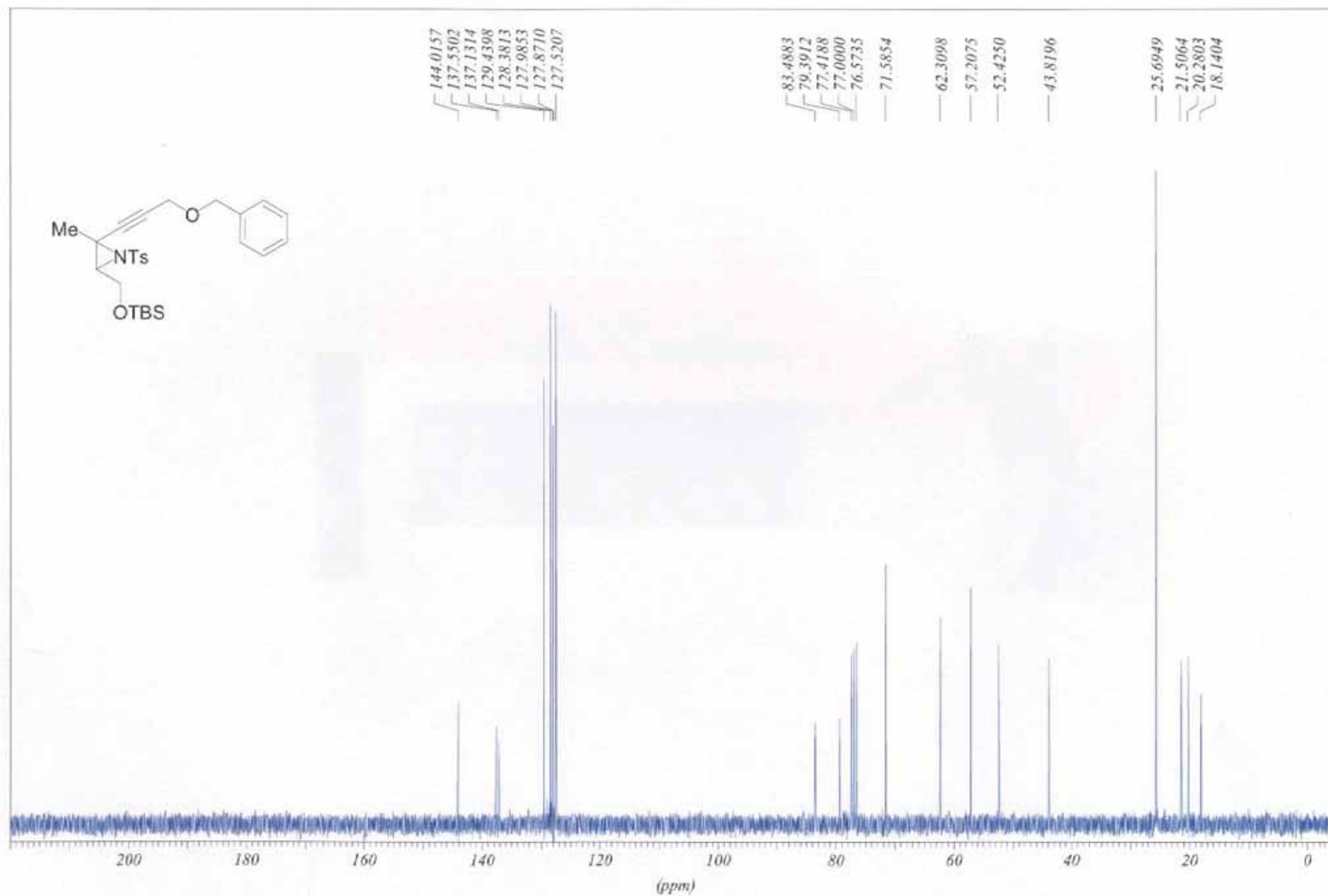


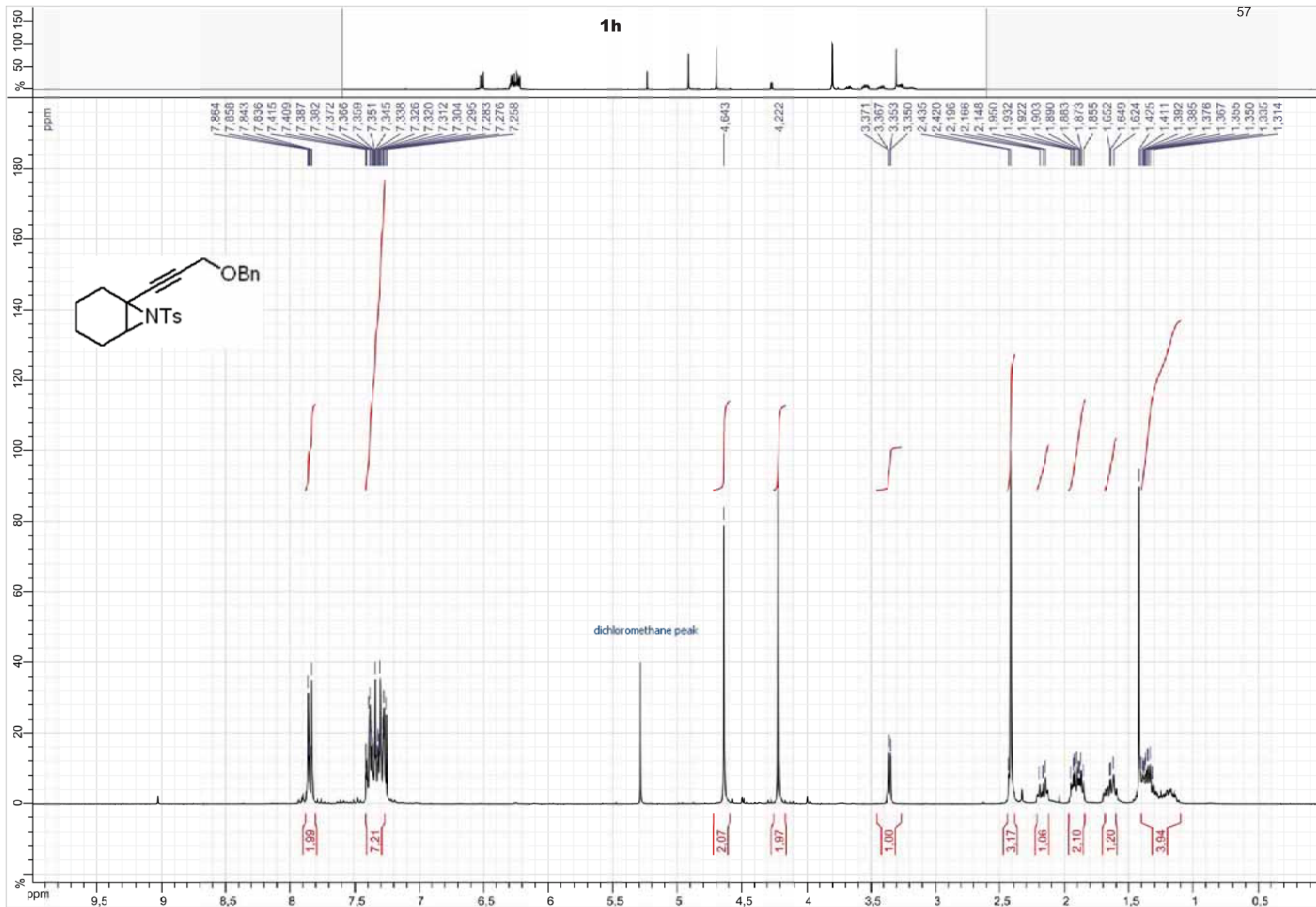
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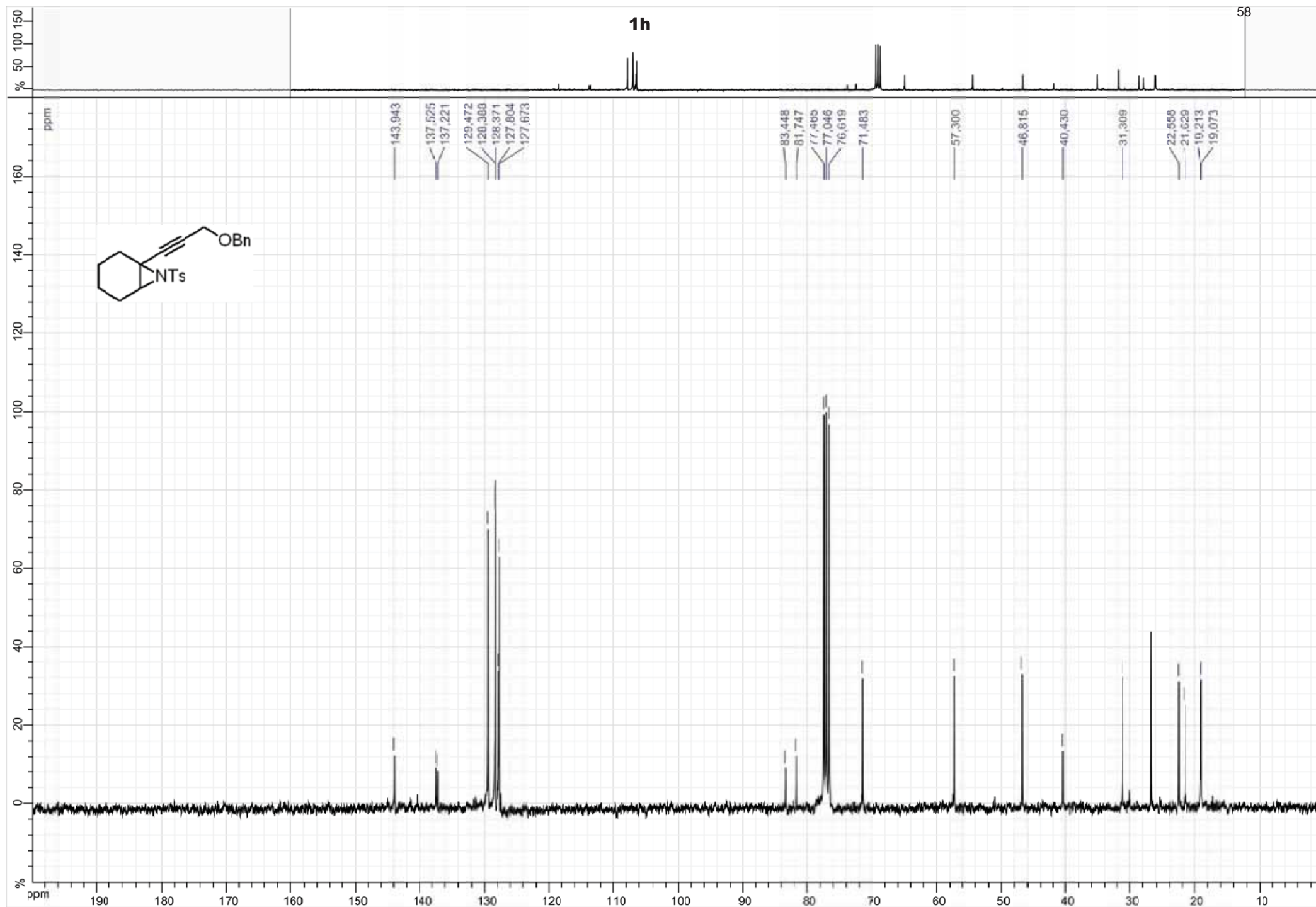
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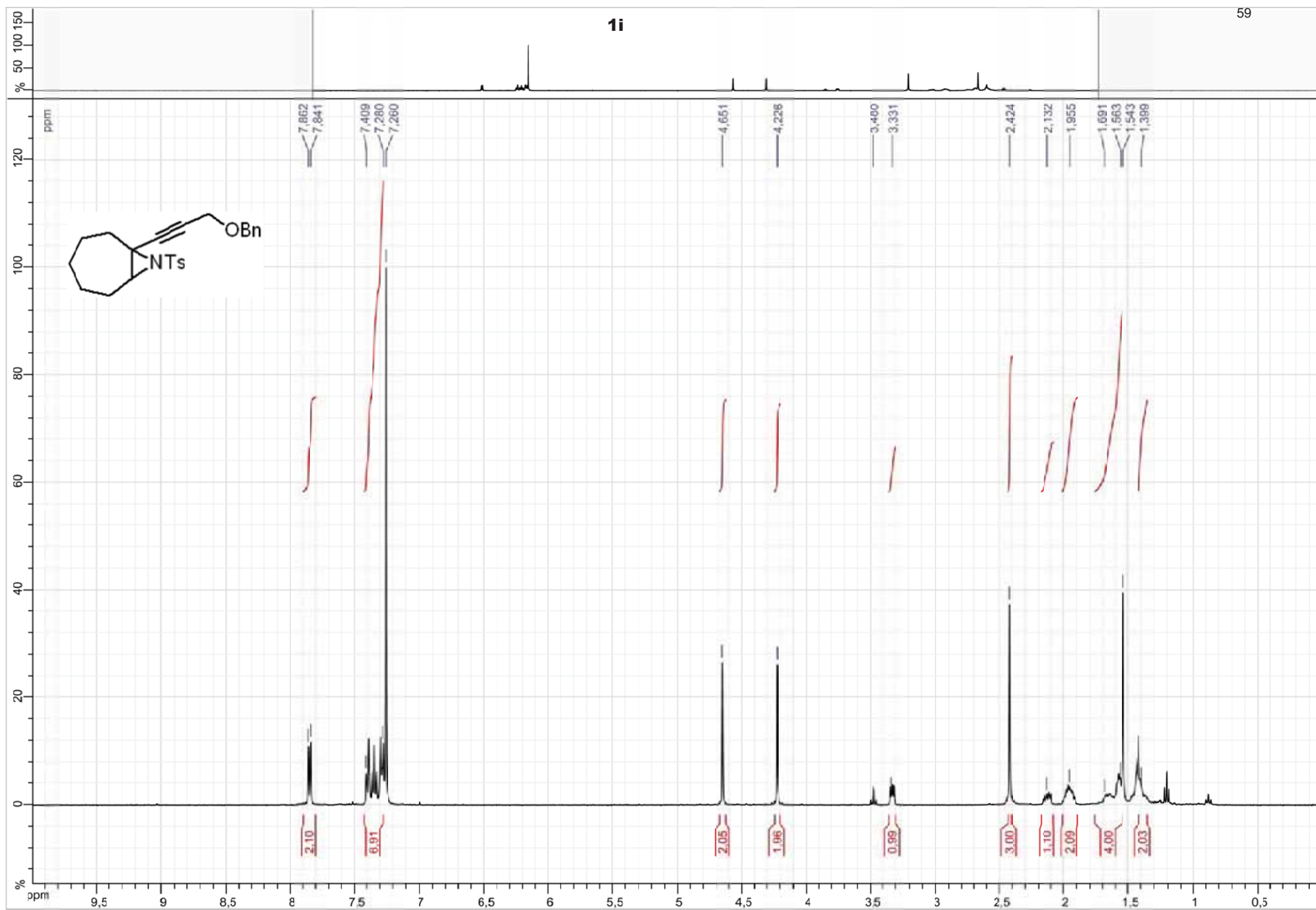


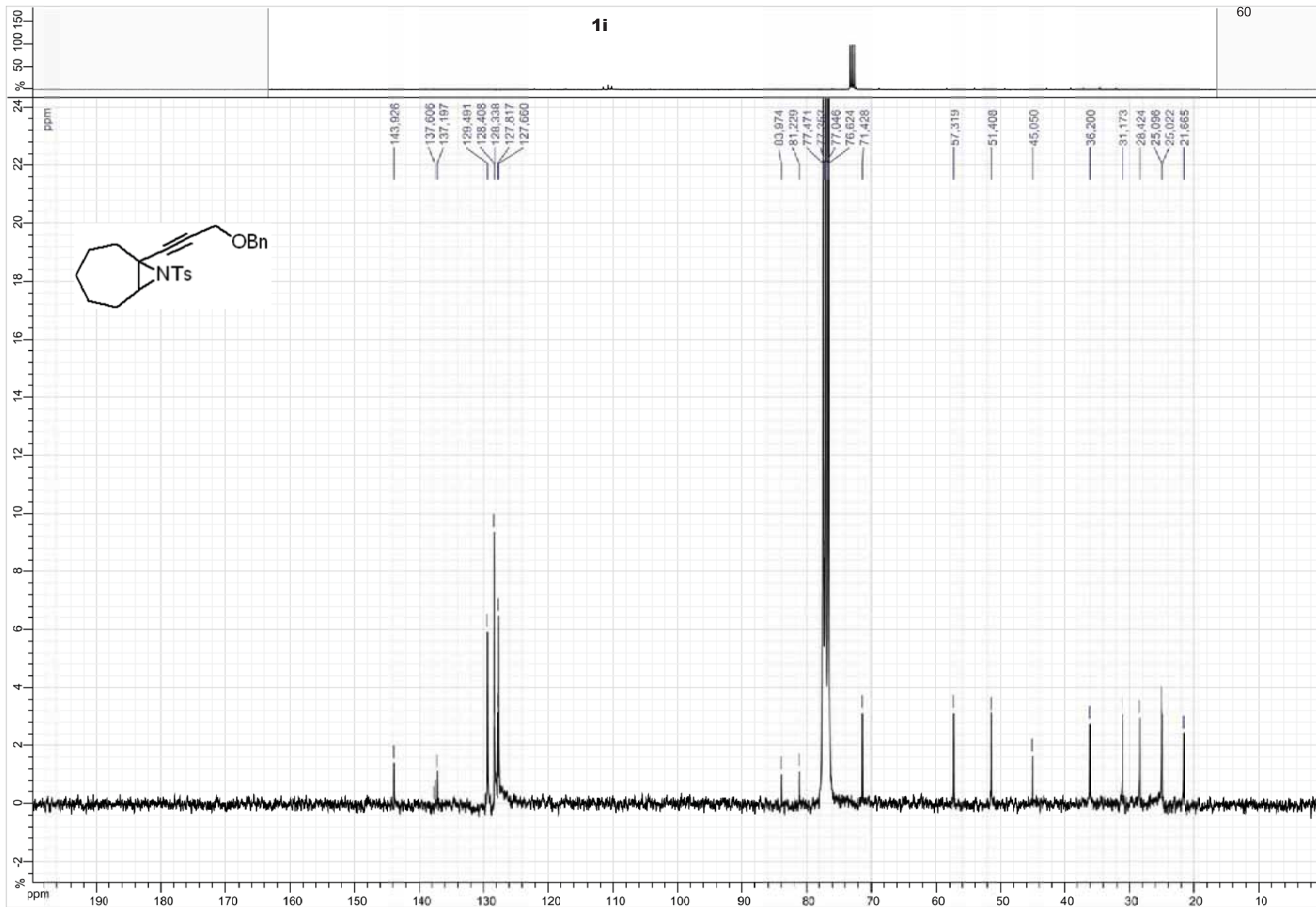


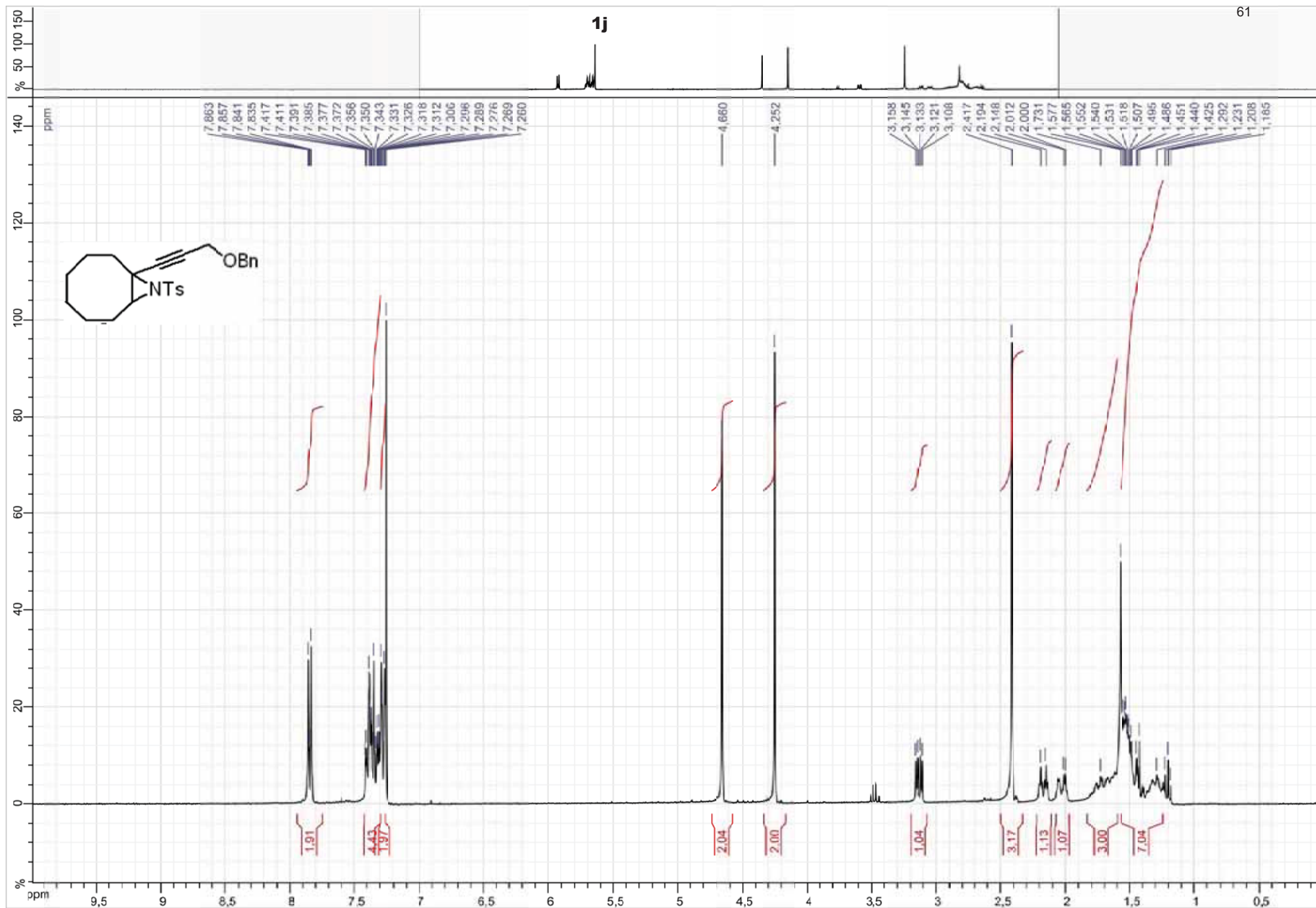


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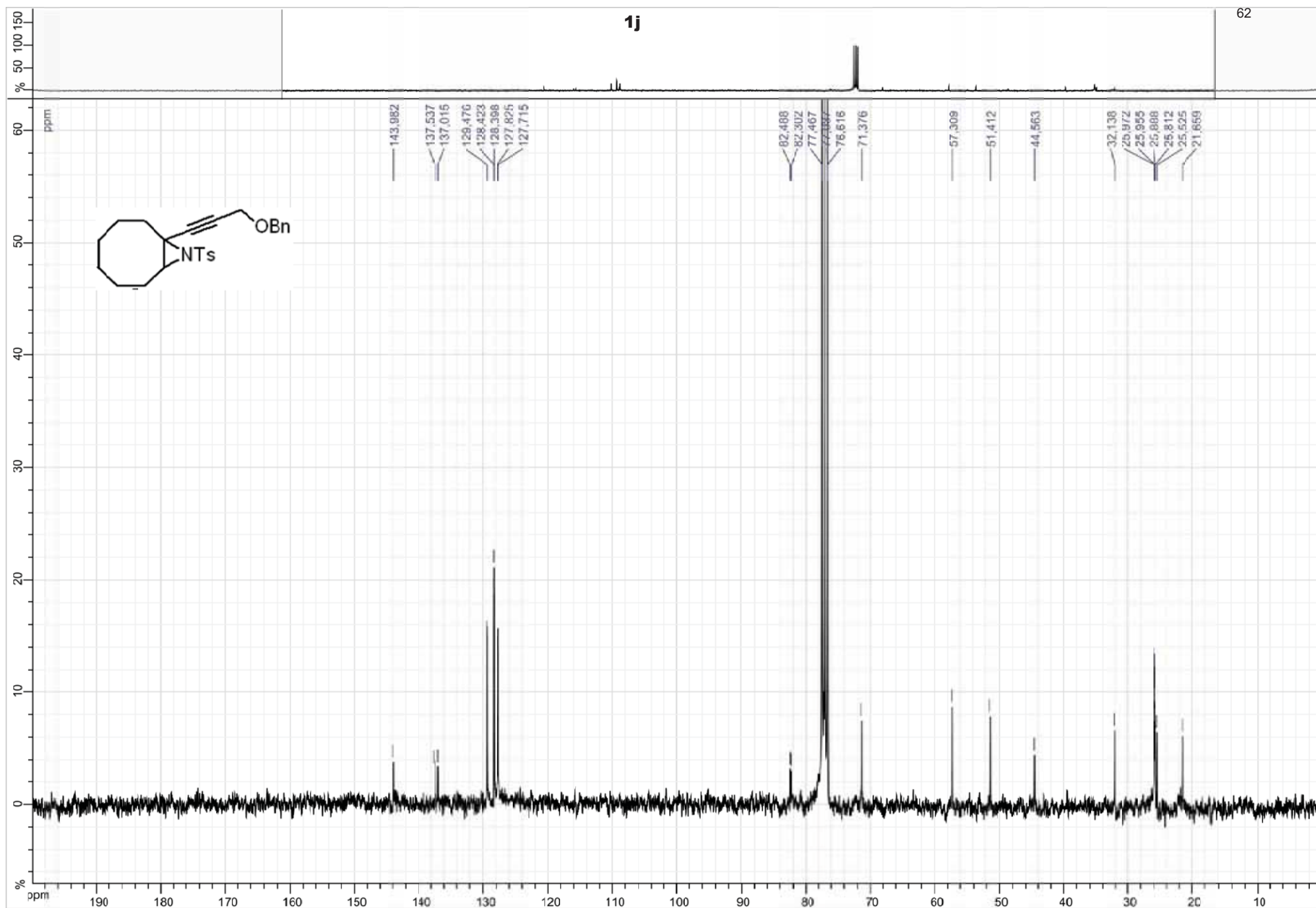






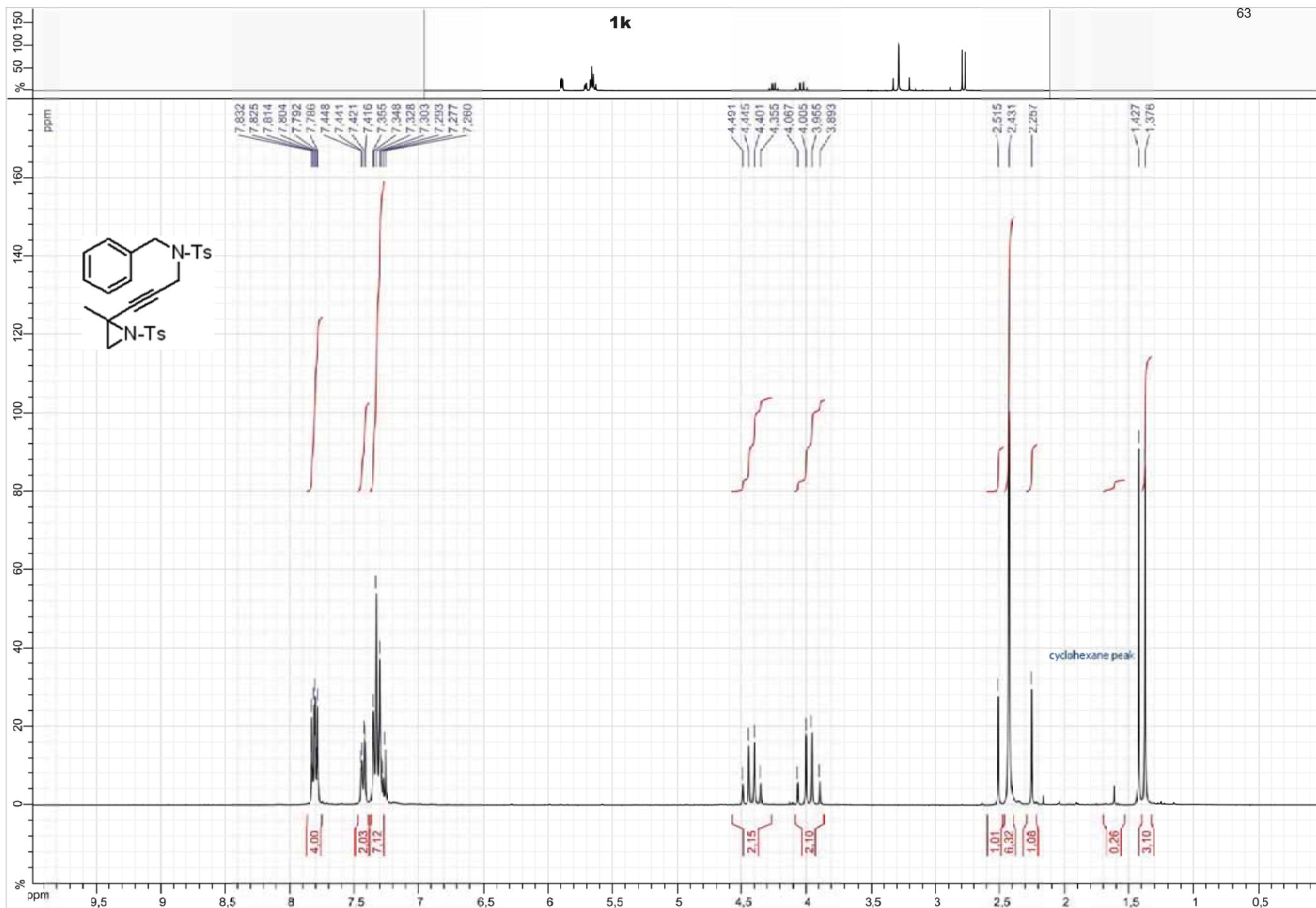
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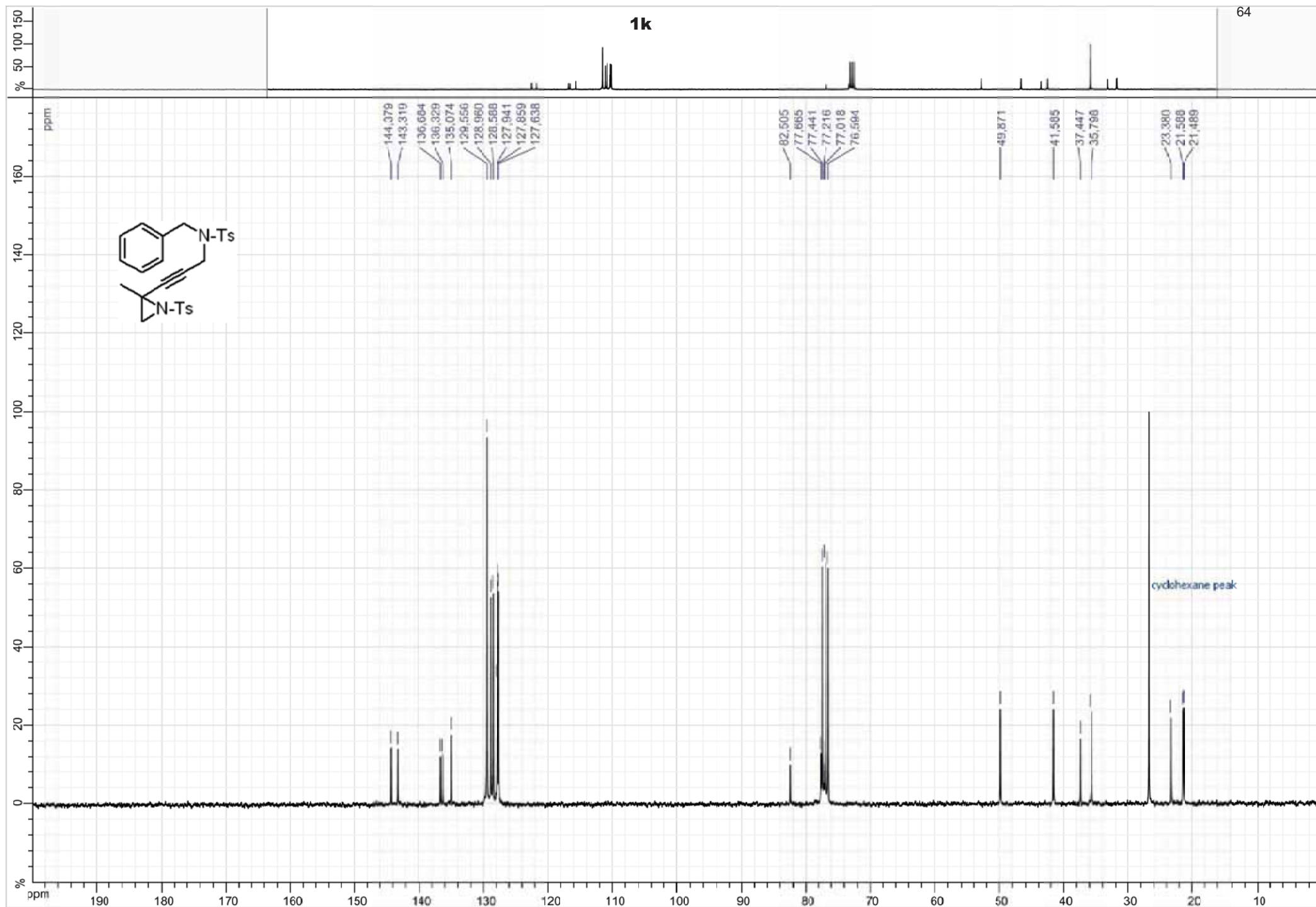
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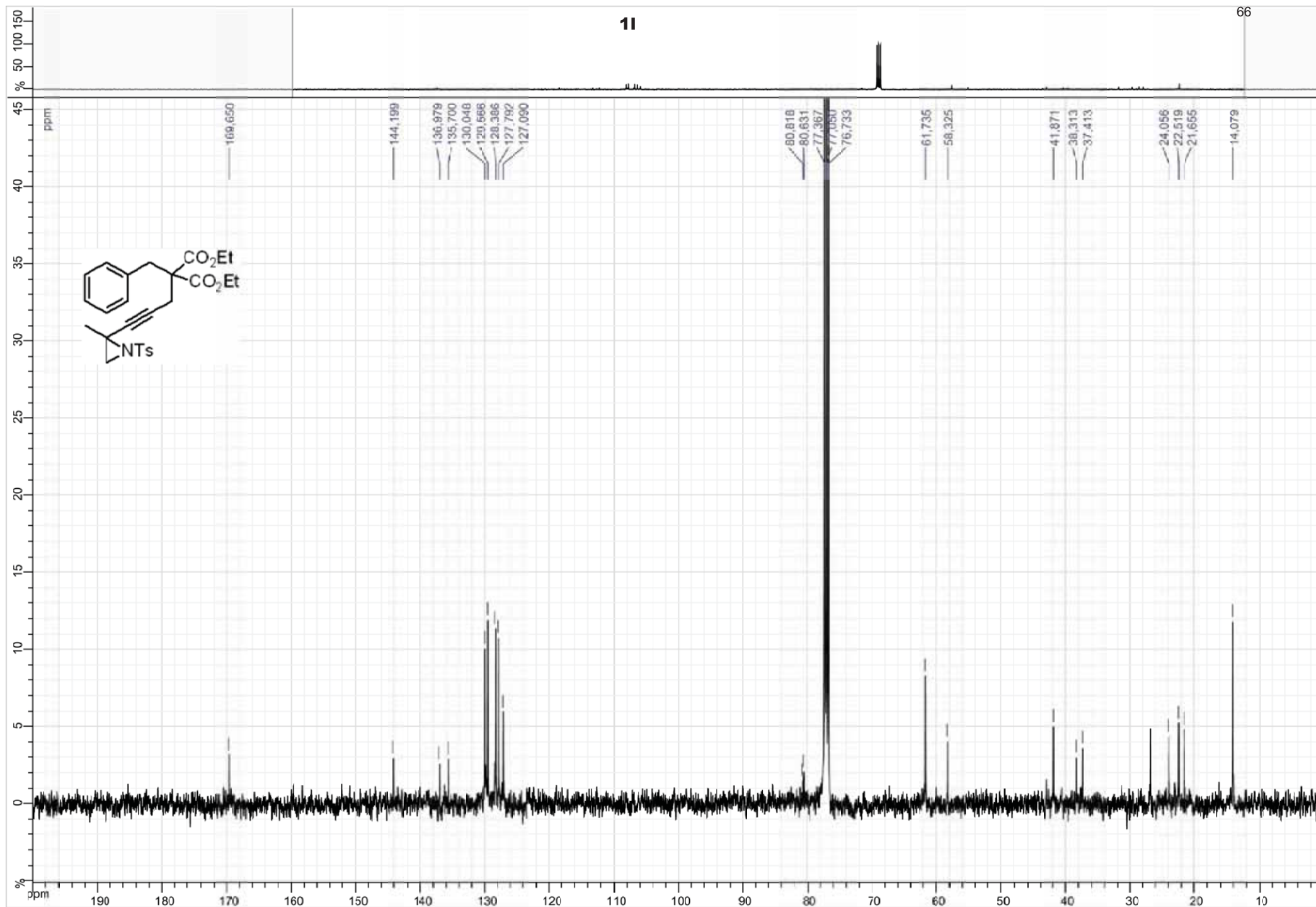


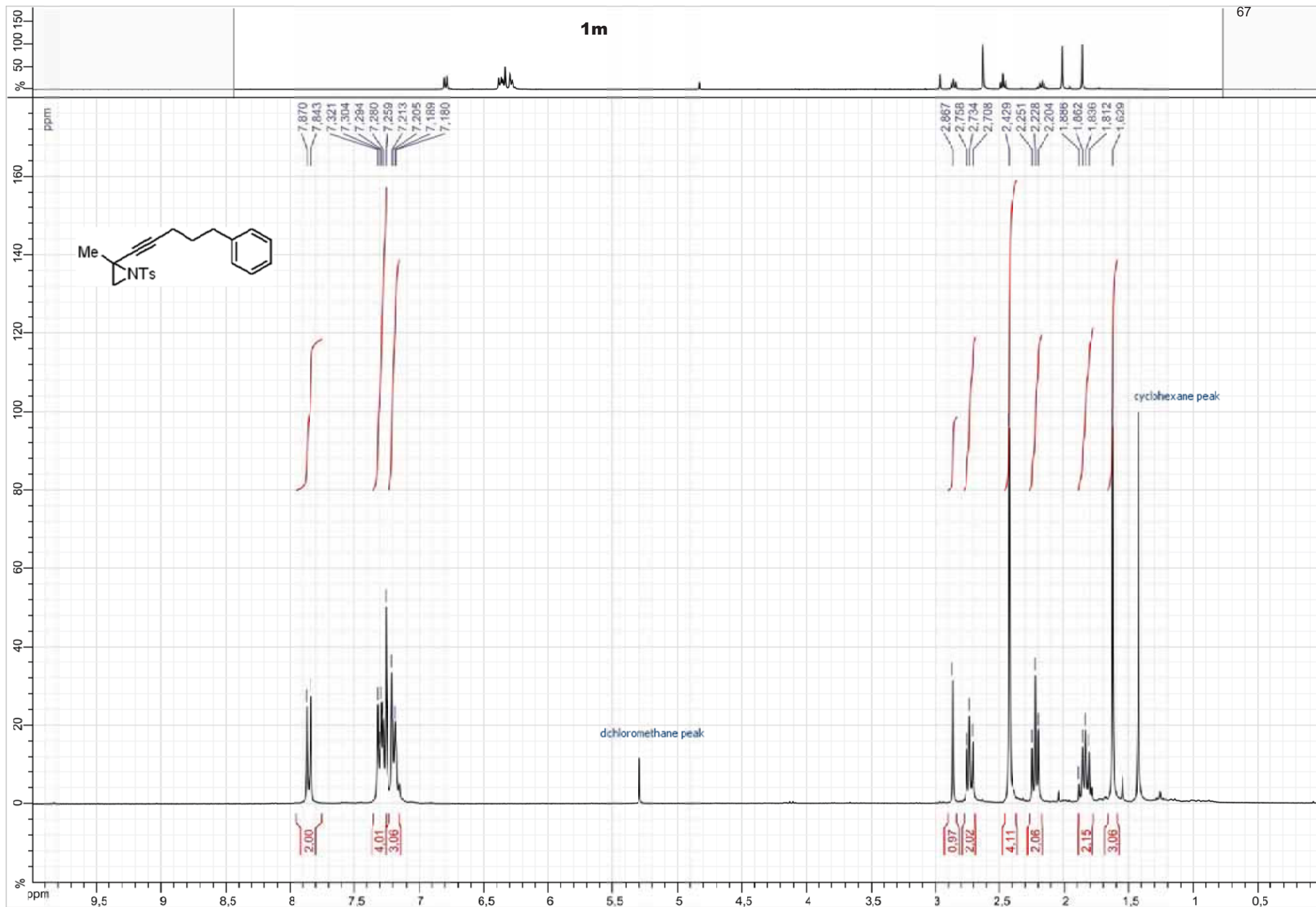
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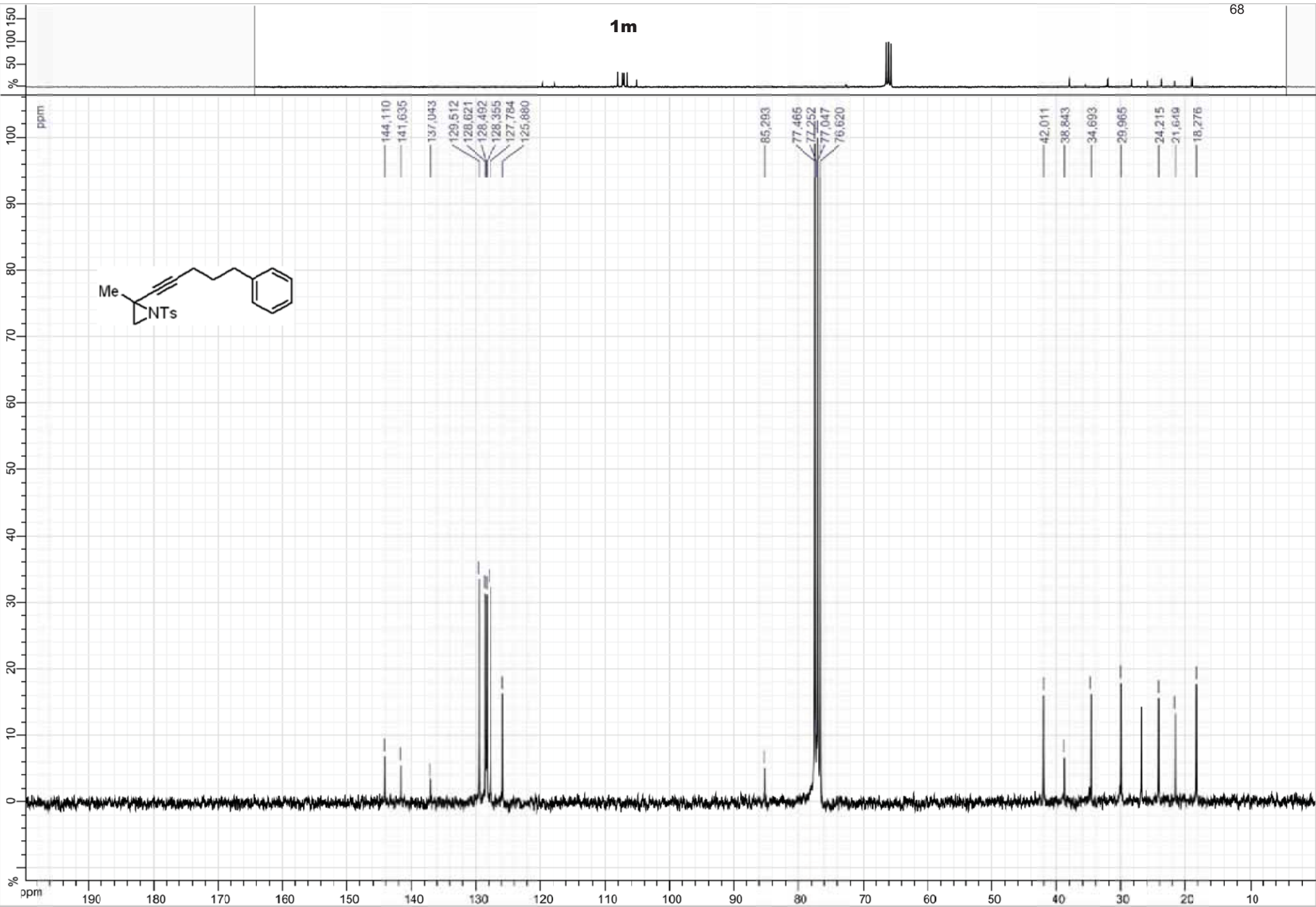
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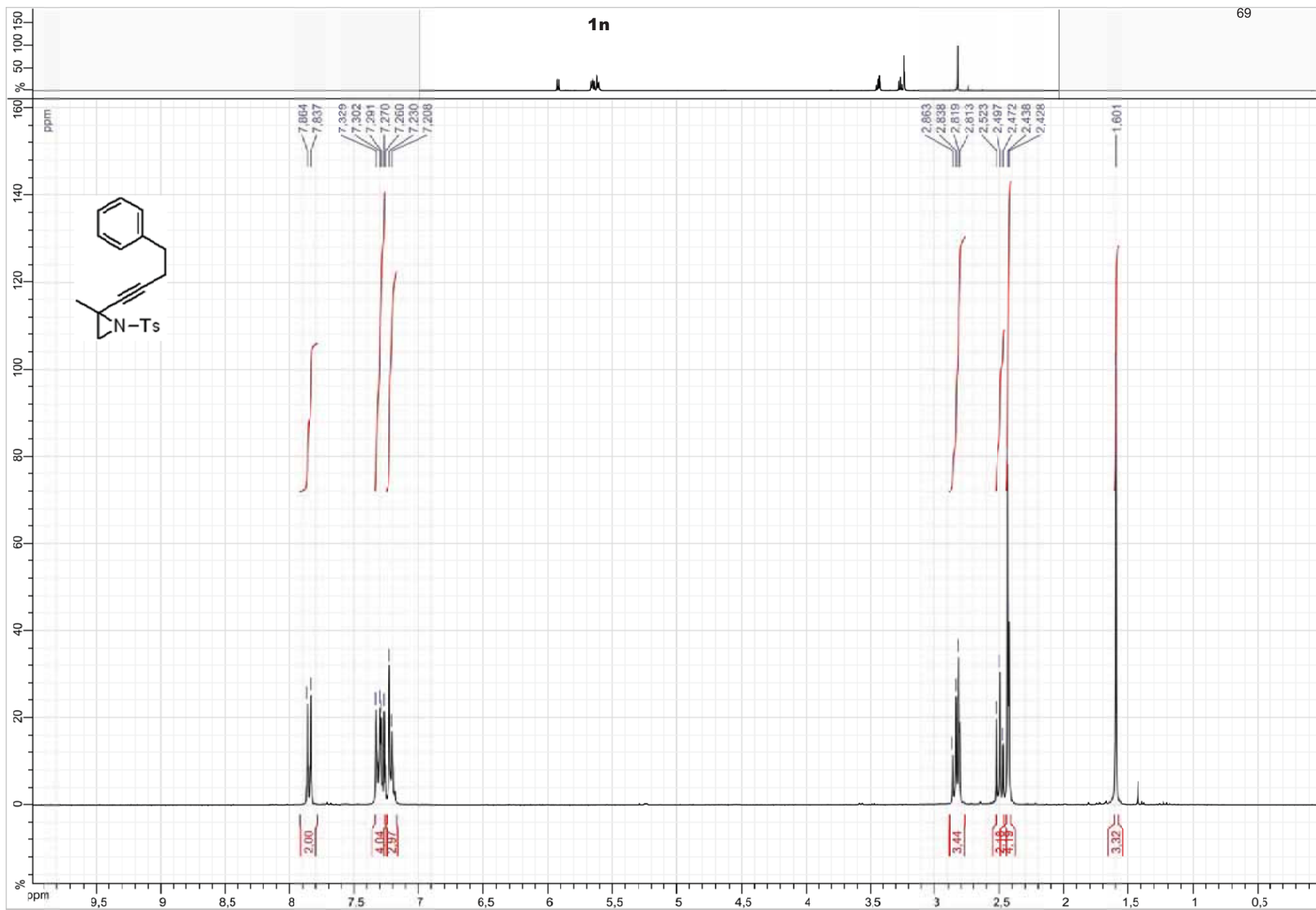


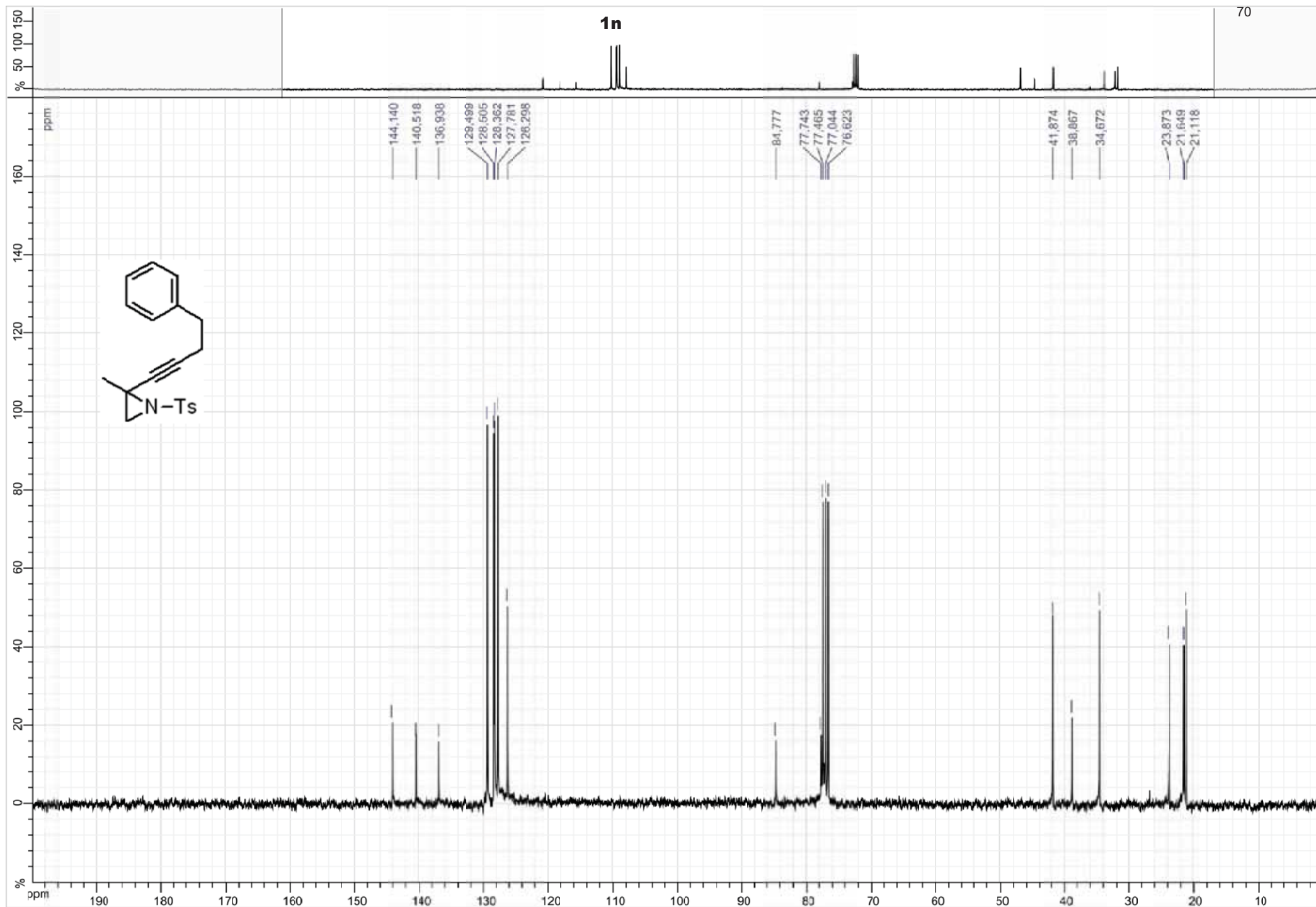


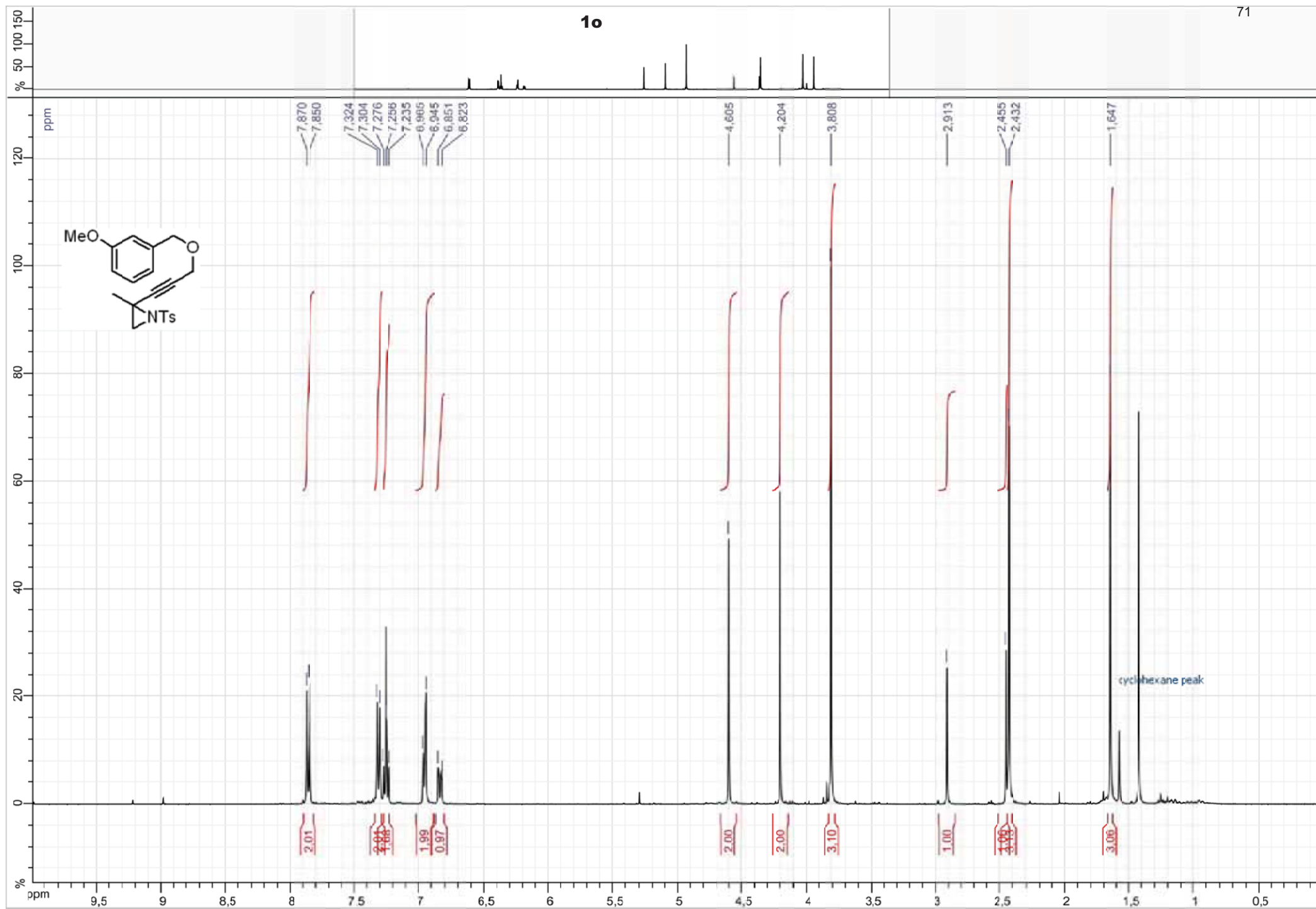


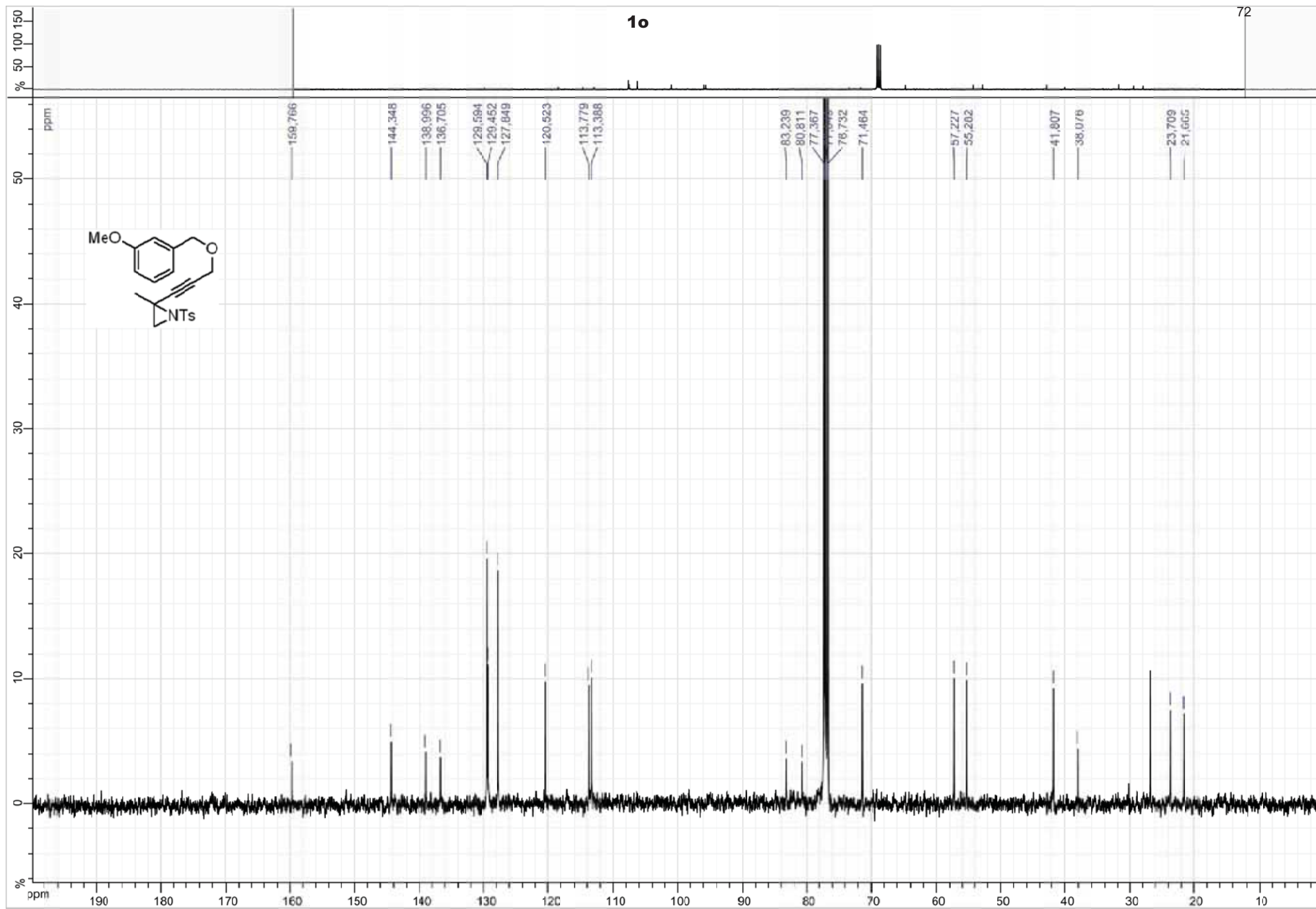
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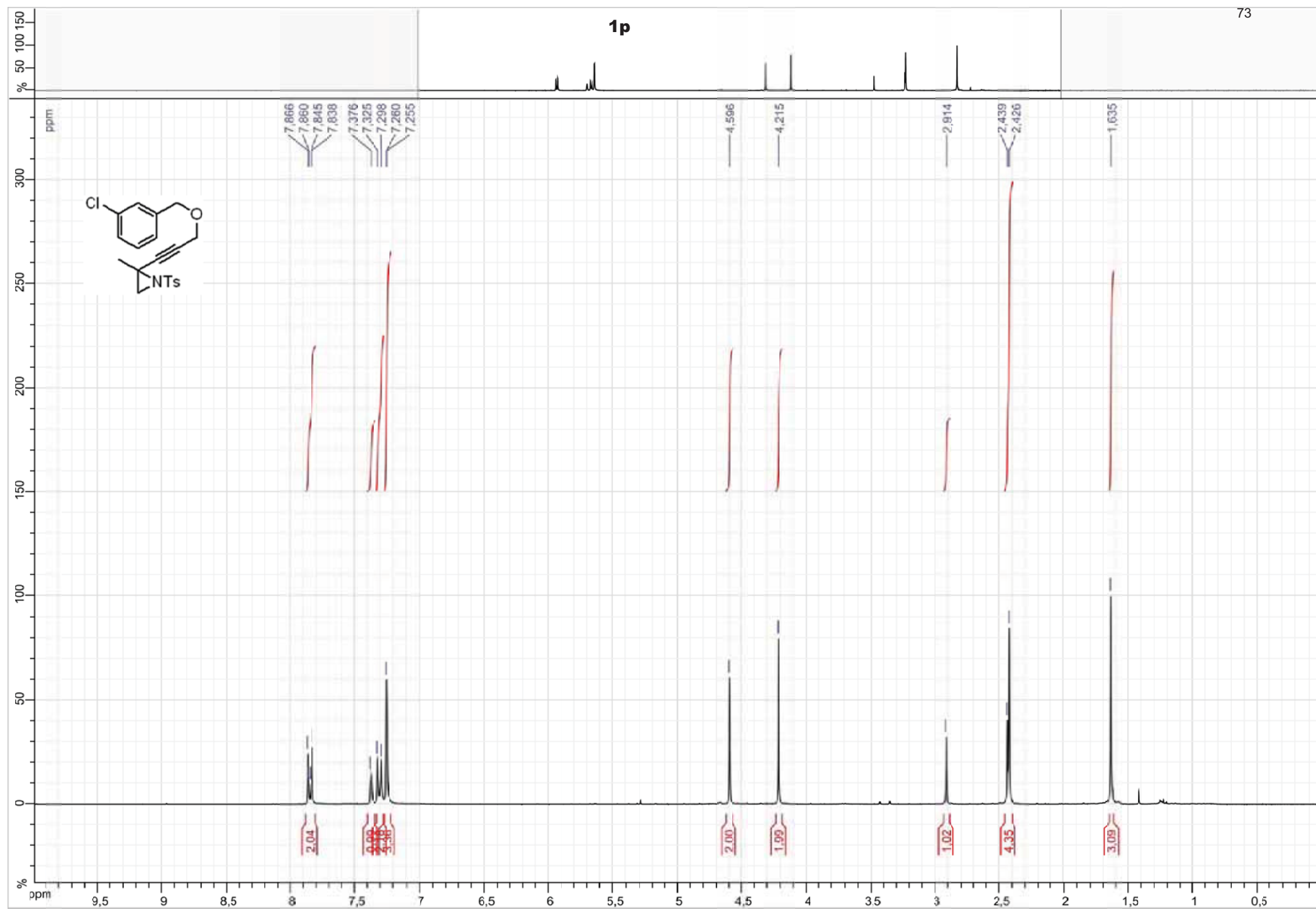
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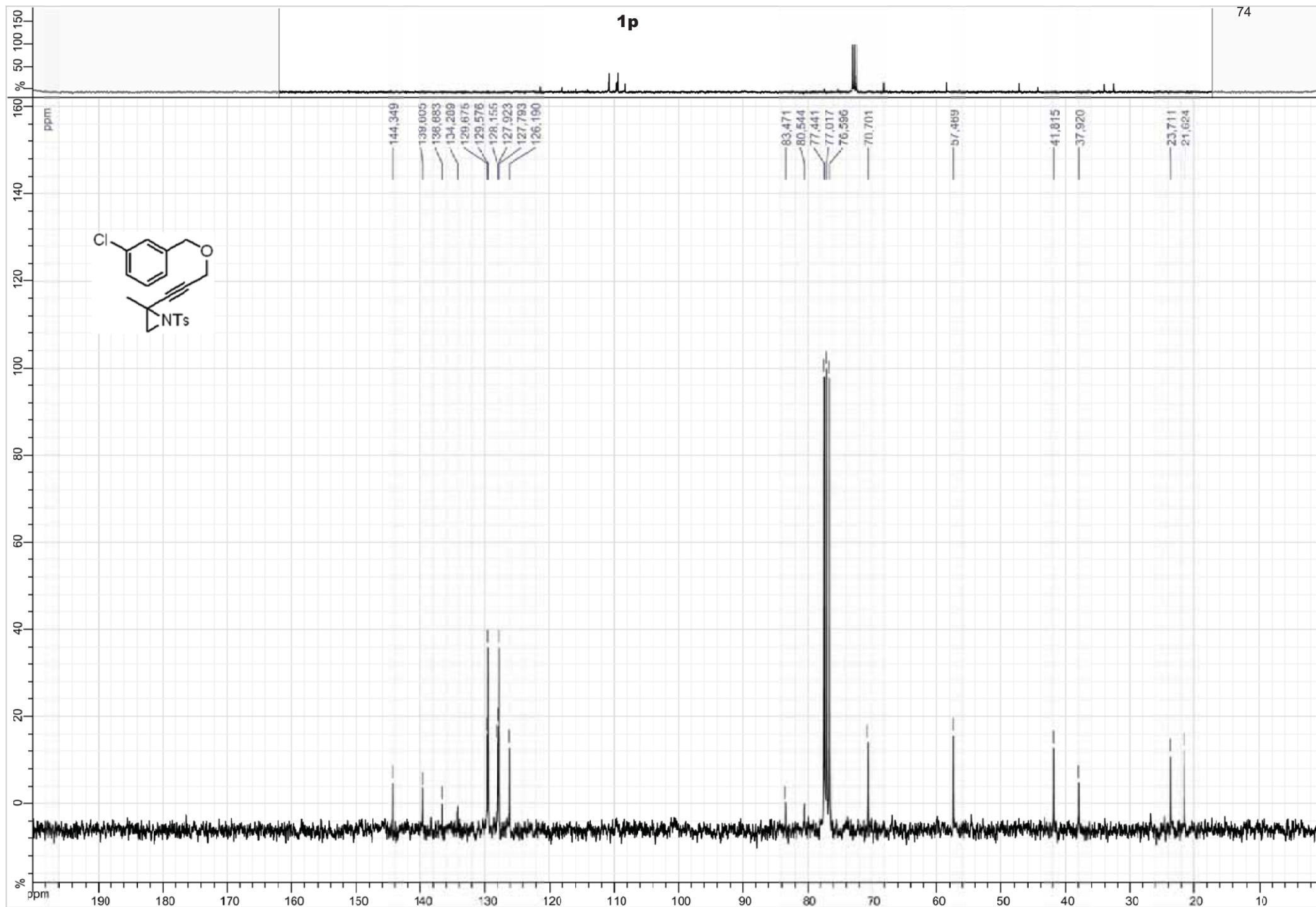


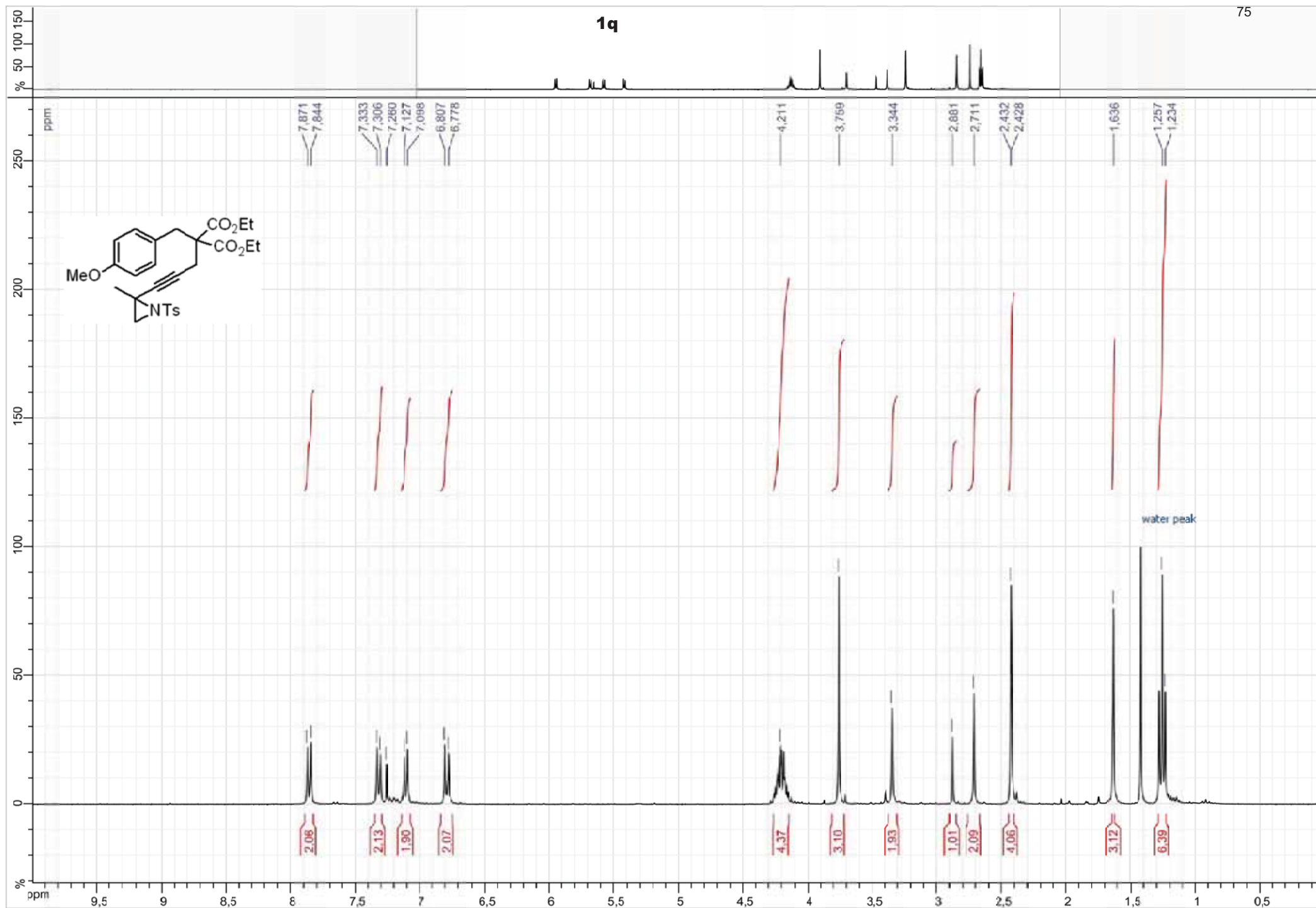


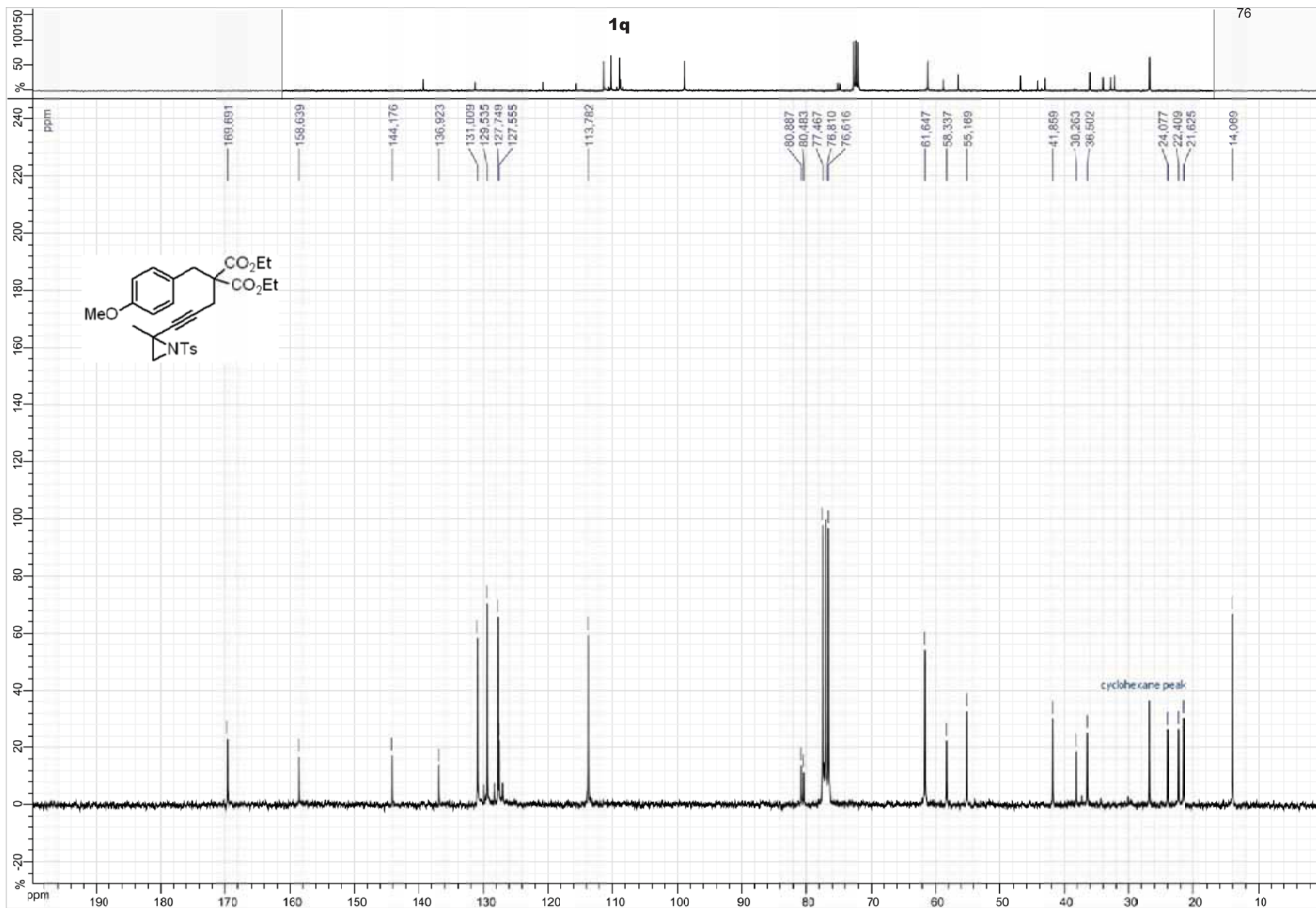


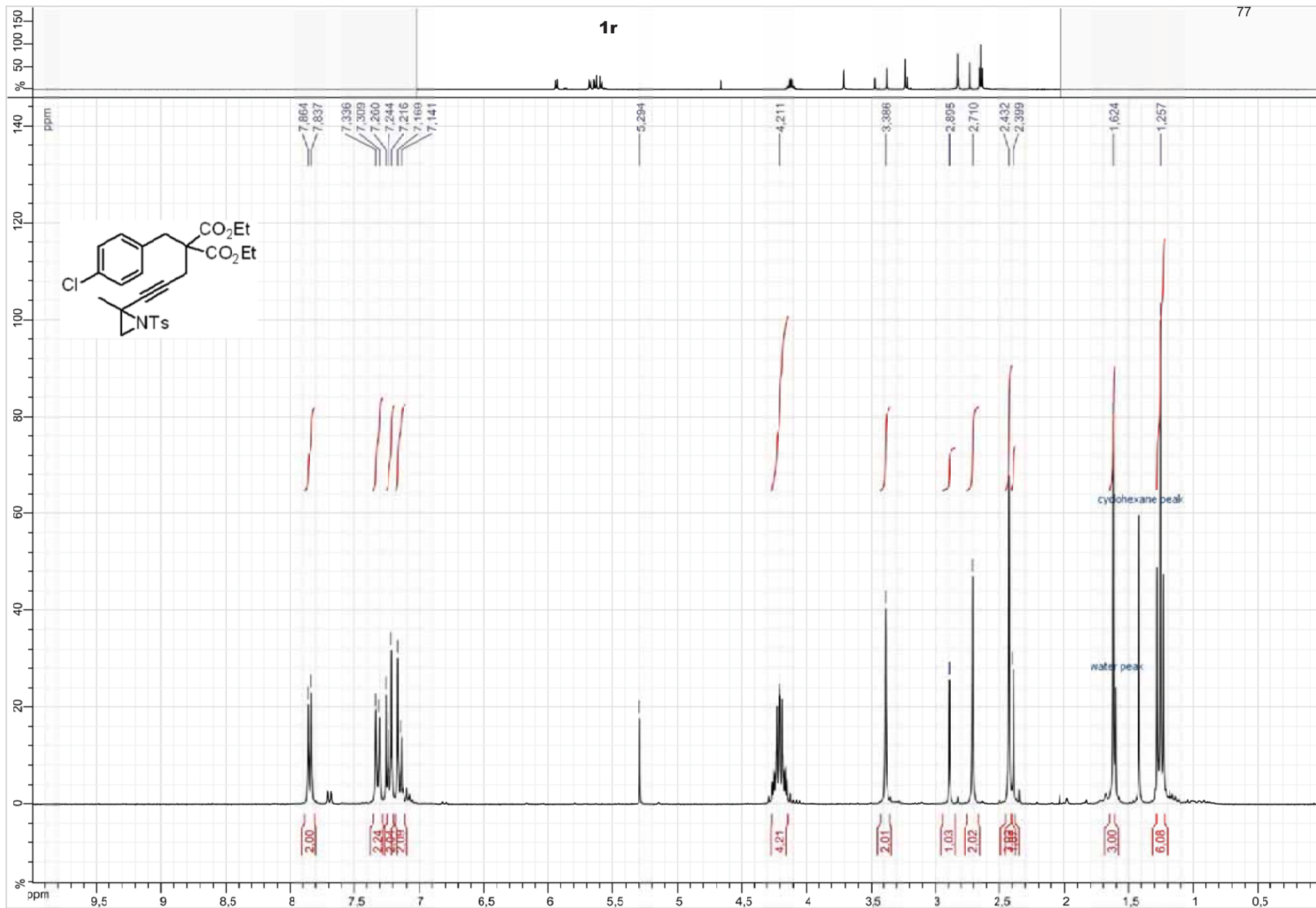


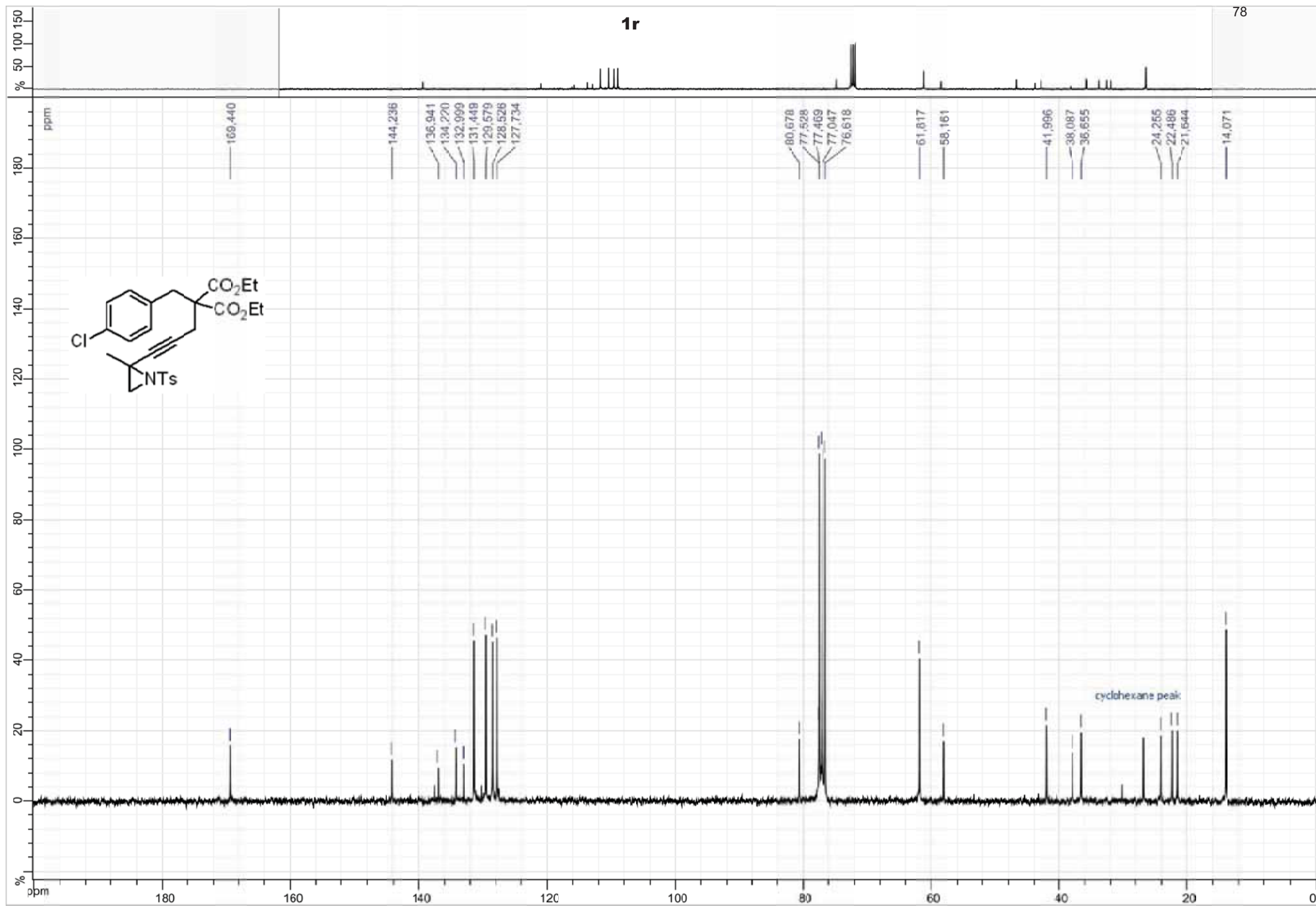


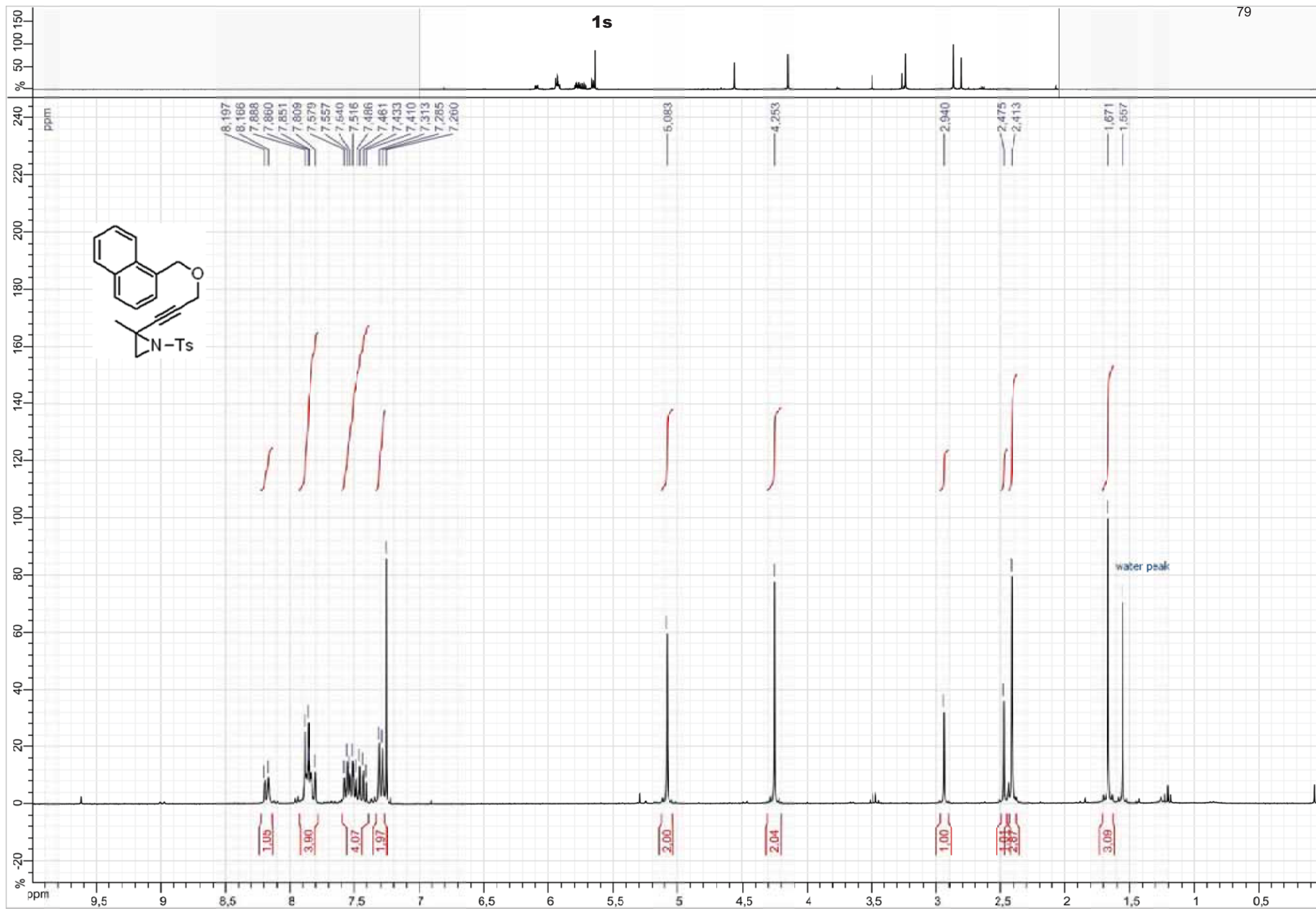


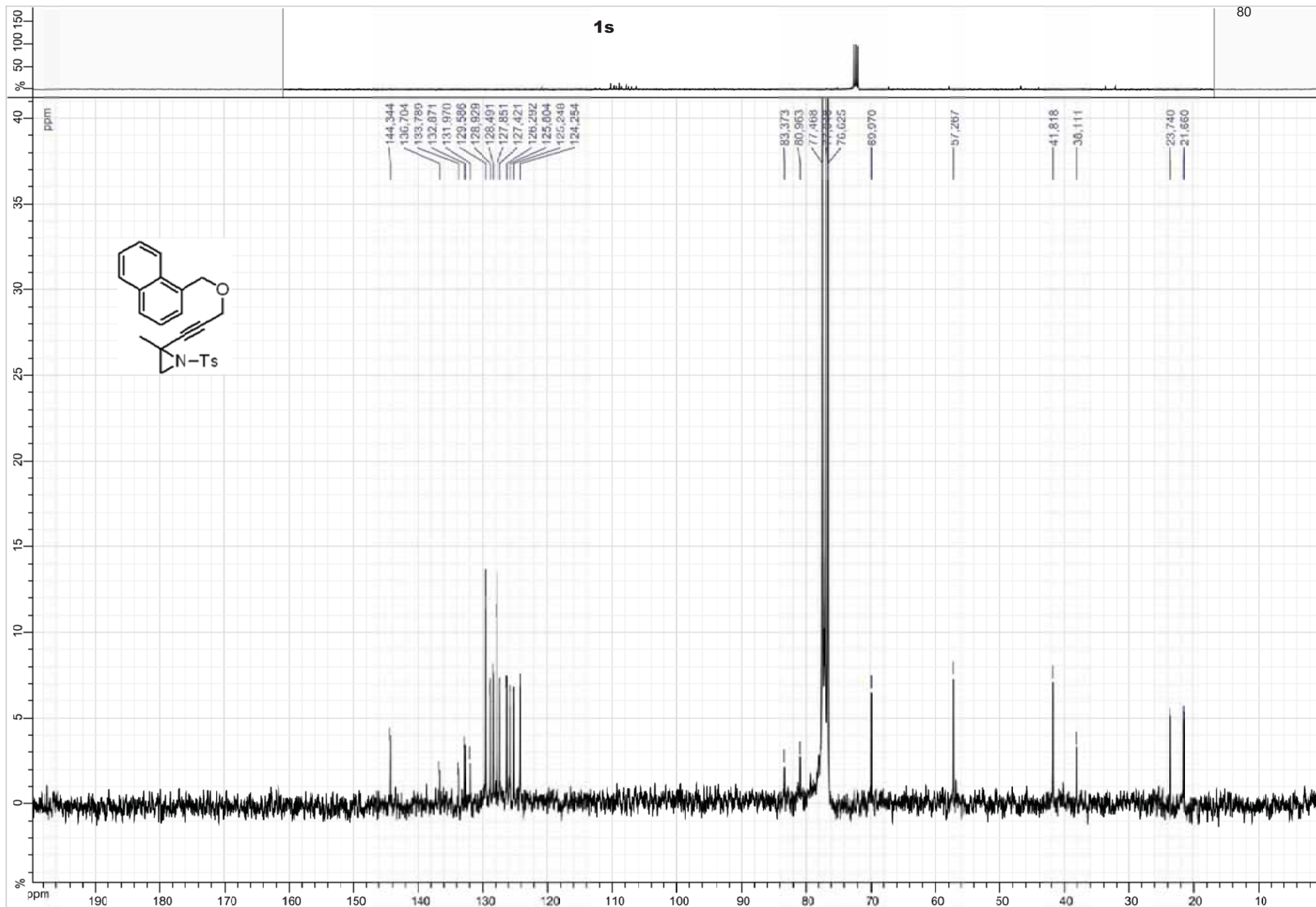


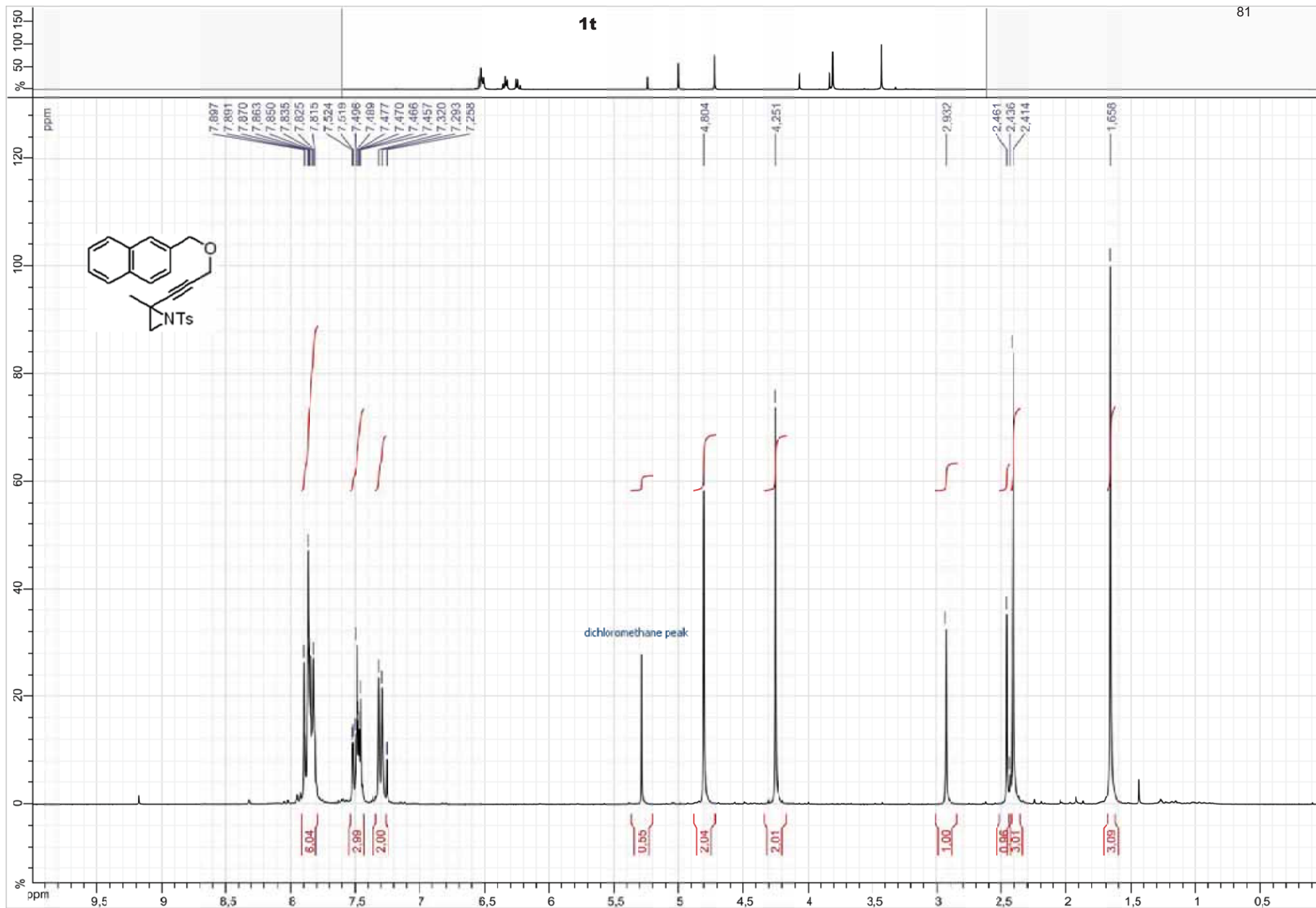


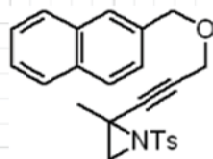
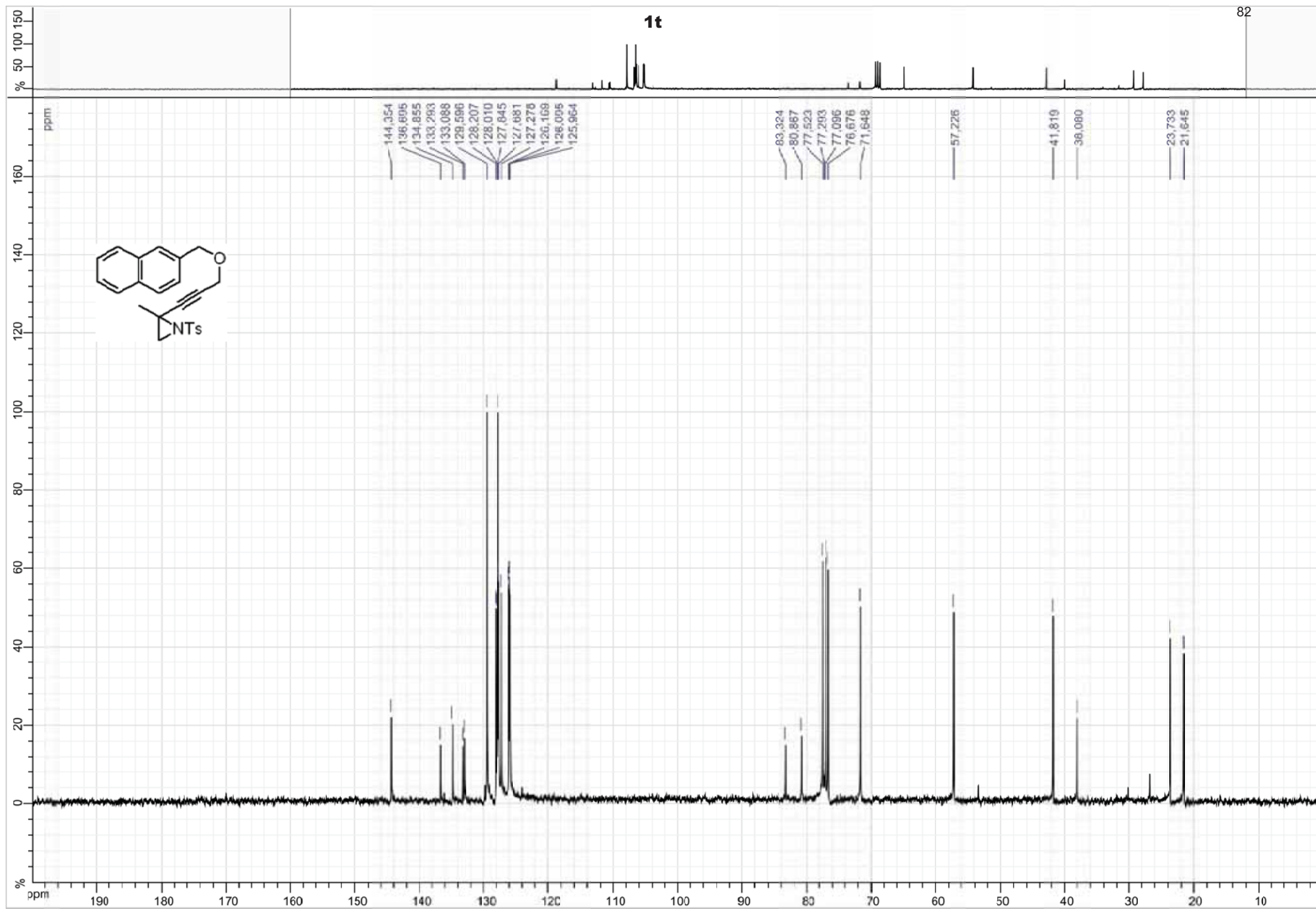


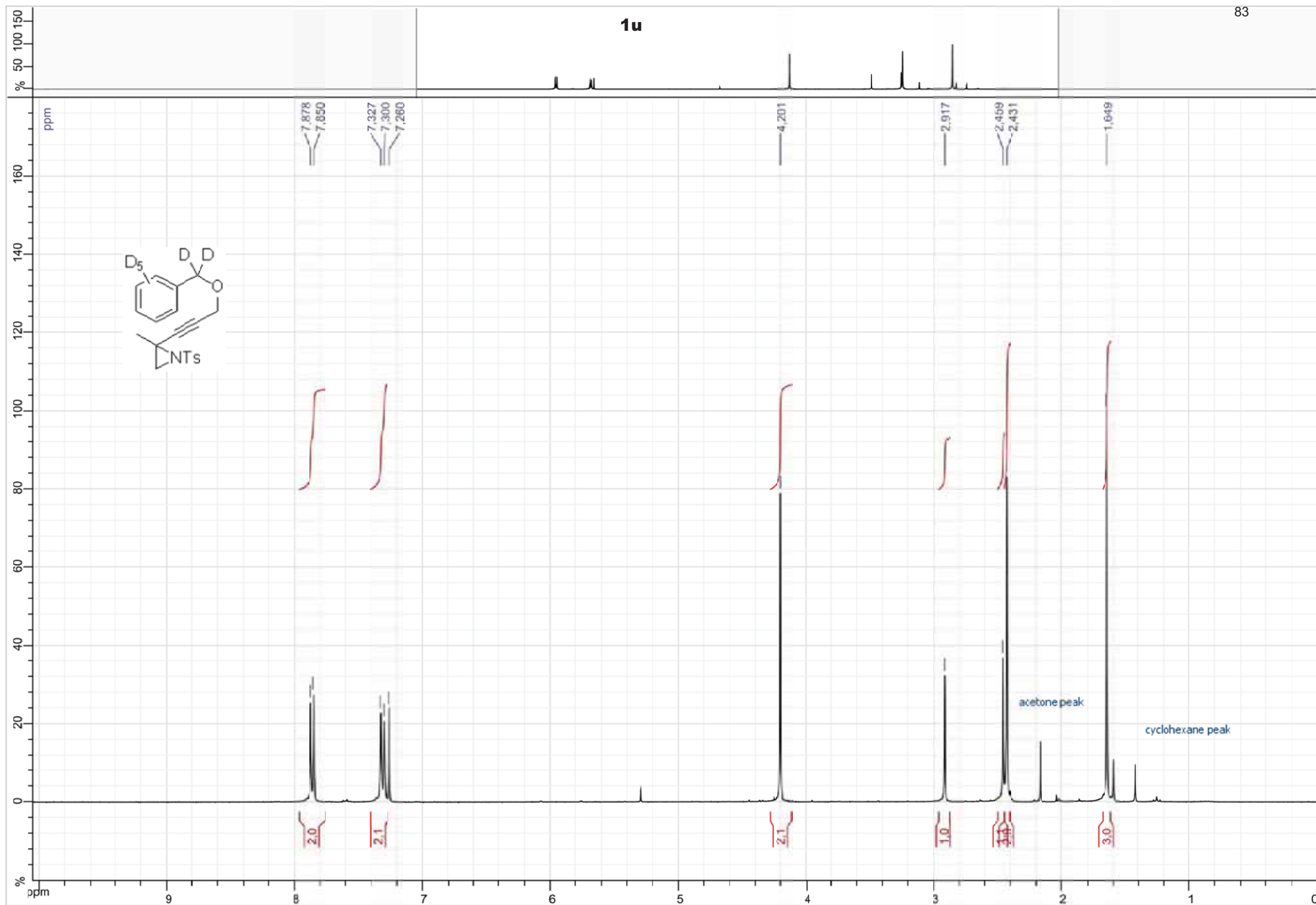


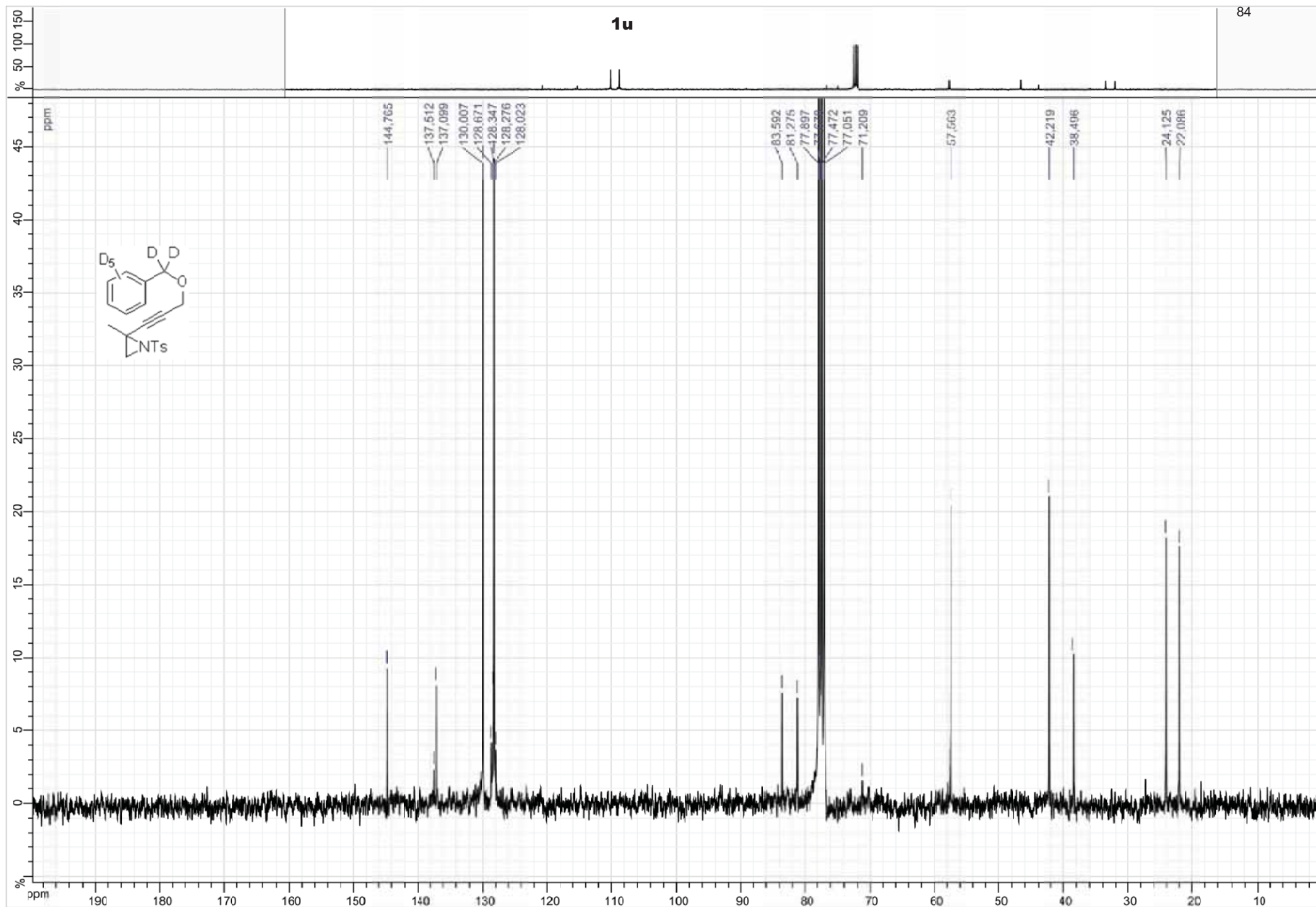


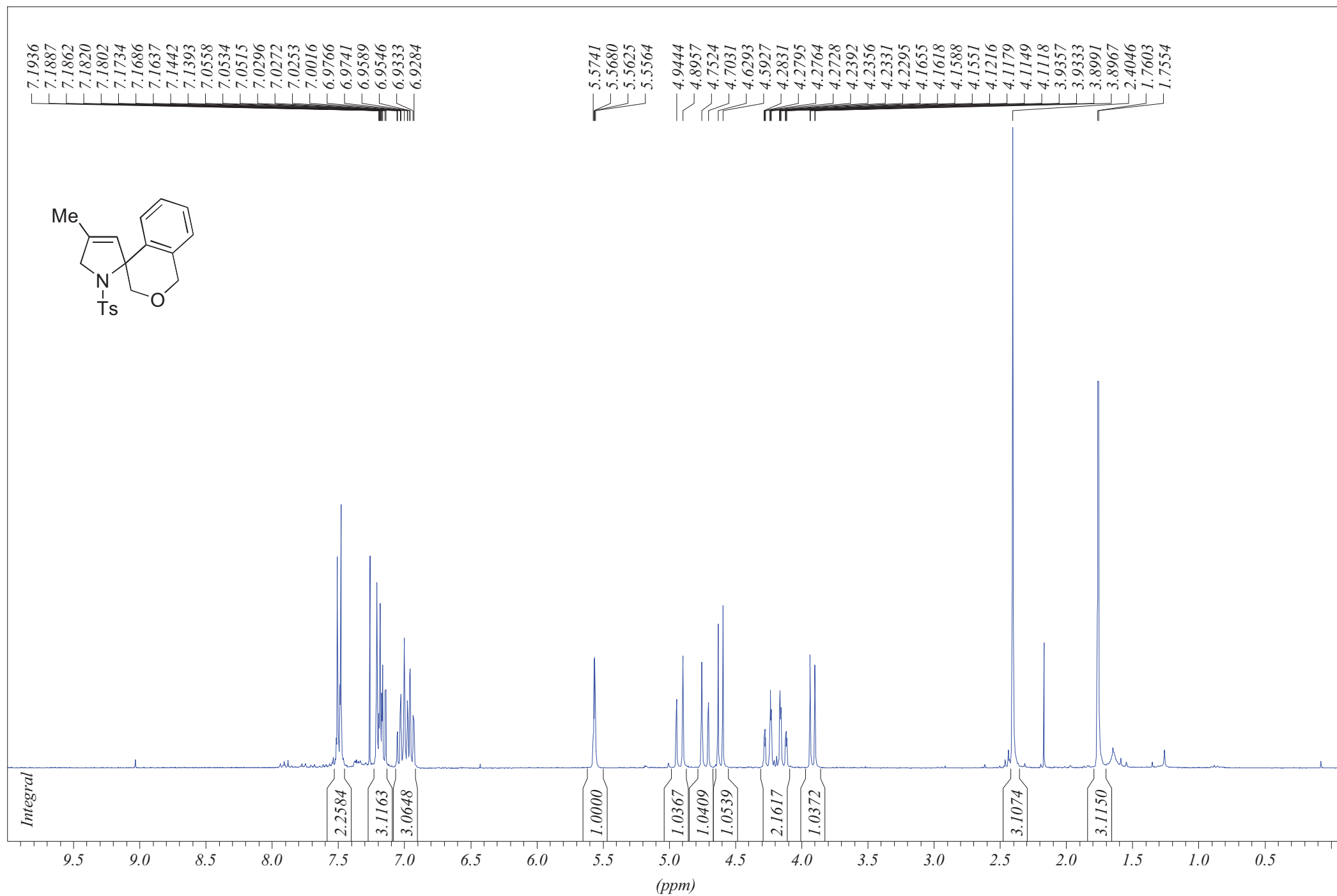


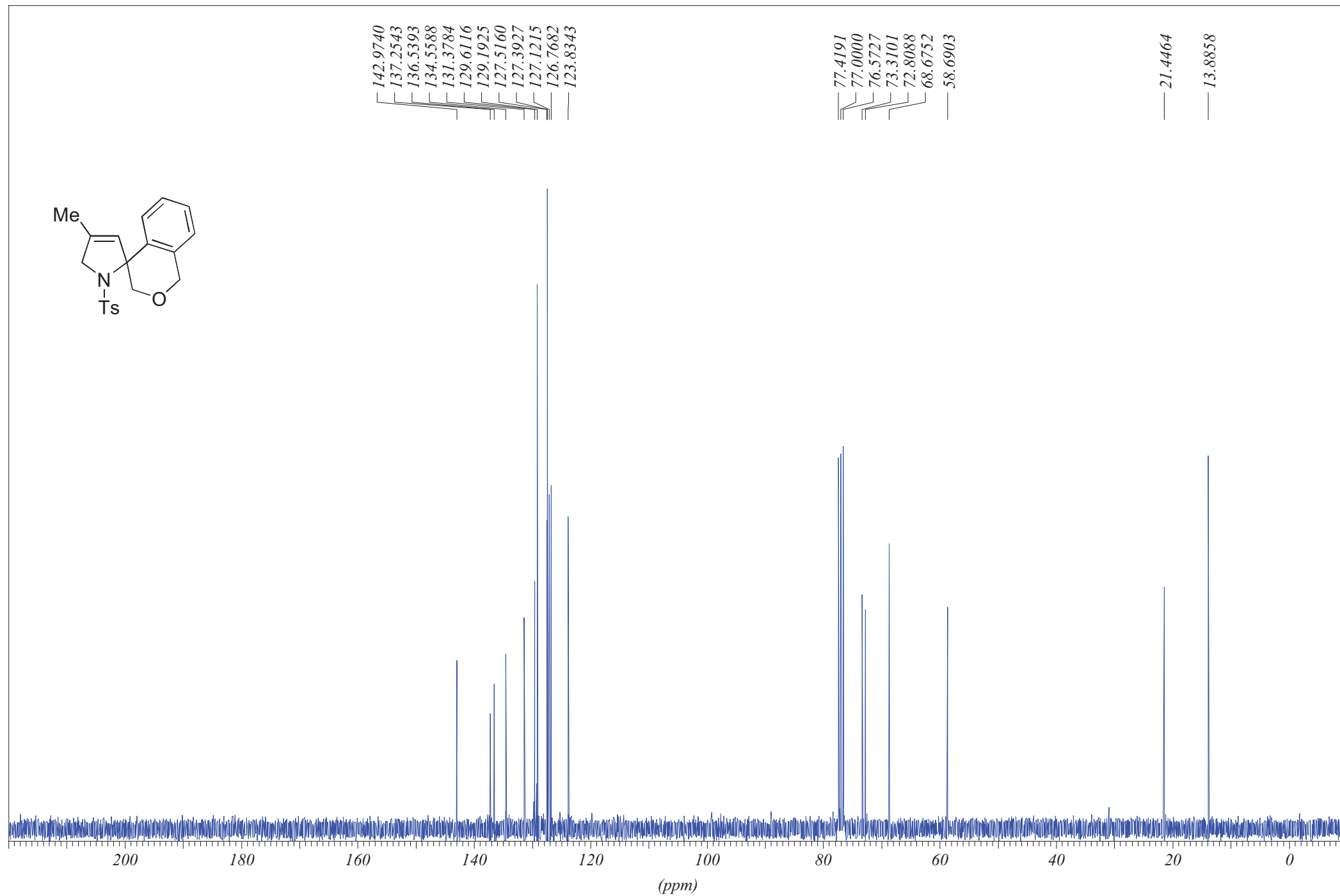


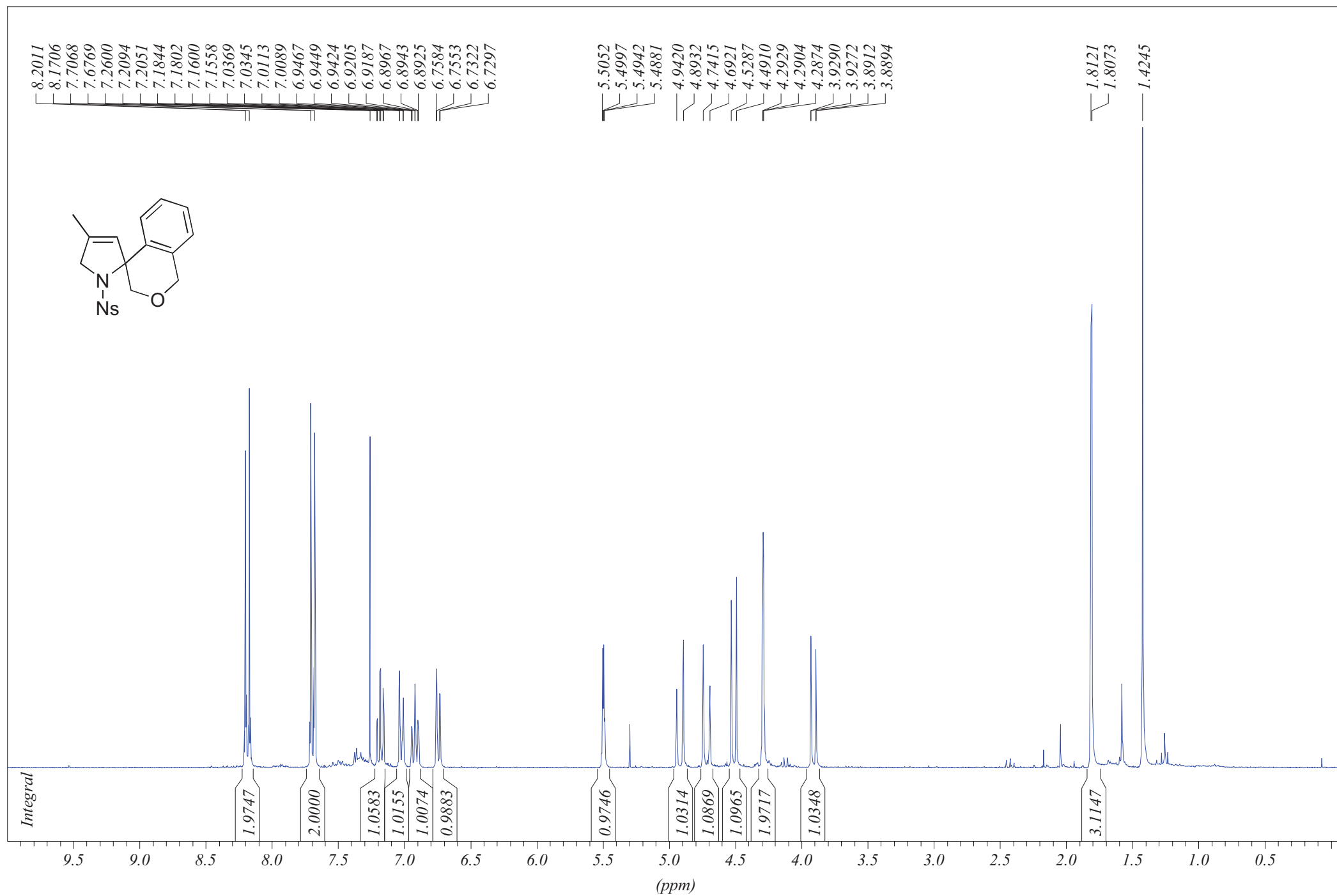


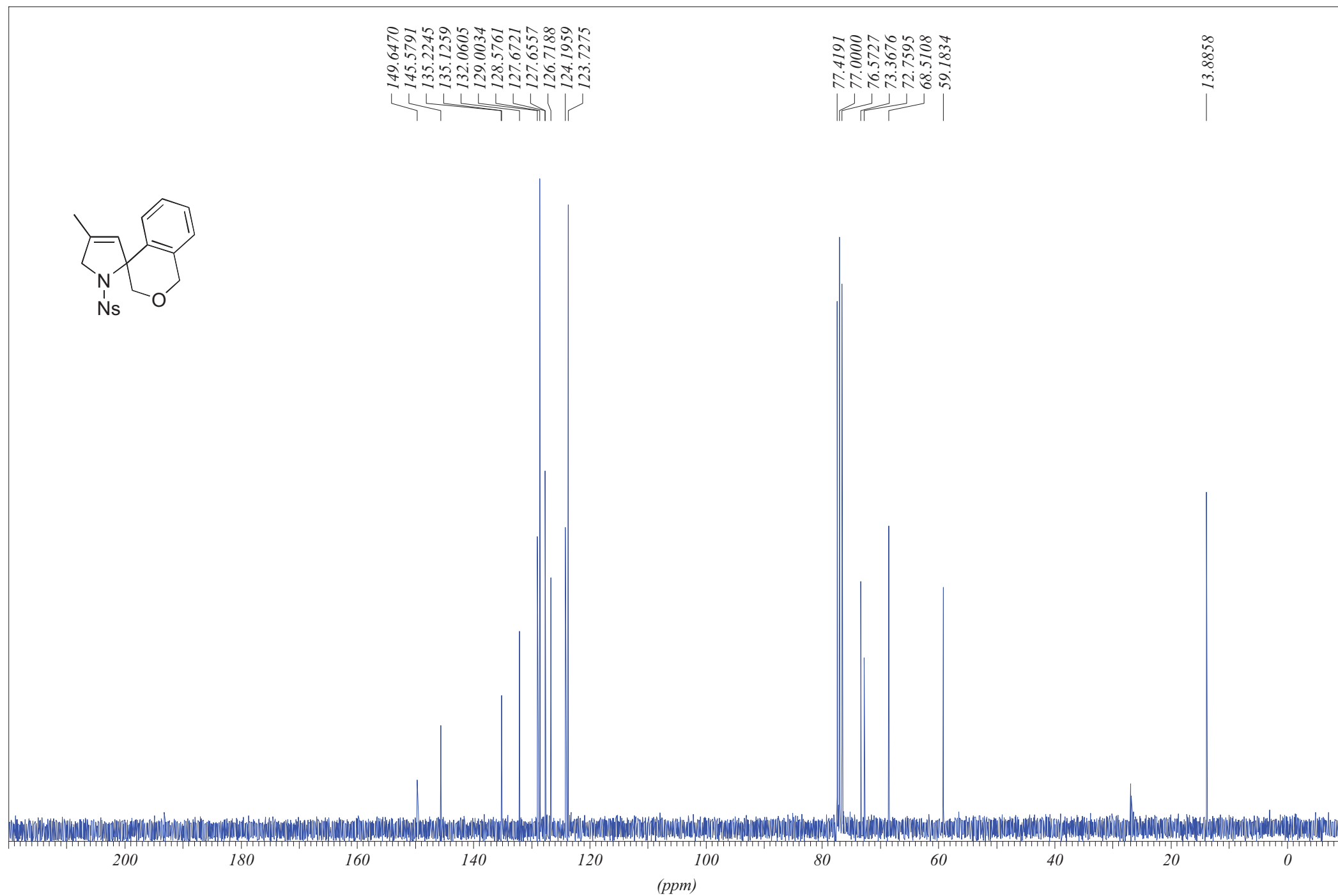


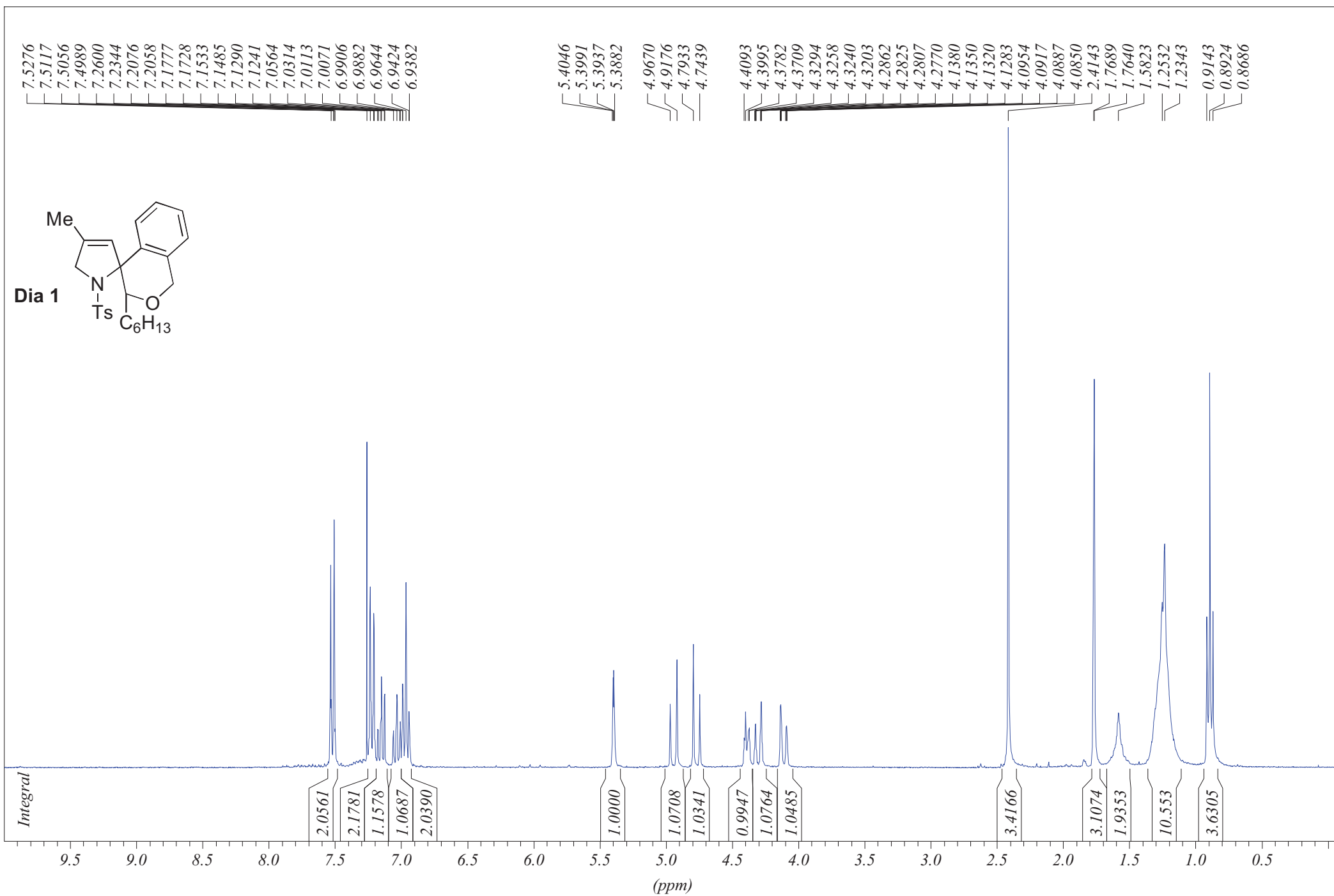


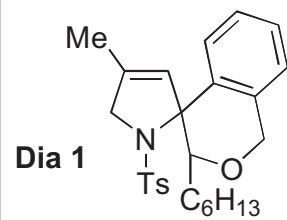










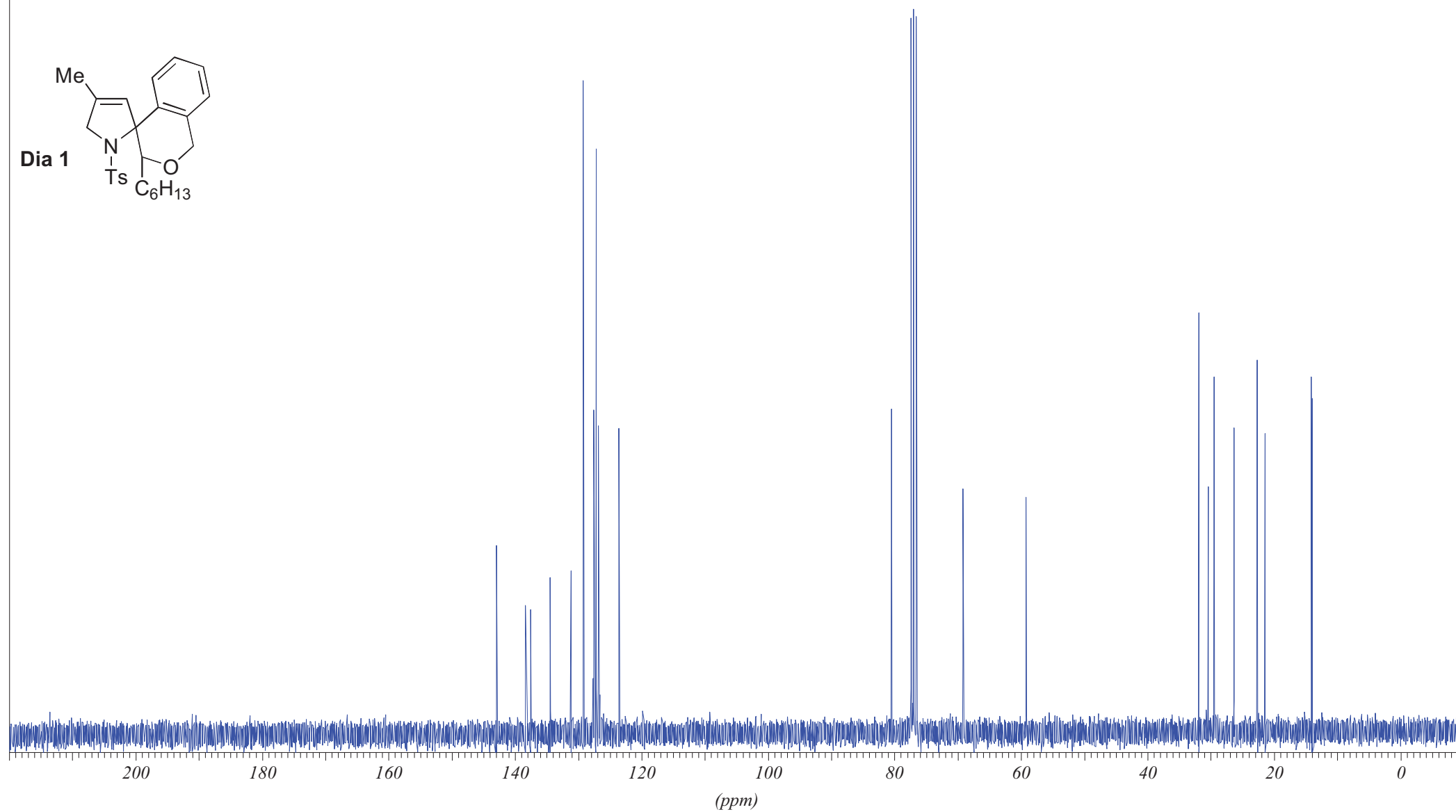


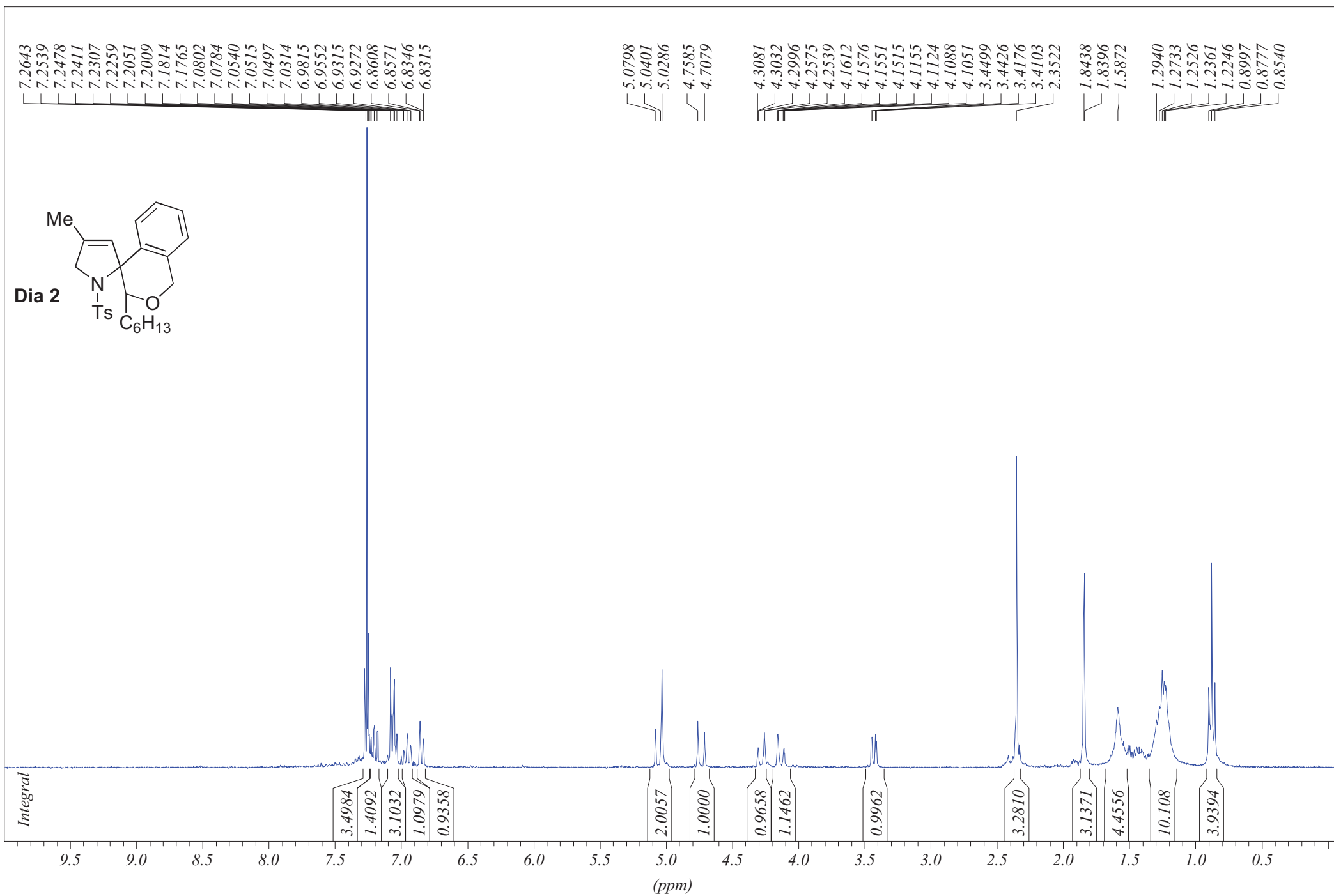
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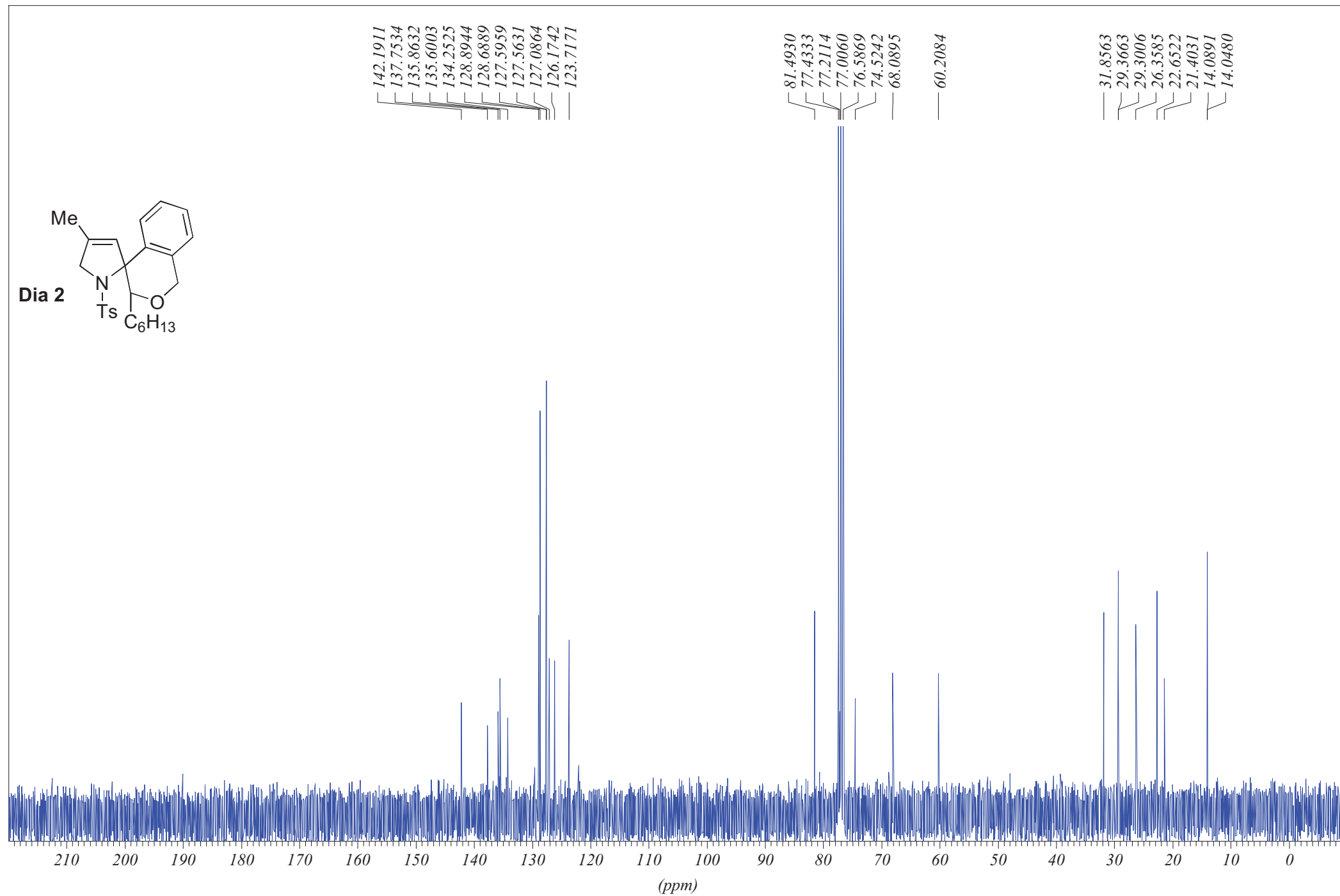
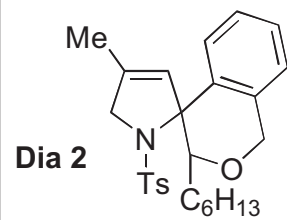
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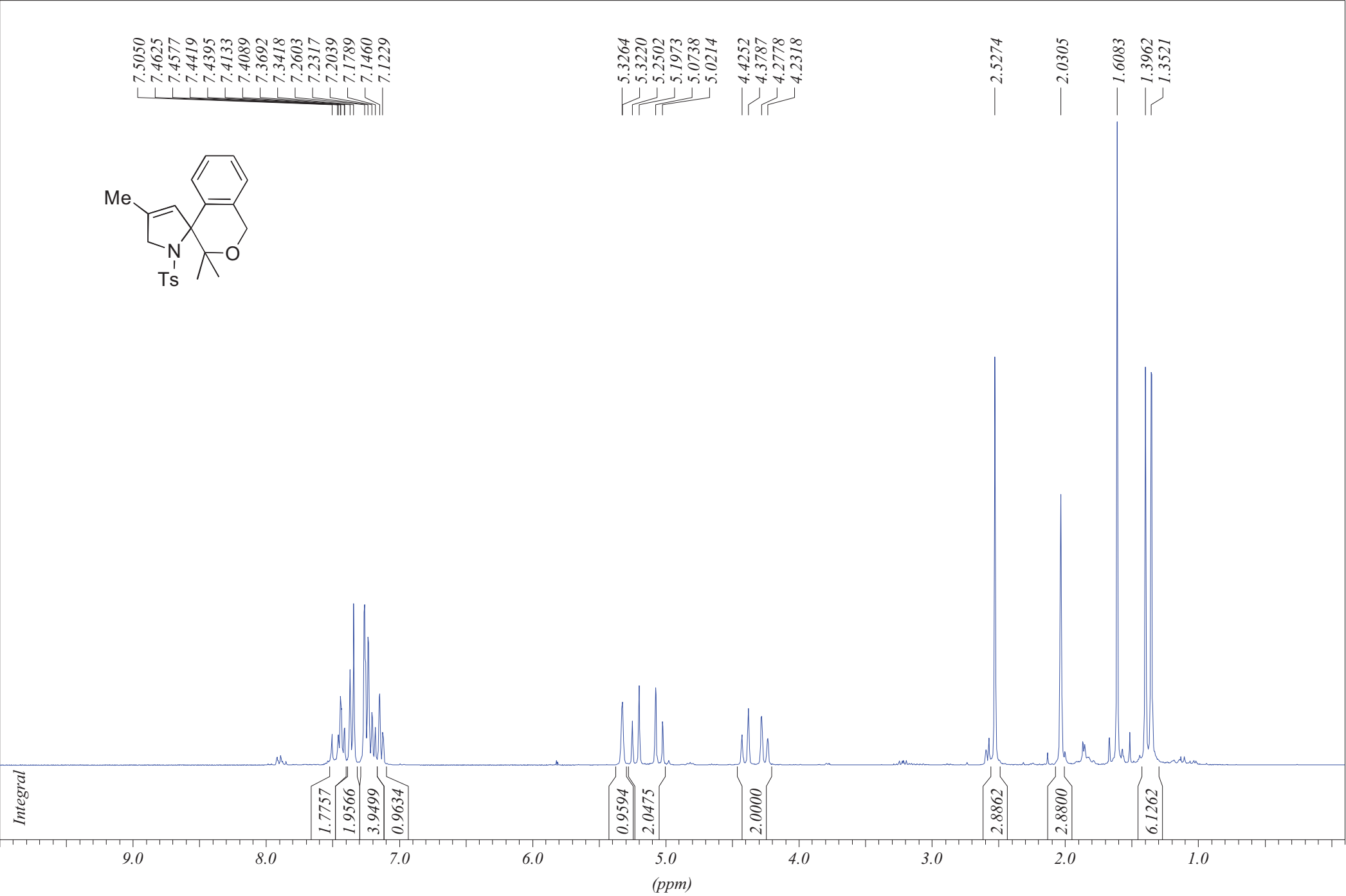
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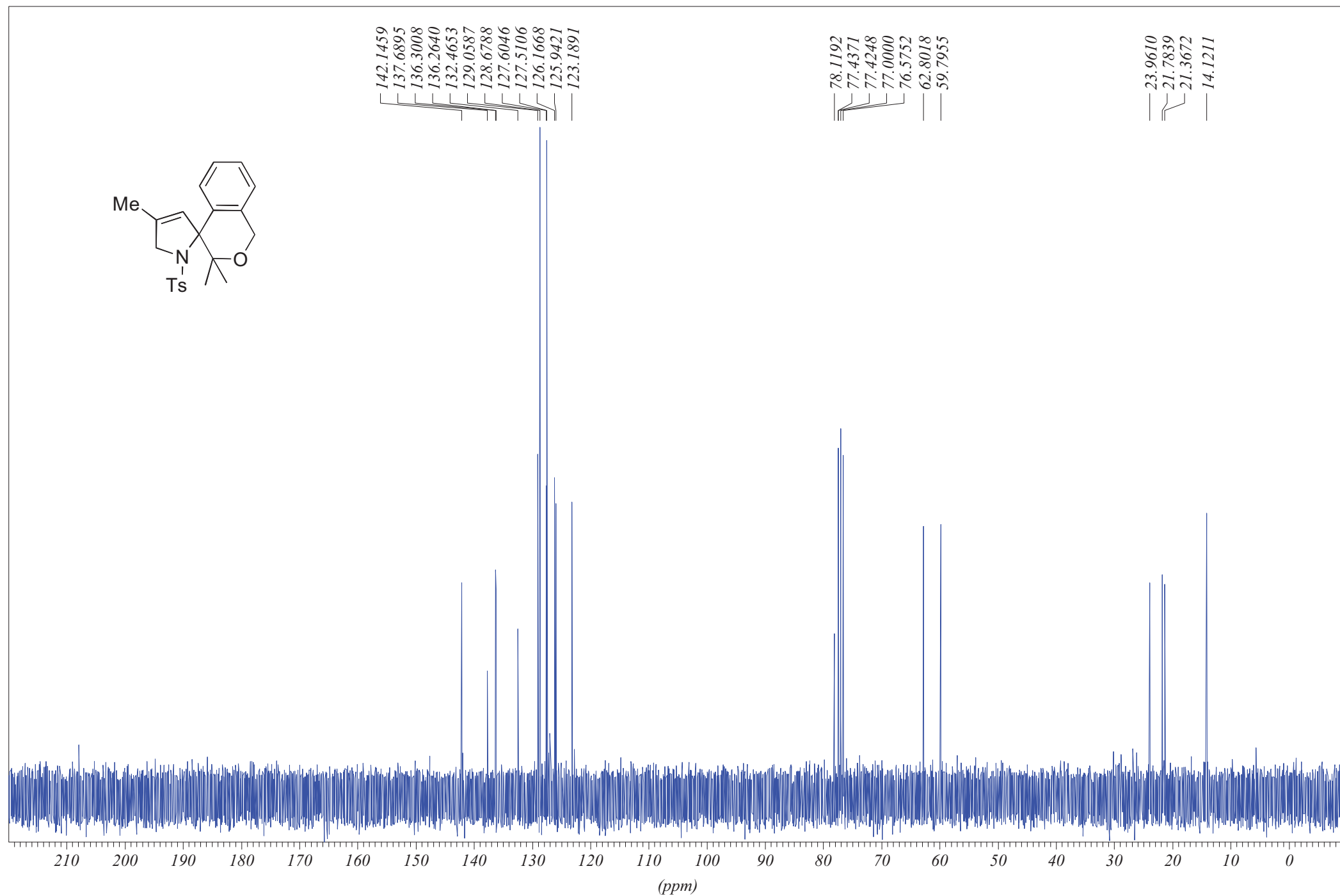
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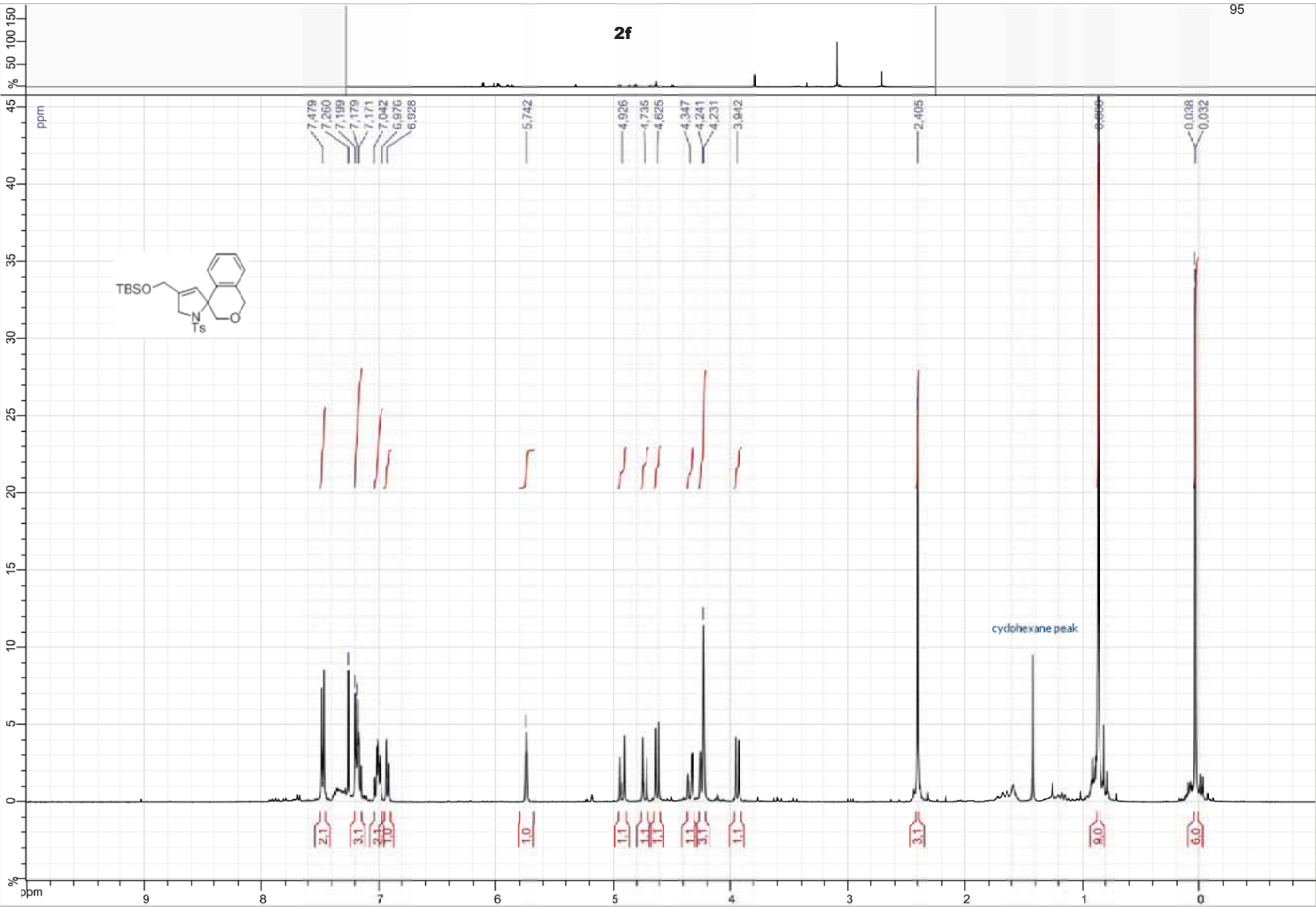


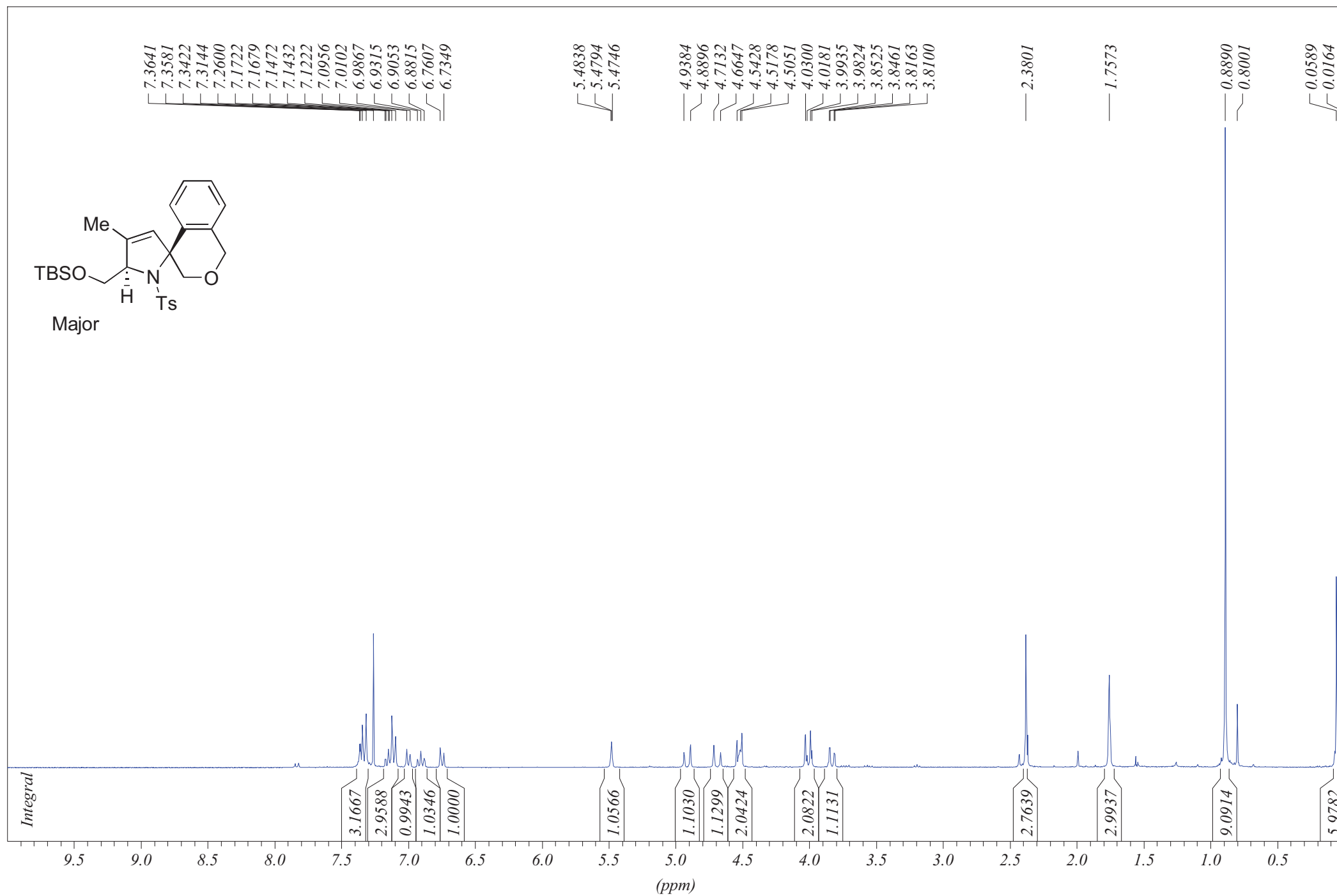


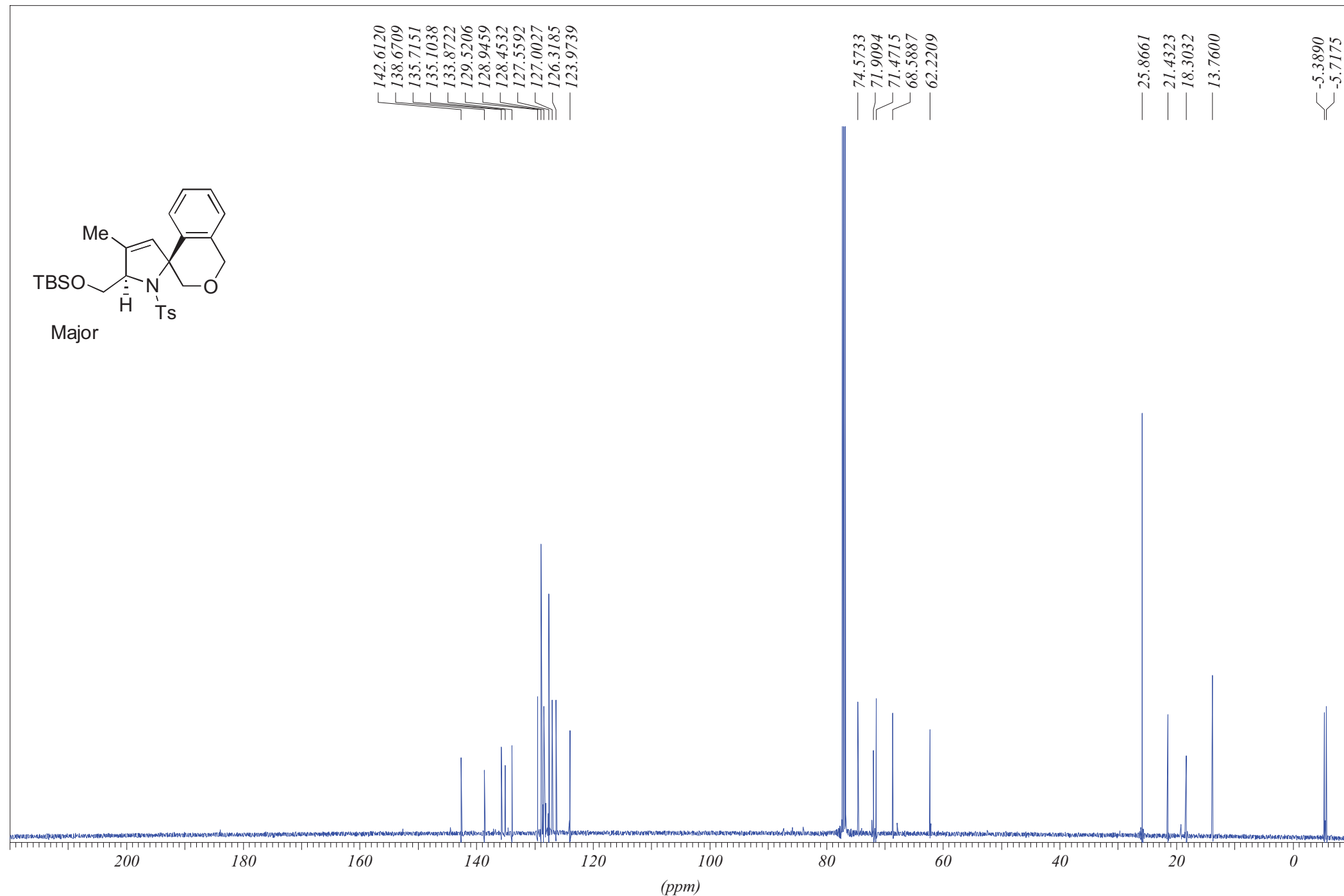
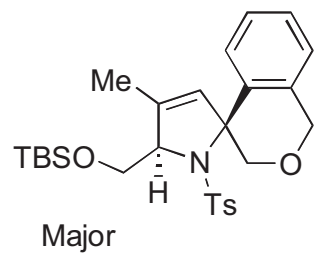


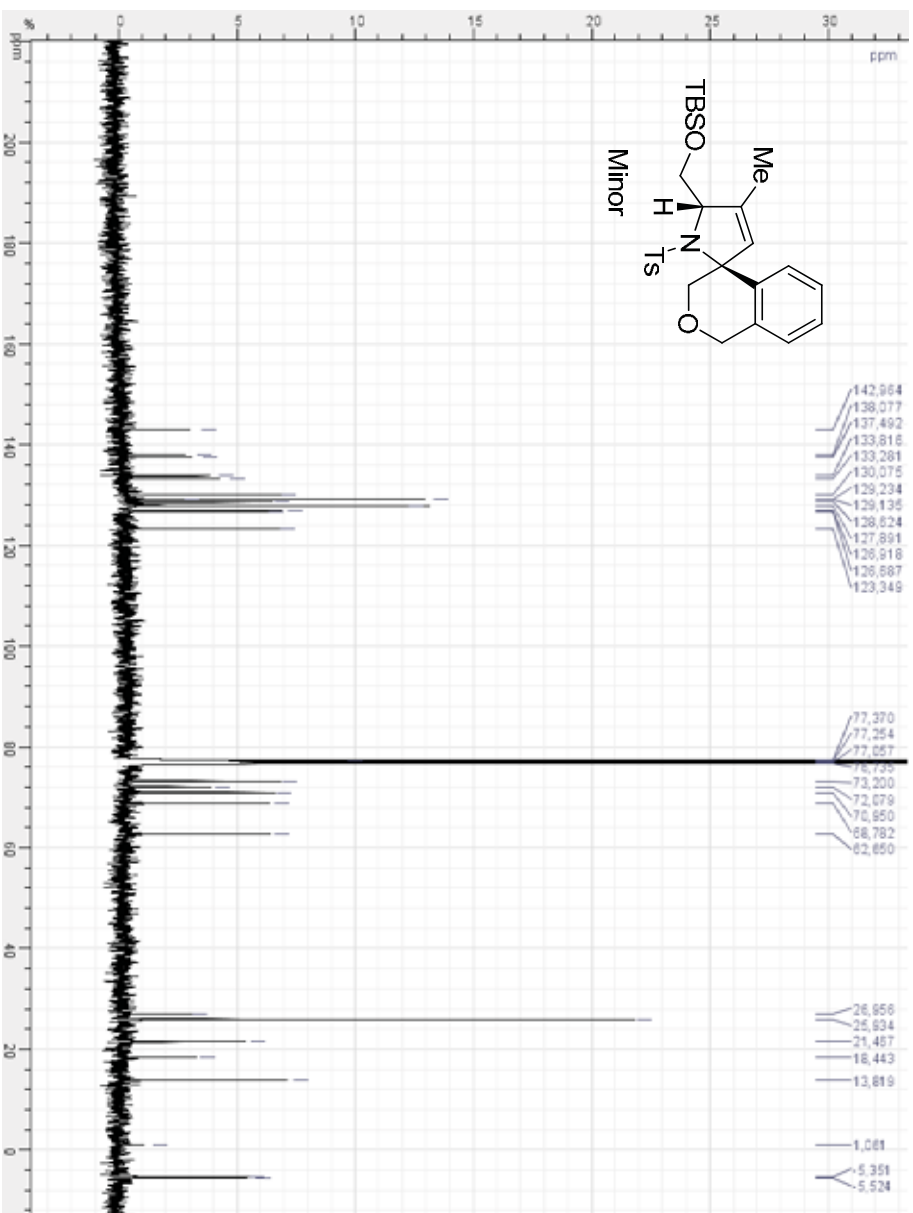
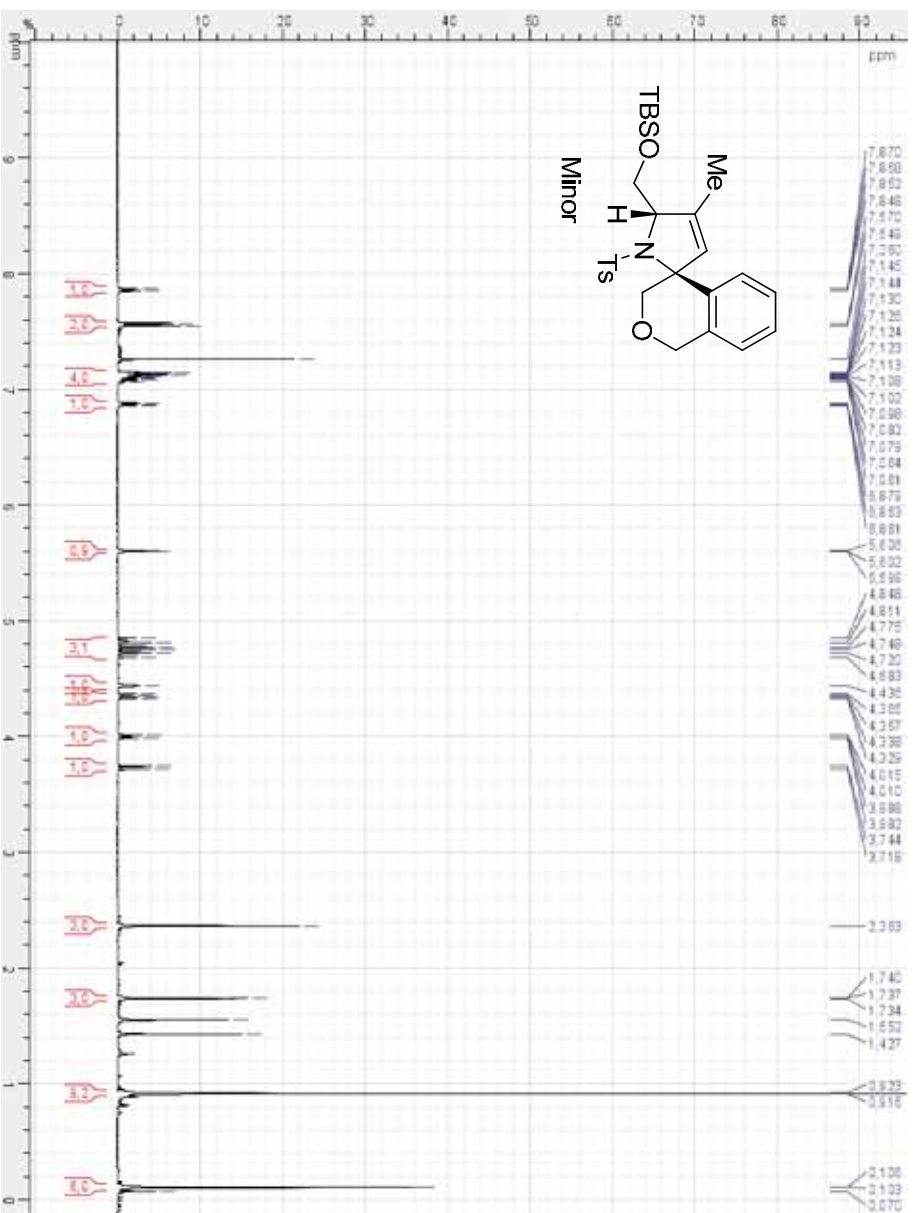


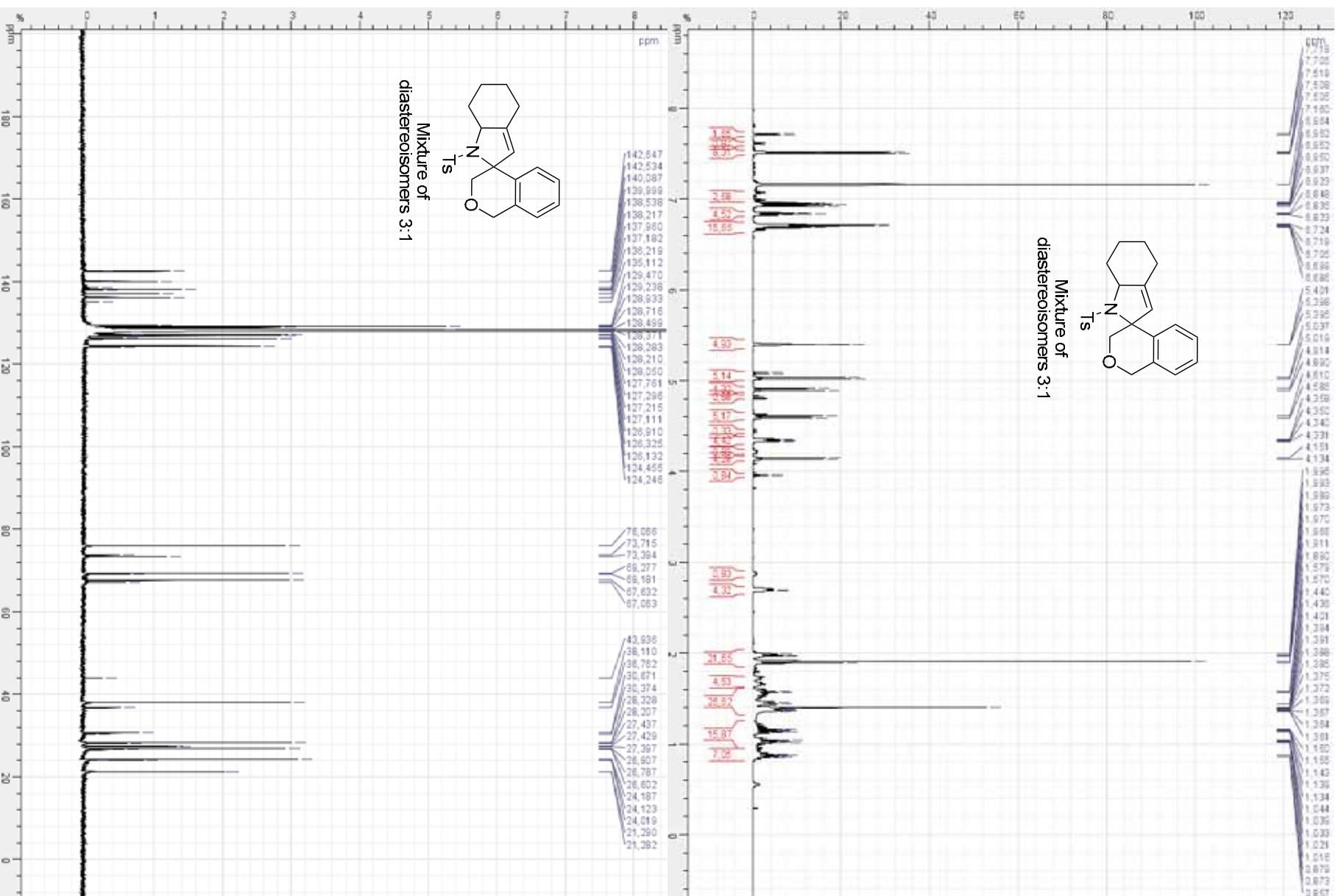


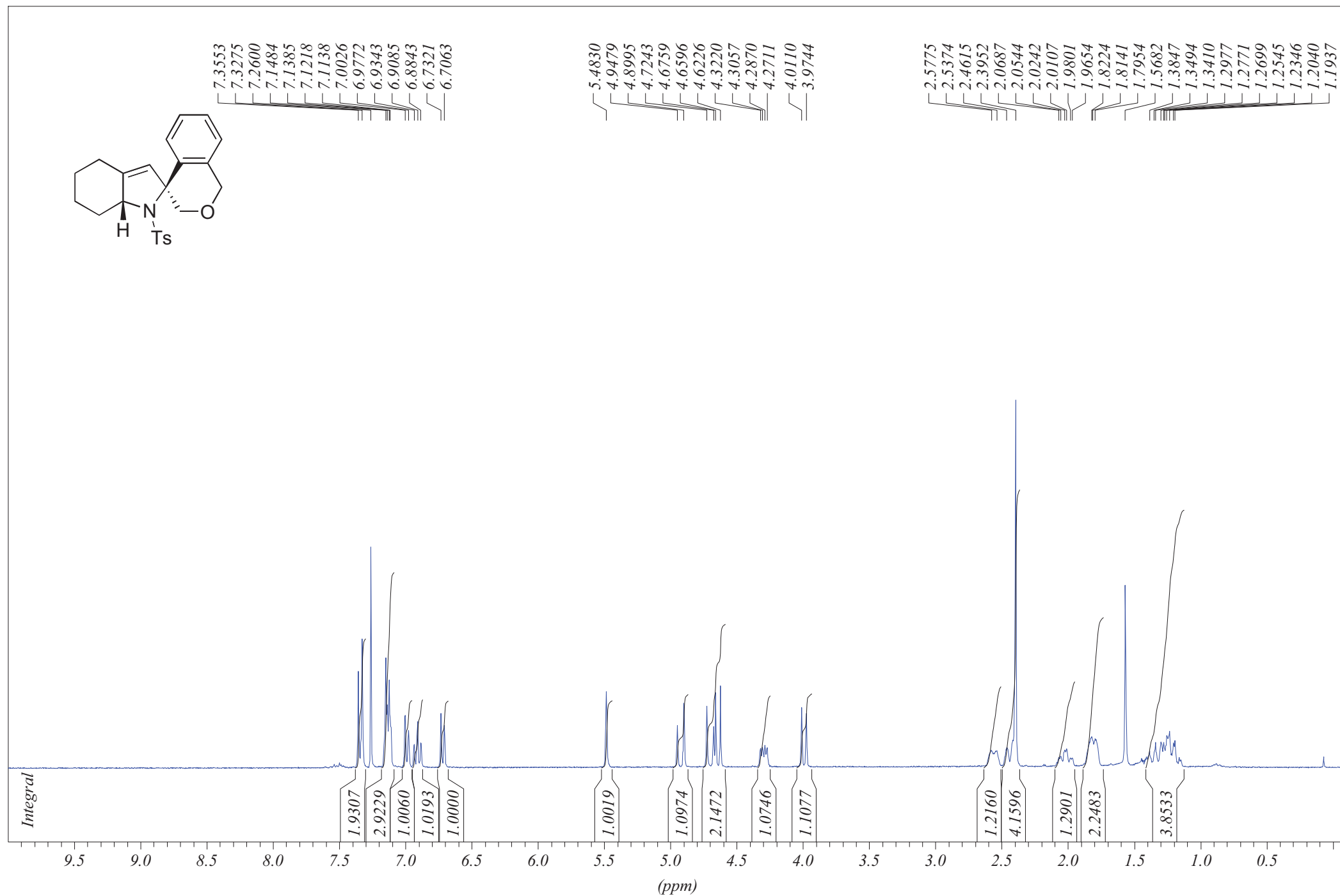


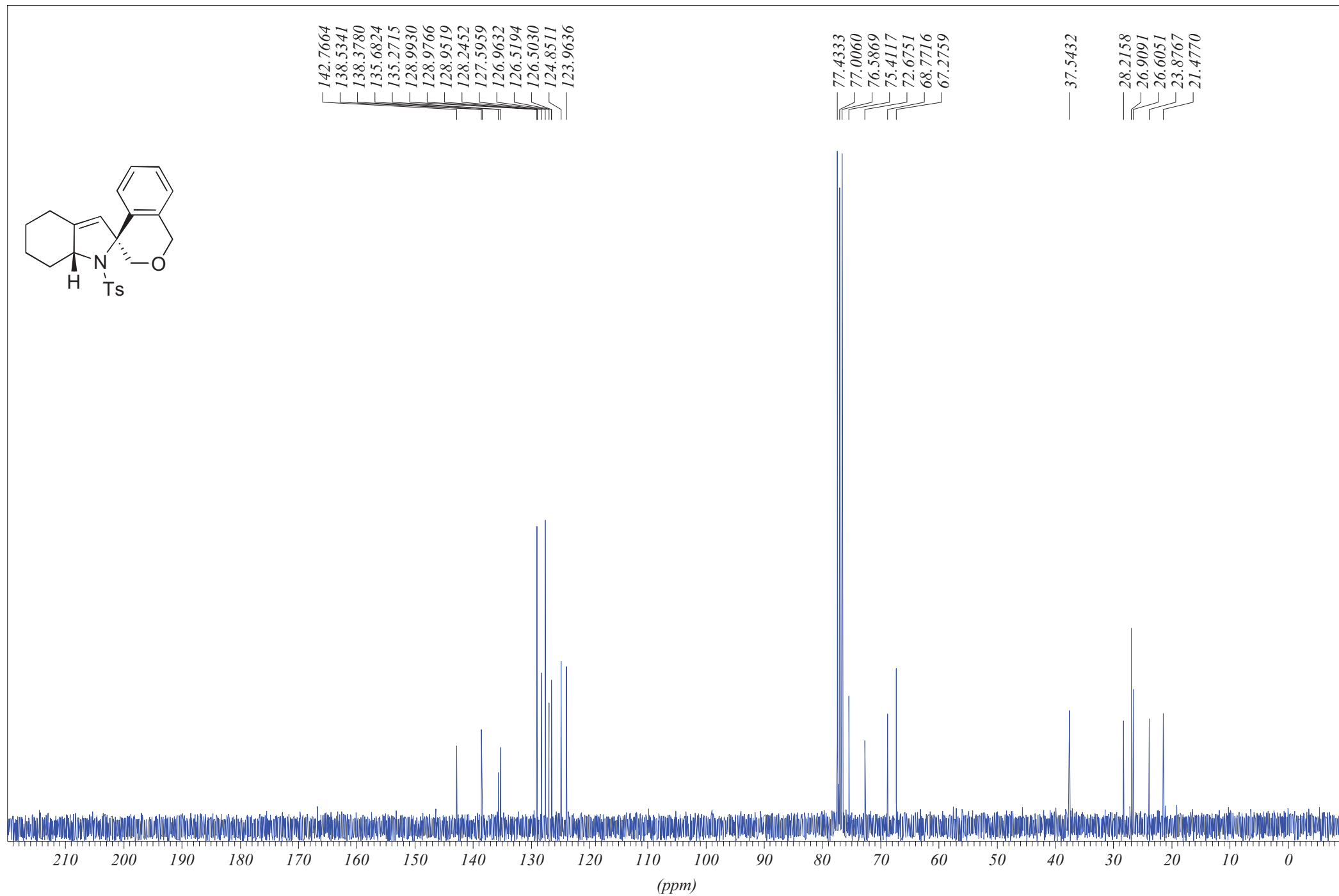


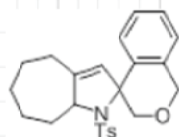
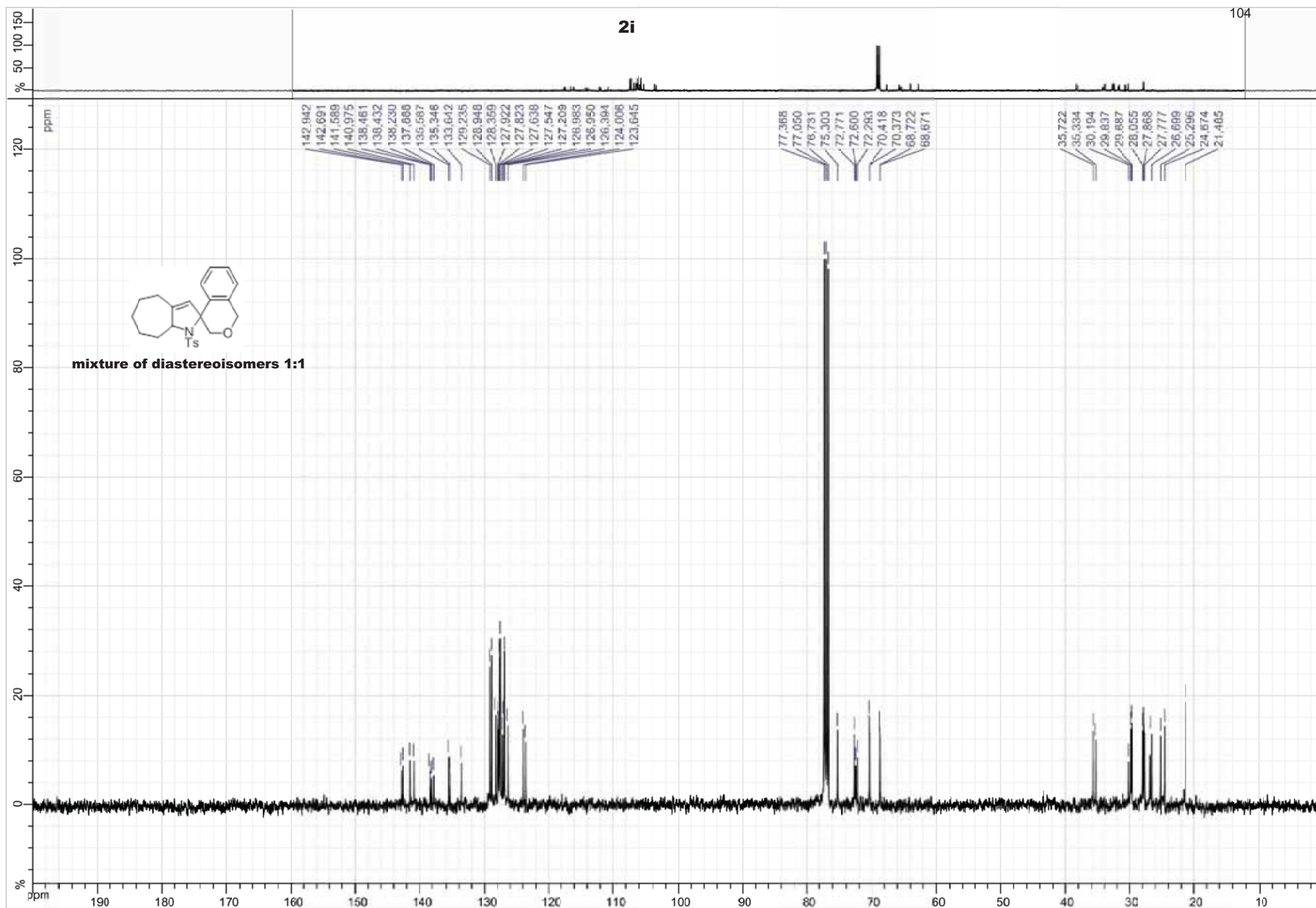




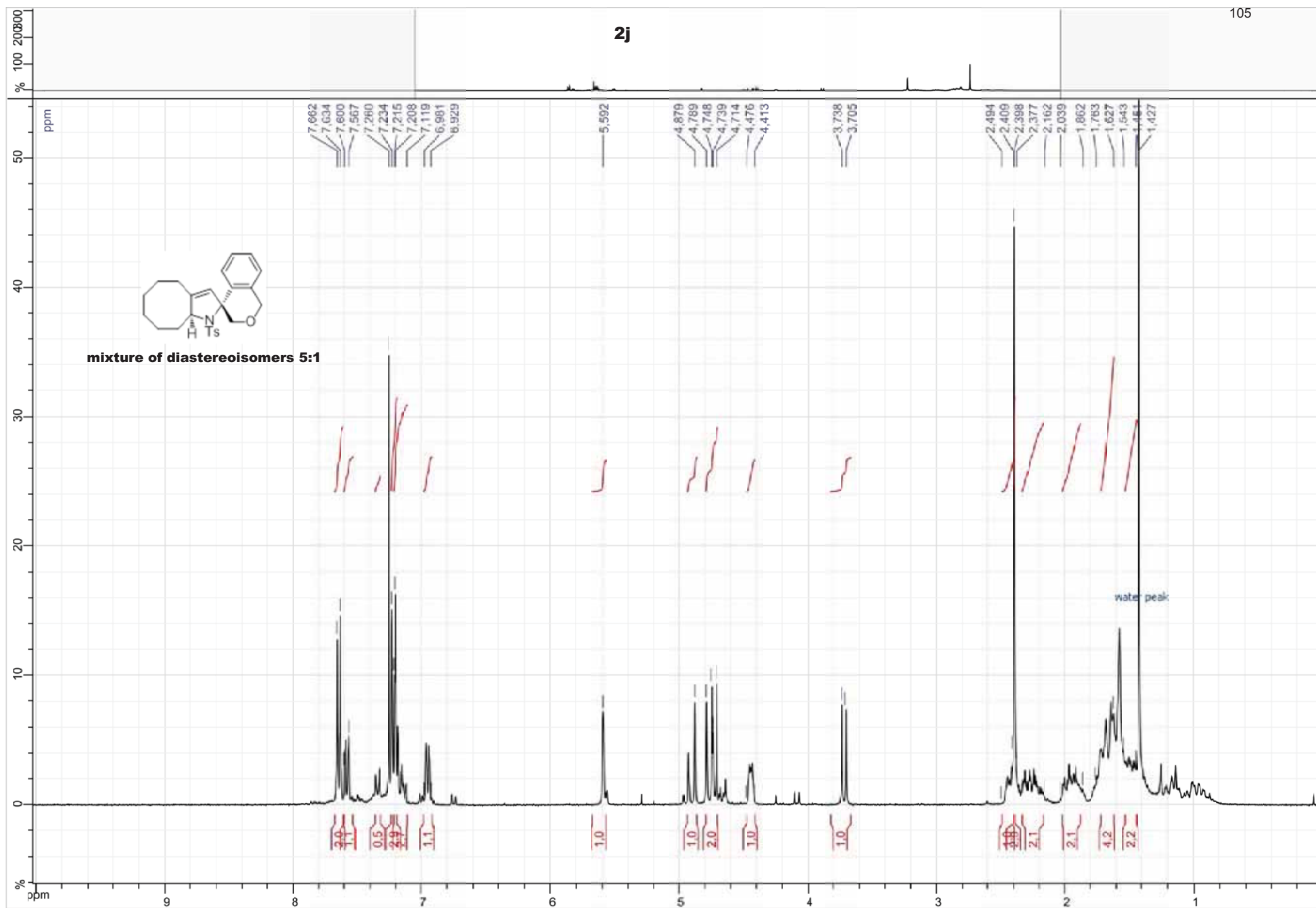


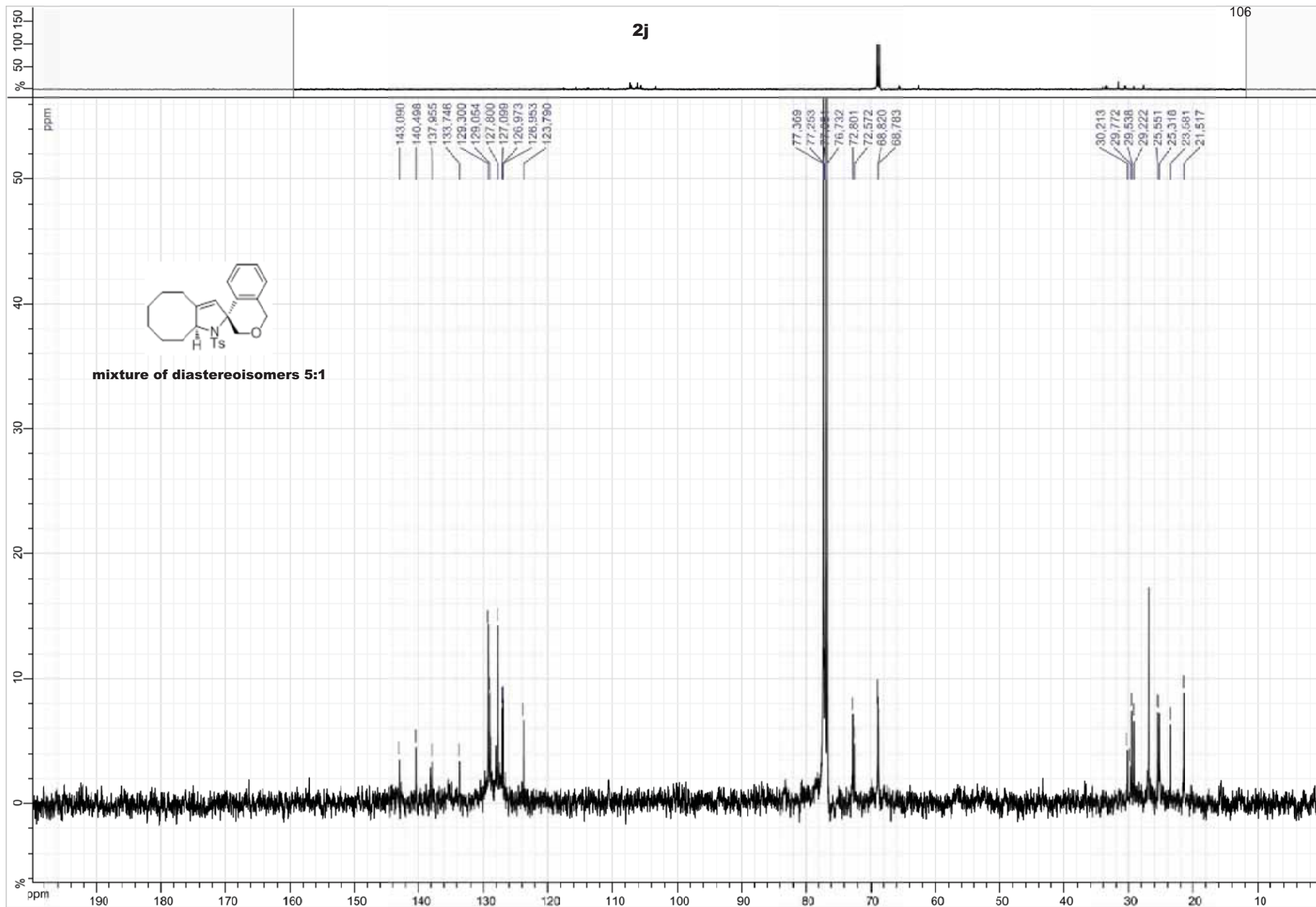


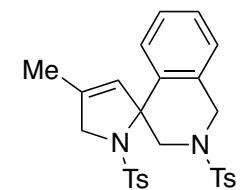




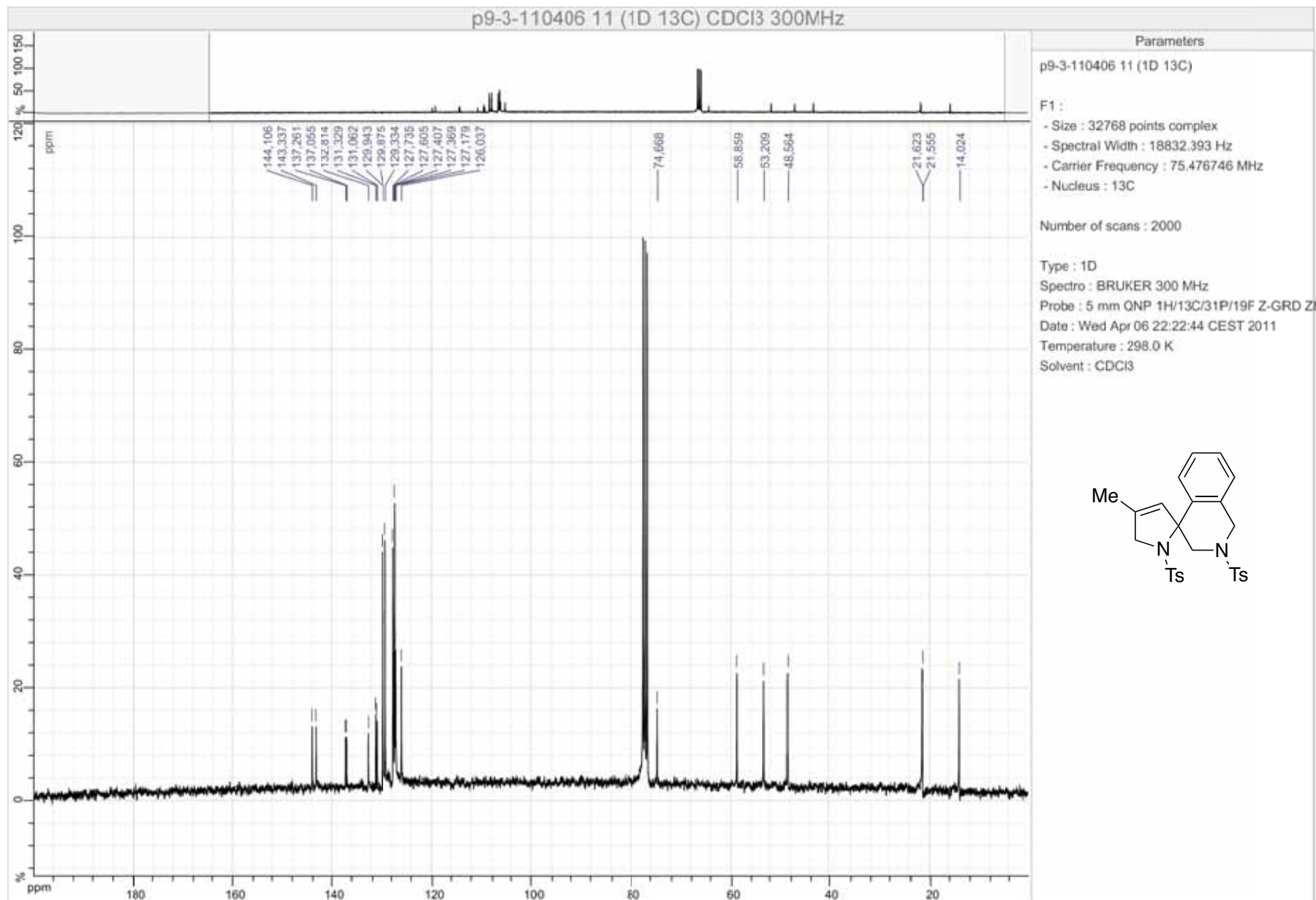
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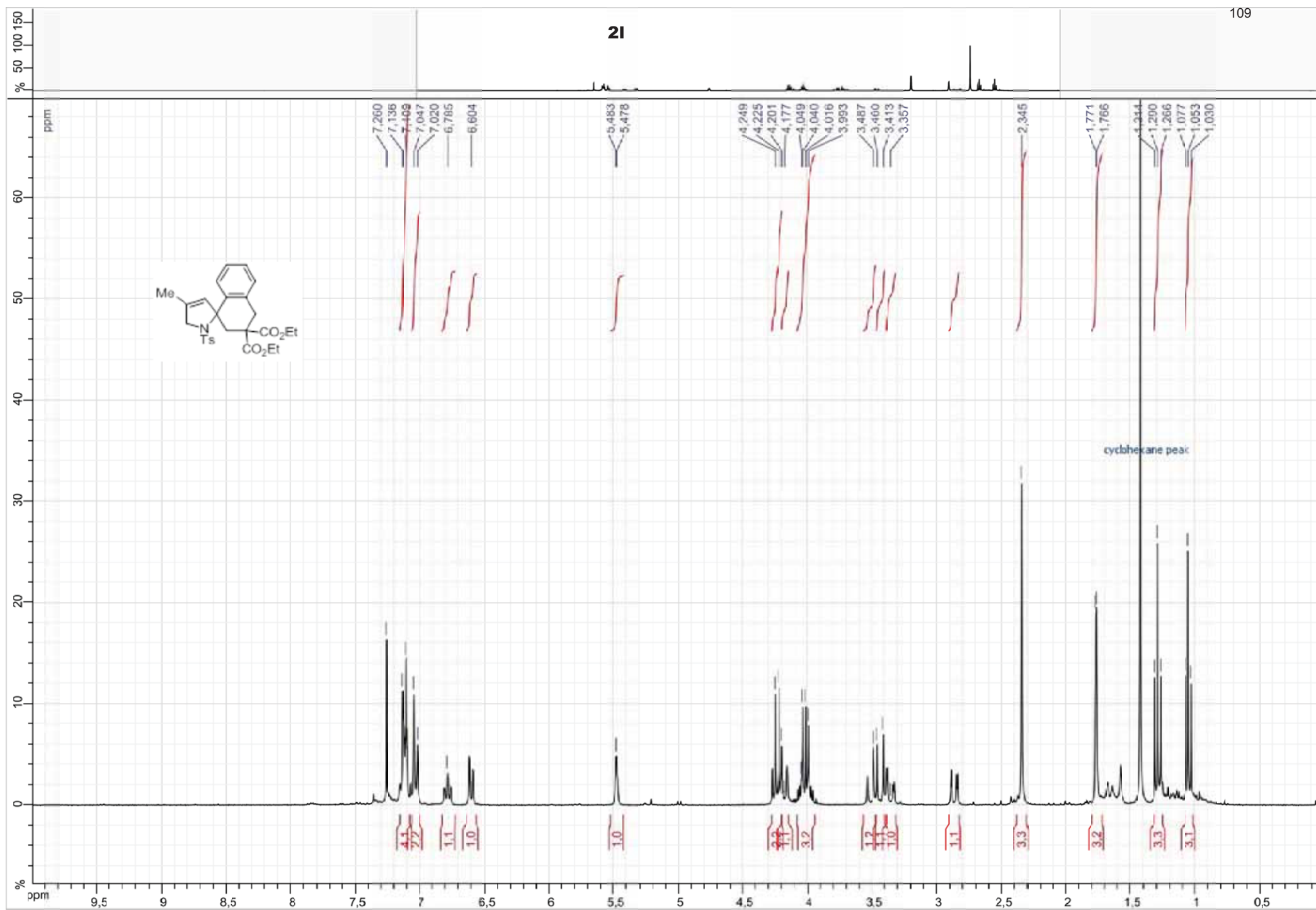


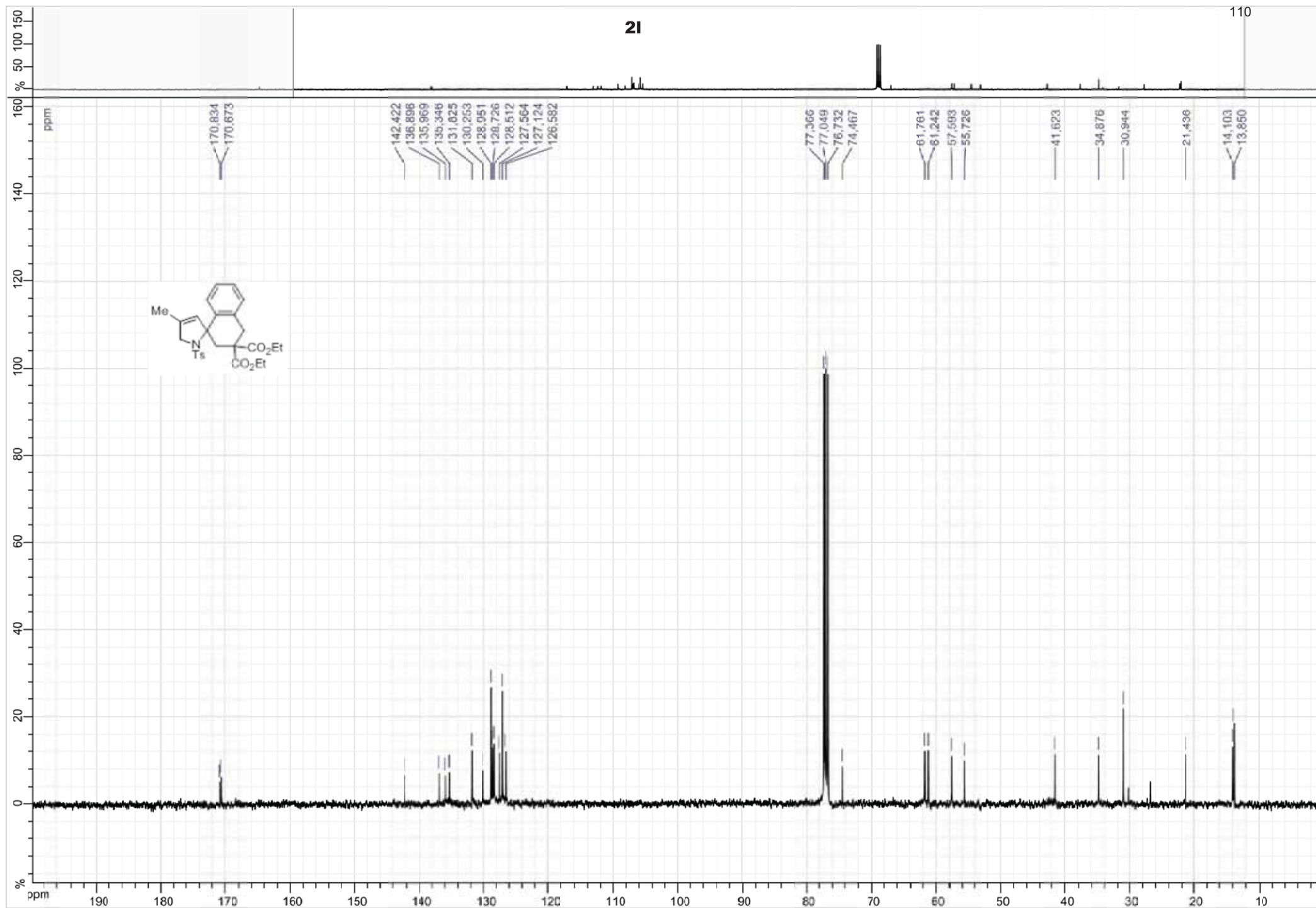


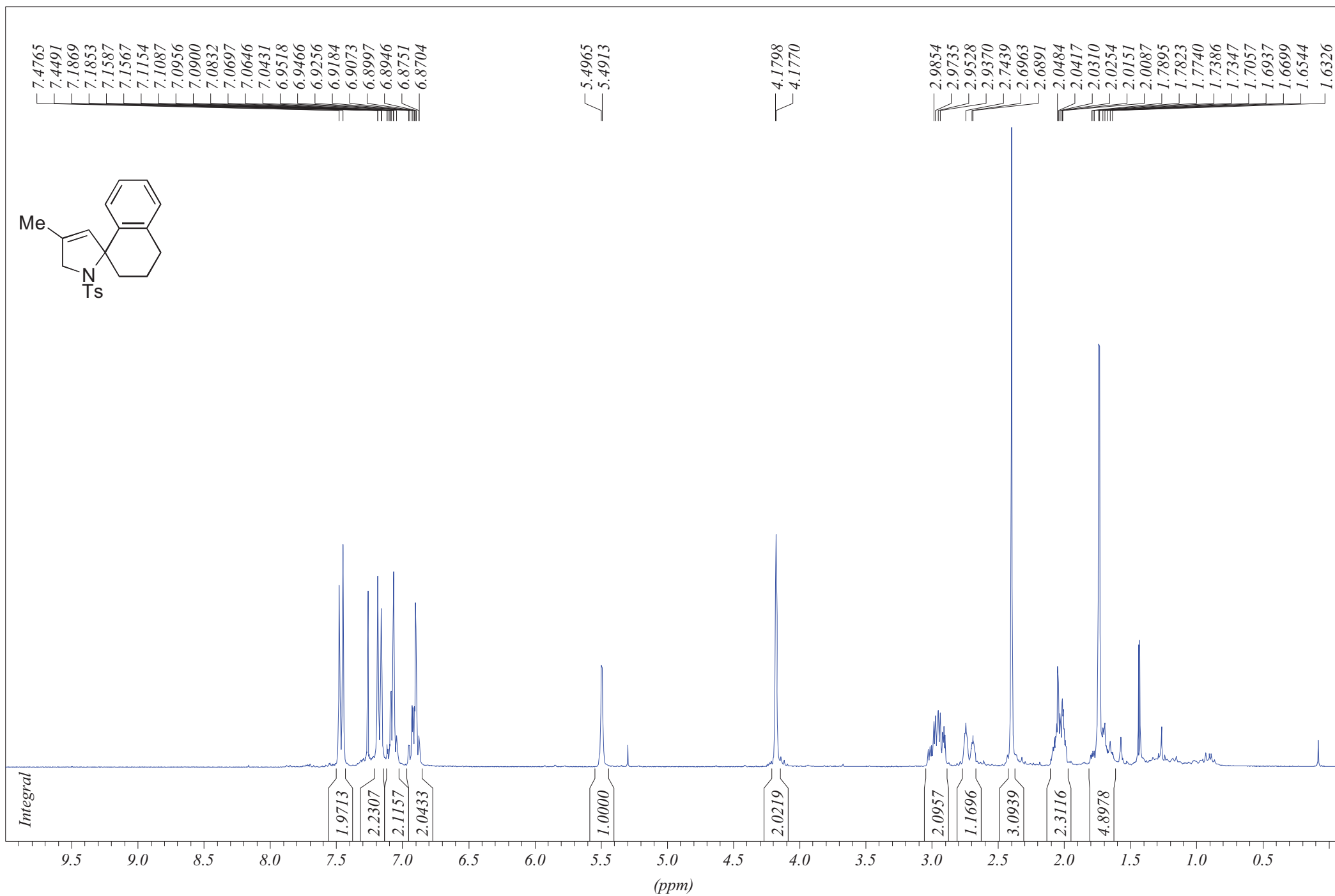


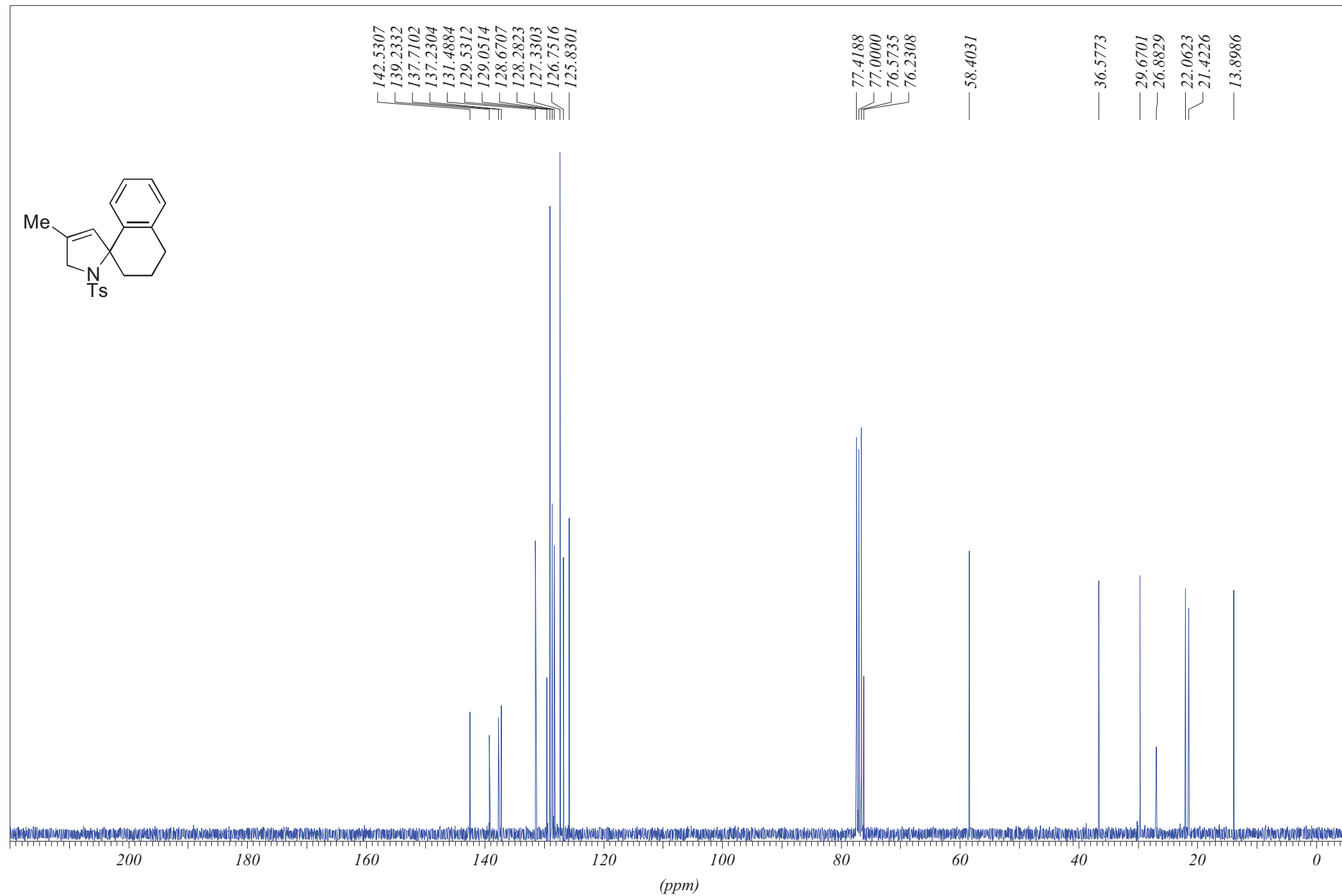
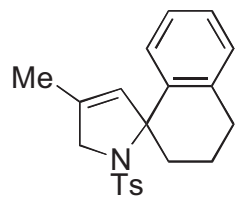
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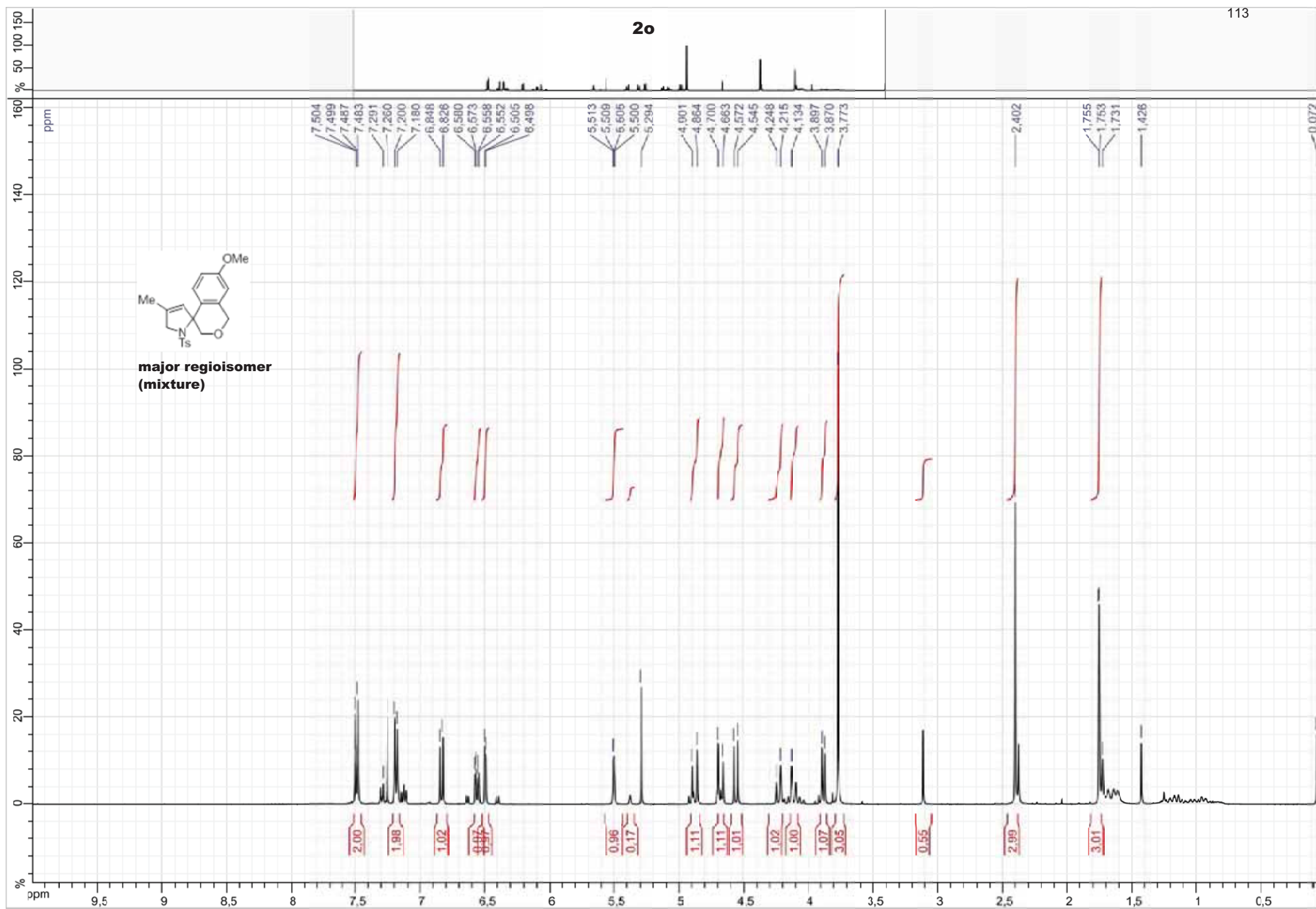


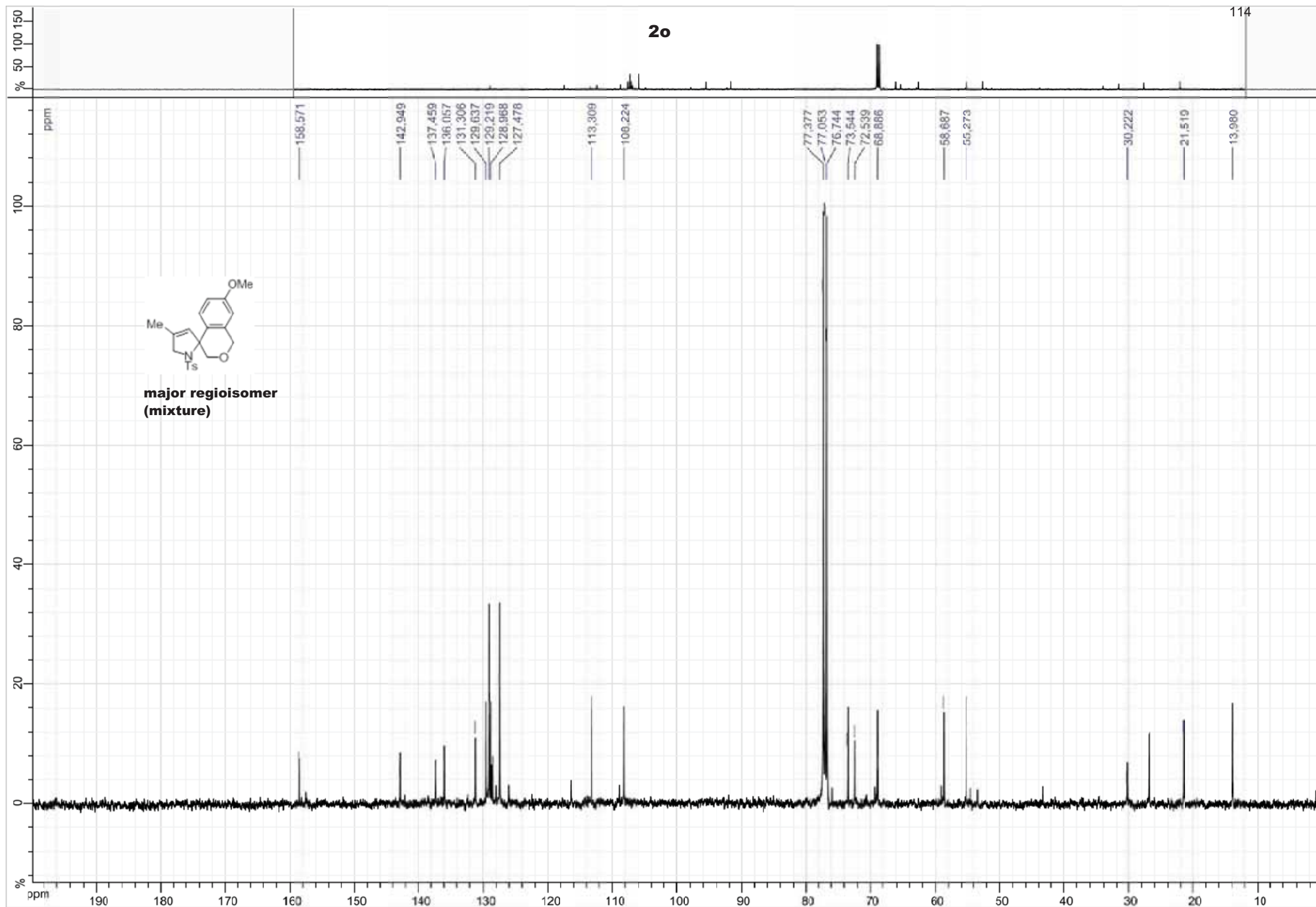
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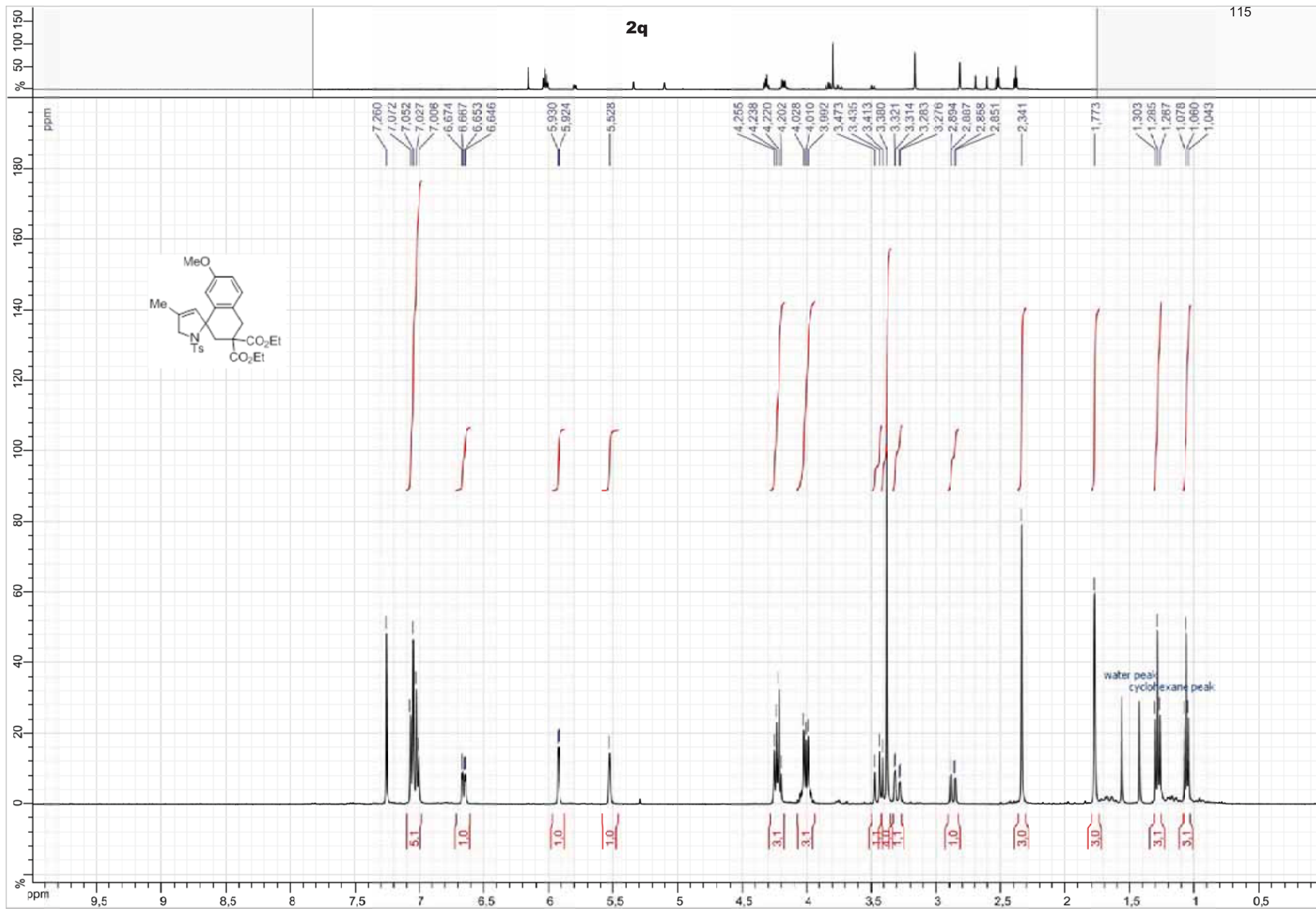


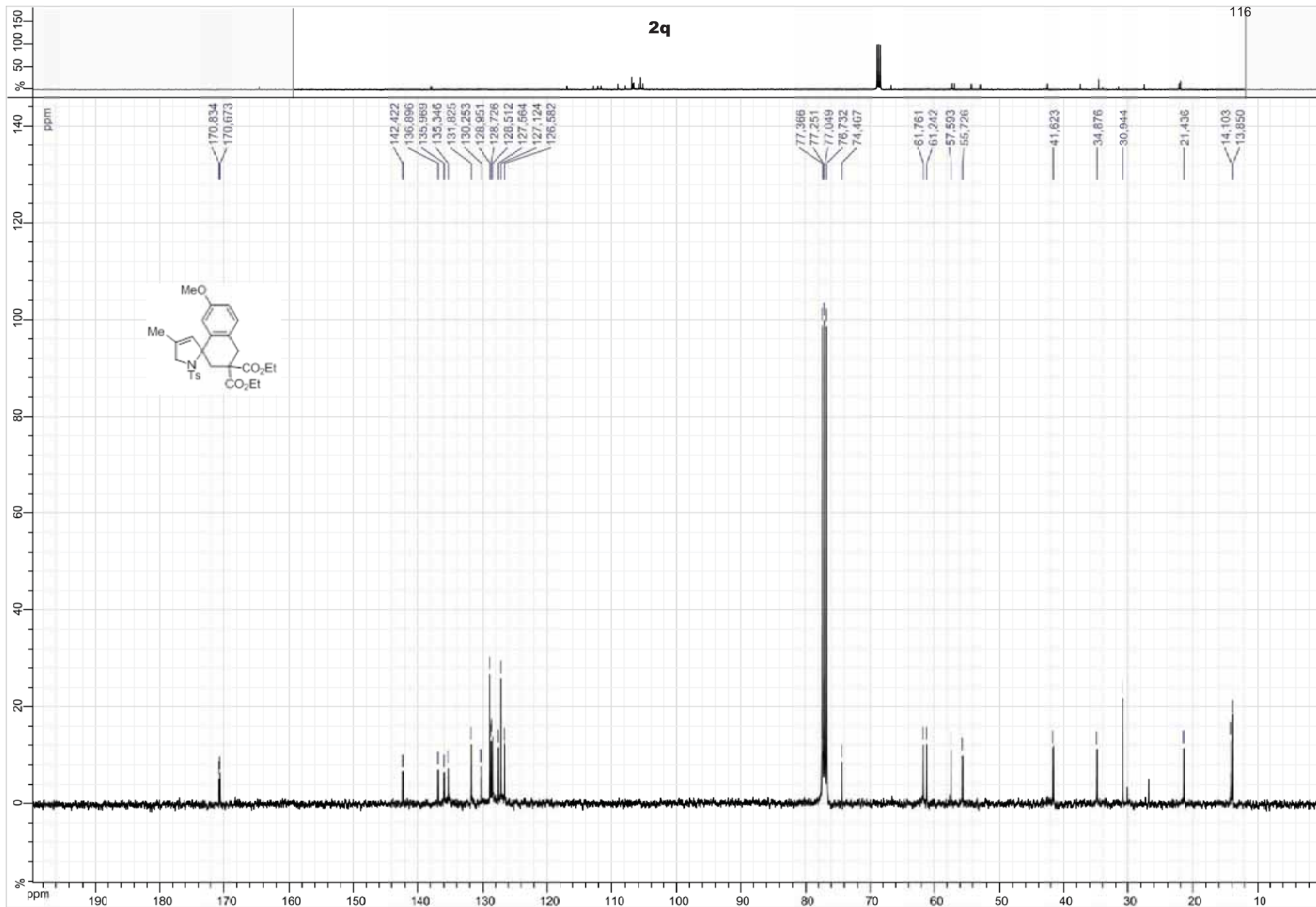






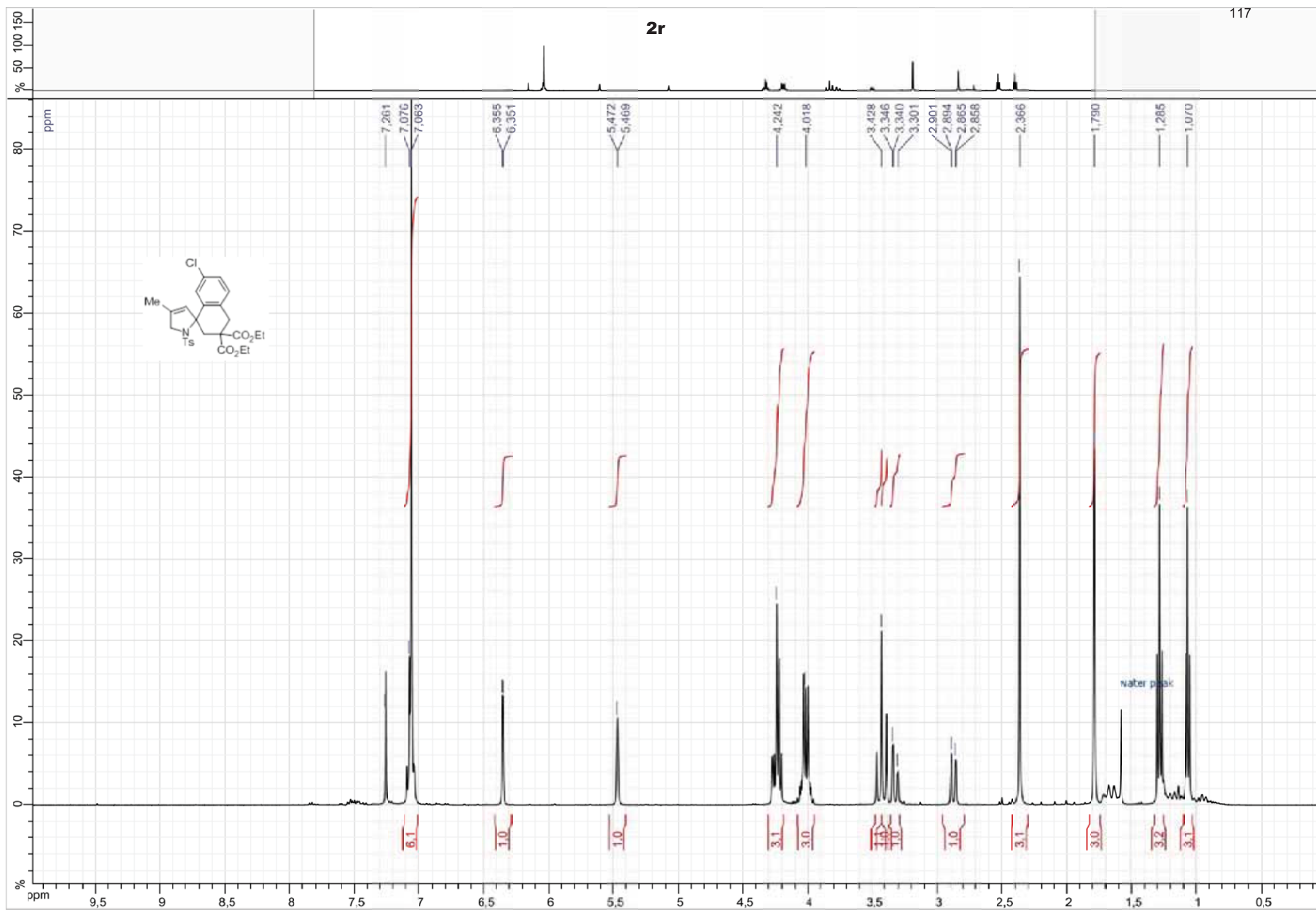


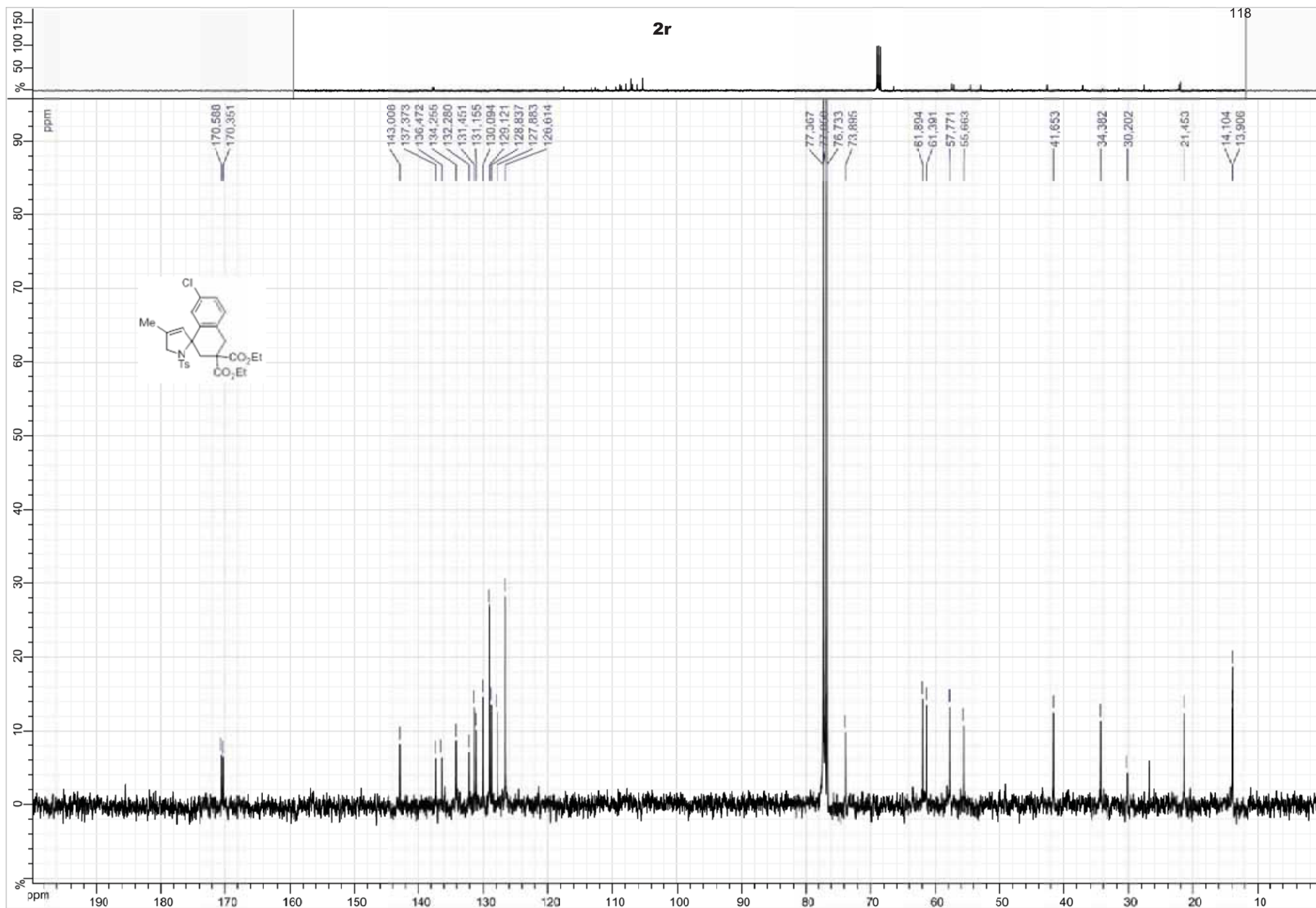


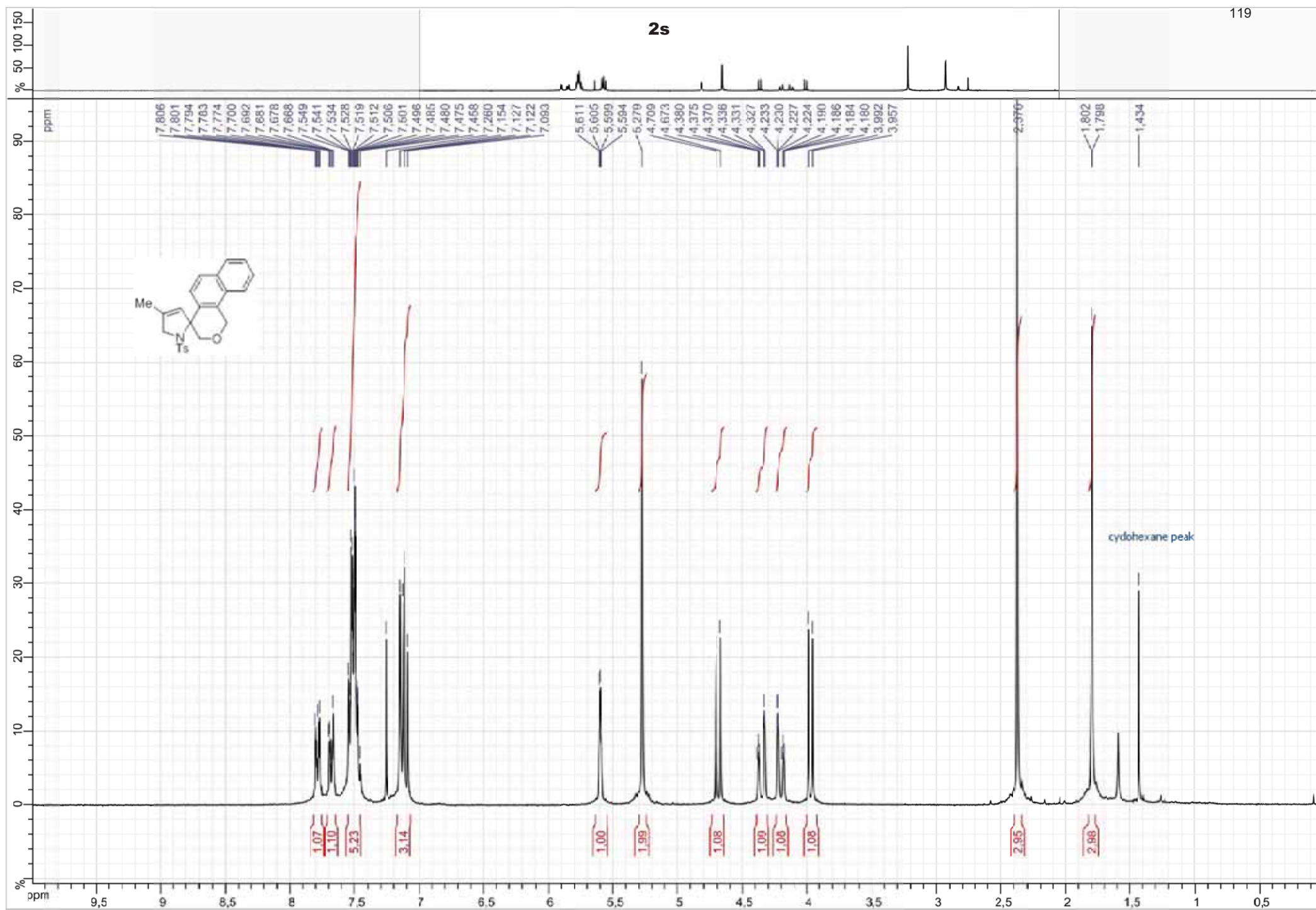


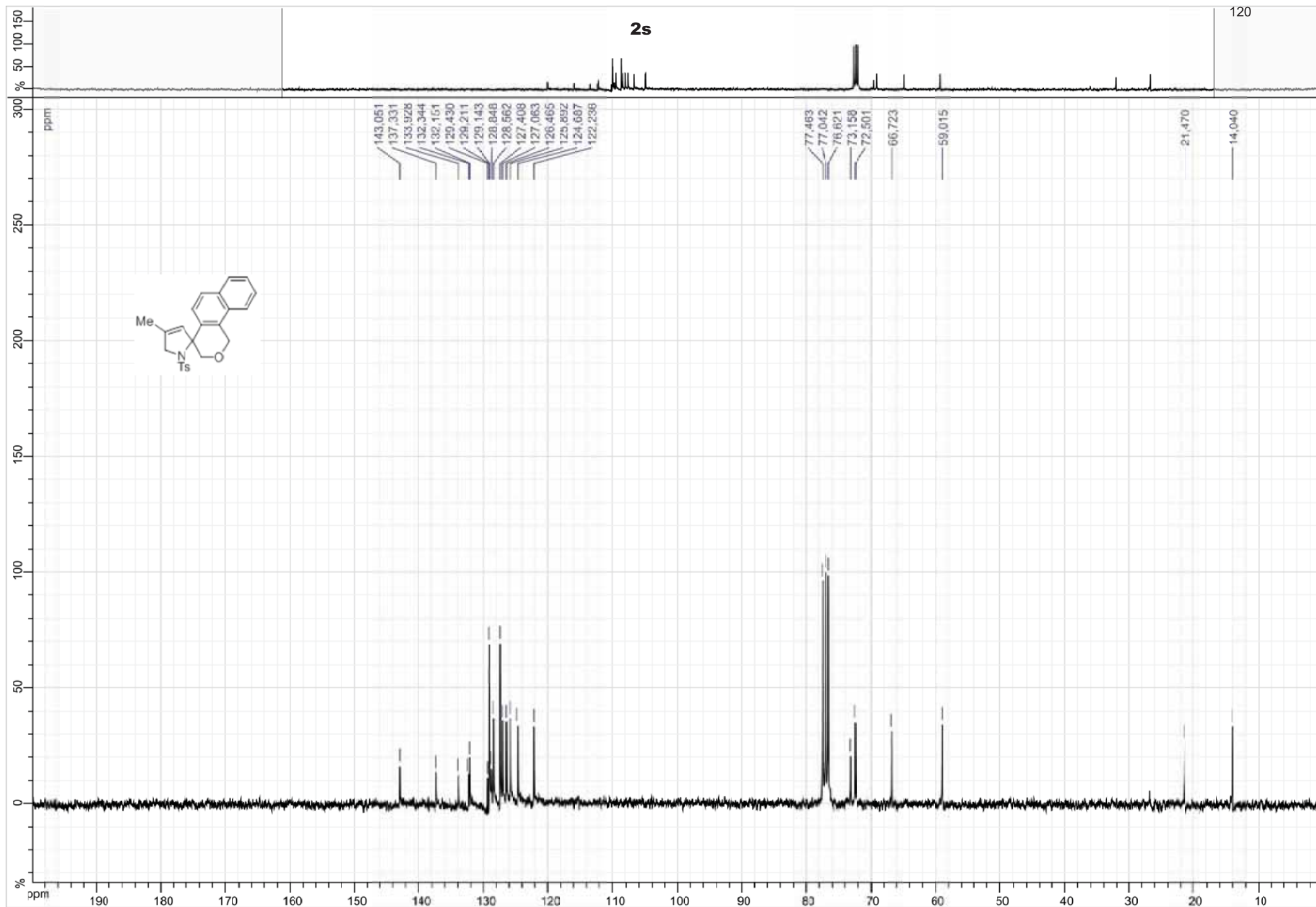
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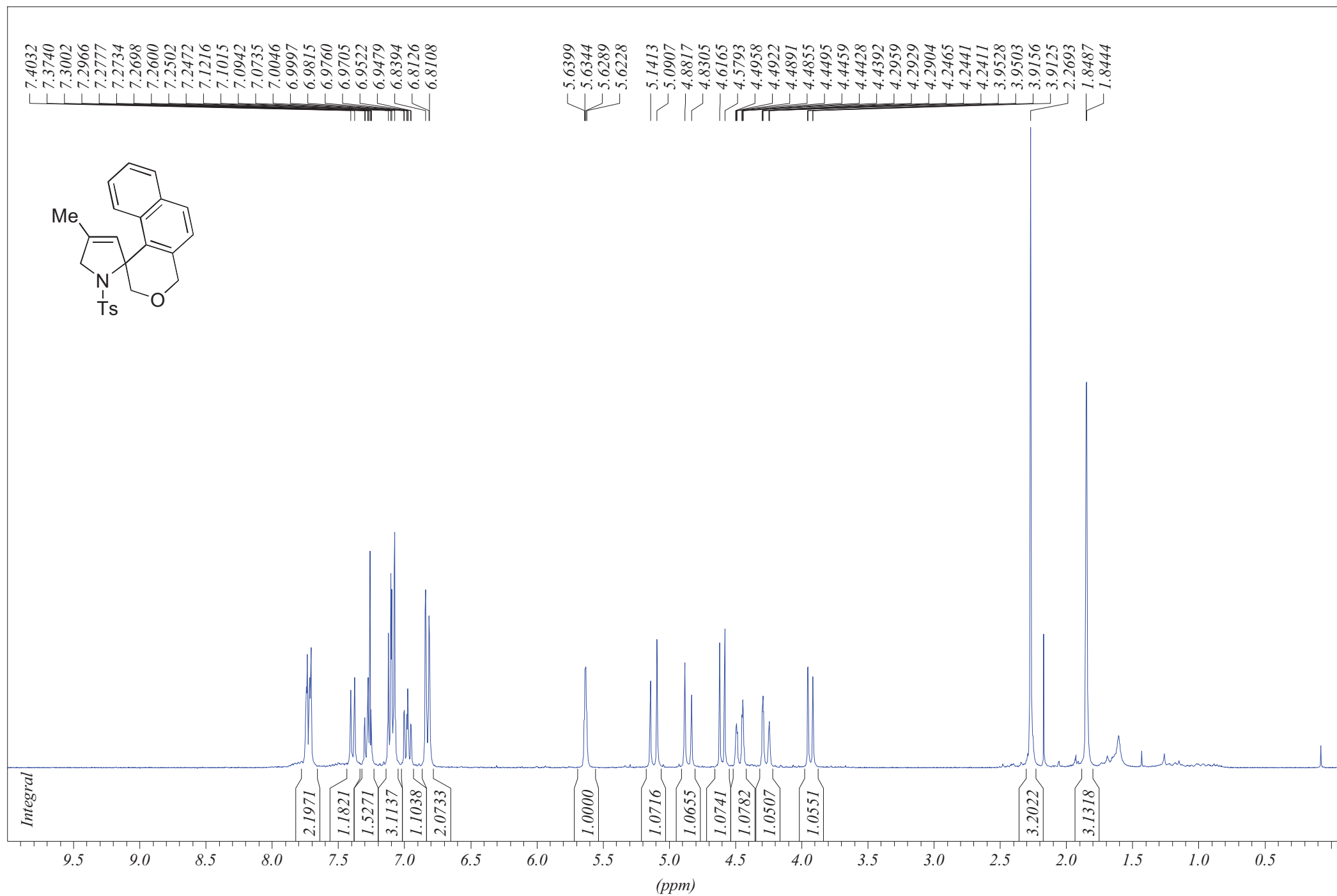
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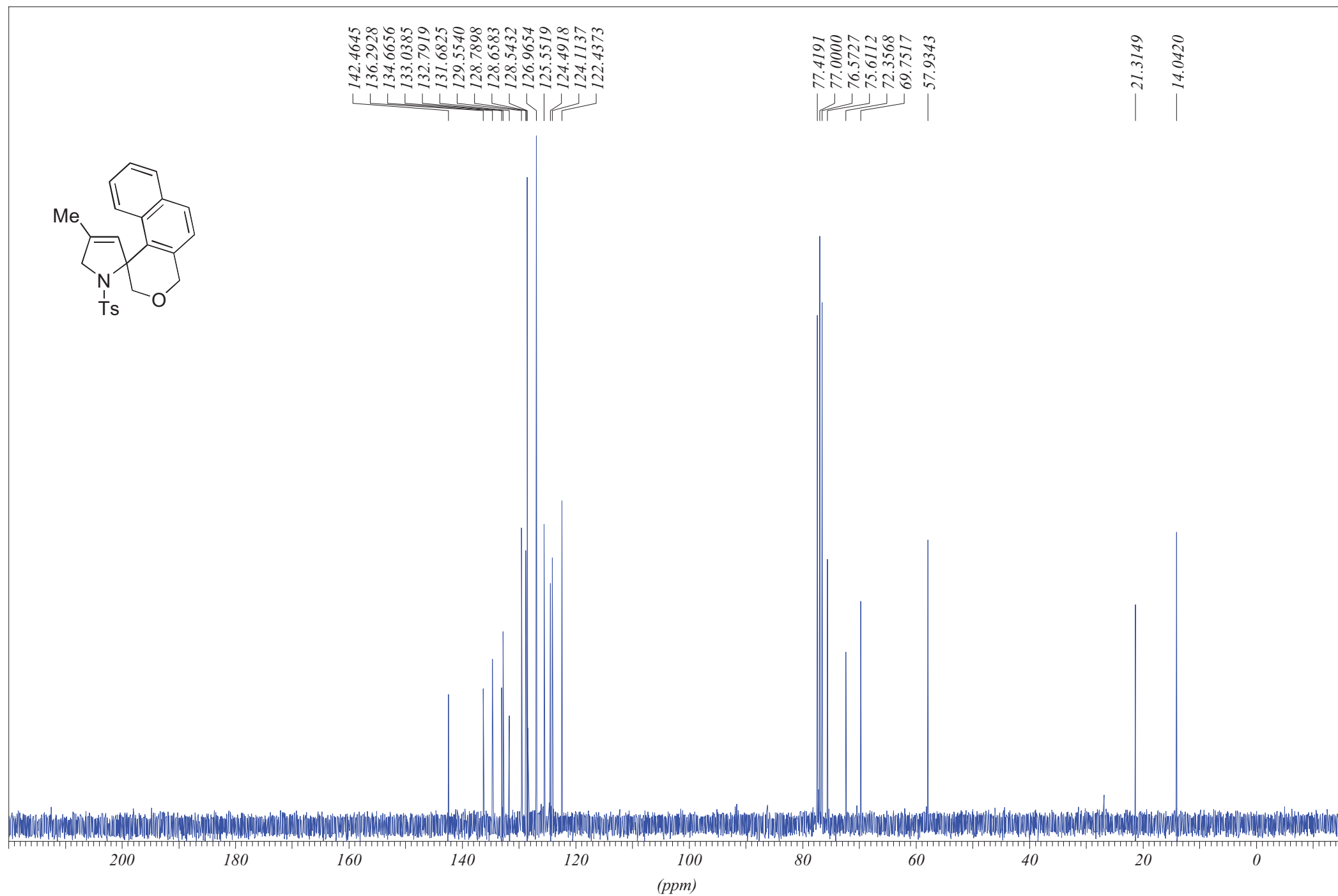


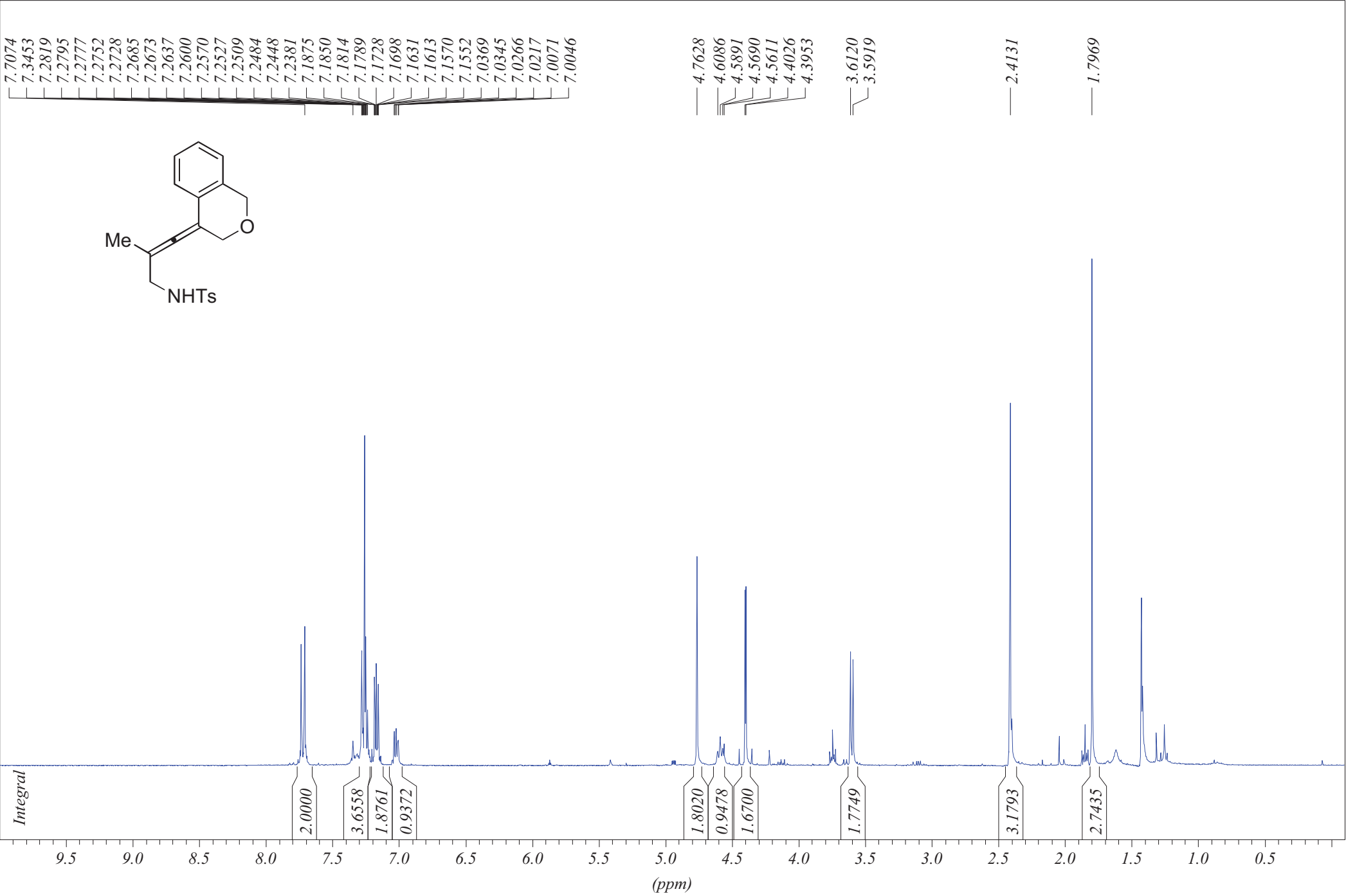


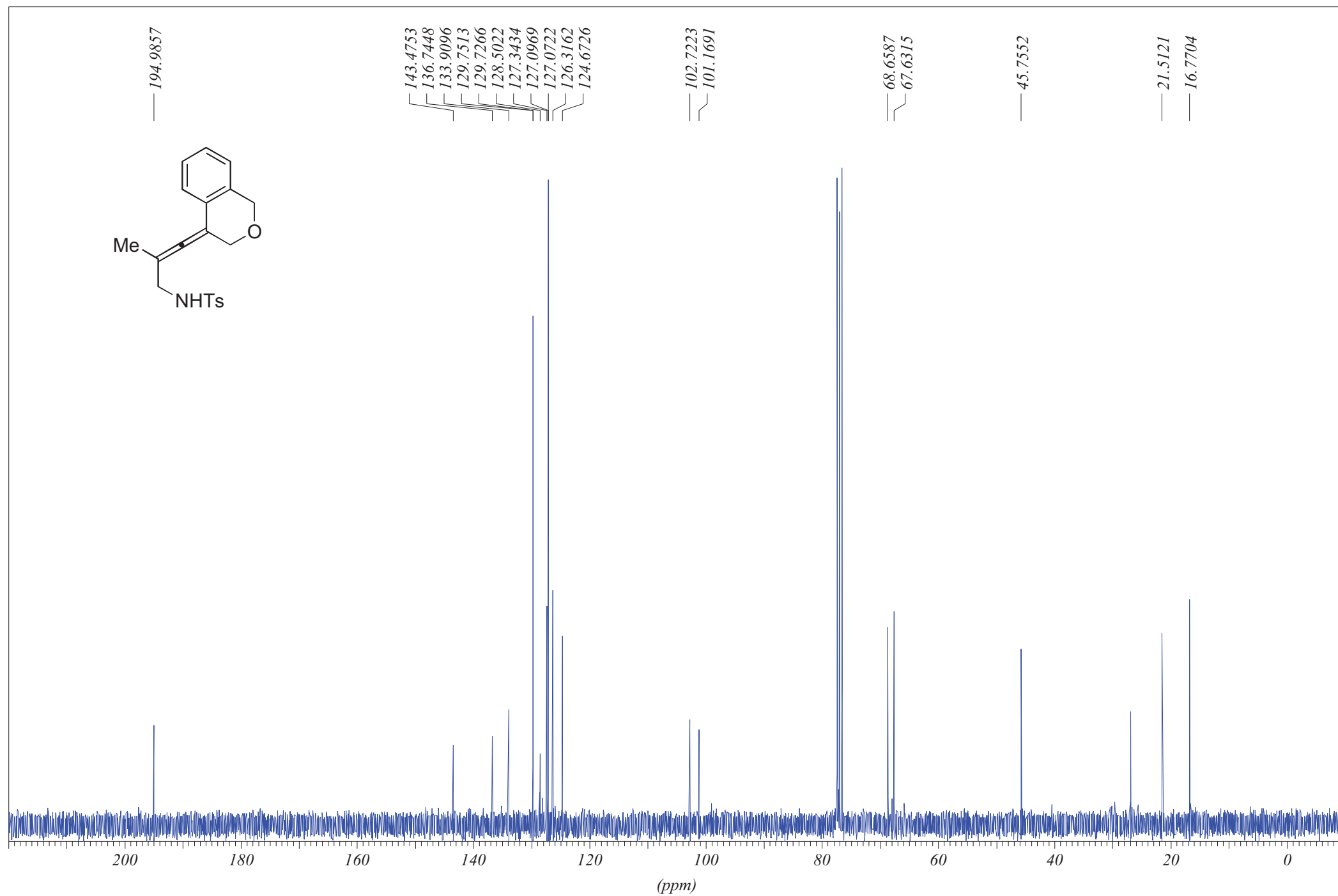


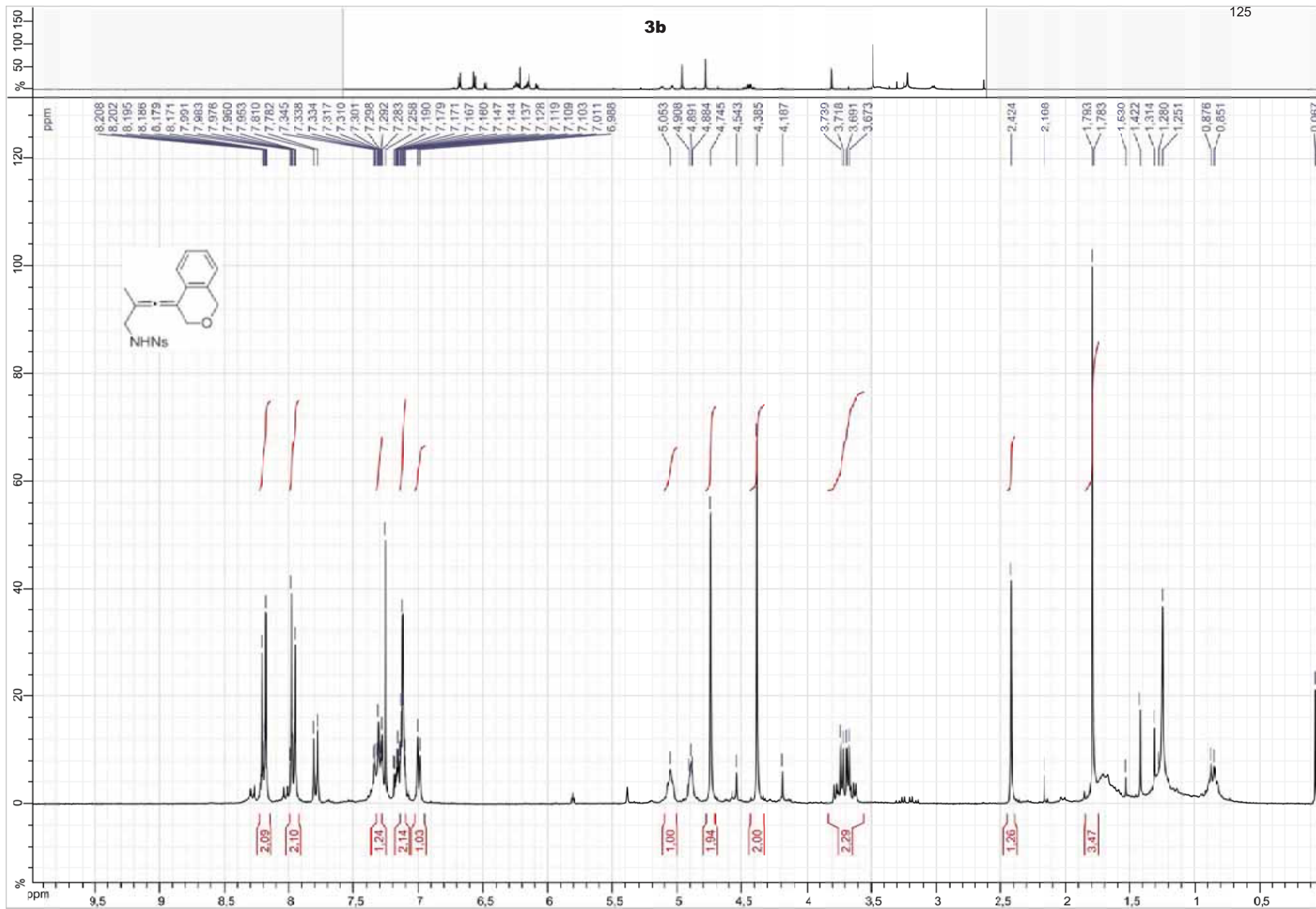


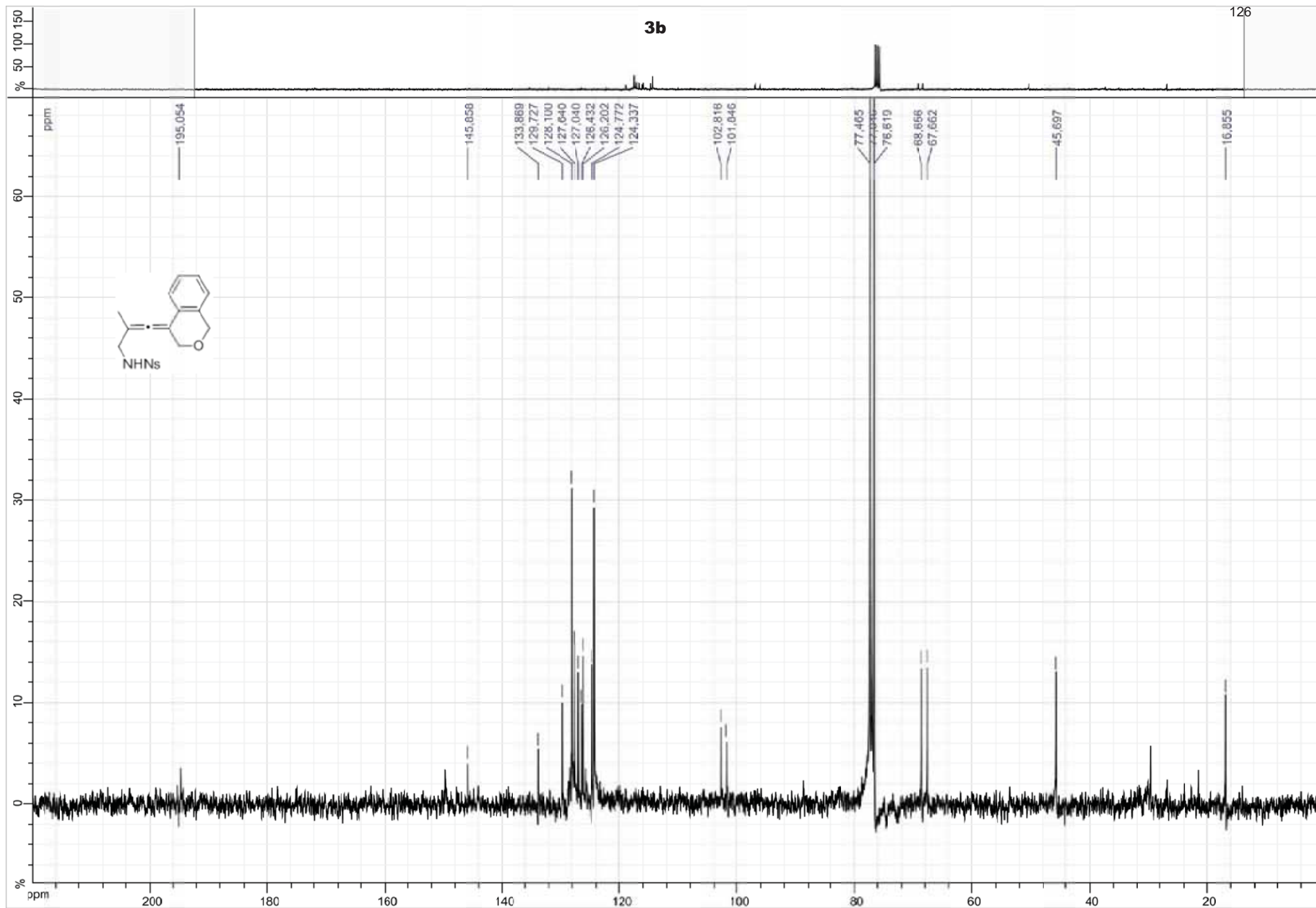


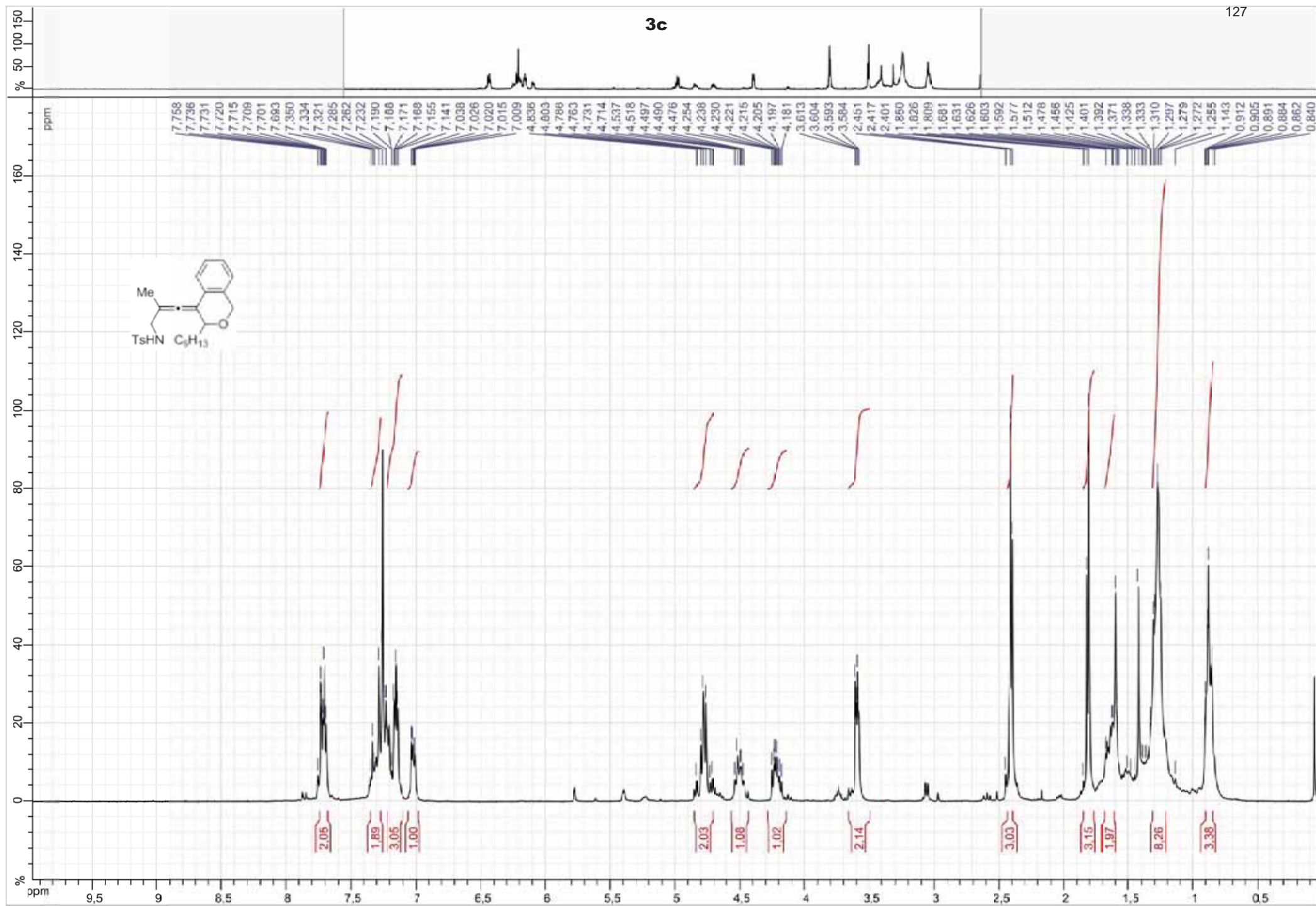


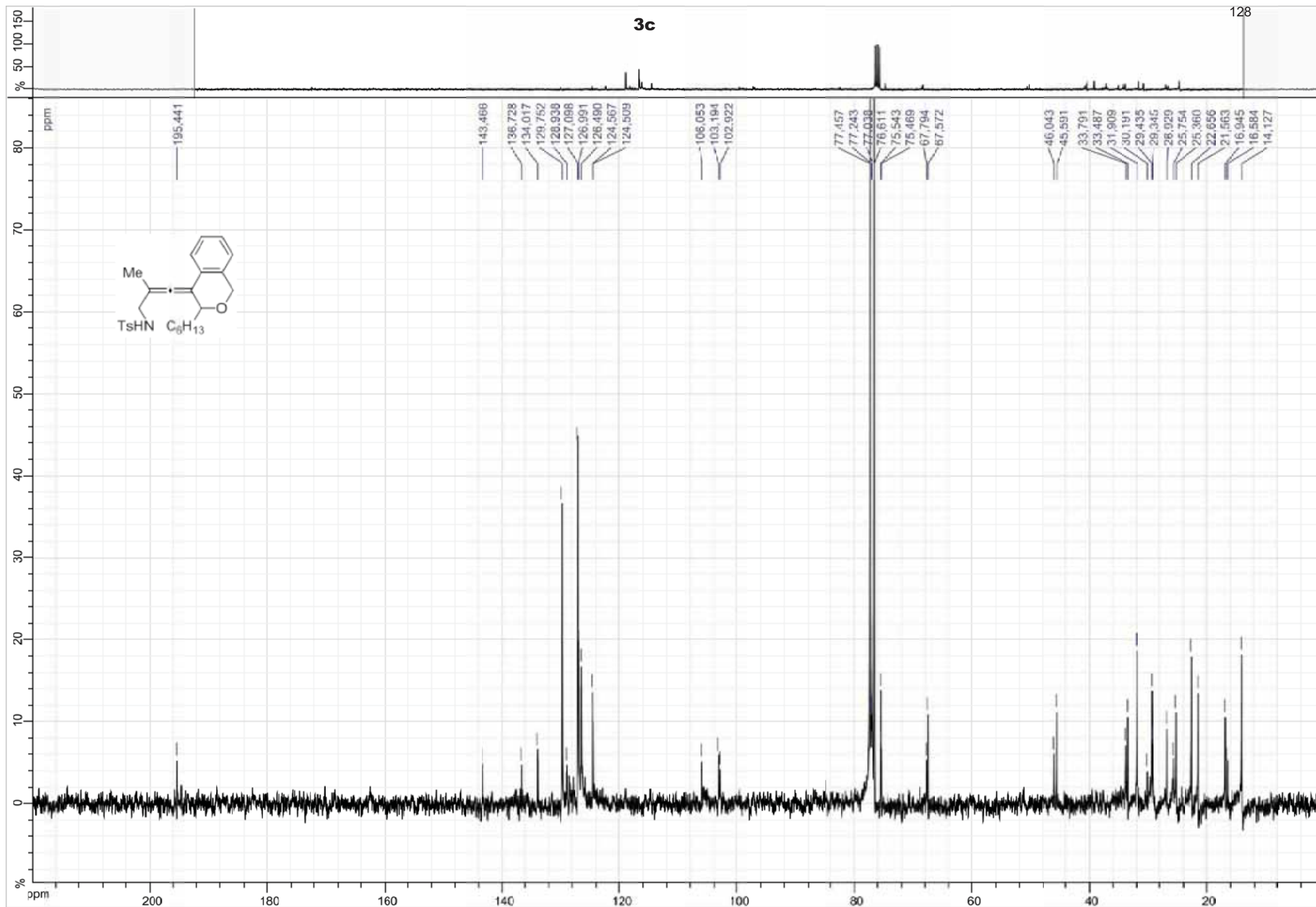


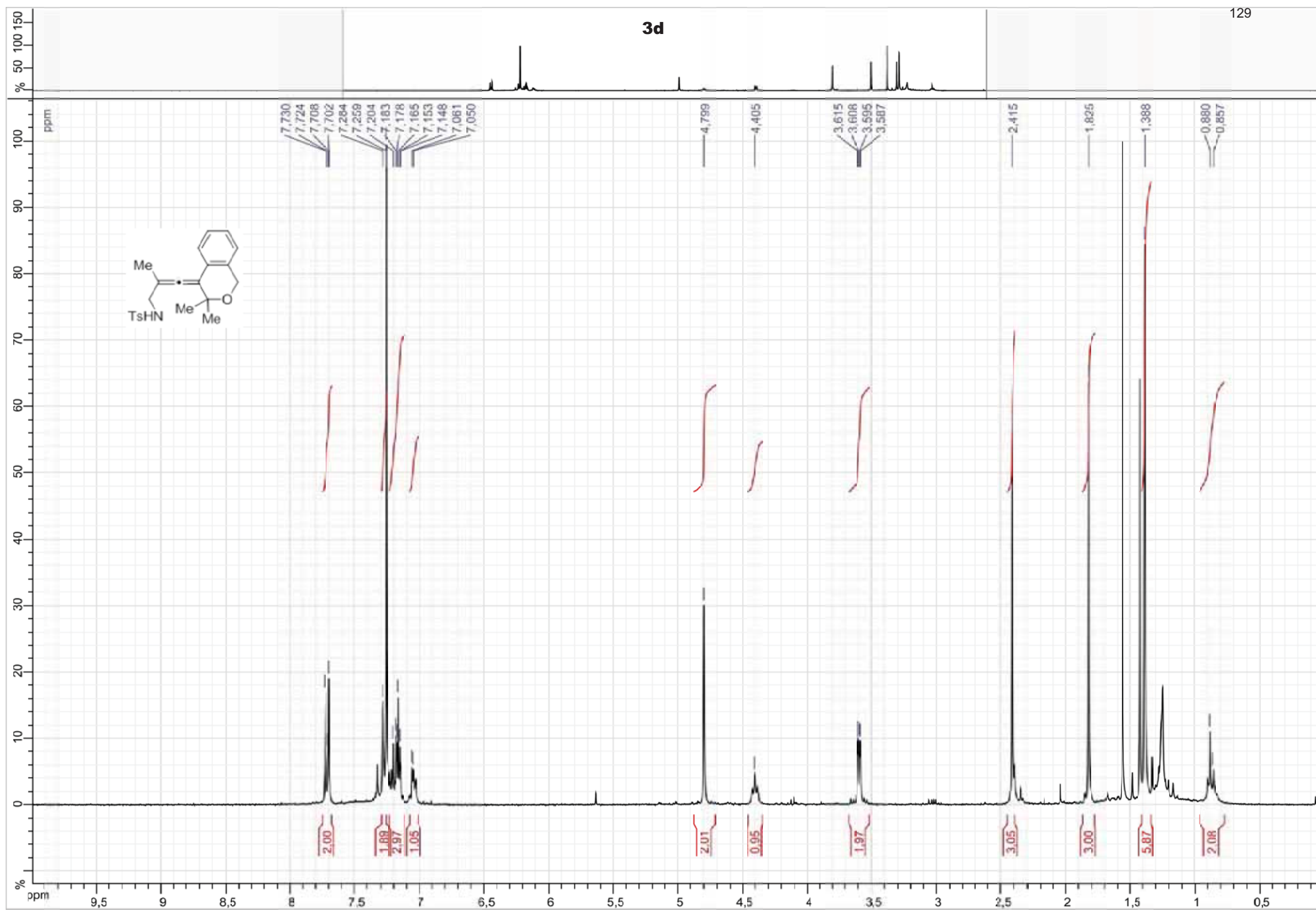


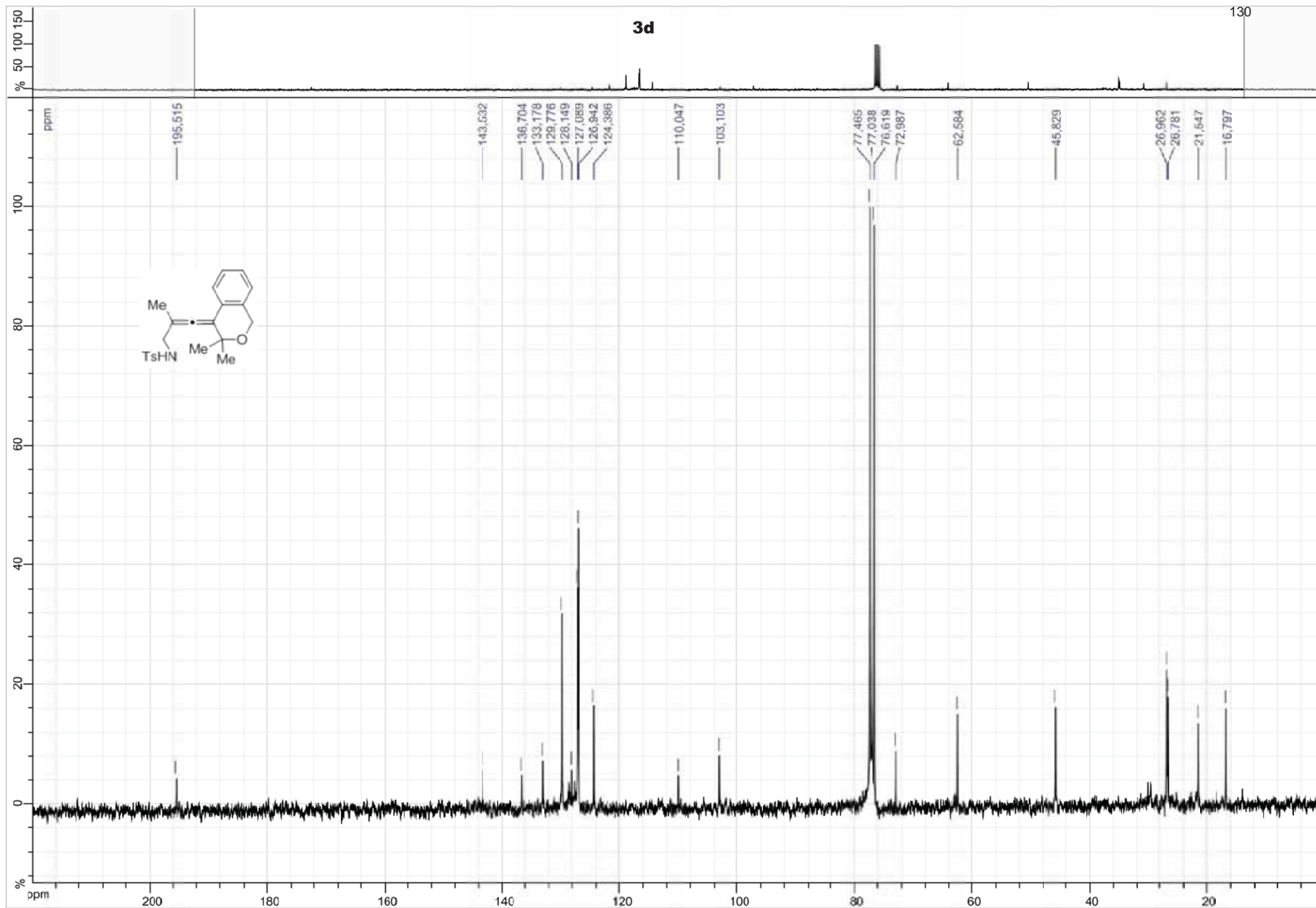


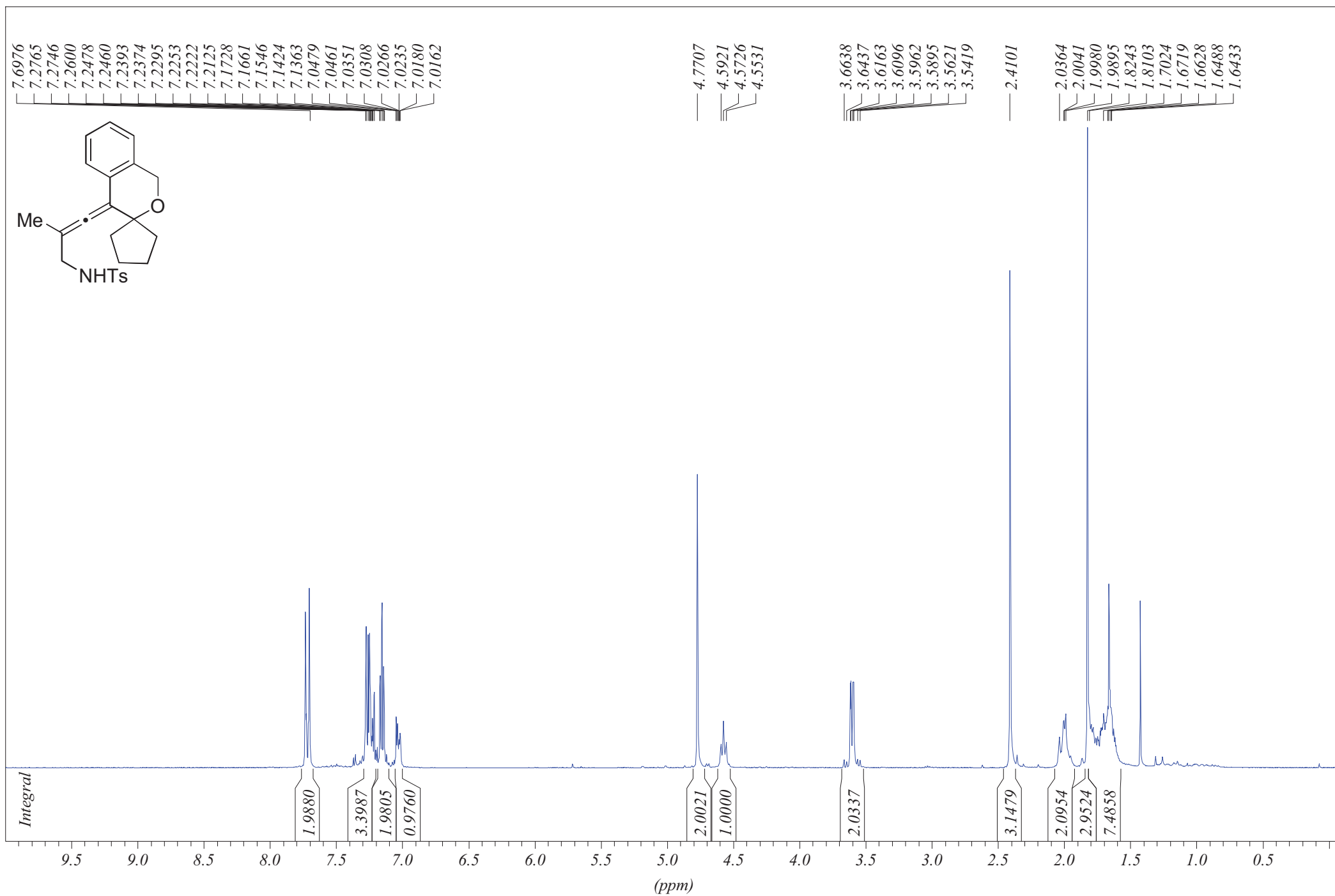


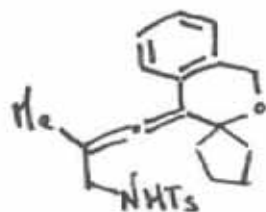
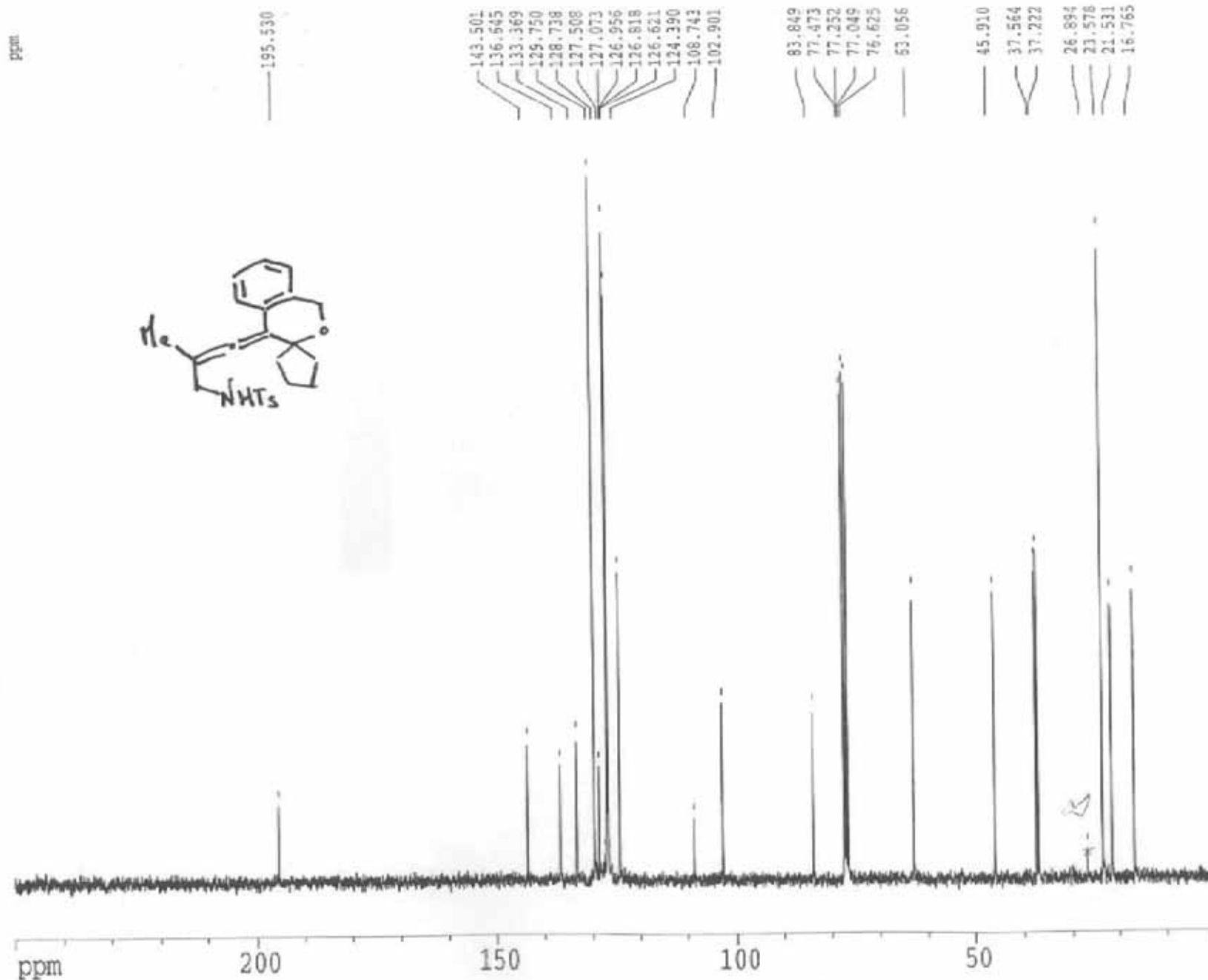












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PROCNO 1

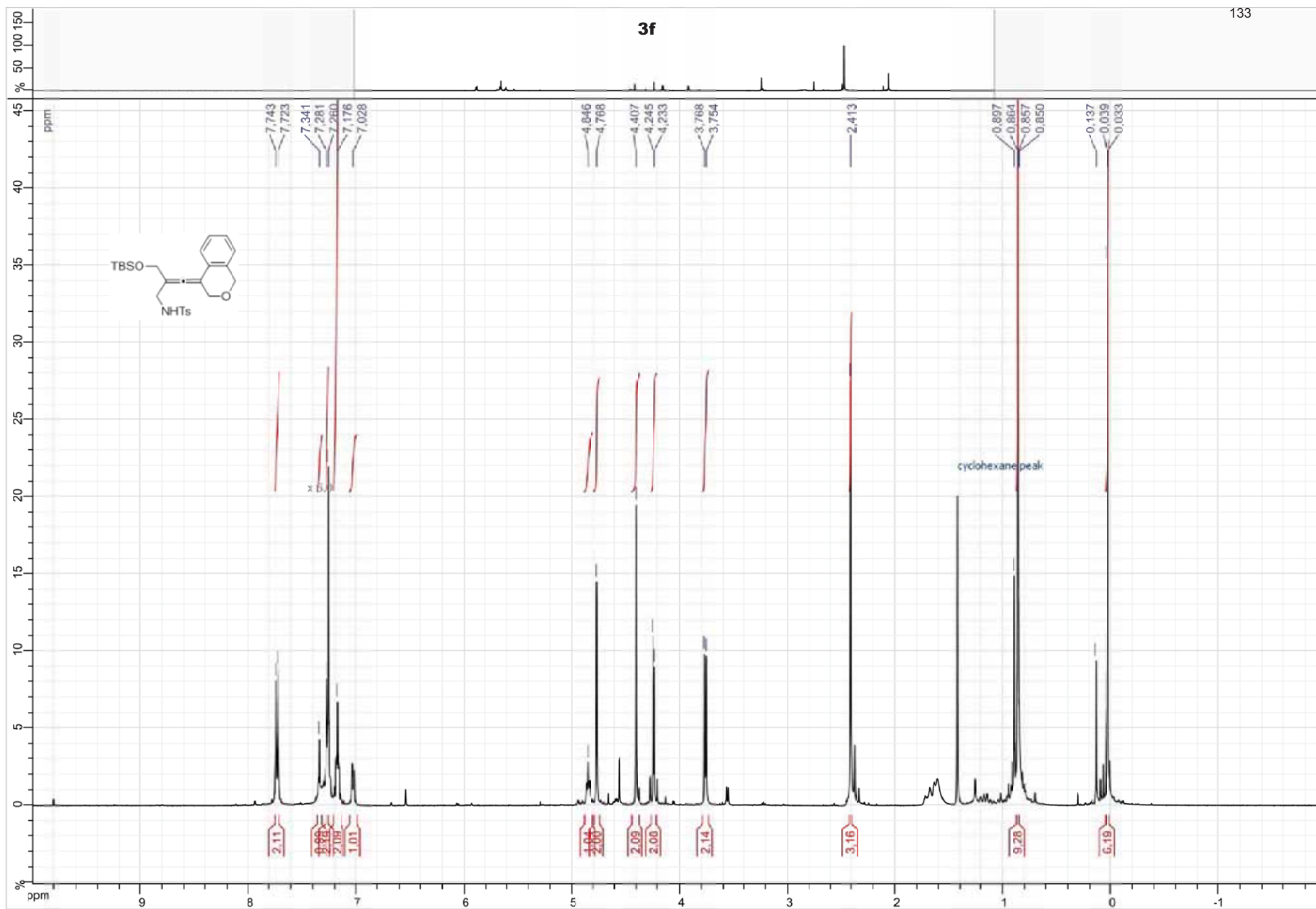
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DS 4
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FIDRES 0.620275 Hz
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DE 18.00 usec
TE 297.0 K
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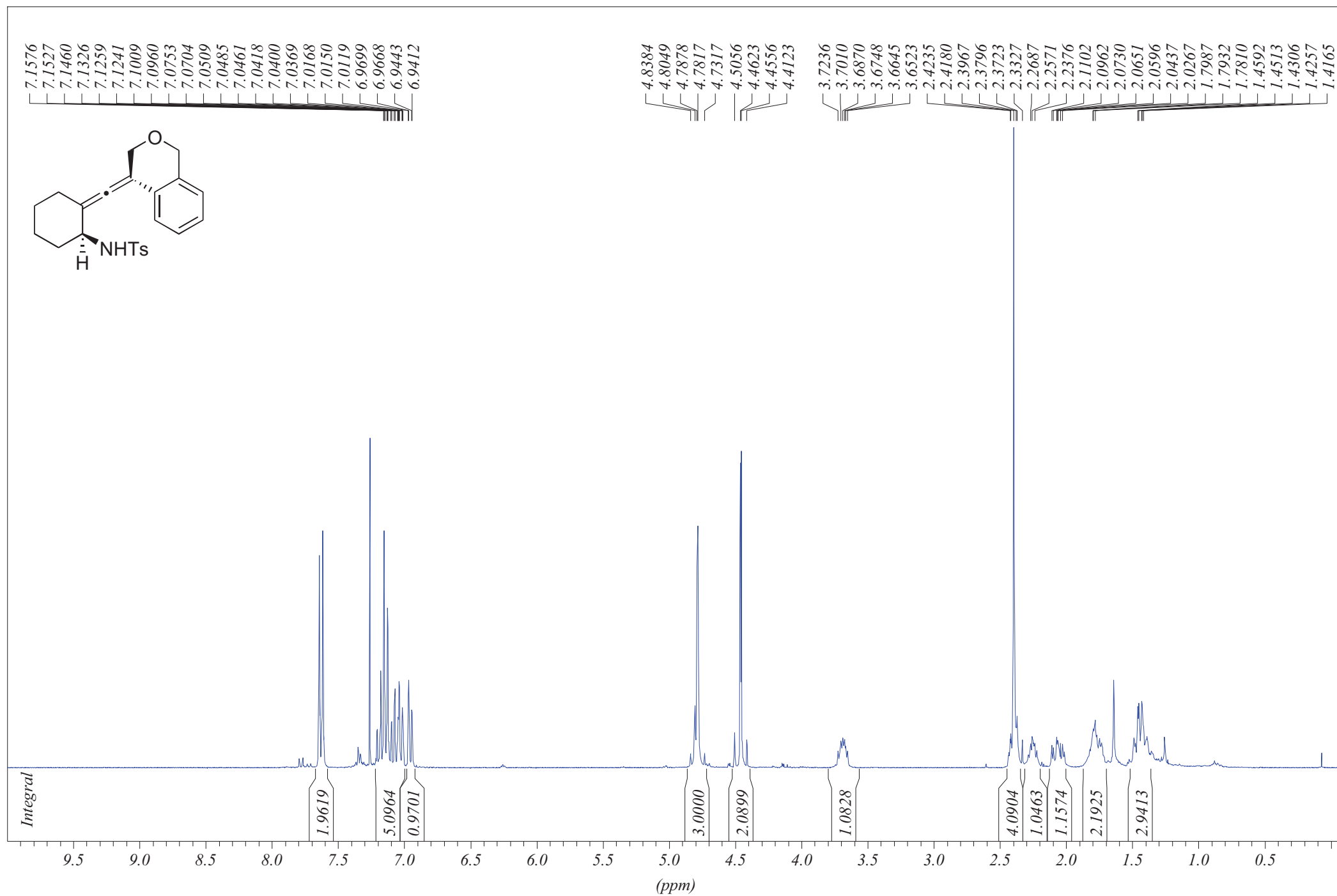
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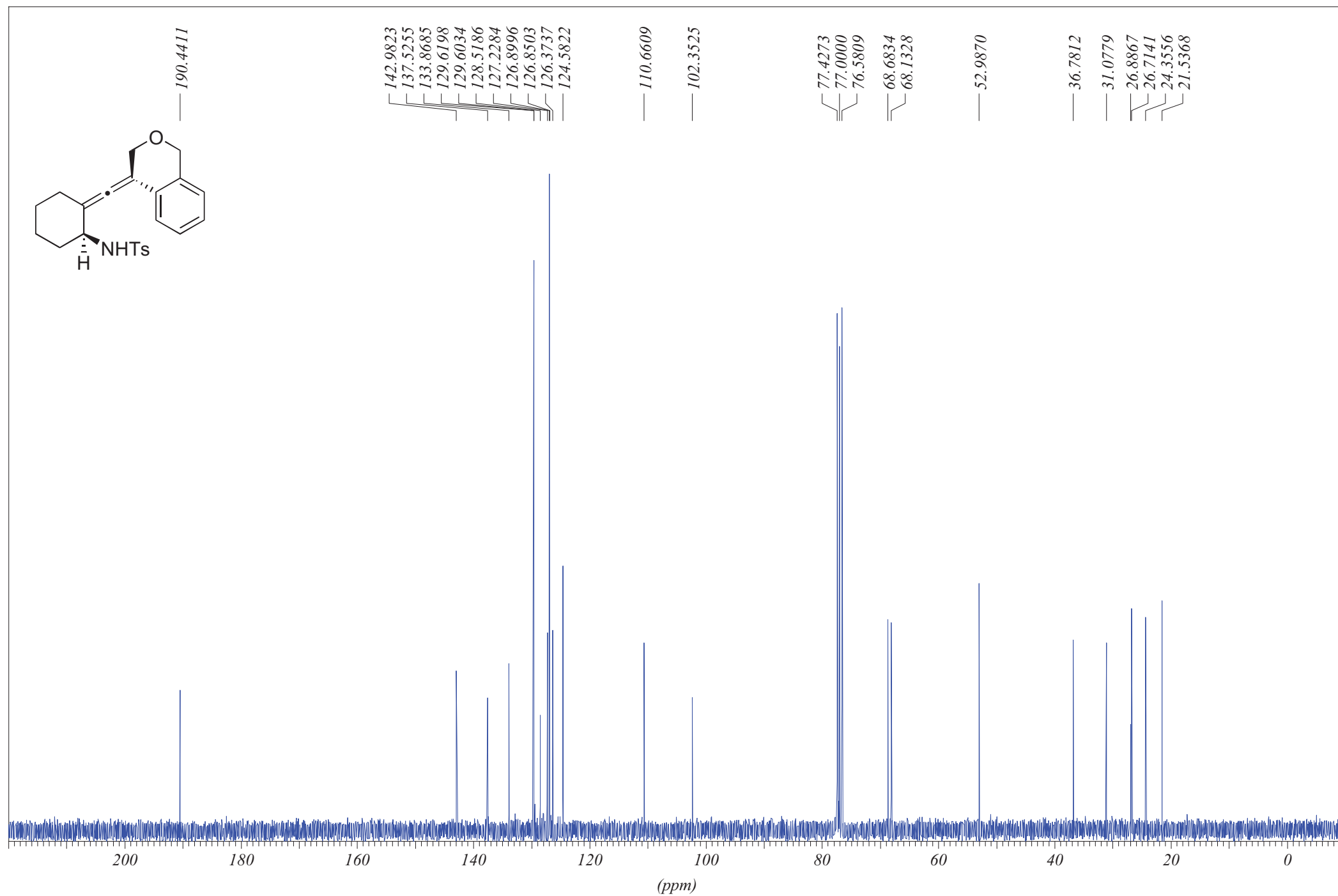
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PL2 2.00 dB
PL12 22.92 dB
SFO2 100.1711607 MHz

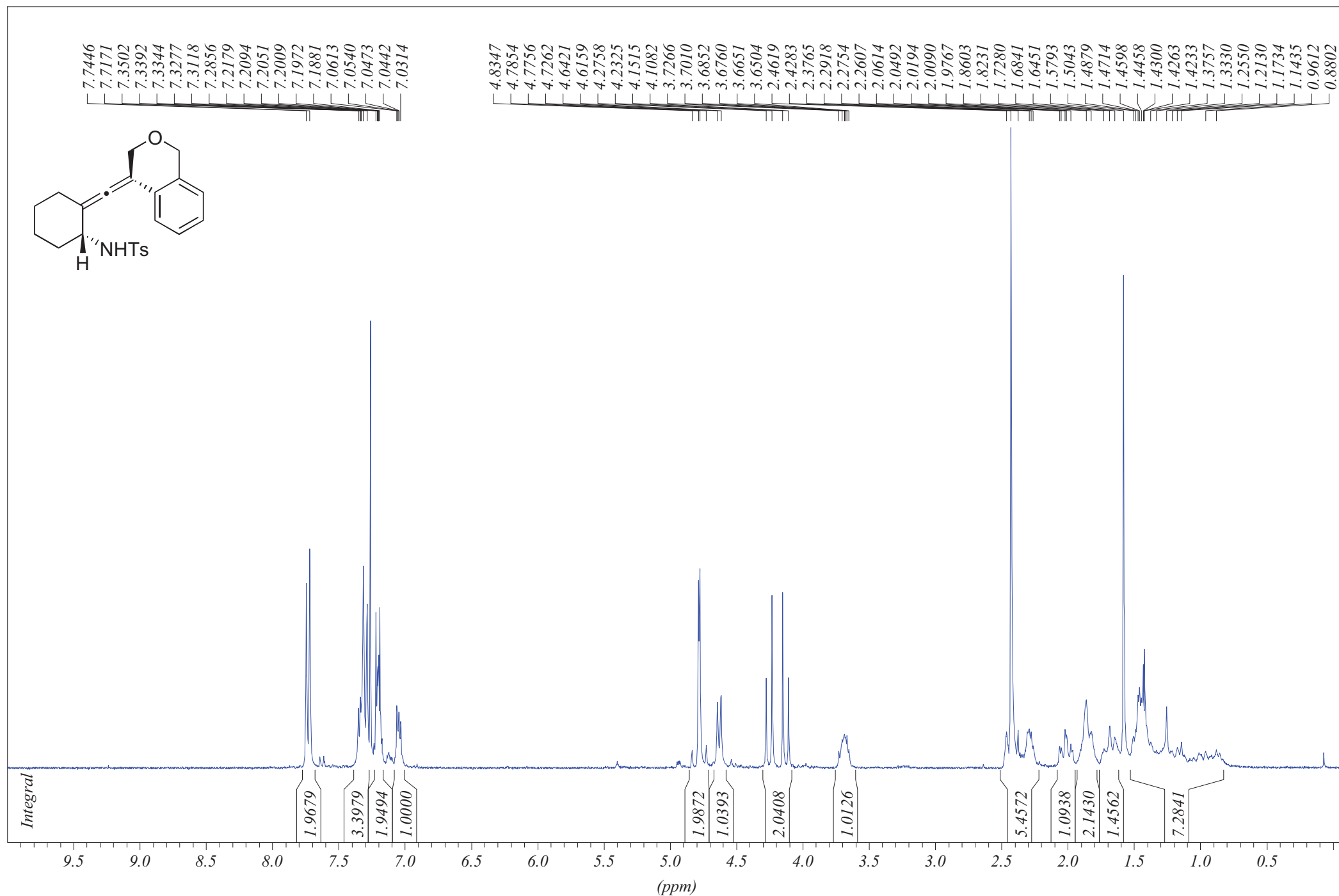
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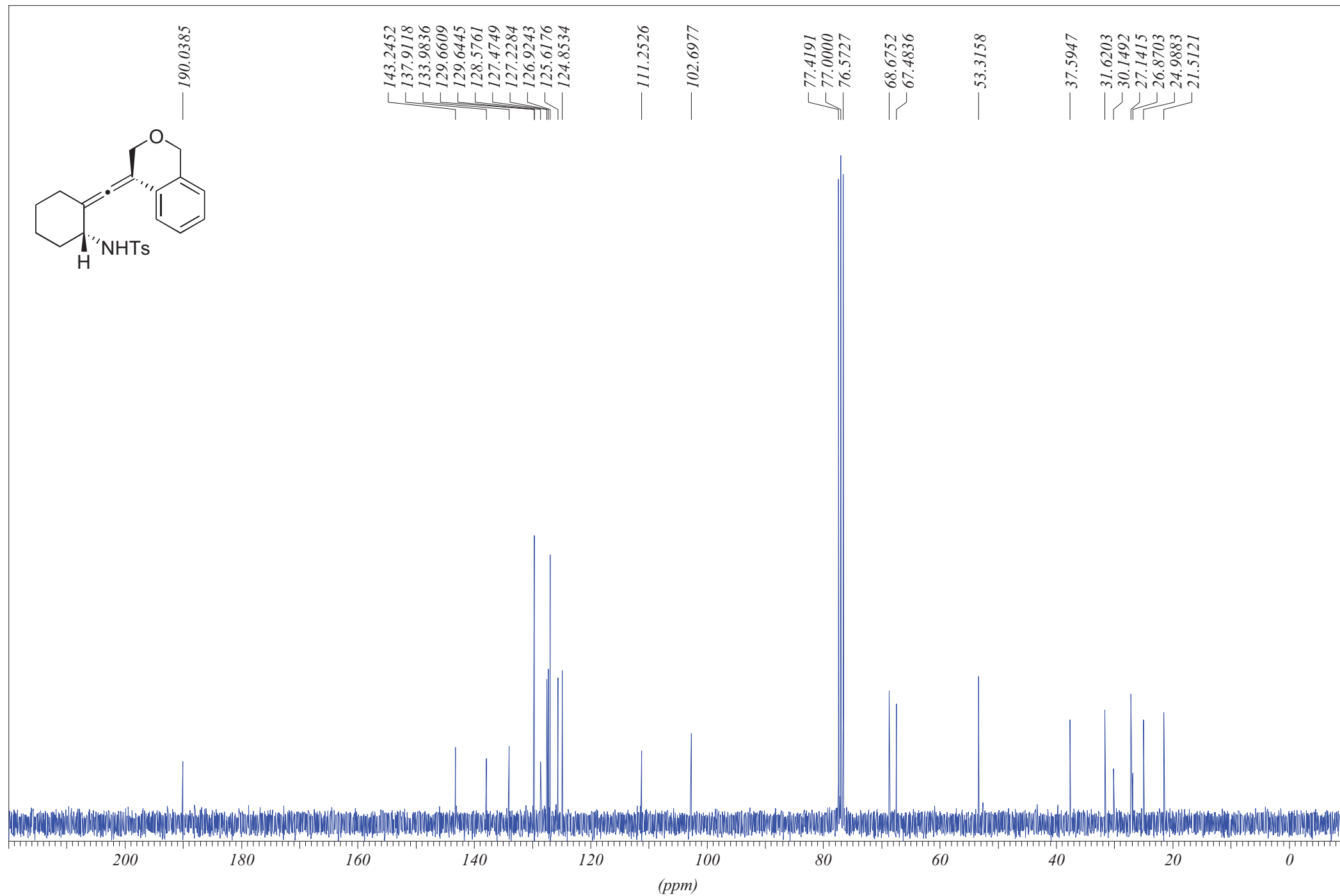
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CY 13.00 cm
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F2 0.00 Hz
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HSCM 857.70245 Hz/cm





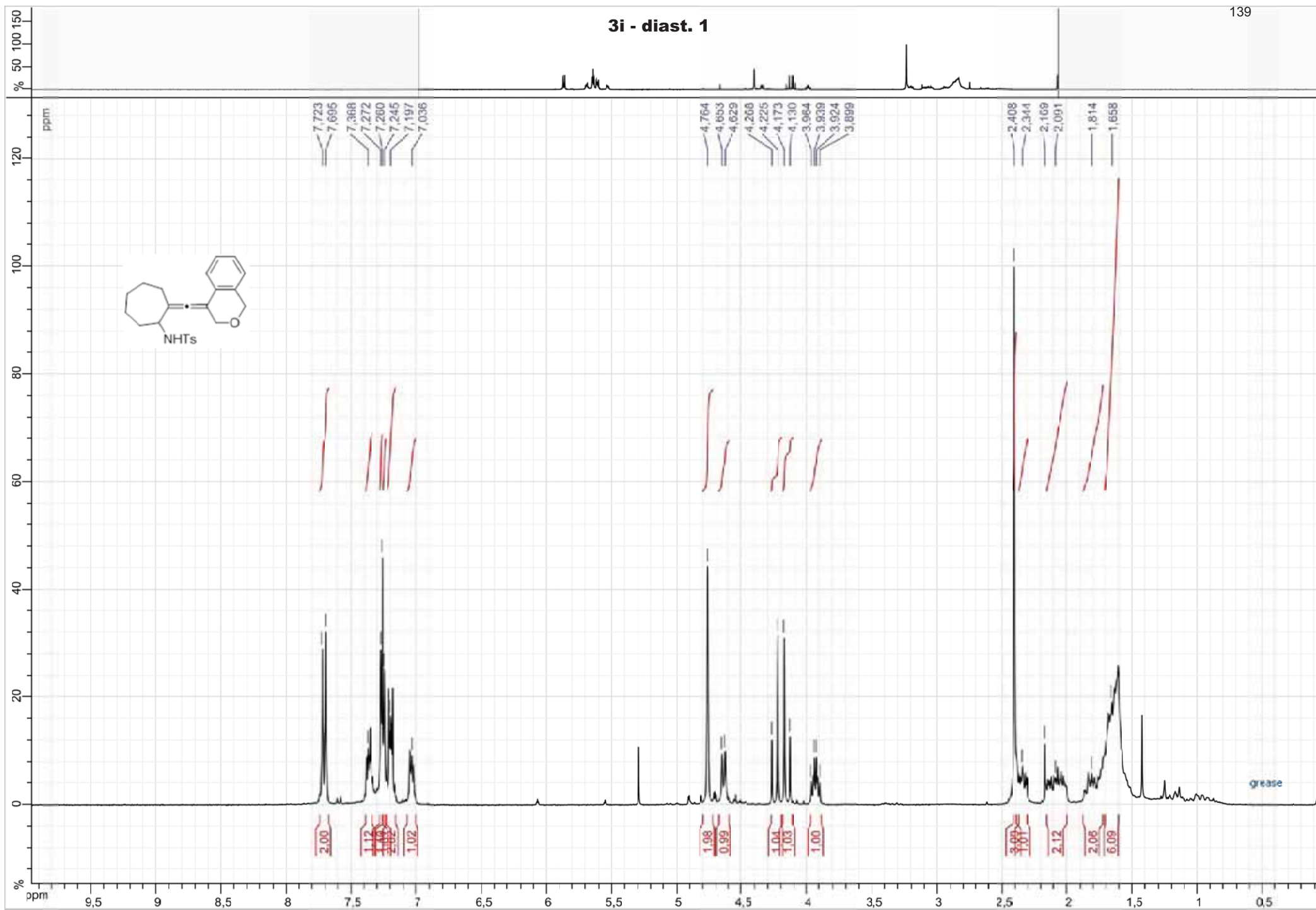
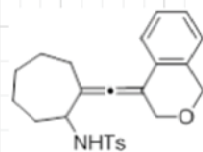






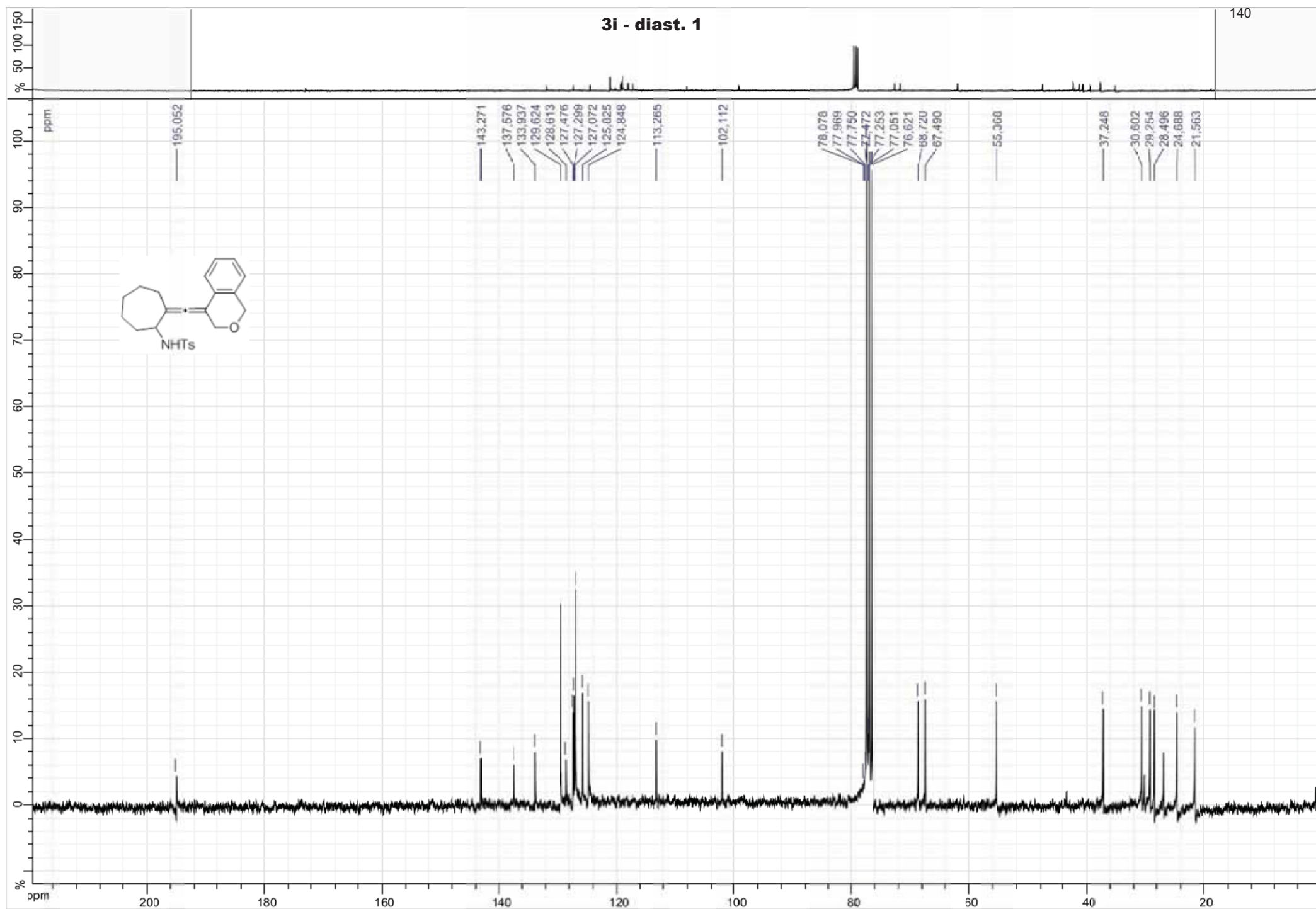
3i - diast. 1

139



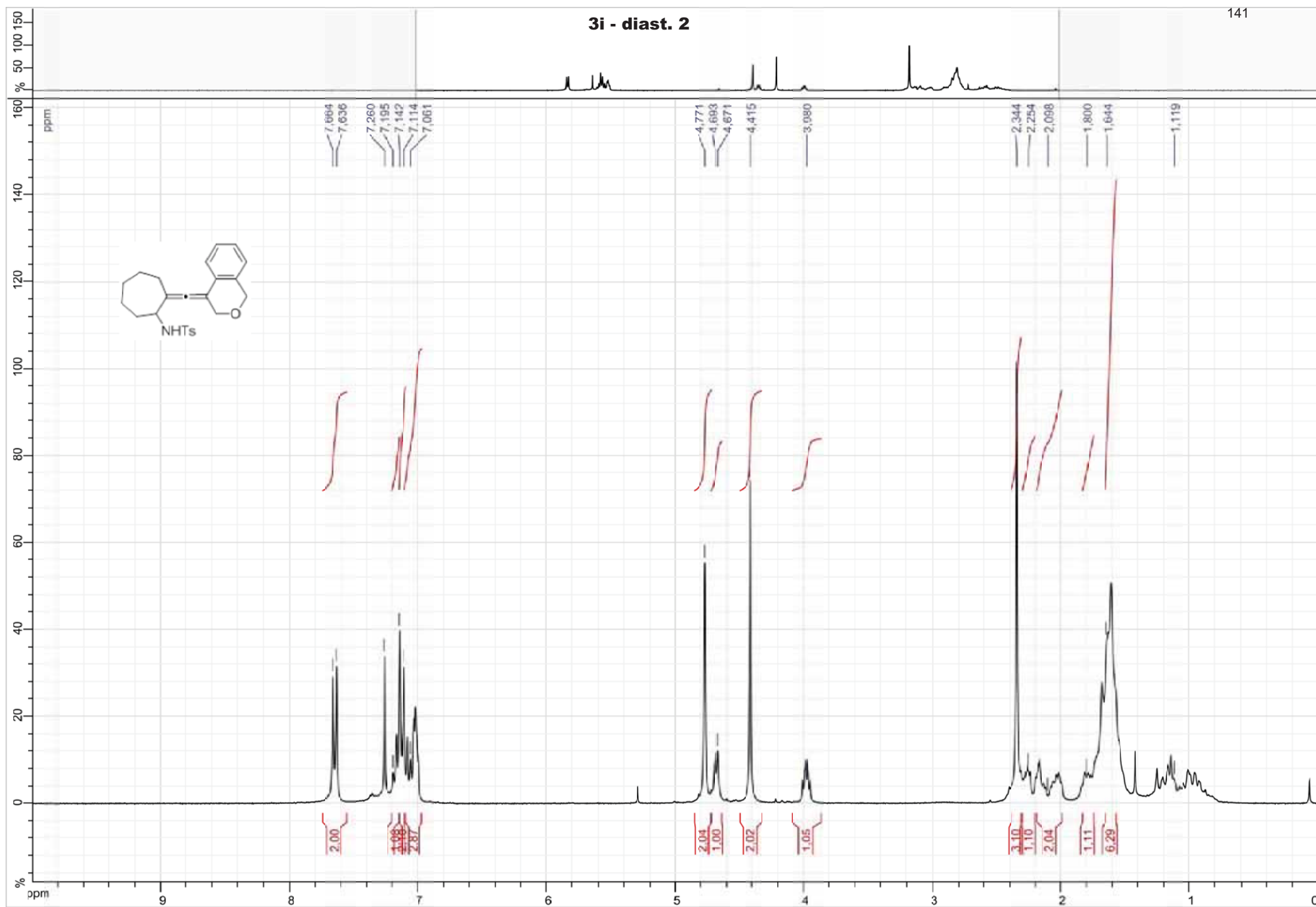
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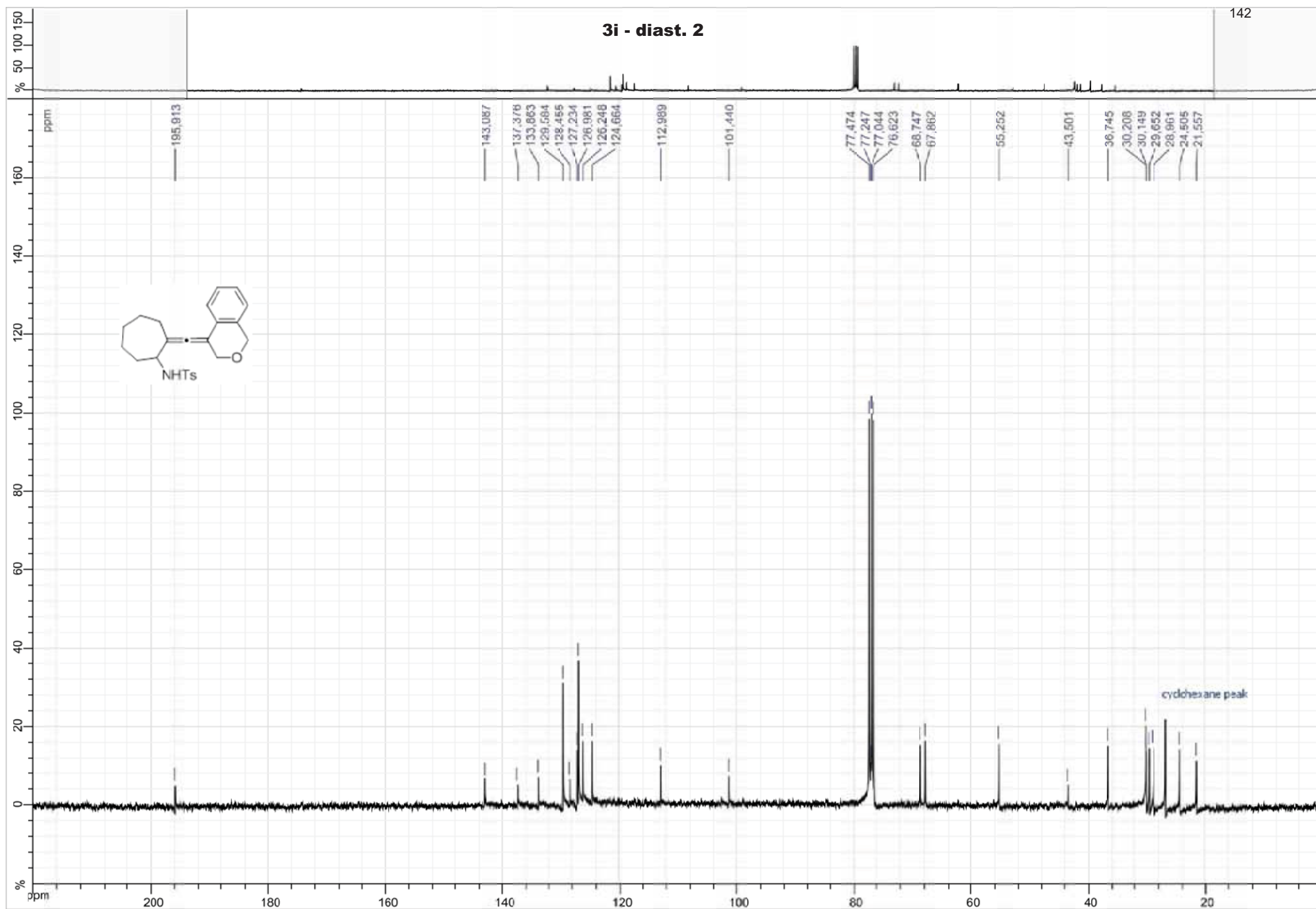
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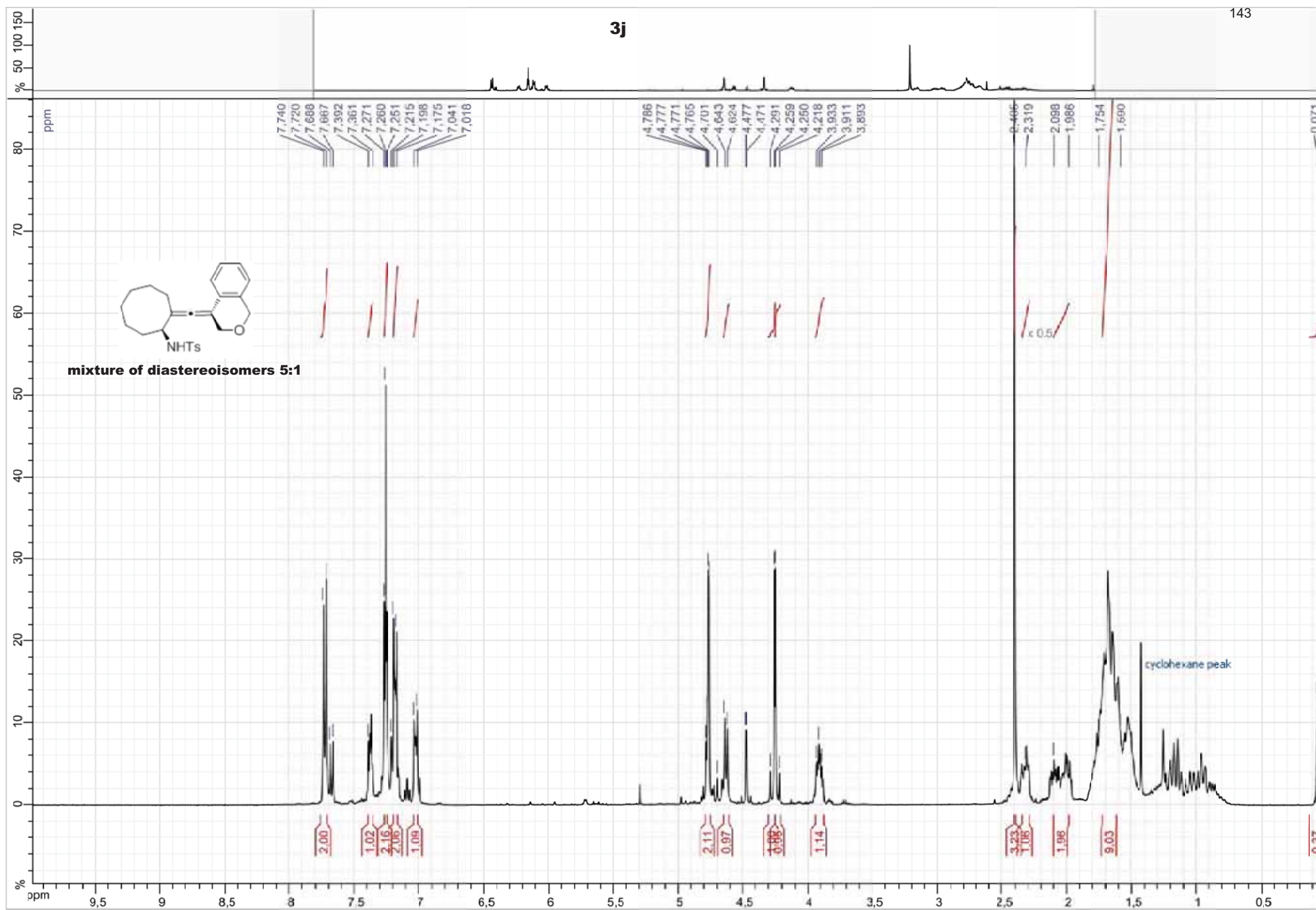
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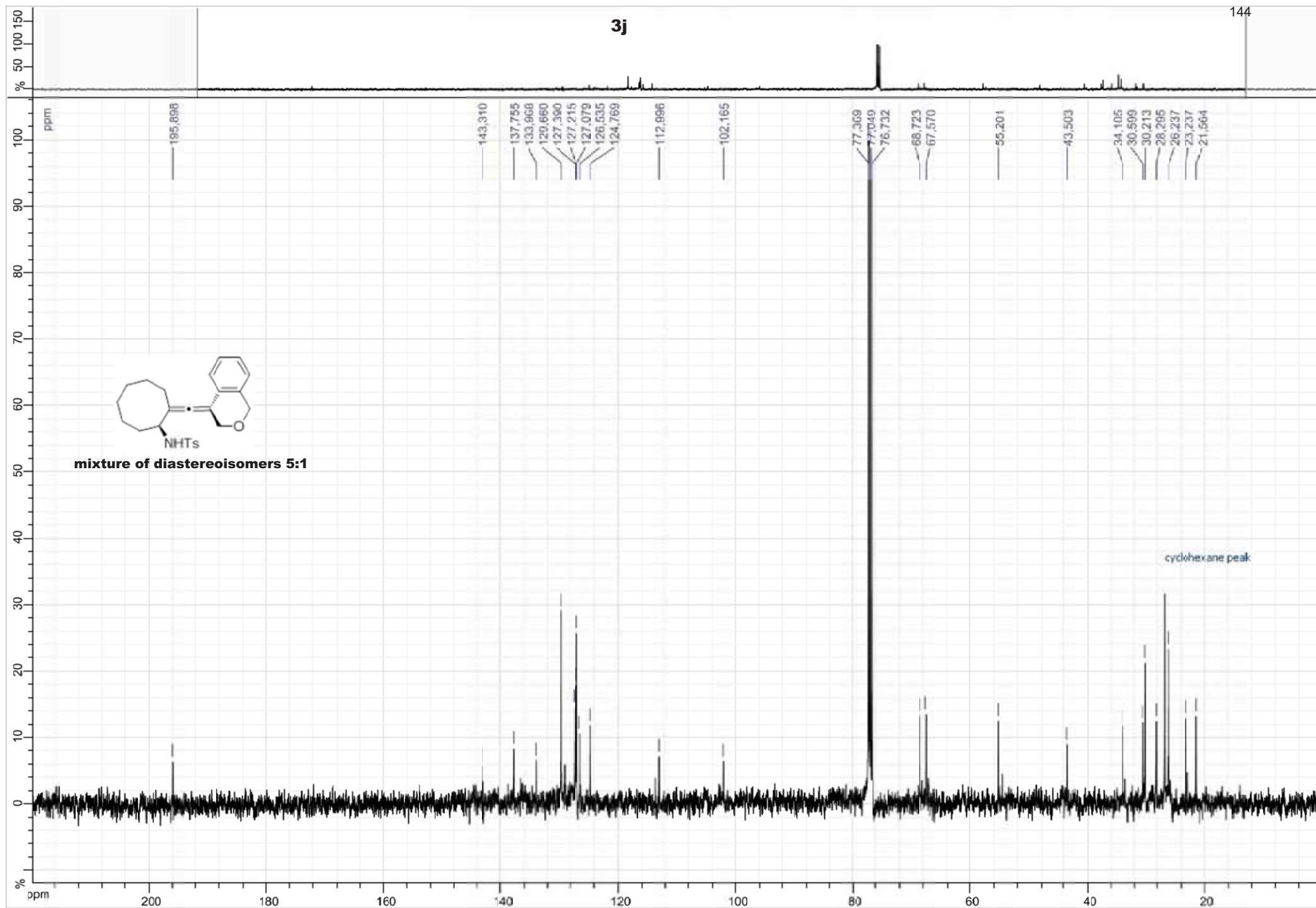
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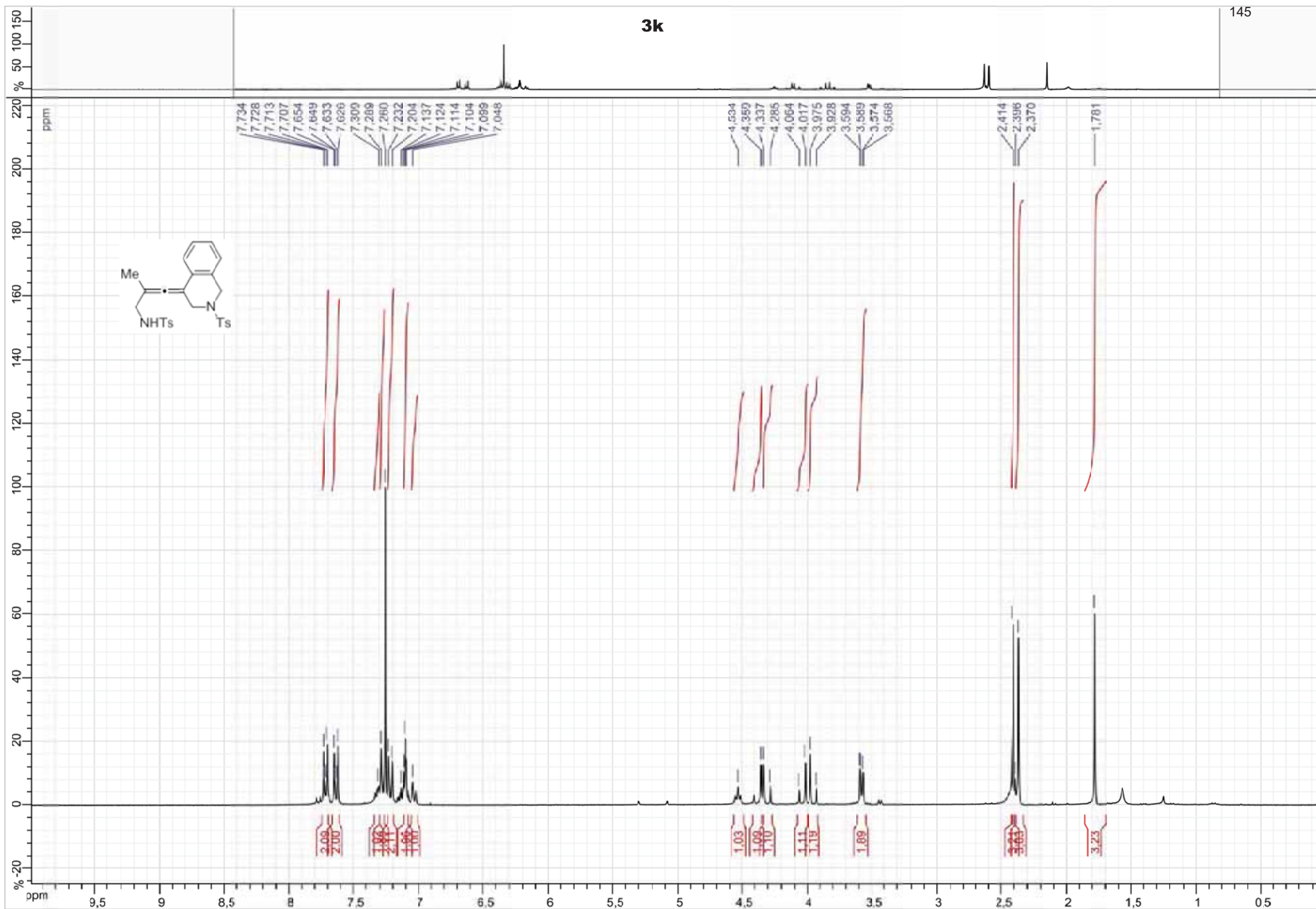
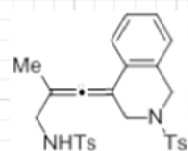
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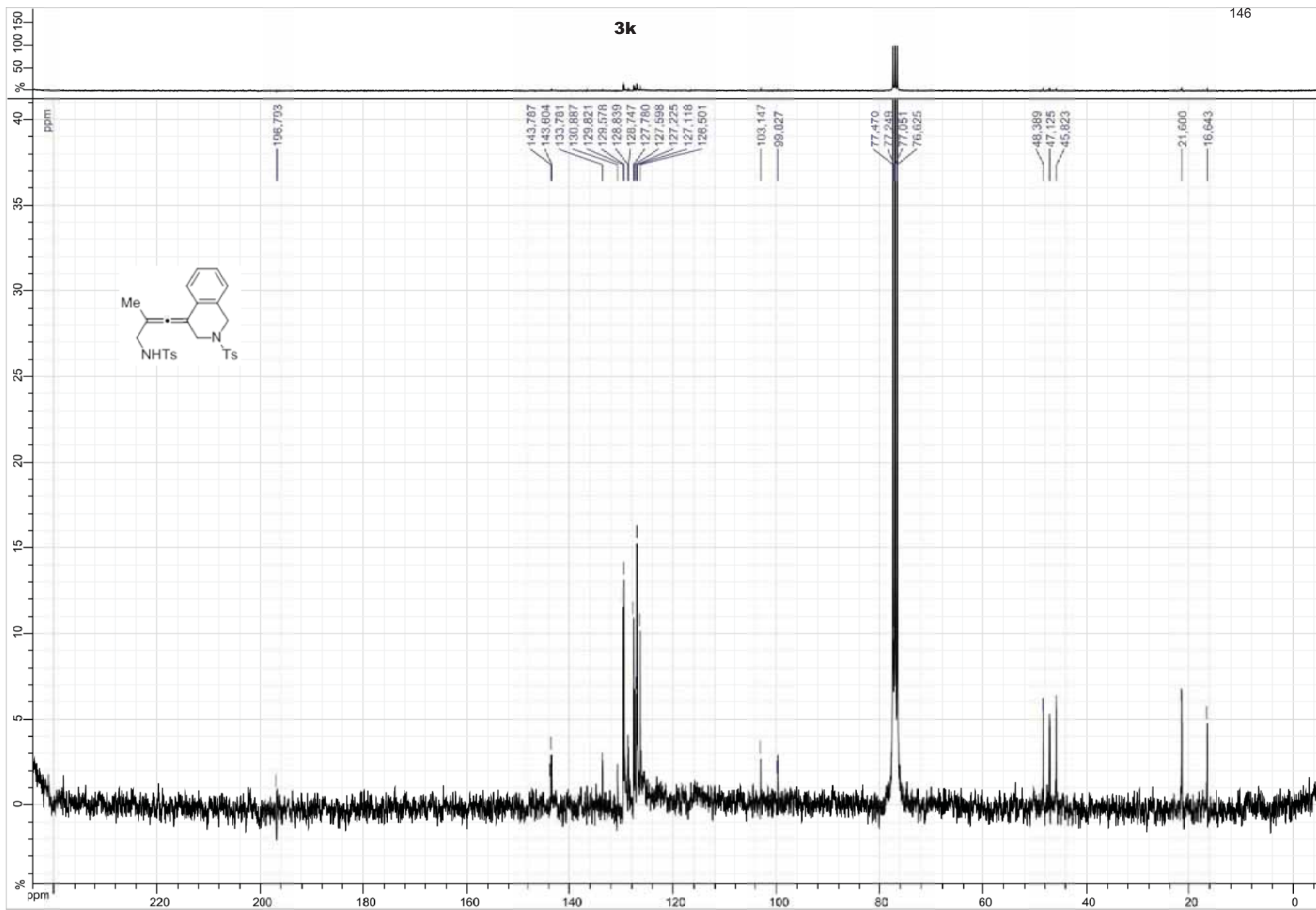


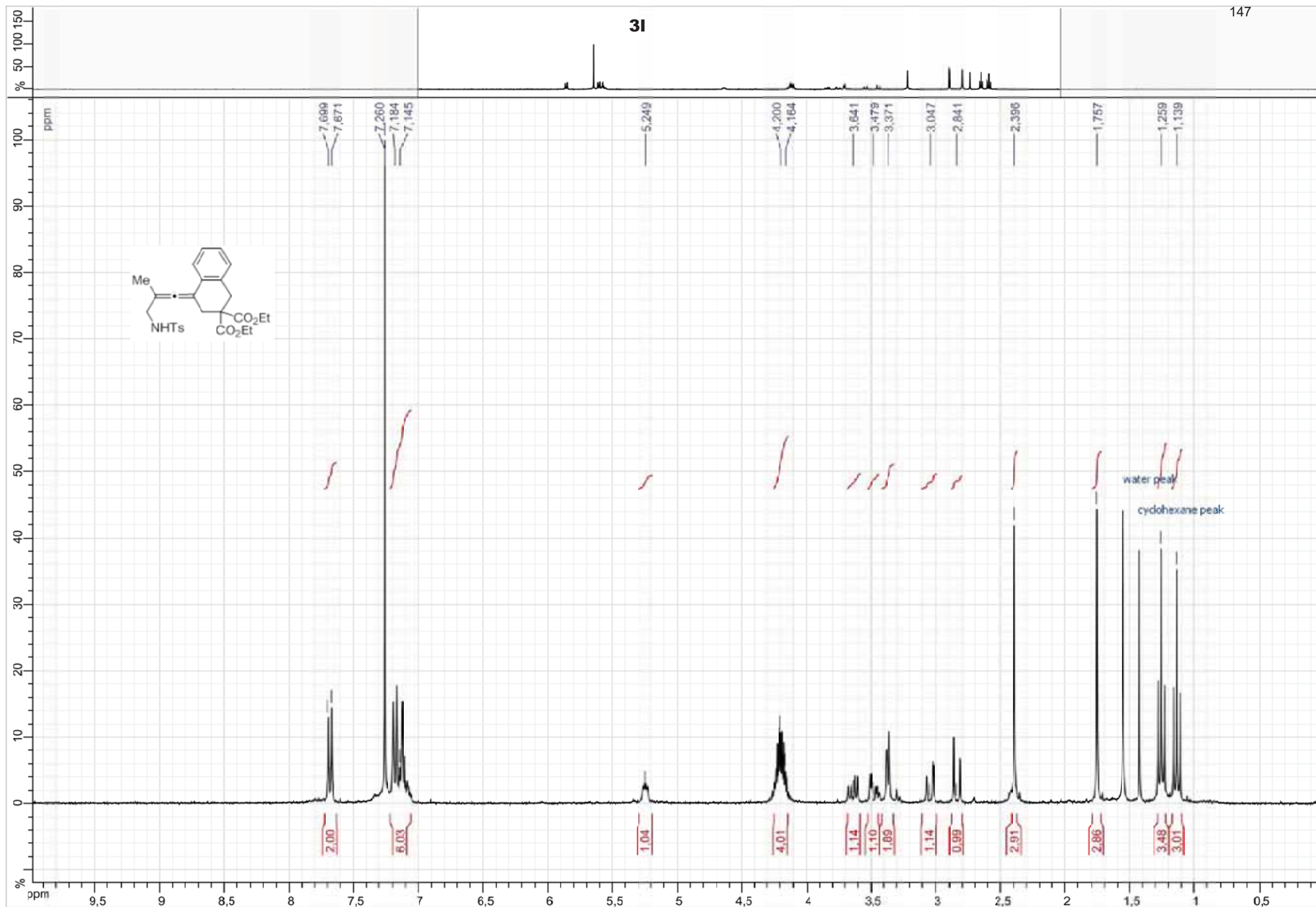


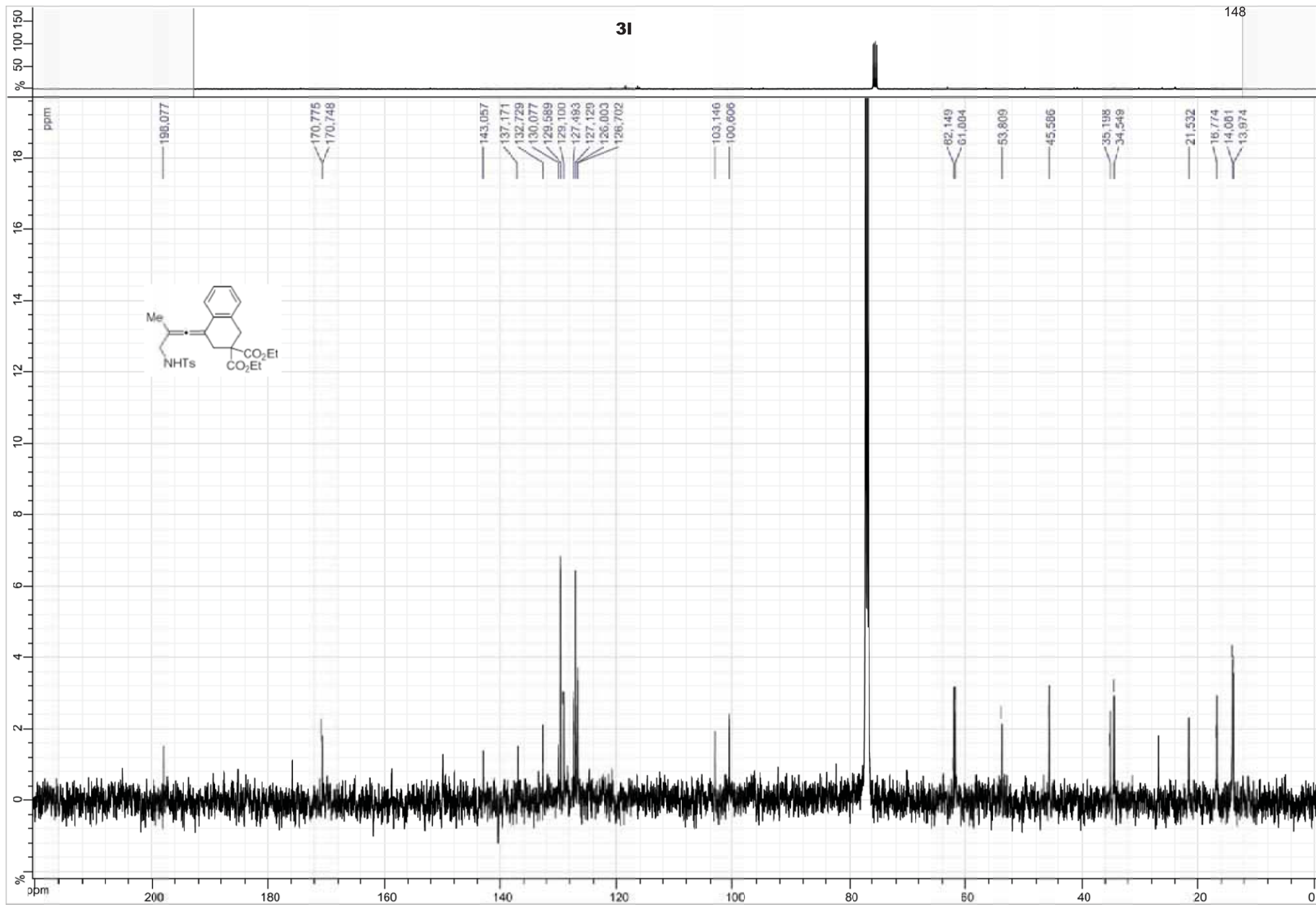
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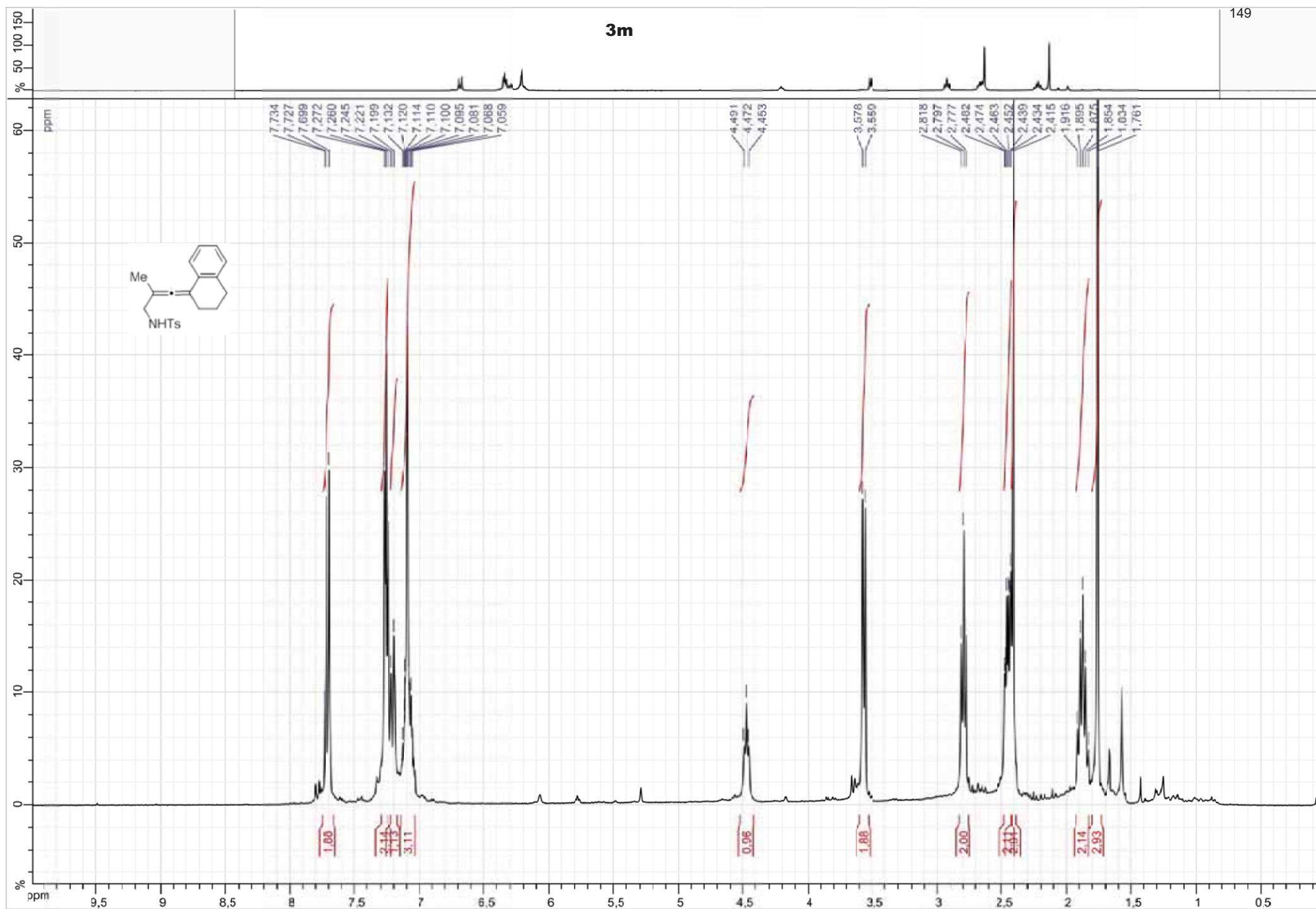
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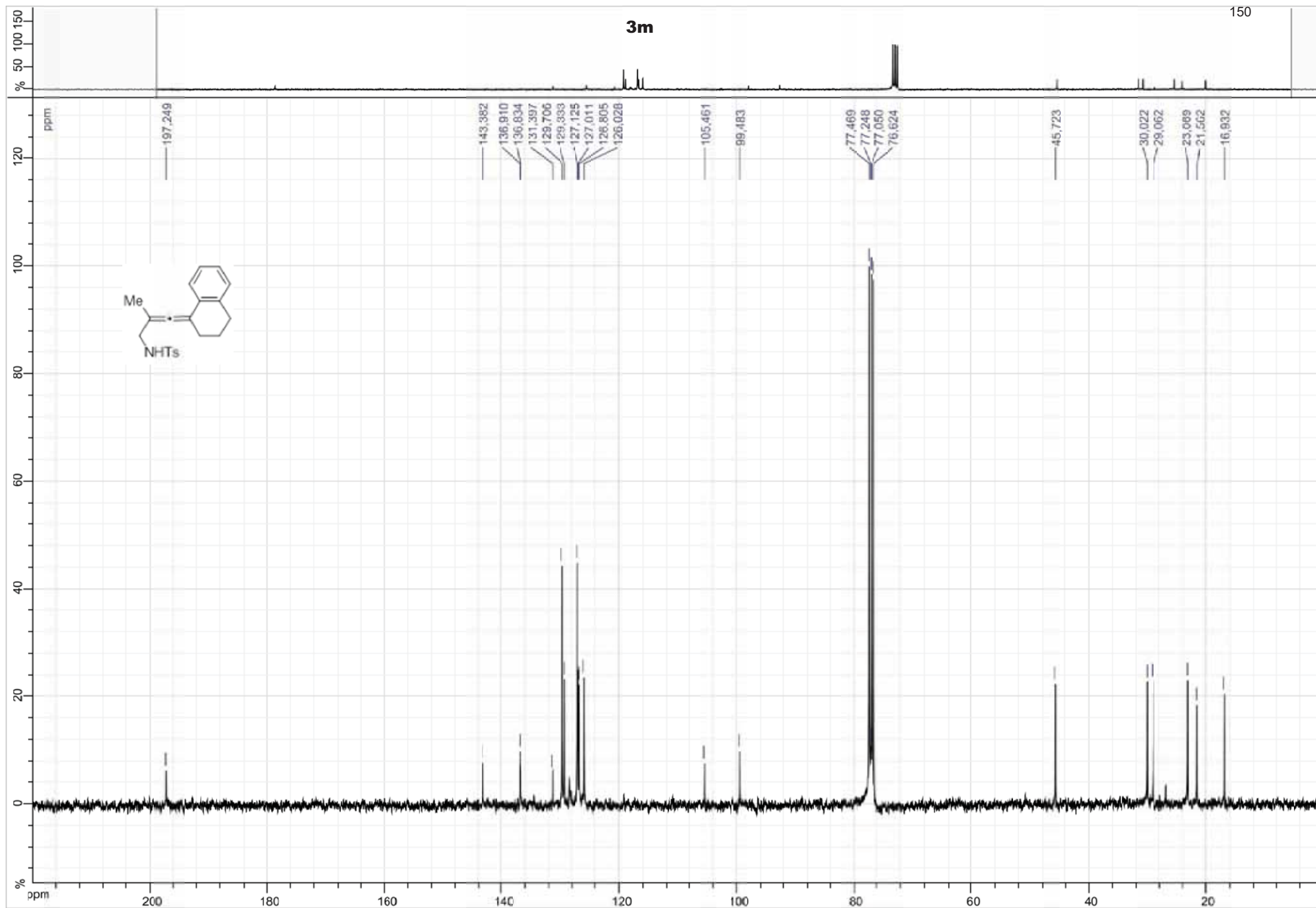


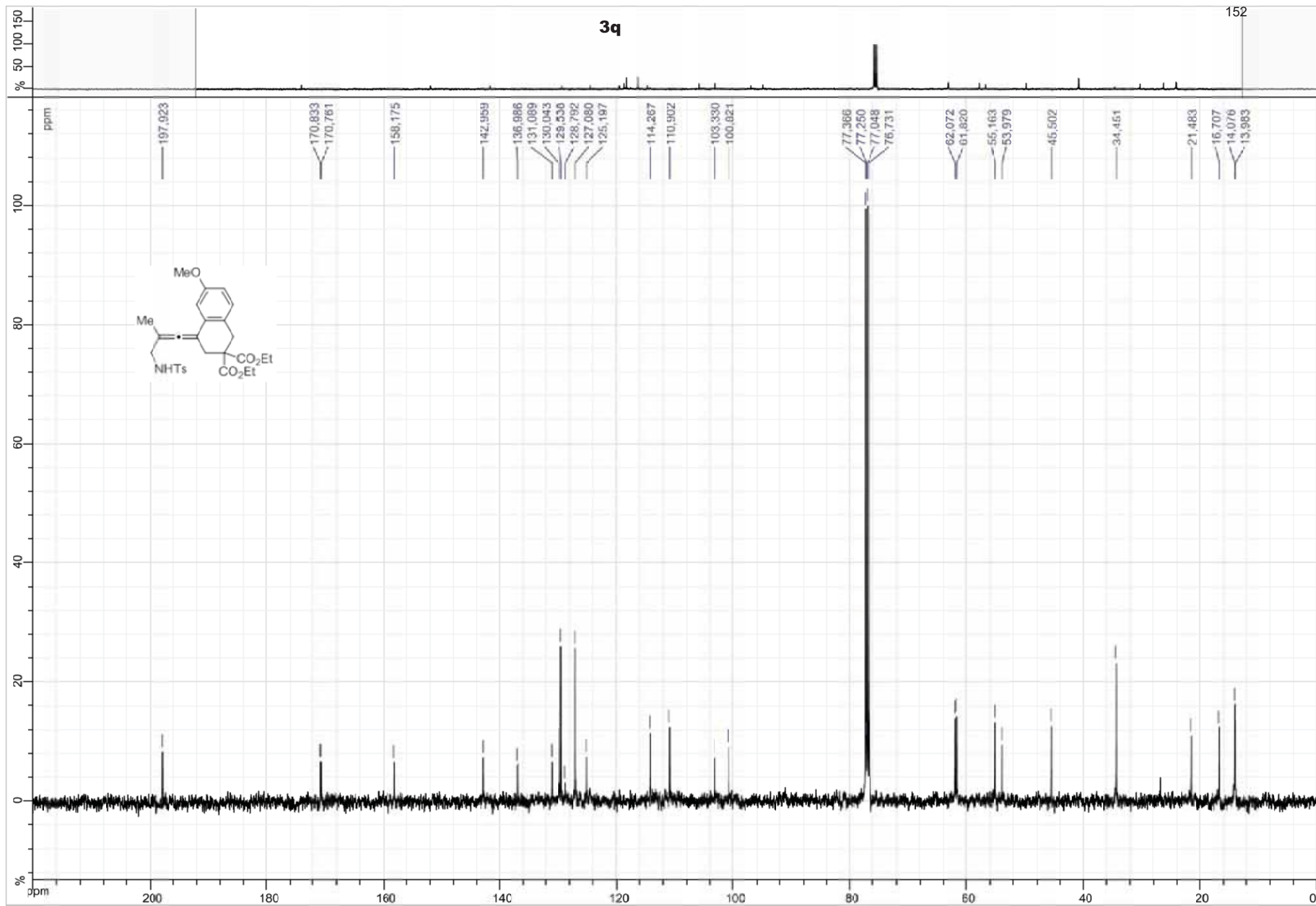
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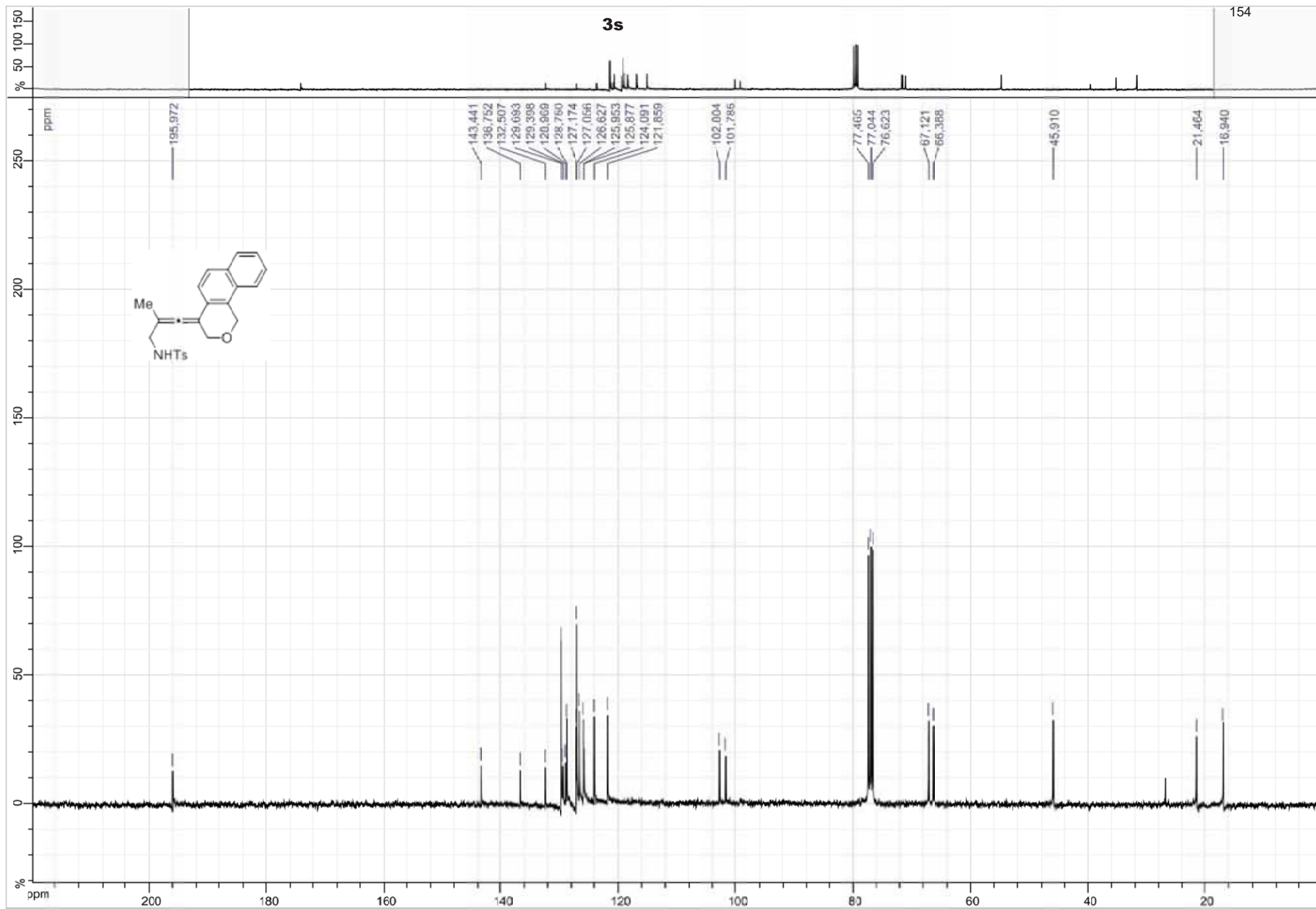


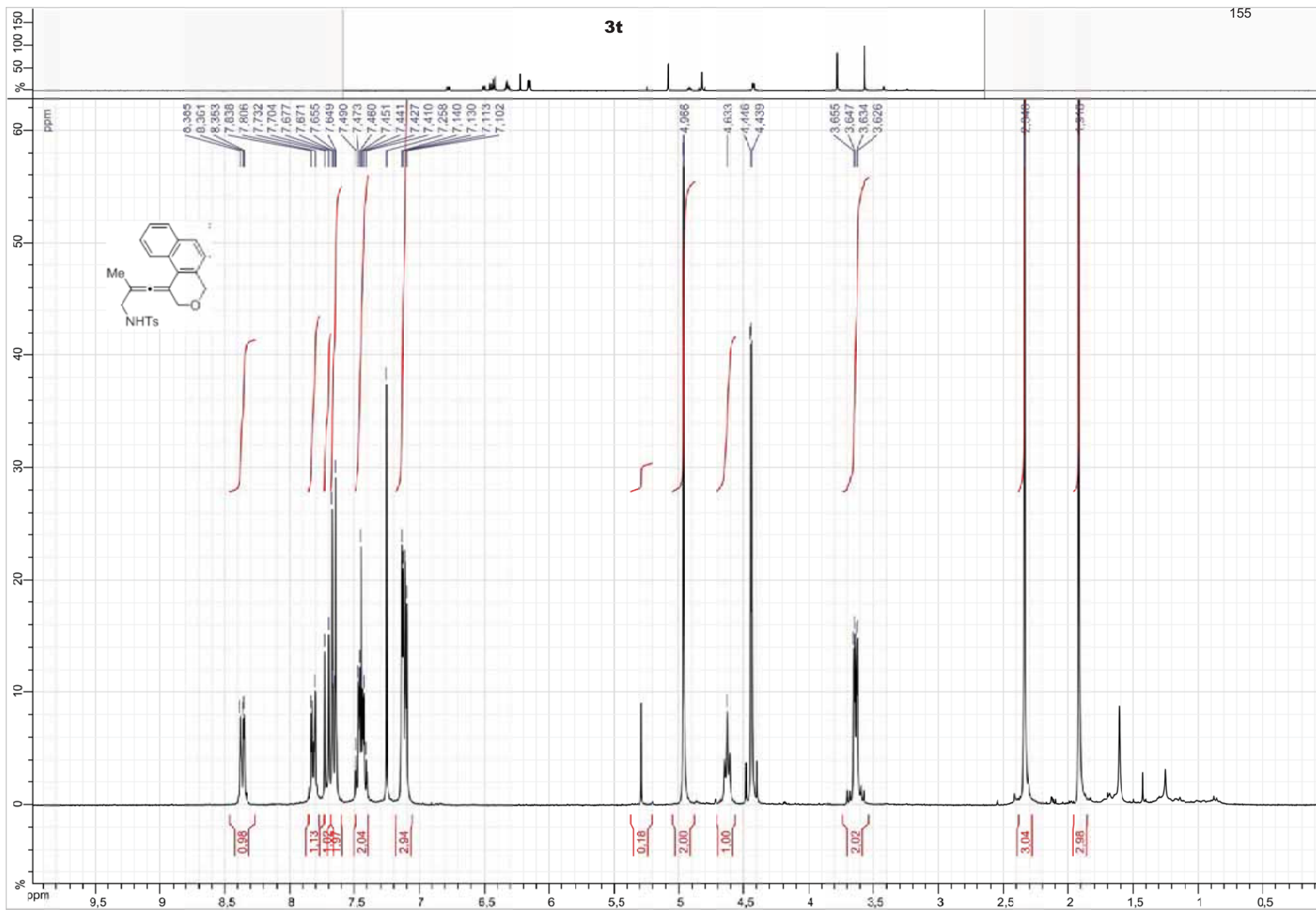












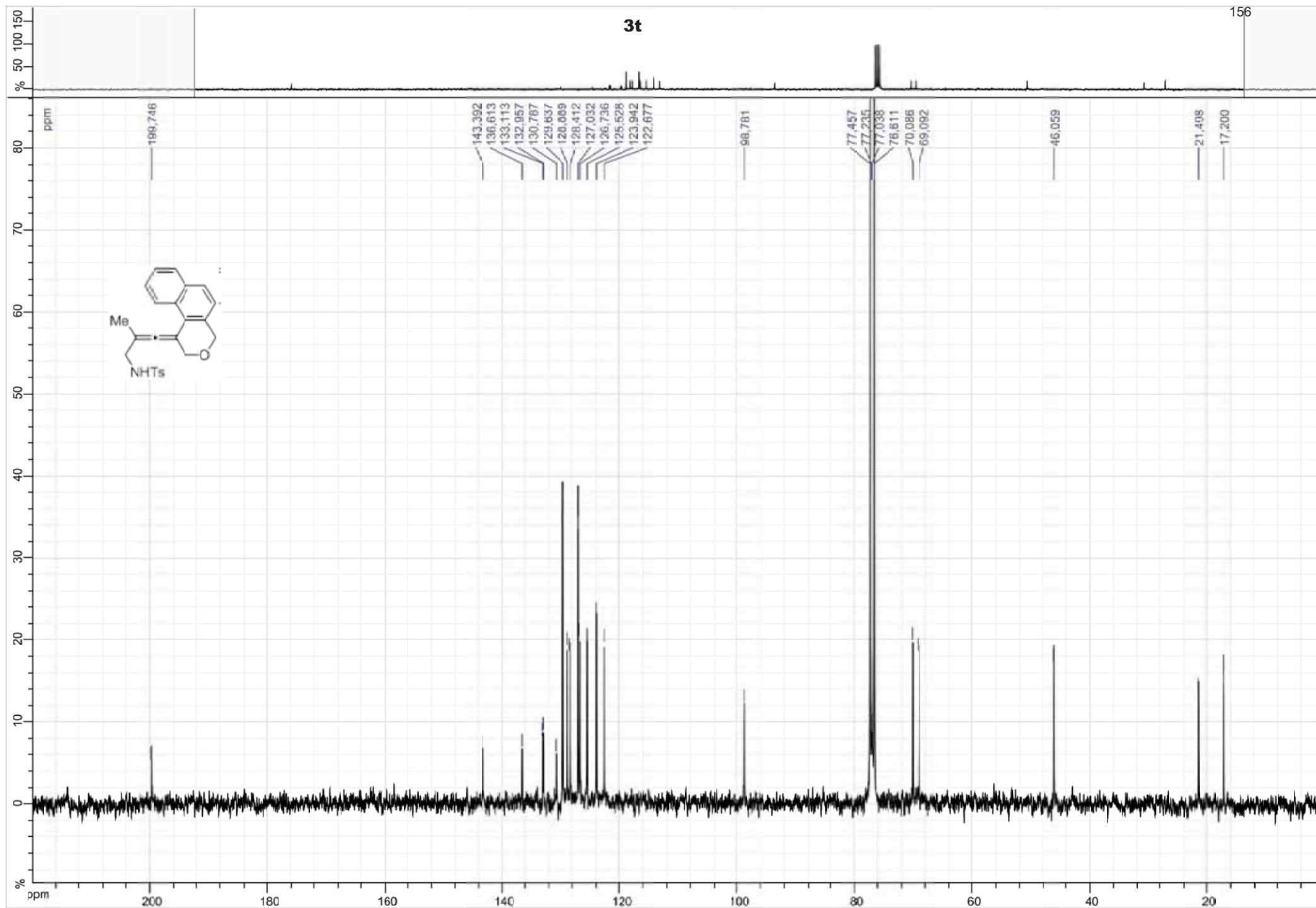
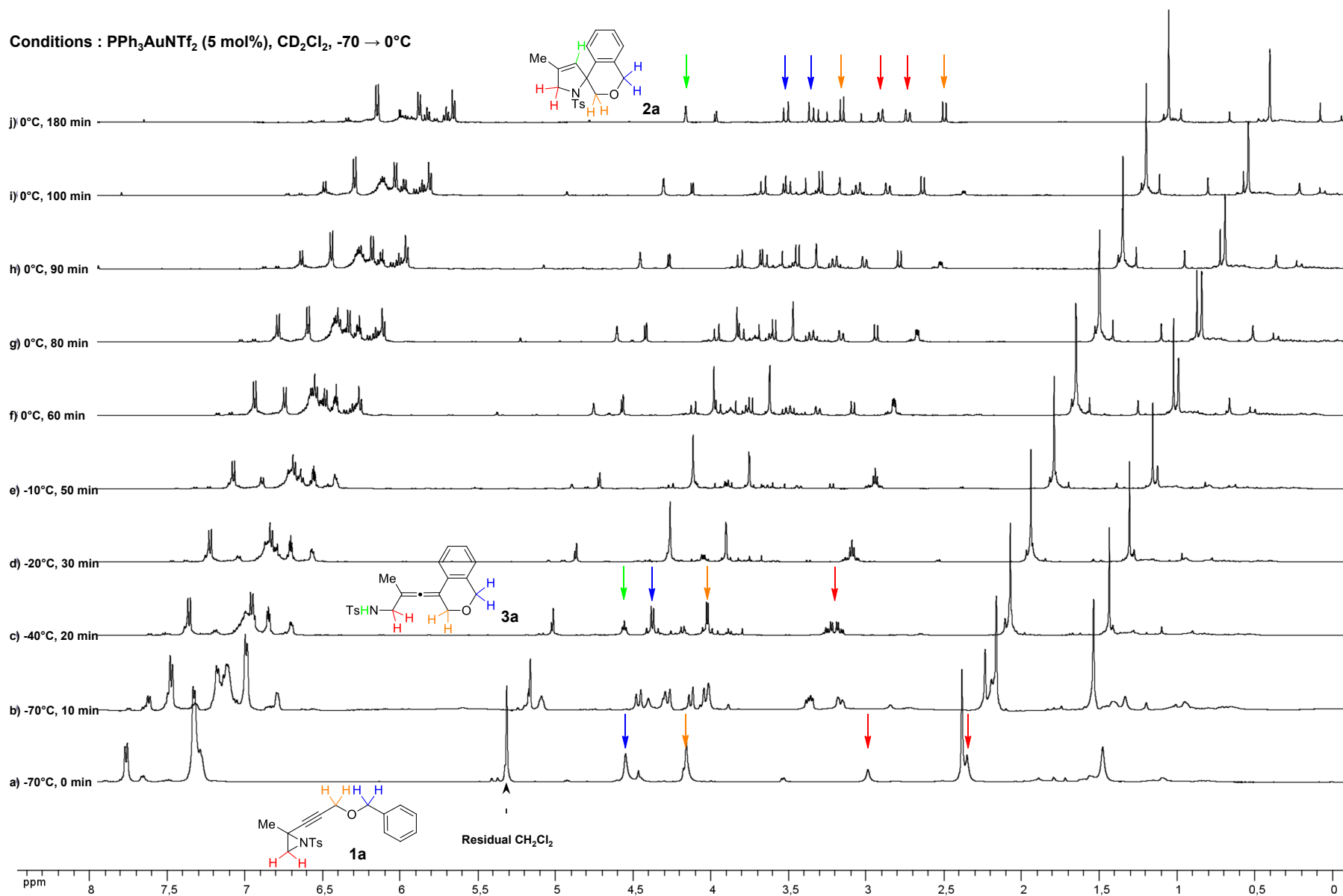
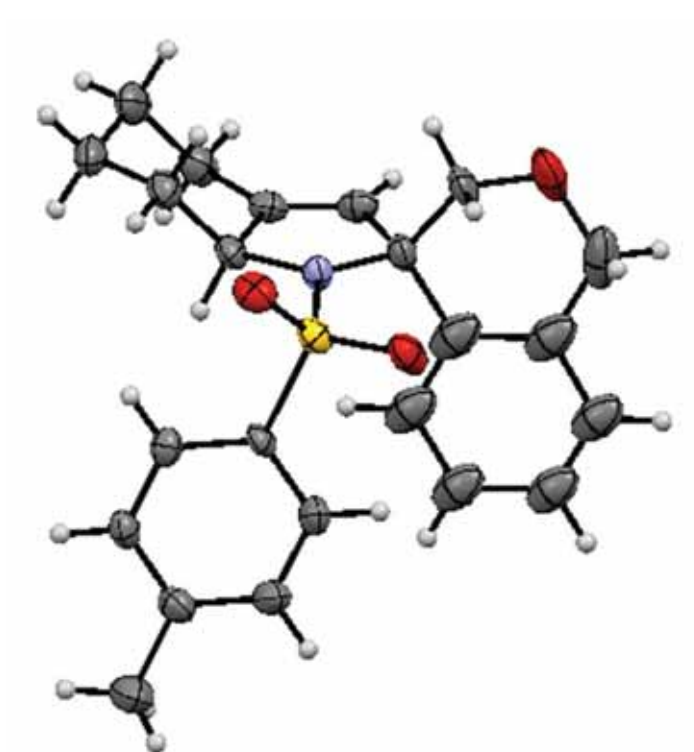


Figure 2: ^1H NMR spectra of the transient intermediates **3a** in gold(I)-catalyzed formation of spiro[isochroman-4,2-pyrrole] **2a** (The ppm scale was calibrated on residual CH_2Cl_2 of the **a** spectrum).





Structure of
2h-maj determined from
X-ray diffraction

Table of crystal data and refinement details for **2h-maj**:

Table 1. Crystal data and structure refinement for ppnk110706.

Identification code	ppnk110706
Empirical formula	C ₂₃ H ₂₅ N O ₃ S
Formula weight	395.50
Temperature	173(2) K
Wavelength	0.71073 Å
Crystal system, space group	Tetragonal, I -4
Unit cell dimensions	a = 22.5705(17) Å alpha = 90 deg. b = 22.5705(17) Å beta = 90 deg. c = 8.0007(6) Å gamma = 90 deg.
Volume	4075.8(5) Å ³
Z, Calculated density	8, 1.289 Mg/m ³
Absorption coefficient	0.182 mm ⁻¹
F(000)	1680
Crystal size	0.50 x 0.08 x 0.06 mm
Theta range for data collection	1.80 to 28.02 deg.
Limiting indices	-26<=h<=29, -29<=k<=22, -10<=l<=9
Reflections collected / unique	25172 / 4936 [R(int) = 0.0911]
Completeness to theta = 28.02	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9891 and 0.9143
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	4936 / 0 / 224
Goodness-of-fit on F ²	1.017
Final R indices [I>2sigma(I)]	R1 = 0.0597, wR2 = 0.1203
R indices (all data)	R1 = 0.1159, wR2 = 0.1413
Absolute structure parameter	0.12(12)
Largest diff. peak and hole	0.808 and -0.515 e.Å ⁻³

Table 2. Atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters (Å² x 10³) for ppnk110706.
U(eq) is defined as one third of the trace of the orthogonalized

Uij tensor.

	x	y	z	U (eq)
C (1)	8220 (2)	4428 (2)	4041 (4)	28 (1)
C (2)	8601 (2)	4978 (2)	4343 (5)	35 (1)
C (3)	9073 (2)	4834 (2)	5659 (5)	41 (1)
C (4)	9468 (2)	4326 (2)	5093 (5)	42 (1)
C (5)	9101 (2)	3763 (2)	4825 (5)	38 (1)
C (6)	8598 (2)	3897 (2)	3642 (5)	30 (1)
C (7)	8441 (2)	3643 (2)	2242 (5)	31 (1)
C (8)	7918 (2)	3930 (2)	1404 (4)	29 (1)
C (9)	8072 (2)	4161 (2)	-355 (4)	39 (1)
C (10)	7488 (2)	3479 (2)	-1778 (6)	60 (1)
C (11)	7193 (2)	3303 (2)	-198 (7)	54 (1)
C (12)	7397 (2)	3506 (2)	1342 (7)	54 (1)
C (13)	7140 (2)	3297 (2)	2810 (7)	54 (1)
C (14)	6660 (2)	2910 (2)	2790 (7)	54 (1)
C (15)	6445 (2)	2736 (2)	1234 (7)	54 (1)
C (16)	6700 (2)	2915 (2)	-194 (7)	54 (1)
C (17)	6828 (2)	4760 (2)	4247 (4)	24 (1)
C (18)	6954 (2)	5027 (2)	5765 (5)	29 (1)
C (19)	6600 (2)	4907 (2)	7139 (4)	30 (1)
C (20)	6119 (2)	4529 (2)	7007 (4)	33 (1)
C (21)	5990 (2)	4278 (2)	5471 (5)	37 (1)
C (22)	6341 (2)	4388 (2)	4091 (5)	32 (1)
C (23)	5733 (2)	4394 (2)	8496 (6)	51 (1)
O (1)	8078 (1)	3686 (1)	-1520 (4)	52 (1)
O (2)	7501 (1)	5499 (1)	2740 (3)	36 (1)
O (3)	6948 (1)	4779 (1)	1021 (3)	38 (1)
S (1)	7277 (1)	4911 (1)	2496 (1)	28 (1)
N (1)	7833 (1)	4455 (1)	2539 (4)	25 (1)

Table 3. Bond lengths [Å] and angles [deg] for ppnk110706.

C (1) -N (1)	1.486 (4)
C (1) -C (6)	1.506 (5)
C (1) -C (2)	1.531 (5)
C (1) -H (1)	1.0000
C (2) -C (3)	1.533 (5)
C (2) -H (2A)	0.9900
C (2) -H (2B)	0.9900
C (3) -C (4)	1.521 (6)
C (3) -H (3A)	0.9900
C (3) -H (3B)	0.9900
C (4) -C (5)	1.532 (6)
C (4) -H (4A)	0.9900
C (4) -H (4B)	0.9900
C (5) -C (6)	1.508 (5)
C (5) -H (5A)	0.9900
C (5) -H (5B)	0.9900
C (6) -C (7)	1.306 (5)
C (7) -C (8)	1.504 (5)
C (7) -H (7)	0.9500
C (8) -N (1)	1.505 (4)
C (8) -C (12)	1.518 (6)

C(8)-C(9)	1.541(5)
C(9)-O(1)	1.420(5)
C(9)-H(9A)	0.9900
C(9)-H(9B)	0.9900
C(10)-O(1)	1.426(5)
C(10)-C(11)	1.482(7)
C(10)-H(10A)	0.9900
C(10)-H(10B)	0.9900
C(11)-C(12)	1.392(7)
C(11)-C(16)	1.417(6)
C(12)-C(13)	1.392(7)
C(13)-C(14)	1.392(6)
C(13)-H(13)	0.9500
C(14)-C(15)	1.393(7)
C(14)-H(14)	0.9500
C(15)-C(16)	1.342(7)
C(15)-H(15)	0.9500
C(16)-H(16)	0.9500
C(17)-C(18)	1.384(5)
C(17)-C(22)	1.390(5)
C(17)-S(1)	1.762(3)
C(18)-C(19)	1.385(5)
C(18)-H(18)	0.9500
C(19)-C(20)	1.385(5)
C(19)-H(19)	0.9500
C(20)-C(21)	1.385(5)
C(20)-C(23)	1.507(5)
C(21)-C(22)	1.381(5)
C(21)-H(21)	0.9500
C(22)-H(22)	0.9500
C(23)-H(23A)	0.9800
C(23)-H(23B)	0.9800
C(23)-H(23C)	0.9800
O(2)-S(1)	1.434(2)
O(3)-S(1)	1.426(3)
S(1)-N(1)	1.626(3)

N(1)-C(1)-C(6)	101.1(3)
N(1)-C(1)-C(2)	115.1(3)
C(6)-C(1)-C(2)	111.2(3)
N(1)-C(1)-H(1)	109.7
C(6)-C(1)-H(1)	109.7
C(2)-C(1)-H(1)	109.7
C(1)-C(2)-C(3)	109.1(3)
C(1)-C(2)-H(2A)	109.9
C(3)-C(2)-H(2A)	109.9
C(1)-C(2)-H(2B)	109.9
C(3)-C(2)-H(2B)	109.9
H(2A)-C(2)-H(2B)	108.3
C(4)-C(3)-C(2)	111.3(3)
C(4)-C(3)-H(3A)	109.4
C(2)-C(3)-H(3A)	109.4
C(4)-C(3)-H(3B)	109.4
C(2)-C(3)-H(3B)	109.4
H(3A)-C(3)-H(3B)	108.0
C(3)-C(4)-C(5)	110.4(3)
C(3)-C(4)-H(4A)	109.6
C(5)-C(4)-H(4A)	109.6
C(3)-C(4)-H(4B)	109.6
C(5)-C(4)-H(4B)	109.6
H(4A)-C(4)-H(4B)	108.1

C (6) -C (5) -C (4)	109.3 (3)
C (6) -C (5) -H (5A)	109.8
C (4) -C (5) -H (5A)	109.8
C (6) -C (5) -H (5B)	109.8
C (4) -C (5) -H (5B)	109.8
H (5A) -C (5) -H (5B)	108.3
C (7) -C (6) -C (1)	112.1 (3)
C (7) -C (6) -C (5)	130.9 (3)
C (1) -C (6) -C (5)	116.9 (3)
C (6) -C (7) -C (8)	114.0 (3)
C (6) -C (7) -H (7)	123.0
C (8) -C (7) -H (7)	123.0
C (7) -C (8) -N (1)	99.8 (3)
C (7) -C (8) -C (12)	110.6 (3)
N (1) -C (8) -C (12)	114.7 (3)
C (7) -C (8) -C (9)	112.1 (3)
N (1) -C (8) -C (9)	108.3 (3)
C (12) -C (8) -C (9)	111.0 (3)
O (1) -C (9) -C (8)	110.2 (3)
O (1) -C (9) -H (9A)	109.6
C (8) -C (9) -H (9A)	109.6
O (1) -C (9) -H (9B)	109.6
C (8) -C (9) -H (9B)	109.6
H (9A) -C (9) -H (9B)	108.1
O (1) -C (10) -C (11)	112.6 (4)
O (1) -C (10) -H (10A)	109.1
C (11) -C (10) -H (10A)	109.1
O (1) -C (10) -H (10B)	109.1
C (11) -C (10) -H (10B)	109.1
H (10A) -C (10) -H (10B)	107.8
C (12) -C (11) -C (16)	117.4 (5)
C (12) -C (11) -C (10)	121.3 (4)
C (16) -C (11) -C (10)	121.3 (5)
C (13) -C (12) -C (11)	119.9 (4)
C (13) -C (12) -C (8)	120.6 (4)
C (11) -C (12) -C (8)	119.5 (5)
C (12) -C (13) -C (14)	121.8 (5)
C (12) -C (13) -H (13)	119.1
C (14) -C (13) -H (13)	119.1
C (13) -C (14) -C (15)	117.3 (5)
C (13) -C (14) -H (14)	121.4
C (15) -C (14) -H (14)	121.4
C (16) -C (15) -C (14)	121.8 (4)
C (16) -C (15) -H (15)	119.1
C (14) -C (15) -H (15)	119.1
C (15) -C (16) -C (11)	121.7 (5)
C (15) -C (16) -H (16)	119.1
C (11) -C (16) -H (16)	119.1
C (18) -C (17) -C (22)	120.2 (3)
C (18) -C (17) -S (1)	119.7 (3)
C (22) -C (17) -S (1)	120.0 (3)
C (17) -C (18) -C (19)	119.5 (3)
C (17) -C (18) -H (18)	120.2
C (19) -C (18) -H (18)	120.2
C (18) -C (19) -C (20)	120.8 (3)
C (18) -C (19) -H (19)	119.6
C (20) -C (19) -H (19)	119.6
C (21) -C (20) -C (19)	119.0 (3)
C (21) -C (20) -C (23)	119.8 (4)
C (19) -C (20) -C (23)	121.2 (3)
C (22) -C (21) -C (20)	121.0 (4)

C (22) - C (21) - H (21)	119.5
C (20) - C (21) - H (21)	119.5
C (21) - C (22) - C (17)	119.4 (3)
C (21) - C (22) - H (22)	120.3
C (17) - C (22) - H (22)	120.3
C (20) - C (23) - H (23A)	109.5
C (20) - C (23) - H (23B)	109.5
H (23A) - C (23) - H (23B)	109.5
C (20) - C (23) - H (23C)	109.5
H (23A) - C (23) - H (23C)	109.5
H (23B) - C (23) - H (23C)	109.5
C (9) - O (1) - C (10)	109.5 (3)
O (3) - S (1) - O (2)	119.40 (17)
O (3) - S (1) - N (1)	106.67 (16)
O (2) - S (1) - N (1)	108.12 (14)
O (3) - S (1) - C (17)	108.59 (15)
O (2) - S (1) - C (17)	105.88 (17)
N (1) - S (1) - C (17)	107.72 (17)
C (1) - N (1) - C (8)	112.4 (2)
C (1) - N (1) - S (1)	119.8 (2)
C (8) - N (1) - S (1)	125.8 (2)

Symmetry transformations used to generate equivalent atoms: