

Supporting Information

Commercial Zinc Oxide: A Facile, Efficient and Eco-friendly Catalyst for the One-pot Three-component Synthesis of Multi-substituted 2-Aminothiophenes Via Gewald Reaction

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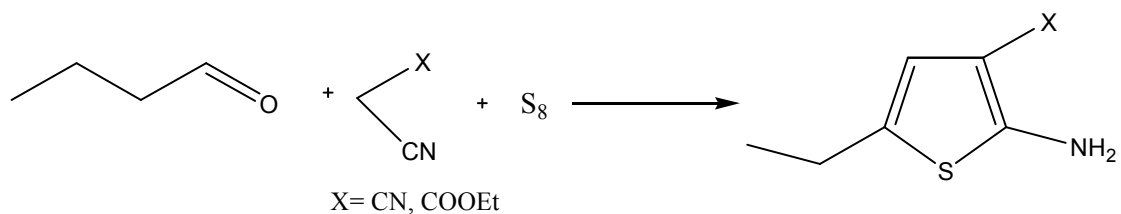
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Table S2. Effect of Reaction Time on the Condensation of Cyclohexanone, Malonodinitrile, and Elemental Sulfur in the Presence of ZnO (5 mol%).

Entry	Time (h)	Yield (%)
6	6	66
7	10	70
8	14	72
9	18	73

Reaction conditions is described below Table 1.

Table S3. Comparison of the Catalytic Activity of ZnO with some Reported Catalysts Used for the Synthesis of 2-Aminothiophene Derivatives



Entry	Catalyst	Mol%	Time (h)	Yield%	Condition	X	Ref.
30	ZnO	5	10	70	Solvent-free	CN	This work
31	Diethylamine	10	10	43	in DMF	CN	12
32	Morpholine	10	11	49	in DMF	CN	12
33	L-Proline	10	10	64	in DMF	CN	12
34	Imidazole	10	10	75	in DMF	CN	12
35	KF-Alumina	0.2 gr	6 min	55	MW	COOEt	15

Copies of NMR Spectra of Multisubstitued 2-Amimothiophenes

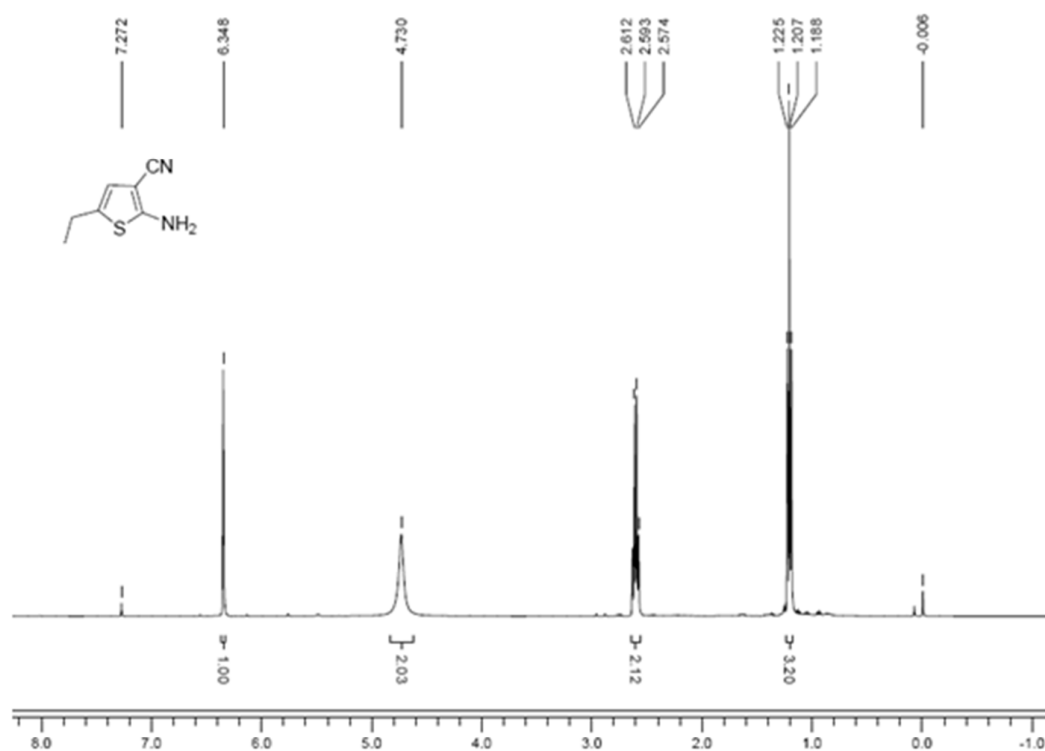


Figure S4: ^1H NMR of 2-amino-5-ethylthiophene-3-carbonitrile.

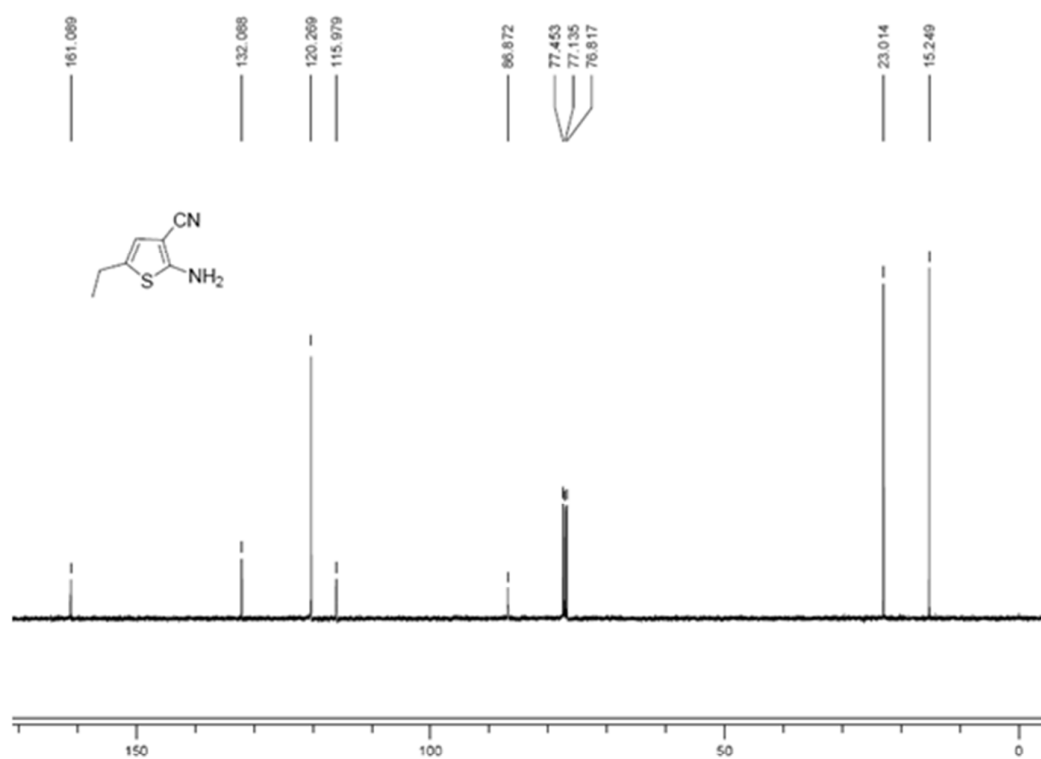


Figure S5: ^{13}C NMR of 2-amino-5-ethylthiophene-3-carbonitrile.

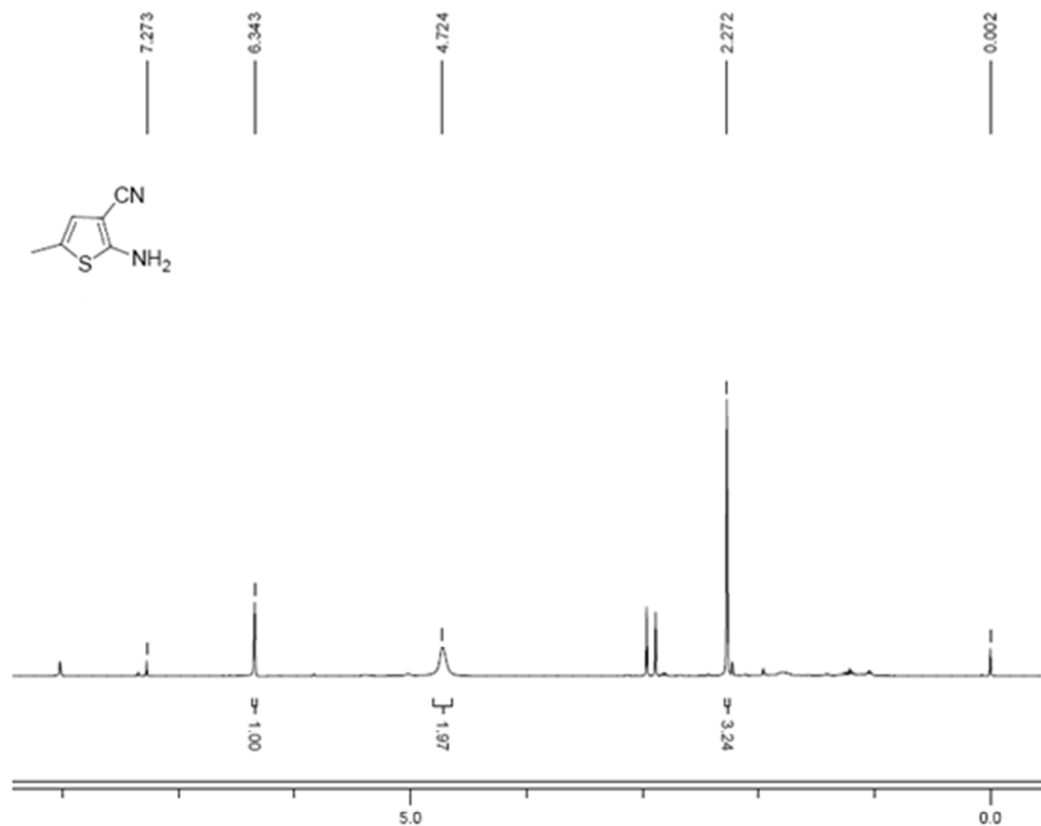


Figure S6: ^1H NMR of 2-amino-5-methylthiophene-3-carbonitrile.

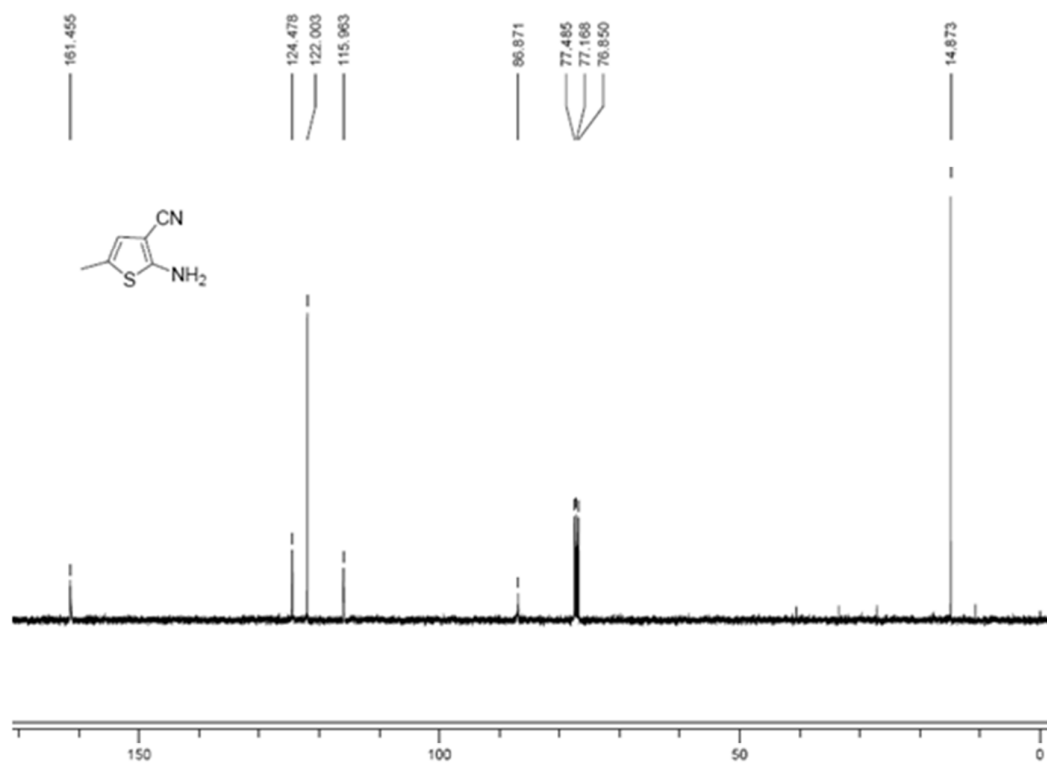


Figure S7: ^{13}C NMR of 2-amino-5-methylthiophene-3-carbonitrile.

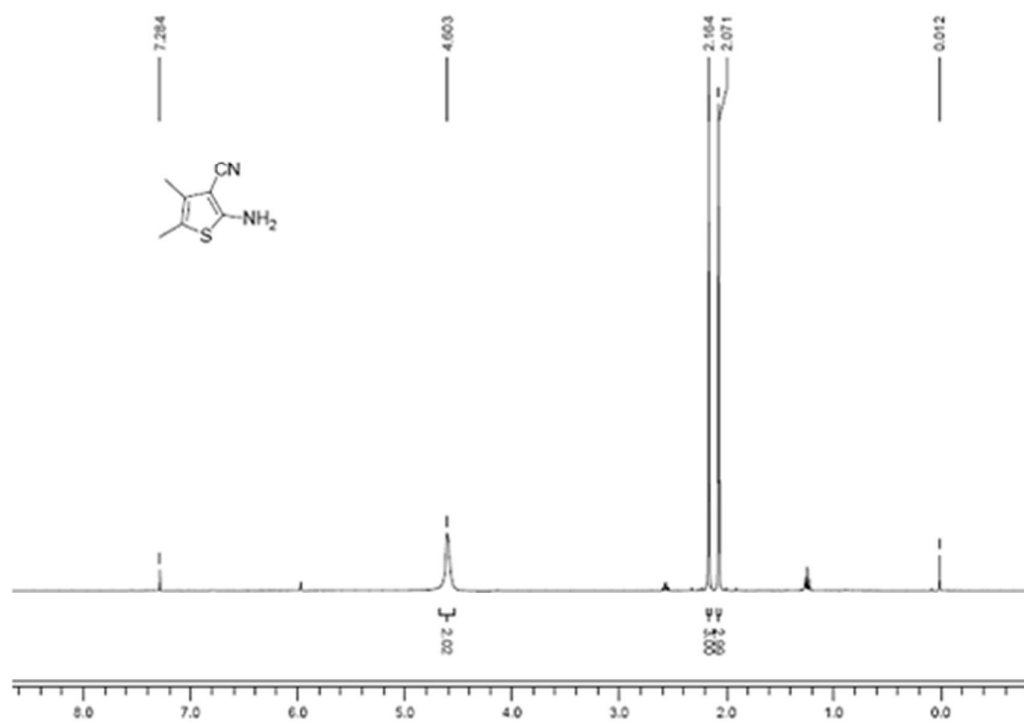


Figure S8: ^1H NMR of 2-amino-4,5-methylthiophene-3-carbonitrile.

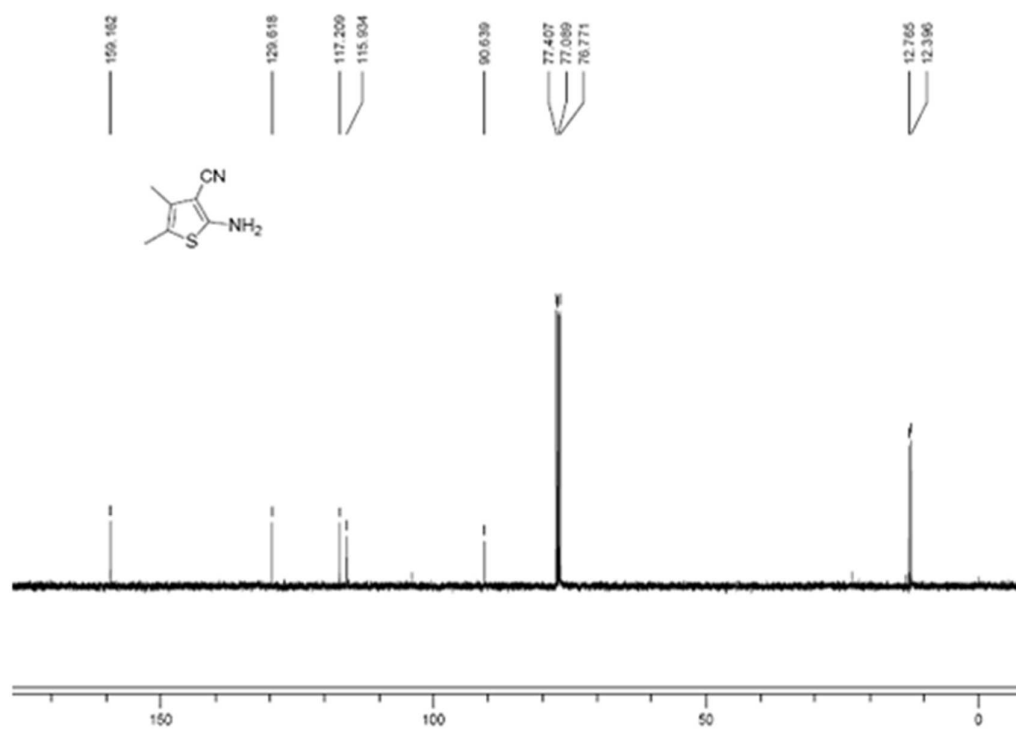


Figure S9: ¹³C NMR of 2-amino-4,5-methylthiophene-3-carbonitrile.

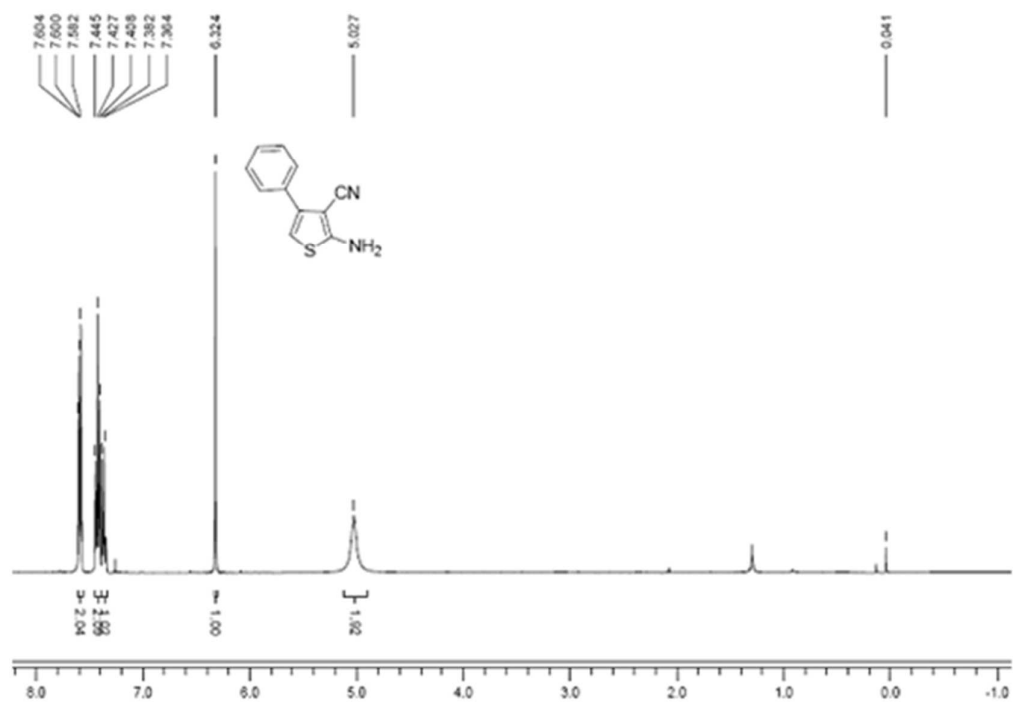


Figure S10: ^1H NMR of 2-amino-4-phenylthiophene-3-carbonitrile.

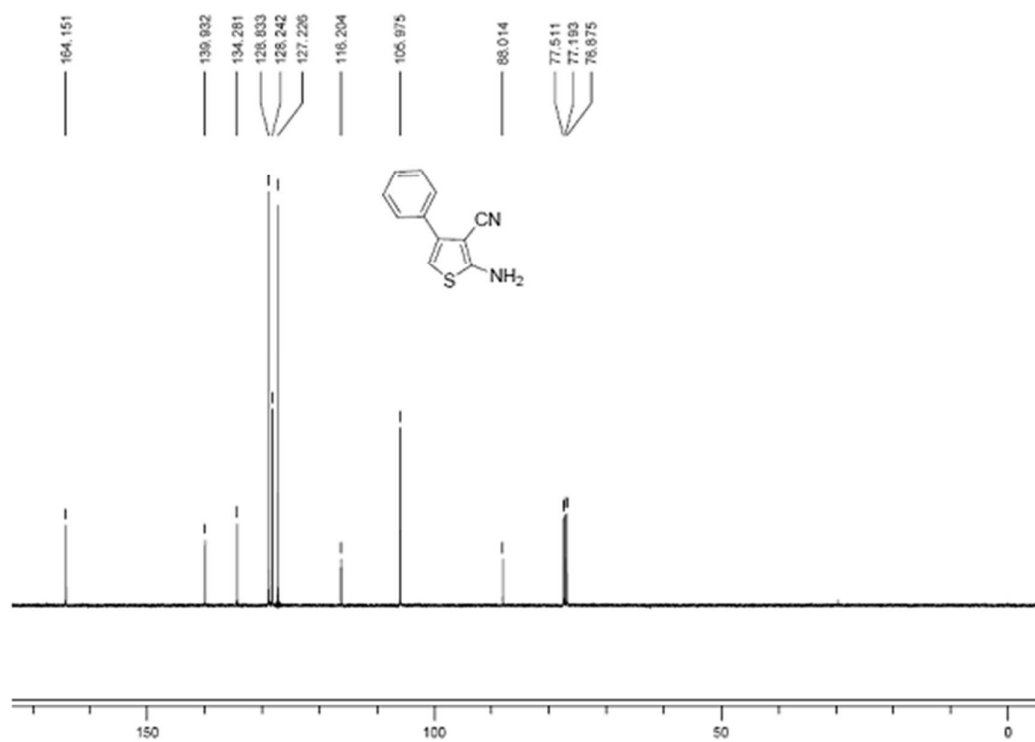


Figure S11: ^{13}C NMR of 2-amino-4-phenylthiophene-3-carbonitrile.

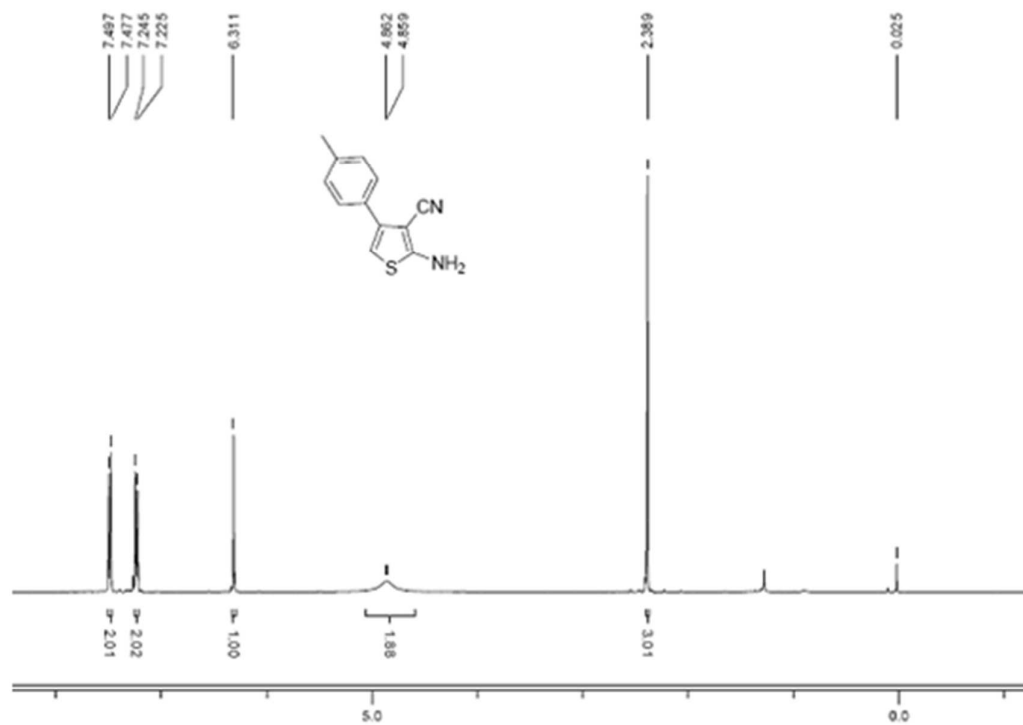


Figure S12: ^1H NMR of 2-amino-4-(4-methylphenyl)thiophene-3-carbonitrile.

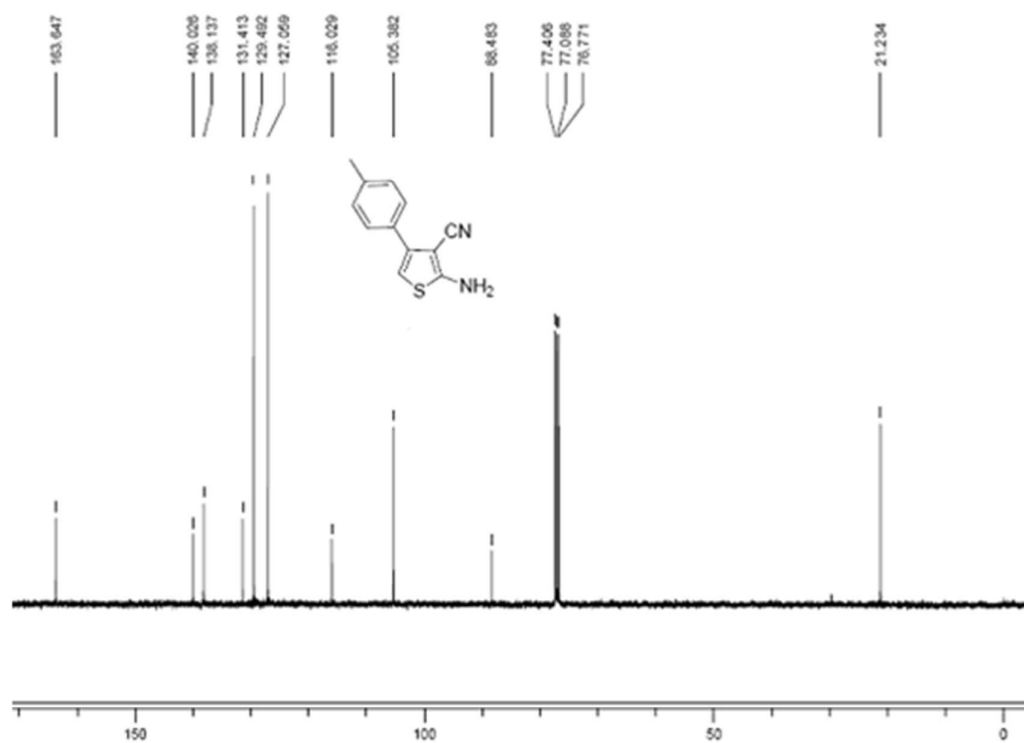


Figure S13: ^{13}C NMR of 2-amino-4-(4-methylphenyl)thiophene-3-carbonitrile.

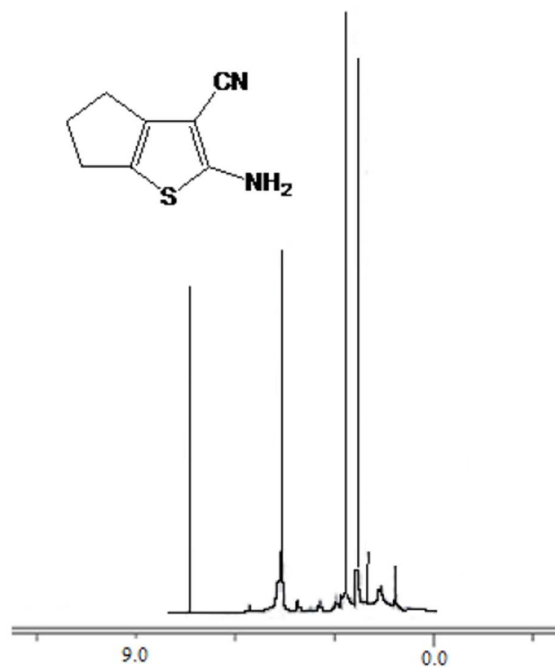
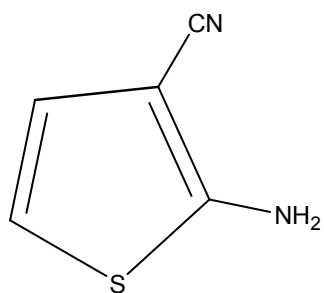
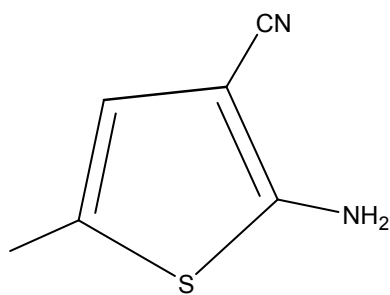


Figure S14: ^1H NMR of 2-amino-4H,5H,6H-cyclopenta[b]thiophene-3-carbonitrile.



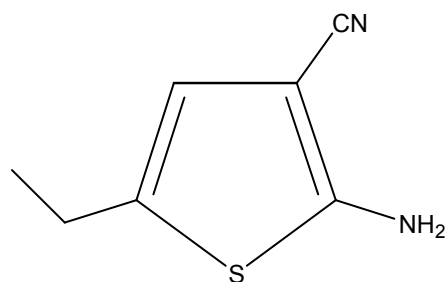
Vibration	NH ₂	=C-H	CN	C=C
$\nu(\text{cm}^{-1})$	3440-3360	3210	2220	1660

Figure S15: FT-IR band assignments of 2-aminothiophene-3-carbonitrile.



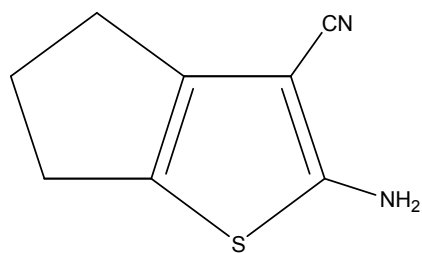
Vibration	NH ₂	=C-H	C-H	CN	C=C
$\nu(\text{cm}^{-1})$	3440-3360	3210	2960	2220	1660

Figure S16: FT-IR band assignments of 2-amino-5-methylthiophene-3-carbonitrile.



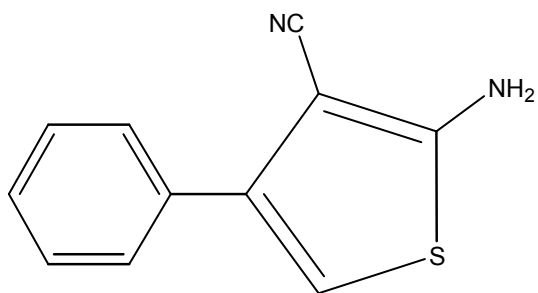
Vibration	NH ₂	=C-H	C-H	CN	C=C	-CH ₂
$\nu(\text{cm}^{-1})$	3440-3360	3210	2960	2220	1660	1465

Figure S17: FT-IR band assignments of 2-amino-5-ethylthiophene-3-carbonitrile.



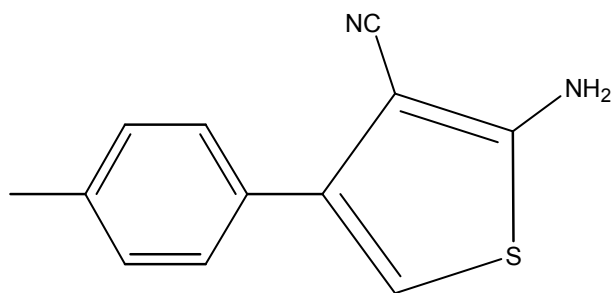
Vibration	NH ₂	C-H	CN	C=C
$\nu(\text{cm}^{-1})$	3440-3340	2840-2920	2220	1630

Figure S18: FT-IR band assignments of 2-amino-4H, 5H, 6H-cyclopenta[b]thiophene-3-carbonitrile.



Vibration	NH₂	=C-H	CN	C=C
$\nu(\text{cm}^{-1})$	3320-3200	3040	2220	1440,1570

Figure S19: FT-IR band assignments of 2-amino-4-phenylthiophene-3-carbonitrile.



Vibration	NH₂	=C-H	CN	C=C	-CH₂	
$\nu(\text{cm}^{-1})$	3320-3200	3040	2220	1440,1560	1375	

Figure S20: FT-IR band assignments of 2-amino-4-(4-methylphenyl)thiophene-3-carbonitrile.

References:

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