#### **Supporting Information for**

# Catalysed Tandem C-N/C-C Bond Formation for the Synthesis of Tricyclic Indoles using Ir(III) Pyrazolyl-1,2,3-Triazolyl Complexes

Chin-Min Wong,<sup>a</sup> Khuong Q. Vuong,<sup>a</sup> Mark R. D. Gatus,<sup>a</sup> Carol Hua,<sup>a</sup> Mohan Bhadbhade<sup>a,b</sup> and Barbara A. Messerle\*, <sup>a</sup>

<sup>a</sup> School of Chemistry, and <sup>b</sup> X-ray Diffraction Laboratory, Mark Wainwright Analytical Centre, The University of New South Wales, Kensington, NSW 2052, Australia.

Telephone: +61-2-9385 4653

Fascimile: +61-2- 9385 6141

Email: b.messerle@unsw.edu.au

### Part A: Synthesis of Ligands

A.1 Synthesis of 4-((1 <i>H</i> -pyrazol-1-yl)methyl)-1-(4-nitrophenyl)-1H-1,2,3-triazole ( <b>2c</b> )	S6
A.2 Synthesis of 4-((3,5-dimethyl-1 <i>H</i> -pyrazol-1-yl)methyl)-1-phenyl-1 <i>H</i> -1,2,3-triazole ( <b>3a</b> )	S7
A.3 Synthesis of 4-((3,5-dimethyl-1 <i>H</i> -pyrazol-1-yl)methyl)-1-(4-(trifluoromethyl)phenyl)-	S8
1 <i>H</i> -1,2,3-triazole ( <b>3b</b> )	
A.4 Synthesis of 4-((3,5-dimethyl-1 <i>H</i> -pyrazol-1-yl)methyl)-1-(4-nitrophenyl)-1 <i>H</i> -1,2,3-	S9
triazole (3c)	
A.5 Synthesis of 4-((3,5-dimethyl-1 <i>H</i> -pyrazol-1-yl)methyl)-1-( <i>p</i> -tolyl)-1 <i>H</i> -1,2,3-triazole	S10
(3d)	
<b>Table S1</b> : Chemical shifts of pyrazolyl C4 (Pz-C4) and 1,2,3-triazolyl C4' (Tz-C4') <sup>13</sup> C	S11
resonances of pyrazolyl-1,2,3-triazolyl bidentate ligands in the <sup>13</sup> C NMR spectra.	
Part B: Synthesis of Ir and Rh Complexes	
B.1 Synthesis of [Ir(1)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (5)	S12
B.2 Synthesis of [Ir(1)Cp*Cl]BPh <sub>4</sub> (5')	S13
B.3 Synthesis of [Ir(2a)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (6a)	S14
B.4 Synthesis of [Ir(2a)Cp*Cl]BPh <sub>4</sub> (6a')	S15
B.5 Synthesis of [Ir(2c)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (6c)	S16
B.6 Synthesis of [Ir(3a)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (7a)	S17
B.7 Synthesis of [Ir( <b>3b</b> )Cp*Cl]BAr <sup>F</sup> <sub>4</sub> ( <b>7b</b> )	S18
B.8 Synthesis of [Ir(3c)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (7c)	S19
B.9 Synthesis of [Ir(3d)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (7d)	S20

B.10 Synthesis of [Ir(4)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (8)	S21
B.11 Synthesis of [Rh(1)Cp*C1]BAr <sup>F</sup> <sub>4</sub> (9)	S22
B.12 Synthesis of [Rh(1)Cp*Cl]BPh <sub>4</sub> (9')	S23
B.13 Synthesis of [Rh(2a)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (10a)	S24
B.14 Synthesis of [Rh(2a)Cp*Cl]BPh <sub>4</sub> (10a')	S25
B.15 Synthesis of [Rh( <b>4</b> )Cp*Cl]BAr <sup>F</sup> <sub>4</sub> ( <b>11</b> )	S26
B.16 Synthesis of $[Rh(2c)(CO)_2)]BAr_4^F$ (14c)	S27
B.17 Synthesis of [Rh( <b>3a</b> )(CO) <sub>2</sub> ]BAr <sup>F</sup> <sub>4</sub> ( <b>15a</b> )	S28
B.18 Synthesis of [Rh( <b>3b</b> )(CO) <sub>2</sub> ]BAr <sup>F</sup> <sub>4</sub> ( <b>15b</b> )	S29
B.19 Synthesis of $[Rh(3c)(CO)_2]BAr^F_4$ (15c)	S30
B.20 Synthesis of $[Rh(3d)(CO)_2]BAr^F_4$ (15d)	S31
<b>Table S2</b> : vCO vibrational frequencies and <sup>13</sup> C chemical shifts of <sup>13</sup> CO of	S32
$[Rh(N-N^2)(CO)_2]BAr^F_4$ (13-16).	
Part C: X-ray Crystallography	
General experimental for X-ray Crystallography	S33
<b>Figure S1</b> : ORTEP depiction of [Ir(1)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (5) at 40% thermal ellipsoids for the	S34
non-hydrogen atoms.	
Figure S2: ORTEP depiction of [Ir(2a)Cp*Cl]BPh <sub>4</sub> (6a') at 40% thermal ellipsoids for the	S34
non-hydrogen atoms.	
Figure S3: ORTEP depiction of [Ir(4)Cp*Cl]BPh <sub>4</sub> (8) at 40% thermal ellipsoids for the non-	S35
hydrogen atoms.	
Figure S4: ORTEP depiction of [Rh(1)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (9) at 40% thermal ellipsoids for the	S35

non-hydrogen atoms.

Figure S5: ORTEP depiction of [Rh(2a)Cp*Cl]BAr <sup>F</sup> <sub>4</sub> (10a) at 40% thermal ellipsoids for	S36
the non-hydrogen atoms.	
Figure S6: ORTEP depiction of [Rh(2a)Cp*Cl]BPh <sub>4</sub> (10a') at 40% thermal ellipsoids for	S36
the non-hydrogen atoms.	
Figure S7: ORTEP depiction of [Rh(4)Cp*Cl]BPh <sub>4</sub> (11) at 40% thermal ellipsoids for the	S37
non-hydrogen atoms.	
Table S3: Crystal structural data for the single crystal X-ray structures of Iridium complexes	S38
8, 6b, 7b, 6a' and 5.	
Table S4: Crystal structural data for the single crystal X-ray structures of Rhodium	S39
complexes 11, 10a, 10a', 9 and 15c.	
Part D: Synthesis of 2-(hydroxyalk-1-ynyl)aniline substrates	S40
D.1 Synthesis of 2-(6-hydroxy-1-hexyn-1-yl)aniline, 17S	S41
D.2 Synthesis of 2-(5- hydroxy-1-pentyn-1-yl)aniline, 18S	S41
D.3 Synthesis of 2-(6-hydroxyhept-1-ynyl)aniline, 19S	S41
D.4 Synthesis of 2-(6 Hydroxy-6-methylhex-1-ynyl)aniline, <b>20S</b>	S42
Part E: General Catalytic Procedure, Time Course Profiles and NMR of	S43
Intermediates and Products	
E.1 General Procedure for the Tandem Hydroamination/C-C Bond Formation Reactions	S43
E.2 Time Course Profiles for Selected Tandem Hydroamination/C-C Bond Formation	S44
Reactions	
<b>Figure S8</b> : Time course profile for [Ir(3c)(Cp*)Cl]BAr <sup>F</sup> <sub>4</sub> (7c) catalysed one-pot tandem C-N	S44

and C-C bond formation in the synthesis of 1,2,3,4-tetrahydrocarbazole (17P) from	
2-(6-Hydroxyhex-1-ynyl)aniline (17S)	
Figure S9: Time course profile for [Ir(4)(Cp*)Cl]BAr <sup>F</sup> <sub>4</sub> (8) catalysed one-pot tandem C-N	S45
and C-C bond formation in the synthesis of 1,2,3,4-tetrahydrocarbazole (17P) from	
2-(6-Hydroxyhex-1-ynyl)aniline (17S)	
E.3 Typical Isolation Procedure for 2-(Hydroxyalkyl)indole intermediates <b>17I-20I</b>	S45
E.4 2-(4-Hydroxybut-1-yl)indole, 17I	S46
E.5 2-(3-Hydroxyprop-1-yl)indole, <b>18I</b>	S46
E.5' 2-[(Z)-(Dihydrofuran-2(3H)-yliden)methyl]benzenamine, <b>I1</b> (from <b>18S</b> )	S46
Figure S10: <sup>1</sup> H NMR stacked-plot showing the formation and disappearance of I1 (from	S47
18S) in the catalysed cyclisation of 2-(5- hydroxy-1-pentyn-1-yl)aniline, 18S using	
[Ir(4)Cp*Cl]BAr $_4$ (8) in toluene- $d_8$ at 60 °C.	
E.6 2-(5-Hydroxypent-1-yl)indole, 19I	S48
E.7 2-(4-Hydroxy-4-methylbut-1-yl)indole, <b>20I</b>	S48
E.8 NMR Data of the Final Products	S48
E.9 1,2,3,4-Tetrahydrocarbazole, <b>17P</b>	S49
E.10 1,2,3,4-Tetrahydrocyclopent[b] indole, <b>18P</b>	S49
E.11 5,6,7,8,9,10-Hexahydrocyclohept[ <i>b</i> ]indole, <b>19P</b>	S49
E.12 4-Methyl-1,2,3,4-tetrahydrocarbazole, <b>20P</b>	S50
References	S50

#### Part A: Synthesis of Ligands

#### A.1 Synthesis of 4-((1*H*-pyrazol-1-yl)methyl)-1-(4-nitrophenyl)-1*H*-1,2,3-triazole (2c)

1-(Prop-2-yn-1-yl)-1*H*-pyrazole (96 mg, 0.90 mmol) and 1-azido-4-nitrobenzene (148 mg, 0.902 mmol) were added into a deoxygenated mixture of 2-propanol and water (2:1 (v/v), 8 mL). Sodium L-ascorbate (39 mg, 0.20 mmol, 20 mol%) was added and the reaction stirred for 5 mins prior to the addition of CuSO<sub>4</sub>.5H<sub>2</sub>O (7.2 mg, 0.029 mmol, 3 mol%). The reaction mixture was stirred overnight at room temperature under nitrogen, during which time a dark brown precipitate formed. The 2-propanol was removed *in vacuo* and the residue was filtered and washed with a saturated aqueous solution of Na<sub>2</sub>EDTA until the filtrate became colourless. The crude product was then dissolved in dichloromethane, dried with anhydrous magnesium sulphate and filtered through celite. The solvent was removed *in vacuo* to yield ligand 2c as a yellow solid (206 mg, 85 %). m.p. 129-132 °C (decomposed). HR-MS (ESI<sup>+</sup>, MeOH): *m/z* (%): 293.0833 (100 %) [M+Na]<sup>+</sup> (Calculated [M+Na]<sup>+</sup> = 293.0757) amu.Elemental Analysis: Found: C, 52.80; H, 3.71 and N, 30.07; Calculated for C<sub>12</sub>H<sub>10</sub>N<sub>6</sub>O<sub>2</sub>.0.25H<sub>2</sub>O: C, 52.46; H, 3.85 and N, 30.59 %.

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  8.40 (d, <sup>3</sup> $J_{\text{H-H}}$  = 9.1 Hz, 2H, *m*-C**H** of **Ph**NO<sub>2</sub>), 8.07 (s, 1H, **H5**'), 7.94 (d, <sup>3</sup> $J_{\text{H-H}}$  = 9.1 Hz, 2H, *o*-C**H** of **Ph**NO<sub>2</sub>), 7.59 (d, <sup>3</sup> $J_{\text{H4-H5}}$  = 2.1 Hz, 1H, **H5**), 7.57 (d, <sup>3</sup> $J_{\text{H3-H4}}$  = 2.1 Hz, 1H, **H3**), 6.31 (t, <sup>3</sup> $J_{\text{H-H}}$  = 2.1 Hz, 1H, **H4**), 5.56 (s, 2H, C**H**<sub>2</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  147.5 (C<sub>q</sub> of **Ph**NO<sub>2</sub>), 145.5 (C<sub>q</sub> of Triaz), 141.1 (*ipso-*C to NO<sub>2</sub> of **Ph**NO<sub>2</sub>), 140.5 (C3), 129.9 (C5), 125.7 (*m*-CH of **Ph**NO<sub>2</sub>), 120.9 (C5'), 120.7 (*o*-CH of **Ph**NO<sub>2</sub>), 106.5 (C4), 47.4 (CH<sub>2</sub>) ppm.

#### A.2 Synthesis of 4-((3,5-dimethyl-1*H*-pyrazol-1-yl)methyl)-1-phenyl-1*H*-1,2,3-triazole (3a)

3,5-Dimethyl-1-(prop-2-yn-1-yl)-1*H*-pyrazole (566 mg, 4.22 mmol) and azidobenzene (517 mg, 4.34 mmol) were added into a deoxygenated mixture of 2-propanol and water (2:1 (v/v),12 mL). Sodium L-ascorbate (171 mg, 0.862 mmol, 20 mol %) was added and the reaction mixture was stirred for 5 mins prior to the addition of CuSO<sub>4</sub>.5H<sub>2</sub>O (34 mg, 0.13 mmol, 3 mol%). The reaction mixture was then stirred for 3 days at room temperature under nitrogen, during which time a pale yellow precipitate formed. The 2-propanol was removed *in vacuo* and the residue was filtered and washed with a saturated aqueous solution of Na<sub>2</sub>EDTA until the filtrate became colourless. The crude product was then dissolved in dichloromethane, dried with anhydrous magnesium sulphate and filtered through celite<sup>®</sup>. The solvent was removed *in vacuo* to yield ligand 3a as a pale yellow solid (812 mg, 76 %). m.p. 101-113 °C.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 276.1667 (100 %) [M + Na]<sup>+</sup> (Calculated [M + Na]<sup>+</sup> = 276.1220) amu.

Elemental Analysis: Found: C, 65.37; H, 5.93 and N, 27.46; Calculated for  $C_{14}H_{15}N_5.0.25H_2O$ : C, 65.22; H, 6.06 and N, 27.17 %.

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.87 (s, 1H, **H5'**), 7.69 (d,  ${}^{3}J_{\text{H-H}}$  = 7.8 Hz, 2H, *o*-C**H** of **Ph**), 7.50 (t,  ${}^{3}J_{\text{H-H}}$  = 7.8 Hz, 2H, *m*-C**H** of **Ph**), 7.42 (t,  ${}^{3}J_{\text{H-H}}$  = 7.8 Hz, 1H, *p*-C**H** of **Ph**), 5.82 (s, 1H, **H4**), 5.38 (s, 2H, C**H**<sub>2</sub>), 2.34 (s, 3H, C5-C**H**<sub>3</sub>), 2.23 (s, 3H, C3-C**H**<sub>3</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  148.3 (C3), 145.3 (C<sub>q</sub> of Triaz), 137.1 (C<sub>q</sub> of **Ph**), 139.6 (C5), 129.9 (*m*-CH of **Ph**), 129.0 (*p*-CH of **Ph**), 120.7 (C5'), 120.5 (*o*-CH of **Ph**), 105.8 (C4), 44.6 (CH<sub>2</sub>), 13.7 (C3-CH<sub>3</sub>), 11.3 (C5-CH<sub>3</sub>) ppm.

## A.3 Synthesis of 4-((3,5-dimethyl-1*H*-pyrazol-1-yl)methyl)-1-(4-(trifluoromethyl)phenyl)-1*H*-1,2,3-triazole (3b)

3,5-Dimethyl-1-(prop-2-yn-1-yl)-1*H*-pyrazole (320 mg, 2.39 mg) and 1-azido-4-(trifluoromethyl)benzene (394 mg, 2.11 mmol) were added into a deoxygenated mixture of 2-propanol and water (2:1 (v/v), 12 mL). Sodium L-ascorbate (104 mg, 5.27 mmol, 20 mol%) was added and the reaction stirred for 5 mins prior to the addition of CuSO<sub>4</sub>.5H<sub>2</sub>O (22 mg, 0.14 mmol, 4 mol%). The reaction mixture was stirred overnight at room temperature under nitrogen, during which time a pale yellow precipitate formed. The 2-propanol was removed *in vacuo* and the residue was filtered and washed with a saturated aqueous solution of Na<sub>2</sub>EDTA until the filtrate became colourless. The crude product was then dissolved in dichloromethane, dried with anhydrous magnesium sulphate and filtered through celite<sup>®</sup>. The solvent was removed *in vacuo* to yield ligand **6** as a yellow solid (615 mg, 91 %). m.p. 134-136 °C.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 322.2500 (100 %) [M+H]<sup>+</sup> (Calculated [M+H]<sup>+</sup> = 322.1274) amu. Elemental Analysis: Found: C, 55.85; H, 4.62 and N, 21.46; Calculated for  $C_{15}H_{14}F_3N_5$ : C, 56.07; H, 4.39 and N, 21.80 %.

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.94 (s, 1H, **H5**'), 8.39 (d, <sup>2</sup> $J_{\text{H-H}}$  = 8.6 Hz, 2H, *m*-C**H** of **Ph**CF<sub>3</sub>), 7.93 (d, <sup>2</sup> $J_{\text{H-H}}$  = 8.6 Hz, 2H, *o*-C**H** of **Ph**CF<sub>3</sub>), 5.83 (s, 1H, **H4**), 5.38 (s, 2H, C**H**<sub>2</sub>), 2.35 (s, 3H, C5-C**H**<sub>3</sub>), 2.23 (s, 3H, C3-C**H**<sub>3</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz): δ 148.5 (C3), 145.8 (C<sub>q</sub> of Triaz), 139.7 (C5), 139.4 (C<sub>q</sub> of **Ph**CF<sub>3</sub>), 130.9 (q,  ${}^{2}J_{\text{C-F}} = 32.9$ , *ipso*-C to CF<sub>3</sub> of **Ph**CF<sub>3</sub>), 127.2 (*m*-CH of **Ph**CF<sub>3</sub>), 123.6 (q,  ${}^{1}J_{\text{C-F}} = 269.0 \text{ Hz}$ , CF<sub>3</sub>), 120.7 (C5'), 120.6 (*o*-CH of **Ph**CF<sub>3</sub>), 105.8 (C4), 44.4 (CH<sub>2</sub>), 13.7 (C3-CH<sub>3</sub>), 11.3 (C5-CH<sub>3</sub>) ppm.

## A.4 Synthesis of 4-((3,5-dimethyl-1*H*-pyrazol-1-yl)methyl)-1-(4-nitrophenyl)-1*H*-1,2,3-triazole (3c)

3,5-Dimethyl-1-(prop-2-yn-1-yl)-1*H*-pyrazole (230 mg, 1.71 mol) were added into a deoxygenated mixture of 2-propanol and water (2:1 (v/v), 12 mL). Sodium L-ascorbate (79 mg, 0.40 mmol, 20 mol %) was added and the reaction stirred for 5 mins prior to the addition of CuSO<sub>4</sub>.5H<sub>2</sub>O (14 mg, 0.086 mmol, 3 mol%). The reaction mixture was stirred overnight at room temperature under nitrogen, during which time a pale yellow precipitate formed. The 2-propanol was removed *in vacuo* and the residue was filtered and washed with a saturated aqueous solution of Na<sub>2</sub>EDTA until the filtrate became colourless. The crude product was then dissolved in dichloromethane, dried with anhydrous magnesium sulphate and filtered through celite<sup>®</sup>. The solvent was removed *in vacuo* to yield ligand 3c as a pale yellow solid (357 mg, 70 %). m.p. 155-157 °C.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 299.1667 (100 %) [M+H]<sup>+</sup> (Calculated [M+H]<sup>+</sup> = 299.1251) amu. Elemental Analysis: Found: C, 55.46; H, 4.78 and N, 27.49; Calculated for  $C_{14}H_{14}N_6O_2.0.25H_2O$ : C, 55.53; H, 4.83 and N, 27.75 %.

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz):  $\delta$  8.39 (d, <sup>3</sup> $J_{\text{H-H}}$  = 8.6 Hz, 2H, m-C**H** of **Ph**NO<sub>2</sub>), 8.00 (s, 1H, **H5'**), 7.93 (d, <sup>3</sup> $J_{\text{H-H}}$  = 8.6 Hz, 2H, o-C**H** of **Ph**NO<sub>2</sub>), 5.84 (s, 1H, **H4**), 5.39 (s, 2H, C**H**<sub>2</sub>), 2.35 (s, 3H, C5-C**H**<sub>3</sub>), 2.23 (s, 3H, C3-C**H**<sub>3</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 150 MHz):  $\delta$  148.6 (C3), 147.4 (C<sub>q</sub> of **Ph**), 146.2 (C<sub>q</sub> of Triaz), 141.2 (*ipso*-C to NO<sub>2</sub> of **Ph**NO<sub>2</sub>), 139.7 (C5), 125.7 (*m*-CH of **Ph**NO<sub>2</sub>), 120.7 (C5'), 120.6 (*o*-CH of **Ph**NO<sub>2</sub>), 105.9 (C4), 44.3 (CH<sub>2</sub>), 13.7 (C3-CH<sub>3</sub>), 11.3 (C5-CH<sub>3</sub>) ppm.

#### A.5 Synthesis of 4-((3,5-dimethyl-1*H*-pyrazol-1-yl)methyl)-1-(p-tolyl)-1*H*-1,2,3-triazole (3d)

3,5-Dimethyl-1-(prop-2-yn-1-yl)-1*H*-pyrazole (249 mg, 1.86 mmol) and 1-azido-4-methylbenzene (249 mg, 1.87 mmol) were added into a deoxygenated mixture of 2-propanol and water (2:1 (v/v), 12 mL). Sodium L-ascorbate (75 mg, 0.38 mmol, 20 mol %) was added and the reaction stirred for 5 mins prior to the addition of CuSO<sub>4</sub>.5H<sub>2</sub>O (15 mg, 0.060 mmol, 3 mol%). The reaction mixture was stirred for 4 days at room temperature under nitrogen, during which time a pale yellow precipitate formed. The 2-propanol was removed *in vacuo* and the residue was filtered and washed with a saturated aqueous solution of Na<sub>2</sub>EDTA until the filtrate became colourless. The crude product was then dissolved in dichloromethane, dried with anhydrous magnesium sulphate and filtered through celite<sup>®</sup>. The solvent was removed *in vacuo* to yield ligand 3d as a white solid (372 mg, 75 %). m.p. 142-143 °C. HR-MS (ESI<sup>+</sup>, MeOH): *m/z* (%): 290.2500 (100 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 290.1376) amu.

Elemental Analysis: Found: C, 67.55; H, 6.46 and N, 26.24; Calculated for  $C_{13}H_{13}N_5$ : C, 67.39; H, 6.41 and N, 26.20 %.

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.82 (s, 1H, **H5**'), 7.56 (d,  ${}^{3}J_{\text{H-H}}$  = 8.3 Hz, 2H, *o*-C**H** of **Ph**CH<sub>3</sub>), 7.28 (d,  ${}^{3}J_{\text{H-H}}$  = 8.3 Hz, 2H, *m*-C**H** of **Ph**CH<sub>3</sub>), 5.82 (s, 1H, **H4**), 5.37 (s, 2H, C**H**<sub>2</sub>), 2.40 (s, 3H, C**H**<sub>3</sub> of **Ph**CH<sub>3</sub>), 2.34 (s, 3H, C5-C**H**<sub>3</sub>), 2.23 (s, 3H, C3-C**H**<sub>3</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  148.2 (C3), 145.1 (C<sub>q</sub> of Triaz), 139.6 (C5), 139.1 (C<sub>q</sub> of PhCH<sub>3</sub>), 134.8 (*ipso*-C to CH<sub>3</sub> of PhCH<sub>3</sub>), 130.3 (*m*-CH of PhCH<sub>3</sub>), 120.7 (C5'), 120.6 (*o*-CH of PhCH<sub>3</sub>), 105.7 (C4), 44.6 (CH<sub>2</sub>), 21.2 (CH<sub>3</sub> of PhCH<sub>3</sub>), 13.7 (C3-CH<sub>3</sub>), 11.3 (C5-CH<sub>3</sub>) ppm.

**Table S1**: Chemical shifts of pyrazolyl C4 (Pz-C4) and 1,2,3-triazolyl C4' (Tz-C4') <sup>13</sup>C resonances of Pyrazolyl-*1,2,3*-Triazolyl bidentate ligands in the <sup>13</sup>C NMR spectra.<sup>a</sup>

Ligands	$\delta$ <sup>13</sup> C	$\delta$ <sup>13</sup> C
	(Pz-C4)	(Tz-C4')
1 <sup>b</sup> C4 N N=N N=N N N=N N N N N N N N N N N N	106.2	144.4
2ab N N N N N N N N N N N N N N N N N N N	106.4	144.5
$\begin{array}{c c} \mathbf{2b} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$	106.5	145.1
$2c \underset{C4}{\overset{N}{\smile}} \underset{N}{\overset{N=N}{\smile}} \underset{N}{\overset{N=N}{\smile}} -NO_2$	106.5	145.5
3d N=N N=N C4' N-CH <sub>3</sub>	105.7	145.1
3a N N N N N N N N N N N N N N N N N N N	105.8	145.3
$\begin{array}{c c} 3b & & \\  & & $	105.8	145.8
$\begin{array}{c c} 3c & & & \\ & $	105.9	146.2

<sup>&</sup>lt;sup>a</sup> The <sup>1</sup>H NMR spectra were acquired in CDCl<sub>3</sub>. <sup>b</sup> From reference <sup>1</sup>.

#### Part B: Synthesis of Ir and Rh Complexes

#### B.1 Synthesis of [Ir(1)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (5)

[Ir(1)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (5) was synthesised following the method used for the synthesis of rhodium complex 10a from [IrCp\*Cl<sub>2</sub>]<sub>2</sub> (150 mg, 1.88 x 10<sup>-4</sup> mol), ligand 1 (84.8 mg, 3.54 x 10<sup>-4</sup> mol) and NaBAr<sup>F</sup><sub>4</sub> (367 mg, 4.14 x 10<sup>-4</sup> mol) to yield complex 5 as a yellow solid (473 mg, 86%). m.p. 87-89 °C (decomposed).

Elemental Analysis: Found: C, 44.52; H, 2.99 and N, 4.64; Calculated for  $C_{55}H_{40}BF_{24}IrN_5$ : C, 45.08; H, 2.75 and N, 4.78%.

ESI-MS (ESI<sup>+</sup>, MeOH): 602.00 ([M]<sup>+</sup>, 100%) amu.

<sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>, 500 MHz):  $\delta$  7.78 (d, <sup>3</sup> $J_{\text{H4-H5}}$  = 2.1 Hz, 1H, **H5**), 7.75 (s, 1H, **H5'**), 7.72 (br s, 8H, o-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.68 (d, <sup>3</sup> $J_{\text{H4-H3}}$  = 2.4 Hz, 1H, **H3**), 7.56 (br s, 4H, p-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.44-7.43 (m, 3H, p & m-C**H** of **Ph**), 7.35-7.33 (m, 2H, o-C**H** of **Ph**), 6.53 (t, <sup>3</sup> $J_{\text{H-H}}$  = 2.4 Hz, 1H, **H4**), 5.70-5.61 (m, 2H, C**H**<sub>2</sub><sup>b</sup>), 5.55 (d, <sup>2</sup> $J_{\text{H-H}}$  = 18.0 Hz, 1H, C**H**<sub>A</sub> of CH<sub>A</sub>H<sub>B</sub><sup>a</sup>), 5.02 (d, <sup>2</sup> $J_{\text{H-H}}$  = 18.0 Hz, 1H, C**H**<sub>B</sub> of CH<sub>A</sub>H<sub>B</sub><sup>a</sup>), 1.66 (s, 15 H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CD<sub>2</sub>Cl<sub>2</sub>, 150 MHz): δ 162.0 (q,  ${}^{1}J_{B-C} = 50.0$  Hz, *ipso-*C to B, BAr<sup>F</sup><sub>4</sub>), 145.3 (C5), 140.0 (C<sub>q</sub> of Triaz), 134.2 (*o*-CH to B, BAr<sup>F</sup><sub>4</sub>), 134.8 (C3), 132.7 (C<sub>q</sub> of **Ph**), 130.2 (*p*-CH of **Ph**), 129.9 (*m*-CH of **Ph**), 129.4 (*o*-CH of **Ph**), 129.0 (app. q,  ${}^{2}J_{F-C} = 35.0$  Hz, CCF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 125.0 (q,  ${}^{1}J_{F-C} = 271.0$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.2 (C5'), 117.9 (br s, *p*-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.1 (C4), 97.6 (d,  ${}^{2}J_{Ir}$  C = 8.0 Hz, C<sub>q</sub> of Cp\*), 56.7 (CH<sub>2</sub><sup>b</sup>), 45.4 (CH<sub>2</sub><sup>a</sup>), 9.5 (s, CH<sub>3</sub> of Cp\*) ppm.

#### B.2 Synthesis of [Ir(1)Cp\*Cl]BPh<sub>4</sub> (5')

5' as a pale yellow solid (93 mg, 84%). m.p. 247-249 °C.

Elemental Analysis: Found: C, 61.49; H, 5.11 and N, 7.34; Calculated for C<sub>47</sub>H<sub>48</sub>BIrN<sub>5</sub>: C, 61.27; H, 5.25 and N, 7.60%.

ESI-MS (ESI<sup>+</sup>, MeOH): 602.06 ([M]<sup>+</sup>, 100%) amu.

<sup>1</sup>H NMR (Acetone- $d_6$ , 400 MHz): δ 8.36 (s, 1H, **H5**'). 8.05 (d,  ${}^3J_{\text{H4-H5}} = 2.8$  Hz, 1H, **H5**), 7.86 (d,  ${}^3J_{\text{H4-H3}} = 2.4$  Hz, 1H, **H3**), 7.45-7.41 (m, 5H, **Ph**), 7.36-7.32 (m, 8H, *o*-C**H** of B**Ph**<sub>4</sub>), 6.92 (t,  ${}^3J_{\text{H-H}} = 7.2$  Hz, 8H, *m*-C**H** of B**Ph**<sub>4</sub>), 6.77 (t,  ${}^3J_{\text{H-H}} = 7.2$  Hz, 4H, *p*-C**H** of B**Ph**<sub>4</sub>), 6.60 (t,  ${}^3J_{\text{H-H}} = 2.4$  Hz, 1H, **H4**), 5.99 (d,  ${}^2J_{\text{H-H}} = 16.4$  Hz, 1H, C**H**<sub>A</sub> of CH<sub>A</sub>H<sub>B</sub><sup>a</sup>), 5.85 (s, 2H, C**H**<sub>2</sub><sup>b</sup>), 5.19 (d,  ${}^2J_{\text{H-H}} = 16.4$  Hz, 1H, C**H**<sub>B</sub> of CH<sub>A</sub>H<sub>B</sub><sup>a</sup>), 1.69 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (Acetone- $d_6$ , 150 MHz): δ 165.0 (q,  ${}^{1}J_{B-C} = 50.0$ , ipso-C to B, BPh<sub>4</sub>), 145.8 (C3), 137.09 (o-CH of BPh<sub>4</sub>), 135.6 (C5), 135.4 (C<sub>q</sub> of Triaz), 130.0 (CH of Ph), 129.9 (CH of Ph), 129.3 (CH of Ph), 126.1 (m-CH of BPh<sub>4</sub>), 126.1 (C5<sup>2</sup>), 125.9 (C<sub>q</sub> of Ph), 122.29 (p-CH of BPh<sub>4</sub>), 109.1 (C4), 89.7 (C<sub>q</sub> of Cp\*), 56.8 (CH<sub>2</sub><sup>b</sup>), 46.1 (CH<sub>2</sub><sup>a</sup>), 9.1 (CH<sub>3</sub> of Cp\*) ppm.

#### B.3 Synthesis of [Ir(2a)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (6a)

Elemental Analysis: Found: C, 44.41; H, 2.67 and N, 4.88; Calculated for  $C_{54}H_{38}BClF_{24}IrN_5$ : C, 44.69; H, 2.64 and N, 4.83 %.

HR-MS (ESI<sup>+</sup>, MeOH):  $[M]^+$  = 588.1499 (Calculated  $[M]^+$  = 588.1506) amu.

<sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>, 600 MHz):  $\delta$  8.22 (s, 1H, **H5'**), 7.78 (d, <sup>3</sup>*J*<sub>H4-H5</sub> = 2.4 Hz, 1H, **H5**), 7.74 (d, <sup>3</sup>*J*<sub>H4-H5</sub> = 2.4 Hz, 1H, **H3**), 7.72 (br s, 8H, *o*-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.70-7.69 (m, 2H, *o*-C**H** of **Ph**), 7.65-7.63 (m, 3H, *p* & *m*-C**H** of **Ph**), 7.56 (br s, 4H, *p*-C**H** of BAr<sup>F</sup><sub>4</sub>), 6.60 (t, <sup>3</sup>*J*<sub>H-H</sub> = 2.4 Hz, 1H, **H4**), 5.66 (d, <sup>2</sup>*J*<sub>H-H</sub> = 15.6 Hz, 1H, C**H**<sub>A</sub> of CH<sub>A</sub>H<sub>B</sub>), 5.04 (d, <sup>2</sup>*J*<sub>H-H</sub> = 15.6 Hz, 1H, C**H**<sub>B</sub> of CH<sub>A</sub>H<sub>B</sub>), 1.68 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CD<sub>2</sub>Cl<sub>2</sub>, 150 MHz): δ 162.2 (q,  ${}^{1}J_{B-C} = 51.0$  Hz, *ipso-*C to B, BAr<sup>F</sup><sub>4</sub>), 145.6 (C5), 139.6 (C<sub>q</sub> of Triaz), 135.9 (C<sub>q</sub> of **Ph**), 135.2 (*o*-CH to B, BAr<sup>F</sup><sub>4</sub>), 134.5 (C3), 131.7 (*p*-CH of **Ph**), 130.9 (*m*-CH of **Ph**), 129.3 (q,  ${}^{3}J_{B-C} = 31.5$  Hz, CCF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 125.0 (q,  ${}^{1}J_{F-C} = 270.0$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 122.4 (C5<sup>2</sup>), 121.5 (*o*-CH of **Ph**), 117.9 (br s, *p*-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.6 (C4), 89.8 (C<sub>q</sub> of Cp\*), 45.8 (CH<sub>2</sub>), 9.3 (CH<sub>3</sub> of Cp\*) ppm.

#### B.4 Synthesis of [Ir(2a)Cp\*Cl]BPh<sub>4</sub> (6a')

Elemental Analysis: Found: C, 60.79; H, 4.83 and N, 7.71; Calculated for  $C_{46}H_{46}BClIrN_5$ : C, 60.89; H, 5.11 and N, 7.72%.

HR-MS (ESI<sup>+</sup>, MeOH):  $[M]^+$  = 588.1502 (Calculated  $[M]^+$  = 588.1506) amu.

<sup>1</sup>H NMR (Acetone- $d_6$ , 600 MHz): δ 8.78 (s, 1H, **H5**' (triazole)). 8.12 (d,  ${}^3J_{\text{H4-H3}} = 2.4 \text{ Hz}$ , 1H, **H3**), 7.91 (d,  ${}^3J_{\text{H4-H5}} = 2.4 \text{ Hz}$ , 1H, **H5**), 7.91 (br s, 2H, o-C**H** of **Ph**), 7.70 (t,  ${}^3J_{\text{H-H}} = 7.2 \text{ Hz}$ , 2H, m-C**H** of **Ph**), 7.64 (t,  ${}^3J_{\text{H-H}} = 7.2 \text{ Hz}$ , 1H, p-C**H** of **Ph**), 7.36-7.34 (m, 8H, o-C**H** of B**Ph**<sub>4</sub>), 6.92 (t,  ${}^3J_{\text{H-H}} = 7.8 \text{ Hz}$ , 8H, m-C**H** of B**Ph**<sub>4</sub>), 6.77 (t,  ${}^3J_{\text{H-H}} = 7.2 \text{ Hz}$ , 4H, p-C**H** of B**Ph**<sub>4</sub>), 6.64 (t,  ${}^3J_{\text{H-H}} = 2.4 \text{ Hz}$ , 1H, **H4**), 6.08 (d,  ${}^2J_{\text{H-H}} = 16.2 \text{ Hz}$ , 1H, C**H**<sub>A</sub> of CH<sub>A</sub>H<sub>B</sub>), 5.30 (d,  ${}^2J_{\text{H-H}} = 16.2 \text{ Hz}$ , 1H, C**H**<sub>B</sub> of CH<sub>A</sub>H<sub>B</sub>), 1.77 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (Acetone- $d_6$ , 150 MHz): δ 165.0 (q,  ${}^{1}J_{B-C} = 49.5$  Hz, ipso-C to B, B**Ph**<sub>4</sub>), 145.8 (C5), 141.1 (C5'), 137.0 (o-CH of B**Ph**<sub>4</sub> & C<sub>q</sub> of Triaz), 135.7 (C3), 131.4 (p-CH of **Ph**), 131.0 (m-CH of **Ph**), 126.0 (m-CH of B**Ph**<sub>4</sub>), 124.2 (C<sub>q</sub> of **Ph**), 122.3 (p-CH of B**Ph**<sub>4</sub>), 122.1 (o-CH of **Ph**), 109.1 (C4), 90.0 (C<sub>q</sub> of Cp\*), 46.1 (CH<sub>2</sub>), 9.1 (CH<sub>3</sub> of Cp\*) ppm.

#### B.5 Synthesis of [Ir(2c)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (6c)

[IrCp\*Cl<sub>2</sub>]<sub>2</sub> (52 mg, 0.0066 mmol) and ligand **2c** (35 mg, 0.13 mmol) were dissolved in dichloromethane (15 mL). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (117 mg, 0.132 mmol) was added, resulting in the formation of a white precipitate in a yellow

solution. The reaction mixture was stirred for 2 hours before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca.* 3 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **6c** as an orange solid (108 mg, 56 %). m.p. 99-102 °C (decomposed).

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 633.1667 (100 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 633.1357)

Elemental Analysis: Found: C, 43.51; H, 2.68; N, 5.62. Calculated for  $C_{54}H_{37}BClF_{24}IrN_6O_2$ : C, 43.34; H, 2.49; N, 5.62 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 400 MHz): δ 9.28 (s, 1H, **H5'**), 8.56 (d,  ${}^3J_{\text{H-H}}$ = 9.1 Hz, 2H, o-C**H** of **Ph**NO<sub>2</sub>), 8.32 (d,  ${}^3J_{\text{H-H}}$  = 9.1 Hz, 2H, m-C**H** of **Ph**NO<sub>2</sub>), 8.23 (d,  $J_{\text{H4-H5}}$  = 2.4 Hz, 1H, **H5**), 7.93 (d,  ${}^3J_{\text{H4-H3}}$  = 2.4 Hz, 1H, **H3**), 7.79 (br s, 8H, o-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, p-C**H** of BAr<sup>F</sup><sub>4</sub>), 6.67 (apparent t,  ${}^3J_{\text{H-H}}$  = 2.4 Hz, 1H, **H4**), 6.31 (d,  ${}^2J_{\text{H-H}}$  = 16.1 Hz, 1H, C**H**<sub>a</sub> of CH<sub>a</sub>H<sub>b</sub>), 5.42 (d,  ${}^2J_{\text{H-H}}$  = 16.1 Hz, 1H, C**H**<sub>b</sub> of CH<sub>a</sub>H<sub>b</sub>), 1.80 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (acetone- $d_6$ , 100 MHz): δ 162.6 (q,  ${}^1J_{B-C}$  = 49.8 Hz, *ipso-*C to B, BAr<sup>F</sup><sub>4</sub>), 149.5 (C<sub>q</sub> of **Ph**NO<sub>2</sub>), 145.9 (C3), 141.8 (C<sub>q</sub> of Triaz), 141.1 (*ipso-*C to NO<sub>2</sub> of **Ph**NO<sub>2</sub>), 135.8 (C5), 135.2 (*o-*CH to B, BAr<sup>F</sup><sub>4</sub>), 130.0 (q,  ${}^2J_{F-C}$  = 30.6 Hz, *ipso-*C to CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 126.7 (C<sub>q</sub> of CH<sub>2</sub>), 126.5 (*m-*CH of **Ph**NO<sub>2</sub>), 125.4 (q,  ${}^1J_{F-C}$  = 270.7 Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.8 (C5'), 123.1 (*o-*CH of **Ph** NO<sub>2</sub>), 118.5 (br s, *p-*CH to B, BAr<sup>F</sup><sub>4</sub>), 109.2 (C4), 90.2 (C<sub>q</sub> of Cp\*), 46.2 (CH<sub>2</sub>), 9.1 (CH<sub>3</sub> of Cp\*) ppm.

#### B.6 Synthesis of [Ir(3a)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (7a)

[IrCp\*Cl<sub>2</sub>]<sub>2</sub> (52 mg, 0.065 mmol) and ligand **3a** (33 mg, 0.13 mmol) were dissolved in dichloromethane (10 mL). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (121 mg, 0.136 mmol) was added, resulting in the formation of a white precipitate in a yellow solution. The reaction

mixture was stirred for 2 hours before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca.* 2 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **7a** as a pale yellow solid (134 mg, 70 %). m.p. 142-145 °C (decomposed).

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 616.2500 (100 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 616.1813) amu.

Elemental Analysis: Found: C, 45.29; H, 2.97; N, 4.40. Calculated for C<sub>56</sub>H<sub>42</sub>BClF<sub>24</sub>IrN<sub>5</sub>: C, 45.46; H, 2.86; N, 4.73 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 400 MHz): δ 8.96 (s, 1H, **H5'**), 7.93 (d,  ${}^3J_{\text{H-H}}$  = 8.0 Hz, 2H, o-C**H** of **Ph**), 7.79 (br s, 8H, o-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.71 (t,  ${}^3J_{\text{H-H}}$  = 8.0 Hz, 2H, m-C**H** of **Ph**), 7.67 (t,  ${}^3J_{\text{H-H}}$  = 8.0 Hz, 1H, p-C**H** of **Ph**, overlapped with p-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, p-C**H** of BAr<sup>F</sup><sub>4</sub>), 6.31 (s, 1H, **H4**), 6.03 (d,  ${}^2J_{\text{H-H}}$  = 16.3 Hz, 1H, C**H**<sub>a</sub> of CH<sub>a</sub>H<sub>b</sub>), 5.08 (d,  ${}^2J_{\text{H-H}}$  = 16.3 Hz, 1H, C**H**<sub>b</sub> of CH<sub>a</sub>H<sub>b</sub>), 2.55 (s, 3H, C5-C**H**<sub>3</sub>), 2.51 (s, 3H, C3-C**H**<sub>3</sub>), 1.76 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (acetone- $d_6$ , 100 MHz): δ 162.6 (q,  ${}^1J_{\text{B-C}} = 48.5$  Hz, ipso-C to B, BAr<sup>F</sup><sub>4</sub>), 154.3 (C3), 144.5 (C5), 141.5 (C<sub>q</sub> of Triaz), 137.1 (C<sub>q</sub> of **Ph**), 135.5 (br s, o-CH to B, BAr<sup>F</sup><sub>4</sub>), 130.0 (q,  ${}^2J_{\text{F-C}} = 31.4$  Hz,  ${}^3J_{\text{B-C}} = 2.9$  Hz, ipso-C to CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 131.1 (m-CH of **Ph**), 131.0 (p-CH of **Ph**), 125.4 (q,  ${}^1J_{\text{F-C}} = 270.2$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.1 (C5'), 122.1 (o-CH of **Ph**), 118.4 (br s, p-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.1 (C4), 89.8 (C<sub>q</sub> of Cp\*), 42.8 (CH<sub>2</sub>), 15.6 (C3-CH<sub>3</sub>), 11.7 (C5-CH<sub>3</sub>), 9.3 (CH<sub>3</sub> of Cp\*) ppm.

#### B.7 Synthesis of [Ir(3b)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (7b)

[IrCp\*Cl<sub>2</sub>]<sub>2</sub> (53 mg, 0.0066 mmol) and ligand **3b** (43 mg, 0.13 mmol) were dissolved in dichloromethane (15 mL). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (119 mg, 0.134 mmol) was added, resulting in the formation of a white precipitate in a yellow

solution. The reaction mixture was stirred overnight before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca.* 2 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **7b** as a pale yellow solid (127 mg, 62 %). m.p. 188-191 °C (decomposed).

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 684.1667 (79 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 684.1687) amu.

Elemental Analysis: Found: C, 44.65; H, 2.80; N, 4.35. Calculated for C<sub>57</sub>H<sub>41</sub>BClF<sub>27</sub>IrN<sub>5</sub>: C, 44.24; H, 2.67; N, 4.53 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 600 MHz): δ 9.12 (s, 1H, **H5**'), 8.23 (d,  ${}^3J_{\text{H-H}} = 8.5$  Hz, 2H, m-CH of **Ph**), 8.08 (d,  ${}^3J_{\text{H-H}} = 8.5$  Hz, 2H, o-CH of **Ph**CF<sub>3</sub>), 7.79 (br s, 8H, o-CH of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, p-CH of BAr<sup>F</sup><sub>4</sub>), 6.32 (s, 1H, **H4**), 6.06 (d,  ${}^2J_{\text{H-H}} = 16.3$  Hz, 1H, CH<sub>a</sub> of CH<sub>a</sub>H<sub>b</sub>), 5.10 (d,  ${}^2J_{\text{H-H}} = 16.3$  Hz, 1H, CH<sub>b</sub> of CH<sub>a</sub>H<sub>b</sub>), 2.55 (s, 3H, C5-CH<sub>3</sub>), 2.52 (s, 3H, C3-CH<sub>3</sub>), 1.77 (s, 15H, CH<sub>3</sub> of Cp\*) ppm. (acetone- $d_6$ , 150 MHz): δ 162.6 (q,  ${}^1J_{\text{B-C}} = 49.6$  Hz, ipso-C to B, BAr<sup>F</sup><sub>4</sub>), 154.5 (C3), 144.6 (C5), 141.9 (C<sub>q</sub> of Triaz), 139.8 (C<sub>q</sub> of **Ph**CF<sub>3</sub>), 135.5 (o-CH to B, BAr<sup>F</sup><sub>4</sub>), 132.4 (q,  ${}^2J_{\text{C-F}} = 33.2$  Hz, ipso-C to CF<sub>3</sub> of **Ph**CF<sub>3</sub>), 130.0 (q,  ${}^2J_{\text{F-C}} = 31.6$  Hz,  ${}^3J_{\text{B-C}} = 2.8$  Hz, ipso-C<sub>q</sub> to CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 128.1 (o-CH of **Ph**CF<sub>3</sub>), 125.4 (q,  ${}^1J_{\text{F-C}} = 270.3$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.6 (q,  ${}^1J_{\text{C-F}} = 270.7$  Hz, PhCF<sub>3</sub>), 124.5 (C5'), 122.7 (m-CH of **Ph**CF<sub>3</sub>), 118.4 (br s, p-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.2 (C4), 90.0 (Cq of Cp\*), 42.8 (CH<sub>2</sub>), 15.6 (C3-CH<sub>3</sub>), 11.7 (C5-CH<sub>3</sub>), 9.3 (CH<sub>3</sub> of Cp\*) ppm.

#### B.8 Synthesis of [Ir(3c)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (7c)

[IrCp\*Cl<sub>2</sub>]<sub>2</sub> (51 mg, 0.064 mmol) and ligand **3c** (39 mg, 0.13 mmol) were dissolved in dichloromethane (15 mL). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (120 mg, 0.135 mmol) was added, resulting in the formation of a white precipitate in a brown

solution. The reaction mixture was stirred for 2 hours before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca.* 3 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **7c** as an orange solid (96 mg, 50 %). m.p. 107-110 °C (decomposed).

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 661.1667 (100 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 661.1664) amu.

Elemental Analysis: Found: C, 44.26; H, 2.70; N, 5.53. Calculated for  $C_{56}H_{41}BClF_{24}IrN_6O_2$ : C, 44.12; H, 2.71; N, 5.51 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 400 MHz): δ 9.19 (s, 1H, **H5**'), 8.56 (d,  ${}^3J_{\text{H-H}} = 9.2$  Hz, 2H, m-CH of **Ph**NO<sub>2</sub>), 8.30 (d,  ${}^3J_{\text{H-H}} = 9.2$  Hz, 2H, o-CH of **Ph**NO<sub>2</sub>), 7.79 (br s, 8H, o-CH of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, p-CH of BAr<sup>F</sup><sub>4</sub>), 6.32 (s, 1H, **H4**), 6.08 (d,  ${}^2J_{\text{H-H}} = 16.3$  Hz, 1H, CH<sub>a</sub> of CH<sub>a</sub>H<sub>b</sub>), 5.11 (d,  ${}^2J_{\text{H-H}} = 16.3$  Hz, 1H, CH<sub>b</sub> of CH<sub>a</sub>H<sub>b</sub>), 2.55 (s, 3H, C5-CH<sub>3</sub>), 2.52 (s, 3H, C3-CH<sub>3</sub>), 1.77 (s, 15H, CH<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (acetone- $d_6$ , 100 MHz): δ 162.6 (q,  ${}^1J_{B-C} = 48.9$  Hz, ipso-C to B, BAr<sup>F</sup><sub>4</sub>), 154.5 (C3), 149.5 (C<sub>q</sub> of **Ph**NO<sub>2</sub>), 142.1 (C<sub>q</sub> of Triaz), 141.1 (ipso-C to NO<sub>2</sub> of **Ph**NO<sub>2</sub>), 144.6 (C5), 135.5 (br s, o-CH to B, BAr<sup>F</sup><sub>4</sub>), 130.0 (q,  ${}^2J_{F-C} = 29.6$  Hz,  ${}^3J_{B-C} = 3.0$  Hz, ipso-C of CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 126.5 (m-CH of **Ph**NO<sub>2</sub>), 125.4 (q,  ${}^1J_{F-C} = 270.2$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.0 (C5'), 123.0 (o-CH of **Ph**NO<sub>2</sub>), 118.5 (br s, p-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.2 (C4), 90.0 (C<sub>q</sub> of Cp\*), 42.9 (CH<sub>2</sub>), 15.6 (C3-CH<sub>3</sub>), 11.7 (C5-CH<sub>3</sub>), 9.3 (CH<sub>3</sub> of Cp\*) ppm.

#### B.9 Synthesis of [Ir(3d)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (7d)

[IrCp\*Cl<sub>2</sub>]<sub>2</sub> (50 mg, 0.063 mmol) and ligand **3d** (31 mg, 0.13 mmol) were dissolved in dichloromethane (12 mL). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (114 mg, 0.129 mmol) was added, resulting in the formation of a white precipitate in a yellow

solution. The reaction mixture was stirred for 3 hours before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca.* 2 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **7d** as a yellow solid (124 mg, 66 %). m.p. 196-198 °C.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 630.3333 (41 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 630.1970) amu.

Elemental Analysis: Found: C, 46.09; H, 2.96; N, 4.73. Calculated for C<sub>57</sub>H<sub>44</sub>BClF<sub>24</sub>IrN<sub>5</sub>: C, 45.84; H, 2.97; N, 4.69 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 600 MHz): δ 8.90 (s, 1H, **H5**'), 7.80 (d,  ${}^3J_{\text{H-H}} = 8.3$  Hz, 2H, o-C**H** of **Ph**CH<sub>3</sub>, overlapped with o-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.79 (br s, 8H, o-C**H** of BAr<sup>F</sup><sub>4</sub> overlapped with o-C**H** of **Ph**CH<sub>3</sub>), 7.67 (br s, 4H, p-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.50 (d,  ${}^3J_{\text{H-H}} = 8.3$  Hz, 2H, m-C**H** of **Ph**CH<sub>3</sub>), 6.31 (s, 1H, **H4**), 6.00 (d,  ${}^2J_{\text{H-H}} = 16.3$  Hz, 1H, C**H**<sub>a</sub> of CH<sub>a</sub>H<sub>b</sub>), 5.07 (d,  ${}^2J_{\text{H-H}} = 16.3$  Hz, 1H, C**H**<sub>b</sub> of CH<sub>a</sub>H<sub>b</sub>), 2.55 (s, 3H, C5-C**H**<sub>3</sub>), 2.51 (s, 3H, C3-C**H**<sub>3</sub>), 2.45 (s, 3H, C**H**<sub>3</sub> of **Ph**CH<sub>3</sub>), 1.76 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm. (C3), 144.5 (C5), 141.9 (C<sub>q</sub> of Triaz), 141.4 (C<sub>q</sub> of **Ph**CH<sub>3</sub>), 135.5 (br s, o-CH to B, BAr<sup>F</sup><sub>4</sub>), 134.9 (*ipso*-C to CH<sub>3</sub> of **Ph**CH<sub>3</sub>), 131.5 (m-CH of **Ph**CH<sub>3</sub>), 130.0 (q,  ${}^2J_{\text{F-C}} = 31.4$  Hz, *ipso*-C of CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 125.4 (q,  ${}^1J_{\text{F-C}} = 270.8$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.0 (C5'), 122.0 (o-CH of **Ph**NO<sub>2</sub>), 118.5 (br s, o-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.1 (C4), 89.8 (C<sub>q</sub> of Cp\*), 42.8 (CH<sub>2</sub>), 21.1 (CH<sub>3</sub> of **Ph**CH<sub>3</sub>), 15.6 (C3-CH<sub>3</sub>), 11.7 (C5-CH<sub>3</sub>), 9.32 (CH<sub>3</sub> of Cp\*) ppm.

#### B.10 Synthesis of [Ir(4)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (8)

BArF<sub>4</sub>

[IrCp\*Cl<sub>2</sub>]<sub>2</sub> (0.269 g, 0.335 mmol) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (25 mL) prior to the addition of bis(1-pyrazolyl)methane (4, 0.101 g, 0.682 mmol) to the reaction mixture. The dark orange mixture turned bright orange. The reaction mixture was left to stir at room temperature for 30

minutes before NaBAr $_4^F$  (0.657 g, 0.741 mmol) was added to the mixture. The resulting murky pale yellow-orange mixture was stirred at room temperature for 1 hour and then filtered through celite to yield a clear yellow solution. The solution was then reduced in volume to ca.15 mL and n-pentane was added with vigorous stirring to precipitate the product 8, which was collected by filtration as a pale yellow crystalline solid (0.763 g, 83%); m.p. 174-176°C.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 511.1236 (100 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 511.12) amu.

Elemental Analysis: Found: C, 42.66; H, 2.53; N, 4.06. Calculated for C<sub>49</sub>H<sub>35</sub>BClF<sub>24</sub>IrN<sub>4</sub>: C, 42.82; H, 2.57; N, 4.08 %.

<sup>1</sup>H NMR (600 MHz, acetone– $d_6$ , 298 K): δ 8.25 (d,  ${}^3J$ (H3-H4) = 2.4 Hz, 2H, **H**3), 7.94 (d,  ${}^3J$ (H5-H4) = 1.8 Hz, 2H, **H**5), 7.79 (br s, 8H, *ortho*-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, *para*-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.28 (d,  ${}^2J$ (H<sub>B</sub>-H<sub>A</sub>) = 14.5 Hz, 1H, C**H**H), 6.67 (apparent t,  ${}^3J$  = 1.7 Hz, 2H, **H**4), 6.20 (d,  ${}^2J$ (H<sub>B</sub>-H<sub>A</sub>) = 14.5 Hz, 1H, CH**H**), 1.78 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>31</sup>C {<sup>1</sup>H} NMR (150 MHz, acetone– $d_6$ , 298 K): δ 162.63 (q,  ${}^1J$ (B-C) = 49.7 Hz, quat **C** *ipso* to B of BAr<sup>F</sup><sub>4</sub>), 146.65 (s, **C**5 of Pz), 135.75 (s, **C**3 of Pz), 135.58 (br s, *ortho*-**C**H of BAr<sup>F</sup><sub>4</sub>), 130.07 (q,  ${}^2J$ (F-C) = 30.3 Hz, **C** *ipso* to CF<sub>3</sub> of BAr<sup>F</sup><sub>4</sub>), 125.42 (q,  ${}^1J$ (F-C) = 270.15 Hz, **C**F<sub>3</sub> of BAr<sup>F</sup><sub>4</sub>) 118.50 (s,  ${}^3J$ (F-C) = 4.1 Hz, *para*-**C**H of BAr<sup>F</sup><sub>4</sub>), 109.58 (s, **C**4 of Pz), 90.00 (s, quat **C** of Cp\*), 64.24 (s, **C**H<sub>2</sub>), 9.15 (s, **C**H<sub>3</sub> of Cp\*) ppm.

#### B.11 Synthesis of [Rh(1)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (9)

[Rh(PyT)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (**9**) was synthesised following the method for the synthesis of complex **10a** from [RhCp\*Cl<sub>2</sub>]<sub>2</sub> (150 mg, 2.43 x  $10^{-5}$  mol), ligand **10** (109 mg, 4.56 x  $10^{-4}$  mol) and NaBAr<sup>F</sup><sub>4</sub> (473 mg, 5.34 x  $10^{-4}$  mol) to yield **9** as an orange solid (618 mg, 92%).

Elemental Analysis: Found: C, 48.07; H, 3.57 and N, 4.76; Calculated for  $C_{55}H_{40}BClF_{24}N_5O_2Rh$ : C, 48.01; H, 2.93 and N, 5.09 %.

ESI-MS (ESI<sup>+</sup>, MeOH): 511.96 ([M]<sup>+</sup>, 100%) amu.

<sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>, 600 MHz):  $\delta$  7.95 (s, 1H, **H5'**), 7.76 (d,  ${}^{3}J_{\text{H4-H5}} = 1.8$  Hz, 1H, **H5**), 7.72 (br s, 8H, *o*-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.70 (d,  ${}^{3}J_{\text{H4-H3}} = 1.8$  Hz, **H3**), 7.56 (br s, 4H, *p*-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.40 (m, 3H, *p* & *m*-C**H** of **Ph**), 7.34 (d, 2H,  ${}^{3}J_{\text{H-H}} = 3.6$  Hz, *o*-C**H** of **Ph**), 6.50 (t,  ${}^{3}J_{\text{H-H}} = 2.4$  Hz, 1H, **H4**), 5.68-5.62 (m, 3H, C**H**<sub>2</sub><sup>b</sup> & C**H**<sub>A</sub> of CH<sub>A</sub>H<sub>B</sub><sup>a</sup>), 4.85 (d,  ${}^{2}J_{\text{H-H}} = 15.6$  Hz, 1H, C**H**<sub>B</sub> of CH<sub>A</sub>H<sub>B</sub><sup>a</sup>), 1.60 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CD<sub>2</sub>Cl<sub>2</sub>, 150 MHz): δ 162.2 (q,  ${}^{1}J_{B-C} = 41.3$  Hz, *ipso-*C to B, BAr<sup>F</sup><sub>4</sub>), 145.3 (C5), 139.2 (C<sub>q</sub> of Triaz), 135.2 (*o*-CH to B, BAr<sup>F</sup><sub>4</sub>), 134.7 (C3), 132.9 (C<sub>q</sub> of **Ph**), 130.2 (*p*-CH of **Ph**), 130.0 (*m*-CH of **Ph**), 129.3 (qq,  ${}^{2}J_{F-C} = 22.5$  Hz,  ${}^{3}J_{B-C} = 2.5$  Hz CCF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 129.0 (*o*-CH of **Ph**), 125.0 (q,  ${}^{1}J_{F-C} = 226.3$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.6 (C5'), 117.9 (br s, *p*-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.3 (C4), 89.5 (C<sub>q</sub> of Cp\*), 56.8 (CH<sub>2</sub><sup>b</sup>), 45.9 (CH<sub>2</sub><sup>a</sup>), 9.2 (s, CH<sub>3</sub> of Cp\*) ppm.

#### B.12 Synthesis of [Rh(1)Cp\*Cl]BPh<sub>4</sub> (9')

[Rh(1)Cp\*Cl]BPh<sub>4</sub> (9') was synthesised following the method for the synthesis of complex 10a' from [RhCp\*Cl<sub>2</sub>]<sub>2</sub> (50.0 mg, 8.10 x  $10^{-5}$  mol), ligand 1 (36.0 mg, 1.62 x  $10^{-4}$  mol) and NaBPh<sub>4</sub> (60.9 mg, 1.78 x  $10^{-4}$  mol) to yield 9' as a yellow solid (117 mg, 88%). <u>m.p.</u> 222-224

Elemental Analysis: Found: C, 67.76; H, 5.64 and N, 7.95; Calculated for C<sub>47</sub>H<sub>48</sub>BN<sub>5</sub>Rh: C, 67.84; H, 5.81 and N, 8.42%.

ESI-MS (ESI<sup>+</sup>, MeOH): 511.89 ([M]<sup>+</sup>, 100%) amu.

<sup>1</sup>H NMR (Acetone- $d_6$ , 600 MHz): δ 8.40 (s, 1H, **H5'**). 8.07 (d,  ${}^3J_{\text{H4-H5}} = 1.8$  Hz, 1H, **H5**), 7.95 (d,  ${}^3J_{\text{H4-H3}} = 2.4$  Hz, 1H, **H3**), 7.48-7.43 (m, 5H, **Ph**), 7.37-7.35 (m, 8H, *o*-C**H** of B**Ph**<sub>4</sub>), 6.94 (t,  ${}^3J_{\text{H-H}} = 7.2$  Hz, 8H, *m*-C**H** of B**Ph**<sub>4</sub>), 6.79 (t,  ${}^3J_{\text{H-H}} = 7.2$  Hz, 4H, *p*-C**H** of B**Ph**<sub>4</sub>), 6.59 (t,  ${}^3J_{\text{H-H}} = 2.4$  Hz, 1H, **H4**), 5.99 (d,  ${}^2J_{\text{H-H}} = 15.9$  Hz, 1H, C**H**<sub>A</sub> of CH<sub>A</sub>H<sub>B</sub><sup>a</sup>), 5.86 (s, 2H, C**H**<sub>2</sub><sup>b</sup>), 5.40 (d,  ${}^2J_{\text{H-H}} = 15.9$  Hz, 1H, C**H**<sub>B</sub> of CH<sub>A</sub>H<sub>B</sub><sup>a</sup>), 1.73 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (Acetone- $d_6$ , 150 MHz): δ 165.0 (q,  ${}^{1}J_{B-C} = 50.0$  Hz, ipso-C to B, BPh<sub>4</sub>), 146.0 (C<sub>q</sub> of Triaz), 141.4 (C3), 137.1 (o-CH of BPh<sub>4</sub>), 135.9 (C5), 135.5 (C<sub>q</sub> of Ph), 130.0 (CH of Ph), 129.9 (CH of Ph), 129.4 (CH of Ph), 126.1 (m-CH of BPh<sub>4</sub>), 125.9 (C5'), 122.3 (p-CH of BPh<sub>4</sub>), 108.8 (C4), 97.9 (d,  ${}^{2}J_{Rh-C} = 7.5$  Hz, C<sub>q</sub> of Cp\*), 56.1 (CH<sub>2</sub><sup>b</sup>), 45.7 (CH<sub>2</sub><sup>a</sup>), 9.3 (CH<sub>3</sub> of Cp\*) ppm.

#### B.13 Synthesis of [Rh(2a)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (10a)

BAr $^{F_4}$  [RhCp\*Cl<sub>2</sub>]<sub>2</sub> (50.0 mg, 8.09 x 10<sup>-5</sup> mol) and ligand **2a** (36.4 mg, 1.62 x 10<sup>-4</sup> mol) were dissolved in CH<sub>2</sub>Cl<sub>2</sub> (10 mL). NaBAr $^{F_4}$  (158 mg, 1.78 x 10<sup>-4</sup> mol) was added, resulting in the formation of a white precipitate. The reaction mixture was stirred for 2 h before being

filtered through celite. The volume of the filtrate was reduced to *ca*. 2 mL and *n*-pentane added (20 mL) yielding **10a** as an orange solid (154 mg, 71%). m.p. 125-130 °C.

Elemental Analysis: Found: C, 47.82; H, 2.96 and N, 5.19; Calculated for  $C_{54}H_{38}BClF_{24}N_5Rh$ : C, 47.62; H, 2.81 and N, 5.14 %.

HR-MS (ESI<sup>+</sup>, MeOH):  $[M]^+$  = 498.0923 (Calculated  $[M]^+$  = 498.0932) amu.

<sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>, 600 MHz):  $\delta$  8.20 (s, 1H, **H5'**), 7.82 (br s, 1H, **H5/H3**), 7.74 (d,  ${}^{3}J_{\text{H-H}} = 2.4 \text{ Hz}$ , 1H, **H5/H3**), 7.72 (br s, 10H, *o*-C**H** of BAr<sup>F</sup><sub>4</sub> & *o*-C**H** of **Ph**), 7.66-7.62 (m, 3H, *p* & *m*-C**H** of **Ph**), 7.55 (br s, 4H, *p*-C**H** of BAr<sup>F</sup><sub>4</sub>), 6.59 (t,  ${}^{3}J_{\text{H-H}} = 2.4 \text{ Hz}$ , 1H, **H4**), 5.68 (br s, 1H, C**H**<sub>A</sub> of CH<sub>A</sub>H<sub>B</sub>), 5.17 (br s, 1H, C**H**<sub>B</sub> of CH<sub>A</sub>H<sub>B</sub>), 1.70 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CD<sub>2</sub>Cl<sub>2</sub>, 150 MHz): δ 162.2 (q,  ${}^{1}J_{B-C} = 49.5$  Hz, ipso-C to B, BAr<sup>F</sup><sub>4</sub>), 145.4 (C5/3), 140.5 (C<sub>q</sub> of Triaz), 136.0 (C<sub>q</sub> of **Ph**), 135.2 (o-CH to B, BAr<sup>F</sup><sub>4</sub>), 134.9 (C5/3), 131.6 (p-CH of **Ph**), 130.9 (m-CH of **Ph**), 129.3 (q,  ${}^{3}J_{B-C} = 30.0$  Hz, CCF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 125.0 (q,  ${}^{1}J_{F-C} = 270.0$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 122.6 (C5<sup>2</sup>), 121.5 (o-CH of **Ph**), 117.9 (br s, p-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.4 (C4), 97.9 (C<sub>q</sub> of Cp\*), 45.5 (CH<sub>2</sub>), 9.6 (CH<sub>3</sub> of Cp\*) ppm.

#### B.14 Synthesis of [Rh(2a)Cp\*Cl]BPh4 (10a')

[RhCp\*Cl<sub>2</sub>]<sub>2</sub> (50.0 mg, 8.10 x 10<sup>-5</sup> mol) was suspended in MeOH (10 mL) and ligand **2a** (36.4 mg, 1.62 x 10<sup>-4</sup> mol) added, resulting in a colour change from orange to bright yellow. NaBPh<sub>4</sub> (60.9 mg, 1.78 x 10<sup>-4</sup> mol) was added after 5 minutes, resulting in the formation of a

yellow precipitate. The reaction mixture was stirred for 1 h before the volume was reduced to *ca.* 1 mL and cooled in ice. The mixture was filtered, the solid washed with MeOH (3 x 10 mL) and dried to yield **21** as a yellow solid (100 mg, 75%). <u>m.p.</u> 209-211 °C (decomposed).

Elemental Analysis: Found: C, 67.44; H, 5.71 and N, 8.59; Calculated for C<sub>46</sub>H<sub>46</sub>BClN<sub>5</sub>Rh: C, 67.54; H, 5.67 and N, 8.56%.

HR-MS (ESI<sup>+</sup>, MeOH):  $[M]^+$  = 498.0923 (Calculated  $[M]^+$  = 498.0932) amu.

<sup>1</sup>H NMR (Acetone- $d_6$ , 600 MHz): δ 8.78 (s, 1H, **H5'**). 8.11 (d,  ${}^3J_{\text{H-H}} = 3.0 \text{ Hz}$ , 1H, **H5/H3**), 7.95 (d,  ${}^3J_{\text{H-H}} = 2.4 \text{ Hz}$ , 1H, **H3/H5**), 7.92 (d,  ${}^3J_{\text{H-H}} = 8.4 \text{ Hz}$ , 2H, o-C**H** of **Ph**), 7.69 (t,  ${}^3J_{\text{H-H}} = 7.8 \text{ Hz}$ , 2H, m-C**H** of **Ph**), 7.64 (t,  ${}^3J_{\text{H-H}} = 7.2 \text{ Hz}$ , 1H, p-C**H** of **Ph**), 7.36-7.34 (m, 8H, o-C**H** of B**Ph**<sub>4</sub>), 6.92 (t,  ${}^3J_{\text{H-H}} = 7.8 \text{ Hz}$ , 8H, m-C**H** of B**Ph**<sub>4</sub>), 6.77 (t,  ${}^3J_{\text{H-H}} = 7.2 \text{ Hz}$ , 4H, p-C**H** of B**Ph**<sub>4</sub>), 6.60 (t,  ${}^3J_{\text{H-H}} = 2.4 \text{ Hz}$ , 1H, **H4**), 6.05 (d,  ${}^2J_{\text{H-H}} = 16.2 \text{ Hz}$ , 1H, C**H**<sub>A</sub> of CH<sub>A</sub>H<sub>B</sub>), 5.47 (d,  ${}^2J_{\text{H-H}} = 16.2 \text{ Hz}$ , 1H, C**H**<sub>B</sub> of CH<sub>A</sub>H<sub>B</sub>), 1.80 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (Acetone- $d_6$ , 150 MHz): δ 165.0 (q,  ${}^{1}J_{B-C} = 49.5$  Hz, ipso-C to B, B**Ph**<sub>4</sub>), 146.0 (C5/C3), 141.9 (C<sub>q</sub> of Triaz), 137.2 (C<sub>q</sub> of **Ph**), 137.0 (o-CH of B**Ph**<sub>4</sub>), 135.9 (C3/C5), 131.3 (p-CH of **Ph**), 131.0 (m-CH of **Ph**), 126.0 (m-CH of B**Ph**<sub>4</sub>), 124.3 (C5'), 122.3 (p-CH of B**Ph**<sub>4</sub>), 122.0 (o-CH of **Ph**), 108.9 (C4), 98.1 (d,  ${}^{2}J_{Rh-C} = 9.0$  Hz, C<sub>q</sub> of Cp\*), 45.7 (CH<sub>2</sub>), 9.4 (s, CH<sub>3</sub> of Cp\*) ppm.

#### B.15 Synthesis of [Rh(4)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (11)

BArF.

[RhCp\*Cl<sub>2</sub>]<sub>2</sub> (0.209 g, 0.338 mmol) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (25 mL) prior to the addition of *bis*(1-pyrazolyl)methane (0.102 g, 0.688 mmol) to the reaction mixture. The dark orange-red mixture turned slightly lighter in colour. The reaction mixture was left to stir at room

temperature for 30 minutes before NaBAr $_4^F$  (0.658 g, 742 µmol) was added to the mixture. The resulting murky orange mixture was stirred at room temperature for 1 hour and then filtered through celite $_8^{\oplus}$  to yield a clear bright orange solution. The solution was then reduced in volume to ca.15 mL and n-pentane was added with vigorous stirring to precipitate the product  $\mathbf{8}$ , which was collected by filtration as a dark orange crystalline solid (0.687 g, 79 %). m.p. 189-192°C.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%) 421.0661 (100 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 421.07) amu.

Elemental Analysis: Found: C, 46.09; H, 2.78; N, 4.37. Calculated for C<sub>49</sub>H<sub>35</sub>BClF<sub>24</sub>N<sub>4</sub>Rh: C, 45.80; H, 2.75; N, 4.36 %.

<sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K):  $\delta$  7.84 (d, <sup>3</sup>*J*(H3-H4) = 2.8 Hz, 2H, **H**3), 7.82 (d, <sup>3</sup>*J*(H5-H4) = 2.2 Hz, 2H, **H**5), 7.72 (br s, 8H, *ortho*-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.56 (br s, 4H, *para*-C**H** of BAr<sup>F</sup><sub>4</sub>), 6.59 (apparent t, <sup>3</sup>*J* = 2.5 Hz, 2H, **H**4), 6.31 (d, <sup>2</sup>*J*(H<sub>B</sub>-H<sub>A</sub>) = 14.3 Hz, 1H, CH**H**), 6.01 (d, <sup>2</sup>*J*(H<sub>A</sub>-H<sub>B</sub>) = 14.3 Hz, 1H, C**H**H), 1.68 (s, 15H, C**H**<sub>3</sub> of Cp\*) ppm.

<sup>31</sup>C {<sup>1</sup>H} NMR (125 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K): δ 162.69 (q,  ${}^{1}J(B-C) = 49.6$  Hz,, q C *ipso* to B of BAr<sup>F</sup><sub>4</sub>), 148.27 (s, C3 of Pz), 136.20 (br s, *ortho*-CH of BAr<sup>F</sup><sub>4</sub>), 134.78 (s, C5 of Pz), 129.41 (q,  ${}^{2}J(F-C) = 30.4$  Hz, C *ipso* to CF<sub>3</sub> of BAr<sup>F</sup><sub>4</sub>), 125.15 (q,  ${}^{1}J(F-C) = 270.7$  Hz, CF<sub>3</sub> of BAr<sup>F</sup><sub>4</sub>) 118.04 (s,  ${}^{3}J(F-C) = 4.1$  Hz, 4C, *para*-CH of BAr<sup>F</sup><sub>4</sub>), 110.33 (s, C4 of Pz), 97.84 (s, quat C of Cp\*), 63.33 (s, CH<sub>2</sub>), 9.77 (s, CH<sub>3</sub> of Cp\*) ppm.

#### B.16 Synthesis of $[Rh(2c)(CO)_2)[BAr^F_4(14c)]$

OC CO BAr
$$^{F_4}$$

Rh

N=N

N NO

NO

S

[Rh(CO)<sub>2</sub>(Cl)]<sub>2</sub> (38 mg, 0.096 mmol) was dissolved in dichloromethane (15 mL) prior to the addition of the ligand **2c** (52 mg, 0.19 mmol). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (176 mg, 0.198 mmol) was added, resulting in the formation of a

white precipitate and a colour change from yellow to dark brown. The reaction mixture was stirred for 2 hours before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca.* 3 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **23** as a brown solid (157 mg, 63 %). m.p. 121-123 °C.

FTIR (CH<sub>2</sub>Cl<sub>2</sub>) v: 2110 (s, vCO), 2054 (s, vCO) cm<sup>-1</sup>.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 429.08334 (43 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 428.9814) amu.

Elemental Analysis: Found: C, 43.20; H, 1.84; N, 6.21. Calculated for  $C_{46}H_{22}BF_{24}N_6O_4Rh$ : C, 42.75; H, 1.72; N, 6.50 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 600 MHz): δ 9.37 (s, 1H, **H5'**), 8.58 (d,  ${}^3J_{\text{H-H}} = 9.3$  Hz, 2H, m-CH of **Ph**NO<sub>2</sub>), 8.37 (d,  $J_{\text{H4-H5}} = 2.4$  Hz, 1H, **H5**, overlapped with o-CH of **Ph**NO<sub>2</sub>), 8.36 (d,  ${}^3J_{\text{H-H}} = 9.3$  Hz, 2H, o-CH of **Ph**NO<sub>2</sub>, overlapped with H5), 8.32 (d,  ${}^3J_{\text{H4-H3}} = 2.4$  Hz, 1H, **H3**), 7.79 (br s, 8H, o-CH of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, p-CH of BAr<sup>F</sup><sub>4</sub>), 6.65 (apparent t,  ${}^3J_{\text{H-H}} = 2.4$  Hz, 1H, **H4**), 6.14 (s, 2H, C**H**<sub>2</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (acetone- $d_6$ , 150 MHz): δ 183.6 (d,  ${}^{1}J_{Rh-C}$  = 68.7 Hz, 2 x CO (overlapped)), 162.9 (q,  ${}^{1}J_{B-C}$  = 49.6 Hz, *ipso-*C to B, BAr<sup>F</sup><sub>4</sub>), 149.3 (C<sub>q</sub> of **Ph**NO<sub>2</sub>), 147.7 (C3), 143.0 (C<sub>q</sub> of Triaz), 140.8 (*ipso-*C to NO<sub>2</sub> of **Ph**NO<sub>2</sub>), 136.9 (C5), 135.5 (*o-*CH to B, BAr<sup>F</sup><sub>4</sub>), 130.0 (q,  ${}^{2}J_{F-C}$  = 31.6 Hz, *ipso-*C to CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 126.6 (*m-*CH of **Ph**), 125.4 (q,  ${}^{1}J_{F-C}$  = 270.2 Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 125.4 (C5'), 123.6 (*o-*CH of **Ph**NO<sub>2</sub>), 118.4 (br s, *p-*CH to B, BAr<sup>F</sup><sub>4</sub>), 109.0 (C4), 46.1 (CH<sub>2</sub>) ppm.

#### **B.17** Synthesis of [Rh(3a)(CO)<sub>2</sub>]BAr<sup>F</sup><sub>4</sub> (15a)

OC CO 
$$BAr^{F_4}$$

A  $N = N$ 

So  $m$ 
 $M = N$ 
 $M = N$ 

[Rh(CO)<sub>2</sub>Cl]<sub>2</sub> (31 mg, 0.080 mmol) was dissolved in dichloromethane (10 mL) prior to the addition of the ligand **3a** (41 mg, 0.16 mmol). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (148 mg, 0.166 mmol) was added, resulting in the formation of a white

precipitate and a colour change from yellow to brown. The reaction mixture was stirred overnight before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca.* 2 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **15a** as a brown solid (83 mg, 41 %). m.p. 44-47 °C (decomposed).

FTIR (CH<sub>2</sub>Cl<sub>2</sub>) v: 2106 (s, vCO), 2047 (s, vCO) cm<sup>-1</sup>.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 412.0833 (9 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 412.0275) amu.

Elemental Analysis: Found: C, 45.07; H, 2.05; N, 4.97. Calculated for  $C_{48}H_{27}BF_{24}N_5O_2Rh.0.25CH_2Cl_2$ : C, 44.69; H, 2.14; N, 5.40 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 400 MHz): δ 9.09 (s, 1H, **H5'**), 7.96 (d,  ${}^3J_{\text{H-H}} = 7.8$  Hz, 2H, o-C**H** of **Ph**), 7.79 (br s, 8H, o-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.71 (t,  ${}^3J_{\text{H-H}} = 7.8$  Hz, 2H, m-C**H** of **Ph**, overlapped with p-C**H** of **Ph**), 7.70 (t,  ${}^3J_{\text{H-H}} = 7.8$  Hz, 1H, p-C**H** of **Ph**, overlapped with m-C**H** of **Ph**), 7.68 (br s, 4H, p-C**H** of BAr<sup>F</sup><sub>4</sub>), 6.34 (s, 1H, **H4**), 5.89 (s, 2H, C**H**<sub>2</sub>), 2.57 (s, 3H, C3-C**H**<sub>3</sub>), 2.54 (s, 3H, C5-C**H**<sub>3</sub>) ppm.

13C{<sup>1</sup>H} NMR (acetone- $d_6$ , 100 MHz): δ 183.4 (d,  ${}^1J_{\text{Rh-C}} = 68.8$  Hz, 2 x CO (overlapped)), 162.6 (q,  ${}^1J_{\text{B-C}} = 49.5$  Hz, ipso-C to B, BAr<sup>F</sup><sub>4</sub>), 153.9 (C3), 146.2 (C5), 142.8 (Cq of Triaz), 136.8 (Cq of **Ph**), 135.5 (br s, o-CH to B, BAr<sup>F</sup><sub>4</sub>), 130.0 (q,  ${}^2J_{\text{F-C}} = 31.3$  Hz,  ${}^3J_{\text{B-C}} = 2.6$  Hz, ipso-C to CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 131.8 (p-CH of **Ph**), 131.2 (m-CH of **Ph**), 125.4 (q,  ${}^1J_{\text{F-C}} = 270.1$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.8 (C5'), 122.4 (o-CH of **Ph**), 118.4 (br s, p-CH to B, BAr<sup>F</sup><sub>4</sub>), 109.1 (C4), 42.6 (CH<sub>2</sub>), 15.6 (C3-CH<sub>3</sub>), 11.5 (C5-CH<sub>3</sub>) ppm.

#### **B.18** Synthesis of [Rh(3b)(CO)<sub>2</sub>]BAr<sup>F</sup><sub>4</sub> (15b)

OC CO 
$$BAr^{F_4}$$

Rh

 $N = N$ 
 $OC$ 
 $CF_3$ 

[Rh(CO)<sub>2</sub>(Cl)]<sub>2</sub> (40 mg, 0.10 mmol) was dissolved in dichloromethane (15 mL) prior to the addition of ligand **3b** (66 mg, 0.21 mmol). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (191 mg, 0.215 mmol) was added, resulting in the formation of a white

precipitate and a colour change from yellow to brown. The reaction mixture was stirred overnight before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca*. 3 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **15b** as a brown solid (177 mg, 64 %). m.p. 51-54 °C (decomposed).

FTIR (CH<sub>2</sub>Cl<sub>2</sub>) v: 2107 (s, vCO), 2049 (s, vCO) cm<sup>-1</sup>.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 480.1667 (41 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 480.0149) amu.

Elemental Analysis: Found: C, 44.27; H, 2.15; N, 4.77. Calculated for C<sub>49</sub>H<sub>26</sub>BF<sub>27</sub>N<sub>5</sub>O<sub>2</sub>Rh: C, 43.81; H, 1.95; N, 5.21 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 600 MHz):  $\delta$  9.24 (s, 1H, **H5'**), 8.25 (d,  ${}^3J_{\text{H-H}}$  = 8.4 Hz, 2H, m-C**H** of **Ph**CF<sub>3</sub>), 8.09 (d,  ${}^3J_{\text{H-H}}$  = 8.4 Hz, 2H, o-C**H** of **Ph**CF<sub>3</sub>), 7.79 (br s, 8H, o-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, p-C**H** of BAr<sup>F</sup><sub>4</sub>), 6.35 (s, 1H, **H4**), 5.92 (s, 2H, C**H**<sub>2</sub>), 2.57 (s, 3H, C3-C**H**<sub>3</sub>), 2.54 (s, 3H, C5-C**H**<sub>3</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (acetone- $d_6$ , 150 MHz): δ 183.7 (d,  ${}^1J_{\text{Rh-C}}$ = 69.8 Hz, 2 x CO (overlapped)), 162.6 (q,  ${}^1J_{\text{B-C}}$ = 49.6 Hz, *ipso*-C to B, BAr<sup>F</sup><sub>4</sub>), 154.0 (C3), 146.3 (C5), 143.2 (C<sub>q</sub> of Triaz), 139.8 (C<sub>q</sub> of **Ph**CF<sub>3</sub>), 135.5 (br s, *o*-CH to B, BAr<sup>F</sup><sub>4</sub>), 132.7 (q,  ${}^2J_{\text{C-F}}$  = 32.1, *ipso*-C to CF<sub>3</sub> of **Ph**CF<sub>3</sub>), 130.0 (q,  ${}^2J_{\text{F-C}}$  = 31.4 Hz, *ipso*-C of CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 128.5 (*o*-CH of **Ph**CF<sub>3</sub>), 125.4 (q,  ${}^1J_{\text{F-C}}$  = 270.2 Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.2 (q,  ${}^1J_{\text{C-F}}$  = 274.9 Hz, CF<sub>3</sub> of **Ph**CF<sub>3</sub>), 125.1 (C5°), 123.3 (*m*-CH of **Ph**CF<sub>3</sub>), 118.5 (br s, *p*-CH to B, BAr<sup>F</sup><sub>4</sub>), 108.9 (C4), 42.6 (CH<sub>2</sub>), 15.6 (C3-CH<sub>3</sub>), 11.6 (C5-CH<sub>3</sub>) ppm.

#### B.19 Synthesis of [Rh(3c)(CO)<sub>2</sub>]BAr<sup>F</sup><sub>4</sub> (15c)

[Rh(CO)<sub>2</sub>(Cl)]<sub>2</sub> (40 mg, 0.10 mmol) was dissolved in dichloromethane (15 mL) prior to the addition of ligand 3c (61 mg, 0.213 mmol) was added, resulting in the formation of a white precipitate and a colour change from yellow to brown. The reaction mixture was stirred for 2 hours before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca.* 3 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield 26 as a pale brown solid (151 mg, 56 %). m.p. 135-137 °C.

FTIR (CH<sub>2</sub>Cl<sub>2</sub>) v: 2108 (s, vCO), 2050 (s, vCO) cm<sup>-1</sup>.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 457.0152 (21 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 457.0126) amu.

Elemental Analysis: Found: C, 43.62; H, 2.11; N, 6.08. Calculated for C<sub>48</sub>H<sub>26</sub>BF<sub>24</sub>N<sub>6</sub>O<sub>4</sub>Rh: C, 43.66; H, 1.98; N, 6.36 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 600 MHz): δ 9.31 (s, 1H, **H5'**), 8.57 (d,  ${}^3J_{\text{H-H}} = 9.2$  Hz, 2H, m-CH of **Ph**NO<sub>2</sub>), 8.33 (d,  ${}^3J_{\text{H-H}} = 9.3$  Hz, 2H, o-CH of **Ph**NO<sub>2</sub>), 7.79 (br s, 8H, o-CH of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, p-CH of BAr<sup>F</sup><sub>4</sub>), 6.35 (s, 1H, **H4**), 5.90 (s, 2H, CH<sub>2</sub>), 2.57 (s, 3H, C3-CH<sub>3</sub>), 2.54 (s, 3H, C5-CH<sub>3</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (acetone- $d_6$ , 150 MHz): δ 183.9 (d,  ${}^{1}J_{Rh-C} = 68.7$  Hz, 2 x CO (overlapped)), 162.6 (q,  ${}^{1}J_{B-C} = 49.8$  Hz, ipso-C to B, BAr<sup>F</sup><sub>4</sub>), 154.0 (C5), 149.8 (C<sub>q</sub> of **Ph**NO<sub>2</sub>), 146.3 (C3), 143.3 (C<sub>q</sub> of Triaz), 140.9 (ipso-C to NO<sub>2</sub> of **Ph**NO<sub>2</sub>), 135.5 (o-CH to B, BAr<sup>F</sup><sub>4</sub>), 130.0 (q,  ${}^{2}J_{F-C} = 31.2$  Hz, ipso-C to CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 126.3 (m-CH of **Ph**), 125.4 (q,  ${}^{1}J_{F-C} = 270.0$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 125.3 (C5'), 123.5 (o-CH of **Ph**NO<sub>2</sub>), 118.4 (br s, p-CH to B, BAr<sup>F</sup><sub>4</sub>), 108.9 (C4), 42.6 (CH<sub>2</sub>), 15.6 (C5-CH<sub>3</sub>), 11.6 (C3-CH<sub>3</sub>) ppm.

#### B.20 Synthesis of [Rh(3d)(CO)<sub>2</sub>]BAr<sup>F</sup><sub>4</sub> (15d)

OC CO 
$$BAr^{F_4}$$

Rh

 $N = N$ 
 $OC$ 
 $OC$ 

[Rh(CO)<sub>2</sub>(Cl)]<sub>2</sub> (44 mg, 0.11 mmol) was dissolved in dichloromethane (15 mL) prior to the addition of ligand **3d** (54 mg, 0.22 mmol). After 5 minutes of stirring, NaBAr<sup>F</sup><sub>4</sub> (200 mg, 0.225 mmol) was added, resulting in the formation of a white

precipitate and a colour change from yellow to dark brown. The reaction mixture was stirred overnight before being filtered through celite<sup>®</sup> and rinsed with dichloromethane. The volume of the filtrate was reduced to *ca*. 2 mL and *n*-pentane (20 mL) was added with vigorous stirring to yield **15d** as a brown solid (159 mg, 62 %). m.p. 49-52 °C (decomposed).

FTIR (CH<sub>2</sub>Cl<sub>2</sub>) v: 2106 (s, vCO), 2047 (s, vCO) cm<sup>-1</sup>.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 426.0567 (18 %) [M]<sup>+</sup> (Calculated [M]<sup>+</sup> = 426.0432) amu.

Elemental Analysis Found: C, 45.55; H, 2.29; N, 5.38. Calculated for  $C_{49}H_{29}BF_{24}N_5O_2Rh$ : C, 45.64; H, 2.27; N, 5.43 %.

<sup>1</sup>H NMR (acetone- $d_6$ , 600 MHz): δ 9.02 (s, 1H, **H5**'), 7.82 (d,  ${}^3J_{\text{H-H}}$  = 8.4 Hz, 2H, o-C**H** of **Ph**CF<sub>3</sub>), 7.79 (br s, 8H, m-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.67 (br s, 4H, p-C**H** of BAr<sup>F</sup><sub>4</sub>), 7.51 (d,  ${}^3J_{\text{H-H}}$  = 8.0 Hz, 2H, o-C**H** of **Ph**CF<sub>3</sub>), 6.33 (s, 1H, **H4**), 5.87 (s, 2H, C**H**<sub>2</sub>), 2.57 (s, 3H, C3-C**H**<sub>3</sub>), 2.53 (s, 3H, C5-C**H**<sub>3</sub>), 2.45 (s, 3H, C**H**<sub>3</sub> of **Ph**CH<sub>3</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (acetone- $d_6$ , 150 MHz): δ 183.8 (d,  ${}^{1}J_{Rh-C} = 68.7$  Hz, 2 x CO (overlapped)), 162.6 (q,  ${}^{1}J_{B-C} = 49.9$  Hz, ipso-C to B, BAr<sup>F</sup><sub>4</sub>), 154.0 (C3), 146.2 (C5), 142.7 (C<sub>q</sub> of Triaz), 142.4 (C<sub>q</sub> of PhCH<sub>3</sub>), 135.5 (br s, o-CH to B, BAr<sup>F</sup><sub>4</sub>), 134.6 (ipso-C to CH<sub>3</sub> of PhCH<sub>3</sub>), 130.0 (q,  ${}^{2}J_{F-C} = 31.7$  Hz, ipso-C of CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 131.6 (m-CH of PhCH<sub>3</sub>), 125.4 (q,  ${}^{1}J_{F-C} = 270.6$  Hz, CF<sub>3</sub>, BAr<sup>F</sup><sub>4</sub>), 124.5 (C5'), 122.2 (o-CH of PhCH<sub>3</sub>), 118.5 (br s, p-CH to B, BAr<sup>F</sup><sub>4</sub>), 108.8 (C4), 42.6 (CH<sub>2</sub>), 21.1 (CH<sub>3</sub> of PhCH<sub>3</sub>), 15.6 (C3-CH<sub>3</sub>), 11.5 (C5-CH<sub>3</sub>) ppm.

**Table S2**: vCO vibrational frequencies and <sup>13</sup>C chemical shifts of <sup>13</sup>CO of [Rh(N-N')(CO)<sub>2</sub>]BAr<sup>F</sup><sub>4</sub> (13-16).

	Complexes	IR <sup>a</sup> (cm <sup>-1</sup> )	$\delta (CO)^b (ppm)$
[]	Rh(N-N')(CO) <sub>2</sub> ]BAr <sup>F</sup> <sub>4</sub> (13-16);		
	N-N' Ligand =		
13°	N = N	2108, 2050	182.8, 182.0
14a <sup>c</sup>	N N=N N	2108, 2051	182.6, 181.9
14b	N > N $N > N$ $N > N$ $N > N$	2110, 2053	183.7 <sup>d</sup>
14c	N = N $N = N$ $N = N$ $N = N$ $N = N$	2110, 2054	183.6 <sup>d</sup>
15d	N N=N CH <sub>3</sub>	2106, 2047	183.8
15a	N N=N	2106, 2047	183.4
15b	$N = N$ $N = N$ $N = CF_3$	2107, 2049	183.7
15c	N = N $N = N$ $N = N$ $N = N$ $N = N$	2108, 2050	183.6
16	N N N	2109, 2051	183.4 <sup>e</sup>

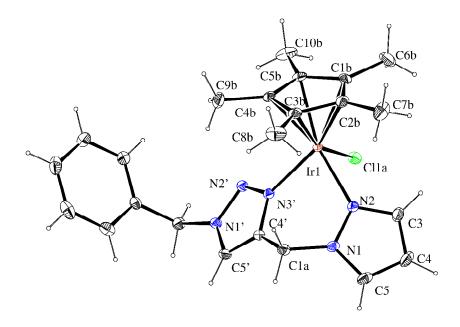
<sup>&</sup>lt;sup>a</sup> FTIR spectra were acquired as solution in dichloromethane. <sup>b</sup> NMR spectra were acquired in acetone- $d_6$  unless otherwise noted. <sup>c</sup> NMR spectra were acquired in CD<sub>2</sub>Cl<sub>2</sub> (reference <sup>1</sup>). <sup>d</sup> The two resonances for <sup>13</sup>CO overlap with each other in acetone- $d_6$  and appear as slightly broad doublet with <sup>1</sup> $J(^{103}\text{Rh}-^{13}\text{C}) \sim 70 \text{ Hz})$ . <sup>e</sup>Spectrum was acquired in THF- $d_8$  (reference 2).

#### Part C: X-ray Crystallography

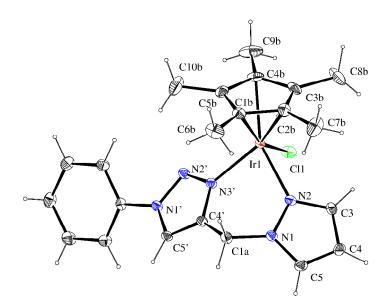
*General experimental for X-ray crystallography:* 

Suitable single crystals of **5**, **6a'**, **6b**, **7b**, **8**, **9**, **10a**, **10a'**, **11** and **15c** selected under the polarizing microscope (Leica M165Z), were picked up on a MicroMount (MiTeGen, USA) consisting of a thin polymer tip with a wicking aperture. The X-ray diffraction measurements were carried out on a Bruker kappa-II CCD diffractometer at ca.150 K by using graphite-monochromated Mo-K $\alpha$  radiation ( $\lambda = 0.710723$  Å). The single crystals, mounted on the goniometer using cryo loops for intensity measurements, were coated with paraffin oil and then quickly transferred to the cold stream using an Oxford Cryo stream attachment. Symmetry related absorption corrections using the program SADABS<sup>3</sup> were applied and the data were corrected for Lorentz and polarisation effects using Bruker APEX2 software.<sup>4</sup> All structures were solved by direct methods and the full-matrix least-square refinements were carried out using SHELXL.<sup>5</sup>

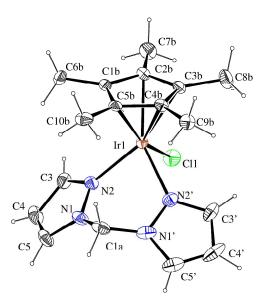
The non-hydrogen atoms were refined anisotropically. The molecular graphic was generated using Mercury.<sup>6</sup> In all the structures, the CF<sub>3</sub> groups of the anion exhibited extensive orientation disorder. Also, crystal lattice contained disordered solvent (mostly Dichloromethane), the treatment of these in the least-squares refinement have been outlined in the cifs.



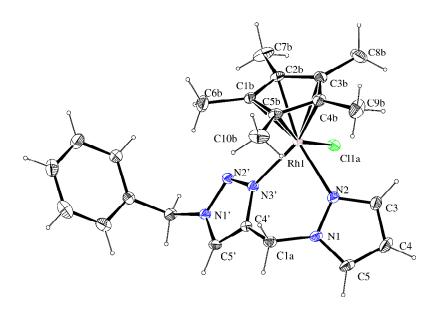
**Figure S1**: ORTEP depiction of [Ir(1)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (5) at 40% thermal ellipsoids for the non-hydrogen atoms.



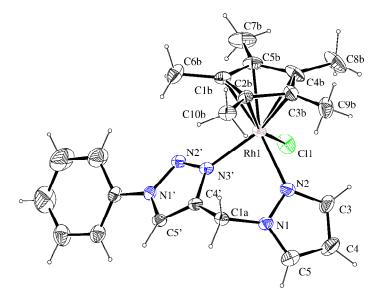
**Figure S2**: ORTEP depiction of [Ir(2a)Cp\*Cl]BPh<sub>4</sub> (6a') at 40% thermal ellipsoids for the non-hydrogen atoms.



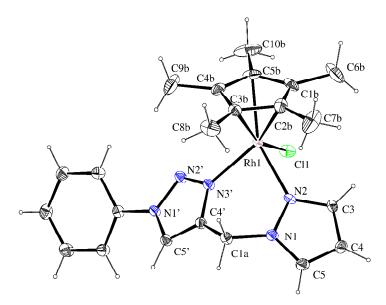
**Figure S3**: ORTEP depiction of [Ir(4)Cp\*Cl]BPh<sub>4</sub> (8) at 40% thermal ellipsoids for the non-hydrogen atoms.



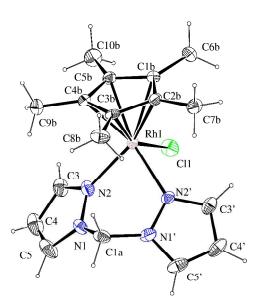
**Figure S4**: ORTEP depiction of [Rh(1)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (9) at 40% thermal ellipsoids for the non-hydrogen atoms.



**Figure S5**: ORTEP depiction of [Rh(1)Cp\*Cl]BPh<sub>4</sub> (10a) at 40% thermal ellipsoids for the non-hydrogen atoms.



**Figure S6**: ORTEP depiction of [Rh(1)Cp\*Cl]BPh<sub>4</sub> (10a') at 40% thermal ellipsoids for the non-hydrogen atoms.



**Figure S7**: ORTEP depiction of [Rh(4)Cp\*Cl]BPh<sub>4</sub> (11) at 40% thermal ellipsoids for the non-hydrogen atoms.

Table S3: Crystal structural data for the single crystal X-ray structures of Iridium complexes 8, 6b, 7b, 6a' and 5.

· ·	8 6b 7b 6a' 5					
	ð	OD	70	oa	3	
Chemical formula	$(C_{17}H_{23}ClIrN_4).$	$(C_{23}H_{25}ClF_3IrN_5).$	$(C_{25}H_{29}ClF_3IrN_5).$	$(C_{22}H_{26}CIIrN_5).$	$(C_{23}H_{28}ClIrN_5)$ .	
	$(C_{32}H_{12}BF_{23.97}).$	$(C_{32}H_{12}BF_{24}).$	$(C_{32}H_{12}BF_{24}).$	$(C_{24}H_{20}B).$	$(C_{32}H_{12}BF_{24}).$	
	$0.375(CCl_2)$	1(O)	$1(CH_2Cl_2)$		$0.5(CH_2Cl_2)$	
M (g mol-1)	1405.36	1535.36	1629.31	907.34	1507.84	
Crystal system	Monoclinic	Monoclinic	Monoclinic	Monoclinic	Triclinic	
Space group	C2/c	C2/c	$P2_1/c$	$P2_1/n$	$P^{-}1$	
Crystal habit	yellow blocks	yellow plates	yellow blocks	yellow blocks	yellow blocks	
Temperature (K)	155	170	170	156	151	
a (Å)	28.8655(12)	29.5110(16)	19.4090 (8)	12.0829(5)	12.5013(14)	
b (Å)	12.9022(5)	13.3179(7)	17.2163(8)	24.4058(10)	16.3750 (8)	
c (Å)	29.8343(12)	32.1453(17)	21.637(1)	13.3522(5)	16.9370 (8)	
α (°)	90	90	90	90	117.541(2)	
β (°)	104.722 (1)	109.255(3)	112.133(2)	93.701(2)	96.775(2)	
γ (°)	90	90	90	90	101.575(2)	
$V(\mathring{A}^3)$	10746.4(7)	11927.2(11)	6697.3(5)	3929.3(3)	2921.7(4)	
$\mathbf{Z}$	8	8	4	4	2	
Radiation type	$MoK_{\alpha}$	$MoK_{\alpha}$	$MoK_{\alpha}$	$MoK_{\alpha}$	$MoK_{\alpha}$	
μ (mm-1)	2.70	2.41	2.23	3.51	2.50	
Crystal size (mm)	$0.36 \times 0.26 \times 0.18$	$0.32 \times 0.10 \times 0.04$	$0.28 \times 0.26 \times 0.14$	$0.35 \times 0.25 \times 0.21$	$0.41 \times 0.38 \times 0.16$	
Tmin, Tmax	0.440, 0.648	0.513, 0.901	0.570, 0.741	0.373, 0.522	0.426, 0.685	
Refl. measured	38085	40150	45958	29202	37589	
Unique reflections	9435	10497	11717	6884	10113	
Obsd. Reflections	8380	7764	9929	6549	9804	
$[I > 2\sigma(I)]$						
Rint	0.023	0.053	0.024	0.026	0.028	
$R[F^2 > 2\sigma(F^2)]$	0.031	0.051	0.051	0.017	0.019	
$wR(F^2)$	0.132	0.154	0.166	0.042	0.047	
S	1.18	1.01	1.15	1.07	1.04	
Reflections used	9435	10497	11717	6884	10113	
Parameters	962	870	915	492	876	
Restraints	341	34	57	492	204	
$\Delta \rho \text{max}, \Delta \rho \text{min} (e \text{ Å}^{-3})$	1.41, -0.80	2.64, -1.39	1.39, -0.83	0.67, -0.62	0.61, -0.86	

Table S4: Crystal structural data for the single crystal X-ray structures of Rhodium complexes 11, 10a, 10a', 9 and 15c.

<u> </u>	11	10a	10a'	a, 10a <sup>2</sup> , 9 and 15c.	15c
	11	Iva	10a	9	130
Chemical formula	$(C_{17}H_{23}ClN_4Rh).$	$(C_{22}H_{26}ClN_5Rh).$	$(C_{22}H_{26}ClN_5Rh).$	$(C_{23}H_{28}ClN_5Rh).$	$(C_{16}H_{24}N_6O_4Rh).$
Chemical formala	$(C_{17}H_{23}CH_{44}CH)$ . $(C_{32}H_{12}BF_{24})$ .	$(C_{32}H_{12}BF_{24}).$	$(C_{24}H_{20}B).$	$(C_{32}H_{12}BF_{23.97}).$	$(C_{32}H_{12}BF_{24}).$
	0.125(CCl <sub>2</sub> )	0.875(CH <sub>2</sub> Cl <sub>2</sub> )	(C241120D).	$0.5(CH_2Cl_2)$	$0.75(CCl_2)$
M (g mol-1)	1295.34	1436.37	818.05	1418.55	1382.65
Crystal system	Monoclinic	Monoclinic	Monoclinic	Triclinic	Monoclinic
Space group	C2/c	C2/c	$P2_1/n$	$P^{-}1$	C2/c
Crystal habit	yellow blocks	Yellow blocks	Red orange blocks	Yellow blocks	Yellow blocks
Temperature (K)	152	151	155	152	160
a (Å)	28.8126(10)	29.9749(17)	12.0994(3)	12.4931(16)	39.396(4)
b (Å)	12.8881(4)	13.0963(8)	24.4422(7)	16.3598(8)	18.3734(16)
c (Å)	29.9049(11)	36.371(2)	13.3308(4)	16.8978 (10)	16.4800(16)
α (°)	90	90	90	117.365(2)	90
β (°)	104.623(2)	111.572(13)	93.565(1)	96.805(3)	106.912(5)
γ (°)	90	90	90	101.470(3)	90
$V(A^3)$	10745.2(6)	13277.7(13)	3934.77(19)	2917.2(4)	11413.0(19)
Z	8	8	4	2	8
Radiation type	$MoK_{\alpha}$	$MoK_{\alpha}$	$MoK_{\alpha}$	$MoK_{\alpha}$	$MoK_{\alpha}$
μ (mm-1)	0.50	0.47	0.54	0.50	0.50
Crystal size (mm)	$0.18\times0.15\times0.05$	$0.34 \times 0.23 \times 0.15$	$0.29 \times 0.17 \times 0.16$	$0.33 \times 0.12 \times 0.06$	$0.21\times0.21\times0.09$
Tmin, Tmax	0.913, 0.974	0.857, 0.931	0.860, 0.921	0.851, 0.972	0.903, 0.957
Refl. measured	39871	40705	25137	33968	123073
Unique reflections	9456	11579	6909	10077	10039
Obsd. Reflections	7065	8562	5495	8671	8611
$[I > 2\sigma(I)]$					
Rint	0.053	0.072	0.148	0.084	0.039
$R[F^2 > 2\sigma(F^2)]$	0.075	0.088	0.027	0.035	0.056
$wR(F^2)$	0.222	0.260	0.068	0.082	0.183
S	1.46	1.10	1.06	1.04	0.76
Reflections used	9456	11579	6909	10077	10039
Parameters	759	922	492	865	819
Restraints	124	336	0	174	36
Δρmax, Δρmin (e Å <sup>-3</sup> )	1.47, -0.94	1.54, -1.49	1.06, -0.76	0.61, -0.82	1.25, -0.99

## Part D: Synthesis of 2-(hydroxyalk-1-ynyl)aniline substrates

2-(6-hydroxyalk-1-ynyl)aniline substrates: 2-(6-hydroxyhex-1-ynyl)aniline (17S), 2-(6-hydroxypent-1-ynyl)aniline (18S), 2-(6-hydroxyhept-1-ynyl)aniline (19S), 2-(6-hydroxy-6-methylhex-1-ynyl)aniline (20S) and were synthesised by the Sonogashira coupling reaction between 2-iodoaniline and the corresponding terminal alkynol. Several reagents used in the synthesis of 17S-20S, namely 6-heptyn-2-ol, 6-heptyn-1-ol and [Pd(PPh<sub>3</sub>)<sub>4</sub>] were prepared according to literature procedures. A typical synthesis procedure is included for compound 17S. The syntheses of 18S-21S were conducted in an analogous manner.

# D.1 Synthesis of 2-(6-Hydroxy-1-hexyn-1-yl)aniline, 17S<sup>10</sup>

A solution of 2-iodoaniline (3.29 g, 15.0 mmol) and 5-hexyn-1-ol (1.65 mL, 15.0 mmol) in triethylamine (50 mL) was deoxygenated  $\frac{3}{6}$   $\frac{2}{1}$   $\frac{1}{1}$   $\frac{1}{1$ 

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.24 (dd, <sup>3</sup>*J*(H4-H3) = 7.9 Hz, <sup>4</sup>*J*(H5-H3) = 1.5 Hz, 1H, **H**3), 7.07 (apparent td, <sup>3</sup>*J*(H6-H5) = <sup>3</sup>*J*(H4-H5) = 7.9 Hz, <sup>4</sup>*J*(H3-H5) = 1.5 Hz, 1H, **H**5), 6.67 (dd, <sup>4</sup>*J*(H5-H6) = 7.8 Hz, <sup>3</sup>*J*(H4-H6) = 1.5 Hz, 1H, **H**6), 6.65 (apparent td, <sup>3</sup>*J*(H5-H4) = 7.5 Hz, <sup>3</sup>*J*(H3-H4) = 7.8 Hz, <sup>4</sup>*J*(H6-H4) = 1.5 Hz, 1H, **H**4), 3.71 (t, <sup>3</sup>*J*(H5'-H6') = 6.3 Hz, 2H, **H**6'), 2.52 (t, <sup>3</sup>*J* = 6.7 Hz, 2H, **H**3'), 1.79-1.69 (m, 4H, H4' & H5') ppm.

## D.2 Synthesis of 2-(5- Hydroxy-1-pentyn-1-yl)aniline, 18S<sup>11</sup>

Starting with 2-iodoaniline (3.29 g, 15.0 mmol) and 4-pentyn-1-ol (1.39 mL, 15.0 mmol). Yield: 2.1 g, 80 %; yellow viscous oil;  $R_f = 0.43$  (Hexane : EtOAc = 4:6 (v/v)).

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  7.23 (dd, <sup>3</sup>*J*(H4-H3) = 8.0 Hz, <sup>4</sup>*J*(H5-H3) = 1.5 Hz, 1H, **H**3), 7.07 (apparent td, <sup>3</sup>*J*(H6-H5) = <sup>3</sup>*J*(H4-H5) = 7.5 Hz, <sup>4</sup>*J*(H3-H5) = 1.5 Hz, 1H, **H**5), 6.67 (dd, <sup>4</sup>*J*(H5-H6) = 7.5 Hz, <sup>3</sup>*J*(H4-H6) = 1.5 Hz, 1H, **H**6), 6.65 (apparent td, <sup>3</sup>*J*(H5-H4) = 7.5 Hz, <sup>3</sup>*J*(H3-H4) = 8.0 Hz, <sup>4</sup>*J*(H6-H4) = 1.5 Hz, 1H, **H**4), 3.76 (t, <sup>3</sup>*J*(H4'-H5') = 6.3 Hz, 2H, **H**5'), 3.66 (br s, 3H, N**H**<sub>2</sub> & O**H**), 2.56 (t, <sup>3</sup>*J*(H4'-H3') = 6.9 Hz, 2H, **H**3'), 1.84 (apparent p, <sup>3</sup>*J*(H3'-H4', H5'-H4') = 6.5 Hz, 2H, **H**4') ppm.

<sup>13</sup>C{1H} NMR (75 MHz, CDCl<sub>3</sub>): δ 147.7 (C2), 132.0 (C3), 129.0 (C5), 118.0 (C4), 114.4 (C6), 108.8 (C1), 94.9 (C2'), 77.5 (C1'), 61.5 (C5'), 31.6 (C4'), 16.2 (C3') ppm.

## D.3 Synthesis of 2-(6-hydroxyhept-1-ynyl)aniline, 19S

Starting with 2-iodoaniline (0.371 g, 1.69 mmol) and 6-heptyn-1-ol (0.190 g, 1.69 mmol). Yield: 0.201 g, 59 %; yellow viscous oil;  $R_f$  = 0.14 (Hexane : EtOAc = 7:3 (v/v)).

<sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  7.24 (dd, <sup>3</sup>*J*(H4-H3) = 7.6 Hz, <sup>4</sup>*J*(H5-H3) = 1.3 Hz, 1H, **H**3), 7.07 (apparent td, <sup>3</sup>*J*(H6-H5) = <sup>3</sup>*J*(H4-H5) = 7.7 Hz, <sup>4</sup>*J*(H3-H5) = 1.5 Hz, 1H, **H**5), 6.68 (dd, <sup>4</sup>*J*(H5-H6) = 7.6 Hz, <sup>3</sup>*J*(H4-H6) = 1.5 Hz, 1H, **H**6), 6.66 (apparent td, <sup>3</sup>*J*(H5-H4) = 7.5 Hz, <sup>3</sup>*J*(H3-H4) = 8.0 Hz, <sup>4</sup>*J*(H6-H4) = 1.5 Hz, 1H, **H**4), 4.16 (br s, 3H, N**H**<sub>2</sub>), 3.66 (t, <sup>3</sup>*J*(H6'-H7') = 6.3 Hz, 2H, **H**7'), 2.49 (t, <sup>3</sup>*J*(H4'-H3') = 7.0 Hz, 2H, **H**3'), 1.66 (apparent p, <sup>3</sup>*J*(H3'-H4', H5'-H4') = 7.3 Hz, 2H, **H**4'), 1.62 (apparent p, <sup>3</sup>*J*(H5'-H6', H6'-H7') = 6.4 Hz, 2H, **H**6'), 1.55 (m, 2H, **H**5') ppm.

<sup>13</sup>C{1H} NMR (150 MHz, CDCl<sub>3</sub>): *δ* 147.7 (C2), 132.2 (C3), 129.0 (C5), 118.0 (C4), 114.3 (C6), 109.0 (C1), 95.5 (C2'), 62.9 (C1'), 32.4 (C7'), 29.8 (C6'), 28.8 (C4'), 25.2 (C5'), 19.7 (C3') ppm.

#### **D.4** Synthesis of 2-(6-Hydroxy-6-methylhex-1-ynyl)aniline, 20S

Starting with 2-iodoaniline (1.0 g, 5.0 mmol) and 6-heptyn-2-ol (0.60 g, 5.0 mmol). Yield: 0.80 g, 81 %; yellow viscous oil;  $R_f = 0.16$  (Hexane : EtOAc = 1:3 (v/v)).

HRMS (ESI, MeOH): m/z: 226.1667. Calculated for  $[C_{13}H_{17}NO + Na]^+ = 226.1203$  amu.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.23 (dd, <sup>3</sup>J(H4-H3) = 7.8 Hz,  $^{4}J(H5-H3) = 1.5 \text{ Hz}$ , 1H, H3), 7.07 (apparent td,  $^{3}J(H6-H5) = ^{3}J(H4-H5) = 7.6 \text{ Hz}$ ,  $^{4}J(H3-H5) = 1.5$ Hz, 1H, H5), 6.68 (dd,  ${}^{4}J(H5-H6) = 7.6$  Hz,  ${}^{3}J(H4-H6) = 1.5$  Hz, 1H, H6), 6.66 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.68 (dd,  ${}^{4}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.68 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  Hz, 1H, H6), 6.88 (apparent td,  ${}^{3}J(H5-H6) = 1.5$  H  $H4) = 7.6 \text{ Hz}, {}^{3}J(H3-H4) = 7.8 \text{ Hz}, {}^{4}J(H6-H4) = 1.5 \text{ Hz}, 1H, H4), 3.85 (m, 1H, H6), 3.28 (br s, 3H, H4) = 7.6 Hz, 1H, H6), 3.85 (m, 1H,$ NH<sub>2</sub> & OH), 2.49 (t,  ${}^{3}J = 7.3$  Hz, 2H, H3'), 1.76-1.58 (m, 2H, H4' and H5'), 1.20 (d,  ${}^{3}J = 6.30$  Hz, 3H, CH<sub>3</sub>) ppm.

<sup>13</sup>C{1H} NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  147.7 (C2), 132.1 (C3), 129.0 (C5), 118.0 (C4), 114.3 (C6), 108.9 (C1), 94.5 (C2'), 77.4 (C1'), 67.7 (C6'), 38.5 (C5'), 25.2 (C4'), 23.7 (CH<sub>3</sub>), 19.7 (C3') ppm.

# Part E: General Catalytic Procedure, Time Course Profiles and NMR of Intermediates and Products

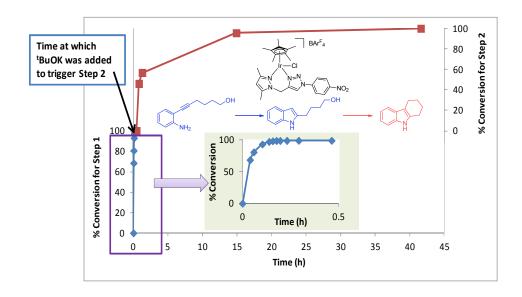
## **E.1** General Procedure for the Tandem Hydroamination/C-C Bond Formation Reactions

The metal catalysed tandem hydroamination/C-C bond formation reactions of the 2-(hydroxyalk-1-ynyl)aniline substrates (17S-20S) were performed on a small scale in J. Youngs NMR tubes. The metal complex (5.0 mol%) were weighed into the NMR tube prior to the addition of deuterated solvent (Toluene- $d_8$ , 0.6 mL) to the tube in a nitrogen filled glovebox. The substrate (concentration 0.17-0.19 M) was then added into the NMR tube. The tube was then removed from the glovebox and immediately placed into an acetone/liquid nitrogen slush-bath. The solution was briefly thawed before the NMR tube was placed into the probe of the NMR spectrometer at 100 °C. The temperature in the NMR magnet was calibrated with neat ethylene glycol using an Omega Microprocessor Thermometer (Model HH23). <sup>1</sup>H NMR spectra were recorded periodically and the products were identified by comparison with reported spectroscopic <sup>1</sup>H NMR data. Upon complete conversion of the 2-(hydroxyalk-1-ynyl)aniline substrates (17S-20S) to the 2-(hydroxylalkyl)indole intermediates (17I-20I) as observed by <sup>1</sup>H NMR spectroscopy, the base (<sup>t</sup>BuOK, KOH, DABCO or K[N(SiMe<sub>3</sub>)<sub>2</sub>], 1 equivalence), was added into the same NMR tube in the nitrogen filled glovebox. The tube was then removed from the glovebox, shaken and placed in a sonicator for 5 minutes. A <sup>1</sup>H NMR spectrum was obtained at 25 °C in the spectrometer before the NMR tube was placed in an oil bath at 110 °C. <sup>1</sup>H NMR spectra were obtained at regular time intervals at 25 °C in the spectrometer. The products were identified following reported <sup>1</sup>H NMR data.

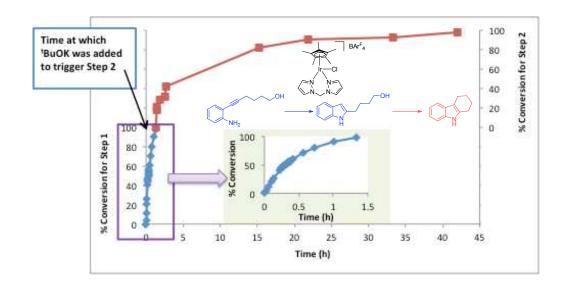
The rate of conversion for all catalytic reactions was determined by the relative integration of <sup>1</sup>H resonances of the product to the starting material in the 1H NMR spectrum. Complete conversion (> 98%) was taken to be the time where no remaining substrate resonances were observed in the <sup>1</sup>H NMR spectrum. The turnover frequency (TOF) was calculated at the point of 50 % conversion as the number of moles of products produced per mole of catalyst used per hour.

The catalysed C-N bond formation (1<sup>st</sup> step) was performed in the same fashion as the catalysed tandem reaction without the addition of base. The efficiency of the catalysed intramolecular C3 alkylation was conducted mainly with isolated 2-(4-hydroxybutyl)indole (21I), which was isolated from the large scale (ca. 0.30 g) catalysed cyclisation of 2-(6-hydroxyhex-1ynyl)aniline (21S) using [Rh(bpm)(CO)<sub>2</sub>]BAr<sup>F</sup><sub>4</sub> (16, bpm = bis(1-pyrazol-1-yl)methane) in toluene. The alkylation reaction was performed in the same way the 2nd step of the tandem C-N/C-C bond formation reaction.

# E.2 Time Course Profiles for Selected Tandem Hydroamination/C-C Bond Formation Reactions



**Figure S8**: Time course profile for [Ir(3c)(Cp\*)Cl]BAr<sup>F</sup><sub>4</sub> (7c) catalysed one-pot tandem C-N and C-C bond formation in the synthesis of 1,2,3,4-tetrahydrocarbazole (17P) from 2-(6-hydroxyhex-1-ynyl)aniline (17S).



**Figure S9**: Time course profile for [Ir(4)(Cp\*)Cl]BAr<sup>F</sup><sub>4</sub> (8) catalysed one-pot tandem C-N and C-C bond formation in the synthesis of 1,2,3,4-tetrahydrocarbazole (17P) from 2-(6-hydroxyhex-1-ynyl)aniline (17S).

## E.3 Typical Isolation Procedure for 2-(Hydroxyalkyl)indole Intermediates, 17I-20I

When the cyclisation of 17S, 18S and 20S to 17I, 18I, and 20I have respectively completed as observed by <sup>1</sup>H NMR spectroscopy, the contents of the NMR tube were poured into a small vessel and the NMR tube rinsed out with dichloromethane (*ca.* 3 x 0.5 mL). The solution was passed through a pad of silica (*ca.* 0.7 cm thick) which was then rinsed with CH<sub>2</sub>Cl<sub>2</sub>: Et<sub>2</sub>O = 1:1 (v/v, 5 mL). The solvent was removed under reduced pressure and the remaining viscous oil or solid dried in a vacuum desiccator to afford the 2-(hydroxylalkyl)indole 17I, 18I, and 20I. NMR data of 2-(5-hydroxypent-1-yl)indole, 19I was obtained without isolation upon completion of the cyclisation form 19S to 19I as observed by <sup>1</sup>H NMR spectroscopy.

#### 2-(4-Hydroxybut-1-yl)indole, 17I<sup>10</sup> **E.4**

Yellow brown solid.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.10 (br s, 1H, N**H**), 7.55 (d, <sup>3</sup>*J*(H5-H4) = 6.6 Hz, 1H, **H**4), 7.29 (d,  ${}^{3}J(\text{H6-H7})$  = 7.2 Hz, 1H, **H**7), 7.11

(m, 2H, H5 and H6), 6.25 (s, 1H, H3), 3.67 (t,  ${}^{3}J$  = 6.1 Hz, 2H, H4'), 2.76 (t,  ${}^{3}J$  = 7.4 Hz, 2H, H1'), 1.98 (br s, 1H, OH), 1.79 (apparent p,  ${}^{3}J = 7.0$  Hz, 2H, H2' or H3'), 1.63 (apparent p,  ${}^{3}J = 7.4$  Hz, 2H, H2') ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  139.7 (C2), 136.0 (C3a), 128.9 (C7a), 121.0 (C6), 119.8 (C4), 119.7 (C5), 110.5 (C7), 99.6 (C3), 62.6 (C4'), 32.1, 28.0, 25. 6 (last three C1', C2' and C3') ppm

#### 2-(3-Hvdroxyprop-1-vl)indole, 18I<sup>12</sup> E.5

5 4 3a 3 2' OH Fellow solid.

5 GC-MS (CI), m/z (%): 176 (76), [M+H]<sup>+</sup>.

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  8.17 (br s. 1H, N**H**), 7.52 (d. <sup>3</sup>*J*(H5-H4) = 7.5 Hz, 1H, H4), 7.30 (d,  ${}^{3}J(H6-H7) = 7.5$  Hz, 1H, H7), 7.12 (apparent td,  ${}^{3}J(H5-H6) = {}^{3}J(H7-H6)$ H6) = 7.5 Hz,  ${}^{4}J(H4-H6) = 1.5$  Hz, 1H, H6), 7.06 (apparent td,  ${}^{3}J(H6-H5) = {}^{3}J(H4-H5) = 7.5$  Hz,  $^{4}J(H7-H5) = 1.5 \text{ Hz}, 1H, H5), 6.25 \text{ (s, } 1H, H3), 3.75 \text{ (t, } ^{3}J(H2'-H3') = 6.0 \text{ Hz, } 2H, H3'), 2.90 \text{ (t, } ^{3}J(H2'-H3') = 6.0 \text{ (t, } ^{3}J(H2'-H3')$  $^{3}J(H2'-H1') = 7.2 \text{ Hz}, 2H, H1'), 1.98 (tt. <math>^{3}J(H3'-H2') = 6.0 \text{ Hz}, ^{3}J(H1'-H2') = 7.2 \text{ Hz}, 2H, H2'),$ 1.45 (br s, 1H, OH) ppm.

<sup>13</sup>C{1H} NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  138.9 (C2), 135.9 (C3a), 128.7 (C7a), 121.0 (C6), 119.7 (C4), 119.5 (C5), 110.3 (C7), 99.6 (C3), 62.0 (C3'), 31.7 (C2'), 24.6 (C1') ppm.

#### 2-[(Z)-(Dihydrofuran-2(3H)-yliden)methyl]benzenamine, I1 (from 18S) E.5'

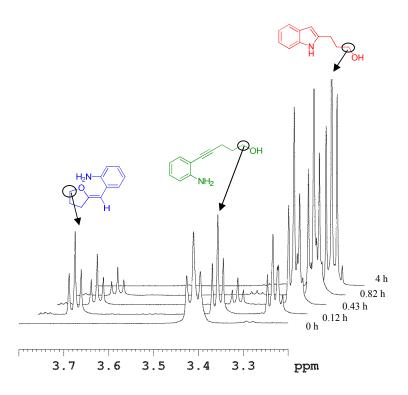
**I1** (from **18S**)

The intermediate I1 (from 18S) was observed following the catalysed cyclisation of 2-(5-hydroxypent-1-ynyl)aniline, 18S, when the reaction was conducted at 60 °C. NMR assignments were determined using 2D NMR

techniques. The NMR spectral data was found to be similar to the data reported for an analogous compound, (Z)-2-benzylidenetetrahydrofuran.<sup>13</sup>

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  7.52 (dd, <sup>3</sup>*J*(H4'-H3') = 7.8 Hz, <sup>4</sup>*J*(H5'-H3') = 1.2 Hz, 1H, **H**3'), 7.01 (apparent td, <sup>3</sup>*J*(H6'-H5') = <sup>3</sup>*J*(H4'-H5') = 7.8 Hz, <sup>4</sup>*J*(H3'-H5') = 1.2 Hz, 1H, **H**5'), 6.80 (apparent td, <sup>3</sup>*J*(H5'-H4') = <sup>3</sup>*J*(H3'-H4') = 7.8 Hz, <sup>4</sup>*J*(H6'-H4') = 1.2 Hz, 1H, **H**4'), 6.70 (dd, <sup>3</sup>*J*(H5'-H6') = 7.8 Hz, <sup>4</sup>*J*(H4'-H6') = 1.2 Hz, 1H, **H**6'), 5.22 (s, 1H, **H**1''), 4.28 (t, <sup>3</sup>*J*(H5-H4) = 6.7 Hz, 2H, **H**4), 3.79 (very br, 2H, N**H**<sub>2</sub>), 2.75 (td, <sup>3</sup>*J*(H4-H3) = 7.2 Hz), <sup>4</sup>*J*(H1''-H3) = 1.5 Hz, 2H, **H**<sub>3</sub>), 2.05 (apparent p, <sup>3</sup>*J*(H4-H5) = <sup>3</sup>*J*(H3-H5) = 7.2Hz, 2H, H5) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 75 MHz): δ 156.8 (C2), 142.7 (C1'), 129.2 (C3'), 126.3 (C2'), 122.6 (C5'), 118.8 (C4'), 116.1 (C6'), 91.7 (C1''), 71.9 (C5), 30.6 (C3), 24.5 (C4) ppm.



**Figure S10**: <sup>1</sup>H NMR stacked-plot showing the formation and disappearance of **I1** (from **18S**) in the catalysed cyclisation of 2-(5- hydroxy-1-pentyn-1-yl)aniline, **18S** using [Ir(**4**)Cp\*Cl]BAr<sup>F</sup><sub>4</sub> (**8**) in toluene- $d_8$  at 60 °C.

### E.6 2-(5-Hydroxypent-1-yl)indole, 19I

**H**1'), 1.44 (apparent p,  ${}^{3}J$  = 7.6 Hz, 2H, **H**2'), 1.34 (apparent p and br s overlap,  ${}^{3}J$  = 6.2 Hz, 3H, **H**4' and O**H**), 1.23 (m, 2H, **H**3') ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, Toluene- $d_8$ ): δ 139.5 (C2), 136.6 (C3a), 129.6 (C7a), 121.1 (C4), 120.2 (C6), 119.9 (C1), 110.7 (C2'), 99.8 (C1'), 62.5 (C7'), 32.7 (C6'), 29.4 (C4'), 28.4 (C5'), 25.8 (C3') ppm.

## E.7 2-(4-Hydroxy-4-methylbut-1-yl)indole, 20I

Yellow-brown oil.

HR-MS (ESI<sup>+</sup>, MeOH): m/z (%): 226.1214 (18 %) [M+Na]<sup>+</sup>

(Calculated [M+Na]<sup>+</sup> = 226.1202) amu.

<sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 8.11 (br s, 1H, N**H**), 7.55 ( ${}^{3}J$ (H5-H4) = 7.7 Hz, 1H, **H**4), 7.30 (d,  ${}^{3}J$ (H6-H7) = 7.9 Hz, 1H, **H**7), 7.14 (apparent td,  ${}^{3}J$ (H5-H6) =  ${}^{3}J$ (H7-H6) = 7.2 Hz,  ${}^{4}J$ (H4-H6) = 1.0 Hz, 1H, **H**6), 7.09 (apparent td,  ${}^{3}J$ (H6-H5) =  ${}^{3}J$ (H4-H5) = 7.2 Hz,  ${}^{4}J$ (H7-H5) = 1.0 Hz, 1H, **H**5), 6.25 (s, 1H, **H**3), 3.85 (apparent sex,  ${}^{3}J$  = 6.1 Hz, 1H, **H**4'), 2.76 (td,  ${}^{3}J$  = 7.5 Hz,  ${}^{4}J$  = 3.7 Hz, 2H, **H**1'), 1.98 (br s, 1H, O**H**), 1.86 (apparent sep, J = 7.6 Hz, 1H, **H**3'a or **H**3'b), 1.76 (apparent sep, J = 7.5 Hz, 1H, **H**3'a or **H**3'b), 1.53 (apparent q,  ${}^{3}J$  = 7.3 Hz, 2H, **H**2'), 1.21 (d,  ${}^{3}J$  = 6.2 Hz, 3H, C**H**<sub>3</sub>) ppm.

<sup>13</sup>C{<sup>1</sup>H} NMR (150 MHz, CDCl<sub>3</sub>): *δ* 139.7 (C2), 136.0 (C3a), 128.9 (C7a), 121.0 (C6), 119.8 (C4), 119.7 (C5), 110.5 (C7), 99.6 (C3), 68.0 (C4'), 38.6 (C2'), 28.2 (C1'), 25.5 (C3'), 23.8 (CH<sub>3</sub>) ppm.

### **E.8** NMR Data for Final Products

The formation of products 17P-20P were confirmed by comparing the <sup>1</sup>H NMR spectra obtained

with that reported in the literature and also by comparing the NMR spectra during the catalytic reaction with the spectra of authentic samples purchased from Aldrich in case of product 17P, 18P.

## E.9 1,2,3,4-Tetrahydrocarbazole, 17P 14 15

<sup>1</sup>H NMR (400 MHz, toluene- $d_8$ ):  $\delta$  7.47 (dd, <sup>3</sup>J = 7.5 Hz, <sup>4</sup>J = 1.3 Hz, 1H, **H**4 or **H**7), 7.15 (apparent p of d, <sup>3</sup>J = 7.2 Hz, <sup>4</sup>J = 1.5 Hz, 2H, **H**5 and **H**6), 7.01 (dd, <sup>3</sup>J = 6.9 Hz, <sup>4</sup>J = 1.3 Hz, 1H, **H**7 or **H**4), 6.27 (br s, 1H, N**H**), 2.59 (m, 2H, **H**1' or **H**4'), 2.27 (m, 2H, **H**1' or **H**4'), 1.67 (m, 4H, **H**2' and **H**3') ppm.

<sup>13</sup>C NMR (100 MHz, toluene- $d_8$ ): δ 136.3, 133.4, 128.6, 121.1, 119.3, 118.1, 110.6, 110.0, 23.8, 23.62 23.3, 21.26 ppm.

## E.10 1,2,3,4-Tetrahydrocyclopent[b] indole, 18P <sup>16</sup>

<sup>3'</sup>
<sup>1</sup>
H NMR (400 MHz, Toluene- $d_8$ ,):  $\delta$  7.47 (m, 1H, H4 or H7), 7.12 (apparent s,  $^3J = 5.9$  Hz,  $^4J = 2.7$  Hz, 2H, H5 and H6), 7.00 (m, 1H, H4 or H7), 6.38 (br s, 1H, NH), 2.71 (apparent t,  $^3J = 6.9$  Hz, 2H, H1' or H3'), 2.44 (apparent t,  $^3J = 7.0$  Hz, 2H, H1' or H3'), 2.27 (apparent p,  $^3J = 7.0$  Hz, 2H, H2') ppm.

<sup>13</sup>C NMR (100 MHz, Toluene- $d_8$ ,): δ 143.1, 141.7, 125.4, 120.7, 119.7, 119.6, 118.9, 111.5, 28.9, 25.9, 24.7 ppm.

# E.11 5,6,7,8,9,10-Hexahydrocyclohept[b]indole,19P<sup>16,14</sup>

<sup>13</sup>C{<sup>1</sup>H} NMR (150 MHz, CDCl<sub>3</sub>): δ 137.5, 134.3, 129.3, 120.5, 118.9, 117.6, 113.6, 110.2, 31.8, 29.6, 28.8, 27.6, 24.7 ppm.

## E.12 4-Methyl-1,2,3,4-tetrahydrocarbazole, 20P<sup>14</sup>

<sup>1</sup>H NMR (400 MHz, toluene- $d_8$ ):  $\delta$  7.57 (d,  $^3J = 7.0$  Hz, 1H, **H**4 or **H**7), 7.14

5 43a3 2 2 6 77a N1 1'

(apparent p of d,  ${}^{3}J = 7.2$  Hz,  ${}^{4}J = 1.2$  Hz, 2H, **H**5 and **H**6), 7.01 (dd,  ${}^{3}J = 6.9$ 

Hz,  ${}^{4}J = 1.0$  Hz, 1H, **H**7 or **H**4),  ${}^{17}6.35$  (br s, 1H, N**H**), 2.99 (m, 1H, **H**4'),

2.26 (m, 2H, H1' or H2'), 1.79 (m, 2H, H2' or H1'), 1.59 (m, 1H, H3'a or

**H**3'b), 1.41 (m, 1H, **H**3'a or **H**3'b), 1.31 (d,  ${}^{2}J$  = 6.9 Hz, 3H, C**H**<sub>3</sub>) ppm.

<sup>13</sup>C NMR (100 MHz, toluene- $d_8$ ): δ 136.5, 133.1, 128.1, 121.0, 119.3, 119.1, 114.9, 110.7, 32.6, 31.2, 27.6, 23.6, 20.9 ppm.

## References

- 1. Hua, C.; Vuong, K. Q.; Bhadbhade, M.; Messerle, B. A., *Organometallics* **2012**, *31* (5), 1790.
- 2. Dabb, S. L.; Ho, J. H. H.; Hodgson, R.; Messerle, B. A.; Wagler, J., *Dalton Trans.* **2009**, (4), 634.
- 3. Bruker, SADABS, Bruker AXS Inc., Madison, Wisconsin, USA. 2001.
- 4. Bruker, *APEX2 and SAINT*, Bruker AXS Inc., Madison, Wisconsin, USA. 2001.
- 5. Sheldrick, G. M., *Acta. Cryst.* **2008**, *A64*, 112.
- 6. Macrae, C. F. B., I. J.; Chisholm, J. A.; Edgington, P. R.; McCabe, P.; Pidcock, E.; Rodriguez-Monge, L.; Taylor, R.; van de Streek, J. and Wood, P. A., *J. Appl. Cryst.* **2008**, *41*, 466.
- 7. Le Drian, C.; Greene, A. E., J. Am. Chem. Soc. **1982**, 104 (20), 5473.
- 8. Li, M. S.; O'doherty, G. A., *Org. Lett.* **2006**, *8* (26), 6087.
- 9. Brandsma, L. V., S. F.; Verkjuijsse, H. D., *Application of Transition Metal Catalysts in Organic Synthesis*. 1st ed.; Springer: New York, 1998.
- 10. Sakai, N.; Annaka, K.; Fujita, A.; Sato, A.; Konakahara, T., J. Org. Chem. 2008, 73 (11), 4160.
- 11. Kawato, H. C.; Nakayama, K.; Inagaki, H.; Ohta, T., Org. Lett. 2001, 3 (22), 3451.
- 12. Smith, A. B.; Visnick, M.; Haseltine, J. N.; Sprengeler, P. A., *Tetrahedron* **1986**, *42* (11), 2957.
- 13. Ley, S. V.; Lygo, B.; Organ, H. M.; Wonnacott, A., *Tetrahedron* **1985**, *41* (18), 3825.
- 14. Sun, K.; Liu, S.; Bec, P. M.; Driver, T. G., Angew. Chem. Int. Ed. 2011, 50 (7), 1702.
- 15. Scott, T. L.; Burke, N.; Carrero-Martinez, G.; Soderberg, B. C. G., *Tetrahedron* **2007**, *63* (5), 1183.
- 16. Banwell, M. G.; Kelly, B. D.; Kokas, O. J.; Lupton, D. W., Org. Lett. 2003, 5 (14), 2497.
- 17. Burling, S.; Field, L. D.; Messerle, B. A.; Turner, P., Organometallics 2004, 23 (8), 1714.