

## Supporting Information for

# Hydrogenated ZnO Core-Shell Nanocables for Flexible Supercapacitors and Self-Powered Systems

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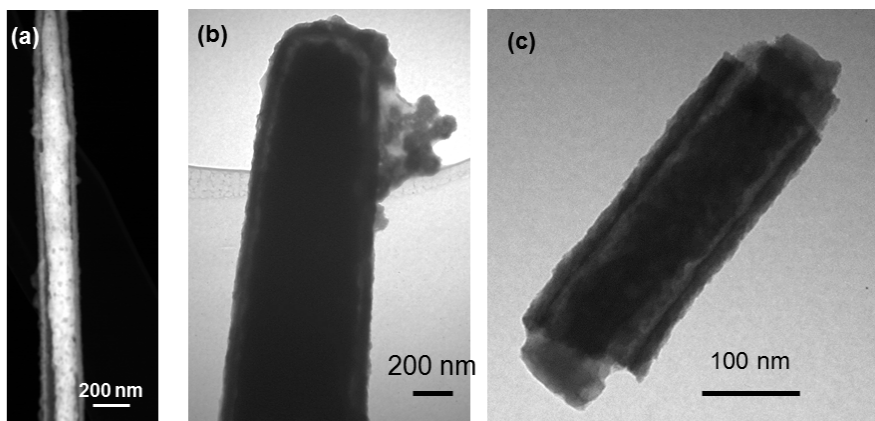
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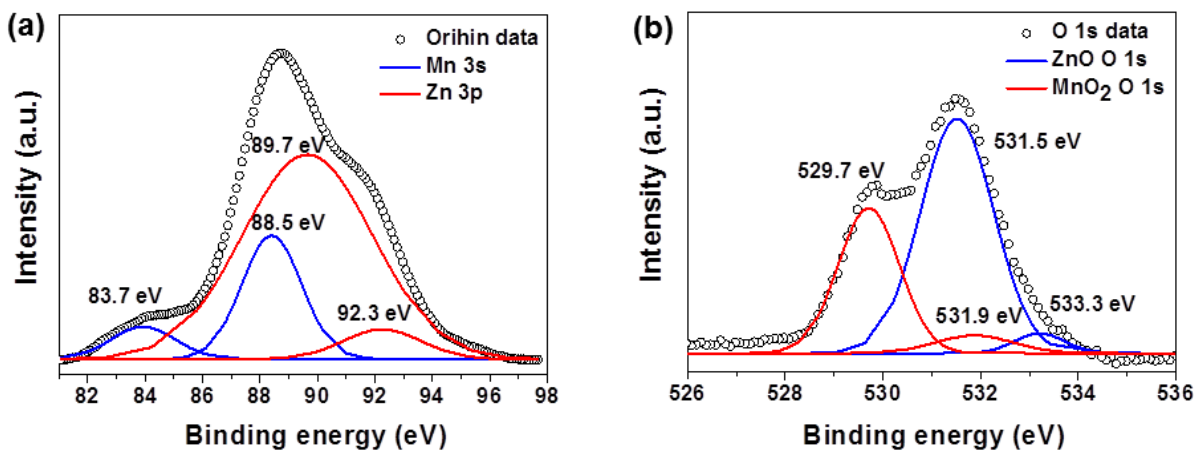
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## Assembly of the DSSCs

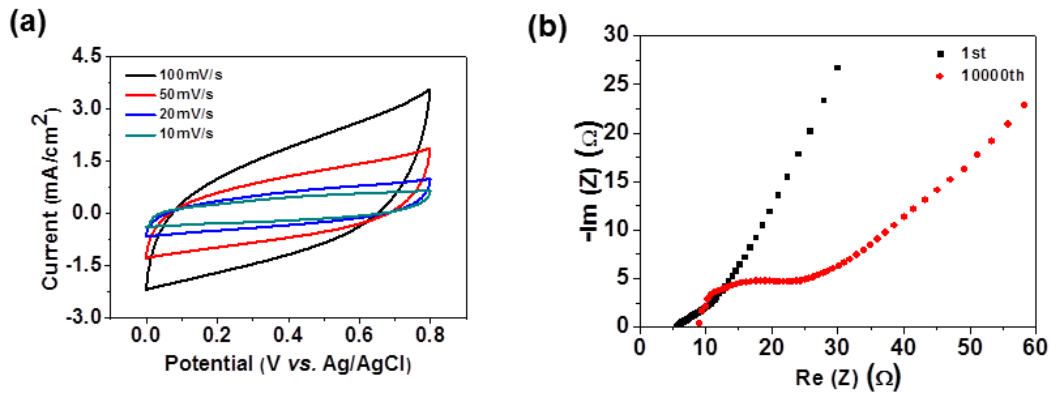
0.1 g P25 TiO<sub>2</sub> nanoparticles were continuously mixed with 0.3 mL deionized water. The resulting colloidal dispersions were coated onto indium tin oxide (ITO) glass to form TiO<sub>2</sub> film. Dye sensitization was continued by immersing the fabricated TiO<sub>2</sub> photoanodes in 0.5 mM N719 ruthenium dye ethanol solution for 12 hours at room temperature in a sealed beaker, and then the sensitized film was placed in open air to evaporate the alcohol. The DSSC was assembled in a typical sandwich-type cell with a platinum-coated ITO as a counter anode. The electrolyte solution was injected into the assembled DSSC from the edges by capillary forces. The effective area of a single DSSC is 0.25 cm<sup>2</sup>.



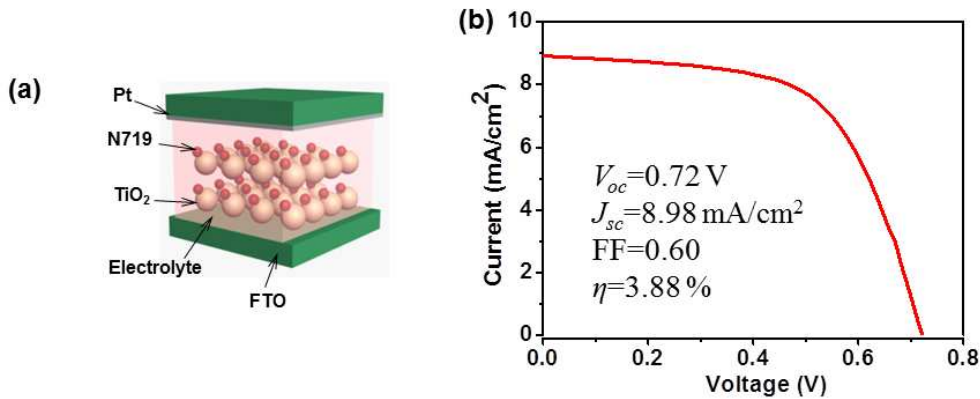
**Figure S1.** (a) STEM image of HZM core/shell nanostructure. TEM images of the (b) HZC nanocable and (c) HZM nanocable. All suggested core/shell structures.



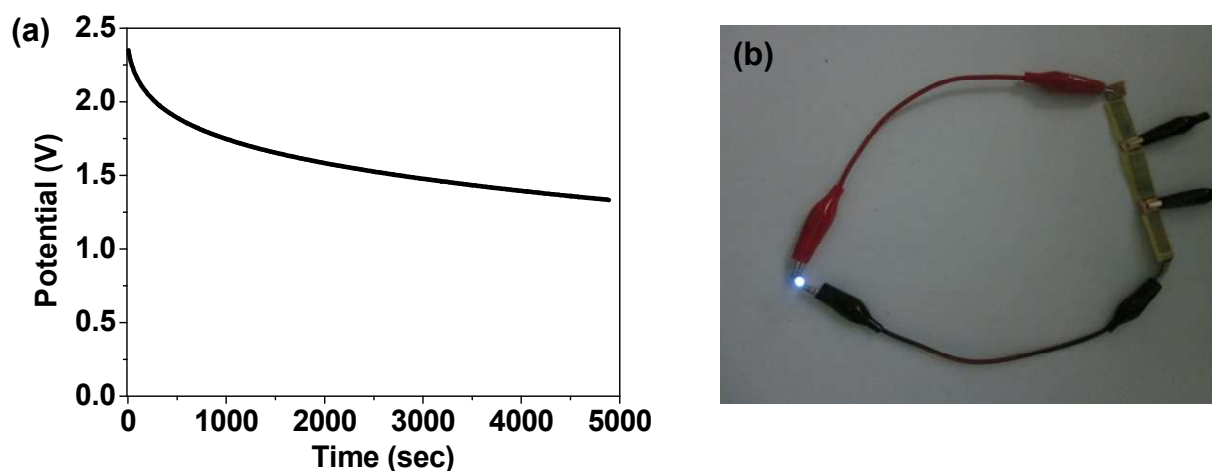
**Figure S2.** XPS spectra of (a) Mn 3s and Zn 3p and (c) O 1s collected from the HZM core/shell nanocable. ZnO had significant signal interference both in Mn 3s and O 1s, therefore they were not be used to determine valence state of Mn.



**Figure S3.** (a) CV curves collected for the HZM SC device at different scan rate. (b) Nyquist plots for the all solid-state HZM SC with an area of 1 cm<sup>2</sup> after 1st and 10000th cycle, respectively.



**Figure S4.** (a) A schematic diagram of a DSSC. (b) Typical current density/voltage curve of a home-made DSSC under the illumination of AM 1.5. The effective area and maximum power output of a DSSC device was about 0.25 cm<sup>2</sup> and 0.97 mW, respectively.



**Figure S5.** (a) Self-discharge curve of three series-wound SCs. (b) A photo showing an LED powered by three series-wound SCs, which were previously charged by four series-wound DSSCs.

**Table S1.** Electrical and electrochemical data for ZnO, AZnO, and HZnO electrodes. Carrier concentration was calculated from M-S plots. Areal capacitance was obtained under a discharge current of  $0.05 \text{ mA/cm}^2$ .

	HZnO	AZnO	ZnO
Carrier concentration ( $\text{cm}^{-3}$ )	$2.65 \times 10^{20}$	$1.86 \times 10^{18}$	$7.04 \times 10^{17}$
Areal capacitance ( $\text{mF/cm}^2$ )	1.38	0.46	0.27
$R_e$ ( $\Omega$ )	2.22	3.83	3.77

**Table S2.** Electrical and electrochemical data for HZM, AZM and ZM electrodes and the HZM-based SC device. Areal capacitance of single electrode was obtained under a discharge current of 1 mA/cm<sup>2</sup>, while 0.5 mA/cm<sup>2</sup> for the SC device.

	HZM	AZM	ZM	Device
Areal capacitance (mF/cm <sup>2</sup> )	138.7	88.2	49.3	23.47
R <sub>ct</sub> (Ω)	0.42	0.86	1.2	2.02
R <sub>s</sub> (Ω)	2.74	2.94	3.8	5.74

**Video S1.** A red LED was lighted up by our 3 series-wound SCs, which were previously charged by our DSSCs. This video was speeded up 10 times faster, showing the first 100 seconds of the process.