

## Supplementary Information

### Pyrrole-Fused Azacoronene Family: The Influence of Replacement with Dialkoxybenzenes on the Optical and Electronic Properties in Neutral and Oxidized States

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## S1. Materials and methods

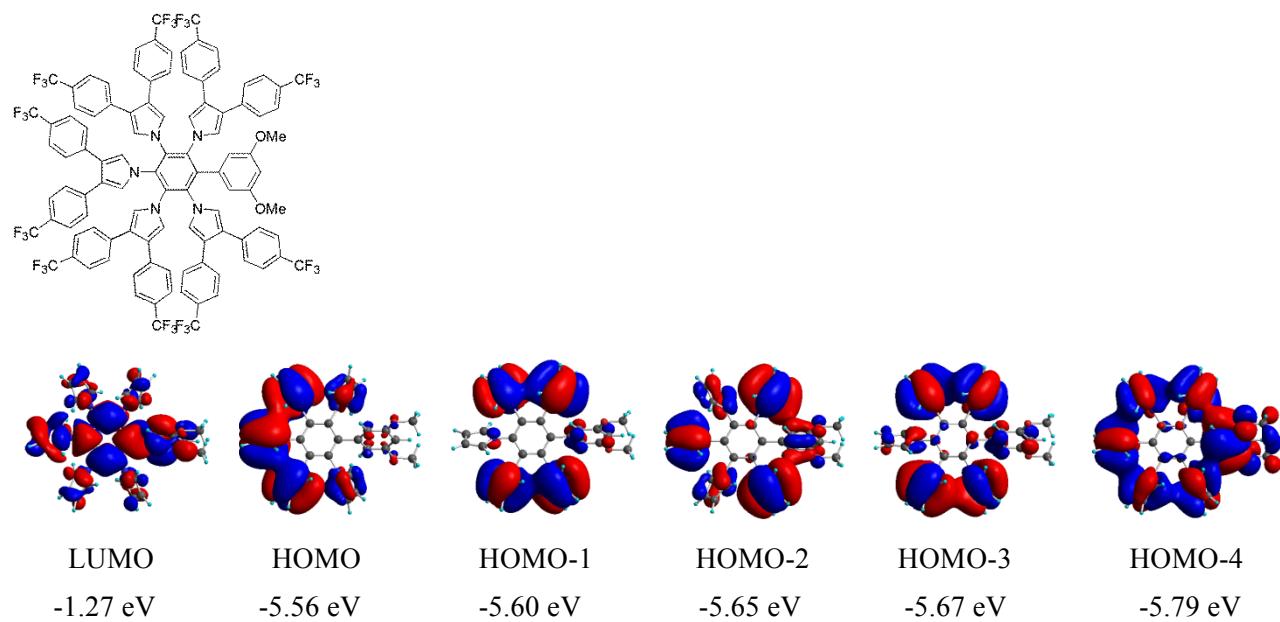
<sup>1</sup>H-, <sup>13</sup>C- and <sup>19</sup>F-NMR spectra were recorded on a Bruker AVANCE III 500 spectrometer operating at 500.13 MHz for <sup>1</sup>H, 125.76 MHz for <sup>13</sup>C, and 470.59 MHz for <sup>19</sup>F with use of tetramethylsilane (0 ppm) or residual solvent for <sup>1</sup>H (5.31 ppm for CD<sub>2</sub>Cl<sub>2</sub>) and <sup>13</sup>C (77.01 ppm for CDCl<sub>3</sub>, 53.85 ppm for CD<sub>2</sub>Cl<sub>2</sub>) signals, and hexafluorobenzene for <sup>19</sup>F signal (-164.9 ppm) as an internal standard, and if not otherwise noted, all spectra measured at 298K. Electron impact (EI) mass spectra and atmospheric pressure chemical ionization (APCI) mass spectra were obtained on SHIMADZU GC-MS QP2020 and Bruker MicrOTOF II, respectively. Melting points were determined with Yanako MP-500D and not corrected. Thermogravimetric analysis (TGA) was carried out with Rigaku DSC8230L under nitrogen atmosphere. Elemental analysis was performed with Exeter Analytical, Inc. CE-440F in the microanalysis laboratory of Tokyo Metropolitan University. Cyclic voltammetry (CV) and differential pulse voltammetry (DPV) measurements were performed on BAS-ALS620B electrochemical analyzer using a standard three-electrode cell consisting of a Pt working electrode, a Pt wire counter electrode, and a Ag/AgNO<sub>3</sub> reference electrode under argon atmosphere. The potentials were calibrated with ferrocene as an external standard. ESR spectra were recorded on JEOL JES-RE3X instruments. Magnetic measurement was carried out on a Quantum Design SQUID MPMS-XL magnetometer on powder samples. Electronic spectra were recorded on SHIMADZU UV-Vis-NIR scanning spectrophotometer (Model UV-3101-PC). Fluorescence and phosphorescence spectra were measured with JASCO FP-6500 spectrofluorometer. Fluorescence quantum yield was determined by comparison with quinine sulfate in 0.5 M H<sub>2</sub>SO<sub>4</sub> solution ( $\Phi_F = 0.51$ ). The calculation of the fluorescence quantum yield of a sample solution ( $\Phi_s$ ), relative to the reference sample of known quantum yield ( $\Phi_r$ ), is given by  $\Phi_s = \Phi_r(A_r/A_s)(I_s/I_r)(n_s/n_r)^2$  where A<sub>s</sub> and A<sub>r</sub> are the absorbances of the sample and the reference solutions, I<sub>s</sub> and I<sub>r</sub> are the corresponding relative integrated fluorescence intensities, and n<sub>s</sub> and n<sub>r</sub> are the measured refractive indexes of solvents (Parker-Ress method). Fluorescence and phosphorescence lifetimes were measured by time-correlated single-photon counting (TCSPC) and multi-channel scaling (MCS) techniques, respectively. These measurements were carried out with HORIBA Jobin Yvon photon counting system (FluoroCube). The excitation wavelength was 375 nm (fluorescence, TCSPC) and 370 nm (phosphorescence, MCS). The total instrument response function has an FWHM of ca. 200 ps for the fluorescence measurements.

All reactions were carried out under nitrogen atmosphere. THF was freshly distilled from sodium benzophenone ketyl before use, and other solvents were purified with standard methods. Column chromatography was carried out using Daiso silica gel 1001W. *N*-Benzenesulfonyl-3,4-dibromopyrrole<sup>[1]</sup> 3,5-didodecyloxy-bromobenzene<sup>[2]</sup> and 3,5-dibuthoxy-bromobenzene<sup>[3]</sup> were synthesized according to the literature procedures.

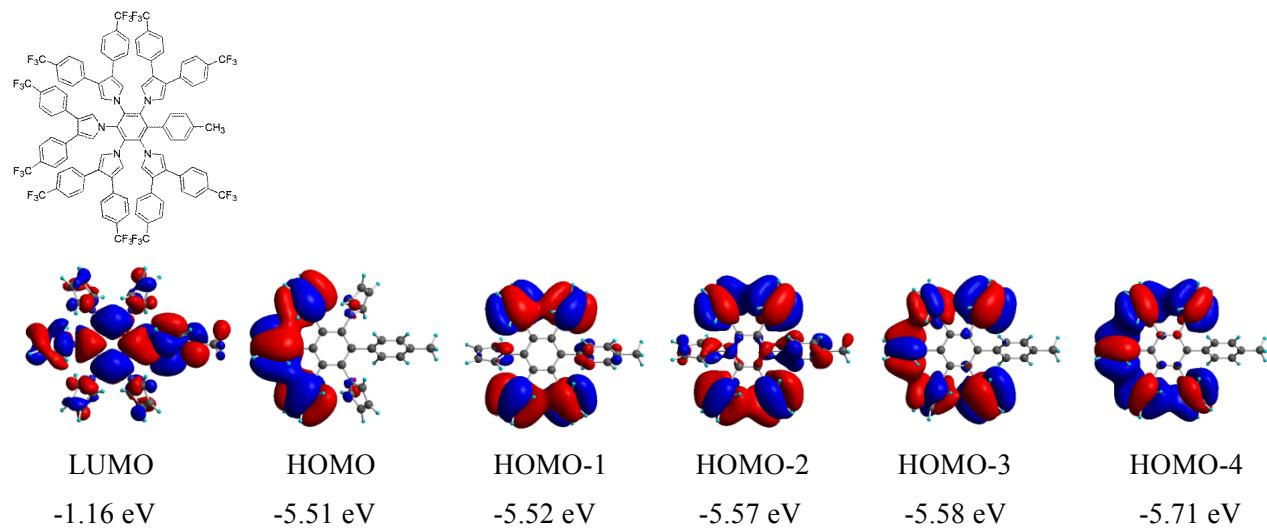
## S2. Molecular orbitals for **9** and **10**

The GAUSSIAN 09<sup>[4]</sup> series of programs was used for all calculations. All molecules were fully optimized using the hybrid density functional at B3LYP level of theory with the 6-31G(d) basis set. Frequency calculations were conducted to ensure that these structures were indeed local minima.

### 1. Molecular orbitals for **9c**

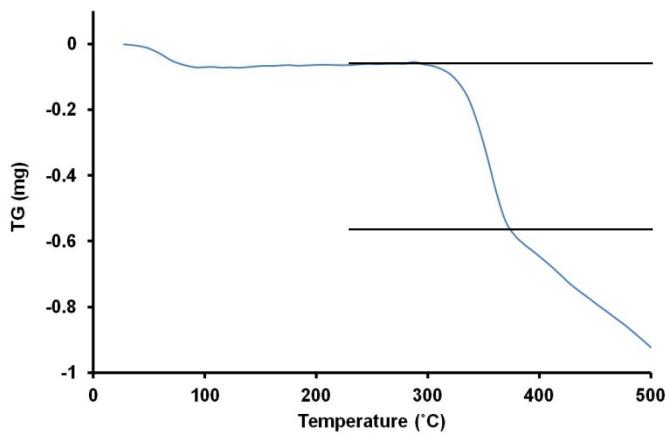


### 2. Molecular orbitals for **10**



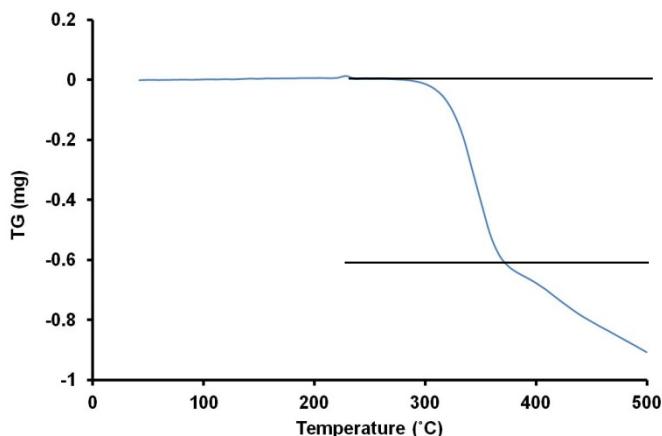
### S3. TGA analyses of 2a, 3a and 4a

#### 1. TGA analysis of 2a



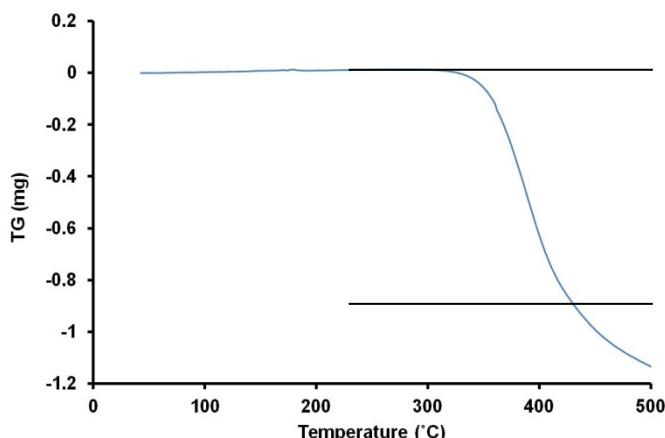
Sample weight: 3.162 mg  
Weight loss: 0.474 mg  
Losing weight%: 15.0%  
Percentages of alkyl chains weight of whole molecule: 14.9%

#### 2. TGA analysis of 3a



Sample weight: 2.143 mg  
Weight loss: 0.585 mg  
Losing weight%: 27.3%  
Percentages of alkyl chains weight of whole molecule: 28.6%

#### 3. TGA analysis of 4a

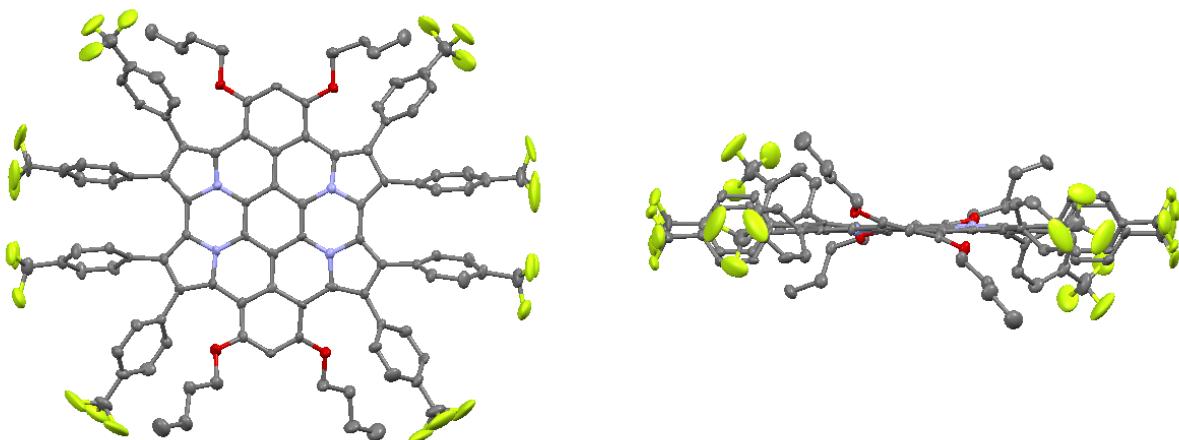


Sample weight: 2.497 mg  
Weight loss: 0.887 mg  
Losing weight%: 35.5 %  
Percentage of alkyl chains weight of whole molecule: 41.3%

#### S4. X-ray crystal structures of **3b**, **4a**, and **12b**

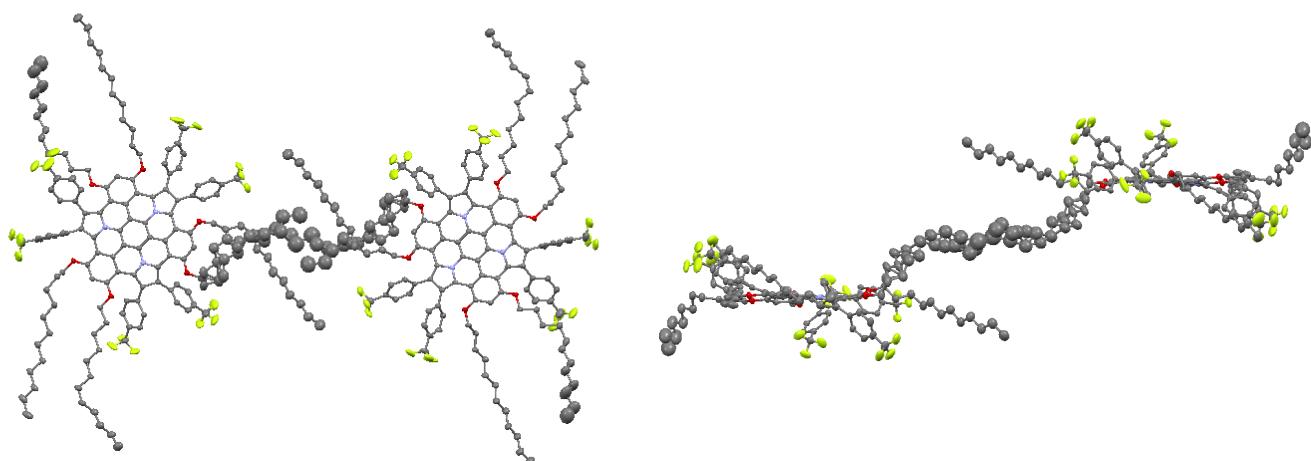
X-ray data were taken on Bruker SMART APEX and SMART APEXII ULTRA diffractometer equipped with a CCD area detector with monochromated Mo-K<sub>α</sub> irradiation ( $\lambda = 0.71073 \text{ \AA}$ ). The structures were solved by direct methods (SHELXS) and refined by the full-matrix least-squares method on  $F^2$  (SHELXL-97). Non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed using AFIX instructions.

**Table S4a.** Crystal data and structure refinement for **3b**.



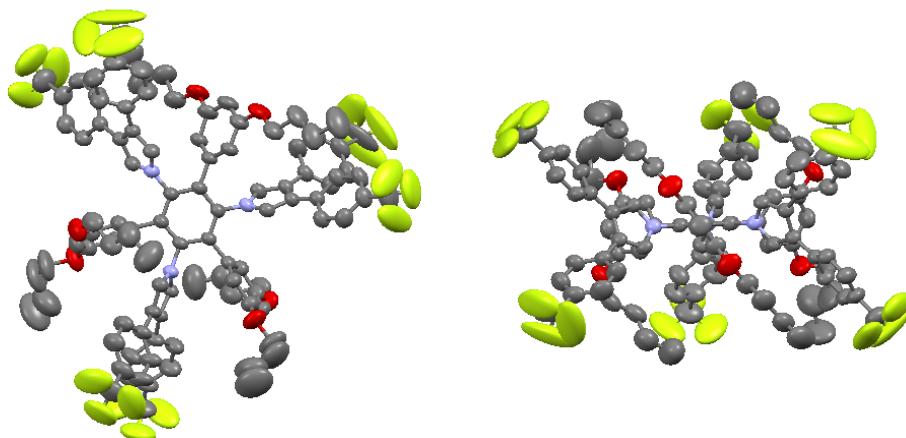
	<b>3b</b>
Identification code	
Empirical formula	$C_{106} H_{70} F_{24} N_4 O_4$
Formula weight	1919.66
Temperature	123 K
Wavelength	$0.71073 \text{ \AA}$
Crystal system	monoclinic
Space group	C2/c
Unit cell dimensions	$a = 18.648(4) \text{ \AA}$ $b = 38.150(8) \text{ \AA}$ $c = 15.294(3) \text{ \AA}$ $\alpha = 90.00^\circ$ . $\beta = 125.87(3)^\circ$ . $\gamma = 90.00^\circ$ .
Volume	$8817(3) \text{ \AA}^3$
Z	4
Density (calculated)	$1.446 \text{ Mg/m}^3$
Absorption coefficient	$0.123 \text{ mm}^{-1}$
F(000)	3928
Crystal size	$0.30 \times 0.20 \times 0.02 \text{ mm}^3$
Theta range for data collection	$1.96 \text{ to } 23.31^\circ$
Index ranges	$-20 \leq h \leq 20, -42 \leq k \leq 37, -15 \leq l \leq 17$
Reflections collected	19072
Independent reflections	6371 [ $R(\text{int}) = 0.1466$ ]
Completeness to theta = $23.31^\circ$	99.7 %
Absorption correction	Empirical
Refinement method	Full-matrix least-squares on $F^2$
Data / restraints / parameters	6371 / 72 / 683
Goodness-of-fit on $F^2$	0.905
Final R indices [ $I > 2\sigma(I)$ ]	$R_1 = 0.0665, wR_2 = 0.1457$
R indices (all data)	$R_1 = 0.1743, wR_2 = 0.1915$
Largest diff. peak and hole	0.515 and -0.304 e. $\text{\AA}^{-3}$

**Table S4b.** Crystal data and structure refinement for **4a**.



	<b>4a</b>	
Identification code		
Empirical formula	$C_{150} H_{177} F_{18} N_3 O_6$	
Formula weight	2459.95	
Temperature	93 K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	$a = 17.1224(17)$ Å	$\alpha = 102.3660(10)^\circ$ .
	$b = 18.0032(17)$ Å	$\beta = 93.7450(10)^\circ$ .
	$c = 23.929(2)$ Å	$\gamma = 113.2540(10)^\circ$ .
Volume	6526.8(11) Å <sup>3</sup>	
Z	2	
Density (calculated)	1.252 Mg/m <sup>3</sup>	
Absorption coefficient	0.092 mm <sup>-1</sup>	
F(000)	2616	
Crystal size	0.42 x 0.10 x 0.05 mm <sup>3</sup>	
Theta range for data collection	2.14 to 25.03°	
Index ranges	$-20 \leq h \leq 16, -20 \leq k \leq 21, -26 \leq l \leq 28$	
Reflections collected	31195	
Independent reflections	22605 [R(int) = 0.0309]	
Completeness to theta = 25.03°	98.1 %	
Absorption correction	Multi-scan	
Refinement method	Full-matrix least-squares on $F^2$	
Data / restraints / parameters	22605 / 3416 / 1850	
Goodness-of-fit on $F^2$	1.040	
Final R indices [ $I > 2\sigma(I)$ ]	R1 = 0.0843, wR2 = 0.2319	
R indices (all data)	R1 = 0.1507, wR2 = 0.2790	
Largest diff. peak and hole	0.995 and -0.611 e.Å <sup>-3</sup>	

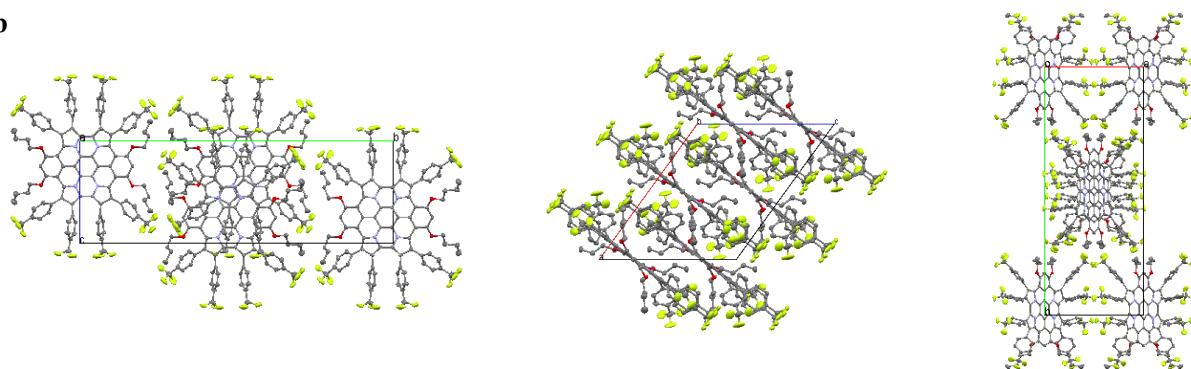
**Table S4c.** Crystal data and structure refinement for **12b**.



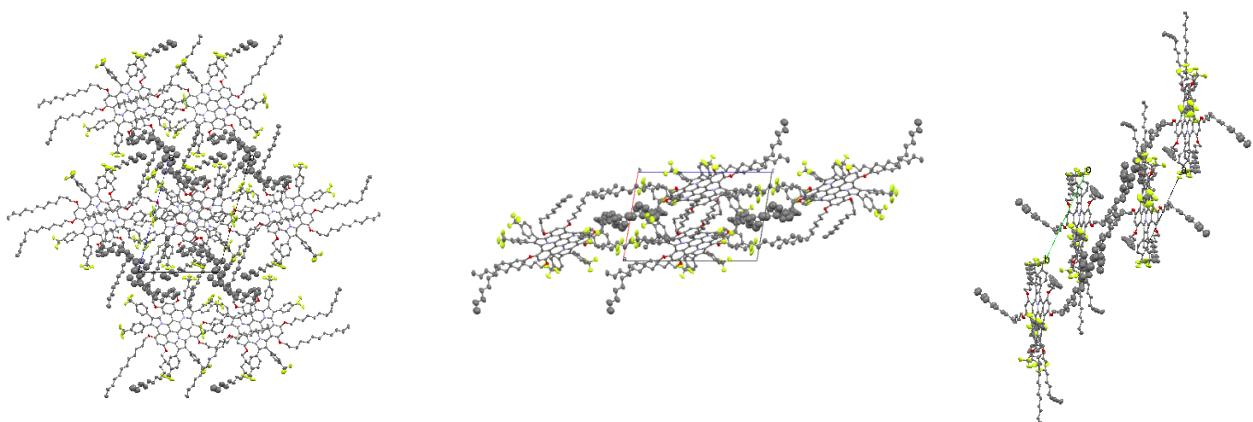
	<b>12b</b>
Identification code	
Empirical formula	C <sub>102</sub> H <sub>93</sub> F <sub>18</sub> N <sub>3</sub> O <sub>6</sub>
Formula weight	1798.79
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system	Triclinic
Space group	P-1
Unit cell dimensions	a = 16.880(3) Å $\alpha$ = 99.12(3) $^\circ$ . b = 16.898(3) Å $\beta$ = 104.64(3) $^\circ$ . c = 20.940(4) Å $\gamma$ = 118.83(3) $^\circ$ .
Volume	4778(3) Å <sup>3</sup>
Z	2
Density (calculated)	1.250 Mg/m <sup>3</sup>
Absorption coefficient	0.101 mm <sup>-1</sup>
F(000)	1872
Crystal size	0.20 x 0.10 x 0.08 mm <sup>3</sup>
Theta range for data collection	1.06 to 23.43 $^\circ$ .
Index ranges	-18 <= h <= 18, -16 <= k <= 18, -23 <= l <= 12
Reflections collected	20809
Independent reflections	13719 [R(int) = 0.0222]
Completeness to theta = 23.43 $^\circ$	97.9 %
Absorption correction	Empirical
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	13719 / 0 / 1168
Goodness-of-fit on F <sup>2</sup>	1.043
Final R indices [I>2sigma(I)]	R1 = 0.0784, wR2 = 0.2253
R indices (all data)	R1 = 0.1333, wR2 = 0.2708
Largest diff. peak and hole	0.436 and -0.370 e.Å <sup>-3</sup>

## S5. Packing structures of 3b and 4a

3b



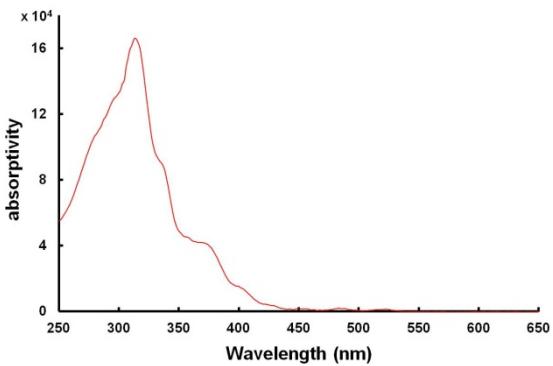
4a



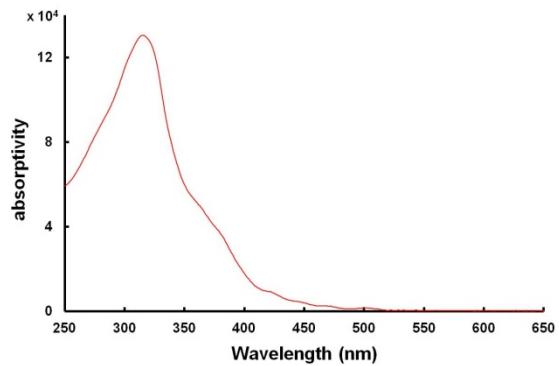
## S6. Absorption spectra of 1a–4a with absorptivity

Absorption spectra were all measured in  $\text{CH}_2\text{Cl}_2$  at room temperature.

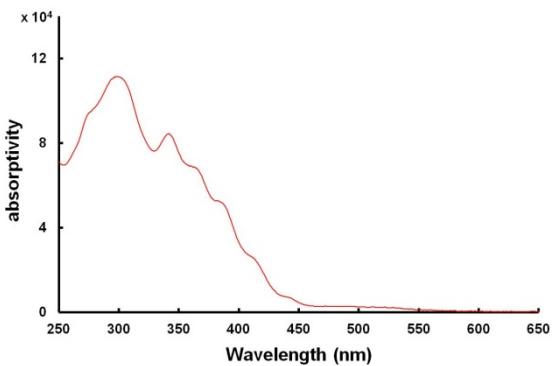
1a



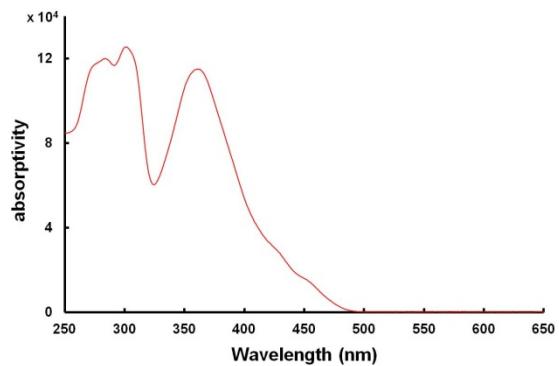
2a



3a

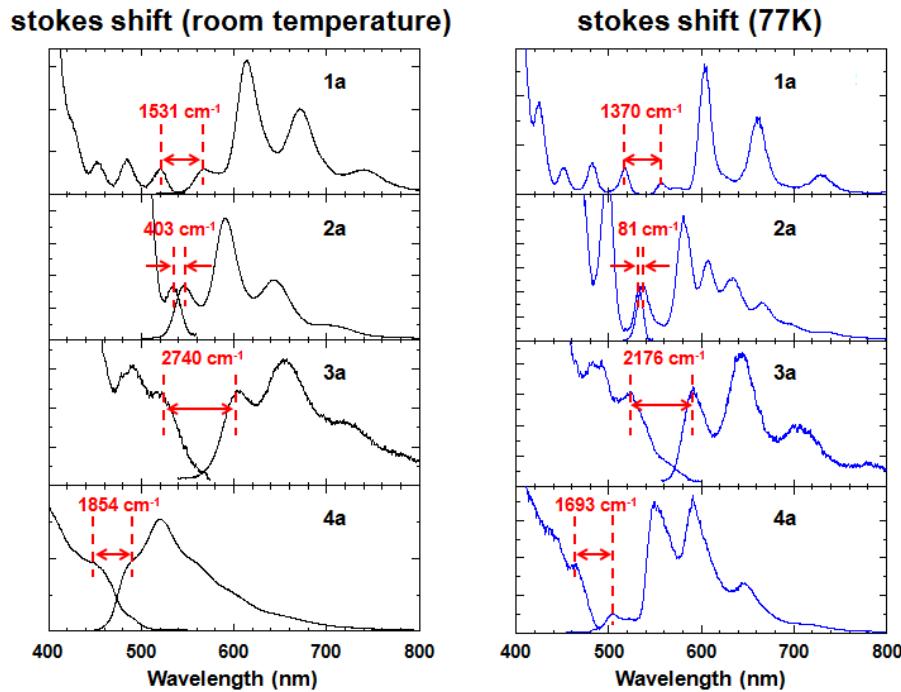


4a



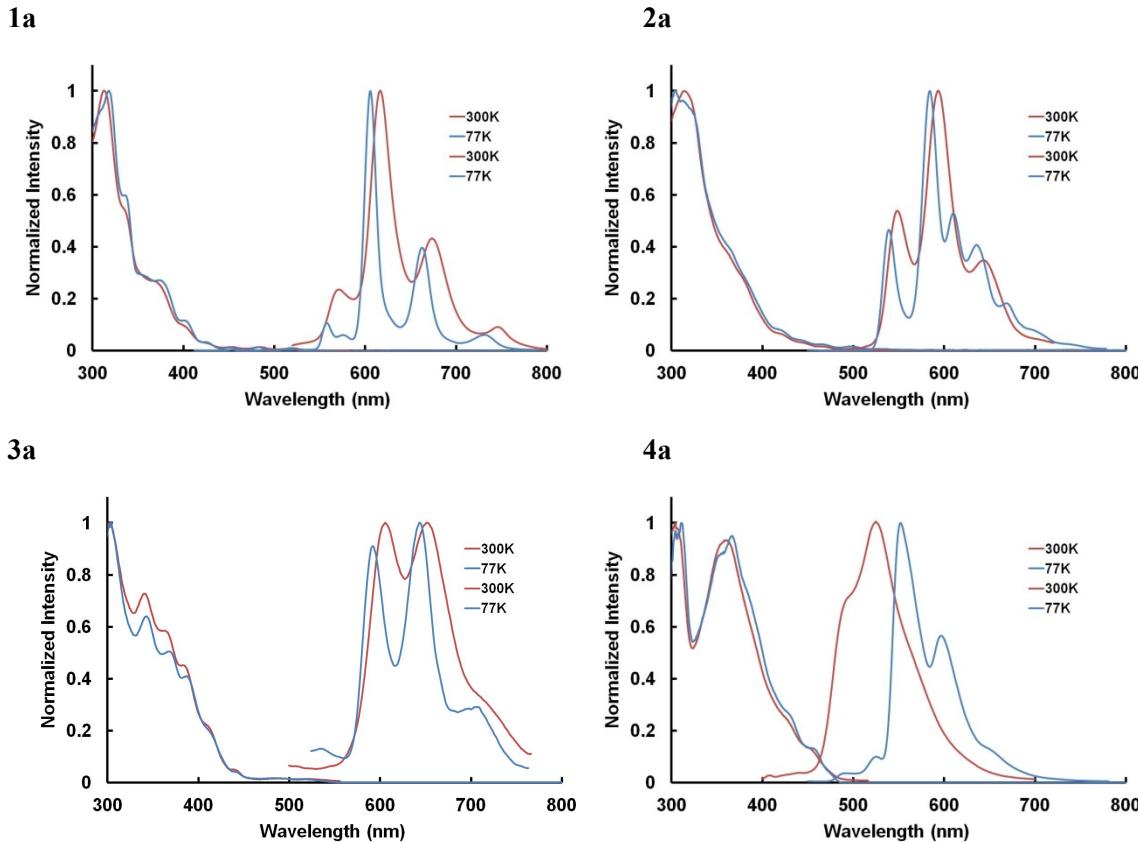
## S7. Stokes shifts of 1a–4a

Stokes shifts of 1a–4a at 300K (left) and 77K (right) in MTHF. All excitation wavelengths are 375 nm.



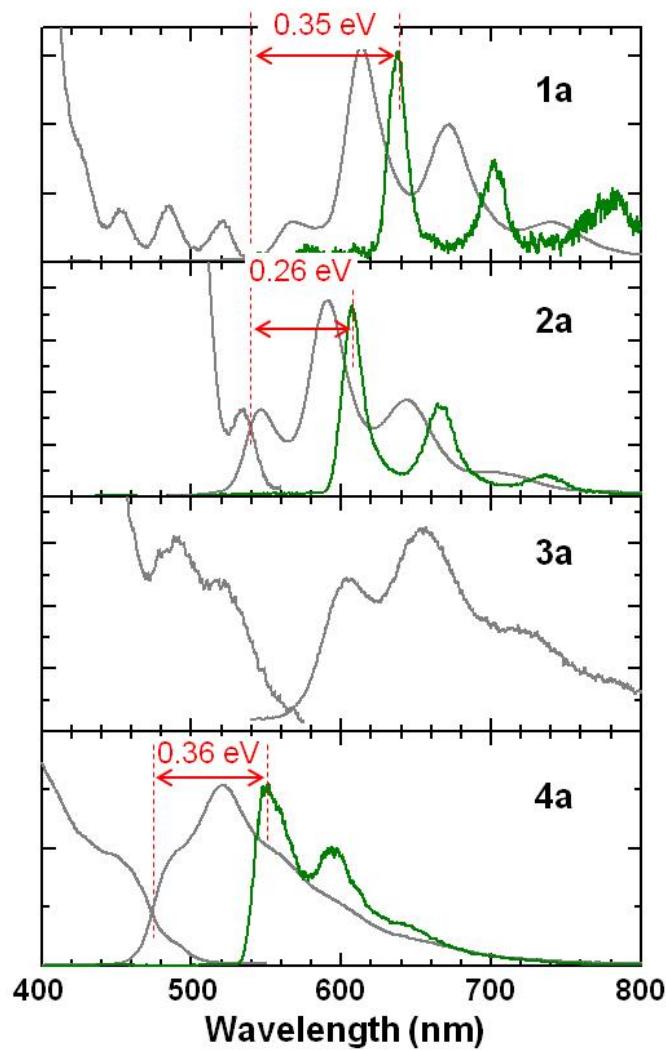
## S8. Low temperature measurements of UV and PL spectra of 1a–4a

Following spectra were all measured in MTHF excited at 375 nm.



### S9. S-T gaps of 1a, 2a, and 4a

Following spectra were all measured in MTHF excited at 375 nm. The S-T gaps were estimated from the 0–0 transition energy and phosphorescence spectra.



## S10. Calculated S-T gaps of 1c–4c

### 1. S–T gap of 1c

S	Energy	Transfer orbitals	T	Energy	Transfer orbitals
S2	3.07 eV	117->119			
			T5	2.75 eV	117->120
			T4	2.75 eV	117->119
			T3	2.65 eV	113->118 115->120 116->118 116->119 117->122
			T2	2.65 eV	114->118 115->118 115->119 116->120 117->121
S1	2.43 eV	117->118			
			T1	2.00 eV	117->118

### 2. S–T gap of 2c

S	Energy	Transfer orbitals	T	Energy	Transfer orbitals
S2	2.86 eV	135->137 136->138			
			T6	2.76 eV	134->137 135->138 136->141
			T5	2.75 eV	135->139 136->138 136->140 136->142
			T4	2.62 eV	135->137 135->139 136->138
			T3	2.53 eV	135->138 136->137 136->139
S1	2.46 eV	136->137			
			T2	2.34 eV	133->138 134->138 135->137 135->139 136->138 136->140
			T1	2.14 eV	136->137 136->139

3. S–T gap of **3c**

	S	Energy	Transfer orbitals	T	Energy	Transfer orbitals
<b>3c</b>	S2	2.32 eV	154->156			
	S1	2.28 eV	155->156			
				T2	2.12 eV	155->156
				T1	2.00 eV	154->156 155->157

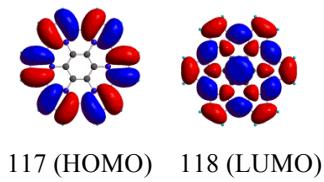
4. S–T gap of **4c**

	S	Energy	Transfer orbitals	T	Energy	Transfer orbitals
<b>4c</b>	S2	2.90 eV	172->176 173->175 173->177 174->176 174->179			
	S1	2.90 eV	172->175 173->176 174->177			
				T9	2.75 eV	172->176 172->179 173->175 173->177 173->178 174->176 174->179
				T8	2.75 eV	172->175 172->177 172->178 173->176 173->179 174->175 174->178
				T7	2.72 eV	172->175 172->178 173->176 173->179 174->177
				T6	2.63 eV	172->176 173->175 173->177 174->176 174->179
				T5	2.63 eV	172->175 172->177 173->176 174->175 174->178
				T4	2.61 eV	172->175 173->176 174->181
				T3	2.44 eV	172->176 173->175 174->180
				T2	2.37 eV	172->179 173->177 173->178 174->176 174->179
				T1	2.37 eV	172->177 172->178 173->179 174->175 174->178

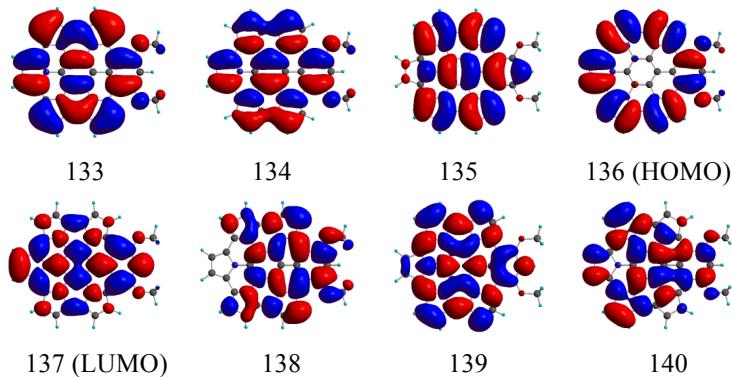
### S11. Molecular orbitals of 1c–4c

The GAUSSIAN 09<sup>[4]</sup> series of programs was used for all calculations. All molecules were fully optimized using the hybrid density functional at B3LYP level of theory with the 6-31G(d) basis set. Frequency calculations were conducted to ensure that these structures were indeed local minima.

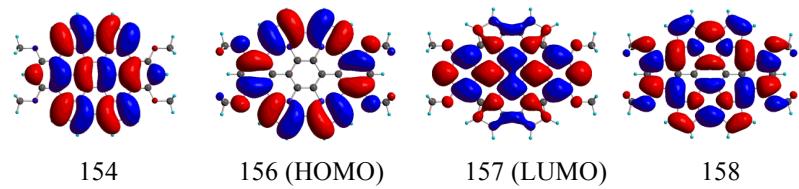
1c



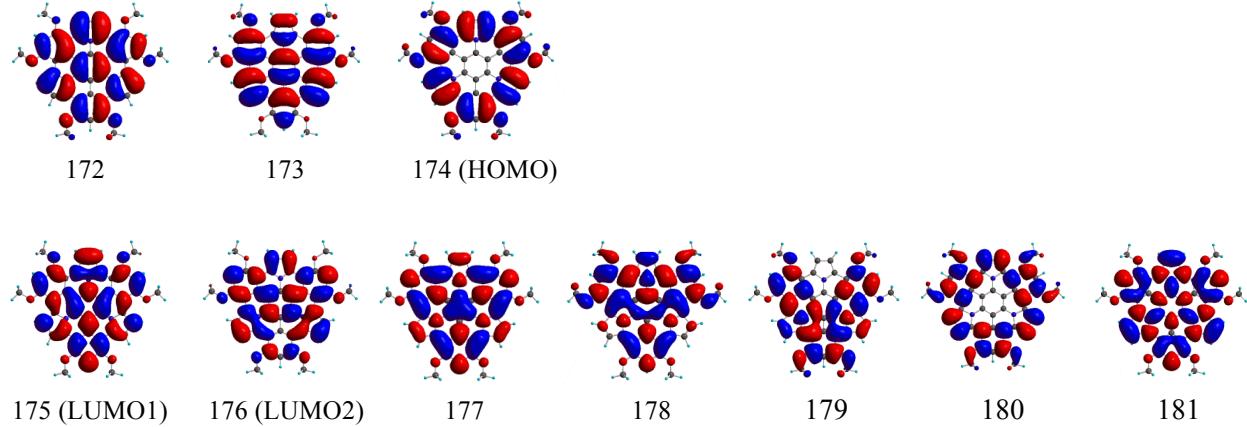
2c



3c



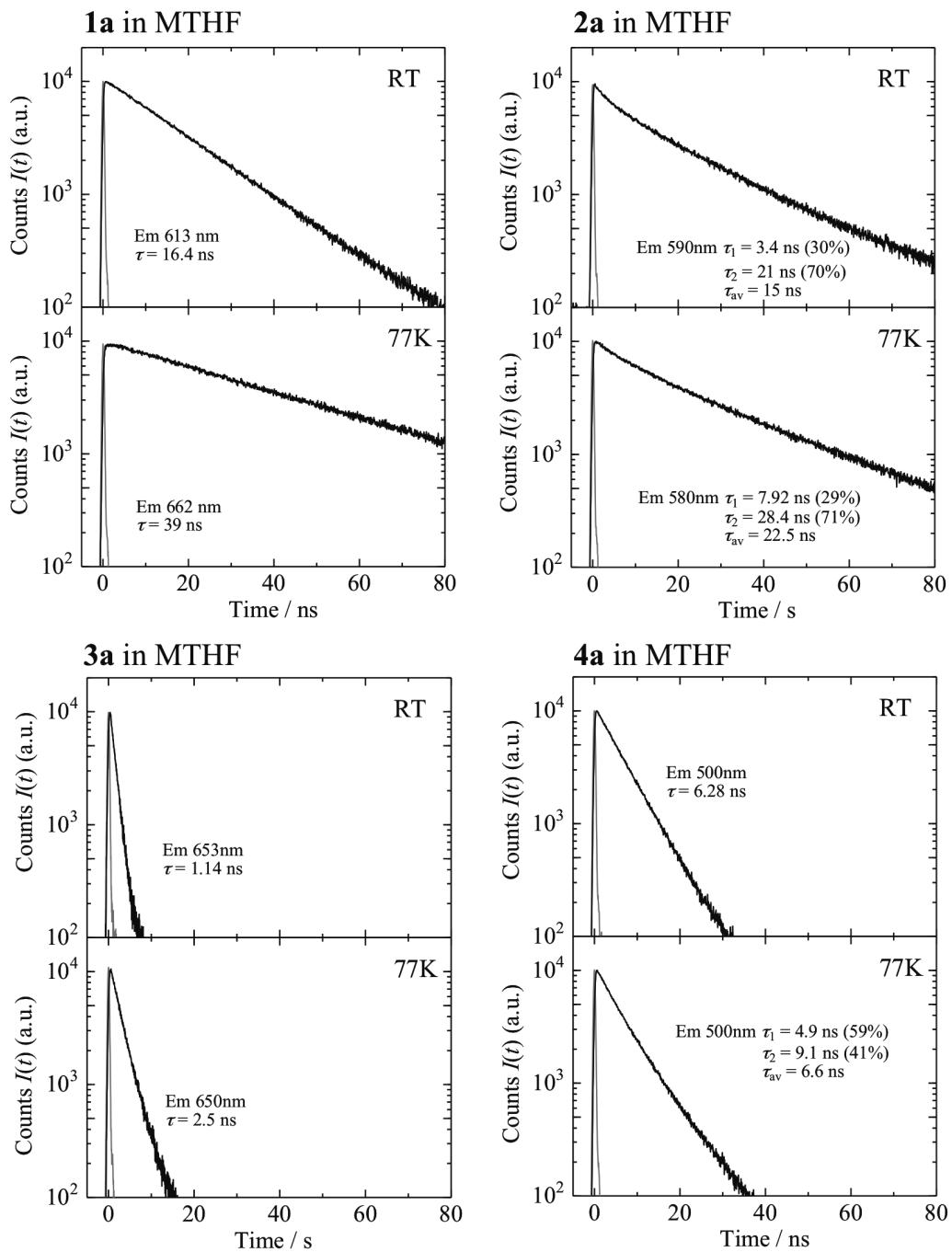
4c



## S12. Fluorescence and phosphorescence lifetimes

### 1. Fluorescence lifetimes

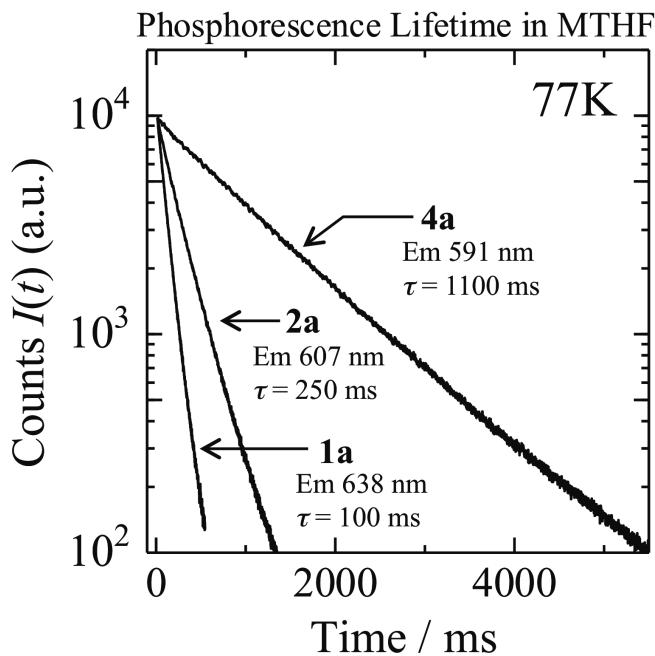
The fluorescence decay curve  $I(t)$  was fitted with a single or double exponential function,  $I(t) = I_0 \sum_i G_i \exp(-t/\tau_i)$ . Excitation wavelength was 375 nm for all samples. Monitor wavelength at RT was 613, 590, 653, and 500 nm for **1a**, **2a**, **3a**, and **4a**, respectively. Monitor wavelength at 77K was 662, 580, 650, and 500 nm for **1a**, **2a**, **3a**, and **4a**, respectively. The gray lines show an excitation laser pulse.



## 2. Phosphorescence lifetimes

The phosphorescence decay curve  $I(t)$  was fitted with a single exponential function,  $I(t) = I_0 \exp(-t/\tau)$ .

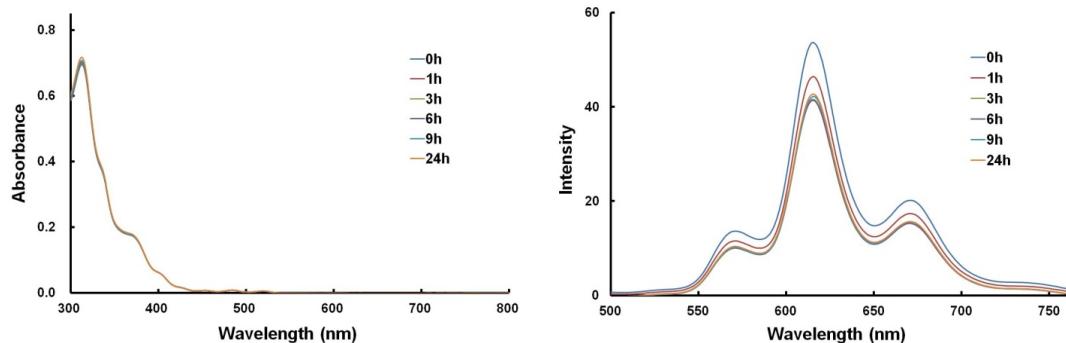
Excitation wavelength was 370 nm for all samples. Monitor wavelength was 638, 607, and 591 nm for **1a**, **2a**, and **4a**, respectively.



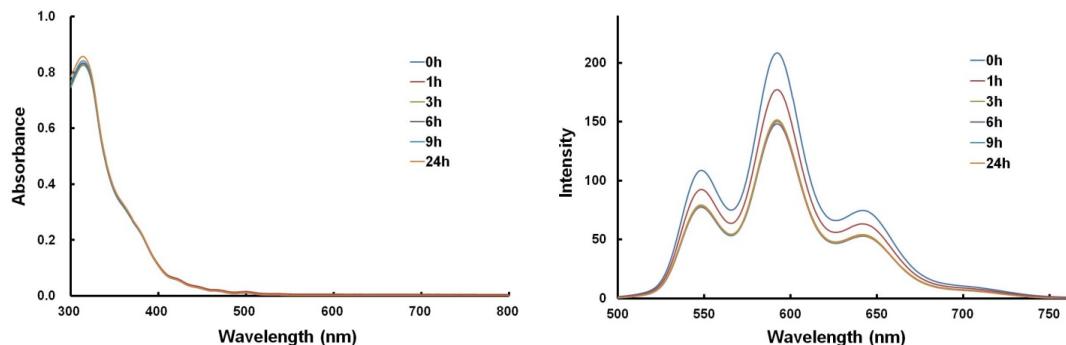
### S13. Time dependent UV and PL spectra of 1a–4a

Time dependent changes of the UV-PL spectra of **1a–4a** were measured in MTHF. The solutions were capped under nitrogen atmosphere and stored in ambient laboratory conditions.

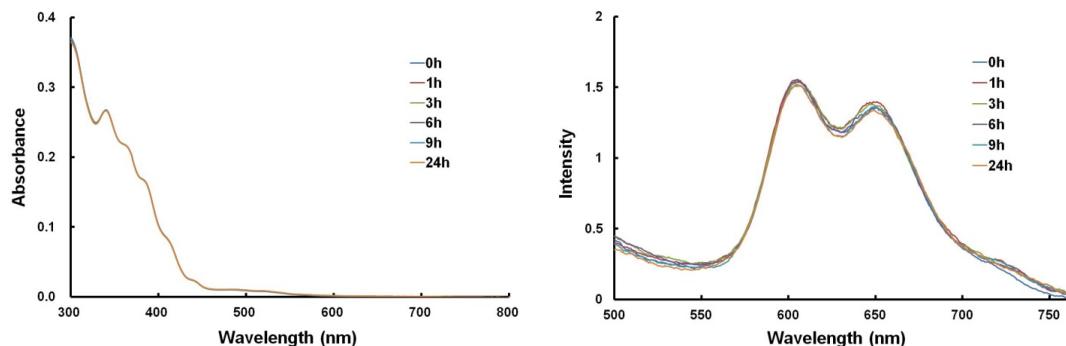
**1a**



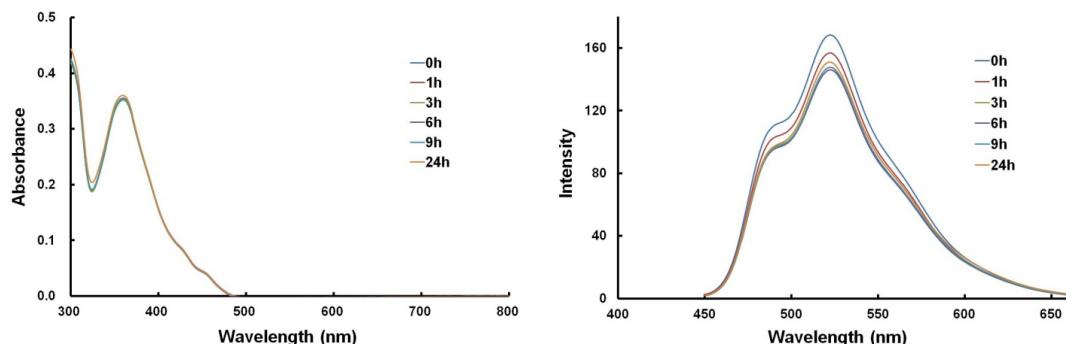
**2a**



**3a**



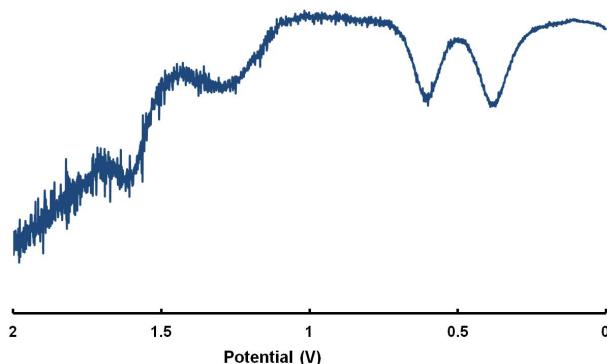
**4a**



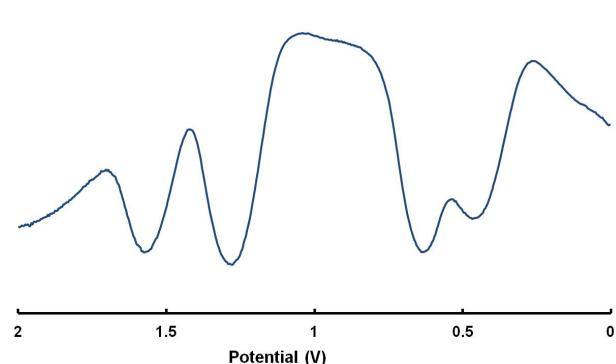
#### S14. Differential pulse voltammograms of **1a–4a**

Differential pulse voltammograms (DPV) of **1a–4a** were measured under following conditions: 0.1 M *n*-Bu<sub>4</sub>NClO<sub>4</sub> in CH<sub>2</sub>Cl<sub>2</sub>, Ag/AgNO<sub>3</sub> reference electrode, Pt working electrode and Pt counter electrode, scan rate: 100 mV s<sup>-1</sup>; V vs. Fc/Fc<sup>+</sup>.

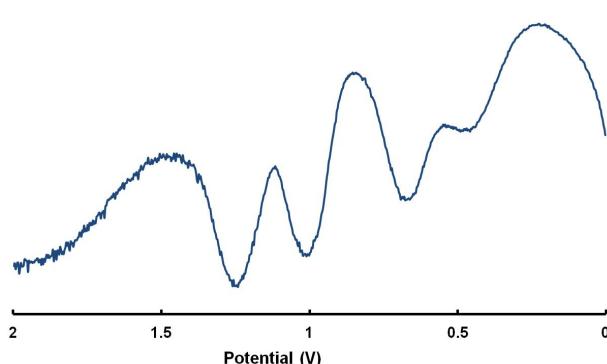
**1a**



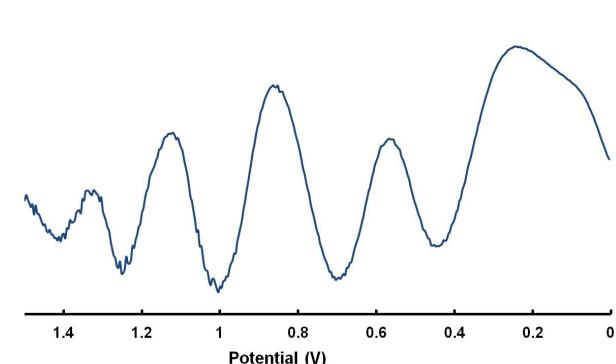
**2a**



**3a**



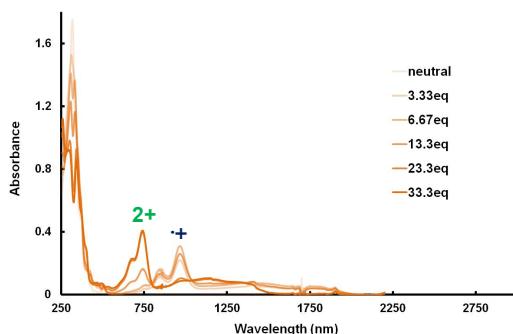
**4a**



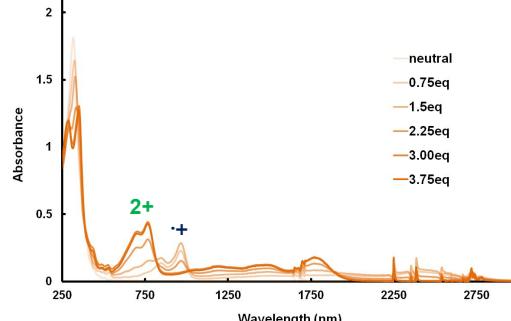
### S15. Stepwise chemical oxidations of **1a** and **2b–4b**

UV-Vis-NIR spectra of **1a** and **2b–4b** in  $\text{CH}_2\text{Cl}_2$  ( $1.05 \times 10^{-5}$  M) oxidized by  $\text{SbCl}_5$ .

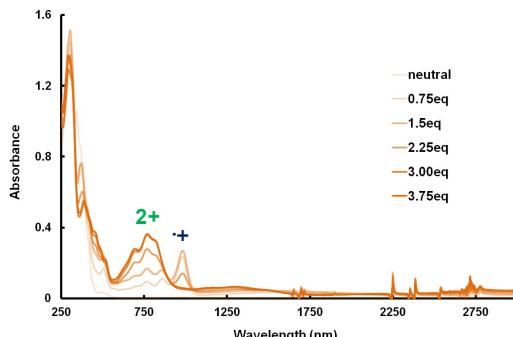
**1a**



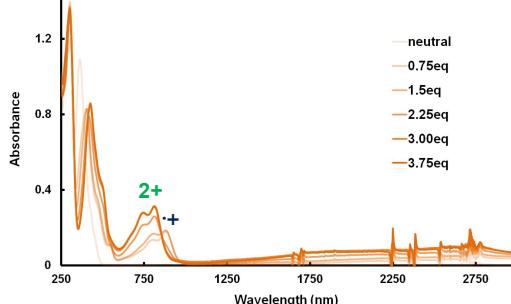
**2b**



**3b**

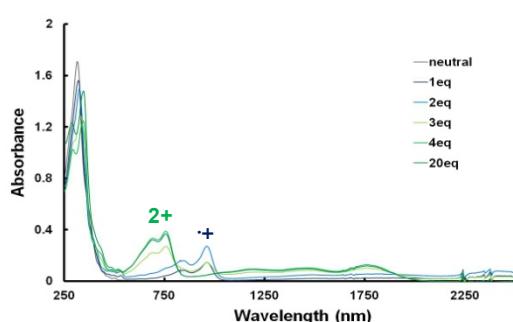


**4b**

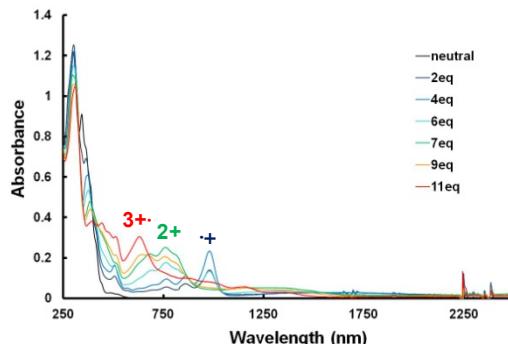


UV-Vis-NIR spectra of **2b–4b** in  $\text{CH}_2\text{Cl}_2$  and  $\text{MeCN}$  (4:1, v/v,  $1.05 \times 10^{-5}$  M) oxidized by  $\text{Fe}(\text{ClO}_4)_3$ .

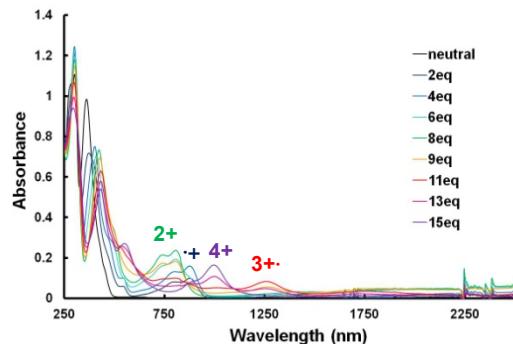
**2b**



**3b**

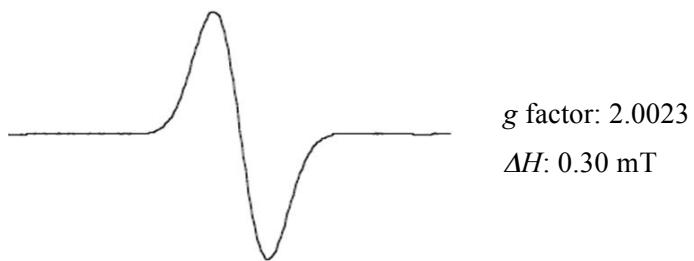


**4b**

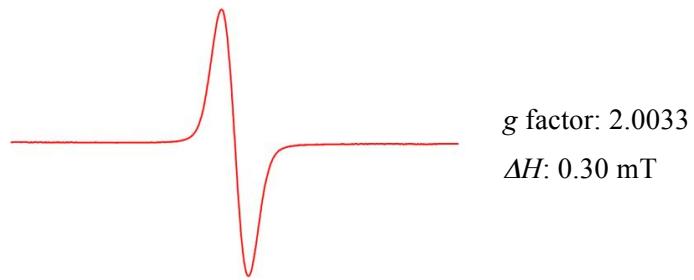


**S16. ESR spectra of  $\mathbf{1a}^+$  and  $\mathbf{2b}^+–\mathbf{4b}^+$  in  $\text{CH}_2\text{Cl}_2$**

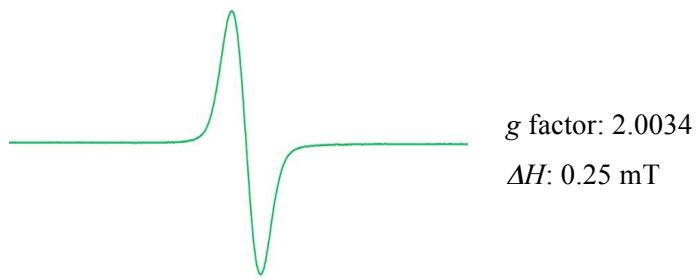
1. ESR spectrum of  $\mathbf{1a}^+$ <sup>[4]</sup>



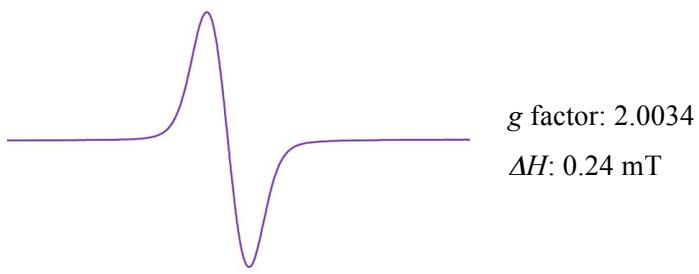
2. ESR spectrum of  $\mathbf{2b}^+$



3. ESR spectrum of  $\mathbf{3b}^+$



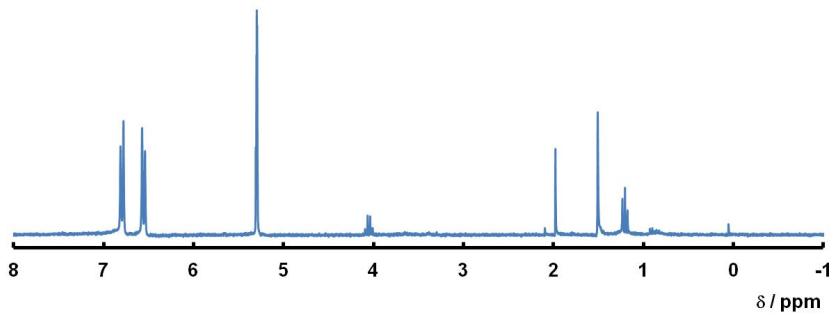
4. ESR spectrum of  $\mathbf{4b}^+$



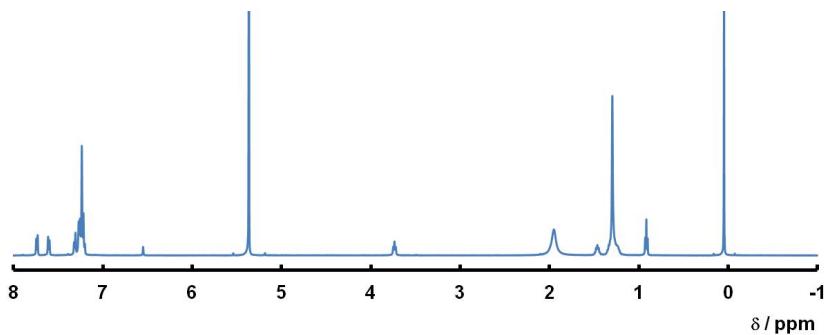
### S17. NMR spectra of $\mathbf{1a}^{2+}$ – $\mathbf{4a}^{2+}$ in $\text{CD}_2\text{Cl}_2$

The 1.0 mM solutions of  $\mathbf{1a}^{2+}$ – $\mathbf{4a}^{2+}$  were prepared by the  $\text{CD}_3\text{CN}$  solutions of  $\text{NO}\text{SbF}_6$  (1 M) and measured at 298 K (for  $\mathbf{1a}^{2+}$  and  $\mathbf{2a}^{2+}$ ) or 233 K (for  $\mathbf{3a}^{2+}$  and  $\mathbf{4a}^{2+}$ ) with or without  $\text{Fc(OAc)}_2\cdot\text{AgBF}_4$  as a radical scavenger. The circles indicate the peaks of radical scavenger.

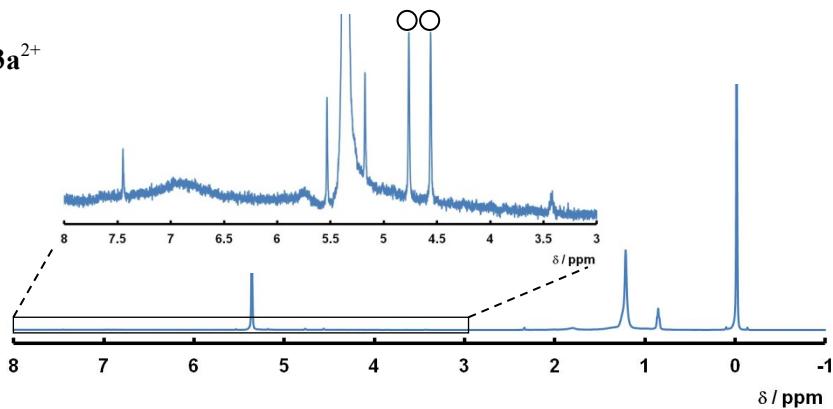
$\mathbf{1a}^{2+}$ <sup>[4]</sup>



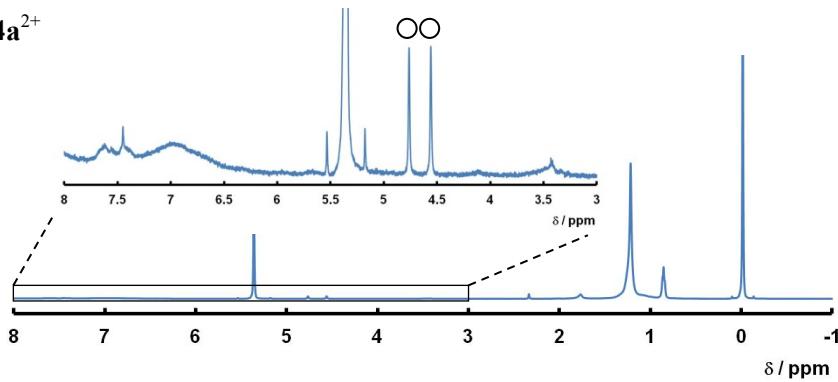
$\mathbf{2a}^{2+}$



$\mathbf{3a}^{2+}$

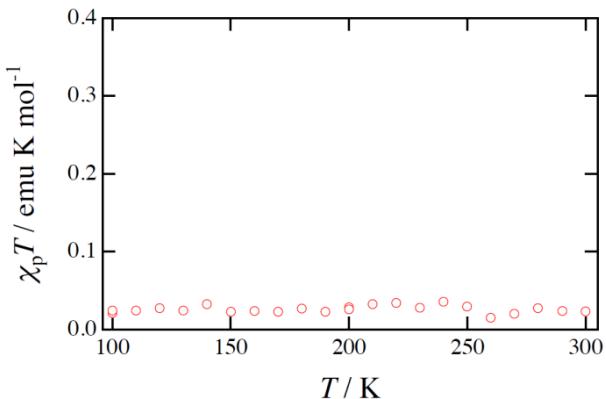


$\mathbf{4a}^{2+}$

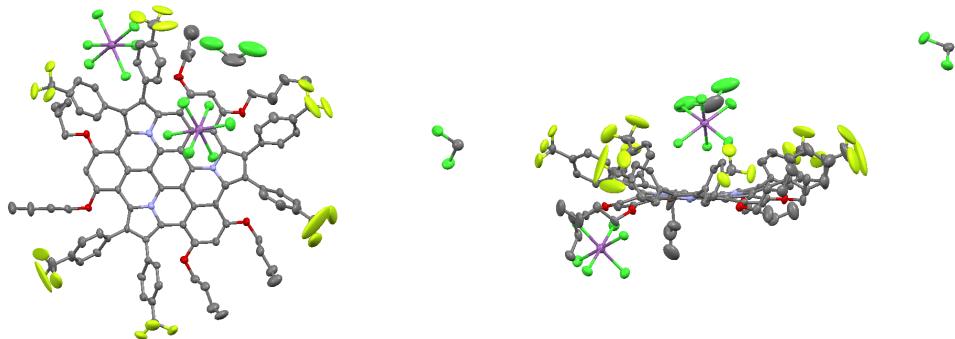


### S18. Thermal dependence of $\chi_M T$ and X-ray crystal structure for $4b^{2+}(SbCl_6)_2$

Magnetic measurements were carried out for a powder sample in a gelatinous capsule under an applied field of 500 Oe. The  $\chi_p$  values were obtained from the dc susceptibility, subtracting the diamagnetic susceptibility including a sample holder,  $-0.0000523 \text{ emu K mol}^{-1}$ , evaluated by assuming that  $\chi_p$  follows the Curie-Weiss law within the temperature range of 100 K – 300 K.



**Table S18.** Crystal data and structure refinement for  $\mathbf{4b}^{2+}(\text{SbCl}_6^-)_2$ .

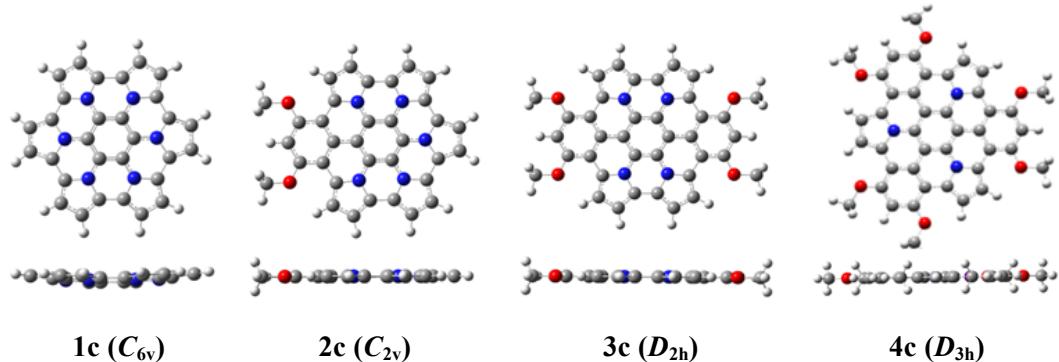


Identification code	<b>4b<sup>2+</sup>(SbCl<sub>6</sub><sup>-</sup>)<sub>2</sub></b>		
Empirical formula	$\text{C}_{103.5}\text{H}_{84}\text{Cl}_{15}\text{F}_{18}\text{N}_3\text{O}_6\text{Sb}_2$		
Formula weight	2582.99		
Temperature	100 K		
Wavelength	0.71070 Å		
Crystal system	Orthorhombic		
Space group	Fdd2		
Unit cell dimensions	$a = 44.931(3)$ Å	$\alpha = 90.00^\circ$ .	
	$b = 47.822(4)$ Å	$\beta = 90.00^\circ$ .	
	$c = 20.0654(15)$ Å	$\gamma = 90.00^\circ$ .	
Volume	$43114(6)$ Å <sup>3</sup>		
Z	16		
Density (calculated)	1.592 Mg/m <sup>3</sup>		
Absorption coefficient	0.959 mm <sup>-1</sup>		
F(000)	20688		
Crystal size	0.10 x 0.05 x 0.02 mm <sup>3</sup>		
Theta range for data collection	3.11 to 25.04°.		
Index ranges	$-53 \leq h \leq 53, -56 \leq k \leq 52, -23 \leq l \leq 23$		
Reflections collected	19030		
Independent reflections	16554 [R(int) = 0.0589]		
Completeness to theta = 25.04°	99.7 %		
Absorption correction	Multi-scan		
Refinement method	Full-matrix least-squares on F <sup>2</sup>		
Data / restraints / parameters	19030 / 1 / 1335		
Goodness-of-fit on F <sup>2</sup>	1.095		
Final R indices [I>2sigma(I)]	R1 = 0.0548, wR2 = 0.1349		
R indices (all data)	R1 = 0.0659, wR2 = 0.1465		
Largest diff. peak and hole	1.431 and -1.048 e.Å <sup>-3</sup>		

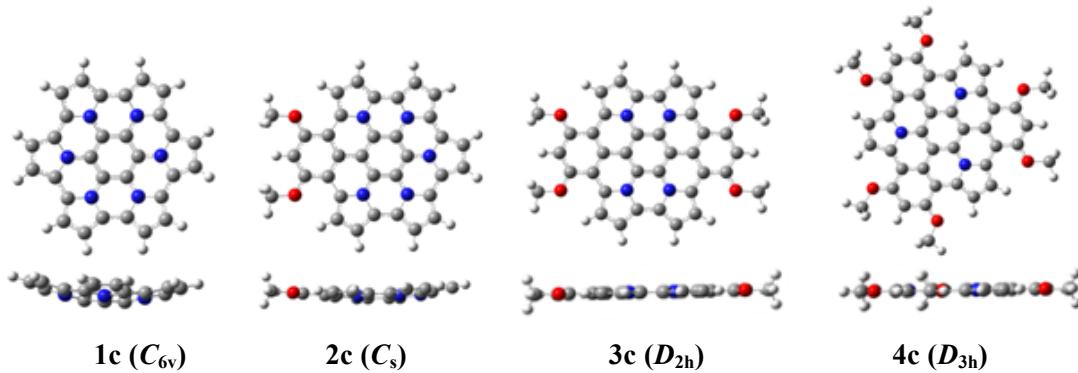
### S19. Optimized structures and their atomic coordinations of **1c–4c** and **1c<sup>2+</sup>–4c<sup>2+</sup>**

DFT calculations were performed with GAUSSIAN 09<sup>[4]</sup> series. All geometry optimizations were carried out at the B3LYP/6–31G(d) basis set. Frequency calculations were conducted to ensure that these structures were indeed local minima. For singlet biradical states, symmetry-broken UB3LYP/6–31G(d) method was employed with a keyword “guess=mix”, and the biradical index was determined on the basis of the LUMO occupation number in natural orbital analysis.

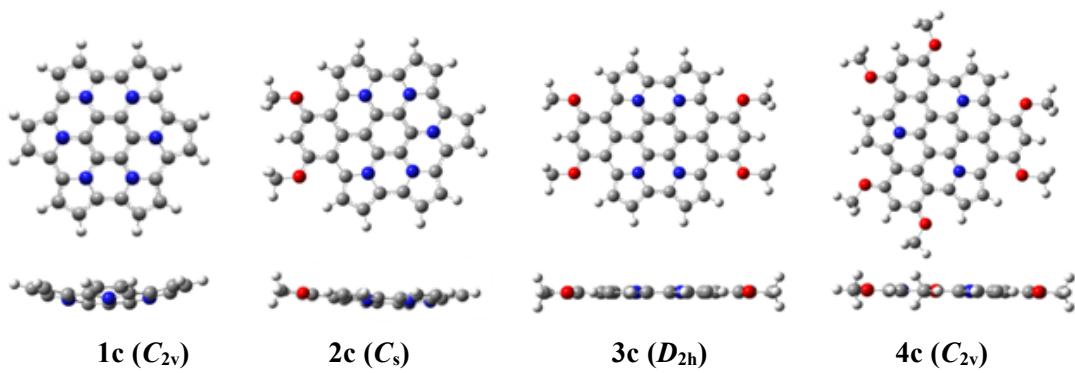
Optimized structures of **1c–4c**.



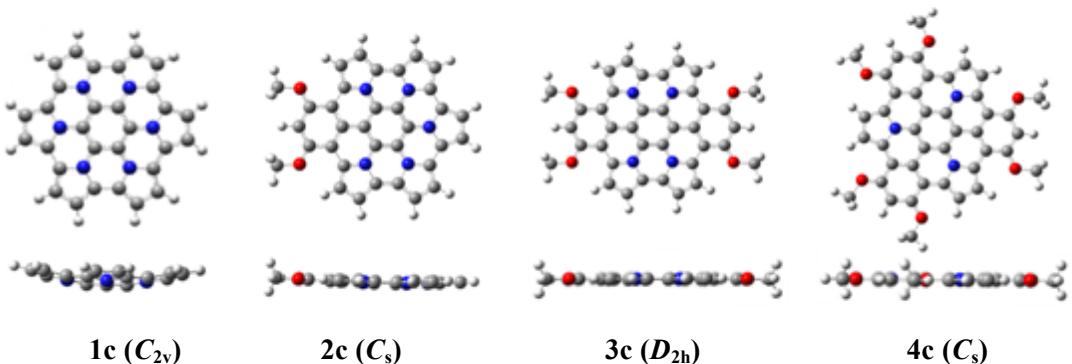
Optimized structures of **1c<sup>2+</sup>–4c<sup>2+</sup>** (closed shell singlet).



Optimized structures of **1c<sup>2+</sup>–4c<sup>2+</sup>** (open shell singlet).



Optimized structures of **1c<sup>2+</sup>**–**4c<sup>2+</sup>** (triplet).



Atomic coordination of **1c** ( $C_{6v}$  symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			24	6	0	1.781858	4.515225	0.158808
			X	Y	Z						
1	6	0	1.378165	0.000000	-0.263505	25	6	0	3.019370	3.800747	0.158808
2	6	0	0.689083	1.193526	-0.263505	26	1	0	1.675046	5.583524	0.286756
3	6	0	-0.689083	1.193526	-0.263505	27	1	0	3.997950	4.242394	0.286756
4	6	0	-1.378165	0.000000	-0.263505	28	7	0	-1.369138	2.371417	-0.136434
5	6	0	-0.689083	-1.193526	-0.263505	29	6	0	-2.758056	2.443324	-0.013385
6	6	0	0.689083	-1.193526	-0.263505	30	6	0	-0.736953	3.610209	-0.013385
7	7	0	-1.369138	-2.371417	-0.136434	31	6	0	-3.019370	3.800747	0.158808
8	6	0	-0.736953	-3.610209	-0.013385	32	6	0	-1.781858	4.515225	0.158808
9	6	0	-2.758056	-2.443324	-0.013385	33	1	0	-3.997950	4.242394	0.286756
10	6	0	-1.781858	-4.515225	0.158808	34	7	0	-2.738276	0.000000	-0.136434
11	6	0	-3.019370	-3.800747	0.158808	35	6	0	-3.495009	-1.166885	-0.013385
12	1	0	-1.675046	-5.583524	0.286756	36	6	0	-3.495009	1.166885	-0.013385
13	1	0	-3.997950	-4.242394	0.286756	37	6	0	-4.801229	-0.714478	0.158808
14	7	0	2.738276	0.000000	-0.136434	38	6	0	-4.801229	0.714478	0.158808
15	6	0	3.495009	1.166885	-0.013385	39	1	0	-5.672996	-1.341129	0.286756
16	6	0	3.495009	-1.166885	-0.013385	40	1	0	-5.672996	1.341129	0.286756
17	6	0	4.801229	0.714478	0.158808	41	7	0	1.369138	-2.371417	-0.136434
18	6	0	4.801229	-0.714478	0.158808	42	6	0	2.758056	-2.443324	-0.013385
19	1	0	5.672996	1.341129	0.286756	43	6	0	0.736953	-3.610209	-0.013385
20	1	0	5.672996	-1.341129	0.286756	44	6	0	3.019370	-3.800747	0.158808
21	7	0	1.369138	2.371417	-0.136434	45	6	0	1.781858	-4.515225	0.158808
22	6	0	0.736953	3.610209	-0.013385	46	1	0	3.997950	-4.242394	0.286756
23	6	0	2.758056	2.443324	-0.013385	47	1	0	1.675046	-5.583524	0.286756
HF = -1478.94481600 hartree											

Atomic coordination of **1c<sup>2+</sup>** (Closed shell,  $C_{6v}$  symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			6	6	0	0.690121	1.195325	0.577874
			X	Y	Z						
1	6	0	1.380243	0.000000	0.577874	7	7	0	-1.361179	2.357631	0.300317
2	6	0	0.690121	-1.195325	0.577874	8	6	0	-0.716361	3.560730	0.017265
3	6	0	-0.690121	-1.195325	0.577874	9	6	0	-2.725502	2.400752	0.017265
4	6	0	-1.380243	0.000000	0.577874	10	6	0	-1.760183	4.443965	-0.373478
5	6	0	-0.690121	1.195325	0.577874	11	6	0	-2.968495	3.746345	-0.373478
						12	1	0	-1.630725	5.482023	-0.647454
						13	1	0	-3.932209	4.153260	-0.647454
						14	7	0	2.722357	0.000000	0.300317

15	6	0	3.441863	-1.159978	0.017265	33	1	0	-3.932209	-4.153260	-0.647454
16	6	0	3.441863	1.159978	0.017265	34	7	0	-2.722357	0.000000	0.300317
17	6	0	4.728678	-0.697620	-0.373478	35	6	0	-3.441863	1.159978	0.017265
18	6	0	4.728678	0.697620	-0.373478	36	6	0	-3.441863	-1.159978	0.017265
19	1	0	5.562933	-1.328762	-0.647454	37	6	0	-4.728678	0.697620	-0.373478
20	1	0	5.562933	1.328762	-0.647454	38	6	0	-4.728678	-0.697620	-0.373478
21	7	0	1.361179	-2.357631	0.300317	39	1	0	-5.562933	1.328762	-0.647454
22	6	0	0.716361	-3.560730	0.017265	40	1	0	-5.562933	-1.328762	-0.647454
23	6	0	2.725502	-2.400752	0.017265	41	7	0	1.361179	2.357631	0.300317
24	6	0	1.760183	-4.443965	-0.373478	42	6	0	2.725502	2.400752	0.017265
25	6	0	2.968495	-3.746345	-0.373478	43	6	0	0.716361	3.560730	0.017265
26	1	0	1.630725	-5.482023	-0.647454	44	6	0	2.968495	3.746345	-0.373478
27	1	0	3.932209	-4.153260	-0.647454	45	6	0	1.760183	4.443965	-0.373478
28	7	0	-1.361179	-2.357631	0.300317	46	1	0	3.932209	4.153260	-0.647454
29	6	0	-2.725502	-2.400752	0.017265	47	1	0	1.630725	5.482023	-0.647454
30	6	0	-0.716361	-3.560730	0.017265	48	1	0	-1.630725	-5.482023	-0.647454
31	6	0	-2.968495	-3.746345	-0.373478						
32	6	0	-1.760183	-4.443965	-0.373478						

HF = -1478.42427922 hartree

### Atomic coordination of $\mathbf{1c}^{2+}$ (open shell singlet, $C_{2v}$ symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			24	6	0	4.443375	1.760385	-0.375592						
			X	Y	Z												
1	6	0	0.000000	1.380198	0.576902	28	7	0	2.357854	-1.361429	0.300750						
2	6	0	1.195261	0.690079	0.576226	29	6	0	2.400604	-2.725900	0.017742						
3	6	0	1.195261	-0.690079	0.576226	30	6	0	3.560291	-0.716295	0.015253						
4	6	0	0.000000	-1.380198	0.576902	31	6	0	3.746069	-2.968761	-0.374020						
5	6	0	-1.195261	-0.690079	0.576226	32	6	0	4.443375	-1.760385	-0.375592						
6	6	0	-1.195261	0.690079	0.576226	33	1	0	4.153481	-3.932472	-0.647277						
7	7	0	-2.357854	-1.361429	0.300750	34	7	0	0.000000	-2.723090	0.302901						
8	6	0	-3.560291	-0.716295	0.015253	35	6	0	-1.160039	-3.442326	0.018936						
9	6	0	-2.400604	-2.725900	0.017742	36	6	0	1.160039	-3.442326	0.018936						
10	6	0	-4.443375	-1.760385	-0.375592	37	6	0	-0.697572	-4.729496	-0.370927						
11	6	0	-3.746069	-2.968761	-0.374020	38	6	0	0.697572	-4.729496	-0.370927						
12	1	0	-5.481291	-1.631335	-0.650317	39	1	0	-1.328469	-5.564174	-0.644221						
13	1	0	-4.153481	-3.932472	-0.647277	40	1	0	1.328469	-5.564174	-0.644221						
14	7	0	0.000000	2.723090	0.302901	41	7	0	-2.357854	1.361429	0.300750						
15	6	0	1.160039	3.442326	0.018936	42	6	0	-2.400604	2.725900	0.017742						
16	6	0	-1.160039	3.442326	0.018936	43	6	0	-3.560291	0.716295	0.015253						
17	6	0	0.697572	4.729496	-0.370927	44	6	0	-3.746069	2.968761	-0.374020						
18	6	0	-0.697572	4.729496	-0.370927	45	6	0	-4.443375	1.760385	-0.375592						
19	1	0	1.328469	5.564174	-0.644221	46	1	0	-4.153481	3.932472	-0.647277						
20	1	0	-1.328469	5.564174	-0.644221	47	1	0	-5.481291	1.631335	-0.650317						
21	7	0	2.357854	1.361429	0.300750	48	1	0	5.481291	-1.631335	-0.650317						
22	6	0	3.560291	0.716295	0.015253												
23	6	0	2.400604	2.725900	0.017742												

HF = -1478.42434503 hartree

### Atomic coordination of $\mathbf{1c}^{2+}$ (triplet, $C_{2v}$ symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			2	6	0	1.200865	0.687518	0.483999						
			X	Y	Z												
1	6	0	0.000000	1.379501	0.475085	3	6	0	1.200865	-0.687518	0.483999						
						4	6	0	0.000000	-1.379501	0.475085						
						5	6	0	-1.200865	-0.687518	0.483999						
						6	6	0	-1.200865	0.687518	0.483999						

7	7	0	-2.360583	-1.365333	0.246224	29	6	0	2.421534	-2.727351	0.012482
8	6	0	-3.589277	-0.715694	0.007155	30	6	0	3.589277	-0.715694	0.007155
9	6	0	-2.421534	-2.727351	0.012482	31	6	0	3.780692	-2.978378	-0.312924
10	6	0	-4.482645	-1.767142	-0.315395	32	6	0	4.482645	-1.767142	-0.315395
11	6	0	-3.780692	-2.978378	-0.312924	33	1	0	4.198696	-3.948773	-0.542713
12	1	0	-5.532548	-1.648232	-0.544992	34	7	0	0.000000	-2.710771	0.230715
13	1	0	-4.198696	-3.948773	-0.542713	35	6	0	-1.165679	-3.461937	0.015383
14	7	0	0.000000	2.710771	0.230715	36	6	0	1.165679	-3.461937	0.015383
15	6	0	1.165679	3.461937	0.015383	37	6	0	-0.706606	-4.752692	-0.284933
16	6	0	-1.165679	3.461937	0.015383	38	6	0	0.706606	-4.752692	-0.284933
17	6	0	0.706606	4.752692	-0.284933	39	1	0	-1.331166	-5.607856	-0.503228
18	6	0	-0.706606	4.752692	-0.284933	40	1	0	1.331166	-5.607856	-0.503228
19	1	0	1.331166	5.607856	-0.503228	41	7	0	-2.360583	1.365333	0.246224
20	1	0	-1.331166	5.607856	-0.503228	42	6	0	-2.421534	2.727351	0.012482
21	7	0	2.360583	1.365333	0.246224	43	6	0	-3.589277	0.715694	0.007155
22	6	0	3.589277	0.715694	0.007155	44	6	0	-3.780692	2.978378	-0.312924
23	6	0	2.421534	2.727351	0.012482	45	6	0	-4.482645	1.767142	-0.315395
24	6	0	4.482645	1.767142	-0.315395	46	1	0	-4.198696	3.948773	-0.542713
25	6	0	3.780692	2.978378	-0.312924	47	1	0	-5.532548	1.648232	-0.544992
26	1	0	5.532548	1.648232	-0.544992	48	1	0	5.532548	-1.648232	-0.544992
27	1	0	4.198696	3.948773	-0.542713						
28	7	0	2.360583	-1.365333	0.246224						

HF = -1478.39815415 hartree

### Atomic coordination of 2c ( $C_{2v}$ symmetry)

Center	Atomic	Atomic	Coordinates (Angstroms)			29	6	0	0.000000	1.165314	4.393056
Number	Number	Type	X	Y	Z	30	6	0	0.000000	-0.714824	5.709731
1	6	0	0.000000	0.000000	2.257863	31	6	0	0.000000	0.714824	5.709731
2	6	0	0.000000	1.191081	1.567652	32	1	0	0.000000	-1.342284	6.590317
3	6	0	0.000000	1.189246	0.185099	33	1	0	0.000000	1.342284	6.590317
4	6	0	0.000000	0.000000	-0.540256	34	7	0	0.000000	-2.367744	2.267450
5	6	0	0.000000	-1.189246	0.185099	35	6	0	0.000000	-3.618630	1.656876
6	6	0	0.000000	-1.191081	1.567652	36	6	0	0.000000	-2.439012	3.661744
7	6	0	0.000000	0.000000	-1.973606	37	6	0	0.000000	-4.528215	2.712738
8	6	0	0.000000	-1.266397	-2.632624	38	6	0	0.000000	-3.803478	3.942032
9	6	0	0.000000	1.266397	-2.632624	39	1	0	0.000000	-5.605136	2.617177
10	6	0	0.000000	-1.225439	-4.041788	40	1	0	0.000000	-4.242260	4.930332
11	6	0	0.000000	1.225439	-4.041788	41	7	0	0.000000	-2.392957	-0.458537
12	6	0	0.000000	0.000000	-4.723264	42	6	0	0.000000	-3.629549	0.192573
13	7	0	0.000000	2.392957	-0.458537	43	6	0	0.000000	-2.511152	-1.846624
14	6	0	0.000000	3.629549	0.192573	44	6	0	0.000000	-4.569654	-0.834811
15	6	0	0.000000	2.511152	-1.846624	45	6	0	0.000000	-3.887419	-2.083480
16	6	0	0.000000	4.569654	-0.834811	46	1	0	0.000000	-5.643073	-0.702761
17	6	0	0.000000	3.887419	-2.083480	47	1	0	0.000000	-4.349622	-3.057716
18	1	0	0.000000	5.643073	-0.702761	48	8	0	0.000000	2.432325	-4.677317
19	1	0	0.000000	4.349622	-3.057716	49	8	0	0.000000	-2.432325	-4.677317
20	7	0	0.000000	2.367744	2.267450	50	6	0	0.000000	2.466634	-6.095534
21	6	0	0.000000	2.439012	3.661744	51	1	0	-0.896198	1.986522	-6.510344
22	6	0	0.000000	3.618630	1.656876	52	1	0	0.000000	3.523651	-6.367117
23	6	0	0.000000	3.803478	3.942032	53	1	0	0.896198	1.986522	-6.510344
24	6	0	0.000000	4.528215	2.712738	54	6	0	0.000000	-2.466634	-6.095534
25	1	0	0.000000	4.242260	4.930332	55	1	0	0.896198	-1.986522	-6.510344
26	1	0	0.000000	5.605136	2.617177	56	1	0	0.000000	-3.523651	-6.367117
27	7	0	0.000000	0.000000	3.624675	57	1	0	-0.896198	-1.986522	-6.510344
28	6	0	0.000000	-1.165314	4.393056	58	1	0	0.000000	0.000000	-5.803653

HF = -1730.09968202 hartree

### Atomic coordination of $2\text{c}^{2+}$ (closed shell, $C_s$ symmetry)

Center	Atomic Number	Atomic Type	Coordinates (Angstroms)							
Number	Number	Type	X	Y	Z	30	6	0	0.032613	-4.350719
1	6	0	-0.330230	-2.253107	0.000000	31	6	0	0.296875	-5.673557
2	6	0	-0.332145	-1.564664	1.187707	32	1	0	0.296875	-5.673557
3	6	0	-0.325143	-0.184711	1.185433	33	1	0	0.481148	-6.532812
4	6	0	-0.321037	0.539447	0.000000	34	7	0	-0.172832	-2.257711
5	6	0	-0.325143	-0.184711	-1.185433	35	6	0	-0.022325	-1.641863
6	6	0	-0.332145	-1.564664	-1.187707	36	6	0	0.017765	-3.643456
7	6	0	-0.186129	1.962651	0.000000	37	6	0	0.220488	-2.715990
8	6	0	-0.099128	2.610295	-1.266766	38	6	0	0.245726	-3.914747
9	6	0	-0.099128	2.610295	1.266766	39	1	0	0.373895	-2.609791
10	6	0	0.028978	4.039373	-1.228459	40	1	0	0.420554	-4.894308
11	6	0	0.028978	4.039373	1.228459	41	7	0	-0.196448	0.455136
12	6	0	0.081155	4.715535	0.000000	42	6	0	-0.052190	-0.209919
13	7	0	-0.196448	0.455136	2.381031	43	6	0	-0.086139	1.828901
14	6	0	-0.052190	-0.209919	3.608933	44	6	0	0.115738	0.831718
15	6	0	-0.086139	1.828901	2.492197	45	6	0	0.092572	2.058925
16	6	0	0.115738	0.831718	4.553709	46	1	0	0.248884	0.695103
17	6	0	0.092572	2.058925	3.885398	47	1	0	0.204640	3.029462
18	1	0	0.248884	0.695103	5.618094	48	8	0	0.092120	4.647520
19	1	0	0.204640	3.029462	4.341935	49	8	0	0.092120	4.647520
20	7	0	-0.172832	-2.257711	2.356283	50	6	0	0.218062	6.082606
21	6	0	0.017765	-3.643456	2.405204	51	1	0	-0.648107	6.568670
22	6	0	-0.022325	-1.641863	3.586251	52	1	0	0.245234	6.301998
23	6	0	0.245726	-3.914747	3.785282	53	1	0	1.147603	6.413248
24	6	0	0.220488	-2.715990	4.495294	54	6	0	0.245234	6.301998
25	1	0	0.420554	-4.894308	4.208602	55	1	0	1.147603	6.413248
26	1	0	0.373895	-2.609791	5.560226	56	1	0	0.092120	-3.562269
27	7	0	-0.158370	-3.610368	0.000000	57	1	0	-0.648107	6.568670
28	6	0	0.032613	-4.350719	-1.163614	58	1	0	0.173370	5.791713

HF = -1729.59523357 hartree

### Atomic coordination of $2\text{c}^{2+}$ (open shell singlet, $C_s$ symmetry)

Center	Atomic Number	Atomic Type	Coordinates (Angstroms)							
Number	Number	Type	X	Y	Z	19	1	0	0.205699	3.029393
1	6	0	-0.332221	-2.253221	0.000000	20	7	0	-0.174107	-2.257773
2	6	0	-0.333797	-1.564729	1.187734	21	6	0	0.017549	-3.643365
3	6	0	-0.326108	-0.184759	1.185442	22	6	0	-0.022564	-1.641875
4	6	0	-0.321844	0.539407	0.000000	23	6	0	0.247153	-3.914538
5	6	0	-0.326108	-0.184759	-1.185442	24	6	0	0.221520	-2.715821
6	6	0	-0.333797	-1.564729	-1.187734	25	1	0	0.423040	-4.894014
7	6	0	-0.186304	1.962582	0.000000	26	1	0	0.375785	-2.609466
8	6	0	-0.099145	2.610234	-1.266729	27	7	0	-0.159893	-3.610490
9	6	0	-0.099145	2.610234	1.266729	28	6	0	0.032302	-4.350630
10	6	0	0.029167	4.039297	-1.228444	29	6	0	0.032302	4.350630
11	6	0	0.029167	4.039297	1.228444	30	6	0	0.298237	-5.673115
12	6	0	0.081328	4.715497	0.000000	31	6	0	0.298237	0.696476
13	7	0	-0.196711	0.455089	2.380979	32	1	0	0.483660	-6.532074
14	6	0	-0.052211	-0.209938	3.608868	33	1	0	0.483660	-6.532074
15	6	0	-0.086001	1.828818	2.492114	34	7	0	-0.174107	-2.257773
16	6	0	0.116309	0.831649	4.553564	35	6	0	-0.022564	-1.641875
17	6	0	0.093173	2.058857	3.885239	36	6	0	0.017549	-3.643365
						37	6	0	0.221520	-2.715821
						38	6	0	0.247153	-3.914538

39	1	0	0.375785	-2.609466	-5.559823	50	6	0	0.218396	6.082447	2.496278
40	1	0	0.423040	-4.894014	-4.208026	51	1	0	-0.647850	6.568743	2.038405
41	7	0	-0.196711	0.455089	-2.380979	52	1	0	0.245596	6.301825	3.562309
42	6	0	-0.052211	-0.209938	-3.608868	53	1	0	1.147801	6.413002	2.023430
43	6	0	-0.086001	1.828818	-2.492114	54	6	0	0.218396	6.082447	-2.496278
44	6	0	0.116309	0.831649	-4.553564	55	1	0	1.147801	6.413002	-2.023430
45	6	0	0.093173	2.058857	-3.885239	56	1	0	0.245596	6.301825	-3.562309
46	1	0	0.249727	0.695004	-5.617916	57	1	0	-0.647850	6.568743	-2.038405
47	1	0	0.205699	3.029393	-4.341673	58	1	0	0.173617	5.791664	0.000000

HF = -1729.5952335 hartree

### Atomic coordination of **2c<sup>2+</sup>** (triplet, *C<sub>s</sub>* symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			29	30	31	32	33	34
			X	Y	Z						
			6	6	6	6	6	6	6	6	6
1	6	0	-0.098949	-2.251509	0.000000	33	1	0	0.145491	-6.578024	1.331013
2	6	0	-0.104343	-1.560533	1.191715	34	7	0	-0.051920	-2.260219	-2.359122
3	6	0	-0.105703	-0.186045	1.189666	35	6	0	-0.007141	-1.639321	-3.613977
4	6	0	-0.105875	0.541434	0.000000	36	6	0	0.007675	-3.642865	-2.429470
5	6	0	-0.105703	-0.186045	-1.189666	37	6	0	0.068432	-2.717979	-4.532157
6	6	0	-0.104343	-1.560533	-1.191715	38	6	0	0.077705	-3.923950	-3.817720
7	6	0	-0.061266	1.955922	0.000000	39	1	0	0.115566	-2.623528	-5.608078
8	6	0	-0.033373	2.615959	-1.274878	40	1	0	0.133405	-4.911943	-4.253199
9	6	0	-0.033373	2.615959	1.274878	41	7	0	-0.064987	0.458390	-2.384256
10	6	0	0.008364	4.039942	-1.233159	42	6	0	-0.017572	-0.207403	-3.627830
11	6	0	0.008364	4.039942	1.233159	43	6	0	-0.029632	1.831128	-2.499716
12	6	0	0.024864	4.712203	0.000000	44	6	0	0.035259	0.841453	-4.579337
13	7	0	-0.064987	0.458390	2.384256	45	6	0	0.027408	2.065786	-3.907638
14	6	0	-0.017572	-0.207403	3.627830	46	1	0	0.077889	0.709675	-5.651628
15	6	0	-0.029632	1.831128	2.499716	47	1	0	0.062769	3.041314	-4.366411
16	6	0	0.035259	0.841453	4.579337	48	8	0	0.030818	4.654928	2.418059
17	6	0	0.027408	2.065786	3.907638	49	8	0	0.030818	4.654928	-2.418059
18	1	0	0.077889	0.709675	5.651628	50	6	0	0.073200	6.093822	2.496774
19	1	0	0.062769	3.041314	4.366411	51	1	0	-0.817339	6.529361	2.033721
20	7	0	-0.051920	-2.260219	2.359122	52	1	0	0.083941	6.321935	3.561435
21	6	0	0.007675	-3.642865	2.429470	53	1	0	0.983772	6.476983	2.026382
22	6	0	-0.007141	-1.639321	3.613977	54	6	0	0.073200	6.093822	-2.496774
23	6	0	0.077705	-3.923950	3.817720	55	1	0	0.983772	6.476983	-2.026382
24	6	0	0.068432	-2.717979	4.532157	56	1	0	0.083941	6.321935	-3.561435
25	1	0	0.133405	-4.911943	4.253199	57	1	0	-0.817339	6.529361	-2.033721
26	1	0	0.115566	-2.623528	5.608078	58	1	0	0.055578	5.792323	0.000000

HF = -1729.58546038 hartree

### Atomic coordination of **3c** (*D<sub>2h</sub>* symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			5	6	7	8	9	10
			X	Y	Z						
			6	6	6	6	6	6	6	6	6
1	6	0	1.422538	0.000000	0.000000	5	6	0	-0.695189	-1.187943	0.000000
2	6	0	0.695189	1.187943	0.000000	6	6	0	0.695189	-1.187943	0.000000
3	6	0	-0.695189	1.187943	0.000000	7	6	0	-2.862183	0.000000	0.000000
4	6	0	-1.422538	0.000000	0.000000	8	6	0	-3.527422	-1.260541	0.000000

13	1	0	-6.700654	0.000000	0.000000	42	7	0	1.357366	-2.387350	0.000000
14	6	0	2.862183	0.000000	0.000000	43	6	0	0.726683	-3.629220	0.000000
15	6	0	3.527422	-1.260541	0.000000	44	6	0	2.746574	-2.500942	0.000000
16	6	0	3.527422	1.260541	0.000000	45	6	0	1.756376	-4.565684	0.000000
17	6	0	4.936683	-1.222718	0.000000	46	6	0	2.997121	-3.873953	0.000000
18	6	0	4.936683	1.222718	0.000000	47	1	0	1.628183	-5.639449	0.000000
19	6	0	5.620267	0.000000	0.000000	48	1	0	3.976398	-4.324872	0.000000
20	1	0	6.700654	0.000000	0.000000	49	8	0	-5.571501	2.431120	0.000000
21	7	0	1.357366	2.387350	0.000000	50	8	0	-5.571501	-2.431120	0.000000
22	6	0	0.726683	3.629220	0.000000	51	8	0	5.571501	-2.431120	0.000000
23	6	0	2.746574	2.500942	0.000000	52	8	0	5.571501	2.431120	0.000000
24	6	0	1.756376	4.565684	0.000000	53	6	0	6.989000	2.465206	0.000000
25	6	0	2.997121	3.873953	0.000000	54	1	0	7.404724	1.985165	0.895699
26	1	0	1.628183	5.639449	0.000000	55	1	0	7.404724	1.985165	-0.895699
27	1	0	3.976398	4.324872	0.000000	56	1	0	7.261001	3.522205	0.000000
28	7	0	-1.357366	2.387350	0.000000	57	6	0	-6.989000	2.465206	0.000000
29	6	0	-0.726683	3.629220	0.000000	58	1	0	-7.404724	1.985165	-0.895699
30	6	0	-2.746574	2.500942	0.000000	59	1	0	-7.404724	1.985165	0.895699
31	6	0	-1.756376	4.565684	0.000000	60	1	0	-7.261001	3.522205	0.000000
32	6	0	-2.997121	3.873953	0.000000	61	6	0	-6.989000	-2.465206	0.000000
33	1	0	-1.628183	5.639449	0.000000	62	1	0	-7.404724	-1.985165	0.895699
34	1	0	-3.976398	4.324872	0.000000	63	1	0	-7.404724	-1.985165	-0.895699
35	7	0	-1.357366	-2.387350	0.000000	64	1	0	-7.261001	-3.522205	0.000000
36	6	0	-0.726683	-3.629220	0.000000	65	6	0	6.989000	-2.465206	0.000000
37	6	0	-2.746574	-2.500942	0.000000	66	1	0	7.404724	-1.985165	-0.895699
38	6	0	-1.756376	-4.565684	0.000000	67	1	0	7.404724	-1.985165	0.895699
39	6	0	-2.997121	-3.873953	0.000000	68	1	0	7.261001	-3.522205	0.000000
40	1	0	-1.628183	-5.639449	0.000000						
41	1	0	-3.976398	-4.324872	0.000000						

HF = -1981.24942783 hartree

### Atomic coordination of $3c^{2+}$ (closed shell, $D_{2h}$ symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			22	6	0	0.714213	3.616305	0.000000
			X	Y	Z						
1	6	0	1.418415	0.000000	0.000000	23	6	0	2.735619	2.486571	0.000000
2	6	0	0.692432	1.181636	0.000000	24	6	0	1.766189	4.566674	0.000000
3	6	0	-0.692432	1.181636	0.000000	25	6	0	2.985356	3.886584	0.000000
4	6	0	-1.418415	0.000000	0.000000	26	1	0	1.638390	5.640199	0.000000
5	6	0	-0.692432	-1.181636	0.000000	27	1	0	3.963751	4.339567	0.000000
6	6	0	0.692432	-1.181636	0.000000	28	7	0	-1.352063	2.378167	0.000000
7	6	0	-2.851763	0.000000	0.000000	29	6	0	-0.714213	3.616305	0.000000
8	6	0	-3.507535	-1.262984	0.000000	30	6	0	-2.735619	2.486571	0.000000
9	6	0	-3.507535	1.262984	0.000000	31	6	0	-1.766189	4.566674	0.000000
10	6	0	-4.944015	-1.226085	0.000000	32	6	0	-2.985356	3.886584	0.000000
11	6	0	-4.944015	1.226085	0.000000	33	1	0	-1.638390	5.640199	0.000000
12	6	0	-5.624468	0.000000	0.000000	34	1	0	-3.963751	4.339567	0.000000
13	1	0	-6.704513	0.000000	0.000000	35	7	0	-1.352063	-2.378167	0.000000
14	6	0	2.851763	0.000000	0.000000	36	6	0	-0.714213	-3.616305	0.000000
15	6	0	3.507535	-1.262984	0.000000	37	6	0	-2.735619	-2.486571	0.000000
16	6	0	3.507535	1.262984	0.000000	38	6	0	-1.766189	-4.566674	0.000000
17	6	0	4.944015	-1.226085	0.000000	39	6	0	-2.985356	-3.886584	0.000000
18	6	0	4.944015	1.226085	0.000000	40	1	0	-1.638390	-5.640199	0.000000
19	6	0	5.624468	0.000000	0.000000	41	1	0	-3.963751	-4.339567	0.000000
20	1	0	6.704513	0.000000	0.000000	42	7	0	1.352063	-2.378167	0.000000
21	7	0	1.352063	2.378167	0.000000	43	6	0	0.714213	-3.616305	0.000000

47	1	0	1.638390	-5.640199	0.000000	59	1	0	-7.405012	2.029383	0.900989
48	1	0	3.963751	-4.339567	0.000000	60	1	0	-7.214947	3.560775	0.000000
49	8	0	-5.554889	2.406688	0.000000	61	6	0	-6.994437	-2.494588	0.000000
50	8	0	-5.554889	-2.406688	0.000000	62	1	0	-7.405012	-2.029383	0.900989
51	8	0	5.554889	-2.406688	0.000000	63	1	0	-7.405012	-2.029383	-0.900989
52	8	0	5.554889	2.406688	0.000000	64	1	0	-7.214947	-3.560775	0.000000
53	6	0	6.994437	2.494588	0.000000	65	6	0	6.994437	-2.494588	0.000000
54	1	0	7.405012	2.029383	0.900989	66	1	0	7.405012	-2.029383	-0.900989
55	1	0	7.405012	2.029383	-0.900989	67	1	0	7.405012	-2.029383	0.900989
56	1	0	7.214947	3.560775	0.000000	68	1	0	7.214947	-3.560775	0.000000
57	6	0	-6.994437	2.494588	0.000000	<hr/>					
58	1	0	-7.405012	2.029383	-0.900989	HF = -1980.75990959 hartree					

### Atomic coordination of $3\text{c}^{2+}$ (open shell singlet, $D_{2h}$ symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	139	140	141	142	143	144	145	146	147	148	149	150	151	152	153	154	155	156	157	158	159	160	161	162	163	164	165	166	167	168	169	170	171	172	173	174	175	176	177	178	179	180	181	182	183	184	185	186	187	188	189	190	191	192	193	194	195	196	197	198	199	200	201	202	203	204	205	206	207	208	209	210	211	212	213	214	215	216	217	218	219	220	221	222	223	224	225	226	227	228	229	230	231	232	233	234	235	236	237	238	239	240	241	242	243	244	245	246	247	248	249	250	251	252	253	254	255	256	257	258	259	260	261	262	263	264	265	266	267	268	269	270	271	272	273	274	275	276	277	278	279	280	281	282	283	284	285	286	287	288	289	290	291	292	293	294	295	296	297	298	299	300	301	302	303	304	305	306	307	308	309	310	311	312	313	314	315	316	317	318	319	320	321	322	323	324	325	326	327	328	329	330	331	332	333	334	335	336	337	338	339	340	341	342	343	344	345	346	347	348	349	350	351	352	353	354	355	356	357	358	359	360	361	362	363	364	365	366	367	368	369	370	371	372	373	374	375	376	377	378	379	380	381	382	383	384	385	386	387	388	389	390	391	392	393	394	395	396	397	398	399	400	401	402	403	404	405	406	407	408	409	410	411	412	413	414	415	416	417	418	419	420	421	422	423	424	425	426	427	428	429	430	431	432	433	434	435	436	437	438	439	440	441	442	443	444	445	446	447	448	449	450	451	452	453	454	455	456	457	458	459	460	461	462	463	464	465	466	467	468	469	470	471	472	473	474	475	476	477	478	479	480	481	482	483	484	485	486	487	488	489	490	491	492	493	494	495	496	497	498	499	500	501	502	503	504	505	506	507	508	509	510	511	512	513	514	515	516	517	518	519	520	521	522	523	524	525	526	527	528	529	530	531	532	533	534	535	536	537	538	539	540	541	542	543	544	545	546	547	548	549	550	551	552	553	554	555	556	557	558	559	560	561	562	563	564	565	566	567	568	569	570	571	572	573	574	575	576	577	578	579	580	581	582	583	584	585	586	587	588	589	590	591	592	593	594	595	596	597	598	599	600	601	602	603	604	605	606	607	608	609	610	611	612	613	614	615	616	617	618	619	620	621	622	623	624	625	626	627	628	629	630	631	632	633	634	635	636	637	638	639	640	641	642	643	644	645	646	647	648	649	650	651	652	653	654	655	656	657	658	659	660	661	662	663	664	665	666	667	668	669	670	671	672	673	674	675	676	677	678	679	680	681	682	683	684	685	686	687	688	689	690	691	692	693	694	695	696	697	698	699	700	701	702	703	704	705	706	707	708	709	710	711	712	713	714	715	716	717	718	719	720	721	722	723	724	725	726	727	728	729	730	731	732	733	734	735	736	737	738	739	740	741	742	743	744	745	746	747	748	749	750	751	752	753	754	755	756	757	758	759	760	761	762	763	764	765	766	767	768	769	770	771	772	773	774	775	776	777	778	779	780	781	782	783	784	785	786	787	788	789	790	791	792	793	794	795	796	797	798	799	800	801	802	803	804	805	806	807	808	809	810	811	812	813	814	815	816	817	818	819	820	821	822	823	824	825	826	827	828	829	830	831	832	833	834	835	836	837	838	839	840	841	842	843	844	845	846	847	848	849	850	851	852	853	854	855	856	857	858	859	860	861	862	863	864	865	866	867	868	869	870	871	872	873	874	875	876	877	878</th

### Atomic coordination of $3c^{2+}$ (triplet, $D_{2h}$ symmetry)

Center	Atomic Number	Atomic Type	Coordinates (Angstroms)								
Number	Number	Type	X	Y	Z	34	1	0	-3.960943	4.341005	0.000000
1	6	0	1.420145	0.000000	0.000000	35	7	0	-1.353093	-2.379246	0.000000
2	6	0	0.691048	1.187071	0.000000	36	6	0	-0.710716	-3.626566	0.000000
3	6	0	-0.691048	1.187071	0.000000	37	6	0	-2.729459	-2.490629	0.000000
4	6	0	-1.420145	0.000000	0.000000	38	6	0	-1.763624	-4.575045	0.000000
5	6	0	-0.691048	-1.187071	0.000000	39	6	0	-2.980265	-3.892865	0.000000
6	6	0	0.691048	-1.187071	0.000000	40	1	0	-1.637190	-5.648717	0.000000
7	6	0	-2.843012	0.000000	0.000000	41	1	0	-3.960943	-4.341005	0.000000
8	6	0	-3.508562	-1.267032	0.000000	42	7	0	1.353093	-2.379246	0.000000
9	6	0	-3.508562	1.267032	0.000000	43	6	0	0.710716	-3.626566	0.000000
10	6	0	-4.933232	-1.229259	0.000000	44	6	0	2.729459	-2.490629	0.000000
11	6	0	-4.933232	1.229259	0.000000	45	6	0	1.763624	-4.575045	0.000000
12	6	0	-5.609201	0.000000	0.000000	46	6	0	2.980265	-3.892865	0.000000
13	1	0	-6.689638	0.000000	0.000000	47	1	0	1.637190	-5.648717	0.000000
14	6	0	2.843012	0.000000	0.000000	48	1	0	3.960943	-4.341005	0.000000
15	6	0	3.508562	-1.267032	0.000000	49	8	0	-5.547896	2.416739	0.000000
16	6	0	3.508562	1.267032	0.000000	50	8	0	-5.547896	-2.416739	0.000000
17	6	0	4.933232	-1.229259	0.000000	51	8	0	5.547896	-2.416739	0.000000
18	6	0	4.933232	1.229259	0.000000	52	8	0	5.547896	2.416739	0.000000
19	6	0	5.609201	0.000000	0.000000	53	6	0	6.985236	2.493953	0.000000
20	1	0	6.689638	0.000000	0.000000	54	1	0	7.396003	2.026760	0.900424
21	7	0	1.353093	2.379246	0.000000	55	1	0	7.396003	2.026760	-0.900424
22	6	0	0.710716	3.626566	0.000000	56	1	0	7.215736	3.558315	0.000000
23	6	0	2.729459	2.490629	0.000000	57	6	0	-6.985236	2.493953	0.000000
24	6	0	1.763624	4.575045	0.000000	58	1	0	-7.396003	2.026760	-0.900424
25	6	0	2.980265	3.892865	0.000000	59	1	0	-7.396003	-2.026760	0.900424
26	1	0	1.637190	5.648717	0.000000	60	1	0	-7.396003	-2.026760	-0.900424
27	1	0	3.960943	4.341005	0.000000	61	6	0	-7.215736	-3.558315	0.000000
28	7	0	-1.353093	2.379246	0.000000	62	1	0	6.985236	-2.493953	0.000000
29	6	0	-0.710716	3.626566	0.000000	63	1	0	-7.396003	-2.026760	-0.900424
30	6	0	-2.729459	2.490629	0.000000	64	1	0	7.396003	-2.026760	0.900424
31	6	0	-1.763624	4.575045	0.000000	65	6	0	7.396003	-2.026760	-0.900424
32	6	0	-2.980265	3.892865	0.000000	66	1	0	7.215736	-3.558315	0.000000
33	1	0	-1.637190	5.648717	0.000000	67	1	0	7.215736	-3.558315	0.000000
						68	1	0	7.215736	-3.558315	0.000000
									HF = -1980.76533037 hartree		

### Atomic coordination of $4c$ ( $D_{3h}$ symmetry)

Center	Atomic Number	Atomic Type	Coordinates (Angstroms)								
Number	Number	Type	X	Y	Z	14	6	0	2.479539	1.431562	0.000000
1	6	0	0.000000	1.381013	0.000000	15	6	0	3.692186	0.685693	0.000000
2	6	0	-1.229891	0.710078	0.000000	16	6	0	2.439921	2.854681	0.000000
3	6	0	-1.195993	-0.690507	0.000000	17	6	0	4.895723	1.419744	0.000000
4	6	0	0.000000	-1.420156	0.000000	18	6	0	3.677396	3.529949	0.000000
5	6	0	1.195993	-0.690507	0.000000	19	6	0	4.881707	2.818455	0.000000
6	6	0	1.229891	0.710078	0.000000	20	1	0	5.817290	3.358614	0.000000
7	6	0	0.000000	-2.863125	0.000000	21	6	0	-2.479539	1.431562	0.000000
8	6	0	1.252266	-3.540374	0.000000	22	6	0	-3.692186	0.685693	0.000000
9	6	0	-1.252266	-3.540374	0.000000	23	6	0	-2.439921	2.854681	0.000000
10	6	0	1.218327	-4.949692	0.000000	24	6	0	-4.895723	1.419744	0.000000
11	6	0	-1.218327	-4.949692	0.000000	25	6	0	-3.677396	3.529949	0.000000
12	6	0	0.000000	-5.636910	0.000000	26	6	0	-4.881707	2.818455	0.000000
						27	1	0	-5.817290	3.358614	0.000000
						28	7	0	-2.381757	-1.375108	0.000000

29	6	0	-3.637676	-0.768427	0.000000	55	6	0	4.831320	5.633205	0.000000
30	6	0	-2.484315	-2.766106	0.000000	56	1	0	5.432263	5.425781	0.895551
31	6	0	-4.558287	-1.815048	0.000000	57	1	0	5.432263	5.425781	-0.895551
32	6	0	-3.851021	-3.040068	0.000000	58	1	0	4.538921	6.684915	0.000000
33	1	0	-5.629479	-1.695565	0.000000	59	6	0	2.462839	-7.000648	0.000000
34	1	0	-4.283142	-4.027490	0.000000	60	1	0	1.982732	-7.417369	0.895551
35	7	0	2.381757	-1.375108	0.000000	61	1	0	1.982732	-7.417369	-0.895551
36	6	0	3.637676	-0.768427	0.000000	62	1	0	3.519846	-7.273279	0.000000
37	6	0	2.484315	-2.766106	0.000000	63	6	0	-7.294159	1.367443	0.000000
38	6	0	4.558287	-1.815048	0.000000	64	1	0	-7.414996	1.991588	0.895551
39	6	0	3.851021	-3.040068	0.000000	65	1	0	-7.414996	1.991588	-0.895551
40	1	0	5.629479	-1.695565	0.000000	66	1	0	-8.058767	0.588364	0.000000
41	1	0	4.283142	-4.027490	0.000000	67	6	0	7.294159	1.367443	0.000000
42	7	0	0.000000	2.750216	0.000000	68	1	0	7.414996	1.991588	-0.895551
43	6	0	1.153361	3.534533	0.000000	69	1	0	7.414996	1.991588	0.895551
44	6	0	-1.153361	3.534533	0.000000	70	1	0	8.058767	0.588364	0.000000
45	6	0	0.707266	4.855116	0.000000	71	6	0	-4.831320	5.633205	0.000000
46	6	0	-0.707266	4.855116	0.000000	72	1	0	-5.432263	5.425781	-0.895551
47	1	0	1.346338	5.723054	0.000000	73	1	0	-5.432263	5.425781	0.895551
48	1	0	-1.346338	5.723054	0.000000	74	1	0	-4.538921	6.684915	0.000000
49	8	0	3.621585	4.895507	0.000000	75	6	0	-2.462839	-7.000648	0.000000
50	8	0	2.428841	-5.584138	0.000000	76	1	0	-1.982732	-7.417369	-0.895551
51	8	0	-6.050426	0.688631	0.000000	77	1	0	-1.982732	-7.417369	0.895551
52	8	0	6.050426	0.688631	0.000000	78	1	0	-3.519846	-7.273279	0.000000

HF = -2232.39686436 hartree

### Atomic coordination of $4\text{e}^{2+}$ (closed shell, $D_{3h}$ symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			25	6	0	-3.682934	3.535846	0.000000
			X	Y	Z						
1	6	0	0.000000	1.374858	0.000000	29	6	0	-3.624839	-0.760753	0.000000
2	6	0	-1.225629	0.707617	0.000000	30	6	0	-2.471251	-2.758826	0.000000
3	6	0	-1.190662	-0.687429	0.000000	31	6	0	-4.557951	-1.827163	0.000000
4	6	0	0.000000	-1.415235	0.000000	32	6	0	-3.861345	-3.033720	0.000000
5	6	0	1.190662	-0.687429	0.000000	33	1	0	-5.629347	-1.709914	0.000000
6	6	0	1.225629	0.707617	0.000000	34	1	0	-4.295502	-4.020200	0.000000
7	6	0	0.000000	-2.851805	0.000000	35	7	0	2.374623	-1.370989	0.000000
8	6	0	1.254036	-3.521893	0.000000	36	6	0	3.624839	-0.760753	0.000000
9	6	0	-1.254036	-3.521893	0.000000	37	6	0	2.471251	-2.758826	0.000000
10	6	0	1.220665	-4.957437	0.000000	38	6	0	4.557951	-1.827163	0.000000
11	6	0	-1.220665	-4.957437	0.000000	39	6	0	3.861345	-3.033720	0.000000
12	6	0	0.000000	-5.642121	0.000000	40	1	0	5.629347	-1.709914	0.000000
13	1	0	0.000000	-6.722035	0.000000	41	1	0	4.295502	-4.020200	0.000000
14	6	0	2.469735	1.425902	0.000000	42	7	0	0.000000	2.741978	0.000000
15	6	0	3.677066	0.674920	0.000000	43	6	0	1.153588	3.519579	0.000000
16	6	0	2.423031	2.846973	0.000000	44	6	0	-1.153588	3.519579	0.000000
17	6	0	4.903599	1.421592	0.000000	45	6	0	0.696606	4.860882	0.000000
18	6	0	3.682934	3.535846	0.000000	46	6	0	-0.696606	4.860882	0.000000
19	6	0	4.886220	2.821061	0.000000	47	1	0	1.333844	5.730114	0.000000
20	1	0	5.821453	3.361017	0.000000	48	1	0	-1.333844	5.730114	0.000000
21	6	0	-2.469735	1.425902	0.000000	49	8	0	3.619588	4.867181	0.000000
22	6	0	-3.677066	0.674920	0.000000	50	8	0	2.405308	-5.568245	0.000000
23	6	0	-2.423031	2.846973	0.000000	51	8	0	-6.024896	0.701065	0.000000
24	6	0	-4.903599	1.421592	0.000000	52	8	0	6.024896	0.701065	0.000000

53	8	0	-3.619588	4.867181	0.000000	67	6	0	7.311473	1.346739	0.000000
54	8	0	-2.405308	-5.568245	0.000000	68	1	0	7.435109	1.956257	-0.900373
55	6	0	4.822047	5.658552	0.000000	69	1	0	7.435109	1.956257	0.900373
56	1	0	5.411723	5.460865	0.900373	70	1	0	8.038769	0.536253	0.000000
57	1	0	5.411723	5.460865	-0.900373	71	6	0	-4.822047	5.658552	0.000000
58	1	0	4.483794	6.693652	0.000000	72	1	0	-5.411723	5.460865	-0.900373
59	6	0	2.489426	-7.005291	0.000000	73	1	0	-5.411723	5.460865	0.900373
60	1	0	2.023386	-7.417122	0.900373	74	1	0	-4.483794	6.693652	0.000000
61	1	0	2.023386	-7.417122	-0.900373	75	6	0	-2.489426	-7.005291	0.000000
62	1	0	3.554976	-7.229905	0.000000	76	1	0	-2.023386	-7.417122	-0.900373
63	6	0	-7.311473	1.346739	0.000000	77	1	0	-2.023386	-7.417122	0.900373
64	1	0	-7.435109	1.956257	0.900373	78	1	0	-3.554976	-7.229905	0.000000
65	1	0	-7.435109	1.956257	-0.900373						
66	1	0	-8.038769	0.536253	0.000000						

HF = -2231.91992422 hartree

### Atomic coordination of **4c<sup>2+</sup>** (open shell singlet, *C<sub>2v</sub>* symmetry)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			37	6	0	0.000000	-4.558060	-1.826975
			X	Y	Z						
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2	6	0	0.000000	1.225782	0.707898	40	1	0	0.000000	-4.295530	-4.019864
3	6	0	0.000000	1.190752	-0.687072	41	7	0	0.000000	0.000000	2.742069
4	6	0	0.000000	0.000000	-1.415035	42	6	0	0.000000	-1.153780	3.520147
5	6	0	0.000000	-1.190752	-0.687072	43	6	0	0.000000	1.153780	3.520147
6	6	0	0.000000	-1.225782	0.707898	44	6	0	0.000000	-0.696909	4.860650
7	6	0	0.000000	0.000000	-2.851248	45	6	0	0.000000	0.696909	4.860650
8	6	0	0.000000	-1.254160	-3.521678	46	1	0	0.000000	-1.333735	5.730180
9	6	0	0.000000	1.254160	-3.521678	47	1	0	0.000000	1.333735	5.730180
10	6	0	0.000000	-1.220811	-4.956907	48	8	0	0.000000	-3.620535	4.866842
11	6	0	0.000000	1.220811	-4.956907	49	8	0	0.000000	-2.405555	-5.567829
12	6	0	0.000000	0.000000	-5.641594	50	8	0	0.000000	6.025710	0.700169
13	6	0	0.000000	-2.469787	1.426266	51	8	0	0.000000	-6.025710	0.700169
14	6	0	0.000000	-3.677500	0.674797	52	8	0	0.000000	3.620535	4.866842
15	6	0	0.000000	-2.423493	2.846944	53	8	0	0.000000	2.405555	-5.567829
16	6	0	0.000000	-4.904341	1.421061	54	6	0	0.000000	-4.823179	5.657688
17	6	0	0.000000	-3.683483	3.535397	55	1	0	-0.900490	-5.412510	5.459525
18	6	0	0.000000	-4.886906	2.820443	56	1	0	0.900490	-5.412510	5.459525
19	1	0	0.000000	-5.821924	3.360843	57	1	0	0.000000	-4.485450	6.692959
20	6	0	0.000000	2.469787	1.426266	58	6	0	0.000000	-2.489348	-7.004848
21	6	0	0.000000	3.677500	0.674797	59	1	0	-0.900391	-2.023148	-7.416514
22	6	0	0.000000	2.423493	2.846944	60	1	0	0.900391	-2.023148	-7.416514
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25	6	0	0.000000	4.886906	2.820443	63	1	0	-0.900458	7.435310	1.955330
26	1	0	0.000000	5.821924	3.360843	64	1	0	0.900458	7.435310	1.955330
27	7	0	0.000000	2.374519	-1.370857	65	1	0	0.900458	-7.435310	1.955330
28	6	0	0.000000	3.624968	-0.760322	66	1	0	-0.900458	-7.435310	1.955330
29	6	0	0.000000	2.471281	-2.758358	67	1	0	0.000000	-8.039464	0.535530
30	6	0	0.000000	4.558060	-1.826975	68	6	0	0.000000	4.823179	5.657688
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32	1	0	0.000000	5.629313	-1.709329	70	1	0	-0.900490	5.412510	5.459525
33	1	0	0.000000	4.295530	-4.019864	71	1	0	0.000000	4.485450	6.692959
34	7	0	0.000000	-2.374519	-1.370857	72	6	0	0.000000	2.489348	-7.004848
35	6	0	0.000000	-3.624968	-0.760322	73	1	0	0.900391	2.023148	-7.416514
36	6	0	0.000000	-2.471281	-2.758358	74	1	0	-0.900391	2.023148	-7.416514

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78	1	0	0.000000	0.000000	-6.721428

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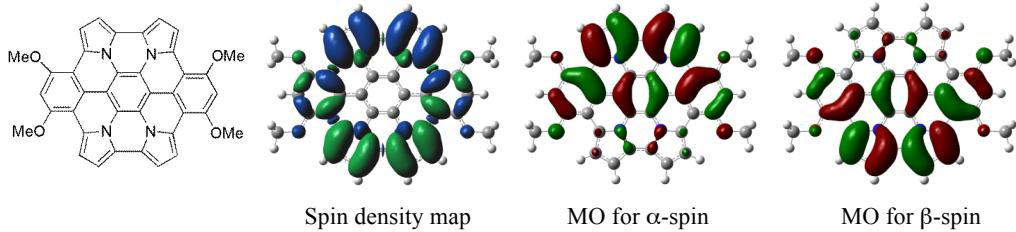
HF = -2231.91993625 hartree

### Atomic coordination of $4c^{2+}$ (triplet, $C_s$ symmetry)

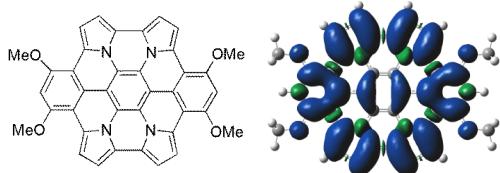
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			40	41	42						
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2	6	0	1.227062	0.719100	0.000000	44	6	0	3.635909	-0.739421	0.000000
3	6	0	0.000000	1.376311	0.000000	45	6	0	3.864496	-3.013160	0.000000
4	6	0	-1.227669	0.699973	0.000000	46	6	0	4.561174	-1.790285	0.000000
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6	6	0	0.006702	-1.422485	0.000000	48	1	0	5.632505	-1.672917	0.000000
7	6	0	-2.465634	1.405472	0.000000	49	8	0	2.423736	-5.556444	0.000000
8	6	0	-3.679740	0.649220	0.000000	50	8	0	-6.023335	0.654685	0.000000
9	6	0	-2.433488	2.836287	0.000000	51	8	0	3.586914	4.909052	0.000000
10	6	0	-4.899887	1.384523	0.000000	52	8	0	-2.395348	-5.588019	0.000000
11	6	0	-3.687976	3.511585	0.000000	53	8	0	6.015286	0.745677	0.000000
12	6	0	-4.884827	2.784895	0.000000	54	8	0	-3.632573	4.850263	0.000000
13	1	0	-5.823325	3.319844	0.000000	55	6	0	2.517297	-6.992371	0.000000
14	6	0	0.016890	-2.856662	0.000000	56	1	0	2.054310	-7.407764	0.900365
15	6	0	-1.244170	-3.539130	0.000000	57	1	0	2.054310	-7.407764	-0.900365
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17	6	0	-1.207859	-4.973995	0.000000	59	6	0	-7.307466	1.300897	0.000000
18	6	0	1.234010	-4.952656	0.000000	60	1	0	-7.433186	1.911021	0.900056
19	6	0	0.014911	-5.647042	0.000000	61	1	0	-7.433186	1.911021	-0.900056
20	1	0	0.024601	-6.727087	0.000000	62	1	0	-8.036856	0.491998	0.000000
21	6	0	2.465983	1.440816	0.000000	63	6	0	4.786164	5.703075	0.000000
22	6	0	2.410703	2.874362	0.000000	64	1	0	5.377133	5.507313	0.900179
23	6	0	3.674145	0.706070	0.000000	65	1	0	5.377133	5.507313	-0.900179
24	6	0	3.663458	3.574062	0.000000	66	1	0	4.446381	6.737790	0.000000
25	6	0	4.889206	1.461587	0.000000	67	6	0	-2.467853	-7.024572	0.000000
26	6	0	4.865352	2.864550	0.000000	68	1	0	-1.998299	-7.433426	-0.900168
27	1	0	5.799642	3.406590	0.000000	69	1	0	-1.998299	-7.433426	0.900168
28	7	0	-0.013733	2.741126	0.000000	70	1	0	-3.531370	-7.259137	0.000000
29	6	0	1.144562	3.531117	0.000000	71	6	0	7.298234	1.397098	0.000000
30	6	0	-1.162744	3.509234	0.000000	72	1	0	7.419639	2.007088	-0.900387
31	6	0	0.676063	4.871286	0.000000	73	1	0	7.419639	2.007088	0.900387
32	6	0	-0.712825	4.859749	0.000000	74	1	0	8.029470	0.590107	0.000000
33	1	0	1.309386	5.743119	0.000000	75	6	0	-4.843168	5.625543	0.000000
34	1	0	-1.361322	5.720662	0.000000	76	1	0	-5.432168	5.422656	-0.900066
35	7	0	-2.365198	-1.384780	0.000000	77	1	0	-5.432168	5.422656	0.900066
36	6	0	-2.454926	-2.784826	0.000000	78	1	0	-4.519208	6.665442	0.000000
37	6	0	-3.611512	-0.787772	0.000000	-----	-----	-----	-----	-----	-----
38	6	0	-3.845946	-3.064115	0.000000	HF = -2231.923508 hartree					

**S20. Spin density maps and possible resonance structures of  $3^{2+}$  and  $4^{2+}$**

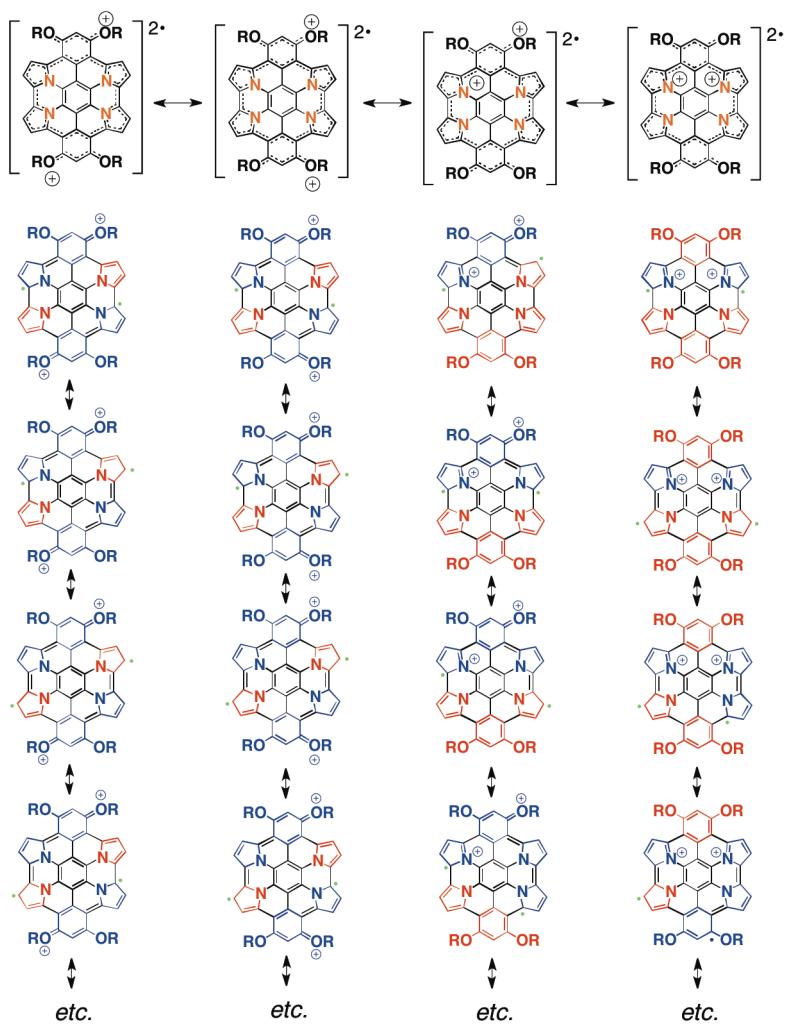
Spin density map and MOs of  $3c^{2+}$  in singlet biradical state.



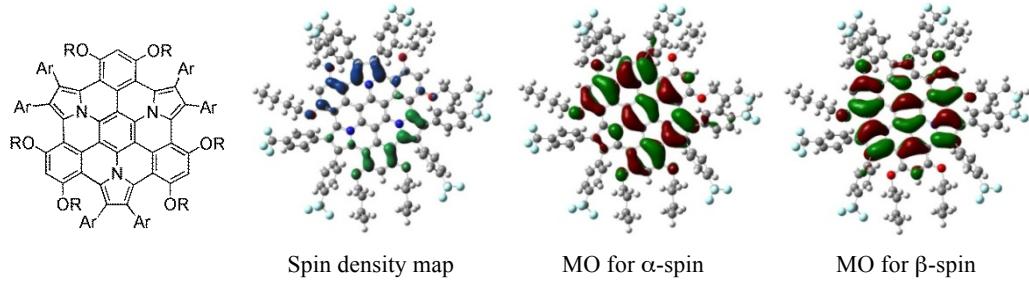
Spin density maps of  $3c^{2+}$  in triplet state.



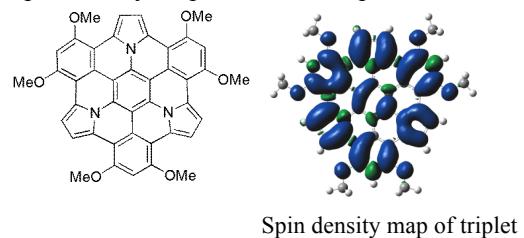
Possible resonance structures of  $3c^{2+}$



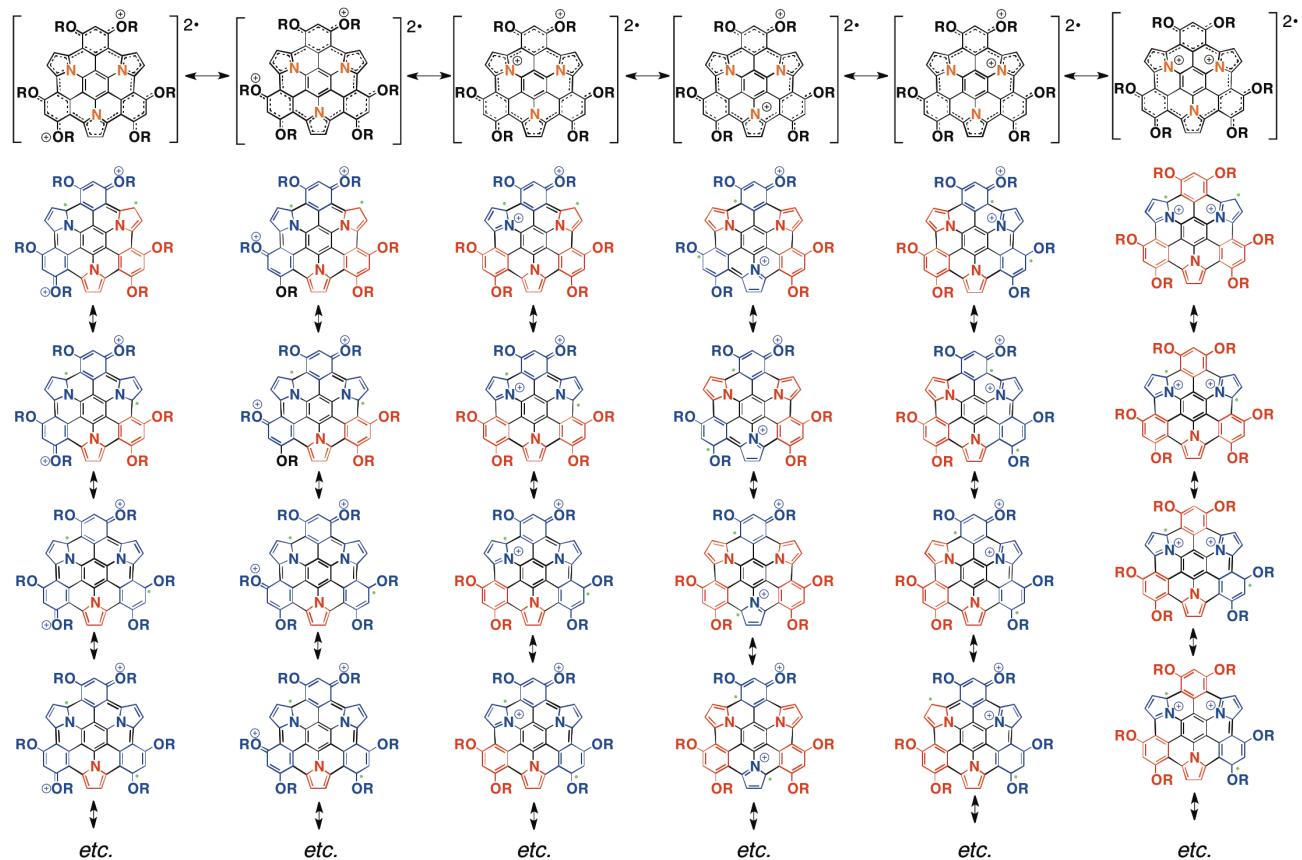
Spin density map and MOs of  $\mathbf{4b}^{2+}$  ( $R = n\text{-C}_4\text{H}_9$ ) in singlet biradical state. The geometry was taken from the crystal structure.



Spin density maps of  $\mathbf{4c}^{2+}$  in triplet state.

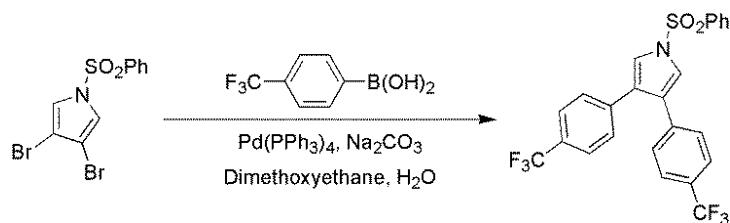


Possible resonance structures of  $\mathbf{4c}^{2+}$



## S21. Synthetic procedures

### 1. *N*-Benzenesulfonyl-3,4-bis(4-trifluoromethylphenyl)pyrrole



In a double-neck round-bottom flask, were added *N*-benzenesulfonyl-3,4-dibromopyrrole (6.40 g, 17.5 mmol), 4-trifluoromethylphenylboronic acid (9.95 g, 52.4 mmol),  $\text{Na}_2\text{CO}_3$  (12.25 g, 115.6 mmol), and  $\text{Pd}(\text{PPh}_3)_4$  (0.93 g, 0.8 mmol), and then evacuated and backfilled with  $\text{N}_2$  three times. Then, degassed dimethoxyethane (140 ml) and  $\text{H}_2\text{O}$  (38 ml) were added. The reaction mixture was stirred at reflux for overnight. After addition of  $\text{H}_2\text{O}$ , the aqueous phase was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic layers were washed with sat- $\text{NaCl}$  aq., and dried over  $\text{MgSO}_4$ , and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/ $\text{AcOEt}$  4:1) to give *N*-benzenesulfonyl-3,4-bis(4-trifluoromethylphenyl)pyrrole (7.66 g, 15.5 mmol) as white solid, yield: 88%.

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ , 500 MHz)  $\delta$  7.99 (d,  $J = 7.5$  Hz, 2H), 7.68 (t,  $J = 7.5$  Hz, 1H), 7.59 (t,  $J = 7.6$  Hz, 2H), 7.53 (d,  $J = 8.2$  Hz, 4H), 7.34 (s, 2H), 7.27 (d,  $J = 8.2$  Hz, 4H).

$^{13}\text{C NMR}$  ( $\text{CDCl}_3$ , 125 MHz)  $\delta$  138.48, 136.72, 134.43, 129.73, 129.50, 129.24, 128.69, 127.20, 127.14, 125.49 (q,  $J = 3.6$  Hz,  $-\text{CF}_3$ ), 119.79.

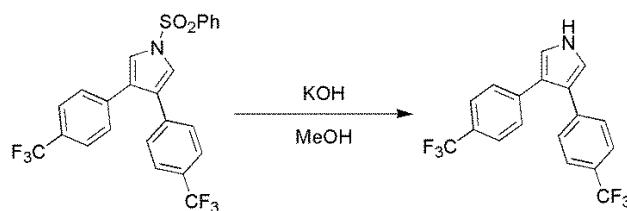
$^{19}\text{F NMR}$  ( $\text{CDCl}_3$ , 470.5 MHz)  $\delta$  -65.71.

EI-MS  $m/z$  495 ( $\text{M}^+$ )

Anal. Calcd for  $\text{C}_{24}\text{H}_{15}\text{F}_6\text{NO}_2\text{S}$ : C, 58.18%; H, 3.05%; N, 2.83%. Found: C, 58.53%; H, 3.06%; N, 2.66%.

Melting point: 125.4 – 127.3 °C

### 2. 3,4-Bis(4-trifluoromethylphenyl)-1*H*-pyrrole



In a round-bottom flask, was added  $\text{MeOH}$  (1000 ml) to *N*-benzenesulfonyl-3,4-bis(4-trifluoromethylphenyl)pyrrole (7.66 g, 15.5 mmol) and KOH (9.15 g, 163.0 mmol). The reaction mixture was stirred at room temperature for 4 h. The reaction was quenched by adding 2N-HCl aq. until the solution to be pH 7. The aqueous phase was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic layers were washed with sat- $\text{NaCl}$  aq., and dried over  $\text{MgSO}_4$ , and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/ $\text{CH}_2\text{Cl}_2$  3:2) to give 3,4-bis(4-trifluoromethylphenyl)-1*H*-pyrrole (5.26 g, 14.8 mmol) as white solid, yield: 96%.

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ , 500 MHz)  $\delta$  8.44 (s, br, 1H), 7.52 (d,  $J = 8.0$  Hz, 4H), 7.34 (d,  $J = 8.0$  Hz, 4H), 6.99 (s,

2H).

$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 125 MHz)  $\delta$  139.03, 128.52, 128.16, 125.49, 125.3 (q,  $J = 3.7$  Hz, - $\text{CF}_3$ ), 122.53, 118.44.

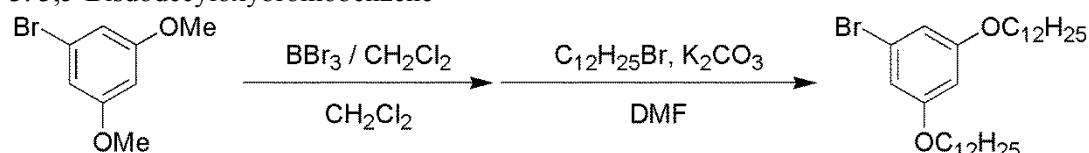
$^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 470.5 MHz)  $\delta$  -65.47.

EI-MS  $m/z$  356 ( $\text{M}^+$ )

Anal. Calcd for  $\text{C}_{18}\text{H}_{11}\text{F}_6\text{N}$ : C, 60.85%; H, 3.12%; N, 3.94%. Found: C, 60.89%; H, 3.38%; N, 3.99%.

Melting point: 139.8 – 145.3 °C

### 3. 3,5-Bisdodecyloxybromobenzene



In a three-necked round-bottom flask equipped with a thermometer, was added  $\text{CH}_2\text{Cl}_2$  (100 ml) to 3,5-dimethoxybromobenzene (10.0 g, 46.2 mmol). The solution of 3,5-dimethoxybromobenzene was cooled to -78 °C, and 1 M- $\text{CH}_2\text{Cl}_2$  solution of  $\text{BBr}_3$  (100 ml, 100.0 mmol) was added slowly. The reaction mixture was allowed to warm up to room temperature for overnight. The stirred solution was again cooled to 0 °C, and then quenched with  $\text{H}_2\text{O}$ . The aqueous phase was extracted with  $\text{Et}_2\text{O}$ . The combined organic layers were washed with  $\text{H}_2\text{O}$ , dried over  $\text{MgSO}_4$ , and evaporated under reduced pressure. Resulting white solids were used to the next reaction without further purification.

In a double-neck round-bottom flask, was added 3,5-dihydroxybromobenzene (8.73 g, 46.2 mmol),  $\text{K}_2\text{CO}_3$  (47.2 g, 341.5 mmol) and DMF (200 ml). The reaction mixture was stirred at room temperature for 30 min. Then dodecylbromide (45 ml, 187.8 mmol) was added and stirred at 60 °C for overnight. Resulting suspension was passed through celite, and the filtrate was evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane) to afford 3,5-bisdodecyloxybromobenzene (19.5 g, 37.1 mmol) as white solid, yield: 80% (2-steps).

$^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 500 MHz)  $\delta$  6.62 (d,  $J = 2.2$  Hz, 2H), 6.35 (t,  $J = 2.2$  Hz, 1H), 3.88 (t,  $J = 6.6$  Hz, 4H), 1.77-1.72 (m, 4H), 1.44-1.39 (m, 4H), 1.33-1.21 (m, 32H), 0.88 (t,  $J = 7.1$  Hz, 6H).

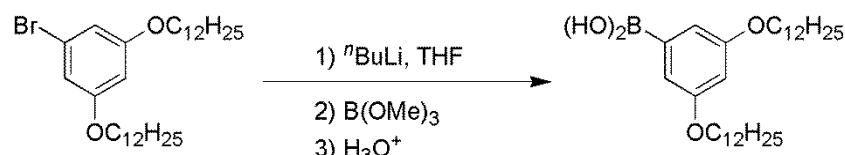
$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 125 MHz)  $\delta$  160.76, 122.82, 110.23, 100.60, 68.31, 31.94, 29.67, 29.65, 29.60, 29.57, 29.36, 29.13, 25.99 (2C), 22.71, 14.13.

EI-MS  $m/z$  524 ( $\text{M}^+$ )

Anal. Calcd for  $\text{C}_{30}\text{H}_{53}\text{BrO}_2$ : C, 68.55%; H, 10.16%. Found: C, 68.19%; H, 9.86%.

Melting point: 32.9 – 34.8 °C

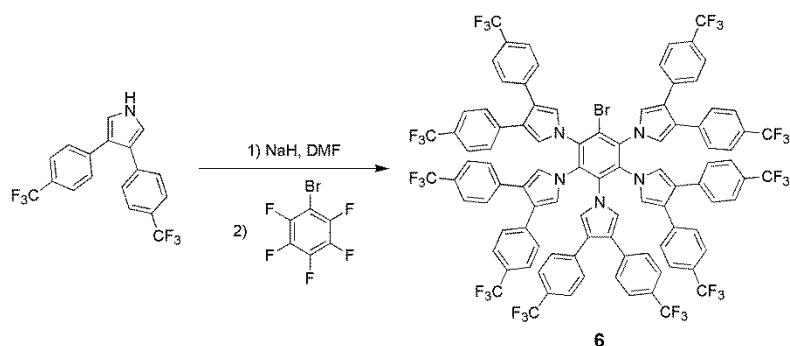
### 4. 3,5-Bisdodecyloxyphenylboronic acid



In a double-neck round-bottom flask, was added THF (10 ml) to 3,5-bisdodecyloxybromobenzene (531 mg,

1.01 mmol). The solution was cooled to -5 °C, and then <sup>7</sup>BuLi (1.61 M in hexane, 0.7 ml, 1.13 mmol) was added dropwise. After completion of the addition, the reaction mixture was further cooled to -78 °C, and stirred for 30 min. Then, B(OMe)<sub>3</sub> (0.2 ml, 1.79 mmol) was added in one portion. After stirring at room temperature for 2 h, the reaction was quenched with 2N-HCl and stirred for 30 min. The aqueous phase extracted with Et<sub>2</sub>O, and the combined organic layers were washed with sat-NaCl aq., dried over MgSO<sub>4</sub>, and evaporated under reduced pressure. Resulting solids were used to the next reaction without further purification.

### 5. Pentakis[3,4-bis(4-trifluoromethylphenyl)-1*H*-pyrrolo]bromobenzene (**6**)



In a double-neck round-bottom flask, was added DMF (4 ml) to 3,4-bis(4-trifluoromethylphenyl)-1*H*-pyrrole (402 mg, 1.13 mmol) and sodium hydride (60% in mineral oil, 55 mg, 1.37 mmol) at 0 °C. The reaction mixture was stirred at 0 °C for 20 min. And then pentafluorobromobenzene (28 µl, 0.22 mmol) was added in one portion. After stirring at room temperature for overnight, the reaction was quenched with H<sub>2</sub>O, and the aqueous phase was extracted with Et<sub>2</sub>O. The combined organic layers were washed with sat-NaCl aq., and dried over MgSO<sub>4</sub>, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/CH<sub>2</sub>Cl<sub>2</sub> 2:1) to afford pentakis[3,4-bis(4-trifluoromethylphenyl)-1*H*-pyrrolo]bromobenzene (**6**, 340 mg, 0.18 mmol) as white solid, yield: 82%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.57 (d, *J* = 8.3 Hz, 8H), 7.52-7.47 (m, 12H), 7.31 (d, *J* = 8.2 Hz, 8H), 7.13-7.09 (m, 12H), 6.96 (s, 4H), 6.56 (s, 4H), 6.55 (s, 2H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 137.35, 137.13, 136.99, 135.26, 132.68, 129.48, 129.35, 129.22, 129.09, 128.51, 128.23, 128.19, 125.88-125.69 (m, -CF<sub>3</sub>, 3C), 125.60, 125.26, 125.20, 125.07, 123.24, 123.04, 122.91, 122.16, 120.92, 120.75.

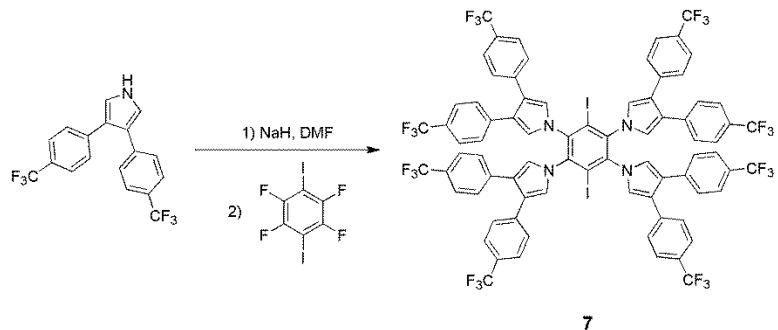
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.56, -65.63, -65.65.

APCI-MS calcd. for C<sub>96</sub>H<sub>50</sub>BrF<sub>30</sub>N<sub>5</sub>: 1921.2770, Found: 1921.2782.

Anal. Calcd for C<sub>96</sub>H<sub>50</sub>BrF<sub>30</sub>N<sub>5</sub>: C, 59.95%; H, 2.62%; N, 3.64%. Found: C, 59.95%; H, 2.72%; N, 3.61%.

Melting Point: 298.5 – 300.2 °C

6. 2,3,5,6-Tetrakis[3,4-bis(4-trifluoromethylphenyl)-*IH*-pyrrolo]-1,4-diiodobenzene (**7**)



In a double-neck round-bottom flask, was added DMF (4 ml) to 3,4-bis(4-trifluoromethylphenyl)-*IH*-pyrrole (397 mg, 1.12 mmol) and sodium hydride (60% in mineral oil, 47 mg, 1.17 mmol) at 0 °C. The reaction mixture was stirred at 0 °C for 20 min. And then 1,4-diiodotetrafluorobenzene (108 mg, 0.48 mmol) was added in one portion. After stirring at room temperature for overnight, the reaction was quenched with H<sub>2</sub>O, and the aqueous phase extracted with Et<sub>2</sub>O. The combined organic layers were washed with sat-NaCl aq., and dried over MgSO<sub>4</sub>, and evaporated under reduced pressure. The residue was dispersed in a small amount of CH<sub>2</sub>Cl<sub>2</sub>, and insoluble solids were collected by filtration to afford 2,3,5,6-tetrakis[3,4-bis(4-trifluoromethylphenyl)-*IH*-pyrrolo]-1,4-diiodobenzene (**7**, 330 mg, 0.19 mmol) as white solid, yield: 70%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.49 (d, *J* = 8.1 Hz, 16H), 7.21 (d, *J* = 8.1 Hz, 16H), 6.80 (s, 8H).

<sup>13</sup>C NMR peaks were not able to be observed due to quite low solubility.

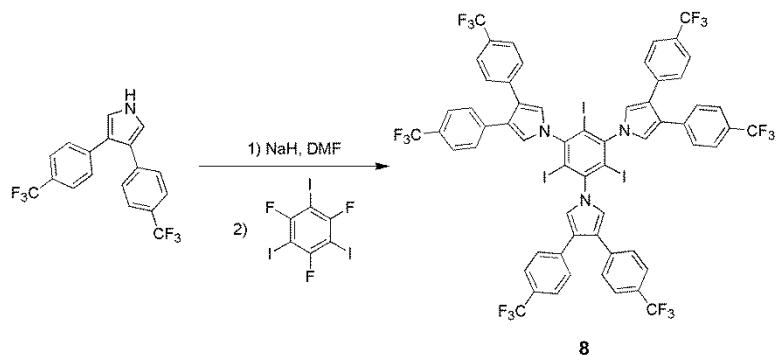
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.63.

APCI-MS calcd. for C<sub>78</sub>H<sub>40</sub>F<sub>24</sub>I<sub>2</sub>N<sub>4</sub>: 1742.0949, Found: 1742.0950.

Anal. Calcd for C<sub>78</sub>H<sub>40</sub>F<sub>24</sub>I<sub>2</sub>N<sub>4</sub>: C, 53.75%; H, 2.31%; N, 3.21%. Found: C, 53.76%; H, 2.25%; N, 3.21%.

Melting Point: > 300 °C

7. 2,4,6-Tris[3,4-bis(4-trifluoromethylphenyl)-*IH*-pyrrolo]-1,3,5-triiodobenzene (**8**)



In a double-neck round-bottom flask, was added DMF (5 ml) to 3,4-bis(4-trifluoromethylphenyl)-*IH*-pyrrole (591 mg, 1.66 mmol) and sodium hydride (60% in mineral oil, 70 mg, 1.75 mmol) at 0 °C. The reaction mixture was stirred at 0 °C for 20 min. And then 2,4,6-trifluoro-1,3,5-triiodobenzene (278 mg, 0.55 mmol) was added in one portion. After stirring at room temperature for overnight, the reaction was quenched with H<sub>2</sub>O, and the aqueous phase extracted with Et<sub>2</sub>O. The combined organic layers were washed with sat-NaCl aq.,

and dried over  $\text{MgSO}_4$ , and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/ $\text{CH}_2\text{Cl}_2$  2:1) to afford 2,4,6-tris[3,4-bis(4-trifluoromethylphenyl)-1*H*-pyrrolo]-1,3,5-tri iodobenzene (**8**, 746 mg, 0.49 mmol) as white solid, yield: 90%.

$^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 500 MHz)  $\delta$  7.57 (d,  $J = 8.2$  Hz, 12H), 7.42 (d,  $J = 8.2$  Hz, 12H), 6.89 (s, 6H).

$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 125 MHz)  $\delta$  148.01, 138.10, 128.63, 125.45 (q,  $J = 3.7$  Hz, - $\text{CF}_3$ ), 125.35, 124.58, 123.18, 120.57, 103.32.

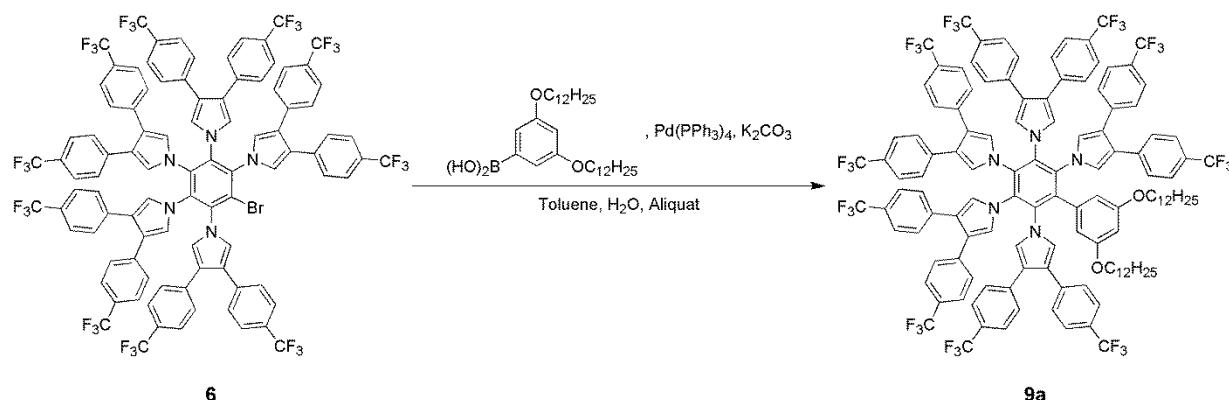
$^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 470.5 MHz)  $\delta$  -65.54.

APCI-MS calcd. for  $\text{C}_{60}\text{H}_{30}\text{F}_{18}\text{I}_3\text{N}_3$ : 1514.9287, Found: 1514.9276.

Anal. Calcd for  $\text{C}_{60}\text{H}_{30}\text{F}_{18}\text{I}_3\text{N}_3$ : C, 47.55%; H, 2.00%; N, 2.77%. Found: C, 48.00%; H, 2.21%; N, 2.71%.

Melting Point: 189.4 – 192.5 °C

### 8. 5AC( $\text{OC}_{12}\text{H}_{25}$ )-prefused (**9a**)



In a double-neck round-bottom flask, were added **6** (313 mg, 0.16 mmol), 3,5-bis(dodecyloxy)phenylboronic acid (493 mg, 1.00 mmol),  $\text{K}_2\text{CO}_3$  (142 mg, 1.03 mmol), toluene (5 ml),  $\text{H}_2\text{O}$  (2 ml), aliquat336 (3 drops). Then evacuated and backfilled with  $\text{N}_2$  three times.  $\text{Pd}(\text{PPh}_3)_4$  (48 mg, 42  $\mu\text{mol}$ ) was added to the reaction mixture. After stirring at reflux for overnight. The reaction mixture was cooled to room temperature and quenched with  $\text{H}_2\text{O}$ , and the aqueous phase was extracted with  $\text{Et}_2\text{O}$ . The combined organic layers were washed with sat- $\text{NaCl}$  aq., and dried over  $\text{MgSO}_4$ , and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/ $\text{CH}_2\text{Cl}_2$  3:1) to afford **9a** (240 mg, 0.10 mmol) as white solid, yield: 64%.

$^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 500 MHz)  $\delta$  7.55 - 7.51 (m, 20H), 7.22 - 7.18 (m, 12H), 7.10 (d,  $J = 8.1$  Hz, 8H), 6.68 (s, 2H), 6.66 - 6.64 (m, 5H), 6.62 (s, 4H), 6.31 (d,  $J = 2.1$  Hz, 2H), 3.77 (t,  $J = 6.5$  Hz, 4H), 1.66 (p,  $J = 6.5$  Hz, 4H), 1.28 - 1.19 (m, 36H), 0.86 (t,  $J = 6.8$  Hz, 6H).

$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 125 MHz)  $\delta$  161.14, 137.76, 137.59, 137.35, 137.29, 135.62, 133.46, 133.07, 132.64, 129.40, 129.18, 129.14, 128.93, 128.39, 128.33, 128.30, 125.84-125.63 (m, - $\text{CF}_3$ , 3C), 125.55, 125.25, 125.19, 124.83, 123.09, 123.02, 121.87, 121.06, 106.75, 103.11, 68.74, 31.91, 29.65, 29.62, 29.60(2C), 29.35, 29.06, 26.07(2C), 22.69, 14.06.

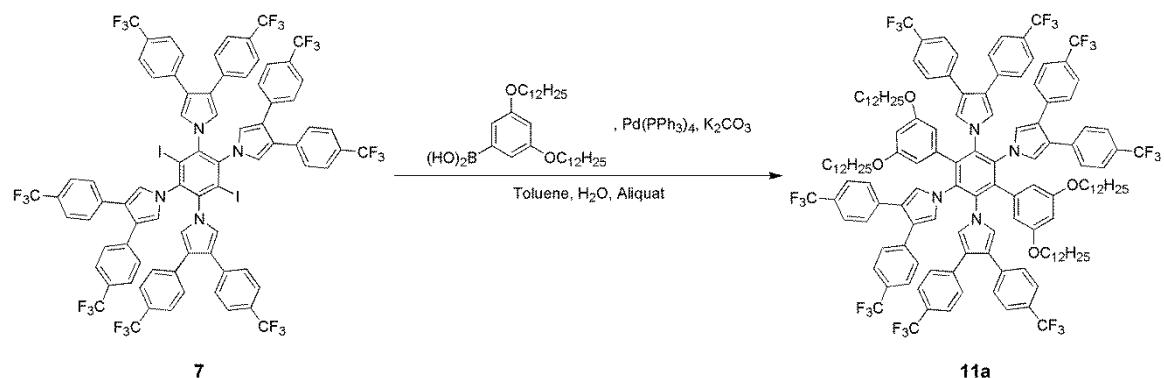
$^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 470.5 MHz)  $\delta$  -65.49, -65.52, -65.53.

APCI-MS calcd. for  $C_{126}H_{103}F_{30}N_5O_2$ : 2287.7633, Found: 2287.7636.

Anal. Calcd for  $C_{126}H_{103}F_{30}N_5O_2$ : C, 66.11%; H, 4.54%; N, 3.06%. Found: C, 65.90%; H, 4.54%; N, 2.97%.

Melting Point: 121.2 – 124.7 °C

#### 9. 4AC(OC<sub>12</sub>H<sub>25</sub>)-prefused (**11a**)



In a double-neck round-bottom flask, were added **7** (310 mg, 0.18 mmol), 3,5-bis(dodecyloxy)phenylboronic acid (493 mg, 1.00 mmol),  $K_2CO_3$  (181 mg, 1.31 mmol), toluene (5 ml),  $H_2O$  (2 ml), aliquat336 (3 drops). Then evacuated and backfilled with  $N_2$  three times.  $Pd(PPh_3)_4$  (48 mg, 42  $\mu$ mol) was added to the reaction mixture. After stirring at reflux for overnight. The reaction mixture was cooled to room temperature and quenched with  $H_2O$ , and the aqueous phase was extracted with  $Et_2O$ . The combined organic layers were washed with sat- $NaCl$  aq., and dried over  $MgSO_4$ , and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/ $CH_2Cl_2$  3:1) to afford **11a** (397 mg, 0.17 mmol) as white solid, yield: 94%.

<sup>1</sup>H-NMR ( $CDCl_3$ , 500 MHz)  $\delta$  7.46 (d,  $J$  = 8.2 Hz, 16H), 7.02 (d,  $J$  = 8.2 Hz, 16H), 6.58 (t,  $J$  = 2.1 Hz, 2H), 6.53 (s, 8H), 6.24 (d,  $J$  = 2.1 Hz, 4H), 3.71 (t,  $J$  = 6.5 Hz, 8H), 1.61 (p,  $J$  = 6.5 Hz, 8H), 1.29 - 1.15 (m, 72H), 0.86 (t,  $J$  = 7.2 Hz, 12H).

<sup>13</sup>C NMR ( $CDCl_3$ , 125 MHz)  $\delta$  160.90, 137.83, 136.88, 135.45, 134.09, 128.28, 125.49 (q,  $J$  = 3.6 Hz, -CF<sub>3</sub>), 125.22, 124.14, 123.06, 122.07, 106.79, 102.88, 68.65, 31.87, 29.60, 29.57 (2C), 29.55 (2C), 29.31, 29.03, 26.01, 22.66, 14.07.

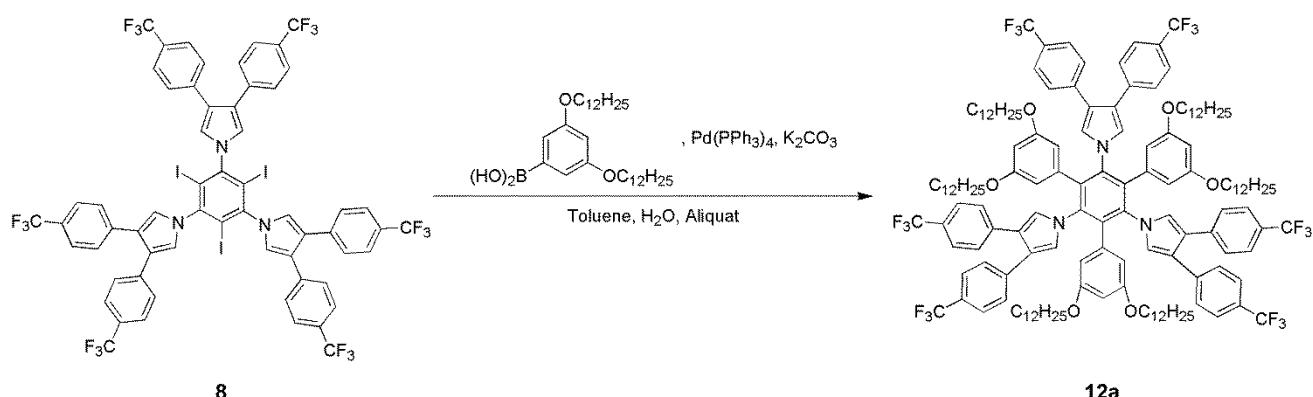
<sup>19</sup>F NMR ( $CDCl_3$ , 470.5 MHz)  $\delta$  -65.61.

APCI-MS calcd. for  $C_{138}H_{146}F_{24}N_4O_4$ : 2379.0961, Found: 2379.0941.

Anal. Calcd for  $C_{138}H_{146}F_{24}N_4O_4$ : C, 69.62%; H, 6.18%; N, 2.35%. Found: C, 69.89%; H, 6.30%; N, 2.27%.

Melting Point: 140.6 – 142.9 °C

10. 3AC(OC<sub>12</sub>H<sub>25</sub>)-prefused (**12a**)



In a double-neck round-bottom flask, were added **8** (384 mg, 0.25 mmol), 3,5-bisdodecyloxyphenyl boronic acid (987 mg, 2.00 mmol), K<sub>2</sub>CO<sub>3</sub> (382 mg, 2.76 mmol), toluene (10 ml), H<sub>2</sub>O (4 ml), aliquat336 (3 drops). Then evacuated and backfilled with N<sub>2</sub> three times. Pd(PPh<sub>3</sub>)<sub>4</sub> (29 mg, 25 μmol) was added to the reaction mixture. After stirring at reflux for overnight. The reaction mixture was cooled to room temperature and quenched with H<sub>2</sub>O, and the aqueous phase was extracted with Et<sub>2</sub>O. The combined organic layers were washed with sat-NaCl aq., and dried over MgSO<sub>4</sub>, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/CH<sub>2</sub>Cl<sub>2</sub> 3:1) to afford **12a** (431 mg, 0.17 mmol) as white solid, yield: 69%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.43 (d, *J* = 7.9 Hz, 12H), 6.94 (d, *J* = 7.9 Hz, 12H), 6.48 (s, 6H), 6.47 (m, 3H), 6.22 (m, 6H), 3.65 (t, *J* = 6.0 Hz, 12H), 1.57-1.55 (m, 12H), 1.27-1.16 (m, 108H), 0.87 (t, *J* = 6.9 Hz, 18H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 160.48, 138.37, 137.71, 136.86, 135.37, 128.33, 128.10, 125.32, 125.19 (q, *J* = 3.3 Hz, -CF<sub>3</sub>), 123.15, 122.92, 106.77, 102.47, 68.54, 31.87, 29.58, 29.56, 29.53 (2C), 29.30 (2C), 29.07, 25.96, 22.65, 14.06.

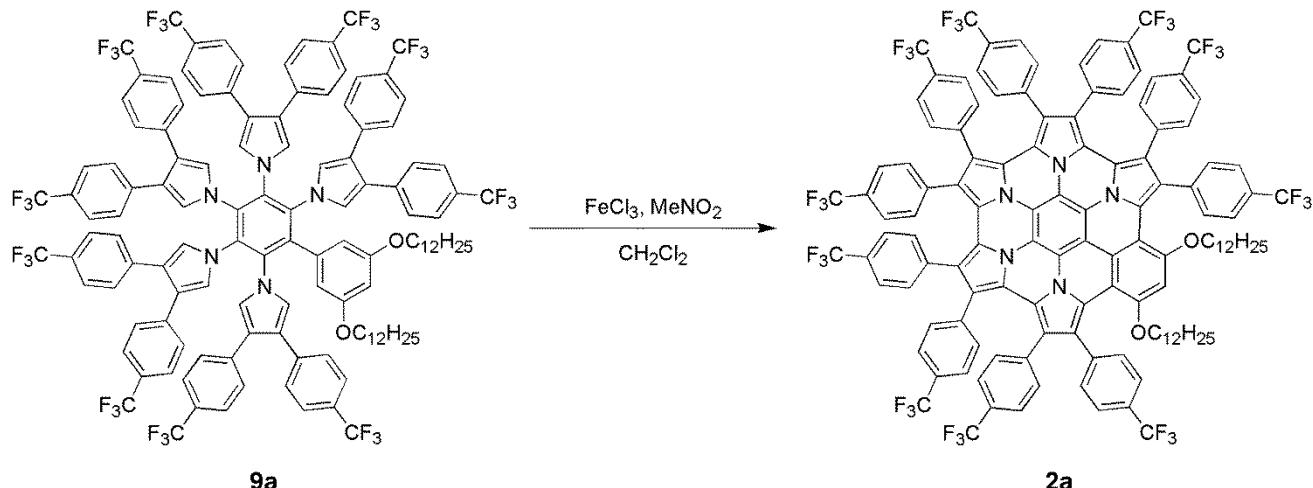
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.56.

APCI-MS calcd. for C<sub>150</sub>H<sub>189</sub>F<sub>18</sub>N<sub>3</sub>O<sub>6</sub>: 2470.4289, Found: 2470.4293.

Anal. Calcd for C<sub>150</sub>H<sub>189</sub>F<sub>18</sub>N<sub>3</sub>O<sub>6</sub>: C, 72.88%; H, 7.71%; N, 1.70%. Found: C, 72.62%; H, 7.72%; N, 1.74%.

Melting Point: 105.2 – 106.3 °C

11. 5AC(OC<sub>12</sub>H<sub>25</sub>) (**2a**)



In a double-neck round-bottom flask, was added CH<sub>2</sub>Cl<sub>2</sub> (120 ml) to **9a** (125 mg, 55 µmol). Then Ar gas was bubbled to the solution for 30min. Under light shielding with foil, FeCl<sub>3</sub> (322 mg, 1.98 mmol) in MeNO<sub>2</sub> (6 ml) was added dropwise. After stirring at room temperature for overnight, the reaction mixture was quenched with Na<sub>2</sub>CO<sub>3</sub> aq. and Na<sub>2</sub>SO<sub>3</sub> aq., and then the aqueous phase was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were washed with sat-NaCl aq., and dried over MgSO<sub>4</sub>, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/CH<sub>2</sub>Cl<sub>2</sub> 3:1), and recrystallization from CH<sub>2</sub>Cl<sub>2</sub>-MeOH to afford **2a** (97 mg, 42 µmol) as orange solid, yield: 78%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.33 (d, *J* = 7.7 Hz, 4H), 7.02 (d, *J* = 7.7 Hz, 4H), 6.97-6.86 (m, 8H), 6.85-6.78 (m, 8H), 6.77-6.69 (m, 4H), 6.66-6.50 (m, 12H), 6.23 (s, br, 1H), 3.22 (m, 4H), 1.30-1.09 (m, 32H), 1.05-0.94 (m, 8H), 0.87(t, *J* = 7.2 Hz, 6H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 142.00, 139.18, 138.14, 137.28, 136.94, 136.74, 131.35, 131.00, 130.70, 130.62, 128.95, 128.68, 128.12, 127.85, 125.41, 125.31, 124.80, 124.77, 124.64, 124.62, 124.44, 124.33-124.13 (m, -CF<sub>3</sub>, 5C), 124.12, 123.54, 123.52, 123.14, 122.76, 122.64, 122.60, 122.59, 122.48, 118.38, 116.89, 116.17, 115.00, 110.56, 108.81, 107.21, 103.09, 67.78, 31.91, 29.62, 29.61, 29.49, 29.47, 29.34, 29.14, 27.78, 25.59, 22.68, 14.09.

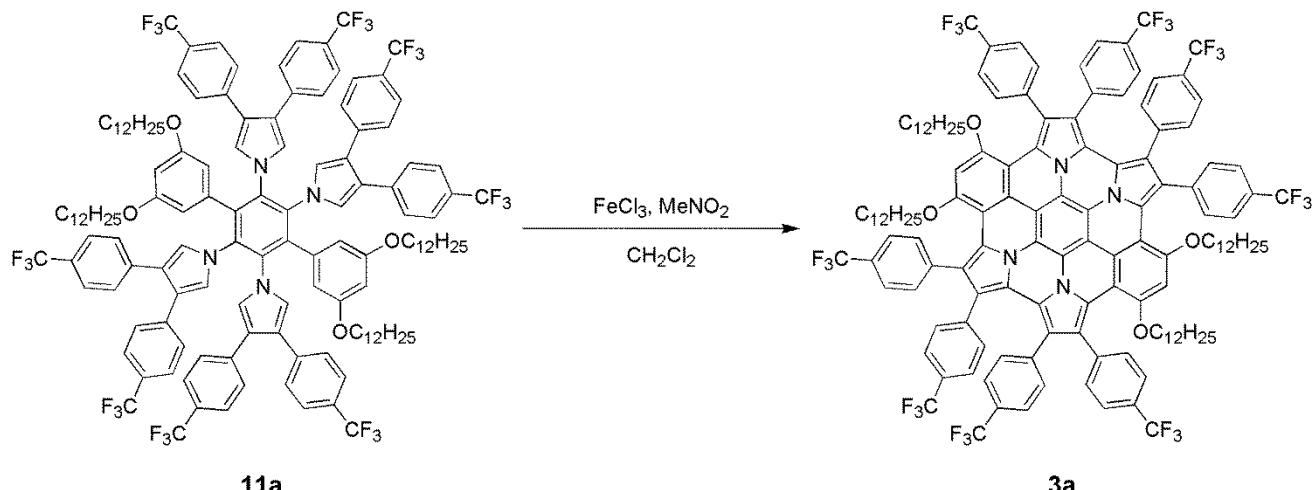
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.73, -66.39, -66.43, -66.52, -66.53.

APCI-MS calcd. for C<sub>126</sub>H<sub>91</sub>F<sub>30</sub>N<sub>5</sub>O<sub>2</sub>: 2275.6694, Found: 2275.6676.

Anal. Calcd for C<sub>126</sub>H<sub>91</sub>F<sub>30</sub>N<sub>5</sub>O<sub>2</sub>: C, 66.46%; H, 4.03%; N, 3.08%. Found: C, 66.74%; H, 4.05%; N, 3.05%.

Melting Point: > 300 °C

12. 4AC(OC<sub>12</sub>H<sub>25</sub>) (**3a**)



In a double-neck round-bottom flask, charged with **11a** (106 mg, 45  $\mu$ mol) and CH<sub>2</sub>Cl<sub>2</sub> (100 ml). Then Ar gas was bubbled to the solution for 30min. Under light shielding with foil, FeCl<sub>3</sub> (274 mg, 1.69 mmol) in MeNO<sub>2</sub> (5 ml) was added dropwise. After stirring at room temperature for overnight, the reaction mixture was quenched with Na<sub>2</sub>CO<sub>3</sub> aq. and Na<sub>2</sub>SO<sub>3</sub> aq., and then the aqueous phase extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were washed with sat-NaCl aq., and dried over MgSO<sub>4</sub>, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/CH<sub>2</sub>Cl<sub>2</sub> 3:1), and recrystallization from CH<sub>2</sub>Cl<sub>2</sub>-MeOH to afford **3a** (86 mg, 36  $\mu$ mol) as red solid, yield: 82%.

<sup>1</sup>H-NMR (CD<sub>2</sub>Cl<sub>2</sub>, 500 MHz)  $\delta$  7.30 (d, *J* = 8.0 Hz, 8H), 7.05 (d, *J* = 8.0 Hz, 8H), 6.98 (d, *J* = 7.8 Hz, 8H), 6.79 (d, *J* = 7.8 Hz, 8H), 6.17 (s, 2H), 3.23-3.16 (m, 8H), 1.32-1.06 (m, 64H), 1.05-0.94 (m, 16H), 0.86 (t, *J* = 7.0 Hz, 12H).

<sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 125 MHz)  $\delta$  154.29, 142.83, 139.35, 132.06, 130.77, 127.49, 127.01, 125.81, 125.57, 125.48, 124.58, 124.35-124.22 (m, -CF<sub>3</sub>), 123.64 (q, *J* = 3.4 Hz, -CF<sub>3</sub>), 123.31, 123.08, 122.89, 120.71, 116.89, 116.09, 109.60, 67.82, 32.16, 29.87, 29.86, 29.74, 29.72, 29.59, 29.36, 28.01, 25.80, 22.92, 14.09.

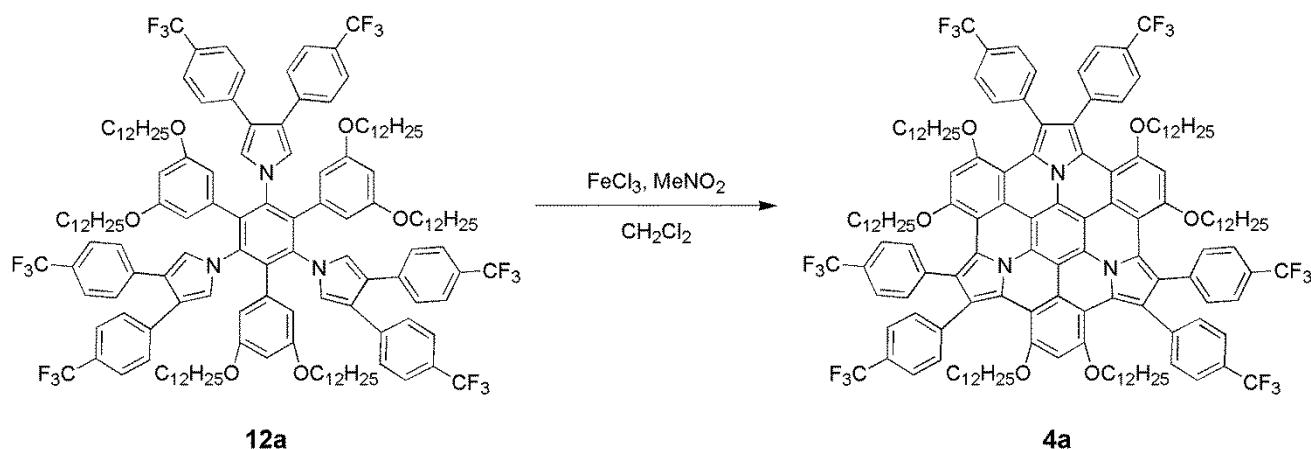
<sup>19</sup>F NMR (CD<sub>2</sub>Cl<sub>2</sub>, 470.5 MHz)  $\delta$  -64.95, -65.45.

APCI-MS calcd. for C<sub>138</sub>H<sub>134</sub>F<sub>24</sub>N<sub>4</sub>O<sub>4</sub>: 2367.0022, Found: 2367.0026.

Anal. Calcd for C<sub>138</sub>H<sub>134</sub>F<sub>24</sub>N<sub>4</sub>O<sub>4</sub>: C, 69.98%; H, 5.70%; N, 2.37%. Found: C, 70.14%; H, 5.54%; N, 2.31%.

Melting Point: > 300 °C

13. 3AC(OC<sub>12</sub>H<sub>25</sub>) (**4a**)



In a double-neck round-bottom flask, charged with **12a** (176 mg, 71  $\mu$ mol) and CH<sub>2</sub>Cl<sub>2</sub> (100 ml). Then Ar gas was bubbled to the solution for 30min. Under light shielding with foil, FeCl<sub>3</sub> (433 mg, 2.67 mmol) in MeNO<sub>2</sub> (5 ml) was added dropwise. After stirring at room temperature for overnight, the reaction mixture was quenched with Na<sub>2</sub>CO<sub>3</sub> aq. and Na<sub>2</sub>SO<sub>3</sub> aq., and then the aqueous phase was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were washed with sat-NaCl aq., and dried over MgSO<sub>4</sub>, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/CH<sub>2</sub>Cl<sub>2</sub> 3:1), and recrystallization from CH<sub>2</sub>Cl<sub>2</sub>-MeOH to afford **4a** (151 mg, 61  $\mu$ mol) as yellow needle, yield: 86%.

<sup>1</sup>H-NMR (CD<sub>2</sub>Cl<sub>2</sub>, 500 MHz)  $\delta$  7.40 (d, *J* = 8.2 Hz, 12H), 7.22 (d, *J* = 8.2 Hz, 12H), 6.28 (s, 3H), 3.23 (t, *J* = 7.1 Hz, 12H), 1.32-0.99 (m, 120H), 0.86 (t, *J* = 7.2 Hz, 18H).

<sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 125 MHz)  $\delta$  153.07, 144.63, 131.57, 127.69, 127.43, 126.98, 126.14, 125.35, 124.78, 123.98, 123.80 (q, *J* = 3.6 Hz, -CF<sub>3</sub>), 120.12, 107.90, 67.73, 32.34, 30.07, 30.01, 29.99, 29.78, 29.68 (2C), 28.40, 26.07, 23.10, 14.27.

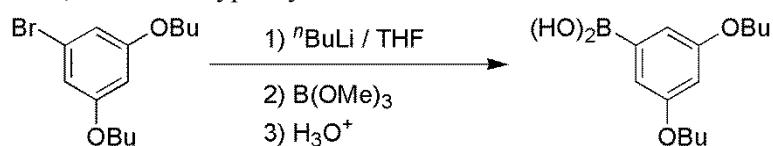
<sup>19</sup>F NMR (CD<sub>2</sub>Cl<sub>2</sub>, 470.5 MHz)  $\delta$  -64.79.

APCI-MS calcd. for C<sub>150</sub>H<sub>177</sub>F<sub>18</sub>N<sub>3</sub>O<sub>6</sub>: 2458.3350, Found: 2458.3294.

Anal. Calcd for C<sub>150</sub>H<sub>177</sub>F<sub>18</sub>N<sub>3</sub>O<sub>6</sub>: C, 73.24%; H, 7.25%; N, 1.71%. Found: C, 73.02%; H, 7.33%; N, 1.73%.

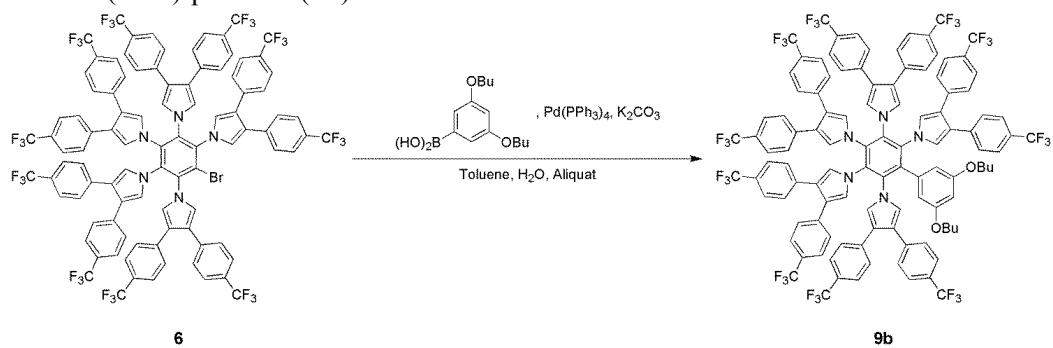
Melting Point: 148.2 – 151.2 °C

14. 3,5-Bisbutoxyphenylboronic acid



This compound was synthesized following the procedure of 3,5-Bisdodecyloxyphenylboronic acid.

### 15. 5AC(OBu)-prefused (9b)



This compound was synthesized following the procedure of **9a**.

White solid, yield: 37%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.56-7.46 (m, 20H), 7.21-7.12 (m, 12H), 7.07 (d, *J* = 8.1 Hz, 8H), 6.65 (s, 2H), 6.64-6.60 (m, 5H), 6.59 (s, 4H), 6.28 (d, *J* = 2.0 Hz, 2H), 3.76 (t, *J* = 6.5 Hz, 4H), 1.67-1.57 (m, 4H), 1.35-1.22 (m, 4H), 0.82 (t, *J* = 7.4 Hz, 6H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 161.09, 137.83, 137.54, 137.30, 137.24, 135.60, 133.45, 133.07, 132.61, 129.49, 129.41, 129.16, 128.90, 128.34, 128.29, 128.26, 126.03-125.54 (m, -CF<sub>3</sub>, 3C), 125.51, 125.22, 125.15, 124.78, 123.06, 122.99, 121.87, 121.03, 106.77, 102.88, 68.32, 30.96, 19.13, 13.61.

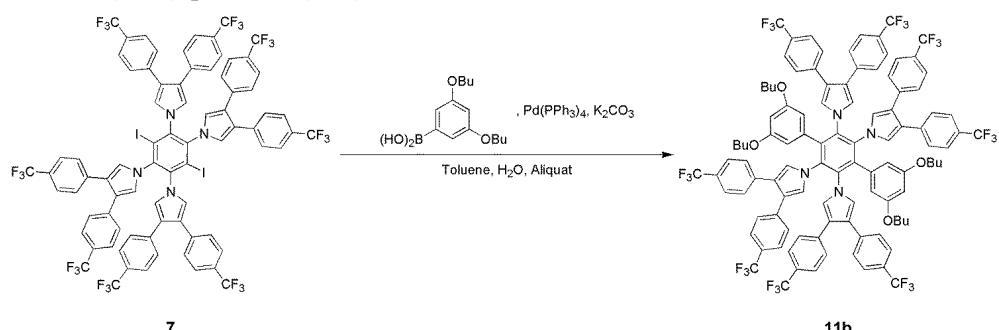
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.89, -65.90, -65.91.

APCI-MS calcd. for C<sub>110</sub>H<sub>71</sub>F<sub>30</sub>N<sub>5</sub>O<sub>2</sub>: 2063.5129, Found: 2063.5130.

Anal. Calcd for  $C_{110}H_{71}F_{30}N_5O_2$ : C, 63.99%; H, 3.47%; N, 3.39%. Found: C, 64.06%; H, 3.11%; N, 3.31%.

Melting Point: 179.2 – 183.7 °C

### 16. 4AC(OBu)-prefused (**11b**)



This compound was synthesized following the procedure of **11a**.

White solid, yield: 60%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.46 (d, *J* = 8.3 Hz, 16H), 7.03 (d, *J* = 8.3 Hz, 16H), 6.58 (t, *J* = 2.2 Hz, 2H), 6.26 (d, *J* = 2.2 Hz, 4H), 3.74 (t, *J* = 6.5 Hz, 8H), 1.64-1.56 (m, 8H), 1.32-1.21 (m, 8H), 0.80 (t, *J* = 7.5 Hz, 12H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 160.92, 137.89, 137.03, 135.55, 134.18, 128.30, 125.51 (q, *J* = 3.5 Hz, -CF<sub>3</sub>), 125.28, 124.15, 123.11, 122.16, 106.90, 102.74, 68.28, 30.98, 19.12, 13.61.

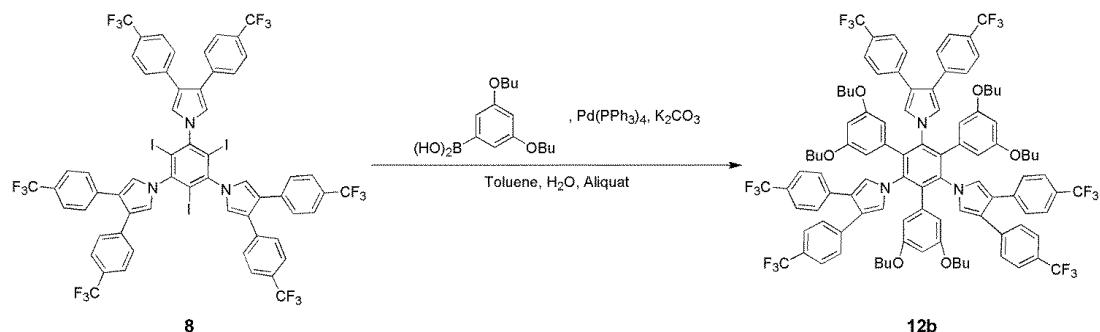
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.58.

APCI-MS calcd. for C<sub>108</sub>H<sub>82</sub>F<sub>24</sub>N<sub>4</sub>O<sub>4</sub>: 1930.5953, Found: 1930.5963.

Anal. Calcd for  $C_{108}H_{82}F_{24}N_4O_4$ : C, 65.91%; H, 4.28%; N, 2.90%. Found: C, 66.14%; H, 4.17%; N, 2.89%.

Melting Point: 231.2 – 232.3 °C

### 17. 3AC(OBu)-prefused (**12b**)



This compound was synthesized following the procedure of **12a**.

Colorless needle, yield: 33%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.43 (d, *J* = 8.1 Hz, 12H), 6.94 (d, *J* = 8.1 Hz, 12H), 6.49 (s, 6H), 6.47-6.44 (m, 3H), 6.24-6.21 (m, 6H), 3.67 (t, *J* = 6.5 Hz, 12H), 1.62-1.49 (m, 12H), 1.29-1.17 (m, 12H), 0.79 (t, *J* = 4.9 Hz, 18H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 160.47, 138.41, 137.74, 137.06, 135.40, 128.32, 128.08, 125.37, 125.20 (q, *J* = 3.5 Hz, -CF<sub>3</sub>), 123.24, 122.90, 106.84, 102.31, 68.15, 31.02, 19.07, 13.59.

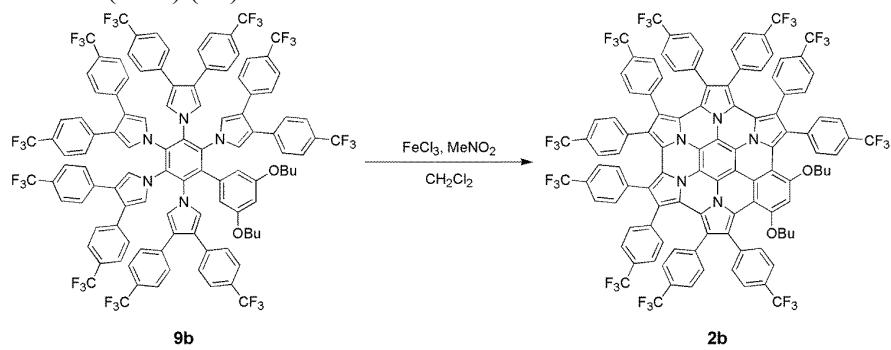
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.80.

APCI-MS calcd. for C<sub>102</sub>H<sub>93</sub>F<sub>18</sub>N<sub>3</sub>O<sub>6</sub>: 1797.6777, Found: 1797.6777.

Anal. Calcd for C<sub>102</sub>H<sub>93</sub>F<sub>18</sub>N<sub>3</sub>O<sub>6</sub>: C, 68.11%; H, 5.21%; N, 2.34%. Found: C, 68.07%; H, 5.04%; N, 2.33%.

Melting Point: 212.7 – 216.3 °C

### 18. 5AC(OBu) (**2b**)



This compound was synthesized following the procedure of **2a**.

Orange solid, yield: 64%.

<sup>1</sup>H-NMR (CD<sub>2</sub>Cl<sub>2</sub>, 500 MHz) δ 7.35 (d, *J* = 7.8 Hz, 4H), 7.07 (d, *J* = 7.8 Hz, 4H), 7.00-6.88 (m, 8H), 6.86-6.77 (m, 12H), 6.69-6.60 (m, 12H), 6.27 (s, 1H), 3.31-3.22 (m, 4H), 1.10-0.95 (m, 8H), 0.75 (t, *J* = 5.2 Hz, 6H).

<sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 125 MHz) δ 155.06, 142.78, 138.91, 138.23, 137.85, 137.62, 136.84, 132.03, 131.66, 131.38, 131.19, 128.88, 128.62, 128.15, 127.90, 127.49, 125.96, 125.84, 125.48, 125.45, 125.33, 124.96,

124.69-124.44 (m, -CF<sub>3</sub>, 4C), 124.44, 123.90 (q, *J* = 4.1 Hz, -CF<sub>3</sub>), 123.79, 123.32, 123.28, 123.17, 123.01, 122.99, 121.90, 118.92, 117.43, 116.73, 115.50, 111.07, 110.60, 109.20, 107.70, 67.89, 30.17, 19.23, 13.74.

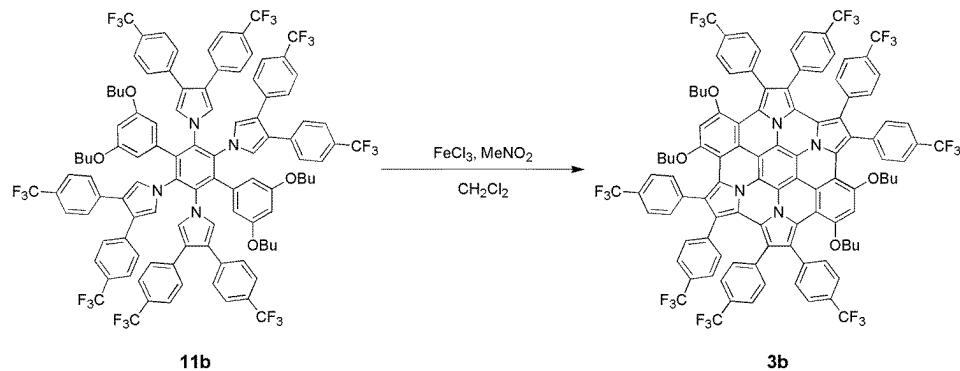
<sup>19</sup>F NMR (CD<sub>2</sub>Cl<sub>2</sub>, 470.5 MHz) δ -65.02, -65.57, -65.60, -65.69, -65.69.

APCI-MS calcd. for C<sub>110</sub>H<sub>59</sub>F<sub>30</sub>N<sub>5</sub>O<sub>2</sub>: 2051.4190, Found: 2051.4187.

Anal. Calcd for C<sub>138</sub>H<sub>134</sub>F<sub>24</sub>N<sub>4</sub>O<sub>4</sub>: C, 64.37%; H, 2.90%; N, 3.41%. Found: C, 64.63%; H, 2.82%; N, 3.39%.

Melting Point: > 300 °C

#### 19. 4AC(OBu) (**3b**)



This compound was synthesized following the procedure of **3a**.

Thin red plate, yield: 41%.

<sup>1</sup>H-NMR (C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 80 °C, 500 MHz) δ 7.27 (d, *J* = 8.0 Hz, 8H), 7.00 (d, *J* = 7.9 Hz, 8H), 6.93 (d, *J* = 8.0 Hz, 8H), 6.73 (d, *J* = 7.9 Hz, 8H), 6.16 (s, 2H) 3.22-3.12 (m, 8H), 1.05-0.95 (m, 16H), 0.77-0.67 (m, 12H).

<sup>13</sup>C NMR (C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 80 °C, 125 MHz) δ 154.43, 147.31, 142.83, 139.19, 132.01, 130.82, 125.71, 125.55, 125.31, 125.21, 124.90, 124.25-124.15 (m, -CF<sub>3</sub>), 123.58 (q, *J* = 4.2 Hz, -CF<sub>3</sub>), 123.50, 123.14, 123.03, 120.61, 116.79, 116.17, 110.00, 67.83, 30.23, 18.99, 13.63.

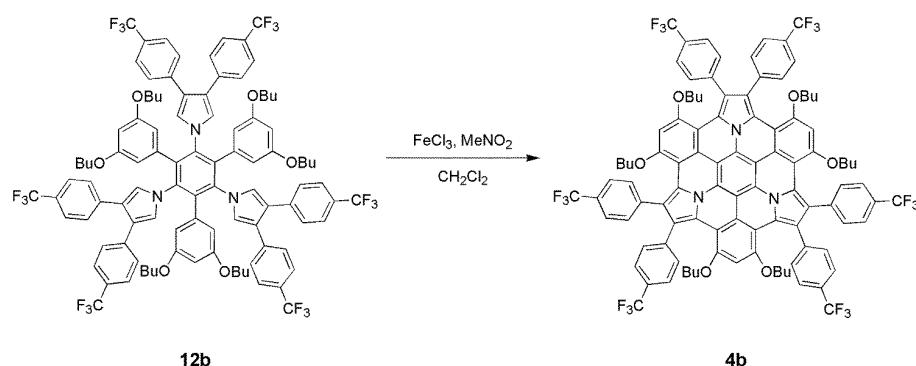
<sup>19</sup>F NMR (C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 80 °C, 470.5 MHz) δ -65.53, -66.03.

APCI-MS calcd. for C<sub>108</sub>H<sub>70</sub>F<sub>24</sub>N<sub>4</sub>O<sub>4</sub>: 1918.5014, Found: 1918.5009.

Anal. Calcd for C<sub>108</sub>H<sub>70</sub>F<sub>24</sub>N<sub>4</sub>O<sub>4</sub>: C, 66.32%; H, 3.68%; N, 2.92%. Found: C, 66.42%; H, 3.57%; N, 2.89%.

Melting Point: > 300 °C

#### 20. 3AC(BuO) (**4b**)



This compound was synthesized following the same procedure for **4a**.

Yellow solid, yield: 62%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.42 (d, *J* = 7.6 Hz, 12H), 7.24 (d, *J* = 7.6 Hz, 12H), 6.32 (s, 3H), 3.38-3.16 (m, 12H), 1.19-1.05 (m, 24H), 0.80 (t, *J* = 7.1 Hz, 18H).

<sup>13</sup>C NMR spectrum could not be record due to low solubility, seriously broadened signals and quite low thermal stability of this compound.

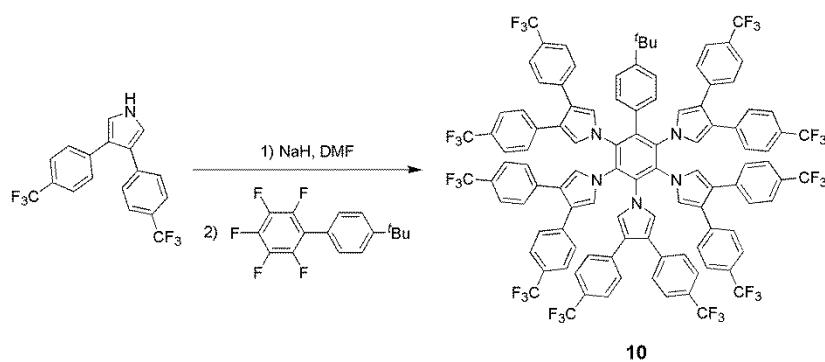
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -64.69.

APCI-MS calcd. for C<sub>102</sub>H<sub>81</sub>F<sub>18</sub>N<sub>3</sub>O<sub>6</sub>: 1785.5838, Found: 1785.5818.

Anal. Calcd for C<sub>102</sub>H<sub>81</sub>F<sub>18</sub>N<sub>3</sub>O<sub>6</sub>: C, 68.57%; H, 4.57%; N, 2.35%. Found: C, 68.49%; H, 4.27%; N, 2.36%.

Melting Point: > 300 °C

## 21. 5AC('Bu)-prefused (**10**)



In a double-neck round-bottom flask, was added DMF (7 ml) to 3,4-bis(4-trifluoromethylphenyl)-1*H*-pyrrole (1.01 g, 2.84 mmol) and sodium hydride (60% in mineral oil, 119 mg, 2.97 mmol) at 0 °C. The reaction mixture was stirred at 0 °C for 20 min. And then, pentafluoro-4'-*tert*-butyl-1,1'-biphenyl (90 mg, 0.30 mmol) was added in one portion. After stirring at room temperature for overnight, the reaction was quenched with H<sub>2</sub>O, and the aqueous phase was extracted with Et<sub>2</sub>O. The combined organic layers were washed with sat-NaCl aq., and dried over MgSO<sub>4</sub>, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (hexane/CH<sub>2</sub>Cl<sub>2</sub> 2:1) to afford **10** (542 mg, 0.27 mmol) as white solid, yield: 92%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.55-7.44 (m, 22H), 7.21-7.14 (m, 12H), 7.10 (d, *J* = 8.3 Hz, 2H), 7.01 (d, *J* = 8.1 Hz, 8H), 6.66 (s, 2H), 6.62 (s, 4H), 6.50 (s, 4H), 1.44 (s, 9H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 153.03, 137.68, 137.59, 137.33, 137.29, 135.80, 132.69, 129.38, 129.12, 129.08, 129.06, 128.82, 128.34, 128.30, 128.26, 128.17, 126.25, 125.84 – 125.49 (m, -CF<sub>3</sub>, 3C), 125.31, 125.22, 125.16, 124.57, 123.06, 123.00, 121.84, 121.05, 121.00, 35.06, 31.37.

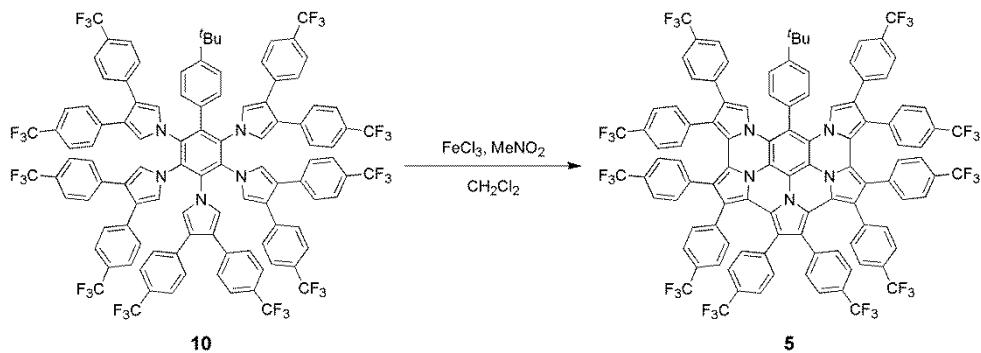
<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.57, -65.60, -65.61.

APCI-MS calcd. for C<sub>106</sub>H<sub>63</sub>F<sub>30</sub>N<sub>5</sub>: 1975.4604, Found: 1975.4692.

Anal. Calcd for C<sub>106</sub>H<sub>63</sub>F<sub>30</sub>N<sub>5</sub>: C, 64.41%; H, 3.21%; N, 3.54%. Found: C, 64.38%; H, 3.17%; N, 3.56%.

Melting Point: > 300 °C

22. Partially-fused 5AC(<sup>t</sup>Bu) (**5**)



This compound was synthesized following the procedure of **2a**.

Recrystallization from hexane-CH<sub>2</sub>Cl<sub>2</sub>. Orange crystal, yield: 91%.

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.91 (d, *J* = 8.0 Hz, 2H), 7.85 (d, *J* = 8.0 Hz, 2H), 7.26 (d, *J* = 7.0 Hz, 4H), 7.15 (d, *J* = 7.8 Hz, 4H), 7.00 (d, *J* = 7.8 Hz, 4H), 6.95-6.88 (m, 12H), 6.82 (d, *J* = 7.8 Hz, 4H), 6.73 (d, *J* = 7.8 Hz, 4H), 6.64 (d, *J* = 7.8 Hz, 4H), 6.57 (d, *J* = 7.8 Hz, 4H), 6.45 (s, 2H), 1.58 (s, 9H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 155.17, 137.86, 137.28, 137.20, 137.09, 136.77, 132.69, 131.44, 131.39, 131.06, 130.82, 129.17, 129.07, 128.90, 128.81, 128.65, 128.39, 127.75, 127.68, 125.97, 125.28 (q, *J* = 3.4 Hz, -CF<sub>3</sub>), 124.97 (q, *J* = 3.0 Hz, -CF<sub>3</sub>), 124.60 (q, *J* = 3.4 Hz, -CF<sub>3</sub>), 124.50 (q, *J* = 3.3 Hz, -CF<sub>3</sub>), 124.31 (q, *J* = 3.4 Hz, -CF<sub>3</sub>), 123.76, 123.31, 122.95, 122.82, 122.70, 122.68, 122.55, 122.22, 121.97, 120.21, 120.11, 118.65, 118.13, 117.60, 116.72, 115.01, 113.29, 112.71, 35.39, 31.50.

<sup>19</sup>F NMR (CDCl<sub>3</sub>, 470.5 MHz) δ -65.73, -66.08, -66.12, -66.30, -66.30.

APCI-MS calcd. for C<sub>106</sub>H<sub>55</sub>F<sub>30</sub>N<sub>5</sub>: 1967.3978, Found: 1967.3994.

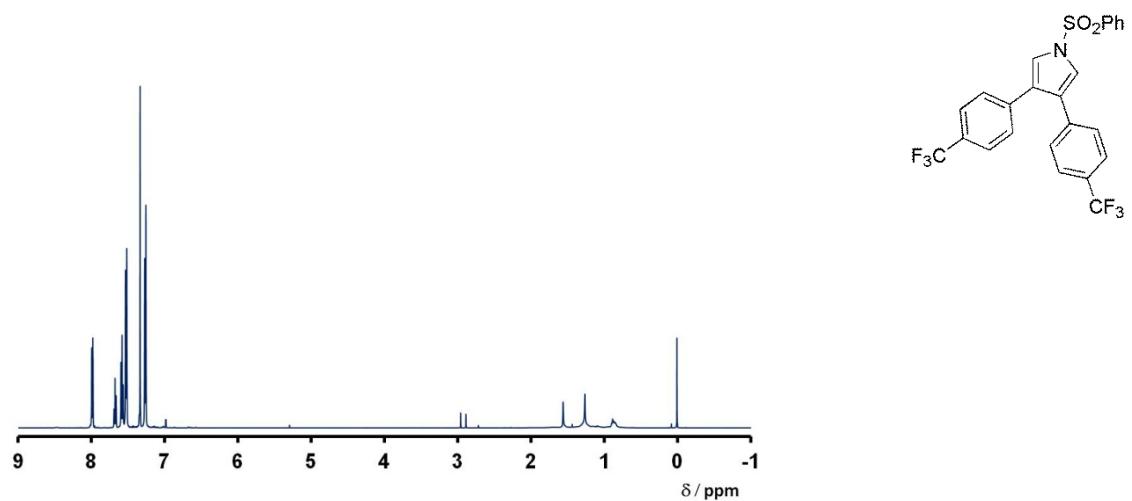
Anal. Calcd for C<sub>106</sub>H<sub>55</sub>F<sub>30</sub>N<sub>5</sub>: C, 64.67%; H, 2.82%; N, 3.56%. Found: C, 64.55%; H, 2.89%; N, 3.55%.

Melting Point: > 300 °C

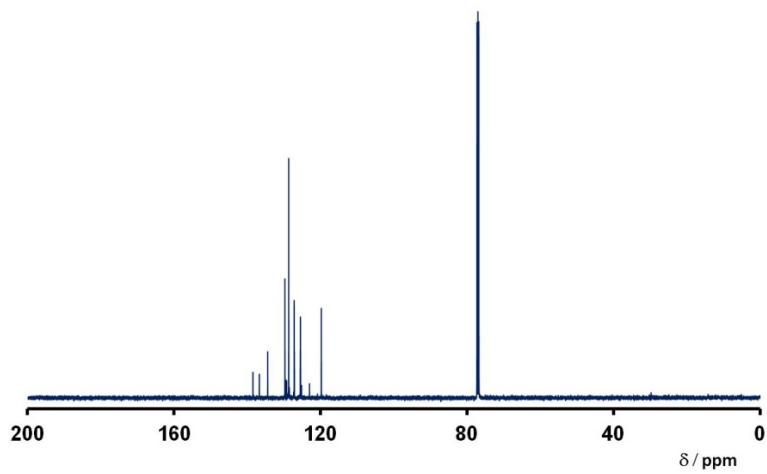
**S22.  $^1\text{H}$ -,  $^{13}\text{C}$ - and  $^{19}\text{F}$ - NMR spectra**

1. *N*-Benzenesulfonyl- 3,4-bis(4-trifluoromethylphenyl)pyrrole in  $\text{CDCl}_3$

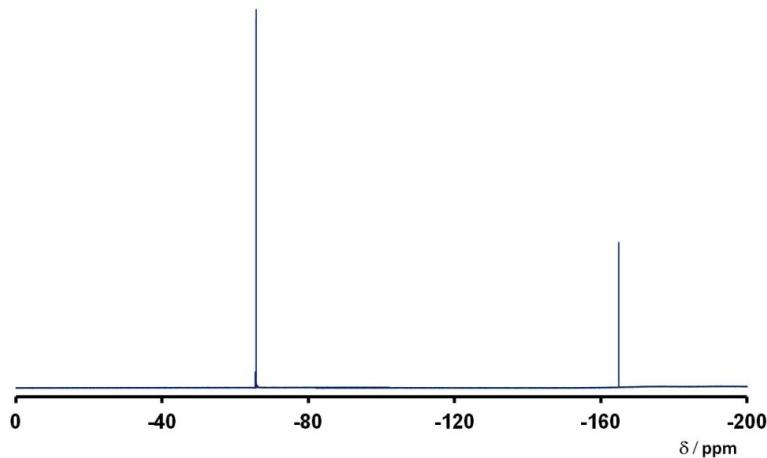
$^1\text{H}$ -NMR



$^{13}\text{C}$ -NMR

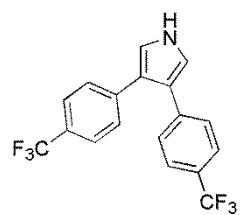
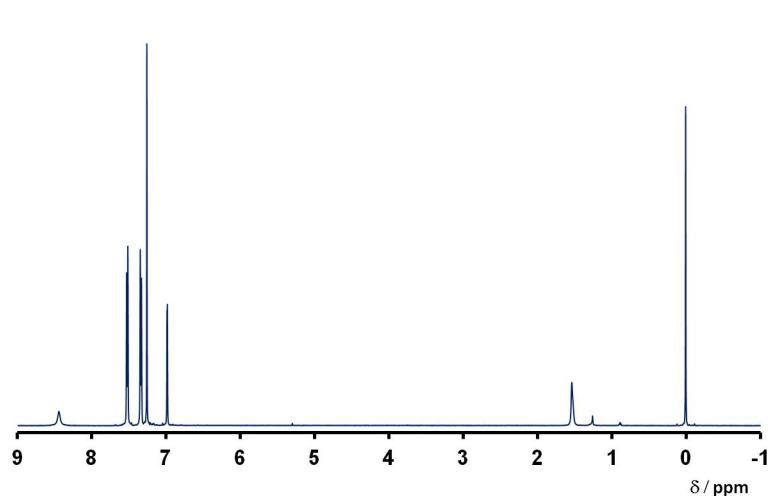


$^{19}\text{F}$ -NMR

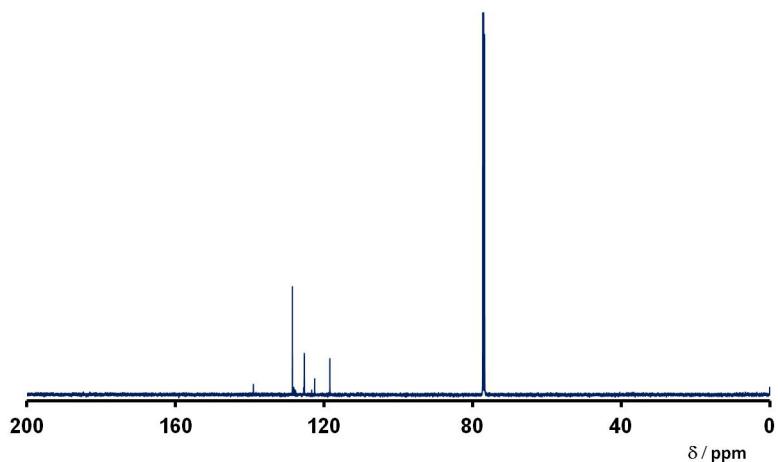


2. 3,4-Bis(4-trifluoromethylphenyl)-*1H*-pyrrole in CDCl<sub>3</sub>

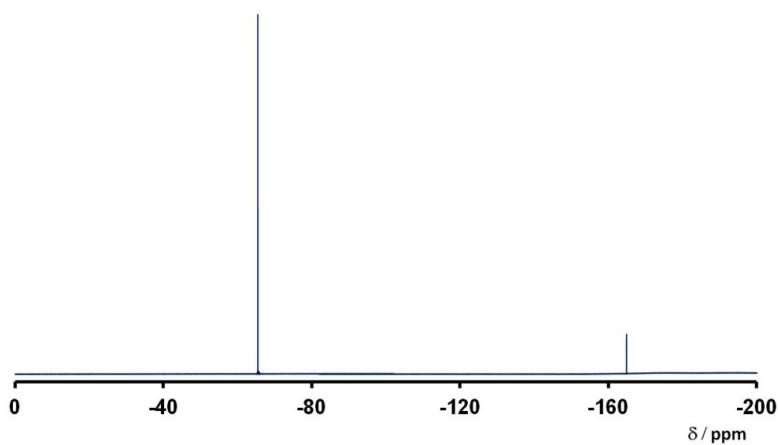
<sup>1</sup>H-NMR|



<sup>13</sup>C-NMR

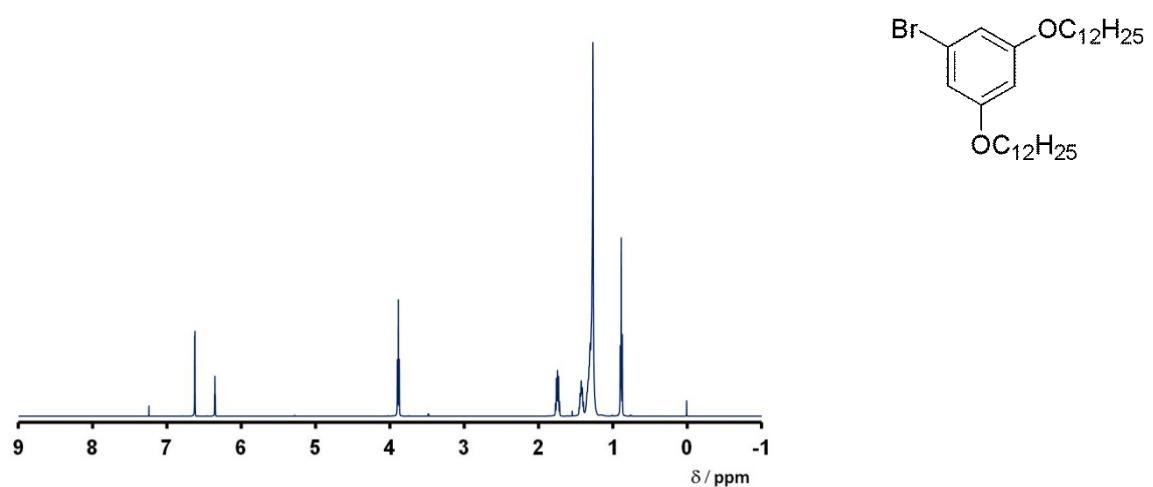


<sup>19</sup>F-NMR

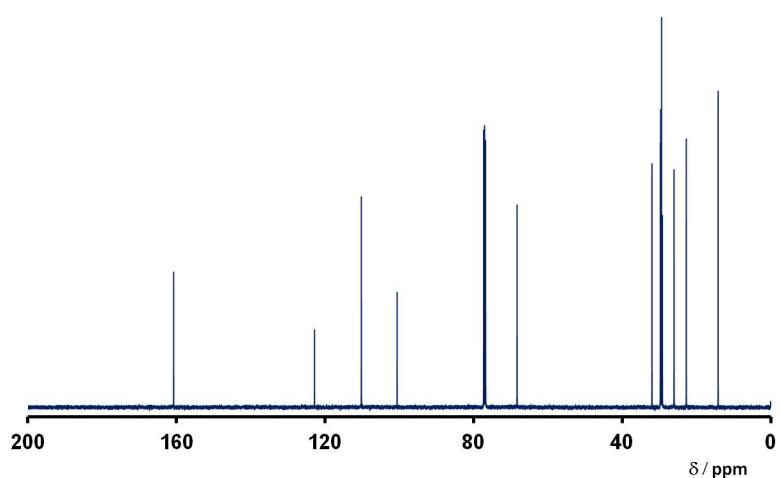


3. 3,5-Bisdodecyloxybromobenzene in CDCl<sub>3</sub>

<sup>1</sup>H-NMR |

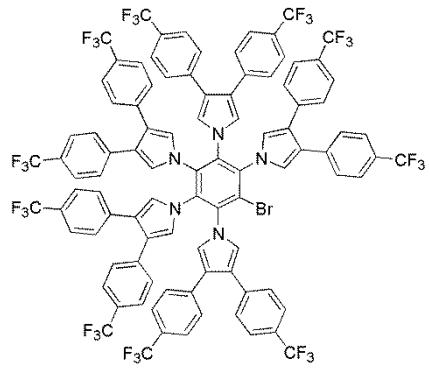
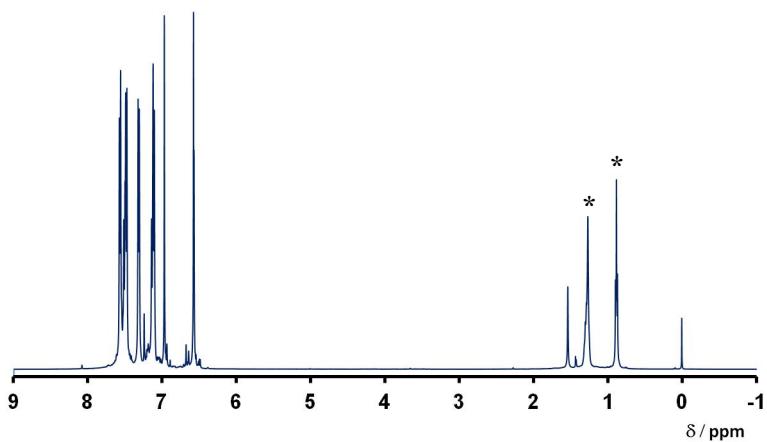


<sup>13</sup>C-NMR



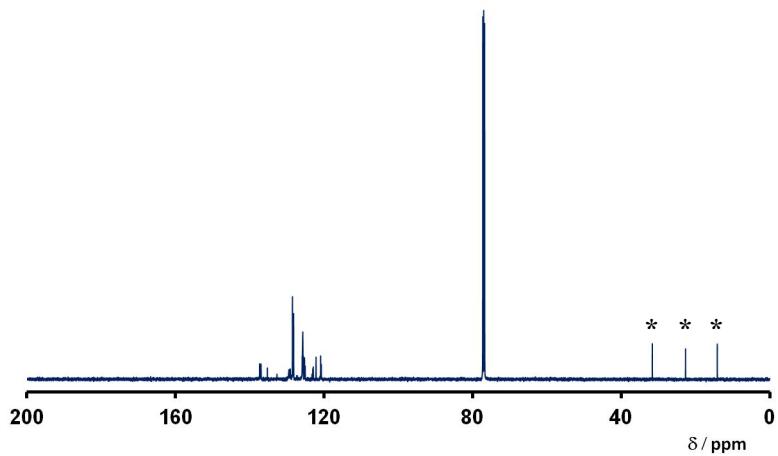
5. 6 in  $\text{CDCl}_3$

## <sup>1</sup>H-NMR

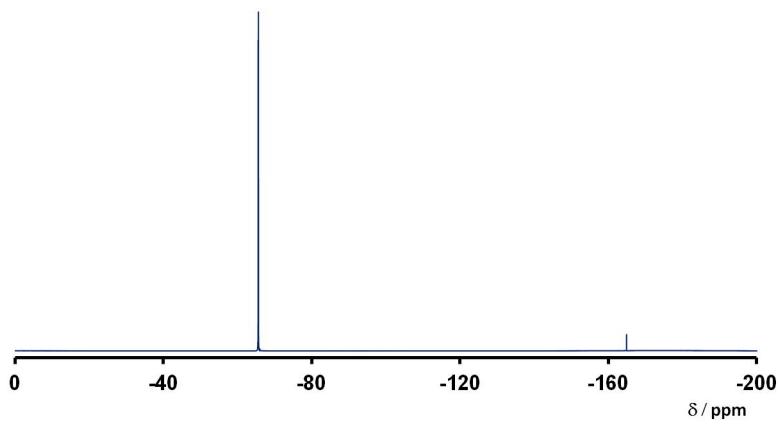


6

<sup>13</sup>C-NMR

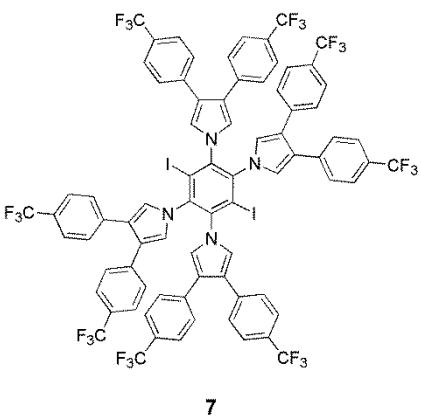
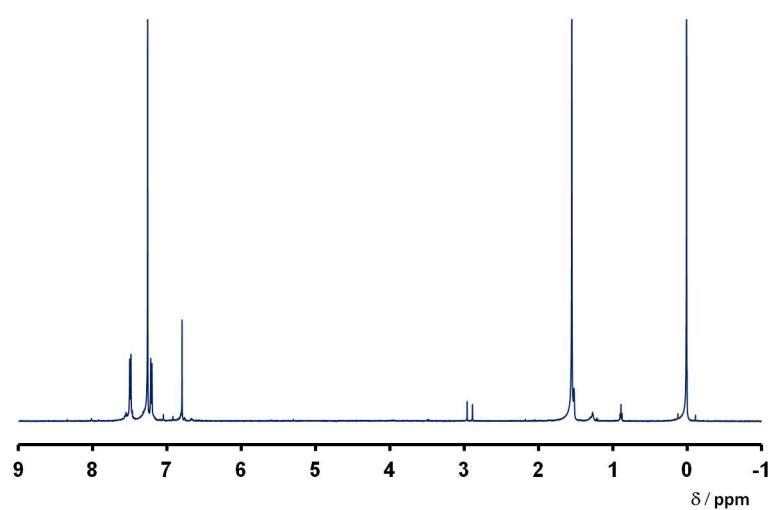


## <sup>19</sup>F-NMR



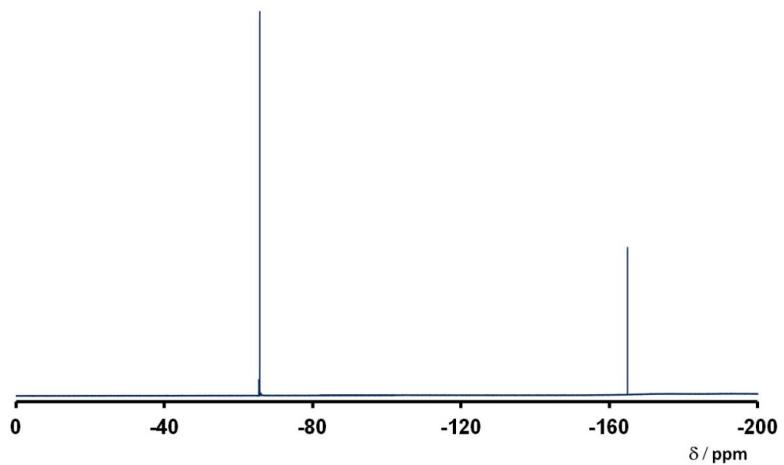
6. **7** in  $\text{CDCl}_3$

$^1\text{H}$ -NMR



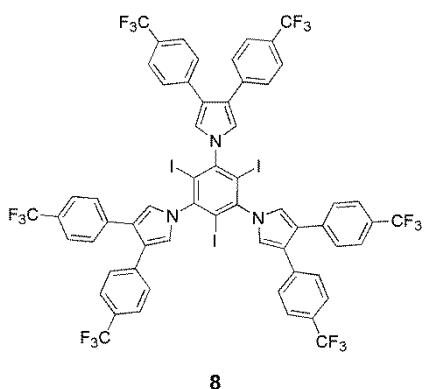
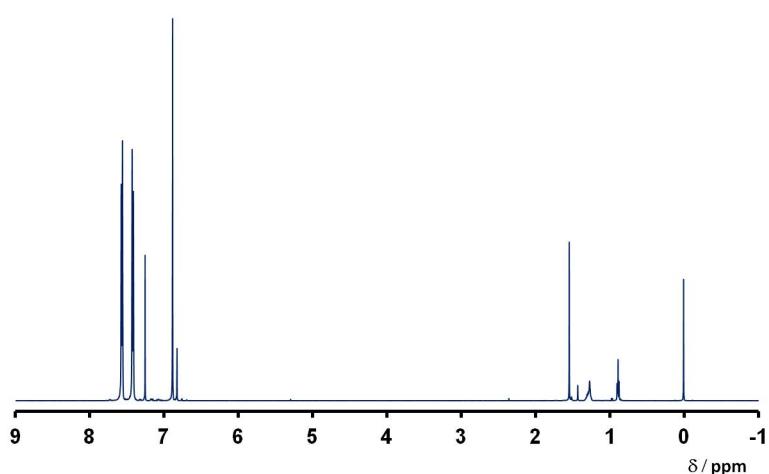
**7**

$^{19}\text{F}$ -NMR



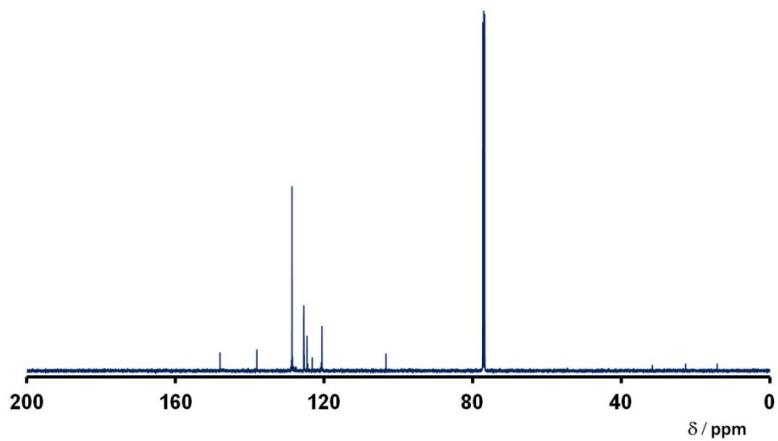
7, 8 in  $\text{CDCl}_3$

$^1\text{H}$ -NMR

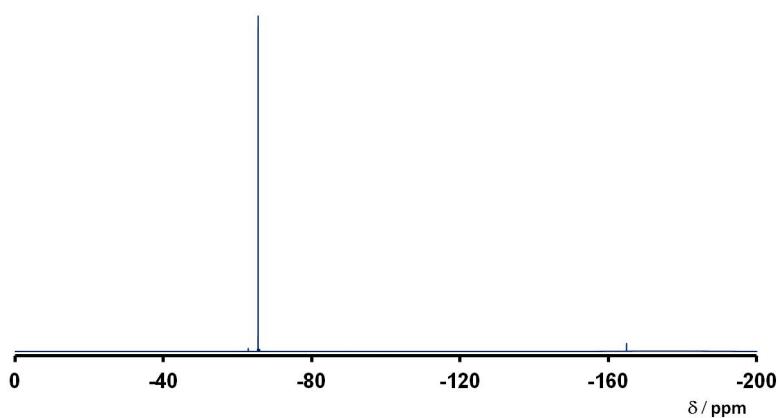


8

$^{13}\text{C}$ -NMR

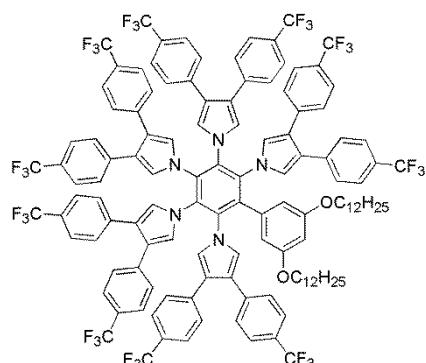
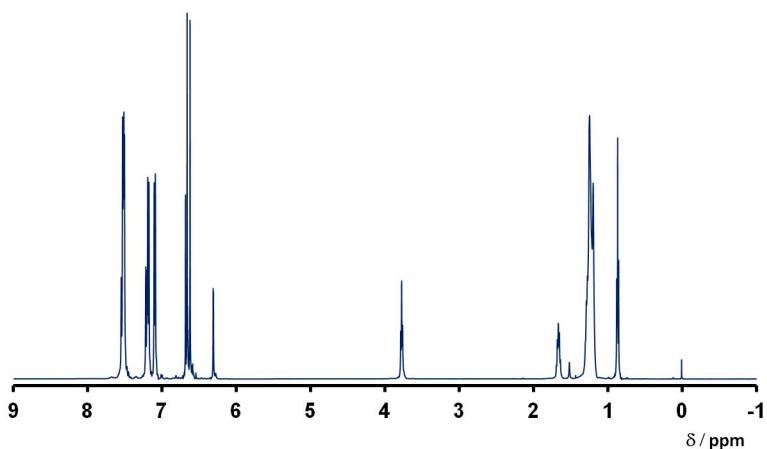


$^{19}\text{F}$ -NMR



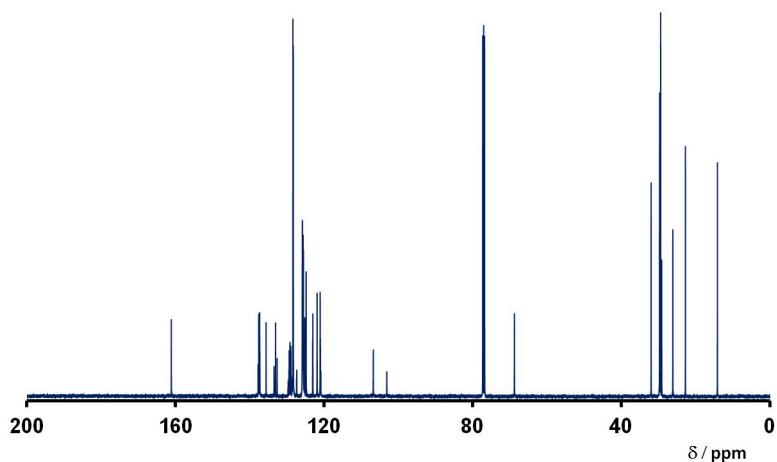
### 8. **9a** in CDCl<sub>3</sub>

## <sup>1</sup>H-NMR

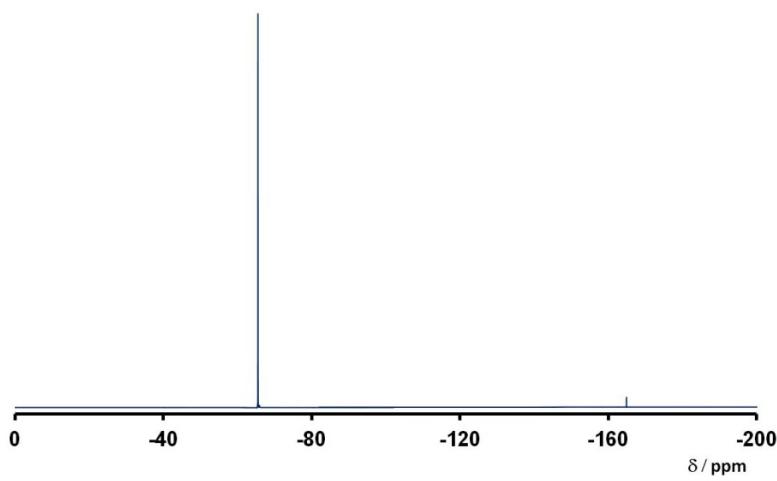


9a

<sup>13</sup>C-NMR

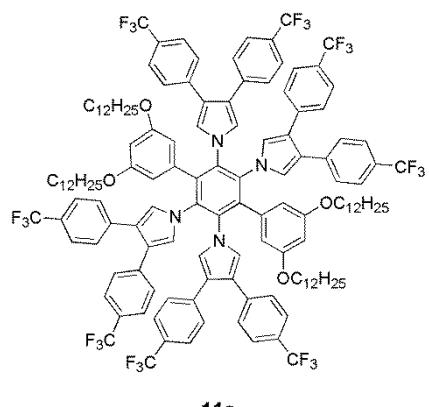
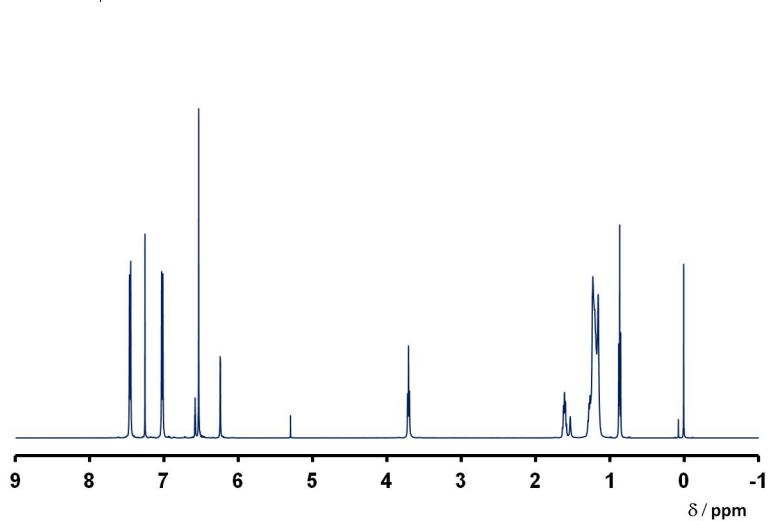


<sup>19</sup>F-NMR

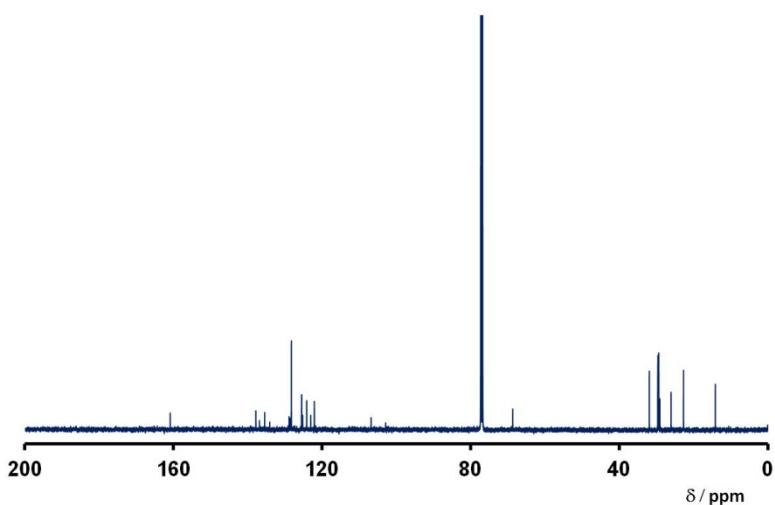


9. **11a** in CDCl<sub>3</sub>

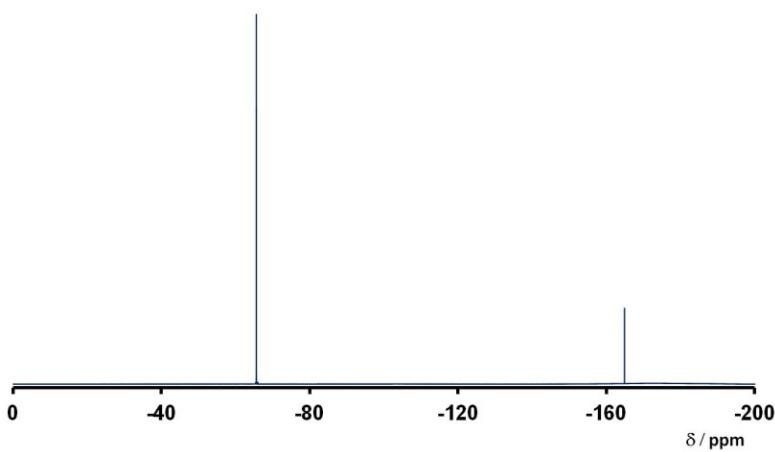
<sup>1</sup>H-NMR |



<sup>13</sup>C-NMR

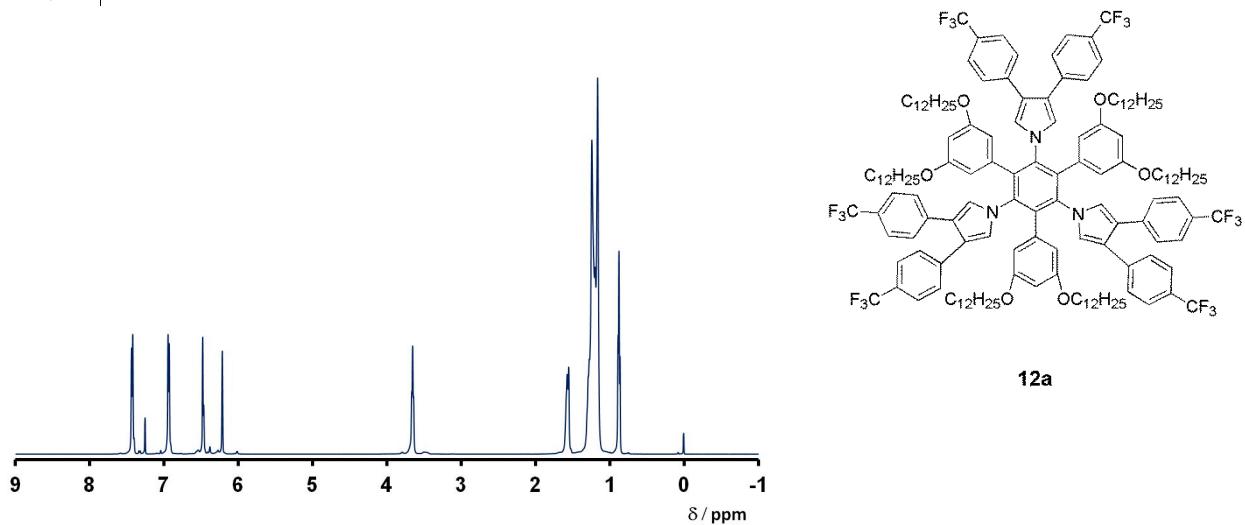


<sup>19</sup>F-NMR

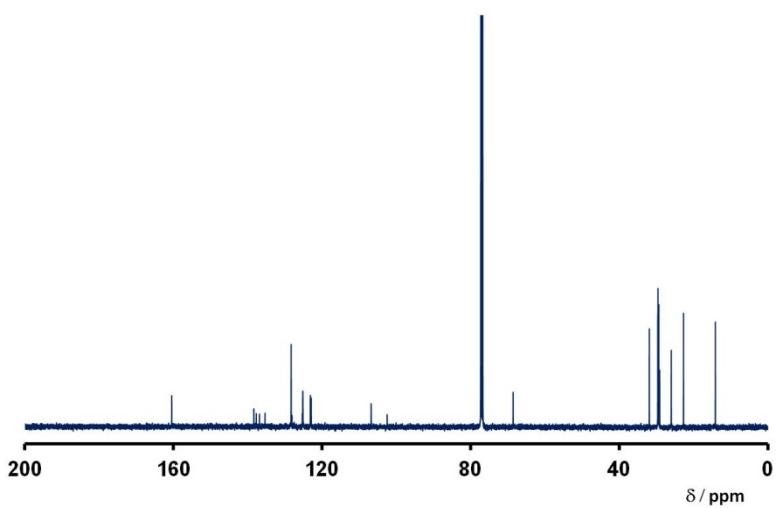


10. **12a** in  $\text{CDCl}_3$

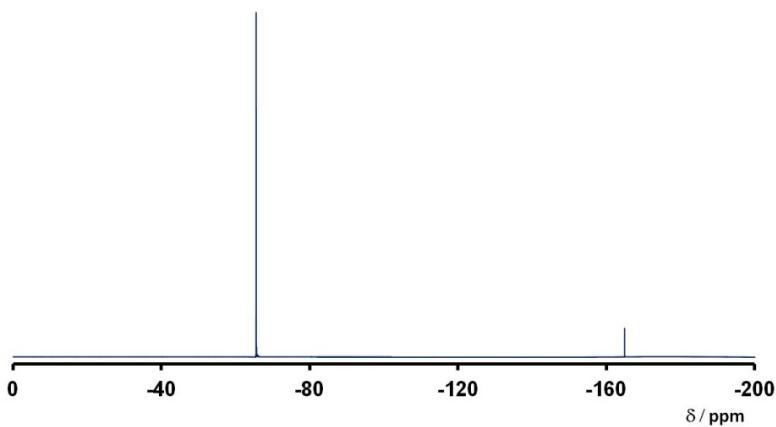
$^1\text{H-NMR}$ |



$^{13}\text{C-NMR}$

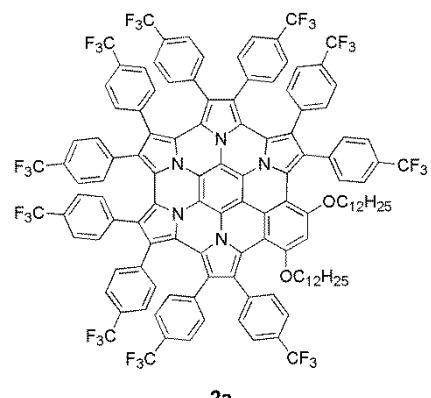
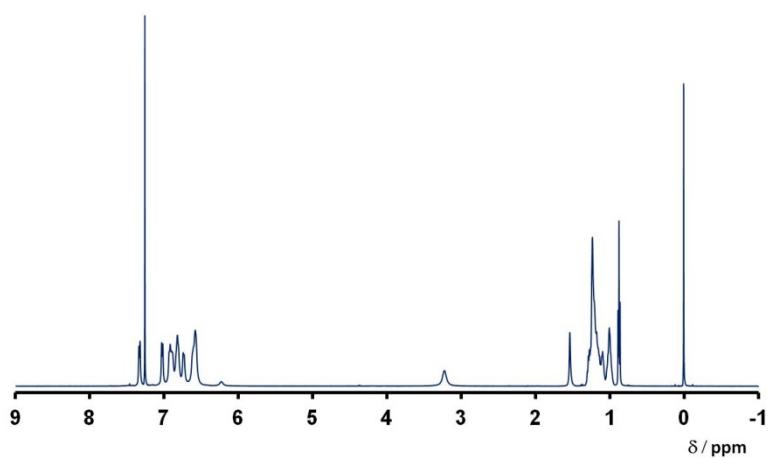


$^{19}\text{F-NMR}$

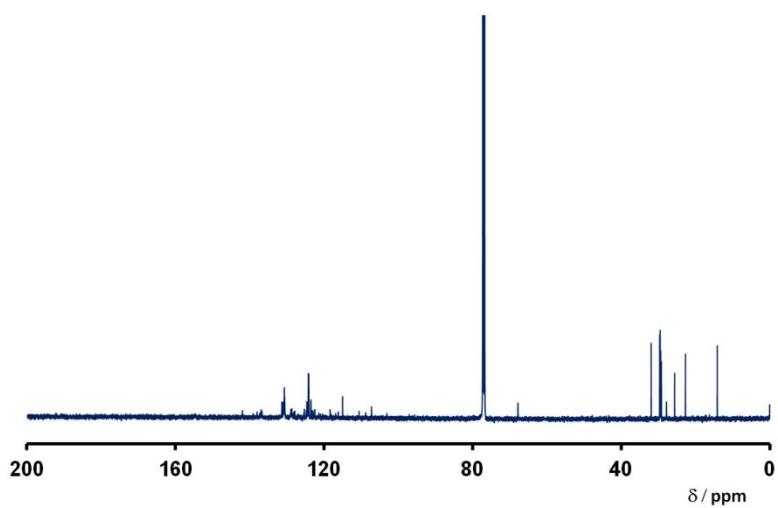


11. **2a** in  $\text{CDCl}_3$

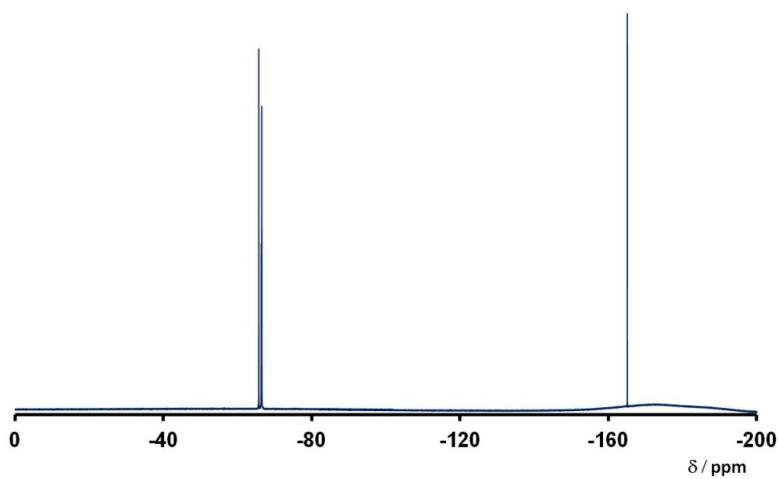
$^1\text{H}$ -NMR



$^{13}\text{C}$ -NMR

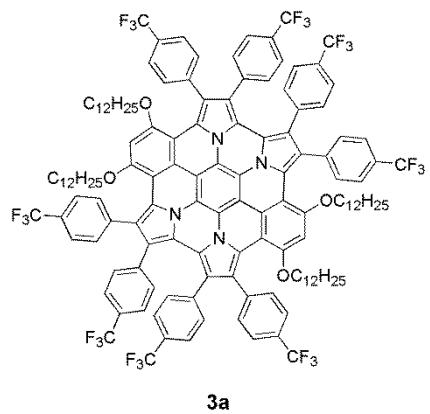
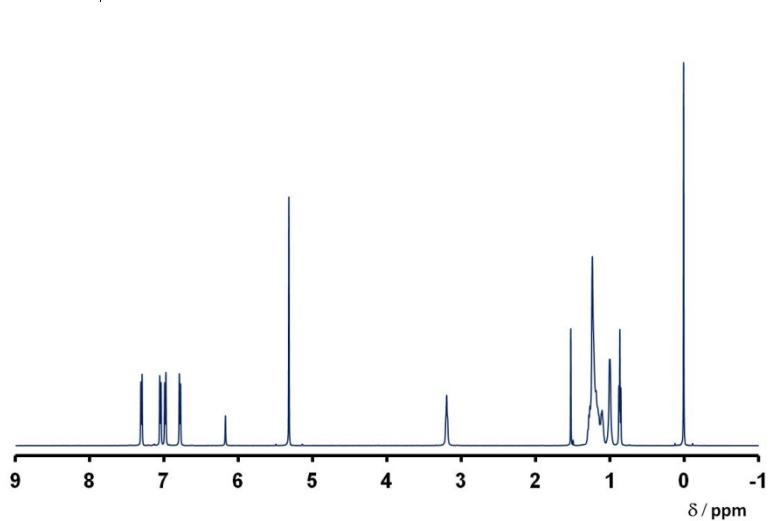


$^{19}\text{F}$ -NMR

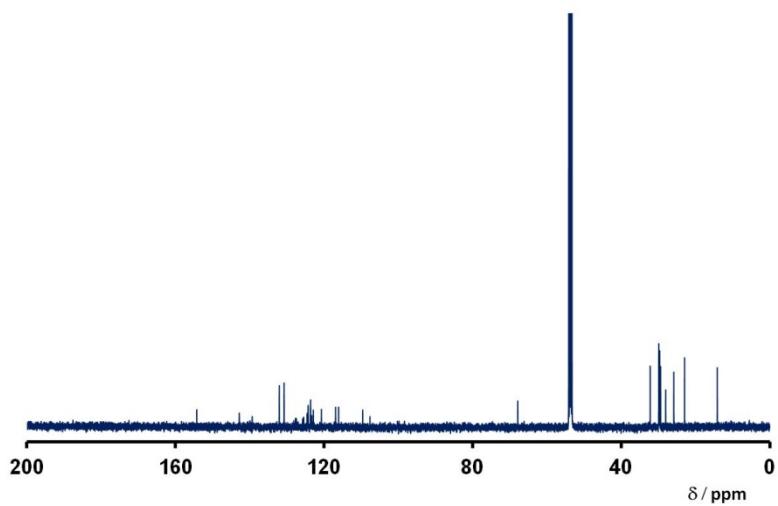


12. **3a** in CD<sub>2</sub>Cl<sub>2</sub>

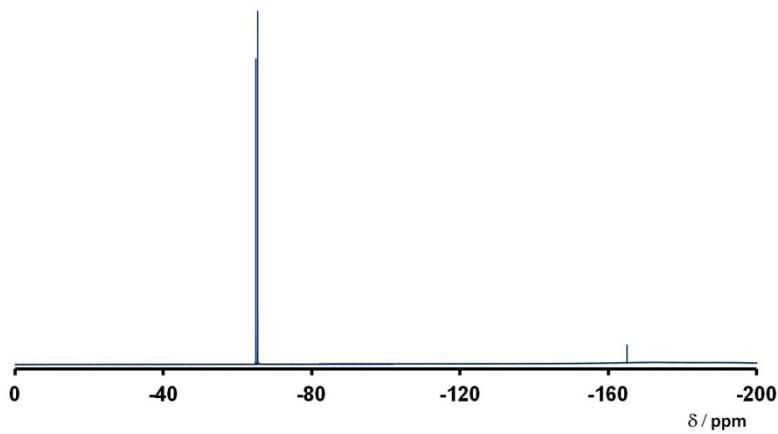
<sup>1</sup>H-NMR |



<sup>13</sup>C-NMR

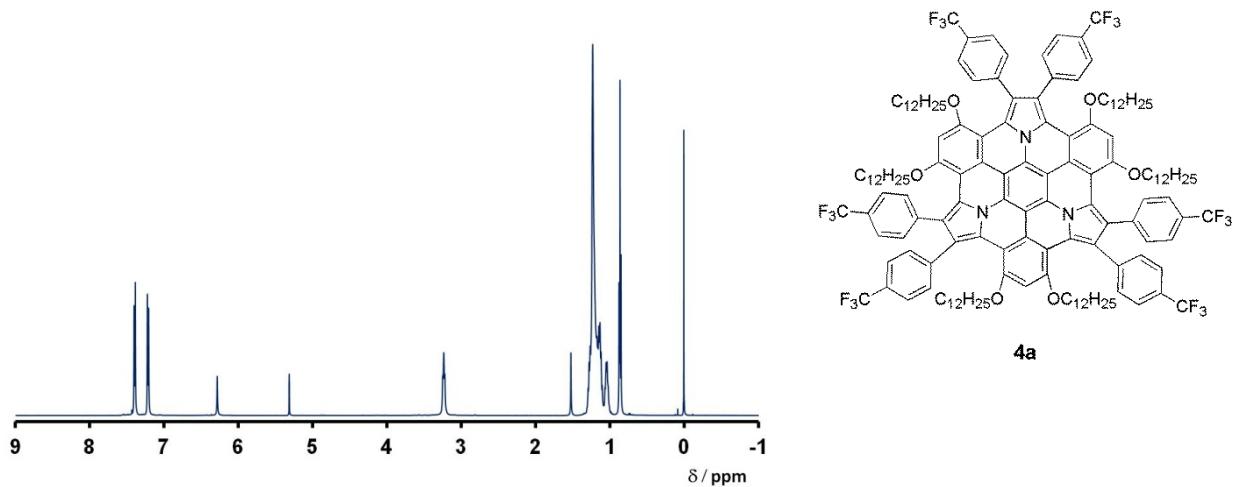


<sup>19</sup>F-NMR

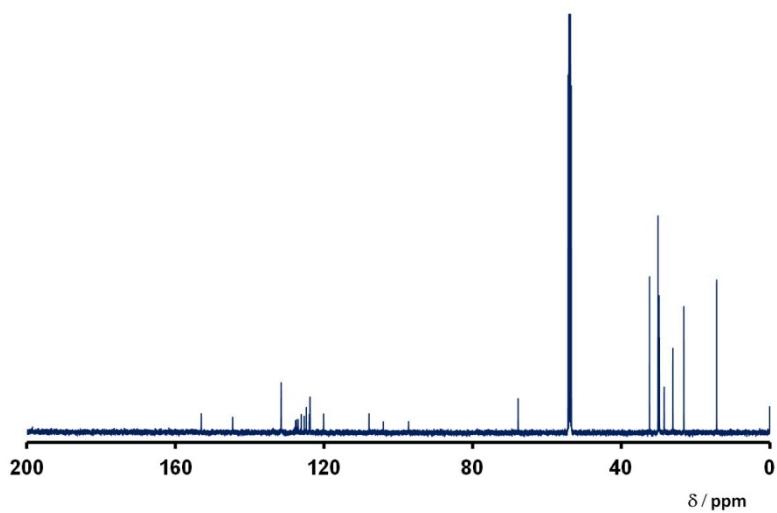


13. **4a** in CD<sub>2</sub>Cl<sub>2</sub>

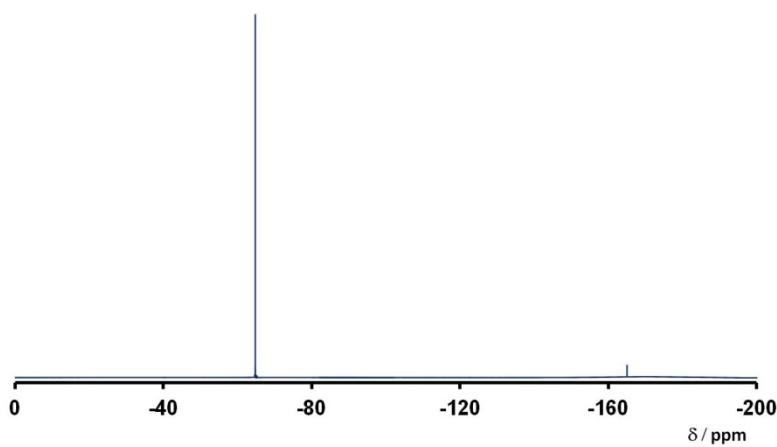
<sup>1</sup>H-NMR |



<sup>13</sup>C-NMR

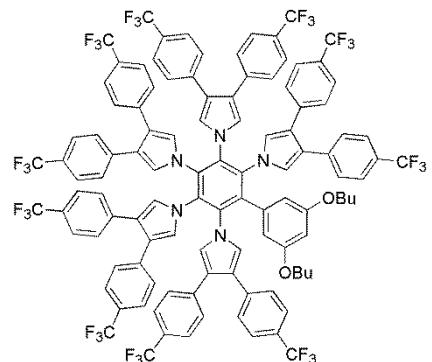
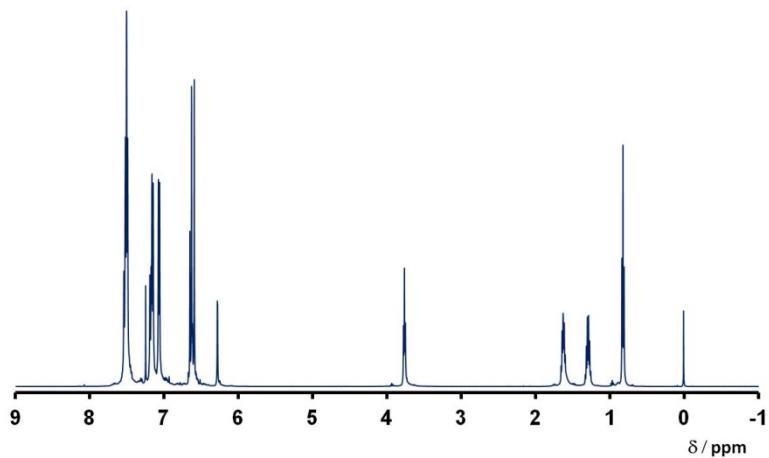


<sup>19</sup>F-NMR

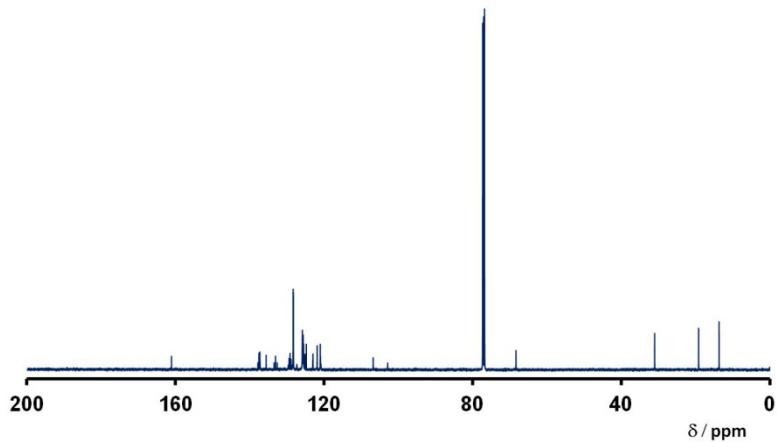


14. **9b** in CDCl<sub>3</sub>

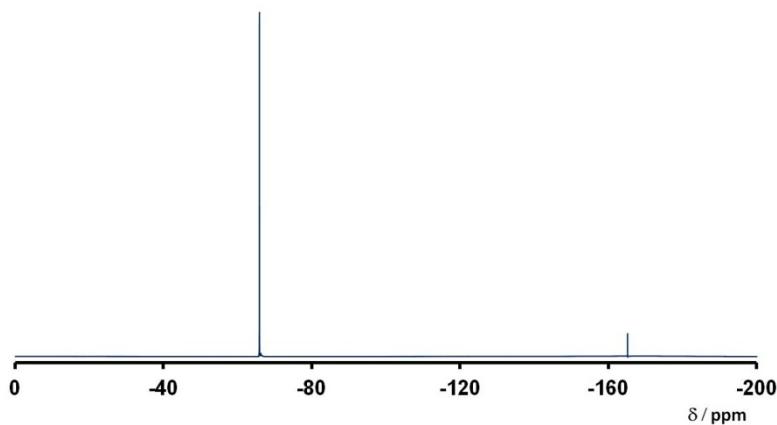
<sup>1</sup>H-NMR |



<sup>13</sup>C-NMR

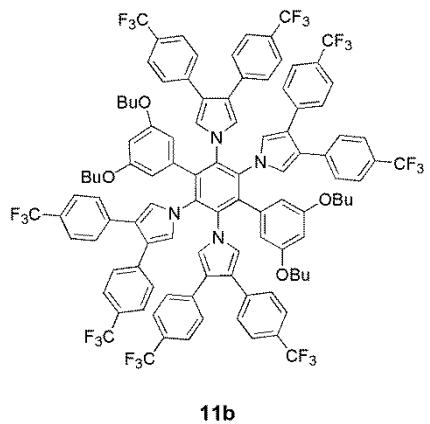
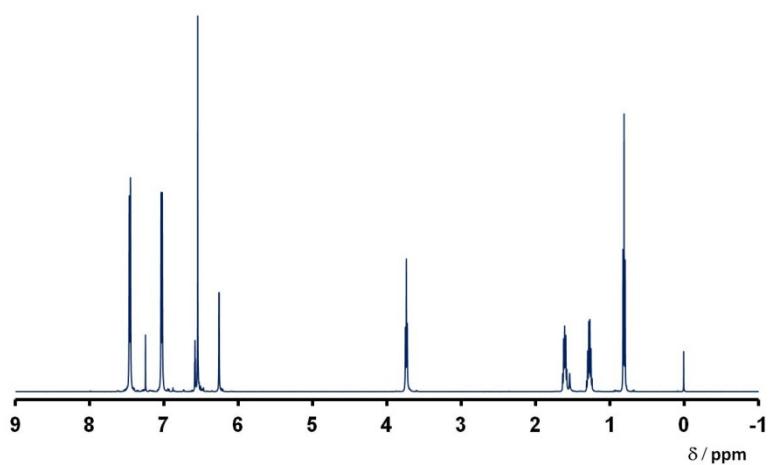


<sup>19</sup>F-NMR

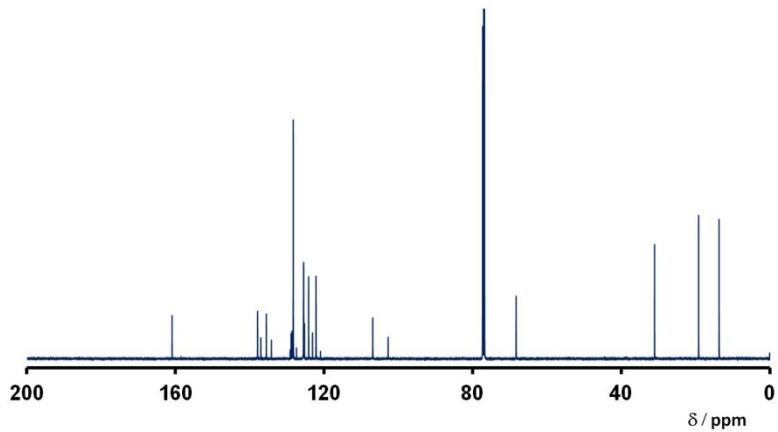


15. **11b** in CDCl<sub>3</sub>

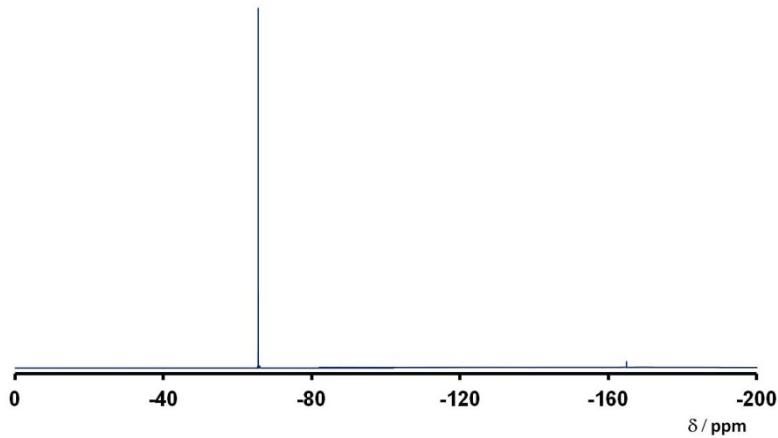
<sup>1</sup>H-NMR |



<sup>13</sup>C-NMR

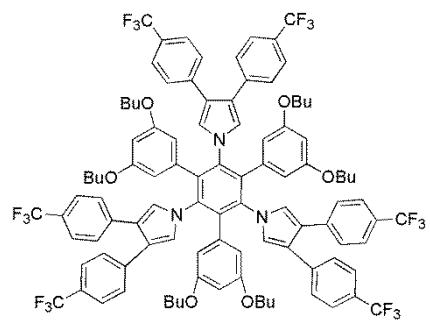
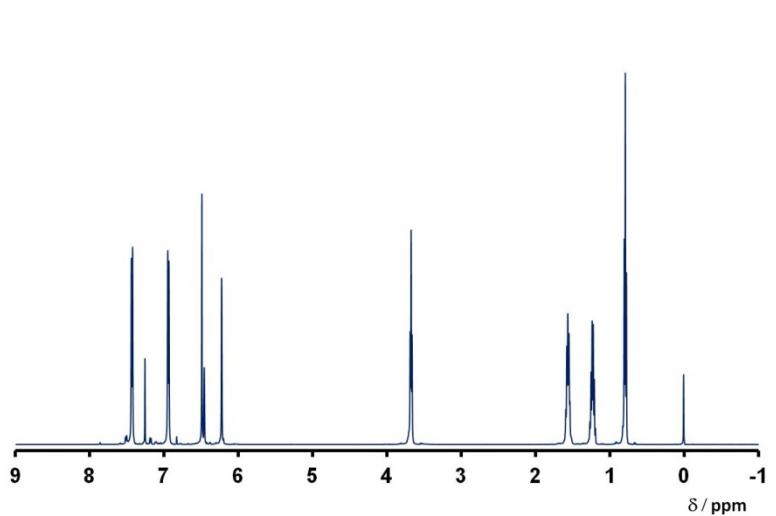


<sup>19</sup>F-NMR



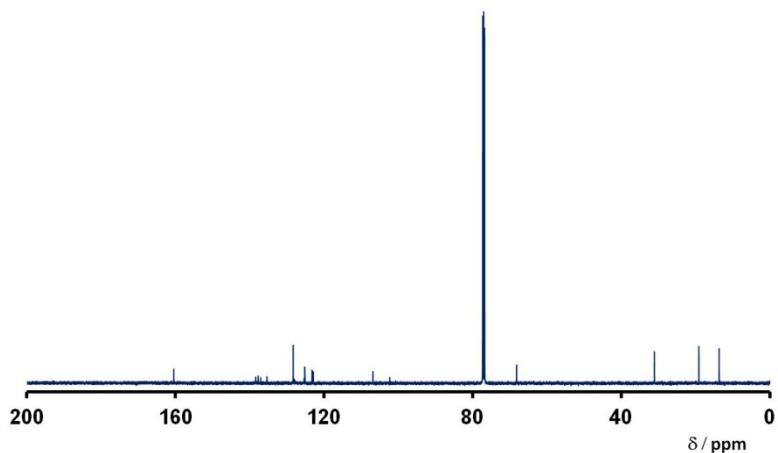
16. **12b** in CDCl<sub>3</sub>

<sup>1</sup>H-NMR

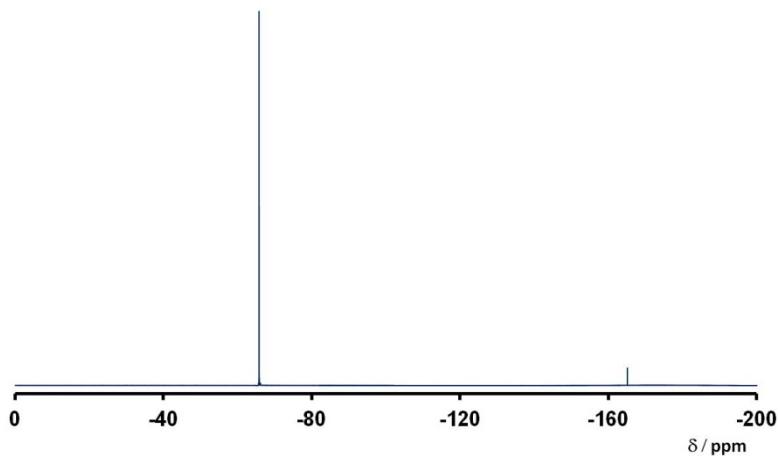


**12b**

<sup>13</sup>C-NMR

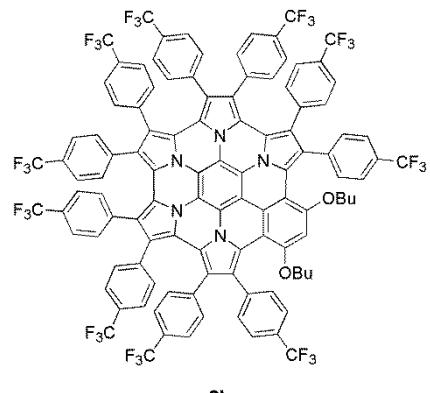
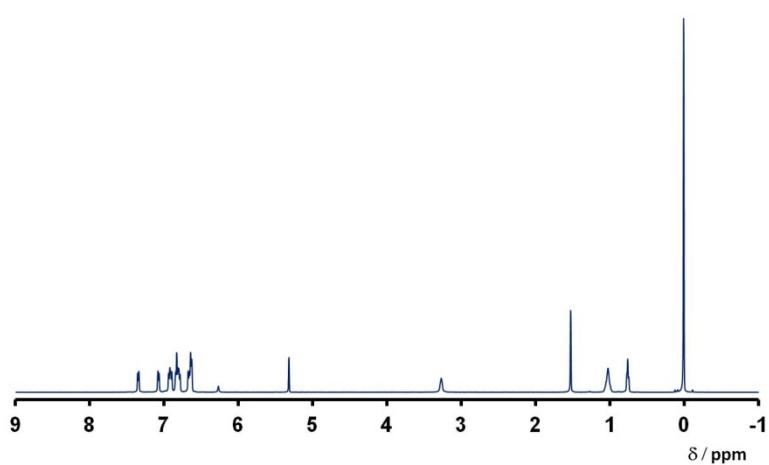


<sup>19</sup>F-NMR



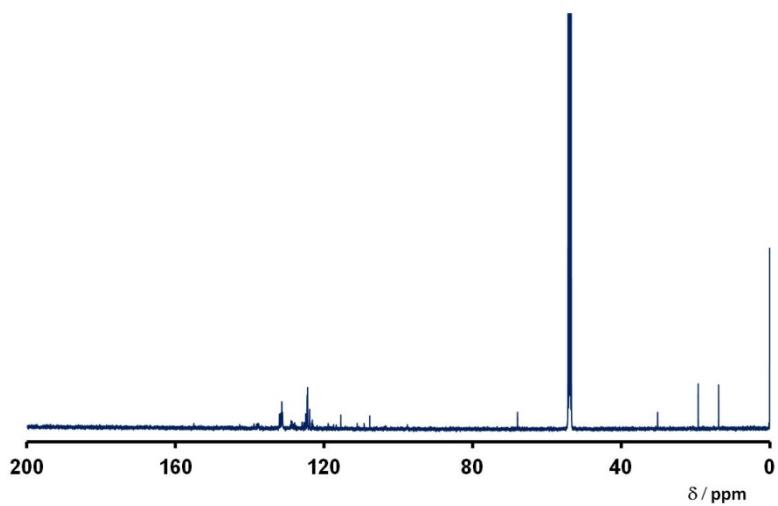
17. **2b** in  $\text{CD}_2\text{Cl}_2$

$^1\text{H}$ -NMR

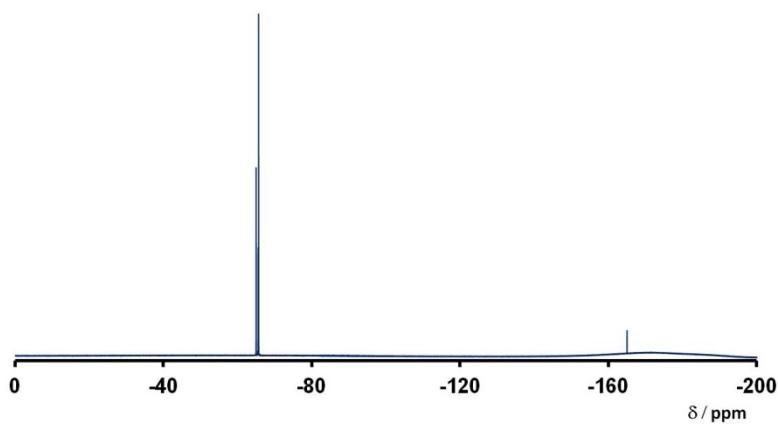


**2b**

$^{13}\text{C}$ -NMR

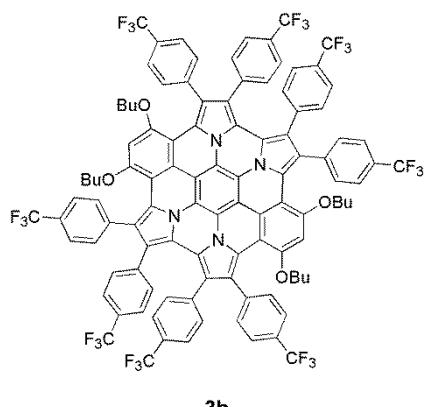
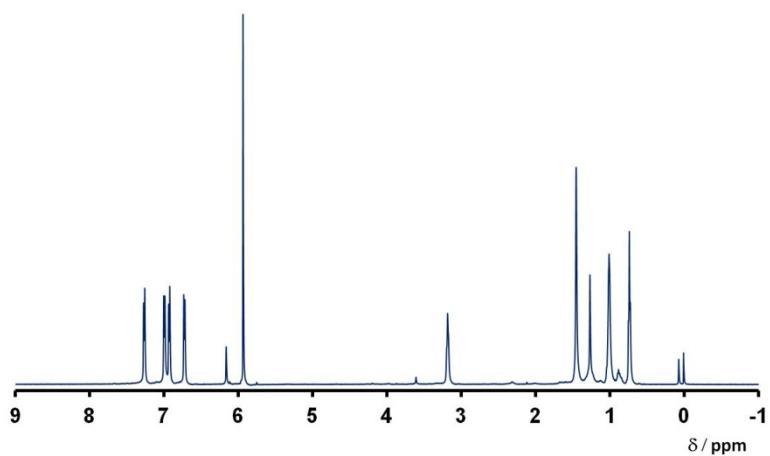


$^{19}\text{F}$ -NMR

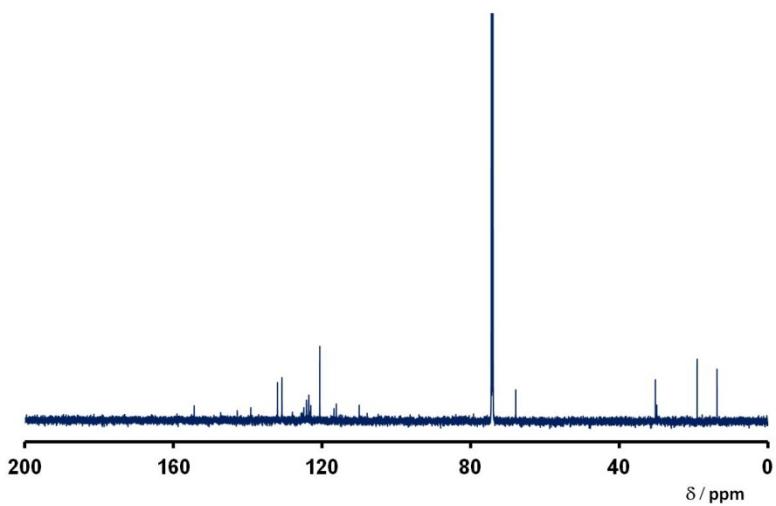


18. **3b** in CD<sub>2</sub>Cl<sub>2</sub>

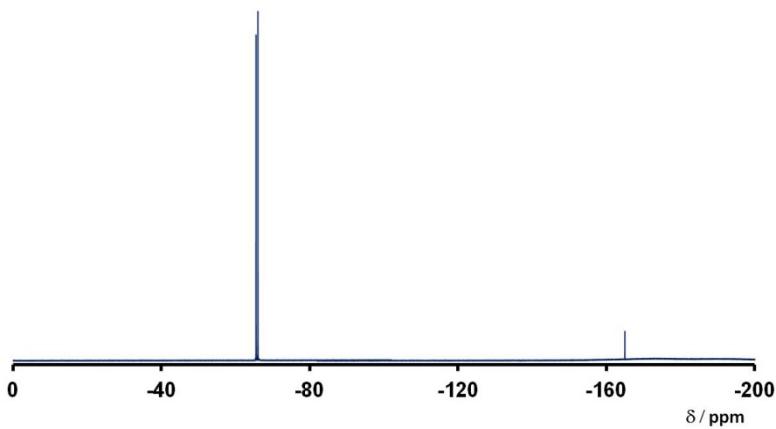
<sup>1</sup>H-NMR |



<sup>13</sup>C-NMR

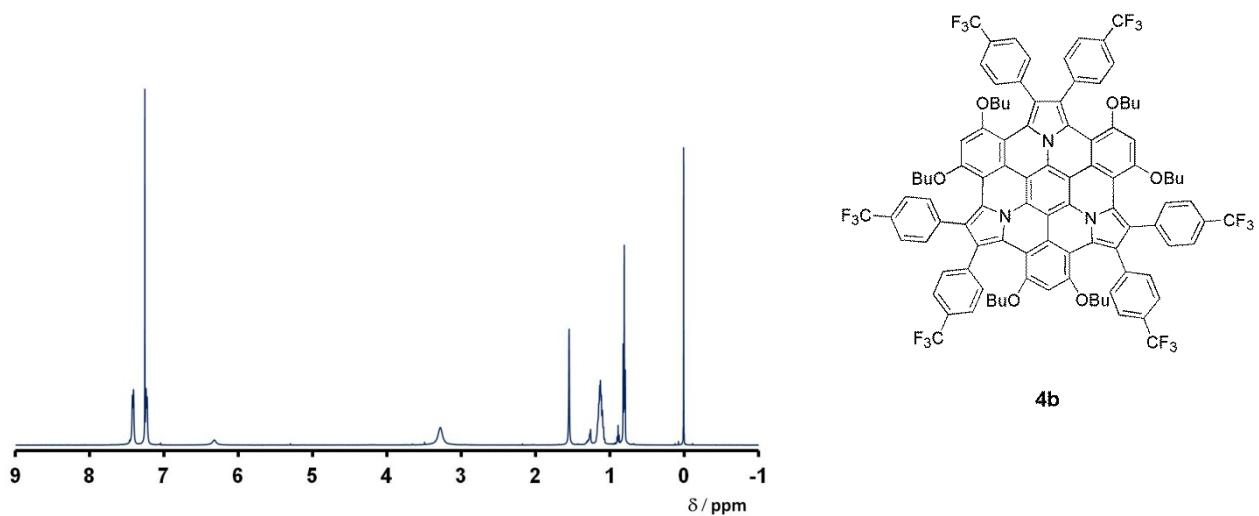


<sup>19</sup>F-NMR

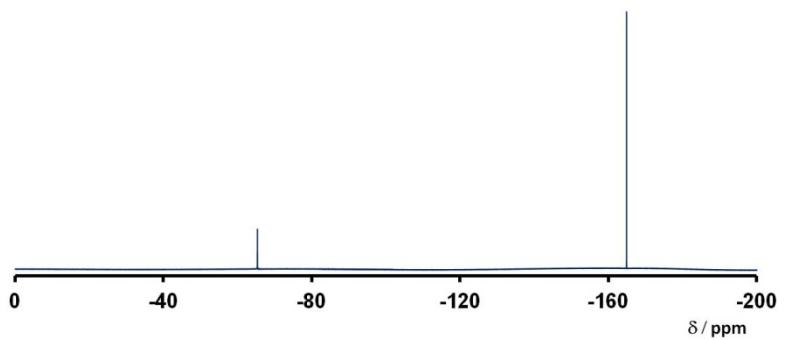


19. **4b** in CD<sub>2</sub>Cl<sub>2</sub>

<sup>1</sup>H-NMR |

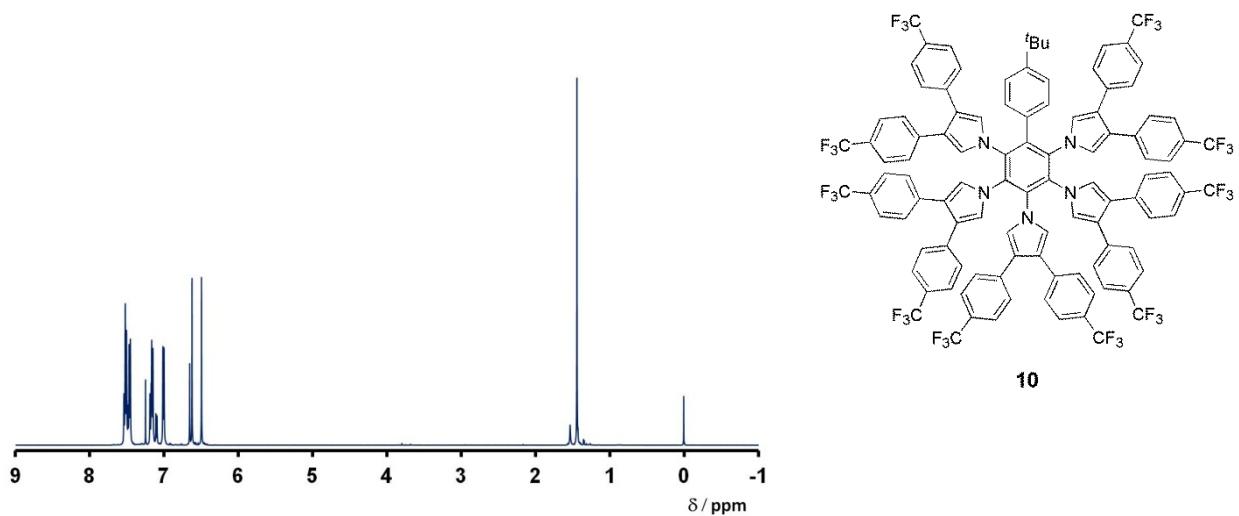


<sup>19</sup>F-NMR

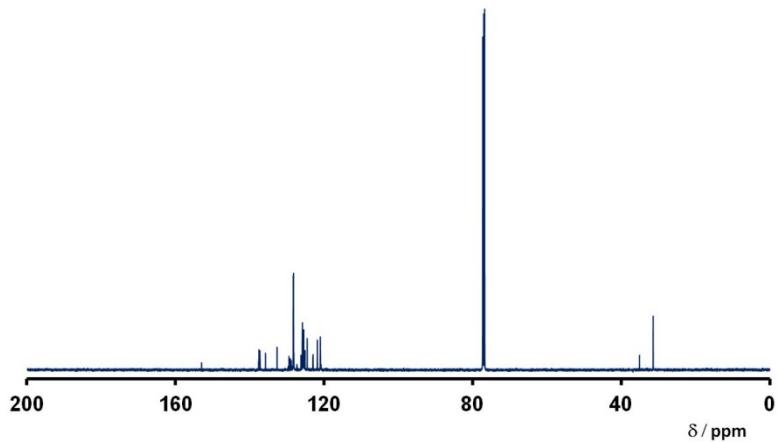


20. **10** in CDCl<sub>3</sub>

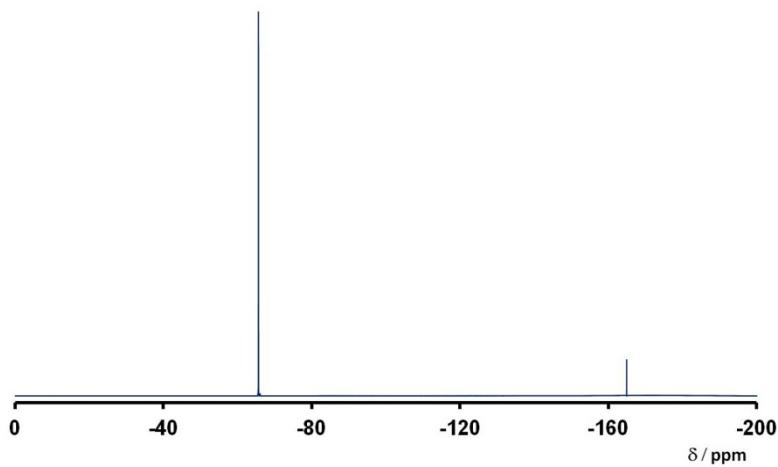
<sup>1</sup>H-NMR |



<sup>13</sup>C-NMR

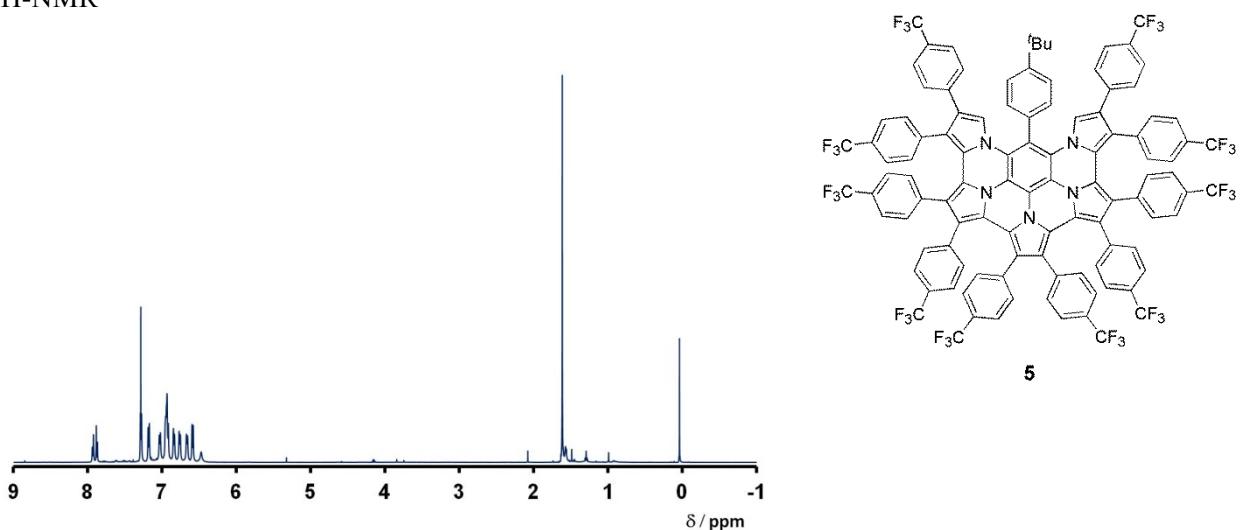


<sup>19</sup>F-NMR

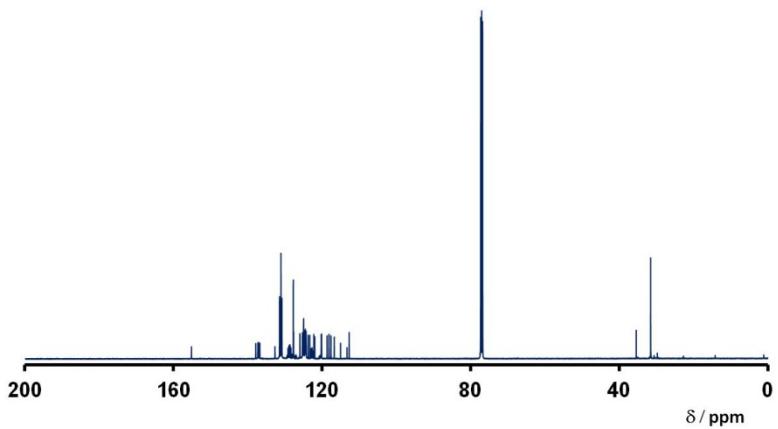


21. **5** in  $\text{CDCl}_3$

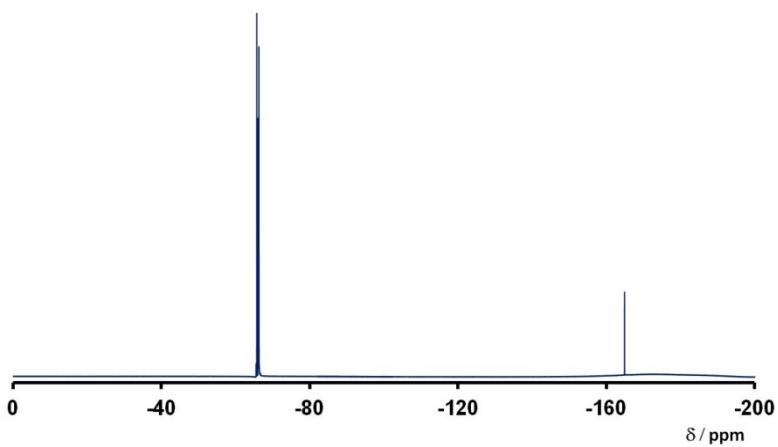
$^1\text{H}$ -NMR



$^{13}\text{C}$ -NMR



$^{19}\text{F}$ -NMR



## S23. References

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