

## Supporting Information

### Chemical Unzipping of WS<sub>2</sub> Nanotubes

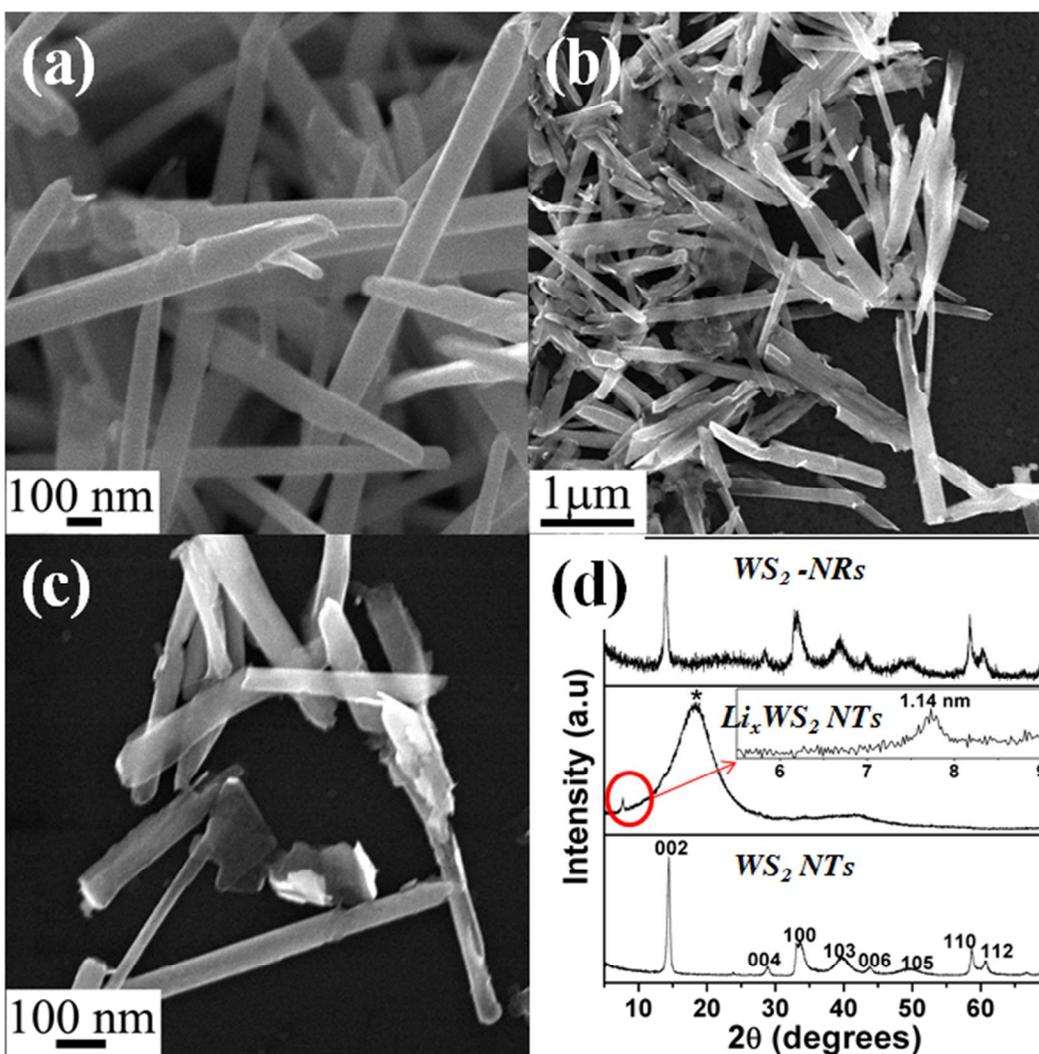
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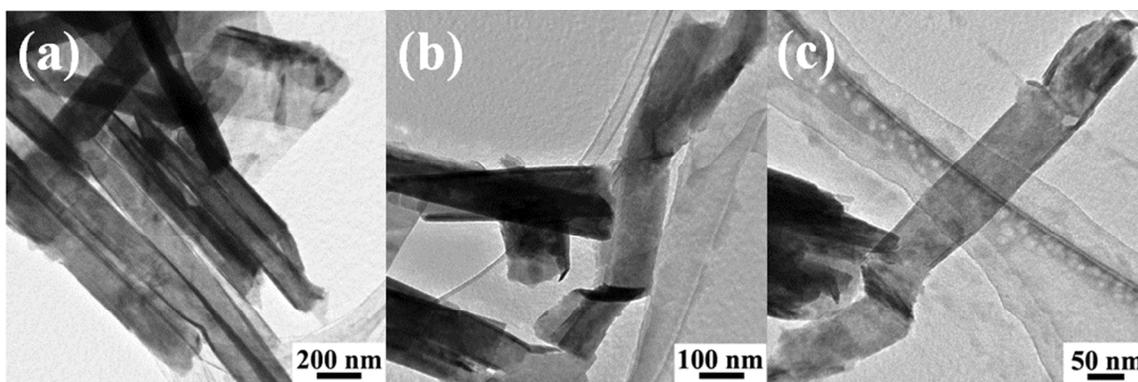
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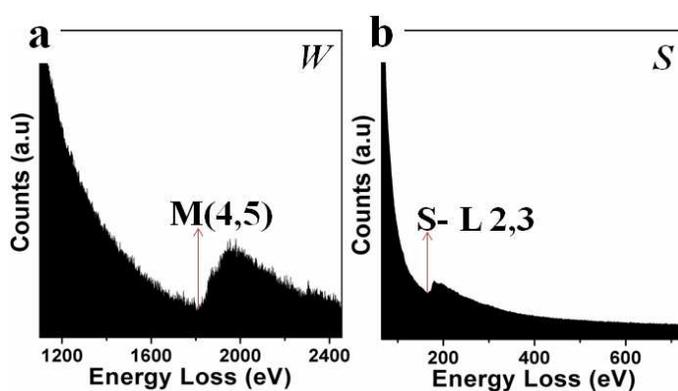
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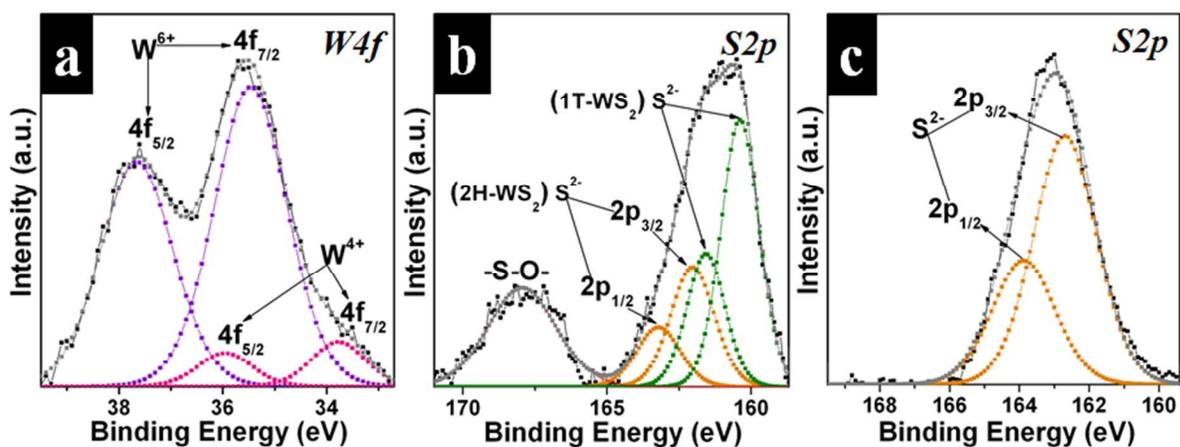
**Figure S1** SEM image of WS<sub>2</sub>-NTs (a), low- (b) and high- (c) magnification SEM images of WS<sub>2</sub>-NRs obtained by unzipping Li<sub>x</sub>WS<sub>2</sub>-NT using octanethiol. XRD patterns (d) of WS<sub>2</sub>-NRs unzipped using octanethiol compared with Li<sub>x</sub>WS<sub>2</sub>-NTs and WS<sub>2</sub>-NTs. (\*) Represents the peak due to paraffin used as a mask to prevent Li<sub>x</sub>WS<sub>2</sub> from reacting with atmospheric moisture.



**Figure S2** (a-c) Bright field TEM images of WS<sub>2</sub>-NRs obtained by unzipping Li<sub>x</sub>WS<sub>2</sub>-NTs using octanethiol.



**Figure S3** Electron energy loss spectra (EELS) of WS<sub>2</sub>-NRs obtained by unzipping Li<sub>x</sub>WS<sub>2</sub>-NTs using octanethiol. W (a) and S (b) edges are separately shown.



**Figure S4** Core level W4f (a) and S2p (b) XPS spectra of Li<sub>x</sub>WS<sub>2</sub>-NTs unzipped using water; and S2p (c) XPS spectra of WS<sub>2</sub>-NRs obtained by unzipping Li<sub>x</sub>WS<sub>2</sub>-NTs using octanethiol.