

Supplementary information

A combined experimental and theoretical study on the formation of crystalline vanadium nitride (VN) in low temperature through a fully solid-state synthesis route

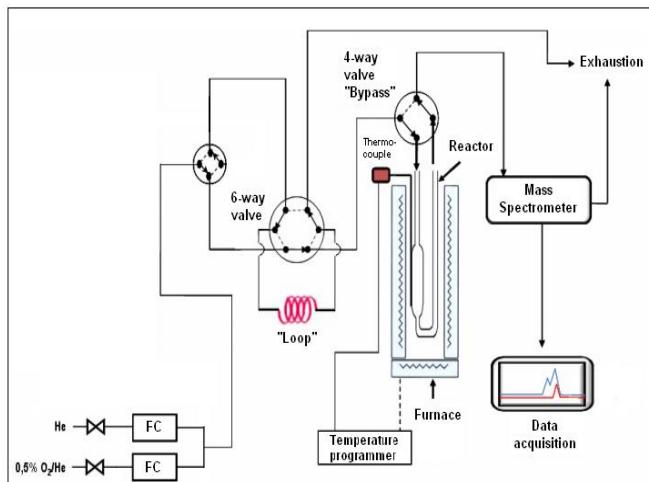
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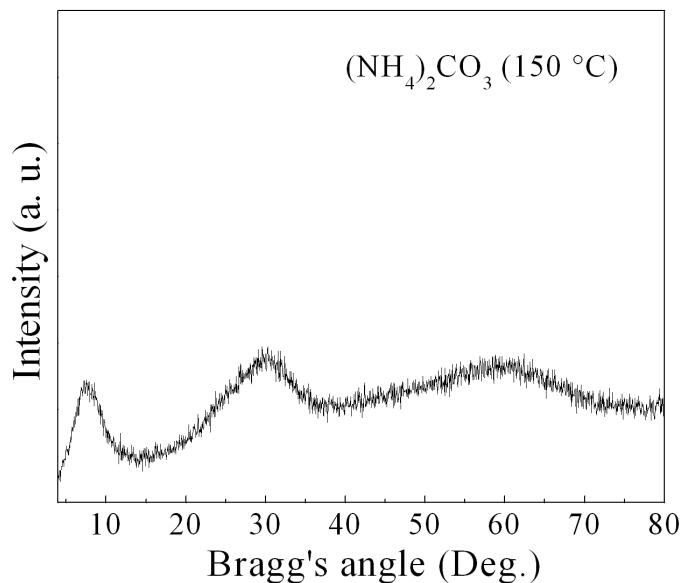
S.1. Multipurpose unit used in the synthesis (FC rely on “flow controller”).



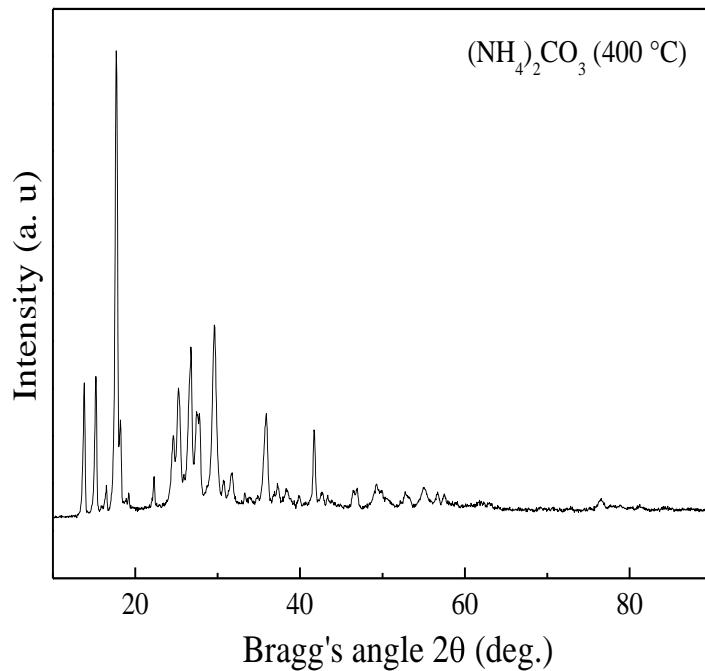
S.2. Diffraction pattern of Ammonium Carbonate (150°C).

It is important to emphasize that the low-crystalline structure obtained for the sub-product, $(\text{NH}_4)_2\text{CO}_3$, was mainly due to the low temperatures employed over the solid-state reaction to form the intermetiate ($\sim 150^\circ\text{C}$), GmV, which consequently led to the formation of a structure predominantly amorphous (S.2). However, when the same sample was subject to higher temperatures (in this case, $\sim 400^\circ\text{C}$) it exhibited high crystallinity and well-characterized diffraction lines of ammonium carbonate in excellent agreement with Sclar and L. C. Garrison (1963)*, as can be seen in S.3.

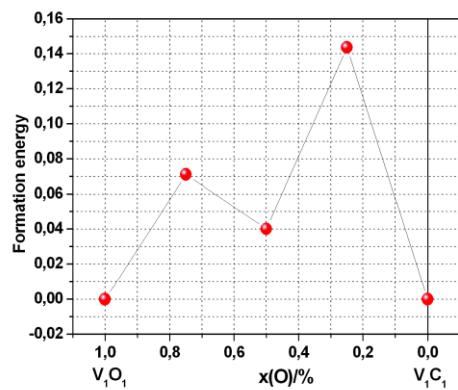
*C. B Sclar and L. C. Garrison, "Phase Composition of Commercial Ammonium Carbonate", *Science*, Vol. 140 no. 3572 pp. 1205-1207 (1963).



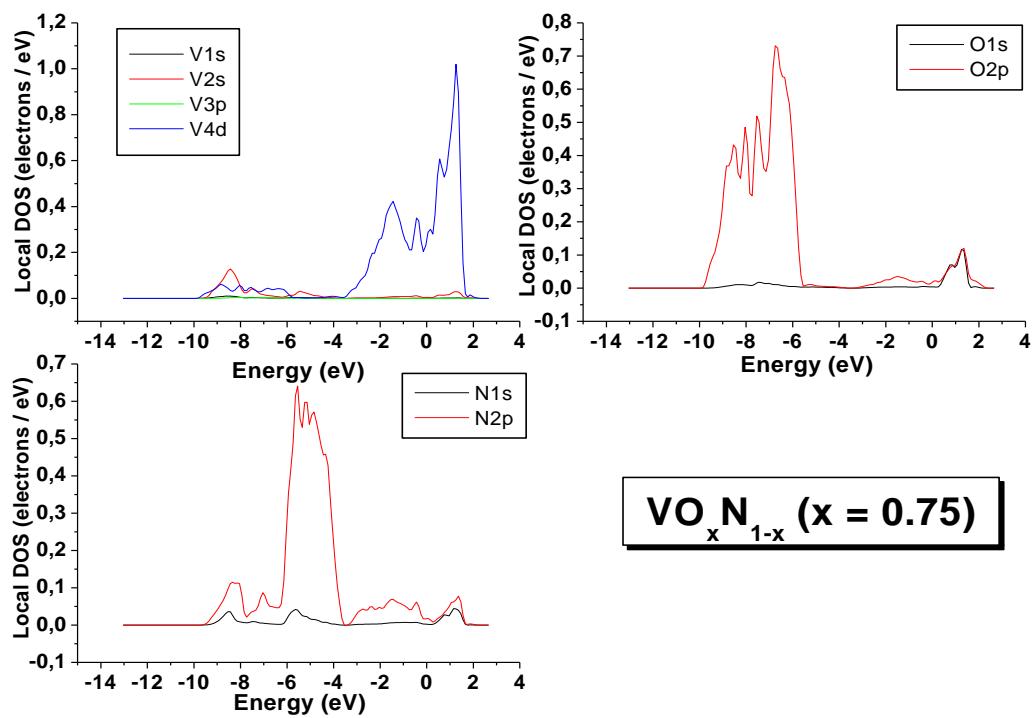
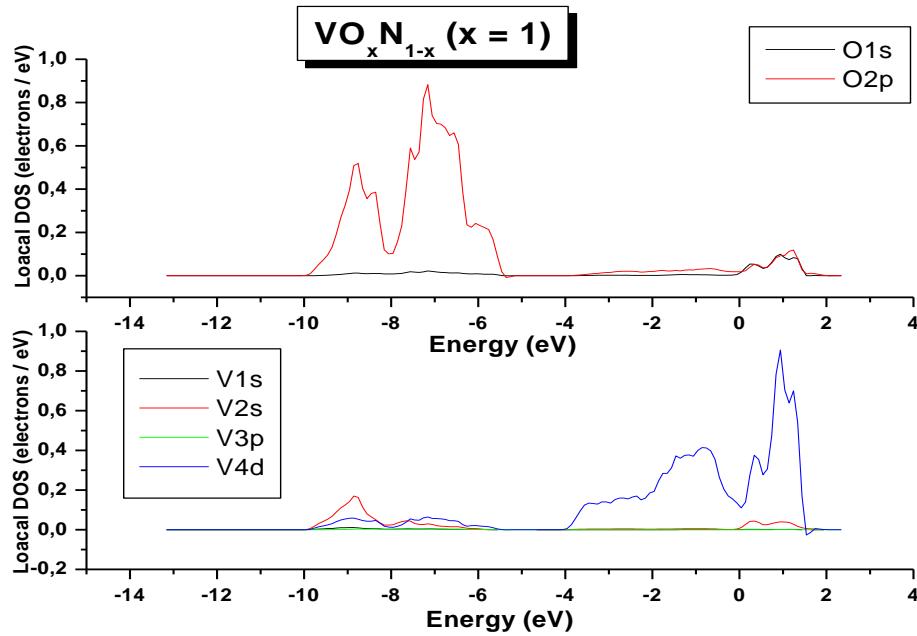
S.3. Diffraction pattern of Ammonium Carbonate (400°C).

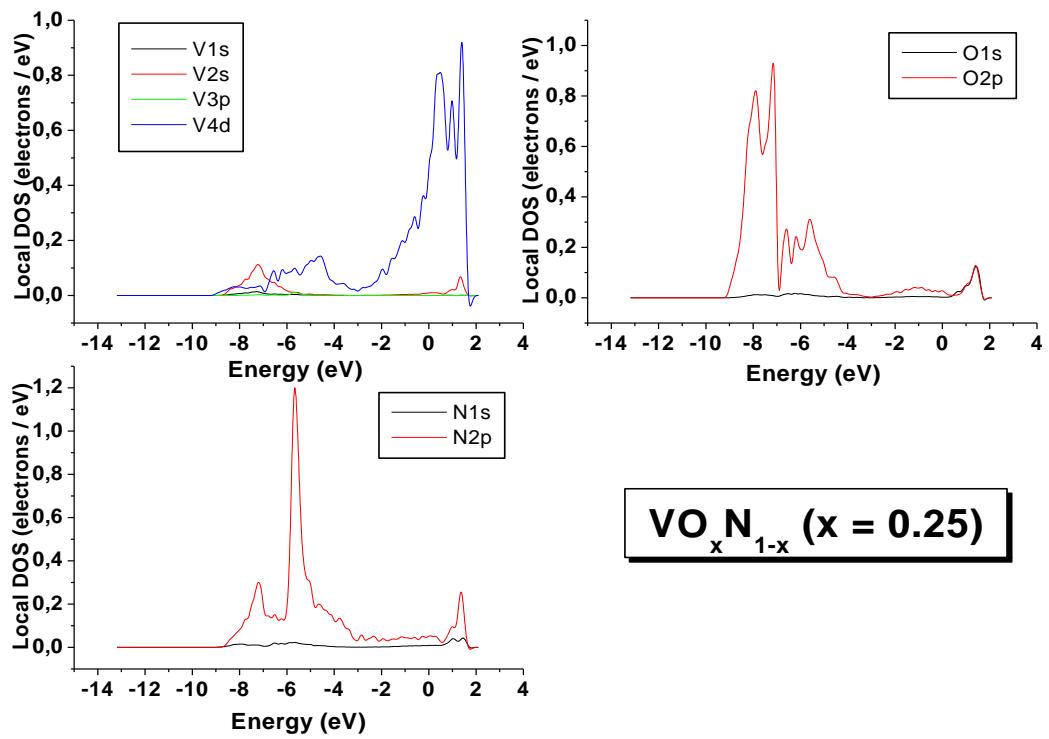
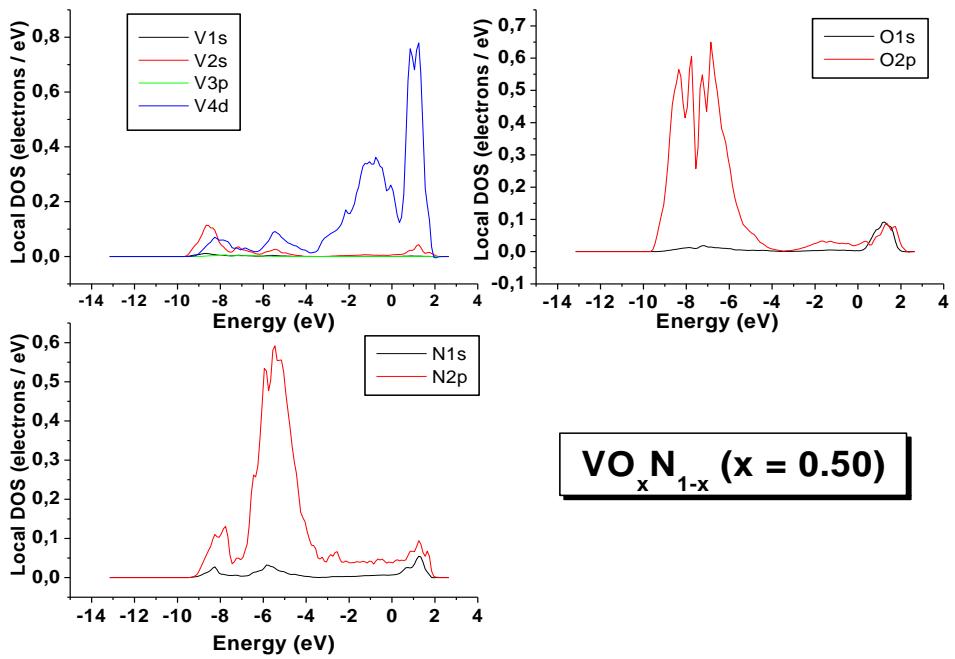


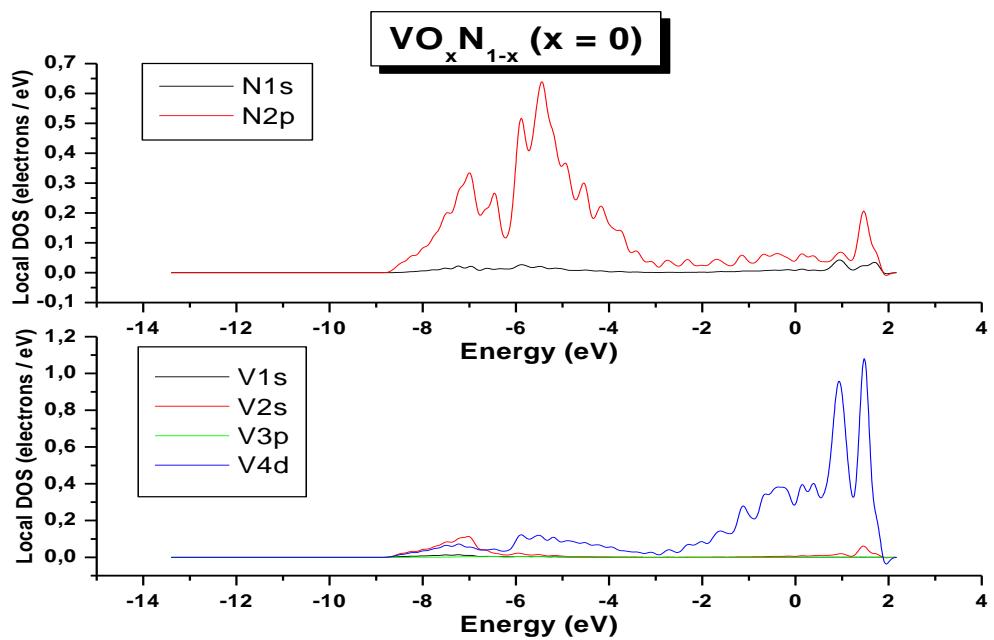
S.4 Calculated formation energy for $\text{VO}_x\text{C}_{1-x}$



S.5 Partial DOS for the $\text{VO}_x\text{N}_{1-x}$ models

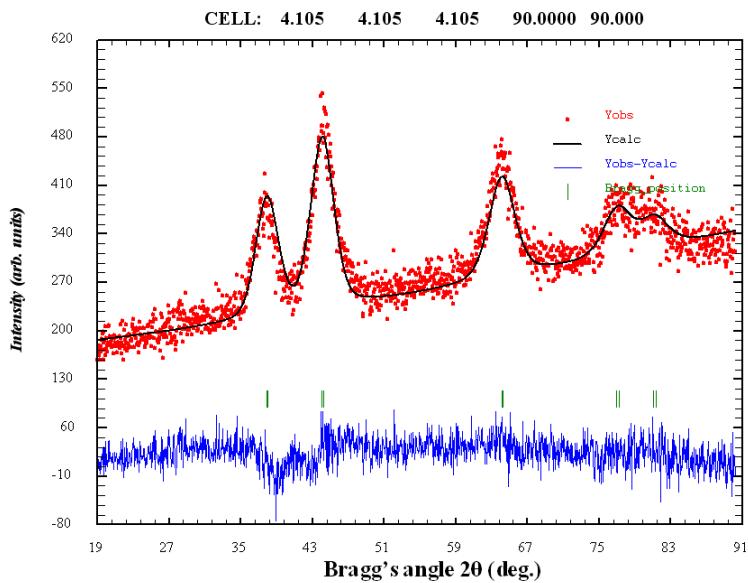




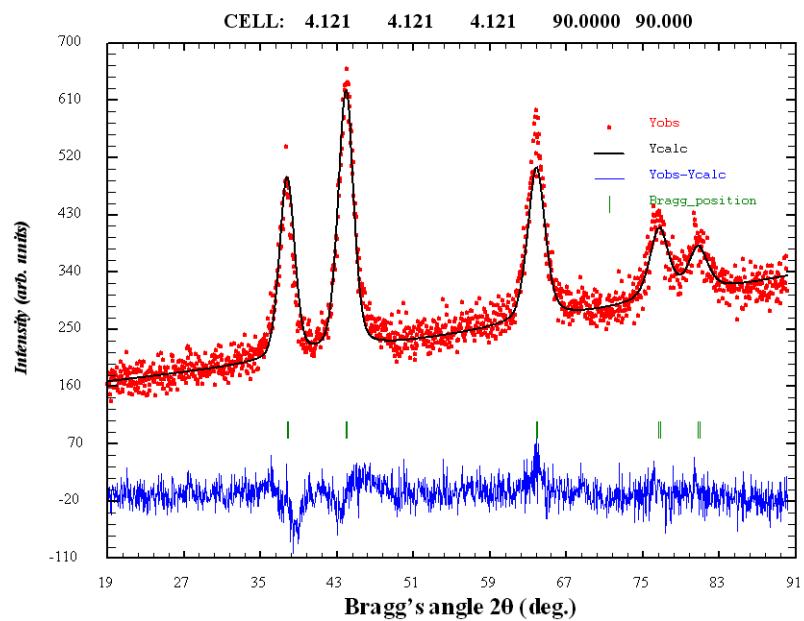


S.6 Diffraction patterns and Rietveld refinement of the thermally treated samples

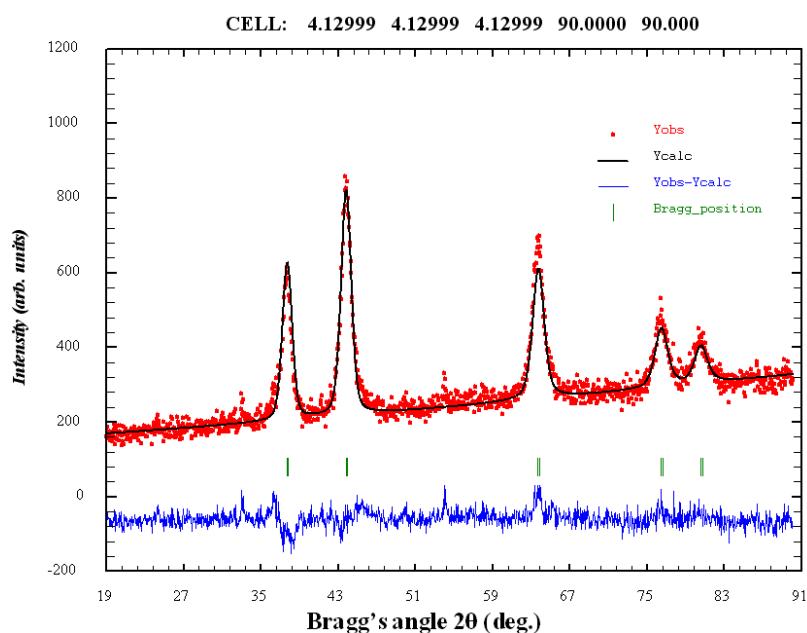
→ 500 °C



→ 600°C



→ 700°C



→ 800°C

