

Hydrogen-Bonded Polymers with Bent-Shaped Side Chains and Poly(4-vinylpyridine) Backbone: Phase Behavior and Thin Film Morphologies.

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Experimental Section

Materials. 3,4,5-trihydroxybenzoic acid methyl ester (98%), 1-bromodecane (98%), K_2CO_3 , KOH, ethanol (99.5%), ethyl acetate (99.8%), acetone (99.9%), dichloromethane (99.5%), 4-(dimethylamino)pyridine (DMAP, 99%), *N,N*-dicyclohexylcarbodiimide (DCC, 95%), 4-benzyloxyphenyl (99%) and palladium carbon catalyst were used as received from Beijing Chemical Co. 1,4-Dioxane (99%) was also purchased from Beijing Chemical Co. and was refluxed over sodium under argon and distilled before use. Poly(4-vinylpyridine) (P4VP, weight-average molecular weight, M_w 5600) was purchased from Polymer Source Inc. Silicon wafers

[p-doped, (100)-oriented, 0.45 mm thick, and 100 mm in diameter] were from Guangzhou Semiconductor Materials (Guangzhou, China).

Nuclear Magnetic Resonance Spectroscopy (^1H NMR) measurements were performed on a Bruker ARX400 MHz spectrometer using with CDCl_3 as solvent, tetramethylsilane (TMS) as the internal standard at room temperature.

10CBP: ^1H NMR(δ , ppm, CDCl_3): 0.86-0.90 (t, 9H, $-\text{CH}_3$), 1.26-1.51 (m, 42H, $-(\text{CH}_2)_7-$), 1.79-1.86 (m, 6H, $3\text{-OCH}_2\text{CH}_2-$), 4.04-4.09 (t, 6H, 3-O-CH_2-), 4.8 (s, 1H, $-\text{OH}$), 6.88-6.91 (d, 2H, Ar-*H*), 7.11-7.14 (d, 2H, Ar-*H*), 7.27-7.31 (m, 2H, Ar-*H*), 7.42 (s, 4H, Ar-*H*), 7.70 (m, 1H, Ar-*H*), 8.46-8.49 (d, 2H, Ar-*H*), 9.02 (s, 1H, Ar-*H*). Elemental Analysis: Anal. Calcd for $\text{C}_{57}\text{H}_{78}\text{O}_{10}$: C, 74.15; H, 8.52. Found: C, 74.17; H, 8.60.

12CBP: ^1H NMR(δ , ppm, CDCl_3): 0.86-0.90 (t, 9H, $-\text{CH}_3$), 1.26-1.54 (m, 54H, $-(\text{CH}_2)_9-$), 1.73-1.87 (m, 6H, $3\text{-OCH}_2\text{CH}_2-$), 4.04-4.09 (t, 6H, 3-O-CH_2-), 5.14 (s, 1H, $-\text{OH}$), 6.87-6.90 (d, 2H, Ar-*H*), 7.09-7.12 (d, 2H, Ar-*H*), 7.27-7.32 (m, 2H, Ar-*H*), 7.41 (s, 4H, Ar-*H*), 7.68-7.71 (m, 1H, Ar-*H*), 8.45-8.47 (d, 2H, Ar-*H*), 9.01 (s, 1H, Ar-*H*). Elemental Analysis: Anal. Calcd for $\text{C}_{63}\text{H}_{90}\text{O}_{10}$: C, 75.11; H, 9.00. Found: C, 75.22; H, 9.16.

14CBP: ^1H NMR(δ , ppm, CDCl_3): 0.86-0.90 (t, 9H, $-\text{CH}_3$), 1.26-1.54 (m, 66H, $-(\text{CH}_2)_7-$), 1.75-1.86 (m, 6H, $3\text{-OCH}_2\text{CH}_2-$), 4.04-4.08 (t, 6H, 3-O-CH_2-), 6.88-6.90 (d, 2H, Ar-*H*), 7.11-7.13 (d, 2H, Ar-*H*), 7.26-7.33 (m, 2H, Ar-*H*), 7.41 (s, 4H, Ar-*H*), 7.68-7.71 (m, 1H, Ar-*H*), 8.45-8.47 (d, 2H, Ar-*H*), 9.01 (s, 1H, Ar-*H*). Elemental Analysis: Anal. Calcd for $\text{C}_{69}\text{H}_{102}\text{O}_{10}$: C, 75.42; H, 9.42. Found: C, 75.63; H, 9.46.

Electron Density Reconstruction

The reconstruction of relative electron density distribution in real space based on XRD data is calculated according to the literature¹. Considering the hexagonal columnar phase, three clear peaks can be assigned for PVP(10CBP)_{0.75}. These peaks are related to reflection (10), (11), (20). The electron density profiles have been calculated using the suitable phase combinations of “+ – –” for the corresponding reflections.

Supporting Figures

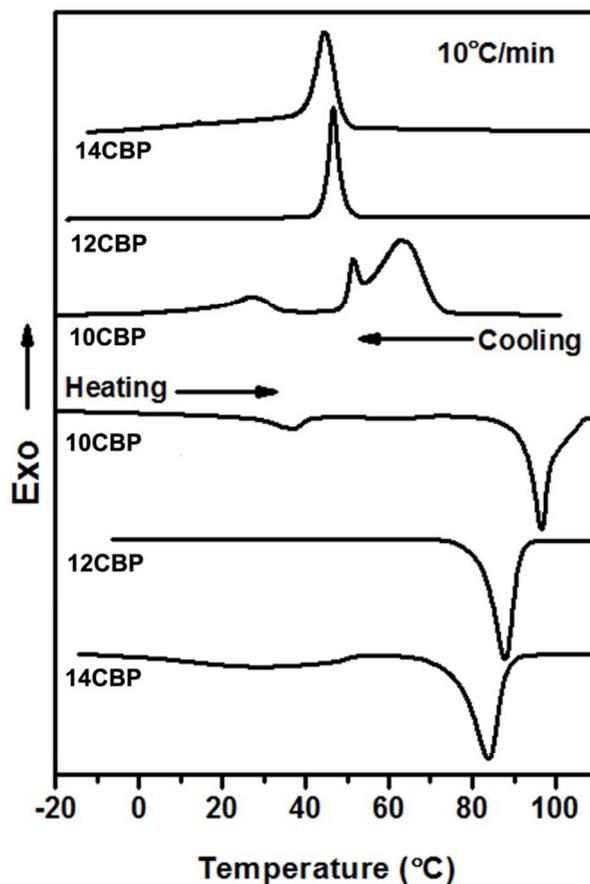


Figure S1.DSC curves of nCBP during first cooling and second heating at a rate of 10 °C/min under N₂ atmosphere

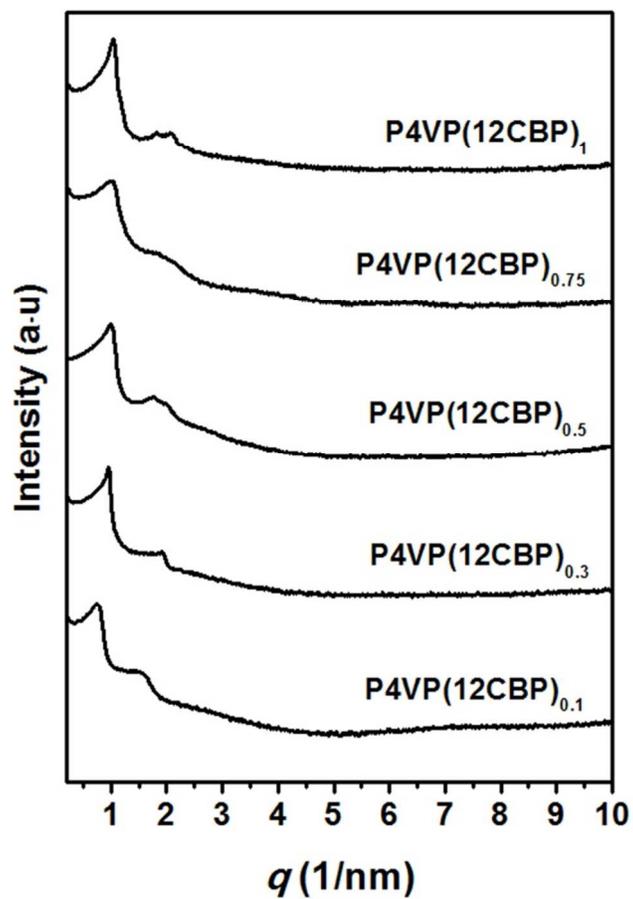


Figure S2. SAXS profiles of P4VP(12CBP)_x complexes recorded at 25 °C

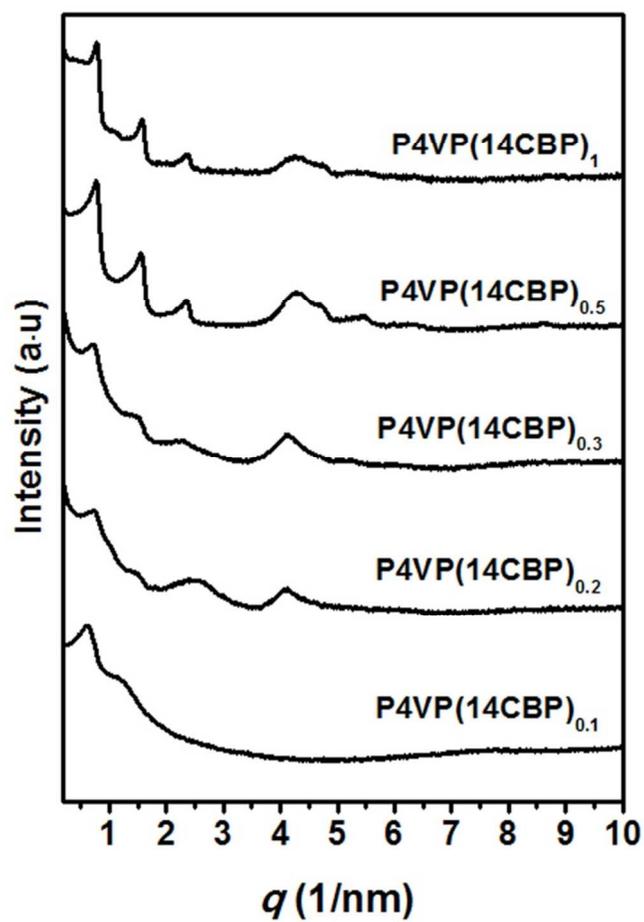


Figure S3. SAXS profiles of P4VP(14CBP)_x complexes recorded at 25 °C.

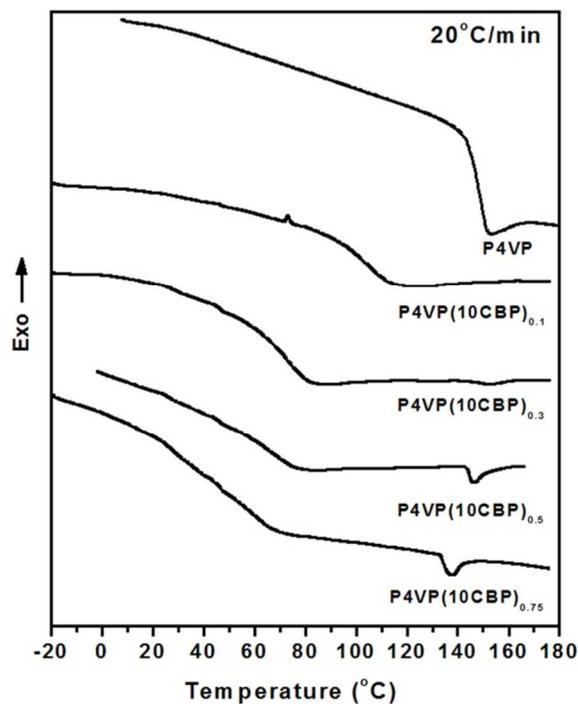


Figure S4. DSC curves of P4VP and P4VP(10CBP)_x during the second heating cycle at a rate of 20 °C/min under N₂ atmosphere.

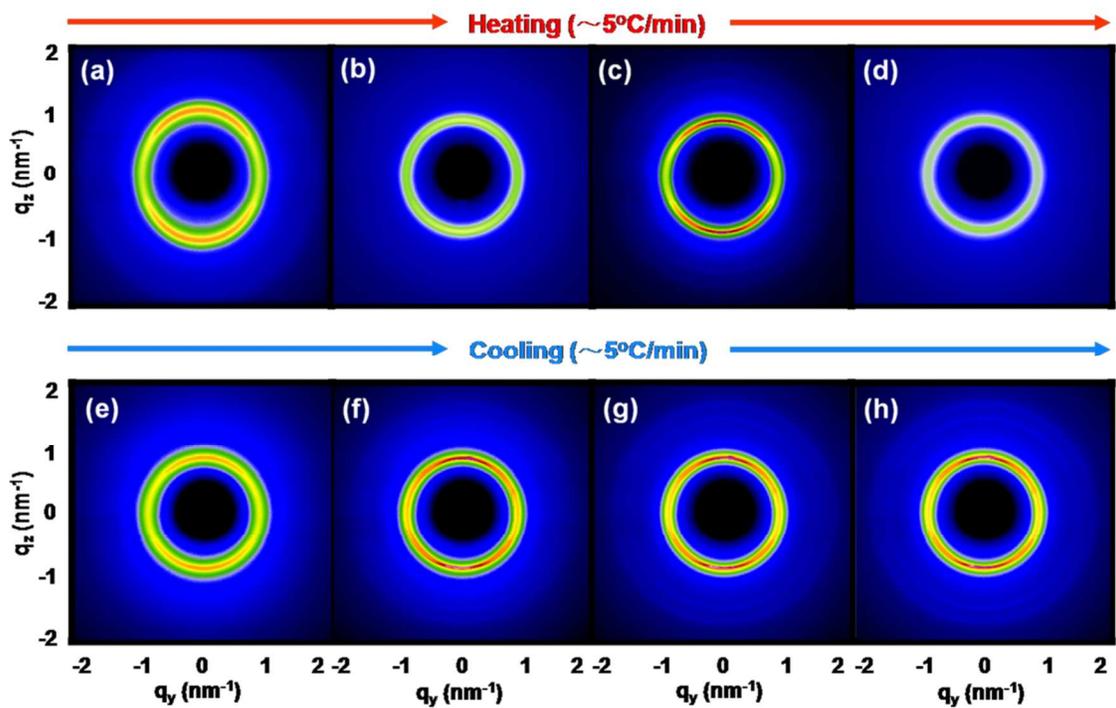


Figure S5. 2D X-ray diffraction patterns of P4VP(12CBP)_{0.3} during the heating and cooling cycles. Data from the heating cycle were taken at (a) 30 °C, (b) 120 °C, (c) 140 °C, and (d) 180 °C. After stabilized ~10 min at 180 °C, data were collected during the cooling cycle at (e) 160 °C, (f) 130 °C, (g) 70 °C, and (h) 30 °C. Heating and cooling rates were ~5 °C/min.

REFERENCE:

- (1) Zheng, J. F.; Liu, X.; Chen, X. F.; Ren, X. K.; Yang, S.; Chen, E. Q. *ACS Macro Letters* **2012**, *1*, 641-645.