

Supporting Information

Mono-, Di- and Tri-borylphosphine Analogues of Triarylphosphines

*Jonathan A. Bailey, Marten Ploeger, Paul G. Pringle**

School of Chemistry, University of Bristol, Cantock's Close, Bristol BS8 1TS, UK

To whom correspondence should be addressed:

e-mail, paul.pringle@bristol.ac.uk; fax: +44 (0)117 929 0509; tel, +44 (0)117 928 8114;
web, www.inchm.bris.ac.uk/people/pringle/welcome.html

Table S1. Crystal data

Compound	I	L₁	L₂
Empirical formula	C ₁₀ H ₈ BClN ₂	C ₂₂ H ₁₈ BN ₂ P	C ₂₆ H ₂₁ B ₂ N ₄ P
Formula weight	202.44	352.16	442.06
Temperature/K	100(2)	100(2)	100(2)
Crystal system	monoclinic	triclinic	orthorhombic
Space group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> -1	<i>Pca</i> 2 ₁
<i>a</i> /Å	8.1003(2)	9.1263(8)	13.6850(7)
<i>b</i> /Å	3.79990(10)	10.6800(10)	19.4779(11)
<i>c</i> /Å	29.2151(7)	11.0452(10)	8.0751(5)
α /°	90	113.860(5)	90
β /°	96.5300(10)	102.299(6)	90
γ /°	90	105.052(6)	90
Volume/Å ³	893.42(4)	886.98(15)	2152.5(2)
<i>Z</i>	4	2	4
ρ_{calc} /mg/mm ³	1.505	1.319	1.364
<i>m</i> /mm ⁻¹	0.378	0.162	0.151
<i>F</i> (000)	416	368	920
Crystal size/mm ³	0.6 × 0.22 × 0.2	0.35 × 0.26 × 0.13	0.4 × 0.25 × 0.13
Radiation	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)
2 Θ range for data collection	2.806 to 55.252°	4.322 to 55.284°	2.09 to 55.062°
Index ranges	-10 ≤ <i>h</i> ≤ 10, -4 ≤ <i>k</i> ≤ 4, -38 ≤ <i>l</i> ≤ 38	-11 ≤ <i>h</i> ≤ 11, -13 ≤ <i>k</i> ≤ 13, -14 ≤ <i>l</i> ≤ 14	-17 ≤ <i>h</i> ≤ 17, -25 ≤ <i>k</i> ≤ 24, -10 ≤ <i>l</i> ≤ 10
Reflections collected	26625	30860	35926
Independent reflections	2069 [<i>R</i> _{int} = 0.0251, <i>R</i> _{sigma} = 0.0101]	4118 [<i>R</i> _{int} = 0.0600, <i>R</i> _{sigma} = 0.0350]	4951 [<i>R</i> _{int} = 0.0849, <i>R</i> _{sigma} = 0.0526]
Data/restraints/parameters	2069/0/135	4118/0/241	4951/5/310
Goodness-of-fit on <i>F</i> ²	1.124	1.014	1.052
Final <i>R</i> indexes [<i>I</i> ≥ 2 σ (<i>I</i>)]	<i>R</i> ₁ = 0.0279, <i>wR</i> ₂ = 0.0820	<i>R</i> ₁ = 0.0371, <i>wR</i> ₂ = 0.0850	<i>R</i> ₁ = 0.0433, <i>wR</i> ₂ = 0.0907
Final <i>R</i> indexes [all data]	<i>R</i> ₁ = 0.0296, <i>wR</i> ₂ = 0.0834	<i>R</i> ₁ = 0.0534, <i>wR</i> ₂ = 0.0933	<i>R</i> ₁ = 0.0625, <i>wR</i> ₂ = 0.0992
Largest diff. peak/hole / e Å ⁻³	0.40/-0.20	0.36/-0.27	0.20/-0.30
Flack Parameter	n/a	n/a	-0.01(7)

Compound	L₃	1
Empirical formula	C ₃₀ H ₂₄ B ₃ N ₆ P	C ₄₈ H ₃₆ B ₂ MoN ₄ O ₄ P ₂
Formula weight	531.95	912.31
Temperature/K	100(2)	100(2)
Crystal system	hexagonal	monoclinic
Space group	<i>P</i> 6 ₃	<i>P</i> 2 ₁ / <i>n</i>
<i>a</i> /Å	14.3773(3)	9.87300(10)
<i>b</i> /Å	14.3773(3)	13.0201(2)
<i>c</i> /Å	7.2074(2)	33.2549(5)
α /°	90	90
β /°	90	90.6290(10)
γ /°	120	90
Volume/Å ³	1290.22(6)	4274.58(10)
<i>Z</i>	2	4
ρ_{calc} /mg/mm ³	1.369	1.418
<i>m</i> /mm ⁻¹	0.141	0.432
<i>F</i> (000)	552	1864
Crystal size/mm ³	0.49 × 0.27 × 0.25	0.42 × 0.36 × 0.28
Radiation	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)
2 Θ range for data collection	3.27 to 55.114°	3.36 to 55.254°
Index ranges	-18 ≤ <i>h</i> ≤ 18, -18 ≤ <i>k</i> ≤ 18, -9 ≤ <i>l</i> ≤ 9	-12 ≤ <i>h</i> ≤ 12, -16 ≤ <i>k</i> ≤ 16, -43 ≤ <i>l</i> ≤ 43
Reflections collected	45115	74957
Independent reflections	1987 [R _{int} = 0.0255, R _{sigma} = 0.0077]	9933 [R _{int} = 0.0365, R _{sigma} = 0.0206]
Data/restraints/parameters	1987/3/128	9933/4/562
Goodness-of-fit on <i>F</i> ²	1.058	1.025
Final <i>R</i> indexes [<i>I</i> ≥ 2 σ (<i>I</i>)]	R ₁ = 0.0270, wR ₂ = 0.0782	R ₁ = 0.0243, wR ₂ = 0.0583
Final <i>R</i> indexes [all data]	R ₁ = 0.0276, wR ₂ = 0.0792	R ₁ = 0.0288, wR ₂ = 0.0605
Largest diff. peak/hole / e Å ⁻³	0.27/-0.17	0.39/-0.39
Flack Parameter	-0.03(13)	n/a

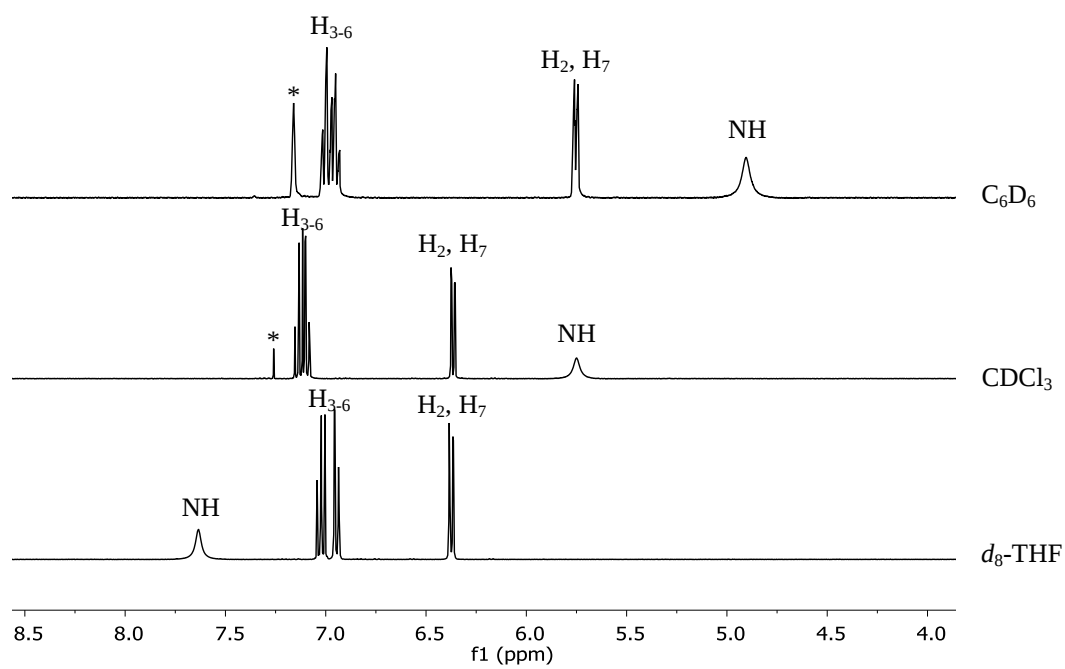
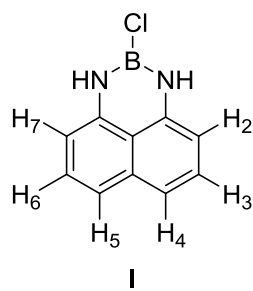


Figure S1. ^1H NMR spectra of **I** in different solvents. Top: C_6D_6 , Middle: CDCl_3 , bottom: $d_8\text{-THF}$. *: residual solvent peak.

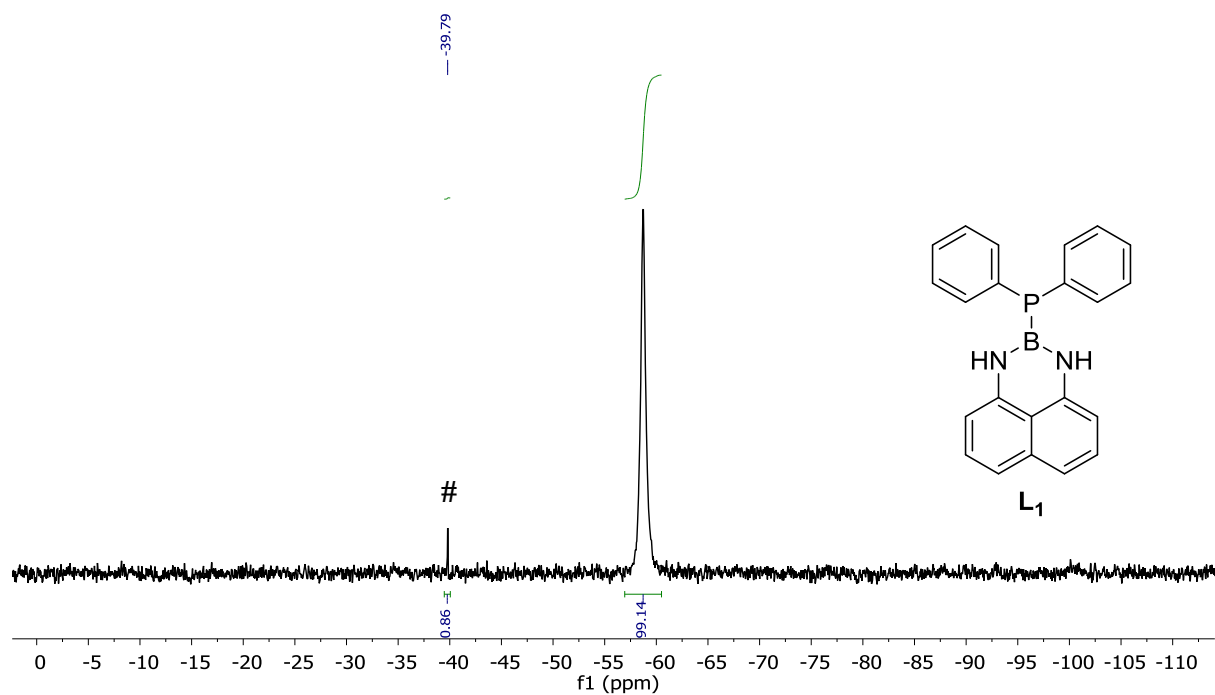


Figure S2. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of **L**₁ in CDCl_3 . Impurity #: Ph_2PH .

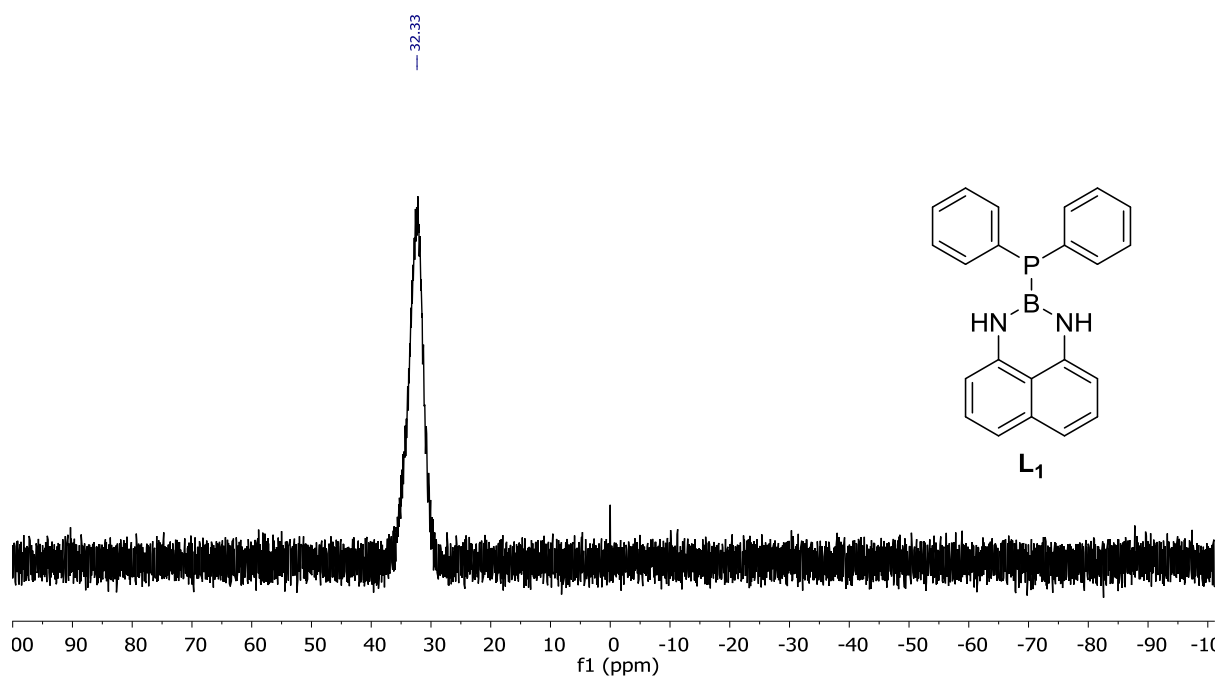


Figure S3. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of **L**₁ in CDCl_3 .

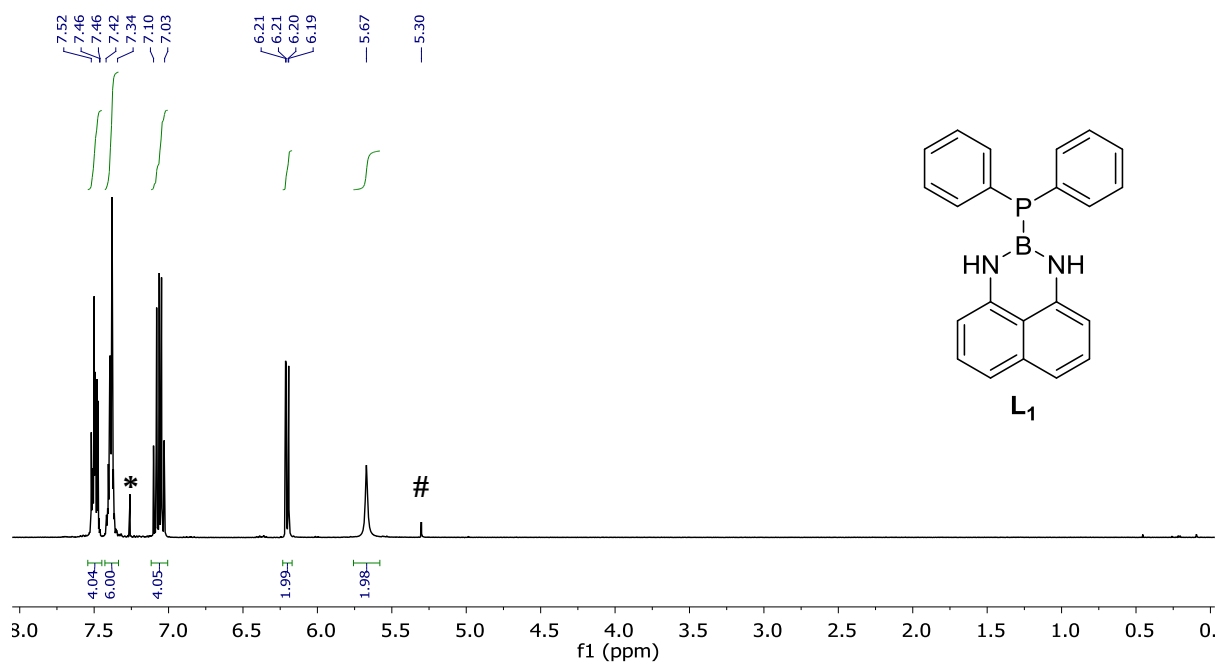


Figure S4. ^1H NMR spectrum of **L₁** in CDCl_3 (*: CHCl_3 ; impurity #: CH_2Cl_2).

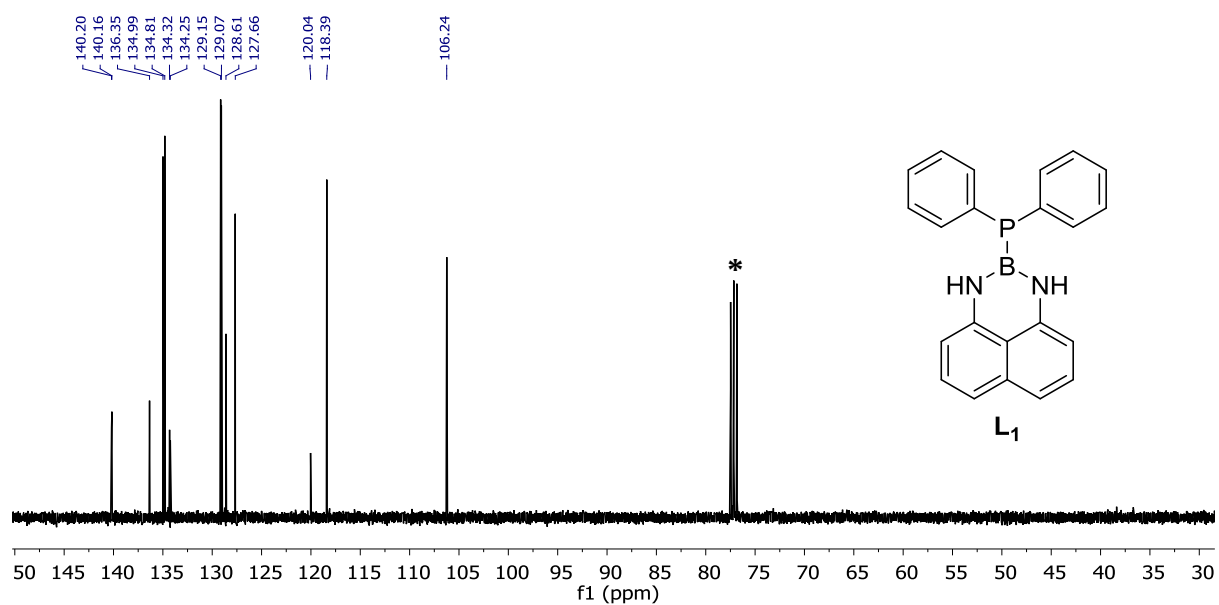


Figure S5. ^{13}C NMR spectrum of **L₁** in CDCl_3 (*: CHCl_3).

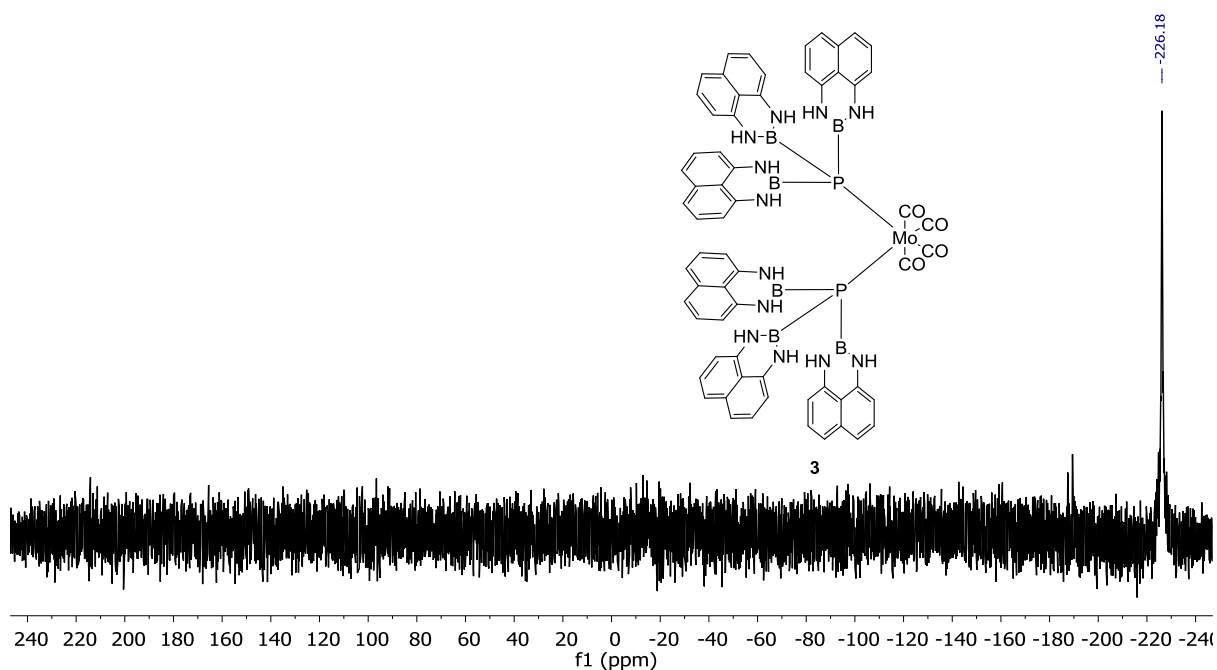


Figure S6. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of **3** in C_6D_6 .

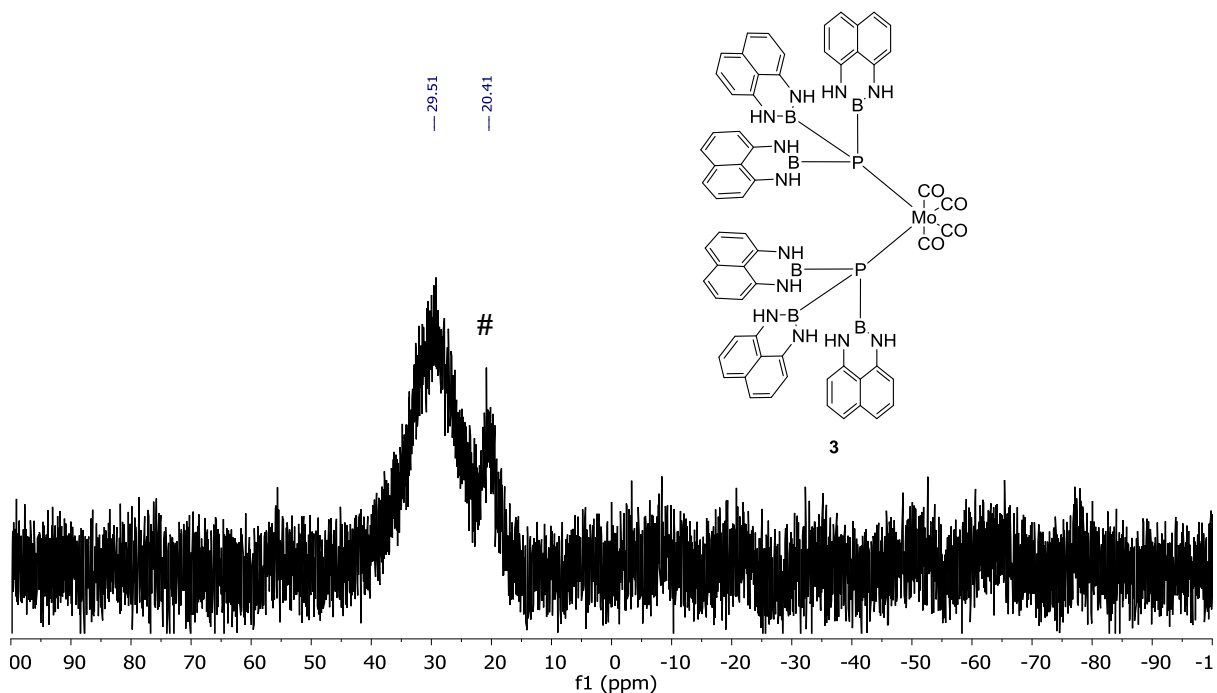


Figure S7. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of **3** in C_6D_6 . Impurity #: tentatively assigned as B-OH hydrolysis product. The undulations in the baseline are due to baseline correction of a weak sample run in a borosilicate tube.

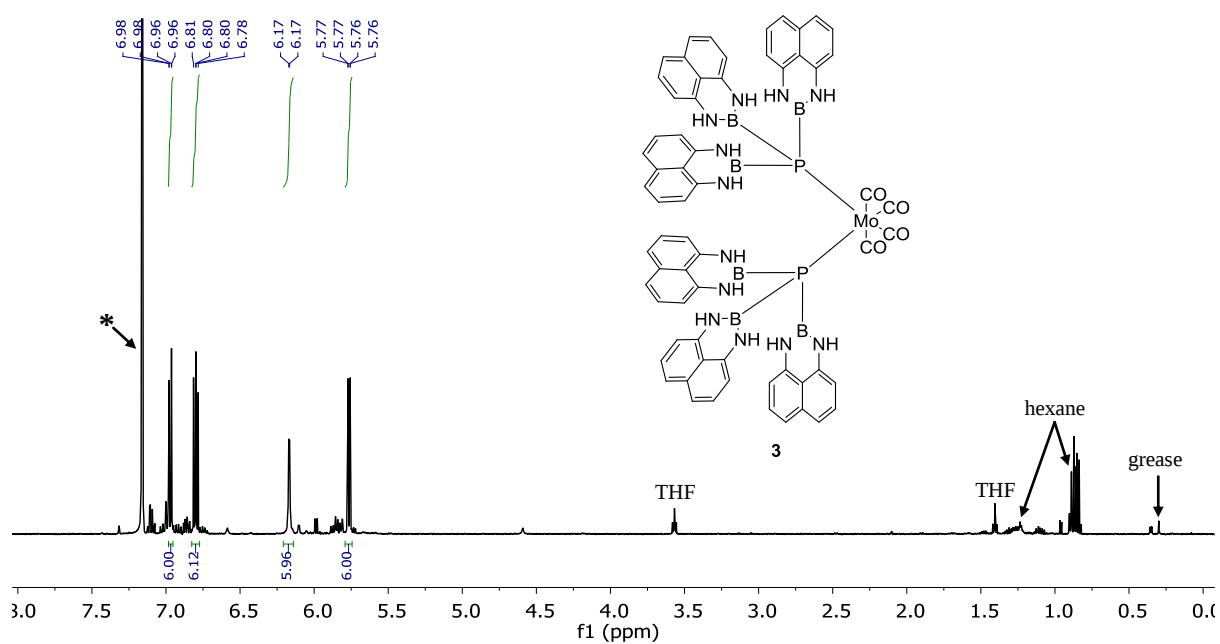


Figure S8. ¹H NMR spectrum of **3** in C₆D₆ (*: C₆D₅H).

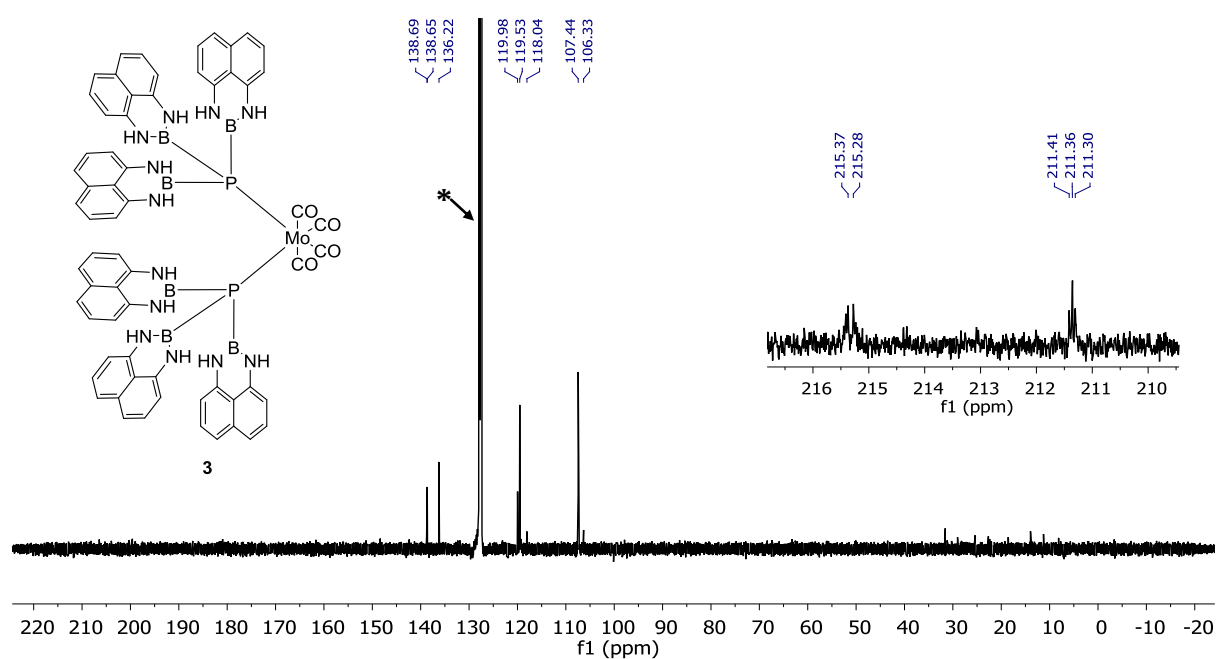


Figure S9. ¹³C NMR spectrum of **3** in C₆D₆ (*: C₆D₅H).