

Supporting Information

Effect of Halogen (Cl, Br) on the Symmetry of Flexible Perovskite-related Framework

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Table S1 Atomic coordinates, equivalent isotropic displacement parameters (\AA^2) and bond valenceanalyses for $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$, $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$, $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$.

$\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$						
Atom	Wyck	x	y	z	U(eq) ^a	BVS ^{b,c}
Na(1)	4a	0.4478(2)	0.2095(1)	0.2429(1)	0.028(1)	1.023
Na(2)	4a	0.9748(2)	0.2590(1)	0.2543(1)	0.030(1)	0.913
Na(3)	4a	0.1312(2)	0.0032(2)	0.4427(1)	0.044(1)	0.819
B(1)	4a	0.2514(4)	0.4990(3)	0.2436(2)	0.012(1)	3.036
B(2)	4a	0.0227(5)	0.7333(3)	0.5195(3)	0.012(1)	3.072
B(3)	4a	0.7522(4)	0.1307(3)	0.5902(3)	0.011(1)	3.074
B(4)	4a	0.1141(4)	0.5023(4)	0.5660(2)	0.012(1)	3.047
B(5)	4a	0.0169(5)	0.2692(3)	0.5237(3)	0.012(1)	3.071
B(6)	4a	0.7600(4)	0.8702(3)	0.5884(3)	0.010(1)	3.065
O(1)	4a	0.1856(2)	0.5001(2)	0.4553(1)	0.009(1)	2.057
O(2)	4a	0.1282(3)	0.2612(2)	0.4395(2)	0.015(1)	2.031
O(3)	4a	0.1360(3)	0.7385(2)	0.4348(2)	0.014(1)	2.027
O(4)	4a	0.2480(3)	0.3782(2)	0.2958(2)	0.013(1)	2.197
O(5)	4a	0.2721(3)	0.6175(2)	0.2974(2)	0.013(1)	2.191
O(6)	4a	0.9973(3)	0.3847(2)	0.5809(2)	0.017(1)	1.976
O(7)	4a	0.2354(2)	0.5017(2)	0.1384(2)	0.019(1)	1.828
O(8)	4a	0.0076(3)	0.6233(2)	0.5811(2)	0.019(1)	1.945
O(9)	4a	0.9307(3)	0.1540(2)	0.5504(2)	0.016(1)	2.010
O(10)	4a	0.9362(3)	0.8501(2)	0.5424(2)	0.015(1)	2.081
Cl(1)	4a	0.1810(1)	0.0163(1)	0.2200(1)	0.030(1)	0.644
$\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$						
Rb(1)	4a	0.4800(1)	0.7522(1)	0.2620(1)	0.038(1)	1.081
Na(1)	4a	0.2708(2)	0.5033(2)	0.0354(2)	0.042(1)	0.769
Na(2)	4a	0.4606(2)	0.7203(1)	0.7576(1)	0.035(1)	0.995
B(1)	4a	0.5484(4)	0.7383(4)	0.5221(2)	0.012(1)	3.085
B(2)	4a	0.4585(4)	0.7655(3)	0.9871(2)	0.012(1)	3.085
B(3)	4a	0.1996(3)	0.0047(4)	0.2270(2)	0.013(1)	3.096

B(4)	4a	0.2085(4)	0.8672(3)	0.0609(2)	0.010(1)	3.084
B(5)	4a	0.1170(3)	0.0007(3)	0.4218(2)	0.010(1)	3.068
B(6)	4a	0.7916(4)	0.6355(3)	0.4402(3)	0.010(1)	3.082
O(1)	4a	0.2812(2)	0.0010(2)	0.0201(1)	0.008(1)	2.024
O(2)	4a	0.2210(2)	0.0044(3)	0.3322(2)	0.023(1)	1.963
O(3)	4a	0.6872(2)	0.7508(2)	0.4664(1)	0.015(1)	2.100
O(4)	4a	0.3221(2)	0.7555(2)	0.0459(2)	0.015(1)	2.111
O(5)	4a	0.5140(3)	0.6229(2)	0.5798(2)	0.016(1)	2.0161
O(6)	4a	0.4494(3)	0.8493(2)	0.5150(2)	0.016(1)	1.999
O(7)	4a	0.1864(2)	0.8824(2)	0.1746(2)	0.014(1)	2.206
O(8)	4a	0.5581(2)	0.6556(2)	0.9920(2)	0.015(1)	1.978
O(9)	4a	0.2040(3)	0.1276(2)	0.1744(2)	0.014(1)	2.213
O(10)	4a	0.4890(3)	0.8784(2)	0.9252(2)	0.020(1)	1.983
Cl(1)	4a	0.7415(1)	0.9872(1)	0.1958(1)	0.044(1)	0.771
RbNa ₂ B ₆ O ₁₀ Br						
Rb(1)	4b	0.5000	0	0	0.043(1)	1.081
Na(1)	4c	0.2048(3)	0.2500	0.2675(5)	0.035(1)	0.829
Na(2)	4a	0.5000	0	0.5000	0.034(1)	0.877
B(1)	4c	0.3288(7)	0.7500	0.3884(1)	0.011(2)	3.037
B(2)	8d	0.2669(5)	0.9875(7)	0.4620(7)	0.014(1)	3.069
B(3)	8d	0.1901(4)	0.8855(6)	0.2162(7)	0.009(1)	3.061
B(4)	4c	0.5242(6)	0.7500	0.3015(1)	0.011(2)	3.167
O(1)	8d	0.2076(3)	0.9981(4)	0.3264(4)	0.014(1)	2.120
O(2)	8d	0.1735(3)	0.1273(4)	0.9934(5)	0.017(1)	1.991
O(3)	4c	0.2297(4)	0.7500	0.2890(6)	0.009(1)	1.995
O(4)	4c	0.4190(4)	0.7500	0.2861(7)	0.021(1)	1.949
O(5)	8d	0.0758(3)	0.8728(4)	0.1977(5)	0.013(1)	2.231
O(6)	8d	0.2575(3)	0.0950(4)	0.5649(4)	0.016(1)	1.966
Br(1)	4c	0.4469(1)	0.2500	0.2622(1)	0.035(1)	0.860

^a U(eq) is defined as the one-third of the trace of the orthogonalized U_{ij} tensor;

^b Bond valences calculated with the program Bond Valence Calculator Version 2.00, C. Hormillosa,

S. Healy, T. Stephen, McMaster University, 1993;

^c Valence sums calculated with the formula: $S_i = \exp[(R_0 - R_i)/B]$, where S_i = valence of bond “i”
and $B = 0.37$.

Table S2Selected bond distances (in Å) and angles (in deg.) for Na₃B₆O₁₀Cl.^a

Na(1)-O(5)#1	2.333(3)	B(1)-O(5)	1.373(4)
Na(1)-O(4)	2.341(3)	B(1)-O(4)	1.374(4)
Na(1)-O(6)#2	2.470(3)	B(2)-O(8)	1.352(4)
Na(1)-O(8)#3	2.676(3)	B(2)-O(10)	1.359(4)
Na(1)-O(9)#2	2.976(4)	B(2)-O(3)	1.375(4)
Na(1)-O(10)#3	2.772(4)	B(3)-O(9)	1.441(4)
Na(1)-Cl(1)	2.777(3)	B(3)-O(4)#12	1.463(4)
Na(2)-O(5)#4	2.405(3)	B(3)-O(2)#12	1.466(4)
Na(2)-O(4)	2.411(3)	B(3)-O(1)#12	1.508(4)
Na(2)-O(3)#4	2.565(4)	B(4)-O(8)	1.453(4)
Na(2)-O(2)	2.632(4)	B(4)-O(7)#10	1.454(4)
Na(2)-Cl(1)#5	2.826(3)	B(4)-O(6)	1.470(4)
Na(2)-Cl(1)	2.892(3)	B(4)-O(1)	1.513(4)
Na(3)-O(10)#8	2.460(3)	B(5)-O(9)	1.356(4)
Na(3)-O(9)	2.524(3)	B(5)-O(2)	1.362(4)
Na(3)-O(2)	2.564(4)	B(5)-O(6)	1.368(4)
Na(3)-O(7)#4	2.917(4)	B(6)-O(10)	1.451(4)
Na(3)-O(6)#2	2.957(4)	B(6)-O(3)#13	1.451(4)
Na(3)-O(3)#8	2.633(4)	B(6)-O(5)#13	1.470(4)
Na(3)-Cl(1)	2.878(4)	B(6)-O(1)#13	1.510(4)
B(1)-O(7)	1.353(4)	O(5)#1-Na(1)-O(4)	155.73(10)
O(5)#1-Na(1)-O(6)#2	85.44(10)	O(4)-Na(1)-Cl(1)	94.16(11)
O(4)-Na(1)-O(6)#2	95.83(10)	O(6)#2-Na(1)-Cl(1)	86.62(8)
O(5)#1-Na(1)-O(8)#3	87.78(9)	O(8)#3-Na(1)-Cl(1)	115.85(8)
O(4)-Na(1)-O(8)#3	81.84(10)	O(10)#3-Na(1)-Cl(1)	88.67(7)
O(6)#2-Na(1)-O(8)#3	157.48(10)	O(5)#1-Na(1)-O(9)#2	114.41(9)
O(5)#1-Na(1)-O(10)#3	55.50(7)	O(4)-Na(1)-O(9)#2	52.35(8)
O(4)-Na(1)-O(10)#3	128.21(9)	O(6)#2-Na(1)-O(9)#2	50.72(9)
O(6)#2-Na(1)-O(10)#3	135.94(9)	O(8)#3-Na(1)-O(9)#2	114.25(11)
O(8)#3-Na(1)-O(10)#3	51.47(9)	O(10)#3-Na(1)-O(9)#2	159.05(9)
O(5)#1-Na(1)-Cl(1)	110.10(10)	Cl(1)-Na(1)-O(9)#2	112.25(7)

O(5)#4-Na(2)-O(4)	172.33(10)	O(2)-Na(3)-O(6)#2	68.26(8)
O(5)#4-Na(2)-O(3)#4	56.55(8)	O(3)#8-Na(3)-O(6)#2	111.11(9)
O(4)-Na(2)-O(3)#4	121.18(10)	Cl(1)-Na(3)-O(6)#2	76.33(5)
O(5)#4-Na(2)-O(2)	125.71(10)	O(7)#4-Na(3)-O(6)#2	145.69(9)
O(4)-Na(2)-O(2)	55.38(8)	O(7)-B(1)-O(5)	119.5(3)
O(3)#4-Na(2)-O(2)	172.01(10)	O(7)-B(1)-O(4)	120.0(3)
O(5)#4-Na(2)-Cl(1)#5	104.36(10)	O(5)-B(1)-O(4)	120.5(3)
O(4)-Na(2)-Cl(1)#5	82.81(9)	O(8)-B(2)-O(10)	121.6(3)
O(3)#4-Na(2)-Cl(1)#5	92.86(7)	O(8)-B(2)-O(3)	122.8(3)
O(2)-Na(2)-Cl(1)#5	93.73(7)	O(10)-B(2)-O(3)	115.4(3)
O(5)#4-Na(2)-Cl(1)	82.84(10)	O(9)-B(3)-O(4)#12	112.5(3)
O(4)-Na(2)-Cl(1)	89.82(10)	O(9)-B(3)-O(2)#12	111.7(3)
O(3)#4-Na(2)-Cl(1)	87.76(7)	O(4)#12-B(3)-O(2)#12	106.8(3)
O(2)-Na(2)-Cl(1)	85.07(7)	O(9)-B(3)-O(1)#12	107.8(3)
Cl(1)#5-Na(2)-Cl(1)	171.75(6)	O(4)#12-B(3)-O(1)#12	109.0(2)
O(5)#4-Na(2)-B(6)#6	28.47(8)	O(2)#12-B(3)-O(1)#12	109.0(2)
O(10)#8-Na(3)-O(9)	74.67(10)	O(9)-B(3)-Na(1)#12	73.91(17)
O(10)#8-Na(3)-O(2)	128.45(10)	O(8)-B(4)-O(7)#10	110.9(2)
O(9)-Na(3)-O(2)	53.83(8)	O(8)-B(4)-O(6)	108.6(2)
O(10)#8-Na(3)-O(3)#8	53.83(8)	O(7)#10-B(4)-O(6)	110.5(2)
O(9)-Na(3)-O(3)#8	128.49(9)	O(8)-B(4)-O(1)	109.2(2)
O(2)-Na(3)-O(3)#8	176.86(8)	O(7)#10-B(4)-O(1)	109.0(2)
O(10)#8-Na(3)-Cl(1)	128.17(8)	O(6)-B(4)-O(1)	108.5(2)
O(9)-Na(3)-Cl(1)	126.19(9)	O(9)-B(5)-O(2)	115.9(3)
O(2)-Na(3)-Cl(1)	86.57(6)	O(9)-B(5)-O(6)	121.5(3)
O(3)#8-Na(3)-Cl(1)	90.29(6)	O(2)-B(5)-O(6)	122.5(3)
O(10)#8-Na(3)-O(7)#4	68.32(10)	O(10)-B(6)-O(3)#13	112.9(3)
O(9)-Na(3)-O(7)#4	69.24(10)	O(10)-B(6)-O(5)#13	111.1(3)
O(2)-Na(3)-O(7)#4	89.50(8)	O(3)#13-B(6)-O(5)#13	107.7(3)
O(3)#8-Na(3)-O(7)#4	89.64(8)	O(10)-B(6)-O(1)#13	107.3(2)
Cl(1)-Na(3)-O(7)#4	76.57(7)	O(3)#13-B(6)-O(1)#13	109.1(3)
O(10)#8-Na(3)-O(6)#2	145.99(9)	O(5)#13-B(6)-O(1)#13	108.7(2)
O(9)-Na(3)-O(6)#2	112.07(10)		

^a Symmetry transformations used to generate equivalent atoms:

(#1) $-x+1, y-1/2, -z+1/2$, (#2) $x+1/2, -y+1/2, -z+1$, (#3) $-x+1/2, -y+1, z-1/2$, (#4) $-x, y-1/2, -z+1/2$, (#5) $-x, y+1/2, -z+1/2$, (#6) $-x-1/2, -y+1, z-1/2$, (#7) $x-1, y, z$, (#8) $x, y-1, z$, (#9) $-x+1, y+1/2, -z+1/2$, (#10) $-x+1/2, -y+1, z+1/2$, (#11) $x, y+1, z$, (#12) $x-1/2, -y+1/2, -z+1$, (#13) $x-1/2, -y+3/2, -z+1$, (#14) $-x-1/2, -y+1, z+1/2$, (#15) $x+1/2, -y+3/2, -z+1$.

Table S3Selected bond lengths (in Å) and angles (in deg.) for RbNa₂B₆O₁₀Cl.^a

Rb(1)-O(7)	2.966(2)	B(1)-O(6)	1.354(4)
Rb(1)-O(9)#1	3.004(2)	B(1)-O(3)	1.362(3)
Rb(1)-O(4)	3.0540(19)	B(1)-O(5)	1.366(4)
Rb(1)-O(6)	3.370(2)	B(2)-O(8)	1.348(4)
Rb(1)-O(2)	3.374(2)	B(2)-O(4)	1.364(4)
Rb(1)-O(3)	3.1252(19)	B(2)-O(10)	1.370(4)
Rb(1)-O(8)	3.627(2)	B(3)-O(2)	1.354(4)
Rb(1)-Cl(1)#1	3.1996(11)	B(3)-O(7)	1.361(4)
Rb(1)-Cl(1)	3.2560(11)	B(3)-O(9)	1.364(4)
Na(1)-O(3)#1	2.462(3)	B(4)-O(8)#4	1.439(4)
Na(1)-O(4)	2.476(3)	B(4)-O(4)	1.447(4)
Na(1)-O(10)#4	2.657(3)	B(4)-O(7)	1.469(4)
Na(1)-O(5)#5	2.725(3)	B(4)-O(1)	1.518(4)
Na(1)-O(6)#1	2.837(3)	B(5)-O(2)	1.434(3)
Na(1)-O(8)	2.862(3)	B(5)-O(10)#10	1.464(4)
Na(1)-Cl(1)#4	2.960(2)	B(5)-O(5)#11	1.470(3)
Na(2)-O(9)#7	2.273(2)	B(5)-O(1)#10	1.513(3)
Na(2)-O(7)#2	2.294(2)	B(6)-O(6)#12	1.440(4)
Na(2)-O(5)#8	2.496(2)	B(6)-O(3)	1.451(4)
Na(2)-O(10)	2.637(2)	B(6)-O(9)#1	1.464(4)
Na(2)-Cl(1)#4	2.7729(19)	B(6)-O(1)#1	1.521(4)
O(7)-Rb(1)-O(9)#1	173.12(6)	O(7)-Rb(1)-Cl(1)	99.10(4)
O(7)-Rb(1)-O(4)	45.70(5)	O(9)#1-Rb(1)-Cl(1)	76.34(4)
O(9)#1-Rb(1)-O(4)	128.66(6)	O(4)-Rb(1)-Cl(1)	92.65(4)
O(7)-Rb(1)-O(3)	140.36(5)	O(3)-Rb(1)-Cl(1)	81.40(4)
O(9)#1-Rb(1)-O(3)	44.76(5)	Cl(1)#1-Rb(1)-Cl(1)	170.50(3)
O(4)-Rb(1)-O(3)	171.99(5)	O(7)-Rb(1)-O(6)	100.36(5)
O(7)-Rb(1)-Cl(1)#1	85.81(4)	O(9)#1-Rb(1)-O(6)	85.36(5)
O(9)#1-Rb(1)-Cl(1)#1	97.99(4)	O(4)-Rb(1)-O(6)	145.98(5)
O(4)-Rb(1)-Cl(1)#1	84.95(4)	O(3)-Rb(1)-O(6)	40.92(5)
O(3)-Rb(1)-Cl(1)#1	100.05(4)	Cl(1)#1-Rb(1)-O(6)	91.01(4)

Cl(1)-Rb(1)-O(6)	96.05(4)	O(6)#1-Na(1)-Cl(1)#4	81.84(7)
O(7)-Rb(1)-O(2)	42.88(5)	O(8)-Na(1)-Cl(1)#4	81.96(7)
O(9)#1-Rb(1)-O(2)	141.00(5)	O(9)#7-Na(2)-O(7)#2	161.70(11)
O(4)-Rb(1)-O(2)	87.54(5)	O(9)#7-Na(2)-O(5)#8	85.76(8)
O(3)-Rb(1)-O(2)	97.73(5)	O(7)#2-Na(2)-O(5)#8	91.93(8)
Cl(1)#1-Rb(1)-O(2)	99.44(4)	O(9)#7-Na(2)-O(10)	93.35(8)
Cl(1)-Rb(1)-O(2)	89.62(5)	O(7)#2-Na(2)-O(10)	82.61(8)
O(6)-Rb(1)-O(2)	59.79(5)	O(5)#8-Na(2)-O(10)	159.83(10)
O(7)-Rb(1)-O(8)	84.24(5)	O(9)#7-Na(2)-Cl(1)#4	99.83(8)
O(9)#1-Rb(1)-O(8)	89.80(5)	O(7)#2-Na(2)-Cl(1)#4	98.40(7)
O(4)-Rb(1)-O(8)	38.98(5)	O(5)#8-Na(2)-Cl(1)#4	92.29(7)
O(3)-Rb(1)-O(8)	133.78(5)	O(10)-Na(2)-Cl(1)#4	107.68(7)
Cl(1)#1-Rb(1)-O(8)	93.29(4)	O(6)-B(1)-O(3)	114.3(3)
Cl(1)-Rb(1)-O(8)	79.21(4)	O(6)-B(1)-O(5)	123.6(3)
O(6)-Rb(1)-O(8)	173.93(6)	O(3)-B(1)-O(5)	122.1(3)
O(2)-Rb(1)-O(8)	123.52(5)	O(8)-B(2)-O(4)	115.5(3)
O(3)#1-Na(1)-O(4)	161.70(9)	O(8)-B(2)-O(10)	122.6(3)
O(3)#1-Na(1)-O(10)#4	123.54(9)	O(4)-B(2)-O(10)	121.8(3)
O(4)-Na(1)-O(10)#4	73.71(7)	O(2)-B(3)-O(7)	119.7(3)
O(3)#1-Na(1)-O(5)#5	71.52(7)	O(2)-B(3)-O(9)	119.1(3)
O(4)-Na(1)-O(5)#5	125.36(9)	O(7)-B(3)-O(9)	121.0(3)
O(10)#4-Na(1)-O(5)#5	52.02(6)	O(8)#4-B(4)-O(4)	113.1(2)
O(3)#1-Na(1)-O(6)#1	50.40(7)	O(8)#4-B(4)-O(7)	111.6(2)
O(4)-Na(1)-O(6)#1	112.70(8)	O(4)-B(4)-O(7)	106.7(2)
O(10)#4-Na(1)-O(6)#1	173.19(9)	O(8)#4-B(4)-O(1)	108.4(2)
O(5)#5-Na(1)-O(6)#1	121.81(9)	O(4)-B(4)-O(1)	109.1(2)
O(3)#1-Na(1)-O(8)	112.83(8)	O(7)-B(4)-O(1)	107.7(2)
O(4)-Na(1)-O(8)	50.27(6)	O(2)-B(5)-O(10)#10	111.6(2)
O(10)#4-Na(1)-O(8)	123.63(9)	O(2)-B(5)-O(5)#11	111.2(2)
O(5)#5-Na(1)-O(8)	175.63(9)	O(10)#10-B(5)-O(5)#11	107.2(2)
O(6)#1-Na(1)-O(8)	62.56(6)	O(2)-B(5)-O(1)#10	108.8(2)
O(3)#1-Na(1)-Cl(1)#4	91.91(8)	O(10)#10-B(5)-O(1)#10	108.8(2)
O(4)-Na(1)-Cl(1)#4	92.17(8)	O(5)#11-B(5)-O(1)#10	109.2(2)
O(10)#4-Na(1)-Cl(1)#4	95.89(7)	O(6)#12-B(6)-O(3)	112.2(2)
O(5)#5-Na(1)-Cl(1)#4	98.58(7)	O(6)#12-B(6)-O(9)#1	112.3(2)

O(3)-B(6)-O(9)#1	106.6(2)	O(3)-B(6)-O(1)#1	109.9(2)
O(6)#12-B(6)-O(1)#1	108.6(2)	O(9)#1-B(6)-O(1)#1	107.3(2)

^a Symmetry transformations used to generate equivalent atoms:

(#1) $-x+1, y-1/2, -z+1/2$, (#2) $x+1/2, -y+3/2, -z$, (#3) $-x+1, y+1/2, -z+1/2$, (#4) $x-1/2, -y+3/2, -z$, (#5) $-x+1/2, -y+1, z-1/2$, (#6) $-x+1/2, -y+1, z+1/2$, (#7) $-x+1/2, -y+2, z-1/2$, (#8) $x, y, z-1$, (#9) $x, y, z+1$, (#10) $-x+1/2, -y+2, z+1/2$, (#11) $x-1/2, -y+3/2, -z+1$, (#12) $x+1/2, -y+3/2, -z+1$.

Table S4Selected bond distances (in Å) and angles (in deg.) for RbNa₂B₆O₁₀Br ^a.

Rb(1)-O(5)#1	2.977(4)	Na(2)-O(2)#6	2.531(4)
Rb(1)-O(5)#2	2.977(4)	Na(2)-O(2)#2	2.531(4)
Rb(1)-O(1)#1	3.025(4)	Na(2)-Br(1)	3.1960(15)
Rb(1)-O(1)#2	3.025(4)	Na(2)-Br(1)#8	3.1960(15)
Rb(1)-Br(1)	3.3275(16)	B(1)-O(4)	1.436(10)
Rb(1)-Br(1)#3	3.3275(16)	B(1)-O(2)#6	1.472(7)
Rb(1)-O(6)#1	3.456(4)	B(1)-O(2)#10	1.472(7)
Rb(1)-O(6)#2	3.456(4)	B(1)-O(3)	1.515(10)
Na(1)-O(1)#4	2.471(4)	B(2)-O(6)	1.352(8)
Na(1)-O(1)	2.471(4)	B(2)-O(2)#6	1.366(7)
Na(1)-O(2)	2.615(6)	B(2)-O(1)	1.371(7)
Na(1)-O(2)#4	2.615(6)	B(3)-O(1)	1.441(7)
Na(1)-O(6)#4	2.983(6)	B(3)-O(6)#1	1.447(7)
Na(1)-O(6)	2.983(6)	B(3)-O(5)	1.472(7)
Na(1)-Br(1)	3.091(4)	B(3)-O(3)	1.525(6)
Na(1)-Br(1)#5	3.303(5)	B(4)-O(4)	1.349(10)
Na(2)-O(5)#6	2.277(4)	B(4)-O(5)#2	1.353(5)
Na(2)-O(5)#2	2.277(4)	B(4)-O(5)#11	1.353(5)
O(5)#1-Rb(1)-O(5)#2	180.00(13)	O(1)#2-Rb(1)-Br(1)#3	97.77(8)
O(5)#1-Rb(1)-O(1)#1	45.75(10)	Br(1)-Rb(1)-Br(1)#3	180
O(5)#2-Rb(1)-O(1)#1	134.25(10)	O(5)#1-Rb(1)-O(6)#1	86.17(10)
O(5)#1-Rb(1)-O(1)#2	134.25(10)	O(5)#2-Rb(1)-O(6)#1	93.83(10)
O(5)#2-Rb(1)-O(1)#2	45.75(10)	O(1)#1-Rb(1)-O(6)#1	40.86(9)
O(1)#1-Rb(1)-O(1)#2	180	O(1)#2-Rb(1)-O(6)#1	139.14(9)
O(5)#1-Rb(1)-Br(1)	101.55(8)	Br(1)-Rb(1)-O(6)#1	83.83(7)
O(5)#2-Rb(1)-Br(1)	78.45(8)	Br(1)#3-Rb(1)-O(6)#1	96.17(7)
O(1)#1-Rb(1)-Br(1)	97.77(8)	B(2)#1-Rb(1)-O(6)#1	22.65(13)
O(1)#2-Rb(1)-Br(1)	82.23(8)	B(2)#2-Rb(1)-O(6)#1	157.35(13)
O(5)#1-Rb(1)-Br(1)#3	78.45(8)	O(5)#1-Rb(1)-O(6)#2	93.83(10)
O(5)#2-Rb(1)-Br(1)#3	101.55(8)	O(5)#2-Rb(1)-O(6)#2	86.17(10)
O(1)#1-Rb(1)-Br(1)#3	82.23(8)	O(1)#1-Rb(1)-O(6)#2	139.14(9)

O(1)#2-Rb(1)-O(6)#2	40.86(9)	O(5)#6-Na(2)-O(5)#2	180
Br(1)-Rb(1)-O(6)#2	96.17(7)	O(5)#6-Na(2)-O(2)#6	84.45(13)
Br(1)#3-Rb(1)-O(6)#2	83.83(7)	O(5)#2-Na(2)-O(2)#6	95.55(13)
O(6)#1-Rb(1)-O(6)#2	180	O(5)#6-Na(2)-O(2)#2	95.55(13)
O(1)#4-Na(1)-O(1)	156.9(3)	O(5)#2-Na(2)-O(2)#2	84.45(13)
O(1)#4-Na(1)-O(2)	128.29(19)	O(2)#6-Na(2)-O(2)#2	180
O(1)-Na(1)-O(2)	74.71(14)	O(5)#6-Na(2)-Br(1)	87.76(11)
O(1)#4-Na(1)-O(2)#4	74.71(14)	O(5)#2-Na(2)-Br(1)	92.24(11)
O(1)-Na(1)-O(2)#4	128.29(19)	O(2)#6-Na(2)-Br(1)	99.43(9)
O(2)-Na(1)-O(2)#4	53.60(19)	O(2)#2-Na(2)-Br(1)	80.57(9)
O(1)#4-Na(1)-O(6)#4	48.73(13)	O(5)#6-Na(2)-Br(1)#8	92.24(11)
O(1)-Na(1)-O(6)#4	108.59(18)	O(5)#2-Na(2)-Br(1)#8	87.76(11)
O(2)-Na(1)-O(6)#4	174.42(18)	O(2)#6-Na(2)-Br(1)#8	80.57(9)
O(2)#4-Na(1)-O(6)#4	123.02(12)	O(2)#2-Na(2)-Br(1)#8	99.43(9)
O(1)#4-Na(1)-O(6)	108.59(18)	Br(1)-Na(2)-Br(1)#8	180
O(1)-Na(1)-O(6)	48.73(13)	O(4)-B(1)-O(2)#6	111.9(4)
O(2)-Na(1)-O(6)	123.02(12)	O(4)-B(1)-O(2)#10	111.9(4)
O(2)#4-Na(1)-O(6)	174.42(18)	O(2)#6-B(1)-O(2)#10	106.5(6)
O(6)#4-Na(1)-O(6)	59.92(18)	O(4)-B(1)-O(3)	109.9(6)
O(1)#4-Na(1)-Br(1)	89.34(13)	O(2)#6-B(1)-O(3)	108.2(4)
O(1)-Na(1)-Br(1)	89.34(13)	O(2)#10-B(1)-O(3)	108.2(4)
O(2)-Na(1)-Br(1)	98.07(14)	O(6)-B(2)-O(2)#6	122.9(5)
O(2)#4-Na(1)-Br(1)	98.07(14)	O(6)-B(2)-O(1)	115.1(5)
O(6)#4-Na(1)-Br(1)	77.67(12)	O(2)#6-B(2)-O(1)	121.8(5)
O(6)-Na(1)-Br(1)	77.67(12)	O(1)-B(3)-O(6)#1	113.2(5)
O(1)#4-Na(1)-Br(1)#5	91.69(13)	O(1)-B(3)-O(5)	106.5(4)
O(1)-Na(1)-Br(1)#5	91.69(13)	O(6)#1-B(3)-O(5)	112.1(4)
O(2)-Na(1)-Br(1)#5	77.35(13)	O(1)-B(3)-O(3)	109.5(4)
O(2)#4-Na(1)-Br(1)#5	77.35(13)	O(6)#1-B(3)-O(3)	108.0(4)
O(6)#4-Na(1)-Br(1)#5	106.75(13)	O(5)-B(3)-O(3)	107.5(4)
O(6)-Na(1)-Br(1)#5	106.75(13)	O(4)-B(4)-O(5)#2	119.1(3)
Br(1)-Na(1)-Br(1)#5	174.85(16)	O(4)-B(4)-O(5)#11	119.1(3)
B(2)#4-Na(1)-Br(1)#5	107.10(14)	O(5)#2-B(4)-O(5)#11	121.6(7)

^a Symmetry transformations used to generate equivalent atoms:

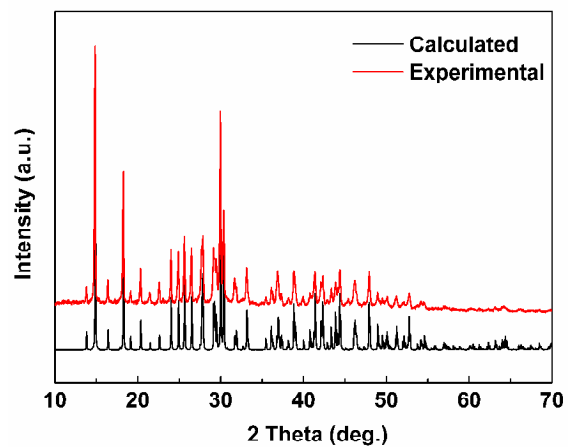
(#1) $-x+1/2, -y+2, z-1/2$, (#2) $x+1/2, y, -z+1/2$, (#3) $-x+1, -y+2, -z$, (#4) $x, -y+5/2, z$, (#5) $x-1/2, y, -z+1/2$, (#6) $-x+1/2, -y+2, z+1/2$, (#7) $x-1/2, -y+5/2, -z+1/2$, (#8) $-x+1, -y+2, -z+1$, (#9) $x, y, z+1$, (#10) $-x+1/2, y-1/2, z+1/2$, (#11) $x+1/2, -y+3/2, -z+1/2$, (#12) $-x+1, y-1/2, -z+1$, (#13) $-x+1, y-1/2, -z$, (#14) $x, -y+3/2, z$, (#15) $-x+1, y+1/2, -z+1$, (#16) $-x+1, y+1/2, -z$.

Table S5 The element analysis of $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$, $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$ and $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$.

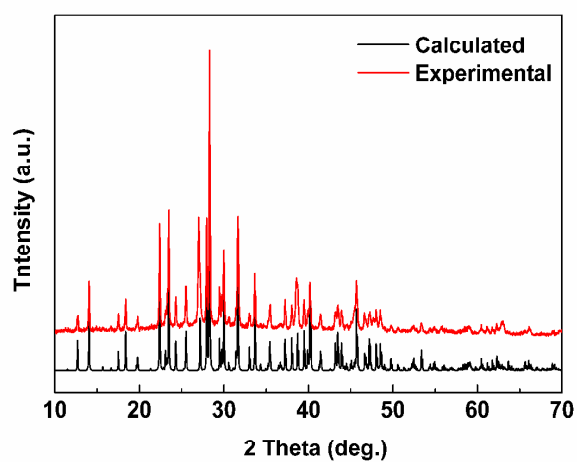
Element	$\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$		$\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$		$\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$	
	Cal.	Exp.	Cal.	Exp.	Cal.	Exp.
Na	20.95	21.13	11.74	11.82	10.54	10.71
B	19.70	19.65	16.56	16.62	14.87	14.93

Table S6 The assignment of infrared spectra for Na₃B₆O₁₀Cl and RbNa₂B₆O₁₀X (X = Cl, Br).

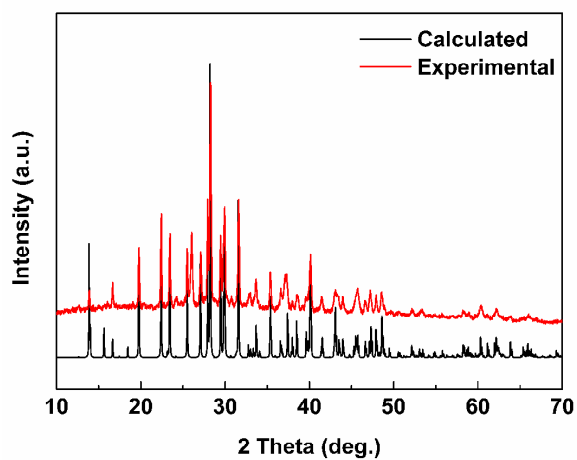
Na ₃ B ₆ O ₁₀ Cl (cm ⁻¹)	RbNa ₂ B ₆ O ₁₀ Cl (cm ⁻¹)	RbNa ₂ B ₆ O ₁₀ Br (cm ⁻¹)	Assignment
1341, 1187	1347, 1201	1346, 1194	asymmetric stretching vibrations of BO ₃
1083, 1017	1081, 1012	1075, 1024	asymmetric stretching vibrations of BO ₄
961, 908	963, 945	962, 945	symmetric stretching vibrations of BO ₃
860, 800	858, 826, 798	856, 829, 797	symmetric stretching vibrations of BO ₄
744, 691, 628	733, 693, 641	727, 686, 641	out-of-bending of BO ₃
587, 563, 519	572, 523	573, 537, 515	bending of BO ₃ and BO ₄



(a) $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$



(b) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$



(c) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$

Figure S1. Powder XRD patterns of calculated and experimental for crystals: (a) $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$, (b)

$\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$, (c) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$.

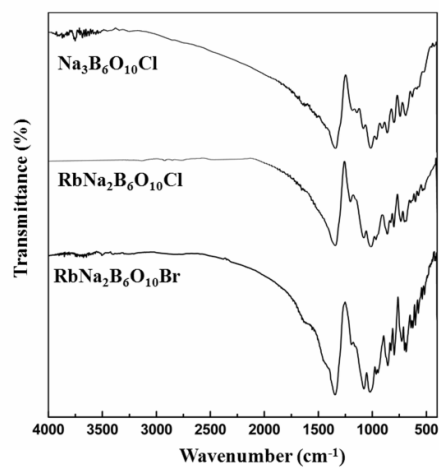


Figure S2. The infrared spectra of $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$, $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$ and $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$.

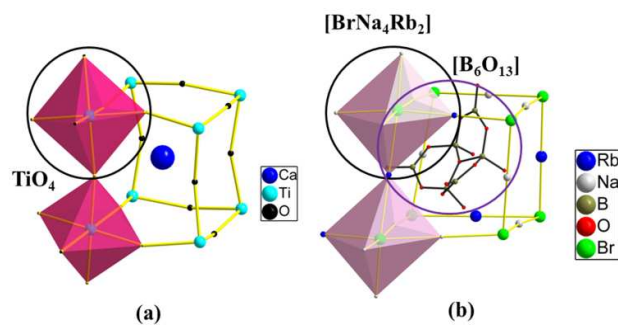


Figure S3 (a) The perovskite structure of CaTiO_3 . (b) The perovskite-related structure of $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$.

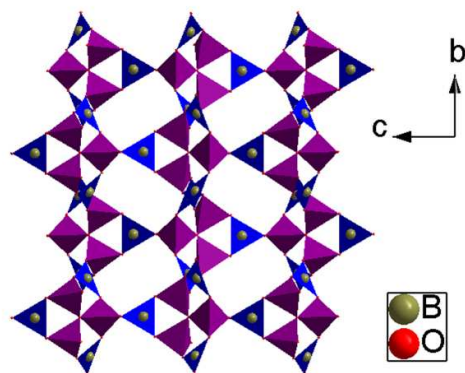


Figure S4.1 The boron-oxygen framework in the crystal structure of $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$ in bc plane

(BO_3 triangles and BO_4 tetrahedra are shown in blue and violet, respectively).

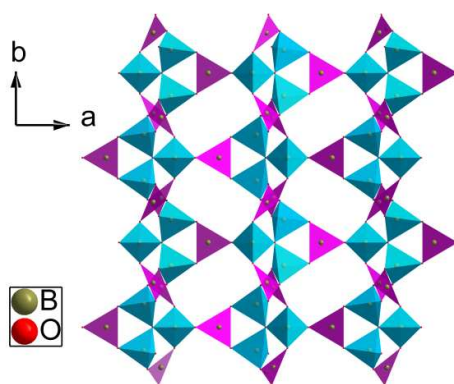
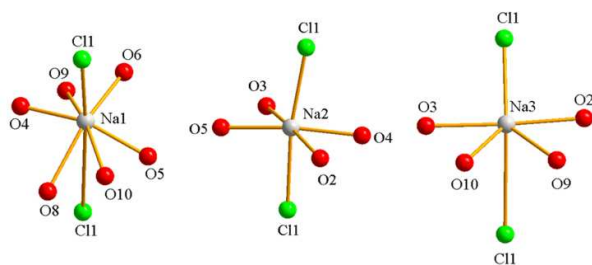
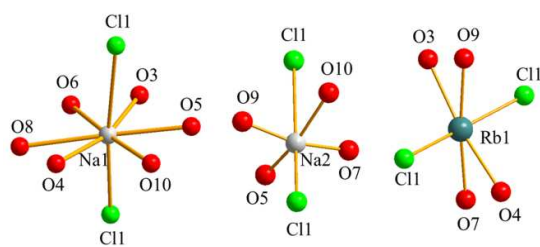


Figure S4.2 The boron-oxygen framework in the crystal structure of $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$ in ab plane

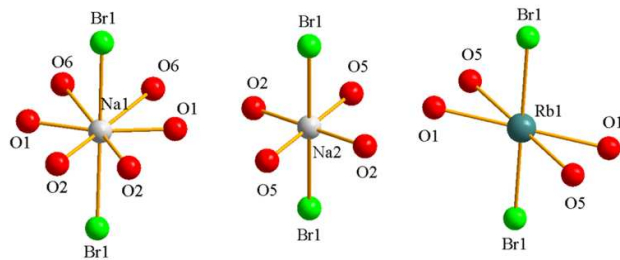
(BO_3 triangles and BO_4 tetrahedra are shown in pink and sky blue, respectively).



(a) $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$



(b) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$



(c) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$

Figure S5. Coordination environment of the cation: (a) $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$, (b) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$, (c)

$\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$

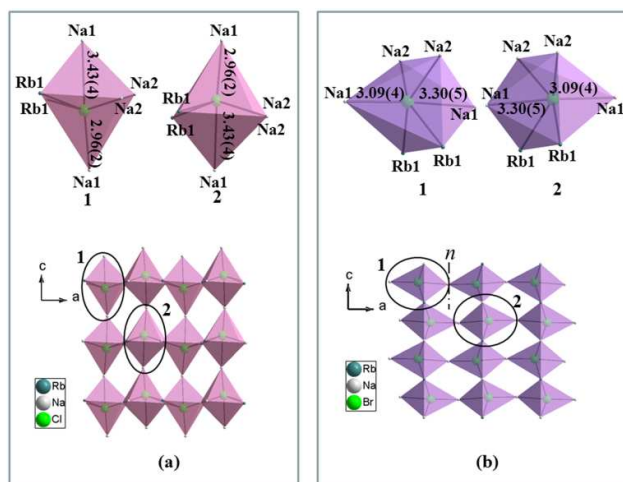


Figure S6.1. The n -glide plane comparison of (a) ClNa_4Rb_2 and (b) BrNa_4Rb_2 networks in $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$ and $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$. (The ClNa_4Rb_2 octahedra are shown in rose and the BrNa_4Rb_2 octahedra are shown in lavender).

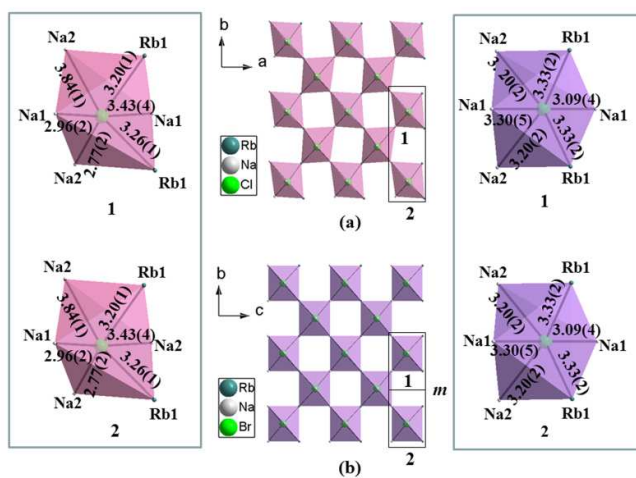


Figure S6.2. The m -mirror plane comparison of (a) ClNa_4Rb_2 and (b) BrNa_4Rb_2 networks in $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$ and $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$. (The ClNa_4Rb_2 octahedra are shown in rose and the BrNa_4Rb_2 octahedra are shown in lavender).

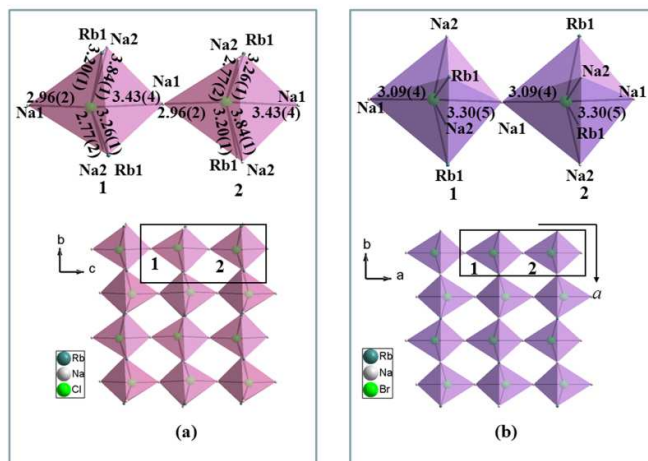


Figure S6.3. The a -glide plane comparison of (a) ClNa_4Rb_2 and (b) BrNa_4Rb_2 networks in $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$ and $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$. (The ClNa_4Rb_2 octahedra are shown in rose and the BrNa_4Rb_2 octahedra are shown in lavender).

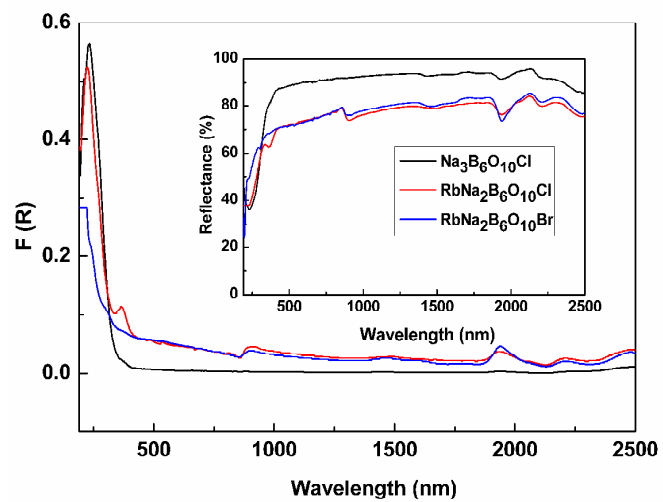
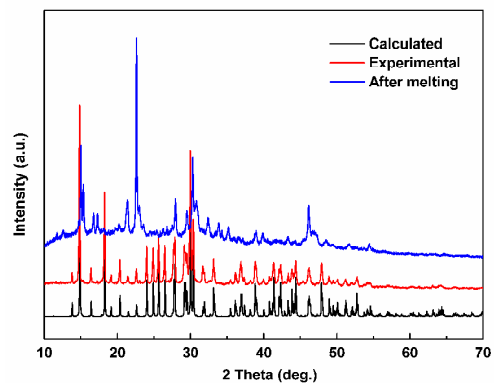
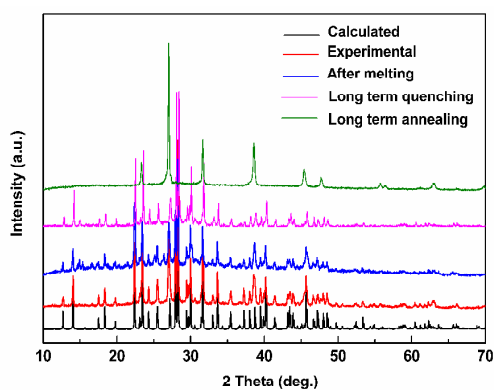


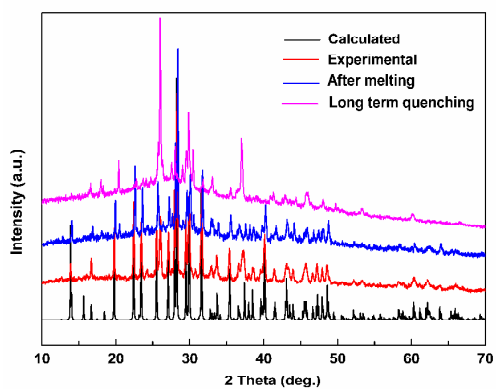
Figure S7. The UV–Vis–NIR diffuse-reflectance spectra of $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$, $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$ and $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$.



(a) $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$



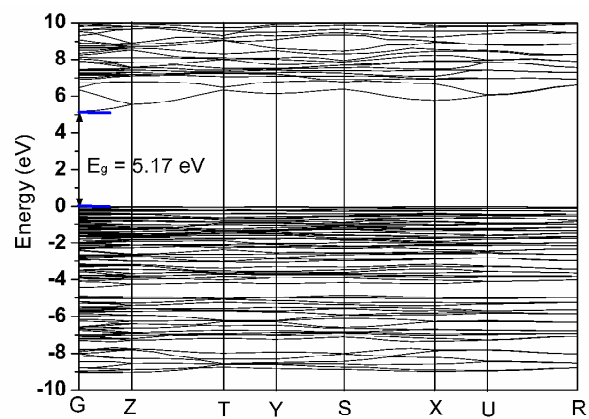
(b) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$



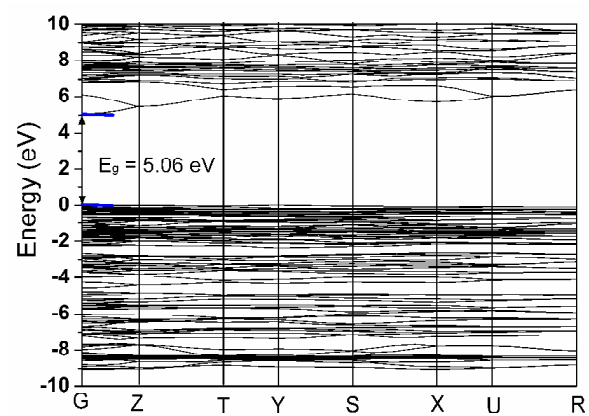
(c) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$

Figure S8. The powder XRD patterns of (a) $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$, (b) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$, (c) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$

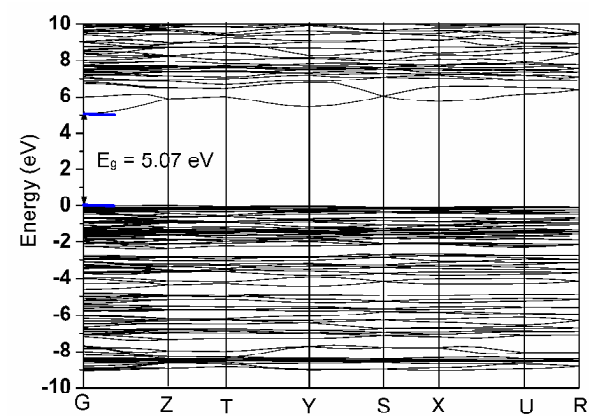
compounds before and after melting, long term quenching and annealing.



(a) $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$

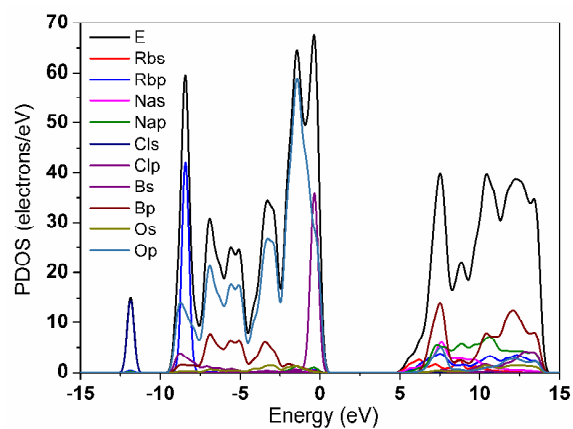


(b) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$

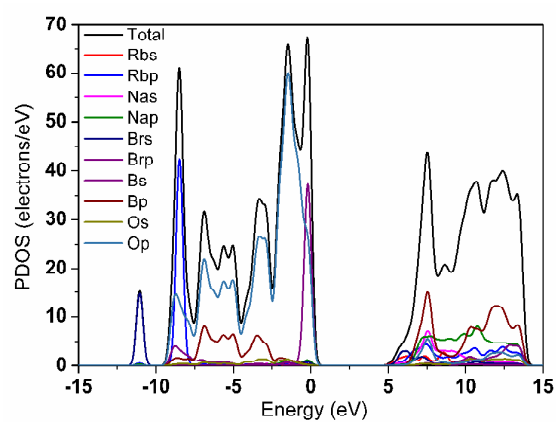


(c) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$

Figure S9. Calculated band structures of (a) $\text{Na}_3\text{B}_6\text{O}_{10}\text{Cl}$, (b) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$, (c) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$.



(a) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$



(c) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$

Figure S10. The full density of states (DOS) and partial density of states (PDOS): (a) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Cl}$, (b) $\text{RbNa}_2\text{B}_6\text{O}_{10}\text{Br}$.