

# **Structure sensitivity of $\text{La}_2\text{O}_2\text{CO}_3$ catalysts in the oxidative coupling of methane**

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**Table S1** The calculated density of sites for O<sub>2</sub> and CO<sub>2</sub> adsorption over different samples based on TPD data.

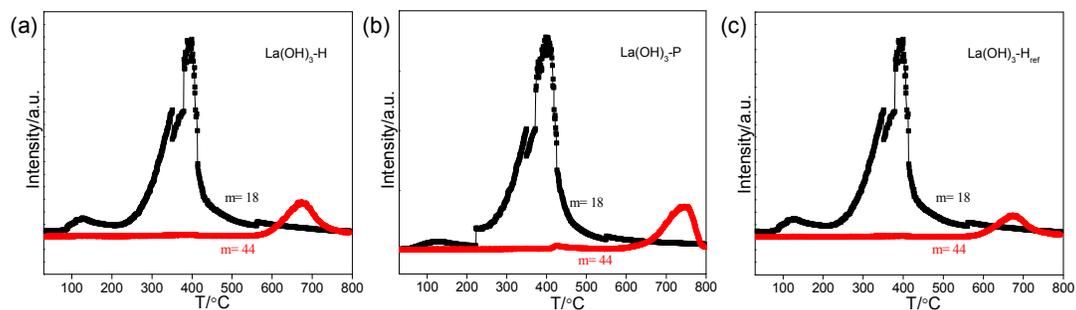
Samples	$\rho_{\text{O}_2}/\mu\text{mol}\cdot\text{m}^{-2}$		$\rho_{\text{CO}_2}/\mu\text{mol}\cdot\text{m}^{-2}$	
	T=200-400 °C	T=400-600 °C	T=400-600 °C	T=600-800 °C
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -H	0.52	0.86	1.24	4.07
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -P	0.03	-	0.24	9.02
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -H <sub>ref</sub>	-	-	-	3.49
La <sub>2</sub> O <sub>3</sub> -H	-	0.11	-	7.41
La <sub>2</sub> O <sub>3</sub> -P	-	-	-	11.90
La <sub>2</sub> O <sub>3</sub> -H <sub>ref</sub>	-	-	-	7.00

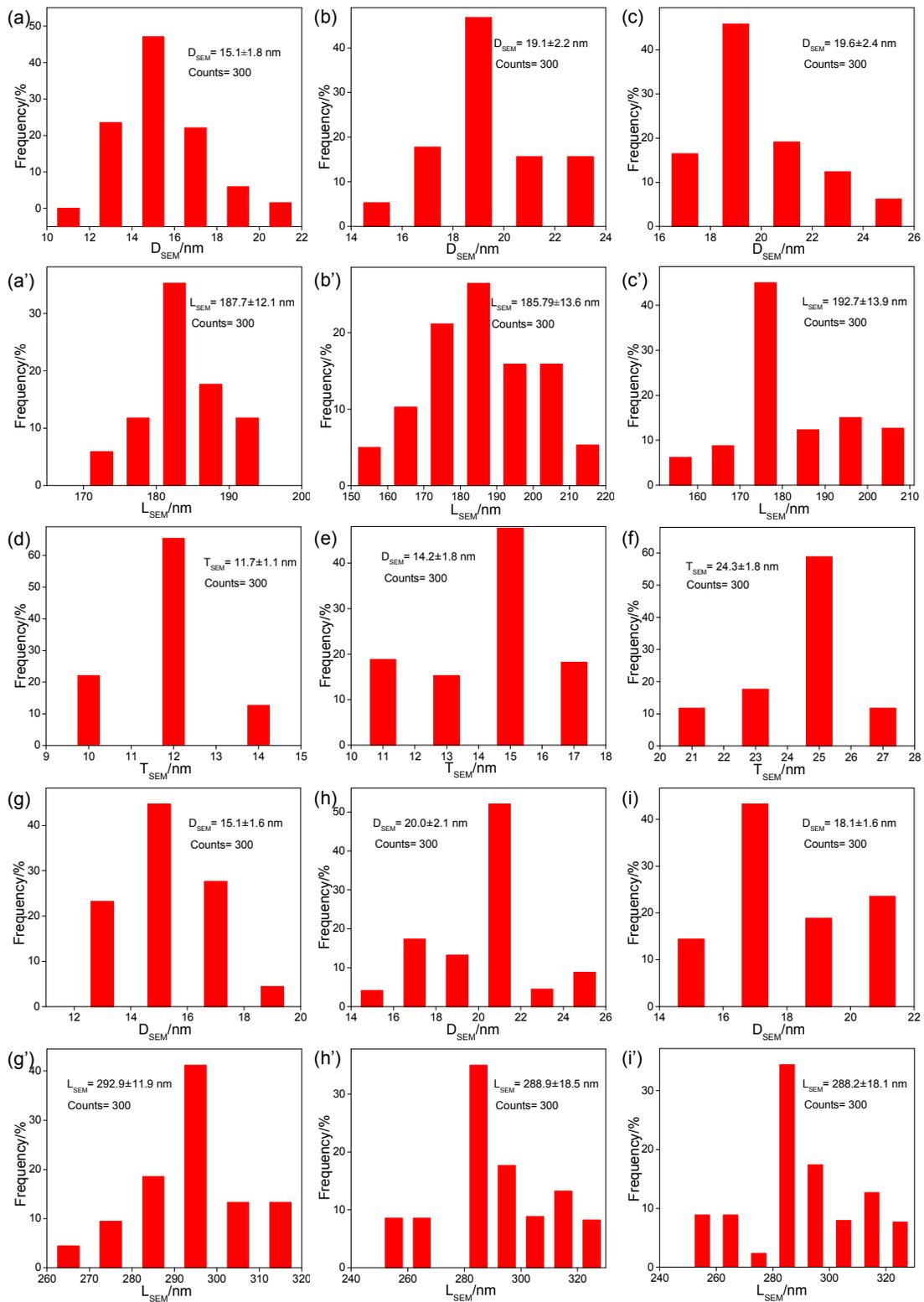
**Table S2** Areas of the oxygen species peaks based on O 1s binding energy spectra of the La<sub>2</sub>O<sub>2</sub>CO<sub>3</sub> samples and the ratio of (O<sup>-</sup>+O<sub>2</sub><sup>2-</sup>)/O<sup>2-</sup>.

Samples	Binding energy/eV				(O <sup>-</sup> +O <sub>2</sub> <sup>2-</sup> )/O <sup>2-</sup>
	528.2 (O <sup>2-</sup> )	530.1 (O <sup>-</sup> )	531.1 (O <sub>2</sub> <sup>2-</sup> )	532.2 (CO <sub>3</sub> <sup>2-</sup> )	
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -H	10156.3	15260.8	50590.7	31589.6	6.5
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -P	9983.7	5844.7	30738.9	4460.4	3.7
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -H <sub>ref</sub>	14290.0	17651.8	35834.2	3153.6	3.7

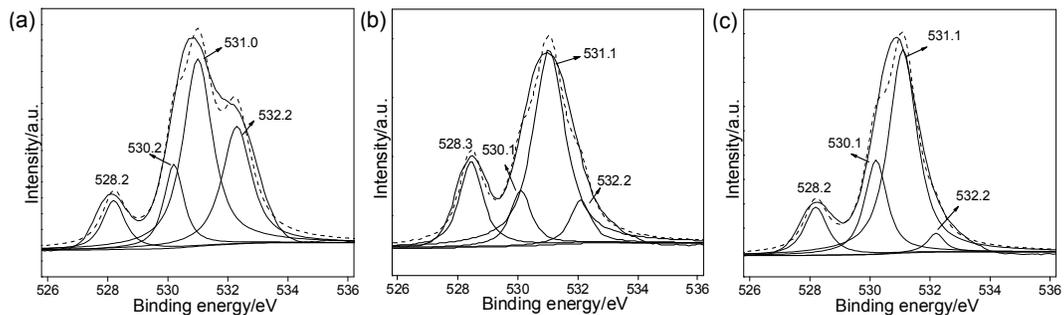
**Table S3** The calculated surface atomic composition of the La<sub>2</sub>O<sub>2</sub>CO<sub>3</sub> samples based on XPS analysis.

Samples	Atomic composition/%		
	O	La	C
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -H	55.5	18.2	26.3
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -P	57.5	18.6	23.9
La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> -H <sub>ref</sub>	61.0	16.5	22.5

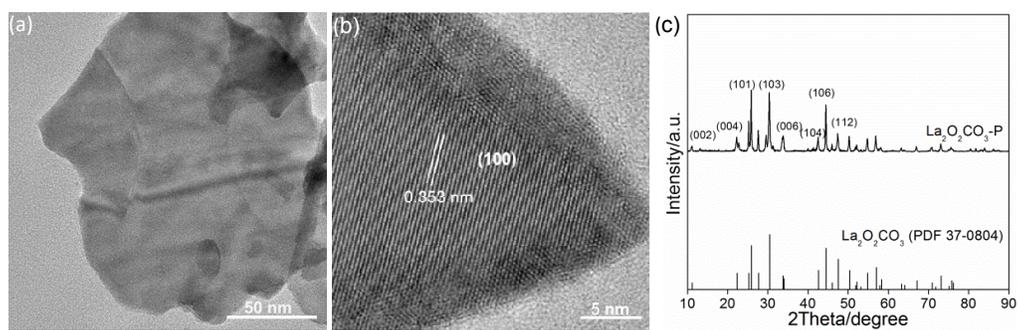
**Figure S1** TPD-MS traces for H<sub>2</sub>O and CO<sub>2</sub> corresponding to the thermal decomposition of samples (a) La(OH)<sub>3</sub>-H, (b) La(OH)<sub>3</sub>-P and (c) La(OH)<sub>3</sub>-H<sub>ref</sub> in air.



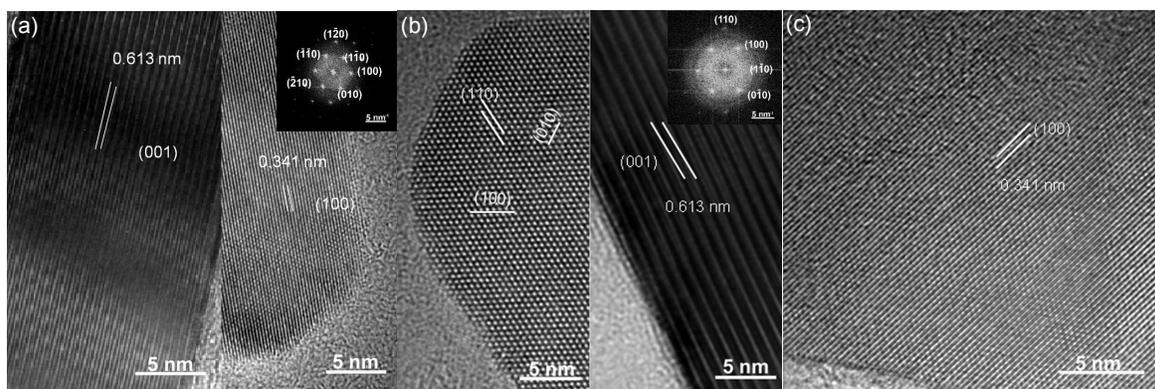
**Figure S2** Particle sizes distribution of samples (a, a')  $\text{La}(\text{OH})_3\text{-H}$ , (b, b')  $\text{La}_2\text{O}_2\text{CO}_3\text{-H}$ , (c, c')  $\text{La}_2\text{O}_3\text{-H}$ , (d)  $\text{La}(\text{OH})_3\text{-P}$ , (e)  $\text{La}_2\text{O}_2\text{CO}_3\text{-P}$ , (f)  $\text{La}_2\text{O}_3\text{-P}$ , (g, g')  $\text{La}(\text{OH})_3\text{-H}_{\text{ref}}$ , (h, h')  $\text{La}_2\text{O}_2\text{CO}_3\text{-H}_{\text{ref}}$ , and (i, i')  $\text{La}_2\text{O}_3\text{-H}_{\text{ref}}$ .  $D_{\text{SEM}}$ ,  $L_{\text{SEM}}$ , and  $T_{\text{SEM}}$  stand for diameters, length, and thickness of the samples, respectively.



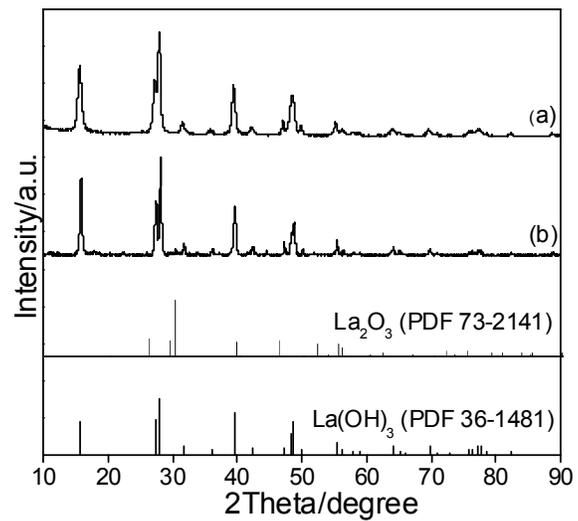
**Figure S3** O 1s binding energy spectra of the samples (a)  $\text{La}_2\text{O}_2\text{CO}_3\text{-H}$ , (b)  $\text{La}_2\text{O}_2\text{CO}_3\text{-P}$ , and (c)  $\text{La}_2\text{O}_2\text{CO}_3\text{-H}_{\text{ref}}$ .



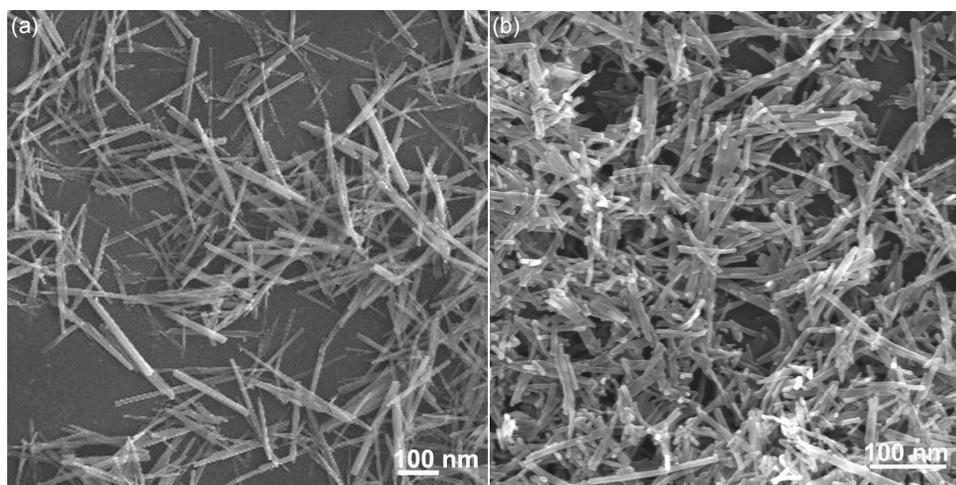
**Figure S4.** TEM, HRTEM images and XRD patterns of the sample  $\text{La}_2\text{O}_2\text{CO}_3\text{-P}$ .



**Figure S5** HRTEM images, SAED patterns of the samples  $\text{La}_2\text{O}_3\text{-H}$  (a) and  $\text{La}_2\text{O}_3\text{-H}_{\text{ref}}$  (b), and HRTEM images of the sample  $\text{La}_2\text{O}_3\text{-P}$  (c).



**Figure S6** XRD patterns of the fresh sample  $\text{La}(\text{OH})_3\text{-H}$  (a) and the hydrolysis-treated sample  $\text{La}_2\text{O}_3\text{-H}$  (b).



**Figure S7** SEM images of the fresh sample  $\text{La}(\text{OH})_3\text{-H}$  (a) and the hydrolysis-treated sample  $\text{La}_2\text{O}_3\text{-H}$  (b).