

Supporting Information

A Facile Access to Novel [60]Fullerenyl Diethers and [60]Fullerene-Sugar Conjugates via Annulation of Diol Moieties

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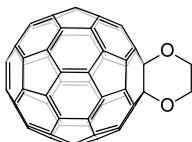
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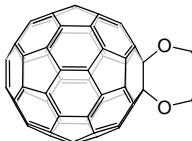
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General Procedure for the Reaction of C₆₀ with Diols 1a–o.

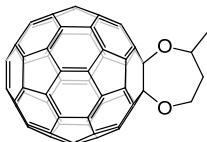
Firstly, C₆₀ (0.05 mmol) was dissolved in anhydrous chlorobenzene (7 mL) by sonication, then iron salt (0.15 mmol FeCl₂ for **1a–g**, **1i**, **1k**, **1l**, **1n** and **1o**; 0.15 mmol FeCl₃ for **1h**; 0.10 mmol FeCl₃ for **1j** and **1m**), the hypervalent iodine(III) reagent (0.075 mmol/0.075 mmol PIDA/PIFA for **1a–e**, **1g–i**, **1k**, **1l** and **1n**; 0.15 mmol PIDA for **1f** and **1o**; 0.10 mmol PIDA for **1j** and **1m**) and diol (0.25 mmol for **1a–f**, **1k–o**; 0.50 mmol for **1g** and **1l**; 0.10 mmol for **1h**; 0.15 mmol for **1j**) were added. After being stirred at the designated reaction time and temperature, then the reaction mixture was directly separated on a silica gel column with CS₂/CH₂Cl₂ or CS₂/EtOAc as the eluent. The desired product **2** was obtained along with recovered C₆₀.



By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1a** (14 μL, 0.25 mmol), FeCl₂ (19 mg, 0.15 mmol), PIDA (24 mg, 0.075 mmol) and PIFA (33 mg, 0.075 mmol) in anhydrous CB (7 mL) at 100 °C for 1 h afforded **2a**^[1] (14.9 mg, 38%) and recovered C₆₀ (20.3 mg, 56%): amorphous brown solid. ¹H NMR (400 MHz, CS₂/CDCl₃) δ 4.99 (s, 4H).

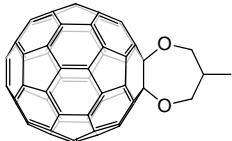


By following the general procedure, the reaction of C₆₀ (36.1 mg, 0.05 mmol) with **1b** (18 μL, 0.25 mmol), FeCl₂ (20 mg, 0.15 mmol), PIDA (25 mg, 0.075 mmol) and PIFA (34 mg, 0.075 mmol) in anhydrous CB (7 mL) at 100 °C for 1 h afforded **2b**^[1] (19.8 mg, 50%) and recovered C₆₀ (10.1 mg, 28%): amorphous brown solid. ¹H NMR (400 MHz, CS₂/CDCl₃) δ 5.06–5.02 (m, 4H), 2.68–2.61 (m, 2H).

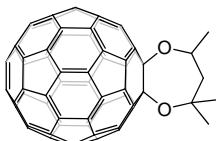


By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1c** (23 μL, 0.25 mmol), FeCl₂ (19 mg, 0.15 mmol), PIDA (24 mg, 0.075 mmol) and PIFA (32 mg, 0.075 mmol) in anhydrous CB (7 mL) at 100 °C for 1 h afforded **2c**^[1] (15.5 mg, 38%) and recovered C₆₀ (14 mg, 39%): amorphous brown solid. ¹H NMR (400 MHz, CS₂/CDCl₃) δ 5.40–5.31 (m, 1H), 5.15 (td, *J* = 12.8, 2.0 Hz, 1H), 4.87

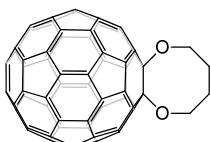
(ddd, $J = 12.8, 4.6, 2.0$ Hz, 1H), 2.82-2.70 (m, 1H), 2.31 (dq, $J = 14.4, 2.0$ Hz, 1H), 1.74 (d, $J = 6.0$ Hz, 3H).



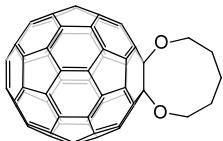
By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1d** (22 μ L, 0.25 mmol), FeCl₂ (19.2 mg, 0.15 mmol), PIDA (24 mg, 0.075 mmol) and PIFA (32 mg, 0.075 mmol) in anhydrous CB (7 mL) at 100 °C for 1 h afforded **2d** (16.3 mg, 40%) and recovered C₆₀ (14.1 mg, 39%): amorphous brown solid. ¹H NMR (400 MHz, CS₂/CDCl₃, broad peaks due to ring flipping) δ 4.98-4.73 (br s, 2H), 4.73-4.64 (m, 2H), 3.00-2.80 (br s, 1H), 1.41-1.24 (br s, 3H); ¹³C NMR (100 MHz, CS₂/CDCl₃, -60 °C) δ 150.71, 148.31, 148.13, 146.18, 146.15, 145.80, 145.78, 145.74, 145.54, 145.11, 145.01, 144.98, 144.80, 144.74, 144.40, 144.22, 144.17, 142.20, 142.17, 141.94, 141.92, 141.51, 141.34, 140.98, 140.27, 139.05, 138.91, 138.47, 135.67, 91.60 (sp³-C of C₆₀), 53.76, 35.70, 11.48; FT-IR ν /cm⁻¹ (KBr) 2925, 2863, 1457, 1430, 1389, 1168, 1124, 1089, 1024, 980, 909, 598, 526, 501; UV-vis (CHCl₃) $\lambda_{\text{max}}/\text{nm}$ 259, 318, 415; HRMS (MALDI-TOF) m/z calcd for C₆₄H₈O₂ [M⁺] 808.0519, found 808.0545.



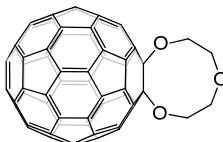
By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1e** (32 μ L, 0.25 mmol), FeCl₂ (19 mg, 0.15 mmol), PIDA (24 mg, 0.075 mmol) and PIFA (32.3 mg, 0.075 mmol) in anhydrous CB (7 mL) at 100 °C for 1 h afforded **2e** (8.9 mg, 21%) and recovered C₆₀ (15 mg, 42%): amorphous brown solid. ¹H NMR (300 MHz, CS₂/CDCl₃) δ 5.40-5.28 (m, 1H), 3.37 (dd, $J = 15.8, 7.7$ Hz, 1H), 3.01 (dd, $J = 15.8, 4.4$ Hz, 1H), 1.93 (s, 3H), 1.91 (s, 3H), 1.87 (d, $J = 6.6$ Hz, 3H); ¹³C NMR (100 MHz, CS₂/CDCl₃) δ 152.92, 152.78, 152.57, 151.60, 148.54, 146.52, 146.46, 146.42, 146.16, 146.12, 145.94, 145.91, 145.84, 145.80, 145.41, 145.35, 145.28, 145.24, 145.08, 145.03, 144.90, 144.82, 142.69, 142.67, 142.64, 142.63, 142.60, 142.35, 141.59, 141.56, 141.52, 141.51, 141.46, 141.39, 141.28, 139.53, 139.51, 139.42, 139.38, 136.79, 136.62, 136.33, 136.19, 88.75 (sp³-C of C₆₀), 88.49 (sp³-C of C₆₀), 80.65, 74.66, 43.56, 31.62, 30.91, 23.93; FT-IR ν /cm⁻¹ (KBr) 2969, 2922, 2860, 1434, 1368, 1152, 1061, 953, 573, 526; UV-vis (CHCl₃) $\lambda_{\text{max}}/\text{nm}$ (log ϵ) 258, 318, 415; HRMS (MALDI-TOF) m/z calcd for C₆₆H₁₂O₂ [M⁺] 836.0832, found 836.0810.



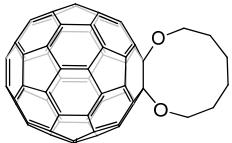
By following the general procedure, the reaction of C₆₀ (36.1 mg, 0.05 mmol) with **1f** (23 µL, 0.25 mmol), FeCl₂ (19.8 mg, 0.15 mmol) and PIDA (49.7 mg, 0.15 mmol) in anhydrous CB (7 mL) at 100 °C for 1 h afforded **2f** (9.3 mg, 23%) and recovered C₆₀ (24.2 mg, 68%): amorphous brown solid. ¹H NMR (300 MHz, CS₂/CDCl₃, broad peaks due to ring flipping) δ 5.70-4.54 (br s, 4H), 2.51-2.30 (br s, 4H); ¹³C NMR (100 MHz, CS₂/CDCl₃, -60 °C) δ 150.63, 149.97, 148.10, 146.14, 146.08, 145.83, 145.79, 145.74, 145.53, 145.38, 145.07, 144.84, 144.82, 144.56, 144.19, 144.07, 142.22, 142.13, 142.07, 141.93, 141.85, 141.46, 141.15, 141.13, 140.83, 139.14, 139.09, 137.93, 135.32, 87.26 (sp³-C of C₆₀), 69.21, 27.81; FT-IR ν/cm⁻¹ (KBr) 2921, 2867, 1461, 1431, 1372, 1221, 1127, 1091, 1071, 976, 526; UV-vis (CHCl₃) λ_{max}/nm (log ε) 259, 318, 415; HRMS (MALDI-TOF) m/z calcd for C₆₄H₈O₂ [M⁺] 808.0519, found 808.0491.



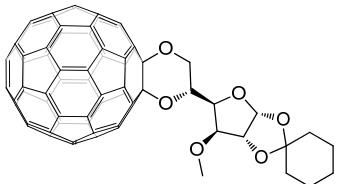
By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1g** (52 µL, 0.5 mmol), FeCl₂ (19 mg, 0.15 mmol), PIDA (24 mg, 0.075 mmol) and PIFA (32 mg, 0.075 mmol) in anhydrous CB (7 mL) at 100 °C for 1 h afforded **2g** (8.1 mg, 20%) and recovered C₆₀ (19.6 mg, 54%): amorphous brown solid. ¹H NMR (400 MHz, CS₂/CDCl₃) δ 5.06-5.01 (m, 4H), 2.32-2.26 (m, 6H); ¹³C NMR (100 MHz, CS₂/CDCl₃) δ 151.25, 148.56, 146.51, 146.20, 145.95, 145.52, 145.47, 145.29, 144.78, 142.59, 142.57, 142.34, 141.58, 141.54, 139.47, 137.36, 89.18 (sp³-C of C₆₀), 71.20, 29.93, 26.31; FT-IR ν/cm⁻¹ (KBr) 2916, 2867, 1508, 1463, 1430, 1180, 1124, 1095, 1041, 601, 573, 565, 526; UV-vis (CHCl₃) λ_{max}/nm (log ε) 257, 316, 415; HRMS (MALDI-TOF) m/z calcd for C₆₅H₁₀O₂ [M⁺] 822.0675, found 822.0699.



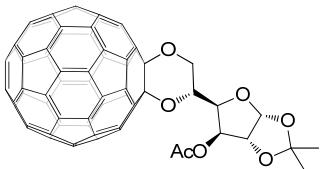
By following the general procedure, the reaction of C₆₀ (37.0 mg, 0.05 mmol) with **1h** (10 µL, 0.10 mmol), FeCl₃ (30 mg, 0.15 mmol), PIDA (25.5 mg, 0.075 mmol) and PIFA (32.8 mg, 0.075 mmol) in anhydrous CB (7 mL) at 130 °C for 2.5 h afforded **2h** (4.0 mg, 10%) and recovered C₆₀ (19 mg, 53%): amorphous brown solid. ¹H NMR (400 MHz, CS₂/CDCl₃) δ 4.99 (s, 8H); ¹³C NMR (75 MHz, CS₂/DMSO-d₆) δ 148.30, 148.11, 146.16, 145.84, 145.81, 145.13, 144.97, 144.85, 144.40, 142.26, 142.18, 141.90, 141.38, 141.06, 139.24, 137.40, 86.68 (sp³-C of C₆₀), 60.91; FT-IR ν/cm⁻¹ (KBr) 2969, 2923, 2869, 1511, 1462, 1428, 1265, 1182, 1132, 1092, 1007, 942, 897, 846, 768, 694, 663, 594, 568, 527, 497; UV-vis (CHCl₃) λ_{max}/nm (log ε) 257, 318, 423; HRMS (MALDI-TOF) m/z calcd for C₆₂H₄O₂ [M-OC₂H₄]⁺ 780.0206, found 780.0173.



By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1i** (62 mg, 0.50 mmol), FeCl₂ (19.3 mg, 0.15 mmol), PIDA (24 mg, 0.075 mmol) and PIFA (32 mg, 0.075 mmol) in anhydrous CB (7 mL) at 100 °C for 1 h afforded **2i** (6.9 mg, 17%) and recovered C₆₀ (19.5 mg, 54%): amorphous brown solid. ¹H NMR (400 MHz, CS₂/C₆D₆) δ 5.36 (t, *J* = 5.0 Hz, 4H), 2.47-2.36 (m, 8H). ¹³C NMR (100 MHz, CS₂/C₆D₆) δ 152.59, 149.37, 147.32, 147.01, 146.75, 146.35, 146.29, 146.10, 145.62, 143.43, 143.20, 142.37, 142.33, 140.24, 137.99, 90.12 (sp³-C of C₆₀), 69.29, 31.24, 20.99; FT-IR ν /cm⁻¹ (KBr) 2925, 2866, 1510, 1428, 1182, 1122, 1098, 1066, 963, 598, 569, 526; UV-vis (CHCl₃) λ_{max} /nm (log ε) 258, 317, 415; HRMS (MALDI-TOF) *m/z* calcd for C₆₆H₁₂O₂ [M⁺] 836.0832, found 836.0863.

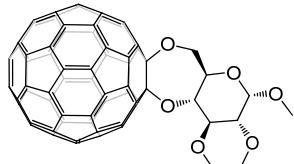


By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1j** (41 mg, 0.15 mmol), FeCl₃ (16.3 mg, 0.10 mmol), and PIDA (33.3 mg, 0.10 mmol) at 100 °C for 2 h afforded **2j** (12.7 mg, 25%) and recovered C₆₀ (19.7 mg, 55%): amorphous brown solid. ¹H NMR (300 MHz, CDCl₃) δ 6.11 (d, *J* = 3.6 Hz, 1H), 5.49-5.41 (m, 1H), 5.16 (dd, *J* = 11.8, 7.0 Hz, 1H), 5.00 (dd, *J* = 11.8, 7.0 Hz, 1H), 4.96 (dd, *J* = 7.6, 3.0 Hz, 1H), 4.71 (d, *J* = 3.6 Hz, 1H), 4.20 (d, *J* = 3.0 Hz, 1H), 1.93-1.83 (m, 2H), 1.81-1.69 (m, 2H), 1.69-1.61 (m, 4H), 1.51-1.41 (m, 2H); ¹³C NMR (100 MHz, CS₂/CDCl₃) δ 148.90, 148.71, 148.54, 148.49, 148.39, 148.26, 146.57, 146.54, 146.51, 146.25, 146.22, 146.19, 146.16, 146.08, 145.65, 145.50, 145.39, 145.31, 145.20, 145.18, 145.15, 144.81, 144.72, 142.67, 142.65, 142.62, 142.57, 142.30, 142.23, 141.83, 141.81, 141.74, 141.72, 141.70, 141.65, 141.62, 139.68, 139.65, 139.59, 139.56, 138.57, 138.47, 137.27, 137.07, 112.55, 105.04, 87.11 (sp³-C of C₆₀), 87.04 (sp³-C of C₆₀), 84.01, 82.22, 81.86, 68.40, 64.53, 58.60, 36.90, 36.12, 25.35, 24.29, 24.02; FT-IR ν /cm⁻¹ (KBr) 2928, 2859, 1451, 1437, 1366, 1159, 1121, 1087, 1023, 935, 839, 753, 709, 562, 526; UV-vis (CHCl₃) λ_{max} nm (log ε) 258, 317, 423; HRMS (MALDI-TOF) *m/z* calcd for C₇₃H₂₀O₆ [M⁺] 992.1254, found 992.1295.

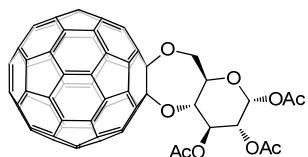


By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1k** (65.5 mg, 0.25 mmol), FeCl₂ (19.1 mg, 0.15 mmol), PIDA (24 mg, 0.075 mmol) and

PIFA (31.8 mg, 0.075 mmol) at 80 °C for 1.5 h afforded **2k** (11.3 mg, 23%) and recovered C₆₀ (19.8 mg, 55%): brown amorphous solid. ¹H NMR (400 MHz, CDCl₃) δ 6.04 (d, *J* = 3.6 Hz, 1H), 5.68 (d, *J* = 3.0 Hz, 1H), 5.29 (dt, *J* = 8.4, 6.6 Hz, 1H), 5.11 (dd, *J* = 11.8, 7.0 Hz, 1H), 4.98 (dd, *J* = 8.4, 3.0 Hz, 1H), 4.90 (dd, *J* = 11.8, 6.2 Hz, 1H), 4.60 (d, *J* = 3.6 Hz, 1H), 1.91 (s, 3H), 1.66 (s, 3H), 1.39 (s, 3H); ¹³C NMR (100 MHz, CS₂/CDCl₃) δ 168.43, 148.62, 148.59, 148.19, 148.14, 147.75, 146.66, 146.65, 146.58, 146.34, 146.33, 146.25, 146.11, 145.70, 145.55, 145.45, 145.44, 145.33, 145.25, 145.20, 145.15, 144.90, 144.82, 144.80, 144.77, 142.77, 142.75, 142.71, 142.70, 142.64, 142.63, 142.38, 142.36, 142.33, 142.25, 142.06, 141.91, 141.82, 141.77, 141.64, 141.16, 140.93, 139.79, 139.75, 139.68, 139.56, 138.49, 138.32, 137.40, 137.07, 112.44, 105.31, 87.31 (sp³-C of C₆₀), 86.88 (sp³-C of C₆₀), 83.56, 81.21, 75.53, 67.99, 64.86, 26.90, 26.30, 20.35; FT-IR ν /cm⁻¹ (KBr) 2980, 2922, 1751, 1460, 1429, 1373, 1220, 1160, 1093, 1029, 935, 528; UV-vis (CHCl₃) λ_{max} nm (log ε) 257, 318, 424; HRMS (MALDI-TOF) *m/z* calcd for C₇₁H₁₆O₇ [M⁺] 980.0891, found 980.0889.

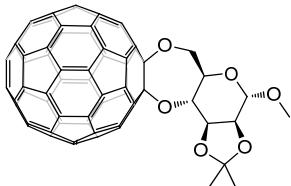


By following the general procedure, the reaction of C₆₀ (36.3 mg, 0.05 mmol) with **1l** (56 mg, 0.25 mmol), FeCl₂ (19 mg, 0.15 mmol), PIDA (23.5 mg, 0.075 mmol) and PIFA (32 mg, 0.075 mmol) at 100 °C for 2 h afforded **2l** (13.3 mg, 28%) and recovered C₆₀ (7 mg, 19%): brown amorphous solid. ¹H NMR (400 MHz, CDCl₃) δ 5.02 (d, *J* = 3.6 Hz, 1H), 4.99 (dd, *J* = 12.0, 10.4 Hz, 1H), 4.88 (dd, *J* = 10.4, 9.2 Hz, 1H), 4.99 (dd, *J* = 12.0, 4.8 Hz, 1H), 4.64 (ddd, *J* = 10.4, 10.4, 4.8 Hz, 1H), 4.09 (t, *J* = 9.2 Hz, 1H), 3.76 (s, 3H), 3.66 (s, 3H), 3.64 (s, 3H), 3.49 (dd, *J* = 9.2, 3.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 151.09, 150.90, 149.06, 148.99, 148.96, 148.30, 146.90, 146.81, 146.55, 146.51, 146.47, 146.45, 146.26, 146.24, 145.83, 145.73, 145.70, 145.69, 145.54, 145.50, 145.43, 145.42, 145.10, 145.06, 144.98, 144.89, 144.87, 144.80, 142.89, 142.86, 142.84, 142.83, 142.62, 142.56, 142.55, 142.11, 142.05, 141.93, 141.72, 141.69, 141.14, 140.05, 139.77, 139.72, 139.60, 139.26, 138.64, 136.36, 136.13, 98.20, 91.72 (sp³-C of C₆₀), 91.54 (sp³-C of C₆₀), 84.67, 81.99, 81.65, 72.67, 66.83, 62.01, 59.86, 55.67; FT-IR ν /cm⁻¹ (KBr) 2922, 2818, 1455, 1437, 1378, 1192, 1088, 1052, 967, 911, 769, 528; UV-vis (CHCl₃) λ_{max} nm (log ε) 257, 318, 423; HRMS (MALDI-TOF) *m/z* calcd for C₆₉H₁₆O₆ [M⁺] 940.0941, found 940.0979.

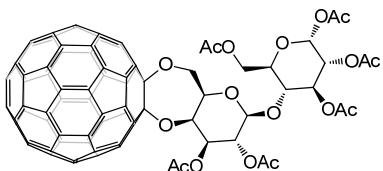


By following the general procedure, the reaction of C₆₀ (36.2 mg, 0.05 mmol) with **1m** (78.7 mg, 0.25 mmol), FeCl₃ (16.1 mg, 0.10 mmol), and PIDA (32 mg, 0.10

mmol) at 80 °C for 3 h afforded **2m** (8.5 mg, 17%) and recovered C₆₀ (12.1 mg, 33%): brown amorphous solid. ¹H NMR (400 MHz, CDCl₃) δ 6.51 (d, *J* = 3.6 Hz, 1H), 6.03 (t, *J* = 9.8 Hz, 1H), 5.37 (dd, *J* = 9.8, 3.6 Hz, 1H), 5.11 (t, *J* = 9.8 Hz, 1H), 5.02 (td, *J* = 11.6, 1.4 Hz, 1H), 4.95-4.85 (m, 2H), 2.36 (s, 3H), 2.10 (s, 3H), 1.90 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 170.17, 170.13, 169.39, 150.68, 150.17, 149.04, 149.03, 147.97, 147.93, 146.94, 146.89, 146.63, 146.59, 146.57, 146.52, 146.44, 146.33, 146.28, 145.79, 145.77, 145.70, 145.57, 145.55, 145.35, 145.15, 145.03, 144.93, 144.88, 144.83, 144.71, 144.29, 142.99, 142.95, 142.94, 142.92, 142.89, 142.66, 142.62, 142.53, 142.13, 142.08, 142.04, 141.76, 141.61, 141.12, 141.09, 139.92, 139.81, 139.77, 139.67, 139.00, 138.91, 136.61, 136.17, 91.68 (sp³-C of C₆₀), 91.58 (sp³-C of C₆₀), 89.54, 81.28, 71.90, 70.23, 69.98, 69.22, 21.25, 21.12, 20.74; FT-IR ν/cm⁻¹ (KBr) 2954, 1755, 1427, 1370, 1212, 1138, 1064, 1012, 979, 935, 773, 731, 599, 526, 484; UV-vis (CHCl₃) λ_{max} nm (log ε) 257, 318, 424; HRMS (MALDI-TOF) *m/z* calcd for C₇₂H₁₆O₉ [M⁺] 1024.0789, found 1024.0766.



By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1n** (58.5 mg, 0.25 mmol), FeCl₂ (19 mg, 0.15 mmol), PIDA (24.5 mg, 0.075 mmol), and PIFA (33 mg, 0.075 mmol) at 120 °C for 1.5 h afforded **2n** (12.6 mg, 26%) and recovered C₆₀ (8.8 mg, 24%): brown amorphous solid. ¹H NMR (400 MHz, CDCl₃) δ 5.15 (s, 1H), 5.12-5.03 (m, 2H), 4.91 (dd, *J* = 12.4, 5.2 Hz, 1H), 4.71 (dd, *J* = 7.6, 5.2 Hz, 1H), 4.57-4.49 (m, 1H), 4.41 (d, *J* = 5.2 Hz, 1H), 3.61 (s, 3H), 1.43 (s, 3H), 1.40 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 151.03, 150.90, 149.28, 149.02, 148.99, 148.38, 146.96, 146.93, 146.91, 146.59, 146.57, 146.53, 146.44, 146.31, 145.78, 145.75, 145.72, 145.52, 145.45, 145.33, 145.15, 145.10, 145.00, 144.94, 144.90, 142.94, 142.91, 142.90, 142.87, 142.83, 142.65, 142.60, 142.17, 142.12, 142.01, 141.75, 141.71, 141.07, 140.94, 139.97, 139.77, 139.59, 139.18, 138.43, 136.47, 136.34, 109.80, 98.48, 91.83 (sp³-C of C₆₀), 91.56 (sp³-C of C₆₀), 82.56, 76.57, 76.34, 72.45, 65.01, 55.50, 27.55, 26.59; FT-IR ν/cm⁻¹ (KBr) 2990, 2982, 1454, 1432, 1376, 1218, 1126, 1086, 1029, 977, 903, 853, 776, 600, 562, 526, 479; UV-vis (CHCl₃) λ_{max} nm (log ε) 258, 318, 422; HRMS (MALDI-TOF) *m/z* calcd for C₇₀H₁₆O₆ [M⁺] 952.0941, found 952.0908.



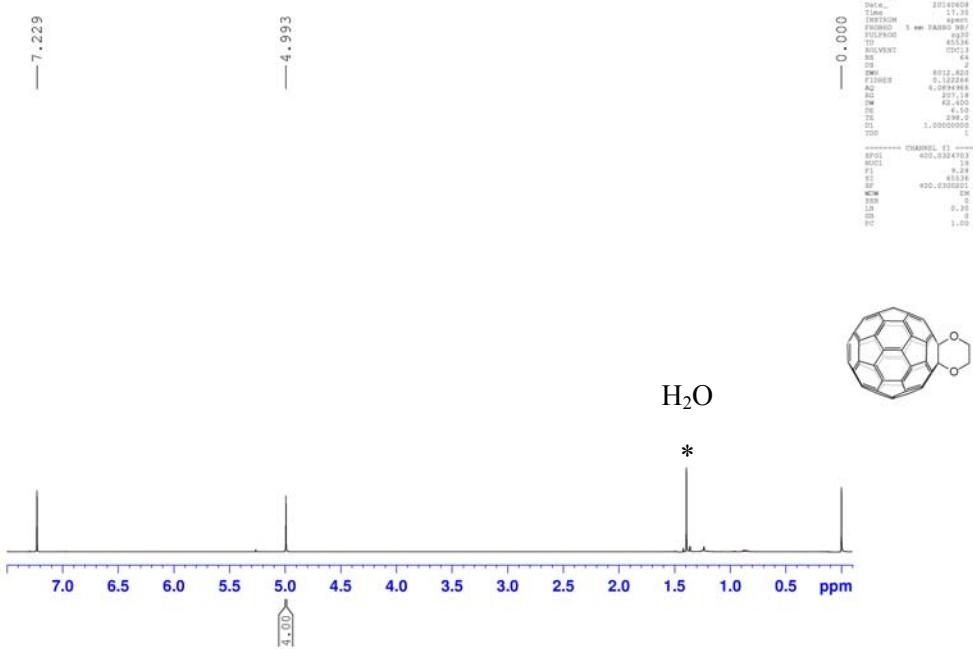
By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1o** (149 mg, 0.25 mmol), FeCl₂ (18.9 mg, 0.15 mmol), and PIDA (48.3 mg, 0.15 mmol) at 100 °C for 2 h afforded **2o** (9.7 mg, 15%) and recovered C₆₀ (18.4 mg, 51%):

brown amorphous solid. ^1H NMR (400 MHz, CDCl_3) δ 6.35 (d, $J = 3.6$ Hz, 1H), 5.86 (d, $J = 3.6$ Hz, 1H), 5.73 (dd, $J = 10.6, 8.0$ Hz, 1H), 5.68 (dd, $J = 10.0, 9.2$ Hz, 1H), 5.31 (d, $J = 13.6$ Hz, 1H), 5.18 (dd, $J = 10.0, 3.6$ Hz, 1H), 5.15 (dd, $J = 10.0, 3.6$ Hz, 1H), 5.01 (dd, $J = 13.6, 1.6$ Hz, 1H), 4.77 (d, $J = 8.0$ Hz, 1H), 4.57 (dd, $J = 12.0, 2.0$ Hz, 1H), 4.29 (dd, $J = 12.0, 4.4$ Hz, 1H), 4.21-4.14 (m, 2H), 4.09 (dd, $J = 10.0, 9.2$ Hz, 1H), 2.53 (s, 3H), 2.25 (s, 3H), 2.20 (s, 3H), 2.14 (s, 3H), 2.07 (s, 3H), 1.86 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 170.91, 170.80, 170.55, 170.24, 169.30, 169.24, 151.06, 150.80, 149.00, 148.98, 148.10, 147.87, 146.97, 146.93, 146.90, 146.63, 146.60, 146.52, 146.46, 146.41, 146.36, 146.23, 145.81, 145.78, 145.76, 145.71, 145.51, 145.26, 145.16, 145.15, 144.85, 144.82, 144.77, 144.72, 144.69, 142.97, 142.89, 142.88, 142.62, 142.60, 142.54, 142.34, 142.31, 142.24, 142.09, 141.68, 141.52, 141.02, 140.94, 139.85, 139.67, 139.58, 139.21, 138.95, 136.98, 136.58, 101.65, 93.23 (sp³-C of C₆₀), 92.45 (sp³-C of C₆₀), 89.35, 77.88, 75.61, 73.39, 72.93, 71.94, 71.23, 69.67, 69.54, 61.89, 21.53, 21.19, 21.08, 21.04, 20.99, 20.75; FT-IR ν/cm^{-1} (KBr) 1750, 1429, 1367, 1219, 1150, 1046, 936, 907, 769, 744, 682, 601, 561, 528; UV-vis (CHCl_3) λ_{\max} nm (log ε) 257, 318, 424; HRMS (MALDI-TOF) m/z calcd for C₇₃H₂₀O₆ [M⁺] 1312.1634, found 1312.1649.

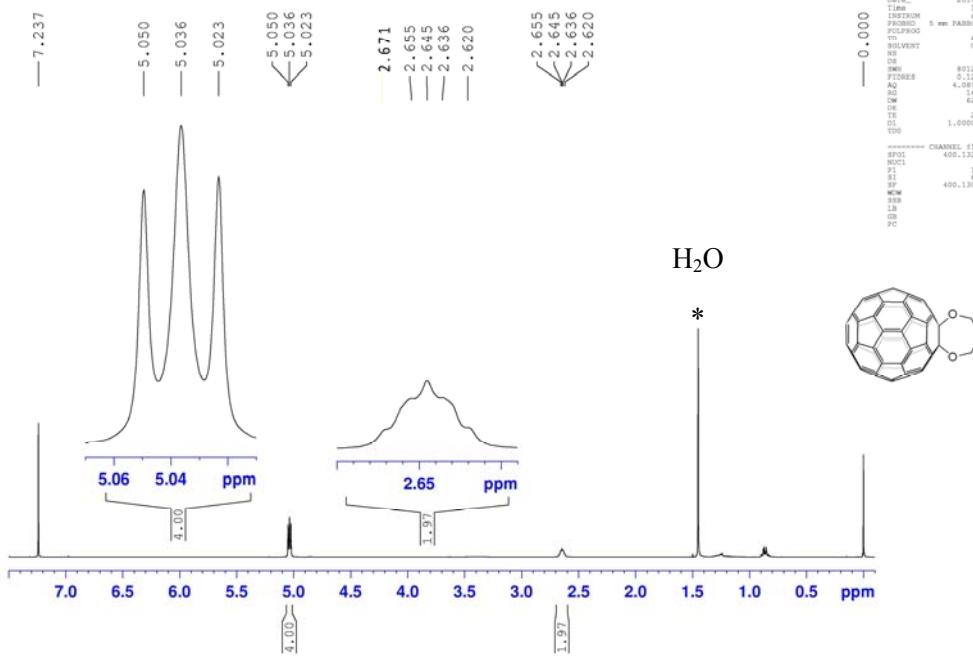
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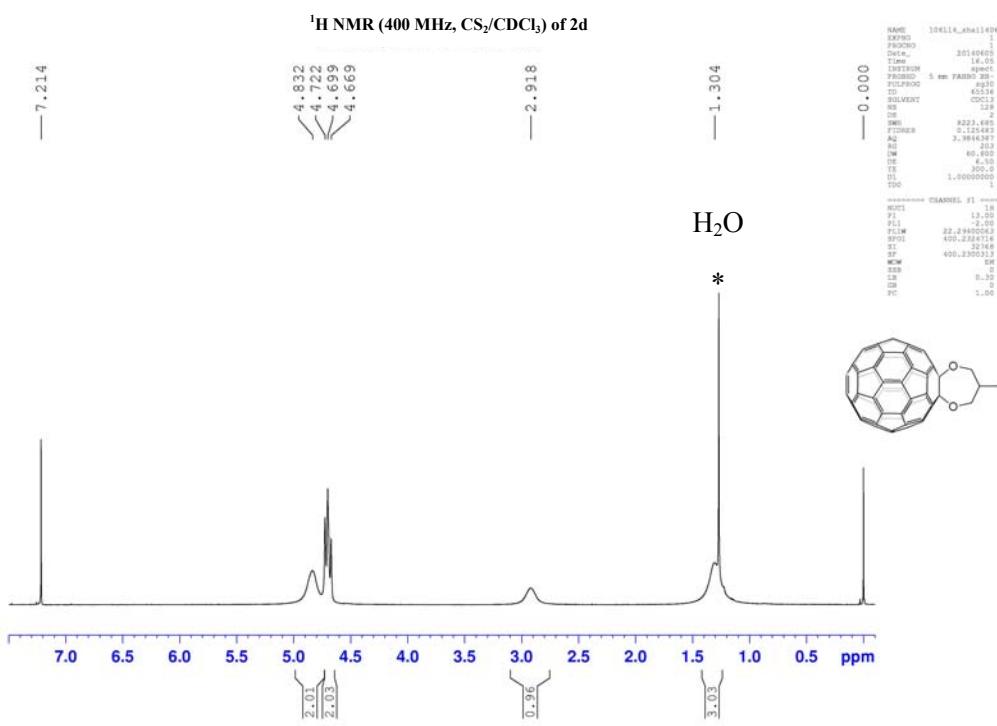
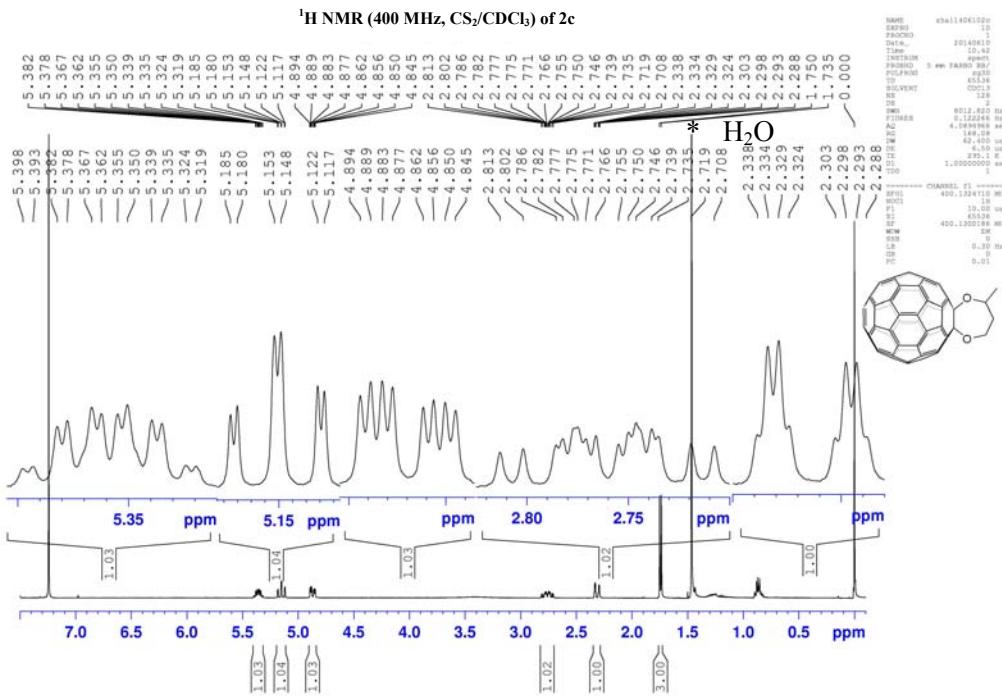
- (1) Li, F.-B.; You, X.; Liu, T.-X.; Wang, G.-W. *Org. Lett.* **2012**, *14*, 1800.

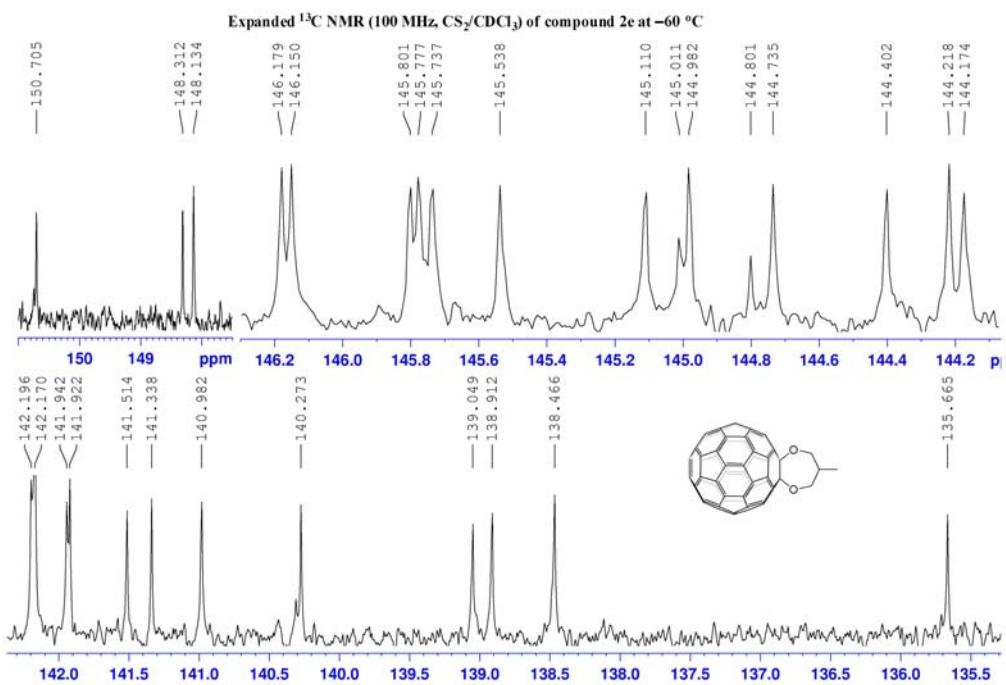
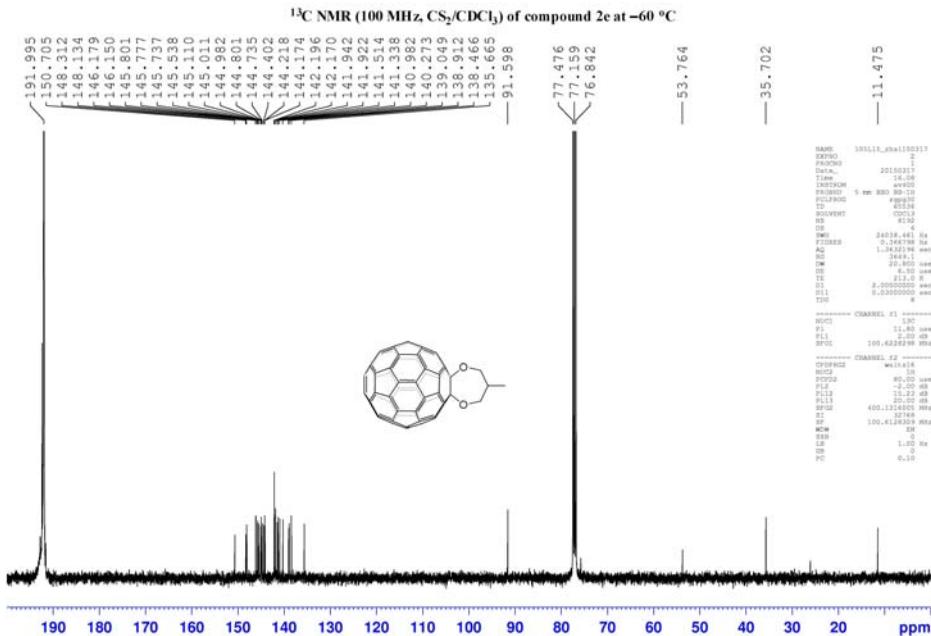
¹H NMR (400 MHz, CS₂/CDCl₃) of 2a



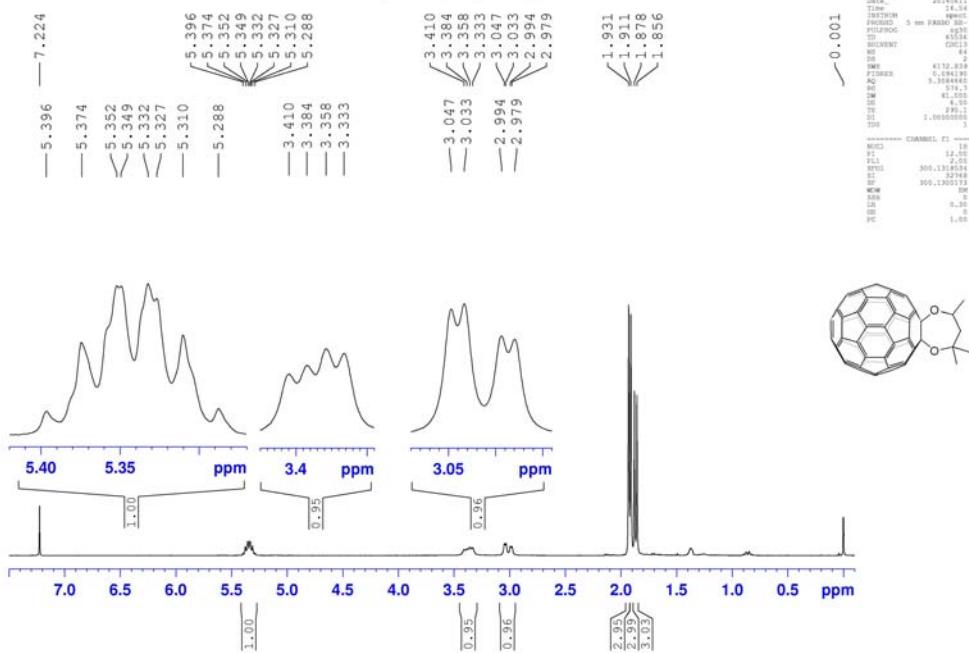
¹H NMR (400 MHz, CS₂/CDCl₃) of 2b



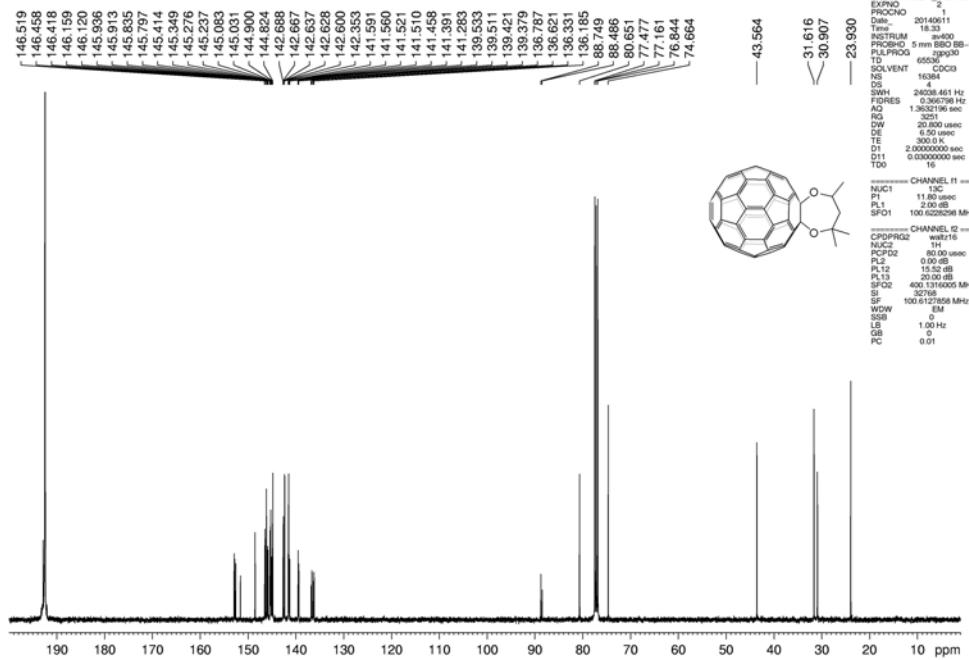




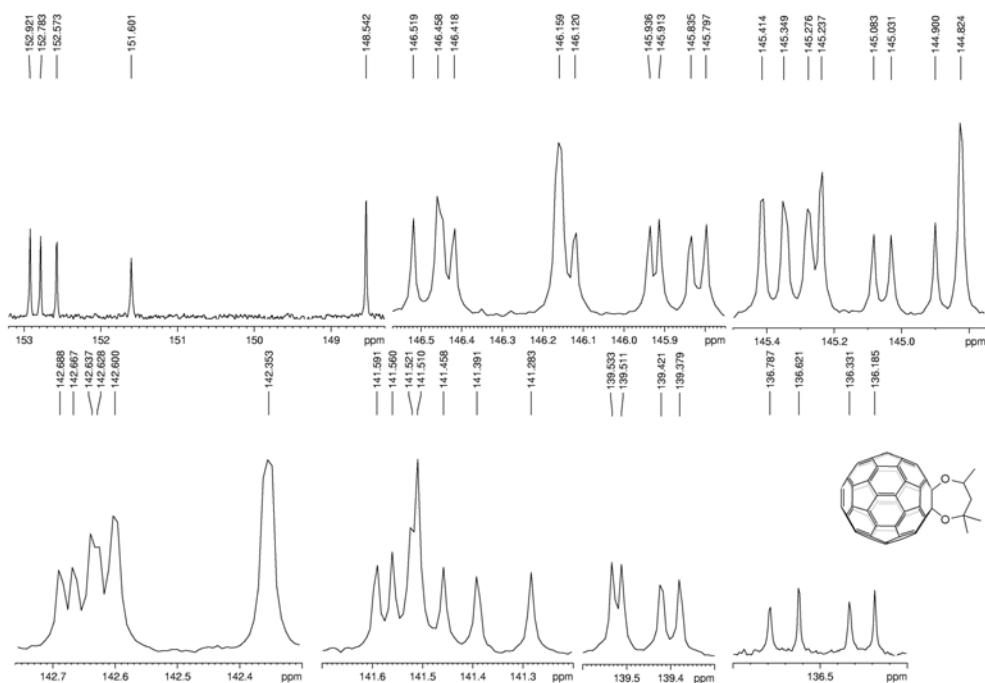
¹H NMR (300 MHz, CS₂/CDCl₃) of 2e



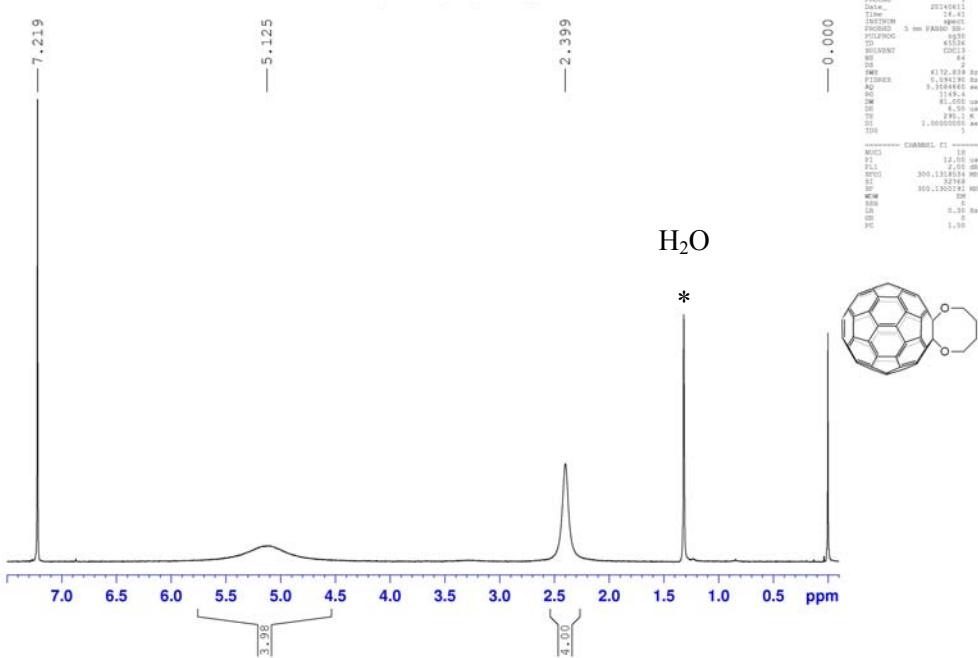
¹³C NMR (100 MHz, CS₂/CDCl₃) of 2e

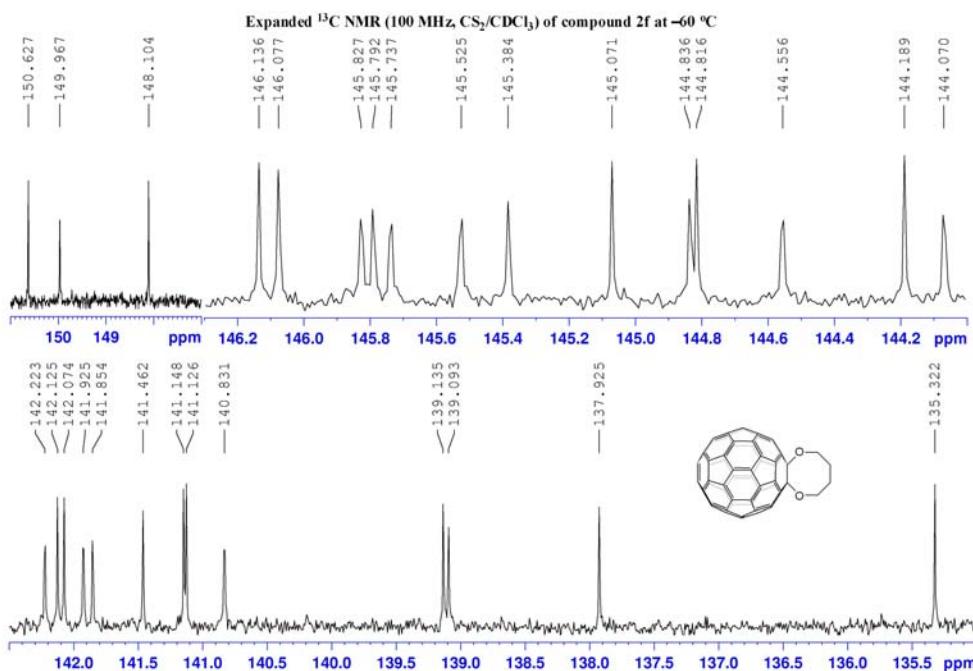
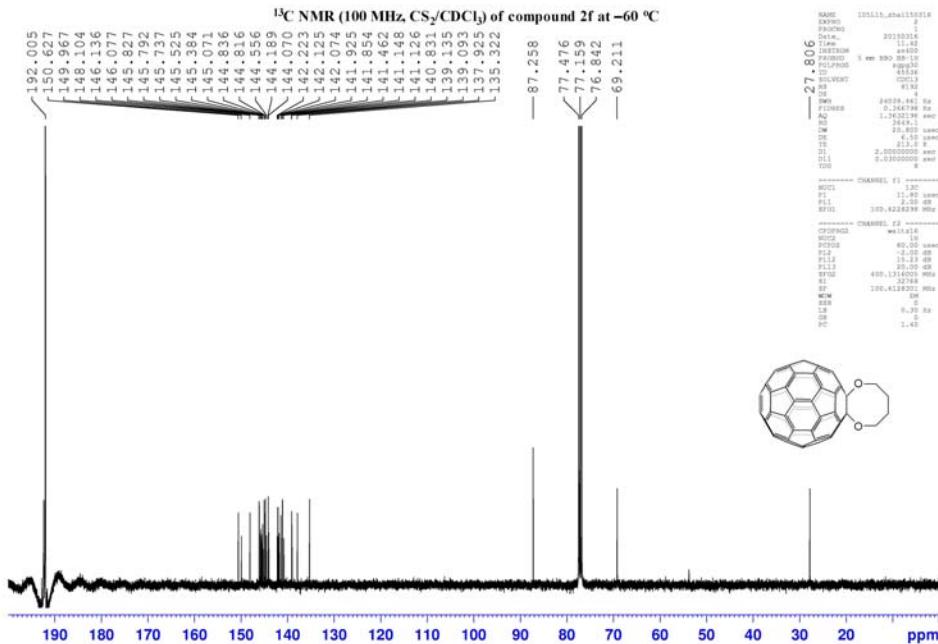


Expanded ^{13}C NMR (100 MHz, $\text{CS}_2/\text{CDCl}_3$) of 2e

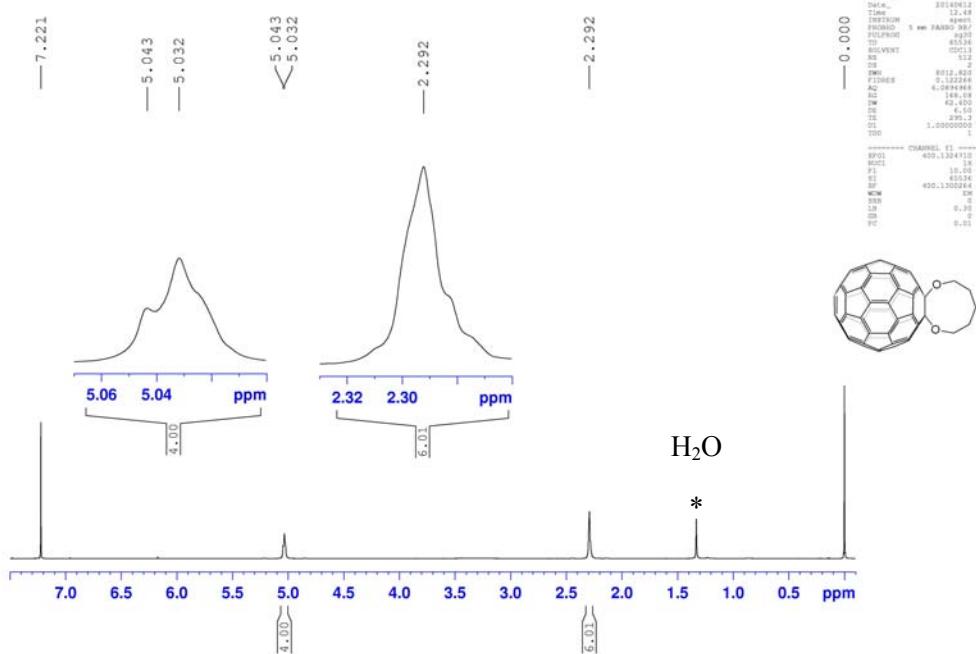


^1H NMR (300 MHz, $\text{CS}_2/\text{CDCl}_3$) of 2f





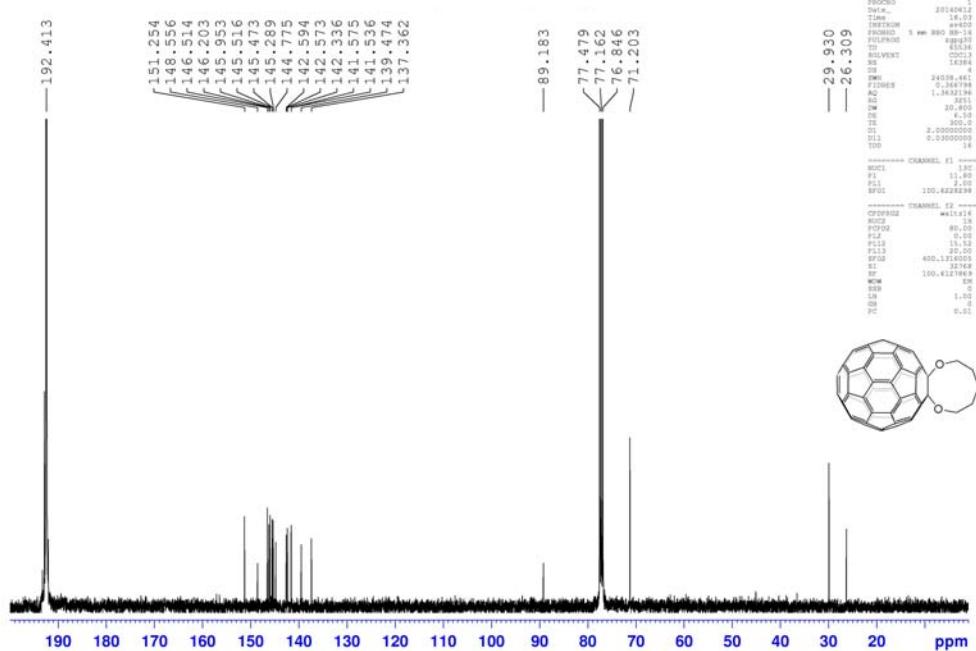
¹H NMR (400 MHz, CS₂/CDCl₃) of 2g



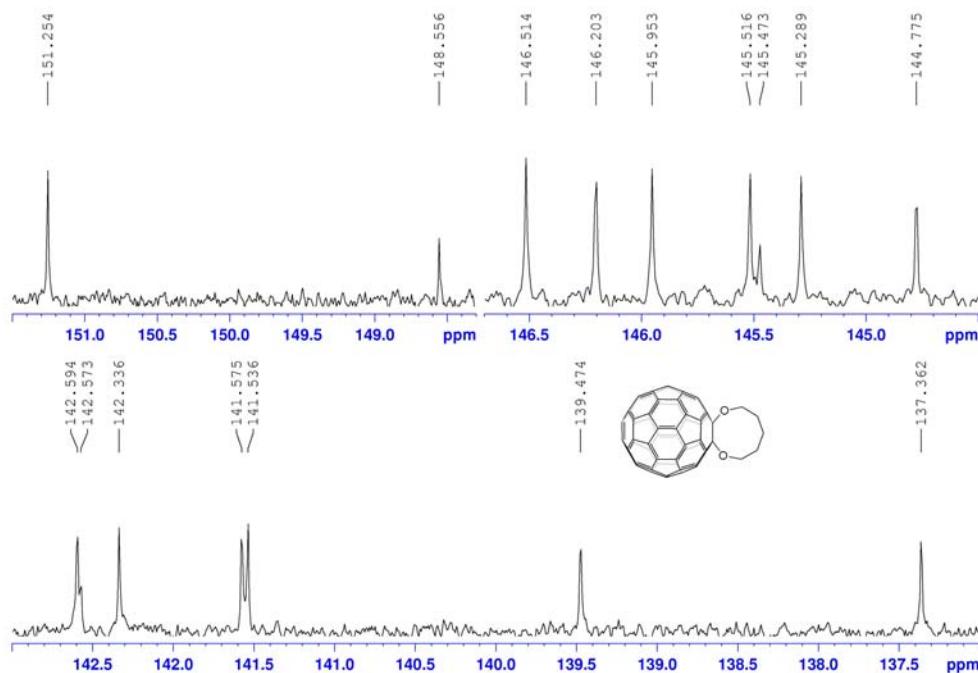
H₂O



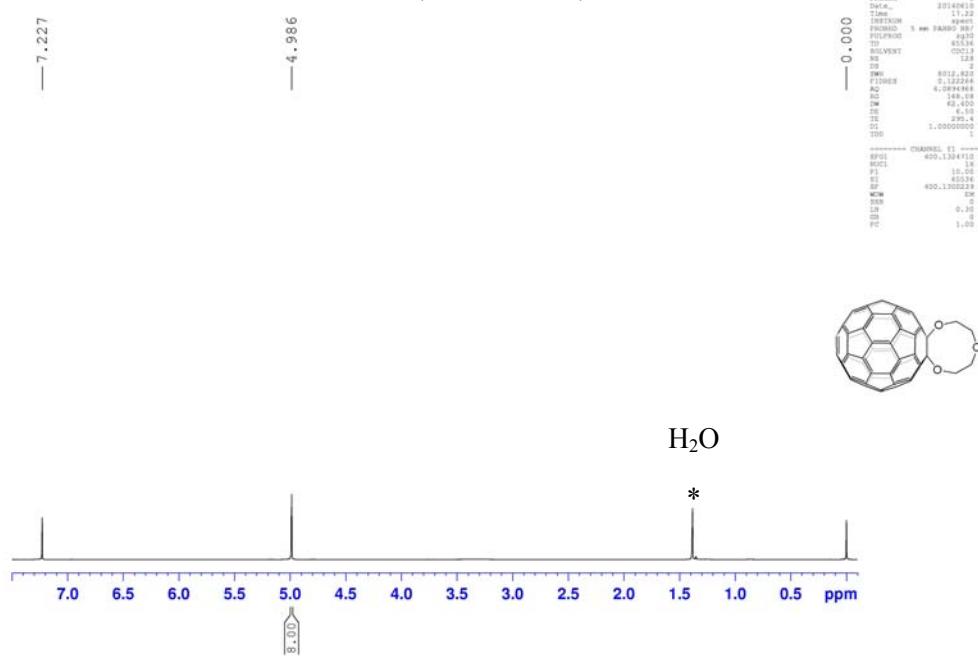
¹³C NMR (100 MHz, CS₂/CDCl₃) of 2g

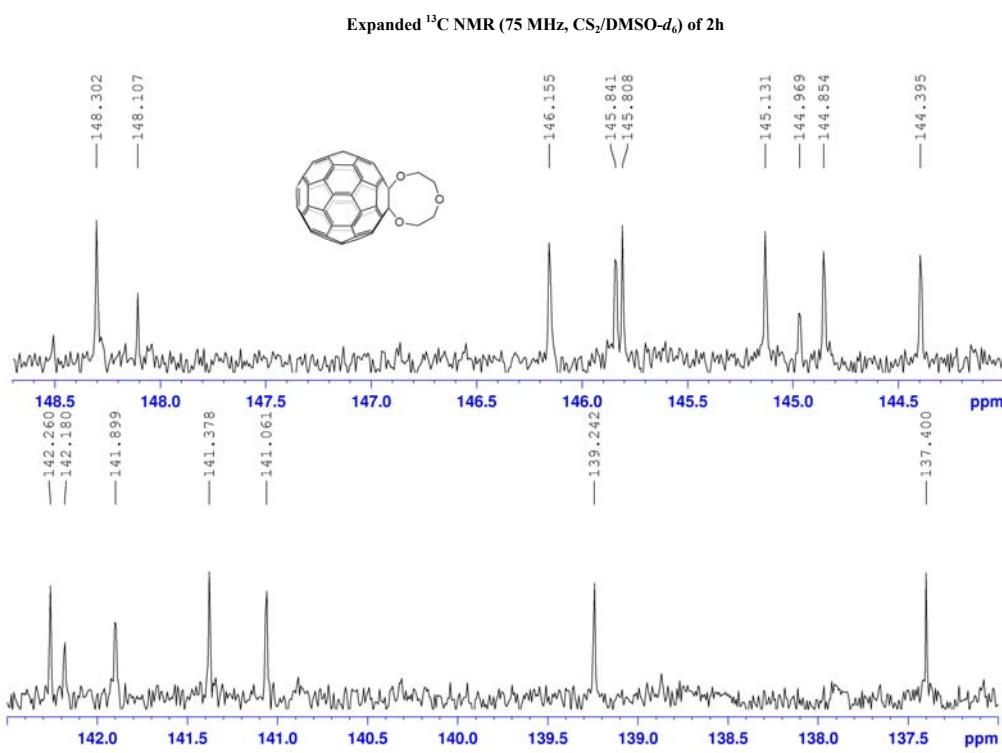
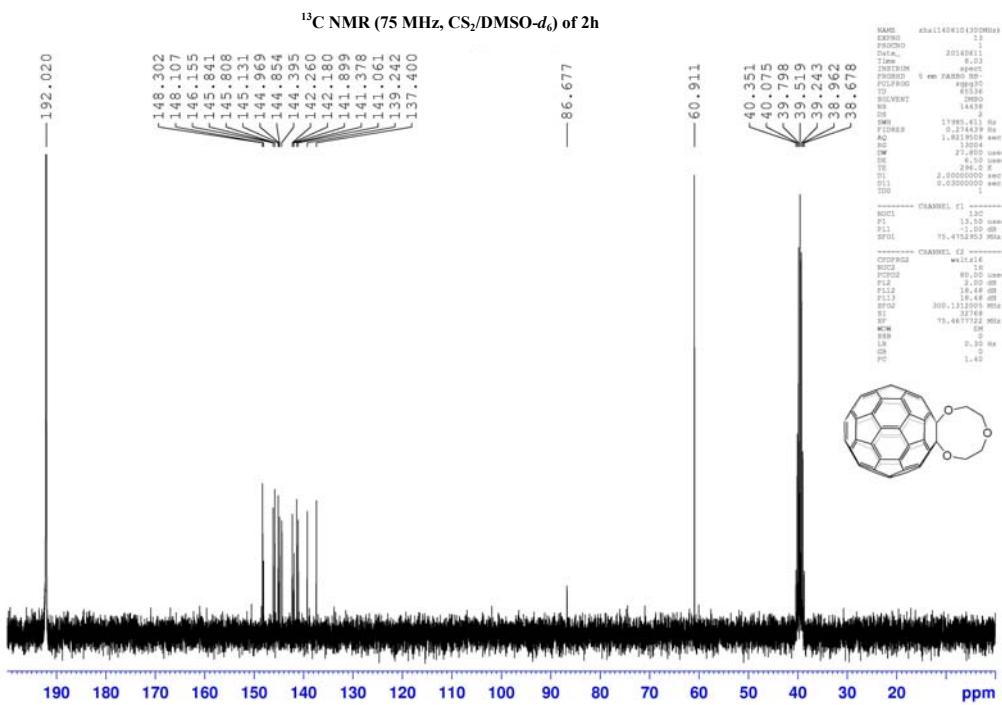


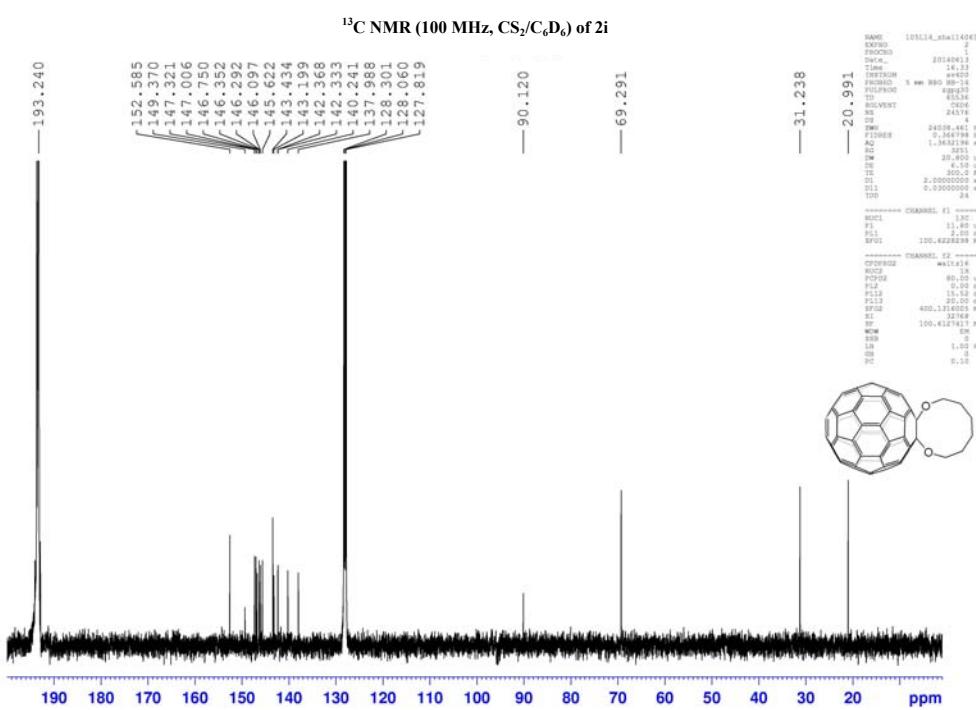
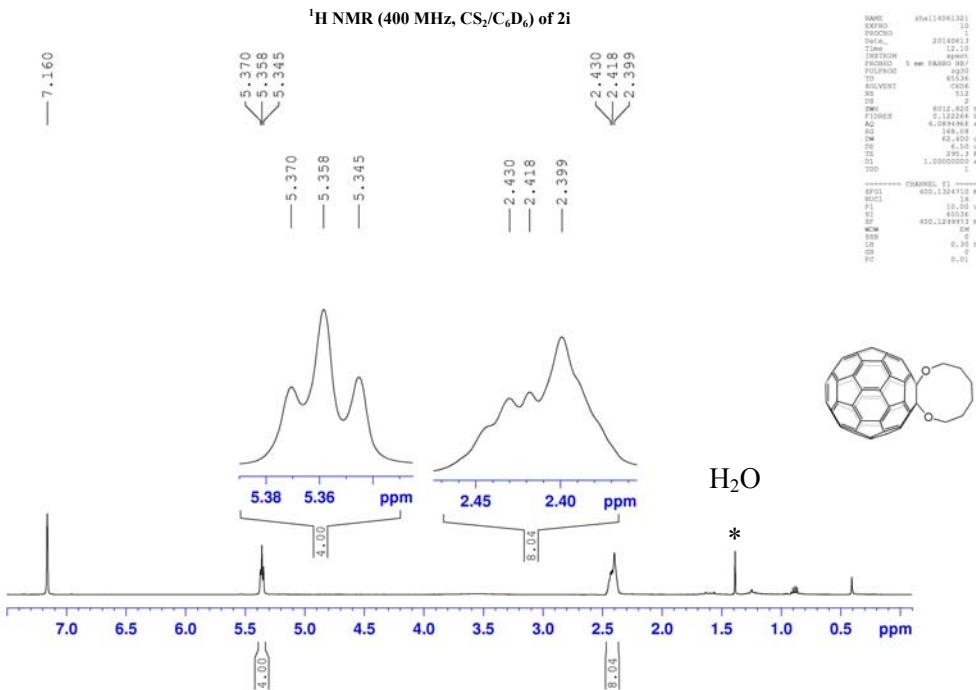
Expanded ^{13}C NMR (100 MHz, $\text{CS}_2/\text{CDCl}_3$) of 2g



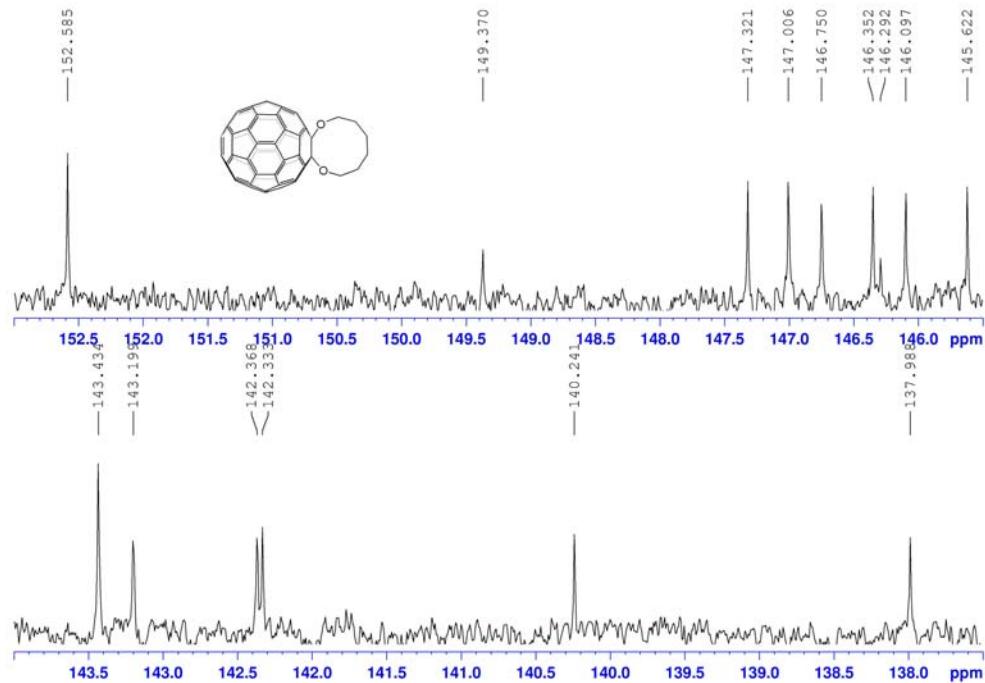
^1H NMR (400 MHz, $\text{CS}_2/\text{CDCl}_3$) of 2h



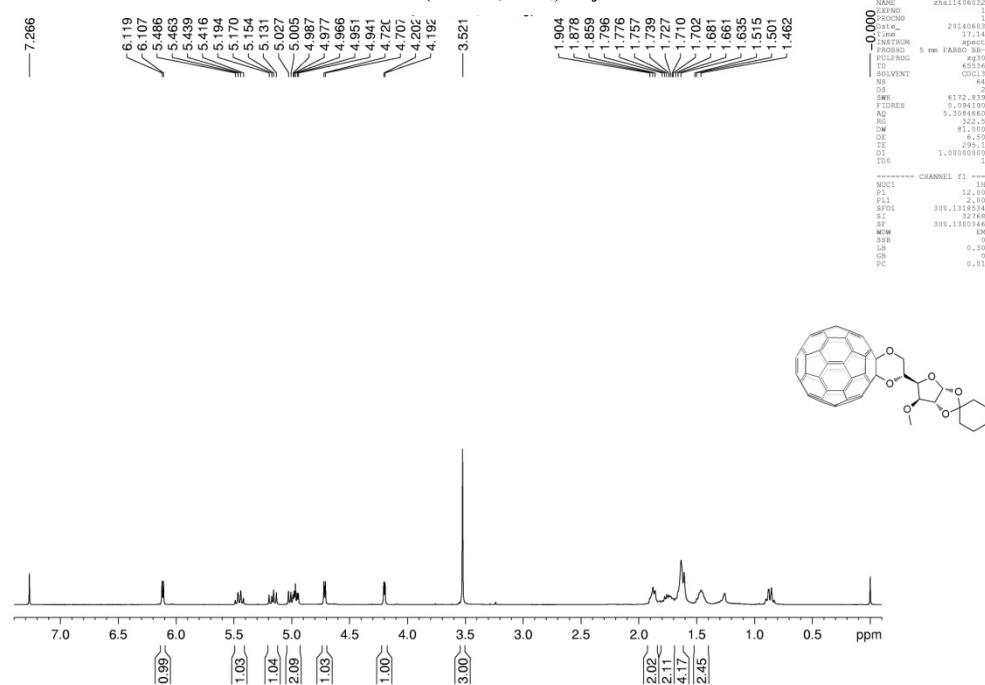




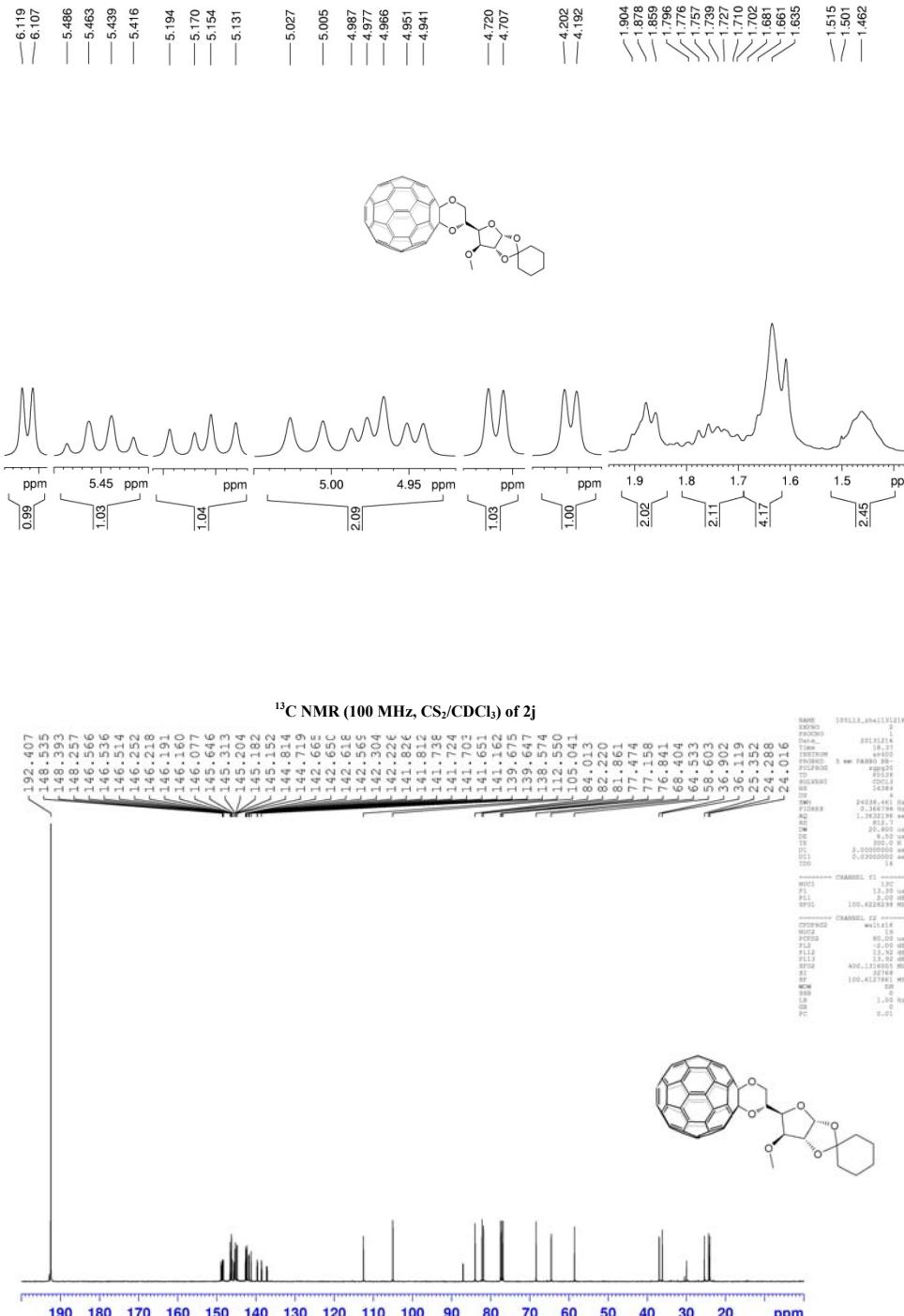
Expanded ^{13}C NMR (100 MHz, $\text{CS}_2/\text{C}_6\text{D}_6$) of 2i

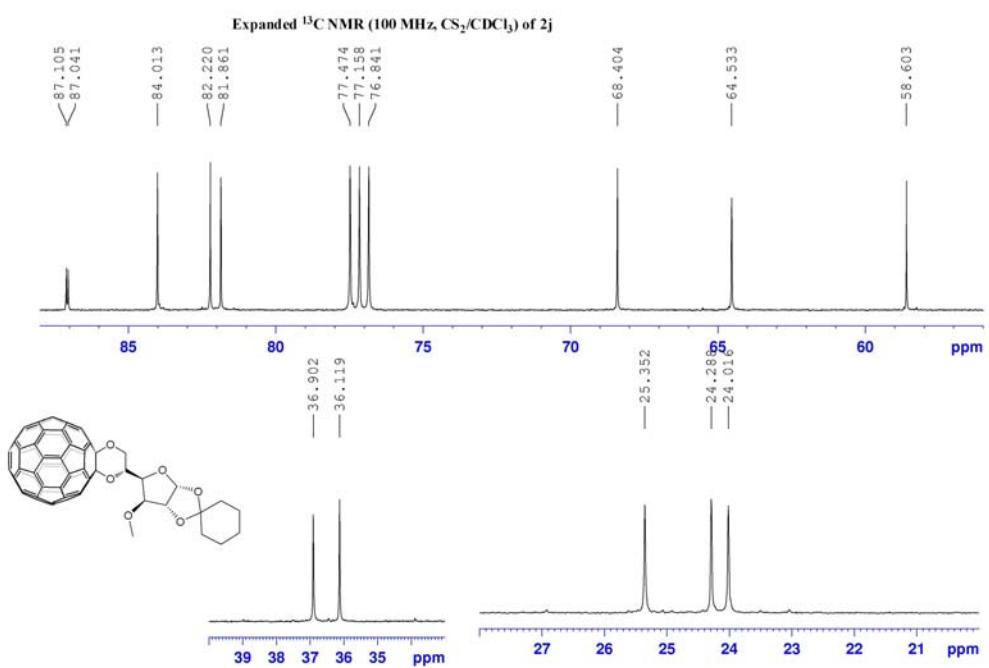
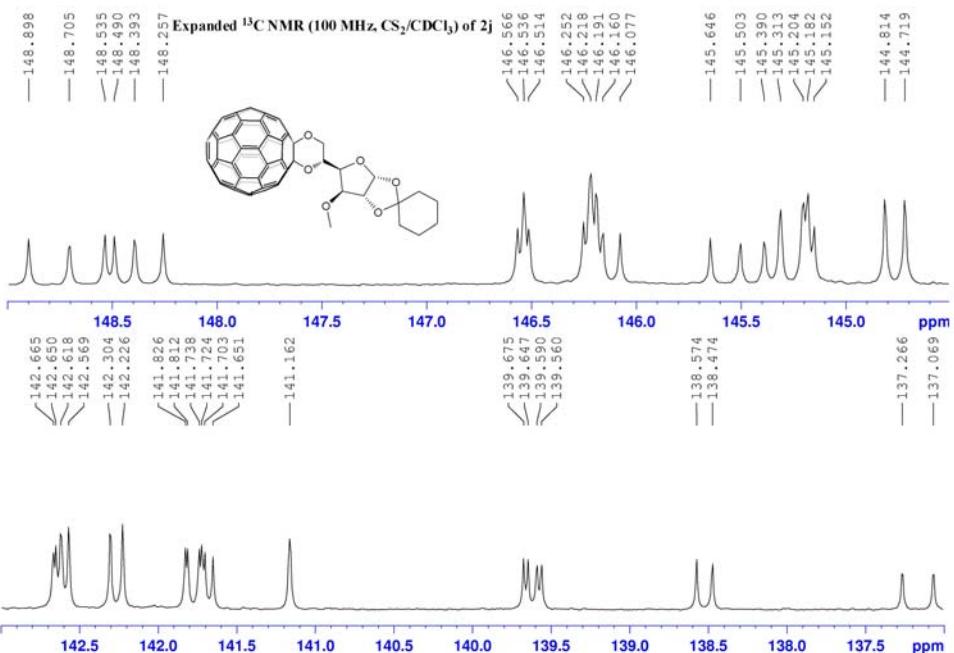


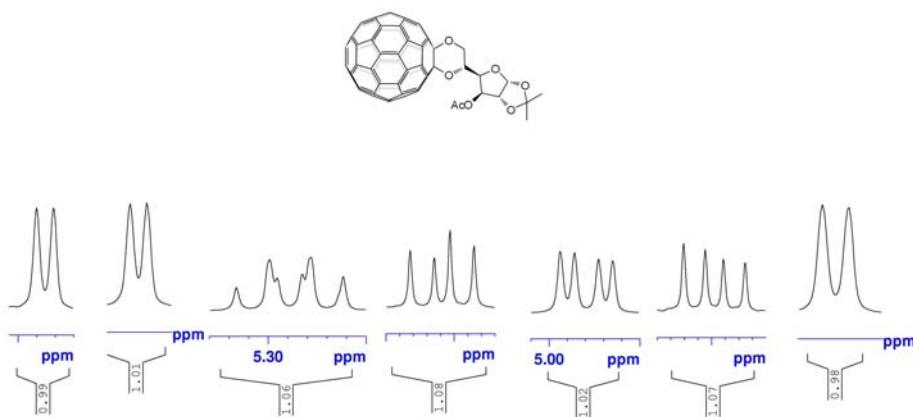
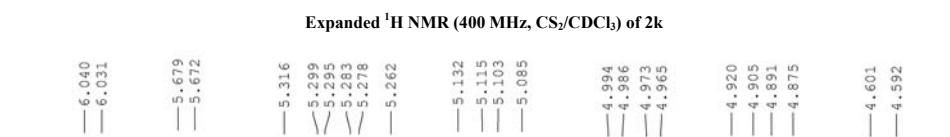
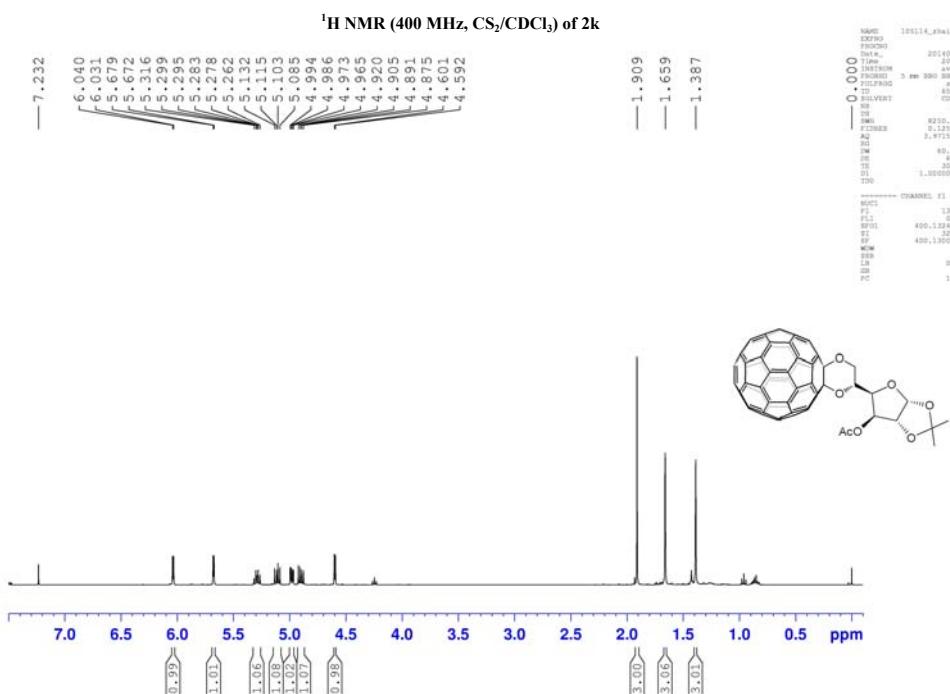
^1H NMR (300 MHz, CDCl_3) of 2j

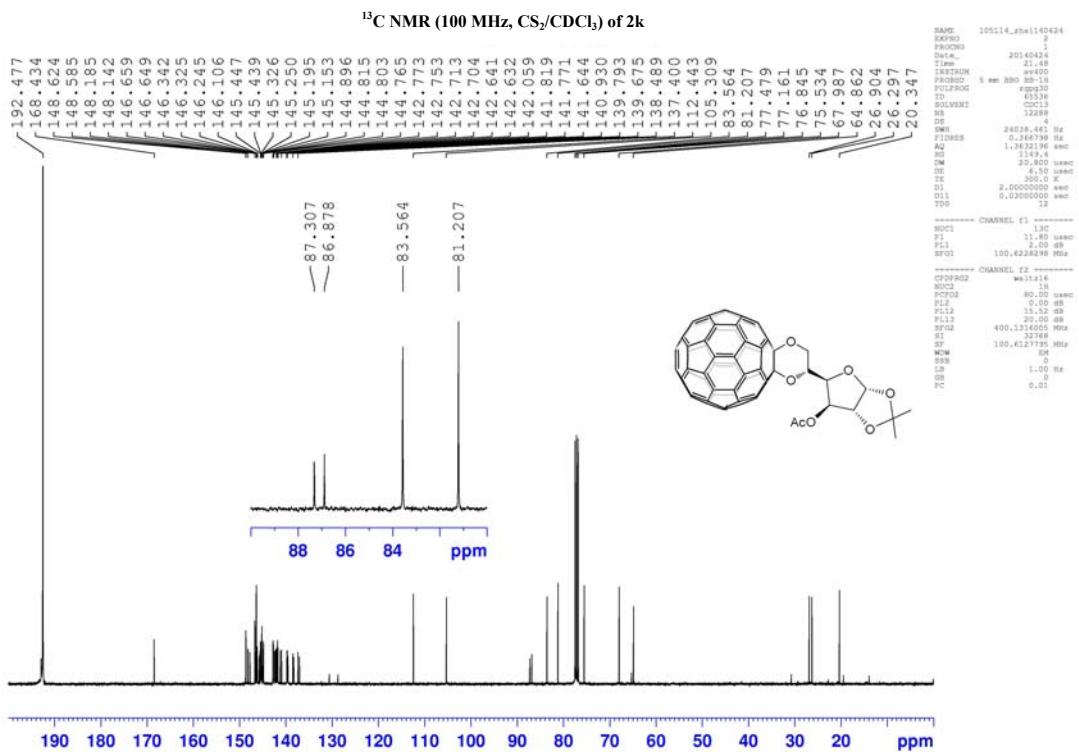


Expanded ^1H NMR (300 MHz, CDCl_3) of 2j

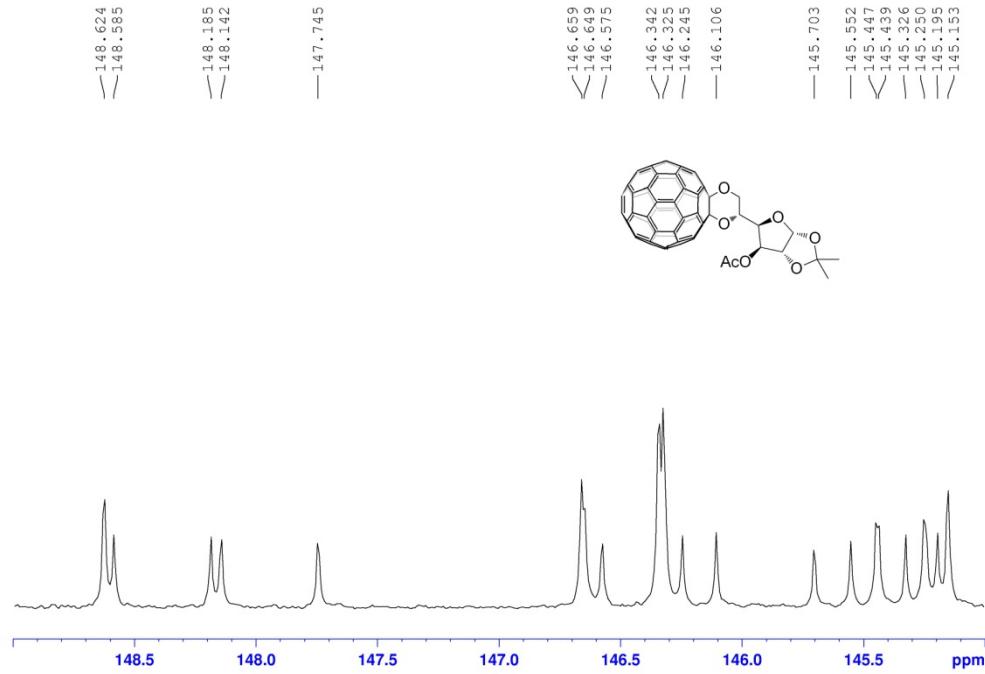




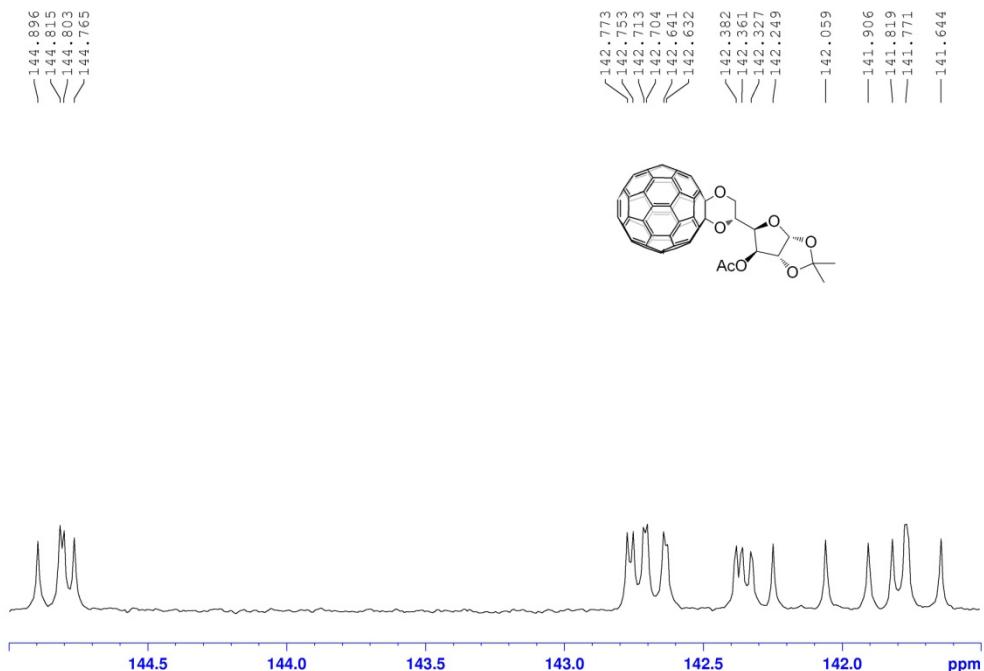




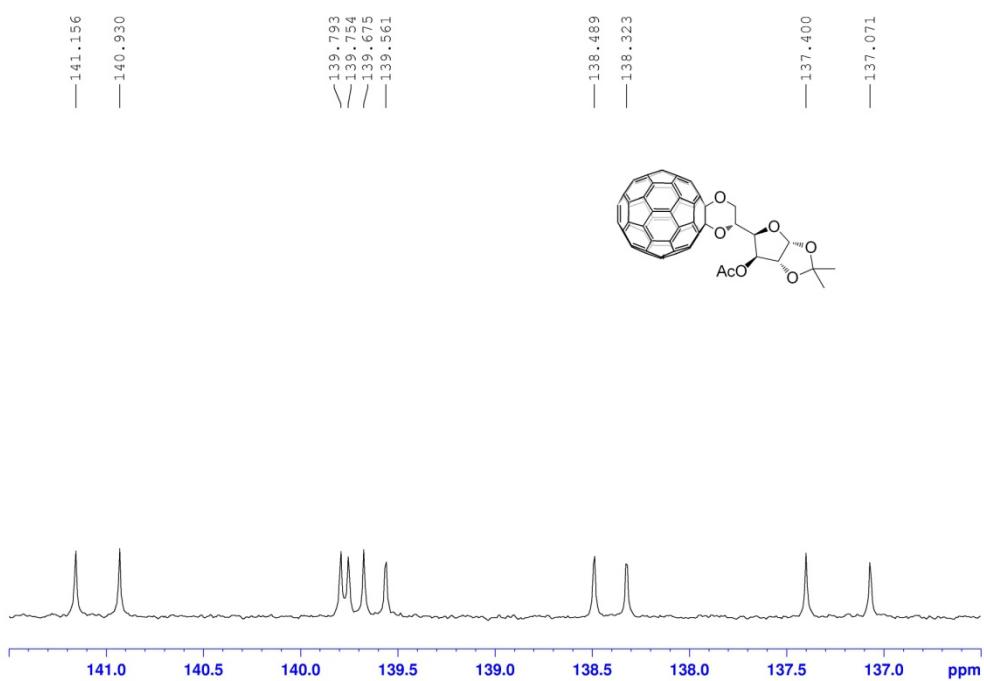
Expanded ¹³C NMR (100 MHz, CS₂/CDCl₃) of 2k

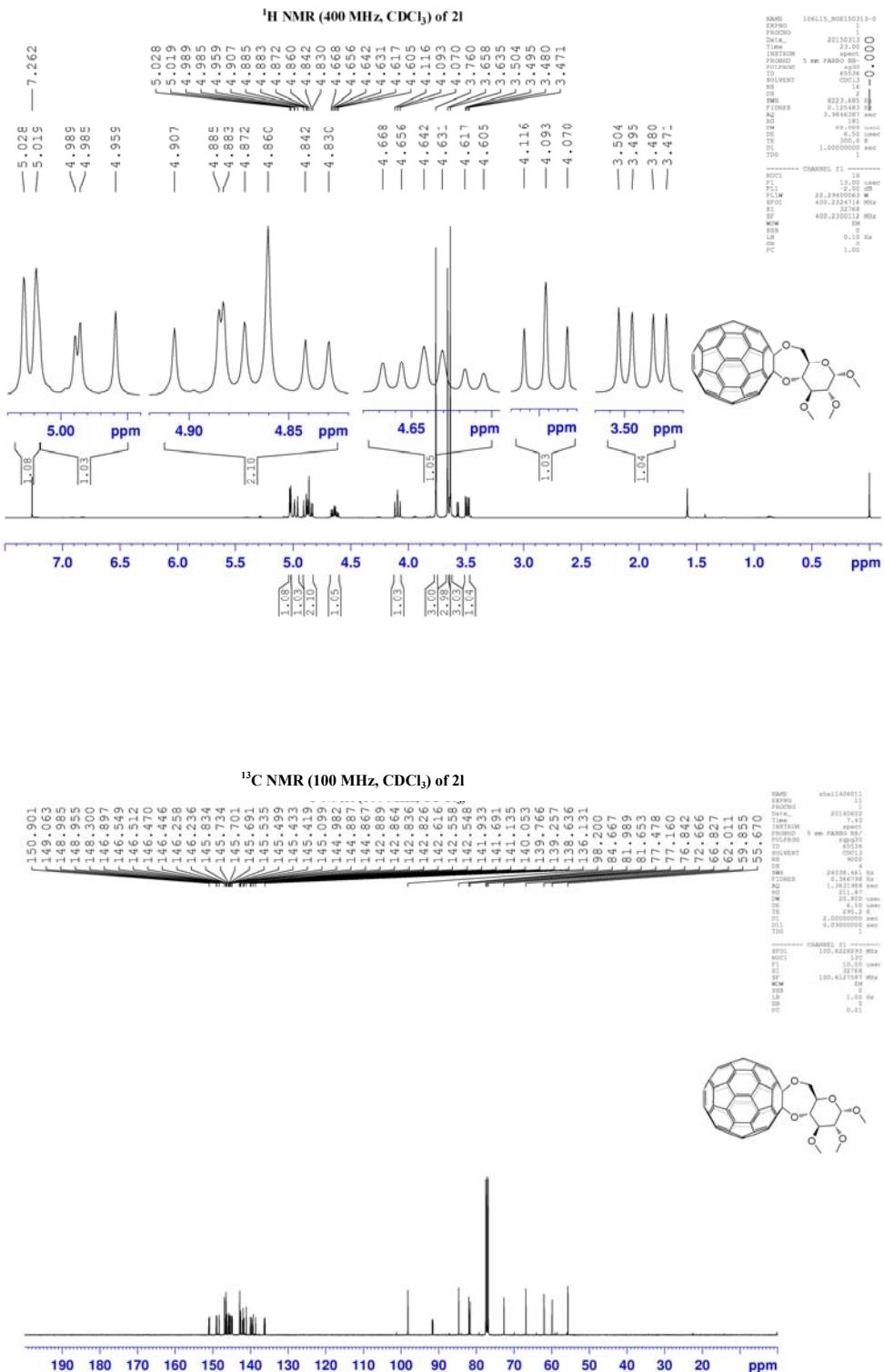


Expanded ^{13}C NMR (100 MHz, $\text{CS}_2/\text{CDCl}_3$) of 2k

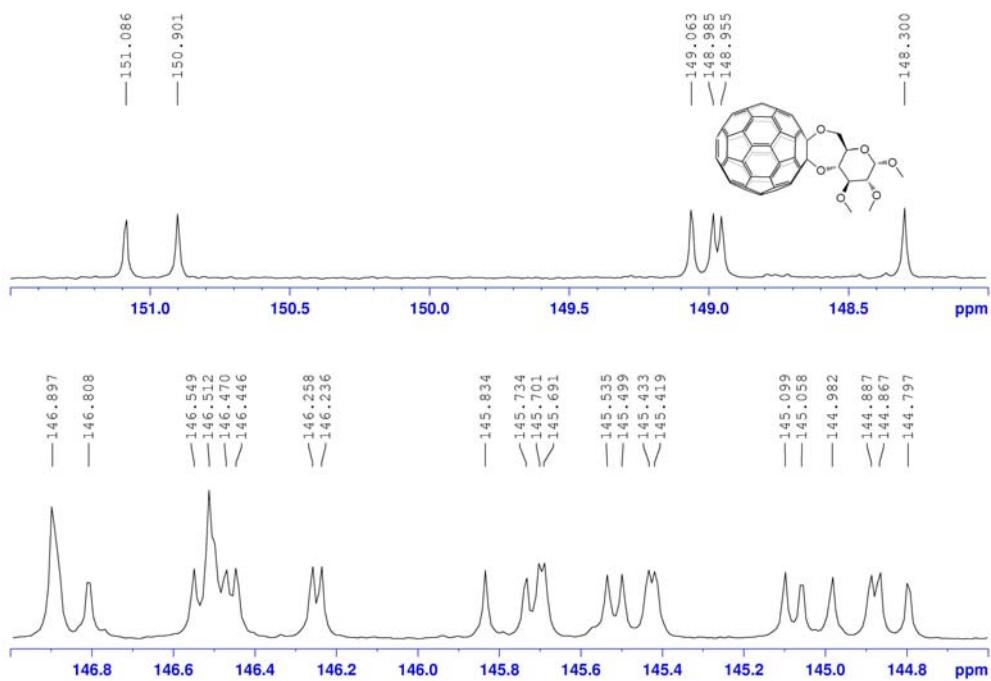


Expanded ^{13}C NMR (100 MHz, $\text{CS}_2/\text{CDCl}_3$) of 2k

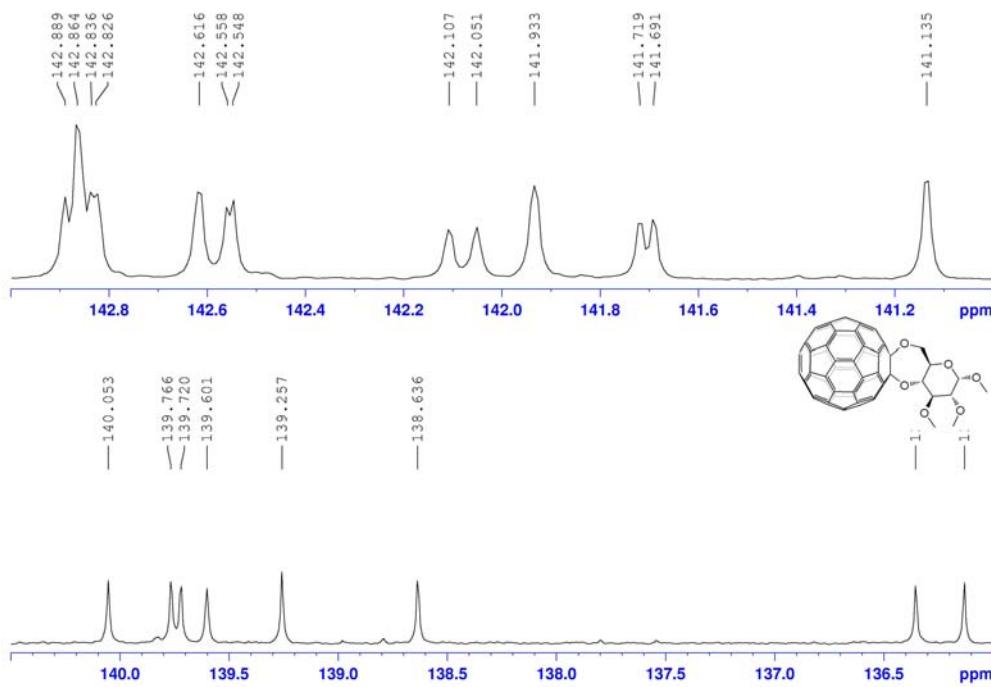




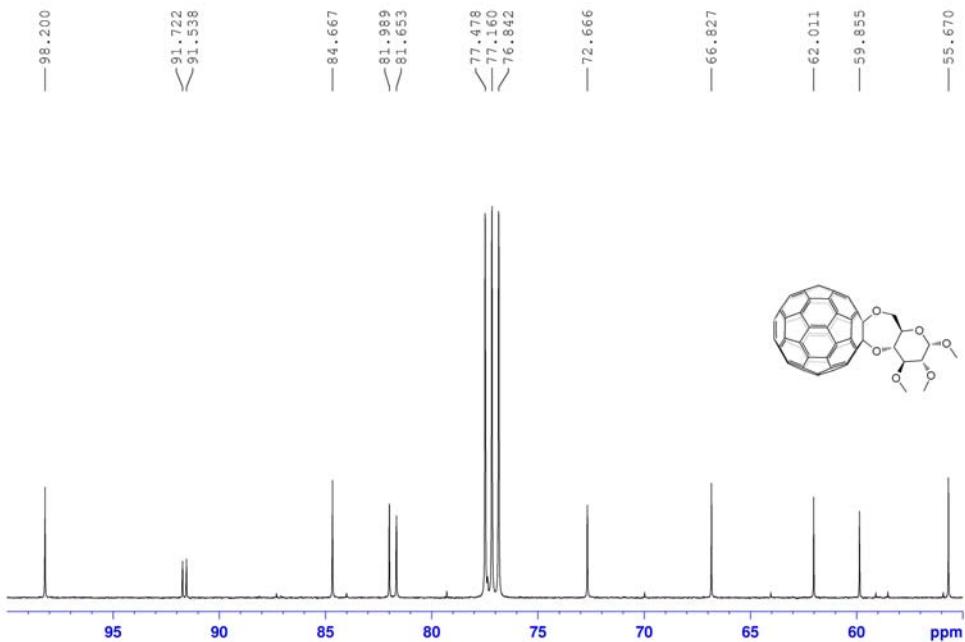
Expanded ^{13}C NMR (100 MHz, CDCl_3) of 2l



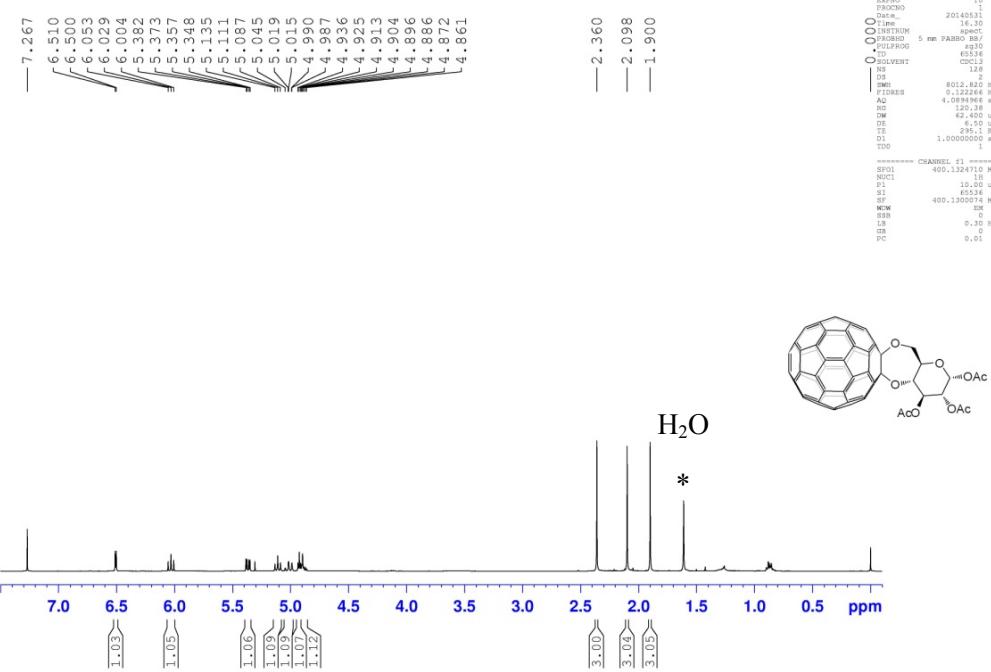
Expanded ^{13}C NMR (100 MHz, CDCl_3) of 2l



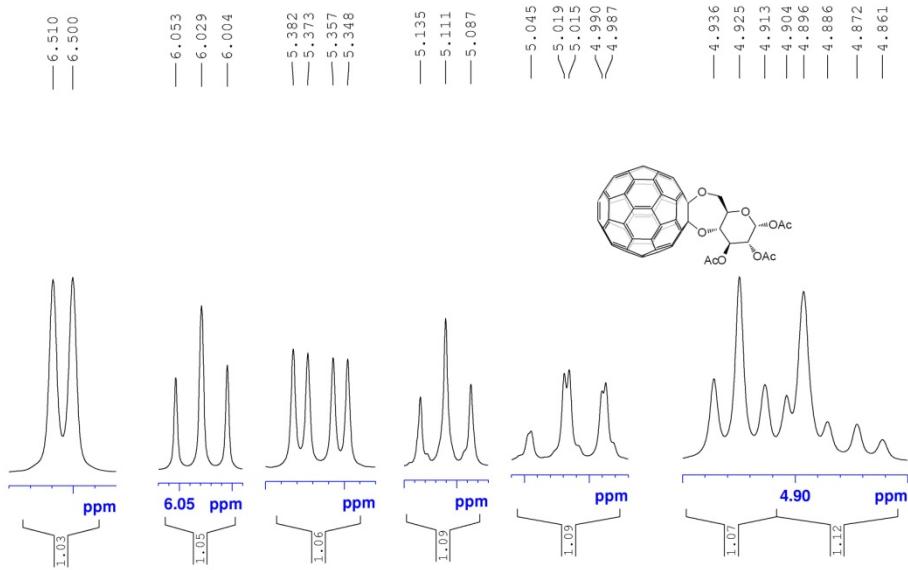
Expanded ^{13}C NMR (100 MHz, CDCl_3) of 2l



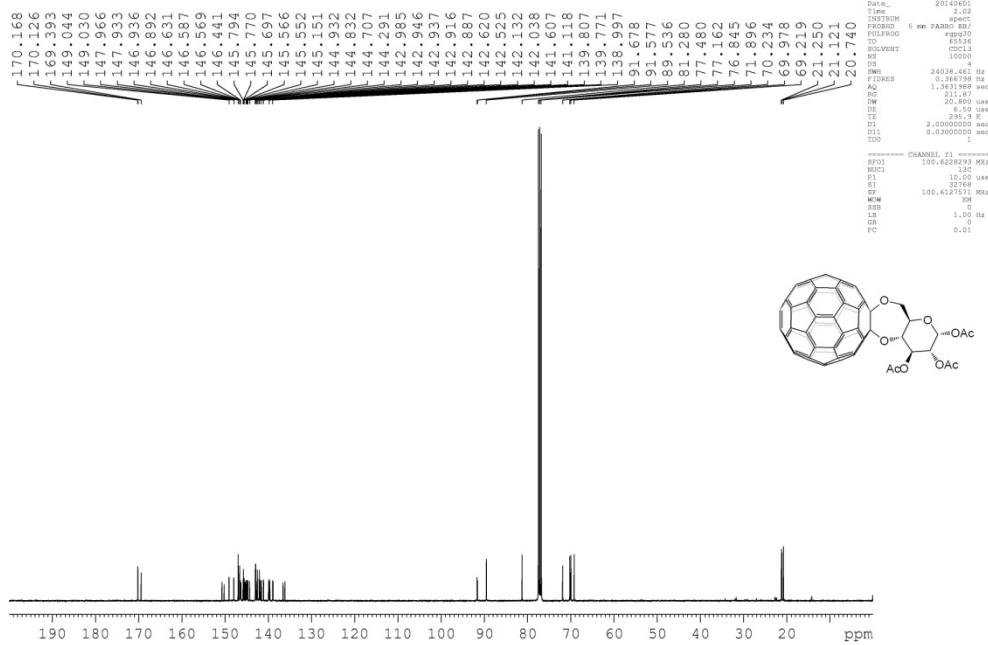
^1H NMR (400 MHz, CDCl_3) of 2m

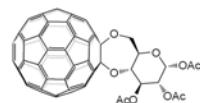
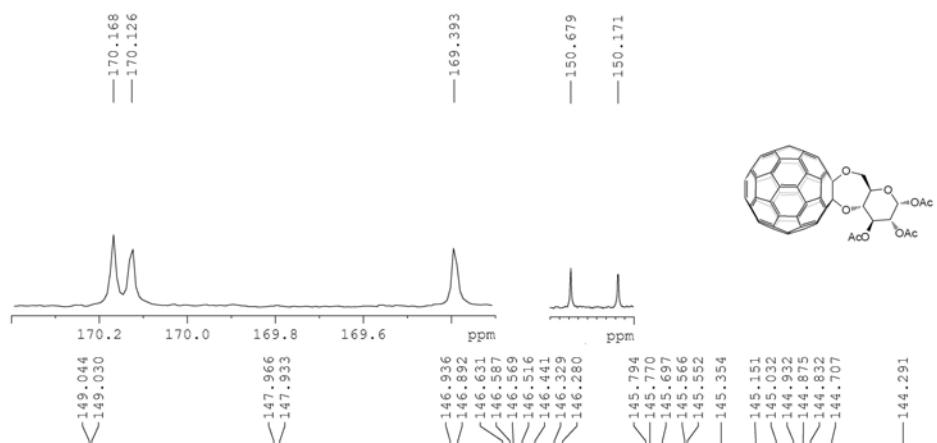
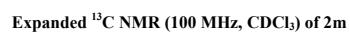


Expanded ^1H NMR (400 MHz, CDCl_3) of 2m

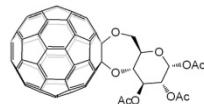
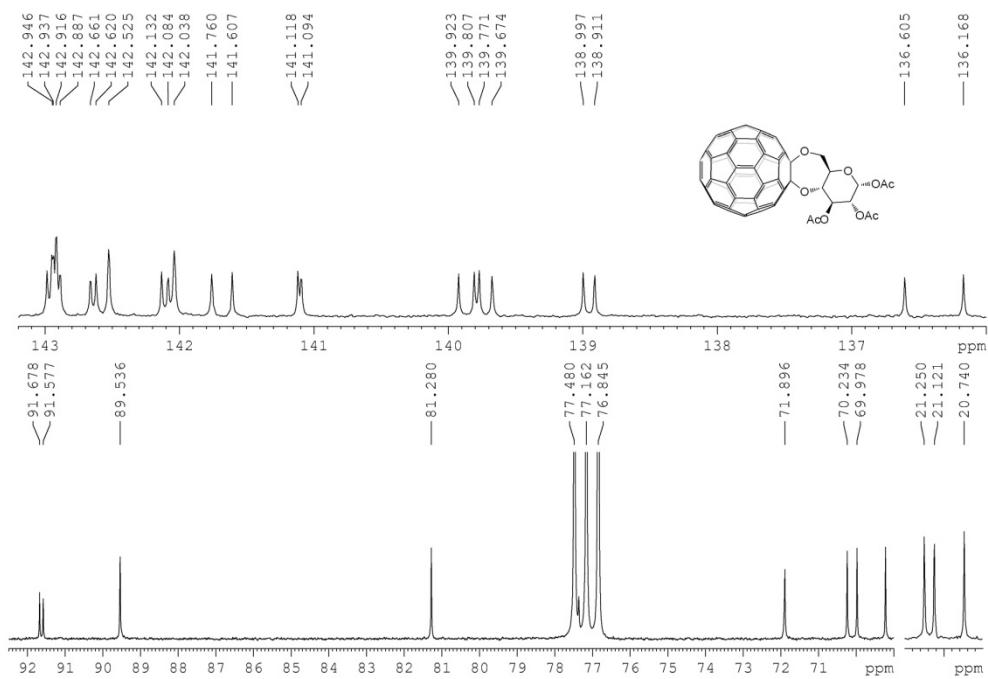


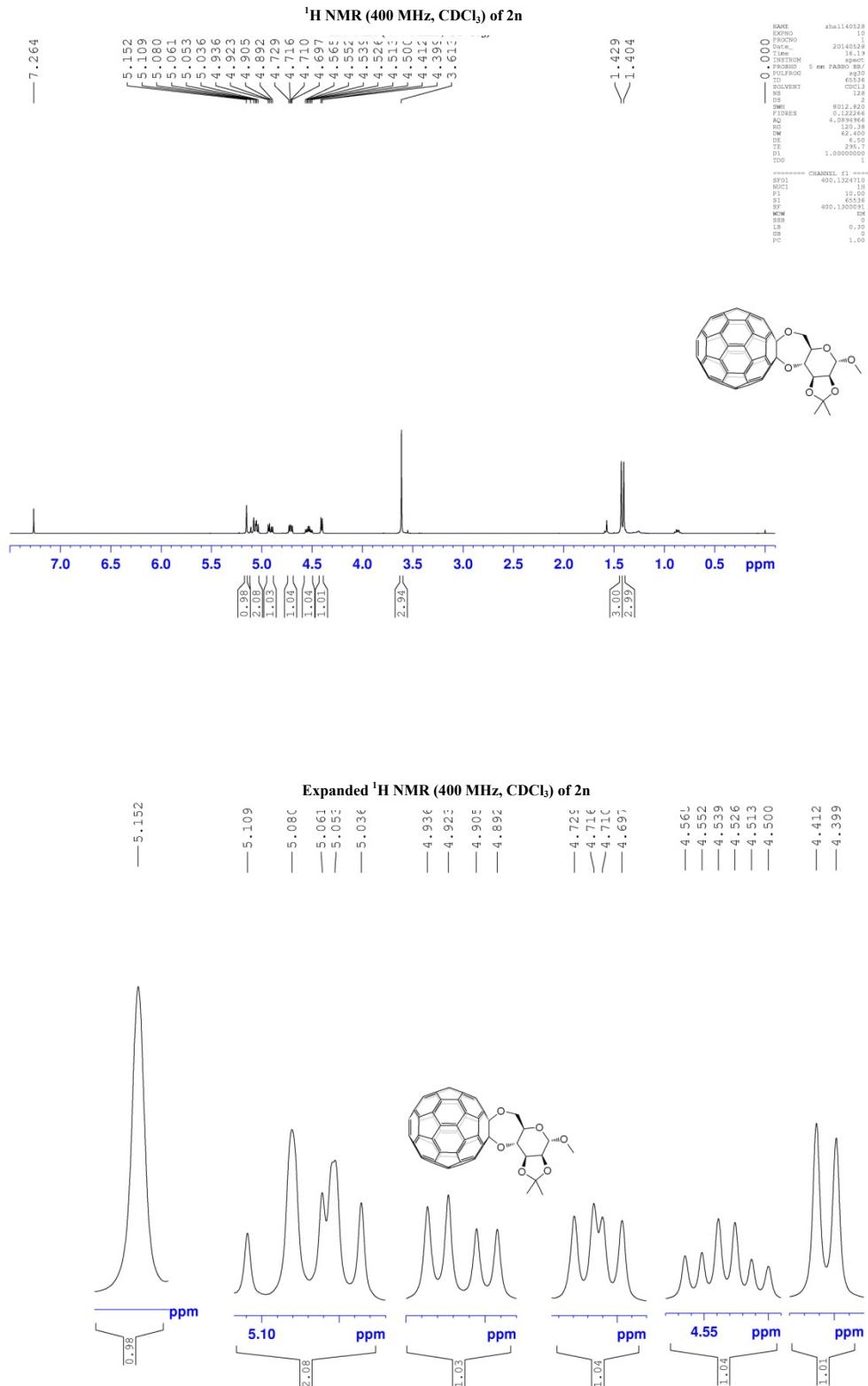
¹³C NMR (100 MHz, CDCl₃) of 2m

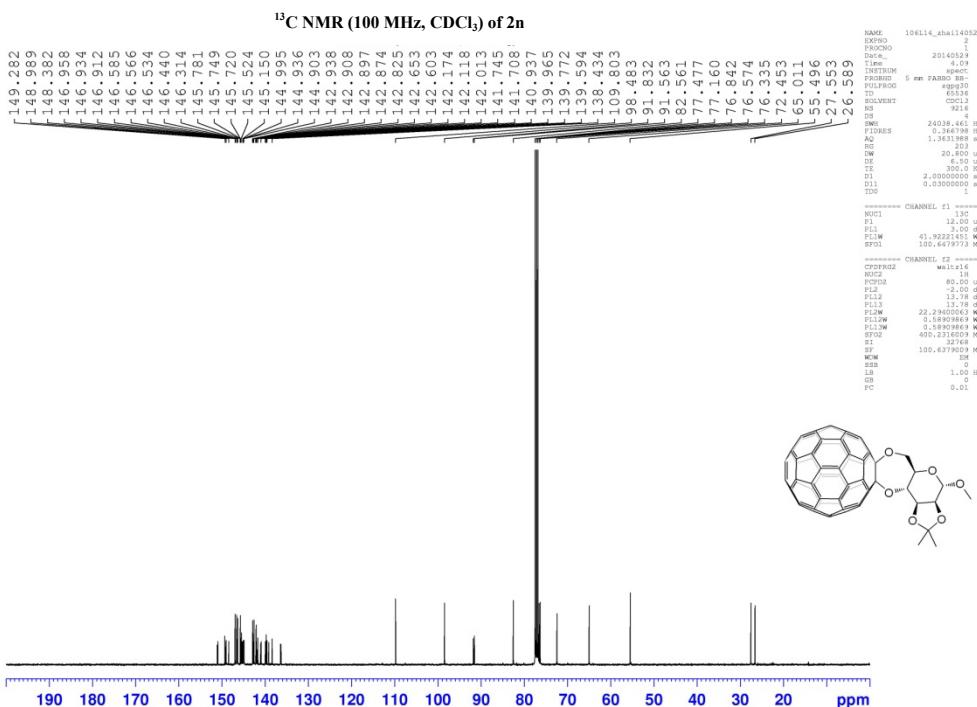




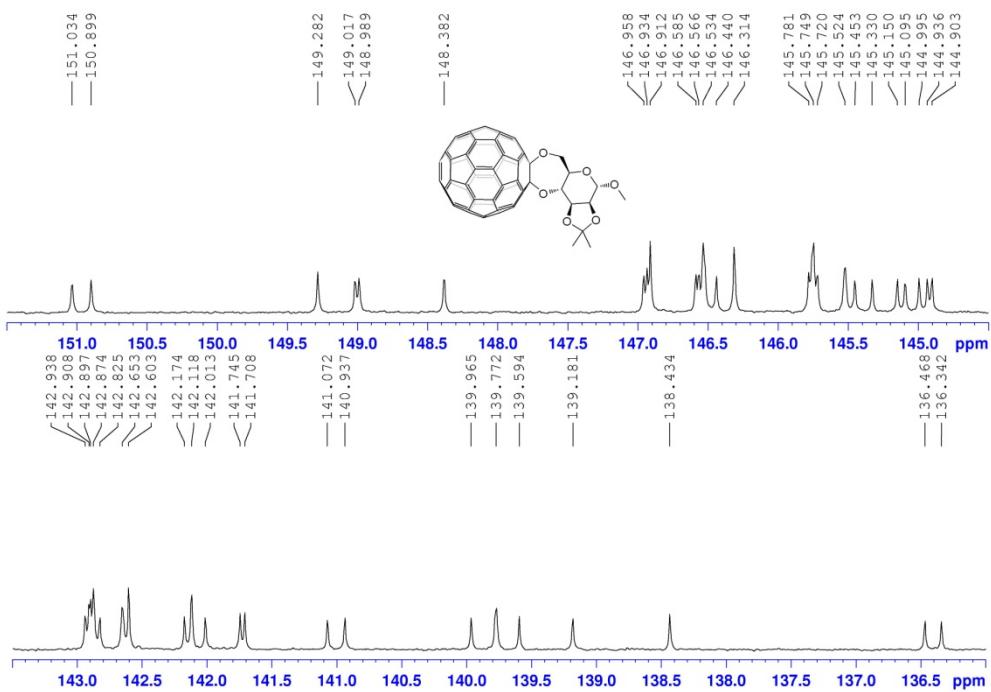
Expanded ^{13}C NMR (100 MHz, CDCl_3) of 2m



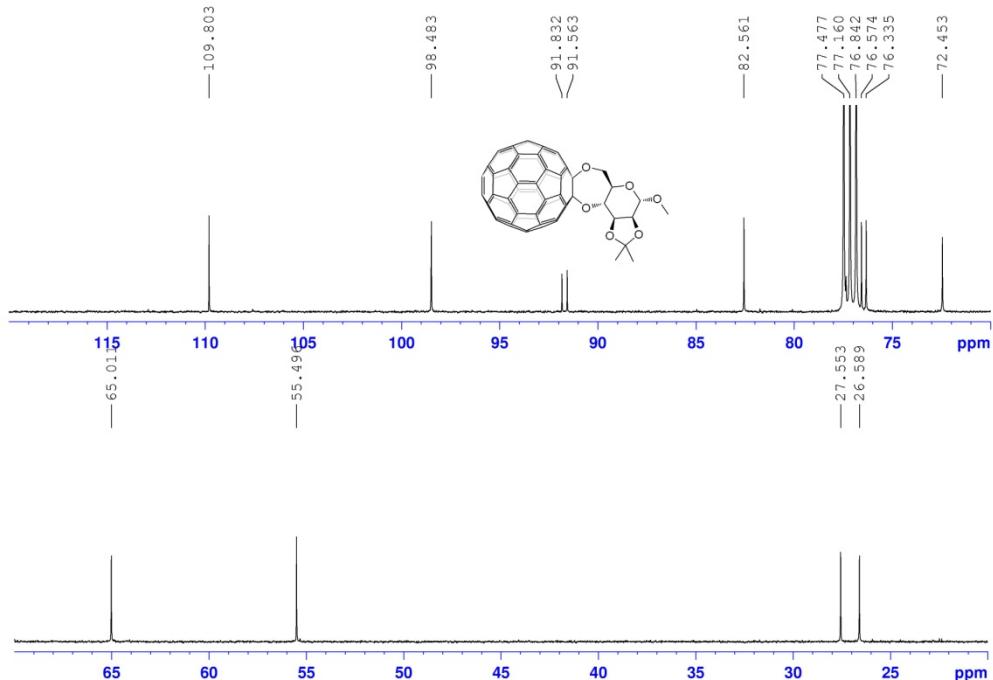




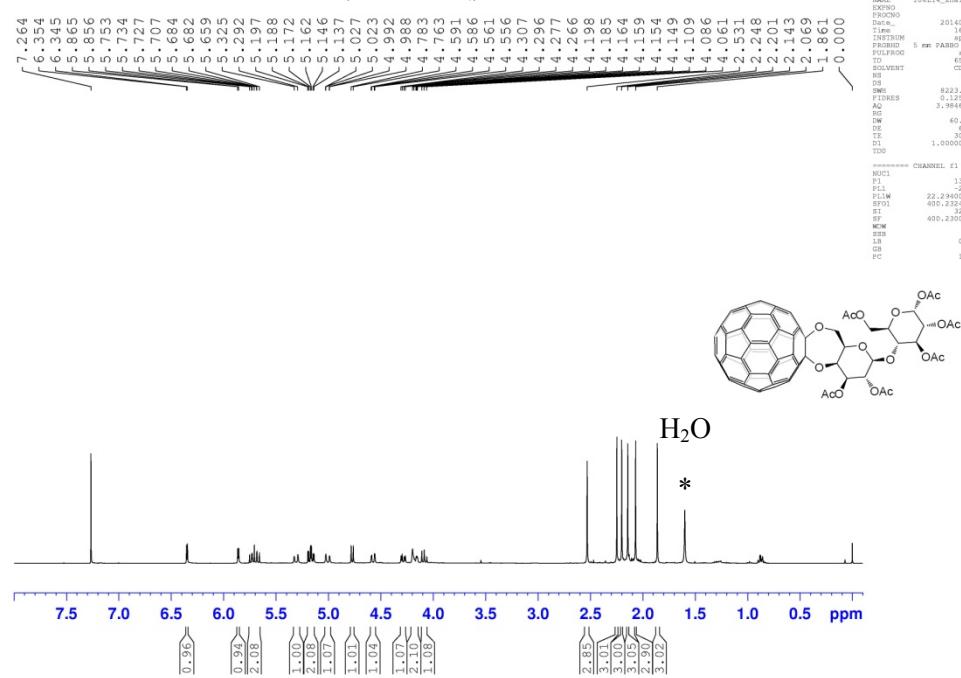
Expanded ^{13}C NMR (100 MHz, CDCl_3) of 2n

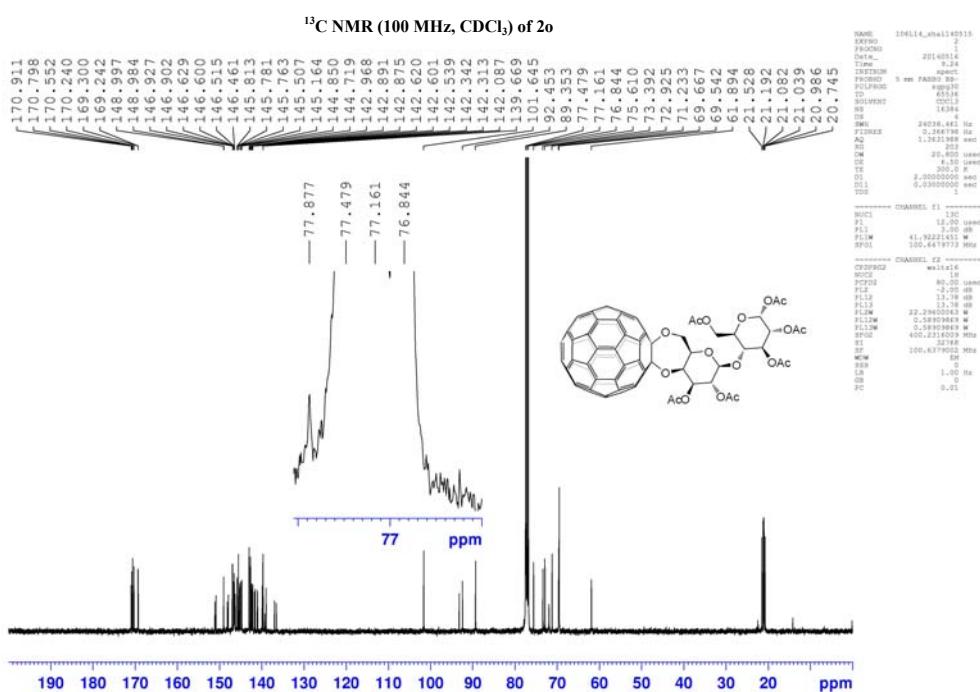
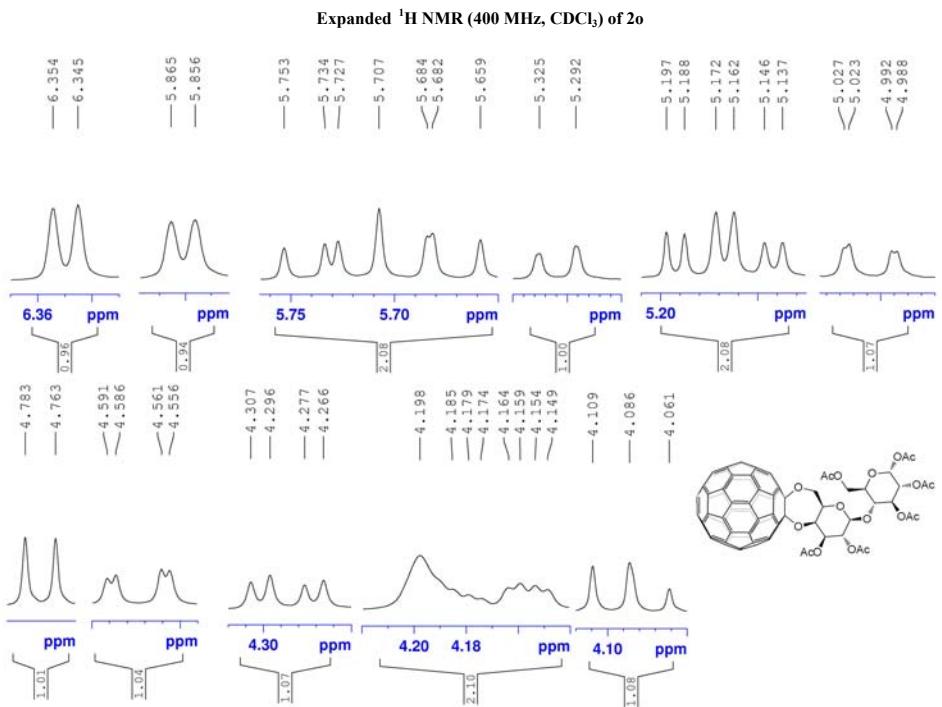


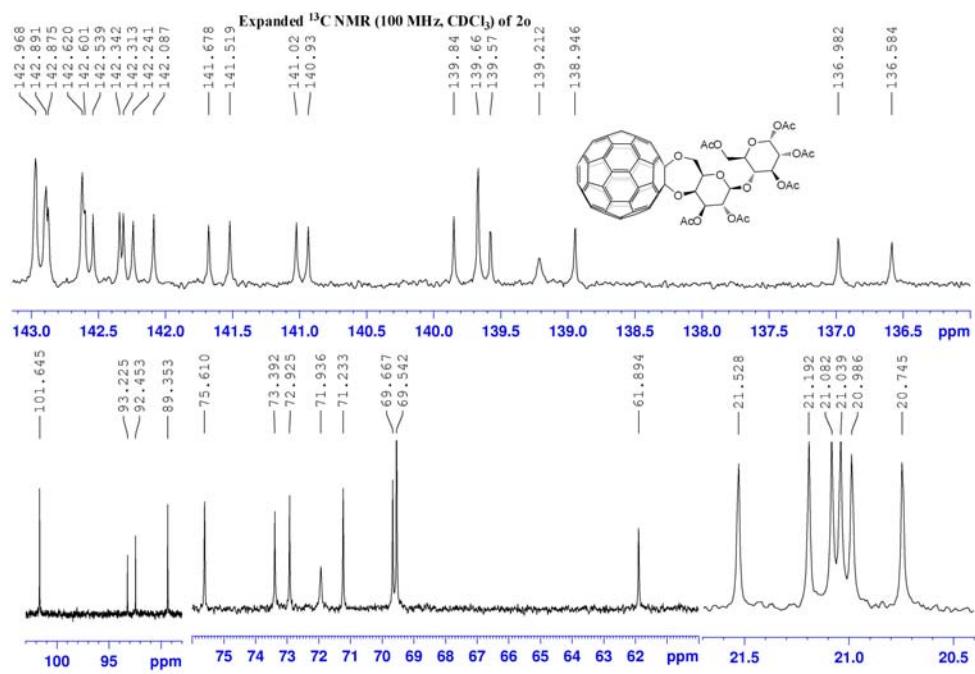
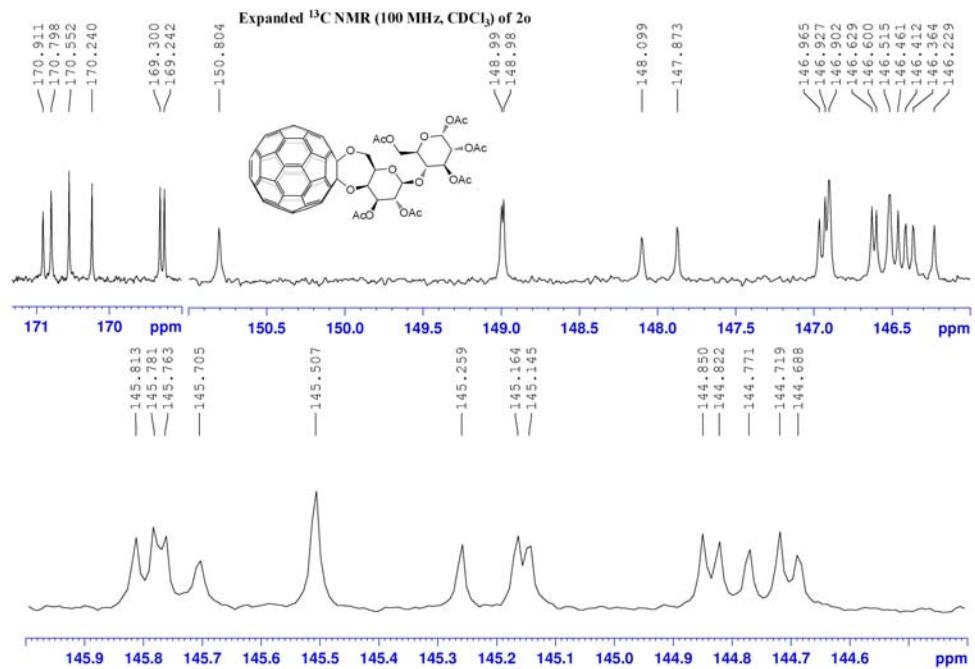
Expanded ^{13}C NMR (100 MHz, CDCl_3) of 2n

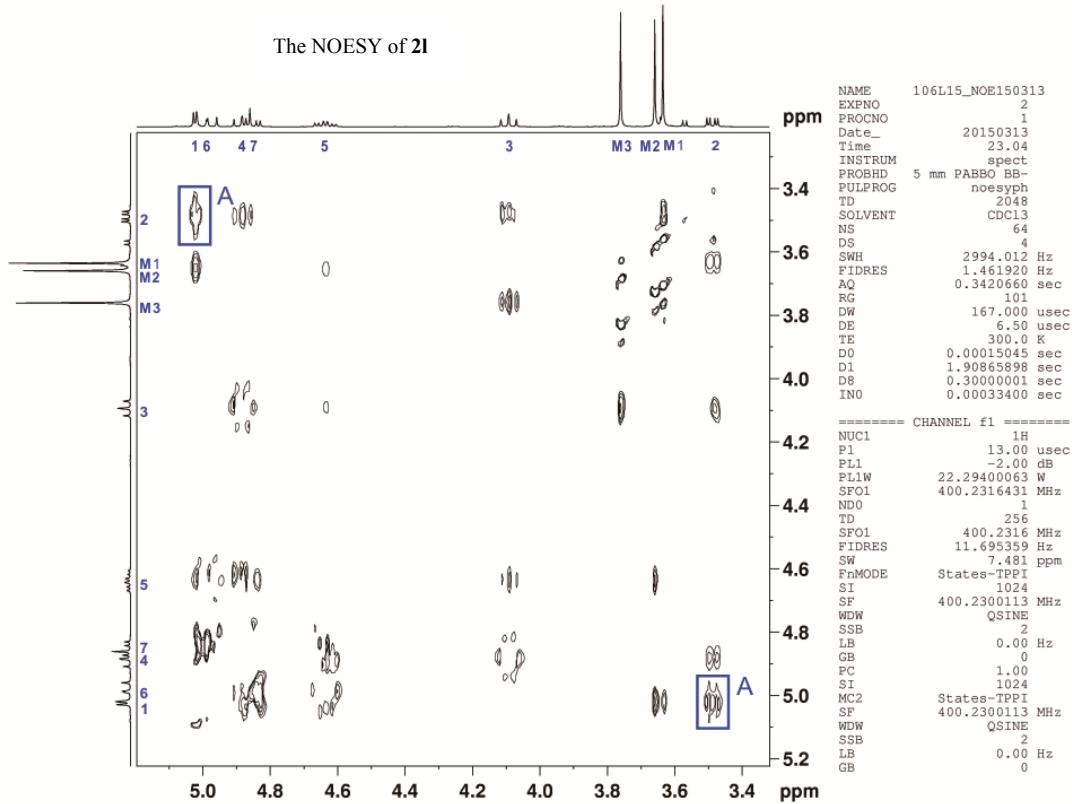


^1H NMR (400 MHz, CDCl_3) of 2o

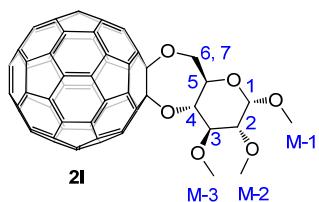








As a representative example, the NOESY spectrum of **2I** was employed to confirm the stereochemistry. The chemical shift and coupling constant of H-1 at 5.02 ppm ($d, J = 3.6$ Hz, 1H) agreed with the configuration of the α -isomer, and the cross peak A in its NOESY spectrum was also consistent with the α -isomer. For the β -isomer, the coupling constant of H-1 is generally in 7-9 Hz (Michigami, K.; Hayashi, M. *Tetrahedron* **2013**, *69*, 4221), which did not correlate with the ^1H NMR spectrum of **2I**. Then, the peak at 3.49 ppm ($dd, J = 9.2, 3.6$ Hz, 1H) was assigned to H-2 on the basis of coupling constant. The three methoxy groups were assigned according to a recent report (Matwiejuk, M.; Thiem, J. *Chem. Commun.* **2011**, *47*, 8379). Subsequently, the chemical shifts and stereochemistry of rest protons were assigned according to the ^1H NMR and NOESY spectra of **2I**, and were labeled in the NOESY spectrum.



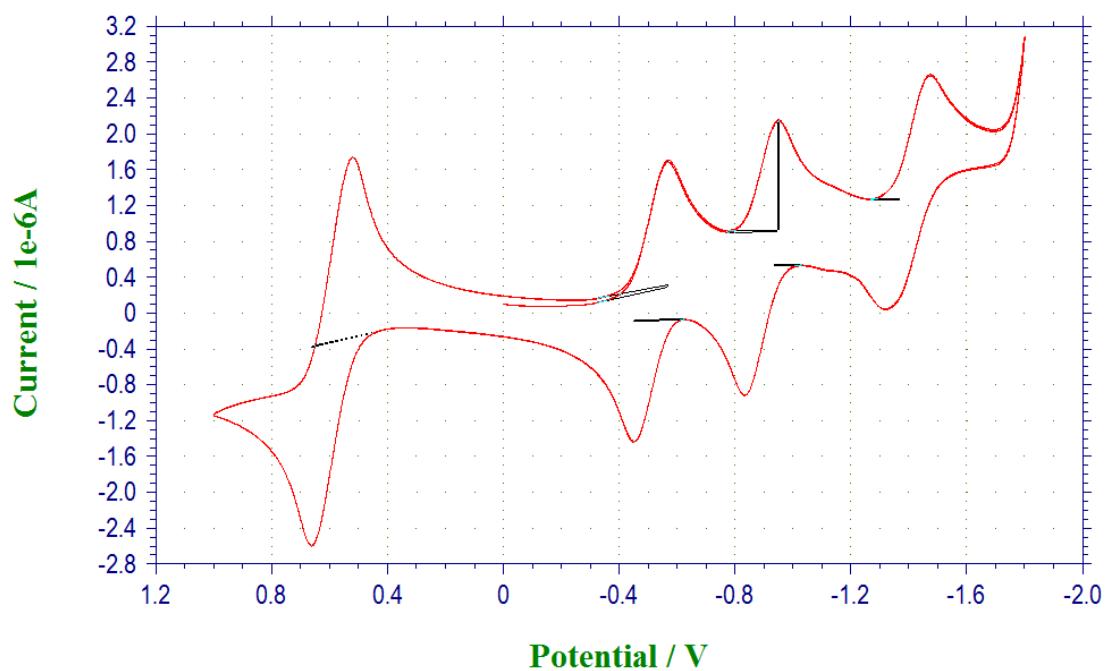
H-1, 5.02 (d, $J = 3.6$ Hz, 1H)	H-2, 3.49 (dd, $J = 9.2, 3.6$ Hz, 1H)
H-3, 4.09 (t, $J = 9.2$ Hz, 1H)	H-4, 4.88 (dd, $J = 10.4, 9.2$ Hz, 1H)
H-5, 4.64 (ddd, $J = 10.4, 10.4, 4.8$ Hz, 1H)	H-6, 4.99 (dd, $J = 12.0, 10.4$ Hz, 1H)
H-7, 4.99 (dd, $J = 12.0, 4.8$ Hz, 1H)	M-3, 3.76 (s, 3H)
M-2, 3.66 (s, 3H)	M-1, 3.64 (s, 3H)

Table S1: Half-wave Reduction Potentials of Compounds **2a–o** and C_{60}^a

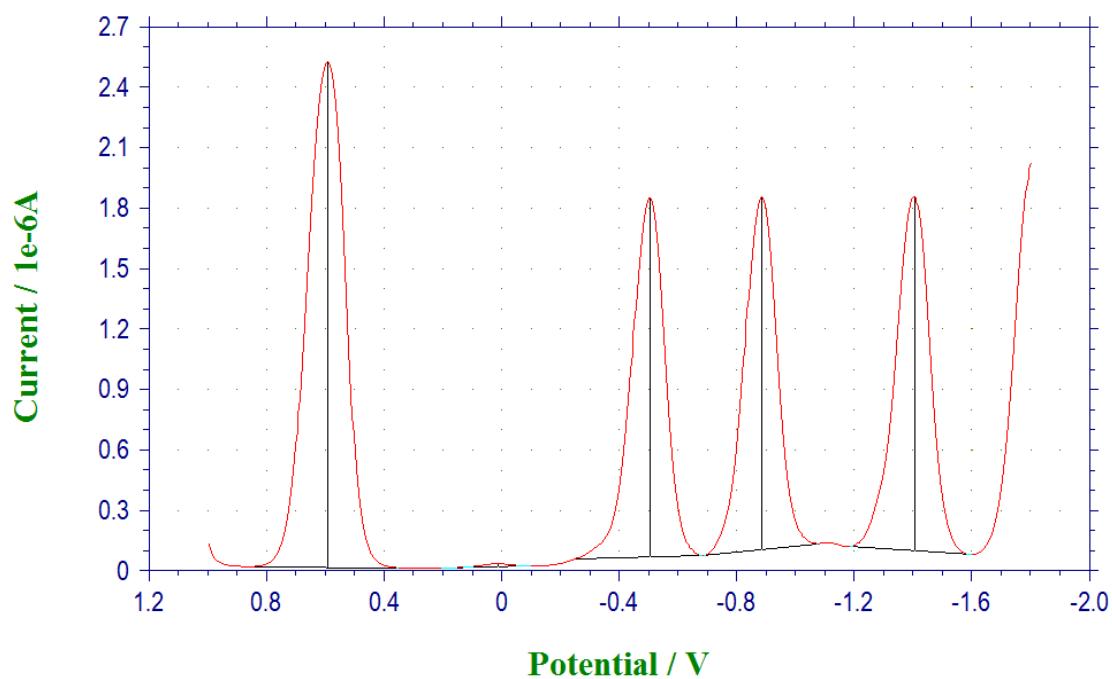
Compd	E_1	E_2	E_3
C_{60}	-1.080	-1.473	-1.934
2a	-1.102	-1.484	-1.990
2b	-1.119	-1.501	-2.032
2c	-1.122	-1.505	-2.036
2d	-1.121	-1.508	-2.043
2e	-1.139	-1.529	-2.075
2f	-1.133	-1.513	-2.048
2g^b	-1.135	-1.518	-2.054
2h	-1.106	-1.492	-2.014
2i^b	-1.138	-1.518	-2.044
2j	-1.107	-1.510	-2.052
2k	-1.083	-1.473	-1.977
2l	-1.109	-1.500	-2.019
2m	-1.074	-1.469	-1.981
2n	-1.106	-1.491	-2.001
2o	-1.080	-1.483	-2.024

^aPotential in V versus a ferrocene/ferrocenium couple. Experimental conditions: 0.1 mM of **2a–o**/ C_{60} and 0.1 M of $^{n}\text{Bu}_4\text{NClO}_4$ in anhydrous ODCB; reference electrode, SCE; working electrode, Pt; auxiliary electrode, Pt wire; scanning rate, 20 mV s⁻¹.

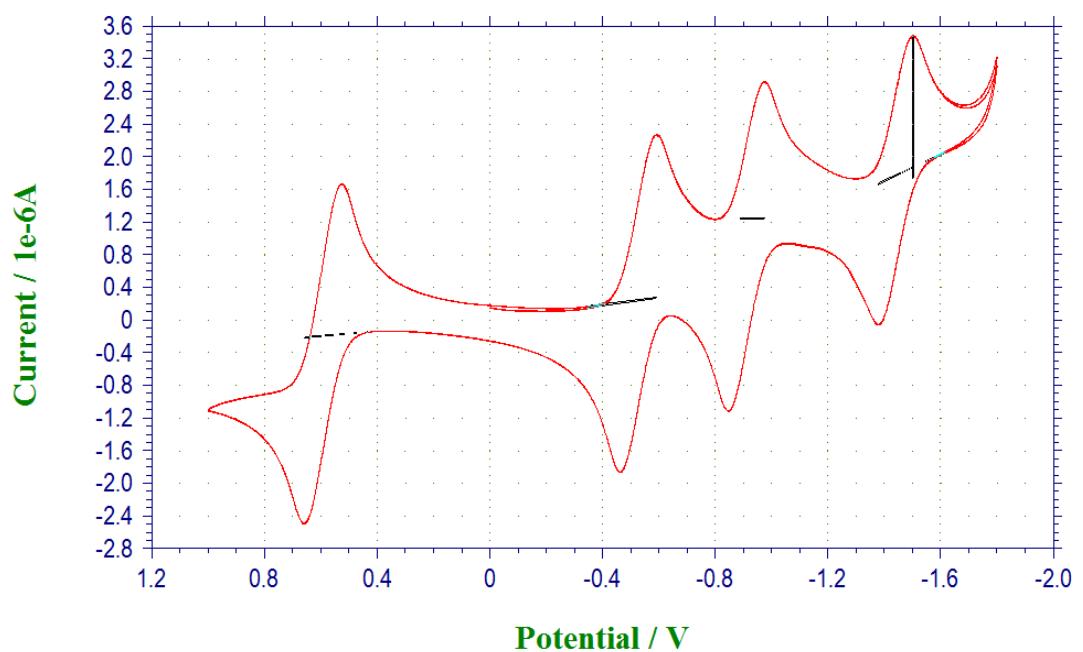
^bsaturated solution.



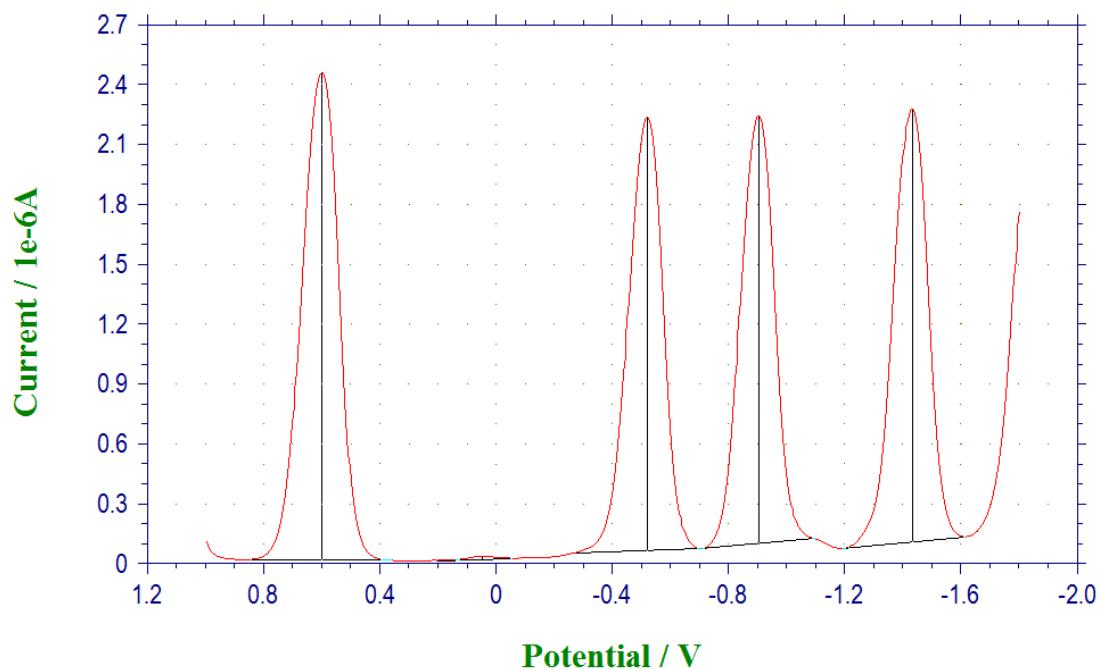
Cyclic Voltammogram of Compound **2a** (scanning rate: 20 mV s⁻¹)



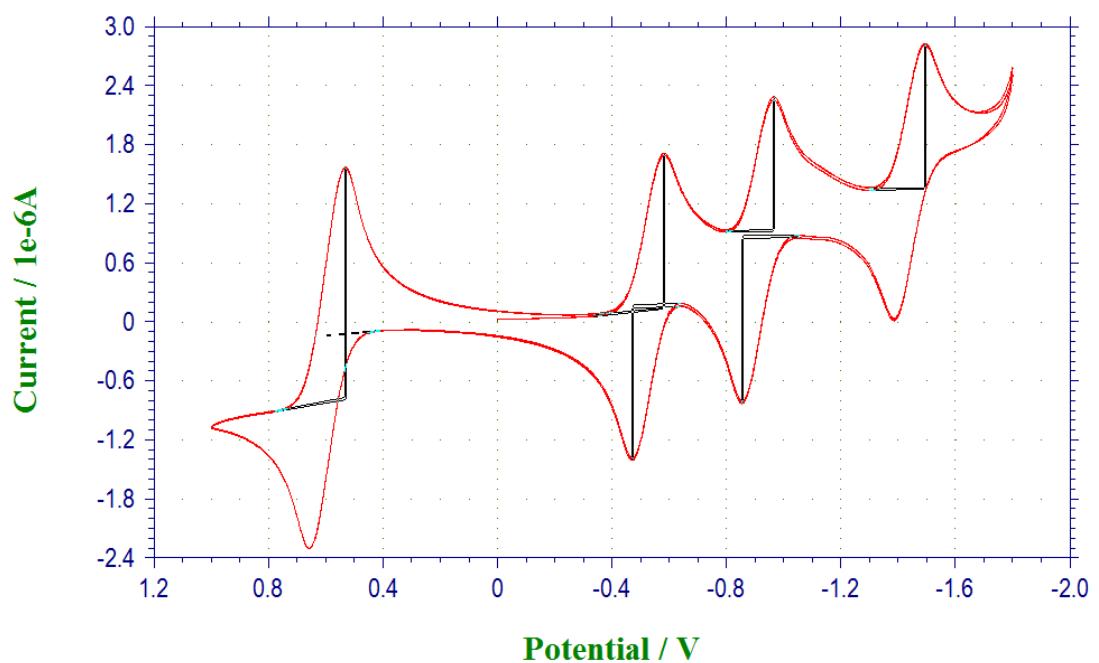
Differential Pulse Voltammogram of Compound **2a**



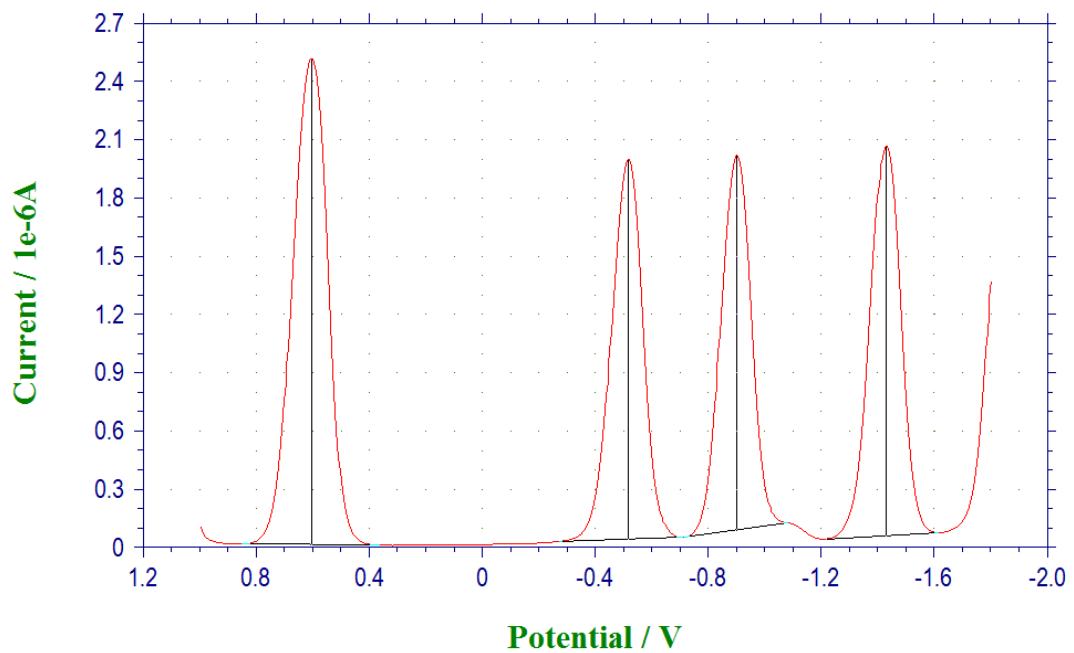
Cyclic Voltammogram of Compound **2b** (scanning rate: 20 mV s⁻¹)



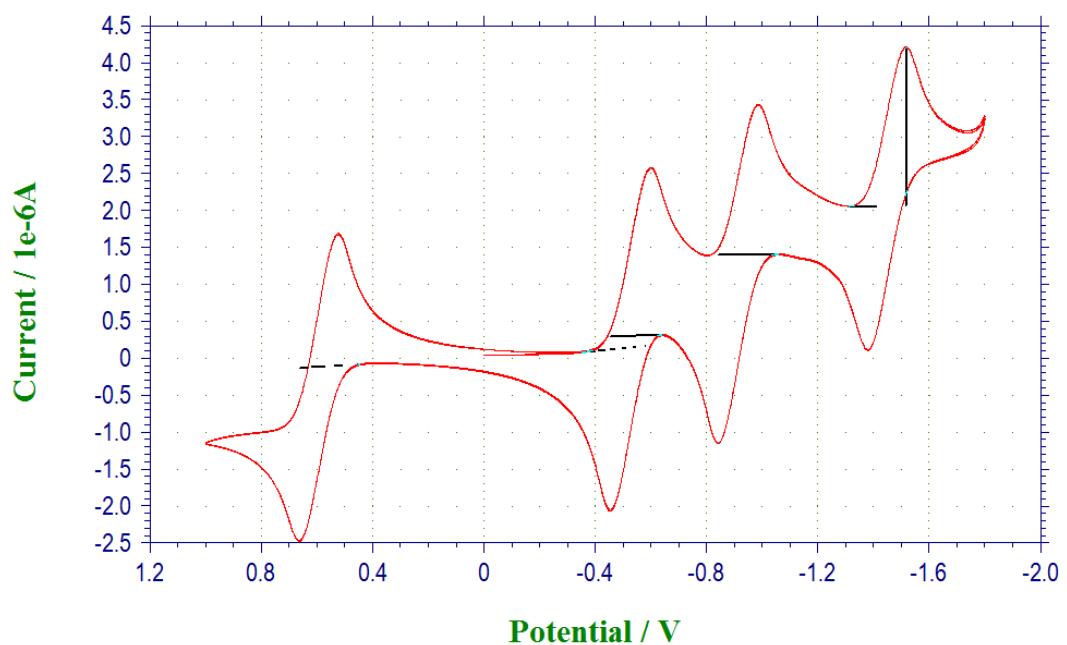
Differential Pulse Voltammogram of Compound **2b**



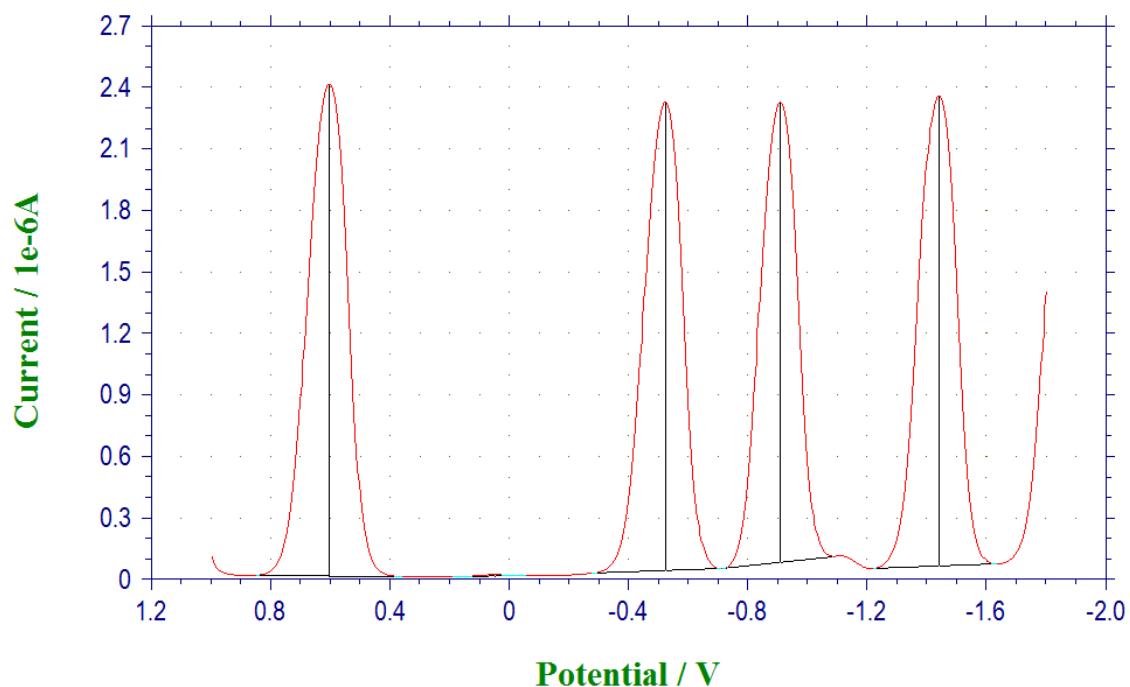
Cyclic Voltammogram of Compound **2c** (scanning rate: 20 mV s⁻¹)



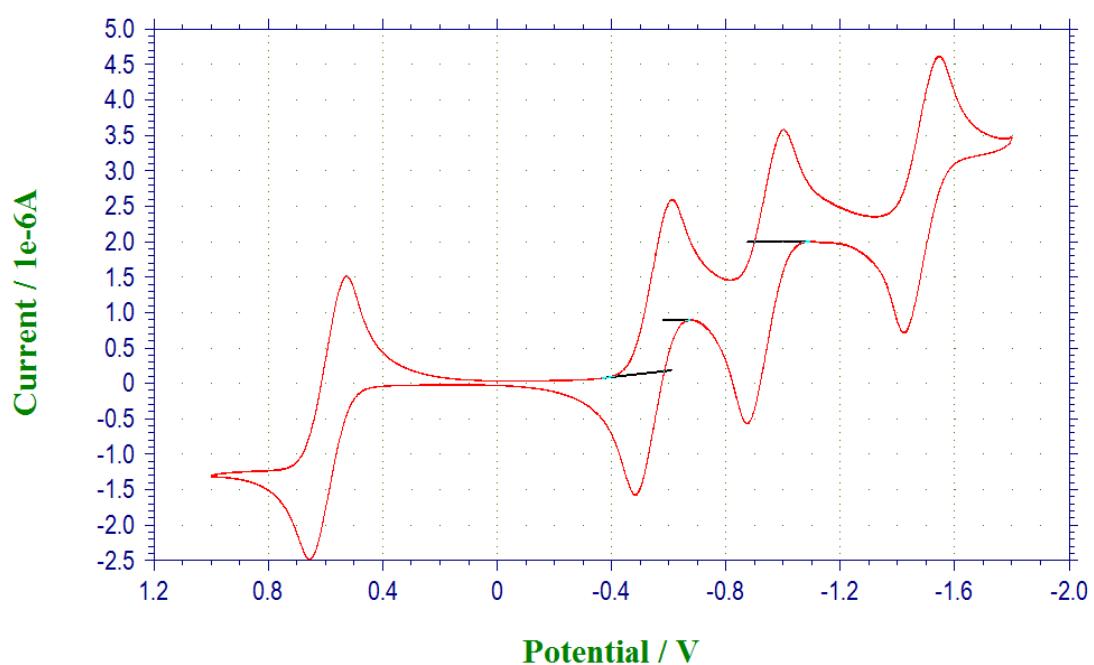
Differential Pulse Voltammogram of Compound **2c**



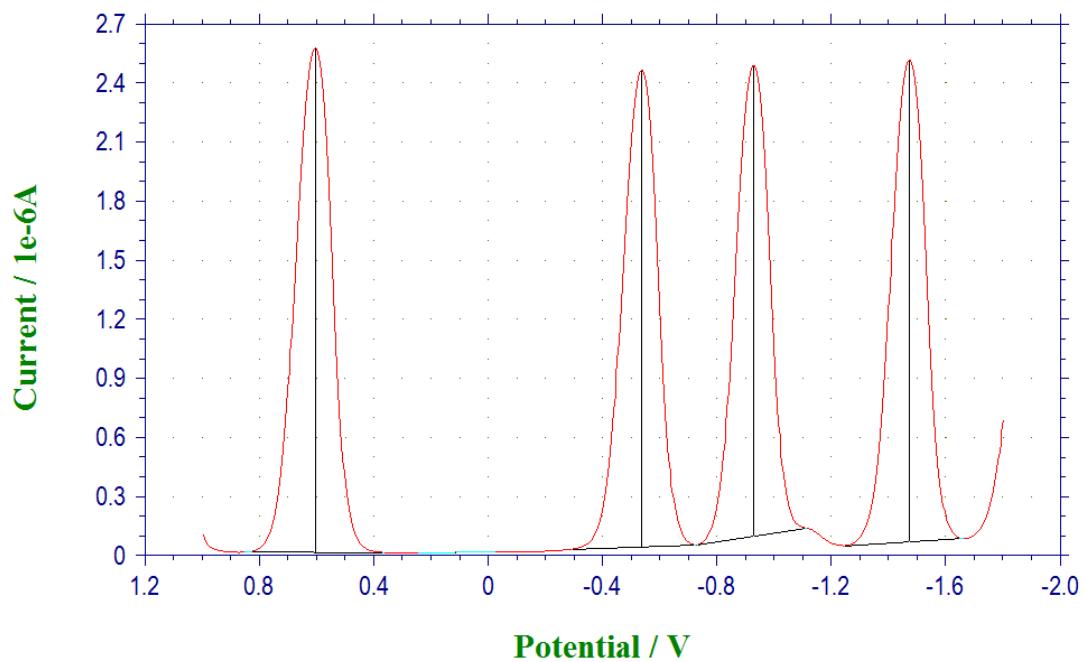
Cyclic Voltammogram of Compound **2d** (scanning rate: 20 mV s⁻¹)



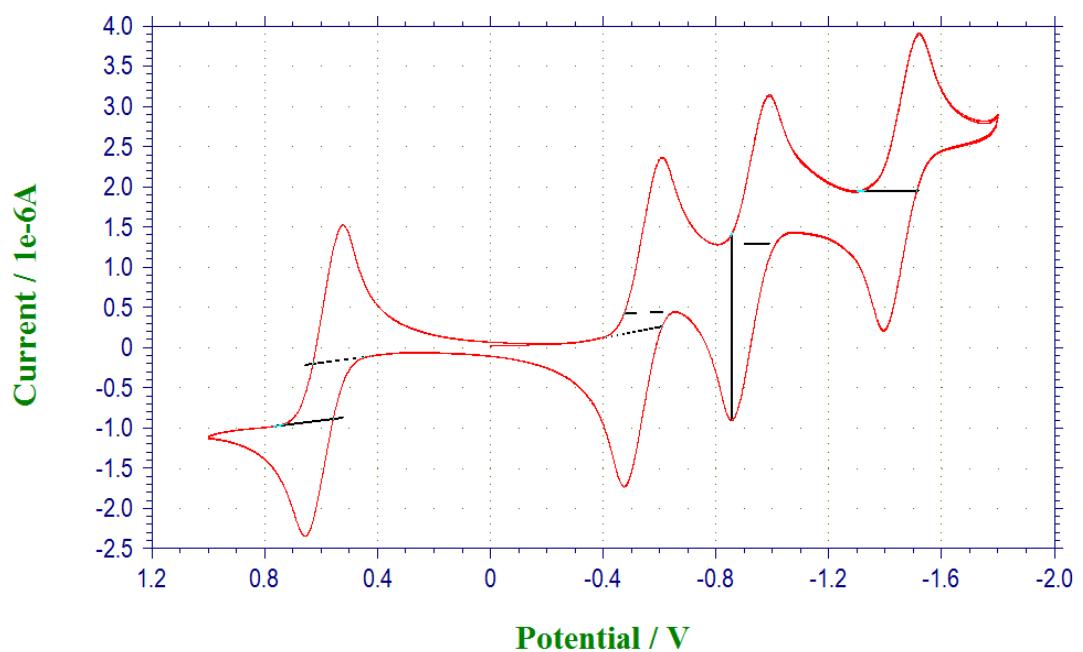
Differential Pulse Voltammogram of Compound **2d**



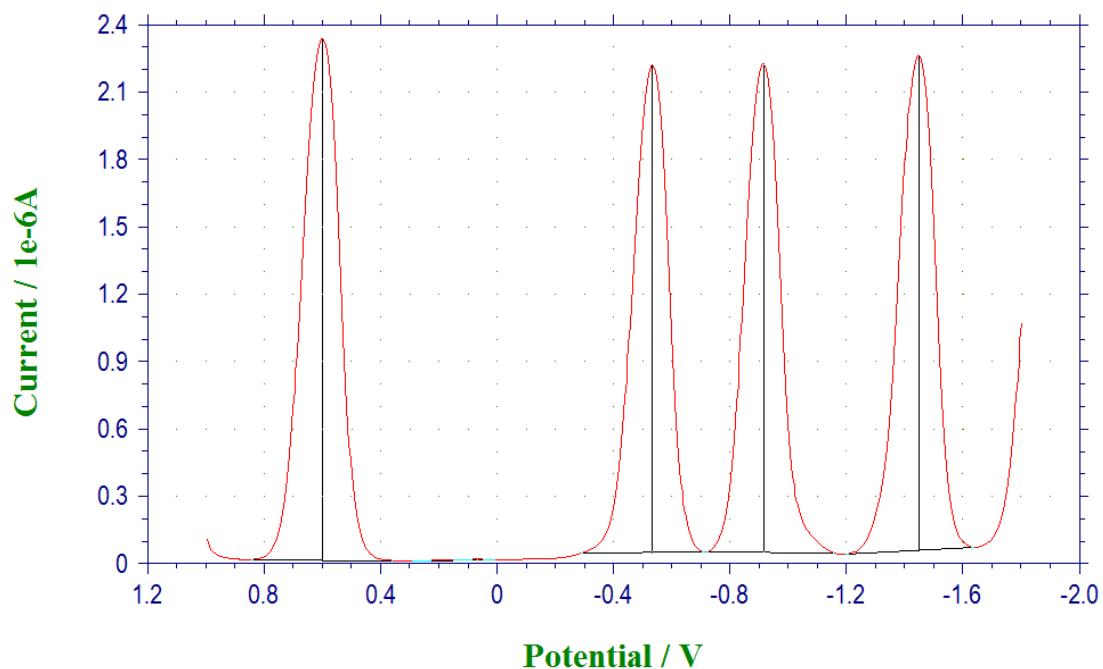
Cyclic Voltammogram of Compound **2e** (scanning rate: 20 mV s⁻¹)



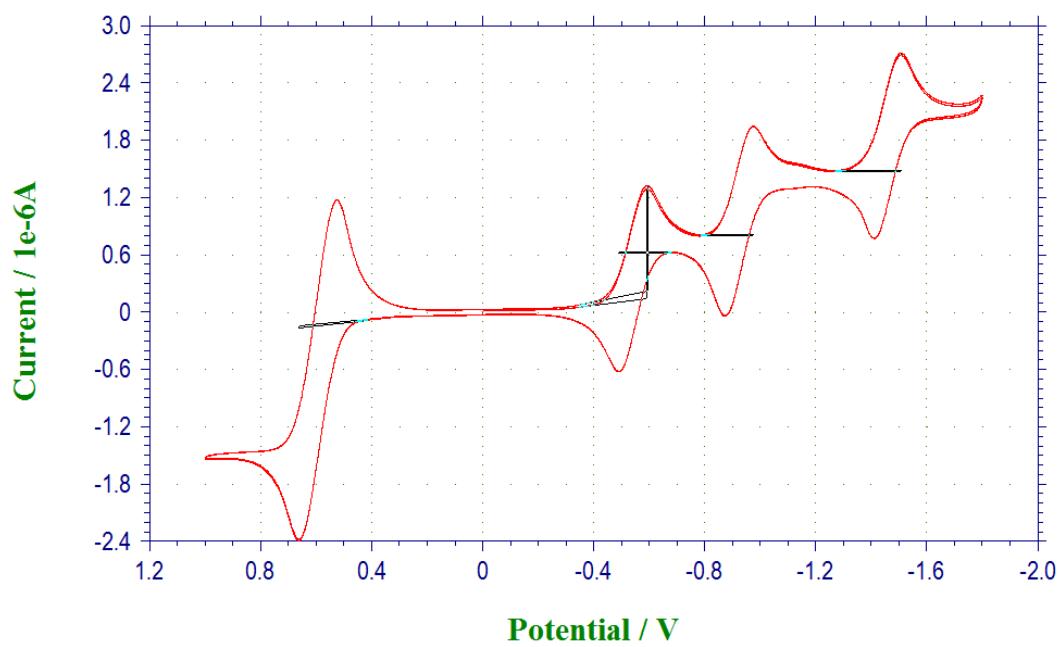
Differential Pulse Voltammogram of Compound **2e**



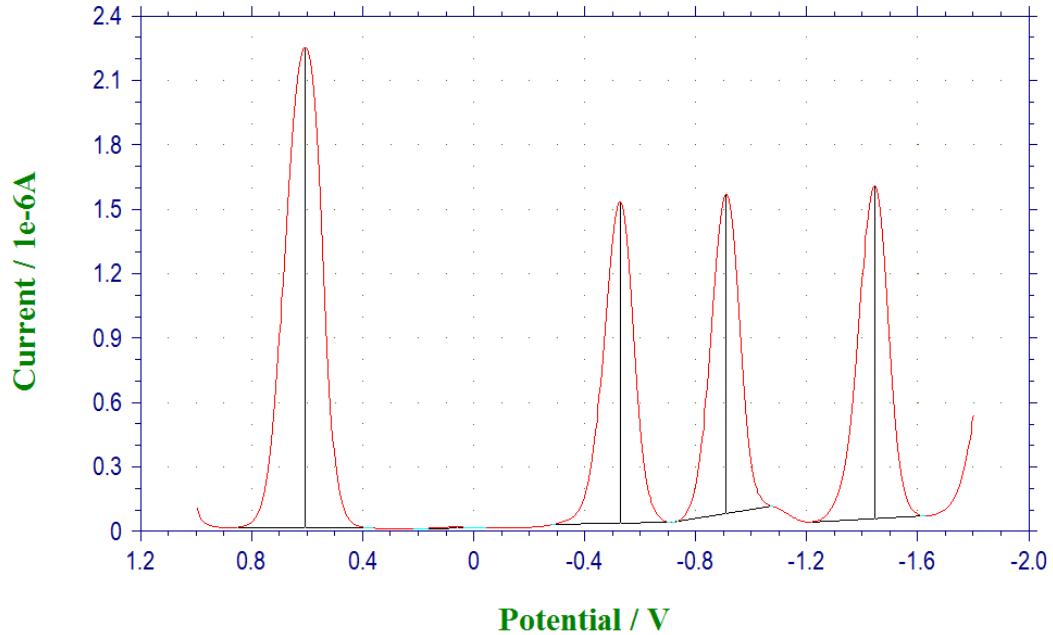
Cyclic Voltammogram of Compound **2f** (scanning rate: 20 mV s⁻¹)



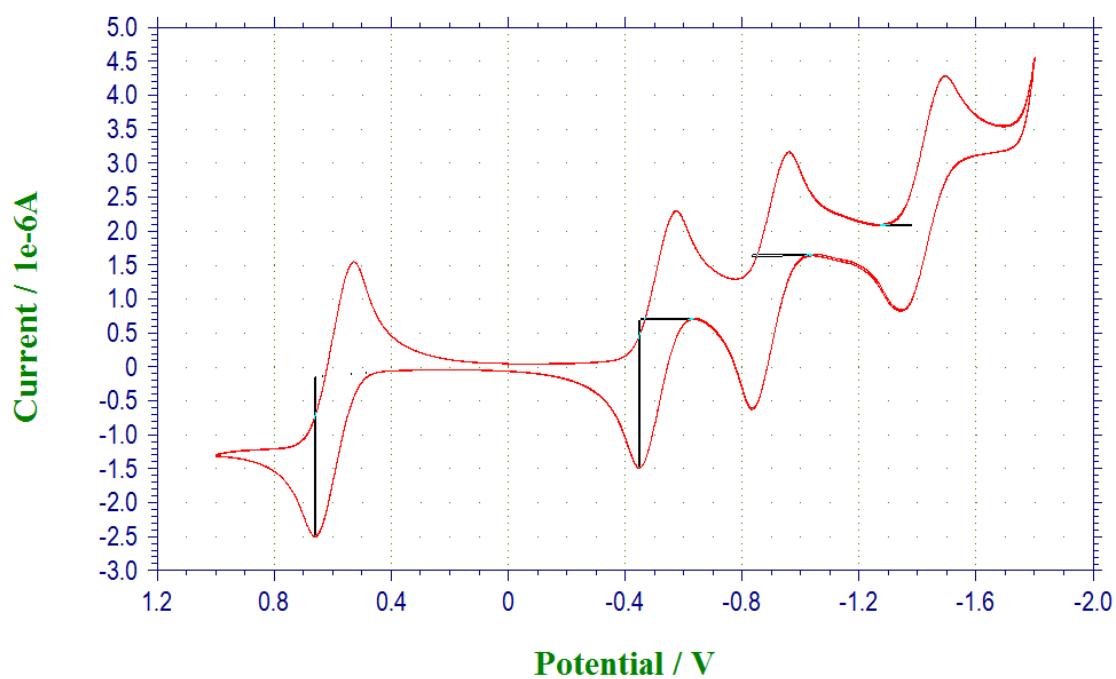
Differential Pulse Voltammogram of Compound **2f**



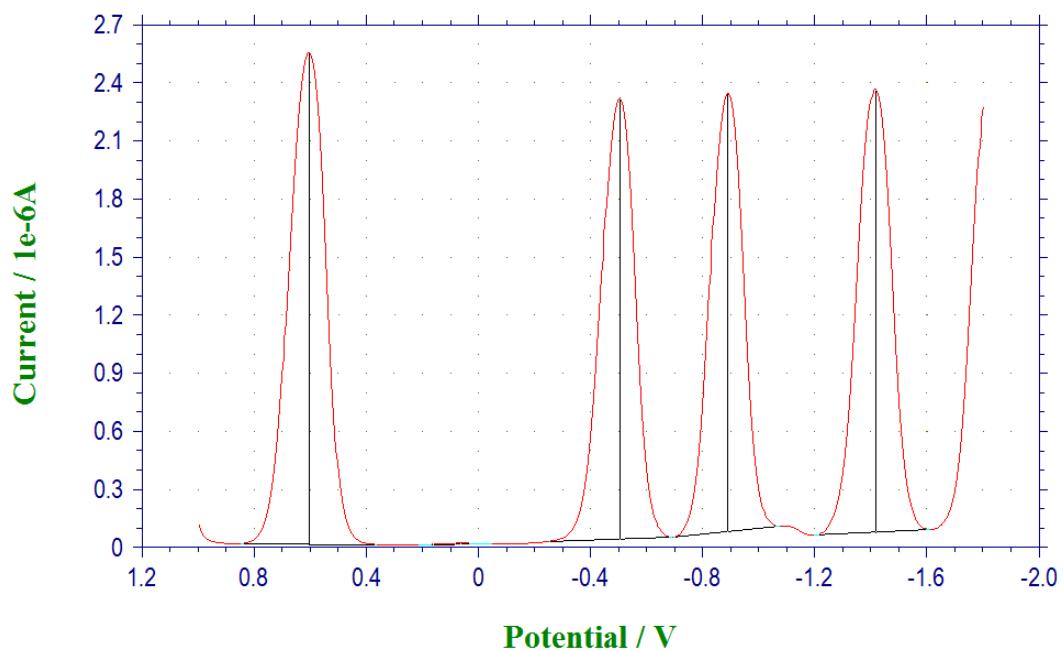
Cyclic Voltammogram of Compound **2g** (scanning rate: 20 mV s⁻¹)



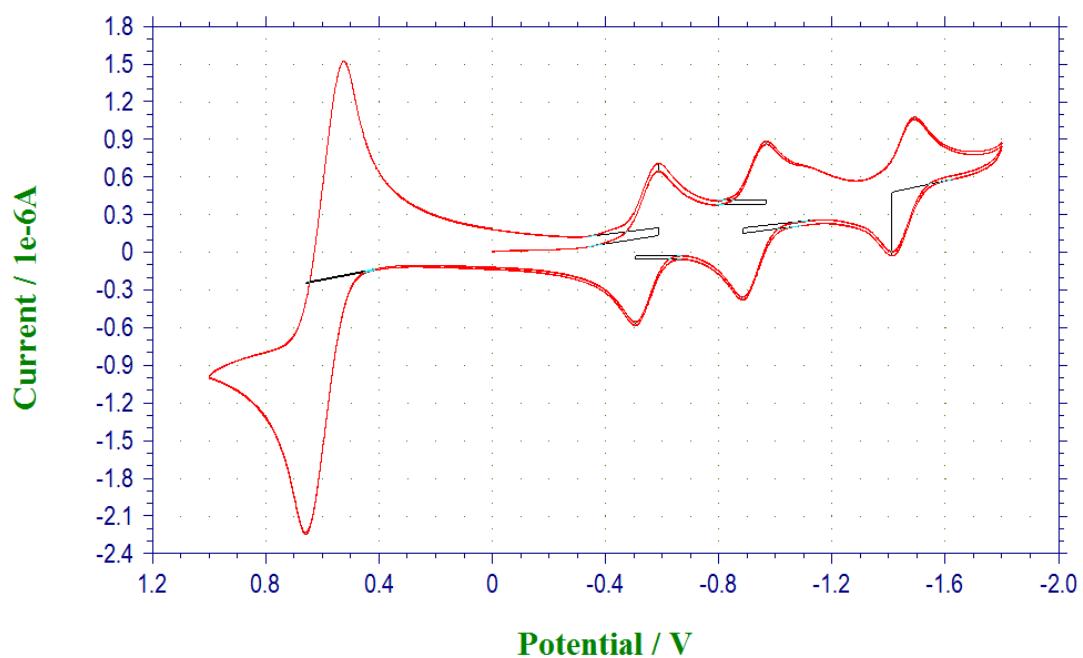
Differential Pulse Voltammogram of Compound **2g**



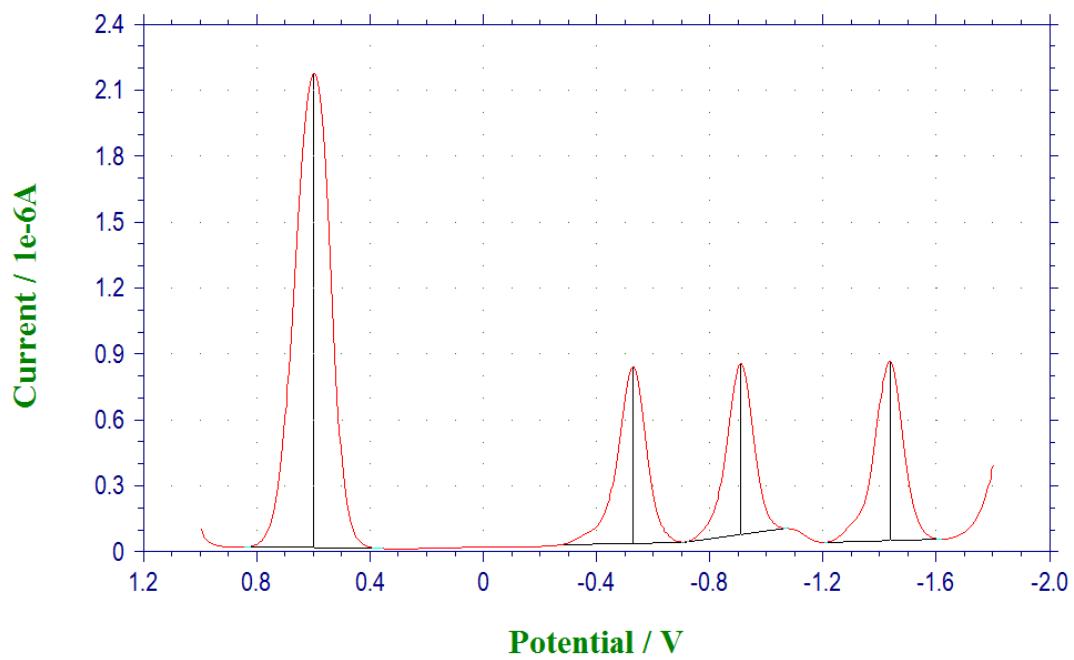
Cyclic Voltammogram of Compound **2h** (scanning rate: 20 mV s⁻¹)



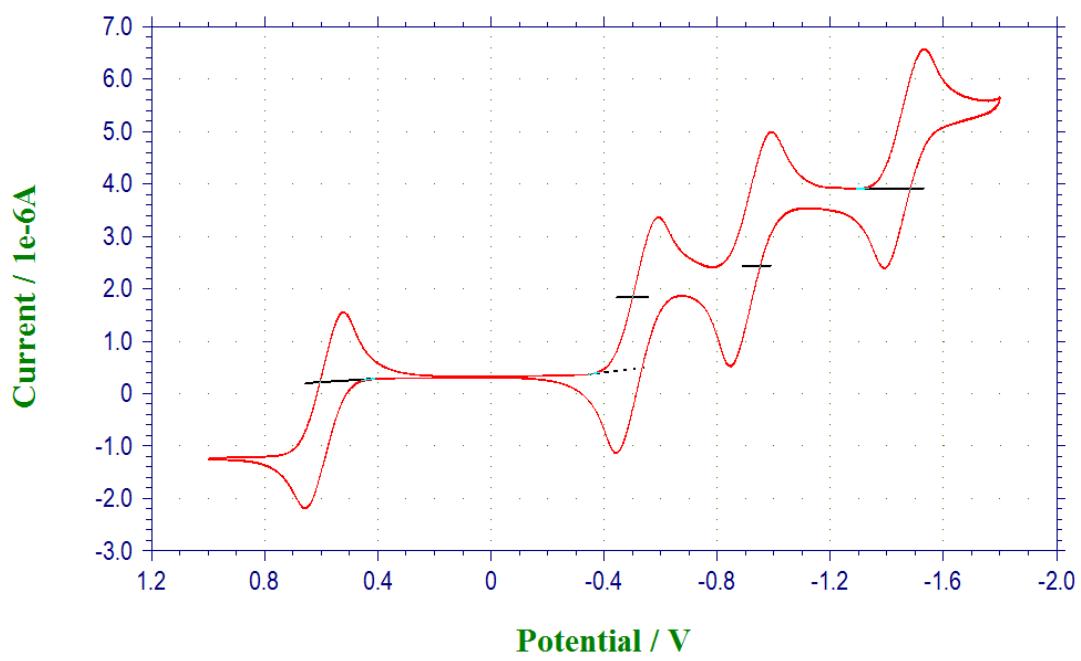
Differential Pulse Voltammogram of Compound **2h**



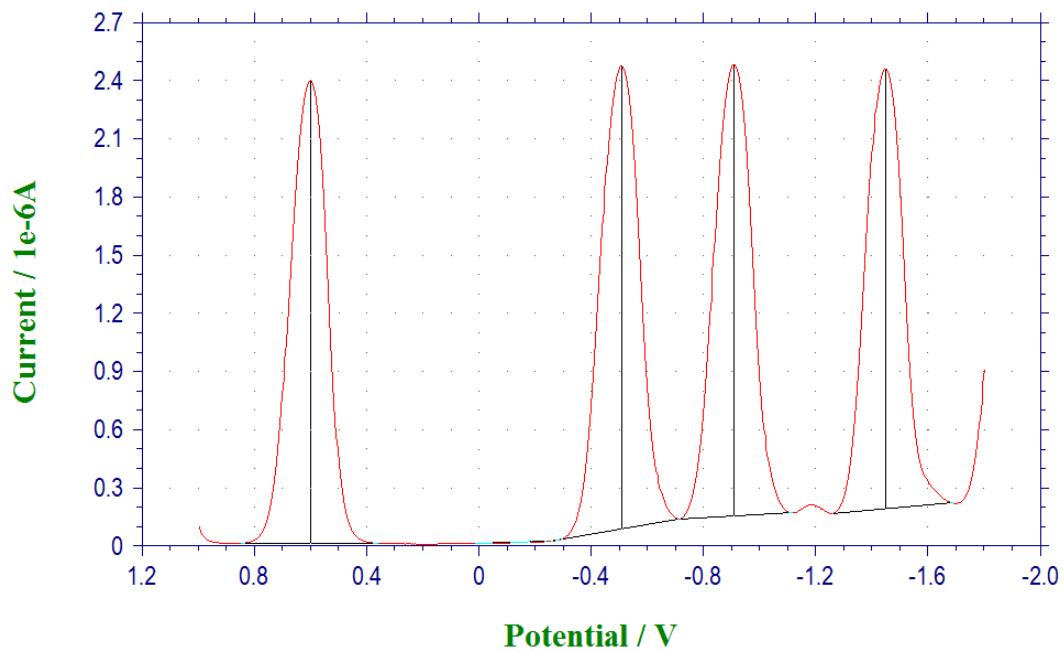
Cyclic Voltammogram of Compound **2i** (scanning rate: 20 mV s^{-1})



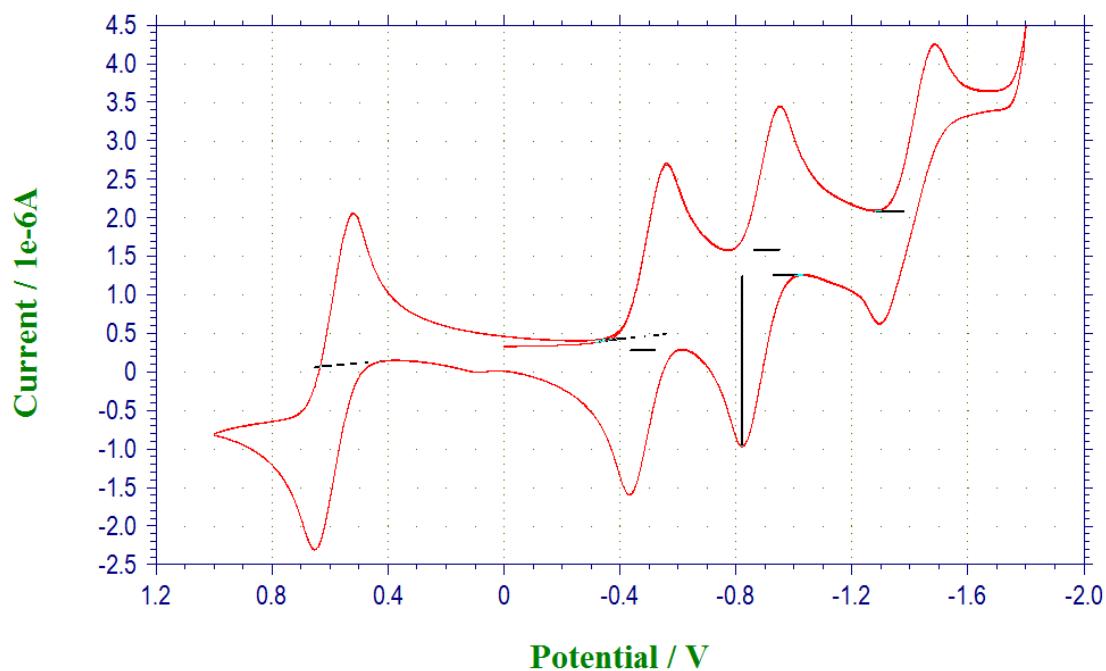
Differential Pulse Voltammogram of Compound **2i**



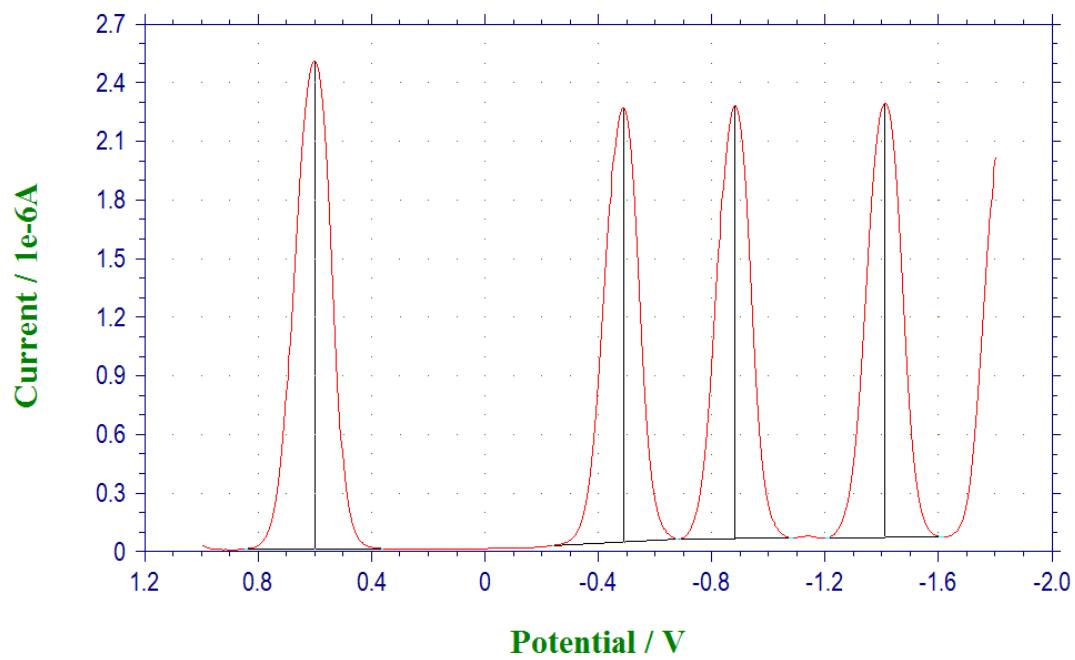
Cyclic Voltammogram of Compound **2j** (scanning rate: 20 mV s^{-1})



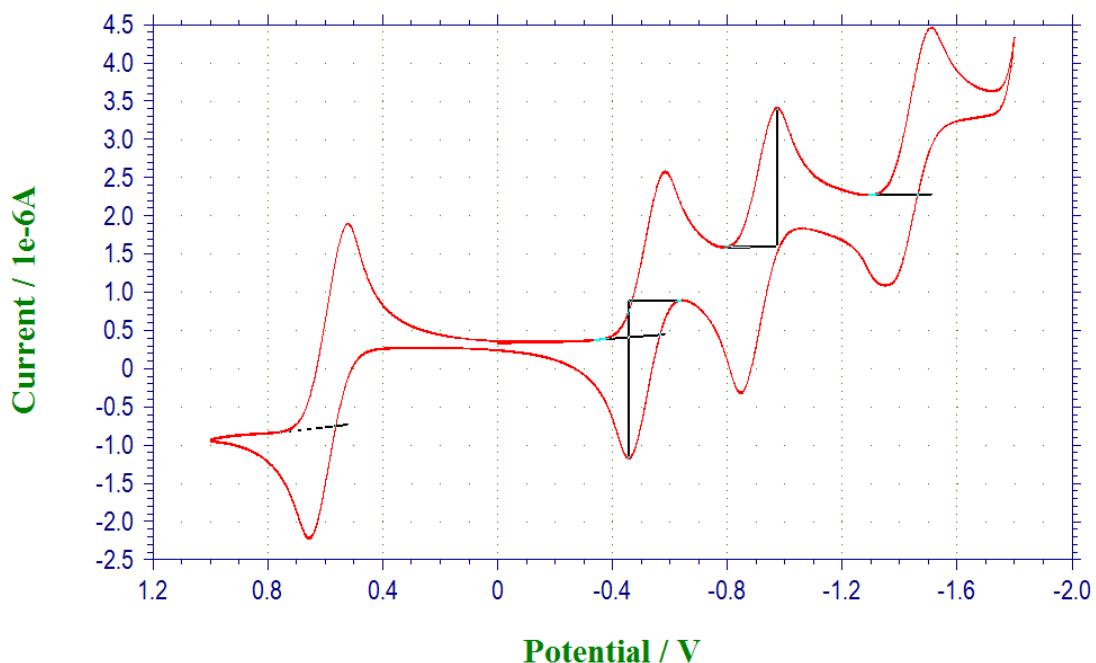
Differential Pulse Voltammogram of Compound **2j**



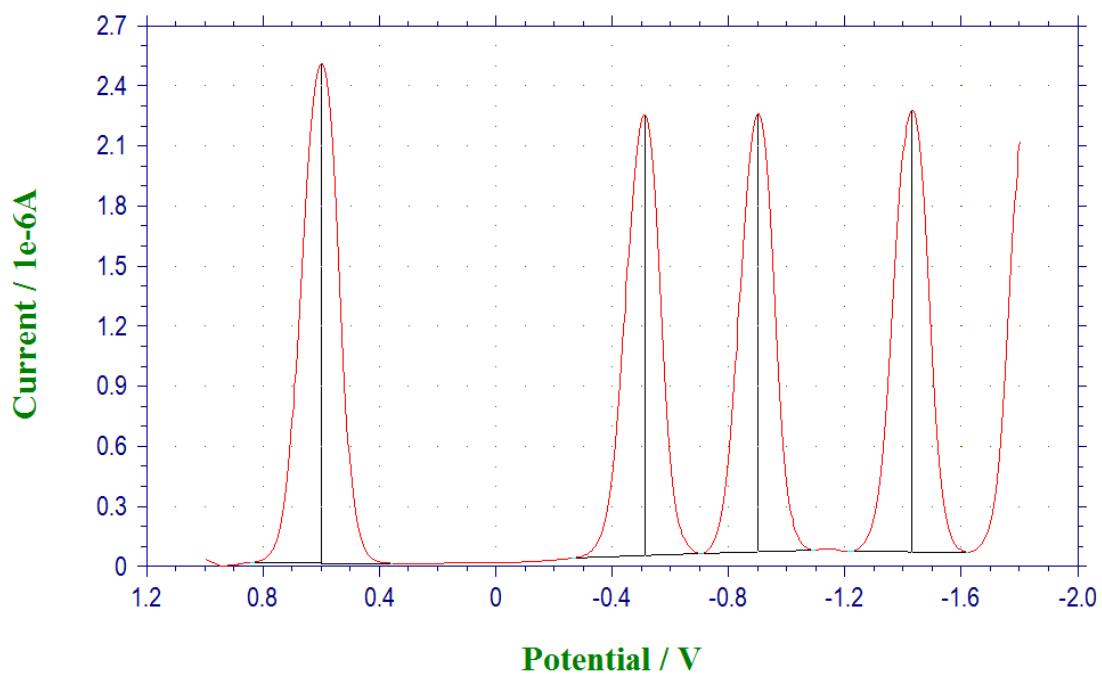
Cyclic Voltammogram of Compound **2k** (scanning rate: 20 mV s^{-1})



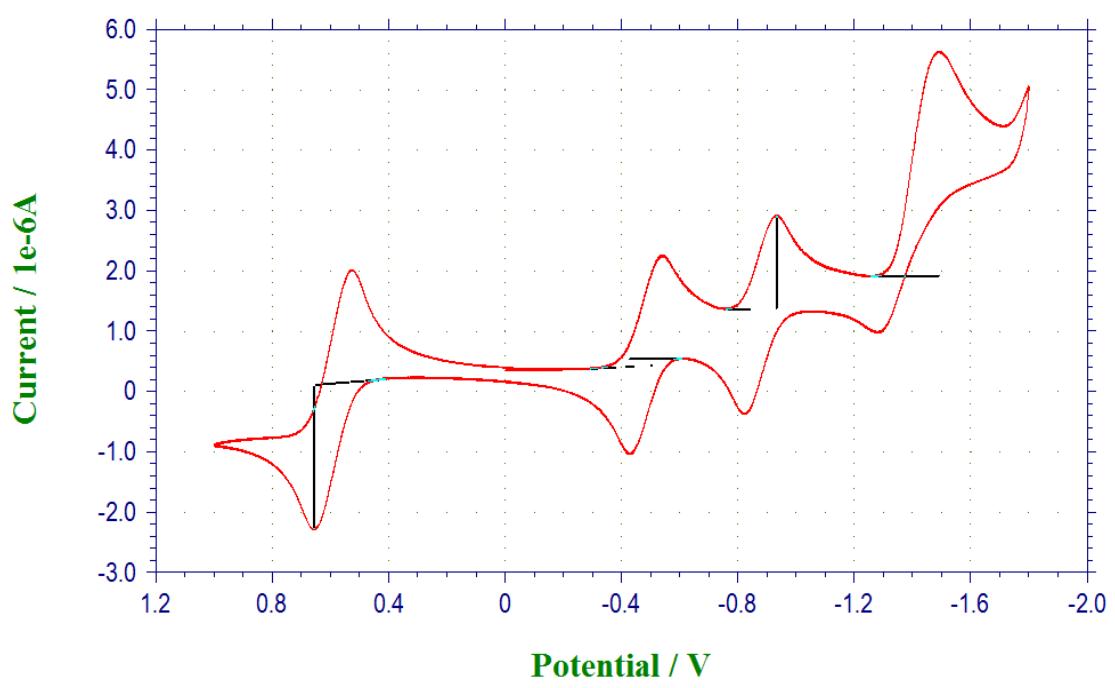
Differential Pulse Voltammogram of Compound **2k**



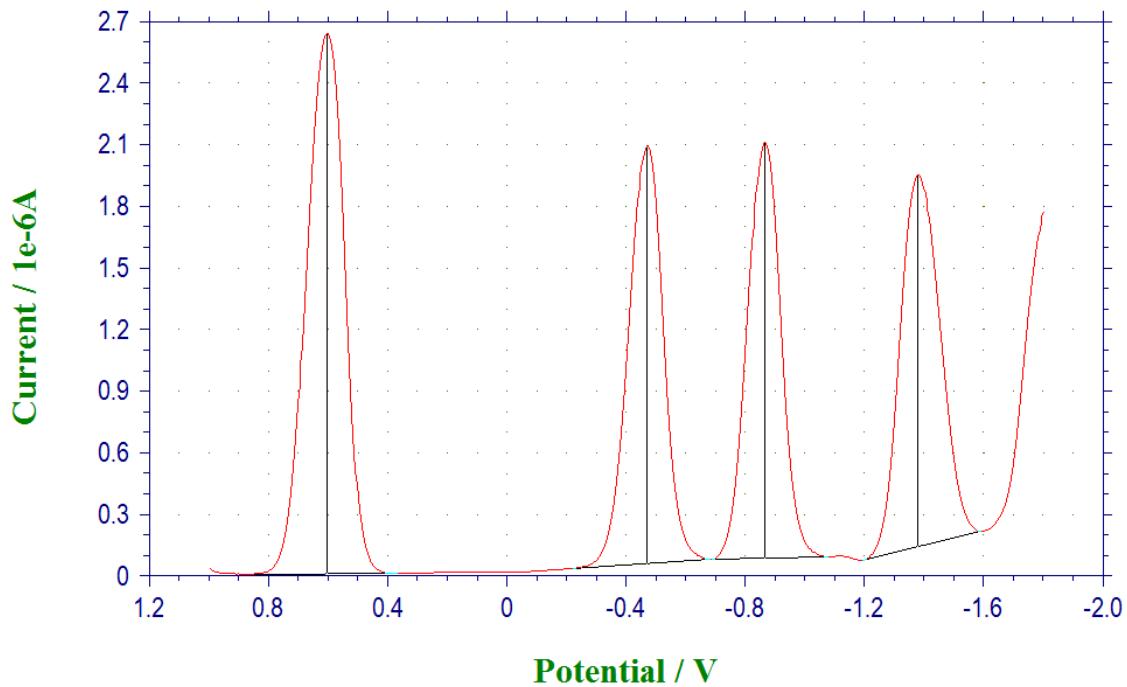
Cyclic Voltammogram of Compound **2I** (scanning rate: 20 mV s⁻¹)



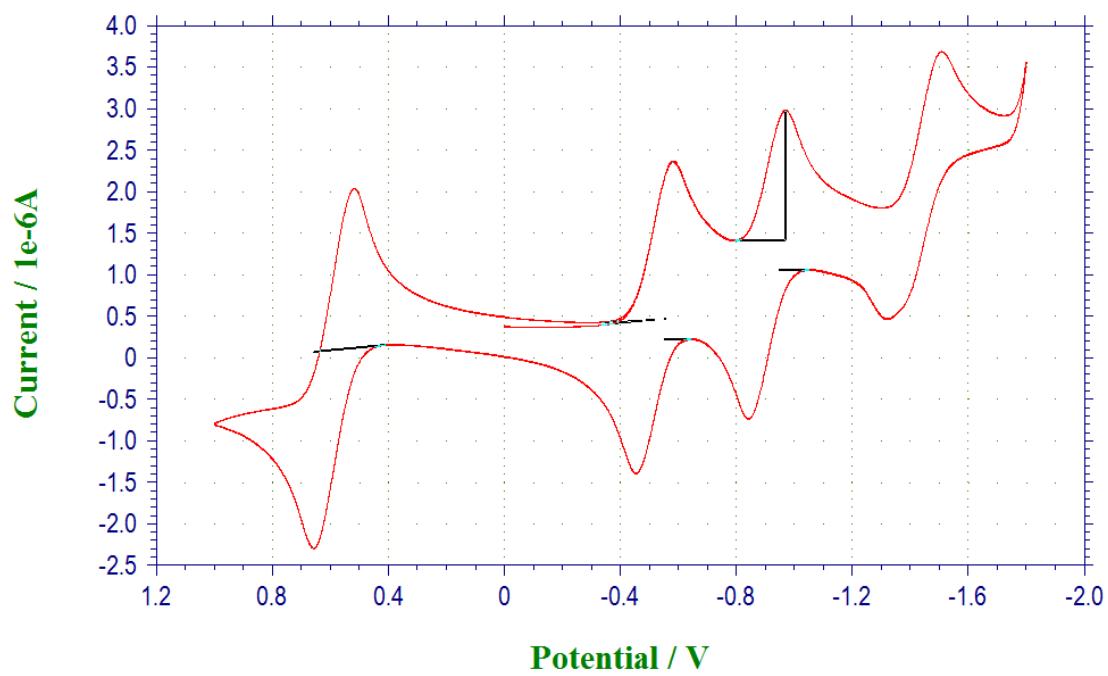
Differential Pulse Voltammogram of Compound **2I**



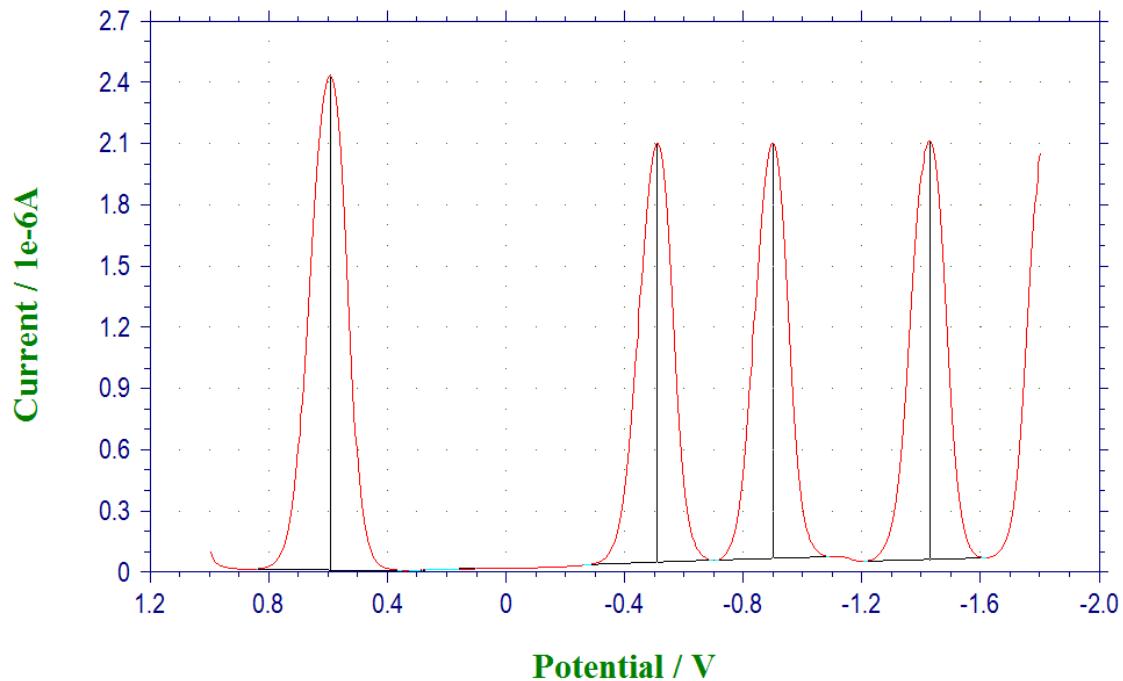
Cyclic Voltammogram of Compound **2m** (scanning rate: 20 mV s⁻¹)



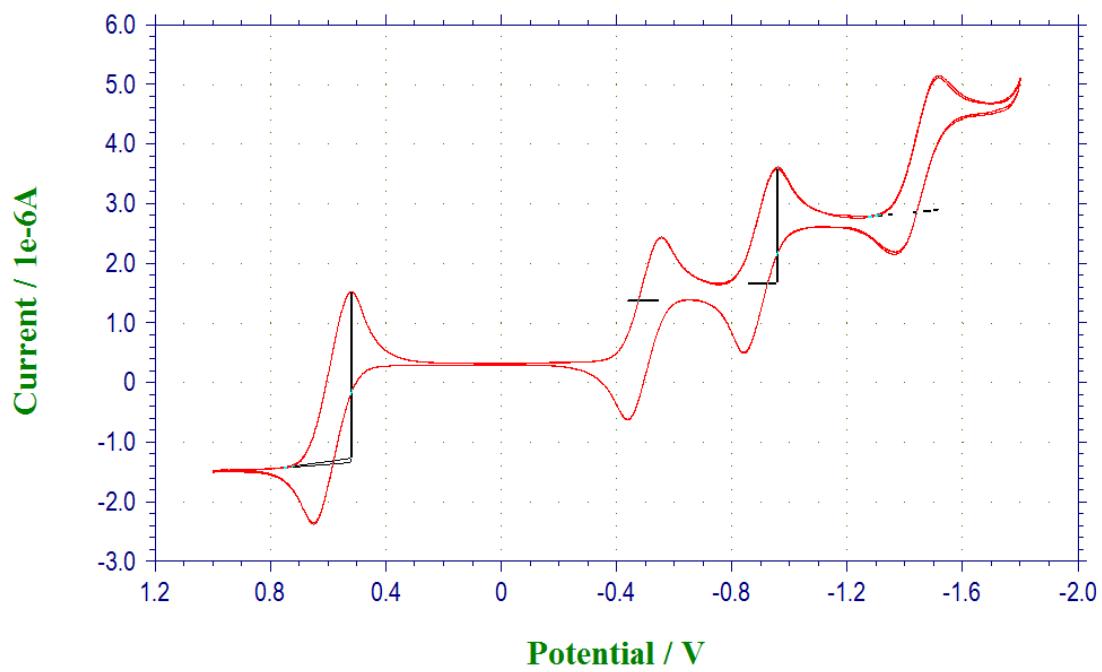
Differential Pulse Voltammogram of Compound **2m**



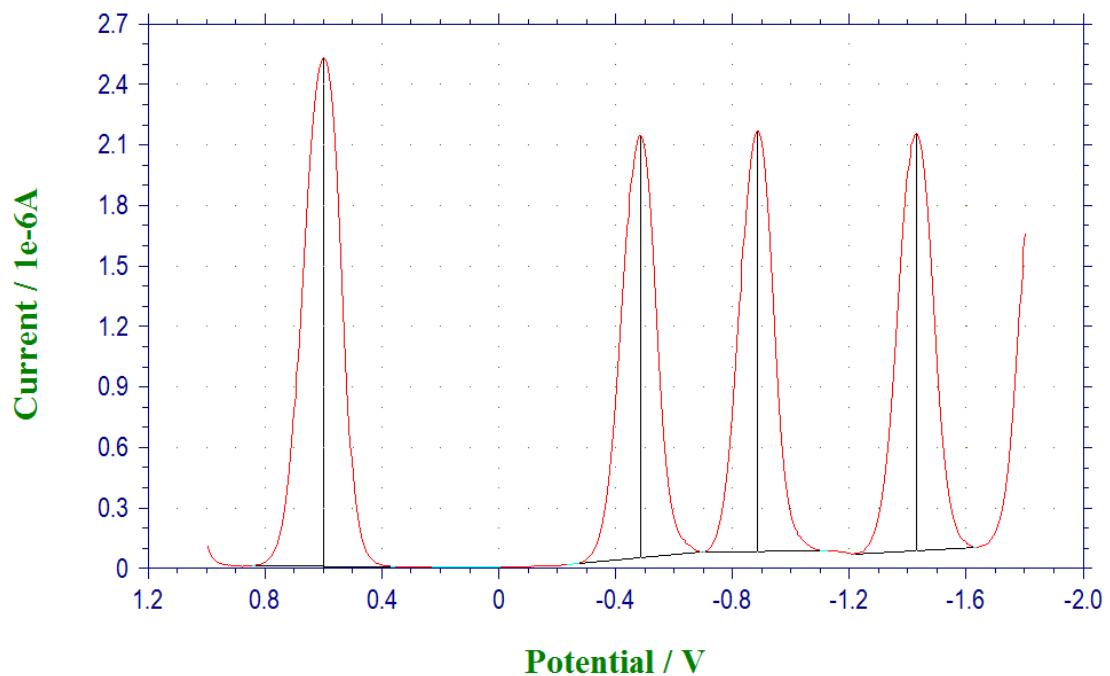
Cyclic Voltammogram of Compound **2n** (scanning rate: 20 mV s⁻¹)



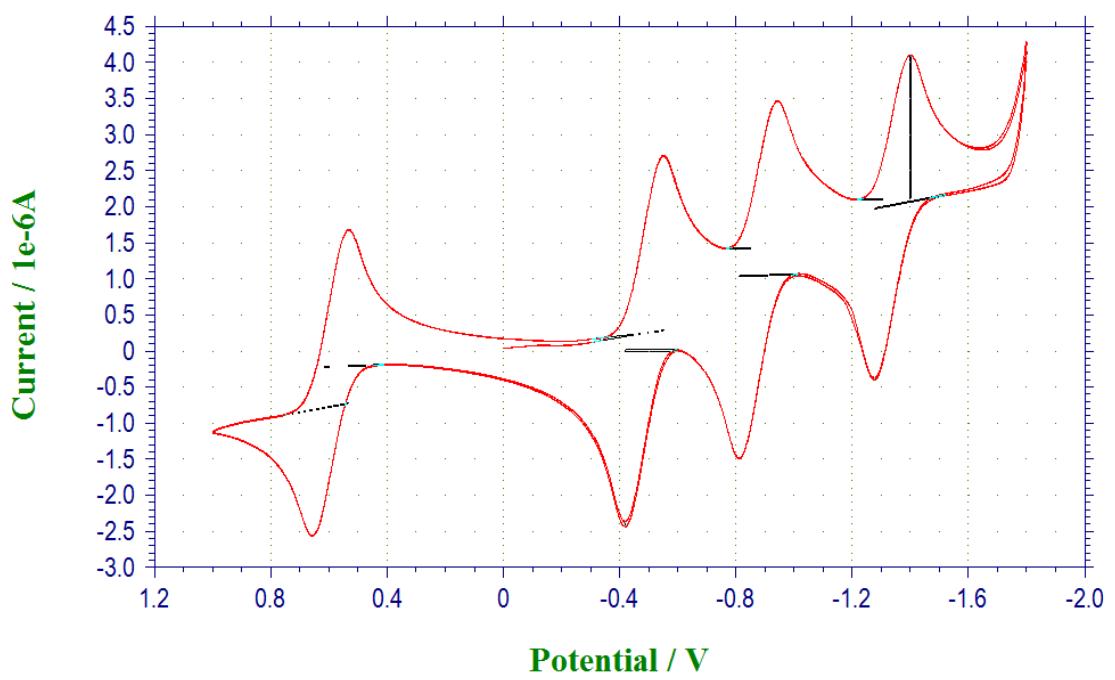
Differential Pulse Voltammogram of Compound **2n**



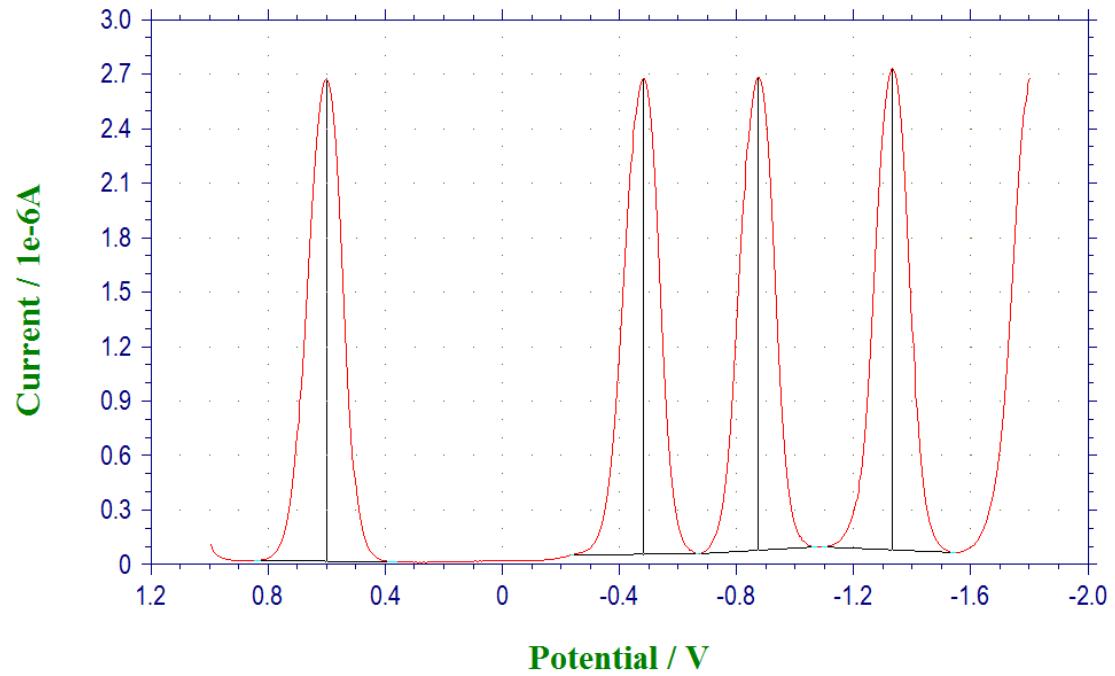
Cyclic Voltammogram of Compound **2o** (scanning rate: 20 mV s⁻¹)



Differential Pulse Voltammogram of Compound **2o**



Cyclic Voltammogram of C_{60} (scanning rate: 20 mV s⁻¹)



Differential Pulse Voltammogram of C_{60}