

Influence of Alkoxy Groups on Rates of Acetal Hydrolysis and Tosylate Solvolysis: Electrostatic Stabilization of Developing Oxocarbenium Ion Intermediates and Neighboring-Group Participation to Form Oxonium Ions

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Supporting Information

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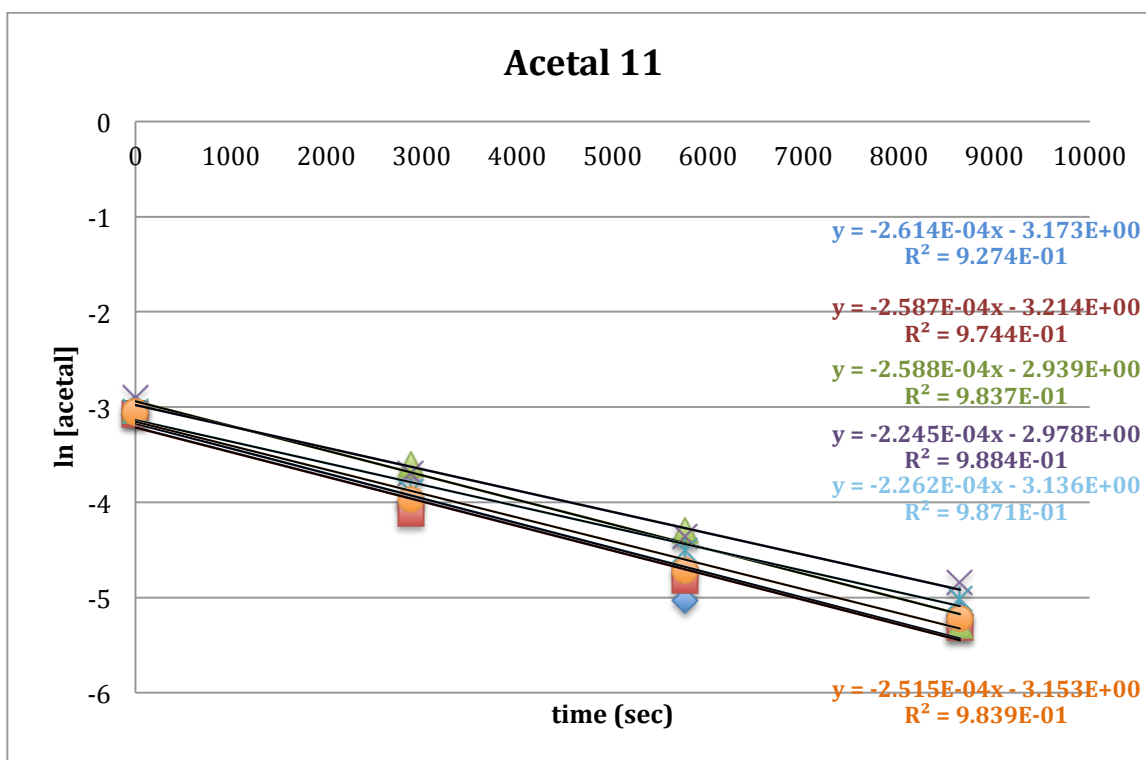
I. General Experimental

Infrared (IR) spectra were obtained by a FT-IR spectrometer using either attenuated total reflectance (ATR) or a thin film on a salt plate, as indicated. High-resolution mass spectra (HRMS) were acquired on an Accurate-Mass time-of-flight spectrometer. At ambient temperature, ¹H and ¹³C NMR spectra were obtained utilizing AV-400 (400 and 100 MHz, respectively), AV-500 (500 and 125 MHz, respectively), or AVIII-600 (600 and 150 MHz, respectively) spectrometers, as indicated. The experimental data is listed as follows: chemical shift in ppm from internal tetramethylsilane or referenced to residual solvent (¹H NMR: C₆D₆ δ 7.16; CDCl₃ δ 7.26. ¹³C NMR: C₆D₆ δ 128.06; CDCl₃ δ 77.16) on the δ scale, multiplicity (appar = apparent, br = broad, s = singlet, d = doublet, t = triplet, q = quartet, quint = quintet, sext = sextet, m = multiplet), coupling constants (Hz), and integration. Microanalyses were performed by an independent facility in GA. Analytical thin layer chromatography was performed on silica gel 60 Å F₂₅₄ plates. Liquid chromatography was performed using forced flow (flash chromatography) of the indicated solvent system on silica gel (SiO₂) 60 (230-400 mesh). THF, CH₂Cl₂, and Et₂O were dried by filtration through alumina following the method by Grubbs.¹ Reactions using solvents THF, CH₂Cl₂, and Et₂O were run under an atmosphere of nitrogen with glassware that was flame-dried under vacuum and cooled under a stream of nitrogen.

II. Hydrolysis of Acetal 11

The procedure for the hydrolysis of acetal **11** followed methods previously reported.² The hydrolysis of acetals **5**, **6**, *trans*-**7**, *trans*-**8**, **9**, **10**, **12**, and **13** have also previously been reported.²

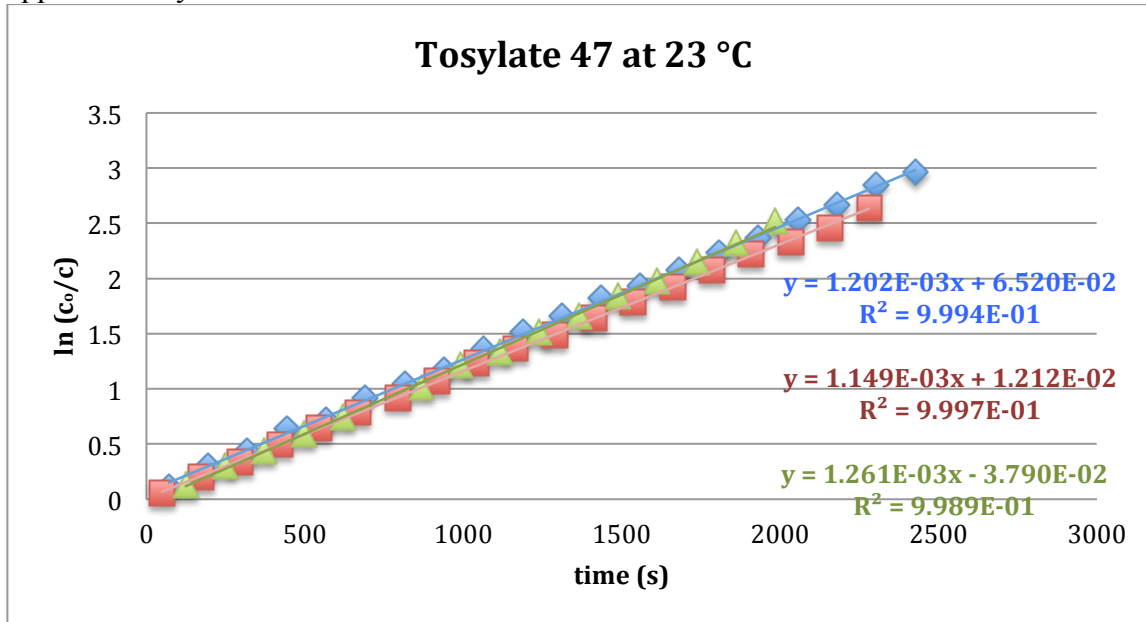
Graph S1. Rate of hydrolysis of acetal **11** plotted as $\ln [\text{acetal}]$ versus t (s) over approximately three half-lives, resulting in $k = 4.0 \times 10^{-3}$.



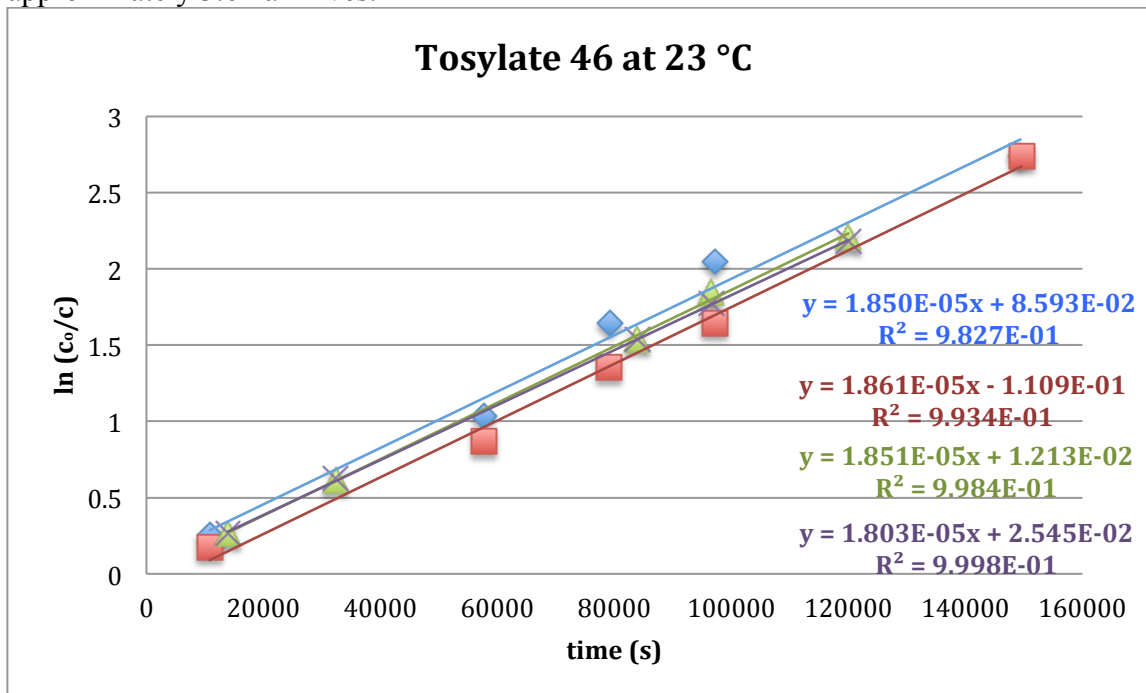
III. Solvolysis Studies

General Procedure: The tosylate was added to a standard NMR tube equipped with 1,4-dimethoxybenzene as an internal standard. Anisole was used for tosylate **45** due to stability issues over long periods of time. The NMR tube was inverted after addition of acetic acid- d_4 . An NMR spectrum was obtained at 25 °C during given time points, depending on the tosylate substrate. The decrease of starting material/tosylate was measured to determine the rate, by plotting $\ln (c_0/c)$ versus time (s). The solvolysis rate of tosylate **45** was determined at three different temperatures (100, 110, and 125 °C) using a J. Young NMR tube, and the Arrhenius equation was applied to determine the relative rate of tosylate **45** at 23 °C.

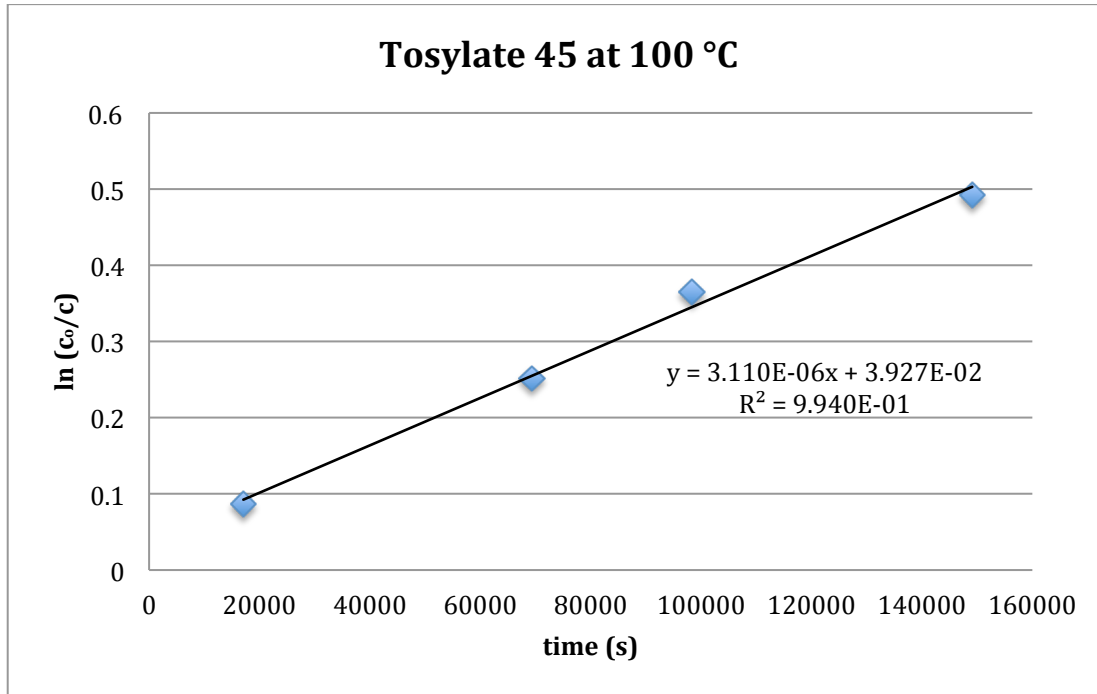
Graph S2. Solvolysis of tosylate **47** plotted as $\ln(c_0/c)$ versus time (s) at 23 °C, over approximately 4 half-lives.



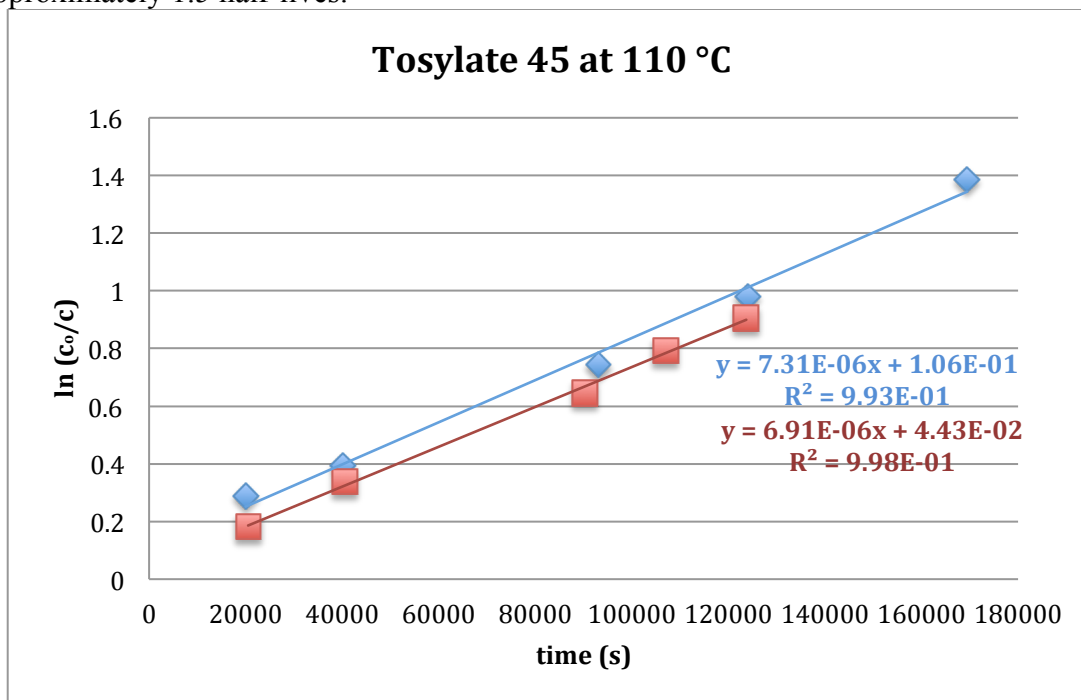
Graph S3. Solvolysis of tosylate **46** plotted as $\ln(c_0/c)$ versus time (s) at 23 °C, over approximately 3.6 half-lives.



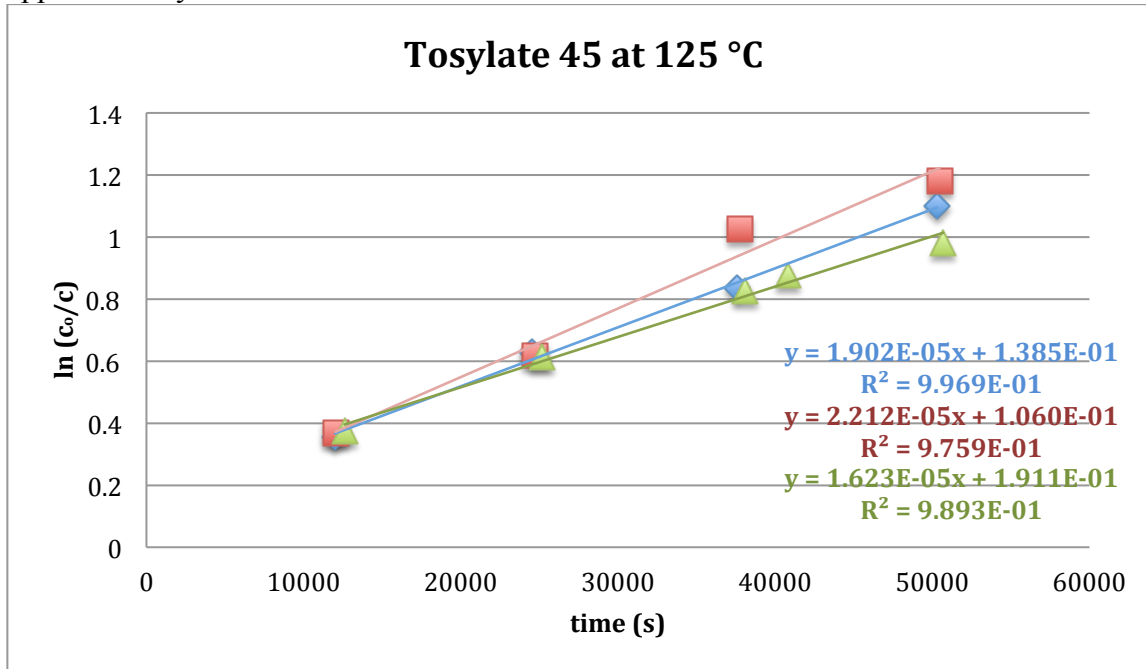
Graph S4. Solvolysis of tosylate **45** plotted as $\ln (c_0 / c)$ versus time (s) at 100 °C, over less than one half-life.



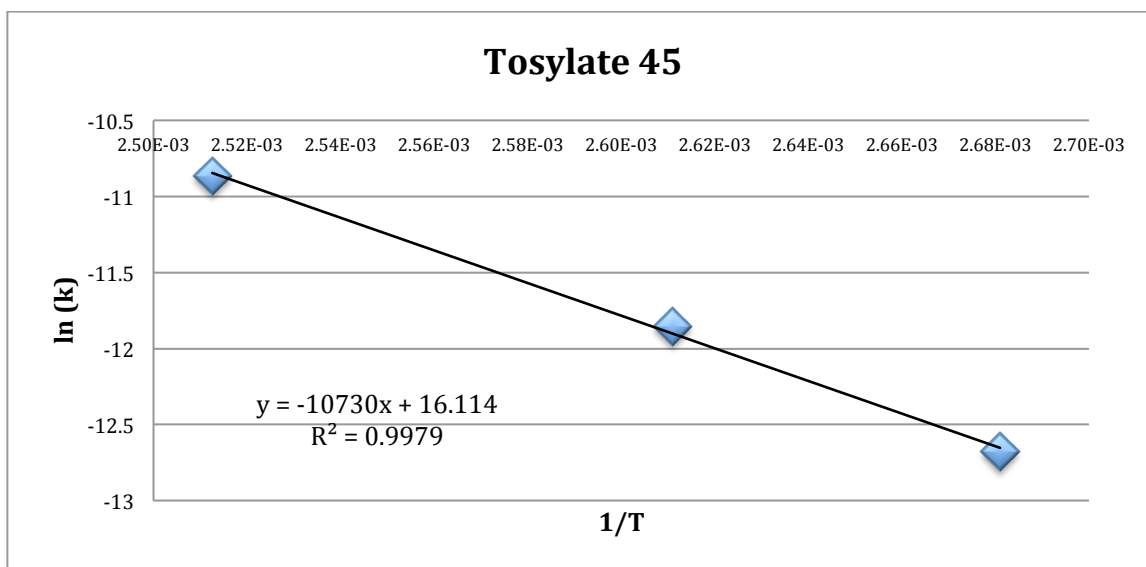
Graph S5. Solvolysis of tosylate **45** plotted as $\ln (c_0 / c)$ versus time (s) at 110 °C, over approximately 1.5 half-lives.



Graph S6. Solvolysis of tosylate **45** plotted as $\ln(c_0/c)$ versus time (s) at 125 °C, over approximately 1.4 half-lives.



Graph S7. Arrhenius plot for tosylate **45**. $\ln(k)$ versus $1/T$ plotted to determine activation energy and pre-exponential factor to apply the Arrhenius equation and determine the rate at 23 °C.



IV. Computational Methods

The oxocarbenium ion intermediates derived from acetals were calculated using Spartan'10 (Wavefunction, Inc., Irvine, CA). An input structure was subjected to a search for low-energy conformers using the conformational search package using semi-empirical methods (AM1). The 35 lowest-energy structures (within 10 kcal/mol) were then minimized using Density Functional methods (B3LYP/6-31G*) with the Gaussian Software Package.

Oxocarbenium ion from acetal *trans*-8

Energy = -536760.45 kcal/mol

ATOM

1 H1	-4.8382	-2.5601	0.7514 H
2 C2	-4.1739	-1.6942	0.6522 C
3 H3	-4.1524	-1.2290	1.6480 H
4 C4	-4.8352	-0.7356	-0.3590 C
5 H5	-5.9185	-0.8419	-0.2320 H
6 H6	-4.6296	-1.0788	-1.3837 H
7 C7	-4.5344	0.7698	-0.2455 C
8 H8	-5.3510	1.3005	-0.7489 H
9 H9	-4.5909	1.0776	0.8081 H
10 C10	-3.2333	1.3096	-0.8640 C
11 H11	-3.3537	2.3950	-0.9650 H
12 H12	-3.1177	0.9276	-1.8886 H
13 C13	-1.9198	1.0958	-0.0823 C
14 H14	-2.1109	1.2280	0.9888 H
15 C15	-1.5400	-1.4138	0.6516 C
16 H16	-0.6755	-2.0878	0.6286 H
17 C17	-2.7830	-2.2625	0.3145 C
18 H18	-2.7516	-2.5556	-0.7446 H
19 H19	-2.6628	-3.1938	0.8792 H
20 C20	-1.2395	-0.2626	-0.2986 C
21 H21	-1.3318	-0.5691	-1.3466 H
22 H22	-1.6043	-1.0282	1.6760 H
23 O23	0.2607	0.0491	-0.1338 O
24 C24	1.1478	-0.6372	-1.1426 C
25 H25	0.7349	-1.6456	-1.1892 H
26 H26	0.9924	-0.1322	-2.0999 H
27 C27	2.5786	-0.6341	-0.7096 C
28 C28	5.2728	-0.6772	0.0590 C
29 C29	3.5139	0.1854	-1.3593 C
30 C30	3.0077	-1.4869	0.3210 C
31 C31	4.3475	-1.5039	0.7067 C
32 C32	4.8572	0.1634	-0.9767 C
33 H33	3.1998	0.8258	-2.1805 H

34 H34	2.2949	-2.1423	0.8150 H
35 H35	4.6739	-2.1691	1.4998 H
36 H36	5.5764	0.7925	-1.4914 H
37 H37	6.3174	-0.6990	0.3533 H
38 C38	-0.8498	2.1209	-0.5123 C
39 H39	-1.0754	3.1272	-0.1523 H
40 H40	-0.7683	2.1567	-1.6050 H
41 C41	0.4480	1.6358	0.0793 C
42 H42	1.3729	1.8291	-0.4699 H
43 O43	0.5231	1.8728	1.3965 O
44 C44	1.8003	1.6853	2.0472 C
45 H45	2.5814	2.2326	1.5107 H
46 H46	2.0482	0.6228	2.0981 H
47 H47	1.6808	2.0938	3.0493 H

Oxocarbenium ion from acetal *trans*-7

Energy = -512012.10 kcal/mol

ATOM

1 C1	-4.2782	-1.4622	-0.1459 C
2 H2	-5.2018	-2.0090	0.0720 H
3 H3	-3.9022	-1.8858	-1.0885 H
4 C4	-4.6506	0.0219	-0.3272 C
5 H5	-5.5090	0.0939	-1.0047 H
6 H6	-4.9960	0.4112	0.6400 H
7 C7	-3.5496	0.9446	-0.8756 C
8 H8	-3.9278	1.9736	-0.8815 H
9 H9	-3.3414	0.6896	-1.9250 H
10 C10	-2.2212	0.9264	-0.0965 C
11 H11	-2.4110	1.1038	0.9686 H
12 C12	-1.7835	-1.5293	0.7090 C
13 H13	-1.3653	-2.4609	0.3099 H
14 C14	-3.2908	-1.7262	1.0099 C
15 H15	-3.4104	-2.7552	1.3636 H
16 H16	-3.5745	-1.0870	1.8547 H
17 C17	-1.4803	-0.4012	-0.2711 C
18 H18	-1.5790	-0.7286	-1.3114 H
19 H19	-1.2565	-1.3193	1.6460 H
20 O20	-0.0046	-0.0398	-0.1316 O
21 C21	0.9019	-0.7449	-1.1084 C
22 H22	0.5258	-1.7687	-1.1011 H
23 H23	0.7271	-0.2961	-2.0902 H
24 C24	2.3322	-0.6661	-0.6811 C
25 C25	5.0246	-0.5611	0.0875 C
26 C26	3.2346	0.1570	-1.3717 C

27 C27	2.7940	-1.4488	0.3902 C
28 C28	4.1328	-1.3918	0.7757 C
29 C29	4.5770	0.2090	-0.9890 C
30 H30	2.8958	0.7421	-2.2238 H
31 H31	2.1077	-2.1088	0.9148 H
32 H32	4.4850	-2.0035	1.6003 H
33 H33	5.2710	0.8402	-1.5348 H
34 H34	6.0688	-0.5257	0.3821 H
35 C35	-1.2164	1.9804	-0.5986 C
36 H36	-1.4789	2.9931	-0.2838 H
37 H37	-1.1548	1.9648	-1.6928 H
38 C38	0.1164	1.5881	-0.0104 C
39 H39	1.0213	1.7774	-0.5929 H
40 O40	0.2107	1.8955	1.2864 O
41 C41	1.5088	1.7978	1.9179 C
42 H42	1.3934	2.2626	2.8955 H
43 H43	2.2558	2.3396	1.3303 H
44 H44	1.7975	0.7502	2.0263 H

Oxocarbenium ion from acetal *trans*-6

Energy = -487264.08 kcal/mol

ATOM

1 H1	0.0000	0.0000	0.0000 H
2 C2	0.0000	0.0000	1.0968 C
3 C3	2.5587	0.0000	1.2943 C
4 C4	1.2956	2.1629	1.1412 C
5 C5	2.5374	1.4802	1.7269 C
6 C6	0.0351	1.4397	1.5976 C
7 C7	1.2491	-0.7309	1.6405 C
8 H8	3.4037	-0.5098	1.7681 H
9 H9	3.4453	1.9903	1.3871 H
10 H10	1.2698	-1.7497	1.2414 H
11 H11	-0.9082	-0.5149	1.4277 H
12 H12	2.7314	-0.0531	0.2111 H
13 H13	1.3498	2.0962	0.0463 H
14 H14	2.5208	1.5537	2.8232 H
15 H15	-0.0193	1.4569	2.6964 H
16 H16	1.1609	-0.8287	2.7310 H
17 O17	-1.0651	2.2739	1.1020 O
18 C18	-2.3165	2.1354	1.8649 C
19 H19	-2.5176	1.0617	1.9159 H
20 H20	-2.1447	2.5092	2.8809 H
21 C21	-3.4361	2.8684	1.1851 C
22 C22	-5.5495	4.2209	-0.0708 C

23 C23	-3.9879	2.3711	-0.0068 C
24 C24	-3.9628	4.0402	1.7469 C
25 C25	-5.0164	4.7147	1.1224 C
26 C26	-5.0356	3.0467	-0.6338 C
27 H27	-3.6035	1.4499	-0.4371 H
28 H28	-3.5687	4.4156	2.6889 H
29 H29	-5.4263	5.6133	1.5728 H
30 H30	-5.4629	2.6509	-1.5500 H
31 H31	-6.3732	4.7384	-0.5528 H
32 C32	1.0738	3.6458	1.5090 C
33 H33	1.8649	4.2814	1.0975 H
34 H34	1.0463	3.7807	2.5949 H
35 C35	-0.2311	4.0736	0.9145 C
36 H36	-1.0087	4.5561	1.5075 H
37 O37	-0.2205	4.3092	-0.3560 O
38 C38	-1.3982	4.8662	-1.0127 C
39 H39	-1.8929	5.5777	-0.3490 H
40 H40	-2.0665	4.0442	-1.2693 H
41 H41	-1.0219	5.3606	-1.9058 H

Oxocarbenium ion from acetal *trans*-**5**

Energy = -462498.62 kcal/mol

ATOM	X	Y	Z
1 H1	0.0000	0.0000	0.0000 H
2 C2	0.0000	0.0000	1.0955 C
3 C3	1.4566	0.0000	1.6596 C
4 C4	1.7712	1.4704	2.0966 C
5 C5	0.6256	2.2773	1.4723 C
6 C6	-0.5671	1.3359	1.5933 C
7 H7	-0.5759	-0.8670	1.4285 H
8 H8	2.1609	-0.3450	0.8989 H
9 H9	1.5542	-0.6802	2.5092 H
10 H10	1.7507	1.5610	3.1891 H
11 H11	2.7552	1.8076	1.7615 H
12 H12	0.8245	2.4114	0.3998 H
13 H13	-0.8602	1.2504	2.6516 H
14 O14	-1.6300	1.9421	0.8572 O
15 C15	-2.9490	1.3258	0.9736 C
16 H16	-2.8104	0.2405	1.0076 H
17 H17	-3.4506	1.5774	0.0372 H
18 C18	-3.7484	1.8168	2.1566 C
19 C19	-5.2623	2.7505	4.3380 C
20 C20	-3.6189	1.2238	3.4235 C
21 C21	-4.6600	2.8723	1.9958 C

22 C22	-5.4145	3.3356	3.0767 C
23 C23	-4.3640	1.6927	4.5099 C
24 H24	-2.9601	0.3696	3.5584 H
25 H25	-4.7957	3.3188	1.0137 H
26 H26	-6.1330	4.1368	2.9312 H
27 H27	-4.2639	1.2158	5.4802 H
28 H28	-5.8576	3.0999	5.1762 H
29 C29	0.2650	3.6488	2.0712 C
30 H30	0.1065	3.5917	3.1511 H
31 H31	1.0857	4.3641	1.8872 H
32 C32	-0.9023	4.2147	1.3575 C
33 H33	-0.9095	4.2361	0.2693 H
34 O34	-1.8661	4.8727	1.8620 O
35 C35	-2.0723	5.0135	3.3119 C
36 H36	-1.1869	5.4643	3.7601 H
37 H37	-2.2986	4.0300	3.7245 H
38 H38	-2.9314	5.6732	3.3950 H

Oxocarbenium ion from acetal *cis*-**5**

Energy = -462507.04 kcal/mol

ATOM

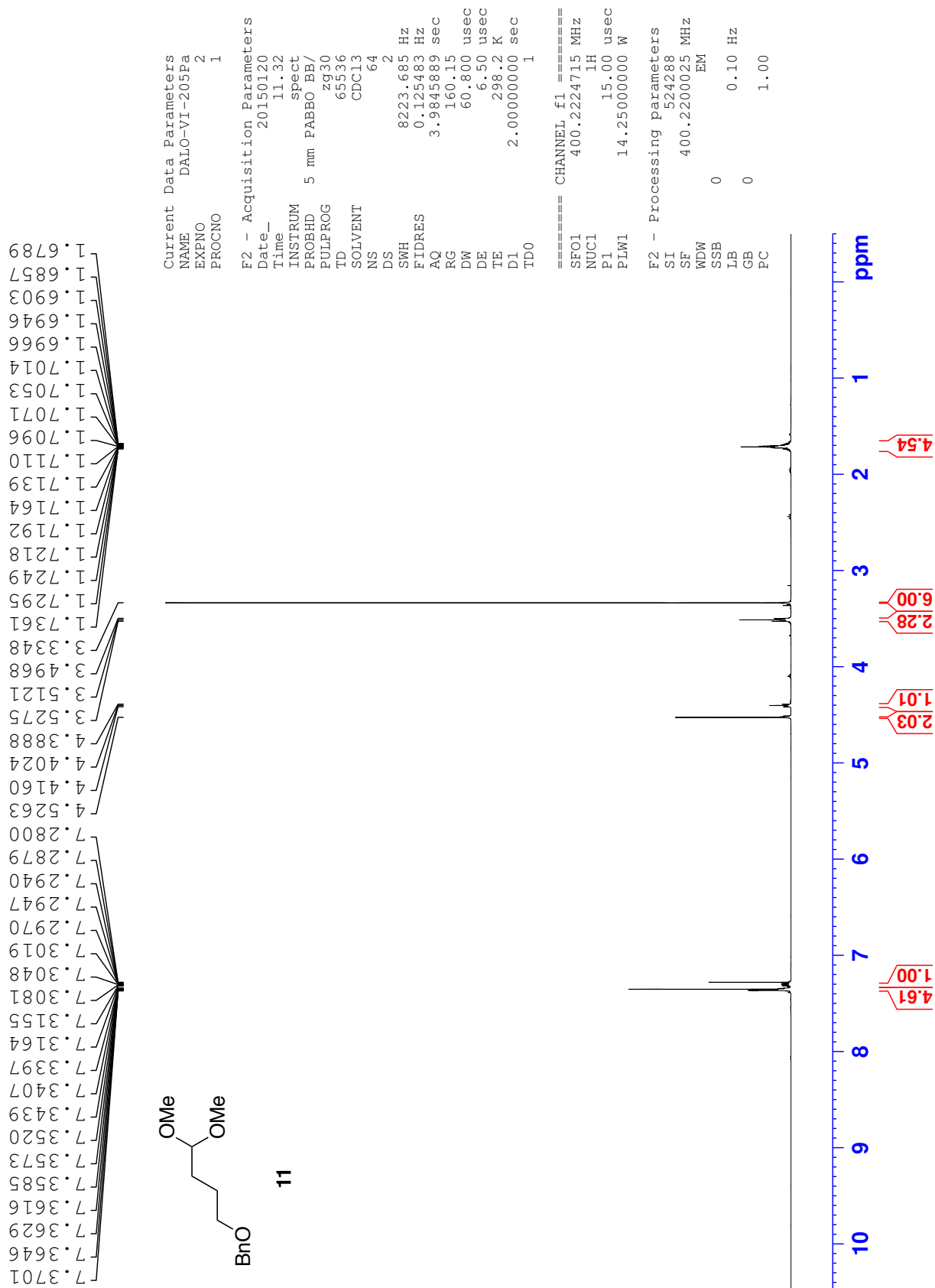
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5 C5	0.1880	1.7288	2.8745 C
6 C6	-0.6693	1.2528	1.6797 C
7 H7	-0.5552	-0.8834	1.4252 H
8 H8	2.0861	0.6429	1.1273 H
9 H9	1.8588	-0.9993	1.7259 H
10 H10	0.7731	-0.1700	3.7817 H
11 H11	2.1239	0.9520	3.5772 H
12 H12	0.7309	2.6327	2.5794 H
13 H13	-0.8661	2.0417	0.9523 H
14 O14	-2.0295	0.9197	2.2616 O
15 C15	-3.1979	1.5263	1.5160 C
16 H16	-3.1667	2.6042	1.6961 H
17 H17	-2.9546	1.3175	0.4735 H
18 C18	-4.4933	0.9057	1.9261 C
19 C19	-6.9512	-0.2342	2.6400 C
20 C20	-4.8372	-0.3784	1.4714 C
21 C21	-5.3989	1.6181	2.7276 C
22 C22	-6.6240	1.0499	3.0835 C
23 C23	-6.0583	-0.9474	1.8319 C

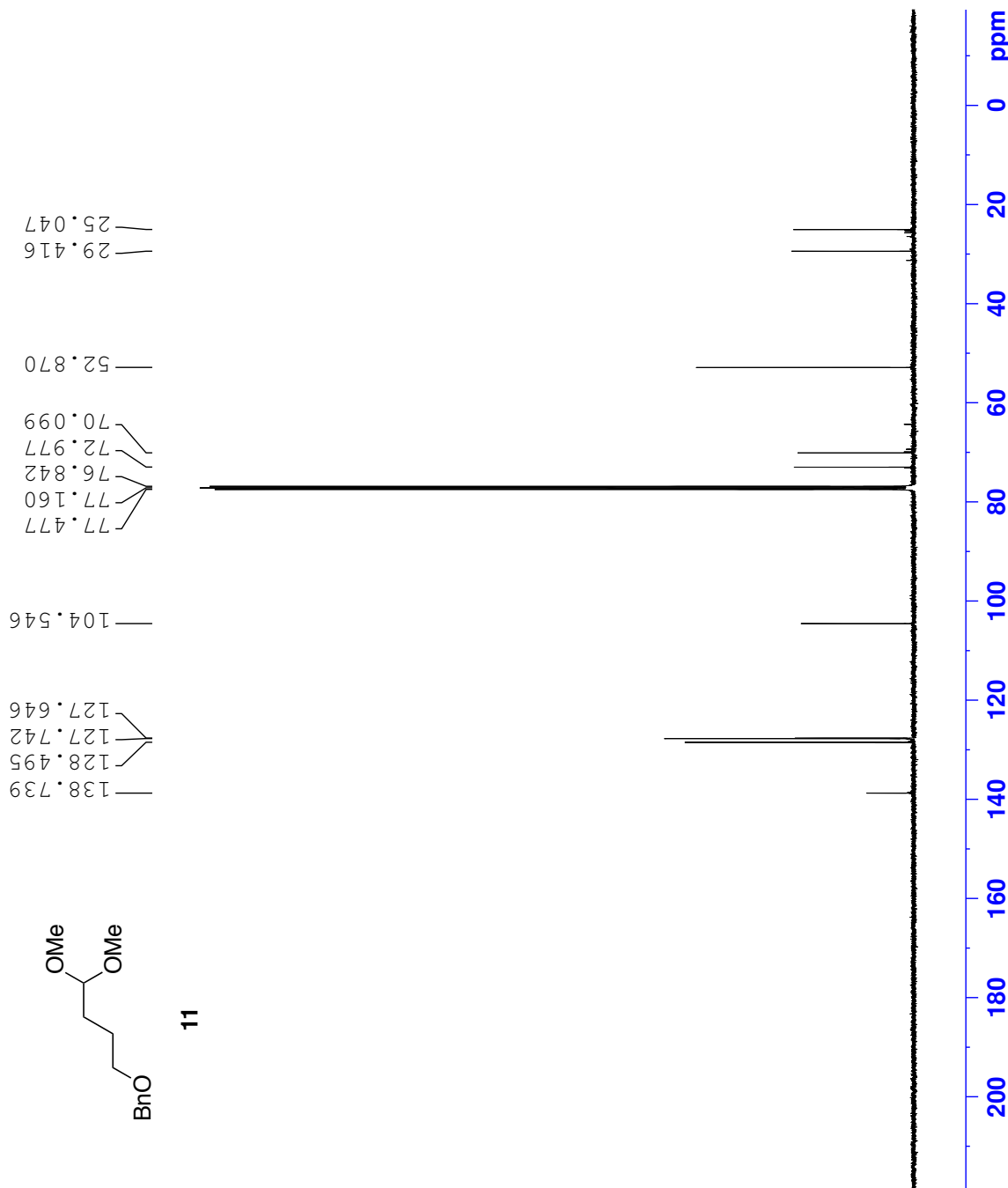
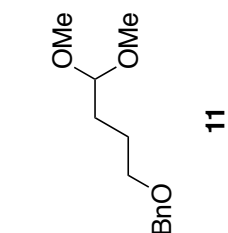
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27 H27	-6.3214	-1.9369	1.4718 H
28 H28	-7.9061	-0.6740	2.9106 H
29 C29	-0.8026	2.0702	4.0022 C
30 H30	-1.1810	3.0909	3.8888 H
31 H31	-0.3539	1.9853	4.9951 H
32 C32	-1.9620	1.1044	3.8872 C
33 H33	-2.9651	1.4848	4.0944 H
34 O34	-1.7027	-0.0921	4.4241 O
35 C35	-2.8093	-1.0059	4.6124 C
36 H36	-2.4210	-1.8109	5.2340 H
37 H37	-3.1437	-1.3967	3.6491 H
38 H38	-3.6331	-0.4997	5.1240 H

V. References

- (1) Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. *Organometallics* **1996**, *15*, 1518.
- (2) Garcia, A.; Otte, D. A. L.; Salamant, W. A.; Sanzone, J. R.; Woerpel, K. A. *Angew. Chem. Int. Ed.* **2015**, in press.

VI. Analytical Data





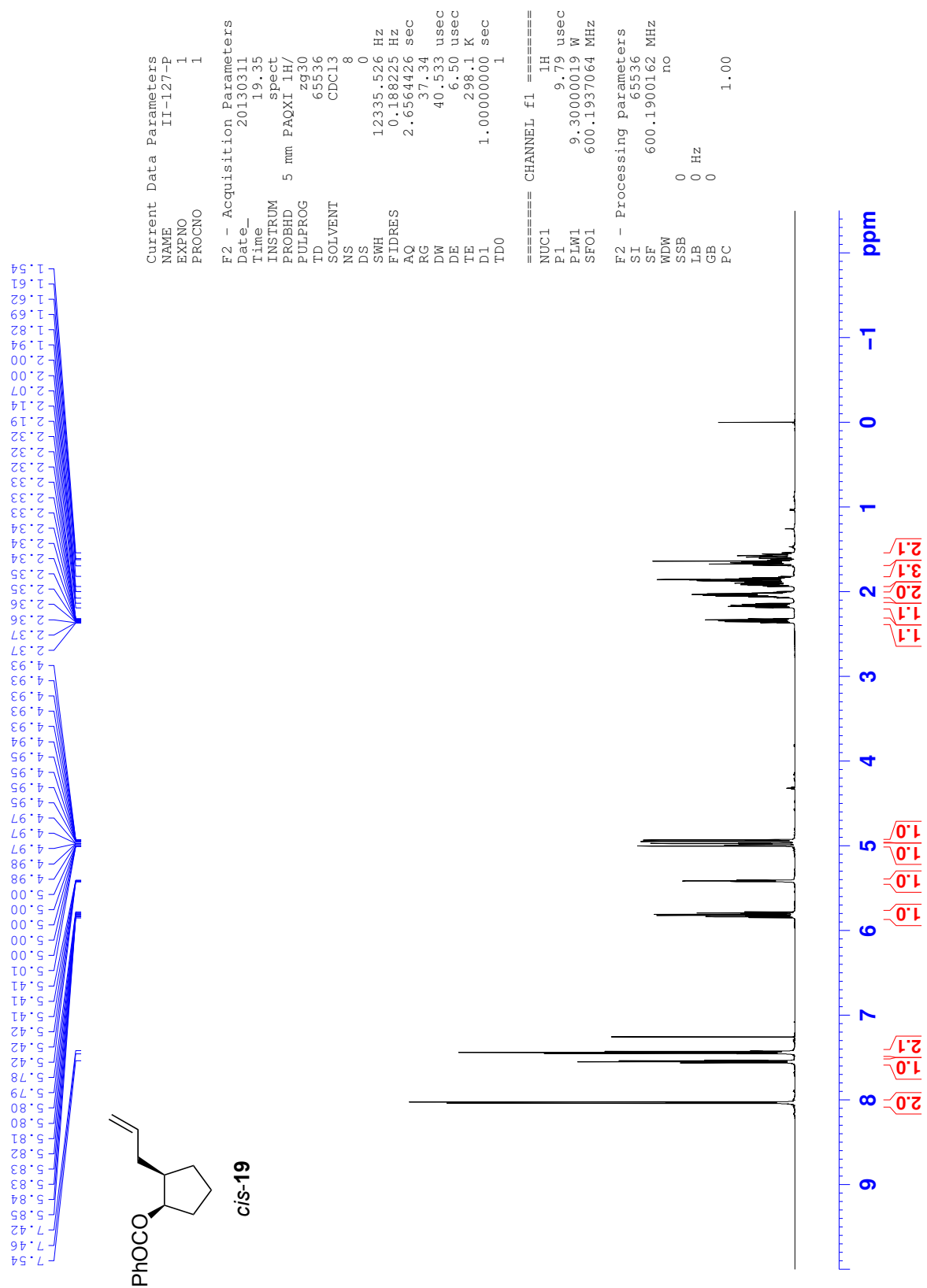
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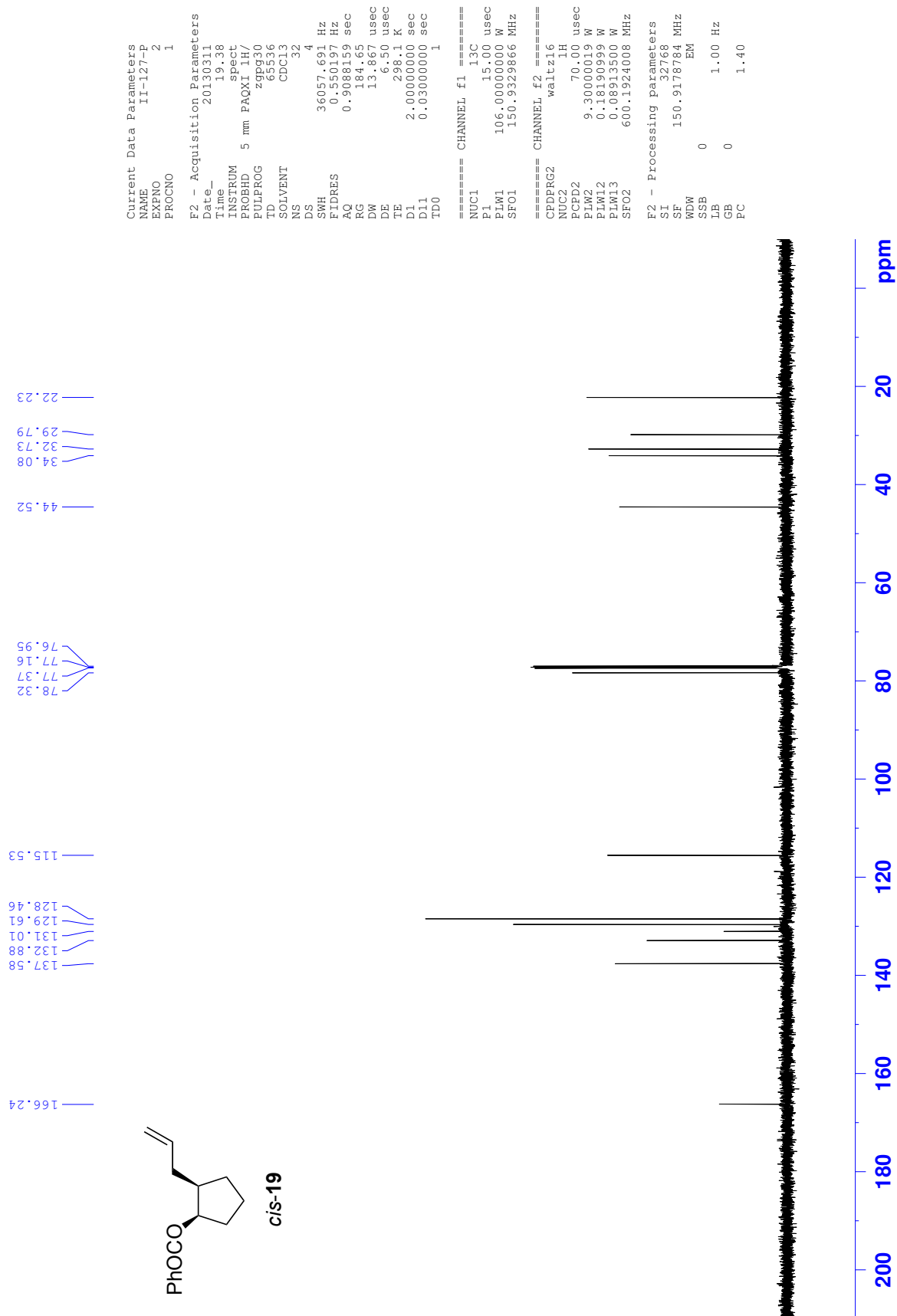
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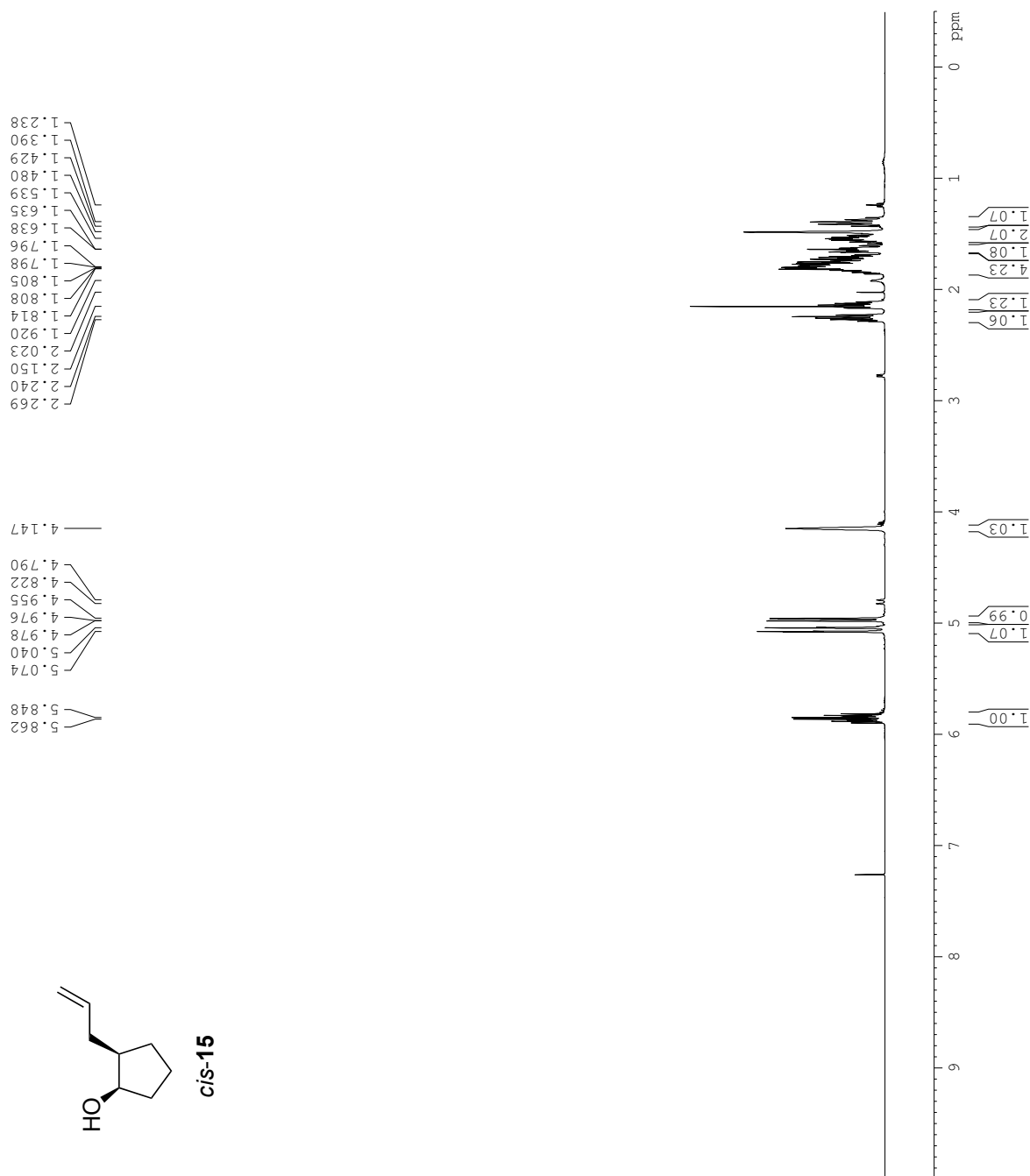
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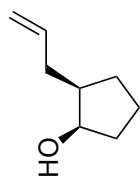
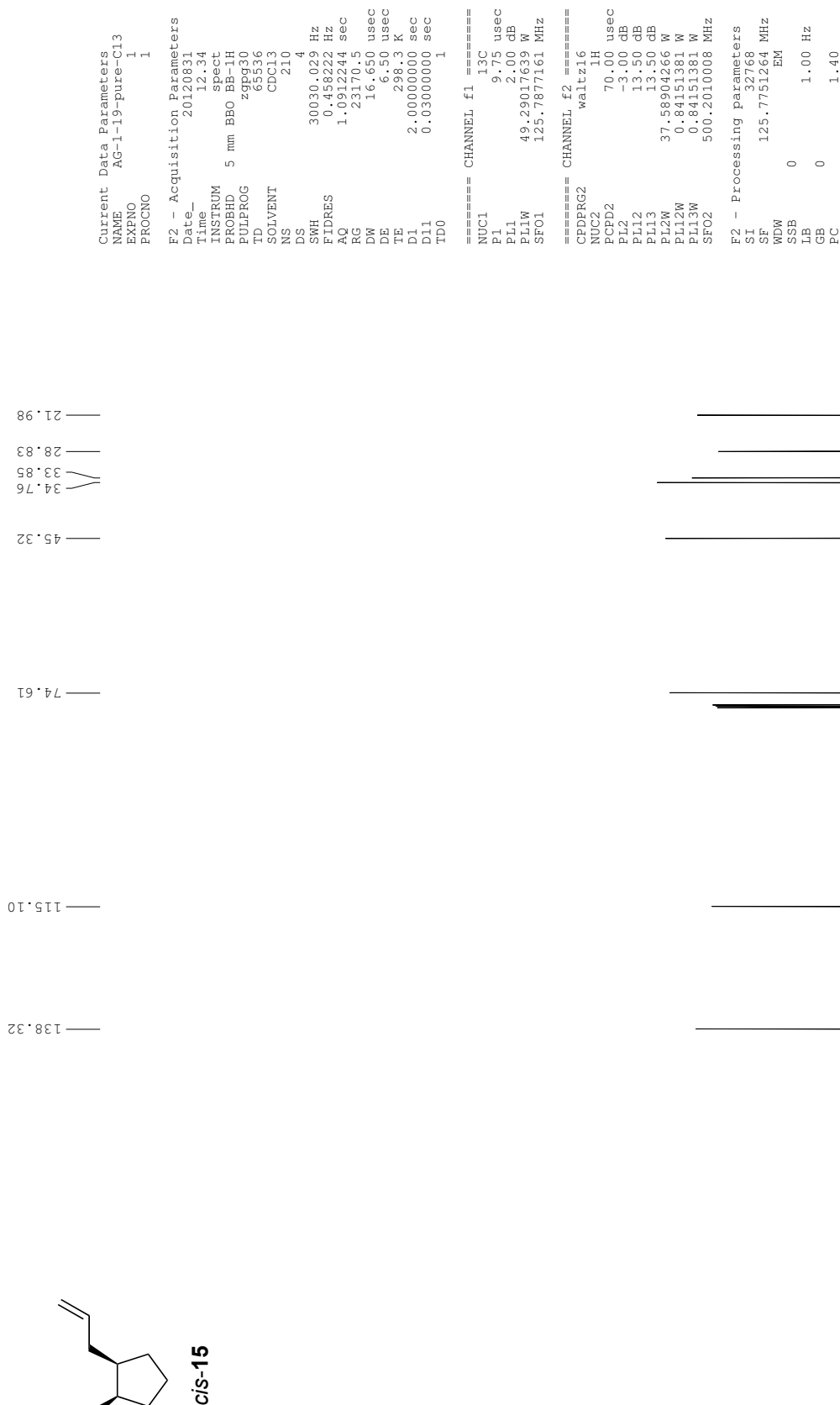
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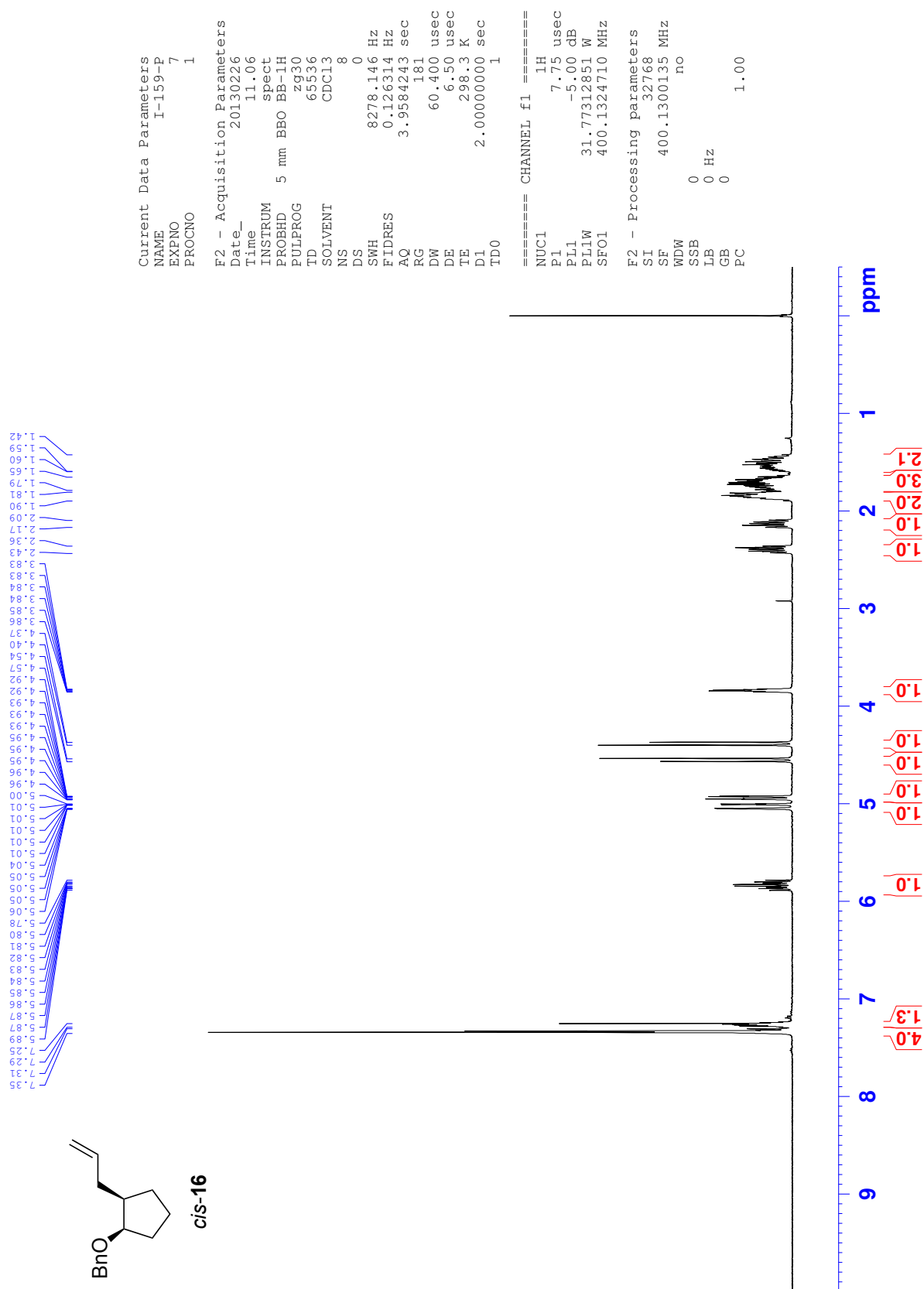
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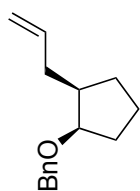






**cis-15**



**cis-16**

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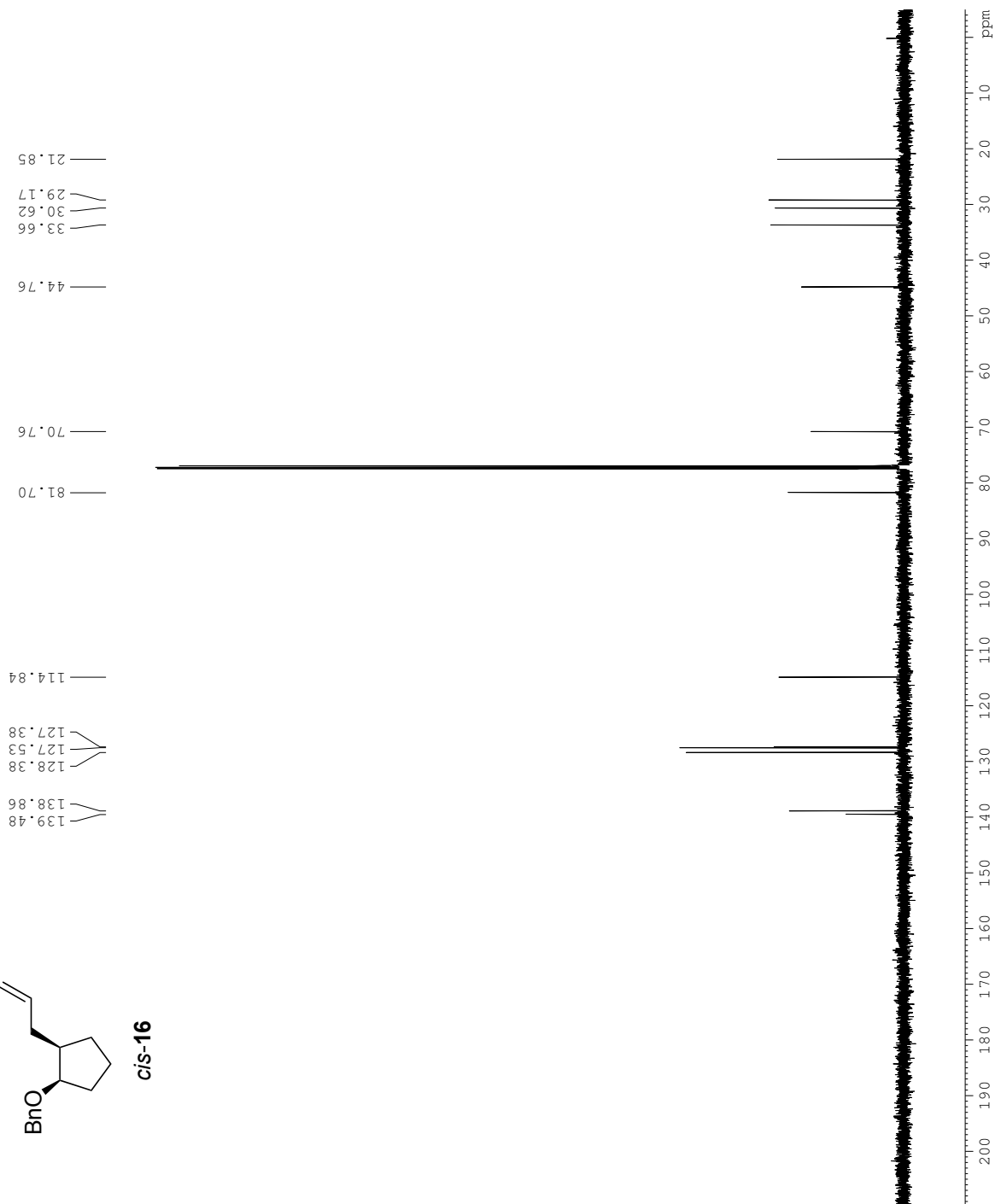
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INSTRUM   spect
PROBHD    5 mm BBO BB-1H
PULPROG   zgpg30
TD        65536
SOLVENT   CDCl3
NS         256
DS         0
SWH        30030.029 Hz
FIDRES     0.458222 Hz
AQ         1.0912244 sec
RG         23170.5
DW         16.650 usec
DE         6.50 usec
TE         298.6 K
D1         2.0000000 sec
D11        0.0300000 sec
TD0        1

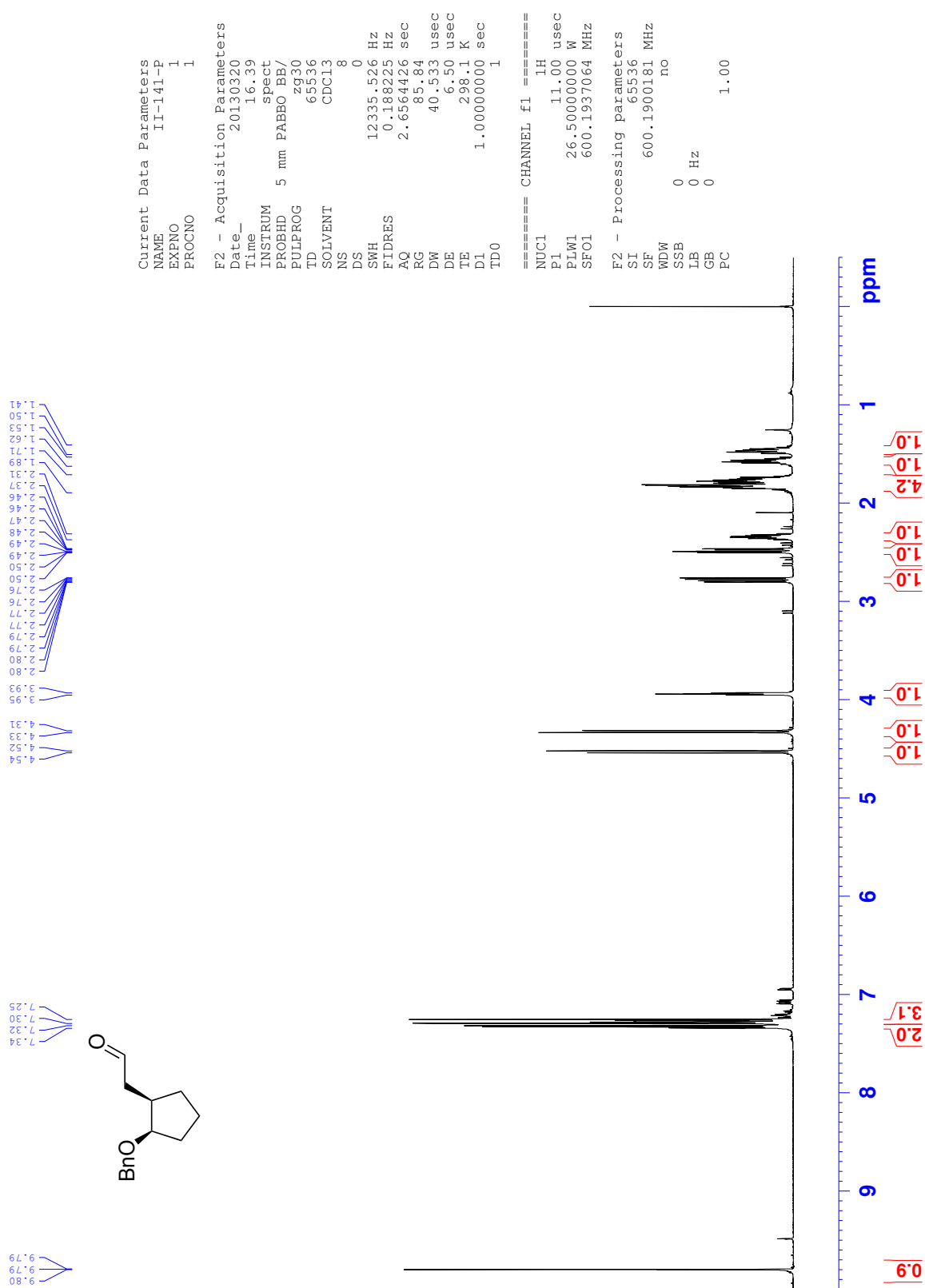
===== CHANNEL f1 =====
NUC1       13C
P1         9.75 usec
PL1        2.00 dB
PL1W       49.29017639 W
SFO1       125.7877161 MHz

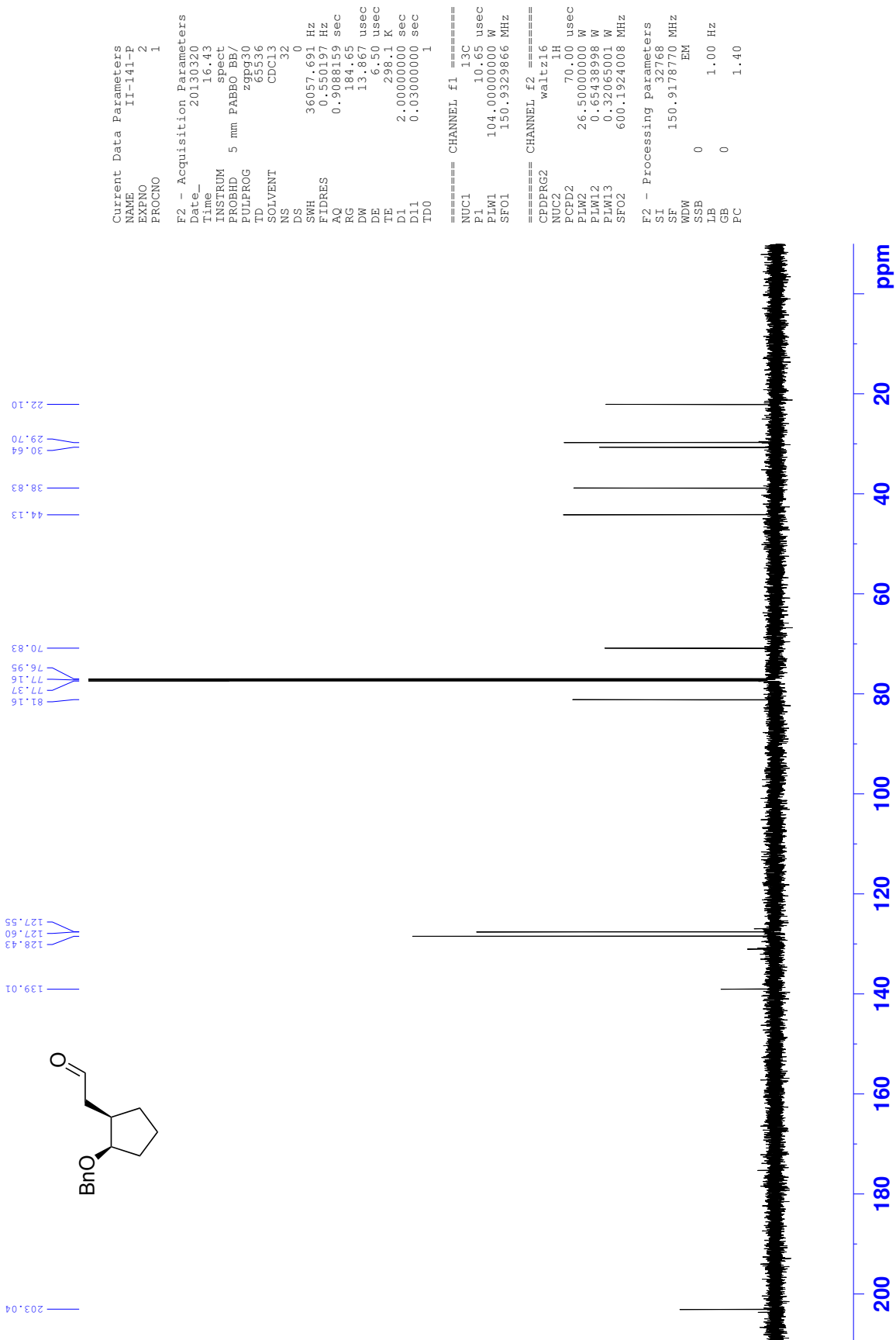
===== CHANNEL f2 =====
CPDPRG2    waltz16
NUC2        1H
PCPD2       70.00 usec
PL2         -3.00 dB
PL12        13.50 dB
PL13        13.50 dB
PL2W        37.58904266 W
PL12W       0.84151381 W
PL13W       0.84151381 W
SFO2       500.2010008 MHz

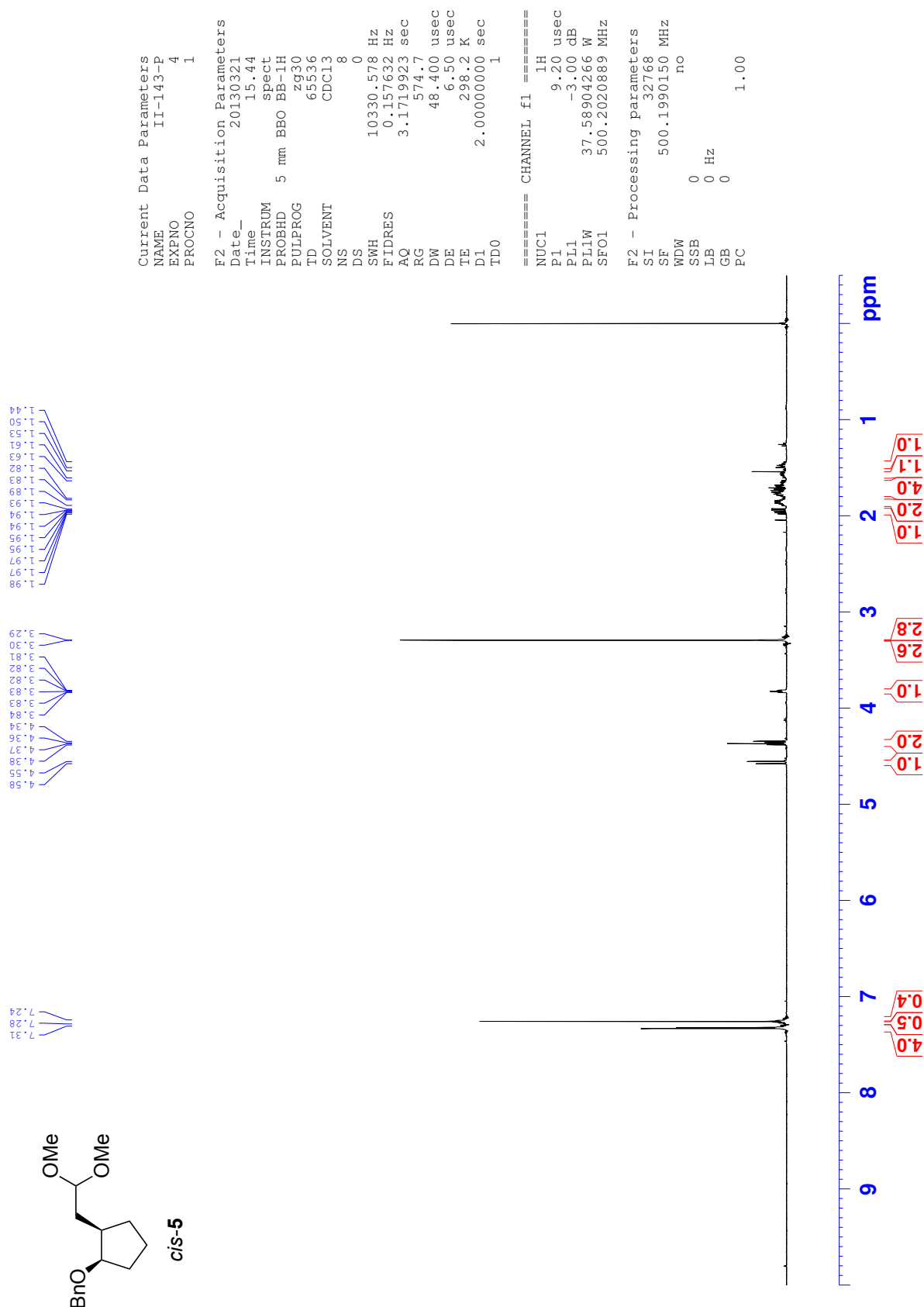
F2 - Processing parameters
SI         32768
SF         125.7751222 MHz
WDW        EM
SSB        0
LB         1.00 Hz
GB         0
PC         1.40

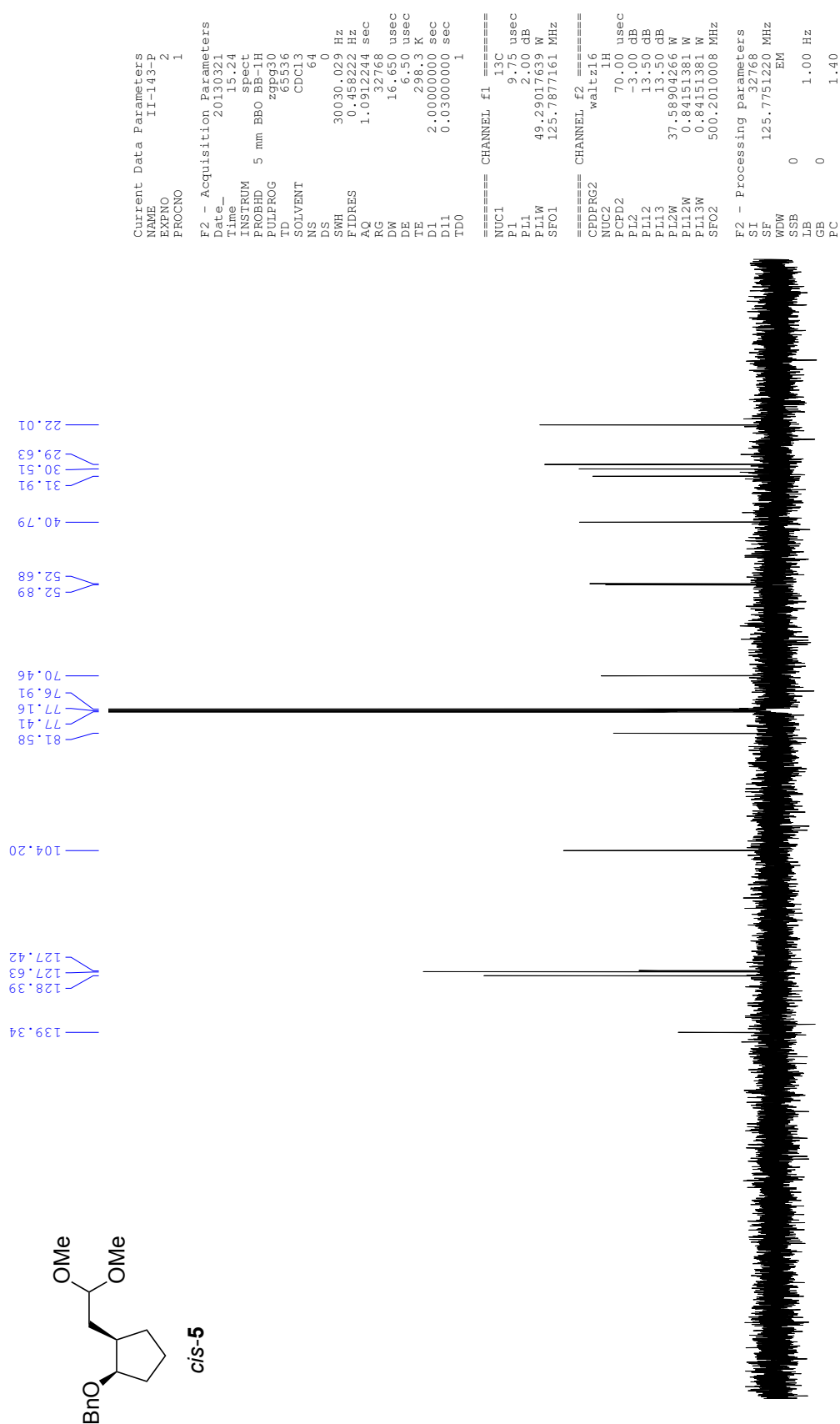
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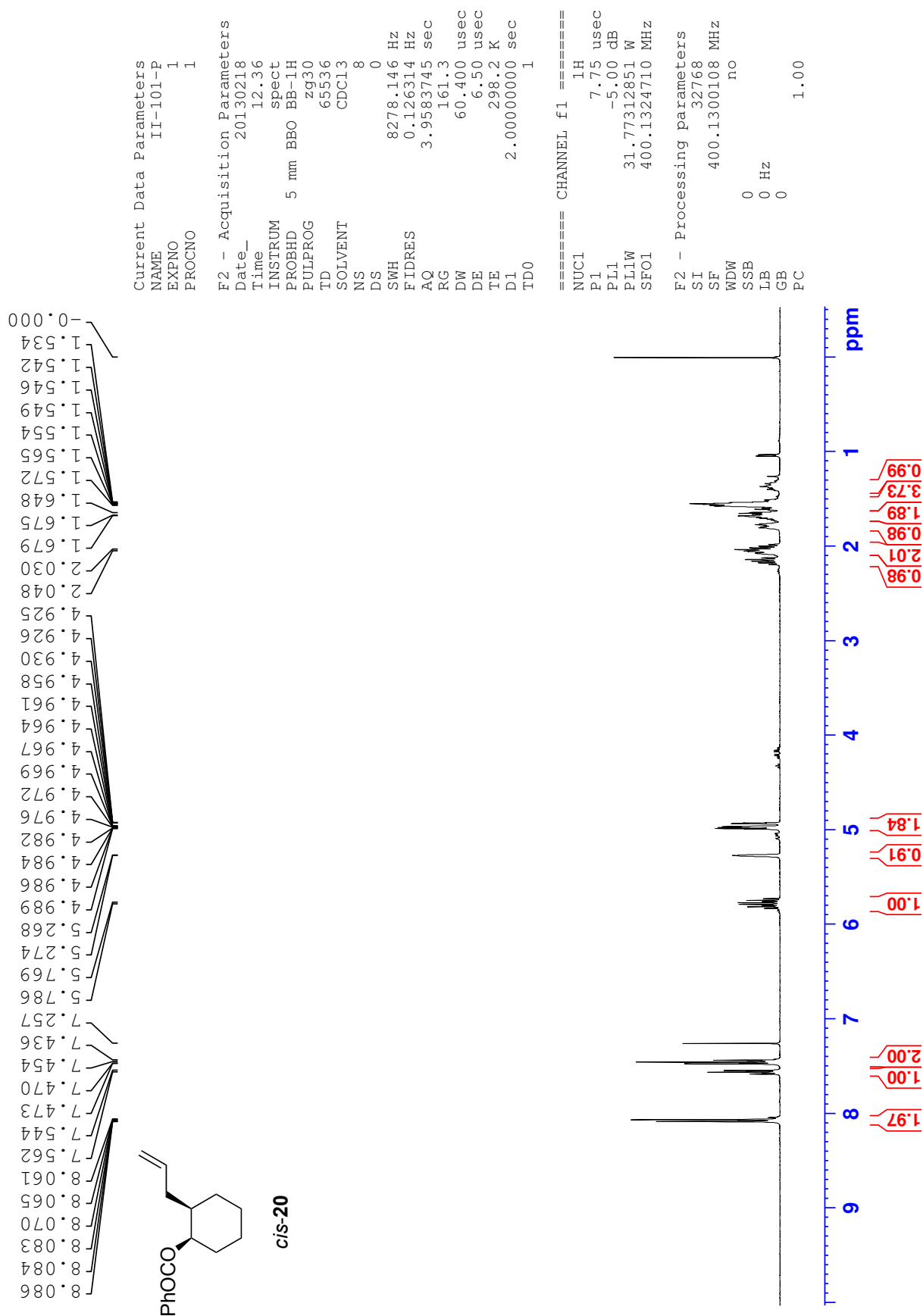


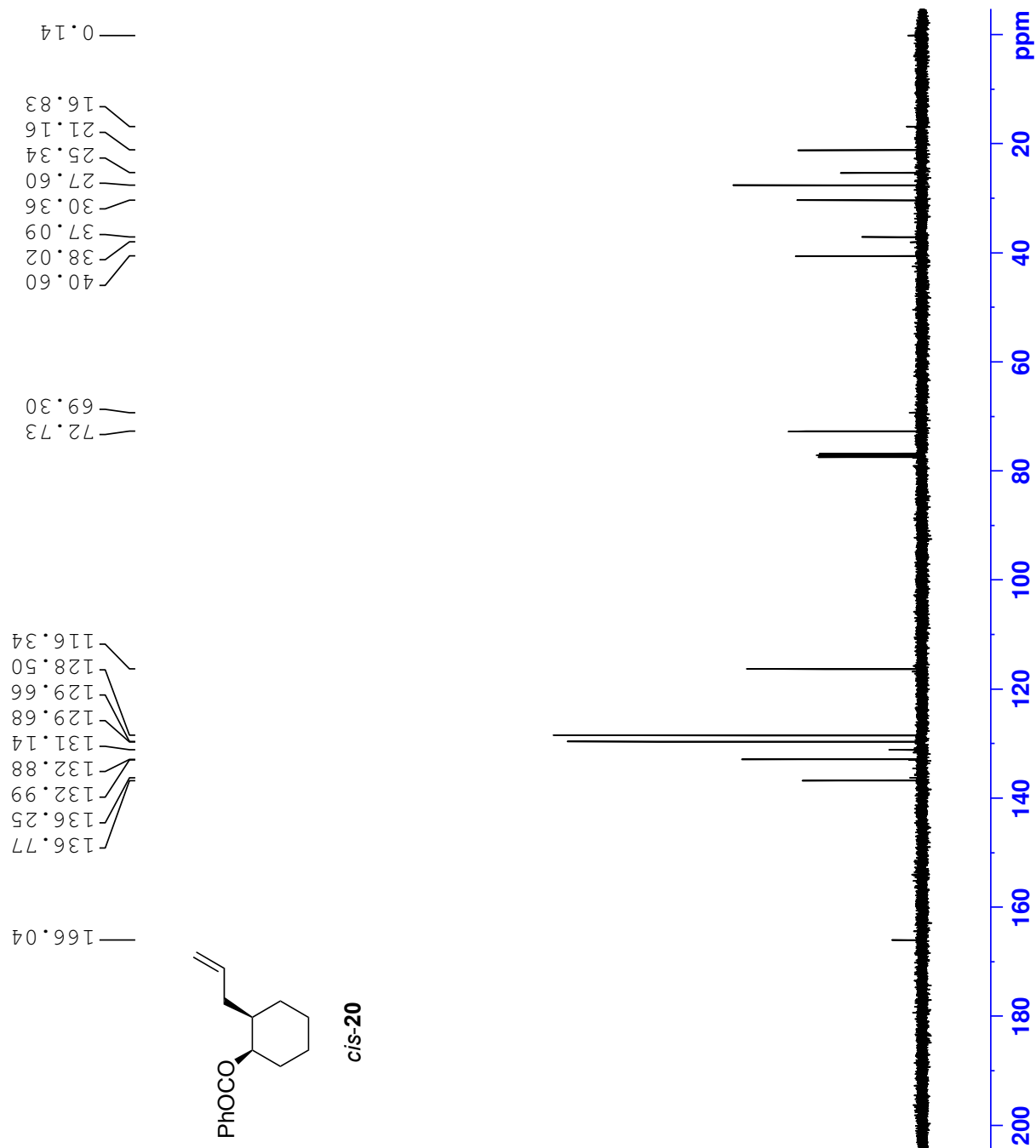


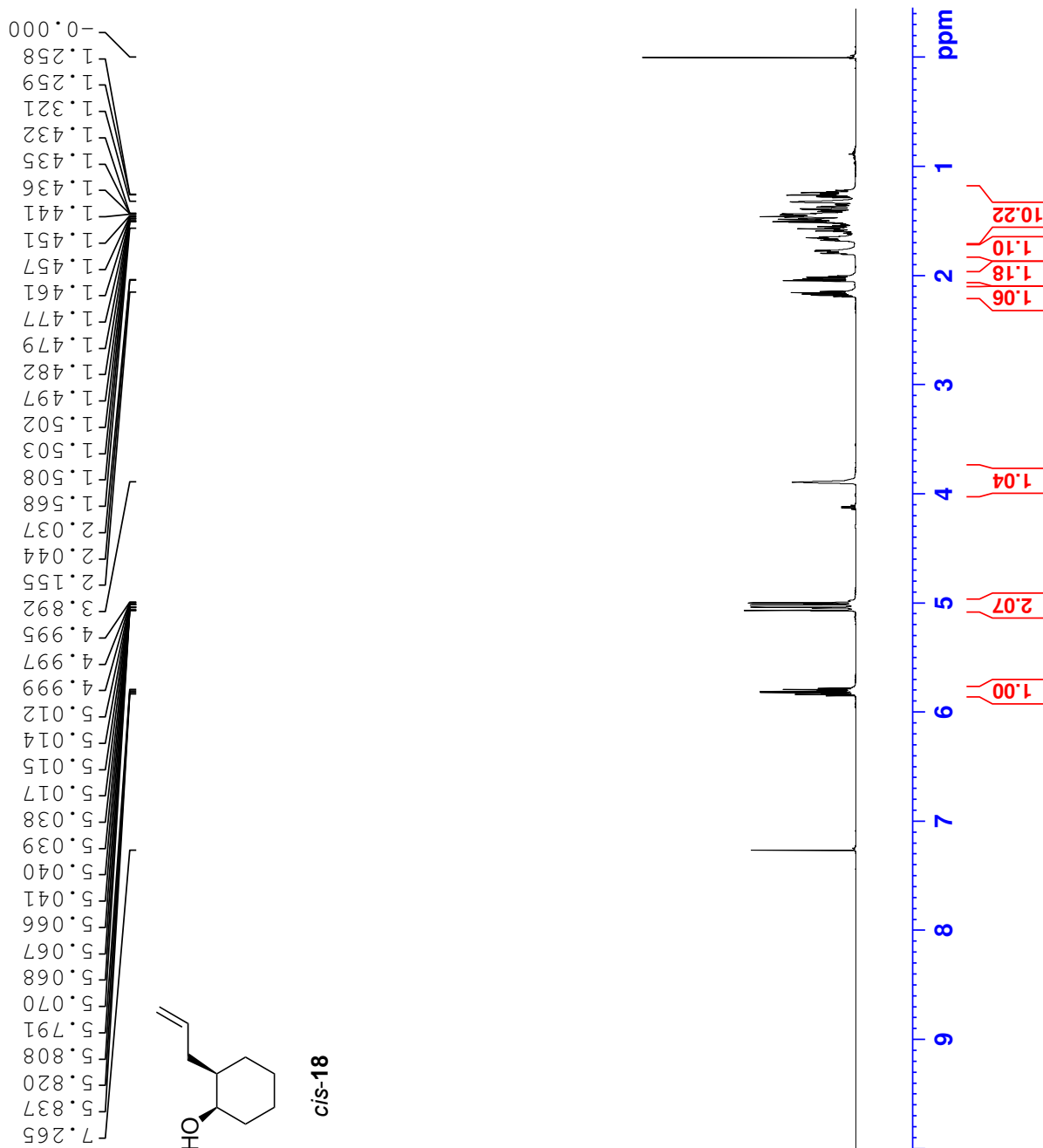


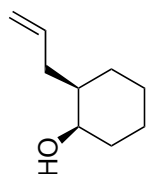
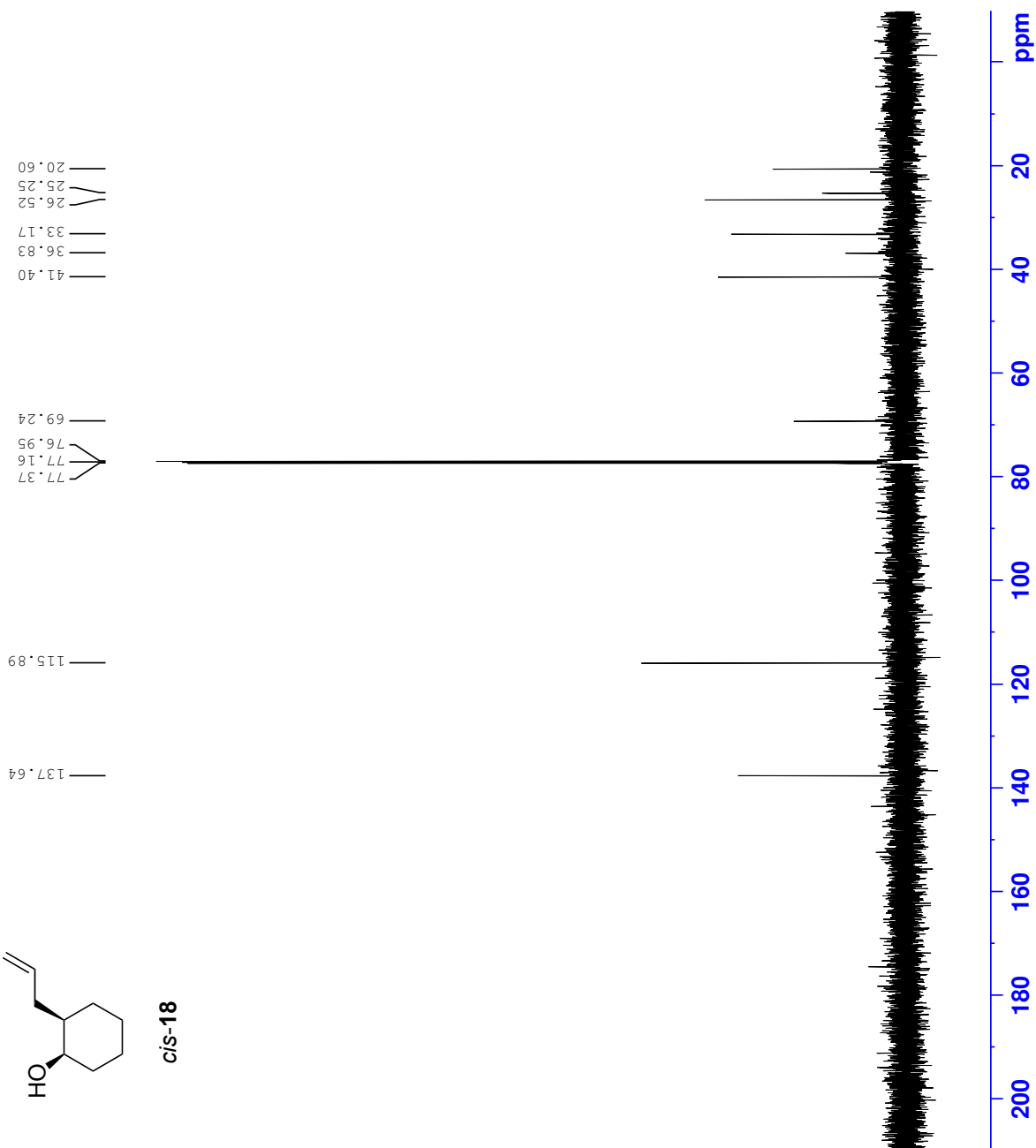










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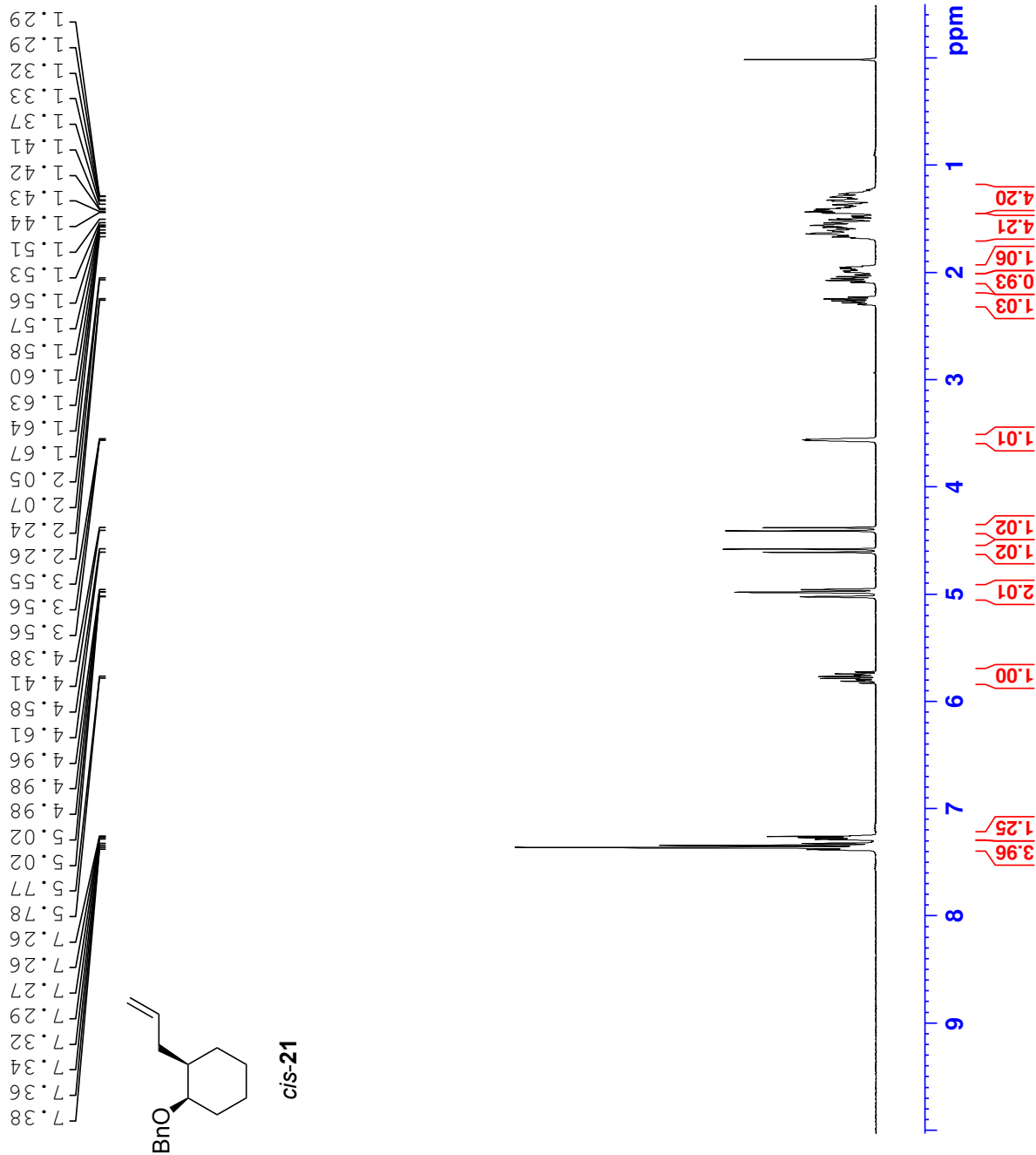
Current Data Parameters
NAME 11-102-P
EXPNO 2
PROCNO 1

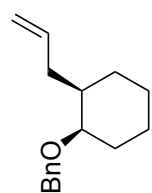
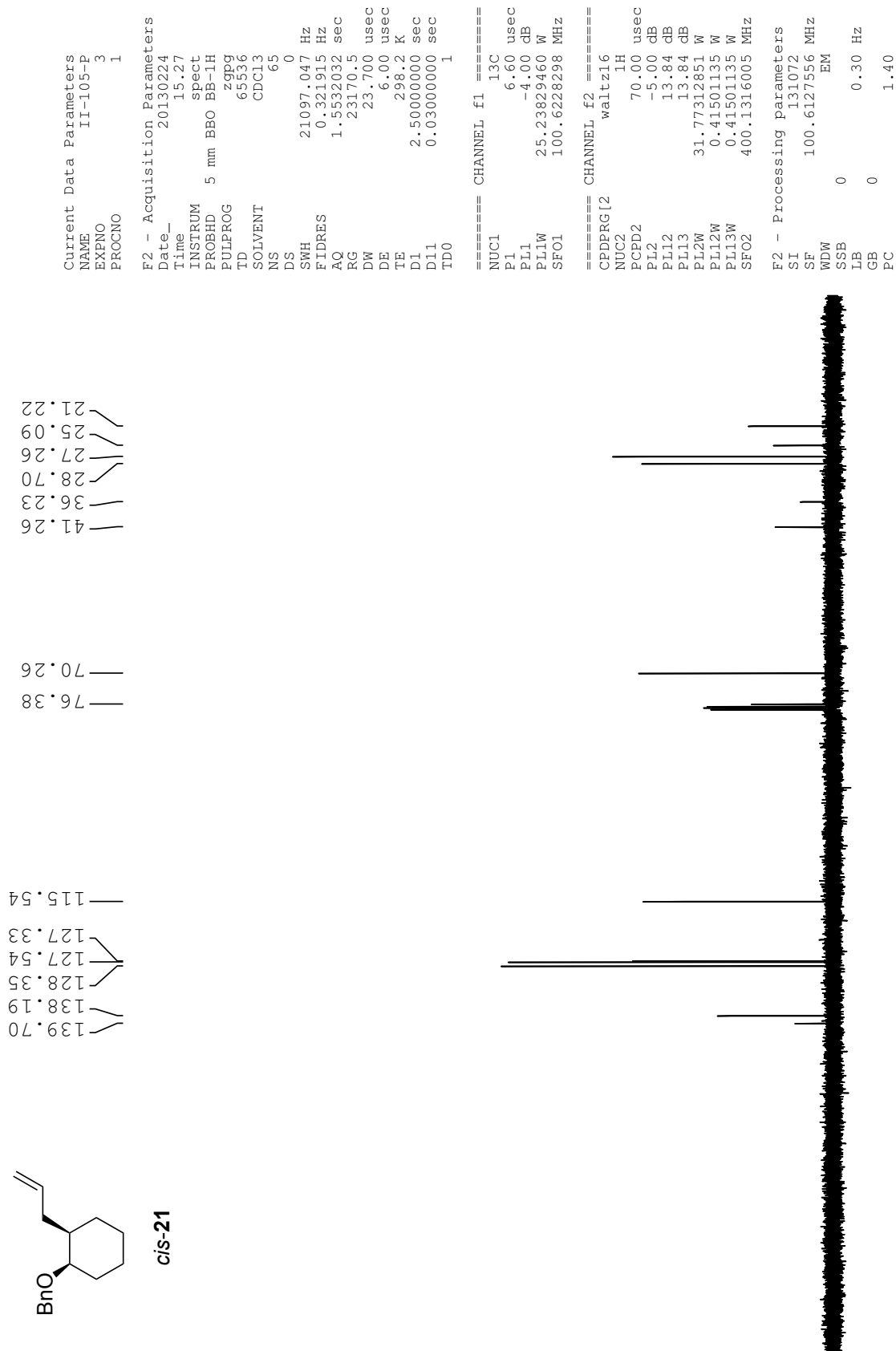
F2 - Acquisition Parameters
Date_ 20130219
Time 15.38
INSTRUM spect
PROBHD 5 mm PAQXI 1H/
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 32
DS 0
SWH 36057.691 Hz
FIDRES 0.550197 Hz
AQ 0.9087659 sec
RG 184.65
DW 13.867 usec
DE 6.50 usec
TE 298.2 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1

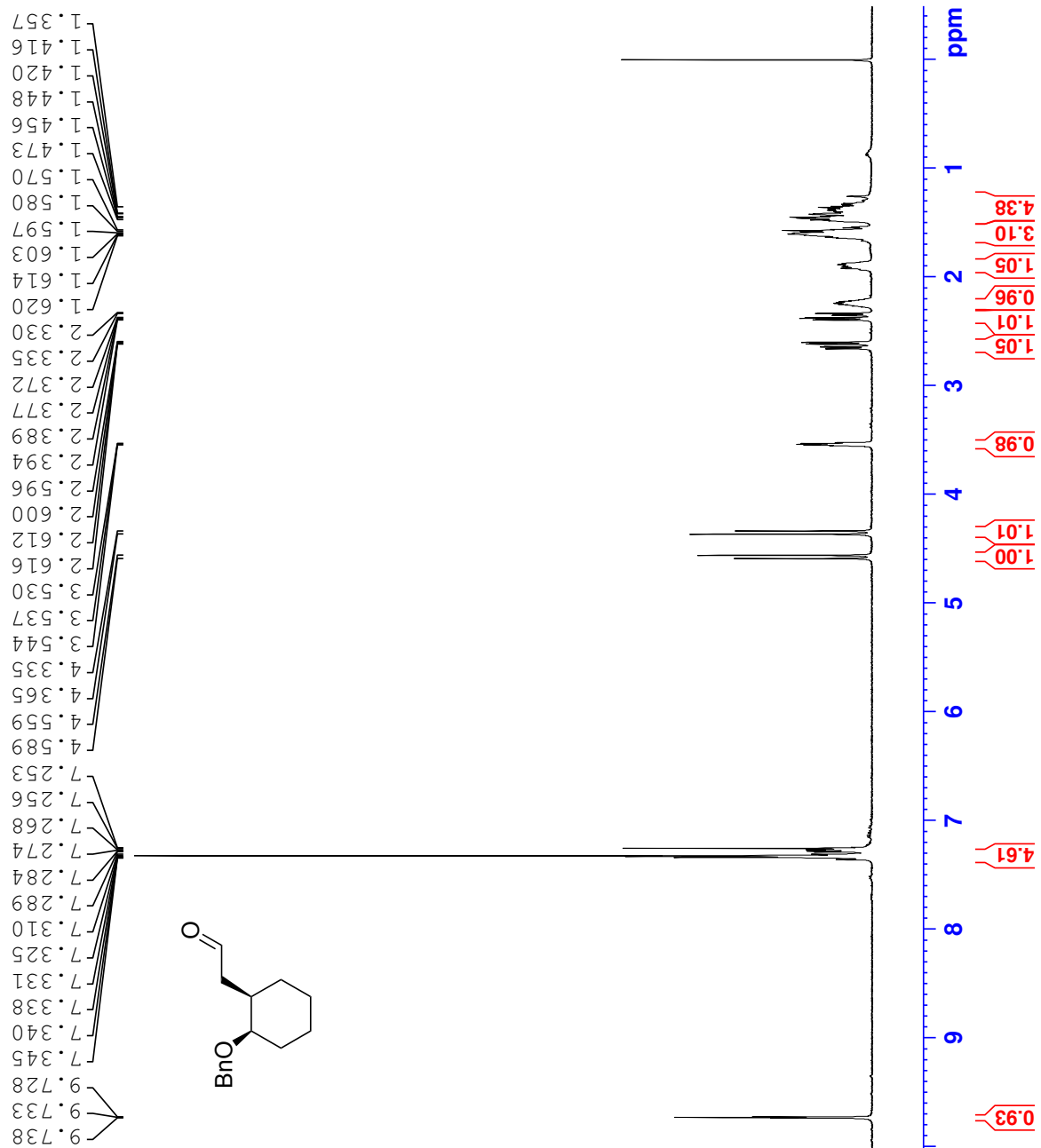
==== CHANNEL f1 =====
NUC1 13C
P1 15.00 usec
PLW1 106.00000000 W
SFO1 150.9329866 MHz

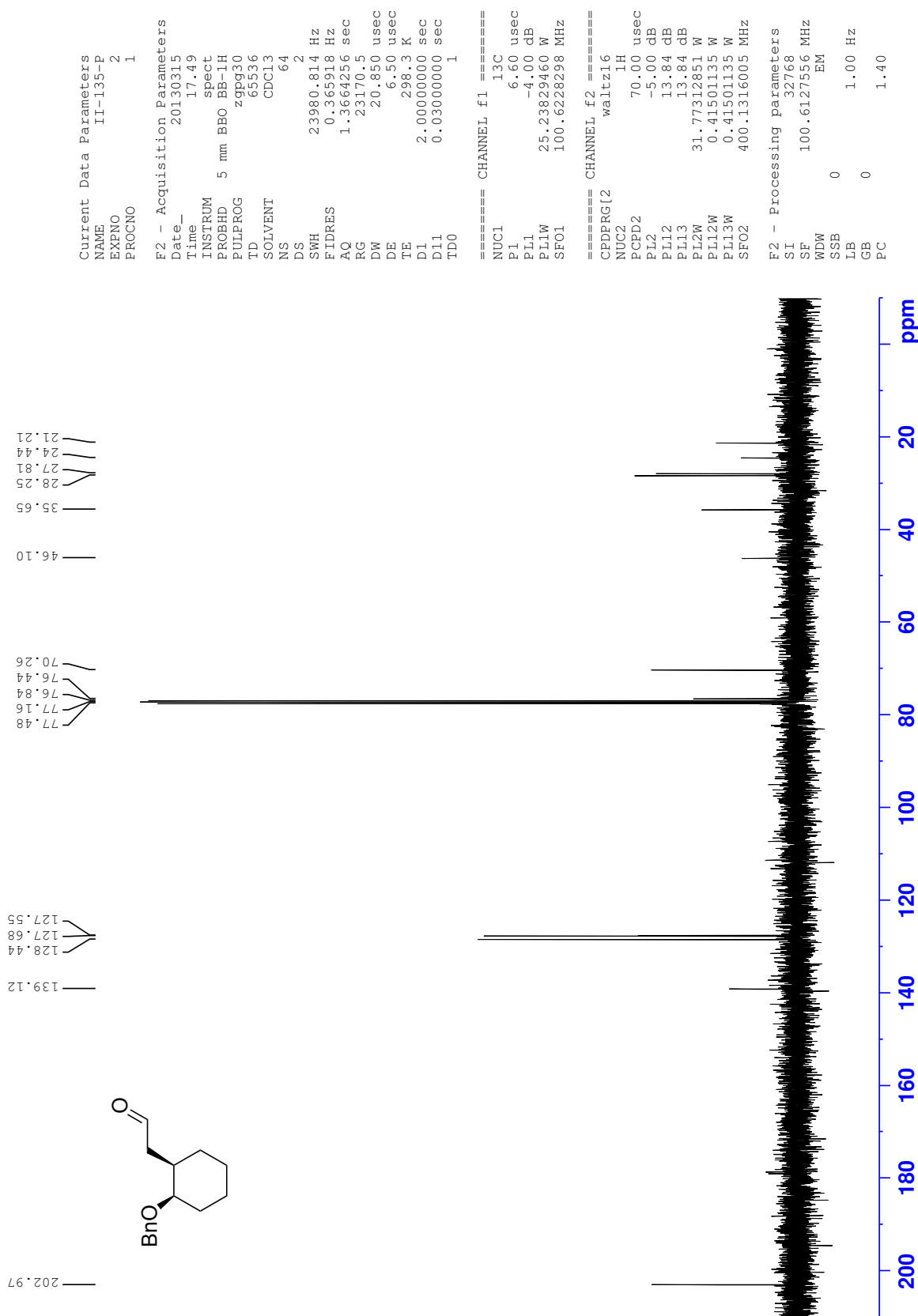
==== CHANNEL f2 =====
CPDPRG[2] waltz16
NUC2 1H
PCPD2 70.00 usec
PLW2 9.30000019 W
PLW12 0.18190999 W
PLW13 0.08913500 W
SFO2 600.1924008 MHz

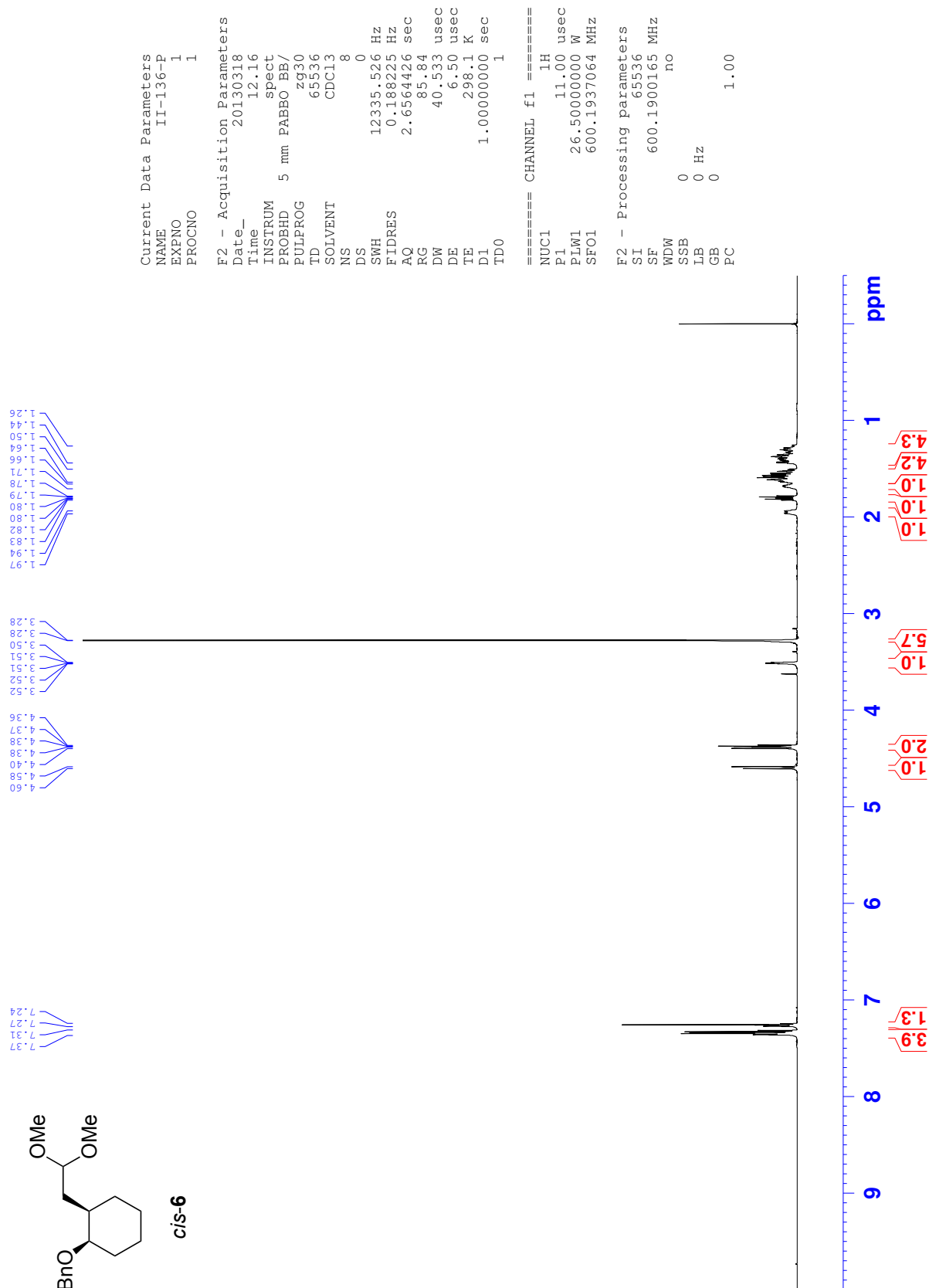
F2 - Processing parameters
SI 32768
SF 150.9178749 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

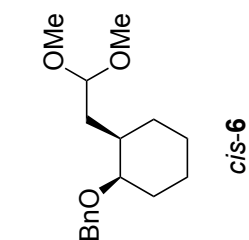


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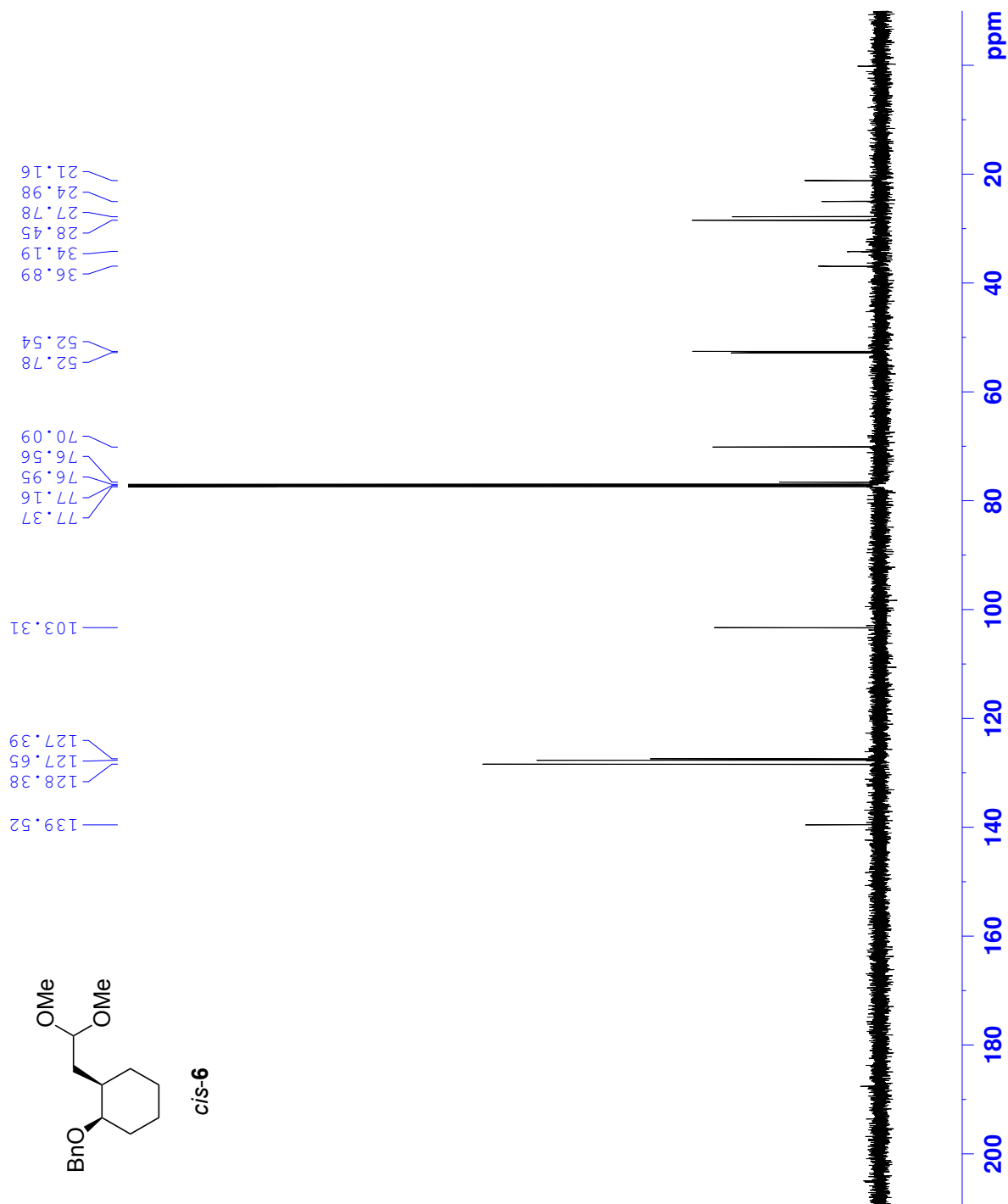
Current Data Parameters
NAME 11-136-P
EXPNO 2
PROCNO 1

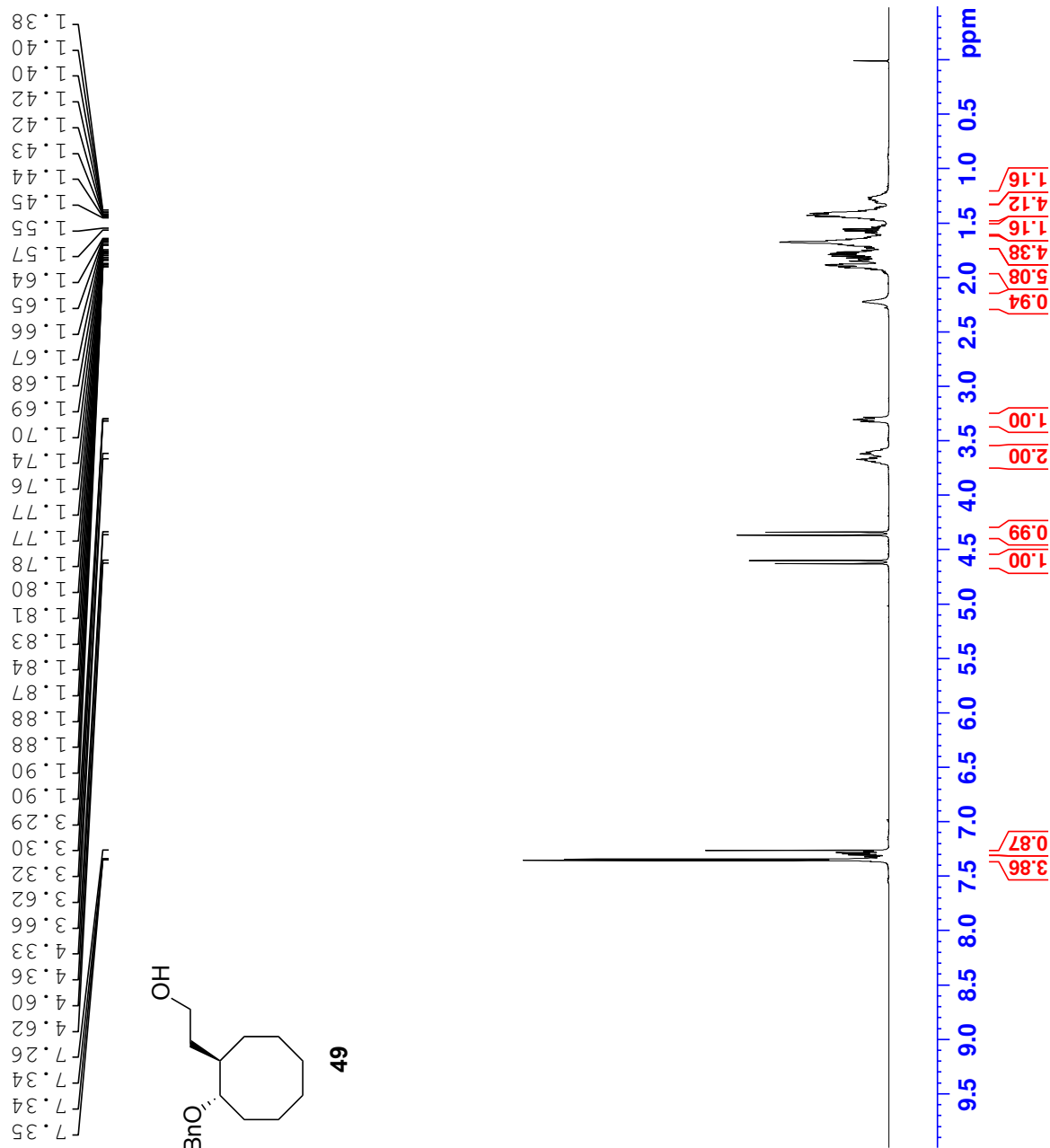
F2 - Acquisition Parameters
Date_ 20130318
Time 12.20
INSTRUM spect
PROBHD 5 mm PABBO BB/
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 64
DS 0
SWH 36057.691 Hz
FIDRES 0.550197 Hz
AQ 0.9088159 sec
RG 184.65
DW 13.867 usec
DE 6.50 usec
TE 298.1 K
D1 2.0000000 sec
D11 0.0300000 sec
TD0 1

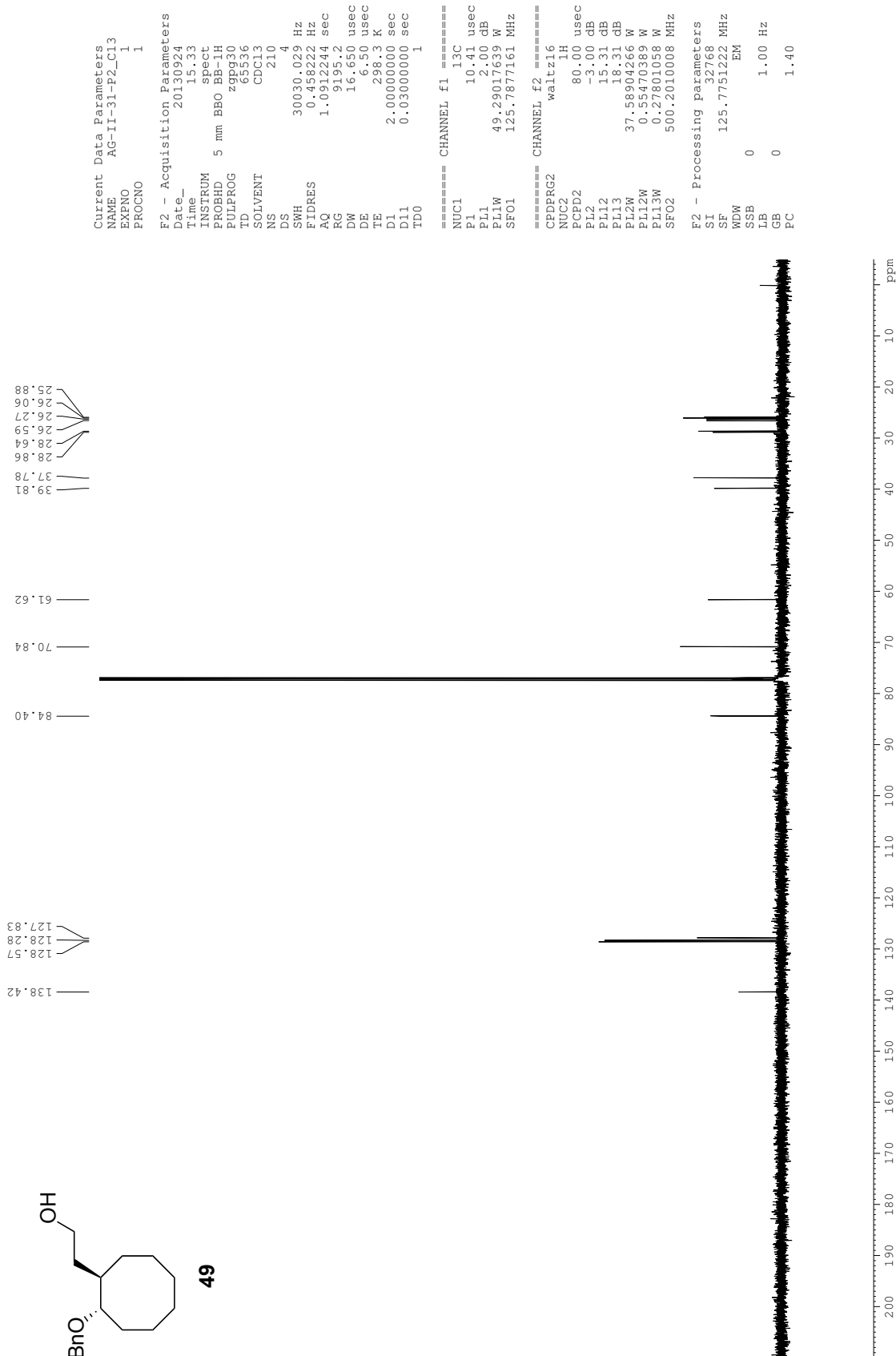
===== CHANNEL f1 =====
NUC1 13C
P1 10.65 usec
PLW1 104.0000000 W
SFO1 150.9329866 MHz

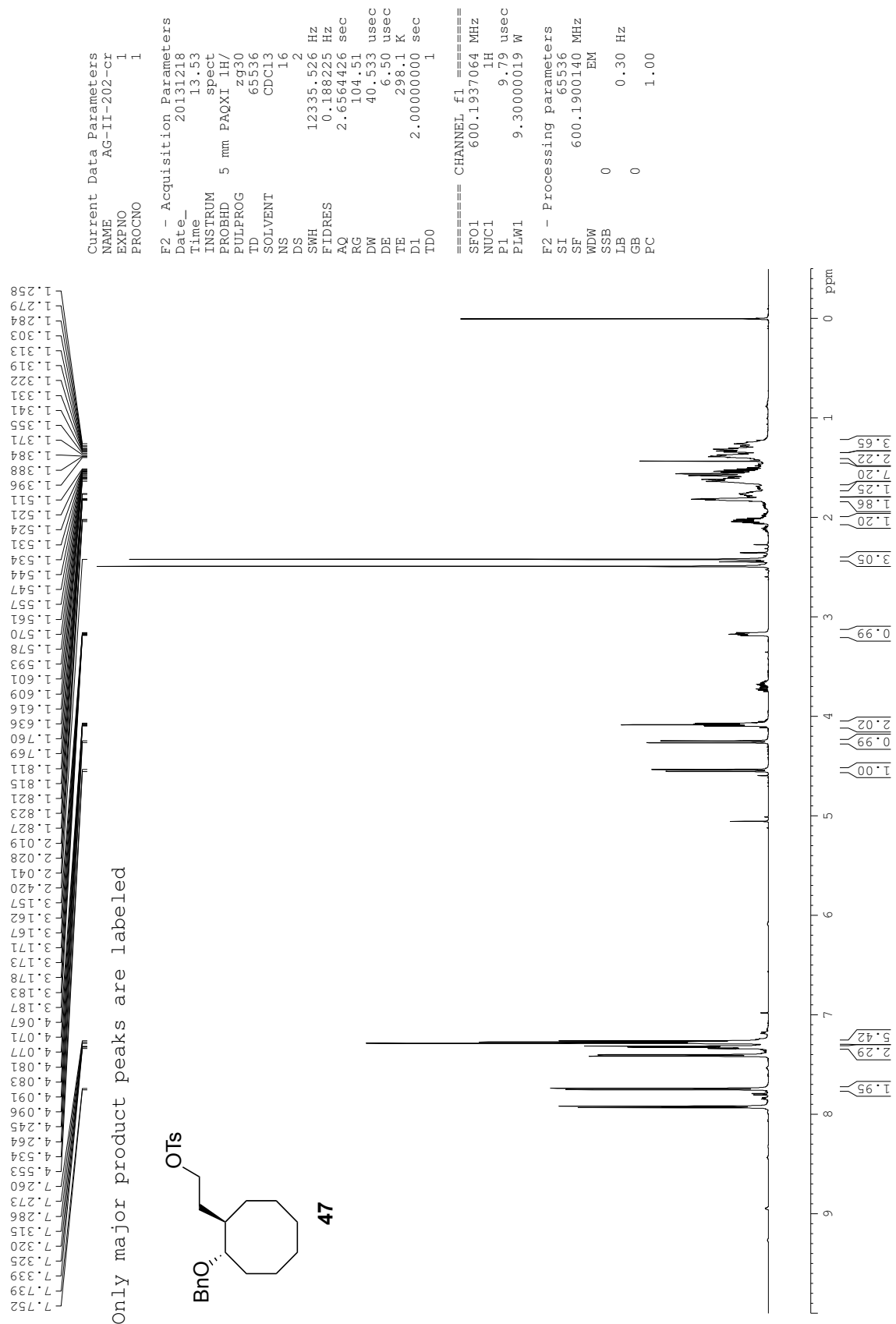
===== CHANNEL f2 =====
CPDPRG2 waltz16
NUC2 1H
PCPD2 70.00 usec
PLW2 26.5000000 W
PLW12 0.65438998 W
PLW13 0.32065001 W
SFO2 600.1924008 MHz

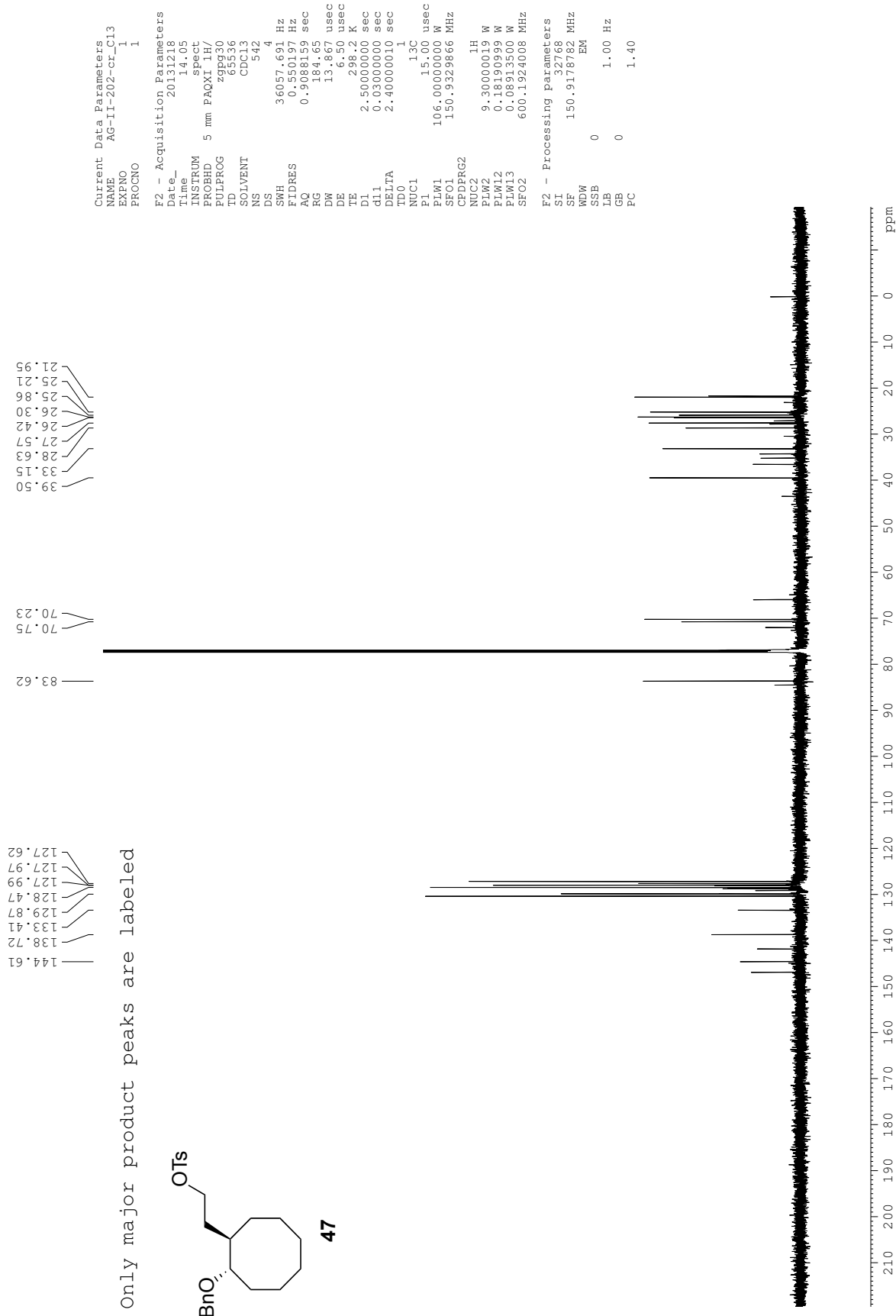
F2 - Processing parameters
SI 32768
SF 150.9178750 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40



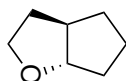




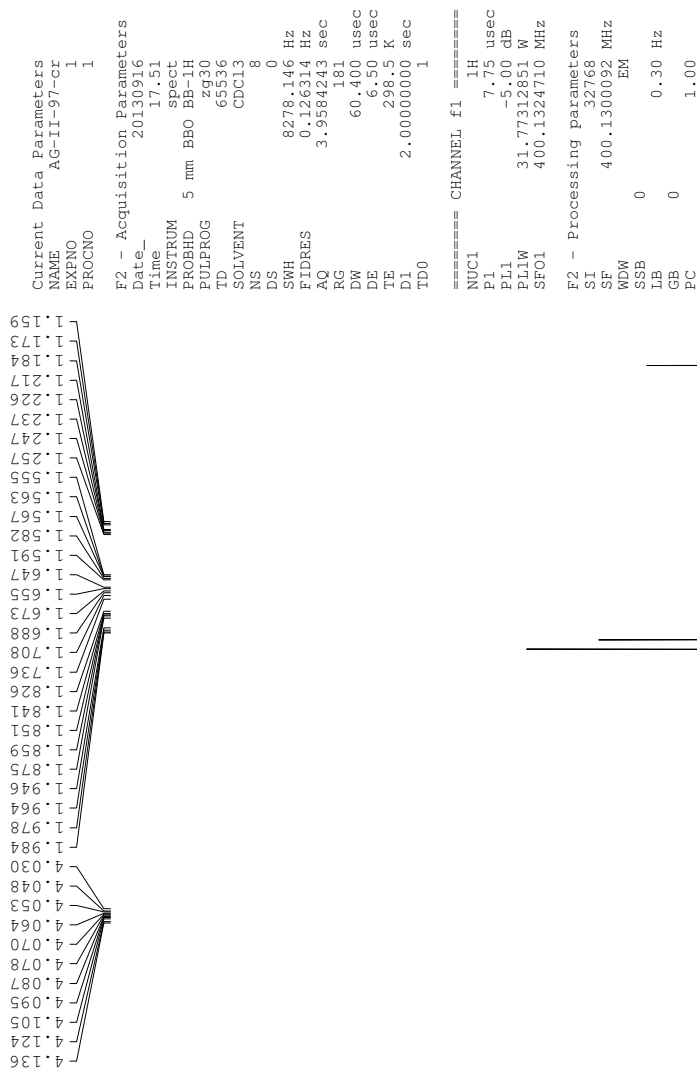




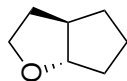
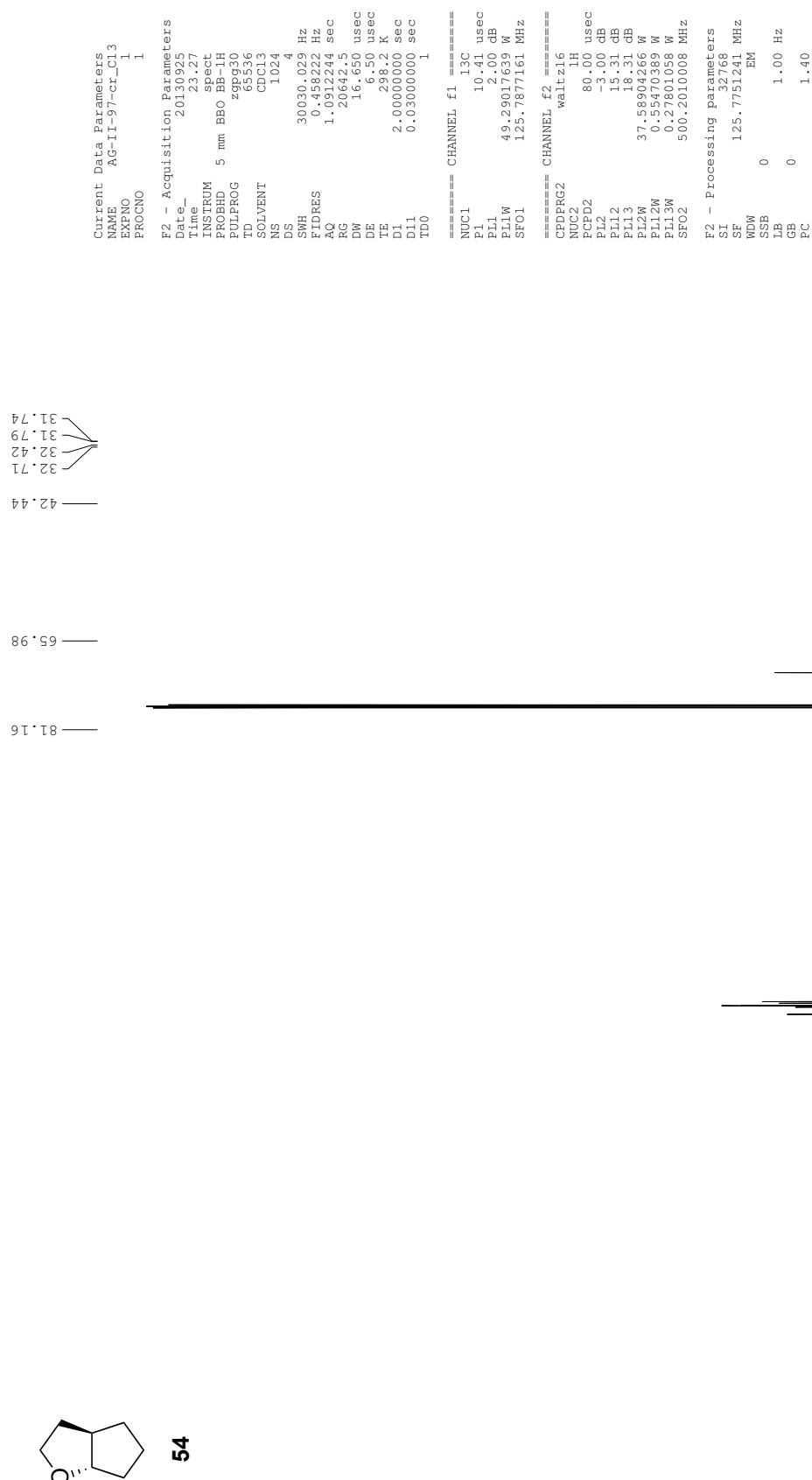
Only peaks for indicated product are labeled

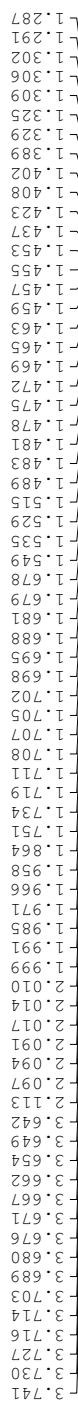


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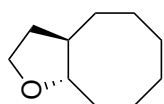


Only peaks for indicated product are labeled

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Only peaks for indicated product are labeled



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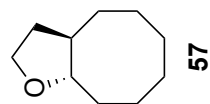
Current Data Parameters
NAME      AG-II-84-C
EXPNO    1
PROCNO   1

F2 - Acquisition Parameters
Date_    20130913
Time     13.27
INSTRUM  spect
PROBHD   5 mm PAQXI LH/
PULPROG  zg30
TD       65536
SOLVENT  CDCl3
NS       8
DS       0
SWH      12335.526 Hz
FIDRES   0.188225 Hz
AQ       2.6564426 sec
RG       76.07
DM       40.533 usec
DE       6.50 usec
TE       298.2 K
D1       2.0000000 sec
TD0      1

===== CHANNEL f1 =====
SFO1    600.1337064 MHz
NUC1     1H
P1       9.79 usec
PLW1    9.30000019 W

F2 - Processing parameters
SI       65536
SF      600.1300147 MHz
WDW      EM
SSB      0
LB       0.30 Hz
GB       0
PC      1.00
  
```

Only peaks for indicated product are labeled



Current Data Parameters
 NAME AG-II-84-C_C13
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20130913
 Time 13.33
 INSTRUM spect
 PROHD 5 mm PAQXI 1H/
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 841
 DS 4
 SWH 36057.691 Hz
 FIDRES 0.550197 Hz
 AQ 0.9088159 sec
 RG 184.65
 DW 13.867 usec
 DE 6.50 usec
 TE 298.1 K
 d11 2.50000000 sec
 d11 0.03000000 sec
 DELTA 2.40000010 sec
 TD0 1
 NUC1 13C
 P1 15.00 usec
 PLW1 106.00000000 W
 SF01 150.9329866 MHz
 CPDPRG2
 NUC2 1H
 PLW2 9.30000019 W
 PLW12 0.18190999 W
 PLW13 0.08913500 W
 SF02 600.1924008 MHz

F2 - Processing parameters
 SI 32768
 SF 150.9178761 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

