Supporting Information for

The Electrochemistry of Iodide, Iodine and Iodine Monochloride in Chloride Containing Nonhaloaluminate Ionic Liquids

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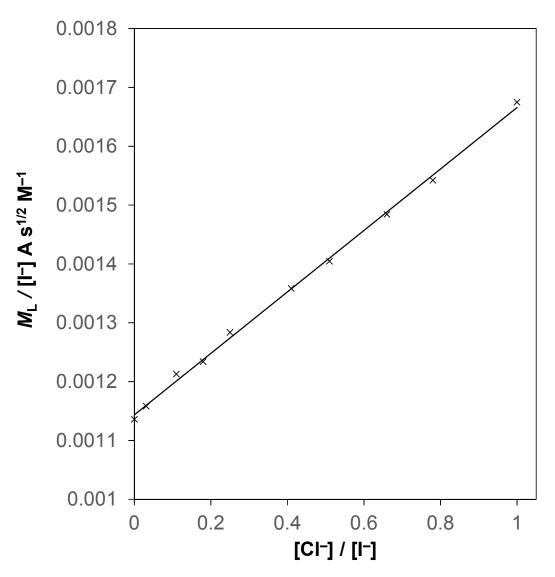


Figure S1. M_L / [I⁻] versus [Cl⁻] / [I⁻] data obtained from a semiintegral voltammogram (Pt macrodisk electrode, scan rate = 50 mV s⁻¹) along with a plot (—) of the least-squares linear regression line (r² = 0.998).

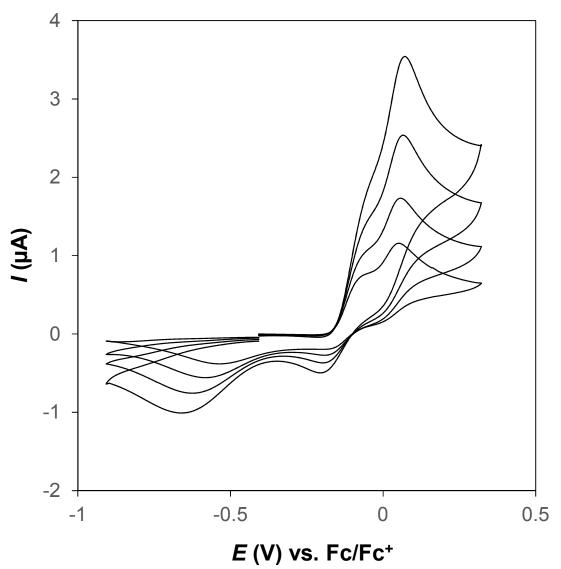


Figure S2. Background-subtracted cyclic voltammograms obtained at a 1.6 mm diameter Pt macrodisk electrode from the electro-oxidation of 10.8 mM Γ (from [C₂mim]I) in a mixture of [C₄mim]Cl and [C₂mim][NTf₂] (2.5:1 mol ratio). Scan rates are (from top to bottom) 100, 50, 25 and 10 mV s⁻¹.

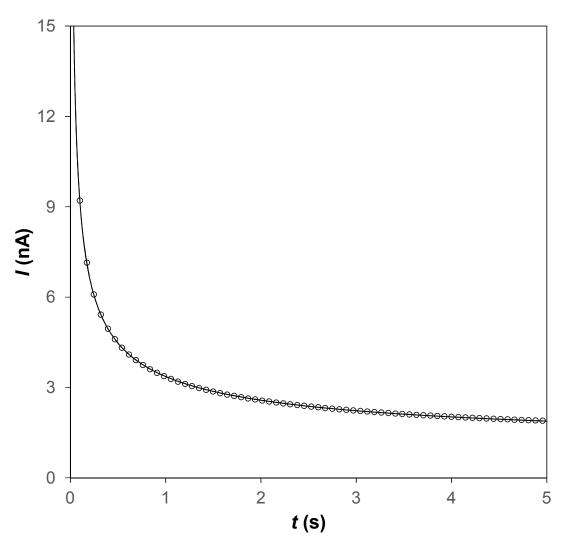
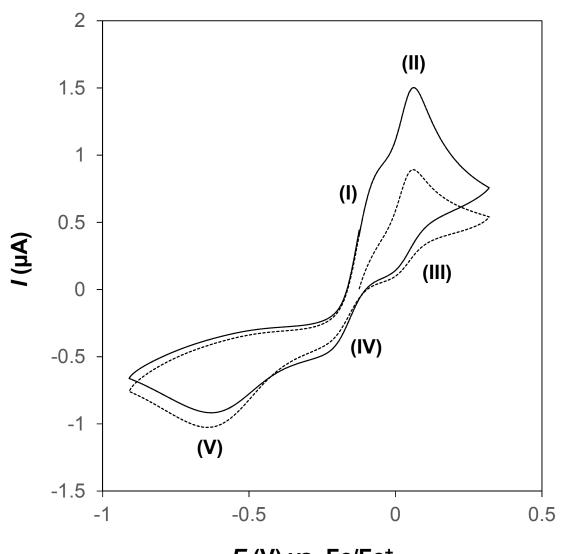


Figure S3. Shoup-Szabo theoretical ($^{\circ}$) and experimental (—) chronoamperograms obtained at a 20 µm diameter Pt microdisk electrode from the diffusion controlled oxidation ($E_{\text{step}} = 0.16 \text{ V vs. Fc/Fc}^+$) of 43.9 mM Γ (from [C₂mim]I]) in a mixture of [C₄mim]Cl and [C₂mim][NTf₂] (2:1 mol ratio). The fitting parameters ($r^2 = 0.99997$) used in the theoretical plot were: [Γ] = 0.0439 M, $r_0 = 0.00112$ cm, n = 1.96 and $D = 2.2 \times 10^{-8}$ cm² s⁻¹.



E (V) vs. Fc/Fc⁺ ted cyclic voltammogram obtained at

Figure S4. A background-subtracted cyclic voltammogram obtained at a 1.6 mm diameter Pt macrodisk electrode from 8.1 mM I₂ in a mixture of [C₄mim]Cl and [C₂mim][NTf₂] (2.5:1 mol ratio) with a scan rate of 10 mV s⁻¹. From open circuit (-0.124 V vs. Fc/Fc⁺), the potential was initially swept positively. Cycles 1 (---) and 2 (—) are shown.

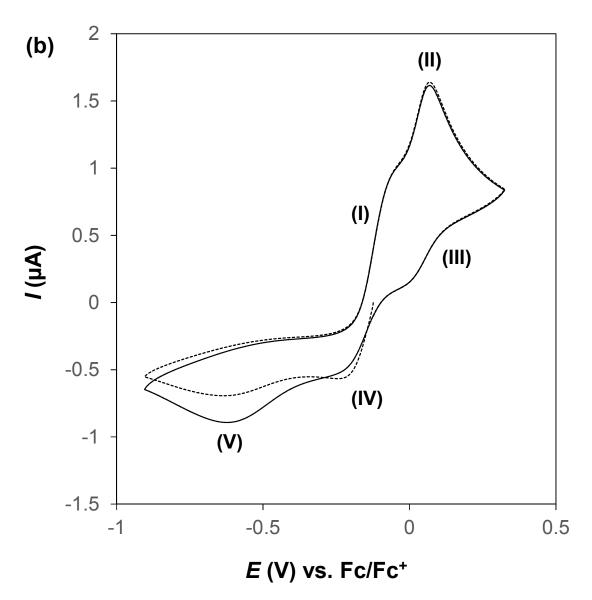


Figure S5. A background-subtracted cyclic voltammogram obtained at a 1.6 mm diameter Pt macrodisk electrode from 8.1 mM I₂ in a mixture of $[C_4mim]Cl$ and $[C_2mim][NTf_2]$ (2.5:1 mol ratio) with a scan rate of 10 mV s⁻¹. From open circuit (-0.124 V vs. Fc/Fc⁺), the potential was initially swept negatively. Cycles 1 (---) and 2 (---) are shown.