

Supporting Information for:

Surface Contributions to Mn^{2+} Spin Dynamics in Colloidal Doped Quantum Dots

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Fitting analysis to obtain T_1 and T_2

Figure S1 shows inversion recovery data for $\text{Zn}_{0.995}\text{Mn}_{0.005}\text{O}$ QDs in toluene (tol, a) or deuterated toluene (tol- d_8 , b), and for $\text{Zn}_{0.900}\text{Mn}_{0.010}\text{O}$ QDs washed with ethanol (EtOH, c) or deuterated ethanol (EtOD, d). The data were fit to both single (equation S1a, dashed blue) and double (equation S1b, dotted orange) exponentials¹ to obtain T_1 . From the double exponential fits, an effective spin-lattice relaxation time was calculated ($T_{1\text{eff}}$) using equation S1c. The results of this analysis are presented in Table S1. The $T_{1\text{eff}}$ values determined from this analysis are reported in the main text as T_1 values.

$$V(\tau) = 1 - V_0 \exp(-\tau / T_1) \quad (\text{S1a})$$

$$V(\tau) = 1 - A \exp(-\tau / T_{1A}) - B \exp(-\tau / T_{1B}) \quad (\text{S1b})$$

$$T_{1\text{eff}} = \frac{A}{A+B} T_{1A} + \frac{B}{A+B} T_{1B} \quad (\text{S1c})$$

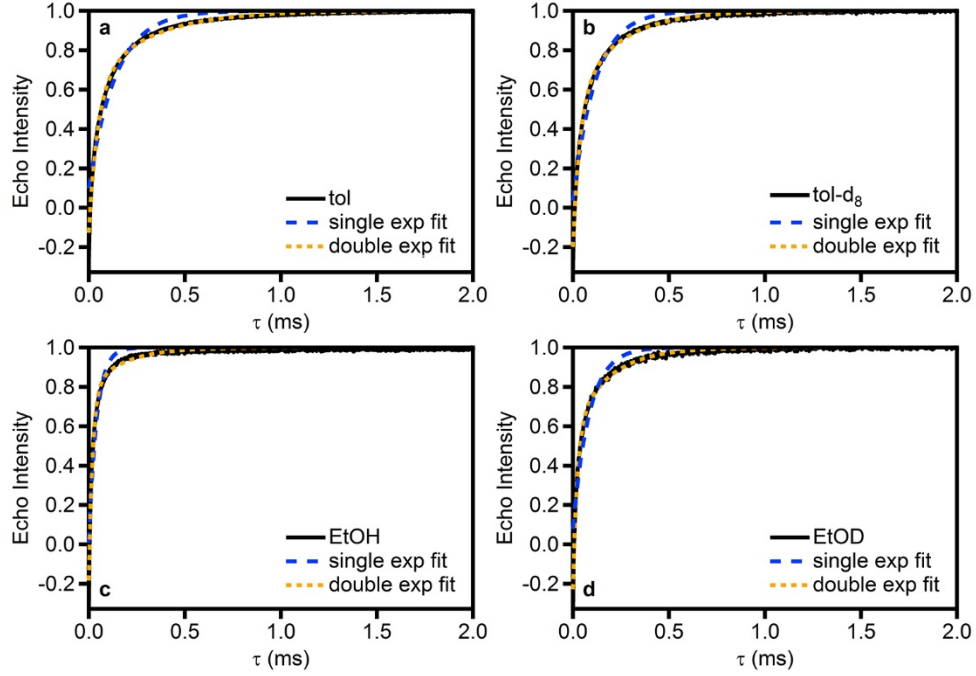


Figure S1. Inversion recovery data (solid black) with single (dashed blue) and double (dotted orange) exponential fits following equations S1a and S1b, respectively. Data are shown for $\text{Zn}_{0.995}\text{Mn}_{0.005}\text{O}$ QDs in tol (a) or tol- d_8 (b) and for $\text{Zn}_{0.990}\text{Mn}_{0.010}\text{O}$ QDs washed with EtOH (c) or EtOD (d).

Table S1. Spin-lattice relaxation times in $\text{Zn}_{0.995}\text{Mn}_{0.005}\text{O}$ (tol vs tol- d_8) and $\text{Zn}_{0.990}\text{Mn}_{0.010}\text{O}$ (EtOH vs EtOD) QDs.

	Single Exponential Fits		Double Exponential Fits			
	T_1 (μs)	T_{1A} (μs)	A	T_{1B} (μs)	B	$T_{1\text{eff}}$ (μs)
In tol	139 ± 2	298 ± 3	0.342 ± 0.004	46.6 ± 0.6	0.658 ± 0.005	151 ± 2
In tol- d_8	117 ± 1	230 ± 2	0.374 ± 0.005	37.4 ± 0.5	0.625 ± 0.005	109 ± 1
With EtOH	41 ± 1	179 ± 5	0.177 ± 0.005	19.7 ± 0.3	0.823 ± 0.007	48 ± 1
With EtOD	84 ± 1	184 ± 2	0.338 ± 0.004	22.4 ± 0.4	0.662 ± 0.006	80 ± 1

Figure S2 shows 2-pulse ESEEM data for $\text{Zn}_{0.995}\text{Mn}_{0.005}\text{O}$ QDs in toluene (tol, a) or deuterated toluene (tol- d_8 , b), and for $\text{Zn}_{0.990}\text{Mn}_{0.010}\text{O}$ QDs washed with ethanol (EtOH, c) or deuterated ethanol (EtOD, d). The data were fit to both single (equation S2a, dashed blue) and double (equation S2b, dotted orange) exponentials¹ to obtain T_2 . From the double exponential fits, effective spin-lattice relaxation times were calculated ($T_{2\text{eff}}$) using equation S2c. The results of this analysis are presented in Table S2. For EtOD, a double exponential fit does not converge when $T_{1A} \neq T_{1B}$, so a single exponential fit was used. In the main text, $T_{2\text{eff}}$ from this analysis is

reported for comparison between tol and tol-d₈, and T_2 from this analysis is reported for comparison between EtOH and EtOD.

$$V(\tau) = V_0 \exp(-2\tau / T_2) \quad (\text{S2a})$$

$$V(\tau) = A \exp(-2\tau / T_{2A}) + B \exp(-2\tau / T_{2B}) \quad (\text{S2b})$$

$$T_{2\text{eff}} = \frac{A}{A+B} T_{2A} + \frac{B}{A+B} T_{2B} \quad (\text{S2c})$$

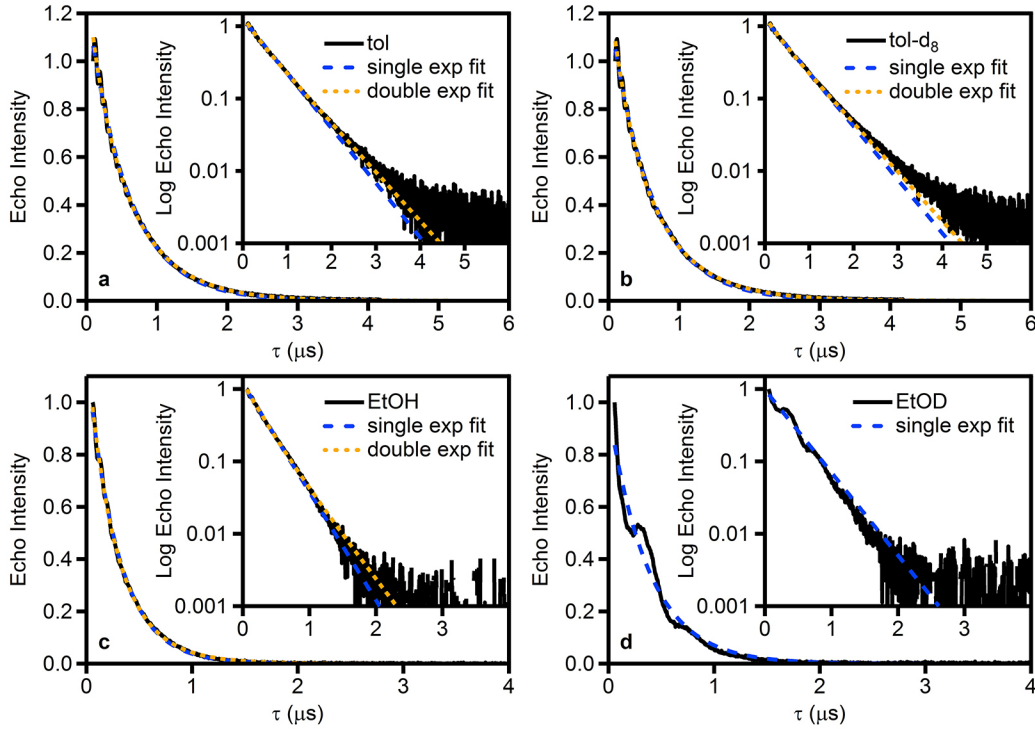


Figure S2. Hahn echo decay data (solid black) with single (dashed blue) and double (dotted orange) exponential fits following equations S2a and S2b, respectively. Decays are shown for Zn_{0.995}Mn_{0.005}O QDs in tol (a) or tol-d₈ (b) and for Zn_{0.990}Mn_{0.010}O QDs washed with EtOH (c) or EtOD (d).

Table S2. Spin-lattice relaxation times in Zn_{0.995}Mn_{0.005}O (tol vs tol-d₈) and Zn_{0.990}Mn_{0.010}O (EtOH vs EtOD) QDs.

	Single Exponential Fits	Double Exponential Fits				
	T_2 (μs)	T_{2A} (μs)	A	T_{2B} (μs)	B	$T_{2\text{eff}}$ (μs)
In tol	1.150 ± 0.002	1.28 ± 0.01	0.77 ± 0.01	0.49 ± 0.02	0.23 ± 0.01	1.10 ± 0.02
In tol-d ₈	1.176 ± 0.002	1.36 ± 0.01	0.70 ± 0.02	0.60 ± 0.03	0.30 ± 0.02	1.13 ± 0.03
With EtOH	0.581 ± 0.001	0.73 ± 0.03	0.47 ± 0.02	0.43 ± 0.02	0.53 ± 0.02	0.57 ± 0.03
With EtOD	0.76 ± 0.01					

Figure S3 shows inversion recovery (a) and 2-pulse ESEEM data (b) collected on $\text{Cd}_{0.996}\text{Mn}_{0.004}\text{Se}$ QDs before (blue) and after (red) CdSe shell growth. To obtain T_1 and T_2 the inversion recovery and 2-pulse ESEEM spectra were fit to equations S1a and S2a, respectively.

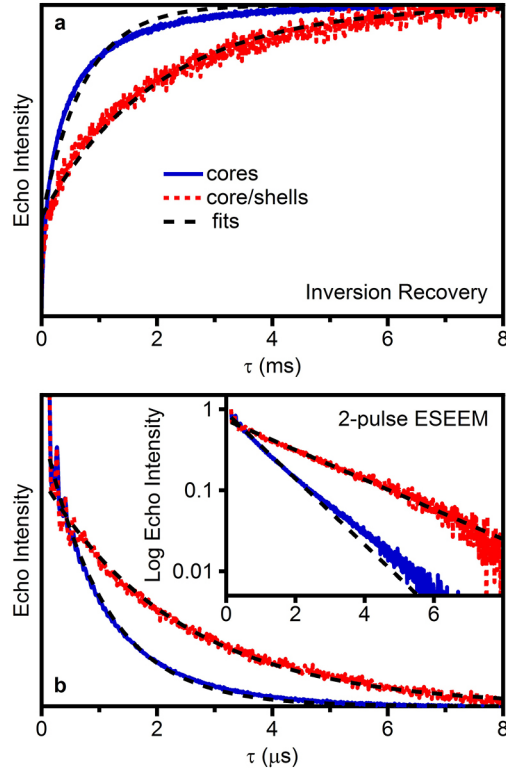


Figure S3. pEPR data collected on $\text{Cd}_{0.996}\text{Mn}_{0.004}\text{Se}$ QDs before (blue) and after (red) CdSe shell growth. (a) Echo-detected inversion recovery (T_1) and (b) 2-pulse spin-echo decay (T_2). The dashed black lines are single exponential fits to the inversion recovery and 2-pulse ESEEM data following equations S1a and S2a, respectively.

Estimated dipole-dipole contributions assuming non-uniformly doped core QDs or dopant migration during shell growth

If an undoped core nucleus volume with $d = 2.0$ nm is assumed,² then the average Mn^{2+} position is ~ 0.2 nm from the nearest QD surface. Upon shell growth this distance increases to ~ 1.3 nm. The $1/r^6$ dependence of dipole-dipole coupling leads to $T_{\text{dip}}^{\text{core/shell}} / T_{\text{dip}}^{\text{core}} \approx 3 \times 10^4$. Table S3 summarizes the results in this scenario. Similar results are obtained even if the Mn^{2+} position is randomized throughout the core/shell volume, for example by diffusion during shell growth,³ although this scenario is not considered likely.

Table S3. Intrinsic and dipolar contributions to spin relaxation times in core and core/shell nanocrystals under the assumption of an undoped core nucleus with $d = 2.0$ nm, determined using equation 1 of the main text.

	T_0	$T_{\text{dip}}^{\text{core}}$	$T_{\text{dip}}^{\text{core/shell}}$
Spin-lattice (T_1)	2.01 ms	1.13 ms	34.0 s
Spin-spin (T_2)	4.70 μs	3.90 μs	117 ms

References

- (1) Schweiger, A.; Jeschke, G. *Principles of Pulse Electron Paramagnetic Resonance*; Oxford University Press: United Kingdom, 2001.
- (2) Norberg, N. S.; Parks, G. L.; Salley, G. M.; Gamelin, D. R. Giant Excitonic Zeeman Splittings in Colloidal Co^{2+} -Doped ZnSe Quantum Dots. *J. Am. Chem. Soc.* **2006**, *128*, 13195-13203.
- (3) Vlaskin, V. A.; Barrows, C. J.; Erickson, C. S.; Gamelin, D. R. Nanocrystal Diffusion Doping. *J. Am. Chem. Soc.* **2013**, *135*, 14380-14389.