Supporting Information of Isolated-Mn²⁺-like Luminescent Behavior in CsMnF₃ Caused by Competing Magnetic Interactions at Cryogenic Temperature

Xinglu Zhu ^a, Yifei Zhao ^{a,d}, Shuai Zhang ^a, Junkun Wu^b, Dong Shao^c, Enhai Song ^a, Qinyuan Zhang ^a, Congling Yin*^b, Shi Ye*^a

^aState Key Laboratory of Luminescent Materials and Devices, and Guangdong Provincial Key Laboratory of Fiber Laser Materials and Applied Techniques, South China University of Technology, Guangzhou 510641, China.

^bMOE Key Laboratory of New Processing Technology for Nonferrous Metal and Materials, Guangxi Key Laboratory of Optic and Electronic Materials and Devices, College of Materials Science and Engineering, Guilin University of Technology, Guilin 541004, P. R. China.

^cHubei Key Laboratory of Processing and Application of Catalytic Materials, College of Che mistry and Chemical Engineering, Huanggang Normal University, Huanggang 438000,

China.

^dDepartment of Chemistry, City University of Hong Kong, Kowloon 999077, Hong Kong, Chi

*E-mail: congling.yin@glut.edu.cn, msyes@scut.edu.cn,

Supplementary Tables and Table captions:

Formula sum	CsMnF ₃		
Crystal system	hexagonal		
Space-group	P 6 ₃ /mmc (194)		
<i>a</i> (Å)	6.2325		
<i>c</i> (Å)	15.1193		
a/b	1		
b/c	0.4122		
c/a	2.4259		
Cell volume ($Å^3$)	508.62		
Z	6		
Rwp (%)	3.58		
Rp (%)	2.71		
Gof	1.48		

Table S1. The refined Crystallographic and structure data summary for CMF.

Table S2. The atomic coordinates of CMF.

	Atom	Ox.	Site	Wyck.	x/a	y/b	z/c
	Cs1	1	-6m2	2b	0	0	0.25
	Cs2	1	<i>3m</i> .	4f	0.33333	0.66667	0.0986
CsMnF ₃	Mn1	2	<i>-3m</i> .	2a	0	0	0
	Mn2	2	<i>3m</i> .	4f	0.33333	0.66667	0.85
	F1	-1	mm2	6h	0.522	0.044	0.25

Bond	Distance (Å)
Mn1-F2	2.136(2)
Mn2-F2	2.118(3)
Mn2-F2	2.118(3)
Mn2-F2	2.118(3)
Mn2-F1	2.173(7)
Mn2-F1	2.173(7)
Mn2-F1	2.173(7)
Mn2-F2	2.118(3)
Mn1-F2	2.136(2)

Table S3. The selected bond distances of CMF.

Table S4. The Mn²⁺-Mn²⁺ distances and Mn²⁺-F⁻-Mn²⁺ angles of CMF.

Species	Distance(Å)	Angle (°)
Mn1-F2-Mn2	4.253(4)	177.41(6)
Mn2-F1-Mn2	3.023(9)	88.14(5)

Table S5. The Mn^{2+} - Mn^{2+} distances and Mn^{2+} - Cl^{-} - Mn^{2+} angles of for CMC.

Species	Distance(Å)	Angle (°)
Mn1-Cl2-Mn2	3.179(7)	77.24(1)
Mn2-Cl1-Mn2	5.025(5)	180

Supplementary Figures and Figure captions:



Figure S1. The excitation spectra of CMF at (a) room temperature and (b) 50 K.



Figure S2. (a-b) The temperature-dependent excitation spectra of CMF. (c-d) The temperature-

dependent decay curves of CMF.



Figure S3. (a) The emission spectrum and (b) the decay curve of CMC at room temperature.



Figure S4. The hypothetic models of Mn²⁺ ion with magnetic moment in CMF. (The red and blue solid spheres denote the opposite magnetic moment directions.)



Figure S5. (a) The EPR spectra of CMF and CMC at 130 K. (b) The temperature-dependent inverse of magnetic susceptibility of CMC measures under ZFC and FC mode. (c) Plots of $\chi_m T$ versus *T* from 2 to 300 K of CMF and CMC. The orange and purple lines across $\chi_m T$ are the fitted lines using the molecular-field theory. (d) The partial photomagnetization curve of CMF.

Here, the magnetic coupling constant *J* can be estimated using Equation S1-3 given by appropriate isotropic Heisenberg Hamiltonian.¹⁻² The molecular-field theory (Equation S3) to fit the plots of $\chi_m T$ versus *T* from 2 to 300 K of CMF and CMC. (Figure S5c, orange and purple lines.)

$$\chi_{chain} = \frac{2Ng^2\beta^2 S(S+1)}{3kT} \frac{(1-u)}{(1+u)}$$
(S1)

$$u = -coth \frac{JS(S+1)}{kT} + \frac{kT}{JS(S+1)}$$
(S2)

$$\chi = \frac{\chi_{chain}}{1 - (zJ/Ng^2\beta^2)\chi_{chain}}$$
(S3)

Where, *N*, *g*, *k*, and β are the Avogadro constant, the Lande factor, the Boltzmann constant, and the Bohr magneton. *J* and *zJ* are the magnetic coupling constants of the intra- and interchain. To obtain a reasonable result, the *S* and *g* are set to 5/2 and 2.05, respectively.

References

 Wang, X. Y.; Li, B. L.; Zhu, X.; Gao, S. Extended Networks of Co²⁺ and Mn²⁺ Bridged by NCS⁻/N³⁻ Anions and Flexible Long Spacers: Syntheses, Structures, and Magnetic Properties. *Eur. J. Inorg. Chem.* **2005**, *16* (8), 3277-3286.

(2) Kahn, O. Molecular magnetism, VCH. Publishers Inc., New York 1993, p251-286.