Supporting Information for

Enhanced kinetic performance of amine-infused hydrogels for separating $CO_2 \; from \; CH_4/CO_2 \; gas \; mixture$

Hyunho Kim ^{a, b} , Jong Yeon Jung ^a , Ki-Heum Park ^a Praveen Linga ^b , Yutaek Seo ^{a*} , and Colin
$D. Wood^{c,d^*}$,
^a Department of Naval Architecture and Ocean Engineering, Research Institute of Marine
Systems Engineering, Seoul National University, 1 Gwanak-ro, Gwanak-gu, Seoul, 08826,
Republic of Korea
^b Department of Chemical and Biomolecular Engineering, National University of Singapore,
Singapore 117585
^c CSIRO Australian Resources Research Centre, Kensington, WA 6152, Australia
^d Curtin Oil and Gas Innovation Centre, Curtin University, Australian Resources Research
Centre, Kensington, WA 6152, Australia
*Corresponding author(s): yutaek.seo@snu.ac.kr (Y. Seo), colin.wood@csiro.au (C.D. Wood).
Including:
4 Figures
1 Table

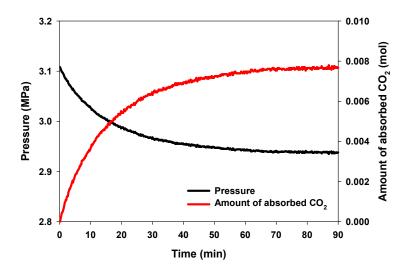


Figure S1. Pressure profile and amount of absorbed CO₂ calculated from pressure change ($\Delta n_{CO_2,pressure}$) during the experiment at 303 K

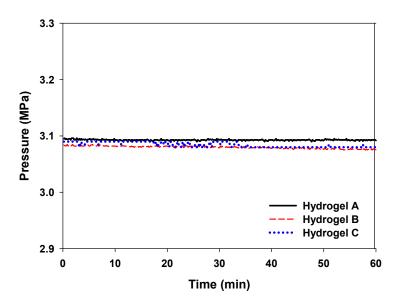


Figure S2. Pressure profile of methane (CH₄) gas on each hydrogel

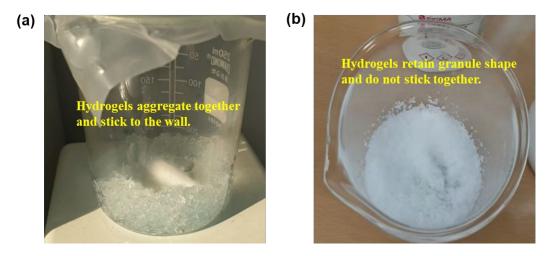


Figure S3. Shape of MEA-hydrogel B (a) and MEA-hydrogel C (b) under mixing condition

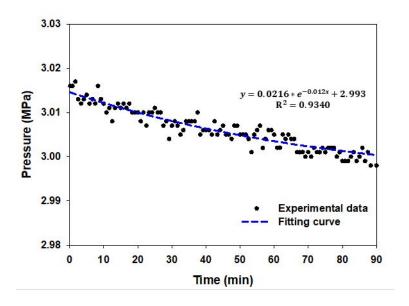


Figure S4. Pressure profile of methane (CH₄) gas in decane phase at 303 K

Table S1. Final composition of CH_4 and CO_2 in gas phase with decane liquid phase at 303 K

Method	Component	Final composition (mol%)
Measured by gas chromatography (GC)	CH ₄	84.92
	CO_2	15.08
Calculated by pressure change	CH ₄	84.93
	CO_2	15.07