

## Supplementary Information

### **TiO<sub>2</sub>-Decorated Titanium Carbide MXene co-Doped with Nitrogen and Sulfur for Oxygen Electroreduction**

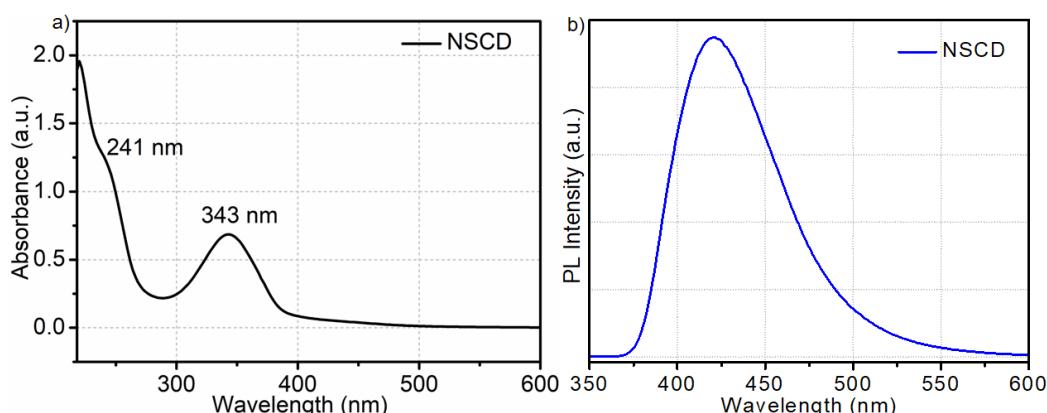
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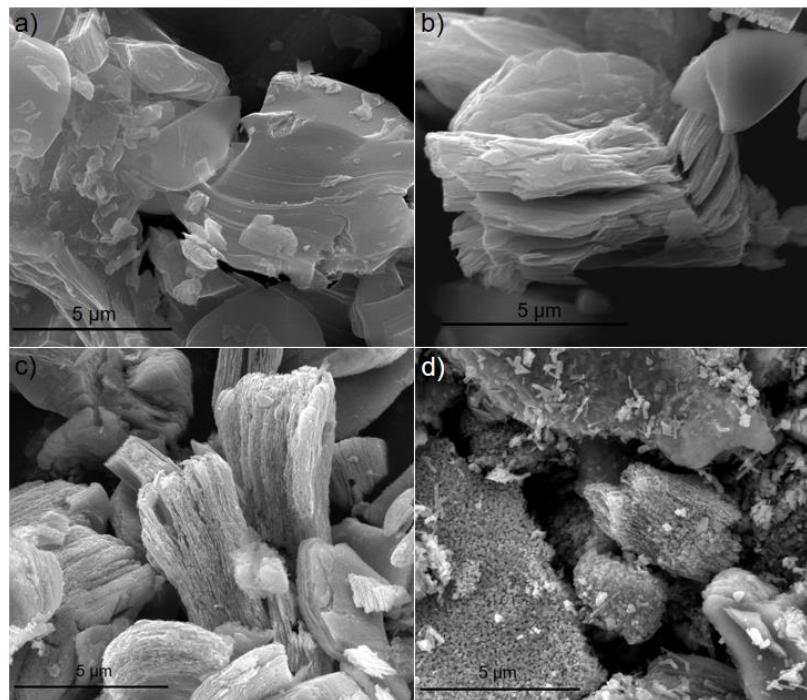
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**Figure S1**



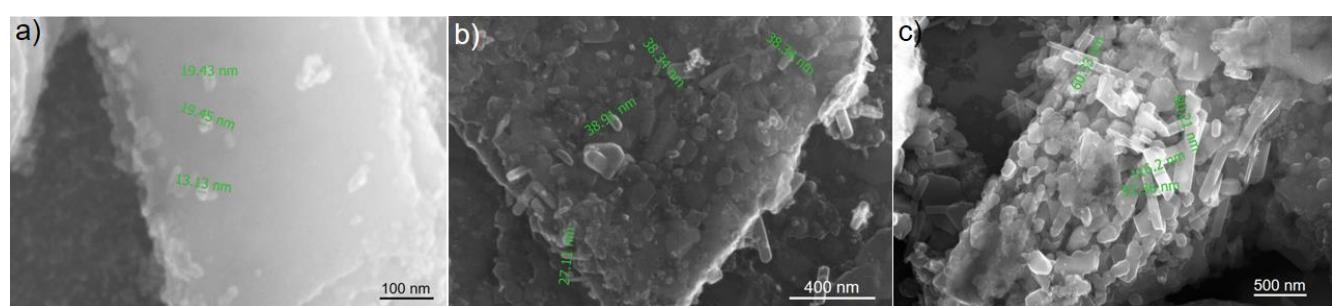
**Figure S1.**(a) UV-visible spectrum of NSCD; (b) photoluminescence spectrum of as-prepared NSCDs

**Figure S2**



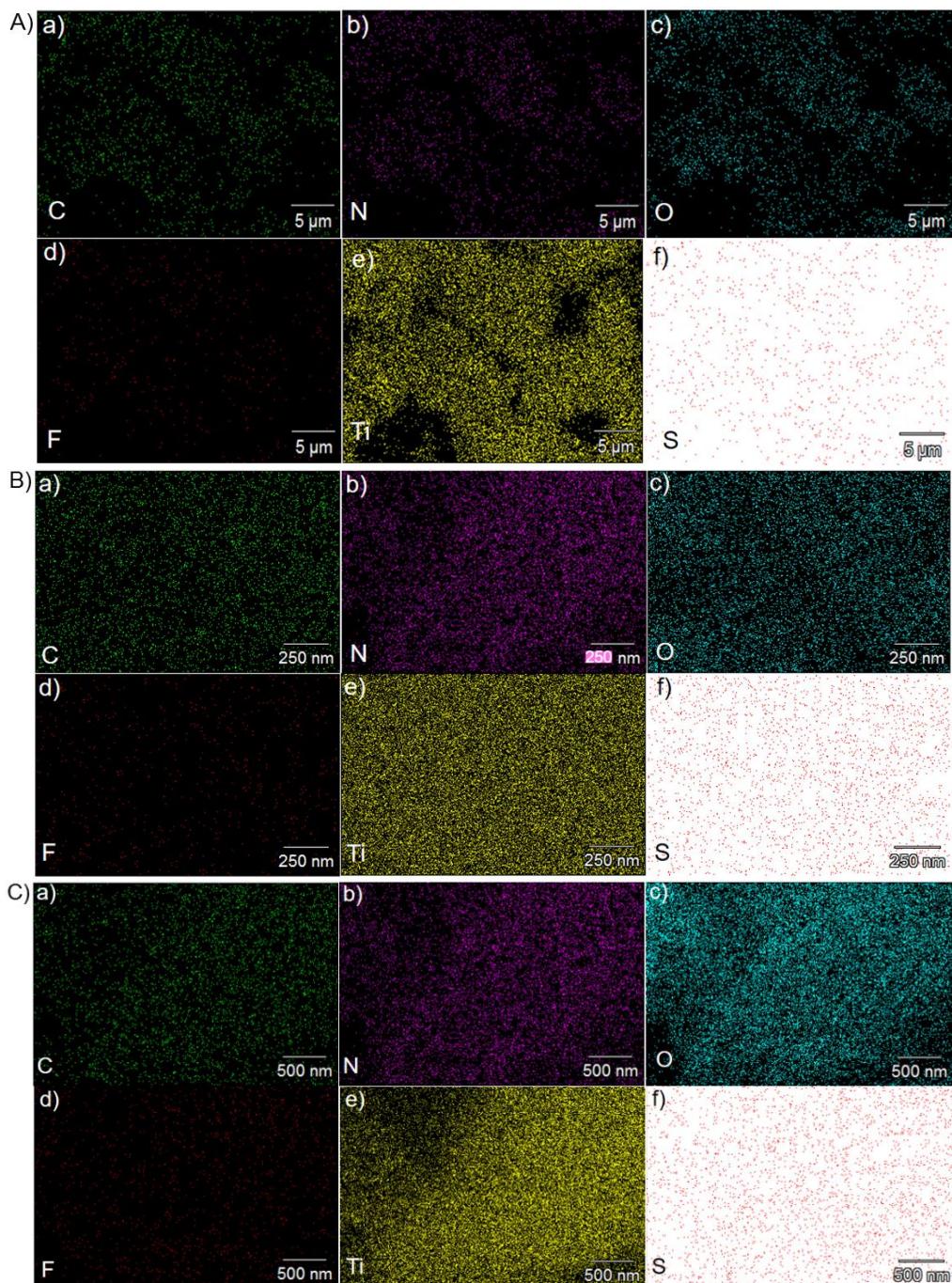
**Figure S2.** (a-b) SEM images of the pristine  $\text{Ti}_3\text{AlC}_2$  (MAX phase) and  $\text{Ti}_3\text{C}_2\text{T}_x$  (MXene); (c-d) SEM images of the  $\text{Ti}_3\text{C}_2/\text{NSCD-500}$  and  $\text{Ti}_3\text{C}_2/\text{NSCD-700}$ .

**Figure S3**



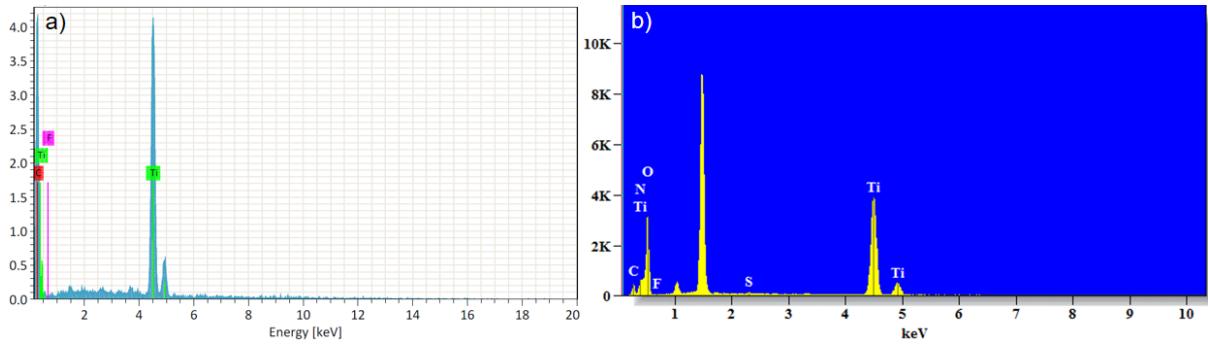
**Figure S3.** (a-c) SEM images of  $\text{Ti}_3\text{C}_2/\text{NSCD-500}$ ,  $\text{Ti}_3\text{C}_2/\text{NSCD-600}$ ,  $\text{Ti}_3\text{C}_2/\text{NSCD-700}$  catalyst (numbers in Fig. indicating the  $\text{TiO}_2$  particle size)

**Figure S4**



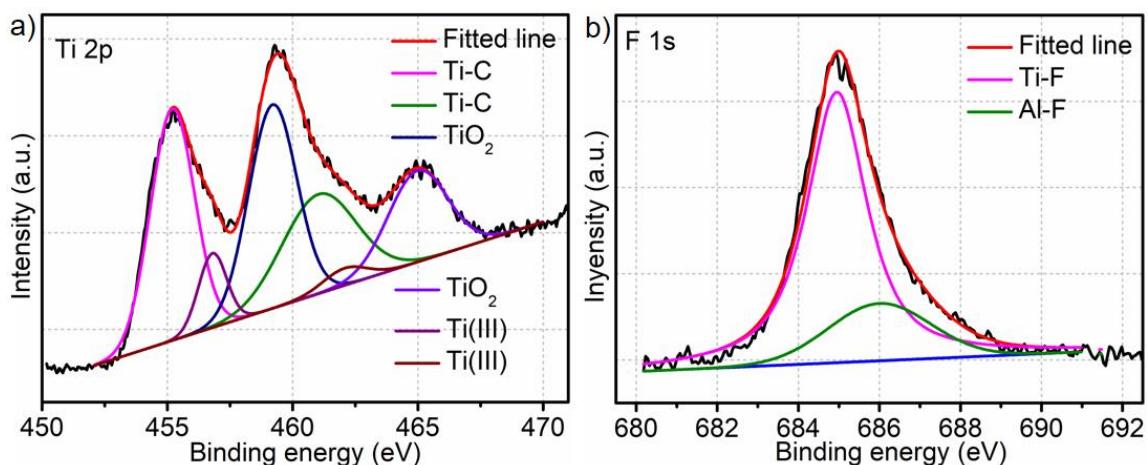
**Figure S4.** SEM-EDS mapping analysis of A)  $\text{Ti}_3\text{C}_2$ /NSCD-500, B)  $\text{Ti}_3\text{C}_2$ /NSCD-600 and C)  $\text{Ti}_3\text{C}_2$ /NSCD-700 which confirms the uniform distribution of the C, N, O, F, Ti, and S (a-f) elements respectively in A), B), C).

**Figure S5**



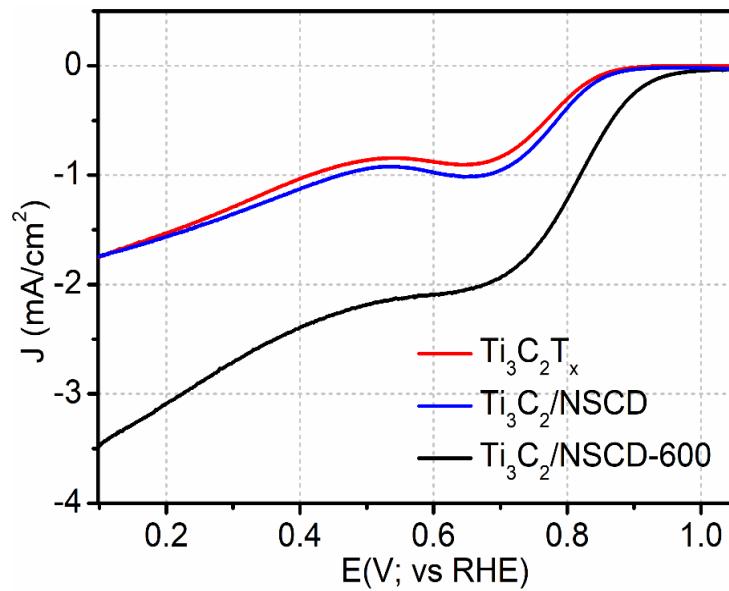
**Figure S5.** (a)EDS spectrum of the pristine  $\text{Ti}_3\text{C}_2\text{T}_x$ ; (b) EDS profile of nanocomposite  $\text{Ti}_3\text{C}_2/\text{NSCD}-600$ .

**Figure S6**



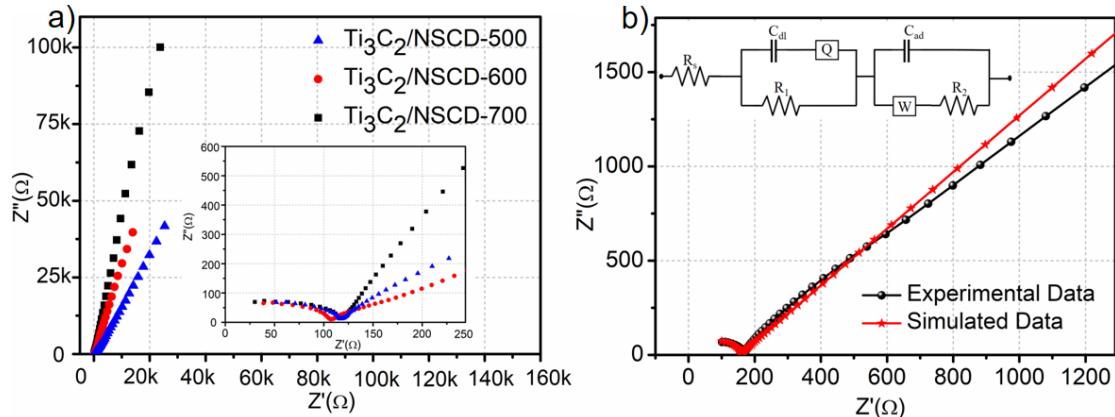
**Figure S6.** XP spectra of  $\text{Ti}_3\text{C}_2\text{T}_x$  deconvoluted core levels for (a)  $\text{Ti} 2\text{p}$  and (b)  $\text{F} 1\text{s}$ .

**Figure S7**



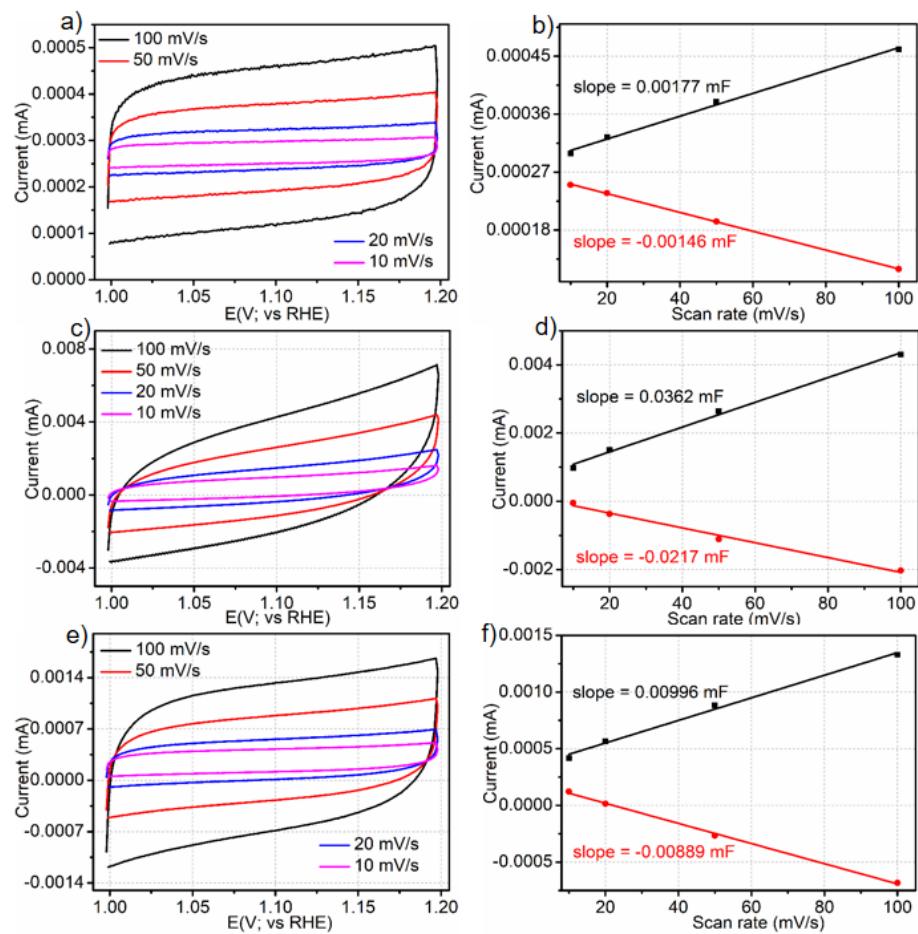
**Figure S7.** Comparative LSV curves for  $\text{Ti}_3\text{C}_2\text{T}_x$ ,  $\text{Ti}_3\text{C}_2/\text{NSCD}$  and  $\text{Ti}_3\text{C}_2/\text{NSCD-600}$  (Step wise electrochemical characterization)

**Figure S8**



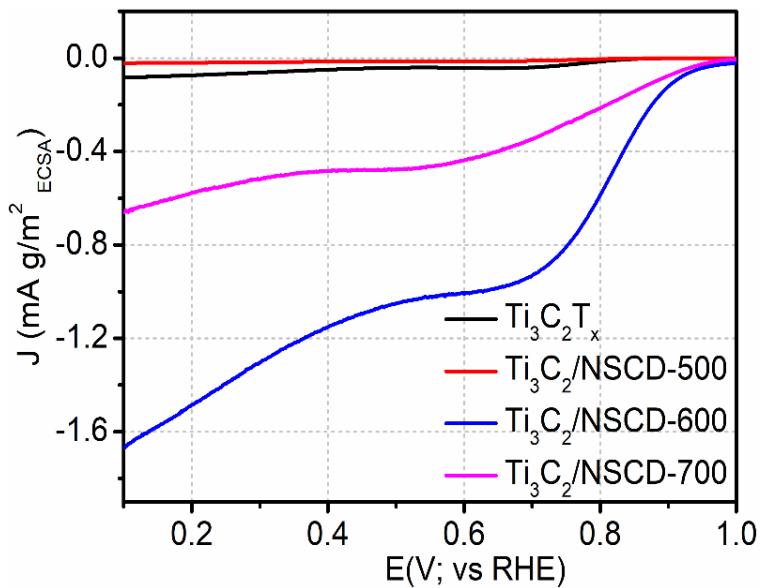
**Figure S8.** (a) Nyquist plots of the  $\text{Ti}_3\text{C}_2/\text{NSCD-500}$ ,  $\text{Ti}_3\text{C}_2/\text{NSCD-600}$  and  $\text{Ti}_3\text{C}_2/\text{NSCD-700}$ , whereas inset shows high frequency region of EIS; (b) Comparison of experimental and with simulated Nyquist plot of  $\text{Ti}_3\text{C}_2/\text{NSCD-600}$  electrocatalyst.

**Figure S9**



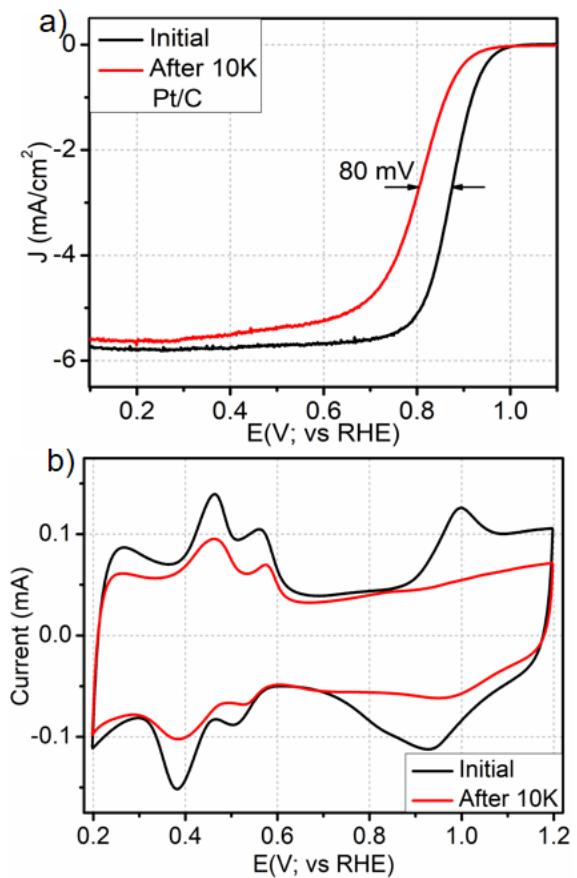
**Figure S9.** CV curves were recorded in a non-Faradaic region at various scan rates (10 to 100 mV/s) in 0.1 M KOH for (a)  $\text{Ti}_3\text{C}_2/\text{NSCD}-500$  (c)  $\text{Ti}_3\text{C}_2/\text{NSCD}-600$  and (e)  $\text{Ti}_3\text{C}_2/\text{NSCD}-700$ . The variations in the current as function of scan rate for (b)  $\text{Ti}_3\text{C}_2/\text{NSCD}-500$  (d)  $\text{Ti}_3\text{C}_2/\text{NSCD}-600$  and (f)  $\text{Ti}_3\text{C}_2/\text{NSCD}-700$  electrode.

**Figure S10**



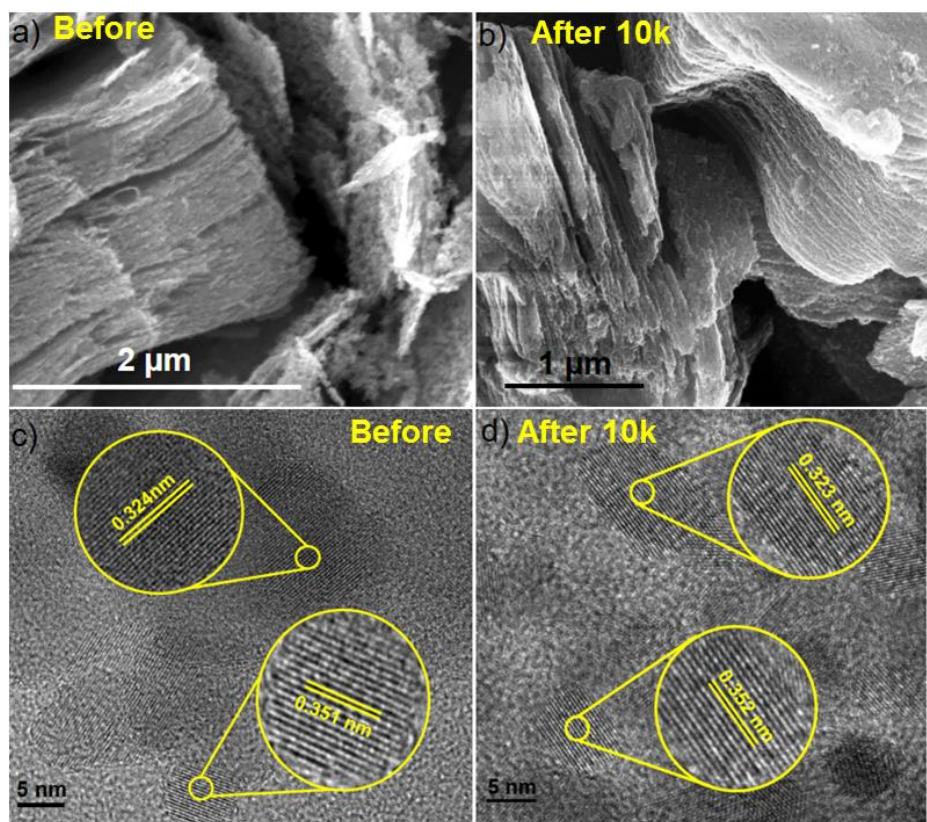
**Figure S10.** Comparative LSV curves of current density normalized with ECSA value for  $\text{Ti}_3\text{C}_2\text{T}_x$  composite prepared at different annealing temperature.

**Figure S11**



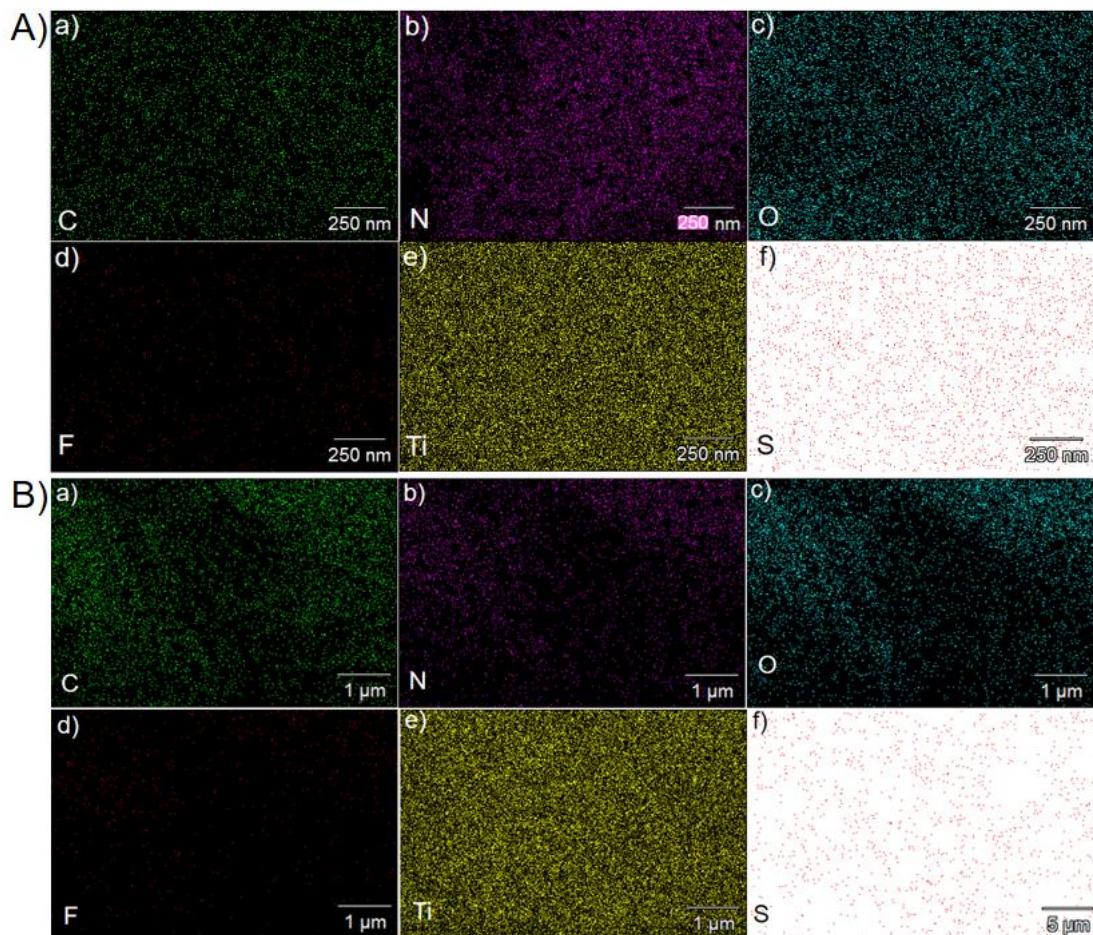
**Figure S11.** (a) Comparative polarization curves of commercial Pt/C before and after 10k electrochemical cycles at 1600 rpm at 10 mV/s scan rate in alkaline regime (0.1 M KOH); (b) CV curves of the state-of-art catalyst before and after 10k cycles.

**Figure S12**



**Figure S12.** (a-b) SEM and (c-d) HR-TEM images of the  $\text{Ti}_3\text{C}_2/\text{NSCD-600}$  before and after stability test.

**Figure S13**



**Figure S13.** Elemental mapping of  $\text{Ti}_3\text{C}_2/\text{NSCD}$  (A) before and (B) after 10k electrochemical cycles which confirms the uniform distribution of the C, N, O, F, Ti, and S (a-f) elements respectively in A) and B) before and after stability .

**Table S1**

Sr. No.	Catalyst	Electrolyte	Mass loading (mg/cm <sup>-2</sup> )	J <sub>L</sub> (mA/cm <sup>2</sup> )	E <sub>1/2</sub> (V) vs RHE	E <sub>onset</sub> (V) vs RHE	Ref.
1	MXene/NW-Ag0.9Ti0.1	0.1 M KOH	-	4.5	0.782	0.921	[1]
2	NiCo <sub>2</sub> O <sub>4</sub> /MXene	0.1 M KOH	0.700	6.0	0.700	0.920	[2]
3	Mn <sub>3</sub> O <sub>4</sub> /MXene	0.1 M KOH	0.100	3.5	0.800	0.890	[3]
4	N-doped porous carbons	0.1 M KOH	0.240	5.5	0.840	0.910	[4]
5	Co/N-CNTs@Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	0.1 M KOH	0.408	5.8	0.815	0.936	[5]
6	FePc/Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	0.1 M KOH	-	5.5	0.850	0.980	[6]
7	Ti-C-T <sub>x</sub>	0.1 M KOH	0.120	4.2	0.75	0.870	[7]
8	NSCD/C-600	0.1 M KOH	0.226	5.0	0.71	0.87	[8]
9	N-GQD	0.1 M KOH	-	2.5	0.4	0.75	[9]
10	TiC@NC(0.2)-800	0.1 M KOH	0.120	4.8	0.92	1.08	[10]
11	CNFe	0.1 M KOH	0.600	5.5	0.901	0.998	[11]
12	GQD-MWCNT	0.1 M KOH	-	7.8	0.79	0.81	[12]
13	G-BGQD	0.1 M KOH	-	4.2	0.8	0.94	[13]
14	N-GQDs	0.1 M KOH	-	2.8	0.69	0.79	[14]
15	Co@CoOx/NCNT	0.1 M KOH	0.200	5.5	0.800	0.940	[15]
16	Co-CoO/N-rGO	0.1 M KOH	0.210	5.8	0.880	0.880	[16]
17	Co@NCNT	0.1 M KOH	0.425	6.2	0.828	1.030	[17]
18	NG/CNT/Co <sub>3</sub> O <sub>4</sub>	0.1 M KOH	-	5.0	-	0.938	[18]
19	<b>Ti<sub>3</sub>C<sub>2</sub>/NSCD-600</b>	0.1 M KOH	<b>0.240</b>	<b>3.5</b>	<b>0.8</b>	<b>0.98</b>	<b>This Work</b>

**Table S1.** Electrochemical activity comparison of optimized catalyst and previously reported catalysts.

**Table S2**

Sr. No.	Sample Code	Sample Name	Mass Activity (mA/mg)
1	A	Ti <sub>3</sub> C <sub>2</sub> @NSCD-500	91.25
2	B	<b>Ti<sub>3</sub>C<sub>2</sub>@NSCD-600</b>	<b>211.8</b>
3	C	Ti <sub>3</sub> C <sub>2</sub> @NSCD-700	180.0
4	D	Ti <sub>3</sub> C <sub>2</sub> (HT)-600	195.5
5	E	Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	77.5
6	F	Pt/C	236.0

**Table S2:** Comparative mass activity of all composites**References:**

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