Supporting Information for

Catalytic membrane reactor of nano (Ag+ZIF-8) @ Poly(tetrafluoroethylene) built by deep-permeation synthesis fabrication

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Figure S1. (A) Schematic illustration of the continuous catalytic reaction experiment,

(B) mechanism illustration of conversion of p-NA to p-PD via DPNS catalytic

membrane.



Figure S2. Particle size distribution of Ag+ZIF-8 NPs inside PTFE

Suspension synthesis of Ag+ZIF-8 particles

Briefly, a clear solution of Zn(NO₃)₂·6H₂O (9.2 g) in DI water (50 mL) was poured into another clear solution of H-MeIM (8.21 g) in DI water (50 mL) under stirring for 30 min. The clear solution turned milky white and showed suspended particles. The as-prepared ZIF-8/water suspension was centrifuged and washed with fresh water for three times to remove residues of unreacted reagents and by-products. The product was dried at 70 °C overnight in a vacuum drying oven ¹. Similarly, 100 mg of pretreated ZIF-8 was dispersed in 10 mL of DI water under ambient conditions, to which 10 mL of AgNO₃ (100 mg) aqueous solution was added under vigorous stirring. The desolvated ZIF-8 was stirred with AgNO₃ aqueous solution for 20 min. Then, 20 ml NaBH₄ (37.8 mg) aqueous solution was added dropwise under vigorous stirring for the complete reduction of Ag⁺. The solid was recovered by filtration and thoroughly washed by water after over 30 min stirring. The sample was finally dried in a vacuum drying oven at room temperature for further use.

Figure S3A showed low magnification TEM image of ethanol-based Ag+ZIF-8 suspensions, which indicated the particle size about 30–50 nm. Those individual particles were agglomerated. The high-resolution TEM (HRTEM) image of Ag+ZIF-8 can be found in Figure S3B. ZIF-8 displayed a mesoporous structure, as mentioned in previous reports ^{2, 3}. Figure S3B exhibited the crystal lattice fringes of Ag+ZIF-8, with the d-spacing of 0.201 nm and 0.242 nm correspond to the crystal plane of Ag (200) and Ag (111), which were close to the d spacing of the (200) and (111) plane of metallic Ag (Joint Committee on Powder Diffraction Standards 040783 File Card) ⁴.

The HRTEM image proved that the Ag and ZIF-8 crystals could coexist. If we consider the membrane pore as a microreactor, the similar individual Ag+ZIF-8 crystal could be obtained. In combination with the other characterizations described above, it can be judged that the Ag was combined with the outer sphere of the ZIF-8 in some way.



Figure S3. TEM image of Ag+ZIF-8 particles (A), HRTEM image of Ag+ZIF-8

particles (B).



Figure S4. High-resolution XPS spectra of Zn 2p (A), Ag 3d (B)

Figure S5. UV-vis absorption spectra of the p-NA solution (1 mM) reduced by

NaBH₄ in different v (ml cm² min⁻¹).

Samples	Ag content (wt %)
(Ag+ZIF-8)@PTFE before catalysis	6.96
(Ag+ZIF-8)@PTFE after deactivation	6.27
Ag@PTFE	0.59

Table S1. Ag content of three catalytic membranes by ICP analysis

Ag-based catalysts	Concentration of p-NA (mM)	NaBH ₄ (mM)	Temperature °C	R	k_{app} (min ⁻¹)	ref
Silver-doped magnetic	1	30	Room	100%, 4min	1.110	5
nanoparticles			temperature			
AgNPs on porous	1	30	50	100%, 10min	0.426	6
glass filters						
Ag-p(NiPA-co-AAc)	0.05	10	23	100%,13min	0.415	7
Hybrid Microgels						
AgNPs/T. indica seed	1	50	Not mentioned	100%, 10min	0.373	8
coat extract						
Ag-Cu nanoshells	0.01	50	Room	100%, 21min	0.12	9
			temperature			
Ag-Au nano-alloy	2	150	Not mentioned	87.2%, 20min	0.038	10
Ag nanoclusters	0.2	100	Room	100%, 12min	0.315	11
@MIL-101(Fe)			temperature			
Ag- poly(N-	0.09	18	22	100%, 11min	0.244	12
isopropylacrylamide-						
acrylicacid-						
acrylamide) microgels						
Ag rhombic	0.1	0.4(NH ₃ ·	50	100%, 1 hour	Not	13
dodecahedra		BH3)			mentioned	
(Ag+ZIF-8)@PTFE	1	200	Room	100%	1128	This
DPNS			temperature	0.015 min		work
(Ag+ZIF-8)@PTFE	0.06	12	Room	100%	991.89	This
DPNS			temperature	0.007 min		work

Table S2. Comparison of p-NA catalytic activities of several Ag-Based systems

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