## **Electronic Supplementary Information**

## Hydrogenation of α-pinene over platinum nanoparticles reduced and stabilized by sodium lignosulfonate

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**Table S5.** Hydrogenation of other alkenes catalyzed by SLS-stabilized Pt NPs.

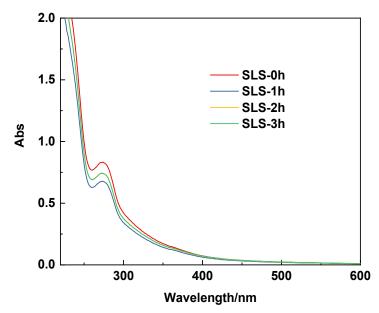


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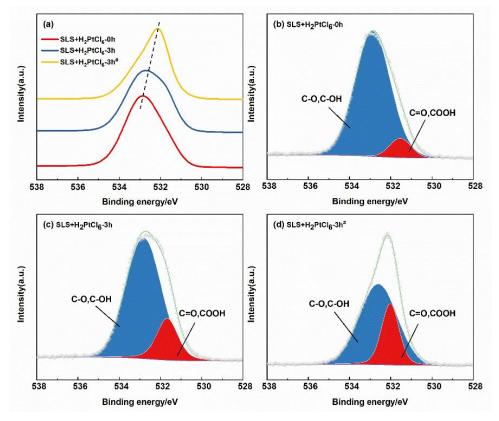


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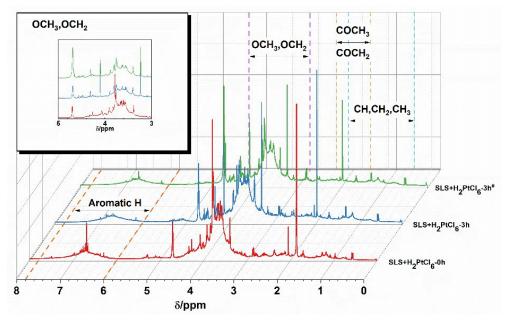
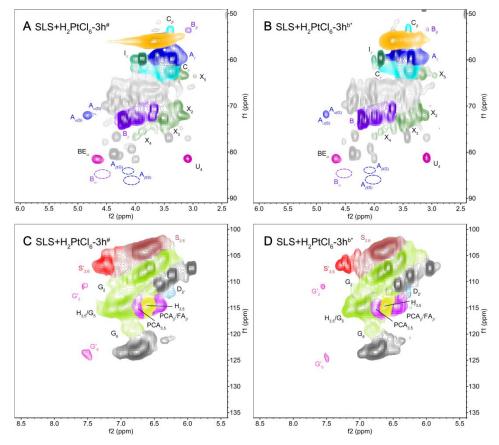
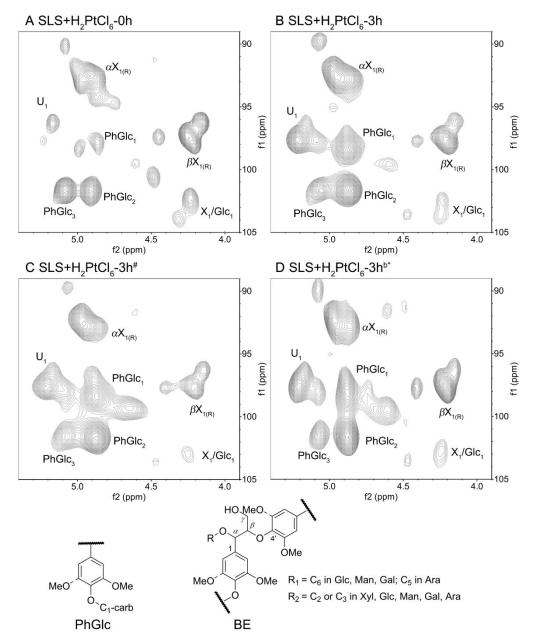


Figure S3. <sup>1</sup>H NMR spectra of SLS+H<sub>2</sub>PtCl<sub>6</sub>-0h, SLS+H<sub>2</sub>PtCl<sub>6</sub>-3h and SLS+H<sub>2</sub>PtCl<sub>6</sub>-3h<sup>#</sup>.



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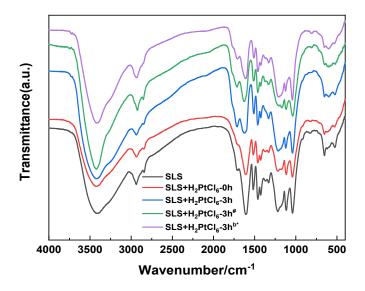


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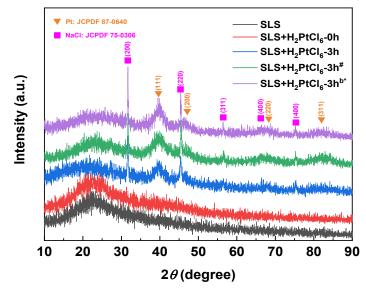
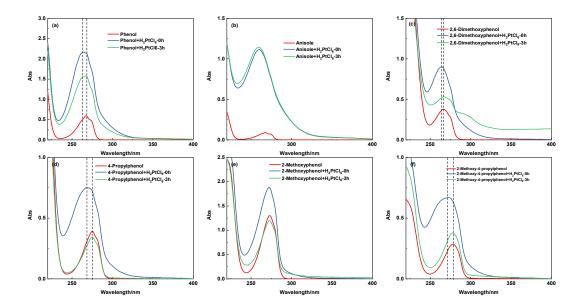


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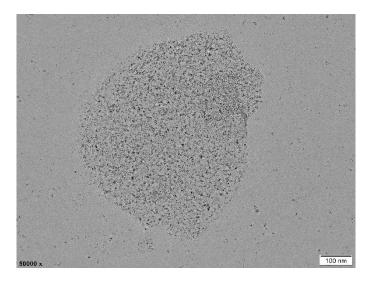
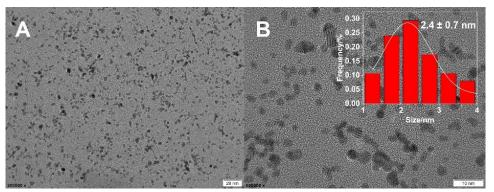


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**Figure S10.** TEM images (A) and Particle Size Distribution (PSD) (B) of the Pt NPs catalyst system after 5 runs.

Sample	Relative Area/% (Binding energy/eV) of C 1s Peaks				Relative Area	/% (Binding	
				C <sub>3</sub> /C <sub>2</sub>	energy/eV) of O 1s Peaks		$O_1/O_2$
Description	C <sub>1</sub> (C-C/C=C)	C <sub>2</sub> (C-O)	C <sub>3</sub> (C=O)		O <sub>1</sub> (C=O)	O <sub>2</sub> (C-O)	-
	35.39/21.24	39.61	3.76	0.09	8.26	91.74	0.09
SLS+H <sub>2</sub> PtCl <sub>6</sub> -0h	(284.7/285.1)	(286.3)	(288.5)		(531.5)	(532.6)	
	43.43/26.06	24.67	5.84	0.24	19.36	80.64	0.24
SLS+H <sub>2</sub> PtCl <sub>6</sub> -3h	(284.7/285.1)	(286.4)	(288.8)		(531.7)	(532.8)	
SLS+H <sub>2</sub> PtCl <sub>6</sub> -3h <sup>#</sup>	51.71/31.03	12.41	4.85	0.39	28.06	71.94	0.39
	(284.7/285.3)	(286.3)	(288.9)		(532.0)	(532.6)	

**Table S1.** Subpeak area fractions of C 1s and O 1s.

Lable	$\delta_{ m C}/\delta_{ m H}$	Assignment
OCH <sub>3</sub>	55.6/3.68	C-H in methoxyls
$A_{\gamma}$	59.8/3.35 and 3.72	$C_{\gamma}$ -H <sub><math>\gamma</math></sub> in $\gamma$ -hydroxylated $\beta$ -O-4' substructures
$A_{\alpha(G)}$	71.8/4.68	$C_{\alpha}$ -H <sub><math>\alpha</math></sub> in $\beta$ -O-4' substructures linked to G-unit
$A_{\alpha(S)}$	72.1/4.85	$C_{\alpha}$ -H <sub><math>\alpha</math></sub> in $\beta$ -O-4' substructures linked to S-unit
$A_{\beta(G)}$	84.1/4.22	$C_{\beta}$ -H <sub><math>\beta</math></sub> in $\beta$ -O-4' substructures linked to G-unit
$A_{\beta(S)}$	86.2/4.09	$C_{\beta}$ -H <sub><math>\beta</math></sub> in $\beta$ -O-4' substructures linked to S-unit
$\mathbf{B}_{a}$	85.5/4.61	$C_{\alpha}$ - $H_{\alpha}$ in $\beta$ - $\beta'$ resinol substructures
$\mathbf{B}_{eta}$	53.5/3.08	$C_{\beta}$ - $H_{\beta}$ in $\beta$ - $\beta'$ resinol substructures
$\mathbf{B}_{\gamma}$	71.2/3.72 and 4.20	$C_{\gamma}$ - $H_{\gamma}$ in $\beta$ - $\beta'$ resinol substructures
$C_{\alpha}$	87.6/5.40	$C_{\alpha}$ -H <sub><math>\alpha</math></sub> in phenylcoumaran substructures
$C_{\beta}$	53.5/3.41	$C_{\beta}$ -H <sub><math>\beta</math></sub> in phenylcoumaran substructures
$C_{\gamma}$	62.7/3.83	$C_{\gamma}$ -H <sub><math>\gamma</math></sub> in phenylcoumaran substructures
$D_{\alpha}$	81.2/5.10	$C_{\alpha}$ -H <sub><math>\alpha</math></sub> in spirodienone substructures
$\mathrm{D}_{eta}$	59.3/2.77	$C_{\beta}$ -H <sub><math>\beta</math></sub> in spirodienone substructures
$\mathrm{D}_{eta'}$	78.5/4.18	$C_{\beta}$ - $H_{\beta'}$ in spirodienone substructures
$D_{2^{\prime}}$	112.7/6.25	C2'-H2' in spirodienone substructures
$D_{6'}$	121.0/6.09	$C_{6'}$ -H <sub>6'</sub> in spirodienone substructures
$I_{\gamma}$	61.6/4.03	$C_{\gamma}$ -H <sub><math>\gamma</math></sub> in cinnamyl alcohol end-groups
S <sub>2,6</sub>	104.2/6.66	C <sub>2,6</sub> -H <sub>2,6</sub> in etherified syringyl units (S)
S <sub>2,6'</sub>	106.3/7.27	$C_{2,6}$ -H <sub>2,6</sub> in oxidized ( $C_{\alpha}$ =O) syringyl units
$G_2'$	110.9/6.89	C <sub>2</sub> -H <sub>2</sub> in guaiacyl units (G, Non-phenolic G)
$G_2$	110.5/7.55	C <sub>2</sub> -H <sub>2</sub> in oxidized (C <sub><math>\alpha</math></sub> =O) guaiacyl units
$G_2$	107.4/6.63	C <sub>2</sub> -H <sub>2</sub> in guaiacyl units (G, Phenolic G)
G <sub>5</sub>	114.9/6.95	C <sub>5</sub> -H <sub>5</sub> in guaiacyl units
$G_6$	118.9/6.78	C <sub>6</sub> -H <sub>6</sub> in guaiacyl units
$G'_6$	123.9/7.55	C <sub>6</sub> -H <sub>6</sub> in oxidized (C <sub><math>\alpha</math></sub> =O) guaiacyl units
$PCA_{\beta}$	114.4/6.43	$C_{\beta}$ - $H_{\beta}$ in <i>p</i> -coumarate
PCA <sub>3,5</sub>	115.1/6.86	$C_{3,5}$ - $H_{3,5}$ in <i>p</i> -coumarate
PCA <sub>α</sub>	143.7/7.60	$C_{\alpha}$ - $H_{\alpha}$ in <i>p</i> -coumarate
H <sub>2,6</sub>	127.7/7.13	C <sub>2,6</sub> -H <sub>2,6</sub> in p-hydroxybenzoate substructures
FA <sub>6</sub>	122.7/7.17	C <sub>6</sub> -H <sub>6</sub> in ferulate

**Table S2.** Assignment of the <sup>1</sup>H-<sup>13</sup>C cross-signals in the 2D HSQC NMR spectra of SLS+H<sub>2</sub>PtCl<sub>6</sub>-0h, SLS+H<sub>2</sub>PtCl<sub>6</sub>-3h, SLS+H<sub>2</sub>PtCl<sub>6</sub>-3h<sup>#</sup> and SLS+H<sub>2</sub>PtCl<sub>6</sub>-3h<sup>b\*</sup>.

Lable	$\delta_{ m C}/\delta_{ m H}$	Assignment		
X <sub>2</sub>	72.2/3.13	$C_2$ -H <sub>2</sub> in $\beta$ -D-xylopyranoside		
$X_3$	73.5/3.42	$C_3$ - $H_3$ in $\beta$ - $_D$ -xylopyranoside		
$X_4$	75.4/3.87	$C_4$ -H <sub>4</sub> in $\beta$ -D-xylopyranoside		
$X_5$	63.2/3.18	$C_5$ - $H_5$ in $\beta$ - $_D$ -xylopyranoside		
$BE_{\alpha}$	81.7/4.68	$C_{\alpha}$ -H <sub><math>\alpha</math></sub> in benzyl ether LCC structures anomeric correlations (C <sub>1</sub> -H <sub>1</sub> )		
$U_4$	81.3/3.10	C <sub>4</sub> -H <sub>4</sub> in 4-O-methyl-α-D-GlcUA		
$\alpha X_{1(R)}$	92.5/4.91	$(1\rightarrow 4)$ - $\alpha$ -D-xylopyranoside (R)		
$\beta X_{1(R)}$	97.4/4.25	$(1\rightarrow 4)$ - $\beta$ -D-xylopyranoside (R)		
$U_1$	97.2/5.20	4-O-methyl-α- <sub>D</sub> -GlcUA		
PhGlc <sub>1</sub>	98.2/4.88	phenyl glycoside linkages		
PhGlc <sub>2</sub>	101.8/4.90	phenyl glycoside linkages		
PhGlc <sub>3</sub>	101.7/5.07	phenyl glycoside linkages		
X <sub>1</sub> /Glc <sub>1</sub>	103.9/4.32	$\beta$ -D-xylopyranoside/ $\beta$ -D-glucopyranoside		

**Table S3.** Assignment of the associated carbohydrate  ${}^{1}H{}^{-13}C$  cross-signals in the 2D HSQC NMR spectra of SLS+H<sub>2</sub>PtCl<sub>6</sub>-0h, SLS+H<sub>2</sub>PtCl<sub>6</sub>-3h, SLS+H<sub>2</sub>PtCl<sub>6</sub>-3h<sup>#</sup> and SLS+H<sub>2</sub>PtCl<sub>6</sub>-3h<sup>b\*</sup>.

Substrates	Structural formulas	Conversion/%	Distribution of products %
<u>.</u>	$\mathbf{i}$	100	82.12
β-pinene	A	100	17.88
3-carene	$\times$	100	100
, .		02.24	26.96
α-terpinene		92.24	73.04
		02.79	33.22
γ- terpinene		92.78	66.78
	= -	99.38	53.54
terpinolene			46.46
limonene		09.42	43.16
millionene	<i>&gt;</i> → 98.43	96.45	56.84
		4.43	47.19
<i>p</i> -cymene		4.45	52.81
			37.42
caryophyllene		97.14	31.22
			31.36
	X .		5.15
longifolene		100	36.25
			58.59

Table S	4. Hydrogenation	of terpenes	catalyzed by	y SLS-stabilized Pt NF	Ps.
-			-		

Reaction conditions:  $n_{(cat.)}$ :  $n_{(\alpha-pinene)} = 1 : 400, 10 \text{ mmol } \alpha$ -pinene, 1 MPa H<sub>2</sub>, 70 °C, 2.0 h.

Substrates	Conversion/%	Products	Selectivity%
	42.43	$\frown$	100
	0.93	$\bigcirc$	100
$\langle $	100	$\bigcirc$	100
$\bigcirc$	100	$\bigcirc$	100
	100	$\sim$	100
$\checkmark \checkmark \checkmark \checkmark \land$	99.84	$\sim$	100

 Table S5. Hydrogenation of other alkenes catalyzed by SLS-stabilized Pt NPs.

Reaction conditions:  $n_{(cat.)}$ :  $n_{(\alpha-pinene)} = 1 : 400, 10 \text{ mmol } \alpha$ -pinene, 1 MPa H<sub>2</sub>, 70 °C, 2.0 h.