Supporting Information

Acetate based Crystallization Kinetics Modulation of CsPbI₂Br for Improved Photovoltaic Performance

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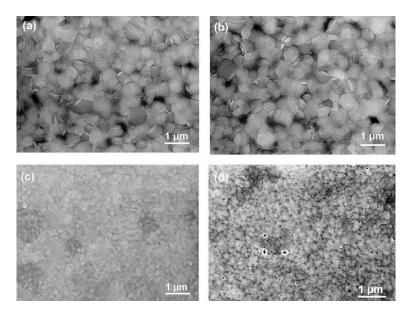


Figure S1. SEM images of the CsPbI₂Br films with different concentration of Co(Ac)₂. (a) 0.1%. (b) 0.25%. (c) 0.5%. (d) 1.5%.

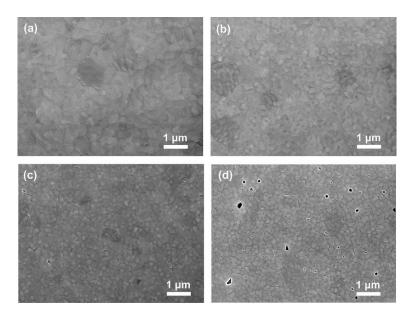


Figure S2. SEM images of the CsPbI₂Br films with different concentration of $Zn(Ac)_2$. (a) 0.5%. (b) 1.0%. (c) 1.5%. (d) 2.0%.

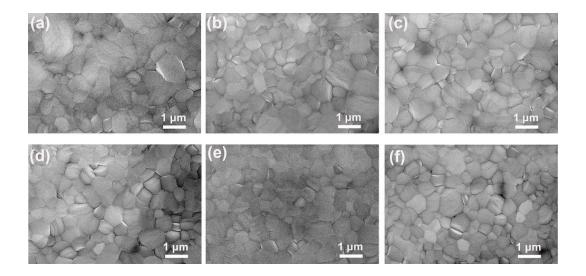


Figure S3. SEM images of the CsPbI₂Br films with different concentration of CoCl₂ or ZnI₂. (a) 0.5% CoCl₂. (b) 1.0% CoCl₂. (c) 2% CoCl₂. (d) 0.5% ZnI₂. (e) 1.0% ZnI₂. (f) 2% ZnI₂.

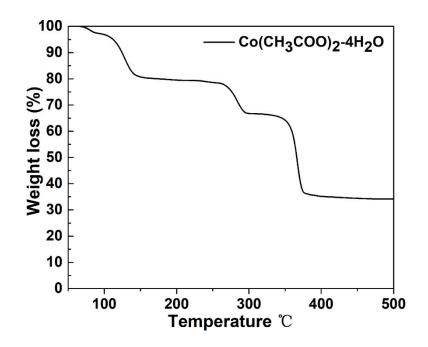


Figure S4. Thermogravimetry (TG) of $Co(Ac)_2$ -4H₂O was measured in nitrogen atmosphere at a temperature rate of 10 °C min⁻¹.

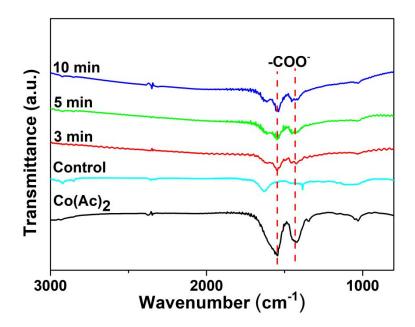


Figure S5. FTIR spectra of CsPbI₂Br/Co(Ac)₂ films with different annealing times at 270 °C.

As indicated in FTIR spectrum of $Co(Ac)_2$ treated at 270 °C for 3 min, two prominent peaks at 1542 and 1413 cm⁻¹ belonging to the asymmetric and symmetric stretching of the carboxylate ion can be observed. To further investigate the thermal stability of Acat such temperature, the annealing time was prolonged to 10 min and the carbonyl signal still remained, which is in agreement with the TG results (Figure S4). This further indicates that $Co(Ac)_2$ is not decomposed in our procedure of preparing the perovskite film at 270 °C for 3 min.

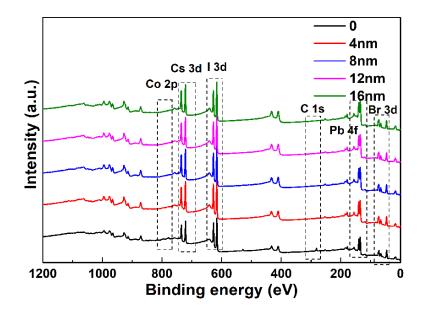


Figure S6. XPS spectra of the CsPbI₂Br/Co(Ac)₂ film in different depth.

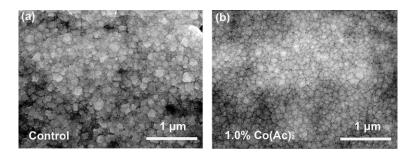


Figure S7. SEM images of the (a) control film and (b) $CsPbI_2Br/Co(Ac)_2$ film annealing at 50 °C.

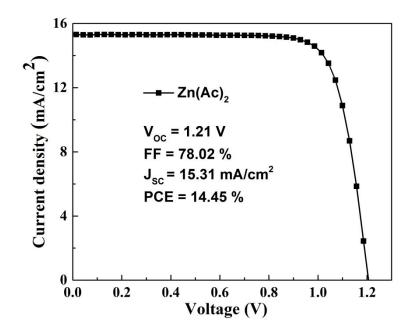


Figure S8. J-V characteristics of the best CsPbI₂Br/Zn(Ac)₂ PSCs under reverse scan

directions.

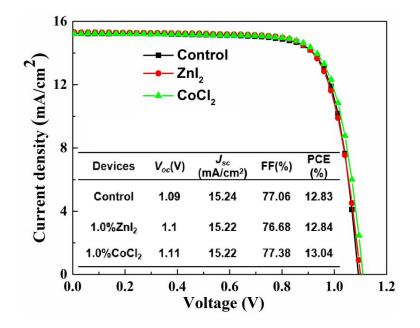


Figure S9. J-V characteristics of the best CsPbI₂Br PSCs with ZnI₂ and CoCl₂.

The *J-V* characteristics for CsPbI₂Br PSCs doped with CoCl₂ and ZnI₂ were almost the same as the control device. By comparing the PCE performance of CoCl₂ and ZnI₂ induced CsPbI₂Br PSCs, it is demonstrated that add a little of Co²⁺ or Zn²⁺ in the precursor solution has also no influence on the performance on the CsPbI₂Br solar cells.

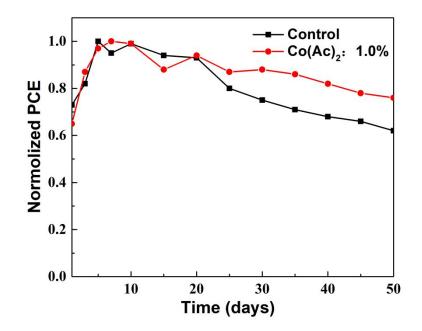


Figure S10. Long-time storage stability of PSC in N_2 glove-box.

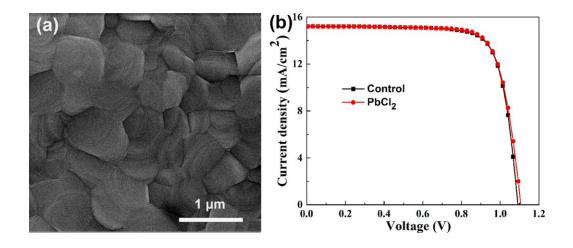


Figure S11. (a) SEM images of the CsPbI₂Br/PbCl₂ film. **(b)** *J-V* characteristics of the best CsPbI₂Br PSCs with 1.0% PbCl₂.

Figure S11a shows SEM images of CsPbI₂Br film with PbCl₂ addition. The morphology of film has not significant change with PbCl₂. The *J-V* characteristics for CsPbI₂Br PSCs with PbCl₂ (shown in Figure S11b) demonstrate that the Cl⁻ has no modification on the performance of CsPbI₂Br PSCs.

Samples	A_{l} (%)	$\tau_l(ns)$	$A_2(\%)$	$\tau_2(ns)$	$\tau_{avg}(ns)$
Control	0.85	2.98	0.16	7.73	4.53
1%-Co(Ac) ₂	0.59	5.19	0.42	11.11	9.57

 Table S1. PL lifetime for the control and CsPbI2Br/Co(Ac)2 film.