

SUPPORTING INFORMATION
Solubilities of Ammonia in Polyethylene Glycols at
(298.2 to 353.2) K and (0 to 200) kPa

Jian-Bo Zhang and Kuan Huang*

Key Laboratory of Poyang Lake Environment and Resource Utilization of Ministry of
Education, School of Resources Environmental and Chemical Engineering, Nanchang
University, Nanchang, Jiangxi 330031, China.

Table S1. Solubilities of CO₂ in PEG200

Measured in this work ^a			Literature data cited from reference 31 ^b		
<i>T</i> /K	<i>P</i> _{CO2} /kPa	<i>m</i> _{CO2} /(mol·kg ⁻¹)	<i>T</i> /K	<i>P</i> _{CO2} /kPa	<i>m</i> _{CO2} /(mol·kg ⁻¹)
313.2	63.5	0.026±0.004	313.2	72.2	0.0312±0.0003
313.2	108.2	0.049±0.004	313.2	91.9	0.0386±0.0003
313.2	147.7	0.068±0.004	313.2	154.8	0.0669±0.0003
313.2	194.5	0.089±0.004	313.2	273.3	0.1240±0.0005
313.2	300.5	0.138±0.004			

^a*T* is the temperature; *P*_{CO2} is the pressure of CO₂; *m*_{CO2} is the solubility of CO₂; standard uncertainties *u* are *u*(*T*)=0.1 K and *u*(*P*_{CO2})=1.2 kPa, *u*(*m*_{CO2}) are reported following the ± sign. ^b*T* is the temperature; *P*_{CO2} is the pressure of CO₂; *m*_{CO2} is the solubility of CO₂ in PEG200; standard uncertainties *u* are *u*(*T*)=0.1 K and *u*(*P*_{CO2})=0.2 kPa, *u*(*m*_{CO2}) are reported following the ± sign.

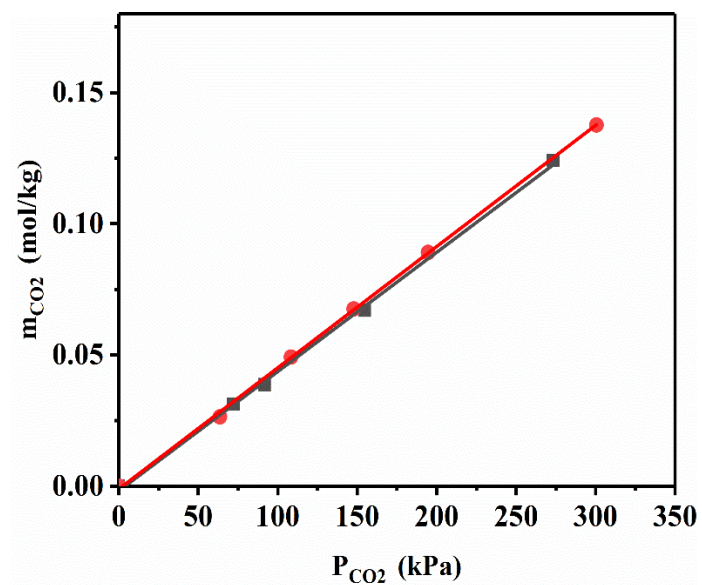


Figure S1. Solubilities of CO₂ in PEG200 (■: literature data cited from reference 31; ●: measured in this work).

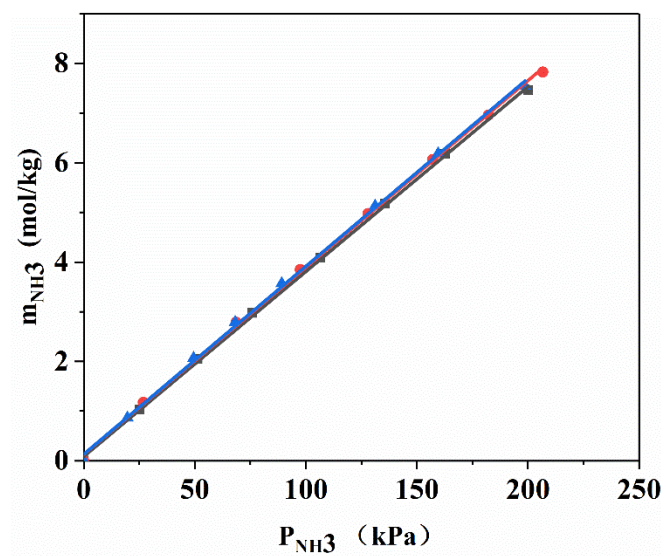


Figure S2. Solubilities of NH₃ in Di-EG at 313.2 K (■: first test; ●: second test; ▲: third test).

Table S2. Experimental and literature densities of PEGs ^a			
Solvent	<i>T</i> (K)	Experimental/(kg/m ³)	Literature/(kg/m ³)
Di-EG	293.2	1116	1116.0
	303.2	1109	1108.9
	313.2	1101	1101.7
	323.2	1095	1094.4
	333.2	1087	1087.1
	343.2	1080	1079.7
	353.2	1072	1072.2
Tri-EG	293.2	1123	1123.1
	303.2	1115	1115.3
	313.2	1108	1107.5
	323.2	1100	1099.6
	333.2	1092	1091.8
	343.2	1084	1083.9
	353.2	1076	1076.1

^aThe literature densities are cited from reference 44.

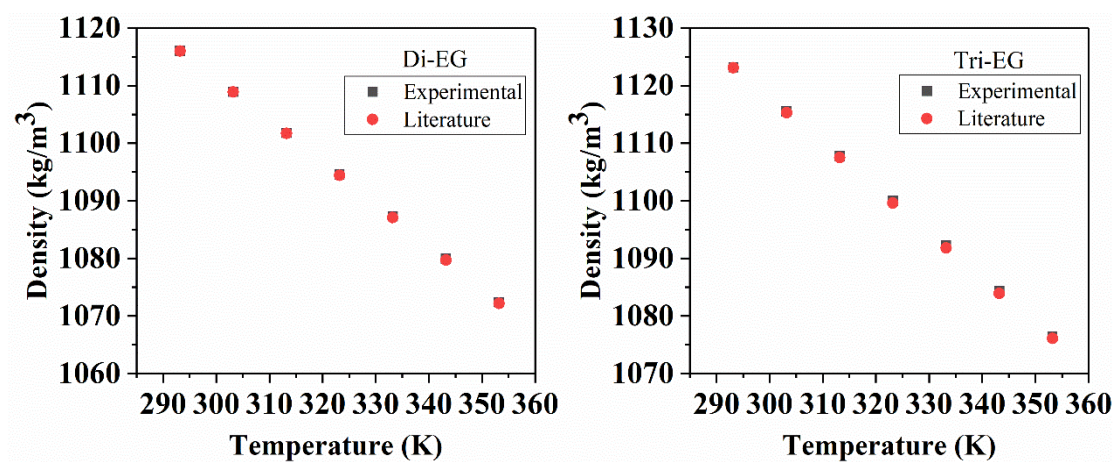


Figure S3. Comparison of experimental and literature densities of PEGs.

Table S3. Experimental and literature surface tensions of PEGs^a

Solvent	<i>T</i> (K)	Experimental/(mN/m)	Literature/(mN/m)
Di-EG	298.2	44.25	45.22
	303.2	43.91	44.08
	313.2	43.38	
	323.2	42.80	
	333.2	42.15	
	343.2	41.47	
Tri-EG	293.2	45.28	
	303.2	44.71	44.71
	313.2	44.16	
	323.2	43.36	
	333.2	42.43	
	343.2	41.55	

^aThe literature surface tensions are cited from references 45 and 46.

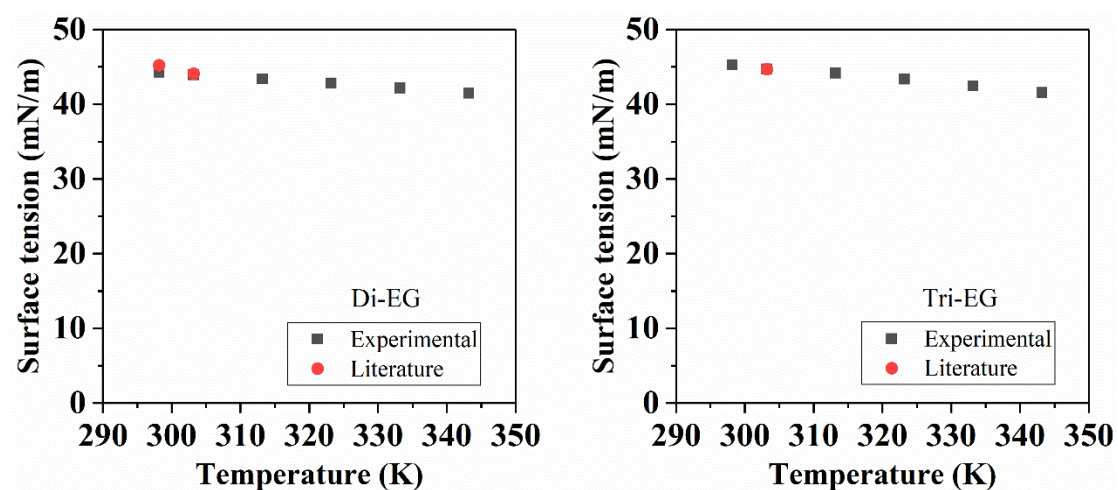


Figure S4. Comparison of experimental and literature surface tensions of PEGs.