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From: Liquid Crystals (onbehalfof@manuscriptcentral.com)

To: dakshinamurthy_potukuchi@yahoo.com

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Dear Professor Potukuchi

Your manuscript entitled "Influence of Meta-Extended Rigid-Core, Complementary Hydrogen Bonding and Flexible Chain on Polymorphism in Schiff-based Hydrogen Bonded Liquid Crystals: (4)MeOBD(3)AmnBA:nOBAs" which you submitted to Liquid Crystals, has been reviewed. The reviewer's comments are included at the bottom of this letter.

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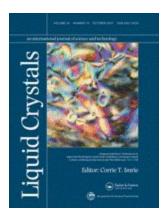
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The reviewer acknowledges the efforts made by the authors in this latter revision. There are still minor aspects that should be addressed, including a detailed proof read of the next version, and possibly language polishing. Some of these are flagged in the document attached, as comments in a pdf file. Sincerely,



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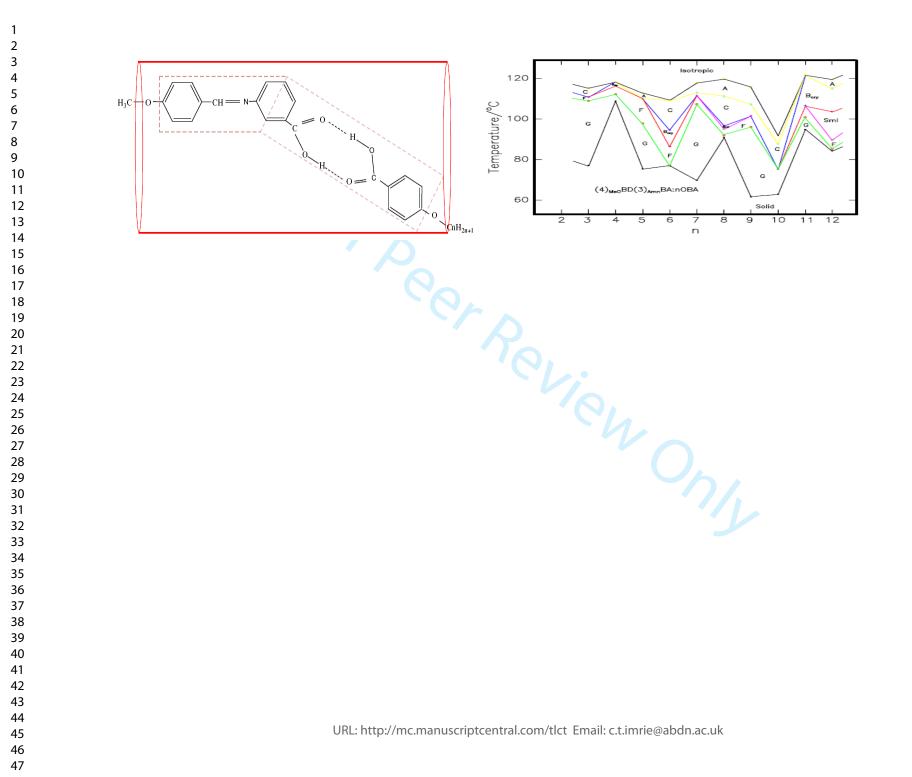
Liquid Crystals



Influence of Meta-Extended Rigid-Core, Complementary Hydrogen Bonding and Flexible Chain on Polymorphism in Schiff-based Hydrogen Bonded Liquid Crystals: (4)MeOBD(3)AmnBA:nOBAs

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Influence of Meta-Extended Rigid-Core, Complementary Hydrogen Bonding and Flexible Chain on Polymorphism in Schiff-based Liquid Crystals: (4)_{MeO}BD(3)_{Am}BA:nOBAs

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Abstract

Schiff based complementary hydrogen bonded liquid crystals (HBLC), viz., (4)_{MeO}BD(3)_{Am}BA:nOBA's with flexible chain length for n=3,4,5,6,7,8,9,10,11 and 12 are reported.¹H;¹³C-NMR and Infra-Red spectroscopy used to confirm the formation of HBLCs. LC phases and transition temperatures (T_c) determined by polarized optical microscopy (POM) and differential scanning calorimetry (DSC). T_c and enthalpy (ΔH) determined by DSC also. Odd-even effect observed at clearing and melting transitions. Influence of Schiff base and Oxygen as bridging atom promote smectic phase abundance. HBLCs exhibit tetra- or penta-phase variance. Maximum (penta) phase variance is exhibited by n=8 and 12 with long flexible chain. Prevalent abundance of quasi-2dimensional (2D) LC phases SmF and SmI are observed. Nematic phase is quenched. Lower (n=4) and intermediate (n=6) members exhibited SmB_{crvst} phase. Predominant occurrence of enantiotropic LC phases is noticed. All members exhibited 3D tilted SmG phase. AC transition exhibited by intermediate homologues (for n=7, 8, 9 and 10) is found to be either second order or with very small enthalpy. Phase diagram reveals the abundance of multi-critical points, some involving exotic symmetries. Influence of meta-extended rigid core, complementary HB and flexibility are studied for the LC phase abundance with characteristic structural order. POM and DSC results are discussed in the wake of reports in other achiral calamitic LCs.

Keywords: Schiff based moieties; Complementary Hydrogen Bonding; Smectic liquid crystal phases, Transition temperatures; Enthalpy; Odd-even effect; Multi-critical point; Phase stability.

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1. Introduction

Liquid Crystals (LC) possess dual [1] properties of isotropic fluids and anisotropic crystals. LC phase structures are interesting as they catered the awaited experimental proof [2,3] for the theoretical predictions of 1D-, 2D- and quasi 2Dcrystal melting transitions and anisotropic scaling hypothesis. Recently, they are identified [4] to mimic the transitions of topological insulators. Inherent large birefringence and high contrast ratios along with the fluid nature exhibited by LC phase structures promoted [5,6] their utility in electro-optic (EO) devices. Interdisciplinary research activity by physicists and chemists in the area of LCs is targeted to realize ambient LC phase structures with large field response. LC physico-chemical research involves design and architectural aspects like chemical moieties, configuration, conjugation, chiral centers, interaction (bonding between moieties) etc., Design of calamitic [7] or rod-like, discotic [8], hydrogen bonded [9-12], bent [13], phasmidic [14] etc., types of LCs are known to exhibit phases of structural diversity. Initially, Nematic LCs (or their structural analogues like cholesterics etc.,) are underlined by long range orientational order [7] are used in displays [15] operating with millisecond speed. Advent of ferroelectric response in chiral calamitic type of LCs stimulated [16] the LC synthetic research activity to realize the layered tilted smectic phases. Chiral versions of tilted smectic phases, SmC* are found to respond [6] at microsecond speed. Hence, realization of tilted smectic phases with wide temperature range and large tilt angle in the vicinity of ambient temperatures occupied the center stage of LC synthetic chemists.

Much of the LC molecular design reported [17,18] owes to the calamitic/rod type molecular structure.Details of chemical moieties (aliphatic/aromatic) used, viz., as rigid core [17,18], bridging(Oxygen atom) groups [18,19] and flexible chains [17,19-21], chemical bonding or interaction [9-12], configuration of polar substitutions [22-26], conjugated environment [17,22,27,28], ester moieties [16,17,22,25,29,30], aromatic moieties [17,19,26,30], Schiff based [17,19,25,31,32] moieties etc., on the molecular body are found to influence the stability and diversity of LC phases towards ambient temperatures. Involvement of hydrogen bonding (HB or soft-covalent) interaction [9-12] on the molecular frame of LC is known as the most effective way of growth of supra molecular (SM) assemblies. Design of HBLCs mediated by soft-covalent interaction and SM assembly of chemical moieties

accelerated the research activity. Selection of proton donor/acceptor moieties, direction of bonding (with respect to the molecular long axis) and bonding nature, i.e., being linear [10,32-35], complementary [9,24,36], alternative [37], double [38-40] types are also reported to influence the stability and diversity of LC phase structures. Presence of HB in LCs is found [9,10,24,28-31,36-40,41-42] to increase phase stability towards room temperature. Complementary type of HB is recognized [9,24,36,38-40,41-42] to enhance the rigidity component in calamitic type of LCs, and thereby to promote LC phase abundance. Inclined configuration of HB (to the molecule's longitudinal direction) is found [39] to enhance the transverse dipole moment μ_t and result for induction of tilted smectic phases. Additionally, the tilted smectic phase stability is found [36,39,40] to get depressed towards room temperatures along with growth of re-entrant SmC phase by the involvement of HB interaction in dimers. Intra-molecular hydrogen bonding is also reported [10] to influence mesomorphic stability. Phase stability is reported [11] to be influenced by the inclusion of HB in main chain, side chain and cross-linked polymeric liquid crystals also.

Continuous increasing of chain length (or flexible component by adding methylene -CH₂- units) is found [17,19-21,43-45] to result for alternating trend of melting and clearing temperatures, known as odd-even (OE) effect. OE effect indirectly influences the LC phase stability and it is reported to be originated [44] by the alternating contributions of axial polarizabilities. Lateral substitutions [18,24,46] with highly polar substitutions (-C=N, -F, -Cl, -Br, -NO₂, -NH₂ etc.,)on the calamitic molecular frame are found to result for large Nematic phase stability. Increasing flexible end chain length is argued [47,48] to induce orientational disorder, which in turn promotes the stability of quasi-2D hexatic smectic phases (SmB_{Hex}, SmI, SmF) with bond-orientational order and exotic symmetries. Involving electro-negative Oxygen as bridging (between flexible chain and rigid core) atom is found [19,49] to promote lateral stacking and lead to the induction and stability of layered smectic phases. Bent type of molecular architecture is found [13,43,50,51-56] to result for ferro-electric (FE) response in achiral molecules also. Discotic molecular design is found [57,58] to result for semiconductor properties and thermoelectric response in LC phases.

Investigations for influence of design and architecture features of molecules capable of exhibiting LC phases of device interest are an ongoing topic of research.

Liquid Crystals

Selection of O-atom as bridging moiety [22,23] and Schiff base [17,19,25,31,32] as core are known to promote lateral stacking and growth of layered smectic phases. HB interaction inLCs is found be inclined with respect to the long molecular axis, which in turn has enhanced device savvy tilted smectic phase stability. Despite the achiral constitution, B_2 and B_5 phases of bent shaped LCs are known [13,28,50] to exhibit ferroelectric response. In calamitic type of LC molecules, a meta-positioned extension is anticipated to result for enhanced μ_t and impart slightly bent shape for the possible enhancement of device savvy tilted phase abundance. In the wake of the expected influence of molecular (Figure-1) design (or architectural features by proper [7-19] selection of chemical moieties) for the growth of device savvy LC phases, an attempt is made to design a series of calamitic type of achiral LCs possessing

i) soft-covalent (HB) interaction,

ii) Schiff based moieties,

iii) electron releasing lateral substitutions and

iv) meta-extended rigid core

in anticipation of enhanced tilted smectic phases.

An indigenously synthesized series of calamitic type of LCs possessing complementary HBcomprised of meta extended core are investigated by nuclear magnetic resonance (NMR) and Fourier-Transform Infrared (FTIR) spectroscopy techniques to confirm their formation. LC phase structures and phase transition temperatures (T_c) are determined [59,60] with the help of polarized optical spectroscopy (POM). Phase transition temperatures and heats of transition (enthalpy Δ H) associated with LC phases are also determined by differential scanning calorimetry (DSC). Total LC phase stability [Δ T]_{LC}, SmC phase stability [Δ T]_C and overall tilted phase stability [Δ T]_{Tilt} is estimated by evaluating the difference between clearing and melting temperatures of the phase of interest. However, [Δ T]_{Tilt} includes the data for all the tilted LC phase thermal stabilities viz., SmC, SmI, SmF, and SmG. Paper is organized in 3-sections comprised of introduction, details of experimental techniques, synthetic route and discussion of results.

2.Experimental Methods and Synthesis 2a. Experimental Techniques A Bruker - Avance NMR spectrometer is used to record ¹H-NMR spectra (300 MHz) and ¹³C-NMR spectra (75 MHz) for the identification [61,62] and tally of H-atoms (Figure-2 as representative) and C-atoms (Figure-3 as representative) positioned at various places of the targeted of HBLC complex. The presence of HB is identified [63] with solid state (KBr pellet) FTIR spectra (ABB Bomem) at room temperature and analyzed with MB3000 software. FTIR spectra is analyzed in the backdrop of reports [64-68] on supra-molecular LCs possessing HB interaction, viz., in aromatic/aliphatic acids, nOBAs and Schiff base compounds involving >C=O, -OH stretching and Fermi resonance of A-, B- and C-bands. POM textures [59,60] are recorded and analyzed with the help under programmed heating/cooling mode at a rate of 0.1°C/min. LC samples targeted for POM study are prepared as microscopic (through rubbing technique). POM studies are carried out over slides homogeneously aligned sample also by introducing the sample into the Device Tech (USA) madepolyimide buffed thin transparent glass cells of 5 µm thickness. Thustextural observations of LC phase structures are carried out using an SD Tech POM equipped with Charged Coupled Device (CCD) camera and u-master software inconjunction with an INSTEC 402 hot stage (STC 200 heating system). Phase transition temperatures recorded with POM technique are accurate to ±0.1°C. A Shimadzu DSC-60 differential scanning calorimeter is used to determine the LC phase transition temperatures ($\pm 0.1^{\circ}$ C) and the heat (Δ H, at ± 0.01 J/g) of transition.

The cooling and heating rates adopted during the DSC are 5°Cmt⁻¹.

2b. Synthesis

The chemical reactants, viz., mesogenic p-n-alkoxy benzoic acids (nOBA's), pmethoxybenzaldehyde and m-aminobenzoicacid are procured from Sigma Aldrich (USA). Solvents of AR grade, viz., tetrahydrofuran (THF), glacial acetic acid and ethyl alcohol are procured from CDH (India).

The **three-step** synthetic path followed during the preparation of targeted HBLC series is presented in Scheme-1. In the **first** step, p-methoxybenzaldehyde and m-amino benzoic acid were mixed in equi-molar ratio (1:1) in ethanol at cold conditions. 4-5 drops of glacial acetic acid is added to enhance the yield of Schiff based intermediate compound in the condensation reaction. Intermediate product, viz.,3-(4-methoxybenzylideneamino)benzoic acid is isolated from solvent by filtration under reduced pressure conditions. The product obtained in the 1st step is found to

be colourless which signifies its initial purity and susceptibility against possible chemical degradation. In the **second** step, the dimeric forms of nOBAs (for n=3 to 12) were converted to monomer form by dissolving in THF. In the **third**step, each monomer form of nOBA is mixed with 3-(4-methoxybenzylideneamino) benzoic acid in 1:1 molar ratio in excess amount of THF and the reactants are refluxed for 5 hours. Solvent is removed after the formation of HBLC complex under reduced pressure conditions. The general molecular formula for the final product HBLC complex is given as $(4)_{MeO}BD(3)_{Am}BA:nOBA$.

3. Results and Discussion

3a. Supra-Molecular Structure of HBLC complexes

Supra-molecular -tructure (Figure-4) of the targeted HBLC compound constituted by HB interaction and meta-position extendedrigid core part is presented in scheme-1. An overview of HBLC seems to entail a rigid core constituted by a Schiff base (possessing meta-situated acid) interacted through HB with an aromatic acid (-COOH of nOBA). Overview of the molecular structure reveals that one of flexible ends contains Oxygen as bridging atom. As O-atom is highly electronegative, it is anticipated to promote lateral stacking [22] to pronounce smectic phase abundance. It is also noticed that Schiff based acid (4)_{MeO}BD(3)_{Amn}BA and nOBA along with the HB interacted space (on the molecular frame) provide the essential rigidity to the LC molecule. However, length of end chain (-CH₂ units in nOBA) is varied to tune the flexibility for the possible induction of orientational disorder.

3b. FTIR Spectra

FTIR study. The FTIR spectra of (4)_{MeO}BD(3)_{Amn}BA:5OBA (as representative) as augmented by the spectra for 5OBA and schiff based intermediate are presented in Figures-5_{a,b,c}. IR spectra of chemical componentscarboxylic acid i.e., 5OBA and intermediate Schiff's base, viz., (4)_{MeO}BD(3)_{Amn}BA are also presented in the same figure for comparative analysis. The data for prominent IR absorptions characteristic underlying functional groups is presented inTable-1.

The FTIR spectrum of $(4)_{MeO}BD(3)_{Amn}BA$ exhibited strong absorption peaks at 2931 and 2869cm⁻¹which correspond to the C-H stretching. A shoulder at 3062cm⁻¹ is identified to correspond to the fundamental mode of vibration of hydroxyl (–OH) group of the carboxylic acid. It is also noticed that the absorption peak expected in

Liquid Crystals

vicinity of 3500cm^{-1} relevant to free hydroxyl group is absent. Absence indicates the involvement of –OH group withan extensive no interaction. Two bands at 2669 and 2561cm^{-1} are also observed which are assigned to the resonance between the fundamental mode of vibration and the in-plane bending of –OH group. These observations are in concurrence with the reported [67] values. A strong absorption peak observed at 1697cm^{-1} is argued due to the>C=O stretching of carboxylic acid. The peak observed at 1596cm^{-1} is assigned to the >CH=N- of the Schiff's base intermediate, $(4)_{\text{MeO}}\text{BD}(3)_{\text{Am}}\text{BA}$.

In the IR spectra of 5OBA Strong absorptionpeaks (as a shoulder) at 3062cm⁻¹ are assigned to –OH group. Peaks observed at 2952cm⁻¹, 2869cm⁻¹are assigned to the C-H stretching modes of 5OBA. Absorption bands observed at 2661cm⁻¹ and 2543cm⁻¹ are assigned as resonant absorptions of fundamental vibrations and inplane bending of the -OH group. An absorption peak observed at 1919cm⁻¹ with low intensity is assigned to the resonance mode of first and the second harmonics of the –OH group. The fundamental mode of vibration of –OH group is argued to merge with the C–H stretching. Absorption peak observed at 1676cm⁻¹due to carbonyl (>C=O) stretching mode (of 50BA) concurs with the reports [68] and confirms the closed HB dimer configuration for 50BA. The peaks observed at 1514, 1577 and 1602cm⁻¹ are attributed to the skeletal vibrations in the aromatic core.

A broad absorption peak in the range of 3055 - 2831cm⁻¹ corresponding to the C-H stretching (in which the fundamental mode of vibration of –OH is merged) is observed for FTIR spectrum of $(4)_{MeO}$ BD(3)_{Am}BA:5OBA. The peaks at 2669cm⁻¹ and 2553cm⁻¹ correspond to the Fermi resonance between the fundamental mode and in-plane bending of –OH moiety. The peak observed at 1897cm⁻¹ is argued due to the resonance between first and second harmonics pertaining to –OH moiety in the HBLC. A strong absorption peak observed at 1681cm⁻¹ is assigned to the >C=O stretching mode. An absorption peak observed at 1604cm⁻¹ is argued to correspond to the Schiff's base. IR absorption peaks observed at 1512cm⁻¹are assigned to the -CH=CH- stretching. The appearance of strong absorption observed at 1681cm⁻¹ (pertaining to carbonyl >C=O stretching of carboxylic acids) are found to accompany with a bathochromic shift of ~5cm⁻¹ (in comparison with that for 5OBA). They are also found to accompany with a hypsochromic shift of ~15cm⁻¹ with respect to (4)_{MeO}BD(3)_{Am}BA. Observed IR shifts infer that the carbonyl groups (of acids) participate in HB interaction. The absorption corresponding to –OH group is

Page 9 of 60

Liquid Crystals

observed to manifest as a shoulder in all the three cases. But, it is observed to accompany with a shift to 3055cm⁻¹ in HB complex, viz., from 3062cm⁻¹observed for respective carboxylic acids. Observed shift infers that the –OH groups of both carboxylic acids participate in HB bond interaction. IR peak observed at 1681cm⁻¹ is argued due to the carbonyl (>C=O) stretching mode and the closed (if not cyclic or nor homocynthon) configuration of constituent carboxylic acids.

941.24cm⁻¹ An IR absorption at is noticed of in spectra (4)_{MeO}BD(3)_{Am}BA:5OBA. This IR absorption is found to be comparable for the torsional γ -modes of –OH moietyand due to Fermi resonance [68] in pure nOBAs with (945cm⁻¹) dimeric (and cyclic) HB interaction. It is also found to agree with the reported [65] torsional γ -modes of -OH moiety pertaining to other organic compounds (948cm⁻¹) with HB interaction. A weak intensity IR absorption observed at 1913.25cm⁻¹ is found to be comparable to that reported [65] for C-band relevant to υ (–OH) mode in Pyridine+CCl₄ system with HB. Hence, observed weak IR absorption is argued due to the C-band pertinent Fermi resonance of -(OH) stretching of -COOH moiety (in the intermediate Schiff base meta-substituted acid) due to its participation in HB interaction.

Observed IR absorptions confirms [63] the presence of HB on the body of $(4)_{MeO}BD(3)_{Am}BA:nOBA$ type of LC complex. Since O-atom in the –COOH of nOBA moieties interacts with H-atom of –COOH of $(4)_{MeO}BD(3)_{Am}BA$ or vice versa, the HB is complementary type.

3c. Phase Characterization by POM Textures and Determination of Transition Temperatures (T_c)

The LC phases exhibited by present series of HBLC complexes are characterized by the POM textural observations [59,60] carried out during heating and cooling runs. The LC filled cells (or slides) are placed in the temperature controller kept on POM hot stage set for crossed polar configuration. The LC phases exhibited by(4)_{MeO}BD(3)_{Am}BA:nOBA series of compounds along with the phase transition temperatures are presented in Table-2. The present series of HBLCs are found to exhibit SmA, SmB_{cry}, SmC, SmI, SmF and SmG phases.

SmA phase is observed to exhibit (Plate-1) focal conic fan, elongated (lathelike) fan board (Plate-2) and spiked fan (Plate-3) textures. The spiked fan texture is found to be analogous to that reported [60] for TGBA version of SmA confined in homeotropic geometry. Although series of HBLCs possess dominant

calamitic type of (Scheme-1) architectural features, the LC core looks like slightly bent structured with more mass on one side. Similarity between the textures exhibited by meta-extended core bearing HBLC and reported chiral LCs is argued due to the similarity of molecular architecture. It is noticed that the latter case [69,70] represents a one-sided mass loaded core with double and/or triple chiral centers. Meta-extension of rigid core is found to be akin to configuration of skewed mass in Cholesteryl Pelargonate. SmC phase is found to exhibit four brushed schlieren texture (Plate-4) as grown from the precursor SmA phase in heating run of(4)_{MeO}BD(3)_{Amn}BA:8OBA. SmC phase is found to exhibit black-on-white natural mosaic texture (Plate-5) also. SmB_{crv} is found to grow as colored mosaic (Plate-6) and glossy focal conic fan batonet textures (Plate-7). Sml is found to exhibit broken focal conic plume-like fan (Plate-8) texture when grown by cooling the SmB_{crv} in (4)_{MeO}BD(3)_{Amn}BA:8OBA. Sml is also found to exhibit colored threaded (Plate-9) texture in (4)_{MeO}BD(3)_{Amn}BA:12OBA. However, SmF is found to exhibit characteristic colored marble threaded texture (Plate-10) and checkered-board of elongated fan textures (Plate-11) in HBLC complex for n=9 and 12, respectively. In addition to the above textures, SmF is also found to exhibit a paramorphotic colored mosaic texture with undulated boundaries (Plate-12). Paramorphotic texture is found to be much akin to the undulated edge texture observed for n=11 in the cooling run. SmG exhibited characteristic colored mosaic (Plate-13) texture in the lower homologues of the present series of HBLCs.

LC phase variance is found to be large for n=8 and 12 compounds with ACIFG and AB_{cry}IFG phase sequences, respectively. It is observed that lower homologues (viz., for n≤6) and higher homologues (viz., for n=6 and 10 or 11) prevalently exhibit lower phase variance. It is also observed (Table-2) that LC clearing temperatures (T_{IC} , T_{IBcry} , T_{IA}) and solid melting temperatures (T_{cry-LC}) exhibit O-E effect. However, O-E effect is rather rigorous at melting transitions to infer the dominant contributions [43-45] of axial polarizabilities. It is also noticed that the LC phase occurrence (in both of the heating and cooling scans) follows the standard hierarchy [59] and thermal order for growth of LC structural order.

It is noticed (Table-2) that although intermediate homologues do not exhibit 3D SmB_{cryst} phase, lower homologues (for n=4 and 6) with short chain exhibit orthogonal (hexagonal) SmB_{cryst} phase. Intermediate homologues (for n=7,8,9 and 10) do not exhibit 3D SmB_{cryst} phase. However, higher homologues (for n=11 and

12) with large flexible chain exhibited this 3D SmB_{cryst} phase with further increase of chain (n) length. Thus, SmB_{cryst} phase appears to be re-entered with increasing flexibility in present HBLCs. Odd lower homologues, all intermediate homologues and higher homologues are found to exhibit enhanced tilted phase stability involving SmC, SmF, SmI and SmG phases. Compounds with n=5,7,8,9,10,11 and 12 exhibited dominant tilted phase stability. However, tilted LC phase abundance is rather predominant in intermediate homologues. Overall tilted phase abundance in HBLCs is observed to witness an enormous amount of increase in comparison with nOBAs. In comparison with [10] nOBA's, meta-extended core LCs (table-2) with O-atom as bridging atom are found to exhibit larger layered smectic polymorphism. Hence, the increased smectic phase abundance is argued due to predominant lateral stacking capability in meta-extended core LCs. It is interesting to notify that meta-extended rigid core, complementary hydrogen bonding and tunable flexibility (by varying length 'n' of end chain) establish an effective set of molecular design parameters to promote tilted LC phase stability in HBLCs.

3d Differential Scanning Calorimetry

3d(i) Phase Transition Temperatures (T_c) and Enthalpy (Δ H) and LC Phase range

DSC thermogram recorded during heating and cooling scans for $(4)_{MeO}BD(3)_{Am}BA:nOBAs$ is presented (for n=7 as representative) in Figure-6. Data for the phase transition temperature (T_c) and the associated heats of transition (ΔH) observed during heating and cooling scans (5°C/min) is presented in Table-2. Some of the peaks/valleys are observed to possess shoulders. In the case of DSC shoulders, the derivative curve is drawn to determine transition temperature T_{c} . Comparison of clearing transition temperatures and melting transition temperatures in nOBA's [10] with that in table-2 reveals that clearing temperatures witness a shift from a maximum value of 160°C (for n = 4 in nOBA's) to 121.8°C (for n=11 in metaextended HBLCs). Nevertheless, the minimum of clearing temperature is also found to be shifted from 137°C (for n = 12 in nOBA's) to 91.8°C (for n=10 in meta-extended HBLCs). Analogously, comparison of maximum of melting transition temperatures in nOBA's and HBLC's also revealed a shift from 145° C (for n = 3 in nOBA's) to 112.2°C (for n=4 in meta-extended HBLCs). The minimum of melting temperature is also found to be shifted from $92^{\circ}C$ (for n = 7 in nOBA's) to ambient temperatures (for n=3 in meta-extended HBLCs). An overview of the observed depression of clearing and melting temperatures suggests that LCs with meta-extended core and complementary HB interaction results for the realization of LC phases structures towards ambient temperatures (rather than in their pristine LC components).

The enthalpy ΔH for DSC shoulders is argued to extend over two transitions. The ΔH for DSC shoulders is also presented in Table-2 with the extended dotted line over two transitions. The phase transition temperatures, viz., T_{I-A}, T_{I-C}, T_{I-BCry}, T_{A-C}, T_{A-F}, T_{C-F}, T_{F-G}, T_{BCry-F}, T_{C-SmI}, T_{Bcry-SmI}, T_{SmI-F} and T_{G-Solid} determined by DSC are found to agree with the data of POM.

The observed finite Δ H across I-A, I-C, I-B_{cry},C-B_{cry},C-SmI,C-F,B_{cry}-SmI,B_{cry}-F, SmI-F, C-G, F-G, F-SId and G-SId transitions infer their first order nature. It may be noticed that A-C transition in present HBLCs is accompanied with no enthalpy or very small enthalpy to agree with the expected [1] to be of second ordered nature.

However, a small, but non-zero and finite (i.e., 0.39J/g across A-C transition for n=10) heat of transition is argued due to the small range of precursor SmA phase. As such, it is argued to be a case of fluctuation induced weak first order A-C transition. SmI-SmF transition involves a marginal change of tilt direction of pseudo hexagonal network (from apex to side). Its weak first ordernature argued due to the small thermal range of precursor SmI phase. However, slightly higher (or large) enthalpy is witnessed across SmI-SmF transition (in cooling scan)for n=8. Large enthalpy across this transition is attributed to the fluctuations.

3d(ii) Enantiotropic/Monotropic Phase Occurrence, LC Phase Variance and Chain length

Monotropic LC phase occurrence is observed for SmC and SmF phases for n=3 (in cooling run); SmG phase for n=6 (in heating run); SmF (in heating run) and SmG (in cooling run) for n=10; SmA for n=11 (in heating run) and SmI and SmF phases (in cooling run) for n=12. Relatively large Monotropic occurrence is found to occur in tilted smectic versions for even higher homologues. Higher homologues contain large flexible chain. Hence, orientational disorder caused by the large flexible component in HBLCs is argued to result for the prevalent monotropic occurrence. An overview of the data reveals that few number (~8) of monotropic phase occurrences appear out of a large number (~70) of total LC phase occurrences for meta-extended core constituted HBLCs, viz., in $(4)_{MeO}BD(3)_{Am}BA:nOBAs$. POM revealed that out of

Page 13 of 60

Liquid Crystals

ten HBLCs, six (viz., for n = 3,4,5,6,10 and 11) exhibited tri-variant LC phase abundance in cooling scan. Compounds with n = 4,5,6 and 11 are found to exhibit enantiotropic LC phases. Two HBLCs (for n=7 and 9) are found to exhibit enantiotropic tetra-variant LC phase abundance. Remaining two HBLCs (for n=8 and 12) exhibit penta-variant phase abundance in cooling scan. However, HBLC with n =8 exhibits enantiotropic phase occurrence. In HBLC with n= 12, SmI and SmF phases appear as monotropic abundance in cooling scan. As Monotropic occurrence is very less, it may be concluded that enantiotropic LC phase occurrence is dominant in meta-extended HBLCs.

3e. Phase Diagram and Phase Stability

A phase diagram (Figure-7) is constructed with the help of data (Table-2 in cooling scan) to investigate the influence of flexible component (i.e., increasing chain length 'n') on the LC phase occurrence. A comparative study is made with the reported [10] data in nOBAs. As per the molecular structure (fig.-4) of (4)_{MeO}BD(3)_{Am}BA:nOBAs, flexible component is effectively tuned by increasing the end chain (n) length. Increasing the flexibility is effectively carried out by increasing the number of methylene units (-CH₂) at nOBA side of HBLC complex. Both of these series contain complementary type of HB interaction, while (4)_{MeO}BD(3)_{Am}BA:nOBAs possess a meta-extended core. Such an analysis is anticipated to throw light on the possible impact of meta-extended core and increase in flexible (n) component for the abundance of LC polymorphism. Although both of them exhibited a decreasing trend of clearing temperatures with increasing flexibility, O-E effect at LC clearing interface (for where LC undergo clearing transition in to isotropic liquid) is observed to be more intense in meta-extended core HBLCs. Relatively decreasing trend of clearing transition infers that meta-extended core LCs are capable of bringing down the LC thermal stability towards ambient temperatures effectively. However, the O-E effect with increase of 'n' being relatively severe at solid melting interface (where solid melts in to LC phase), it reinstates that meta-extended core effectively brings down the LC phase temperatures to the room temperature range. It is noticed that tilted LC phase abundance (i.e., by origin of SmC phase) is initiated for n>6 in nOBA's. But, tilted LC phase abundance is observed for $n \ge 3$ in meta-extended core LCs. It is reckoned back that O-E effect is argued due to the alternating contributions [43-45] of axial polarizabilities, while odd homologues possess relatively more transverse dipole moment (μ_t) to pronounce tilted LC phase occurrence. O-E is expected to be

Liquid Crystals

severe at solid melting transition involving LC phases with smaller inter-particle separation due to low temperature conditions. Hence, it is argued that metaextended core LCs could promote μ_t more effectively to enhance the occurrence of tilted phases. Nematic phase is found to quench at n = 14 in nOBA's, while it is found absent even for n = 3 in the series of meta-extended core LCs. Nematic phase with orientational order is known to be stabilized by the longitudinal (μ_l) dipole moment. This observation clearly signifies that meta-extension of LC core exhibit a dominating suppressed trend of μ_l with a simultaneous increasing trend of μ_t which in turn enhance the occurrence of tilted phases.

Interesting and rare LC phase interface like SmI-SmF transition is witnessed for n equal to 8 and 12. Rare appearance of Iso-SmB_{cry} LC phase transition is also witnessed for n equal to 4 and 11 in the meta-extended core constituted HBLCs.

From the phase diagram (Fig-7), Multi Critical Points (MCP) are noticed, where more than two phase structures with different structural order and symmetries converge. Symmetries with quasi-2D crystalline and pseudo-hexatic ordering in SmI, SmF and SmB_{hex} represent [2] exotic symmetries. Three phases are found to be involved with IAC (for n=6) and IAB_{cryst} (for n=11) MCPs devoid of exotic symmetries. Three phases are also found to be involved with CB_{cryst}F (for n=3), CIF (for n=9) and B_{cryst}IF (for n=11) MCP's involving exotic symmetries. Four different phases are found to be involved with IACB_{cryst}F (for n=4) without exotic symmetries and ACB_{cryst}F (for n=5), CB_{cryst}IF (for n=7), CB_{cryst}FG (for n=10) MCP's with exotic symmetries.

The data for $[\Delta T]_{LC}$, $[\Delta T]_{C}$ and $[\Delta T]_{Tilt}$ estimated for resultant HBLCs are presented in Table-2. A histogram (Figure-8) is drawn for the LC phase abundance in $(4)_{MeO}BD(3)_{Am}BA:nOBA's$. However, for the sake of comparison, reported [9,32] data in nOBA's is also included in the figure.

A detailed analysis for the influence of flexible component (by increasing length (n)of (4)_{MeO}BD(3)_{Am}BA:nOBA) is also carried out (Figure-9) by plotting the relative LC phase stability, viz., through $[\Delta T]_{LC}$, $[\Delta T]_{C}$ and $[\Delta T]_{Tilt}$ with increasing length (n) of end chain. Estimation, presentation and analysis for trends in ΔT_{LC} , ΔT_{tilt} , ΔT_{ortho} and ΔT_{C} provide scope to study the influence of increasing flexibility and meta-extension of LC core. Further, ΔT in present HBLCs is also analyzedin the

Liquid Crystals

wake of the data reported [9,40] for nOBA's. It is observed from the histogram, that pure nOBA's exhibited maximum LC phase stability for n=7 with a specific rigidity to flexibility ratio. With the increasing n, the ΔT_{LC} steeply increases to the maximum value, rather than its gradual trend of decreasing nature with further increase of flexibility for n≥7.

An integrated view of the Fig-8 and -9 reveals that occurrence of maximum $[\Delta T]_{LC}$ is achieved by the involvement of HB, which appears to be slightly pushed towards the higher flexible component (i.e. for n=9) in HBLCs. $[\Delta T_{LC}]$, $[\Delta T]_C$ and $[\Delta T]_{Tilt}$ are found to exhibit O-E effect. O-E effect is observed to be severe for present series of HBLCs regarding ΔT_{LC} and ΔT_{Tilt} . But, ΔT_{LC} for nOBA's and resultant HBLCs seems to follow an analogous trend with n when methylene (-CH₂) unit increment occurs for n equals to 3 to 4; 5 to 6;and 9 to 10 compounds. Commonality is that trends are similarwhen 'n' is increased to even number. However, a reversed trend for ΔT_{LC} is witnessed, when flexible component (n) is increased from even to odd i.e., from n = 10 to 11. ΔT_{LC} is found to exhibit intermediate magnitude of O-E effect in nOBA's. But, a large O-E effect is observed for HBLCs. $[\Delta T]_{Tilt}$ is found to be absent for nSA's, while it is found to be finite for n= 3 onwards (up to 12)

in meta-extended HBLCs of present interest. As both of the series contain complementary type of HB interaction, meta-extended core design feature in latter series is argued to result for increased tilt LC phase abundance.

 $[\Delta T]_{Tilt}$ is found to exhibit an increasing trend for n≥6 for nOBAs. However,

 $[\Delta T]_{\text{Tilt}}$ exhibited a maximum range for n=9 in HBLCs despite the O-E effect. $[\Delta T]_{\text{Tilt}}$ is found to occupy higher values for n=5, 7 and 9 in HBLCs (except for n=3) being the odd numbered intermediate homologues. This observation of higher tilted LC phase stability agrees with the reports [28] of $[\Delta T]_{\text{Tilt}}$ for odd homologues. Odd homologue's contribution for tilted phases is argued due to outward projection of μ_{t} .

It is also argued that the existence of finite and/or enhanced LC phase stability in the HBLCs of pure nOBA's occurred due to meta-extended architecture

(Fig-4) of LC rigid core. Fig-4 reflects the supra-molecular aspects, viz., inclined HB interaction, meta-extended rigid core with underlying bent shape to molecular frame in addition to overall calamitic body/shape. In the present case of HBLCs, $[\Delta T]_{Tilt}$ is found to be higher (except for n≥10) for intermediate homologues than that reported [10] for nOBA's. Thus, meta-extended rigid core in HBLCs is inferred with larger phase stability for device savvy tilted smectic phases.

SmC phase stability $[\Delta T]_{C}$ is found to exhibit (fig-9) almost a monotonical increasing trend with 'n' (from n ≥ 6 onwards) in pure nOBA's. But, $[\Delta T]_{C}$ in metaextended core LCs is found to be accompanied with rigorous O-E effect with increasing 'n'. Hence, the increasing trend of $[\Delta T]_{C}$ behaviour with increase of flexibility seems to be suppressed in meta-extended core LCs. $[\Delta T]_C$ is also observed to bear higher value in higher (for $n \ge 9$) homologues of pure nOBA's. This observation infers the effective role played by orientational disorder in higher homologues, especially, which are not meta-extended core LCs. $[\Delta T]_C$ is also observed to attain higher value for even homologues of meta-extended core LCs. Appearance/existence of non-zero $[\Delta T]_C$ in HBLCs with n=3 and 6 presents a successful attempt by the involvement of meta extended HBLC architecture. Although nOBA's exhibited [10] almost monotonical increase of $[\Delta T]_{C}$ with increasing flexible (for n>6) component, It is remarkable that the odd homologues of metaextended core LCs exhibited lower values for $[\Delta T]_{C}$ in contrast to the general observation that odd homologues in any LC series exhibit enhanced tilted phase stability. This contrast behaviour for $[\Delta T]_C$ is argued due to the superposition of μ_t due to inclined configuration of HB interaction and meta-extended cores in out-ofphase configuration. $[\Delta T]_C$ is found (fig-8) to be almost quenched for 5≤n≥11 in (except n=3) meta-extended core LCs. This observation infers that judicious flexible component is required to realize $[\Delta T]_{C}$ in meta-extended core LCs. $[\Delta T]_{C}$ seems to

Page 17 of 60

Liquid Crystals

be optimized in intermediate even numbered homologues for n = 6, 8 and 10 for meta-extended core LCs.

It is observed [10] that tilted phase or total LC phase range increases effectively with increasing chain length (n) in a series of LC compounds, especially, when chain length (n) crosses a threshold value. However, analogous increasing trends for $[\Delta T]_{LC}$ and $[\Delta T]_{C}$ with 'n' are not observed for meta-extended HBLCs. As meta-extended HBLCs are visualized by a typical hybrid architecture with features (Fig-4) of calamitic and bent cores, trends of $[\Delta T]_{LC}$ and $[\Delta T]_{C}$ with 'n' are argued to be dissimilar with regard to nOBA's.

Integrated enthalpy is estimated from the ΔH values of DSC data (Table-2)as

3f. Integrated Heats of Transitions – Trends of Mesomorphism

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 $[\Sigma \Delta H]_{Heat}$ from heating scans and cooling scans $[\Sigma \Delta H]_{Cool}$; $\Sigma \Delta H$ are plotted (Figure-10) against the increased flexible (n) component in $(4)_{MeO}BD(3)_{Am}BA:nOBA$ HBLC complexes. $[\Sigma \Delta H]$ represents the heat energy transacted by the HBLC with the surroundings during its morphological changes in to the isotropic state. Thus, $[\Sigma \Delta H]_{Heat}$ reflects the heat energy required for the system to completely forego the orderly arrangement. But, $[\Sigma \Delta H]_{Cool}$ reveals the heat energy liberated from the HBLC, when isotropic liquid of HBLC is transformed back into 3D solid. $[\Sigma \Delta H]_{Heat}$ is found to occupy the negative (-ve) value as heat energy is supplied to the system and $[\Sigma \Delta H]_{Cool}$ occupy the positive (+ve) value as heat is evolved from the system during cooling scan. $[\Sigma \Delta H]$ exhibits O-E effect with increasing flexibility (n) component to infer [41,42,49] the relative susceptibility of axial polarizabilities.

It is noticed (fig-10) that $[\Sigma \Delta H]$ exhibits an overall increasing trend with increasing 'n' in both of the heating and cooling scans for meta-extended core LCs. It is also observed that $[\Sigma \Delta H]_{Cool}$ is higher than $[\Sigma \Delta H]_{Heat}$ to infer higher energy transaction in restoring the order by cooling the sample. Differential heat energy consumption, i.e., $\delta(\Sigma H) = [\Delta H_{H} - \Delta H_{C}]$ is estimated for all HBLCs, during the heating followed by cooling scans. Trends of differential enthalpy with 'n' is also presented in Fig-10 for meta-extended core LCs. The observed predominant occupancy in the negative Y-axis differential enthalpy $\delta[\Sigma \Delta H]$ is argued to vouch for the stability and growth of LC phases. More is the magnitude for differential enthalpy; more would be

the stability for expected LC phase appearance. Hence, optimized chain length (n) with largest differential enthalpy is argued as a characteristic value for the LC series under consideration. As such, the preferential order of stability with flexible component 'n' is confirmed for n =3, 7, 9, 10 and 12 in meta-extended LC cores. Hence, it is concluded that odd lower homologue for n=3, intermediate homologue for n =7 and higher homologues for n =9,10 and 12 exhibit large device savvy LC phase stability due to the presence of meta-extended LC core.

In order to investigate the influence of meta-extended core on optimized growth (or stability) of LC phases, a comparative study of integrated and differential enthalpies is carried out (figures-11 and -12) for the meta-extended core LCs (table-2, which bear complementary [71,72] HB interaction with nOBA's and Schiff base containing [25] linear LC hydrogen bonded (4)PyBD(4^I)BrA-nOBA liquid crystal series. It is noticed that integrated enthalpies (fig-11) in both of the heating and cooling scans for complementary HB meta-extended or linear HB Schiff based LCs fall intermediate between the higher values of heating scan and lower values of cooling scan recorded for nOBA's. This observation speaks out the relatively comfortable growth of LC phase structures by exchange of lesser heat energy in meta-extended CL cores. Hence, involving either linear HB interaction or metaextended core are argued to effectively stabilize the growth of LC phases. O-E effect for integrated enthalpy in the heating scan of nOBA's appears more rigorous than that observed for the heating scans of other two series of LCs. Apart from the O-E effect, the differential enthalpy (fig-12) is observed to accompany with relatively lesser values in nOBA's. Lesser values of $\delta(\Sigma H)$ for nOBA's infer involvement of lesser heat exchange in a cyclic scan of heating followed by cooling. Since, $\delta(\Sigma H)$ of meta-extended core LCs is more is least followed by linear HB Schiff based LCs, meta-extended core LC molecular design is argued to be a preferred direction of synthesis to realize LC phase structures of device savvy nature.

4 Conclusions

¹H-NMR and ¹³C-NMR spectral studies could be effective tools to confirm the presence of number and placement of H-atoms and C-atoms respectively, in the targeted LC molecule. IR-spectroscopic study can confirm the presence of hydrogen bonding interaction and its type in HBLC complexes. Preparation followed by phase characterization studies in $(4)_{MeO}BD(3)_{Am}BA:nOBAs$ with meta-extended hybrid molecular architecture infer that presence of Schiff based moiety, meta-directed

Liquid Crystals

 extension of rigid core and complementary type of HB interaction prevails over the occurrence of tilted smectic polymorphism. Involvement of electro-negative O-atom (as bridging atom/group) promotes lateral stacking of molecules and enhanced smectic phase stability. Meta-extended rigid core in calamitic type of LCs witnesses quenching of Nematic phase due to the enhanced transverse dipole moment. Increase of flexible component in calamitic LC molecules induce orientational disorder to result for SmF and SmI like quasi-2D LC phases with exotic symmetries. Meta-extended rigid core results for ideal expectation of unique second order nature for A-C transition due to the effective contributions of transverse dipole moment. Meta extended rigid core of calamitic HBLC series with tunable (for n=3 to 12) flexible component could realize characteristic (i.e., for n=6) IAC Multi Critical Point (MCP) where exotic symmetries converge.

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CAPTIONS Synthetic route for the preparation of (4)_{MeO}BD(3)_{Am}BA:nOBA. Scheme-1: FIGURES Figure-1: Diversity of liquid crystal molecular architecture. H¹ NMR spectra of (4)_{MeO}BD(3)_{Amn}BA:5OBA. Figure-2 : C¹³ NMR spectra of (4)_{MeO}BD(3)_{Amn}BA:5OBA Figure-3: Figure-4 : Hybrid molecular structure for (4)_{MeO}BD(3)_{Am}BA:nOBA's with calamitic and bent features. Figure-5: FTIR spectra of $(4)_{MeO}BD(3)_{Am}BA:5OBA$ and its pristine compounds. Figure-6: DSC thermogram of (4)_{MeO}BD(3)_{Amn}BA:7OBA. Figure-7: Phase diagram for (4)_{MeO}BD(3)_{Amn}BA:nOBAs Figure-8: LC Phase stability for (4)_{MeO}BD(3)_{Am}BA:nOBAs. Figure-9: Trends of Phase stability Δ Twith flexible component n. Figure-10: Integrated heats ($\Sigma \Delta H_H$ and $\Sigma \Delta H_C$) and Differential heat $\delta \Sigma \Delta H$ for (4)_{MeO}BD(3)_{Am}BA:nOBAs. Comparison of integrated enthalpy of (4)_{MeO}BD(3)_{Am}BA:nOBA's with Figure-11: Schiff based (4)PyBD(4¹)BrA-nOBA's and nOBA's. Comparison of differential enthalpy of (4)_{MeO}BD(3)_{Am}BA:nOBA's with Figure-12: Schiff based (4)PyBD(4¹)BrA-nOBA's and nOBA's. TABLES Table-1: Data of IR absorption for $(4)_{MeO}BD(3)_{Am}BA:nOBAs$. Table-2: Data of LC phases, T_{C} and ΔH from POM and DSC for (4)_{MeO}BD(3)_{Am}BA:nOBAs. PLATES Plate-1: SmA focal conic fan texture of (4)_{MeO}BD(3)_{Amn}BA:110BA at 119.3°C(heating). Plate-2: SmA elongated lathe-like extended-board texture of (4)_{MeO}BD(3)_{Amn}BA:7OBA at 115.6°C (cooling). Plate-3: SmA spiked fan texture of $(4)_{MeO}BD(3)_{Amn}BA:10OBA$ at $90.9^{\circ}C$ (heating). Plate-4: SmC four brushed schleiren texture of (4)_{MeO}BD(3)_{Amn}BA:8OBA at109.8°C(heating).

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2 3 4 5	Plate-5:	SmC black-on-white natural mosaic texture of (4) _{MeO} BD(3) _{Amn} BA:6OBA at 107.2°C(heating).
6 7 8 9	Plate-6:	SmB _{cry} colored mosaic texture of $(4)_{MeO}BD(3)_{Amn}BA:11OBA$ at 105.7°C (heating).
9 10 11 12	Plate-7:	SmB _{cry} glossy focal conic fan batonnet texture of (4) _{MeO} BD(3) _{Amn} BA:6OBA at 90.2°C (cooling).
13 14 15	Plate-8:	SmI broken focal conic plume texture of (4) _{MeO} BD(3) _{Amn} BA:8OBA at 95.8°C(cooling).
16 17 18 19	Plate-9:	SmI colored threaded texture of (4) _{MeO} BD(3) _{Amn} BA:12OBA at 102.7 ^o C (cooling).
20 21 22	Plate-10:	SmF colored marble threaded texture of $(4)_{MeO}BD(3)_{Amn}BA:9OBA$ at 98.3°C(heating).
23 24 25 26	Plate-11:	SmF checkered-board texture with elongated fans of $(4)_{MeO}BD(3)_{Amn}BA:12OBA$ at 87.8°C(cooling).
27 28 29	Plate-12:	SmF paramorphotic coloured mosaic texture with undulated boundaries of (4) _{MeO} BD(3) _{Amn} BA:11OBA at 103.3 ^o C(cooling).
30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59	Plate-13:	SmG coloured mosaic texture of (4) _{MeO} BD(3) _{Amn} BA:8OBA at 91.5°C(cooling).

Appendix-A

NMR spectra is analyzed [61,62] to determine the presence of H-atoms on the body of HBLC.Data for observed chemical shifts (δ_H) for all members of (4)_{MeOBD}(3)_{Am}BA:nOBA series of HBLCs (for n=3,4,5,6,7,8,9,10,11 and 12) is:

1H NMR Study

(4)_{MeO}**BD(3)**_{Amn}**BA:3OBA** is observed with chemical shift (δ_H); 1.05 (3H, t, *J* =7.5Hz,-CH₃), 1.78-1.90 (2H, m,-OCH₂CH₂CH₂-), 3.88 (3H, s, -O-CH₃), 3.97-4.01 (2H, t, J = 6.6 Hz, -OCH₂-CH₂-), 6.92-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.89 (1H, s, -COOH).

(4)_{MeO}**BD(3)**_{Amn}**BA:4OBA** is observed with chemical shift (δ_H);0.99 (3H, t, *J* =6.9Hz,-CH₃), 1.44-1.57 (2H, m,-[CH₂]₁-), 1.75-1.82 (2H, m, -OCH₂CH₂CH₂-), 3.88 (3H, s, -O-CH₃), 4.01-4.05 (2H, t, J = 6.6 Hz, -OCH₂-CH₂-), 6.92-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.89 (1H, s, -COOH).

(4)_{MeO}**BD(3)**_{Amn}**BA:5OBA** is observed with chemical shift (δ_H); 0.91 (3H, t, *J* =6.9Hz,-CH₃), 1.33-1.50 (4H, m,-[CH₂]₂-), 1.77-1.86 (2H, m,-OCH₂CH₂CH₂-), 3.88 (3H, s, -O-CH₃), 4.00-4.04 (2H, t, J = 6.6 Hz, -OCH₂-CH₂-), 6.92-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.89 (1H, s, -COOH).

(4)_{MeO}**BD(3)**_{Amn}**BA:6OBA** is observed with chemical shift (δ_H); 0.91 (3H, t, *J* =6.9Hz,-CH₃), 1.32-1.49 (6H, m,-[CH₂]₃-), 1.76-1.85 (2H, m,-OCH₂CH₂CH₂-), 3.88 (3H, s, -O-CH₃), 4.00-4.04 (2H, t, J = 6.6 Hz, -OCH₂-CH₂-), 6.92-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.89 (1H, s, -COOH).

(4)_{Meo}**BD(3)**_{Amn}**BA:70BA** is observed with chemical shift (δ_H); 0.90 (3H, t, *J* =6.9Hz,-CH₃), 1.31-1.49 (8H, m,-[CH₂]₄-), 1.76-1.85 (2H, m,-OCH₂CH₂CH₂-), 3.89 (3H, s, -O-CH₃), 4.00-4.04 (2H, t, J = 6.6 Hz, -OCH₂-CH₂-), 6.92-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.89 (1H, s, -COOH).

(4)_{Meo}**BD(3)**_{Amn}**BA:80BA** is observed with chemical shift (δ_H); 0.89 (3H, t, *J* =6.9Hz,-CH₃), 1.29-1.47 (10H, m,-[CH₂]₅-), 1.76-1.85 (2H, m,-OCH₂CH₂CH₂-), 3.89 (3H, s, -O-CH₃), 4.00-4.04 (2H, t, J = 6.6 Hz, -OCH₂-CH₂-), 6.92-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.89 (1H, s, -COOH).

(4)_{MeO}**BD(3)**_{Amn}**BA:9OBA** is observed with chemical shift (δ_H); 0.88 (3H, t, *J* =6.9Hz,-CH₃), 1.28-1.48 (12H, m,-[CH₂]₆-), 1.76-1.83 (2H, m,-OCH₂CH₂CH₂-), 3.89 (3H, s, -O-CH₃), 4.00-4.04 (2H, t, J = 6.6 Hz, -OCH₂-CH₂-), 6.92-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.89 (1H, s, -COOH).

(4)_{MeO}**BD(3)**_{Amn}**BA:10OBA** is observed with chemical shift (δ_H); 0.88 (3H, t, *J* = 6.9Hz,-CH₃), 1.28-1.46 (14H, m,-[CH₂]₇-), 1.76-1.85 (2H, m,-OCH₂CH₂CH₂-), 3.88 (3H, s, -O-CH₃), 3.99-4.04 (2H, t, *J*= 6.6 Hz, -OCH₂-CH₂-), 6.91-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.89 (1H, s, -COOH).

(4)_{MeO}**BD(3)**_{Amn}**BA:110BA** is observed with chemical shift (δ_H); 0.88 (3H, t, *J* = 6.9Hz,-CH₃), 1.27-1.46 (16H, m,-[CH₂]₈-), 1.76-1.85 (2H, m,-OCH₂CH₂CH₂-), 3.89 (3H, s, -O-CH₃), 4.00-4.02 (2H, t, J = 6.6 Hz, -OCH₂-CH₂-), 6.91-8.07 (12H, m, Ar-H), 8.43 (1H, s, -CH=N), 9.90 (1H, s, -COOH).

(4)_{MeO}**BD(3)**_{Amn}**BA:12OBA** is observed with chemical shift (δ_H); 0.88(3H, t, *J*=6.9Hz,-CH₃), 1.27-1.46 (18H, m,-[CH₂]₉-), 1.76-1.85 (2H, m,-OCH₂CH₂CH₂-), 3.88 (3H, s, -O-CH₃), 3.99-4.02(2H, t, *J*=6.6 Hz,-O-CH₂-CH₂-), 6.91-8.07 (12H, m,Ar-H), 8.43 (1H, s,-CH=N), 9.89(1H, s,-COOH).

¹H NMR study for (4)_{Meo}BD(3)_{Am}BA:50BA HBLC (Fig-2) as Representative:

As per the molecular formula for n=5 in Scheme-1, the number of H-atoms expected to be on the molecular body of $(4)_{MeO}BD(3)_{Am}BA$:50BA is 29.

Salient features of the observed ¹H NMR spectra are;

- a. A triplet at δ 0.91ppm infers the presence of 3-H atoms relevant to end methyl group of 5OBA moiety.
- b. A multiplet at δ 1.33-1.86ppm infers the presence of 6-H atoms relevant to three methylene (-CH₂-)units of 5OBA.
- c. A singlet in the range of δ 3.88ppm infers the presence of 3-H atoms pertaining to methoxy(-OCH₃) group connected to Schiff base intermediate.
- d. A triplet at δ 4.00–4.04ppm infers the presence of 2-H atoms pertaining -OCH₂ group of 50BA.
- e. A multiplet in the range of at δ 6.92–8.07ppm infers the presence of 12-H atoms corresponding to three aromatic rings (Ar–H) of HBLC complex.
- f. A singlet at δ 8.43ppm infers the presence of 1-H atom pertaining to the Schiff base CH=N moiety.
- g. Asignal (by integrated envelope) in the range of δ 10.2-11.6ppm infers the presence of 2-H atoms pertaining to the two acid (-COOH)moieties that participate in HB.

An overview of ¹H NMR spectra (points a-g) infers the presence of TENTY NINE Hatoms on the molecular body of the HBLC. Since observed number of H-atoms is

found to agree with expected number of H-atoms, the targeted synthesis is argued to be successful. The HBLC product is argued to be pure to spectroscopic level.

It is also noticed (Fig-2) that chemical shift relevant to the H-atom pertaining to hydroxyl (-OH) moieties (i.e., Schiff based intermediate and nOBA) of acid groups (– COOH) is found (δ ~10-12) to vouch [61,62] for the de-shielding effect due to their vicinity to the HB interaction.

¹³C NMR Study

Appendix-B

The presence of C-atoms on the body of HBLC complexes is estimated [61,62] from the¹³C-NMR spectra. Data for chemical shift are presented here for all the ten resulting HBLCs (n=3-12).

(4)_{MeO}BD(3)_{Amn}BA:3OBA

 δ_{c} :10.42, 22.41, 55.42, 55.55, 69.72, 114.18, 114.26, 116.27, 120.27, 120.45, 121.36, 121.86, 127.00, 127.20, 128.87, 129.23, 129.35, 130.46, 130.77, 132.00, 132.33, 152.51, 160.86, 162.55, 163.70, 172.04, 190.90.

(4)_{MeO}BD(3)_{Amn}BA:4OBA

 $\delta_{\rm C}$:13.78, 19.16, 31.10, 55.42, 55.55, 67.95, 114.17, 114.26, 116.27, 120.27, 120.45, 121.34, 121.86, 127.01, 127.20, 128.86, 129.23, 129.34, 130.45, 130.77, 132.00, 132.32, 152.51, 160.86, 162.54, 163.71, 172.05, 190.89.

(4)_{MeO}BD(3)_{Amn}BA:50BA

 $\delta_{\rm C}$:13.97, 22.40,28.17, 28.76, 55.42, 55.55, 68.27, 114.19, 114.27, 116.28, 120.28, 120.46, 121.35, 121.87,127.01,127.21, 128.88, 129.24,129.35, 130.46, 130.78, 132.01, 132.33, 146.42, 152.52, 160.86, 162.56, 163.71, 172.05, 172.36, 190.89.

(4)_{MeO}BD(3)_{Amn}BA:60BA

 $\delta_{\rm C}$:13.98, 22.54, 25.62, 29.02, 31.50, 55.40, 55.53, 68.27, 114.17, 114.25, 116.27, 120.27, 120.45, 121.34, 121.86, 127.00, 127.20, 128.86, 129.22, 129.33, 130.46, 130.77, 132.00, 132.32, 146.40, 152.49, 160.86, 162.54, 163.70, 172.08, 172.39, 190.89.

(4)_{MeO}BD(3)_{Amn}BA:70BA

 δ_{C} :14.05, 22.58, 25.92, 29.00, 29.07, 31.73, 55.42, 55.55, 68.28,114.19, 114.26, 116.27,120.27,120.46, 121.35, 121.87, 127.01, 127.20, 128.88, 129.24, 129.35, 130.37, 130.46, 130.78, 132.01, 132.33, 146.42, 152.52, 160.86, 162.55, 163.71, 172.05, 190.89.

(4)_{MeO}BD(3)_{Amn}BA:80BA

 δ_{C} :14.08, 22.64, 25.98, 29.09, 29.20, 29.31, 31.79, 55.45, 55.58, 69.31, 114.21, 114.28, 116.24, 120.23, 120.44, 121.33, 121.87, 127.00, 127.19, 128.93, 129.25, 129.38, 130.34, 130.43, 130.77, 132.01, 132.34, 146.47, 152.59, 160.82, 162.56, 163.72, 171.78, 172.07, 190.86.

(4)_{MeO}BD(3)_{Amn}BA:9OBA

 $\delta_{\rm C}$:14.05, 22.62, 25.93, 29.05, 29.20, 29.32, 29.47, 31.83, 55.40, 55.53, 68.26, 114.17, 114.24, 116.25, 120.24, 120.44, 121.33, 121.85, 126.99, 127.18, 128.87, 129.21, 129.33, 130.44, 130.75, 131.99, 132.31, 146.40, 152.50, 160.83, 162.53, 163.69, 164.61, 172.02, 172.33, 190.86.

(4)_{MeO}BD(3)_{Amn}BA:10OBA

 δ_{C} :14.07, 22.64, 25.94, 29.06, 29.28, 29.32, 29.51, 31.86, 55.40, 55.53, 68.27, 114.17, 114.24, 116.27, 120.23, 120.44, 121.35, 121.84, 126.98, 127.19, 128.87, 129.21, 129.32, 130.50, 130.76, 131.99, 132.31, 146.39, 152.49, 160.82, 162.53, 163.69, 172.02, 172.34, 190.86.

(4)_{MeO}BD(3)_{Amn}BA:110BA

 δ_{c} :14.08, 22.65, 25.94, 29.06, 29.31, 29.56, 31.88, 55.42, 55.55, 68.28, 114.18, 114.25, 116.24, 120.24, 120.44, 121.31, 121.86, 127.00, 127.19, 128.89, 129.23, 129.35, 130.42, 130.76, 131.99, 132.33, 146.42, 152.54, 160.83, 162.54, 163.71, 171.95, 172.26, 190.86.

(4)_{MeO}BD(3)_{Amn}BA:12OBA

 $\delta_{\rm C}$:14.07, 22.65, 25.94, 29.05, 29.31, 29.54, 29.60, 31.88, 55.40, 55.53, 68.26, 114.16, 114.24, 116.25, 120.21, 120.43, 121.37, 121.83, 126.95, 127.18, 128.88, 129.20, 129.32, 130.41, 130.54, 130.75, 131.99, 132.30, 146.40, 152.50, 160.80, 162.52, 163.67, 171.96, 172.28, 190.86.

The number of C-atoms in of $(4)_{MeO}BD(3)_{Am}BA:nOBA$ (for n=3 to 12) is estimated from the observed ¹³C NMR chemical shifts (δ_C) .¹³C-NMR spectra for $(4)_{MeO}BD(3)_{Amn}BA:5OBA$ is presented in Figure-3 (representative). As per the proposed (Scheme-1) molecular formula, $(4)_{MeO}BD(3)_{Amn}BA:5OBA$ is expected to possess TWENTY SEVEN C-atoms on its molecular body.

Salient features of the ¹³C NMR spectrum are;

i)A peak at δ 3.97 ppm infers the presence one C-atom of end methyl carbon of 50BA.

Liquid Crystals

ii)Peaks in the range of δ 22.4–28.8ppm infer the presence of three C-atoms relevant to the three methylene (-CH₂-) groups of 5OBA.

iii)Peak at $\delta 55.9ppm$ infers the presence of one C-atom corresponding to the methoxy (-OCH_3) group of schiff base moiety connected at para position of aromatic ring .

iv)Peak at δ 68.27ppm infers the presence of one C-atom adjacent to O-atom of 50BA moiety.

v)Peaks in the range of δ 114.19 - 152.52ppm infer the presence of eighteen C-atoms of three aromatic rings in HBLC.

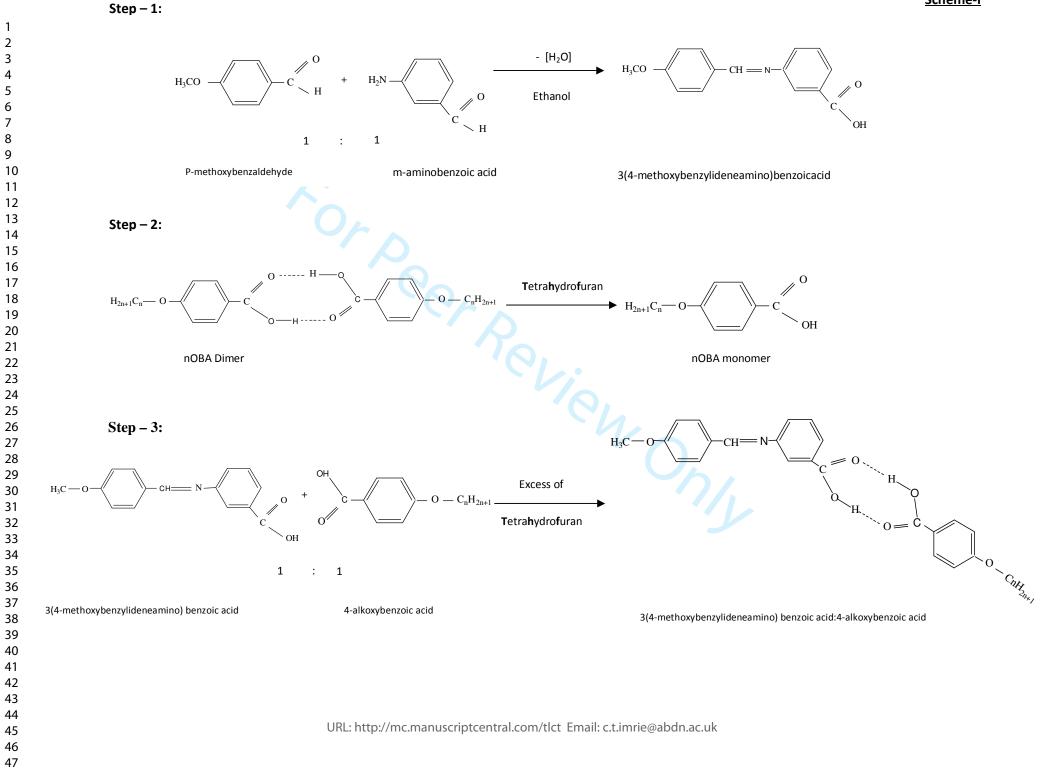
vi)Peak at δ 160.86 ppm infers the presence of one carbon atom in the Schiff base (CH=N-) moiety.

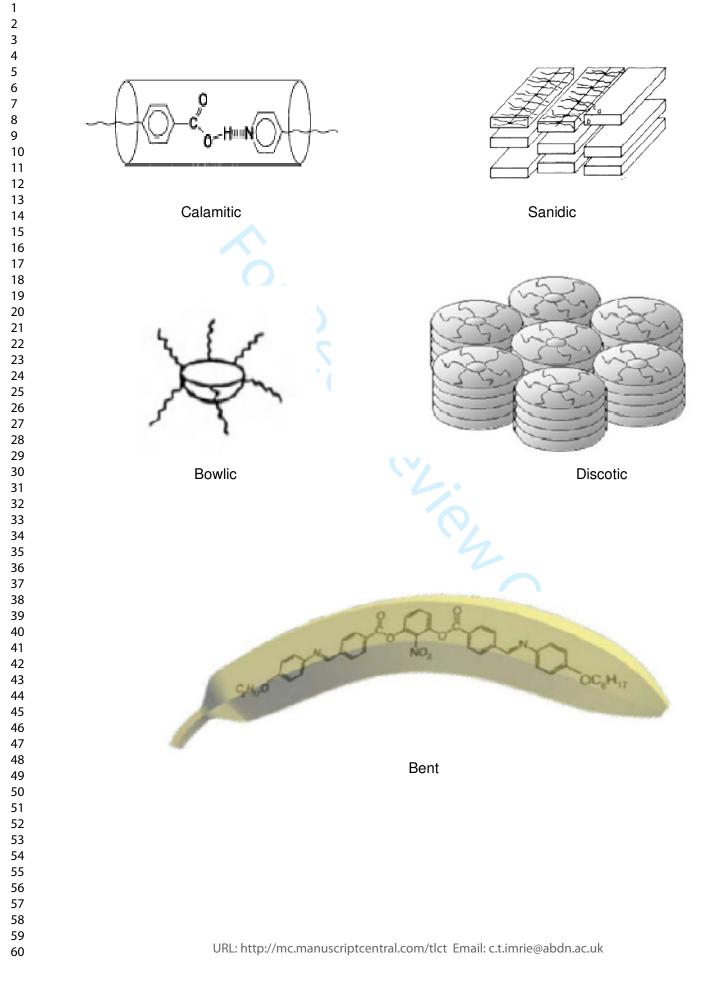
vii)Peaks in range of δ 172.05 - 172.36ppm infer the presence of two carbon atoms pertaining to the two acid (–COOH) moieties.

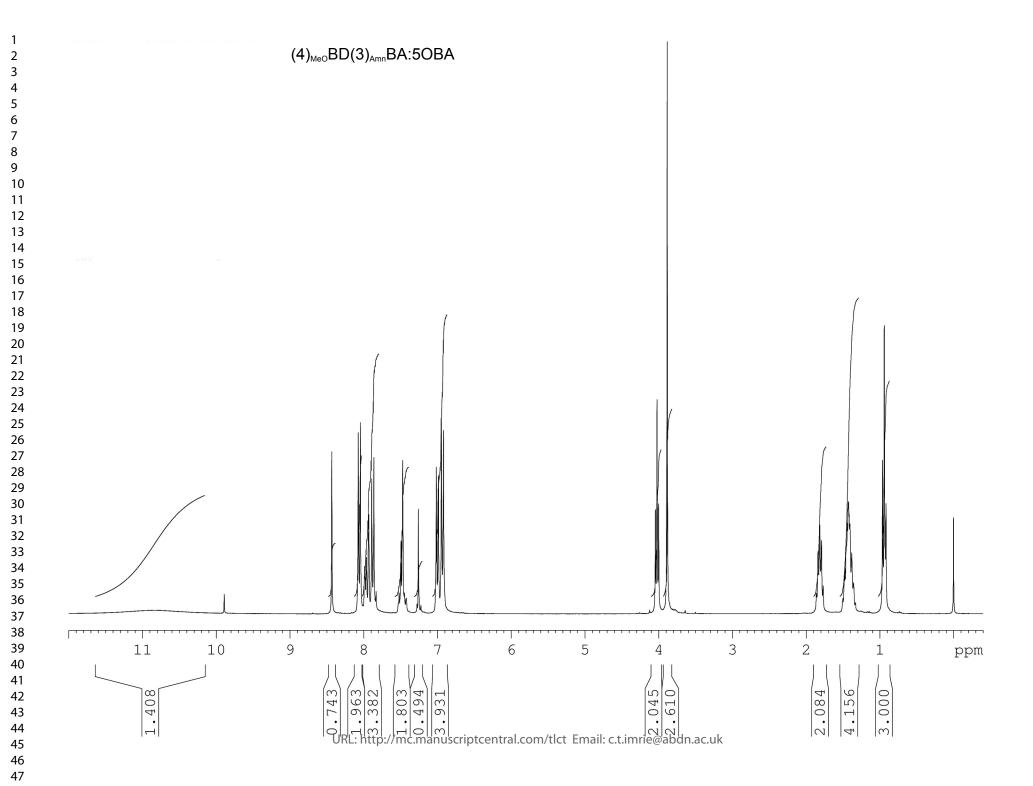
An overview of the NMR (from i to vii) spectrum infers that TWENTY SEVEN Catoms are present on the body of $(4)_{MeO}BD(3)_{Amn}BA:5OBA$. Since, observed number of C-atoms agrees with the expected, the targeted synthesis of HBLC complex is argued to be successful. Further, HBLC is also argued to be pure to spectroscopic level.

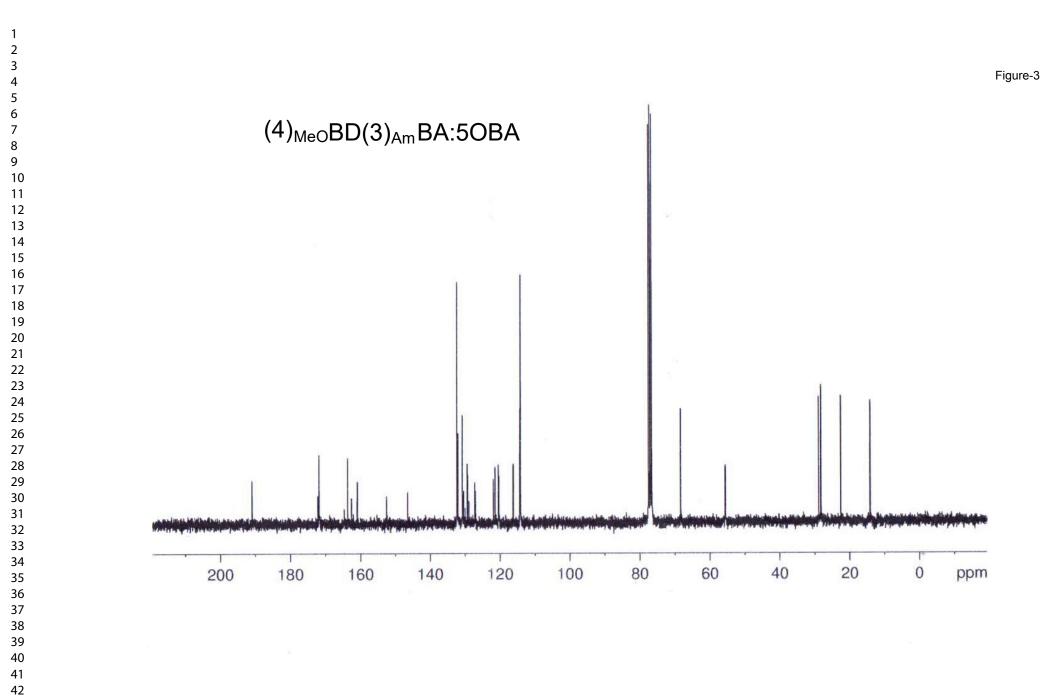
It is also noticed (Fig-3) that the NMR signals (for C-atom) pertaining to the parapositioned (to nOBA) and meta positioned (Schiff based intermediate) acid (– COOH)moieties are found to be witnessed (δ 10.2 – 11.6) to imply [61,62] a shift due to their vicinity to HB interaction.

Scheme-I



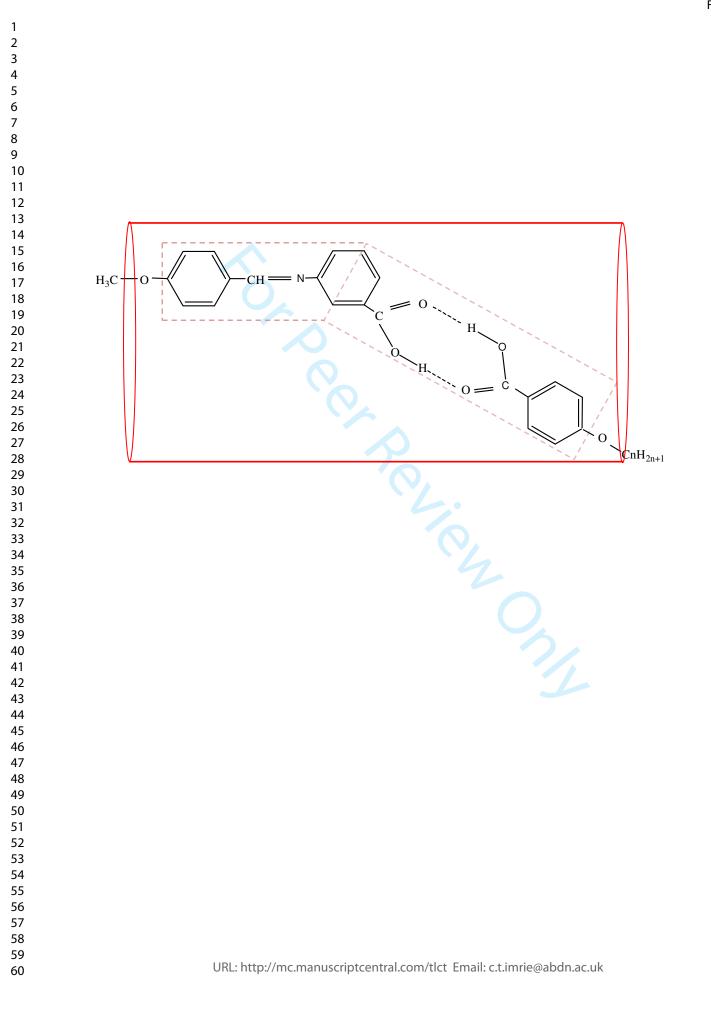


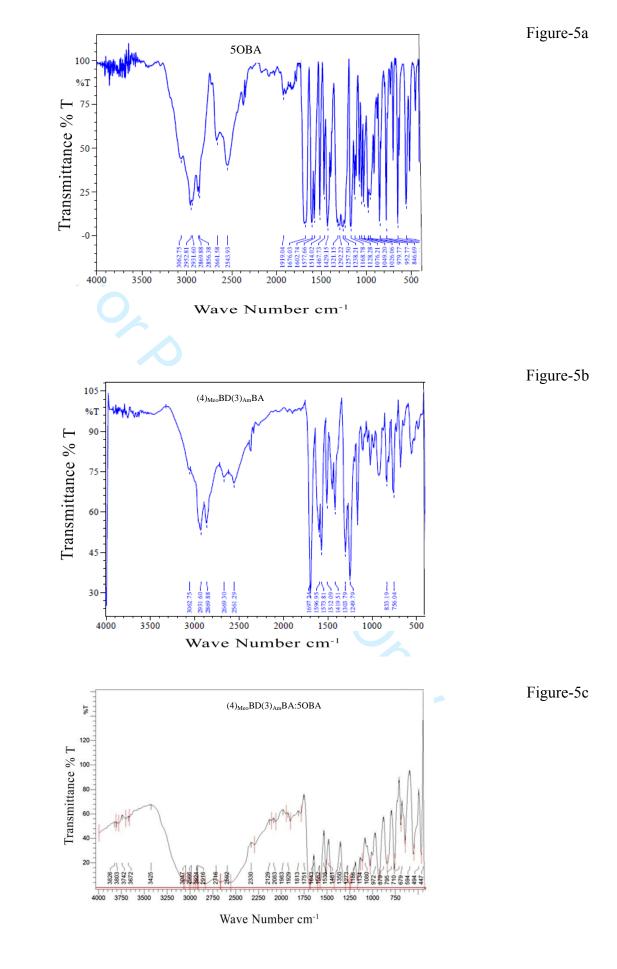




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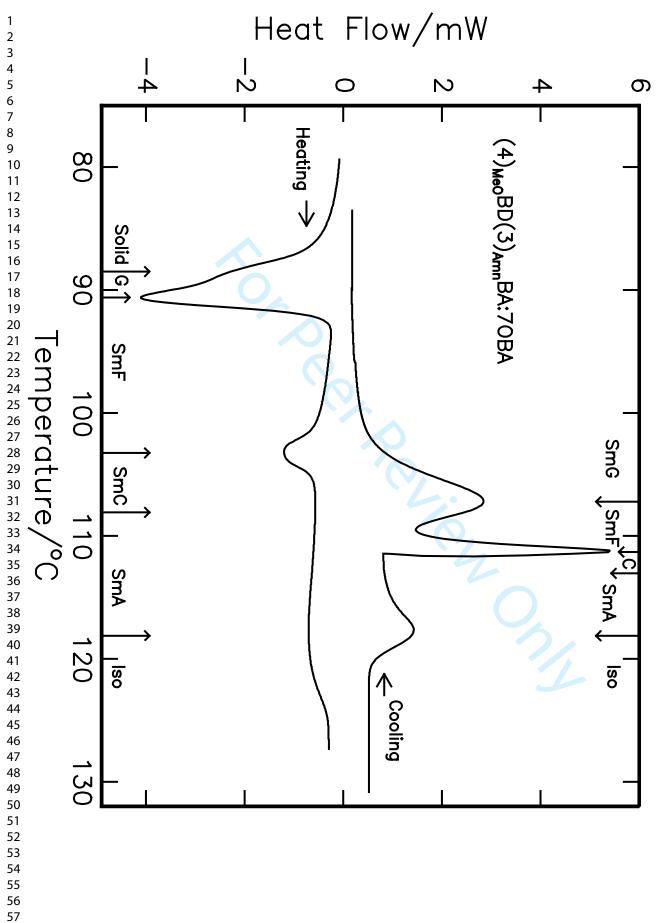
43

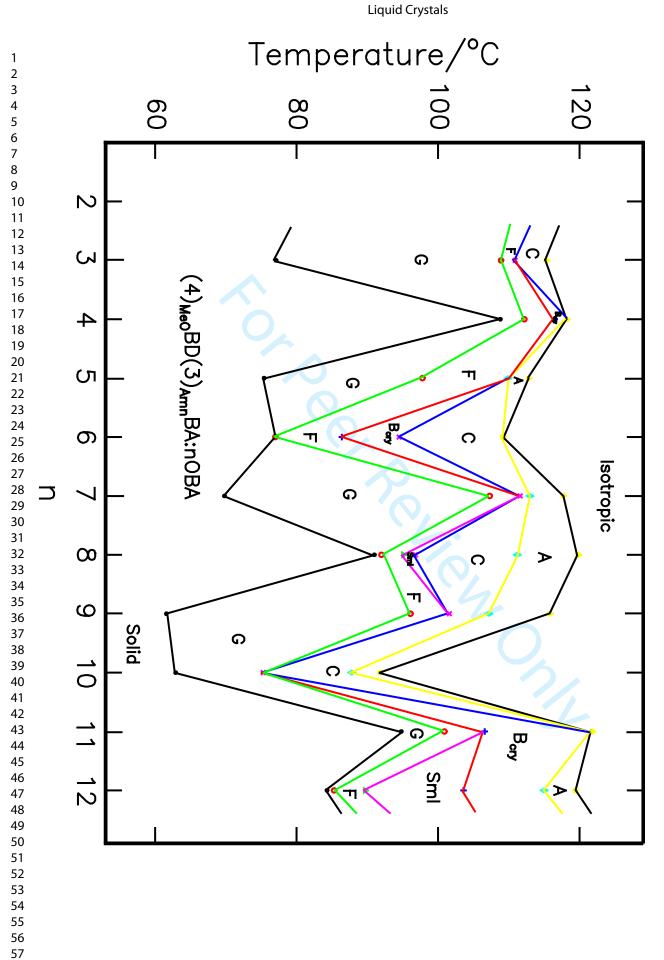


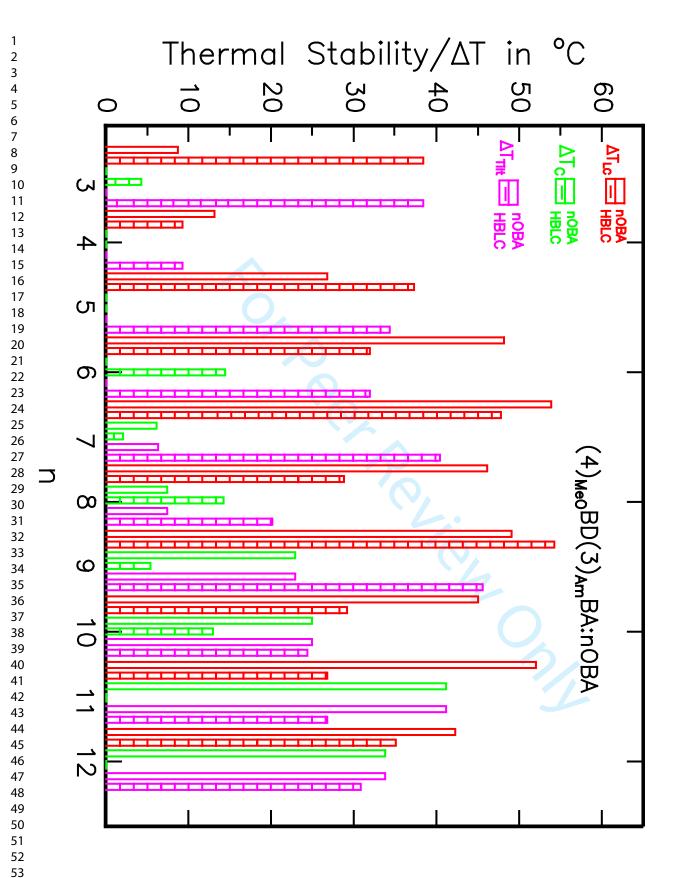


60

Liquid Crystals







Liquid Crystals

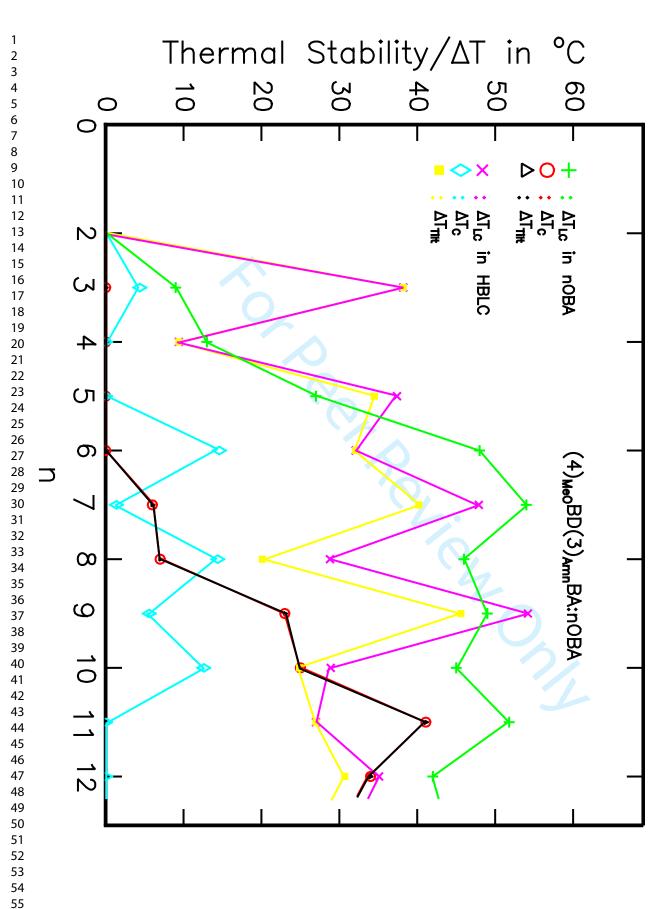
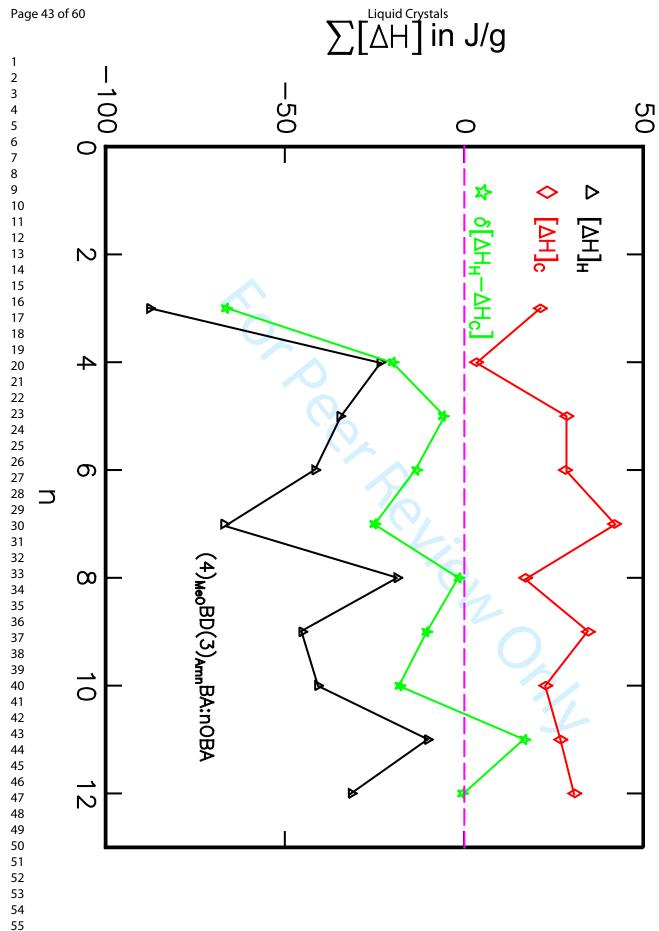
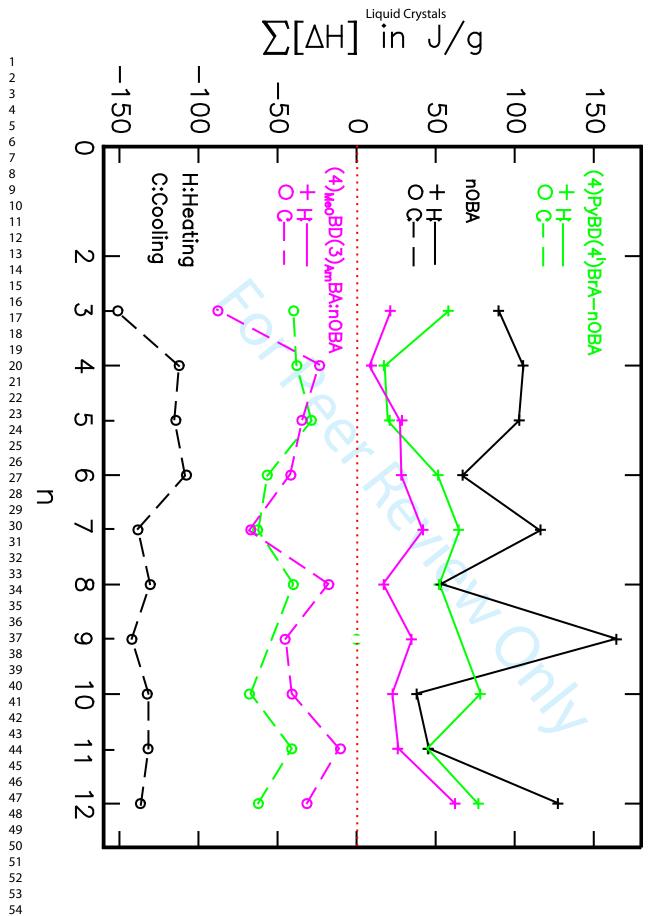
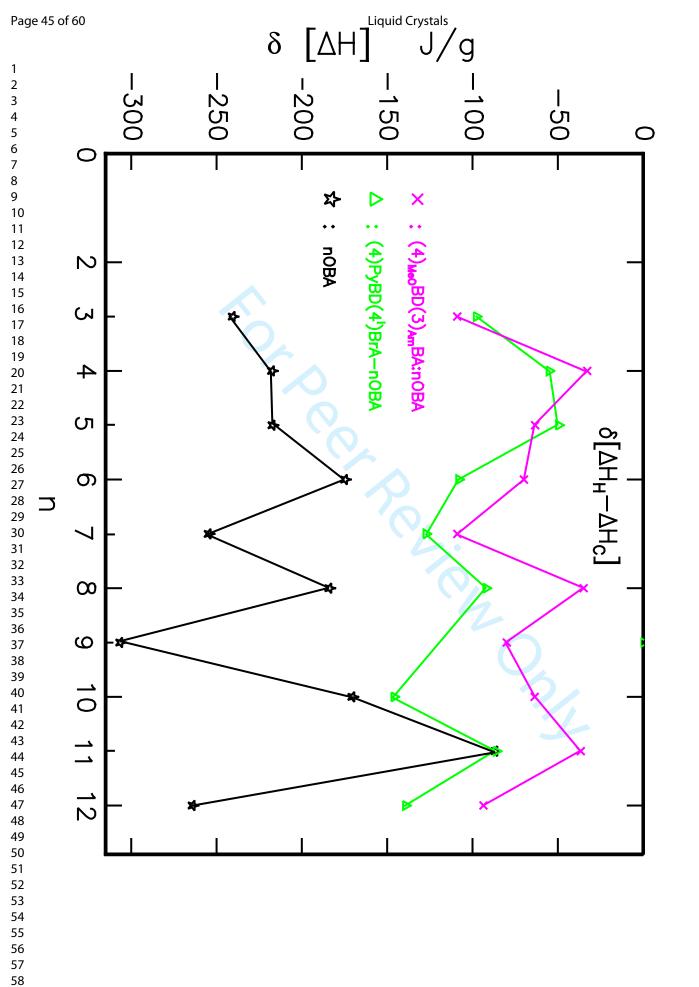


Figure-9





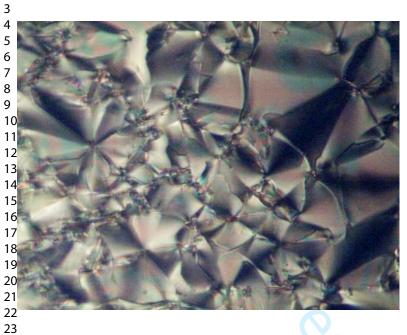




Liquid Crystals

Plate 1

Plate 2



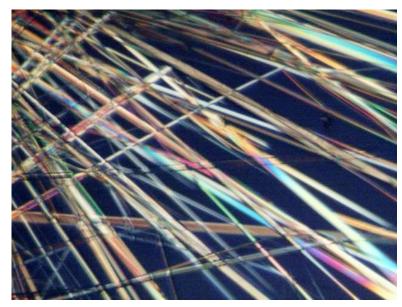


Plate 3

Plate 4

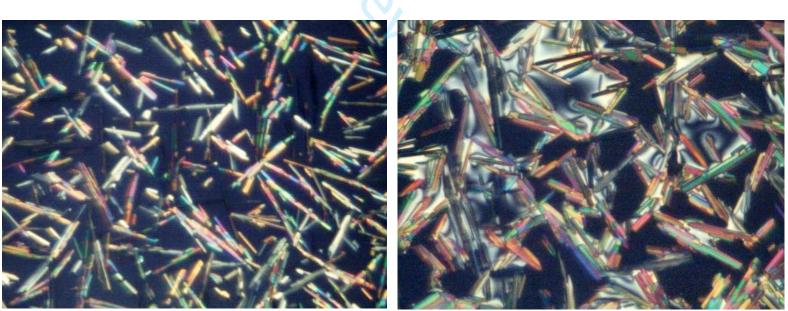


Plate 5

Plate 6

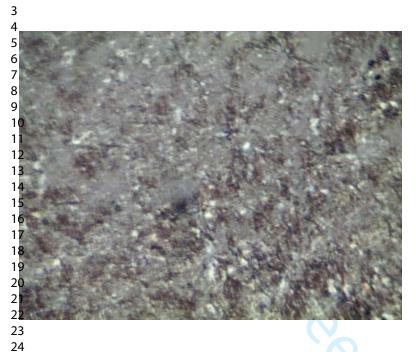




Plate 7



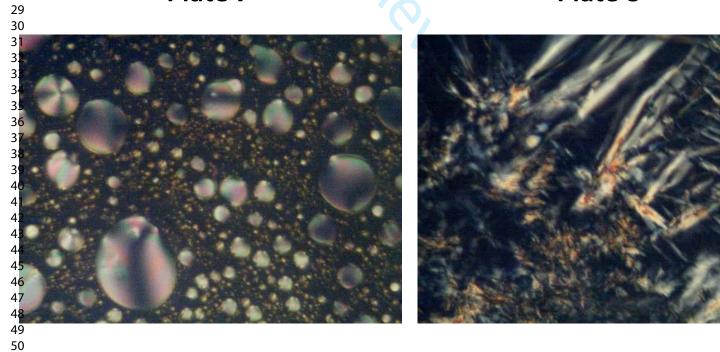
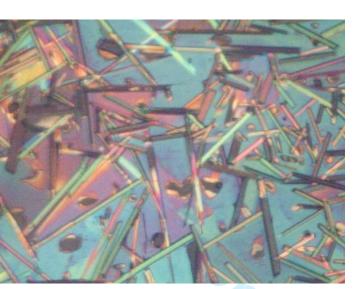


Plate 9



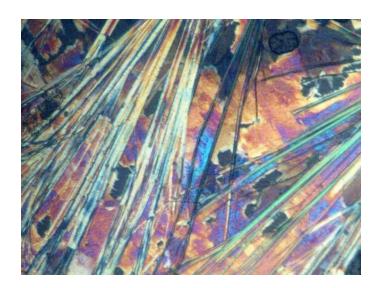


Plate 11

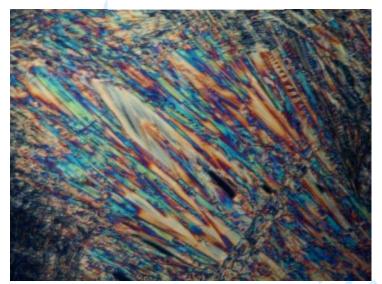
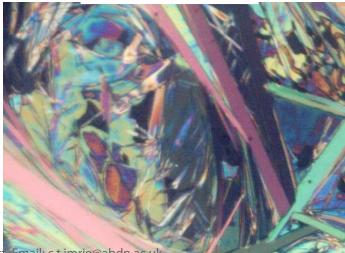


Plate 12



Plate 13



Page 49 of 60		Liquid Crysta	lls		
1 2 3 4		TABLE –	1		
5 6 7		I			
, 8 9	HBLC compound (4) _{MeO} BD(3) _{Am} BA:nOBA	-OH in	R Absorption / o N atom in	Carbonyl	
10 11	for n=	nOBA	-CH=N-	>C=O in nOBA	
12 13	3	2966	1603	1682	
14 15 16	4	2955	1603	1683	
17 18 19	5	2955	1603	1681	
20 21 22	6	2935	1604	1689	
23 24 25	7	2934	1605	1686	
26 27 28	8	2928	1606	1688	
29 30 31	9	2920	1604	1685	
32 33 34	10	2916	1604	1685	
35 36 37	11	2919	1603	1682	
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Table-2

HBLC (4) _{MeO} BD(3) _{Amn} BA:nOBA	MethodPhase VarianceDetails of Transition Temperatures (Tc) in °C and Enthalpy in J/g		[ΔT] _{∟c} (in °C)	[∆T] _c (in °C)	[ΔT] _{⊺∟⊤} (in °C)	[ΔT] _{or} (in °C)		
	T 14	Heating	G	(RT)SmG–(115.4)→lso				
3	ТМ	Cooling	GFC	Sld←(77.0)–SmG←(108.8)–SmF←(110.9)–SmC←(115.4)–Iso				
	DSC	Heating	G	(RT) SmG–(115.27)→ Iso [87.65]				
		Cooling	GFC	$\label{eq:side-(77.08)-SmG} \begin{split} & \text{Sld}\leftarrow(170.08)-\text{SmG}\leftarrow(110.89)-\text{SmC}\leftarrow(115.27)-\text{Iso}\\ & \textbf{[18.14]} \textbf{[2.18]} \textbf{[0.04]} \textbf{[0.94]} \end{split}$	38.19	4.38	38.19	
	ТМ	Heating	GFB _{cry}	SId-(91.4) \rightarrow SmG-(97.4) \rightarrow SmF-(113.4) \rightarrow SmB _{cry} -(118.2) \rightarrow Iso				
	I IVI	Cooling	GFB _{cry}	Sld₁(108.9)SmG(112.2)SmF(116.5)SmB _{cry} (118.2)Iso				
4		Heating	GFB _{cry}	$SId-(91.48) \rightarrow SmG-(97.40) \rightarrow SmF-(113.40) \rightarrow SmB_{cry} -(118.22) \rightarrow Iso$ [1.19] [0.53] [15.86] [5.86]				
DSC	DSC	Cooling	GFB _{cry}	$\label{eq:states} \begin{array}{c} \text{Sld}_1 \leftarrow (108.79) & -\text{SmG} \leftarrow (112.2) - \text{SmF} \leftarrow (116.52) - \text{SmB}_{\text{cry}} \leftarrow (118.22) - \text{lso} \\ \hline [1.51] & [0.12] & [1.64] & [0.26] \end{array}$	9.43		9.43	
	ТМ	Heating	GFA	$SId-(86.0) \rightarrow SmG-(92.5) \rightarrow SmF-(104.2) \rightarrow SmA-(112.8) \rightarrow Iso$				
	1 111	Cooling	GFA	Sld*←(75.4)-SmG←(103.6)-SmF←(109.8)-SmA←(112.8)-Iso				
5	DSC	Heating	GFA	SId–(85.71)→SmG–(91.78)→SmF–(104.28)→SmA–(112.81)→Iso [0.63] [0.57] [31.69] [1.71]				
	200	Cooling	GFA	Sld*←(75.4)-SmG←(97.85)-SmF←(109.87)-SmA←(112.81)-Iso [27.21] [1.1] [0.41]	37.41		34.47	2.94
	тм	$\begin{array}{ c c c c } \hline Heating & GFB_{cry}C & Sld^{*}-(72.1)\rightarrow SmG-(91.8)\rightarrow SmF-(99.7)\rightarrow SmB_{cry}-(103.8)\rightarrow SmC-(109.1)\rightarrow Iso \\ \hline \end{array}$		$Sld^{\star}-(72.1)\rightarrowSmG-(91.8)\rightarrowSmF-(99.7)\rightarrowSmB_{cry}-(103.8)\rightarrowSmC-(109.1)\rightarrowIso$				
I IVI	1 111	Cooling	FB _{cry} C	Sld*←(77.0)–SmF←(86.4)–SmB _{cry} ←(94.5)SmC←(109.1)–lso				
6	DSC	Heating	GFB _{cry} C	$\begin{tabular}{lllllllllllllllllllllllllllllllllll$				
DSC		Cooling	FB _{cry} C	Sld*←(77.0)—SmF←(86.4)—SmB _{cry} ←(94.5)SmC←(109.14)—Iso [14.02] [0.16] [14.07]	32.14	14.64	32.14	
	тм	Heating	GFCA	SId–(88.7)→SmG–(90.5)→SmF–(103.1)→SmC–(107.8)→SmA–(117.7)→Iso				
	1 111	Cooling	GFCA	Sld*←(69.8)–SmG←(107.3)–SmF←(111.6)–SmC←(113.0)–SmA←(117.7)–Iso				
7		nealing Grua [41.		SId–(88.76)→SmG–(90.54)→SmF–(103.13)→SmC–(107.81)→SmA–(117.7)→Iso [41.72] [15.07] [6.69] [0.03] [3.44]				
DSC		Cooling	GFCA	Sld*-(69.8)-SmG-(107.3)-SmF-(111.6)-SmC-(113.0)-SmA-(117.7)-Iso [20.71] [11.34] [0.02] [9.91]	47.90	1.40	40.20	4.70
		Heating	GFICA	Sld₁–(92.0)→SmG–(99.5)→SmF–(104.0)→SmI–(107.2)→SmC–(111.3)→SmA–(119.6)→Iso				
8	ТМ	Cooling	GFICA	Sld-(91.0)-SmG-(92.0)-SmF-(95.1)-SmI-(96.8)-SmC-(111.2)-SmA-(119.6)-lso				
	DSC	Heating	GFICA	$Sid_1-(91.95) \rightarrow SmG-(99.45) \rightarrow SmF-(103.86) \rightarrow SmI-(107.17) \rightarrow SmC-(111.40) \rightarrow SmA-(119.76) \rightarrow Iso$ [10.93] [1.02] [0.29] [0.53] [0.17] [5.73]				
		Cooling	GFICA	Sld←(91.0)-SmG←(92.0)-SmF←(95.1)-SmI←(96.8)-SmC←(111.2)-SmA←(119.76)-Iso [0.36] [3.77] [1.91] [1.82] [0.01] [9.35]	28.76	14.40	20.12	8.64

Liquid Crystals

-					-		
	Heating	GFCA	$SId^*\!\!-\!\!(45.2)\!\!\rightarrow\!\!SmG\!\!-\!\!(92.0)\!\rightarrow\!\!SmF\!\!-\!\!(99.7)\!\rightarrow\!\!SmC-\!\!(103.0)\!\rightarrow\!\!SmA\!\!-\!\!(115.8)\!\rightarrow\!Iso$				
ТМ	Cooling	GFCA	$SId^{*} \leftarrow (61.6) - SmG \leftarrow (96.1) - SmF \leftarrow (101.7) - SmC \leftarrow (107.2) - SmA \leftarrow (115.8) - Iso$				
DSC	Heating	GFCA	Sld*–(45.2)→SmG –(91.97)→SmF–(99.57)→SmC–(103.02)→SmA–(115.8)→Iso [41.37] [2.03] [0.12] [1.64]				
	Cooling	GFCA	Sld*←(61.6)-SmG←(96.12)-SmF←(101.59)-SmC←(107.18)-SmA←(115.8)-lso [15.47] [11.05] [0.05] [8.13]	54.20	5.59	45.58	8.6
ТМ	Heating	FCA	SId*–(49.2)→SmF–(88.7)→SmC–(90.0)→SmA–(91.8)→Iso				
	Cooling	GCA	SId←(62.9)−SmG←(75.3)−SmC←(87.81)−SmA←(91.8)−Iso				
DSC	Heating	FCA	SId*-(49.2)→SmF-(88.7)→SmC-(90.0)→SmA-(91.8)→Iso [6.58] [0.39] [33.81]				
000	Cooling	GCA	SId-(62.9)-SmG-(75.3)-SmC-(87.81)-SmA-(91.8)-Iso [10.92] [3.15] [0.08] [8.70]	28.90	12.59	24.91	3.9
	Heating	GFB _{cry} A	Sld^{*} -(89.0) \rightarrow SmG-(92.4) \rightarrow SmF-(95.8) \rightarrow SmB _{cry} -(107.2) \rightarrow SmA-(121.8) \rightarrow Iso				
ТМ	Cooling	GFB _{cry}	SId←(94.8)–SmG←(100.9)–SmF←(106.6)–SmB _{cry} ←(121.8)–Iso				
DSC	Heating	GFB _{cry} A	$\label{eq:sid} \begin{array}{c c} Sid^*\!$				
	Cooling	GFB _{cry}	SId←(94.8)-SmG←(100.9)-SmF←(106.6)-SmB _{cry} ←(121.8)-Iso [0.12] [12.68] [6.29] [7.94]	27.00		27.00	
	Heating	GB _{cry} A	$SId^*(54.6) \rightarrow SmG(93.68) \rightarrow SmB_{cry}(102.45) \rightarrow SmA(119.36) \rightarrow Iso$				
I IVI	Cooling	GFIB _{cry} A	Sld←(84.2)–SmG←(85.2)–SmF←(89.7)–SmI←(103.6)–SmB _{cry} ←(114.9)–SmA←(119.3)–lso				
DSC	Heating	GB _{cry} A	$ \begin{array}{c c} Sld^* - (54.6) \rightarrow SmG - (93.68) \rightarrow SmB_{cry} - (102.45) \rightarrow SmA - (119.36) \rightarrow lso \\ \hline & [30.87] & [0.49] & [0.03] \end{array} $				
	Cooling	GFIB _{cry} A	Sid←(84.26)-SmG←(85.25)-SmF←(89.73)-SmI←(103.64)-SmB _{cry} ←(114.95)-SmA←(119.36)-Iso [2.04] [15.66] [0.41] [4.3] [7.36] [1.07]	35.10		30.69	4.4
n [] ind	icates Enth	alpy in J/g					
	DSC TM DSC TM DSC TM DSC	TMCoolingDSCHeatingCoolingHeatingTMCoolingDSCHeatingDSCCoolingTMCoolingTMCoolingDSCHeatingTMCoolingDSCHeatingDSCCoolingTMCoolingDSCHeatingDSCCoolingTMCoolingDSCHeatingCoolingCooling	$\begin{array}{c c c c c c } TM & \hline Cooling & GFCA \\ \hline TM & \hline Cooling & GCA \\ \hline DSC & \hline Cooling & GCA \\ \hline Cooling & GCA \\ \hline Cooling & GFB_{cry}A \\ \hline TM & \hline Cooling & GFB_{cry}A \\ \hline \end{array}$	$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$ \begin{array}{c c c c c c c c c c c c c c c c c c c $

Authors' Response to Ms. TLCT-2019-0054

Influence of Meta-Extended Rigid-Core, Complementary Hydrogen Bonding and Flexible Chain on Polymorphism in Schiff-based Hydrogen Bonded Liquid Crystals: (4)MeOBD(3)AmnBA:nOBAs. Krishna Murthy et al.

Authors thank reviewer for his comments which resulted to improve the quality of manuscript. Response to the **general comments** and **particular comments** is partitioned and written in green pen. Accordingly, modified portions of text in the revised script are also marked in green pen. Additional references included in the revised text due to suggestions of the reviewer or circumstantial nature are also listed in green pen.

General comments

The manuscript describes the preparation of new hydrogen-bonded complexes showing liquid crystalline behaviour, a topic of continuous development and interest for the liquid crystal community.

Authors thank reviewer for his encouraging comment regarding the ms. and it's general aspect of interest to the liquid crystal research community.

In this resubmission, authors have tried to clarify the motivation, though the degree of novelty in this approach has not been addressed sufficiently. Some examples of similar supramolecular curved structures could be quoted, as well as future perspectives.

Authors clearly wrote the text for motivation of present work (p-3 & -4, 3rd paragraph in green pen, section-1. However, the sub-heading (as in earlier version) is **removed** in revised version. The novelty aspect of the work is in the revised version addressed (section-1, fig-1, p-4; section-3a, p-6, section-3c, p-10; section-3e, p-13; section-4, p-15 & -16) with respect to the meta-extended core of calamitic type of LC molecules. Nevertheless, the influence of molecular architecture tuned by supra-molecular design involving complementary hydrogen bonding interaction is also analyzed. The type of hydrogen bonding is also mentioned in the title of paper. Influence of meta-extended core and complementary HB for the realization of device savvy tilted LC phases constitutes the overall contributory aspect of the present work.

There are still quite a few style refuses and paragraphs are definitely too long. There is still too much fundamental information. The ideas are difficult to follow and some figures hard to read. The paper could be simplified to attain a higher impact and visibility, highlighting the main ideas (how to maximise SmC and other tilted phases).

Long paragraphs in the old script are partitioned (section-3c,p-8 & -9; section-3d, p-10 & -11; section-3e,p-12,-13 &-14; section-3f, p-14 & -15) in revised version by highlighting the impact of meta-extended LC core and supra-molecular HB architecture for the enhancement of SmC and tilted LC phases.

Fundamental, but, redundant information regarding the DSC peaks to identify T_c and evaluation of ΔH etc., in removed in the revised version.

Clarity of ideas and its readability for ms. are improved by re-writing (p-4; p-6; p-8 & -9; p-11; p-12 & -13; p-15 and p-16) the text. Modifications are incorporated in some figures (old fig-7 is redrawn as Fig-9) to for improve clarity. Additional figures (Fig-11 and Fig-12) are also included to explain the trends with improved clarity.

Considering that the findings are potentially interesting for the journal readers, my recommendation is that the manuscript can be accepted for publication in Liquid Crystals, but only after major corrections are carried out, including, considerable re-writing and restructuration.

As per the reviewer's directive, a major revision is carried out with corrections, re-writing and restructuration to bring out a comprehensive overview for the impact of meta extended core for the LC phase abundance. Modifications are marked in green pen.

Authors should still consider previous comments on figures and tables, as well as a new set of particular issues that are now presented.

Reviewer's earlier comment to increase the clarity for figures and tables is taken care in the revised version. Fig-1 is newly introduced. Fig-4 is a revised version of old fig-7 that conserves the clamitic shape of HBLC complex, despite it being superposed by bent (meta-extended core) shape. Fig-5(a,b,c) represents redrawn old fig-3(a,b,c) with uniformity for X-axis and notation for wave number duly inscribed. To improve the clarity, readability and contrast, trends in ΔT_c and ΔT_{Tilt} of nOBA's in old fig-7, are redrawn as Fig-9. New illustrations, viz., fig-11and fig-12 are also introduced in to the revised script in respond the reviewer's comment of section-3 (p-13 of old script).

Particular comments

Abstract.

Avoid acronyms that have not been introduced earlier.

Abstract is revised by giving full expansion for acronyms. The acronym 'HBLC' in abstract is provided (p-1) with full expansion in the revised script.

Corrections for the text (lines-28 and -30 of old script), which are required as per the experimental observations are also incorporated in the revised version. Correction is made for line-34 (of old script) as many multicritical points are identified in the phase diagram.

Introduction

Page 4. Line 22. I would eliminate "Motivation:" The subheading 'Motivation' is deleted (p-3 and -4) from the text of revised version.

Page 4. Line 22. "Selection of Oxygen atom as bridging moiety and Schiff base as core are known to promote lateral stacking and growth of layered smectic phases." Does require a reference as evidence? Possibly it appears later.

References for the Schiff base moiety [17,19,25,31,32] with Oxygen as bridging atom [18,19] and its influence to enhance lateral stacking trend (viz., for induction and origin of layered smectic phase structures) are included (p-3 and -4) in the revised script.

Page 4. Lines 35 to 45. Please consider rephrasing, in order to clarify what the aim of the study is. I also suggest to include a molecular sketch of the complexes formed.

Aim of the present study is rephrased (p-4) for clarity in the revised text. In accord with suggestion, a sketch for the HBLC complex (fig-4) is also included in the revised text at an appropriate (p-6) place.

Page 4. Line 47. Add "A" at the start of the line.

As correctly pointed out by the reviewer, an indefinite article (An) is used (p-4) to initiate the sentence.

Page 4, Line 3. "Influence of HB and its type are investigated using the results of POM and DSC." Probably not needed.

Probably, the reviewer might have mistyped as p-4 and line-3, instead of p-5 and line-3 of old script. It is understood as a case of redundancy, and with this in view, this sentence is deleted (p-4) in the revised script as per the suggestion of reviewer.

Page 5. Line 7. . The whole paragraph seems a bit redundant.

"Paper is organized in **three** sections. Details of chemical constitution and influence of HB for the occurrence of device relevant LC phases are presented as introduction. Motivation for the targeted work is also presented in the section-1. Selection and procurement of chemical ingredients, synthetic route and details of experimental techniques are presented in section-2. Discussion for observed trends and their analysis are presented section-3"

Cited paragraph is simplified (p-4) as one sentence to detail the organization.

Section 2.

Replace "ingredients" by "reactants".

As suggested, the term – 'ingredients' is now replaced (p-5) by *reactants* in revised version.

Page 5. Line 37. "The product obtained in the 1st step which is found to be white in colour signifies its purity." If NMR is clear, colour needn't to be a purity criteria.

White (or more appropriately colorless) product formed is generally advocated to vouch the initial purity of sample, or else, it's immunity against the possible chemical degradation. Text is modified (p-6) in the revised version in the context of this spirit.

I suggest to include "Techniques" before and the "Synthesis", then stating in this section that the Structures were verified by NMR and FT-IR.

As suggested, the words, viz., *Techniques* and *Synthesis* (sections-2a & -2b) are included in the revised script.

Section 3.

Following my previous comment, from my view, the discussion from Page 6. Line 53 until Page 8. Line 50 should be Supplementary information.

Data for ¹H-NMR and ¹³C-NMR chemical shift, relevant figures and its discussion (at least for one representative out of entire series consisting of 10 compounds) is an accepted practice for this journal. In response to the earlier comments, the data of chemical shift is already shifted to Appendices-A and –B. As per the suggestion, along with the data of chemical shift and tally of H- and C-atoms by the analysis of representative figures is also shifted to the appendices in the revised script. The option of whether –

a) to retain data, figs and discussion (for tally of C- and H-atoms) in Appendices-A and -B or

b) to push in to the supplementary section

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is left to the decision of editor. Since discussion for the account/tally of H-atoms and C-atoms provides a clinching evidence to confirm the formation of targeted HBLC complex, authors strongly opine for the retaining (p-5) the NMR representative figs-2 & -3 in the main text and relevant tally of H- and C-atoms in (p-27-32) appendices.

Page 6. Line 33. I would rename this section as "Complexes structures or formation", since the molecular structure corresponds to section 2. (in any case "Supramolecular Structure"). I would then include the FT-IR discussion, to assess the complex formation.

As suggested, section-3a is renamed (p-6) as "Supra-molecular Structure of HBLC Complexes", while the FTIR discussion follows.

Page 8. Line 52 and following. The FT-IR discussion is now more convincing, and well sustained. Authors should consider using Fig. 3(a) and 3(b) for the pristine components, then Fig. 3(c) for the mixture (for the sake of argument). Authors should also consider shrinking this paragraphs considerably, to keep a sensible size of the paper.

Figures 3(a), 3(b) and 3(c) are still a bit unclear, and Fig. 3(a) needs wavenumber annotations (for the sake of consistency).

As suggested, old figures-3a,b,c are renumbered as figs-5a,b,c and re-organized, such that 5a and -5b correspond to pristine components, while -5c is earmarked for the end product HBLC complex. Shrinking of paragraphs pertaining to IR discussion seems difficult, to hold the integrated and comprehensive outlook of discussion.

Modified figures (-5a,b,c) are clearly annotated with wave number to maintain the consistency. Additionally, consistency of scaling for X-axis (keeping minor ticks) is also preserved throughout the new Figs-5a,b,c. However, since IR absorption for fig-5c seems to get superposed with X-axis scaling, additional version of figs-5a, -5b, and -5c are submitted as supplementary (figs-5a,b,c) information.

Authors are encouraged to merge these three in the same figure, by piling up the curves and shifting them towards the Y-axis.

As per reviewer's suggestion, all the three (-a, -b, and –c) curves are piled up as one figure-5 in both of the versions.

Page 11. Line 5. "LC phases are identified by their comparison with standard POM textures reported", is this needed, or simply common knowledge?

As suggested, the sentence is removed in revised version.

P11. Line 17. "Similarity of molecular architectures between the present meta-extended core and the one side mass loaded core (with double chiral centres) is argued to result for the textural similarity." Specify reference.

As suggested, the relevant references, viz., Refs.-69, -70 are included (p-9) in the text of revised version. They are listed in the reference section also.

P11. Line 21. The following text could appear as a separate paragraph, or could be summarised with the information of the phase assignation, leaving the textures description as SI.

"SmC phase is found to exhibit four brushed schlieren texture (plate-4) as grown from the precursor SmA phase in heating run of (4)MeOBD(3)AmnBA:8OBA. SmC phase is found to exhibit blackonwhite natural mosaic texture (plate-5) also. SmBcry is found to grow as colored mosaic (plate-6) and glossy focal conic fan batonnet textures (plate-7). SmI is found to exhibit broken focal conic plume-like fan (plate-8) texture when grown by cooling the SmBcry in (4)MeOBD(3)AmnBA:8OBA. SmI is also found to Liquid Crystals

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exhibit colored threaded (plate-9) texture in (4)MeOBD(3)AmnBA:12OBA. However, SmF is found to exhibit characteristic colored marble threaded texture (plate-10) and checkered board of elongated fan textures (plate-11) in HBLC complex for n=9 and 12, respectively. In addition to the above textures, SmF is also found to exhibit a paramorphotic colored mosaic texture with undulated boundaries (plate-12). This texture is found to be much akin to the undulated edge texture observed for n=11 in the cooling run. SmG exhibited characteristic colored mosaic (plate-13) texture in the lower homologues of the present series of HBLCs." Since, this paragraph (section-3c, 2nd paragraph, p-8&9) entails continuity of discussion for POM observations (i.e., from plate-1 to -13), authors opine that splitting the paragraph would be improper. Inability to comply with reviewer's opinion may be comprehended as an effort to bring a comprehensive view and integrity to the text for POM observations. However, in accord with the suggestion of the referee, summary of POM textural observations and the phase variance with respect to the molecular architecture is scribed as separated paragraphs (p-8 and -9) for clarity. Authors may be given an opportunity to express and submit their objection to shift POM textures and relevant discussion in to the supplementary information file. After all, POM textures court the fulcrum and fundamental experimental evidence to confirm the occurrence of LC phases with a variety and diversity of structural/orientational orders. P12. Line 7. Please clarify/rewrite this text: "Although intermediate and higher homologues do not exhibit 3D SmBcryst phase, the lower homologues are found to exhibit orthogonal (hexagonal) SmBcryst phase. But, higher homologues (for n=11 and 12) are found to exhibit SmBcryst again." Cited text is re-written (p-8 and -9) to improve the clarity in the revised version. P12. Line 19. The text below is an interesting observation. Can the authors specify what conditions promote the appearance of tilted phases in nOBAs, and compare to the present complexes (perhaps as a function of alkyl chain length "n)" "Overall tilted phase abundance in HBLCs is observed to witness an enormous amount of increase in comparison with nOBAs. Hence, meta extended LC core is argued to promote tilted phase stability for device applications." Last sentence of 1st paragraph of p-9 is also re-written to specify the conditions that prevailed for the appearances of tilted LC phases. Especially, the influence of meta-extended core and increased flexibility (by end chain length "n") are also included to justify the observations. Page 12. Line 30. Not sure this is needed: "Temperatures at which DSC thermo gram exhibits a dips (in heating scan) or peaks in cooling scan is identified as phase transition temperatures." As per reviewer's directive, line (p-10) is removed in revised version. Page 12. Line 24 to Page 13 Line 36. Authors are strongly suggested to break down this paragraph into three main topics, and simplify the discussion: -General description of DH and Ti values (comparison with literature). -Small enthalpy changes associated to small structures of tilted phases. -Monotropic character of the longer-chains analogues. I would shrink the text and refer to the tables.

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3	Entire paragraph of old version is split in to three paragraphs (p-10 & -11) with different
4	side-headings (subsections-3d _{i,ii}) as suggested by the reviewer.
5	In the revised version, i.e., section-3d(i), the shifting trend of LC phase range towards ambient
6	temperatures due to the meta-extended core and complementary HB is also introduced in p-10.
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8 9	Page 13. Line 38 to line 51. The paragraph can be considerably reduced, as it is clear that increasing the
9 10	alkyl terminations may tune the flexibility. It'd be more interesting to actually compare a similar
10	variation in "n" on other systems (nOBAs, for example).
12	Reviewer's comment is properly responded by the authors. And so, effort in this direction
12	yielded finer aspects of influence of molecular design. As suggested, the interesting aspects of
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14	variations of flexible alkyl chain length (n) in meta (table-2) extended HBLC's and in pure [10]
16	dimeric nOBA's is introduced in the beginning of section-3e in p-12 and -13 of revised script.
17	The impact of meta-extended LC core now clearly exemplifies enhanced abundance of device
17	savvy tilted LC phases.
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20	Page 14. Lines 17 to 46. Text appears repetitive and could be merged as part of the introduction, then
21	focus on the results in this (already) long section.
22	As suggested, the seemingly repetitive text is shifted in to the introduction (section-1, p-4).
23	Some redundant part of the text is also removed.
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25	Page 15. Line 21. To my view, this sentence is not clear, considering that nOBAs are indeed HBLCs:
26	" Δ T Tilt is found to be zero for n ≤6 in pure nOBAs, while it is found to attain a finite value in HBLCs to
27	infer the influence of HB for inducing tilted LC phases."
28	The sentence is rewritten (2 nd paragraph, p-14) in the revised version to improve the clarity.
29	The sentence is rewritten (2 paragraph, p-14) in the revised version to improve the clarity.
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31	Please rephrase the following sentences for clarity:
32	Page 15. Line 52. "SmC phase stability [ΔT] _c is found to exhibit almost a monotonical
33	increase for n≥6 in (except for the increment of n=11 to n=12) than in pure nOBAs. But it is
34	also observed that the $[\Delta T]_c$ attains higher values for even homologues in HBLCs."
35	Sentences regarding the discussion of $[\Delta T]_{c}$ is rewritten (p-15) with improved clarity.
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37	Page 16. Line 11. "It is also noticed that [ΔT]C is found to attain either a lesser value or zero value for
38	all odd homologues with $n \ge 4$ for HBLCs"
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40	Sentence is rewritten (p-15) for clarity regarding the contrast behaviour of $[\Delta T]_{C}$ in odd
41	homologues of meta-extended core LCs.
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43	Page 16. Lines 11 to 34. This discussion requires major polishing, trimming and organising. (check the
44	typo in $5 \ge n \le 11$). I suggest that just a few ideas are shown, expressing clearly when the discussion refers
45	to nOBAs and when to the new complexes, considering that both series are HBLCs.
46	Cited text (Page 16. Lines 11 to 34 of old text) by the reviewer is re-written (p-15) with enhanced
47	clarity in the revised script by considering the contributions of μ_t for meta-extended core,
48	inclined HB interaction and orientational disorder by odd homologues. Typographical mistake
49	for greater/lesser inequality is corrected in revised version.
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51	Page 16. Line 32. The following paragraph looks contradicting?
52	"LC phase stability with the underlying structural diversity is found to be maximum for n=9 and 10
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54	compounds. The observed higher LC phase stability for the n=7 and 9 is argued due to the chain number
55	being odd in these compounds."
56	In accord with reviewer's previous directive, text (p-15) is re-written to eliminate the alleged
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59 60	URL: http://mc.manuscriptcentral.com/tlct Email: c.t.imrie@abdn.ac.uk
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contradiction is eliminated.

The whole page 16 is one single paragraph, which makes following the ideas rather difficult. Entire paragraph is split (p-15 and -16) into three small paragraphs and sentences are restructured. Almost entire text is re-written to improve the clarity by comparing with nOBA's and meta-extended core LCs.

Page 16. Line 53 and following. Almost all section 3h is stating the evident, including the physical meaning of phase transitions. I personally do not see what the contribution of this section is.

Old version of section-3h is now re-written as section-3f. The contributory aspects of comparative study (through figs-10, -11 and -12) of integrated and differential enthalpies is to emphasize the impact of meta-extended core design on the growth of LC phase structures optimized over the heat energy exchange during heating and cooling scans. Sumptuous amount of discussion is introduced in the re-written text (p-16 & -17) in the revised script.

In fact, reviewer's comments regarding

a) contributory nature of the fig-10 with discussion of integrated enthalpies and

- b)influence of meta-extended architecture
- c)exclusion/isolation of HB interaction
- d)inclusion of Schiff base LC architecture

rendered the authors to introduce figs-11 and -12 for to carve out a meaningful text in the revised text.

Conclusions.

As a general comment, I would avoid bullet points. As suggested the bullet point version is changed (p-15) in to a single paragraph for conclusion.

Page 17. Line 41. I would rather say "preparation" than "synthesis" of the new HBLC. As suggested, the term "synthesis" is replaced (p-17) by "preparation".

Bullet point i) The phrase does not make sense, but, in any case, the complex formation is assessed by FT-IR.

Authors modified the text regarding spectroscopic methods. The first two sentences (in conclusions, p-17) are written as to how these methods can be used to confirm the formation of targeted HBLC complex.

Bullet point ii) is not clear. The effect of extension of supramolecular core should be clarified, and also, "extended" with respect to what? Respect to the nOBAs? Some examples on other supramolecular liquid crystal having curved geometries and resulting dipoles could be mentioned, such as hydrogen-bonded bent-core materials (see work by Serrano et al., Zaragoza-Spain) or twist-bend nematogens (see work by Imrie et al., Aberdeen-UK).

Meta-extension of LC core is claimed with respect to the centre of mass situated at the aniline relevant aromatic ring in the Schiff based intermediate. It is clearly illustrated in fig-4. As per the reviewer's directive, other examples of supra-molecular HBLCs, Bent LCs exhibiting T_{NB} phases are also referred (Ref-52-57) in modified text (Introduction, 2nd paragraph in p-3; along with the Serrano et al and Imrie et al reports) in the list of references also.

Bullet point iii) has not been addressed in the text, but rather argued, in liaison of bullet point Text regarding electro-negative O-atom to promote lateral stacking, and hence the predominant growth of layered smectic phases is introduced in section-3c, p-10, before

drawing conclusions. Reasoning for predominance of smectic phases is attempted for its connection to meta-extended core.

iv).Please refer more clearly in bullet point

Explanation for the observed quenching Nematic phase is written (1st paragraph of p-13), while the conclusion is written in p-17 & 18.

v) to the effect of "n" on the phase behaviour of the HBLCs, since, it was one of the main aims. This must include bullet point vii).

Appearance of quasi-2D phases with increasing flexible component 'n' is entirely different issue than that in bullet point-vii (of old version) regarding the realization of multicritical points. Hence, multicritical points and exotic symmetries) and orientaional disorder by long chains are scribed in two different sentences. Text is re-written to bring proper connectivity between the appearance of exotic phases (in p-11, section-3dii) induced by long chain (increasing 'n') through orientational disorder, and multicritical points (p-13, section-3e) in whose vicinity rare symmetries converge.

Page 18. Line 12. Define "exotic" symmetries.

As suggested by reviewer, the explanation (p-13, in section-3e for Phase diagram, 3rd para--graph) for exotic symmetries and realization of multi-critical points is introduced.

As all the comments made by the reviewer are responded, answered and properly defended by reports of contemporary relevance, revised script may be considered for publication.

Transmittance %1

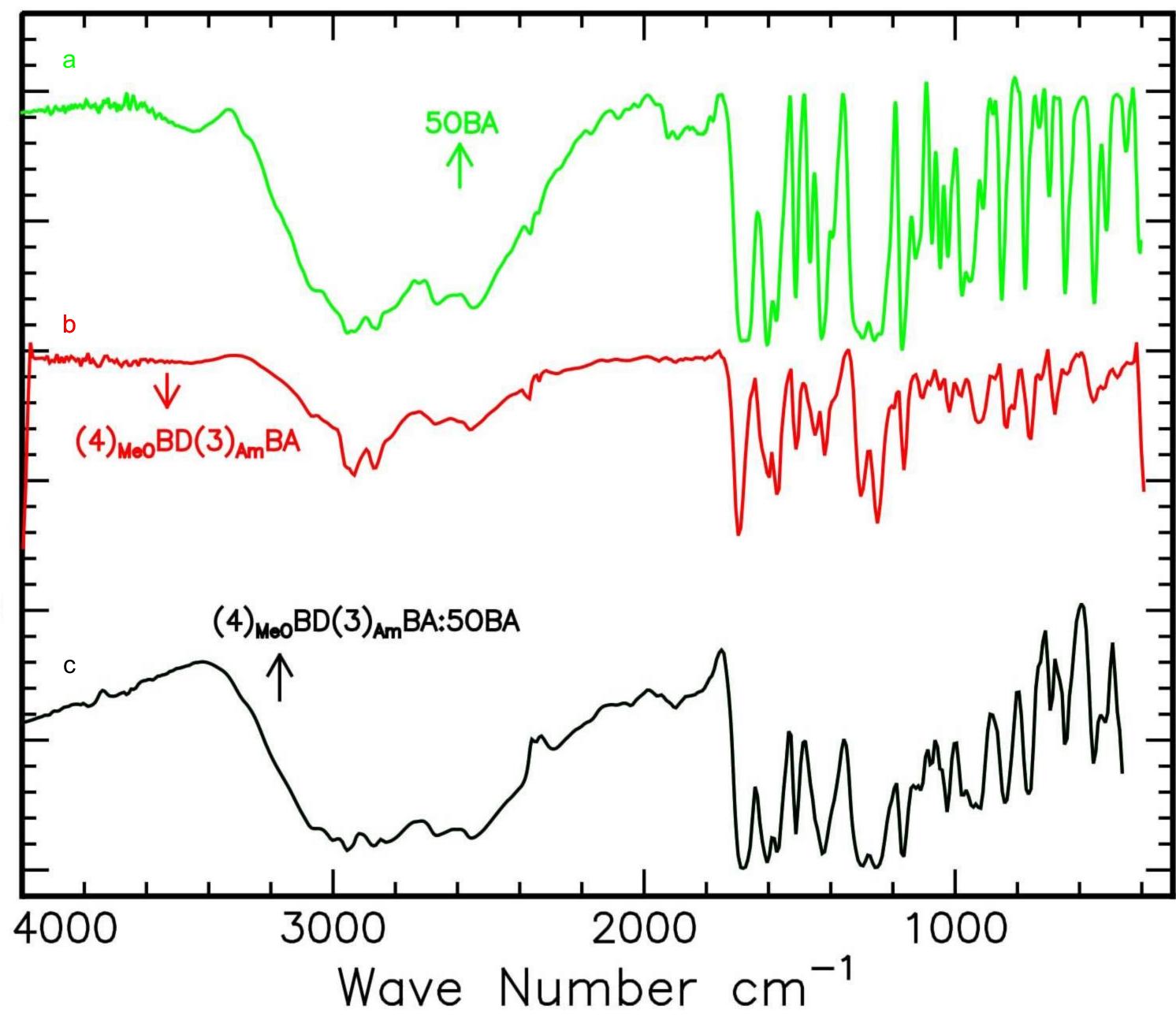


Figure-5