Supporting Information

An Easily Available Ratiometric Reaction-based AIE Probe for Carbon Monoxide Light-up Imaging

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Table of Contents

Experiment section:

Materials and instrumentation	S-3
General procedures for the detection of CO	S-3
Determination of the detection limit of BTCV-CO toward addition of CO	S-4
Cell Culture	S-4
Cell viability	S-4
Cell treatment and cell imaging	S-5
In vivo animal imaging	S-5
Scheme S1	S-5
Synthesis of compound 3	S-5
Synthesis of BTCV-CO	S-6
Synthesis of BTIC	S-6

Figures and tables:

Figure S1
Figure S2
Figure S3
Figure S4
Figure S5
Figure S6
Figure S7
Figure S8
Figure S9
Figure S10
Figure S11
Figure S12
Figure S13
Figure S14
Figure S15
Figure S16
Figure S17
Figure S18
Figure S19
Figure S20
Figure S21
Figure S22
Figure S23
Table S1S-17
References

Experiment section:

Materials and instrumentation

Chemicals were purchased from Energy-Chemical, Sigma-Aldrich, J&K and used without further purification. Solvents and other common reagents were obtained from Sigma-Aldrich. ¹H NMR and ¹³C NMR spectra were measured on a Bruker ARX 400 MHz spectrometer. High-resolution mass spectra (HRMS) were recorded on a GCT Premier CAB 048 mass spectrometer operating in MALDI-TOF mode. UV-vis absorption spectra were recorded on a Rarian 50 Conc UV-Visible spectrophotometer. Fluorescence emission spectra were recorded on a Edinburgh FS5 fluorescence spectrophotometer. Absolute fluorescence quantum yield were measured on Hamamatsu C11347-11 Quantaurus-QY Absolute PL quantum yield spectrometer. Cellular imaging experiments were performed with confocal laser scanning microscope (TCS SP5, Leica, Germany) equipped with Argon, red HeNe, and green HeNe lasers.

General procedures for the detection of CO

Unless otherwise noted, all the spectral measurements were performed in 5 mM phosphate buffer (pH 7.4, containing 5% DMSO) according to the following procedure. The stock solution (1.0 mM) of probe BTCV-CO was first prepared in DMSO. 10 μ L of BTCV-CO stock solution was added to 2 mL PBS followed by addition of different volume of CO solution. The mixture was incubated for certain times at 37 °C and then, the reaction solution was transferred to a quartz cell with 1 cm optical length for measurements. In the meantime, the blank solution without CO was also prepared and measured under the same conditions for comparison.

CO was prepared from $[Ru(CO)_3Cl_2]_2$ (CORM-2) in aqueous solution. Hydrogen peroxide (H₂O₂) and hypochlorite (ClO⁻) were delivered from 30% and 5% aqueous solutions, respectively. Hydroxyl radical (•OH) was generated by reaction of 100 µM Fe²⁺ with 100 µM H₂O₂. ROO• was generated from 2,2'-azobis(2-amidinopropane)dihydrochloride (AAPH), which was dissolved in deionizer water and then stirred at 25 °C for 30 min. Theoretically, 1 equiv of AAPH will generate 2 equiv of ROO•. R stands for 2-amidinoisopropyl. ONOO⁻ was prepared according to the literature and modified.^{S1} A mixture of 50 mM sodium nitrite and 50 mM hydrogen peroxide in 50 mL of ice-cold water was rapidly stirred in a 250 mL beaker. A solution of hydrochloric acid (1 M, 25 mL) is rapidly thrown into the nitrite/peroxide solution followed by a solution of sodium hydroxide (1.5

M, 25 mL) approximately 1 second later. The concentration of ONOO⁻ was determined by UV spectrophotometry ($\varepsilon_{302} = 1670 \text{ L} \cdot \text{mol}^{-1} \cdot \text{cm}^{-1}$). Angeli's salt (AS) was purchased from Santa Cruz Biotechnology, and HNO was prepared from AS using double-distilled water.

Determination of CO concentration released from CORM-2

The release of CO from CORM-2 was assessed spectrophotometrically by measuring the conversion of deoxymyoglobin (deoxy-Mb) to carbonmonoxy myoglobin (MbCO). The amount of MbCO formed was quantified by measuring the absorbance at 540 nm (extinction coefficient=15.4 mmol·L⁻¹·cm⁻¹). Myoglobin solutions (66 μ mol/L final concentration) were prepared fresh by dissolving the protein in 0.04 mol/L phosphate buffer (pH=6.8). Sodium dithionite (0.1%) was added to convert myoglobin to deoxy-Mb. CO released from CORM-2 was quantified by adding aliquots of stock solutions (10 μ L) of CORM-2 in DMSO directly to the myoglobin solution.

Determination of the detection limit of BTCV-CO toward addition of CO

Based on the linear fitting in Figure 2C and Figure S14, the detection limit (C) is estimated as follows:

$$C = 3\sigma/B \tag{1}$$

Where σ is the standard deviation obtained from three individual fluorescent intensity ratio (I_{546}/I_{710}) or (($I-I_0$)/ I_0) at 546 nm of BTCV-CO (5 µM) without any CO and *B* is the slope obtained after linear fitting the titration curves within certain ranges.

Cell cultures

The MCF-7 cells were cultured in DMEM (containing 10% heat-inactivated FBS, 100 mg·mL⁻¹ penicillin and 100 mg·mL⁻¹ streptomycin) at 37 °C in a humidified incubator with 5% CO₂. Before the experiments, the cells were pre-cultured until confluence was reached.

Cell viability

Cell viability was determined by using MTT assay which is based on the reduction of 3-(4,5-dimethythiazol-2-yl)-2,5-diphenyl tetrazolium bromide (MTT, yellow in color) into formazan (blue color) by mitochondrial succinate dehydrogenase. Dispense 100 µL of cell suspension (5000 cells/well) in a 96-well plate. Pre-incubate the plate for 24 h at 37 °C in a humidified incubator with 5% CO₂. Add 10 µL of various concentrations of BTCV-CO into the culture media in the plate. After incubating the plate for 24 h in the incubator, the cell medium were

exchanged with fresh medium (100 μ L), and then 20 μ L of the MTT (5 mg/mL) solution was then added. Medium was removed after the incubation period of 4 hours followed by the addition of 100 μ L of DMSO to dissolve the formazan crystals. Absorbance was taken at 600 nm by an ELISA Plate Reader (Biotek Synergy HT). Untreated cells were taken as control. All the experiments were performed in triplicate. Cell viability was determined by using given formula:

Cell viability (%) =
$$\frac{\text{Absorbance of treated cells}}{\text{Absorbance of untreated cells}}$$
 (2)

Cell treatment and cell imaging

For the imaging of CO, the MCF-7 cells were incubated with 5 μ M BTCV-CO and 5 μ M PdCl₂ for 30 min at 37 °C, then the media was replaced with PBS buffer and the cells were treated with different concentrations of CORM-2 (0, 10, 20 and 50 μ M) for 30 min. The imaging was acquired using a confocal laser scanning microscope (TCS SP5, Leica, Germany). For cell imaging, the cells were washed with PBS three times. A 476 nm laser was used as the light source and emission was collected from 500 to 580 nm (green channel) and 580 to 700 nm (red channel).

In vivo animal imaging

Female BALB/c white mice (4–6 weeks old and weighted 25–30 g) were purchased from Vital River Laboratory Animal Technology Co. Ltd (Beijing, China) and all animals received care incompliance with the guidelines outlined in the Guide for the Care and Use of Laboratory Animals. The procedures were approved by Shenzhen Institutes of Advanced Technology, Chinese Academy of Sciences Animal Care and Use Committee. The probe solution (25 μ L, 1 mM in DMSO) was injected to white mice *via* intraperitoneal injection, followed by an injection of CORM-2 (75 μ L, 2 mM). Control group was injected with 75 μ L of PBS. Images were taken after injection using the Maestro *in vivo* imaging system.

Scheme S1. Synthetic route to BTIC.



Synthesis of compound 3. 4-(Diethylamino)salicylaldehyde (compound 1, 193.0 mg, 1.0 mmol), allyl bromide (compound 2, 242.0 mg, 2.0 mmol) and K₂CO₃ (415.0 mg, 3.0 mmol) was mixed in 10 mL CH₃CN. The mixture was refluxed under the nitrogen atmosphere. The reaction was monitored by TLC. When compound 1 disappeared, the mixture was cooled to room temperature and filtrated to remove any solids. The obtained solution was washed with brine and extracted with dichloromethane. The organic layer was dried over anhydrous Na₂SO₄. After concentrated, the residue was purified by silica gel chromatography with petroleum ether/dichloromethane (1:2 \sim 2:1, ν/ν) to give compound 3 as a white solid (212.0 mg, 91%). ¹H NMR (400 MHz, CDCl₃) δ 10.09 (s, 1H), 7.59 (d, *J* = 8.9 Hz, 1H), 6.16 (d, *J* = 10.4 Hz, 1H), 5.94 (d, *J* = 2.0 Hz, 2H), 5.35 (d, *J* = 18.3 Hz, 1H), 5.21 (d, *J* = 10.6 Hz, 1H), 4.52 (d, *J* = 5.0 Hz, 2H), 3.32-3.27(m, 4H), 1.17-1.08 (m, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 186.77, 163.21, 153.79, 132.97, 130.18, 117.35, 114.18, 104.41, 93.67, 68.78, 44.74, 12.52.

Synthesis of BTCV-CO. A mixture of compound **3** (117.1 mg. 0.5 mmol), benzothiazole-2-yl-acetonitrile (174.2 mg, 1.0 mmol), piperidine (150 µL, 1.6 mmol), acetic acid (150 µL, 2.6 mmol) in 25 mL toluene was refluxed under nitrogen atmosphere for 16 h. After cooling to room temperature, the mixture was washed with brine and extracted with dichloromethane. The organic layer was dried over anhydrous Na₂SO₄. After concentrated, ethyl ether/dichloromethane (15:1, v/v) was added and BTCV-CO was precipitated and collected as a red solid (166.1 mg, 85%). ¹H NMR (400 MHz, CDCl₃) δ 8.57 (s, 1H), 8.43 (d, J = 9.2 Hz, 1H), 8.01 (d, J = 8.1 Hz, 1H), 7.83 (d, J = 7.9 Hz, 1H), 7.49 - 7.41 (m, 1H), 7.38 - 7.30 (m, 1H), 6.37 (m, 1H1H), 6.20 - 6.03 (m, 2H), 5.50 (d, J = 18.8 Hz, 1H), 5.37 (d, J = 11.9 Hz, 1H), 4.67-4.65 (m, 2H), 3.45-3.40 (m, 4H), 1.24-1.21 (m, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 165.91, 160.46, 153.99, 152.58, 141.21, 134.42, 133.01, 130.53, 126.32, 124.85, 122.90, 121.26, 118.63, 117.65, 110.04, 105.25, 96.42, 94.36, 69.28, 44.90, 12.72. HRMS (MALDI-TOF): m/z: [M]⁺ calcd for C₂₃H₂₃N₃OS⁺: 389.1562; found: 389.1602.

Synthesis of BTIC. A mixture of compound 1 (97.1 mg, 0.5 mmol), benzothiazole-2-yl-acetonitrile (174.3 mg, 1.0 mmol) piperidine (150 μ L, 1.6 mmol), acetic acid (150 μ L, 2.6 mmol) in 25 mL toluene was refluxed under nitrogen atmosphere for 16 h. After cooling to room temperature, the mixture was washed with brine and extracted with dichloromethane. The organic layer was dried over anhydrous Na₂SO₄. After concentrated, ethyl ether/dichloromethane (30:1, *v/v*) was added and BTIC was precipitated and collected as a red solid (161.2 mg, 92%). ¹H NMR (400 MHz, CDCl₃) δ

8.87 (s, 1H), 8.00 (d, J = 8.1 Hz, 1H), 7.92 (d, J = 7.6 Hz, 1H), 7.50 – 7.42 (m, 2H), 7.37 – 7.31 (m, 1H), 6.63 (d, J = 11.4 Hz, 1H), 6.52 (d, J = 2.3 Hz, 1H), 3.45-3.40 (m, 4H), 2.60 (s, 1H), 1.24-1.21 (m, 6H). ¹³C NMR (100 MHz, CDCl₃), δ 161.80, 161.06, 157.03, 152.57, 152.10, 142.03, 136.26, 130.77, 126.06, 124.42, 122.12, 121.56, 112.40, 109.98, 108.67, 96.97, 45.07, 40.96, 12.48. HRMS (MALDI-TOF): m/z: [M+H]⁺ calcd for C₂₀H₂₀N₃OS⁺: 350.1327; found: 350.1321.

Figures and tables:



Figure S2. ¹³C NMR spectrum of compound 3 in CDCl₃.



Figure S4. ¹³C NMR spectrum of BTCV-CO in CDCl₃.



Figure S6. ¹H NMR spectrum of BTIC in CDCl₃.





Figure S7. ¹³C NMR spectrum of BTIC in CDCl₃.



Figure S8. HRMS spectrum of BTIC.



Figure S9. (A) PL spectra of BTIC in toluene/DMSO mixtures with different toluene fractions ($f_{\rm H}$). (B) Plot of PL intensity of BTIC at maximum emission wavelength *vs.* $f_{\rm H}$ in the toluene/DMSO mixtures.



Figure S10. (A) UV/Vis spectra and (B) PL spectra of BTCV-CO (5 μ M), BTCV-CO + PdCl₂ (5 μ M) and BTCV-CO + PdCl₂ + CORM-2 (50 μ M) at 37 °C for 20 min in 5% DMSO/PBS solution. Excitation wavelength (λ_{ex}) = 465 nm.



Figure S11. HRMS spectrum of BTCV-CO after incubation with PdCl₂ and CORM-2 (125 μ M) at 37 °C for 20 min.



Figure S12. PL spectra of BTCV-CO (5 μ M in PBS solution, pH 7.4, containing 5% DMSO) after incubation with CORM-2 (125 μ M) in presence of PdCl₂ with different concentrations for 20 min.



Figure S13. (A) The PL intensity ratios $((I-I_0)/I_0)$ of BTCV-CO (5 μ M) + PdCl₂ (5 μ M) at 546 nm in the presence of different concentrations of CORM-2 as a function of incubation time. (B) PL internsity ratios (I_{546}/I_{710}) of BTCV-CO (5 μ M) + PdCl₂ (5 μ M) in the presence of different concentrations of CORM-2 as a function of incubation time. $\lambda_{ex} = 465$ nm.



Figure S14. The PL intensity ratios $((I-I_0)/I_0)$ of BTCV-CO (5 μ M) + PdCl₂ (5 μ M) at 546 nm as a function of the concentration of CORM-2. Inset shows the linear relationship of the PL intensity ratios (I_{546}/I_{710}) as a function of the concentration of CORM-2. $\lambda_{ex} = 465$ nm.



Figure S15. (A) Absorption spectrum of deoxy-Mb and MbCO before and after interaction of myoglobin with the different concentrations of CORM-2. (B) Determination of the amount of MbCO formed after interaction of deoxy-Mb with various concentrations of CORM-2.



Figure S16. Variations of PL intensity ratios (I_{546}/I_{710}) of BTCV-CO (5 µM) + PdCl₂ (5 µM) after incubation with 100 µM CORM-2 or other biologically-relevant species. a: AcO⁻; b: ClO⁻; c: Glu; d: Cys; e: GSH; f: H₂O₂; g: H₂S; h: Hcy; i: HNO; j: Ile; k: Leu; l: Trp; m: Br⁻; n: Cl⁻; o: F⁻; p: HCO₃⁻; q: NO₂⁻; r: ·OH; s: ONOO⁻; t: PdCl₂; u: Phe; v: ROO·; w: Ser; x: SO₄²⁻; y: CORM-2. λ_{ex} = 465 nm.



Figure S17. Variations of PL intensity ratios (I_{546}/I_{710}) of BTCV-CO + PdCl₂ (black bars) and BTCV-CO + PdCl₂ + CORM-2 (red bars) after incubation with 100 µM other biologically-relevant species. a: AcO⁻; b: Glu c:Cys; d:GSH; e:H₂S; f: Hcy; g: Ile; h: Leu; i: Trp; j: Br⁻; k: Cl⁻; l: F⁻; m: HCO₃⁻; n: NO₂⁻; o: PdCl₂; p: Phe; q: Ser; r: SO₄²⁻. Concentrations: BTCV-CO and PdCl₂: 5 µM; CORM-2: 100 µM. λ_{ex} = 465 nm..



Figure S18. The PL intensity ratios (I_{546}/I_{710}) of BTCV-CO (5 μ M, black bars) and BTCV-CO (5 μ M) + PdCl₂ (5 μ M) + CORM-2 (100 μ M, red bars) in different pH buffers. λ_{ex} = 465 nm.



Figure S19. Cell viability of MCF-7 cells at varied concentrations of BTCV-CO by using standard MTT method.



Figure S20. The PL intensity ratios $(I_{\text{green}}/I_{\text{red}})$ of the green channel and red channel in Figure 3.



Figure S21. Time-dependent fluorescent images in live mice using probe BTCV-CO. BTCV-CO (20 nM) and PdCl₂ (20 nM) was injected in an intraperitoneal manner, followed by an injection of PBS (30 μ L). Fluorescent emissions were collected from 580 to 700 nm. $\lambda_{ex} = 523$ nm.



Figure S22. Time-dependent fluorescent images of CO in live mice using probe BTCV-CO. BTCV-CO (20 nM) and PdCl₂ (20 nM) was injected, followed by an injection of CORM-2 (150 nM). Fluorescent emissions were collected from 580 to 700 nm. $\lambda_{ex} = 523$ nm.



Figure S23. Time-dependent changes of average PL intensity (A) and PL intensity ratio $((I-I_0)/I_0)$ (B) in Figure S18 and S19.

References	Response mode	Response time	Detection limit	λ _{em}	Imaging application
Angew. Chem. Int. Ed. 51 (2012) 9652	Turn-on (ACQ)	10 min	0.5 μΜ	528	cell
J. Am. Chem. Soc. 134 (2012) 15668	Turn-on (ACQ)	60 min	1 µM	507	cell
Chem. Sci. 5 (2014) 3439	Turn-on (ACQ)	40 min	0.653 μM	477	cell/tissue
Chem. Commun. 51 (2015) 4410	Turn-on (ACQ)	30 min	8.49 nM	450	cell
Tetrahedron Lett. 26 (2016) 2927	Turn-on (ACQ)	30 min	26.3 nM	549	cell
RSC Adv. 6 (2016) 65373	Turn-on (ACQ)	45 min	0.127 μΜ	670	cell
Anal. Chem. 88 (2016) 11154	Turn-on (ACQ)	30 min	0.72 μΜ	512	cell
Anal. Chem. 88 (2016) 10648	Turn-on (ACQ)	15 min	37 nM	520	cell
Anal. Chem. 89 (2017) 3754	Turn-on (ACQ)	20 min	1: 46 nM 2: 29 nM	1: 516 2: 527	cell
Sens. Actuators B Chem. 240 (2017) 625	Turn-on (ACQ)	15 min	25 nM	496	cell
J. Am. Chem. Soc. 139 (2017) 18484	Turn-on (ACQ)	ND	ND	450	cell/ex vivo
Angew. Chem. Int. Ed. 56 (2017) 13489	Turn-on (ACQ)	30 min	0.5 nM	660	cell/zebrafish/tissue/ organ/mice
Anal. Chem. 90 (2018) 2933	Turn-on (ACQ)	45 min	0.6 μΜ	525	cell
ACS sens. 3 (2018) 285	Turn-on (ACQ)	20 min	0.06 µM	415	cell
Spectrochim. Acta A 202 (2018) 284	Turn-on (ACQ)	40 min	57 nM	685	cell
Talanta 188 (2018) 691	Turn-on (ACQ)	18 min	0.17 μΜ	736	cell/tissue/mice
Sens. Actuators B Chem. 255 (2018) 2314	Turn-on (ACQ)	3 min	3.2 nM	714	cell/mice
New J. Chem. 42 (2018) 13497	Turn-on (ACQ)	45 min	123 nM	520	cell
Anal. Chem. 90 (2018) 7117	Turn-on (ACQ)	40 min	0.18 μΜ	550	cell/zebrafish
Anal. Methods, 11 (2019) 288	Turn-on (ACQ)	20 min	38.9 nM	630	cell
Chem. Sci. 10 (2019) 320	Turn-on (ACQ)	30 min	0.23 μΜ	650	cell/tissue/mice
Chem. Eur. J. 25 (2019) 2069	Turn-on (ACQ)	ND	ND	450	cell
Sens. Actuators B Chem. 291 (2019) 329	Turn-on (ACQ)	10 min	6.1 nM	665	cell/ zebrafish
Sens. Actuators B Chem. 251 (2017) 389	Ratiometric (ACQ)	20 min	58 nM	472/545	cell
New J. Chem. 42 (2018) 14417	Ratiometric (ACQ)	25 min	17.9 nM	545/455	cell
Anal. Chem. 91 (2019) 2939	Ratiometric (ACQ)	60 min	0.44 μM	540/610	cell/tissue/D. magna
This work	Ratiometric (AIE)	20 min	30.8 nM	564/710	cell/mice

Table 51. Comparison of some multication probes for CO detection	Tab	ole S1	. Com	parison	of	some	fluorescent	probes	for	CO	detectio
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