

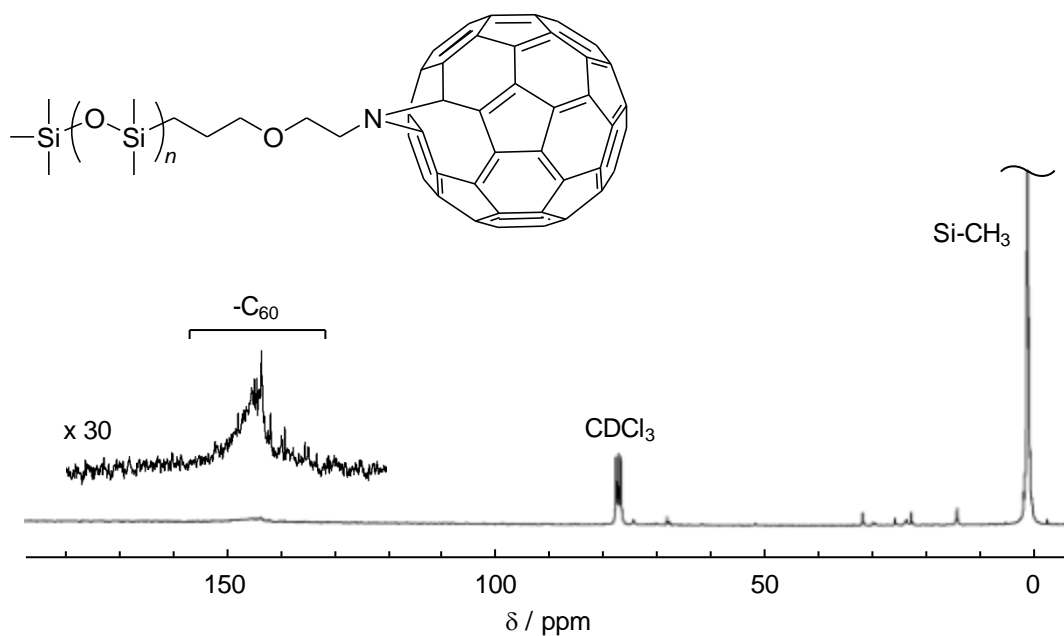
<Supporting Information>

**Polymer Alloying of Helical Syndiotactic PMMA with [60]Fullerene-End-Capped Polymers through Inclusion Complex Formation of the Helical Cavity with Fullerene Units**

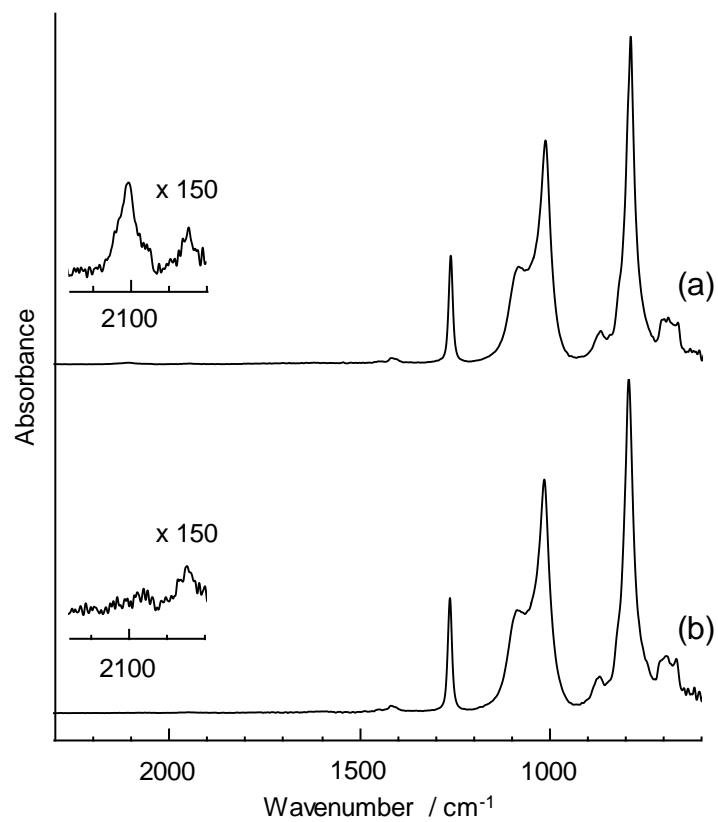
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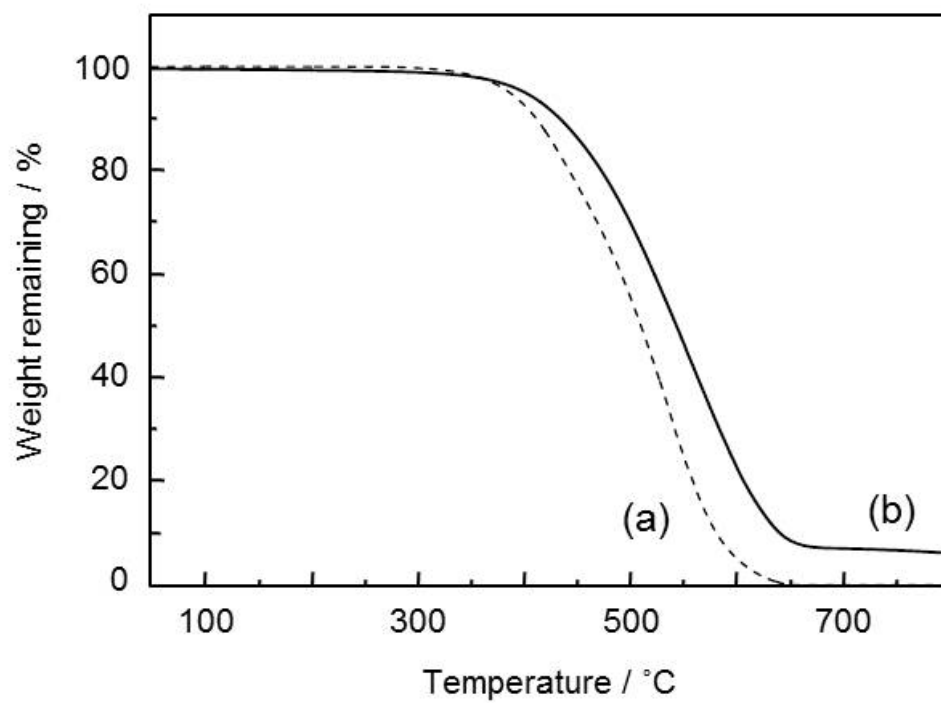
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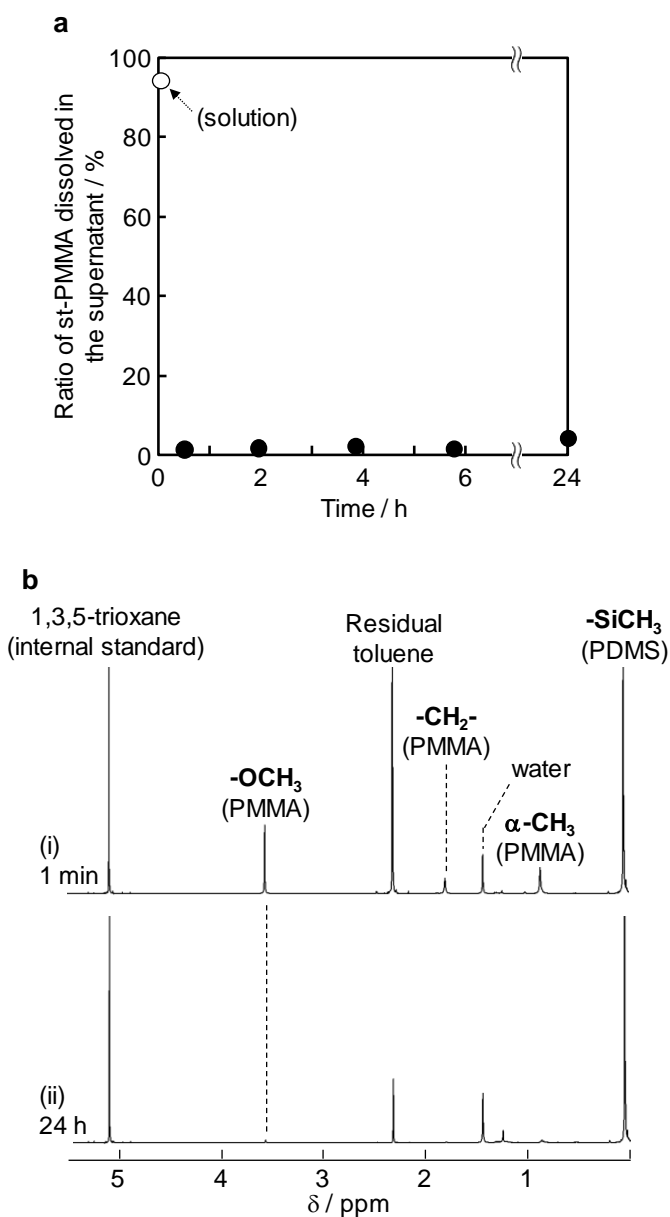
**Figure S1.** <sup>13</sup>C NMR spectrum of as-prepared PDMS-C<sub>60</sub> in CDCl<sub>3</sub> at 25 °C.



**Figure S2.** IR spectra of PDMS-N<sub>3</sub> (a) and as-prepared PDMS-C<sub>60</sub> (b).



**Figure S3.** TGA thermograms of PDMS-OH (a, dotted line) and as-prepared PDMS-C<sub>60</sub> (b, solid line).



**Figure S4.** (a) Changes in the ratio of st-PMMA dissolved in the supernatant toluene layer with time. The st-PMMA (5 mg) was dissolved in the toluene solution of as-prepared PDMS-C<sub>60</sub> ( $f_{\text{C60}} = 42\%$ , 10 mg/mL, 1 mL) with heating at 110 °C. The solution was then stored in an incubator at 25 °C to form the st-PMMA/PDMS-C<sub>60</sub> complex gel. After predetermined time, the toluene layer of 300  $\mu\text{L}$  was sampled. The sample was dried under vacuum and then subjected to NMR measurement using 1,3,5-trioxane as the internal standard to determine the amount of the residual st-PMMA. (b)  $^1\text{H}$  NMR spectra of the toluene layer obtained from the st-PMMA/PDMS-C<sub>60</sub> gel stored at room temperature for 1 min (i) and 24 h (ii).