

## **Supporting Information**

# **Template-Free, Self-doped Approach to Porous Carbon Spheres with High N/O Contents for High-Performance Supercapacitors**

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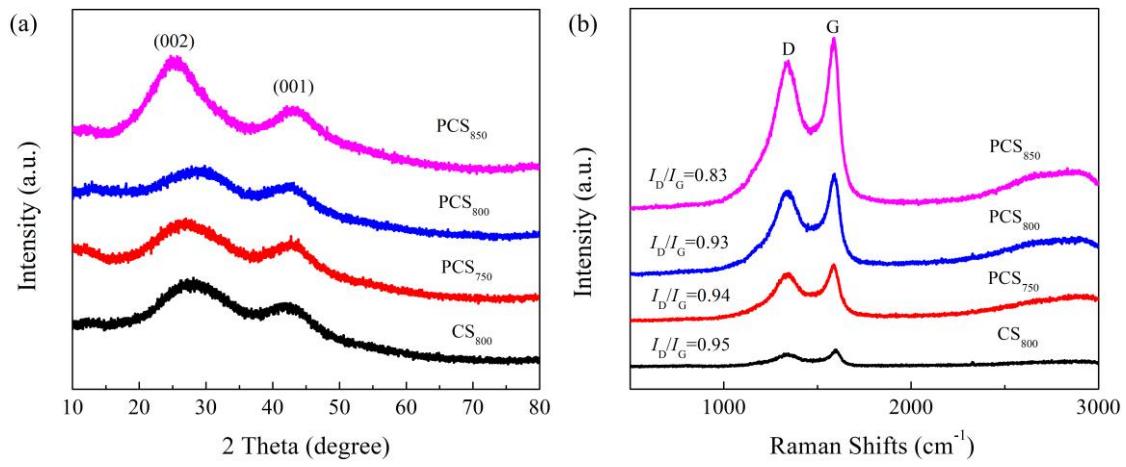
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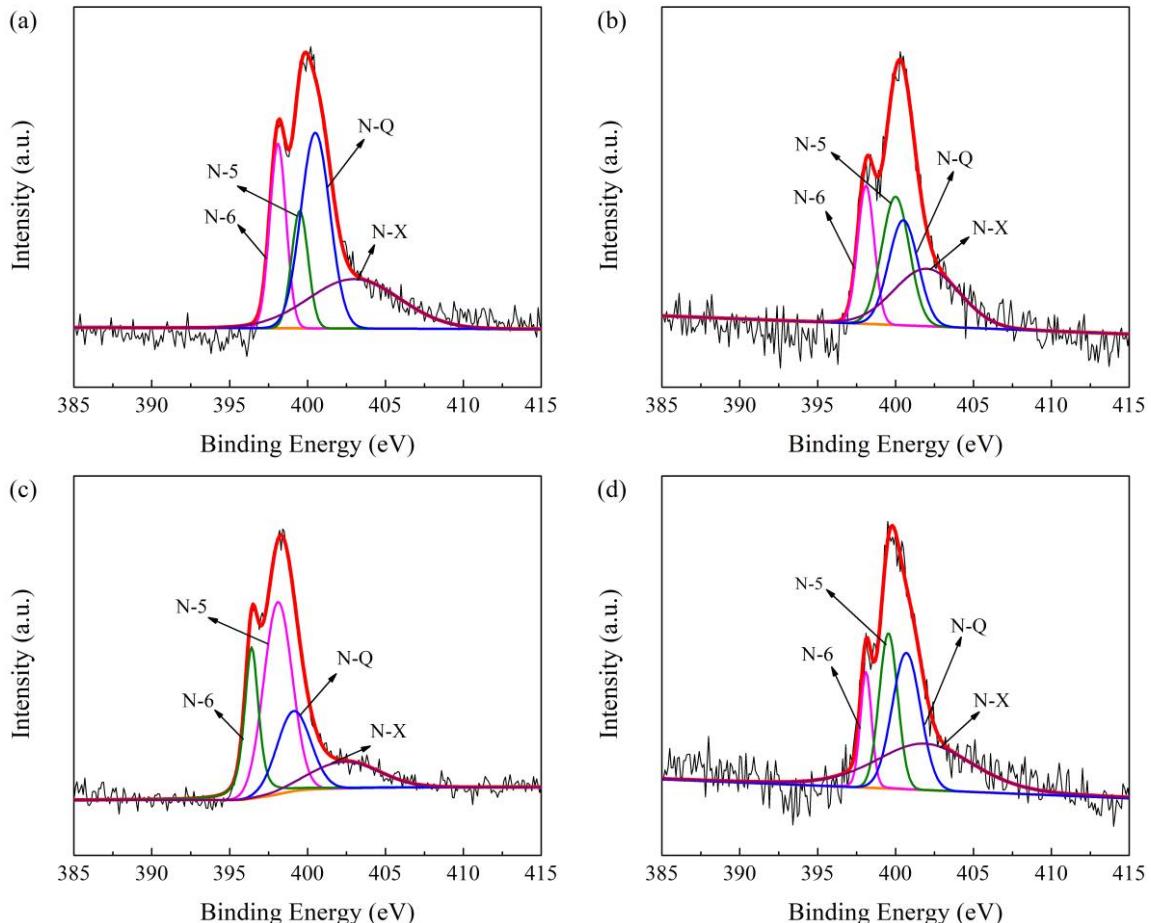
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The supporting information contains 10 pages, 6 Figures and 4 Tables.



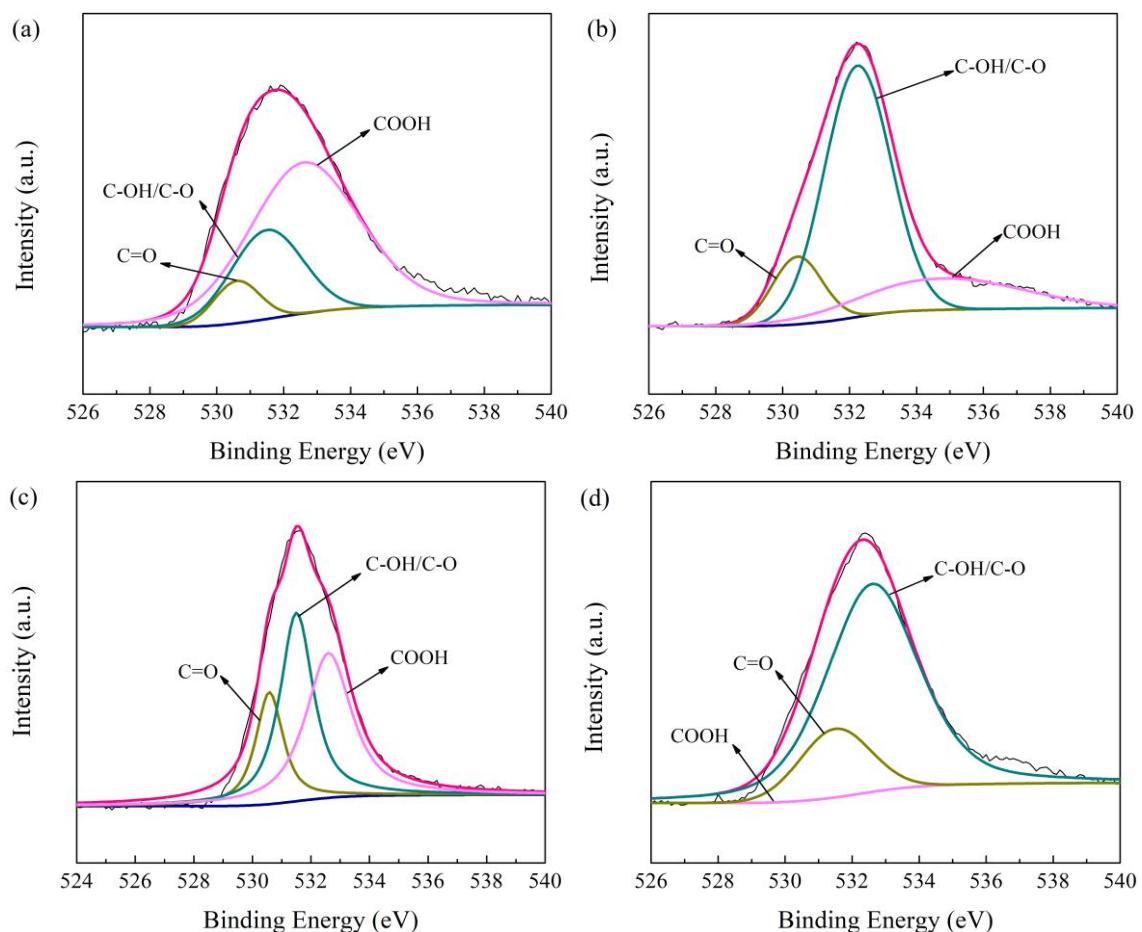
**Figure S1.** XRD patterns (a) and Raman spectra (b) of CS<sub>800</sub> and PCSSs.



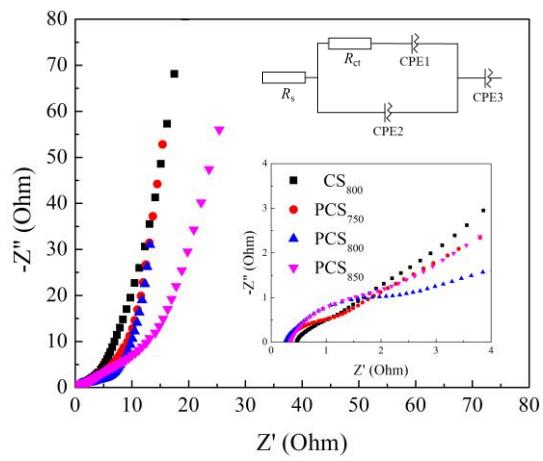
**Figure S2.** N1s high-resolution spectra of CS<sub>800</sub> (a), PCS<sub>750</sub> (b), PCS<sub>800</sub> (c) and PCS<sub>850</sub> (d).

**Table S1.** Summary of XPS analysis of PCSSs.

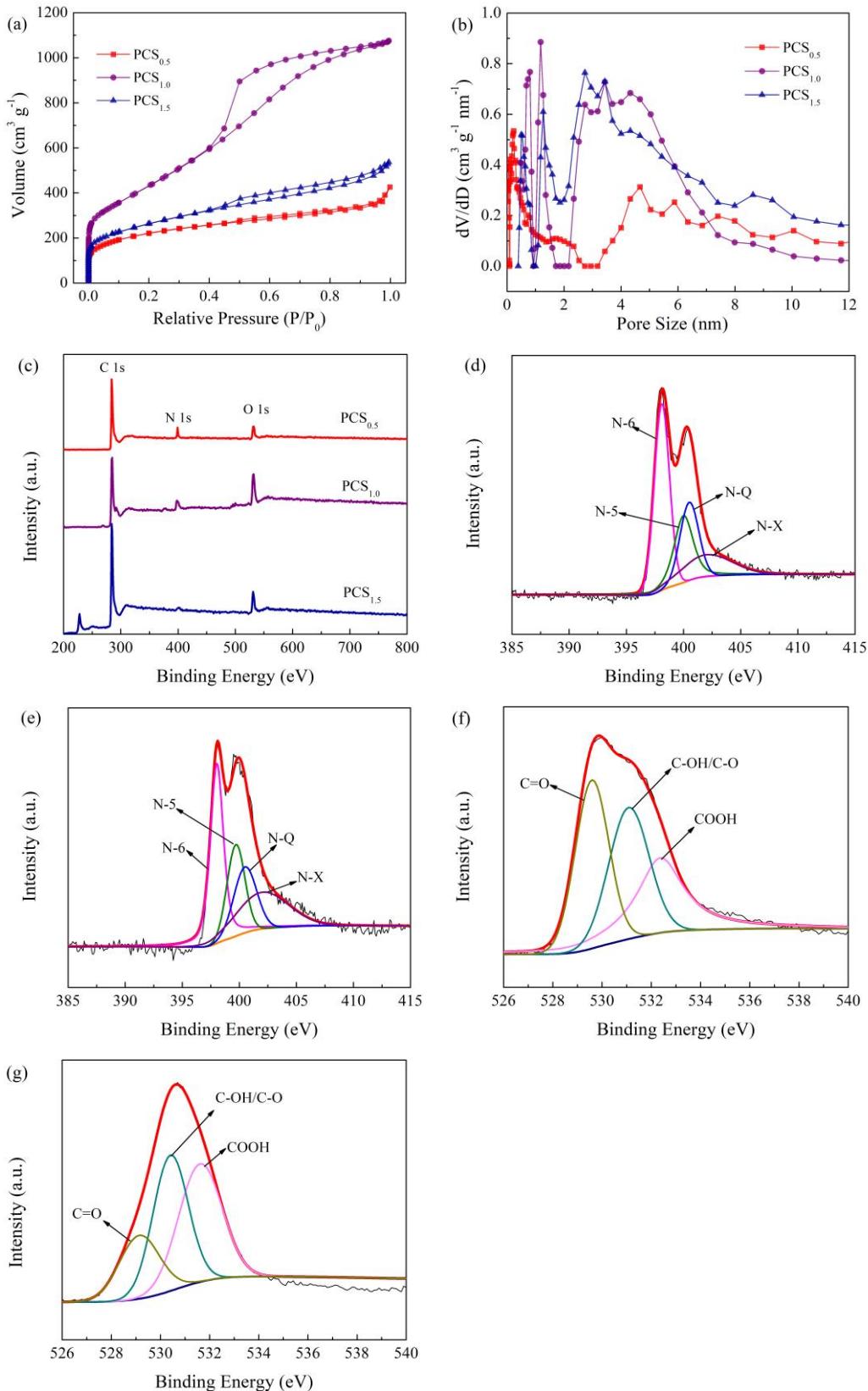
Samples	The relative intensity of the dopant species (%)			
	N-6	N-5	N-Q	N-X
CS <sub>800</sub>	21.5	15.7	48.1	14.7
PCS <sub>750</sub>	34.7	31.7	21.8	11.8
PCS <sub>800</sub>	30.9	41.0	13.3	14.8
PCS <sub>850</sub>	14.4	34.2	25.2	26.2



**Figure S3.** O1s high-resolution spectra of CS<sub>800</sub> (a), PCS<sub>750</sub> (b), PCS<sub>800</sub> (c) and PCS<sub>850</sub> (d).



**Figure S4.** The equivalent circuit model of electrode, Nyquist plots of CS<sub>800</sub> and PCSs electrodes in 6 M KOH electrolyte.

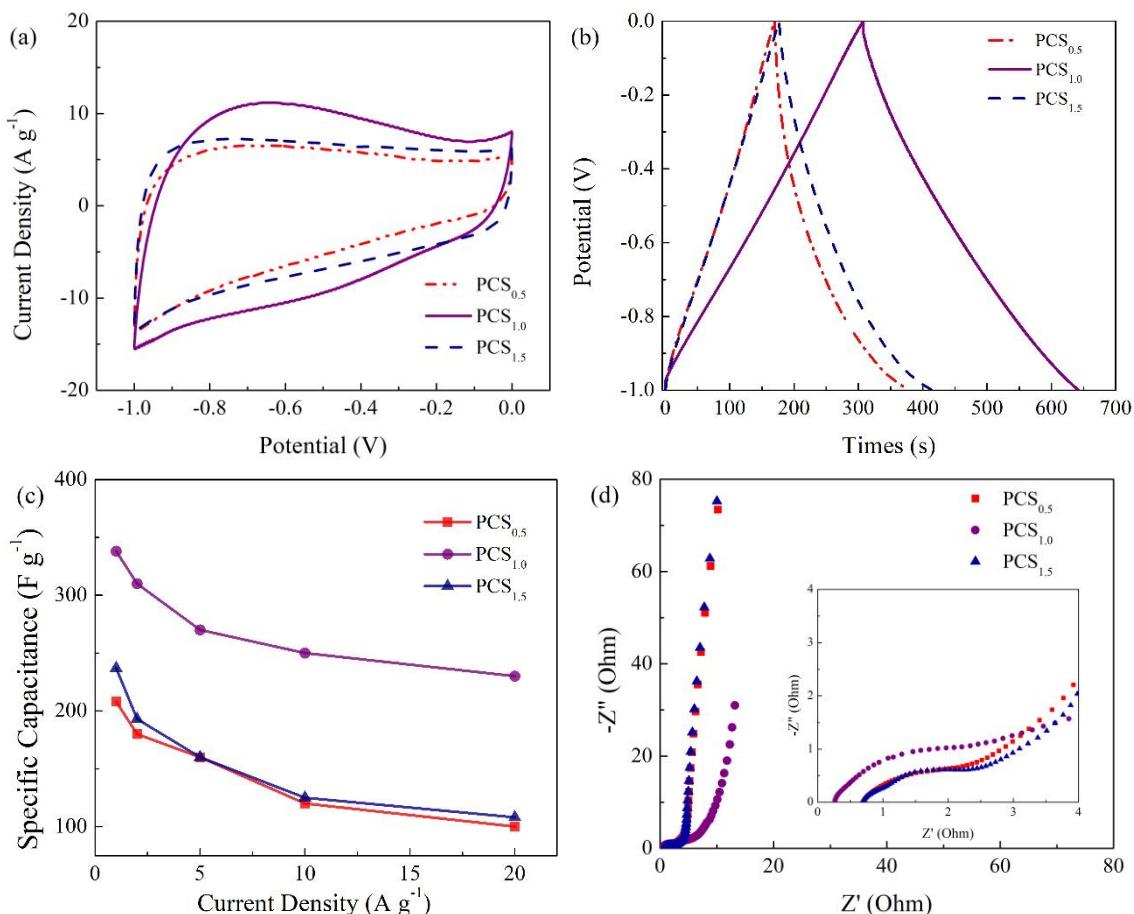


**Figure S5.** Nitrogen sorption isotherms (a), and the curves of pore size distribution (b) of PCSSs. Wide-scan XPS data of PCSSs (c). N1s and O1s high-resolution spectra of PCSSs (d–g).

**Table S2.** Porosity Parameters and Surface Element Contents of PCSs.<sup>a</sup>

Samples	$S_{\text{BET}}$ ( $\text{m}^2 \text{ g}^{-1}$ )	$S_{\text{micro}}$ ( $\text{m}^2 \text{ g}^{-1}$ )	$S_{\text{meso}}$ ( $\text{m}^2 \text{ g}^{-1}$ )	$V_{\text{total}}$ ( $\text{cm}^3 \text{ g}^{-1}$ )	C (wt %)	N (wt %)	O (wt %)
PCS <sub>0.5</sub>	657	404	253	0.97	83.16	9.11	7.73
PCS <sub>1.0</sub>	1321	680	641	1.66	82.88	7.97	10.16
PCS <sub>1.5</sub>	768	406	362	0.77	88.9	2.45	8.65

<sup>a</sup> $S_{\text{BET}}$ ,  $S_{\text{micro}}$  and  $S_{\text{meso}}$ , specific surface area, micropore surface area and mesopore surface area;  $V_{\text{total}}$ , the total pore volume.



**Figure S6.** CV curves of PCSs electrodes at  $50 \text{ mV s}^{-1}$  (a), GCD curves of PCSs electrodes at  $1.0 \text{ A g}^{-1}$  (b), capacitances vs current densities of the carbon-based electrodes (c), and Nyquist plots of PCSs electrodes (d) in 6 M KOH electrolyte.

**Table S3. Electrochemical performance of carbon based KOH-electrolyte symmetric supercapacitors.**

Materials	Current Density (A g <sup>-1</sup> )	Device Capacitance (F g <sup>-1</sup> )	Energy Density (Wh kg <sup>-1</sup> )	Power Density (W kg <sup>-1</sup> )	Ref.
ZTC-P300	1.25	216	7.5	625	1
800AC	0.3	105	3.0	220	2
PNC-900	0.1	236	8.2	250	3
NOCS-1/10	0.5	130	4.3	250	4
a-CNS/EG-10	0.3	210	7.3	500	5
ACG-200	1.0	162	7.5	200	6
200-HTC-800-3	0.4	239	8.11	400	7
N-OMCN@GN	1	242.3	6.68	250	8
MOLC	0.5	211	3.85	27.7	9
CDC	1	173	—	—	10
a-CA	0.5	198.4	6.85	400	11
CNT	1	7.7	—	—	12
PCS <sub>800</sub>	0.5	267	9.23	400	This work

**Table S4. Electrochemical performance of carbon based Na<sub>2</sub>SO<sub>4</sub>-electrolyte symmetric supercapacitors.**

Materials	Current Density (A g <sup>-1</sup> )	Device Capacitance (F g <sup>-1</sup> )	Energy Density (Wh kg <sup>-1</sup> )	Power Density (W kg <sup>-1</sup> )
PCF-700	0.5	32	13.9	460
WJC-800	0.5	44	18.6	400
G/CNTs-200	0.2	33	8	900
IPC <sub>2-0.2-8</sub>	0.5	50	21.9	461
BNDC	0.2	45	22.7	200
HPCs	0.5	30	14.2	444.5
PCACM	0.5	44.5	10.42	8928
PCS <sub>800</sub>	0.5	74.2	33.37	450

## Reference

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