

Supporting information

Synthesis of highly stable Pd@CeO₂ catalyst for methane combustion with the synergistic effect of urea and citric acid

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1. Synthesis of I-Pd/CeO₂-*T* catalyst by impregnation method

1.1. Synthesis of catalyst carriers

In a typical synthesis, 0.21 g of cerium nitrate, 0.03 g of urea and 0.051 g of citric acid were successively dissolved in a water/ethanol mixture with a volume ratio of 7:1 at room temperature. The resultant mixture was vigorously stirred for 20 min, then 0.3 mL of hydrogen peroxide solution was added. Afterward, the final solution was sealed into an autoclave with a Teflon-lined, heated to 120 °C and maintained for 24 h. The as-prepared product was centrifuged, washed with deionized water and ethanol repeatedly until pH \approx 7, and subsequently dried at 60 °C for 7 h. The product was further calcined at 400 °C with a ramping rate of 1 °C min⁻¹ and held for 2 h. After naturally cooling to room temperature, the catalyst carriers CeO₂ was obtained.

1.2. Synthesis of catalysts

Taking the as-synthesized CeO₂ as a carrier, and Pd(NO₃)₂ solution as a precursor, the supported Pd/CeO₂ catalysts with a constant Pd loading of 1 wt% (theoretic value) were prepared by the incipient wetness impregnation method at room temperature. The product was calcined at 400 °C for 4 h (1 °C min⁻¹ ramping rate), then calcined at 600 °C or 1000 °C for 2 h at the same speed. The prepared catalysts were referred to I-Pd/CeO₂-*T*, where *T* represented the calcination temperature (*T* = 600, 1000 °C).

2. Synthesis of C-Pd/CeO₂-*T* catalyst by coprecipitation method

In a typical synthesis, 2.0 g of cerium nitrate and a specific amount of Pd(NO₃)₂ solution with a theoretic Pd loading of 1 wt% were mixed and stirred for 20 min at room temperature. The obtained solution and NaOH solution were added into the distilled water, adjusting the final pH to 11. The as-prepared product was centrifuged, washed with deionized water and ethanol,

and then dried at 60 °C for 7 h. The product was calcined at 400 °C for 4 h (1 °C min⁻¹ ramping rate), then calcined at 600 °C or 1000 °C for 2 h with the same heating rate. The prepared samples were denoted as C-Pd/CeO₂-*T*, where *T* represented the calcination temperature (*T* = 600, 1000 °C).

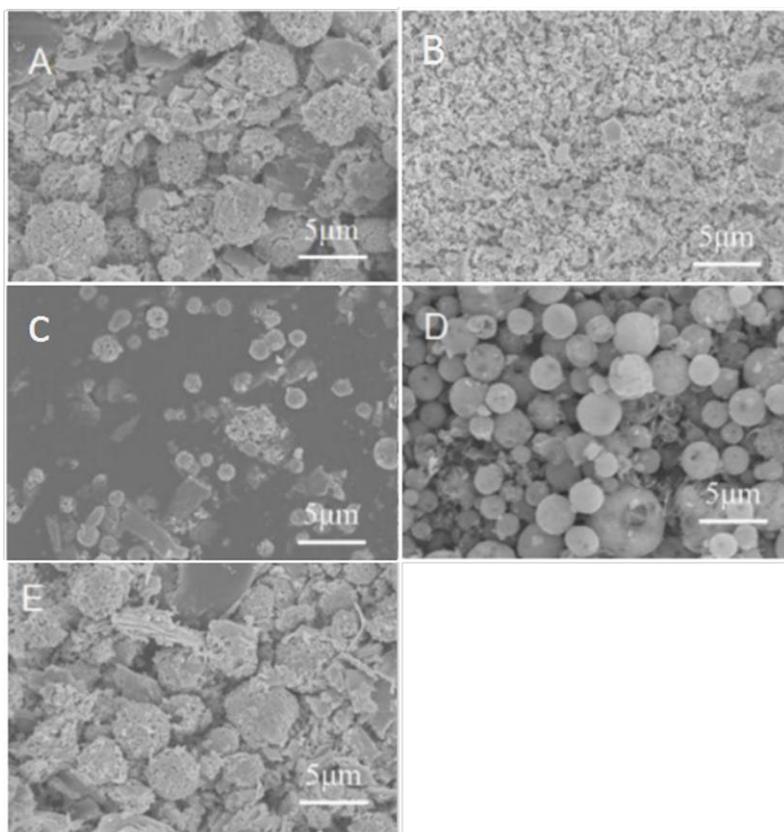


Figure S1. SEM images of H-Pd@CeO₂-600(t) catalysts (A–E: t = 6, 12, 18, 24, 30).

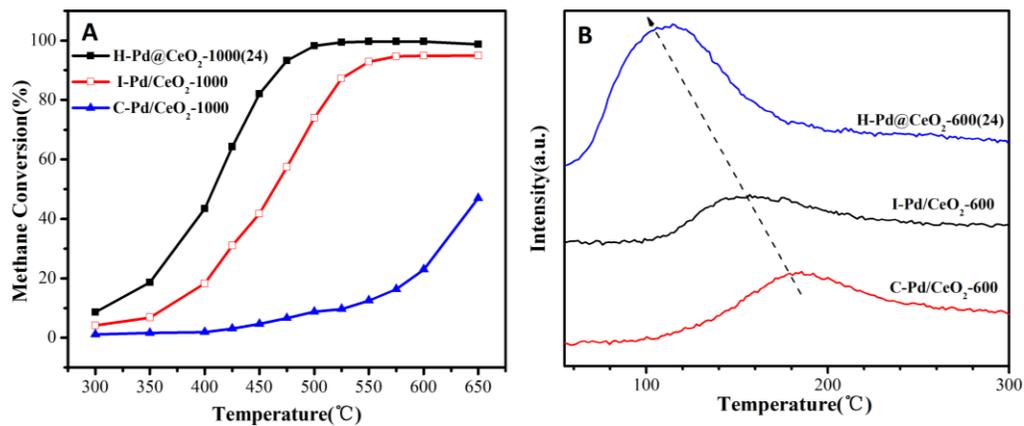


Figure S2. (A) Methane conversion and (B) CO₂-TPD curves of H-Pd@CeO₂-600(24), I-Pd/CeO₂-600 and C-Pd/CeO₂-600.

Table S1 CO pulse adsorption results of the catalysts

Catalyst	Pd Dispersion(%)	Pd particle size (nm)
H-Pd/CeO ₂ -600(6)	5.34	21.0
H-Pd/CeO ₂ -600(12)	42.4	2.6
H-Pd/CeO ₂ -600(18)	40.3	2.8
H-Pd@CeO ₂ -600(24)	43.4	2.6
H-Pd/CeO ₂ -600(30)	42.5	2.6
I-Pd/CeO ₂ -600	16.1	7.0
C-Pd/CeO ₂ -600	6.1	18.6

Table S2 The relevant parameters of the catalysts obtained from XPS results

Catalyst	Pd/Ce atomic ratio	Ce ³⁺ in Ce(%)	Ce ⁴⁺ /Ce ³⁺ atomic ratio
H-Pd/CeO ₂ -600(6)	0.0189	5.74	16.4
H-Pd/CeO ₂ -600(12)	0.0212	4.48	21.3
H-Pd/CeO ₂ -600(18)	0.0145	5.51	17.1
H-Pd@CeO ₂ -600(24)	0.0138	12.6	6.9
H-Pd/CeO ₂ -600(30)	0.0214	12.5	7.0