

Synthesis of 2-Hydroxydodecyl Starch Ethers: Importance of the Purification Process

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I) Characterizations

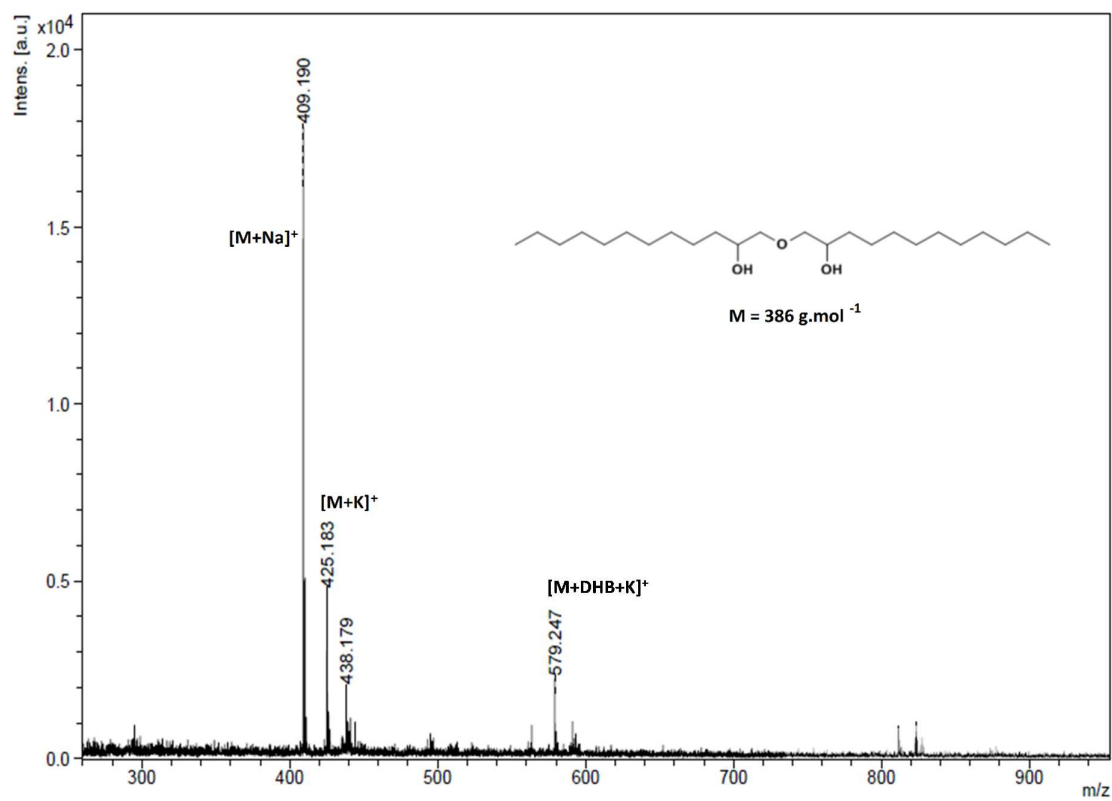


Figure S1: MALDI-TOF spectra of bis-(2-hydroxydodecyl)oxide (positive mode, DHB matrix; DHB=2,5-dihydroxybenzoic acid (154 g.mol^{-1}))

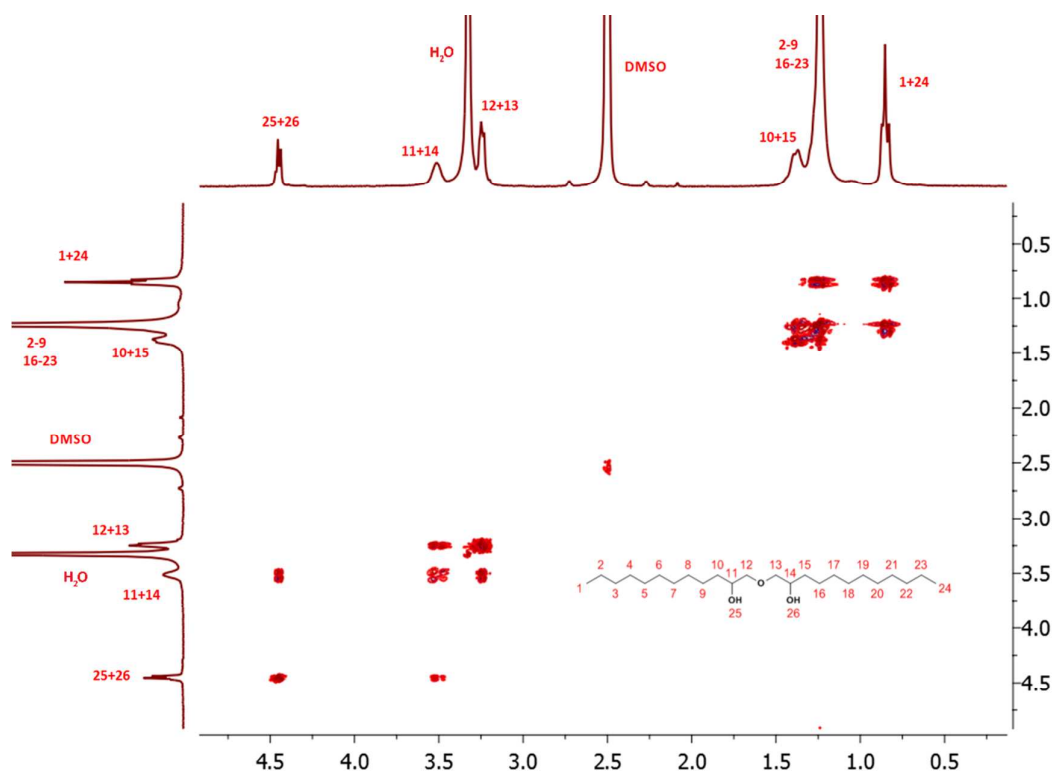


Figure S2: COSY spectrum of bis-(2-hydroxydodecyl)oxide (300 MHz, DMSO, 25°C)

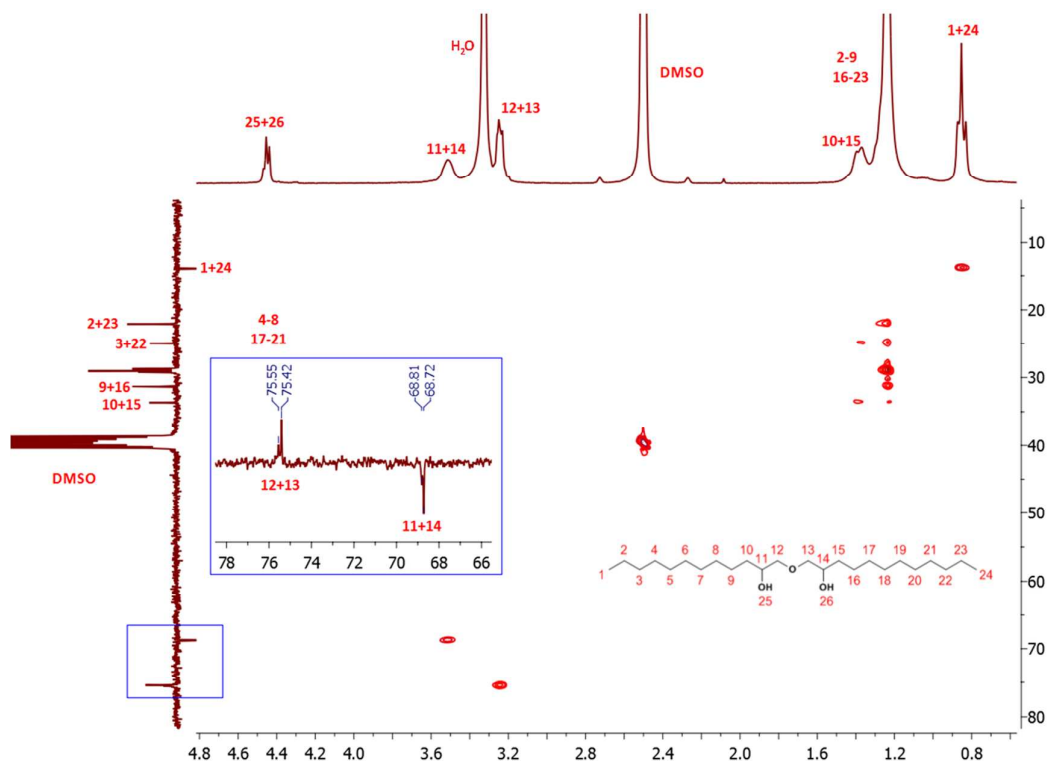


Figure S3: HSQC spectrum of bis-(2-hydroxydodecyl)oxide (300 MHz, DMSO, 25°C)

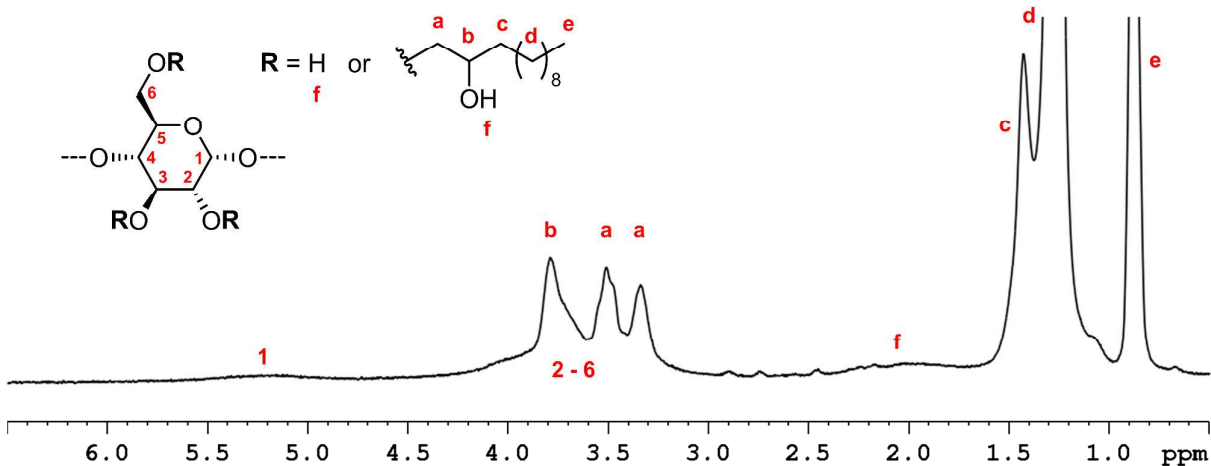


Figure S4: ^1H -NMR analysis of the product obtained after evaporation of the filtrate resulting from the Soxhlet extractions done on the SE-G sample (300 MHz, CDCl_3 , 25°C)

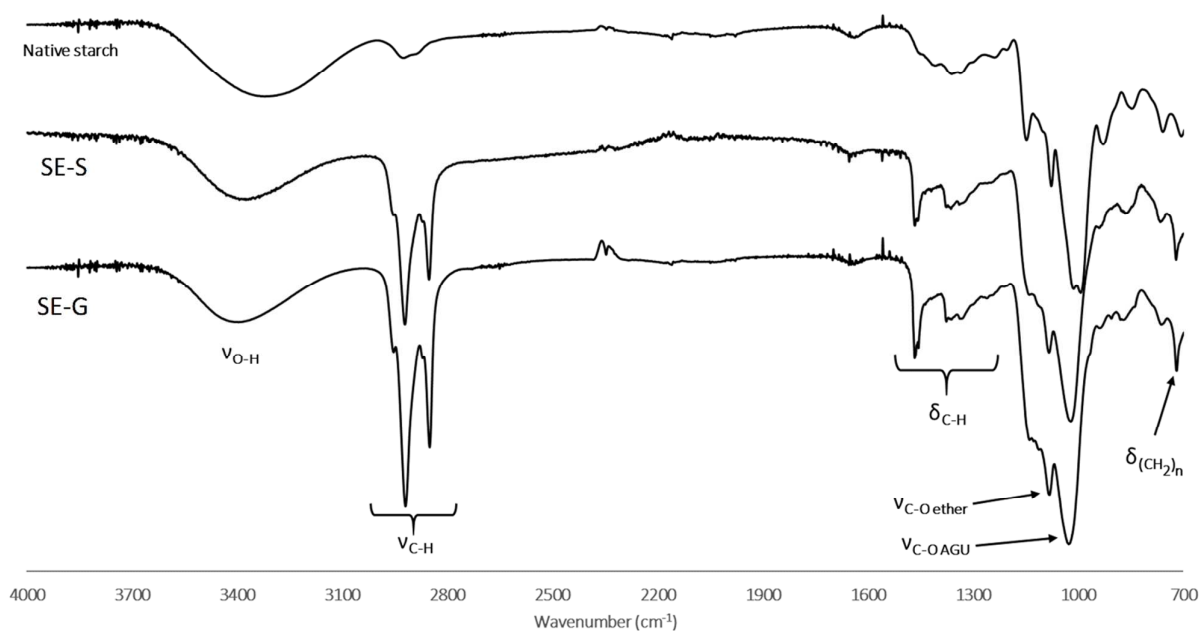


Figure S5: FTIR spectra of native starch and modified starch purified by Soxhlet (SE-S) and purified by grinding (SE-G)

II) Determination of the molar substitution (MS) of the 2-hydroxydodecyl potato starches by elemental analysis

The molar substitution of the 2-hydroxydodecyl potato starches (**MS**) was defined as the average number of 2-hydroxydodecyl groups per anhydroglucose unit in the modified starches.

1) Determination of the molar substitution deduced from the %C in the 2-hydroxydodecyl starches: $MS_{(C)}$

The expression of the $MS_{(C)}$, i.e. the number of 2-hydroxydodecyl groups (HDo) per AGU determined by elemental analysis from wt% of C (%C) in the 2-hydroxydodecyl starches, can be easily demonstrated:

$$\%C = \frac{\text{Mass of C in a modified AGU (g. mol}^{-1}\text{)}}{\text{Mass of a modified AGU (g. mol}^{-1}\text{)}} \times 100$$

$$\%C = \frac{\text{Mass of C in an unmodified AGU (g. mol}^{-1}\text{)} + \text{Mass of C due to the HDo graft presence (g. mol}^{-1}\text{)}}{\text{Mass of a unmodified AGU (g. mol}^{-1}\text{)} + \text{Mass coming from the HDo graft presence (g. mol}^{-1}\text{)}} \times 100$$

$$\%C = \frac{6 M_C + MS \times 12 M_C}{M_{AGU} + MS \times (M_{HDo} - M_H)} \times 100$$

M_C , M_H , M_{AGU} and M_{HDo} = Molecular mass (g.mol⁻¹) of C, H, AGU (C₆H₁₀O₅) and 2-hydroxydodecyl graft (C₁₂H₂₅O)

$$M_C = 12.010736 \text{ g.mol}^{-1}$$

$$M_H = 1.007941 \text{ g.mol}^{-1}$$

$$M_{AGU} = 162.140600 \text{ g.mol}^{-1}$$

$$M_{HDo} = 185.326300 \text{ g.mol}^{-1}$$

$$\%C = \frac{600 M_C + MS \times 1200 M_C}{M_{AGU} + MS \times (M_{HDo} - M_H)}$$

$$MS \times [(M_{HDo} - M_H) \times \%C - 1200 M_C] = 600 M_C - \%C \times M_{AGU}$$

and finally

$$MS_{(C)} = \frac{\%C \times M_{AGU} - 600 M_C}{1200 M_C - \%C \times (M_{HDo} - M_H)}$$

2) Determination of the molar substitution deduced from the %H in the 2-hydroxydodecyl starches: $MS_{(H)}$

The expression of the $MS_{(H)}$, i.e. the number of 2-hydroxydodecyl (HDo) groups per AGU determined by elemental analysis from wt% of H (%H) in the 2-hydroxydodecyl starches, can be easily demonstrated:

$$\%H = \frac{\text{Mass of H in a modified AGU (g. mol}^{-1}\text{)}}{\text{Mass of a modified AGU (g. mol}^{-1}\text{)}} \times 100$$

$$\%H = \frac{\text{Mass of H in an unmodified AGU (g. mol}^{-1}\text{)} + \text{Mass of H due to the HDo graft presence (g. mol}^{-1}\text{)}}{\text{Mass of a unmodified AGU (g. mol}^{-1}\text{)} + \text{Mass coming from the HDo graft presence (g. mol}^{-1}\text{)}} \times 100$$

$$\%H = \frac{10 M_H + MS \times 24 M_H}{M_{AGU} + MS \times (M_{HDo} - M_H)} \times 100$$

$$\%H = \frac{1000 M_H + MS \times 2400 M_H}{M_{AGU} + MS \times (M_{HDo} - M_H)}$$

$$MS \times [(M_{HDo} - M_H) \times \%H - 2400 M_H] = 1000 M_H - \%H \times M_{AGU}$$

and finally

$$MS_{(H)} = \frac{\%H \times M_{AGU} - 1\,000 M_H}{2\,400 M_H - \%H \times (M_{HDo} - M_H)}$$