

A Comprehensive Scope of Peripheral and Axial Substituent Effect on the Spectroelectrochemistry of Boron Subphthalocyanines.

Supporting Information

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Table S1. Extinction Coefficients of Investigated BsubPcs Measured in 1,2-Dichlorobezene.

Compound	Extinction Coefficient ($\epsilon \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$)	
	Soret Band Peak (~300 nm)	Q Band Max Peak (~560 – 590 nm)
PhO-BsubPc (1)	4.20 ± 1.10	8.63 ± 1.28
3F ₁ PhO-BsubPc (2)	4.29 ± 1.17	9.40 ± 1.17
35F ₂ PhO-BsubPc (3)	3.59 ± 1.77	7.30 ± 1.77
246F ₃ PhO-BsubPc (4)	2.89 ± 1.49	5.69 ± 1.49
345F ₃ PhO-BsubPc (5)	4.30 ± 1.42	8.71 ± 1.42
2356F ₄ PhO-BsubPc (6)	5.08 ± 3.10	10.20 ± 3.10
F ₅ BsubPc (7)	4.33 ± 0.82	8.51 ± 0.82
3-MePhO-BsubPc (8)	4.01 ± 1.68	8.93 ± 1.68
PhO-Cl ₆ BsubPc (9)	6.25 ± 4.10	15.6 ± 4.10
PhO-Cl ₁₂ BsubPc (10)	3.86 ± 1.05	10.9 ± 1.10
Ph-BsubPc (11)	6.20 ± 3.55	10.0 ± 3.55
F-BsubPc (12)	4.55 ± 2.72	9.78 ± 2.72
Cl-BsubPc (13)	3.73 ± 1.28	7.35 ± 1.28
Cl-Cl ₆ BsubPc (14)	4.57 ± 1.62	9.99 ± 1.62

Table S2. UV-visible Spectral Data of Neutral and Anion Radicals of Investigated Fluorophenoxy-BsubPcs in CH₂Cl₂ Containing 0.1 M TBAP.^a

Compound	λ_{max} (nm)							
	Neutral			Anion Radical				
PhO-BsubPc (1)	307	513 ^{sh}	561	301	483	513	566	600-800 ^{br}
3F ₁ PhO-BsubPc (2)	307	515 ^{sh}	562	301	483	515	569	600-800 ^{br}
35F ₂ PhO-BsubPc (3)	307	515 ^{sh}	561	302	484	515	564	600-800 ^{br}
246F ₃ PhO-BsubPc (4)	307	515 ^{sh}	561	304	483	515	567	600-800 ^{br}
345F ₃ PhO-BsubPc (5)	307	517 ^{sh}	561	304	488	517	563	600-800 ^{br}
2356F ₄ PhO-BsubPc (6)	307	517 ^{sh}	561	305	488	517	563	600-800 ^{br}
F ₅ BsubPc (7)	307	515 ^{sh}	561	301	483	515	567	600-800 ^{br}

^ash = shoulder peak, br = broad peak

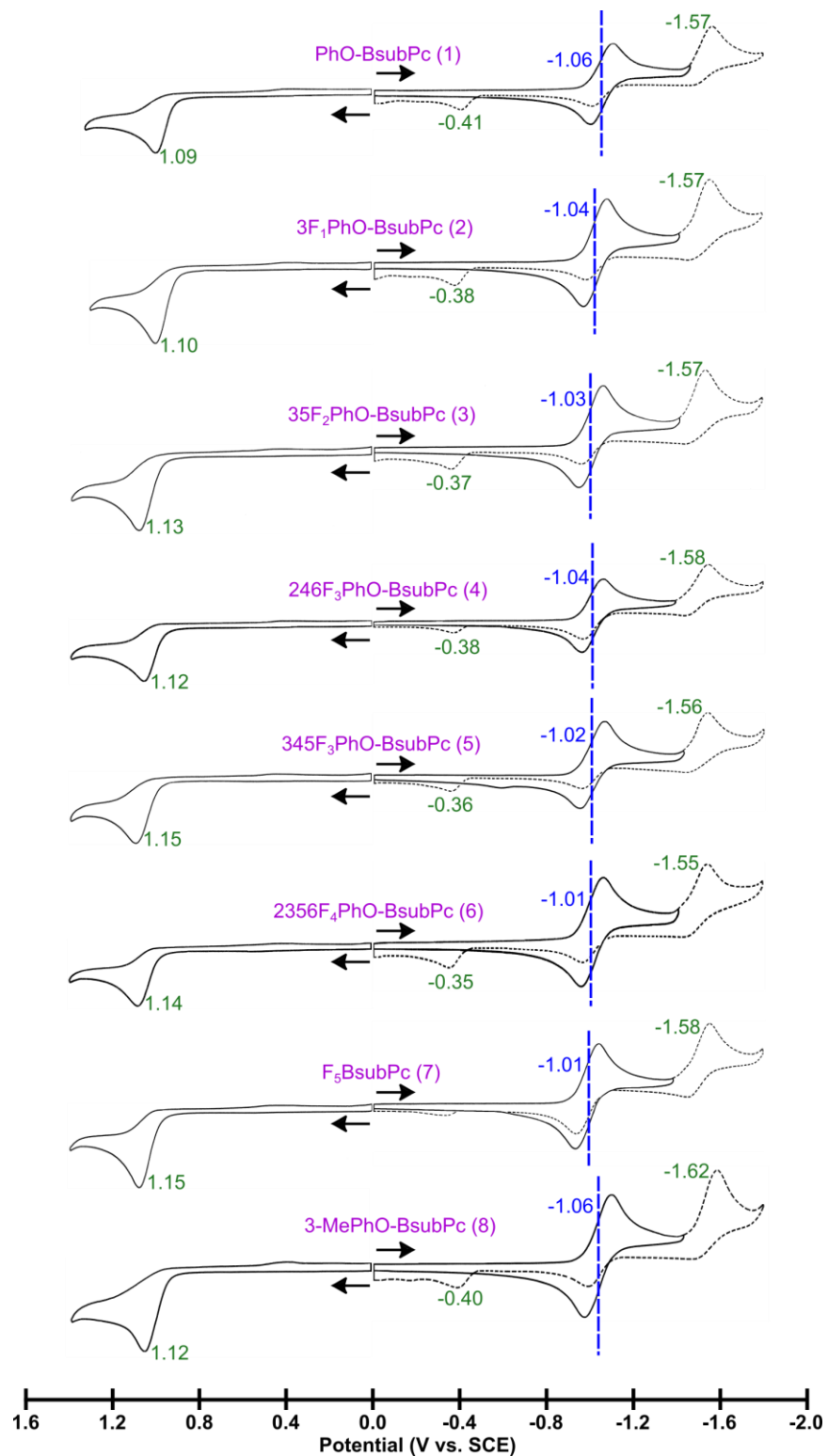


Figure S1. Cyclic voltammograms of axial phenoxy-substituted BsubPcs (PhO-BsubPc, **1**; 3F₁PhO-BsubPc, **2**; 35F₂PhO-BsubPc, **3**; 246F₃PhO-BsubPc, **4**; 345F₃PhO-BsubPc, **5**; 2356F₄PhO-BsubPc, **6**; F₅BsubPc, **7**; and 3-MePhO-BsubPc, **8**) in CH₂Cl₂ containing 0.1 M TBAP.

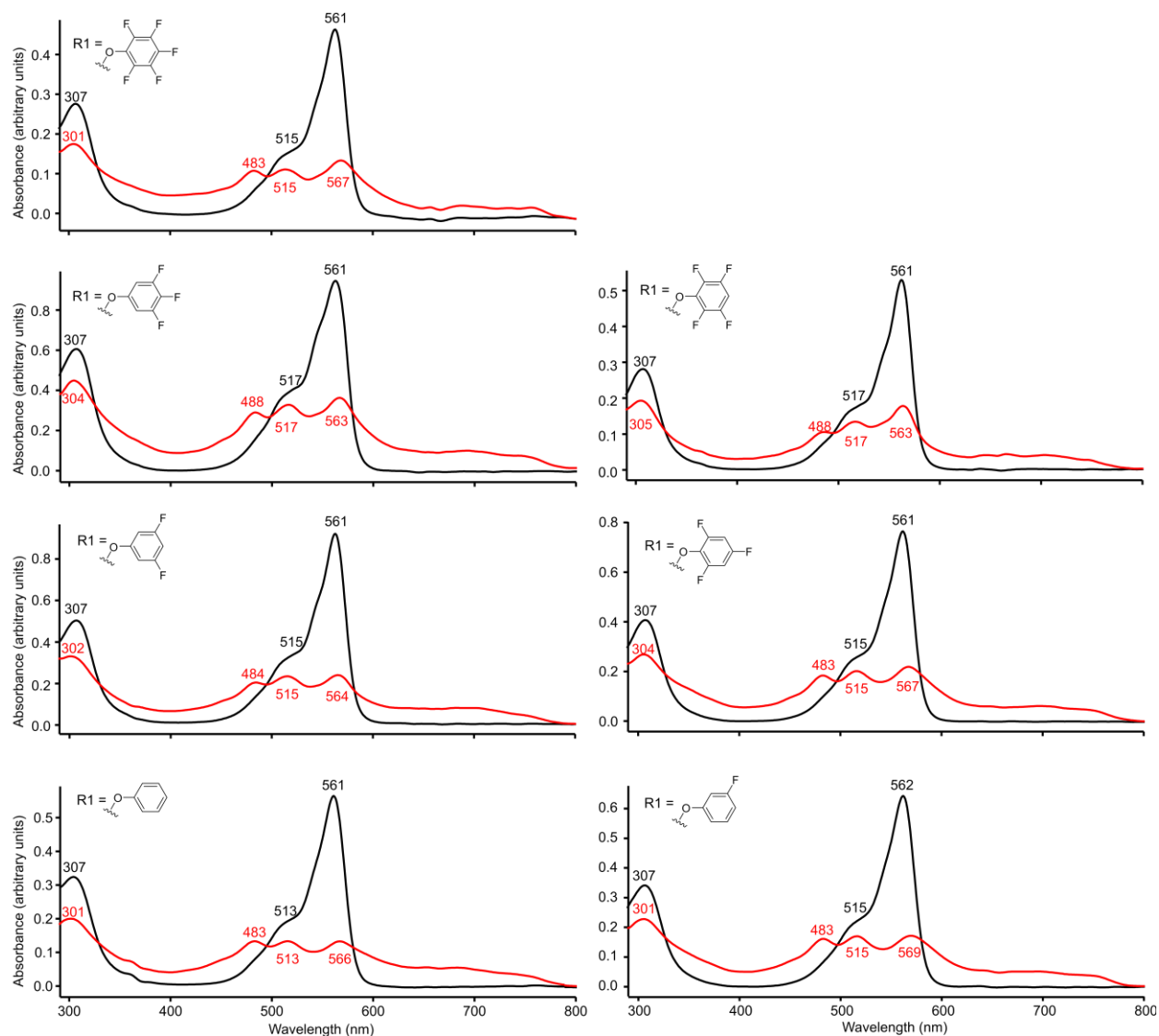


Figure S2. UV-vis spectra of neutral (black line) and singly reduced (red line) fluorophenoxy-BsubPcs at -1.4 V (-1.3 V for F₅BsubPc), where n = 0-5, in CH₂Cl₂ containing 0.1 M TBAP.

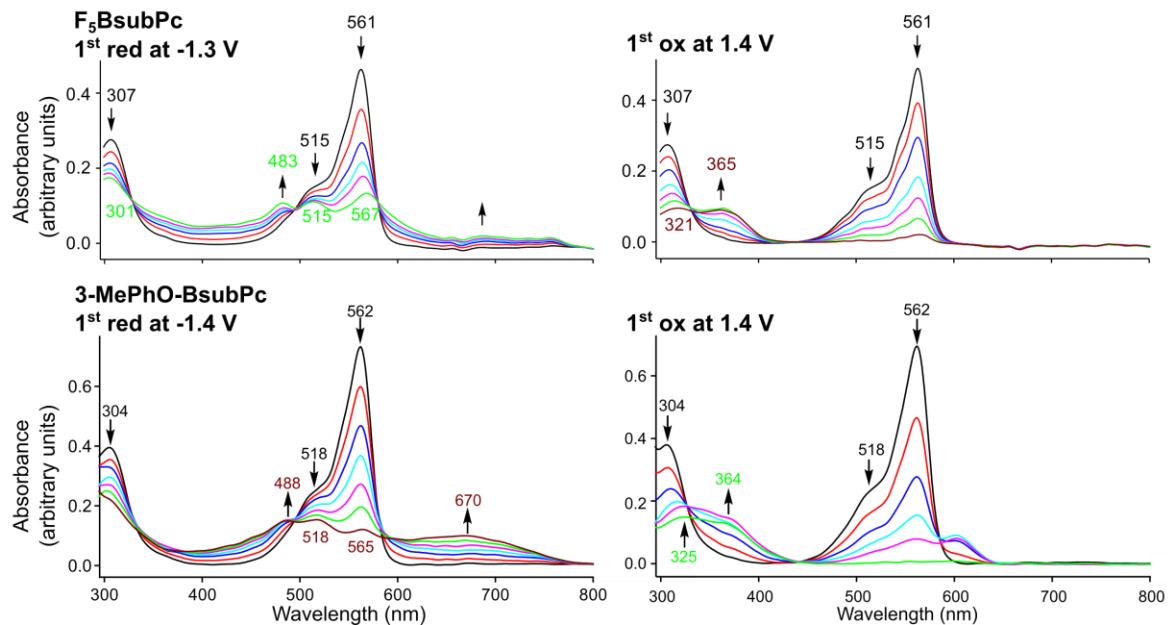


Figure S3. UV-vis spectral changes of $F_5BsubPc$ (top, **7**) and 3-MePhO-BsubPc (bottom, **8**) during first reduction (left spectra) and first oxidation (right spectra) at the respective potentials, in CH_2Cl_2 containing 0.1 M TBAP.