Supporting information

Theoretical model for the Frank elastic moduli in the intercalated SmA_b phase of bent-shaped dimers

<u>Claire Meyer</u>^{1*}, Tatiana Sergan², Vassili Sergan², Daniel Stoenescu³, Patrick Davidson⁴, Anamarija Knežević⁵, Irena Dokli⁵, Andreja Lesac⁵ and Ivan Dozov^{1,4}

 ¹Physique des Systèmes Complexes, Université de Picardie Jules Verne, 80039 Amiens, France
²California State University, Sacramento, 6000 J Street, Sacramento, California 95608, USA
³Optics Department, IMT Atlantique, CS 83818, 29238 Brest cedex, France
⁴ Laboratoire de Physique des Solides, Université Paris-Saclay, CNRS, 91405 Orsay, France
⁵Ruder Bošković Institute, Bijenička 54, 10000 Zagreb, Croatia

Synthesis

General Methods

All reactions were conducted under argon atmosphere unless stated otherwise. THF and Et₂O were dried following standard methods. The commercial grade reagents and solvents were used without further purification. TLC was performed on aluminum-baked silica plates (60 F254, Merck). UV light (254 nm) or phosphomolybdic acid reagent was used for visualizing. Column chromatography was performed on silica gel (Silicagel 60, 70–230 mesh, Merck) or flash silica gel (Silicagel 60, 230-400 mesh, Merck). Transition temperatures were determined from thermograms recorded on Perkin-Elmer Diamond DSC, operated at scanning rates of 5 °C min⁻¹. ¹H and ¹³C NMR spectra were recorded on a Bruker AV 300 and 600 spectrometers in CDCl₃ or d₆-DMSO. Chemical shifts (δ) are given in ppm referenced to TMS or solvent. Coupling constants are given in Hz. Chemical purity and reaction progress were monitored by HPLC on an Agilent 1260 Infinity instrument on an InfinityLab Poroshell 120 EC-C18, 4.6 x 100 mm, 2.7 μ column with DAD detector, 0.5 mL/min flow. CHN analyses were done on Perkin Elmer 2400 Series II CHNS analyser.

Experimental procedures

Scheme 1



1-(6-((*tert*-Butyldimethylsilyl)oxy)naphthalen-2-yl)hexan-1-ol (S2)

Magnesium turnings (510 mg, 20.9 mmol) were suspended in dry diethyl ether (10 mL), flushed with argon, and activated by addition of a single iodine crystal. 1-Bromopentane (1.3 mL, 10.5 mmol) was added, and the reaction mixture was refluxed for 4 h. A solution of 6-((*tert*-butyldimethylsilyl)oxy)-2-naphthaldehyde (**S1**) (2 g, 7.0 mmol) in diethyl ether (10 mL) was added dropwise and the mixture was refluxed for another 1 h. After cooling, a 5% aqueous solution of NH₄Cl (30 mL) was added and the mixture extracted with CH₂Cl₂ (3 x 40 mL). Combined organic layers were washed with brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Product was purified by column chromatography on silica gel (CH₂Cl₂) to obtain 2.29 g (92 %) of compound **S2** as a white solid. ¹H NMR (600 MHz, CDCl₃) δ 7.73-7.65 (m, 3H), 7.42 (dd, *J* = 8.5, 1.7 Hz, 1H), 7.17-7.18 (m, 1H), 7.07 (dd, *J* = 8.8, 2.4 Hz, 1H), 4.80-4.75 (m, 1H), 1.93-1.82 (m, 2H), 1.82-1.72 (m, 1H), 1.49-1.38 (m, 1H), 1.35-1.24 (m, 5H), 1.02 (s, 9H), 0.92-0.78 (m, 3H), 0.24 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 153.5, 140.2, 134.2, 129.3, 129.0, 127.1, 124.5, 122.3, 114.8, 74.9, 38.9, 31.8, 25.7, 25.6, 22.6, 18.3, 14.0, -4.3.

6-Hexylnaphthalen-2-ol (S3)

The mixture of trifluoroacetic acid (5 ml) and triethyilsilane (10 ml) was stirred for 20 min at rt. Compound S2 (2.2 g, 6.1 mmol) in CH_2Cl_2 (15 ml) was added dropwise, and the mixture

was stirred for another 0.5 h at rt. Solvent was evaporated. TBAF solution (1M in THF, 7 mL) was added and the mixture stirred overnight at rt. Water was added (30 mL) and the mixture extracted to EtOAc (3 x 30 mL). Combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Product was purified by column chromatography on silica gel (CH₂Cl₂) to obtain 1.15 g (82 % over two steps) of compound **S3** as a white solid. ¹H NMR (600 MHz, CDCl₃) δ 7.69 (d, *J* = 8.8 Hz, 1H), 7.61 (d, *J* = 8.4 Hz, 1H), 7.57-7.52 (m, 1H), 7.30 (dd, *J* = 8.4, 1.8 Hz, 1H), 7.12 (d, *J* = 2.6 Hz, 1H), 7.08 (dd, *J* = 8.8, 2.5 Hz, 1H), 5.07 (s, 1H), 2.77-2.65 (m, 2H), 1.77-1.59 (m, 2H), 1.45-1.24 (m, 6H), 0.95-0.78 (m, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 152.8, 138.3, 133.0, 129.4, 129.3, 128.3, 126.4, 126.3, 117.7, 109.5, 36.1, 31.9, 31.5, 29.2, 22.8, 14.2.

6-Hexylnaphthalen-2-yl 4-(hexyloxy)benzoate (BNAM_6-6)

4-(Hexyloxy)benzoic acid (0.98 g, 4.4 mmol) was suspended in dry toluene (10 mL) under argon, and at 0 °C. Oxalyl chloride (1.9 mL, 22 mmol) was added followed by DMF (1 drop). The mixture was stirred for 1 hour at room temperature, solvent was evaporated, and the residue dissolved in CH₂Cl₂ (15 mL). This solution was added dropwise to a premixed solution of compound S3 (1.1 g, 4.8 mmol), Et₃N (6.1 mL, 44 mmol) and DMAP (270 mg, 2.2 mmol) in CH₂Cl₂ (10 mL). Reaction mixture was stirred overnight at room temperature. Water (30 mL) was added, and the mixture extracted with CH₂Cl₂ (3 x 20 mL). Organic extracts were washed with brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Crude product was purified using silica gel column chromatography (CH_2Cl_2/n -hexane = 2:1) to obtain ester **BNAM 6-6** (1.81 g, 97%) as a white solid which was further crystalized from acetone (1.52 g, 80%). PT (°C): Cr 84 N 118 Iso. ¹H NMR (600 MHz, CDCl₃) δ 8.18 (d, *J* = 8.9 Hz, 2H), 7.81 (d, J = 9.5 Hz, 1H), 7.73 (d, J = 8.4 Hz, 1H), 7.63-7.61 (m, 2H), 7.34 (dd, J = 8.3, 1.6 Hz, 1H), 7.30 (dd, J = 8.8, 2.4 Hz, 1H), 6.98 (d, J = 8.9 Hz, 2H), 4.04 (t, J = 6.6 Hz, 2H), 2.79-2.74 (m, 2H), 1.85-1.78 (m, 2H), 1.73 – 1.66 (m, 2H), 1.51-1.44 (m, 2H), 1.40-1.27 (m, 10H), 0.94-0.90 (m, 3H), 0.90-0.87 (m, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 165.3, 163.7, 148.3, 140.4, 132.44, 132.37, 131.8, 128.9, 128.3, 127.6, 126.3, 121.8, 121.5, 118.6, 114.4, 68.5, 36.2, 31.9, 31.7, 31.5, 29.2, 29.1, 25.8, 22.8, 22.7, 14.25, 14.17. Anal. Calcd. for C₂₉H₃₆O₃: C, 80.52; H, 8.39; Found: C, 80.95; H, 8.50.