

1 **SUPPLEMENTARY MATERIAL**

2
3 **Insecticidal activity of isolated phenylpropanoids from *Alpinia galanga* rhizomes against**
4 ***Spodoptera litura***

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17
18 **Abstract**

19 Botanical insecticides as a means of controlling insects present an alternative approach that is
20 safer than the use of synthetic insecticides. The present study identified the insecticidal
21 activity of extracts of the rhizomes of *Alpinia galanga* (L.) Willd. and seven isolated
22 phenylpropanoids against the second instar of *Spodoptera litura* Fab. by topical application.
23 The ethyl acetate extract had the highest toxicity on this insect with LD₅₀ values of 1.68 and
24 1.25 µg/larva after 24 and 48 h posttreatment, respectively. Among the seven
25 phenylpropanoids separated from the ethyl acetate extract, 1'S-1'-acetoxychavicol acetate was
26 identified as the most active compound with LD₅₀ values of 1.63 and 1.40 µg/larva after 24
27 and 48 h posttreatment, respectively, followed by *p*-coumaryl diacetate. In addition, the two
28 active compounds decreased glutathione *S*-transferase activity and increased
29 acetylcholinesterase activity. *p*-Coumaryl diacetate also decreased carboxylesterase activity.

30
31 **Keywords:** *Alpinia galanga*, Detoxification enzymes, Insecticidal activity,
32 Phenylpropanoids, *Spodoptera litura*

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34 **Experimental**

35 ***Insect Rearing***

36 *S. litura* were reared on an artificial diet consisting of green bean (360 g), agar (37.5 g), yeast
37 (30 g), ascorbic acid (7.5 g), methylparaben (7.5 g), sorbic acid (4.5 g), water (2.12 L),
38 mixed vitamin solution (60 mL), amoxicillin solution (60 mL) and 40% formalin (6 mL)
39 (Yooboon et al. 2019). The larvae were kept at 27°C, 65 ± 5% RH and L16:D8 photoperiod
40 before testing for insecticidal activity.

41

42 ***Plant Material***

43 *Alpinia galanga* rhizomes were obtained from Prachin Buri province, Thailand, in August
44 2018. A voucher specimen (BK No. 070911) was preserved in Bangkok Herbarium, Plant
45 Varieties Protection Office, Department of Agriculture, Bangkok, Thailand.

46

47 ***Extraction***

48 A sample (1.6 kg) of dried powder of *A. galanga* rhizomes was produced using sequential
49 extraction over seven days with either hexane, ethyl acetate or ethanol by maceration at room
50 temperature. Each solvent extract was filtered and dried using a rotary evaporator (Heidolph,
51 Germany) and stored at 4°C.

52

53 ***Isolation***

54 A sample (9 g) of the ethyl acetate extract was fractionated into twelve fractions using silica
55 gel column chromatography (Kiesel gel 60G, Merck, Thailand) with gradients of
56 hexane/ethyl acetate and ethyl acetate/methanol, respectively. At first, 1'S-1'-acetoxychavicol
57 acetate (**1**, 1.06 g) was isolated from fraction 4 (2.16 g) using silica gel column
58 chromatography and eluting with hexane/ethyl acetate (3/1). Fraction 6 (0.40 g) was
59 subjected to silica gel column chromatography using hexane/ethyl acetate (8/1) to provide
60 subfraction, 6-1 (0.21 g). This subfraction was purified using preparative thin layer
61 chromatography (PTLC) with hexane/ethyl acetate (5/1) to give *p*-coumaryl diacetate (**2**, 0.14
62 g) and *p*-coumaryl alcohol ethyl ether (**3**, 0.02 g). Fraction 8 (0.63 g) was further fractionated
63 using silica gel column chromatography with hexane/ethyl acetate (4/1) to obtain four
64 subfractions. Hydroxychavicol acetate (**4**, 0.16 g) was isolated from subfraction 8-3 (0.54 g)
65 using PTLC and eluting with hexane/ethyl acetate (3/1). The purification of fraction 9 (0.22
66 g) produced *trans-p*-hydroxycinnamaldehyde (**5**, 0.02 g) using PTLC with hexane/ethyl
67 acetate (1/1). Fraction 10 (1.07 g) was fractionated using column chromatography with

68 hexane/ethyl acetate (1/3) to obtain five subfractions. Subfraction 10-3 was identified as
69 *trans-p*-acetoxycinnamyl alcohol (**6**, 0.38 g). Subfraction 10-4 was further purified using
70 PTLC and eluting with hexane/ethyl acetate (1/1) to produce *p*-coumaryl alcohol (**7**, 0.005 g).

71

72 **Identification**

73 The chemical structures of all isolated phenylpropanoids were identified using ^1H and ^{13}C
74 NMR on a Bruker 400 MHz AVANCE III HD (Karlsruhe, Germany) and high resolution
75 mass spectrometry (HRMS) on a Maxis Bruker spectrometer (Karlsruhe, Germany).

76 1'S-1'-Acetoxychavicol acetate (**1**): Colorless oil; $[\alpha]_{\text{D}}^{20}$ -50 ($c=0.5$, CH_2Cl_2):
77 Colorless solid; ^1H NMR (CDCl_3 , 400 MHz) δ 7.36 (2H, d, $J = 8.6$ Hz), 7.07 (2H, d, $J = 8.6$
78 Hz), 6.26 (1H, d, $J = 5.8$ Hz), 5.98 (1H, ddd, $J = 17.1, 10.4, 5.8$ Hz), 5.30 (1H, dt, $J = 17.2,$
79 1.3 Hz), 5.25 (1H, dt, $J = 10.5, 1.2$ Hz), 2.29 (3H, s), 2.10 (3H, s); ^{13}C NMR (CDCl_3 , 100
80 MHz) δ 169.8, 169.3, 150.4, 136.4, 136.0, 128.4, 121.7, 117.0, 75.5, 21.1, 21.0. HRMS (ESI)
81 Calculated for $\text{C}_{13}\text{H}_{14}\text{NaO}_4$ 257.0790 ($[\text{M}+\text{Na}]^+$), Found 257.0825. NMR data were identical
82 to the literature values (Noro et al. 1988).

83 *p*-Coumaryl diacetate (**2**): Pale yellow oil; ^1H NMR (CDCl_3 , 400 MHz) δ 7.39 (2H, d,
84 $J = 8.6$ Hz), 7.05 (2H, d, $J = 8.6$ Hz), 6.62 (1H, d, $J = 16.1$ Hz), 6.22 (1H, dt, $J = 15.9, 6.4$
85 Hz), 4.75 (2H, dd, $J = 6.4, 1.3$ Hz), 2.30 (3H, s), 2.10 (3H, s); ^{13}C NMR (CDCl_3 , 100 MHz) δ
86 170.8, 169.4, 150.4, 134.0, 133.1, 127.6, 123.4, 121.7, 64.9, 21.1, 21.0. HRMS (ESI)
87 Calculated for $\text{C}_{13}\text{H}_{14}\text{NaO}_4$ 257.0790 ($[\text{M}+\text{Na}]^+$), Found 257.0796. NMR data were identical
88 to the literature values (Noro et al. 1988).

89 *p*-Coumaryl alcohol ethyl ether (**3**): Yellow oil; ^1H NMR (CD_3OD , 400 MHz) δ 7.25
90 (2H, d, $J = 8.6$ Hz), 6.75 (2H, d, $J = 8.6$ Hz), 6.52 (1H, d, $J = 15.9$ Hz), 6.12 (1H, dt, $J = 15.9,$
91 6.3 Hz), 4.09 (2H, dd, $J = 6.3, 1.3$ Hz), 3.54 (2H, q, $J = 7.0$ Hz), 1.21 (3H, t, $J = 7.0$ Hz); ^{13}C
92 NMR (MeOD , 100 MHz) δ 158.4, 133.9, 129.7, 128.8, 123.7, 116.3, 72.5, 66.4, 15.4. HRMS
93 (ESI) Calculated for $\text{C}_{11}\text{H}_{14}\text{NaO}_2$ 201.0891 ($[\text{M}+\text{Na}]^+$), Found 201.0886. NMR data were
94 identical to the literature values (Sukhirun et al. 2011).

95 Hydroxychavicol acetate (**4**): Yellow oil; ^1H NMR (CDCl_3 , 400 MHz) δ 7.37 (2H, d,
96 $J = 8.5$ Hz), 7.06 (2H, d, $J = 8.6$ Hz), 6.00 (1H, ddd, $J = 17.1, 10.3, 6.1$ Hz), 5.33 (1H, dt, $J =$
97 17.1, 1.4 Hz), 5.20-5.17 (2H, m), 2.28 (3H, s); ^{13}C NMR (CDCl_3 , 100 MHz) δ 169.6, 150.0,
98 140.2, 140.0, 127.5, 121.6, 115.4, 74.7, 21.1. HRMS (ESI) Calculated for $\text{C}_{11}\text{H}_{12}\text{NaO}_3$
99 215.0684 ($[\text{M}+\text{Na}]^+$), Found 215.0679. NMR data were identical to the literature values
100 (Janssen and Scheffer 1985).

101 *trans-p*-Hydroxycinnamaldehyde (**5**): Yellow solid; ¹H NMR (CDCl₃, 400 MHz) δ
102 9.67 (1H, d, *J* = 7.8 Hz), 7.51 (2H, d, *J* = 8.7 Hz), 7.44 (1H, d, *J* = 15.8 Hz), 6.92 (2H, d, *J* =
103 8.7 Hz), 6.63 (1H, dd, *J* = 15.8, 7.8 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 194.1, 158.8, 153.1,
104 130.8, 127.0, 126.6, 116.3. HRMS (ESI) Calculated for C₉H₈NaO₂ 171.0422 ([M+Na]⁺),
105 Found 171.0417. NMR data were identical to the literature values (Stange et al. 1999).

106 *trans-p*-Acetoxycinnamyl alcohol (**6**): Pale yellow solid; ¹H NMR (CDCl₃, 400 MHz)
107 δ 7.36 (2H, d, *J* = 8.6 Hz), 7.03 (2H, d, *J* = 8.6 Hz), 6.57 (1H, d, *J* = 15.9 Hz), 6.28 (1H, dt, *J*
108 = 15.9, 5.7 Hz), 4.29 (2H, dd, *J* = 5.7, 1.3 Hz), 2.28 (3H, s); ¹³C NMR (CDCl₃, 100 MHz) δ
109 169.6, 150.0, 134.4, 129.9, 128.7, 127.3, 121.6, 63.4, 21.0. HRMS (ESI) Calculated for
110 C₁₁H₁₂NaO₃ 215.0684 ([M+Na]⁺), Found 215.0681. NMR data were identical to the literature
111 values (Zhu et al. 2013).

112 *p*-Coumaryl alcohol (**7**): White solid; ¹H NMR (Acetone-d₆, 400 MHz) δ 7.26 (2H, d,
113 *J* = 8.6 Hz), 6.80 (2H, d, *J* = 8.6 Hz), 6.52 (1H, d, *J* = 15.9 Hz), 6.21 (1H, dt, *J* = 15.9, 5.6
114 Hz), 4.22 (2H, dd, *J* = 5.6, 1.6 Hz); ¹³C NMR (Acetone-d₆, 100 MHz) δ 157.8, 129.6, 128.3,
115 116.2, 63.5. HRMS (APCI) Calculated for C₉H₁₃O₂ 151.0759 ([M+H]⁺), Found 157.0754.
116 NMR data were identical to the literature values (Ly et al. 2003).

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118 **Bioassay**

119 Contact toxicity assays to second instars of *S. litura* were examined to determine the
120 mortality and median lethal dose (LD₅₀) using a topical application. Sample concentrations of
121 0.5, 1, 2, 5, 10 and 20 µg/larva were prepared in acetone. Then, 2 µL of each sample solution
122 was slowly applied to the thorax region of *S. litura*. Each sample was repeated in five
123 replicates each consisting of a population of ten larvae. The treated larvae were transferred
124 into Petri dishes (diameter 100 mm) under controlled conditions with an artificial feeding diet
125 and then mortality was recorded after 24 and 48 h posttreatment. The LD₅₀ values were
126 determined using Probit analysis in the StatPlus Program (2018 version).

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128 **Enzyme assays**

129 *Enzyme preparation method*

130 After 24 h posttreatment of active phenylpropanoid at the LD₅₀ level, the surviving *S. litura*
131 larvae were used to determine detoxification and neuron enzyme activities. The larvae were
132 placed in a microcentrifuge tube and crushed in homogenised buffer (pH 7.2) prepared from
133 0.1 M potassium phosphate buffer and 1 mM ethylenediaminetetraacetic acid for enzyme

134 extraction. The homogenate was then centrifuged at 15,520 g (12,000 rpm) for 15 minutes at
135 4°C to produce the supernatant for analysis of the enzyme activities.

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137 *Carboxylesterase activity*

138 The supernatant (10 µL) was examined for carboxylesterase activity by mixing with
139 potassium phosphate buffer (190 µL, 50 mM, pH 7.4) and *p*-nitrophenyl acetate in DMSO
140 (40 µL, 10 mM). The reaction was analysed at 410 nm and 37°C for 90 seconds with a Biotek
141 PowerWave XS microplate spectrophotometer (Winooski, VT, USA). The extinction
142 coefficient of *p*-nitrophenyl acetate was 176.4705 (Bullangpoti et al. 2012).

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144 *Glutathione S-transferase activity*

145 The supernatant (100) µL was examined for glutathione *S*-transferase activity by mixing with
146 potassium phosphate buffer (100 µL, 50 mM, pH 7.2), glutathione solution (10 µL, 10 mM
147 GSH reduced form) and 1-chloro-2,4'-dinitrobenzene (CDNB) (100 µL, 150 mM). The
148 reaction was analysed at 340 nm and 25°C for 3 min on the microplate spectrophotometer.
149 The extinction coefficient of CDNB was 0.000137 (Oppenoorth F. J. et al. 1979).

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151 *Acetylcholinesterase activity*

152 The mixture of supernatant (50 µL) and potassium phosphate buffer (50 µL, 100 mM, pH
153 7.2) was incubated at 30°C for 30 minutes, followed by adding a mixture of 5,5'-dithiobis(2-
154 nitrobenzoic acid) (2µL, 100 mM), ATCI (2µL, 100 mM) and potassium phosphate buffer
155 (46 µL, 100 mM, pH 7.2)). The reaction was analysed at 412 nm to determine the
156 acetylcholinesterase activity on the microplate spectrophotometer. The extinction coefficient
157 was $1.36 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$ (Kumrungsee et al. 2014).

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Table S1. Characteristics and amounts of extracts from *A. galanga* rhizomes

Extract	Weight (g)	Yield ^a (% wt/wt)	Characteristics
Hexane	6.83	0.43	Pale yellow gum
Ethyl acetate	19.11	1.19	Brown gum
Ethanol	82.86	5.18	Dark brown gum

^a(crude extract / weight of dried plant) × 100

Table S2. Toxicity of the rhizomes of *A. galanga* extracts against *S. litura*

Extract	Time (h)	LD ₅₀	LCL	UCL	<i>P</i> -value	χ^2
Hexane	24	8.02	6.78	9.68	0.36	4.36
	48	6.87	4.40	12.29	0.04	9.78
Ethyl acetate	24	1.68	0.78	2.89	0.27	5.15
	48	1.25	1.02	1.50	0.78	1.74
Ethanol	24	NA	-	-	-	-
	48	NA	-	-	-	-

LD₅₀: Lethal Dosage that kills 50% of the exposed larvae, expressed in µg/larva.

LCL: Lower Confidence Limit, UCL: Upper Confidence Limit.

NA: Not Active at the highest concentration tested (20 µg/larva).

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Table S3. Toxicity of 1'S-1'-acetoxy-chavicol acetate (**1**), *p*-coumaryl diacetate (**2**) and racemic mixture of **1** ((±)-**1**) to the second instar of *S. litura* after 24 and 48 h posttreatment

Compound	Time (h)	LD ₅₀	LCL	UCL	<i>P</i> -value	χ^2
1	24	1.63	1.26	2.03	0.58	2.87
	48	1.40	1.09	1.74	0.95	0.71
2	24	2.37	1.32	3.89	0.68	2.28
	48	2.14	1.25	3.32	0.59	2.81
(±)- 1	24	13.31	11.06	16.38	0.9	1.08
	48	11.41	9.59	13.79	0.85	1.36

LD₅₀: Lethal Dosage that kills 50% of the exposed larvae, expressed in µg/larva.

LCL: Lower Confidence Limit, UCL: Upper Confidence Limit.

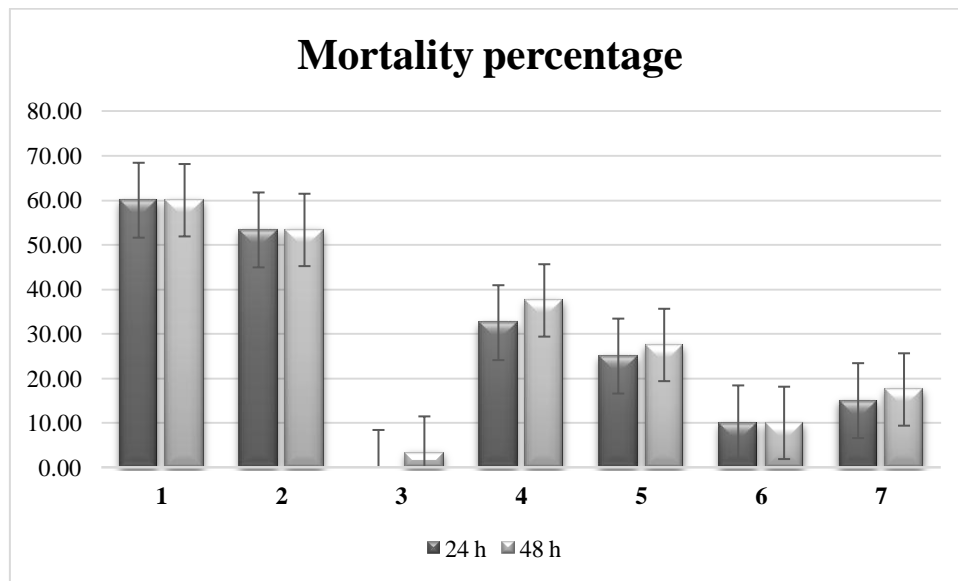
Table S4. Effect of 1'S-1'-acetoxychavicol acetate (**1**) and *p*-coumaryl diacetate (**2**) on enzyme activities

Treatment	Carboxylesterase ^{a,b}	Glutathione S-transferase ^{a,b}	Acetylcholinesterase ^{a,b}
Control	1.007 ± 0.143 a	0.953 ± 0.129 a	0.119 ± 0.003 a
1	0.994 ± 0.082 ab	0.794 ± 0.035 b	0.136 ± 0.018 b
2	0.774 ± 0.096 b	0.815 ± 0.053 b	0.137 ± 0.002 b

^aMeans ± SE within a column followed by the same lowercase letter are not significantly different (*P* > 0.05, ANOVA).

^bCarboxylesterase activity (nM *p*-nitrophenol/min/mg protein); Glutathione-*S*-transferase activity (CDNB conjugated product/mg protein/min); Acetylcholinesterase (acetylcholinesterase activity/mg protein/mL).

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Figure S1. Toxicity of seven isolated phenylpropanoids against *S. litura* at a concentration of 2 µg/larva after 24 and 48 h posttreatment.