

## **Supporting Information**

### **Deuterated Malonamide Synthesis for Fundamental Research on Solvent Extraction Systems**

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### **S1. $^1\text{H}$ NMR data of entries 2, 3, 5–8 in the Table 1**

The  $^1\text{H}$  NMR spectra of entries 2, 3, 5, 6, 7, and 9 in the Table 1 were recorded using a 400 MHz NMR spectrometer (JMTC-400/54/JJ/YH, JEOL Ltd., Tokyo, Japan) at 400 MHz. The

$^1\text{H}$  NMR data are shown in below:

**Entry 2:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ , TMS, 1,4-dioxane)  $\delta$  0.82–0.91 (m, 2.48H), 1.24–1.27 (m, 5.77H), 1.47–1.53 (m, 1.41H), 3.16–3.28 (m, 1.42H), 3.43 (s, 0.18H).

**Entry 3:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ , TMS, 1,4-dioxane)  $\delta$  0.84–0.90 (m, 3.97H), 1.24–1.27 (m, 9.05H), 1.47–1.52 (m, 2.20H), 3.25–3.33 (m, 2.48H), 3.43 (s, 1.06H).

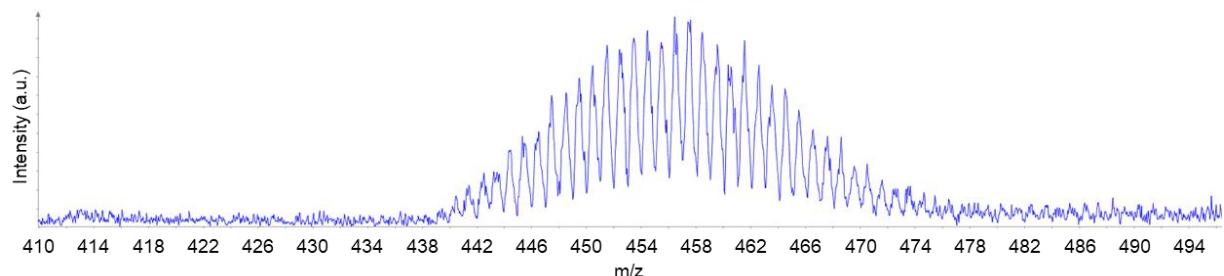
**Entry 5:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ , TMS, 1,4-dioxane)  $\delta$  0.78–0.88 (m, 8.27H), 1.22–1.29 (m, 17.7H), 1.41–1.55 (m, 4.05H), 3.12–3.27 (m, 5.56H), 3.48 (br, 0.26H).

**Entry 6:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ , TMS, 1,4-dioxane)  $\delta$  0.82–0.95 (m, 3.11H), 1.23–1.31 (m, 8.40H), 1.47–1.50 (m, 1.04H), 1.82–1.86 (m, 0.69H), 2.87–2.95 (m, 0.36H), 3.19–3.36 (m, 0.72H), 3.50–3.52 (m, 0.20H).

**Entry 7:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ , TMS, 1,4-dioxane)  $\delta$  0.82–0.93 (m, 2.63H), 1.15–1.30 (m, 14.3H), 1.38–1.59 (m, 2.42H), 1.75–1.86 (m, 0.82H), 2.77–3.04 (m, 0.24H), 3.19–3.39 (m, 1.12H), 3.52 (br, 0.15H).

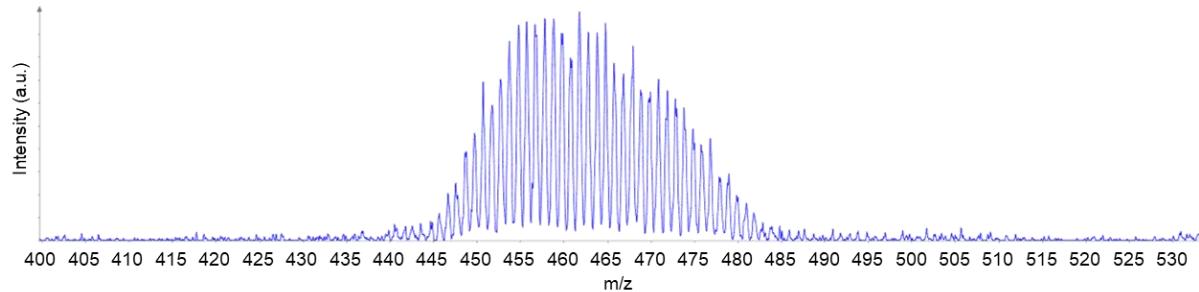
**Entry 9:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ , TMS, 1,4-dioxane)  $\delta$  0.86–0.97 (m, 9.00H),  $\delta$  1.25–1.31 (m, 27.3H), 1.49–1.52 (m, 2.82H), 1.86 (m, 1.41H), 2.86–3.01 (m, 1.60H), 3.17–3.37 (m, 3.43H), 3.48–3.61 (m, 0.58H).

## S2. ESI-MS data of THMA- $d_n$



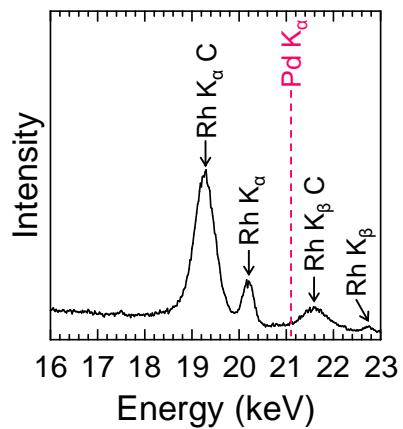
**Figure S1.** Electrospray ionization mass spectra in positive mode of THMA- $d$  cation showing the mass distribution of the different isotopologues, which ranges from  $d_2$ – $d_{40}$ . The distribution of the isotopologues is as follows ( $M^+$ ): 0.54 %,  $d_2$ ; 1.00 %,  $d_3$ ; 1.00 %,  $d_4$ ; 1.22 %,  $d_5$ ; 1.68 %,  $d_6$ ; 2.18 %,  $d_7$ ; 2.13 %,  $d_8$ ; 2.68 %,  $d_9$ ; 2.94 %,  $d_{10}$ ; 3.36 %,  $d_{11}$ ; 4.17 %,  $d_{12}$ ; 4.08 %,  $d_{13}$ ; 4.54 %,  $d_{14}$ ; 4.99 %,  $d_{15}$ ; 4.99 %,  $d_{16}$ ; 4.99 %,  $d_{17}$ ; 5.44 %,  $d_{18}$ ; 5.44 %,  $d_{19}$ ; 4.54 %,  $d_{20}$ ; 4.40 %,  $d_{21}$ ; 4.35 %,  $d_{22}$ ; 4.04 %,  $d_{23}$ ; 3.58 %,  $d_{24}$ ; 3.13 %,  $d_{25}$ ; 2.72 %,  $d_{26}$ ; 2.54 %,  $d_{27}$ ; 1.91 %,  $d_{28}$ ; 1.68 %,  $d_{29}$ ; 1.36 %,  $d_{30}$ ; 1.27 %,  $d_{31}$ ; 1.09 %,  $d_{32}$ ; 1.00 %,  $d_{33}$ ; 1.00 %,  $d_{34}$ ; 0.95 %,  $d_{35}$ ; 0.91 %,  $d_{36}$ ; 0.68 %,  $d_{37}$ ; 0.59 %,  $d_{38}$ ; 0.44 %,  $d_{39}$ ; 0.45 %,  $d_{40}$ . The deuteration ratio was estimated to be 34.6 %.

### S3. ESI-MS data of DBMA- $d_n$



**Figure S2.** Electrospray ionization mass spectra in positive mode of DBMA- $d$  cation showing the mass distribution of the different isotopologues, which ranges from  $d_5$ – $d_{38}$ . The distribution of the isotopologues is as follows ( $M^+$ ):  $d_5$ ; 0.46 %,  $d_6$ ; 0.80 %,  $d_7$ ; 1.17 %,  $d_8$ ; 1.76 %,  $d_9$ ; 2.13 %,  $d_{10}$ ; 2.64 %,  $d_{11}$ ; 2.80 %,  $d_{12}$ ; 3.26 %,  $d_{13}$ ; 3.89 %,  $d_{14}$ ; 4.19 %,  $d_{15}$ ; 4.19 %,  $d_{16}$ ; 4.60 %,  $d_{17}$ ; 4.19 %,  $d_{18}$ ; 4.19 %,  $d_{19}$ ; 4.19 %,  $d_{20}$ ; 3.64 %,  $d_{21}$ ; 4.19 %,  $d_{22}$ ; 3.93 %,  $d_{23}$ ; 3.93 %,  $d_{24}$ ; 3.68 %,  $d_{25}$ ; 3.43 %,  $d_{26}$ ; 3.22 %,  $d_{27}$ ; 3.81 %,  $d_{28}$ ; 3.10 %,  $d_{29}$ ; 3.14 %,  $d_{30}$ ; 2.97 %,  $d_{31}$ ; 2.85 %,  $d_{32}$ ; 2.72 %,  $d_{33}$ ; 2.51 %,  $d_{34}$ ; 2.26 %,  $d_{35}$ ; 2.01 %,  $d_{36}$ ; 1.63 %,  $d_{37}$ ; 1.30 %,  $d_{38}$ ; 1.21 %. The deuteration ratio was estimated to be 39.6 %.

**S4. X-ray fluorescence of by-product in the deuteration of THMA-*h*<sub>54</sub>.**



**Figure S3.** X-ray fluorescence (XRF) spectrum of the white precipitate obtained in the

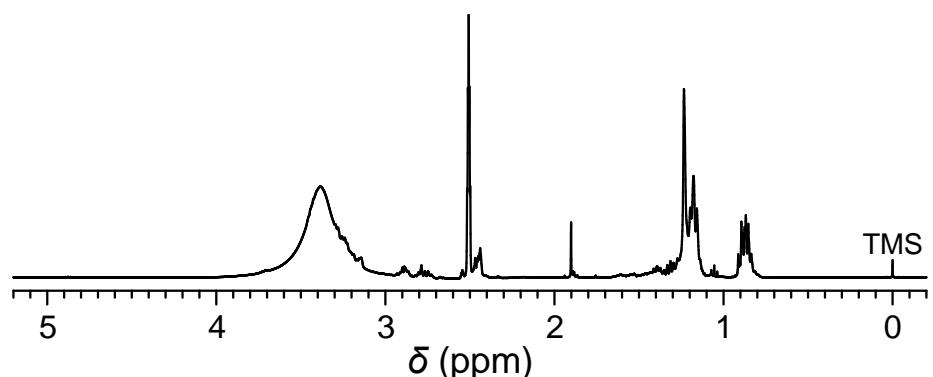
deuteration reaction of THMA-*h*<sub>54</sub>, where the spectrum was recorded using a wavelength-

dispersion-type XRF spectrometer (ZSX Primus II, Rigaku Corporation, Tokyo, Japan). XRF

peaks of K<sub>α</sub> C, K<sub>α</sub>, K<sub>β</sub> C, and K<sub>β</sub> of Rh were observed at 19.3, 20.2, 21.6, and 22.7 keV,

respectively, whereas the XRF peak of K<sub>α</sub> for Pd was not observed.

**S5.  $^1\text{H}$  NMR of by-product in the deuteration of THMA-*h*<sub>54</sub>.**



**Figure S4.**  $^1\text{H}$  NMR spectrum of the white precipitate, obtained in the deuteration reaction of THMA-*h*<sub>54</sub>, in  $\text{CDCl}_3$  with TMS (chemical shift reference).