Magnesium-doped Zinc Oxide as a High Resistance Transparent Layer for thin film CdS/CdTe solar cells

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Abstract—Magnesium-doped Zinc Oxide (MZO) was used as an alternative high resistance transparent layer for CdS/CdTe thin film solar cells. Thin films of MZO were deposited by RF magnetron sputtering and deposited on an Indium Tin Oxide contact (ITO). Thin film CdTe devices including a MZO high resistance transparent layer deposited at above 300 °C yielded a mean efficiency exceeding 10.5 %. This compares with an efficiency of 8.2 % without the MZO layer. The improvement in efficiency was due to a higher open circuit voltage and fill factor. Lowering the deposition temperature of MZO reduced the performance of the devices.

Index Terms—magnesium-doped zinc oxide, high resistance transparent layer, thin film, solar cells, CdTe.

I. INTRODUCTION

Doping ZnO with MgO ($E_G = 7.7$ eV) leads to an energy bandgap increase ($E_G \approx 3.3$ eV [1]–[3]) through the formation of $\text{Zn}_{1-x}\text{Mg}_x\text{O}$ (MZO) [4]. The increase depends linearly upon the Mg content in the film, up to a Mg content of $x = 0.46$, at which point the band gap becomes $E_G = 4.2$ eV [5]. Experimental determination of the band alignment of MZO indicates that the larger band gap of MZO is almost exclusively due to an upshifting of the conduction band energy level [6], [7]. A similar behaviour occurs when ZnO is doped with Ca rather than Mg [8], [9]. In the range $x = 0$ to $x = 0.46$ $\text{Zn}_{1-x}\text{Mg}_x\text{O}$ maintains the typical hexagonal structure of ZnO. Above this doping level there is a gradual transition to the cubic structure of MgO [5]. There are several studies highlighting the importance of controlling the alignment between the band levels of absorber and window layers in chalcogenide solar cells [10]–[12]. These emphasize the importance of a slightly positive conduction band at the buffer/absorber interface to control the inversion of the absorber and recombination at the interface. The first applications of MZO were reported with copper indium gallium selenide (CIGS) thin film solar cells [13]–[17], where MZO was used as a replacement for the CdS buffer layer. The successful application of magnesium-doped zinc oxide to CdTe thin film solar cells has been reported recently [18]. Other semiconductors have been investigated for the tuneability of their energy band structure [19]–[23]. This study has focused on using MZO as an HRT layer with an emphasis on the effect of changing its conduction band

alignment with the adjacent semiconductors by varying the temperature.

II. EXPERIMENTAL DETAILS

Tin-doped indium oxide and magnesium-doped zinc oxide thin films were deposited by Radio-Frequency (RF) magnetron sputtering. Soda lime glass (SLG) was used as a superstrate. The glass was cleaned using a solution composed of 1/3 isopropanol, 1/3 acetone and 1/3 deionized water in an ultrasonic bath at 50 °C for 60 min. Thin films were deposited using an Orion 8 HV magnetron sputtering system (AJA international, USA) equipped with an AJA 600 series RF power supply. The 3" diameter ITO target contained 10 % SnO_2 and 90 % In_2O_3 Wt %. The 3" diameter MZO target contained 11 % MgO and 89 % ZnO Wt %. The glass superstrates were rotated at 10 rpm during deposition to enhance the uniformity of the films. The sputtering process was carried out at a constant power density of 3.5 W.cm^{-2} and at a pressure of 133.3 Pa using Ar as the working gas. Sputtering of MZO films was carried out in a 1 % O_2 in an Ar atmosphere. The temperature of the superstrate was kept constant at 450 °C for the deposition of ITO and ranged from 20 to 400 °C for MZO. The optical properties were investigated using a Cary Varian 5000 UV-VIS-NIR spectrophotometer. The composition of the films was measured using an X-ray photoelectron spectrometer (XPS) (Thermo Scientific K-alpha). Samples were processed into complete CdTe solar cells in the laboratories of University of Verona. ITO/MZO superstrates were coated with CdS and CdTe by thermal evaporation. The deposition process was carried out in a vacuum chamber at a pressure of 10^{-4} Pa with a Edwards XDS10 roughing pump and a Edwards ST-451 turbo-molecular pump. CdS was evaporated from a tungsten crucible at a deposition rate of 0.15 nm/sec. During deposition the substrate temperature was kept at 100 °C using halogen lamps. Before and after CdS deposition, the stack was annealed in vacuum at 450 °C for 30 minutes. CdTe was deposited from a special graphite Knudsen cell with an evaporation rate of 40 Å/sec. The deposition rate was controlled using an Intelmetrics IL-150 quartz controller. The CdTe activation treatment was performed using a CdCl_2

wet treatment. The solution was prepared by dissolving the CdCl_2 powder in methanol to form a saturated solution. The CdCl_2 powder was dried in a furnace at 0.1 Pa before processing in solution. Typically, $\approx 250 \mu\text{l}$ is deposited in form of drops on the CdTe surface. The stack is then annealed in air at 380°C for 30 minutes after a 15 minutes ramp from room temperature. Prior to back contact formation, the CdTe surface is treated with a solution of bromine (50 μl) and methanol (50 ml). This process removes residual CdCl_2 and forms a Te-rich layer. Subsequently, a 2 nm thick layer of Cu and a 50 nm thick layer of Au are deposited by thermal evaporation at room temperature in a vacuum of 10^{-3} Pa. The process is finished by annealing the structure for 20 min at 190°C in air. Devices were characterized using current density-voltage (J-V) characteristics and cross-section images were obtained using transmission electron microscopy (TEM). Samples for TEM were prepared by focused ion beam milling using a dual beam FEI Nova 600 Nanolab. A standard in situ lift out method was used to prepare cross-sectional samples. An electron beam assisted platinum(e-Pt) over-layer was deposited onto the sample surface above the area to be analysed followed by an ion assisted layer to define the surface and homogenize the final thinning of the samples down to 100 nm. TEM analysis was carried out using a Tecnai F20 operating at 200 kV to investigate the detailed microstructure of the cell cross sections. Images were obtained using the bright field (BF) detector.

III. CHARACTERIZATION OF MAGNESIUM-DOPED ZINC OXIDE FILMS

Transmission plots of MZO films deposited at increasing temperatures are shown in Fig. 1(a). From the absorption edge in the UV region it is clear that the band gap of the films varies with the temperature of the substrate during deposition. The band gap of MZO films was estimated using the Tauc plot technique (Fig. 1(b)). The energy band gap of MZO films deposited at room temperature was estimated to be $E_G = 3.65$ eV, 0.35 eV higher than ZnO. This confirms that doping ZnO with Mg widens the optical band gap of the semiconductor. It was also observed that raising the temperature assists further increase in E_G . E_G increased from 3.65 eV at room temperature to 3.95 eV at 400°C , as previously reported [24]. XPS analysis showed that increasing the temperature during MZO film deposition leads to an increased concentration of Mg atoms as shown in Fig. 2. The Mg concentration increases up to 400°C while the Zn concentration decreases up to 300°C . The oxygen concentration remains $\approx 41\%$ at 20°C , 100°C and 400°C and significantly increases at 200 and 300°C . These results suggest that temperature enhances the inclusion of Mg ions in the MZO crystal structure. Evaporation of Zn or Mg during deposition can have an important role in this process. The vapourization temperature of Zn and Mg at a pressure of 1 mTorr (used during deposition) is $\approx 290^\circ\text{C}$ and $\approx 380^\circ\text{C}$ respectively [25]. At 300°C and 400°C , the Mg/Zn ratio increases significantly compared to films deposited at lower temperatures. Zn ions evaporate leaving free lattice

Fig. 1. The transmission spectra (a) and the Tauc plots (b) of MZO films deposited at temperature ranging between 20°C and 400°C showing respectively the shift of the absorption edge and the widening of the energy band gap in MZO films caused by the variation of the substrate temperature.

sites for Mg ions to occupy. Up to 200°C the increasing Mg atomic concentration is not related to Zn evaporation but some other mechanism. Hwang et al have reported [24] that the increasing temperature causes a reduction of the Mg content in the films due to evaporation of Mg as measured using electron probe microanalysis. This is opposite to our findings. They also suggested that the increase in optical energy band gap occurs because raising temperature assists the replacement of Zn ions with Mg ions. Our work confirmed that temperature helps Mg ions replacing Zn ions. It was also confirmed that higher film deposition temperatures increase the MZO band gap. However the Mg concentration was found to grow by increasing temperature. The crystal structure of MZO has been investigated by XRD (Fig. 3). The (002) peak was observed, and the other peaks are associated with the underlying ITO film. The (002) peak was also observed for ZnO films [26] and is indexed from the crystallographic

Fig. 2. The atomic percentage of Mg,ZnO and O in MZO films estimated by XPS analysis showing the variation of the atomic ratios in MZO films caused by the variation of the substrate temperature.

Fig. 3. The XRD patterns of MZO films deposited at different temperatures on ITO. The line highlights the position of the (002) peak, typical of the hexagonal structure of ZnO.

data of the ZnO hexagonal structure. Peaks (200) and (220) indexed from the crystallographic data of the cubic structure of MgO, are not visible [5] (ICDD 00-003-0752). ZnO has a band gap of 3.3 eV and a negative CBO with CdS of -0.3 eV. Depositing MZO at room temperature provides an MZO film with a band gap of 3.65 eV [7]. This corresponds to an almost flat conduction band alignment with CdS. Raising the deposition temperature of MZO increases the band gap to 3.95 eV at 400 °C. This corresponds to in a CBO with CdS of +0.35 eV.

A. Thermal Stability of MZO

Multiple annealing steps are repeated during the fabrication process of thin film CdTe solar cells. The band gap of MZO films is sensitive to the temperature of the substrate during

Fig. 4. Tauc plot of MZO films deposited at different temperatures (dashed lines) and of the same films annealed at 450 °C for 30 minutes (solid lines) showing the slight decrease of optical band gap that occurs after the MZO films are annealed at 450 °C for 30 minutes.

deposition. To test if the band gap of MZO can change during the fabrication process, a thermal annealing step was carried out at 450 °C for 30 minutes. This step simulates an equivalent annealing step used during the fabrication process. The band gaps of the films were extrapolated using the Tauc Plot technique (Fig. 4). The plot reveals that annealing the films at 450 °C slightly decreases the band gap of the films. The films deposited at lower temperatures result in a greater change in band gap. The change in all cases is small. The thin film deposited at room temperature changed the energy band gap from 3.65 eV before annealing to 3.63 eV post annealing. Similarly the film deposited at 100 °C decreased from 3.67 eV to 3.65 eV. These results suggest that the annealing steps carried out during the fabrication of CdS/CdTe devices do not significantly change the energy band gap of the MZO layers.

IV. CDS/CdTe SOLAR CELLS WITH MZO HRTs

MZO films have been tested as HRT layers in CdS/CdTe solar cells with the structure ITO/MZO/CdS/CdTe/back contact. The performance of the devices was significantly affected by the deposition temperature of the MZO HRTs (Fig. 5). This suggests that the band gap of the MZO layer is playing a key role. The mean device efficiency increases from slightly below 5% achieved with a room temperature deposited MZO to ≈10.6 % at 400 °C. The higher efficiency is a consequence of improved Voc and FF. Voc and FF of 0.82 V and 66% respectively were achieved with a deposition temperature of 400 °C for the MZO layer. Temperature has a detrimental effect on Jsc at 100°C while a further increase of temperature gradually improves the current density. The current density reaches a maximum at 300 °C and 400 °C (≈20 mA/cm²). Devices not containing the MZO HRTs were fabricated and the results have been compared. Efficiency, Voc and FF all benefit from the addition of MZO deposited at high temper-

ature. The highest current densities achieved with MZO are similar to those achieved without the layer. This indicates that the high transparency of the films deposited at high temperature does not greatly affect the device current density.

V. QUANTUM EFFICIENCY, TEM AND EDX ANALYSIS

The EQE of the highest performing device, with an MZO layer deposited at 400 °C is shown in Fig. 6. The absorption edge in the near infra-red (NIR) region lies at 850 nm, which corresponds to a typical band gap for the CdTe absorber of 1.45 eV [27]. Whilst the device performance improves with MZO deposition temperature with improved Voc and FF, the current density is limited by the high absorption in the blue region by the CdS. It is clear that to improve the device further, a reduction of thickness of the CdS layer is required. This would lead to higher photocurrent generation in the device. TEM images of the cross section of the sample show the layer by layer microstructure of the solar cell (Fig. 7). The thickness of the CdS layer has been estimated from the images to be in the range 250 nm - 300 nm. The CdS layer is sufficiently thick to absorb most of the radiation in the blue visible region. The CdTe layer is $\approx 7 \mu\text{m}$ in thickness. The CdTe grains develop across the full width of the layer.

VI. CONCLUSIONS

MZO films were deposited by magnetron sputtering at different substrate temperatures. It was confirmed that the addition of Mg to ZnO widens the band gap of the films. Increasing substrate temperature causes the band gap to widen further. These properties were utilized to produce MZO films with different E_g and to test them as HRT layers for CdTe thin film solar cells. The band gap of the MZO films ranged between 3.65 eV and 3.95 eV. A larger band gap was found to be beneficial for the Voc, FF and efficiency of devices. This suggests that the energy band alignment between HRT and CdS plays an important role in the performance of the solar cells. A thick CdS layer ($\approx 300\text{nm}$) was used as a buffer layer leading to a low EQE response below 650 nm. A thinner CdS should significantly improve the current density and the efficiency of these devices. This study provides evidence that MZO is a promising material as a HRT layer for thin film CdTe solar cells and highlights the importance of energy band alignment between adjacent semiconductors.

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Fig. 5. Box Plots giving a statistical representation of Voc (a), FF (b), Jsc (c) and efficiency (d) of devices containing MZO deposited at increasing substrate temperature.

Fig. 6. The EQE spectra of a ITO/MZO/CdS/CdTe solar cell with MZO $E_g = 3.94$ eV showing the low spectral response of the best performing device below 600 nm is due to the thick CdS layer absorption losses.

Fig. 7. A TEM cross-section of a full device. Starting from the top: the CdTe absorber, the CdS layer, the MZO HRT layer and the ITO layer.

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