Self-Assembly of Molecular Prisms *via* Pt₃ Organometallic Acceptors and a Pt₂ Organometallic Clip

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Fig. S1: ¹H NMR spectra of 2a, (top left) 1a (top right) and 3a (bottom).



Fig. S2: ESI-MS spectrum of **3a** in MeOH. Inset: Theoretical and experimental isotopic distribution patterns of the [M-3NO₃]³⁺ peak.





Fig. S3. ¹H NMR (above) and ³¹P NMR (below) of the supramolecular prism **3b**.



Fig. S4. ESI MS of the supramolecular prism **3b** and the experimental isotropic distribution pattern of the peaks.



Fig. S5: ¹H and ³¹P{¹H} (inset) NMR spectra of the prism 3c in CDCl₃.



Fig. S6: ESI mass spectrum (left) of 3c and the experimental isotopic distribution pattern of the peak C_3 (right).



Fig. S7: ESI-Mass spectrum of 1c in methanol.



Fig. S8: ³¹P NMR spectrum of **3d** in CDCl₃.



Fig. S9: ESI mass spectrum of **3d** (left) and the isotopic distribution pattern of the peak due to [M-3NO₃⁻] (right).



Fig.S10: Fluorescence spectra of the prisms 3b-d in DMF.



Fig. S11. Binding study of supramolecular prism 3d by titration of molecular prism 3d with TNT (from 0.5×10^{-3} M to 2.5×10^{-3} M) and monitoring the change in absorption spectra.



Fig. S12. Plot of inverse of the change of absorption of supramolecular prism **3d** vs the inverse of concentration of TNT to determination the binding constant for the binding of supramolecular prism **3d** and TNT.



Fig. S13. Quenching of fluorescence intensity of **3d** (1.5×10^{-5} M) on gradual addition of dilute TNT solution (~ 10^{-5} M).



Fig. S14: Fluorescence quenching of the spin cast film of **3d** after exposing to TNT vapour. Plot of percentage of fluorescence quenching *vs* time (inset).

Table S1. Photophysical Data of the Supramolecular Prisms (3b-d).

Complex	Solvent	Absorbance	Fluorescence	Φ_{F}
		$\lambda_{max}(nm)$	$\lambda_{max}(nm)$	
3b	DMF	283, 365	425	0.04
3c	DMF	430, 383, 270	472	0.48
3d	DMF	421, 398, 309.	430, 460	0.35