

Supporting Information

Clathrate Ba₈Au₁₆P₃₀: The “Gold Standard” for Lattice Thermal Conductivity

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Table SI 1. Atomic coordinates and isotropic displacement parameters for Ba₈Au₁₆P₃₀.

Atom	Site	x/a	y/b	z/c	U_{so}
Ba1	4b	0	0.5	0	0.0139(8)
Ba2	4c	0	0.0062(3)	0.25	0.0125(8)
Ba3	8d	0.3757(2)	0.9915(3)	0.18733(9)	0.0287(7)
Ba4	8d	0.1246(2)	0.0085(3)	0.06050(9)	0.0224(7)
Ba5	8d	0.2486(2)	0.2540(3)	0.3763(1)	0.0256(6)
Au1	8d	0.1858(1)	0.1889(2)	0.24960(6)	0.0180(4)
Au2	8d	0.0020(1)	0.3196(2)	0.09348(5)	0.0180(4)
Au3	8d	0.3799(1)	0.0045(2)	0.31054(5)	0.0224(4)
Au4	8d	0.5522(1)	0.3159(2)	0.03118(6)	0.0185(4)
Au5	8d	0.9014(1)	0.0037(2)	0.14210(5)	0.0180(4)
Au6	8d	0.7164(1)	0.0100(2)	0.04688(5)	0.0163(4)
Au7	8d	0.6523(1)	0.3810(2)	0.32760(6)	0.0228(5)
Au8	8d	0.1612(1)	0.6155(2)	0.07944(6)	0.0182(5)
P1	8d	0.0050(7)	0.192(1)	0.1603(3)	0.0049(5)
P2	8d	0.4863(7)	0.309(0)	0.3402(3)	0.0049(5)
P3	8d	0.6835(7)	0.192(1)	0.9980(3)	0.0049(5)
P4	8d	0.1881(7)	0.679(1)	1.0002(3)	0.0049(5)
P5	8d	0.4990(8)	0.179(1)	0.0951(3)	0.0049(5)
P6	8d	0.2481(9)	0.260(1)	0.1243(4)	0.0049(5)
P7	8d	0.5623(6)	0.69631	0.0304(4)	0.0049(5)
P8	8d	0.5969(6)	0.004(1)	0.1063(3)	0.0049(5)
P9	8d	0.2132(6)	0.011(1)	0.2961(3)	0.0049(5)
P10	8d	0.1822(6)	0.817(1)	0.2570(3)	0.0049(5)
P11	8d	0.1237(6)	0.018(1)	0.4414(3)	0.0049(5)
P12	8d	0.1523(7)	0.125(0)	0.1694(3)	0.0049(5)
P13	8d	0.5537(6)	0.808(1)	0.2209(3)	0.0049(5)
P14	8d	0.1528(7)	0.3690(9)	0.0769(4)	0.0049(5)
P15	8d	0.0661(7)	0.7074(9)	0.2830(3)	0.0049(5)

Table SI 2. Selected interatomic distances in Ba₈Au₁₆P₃₀.

Atoms	$d, \text{\AA}$	Atoms	$d, \text{\AA}$
Ba1 Au2	3.302(2)	Au1 P9	2.34(1)
Ba1 Au4	3.511(2)	Au1 P10	2.36(1)
Ba1 Au6	3.448(1)	Au1 P13	2.46(1)
Ba1 Au8	3.511(2)	Au1 P12	2.47(1)
Ba1 P3	3.36(1)	Au2 P1	2.35(1)
Ba1 P4	3.33(1)	Au2 P7	2.43(1)
Ba1 P5	3.34(1)	Au2 P8	2.45(1)
Ba1 P7	3.42(1)	Au2 P14	2.32(1)
Ba1 P8	3.391(8)	Au3 P8	2.44(1)
Ba1 P14	3.44(1)	Au3 P9	2.48(1)
Ba1 P14	3.44(1)	Au3 P13	2.46(1)
Ba2 Au1	3.325(3)	Au3 P15	2.39(1)
Ba2 Au5	3.444(1)	Au4 P3	2.51(1)
Ba2 Au7	3.427(2)	Au4 P5	2.47(1)
Ba2 P1	3.25(1)	Au4 P7	2.45(1)
Ba2 P2	3.32(1)	Au4 P11	2.50(1)
Ba2 P9	3.393(9)	Au5 P1	2.54(1)
Ba2 P10	3.33(1)	Au5 P2	2.42(1)
Ba2 P13	3.36(1)	Au5 P9	2.45(1)
Ba2 P12	3.46(1)	Au5 P11	2.45(1)
Ba2 P15	3.43(1)	Au6 P3	2.42(1)
Ba3 Au1	3.755(4)	Au6 P4	2.44(1)
Ba3 Au2	3.721(3)	Au6 P8	2.45(1)
Ba3 Au3	3.573(3)	Au6 P11	2.36(1)
Ba3 Au3	3.580(3)	Au7 P2	2.58(1)
Ba3 Au7	3.496(3)	Au7 P6	2.38(1)
Ba3 Au8	3.428(3)	Au7 P10	2.58(1)
Ba3 P13	3.38(1)	Au7 P12	2.55(1)
Ba3 P13	3.44(10)	Au8 P4	2.43(1)
Ba3 P14	3.48(10)	Au8 P5	2.48(1)
Ba4 Au2	3.839(3)	Au8 P6	2.39(1)
Ba4 Au4	3.401(3)	Au8 P14	2.58(1)
Ba4 Au4	3.389(3)	P1 P13	2.30(1)
Ba4 Au7	3.530(3)	P1 P12	2.28(1)
Ba4 Au8	3.374(3)	P2 P5	2.33(1)
Ba4 P7	3.48(1)	P2 P15	2.12(1)
Ba4 P11	3.463(9)	P3 P4	2.31(1)
Ba4 P12	3.40(1)	P3 P14	2.30(1)
Ba5 Au1	3.844(3)	P4 P7	2.42(1)

Ba5	Au2	3.832(4)	P5	P8	2.35(1)
Ba5	Au3	3.760(3)	P6	P12	2.38(1)
Ba5	Au3	3.746(3)	P6	P14	2.27(1)
Ba5	Au5	3.458(4)	P7	P11	2.22(1)
Ba5	Au5	3.482(4)	P9	P10	2.36(1)
Ba5	Au6	3.427(3)	P10	P15	2.18(1)
Ba5	Au6	3.516(3)	P13	P13	2.30(1)
Ba5	P8	3.49(1)	P15	P15	2.72(2)
Ba5	P8	3.47(1)			
Ba5	P9	3.49(1)			

Table SI 3. Au–P bond distance, gold coordination number (CN) and shape of AuP_n fragments in the selected inorganic compounds.

Compound	$d(\text{Au-P}), \text{\AA}$	Au CN	AuP _n fragment	Reference
Au(PF ₃) ₂ ⁺	2.25	2	linear	S1
Au ₂ P ₃	2.33-2.34	2	linear	23
MAu ₂ P ₂ (M = Hg, Tl, Pb)	2.33-2.34	2	linear	S2
K ₅ AuP ₂	2.37	2	linear	S3
K ₂ AuP	2.40	2	linear	S4
BaAu ₂ P ₄	2.36	2	linear	16
Au ₇ P ₁₀ I	2.32-2.34	2, 3	linear, trigonal planar	S5
CaAuP	2.45	3	trigonal planar	S6
EuAuP	2.49	3	trigonal planar	S7
BaAuP	2.54	3	trigonal planar	S8
La ₂ AuP ₂ O	2.42-2.48	3	trigonal planar	31b, S9
Au ₃ SnP ₇	2.41-2.55	3	trigonal pyramid	S10
R ₂ AuP ₃ (R = Ce, Pr, Nd)	2.46-2.53	3	tetrahedral	S11
Au _{1-x} Sn _{1+x} P ₁₄	2.46-2.53	4	tetrahedral	S12
Ba ₈ Au ₁₆ P ₃₀	2.34-2.58	4	tetrahedral	current work

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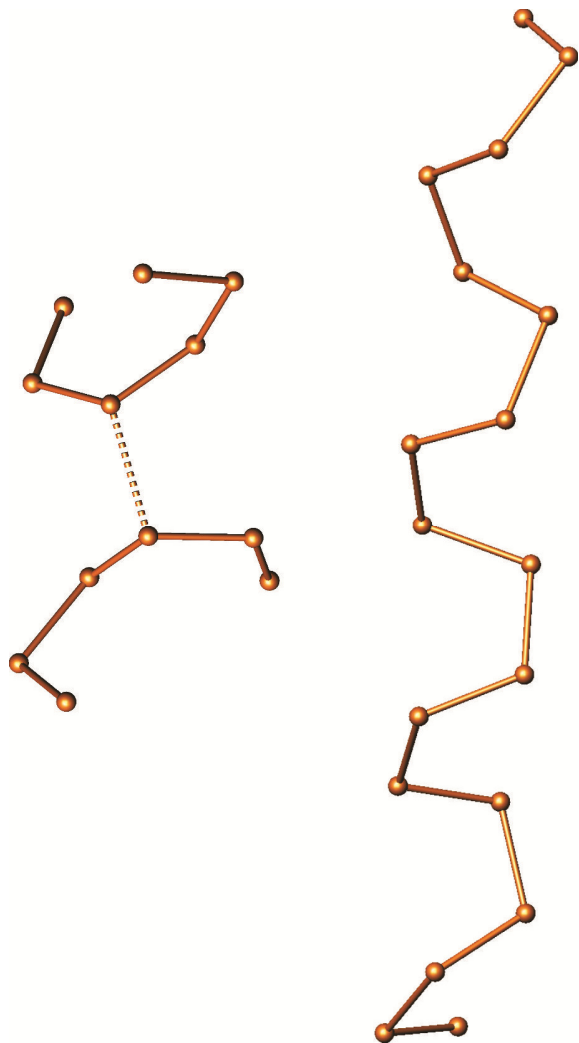


Figure SI 1. P₁₂ (left) and P₁₈ (right) polyphosphide fragments in the crystal structure of Ba₈Au₁₆P₃₀.

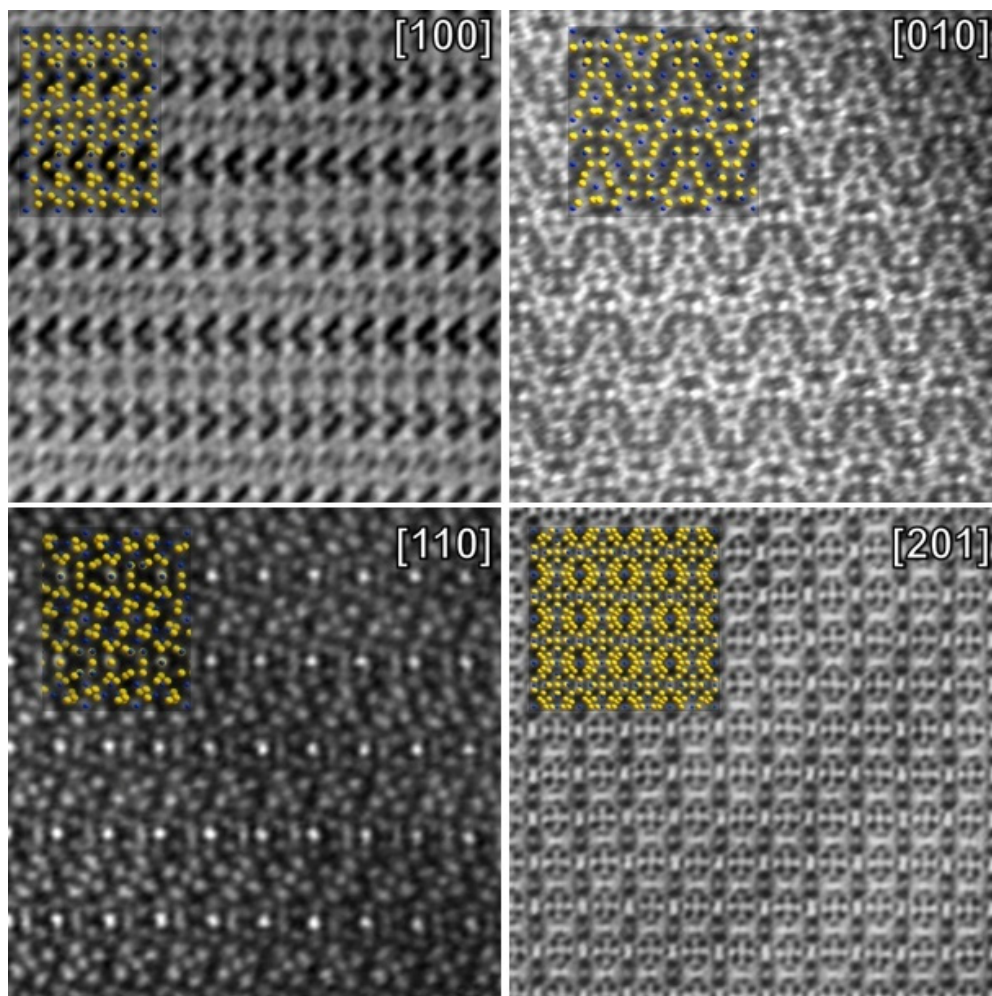


Figure SI 2. High resolution HAADF-STEM images along main zone axis and corresponding overlay structural model of $\text{Ba}_8\text{Au}_{16}\text{P}_{30}$. Only Au (yellow) and Ba (blue) atoms shown for clarity.

Table SI 4. Chemical shifts and relative intensity of the ^{31}P NMR signals for $\text{Ba}_8\text{Cu}_{16}\text{P}_{30}$.

Type of P atoms	Structural ratio	δ , ppm
1 <i>b</i> -P: 3Cu+P	6	132 and 388
2 <i>b</i> -P: 2Cu+2P	22	710 and 364
3 <i>b</i> -P: 1Cu+3P	2	51

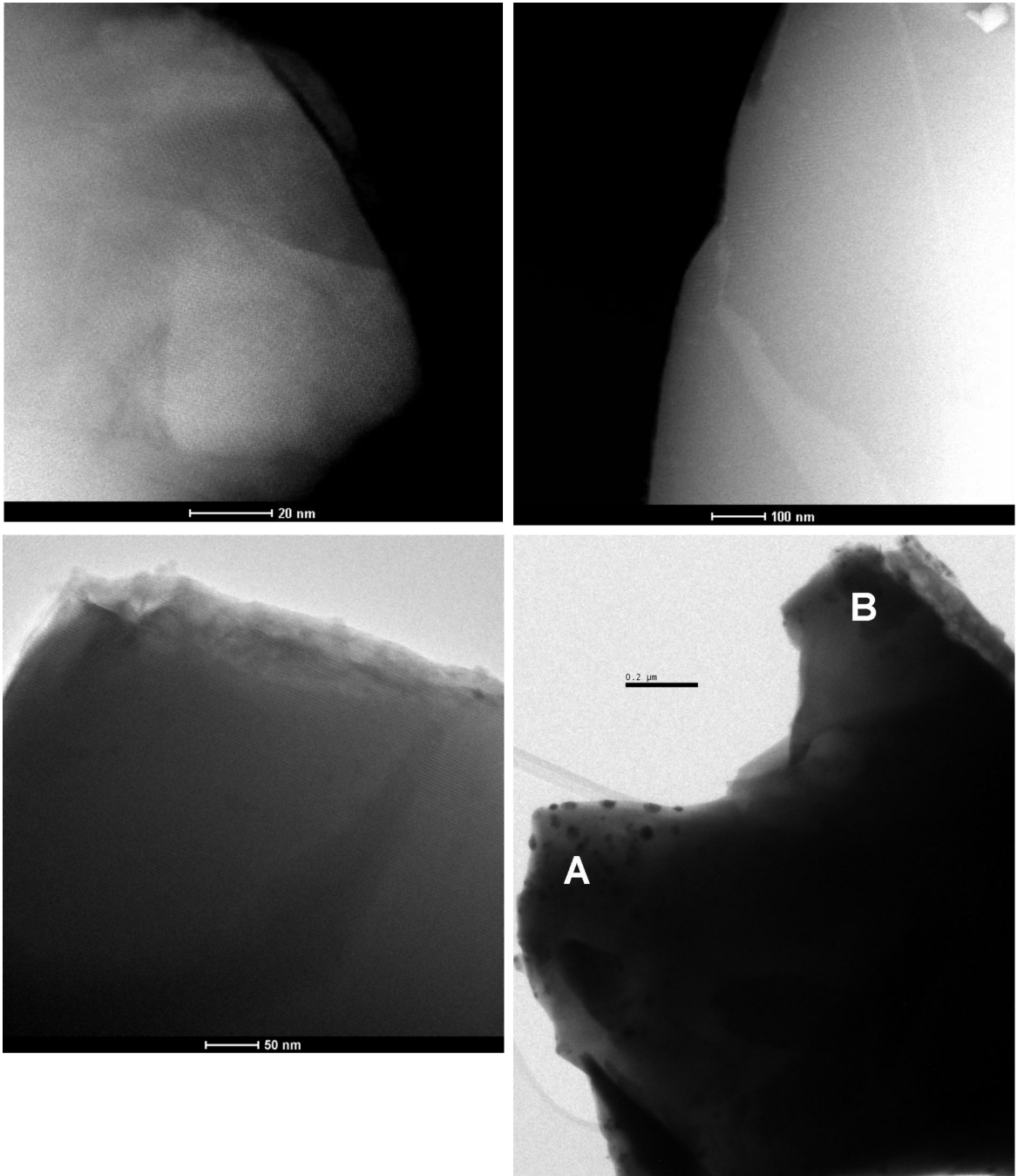


Figure SI 3. Low resolution HAADF-STEM (upper row) and TEM (bottom row) images illustrating average domain size for $\text{Ba}_8\text{Au}_{16}\text{P}_{30}$.